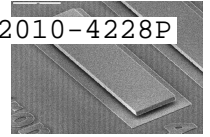




SAND2010-4228P



DSMC Moving-Boundary Algorithms for Simulating MEMS Geometries with Opening and Closing Gaps

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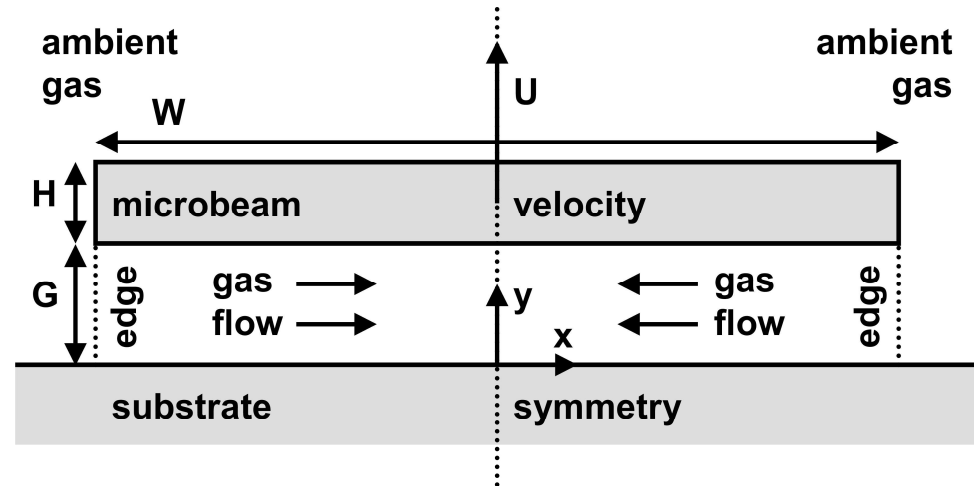
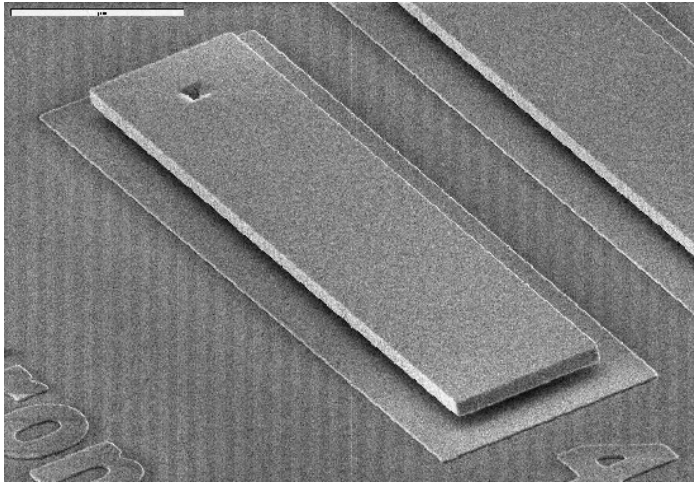
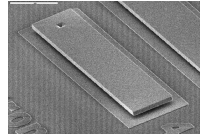
***Rarefied Gas Dynamics: 27th International Symposium
Pacific Grove, California, USA; July 10-15, 2010***



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Motivation



MicroElectroMechanical Systems (MEMS) devices in air

- Beams often oscillate out-of-plane at high frequencies
- Gas in gap between beam and substrate damps motion

Gas motion producing damping force is noncontinuum

- Gaps are small: $\sim 2 \mu\text{m}$ nominal, smaller while closing
- Mean free path: $\sim 0.07 \mu\text{m}$, larger in low-pressure package

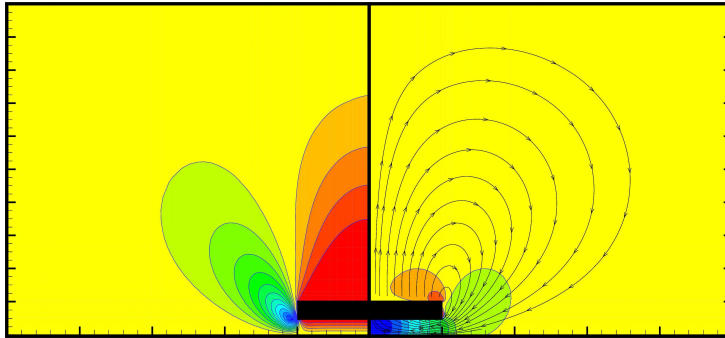
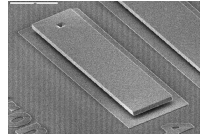
Gas motion is driven by time-varying geometry

- Closing and opening gaps, gas motion to/from ambient

Simulate noncontinuum gas with moving object



Previous Work



Alexeenko et al. (2006): Ellipsoidal Statistical

Several approaches have been used

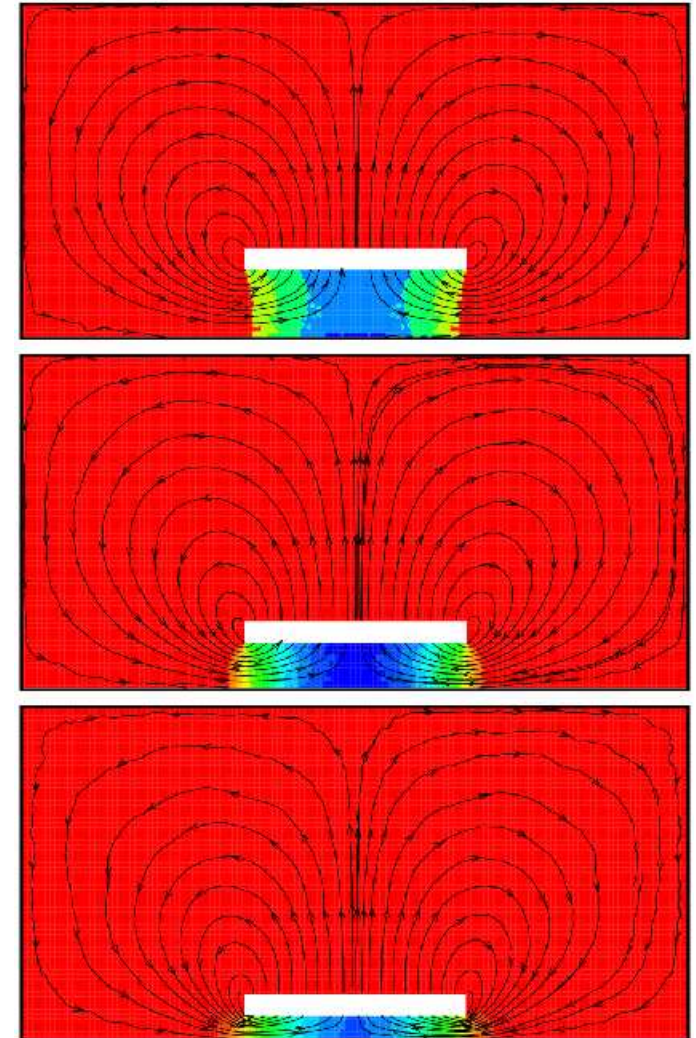
Continuum: Navier-Stokes slip-jump,
Torczynski et al. (2002)

Quasi-static DSMC: fixed geometries,
Gallis et al. (2003)

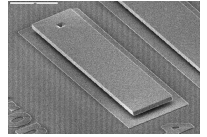
Simpler physics: Reynolds equation,
Gallis and Torczynski (2004)

Kinetic model: Ellipsoidal statistical,
Alexeenko et al. (2006)

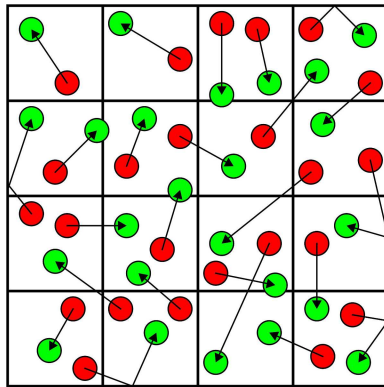
DSMC for noncontinuum gas with moving object



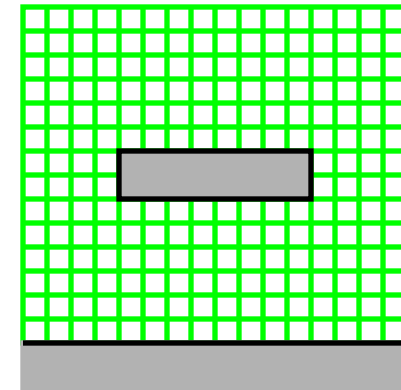
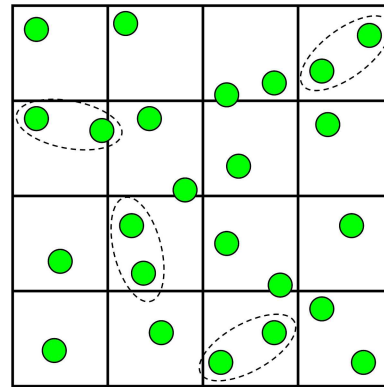
Gallis et al. (2003): Quasi-Static DSMC



Current Approach



DSMC molecules move over mesh, collide within cells



Object moves over mesh

Direct Simulation Monte Carlo (DSMC) with moving object

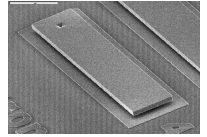
- Object moves over mesh just like molecules do
- Molecules reflect from moving object (diffuse, specular)

Advantages and disadvantages

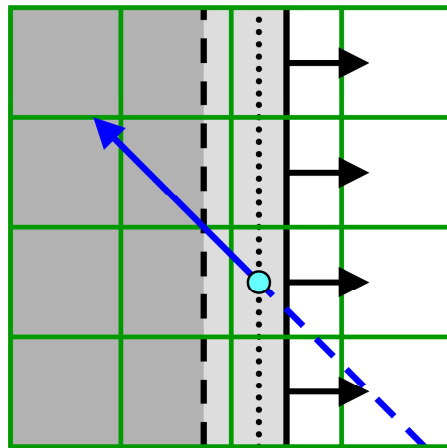
- Gas motion is inherently transient and noncontinuum
- Requires significant computational resources

Surface fluxes to object are most important quantities

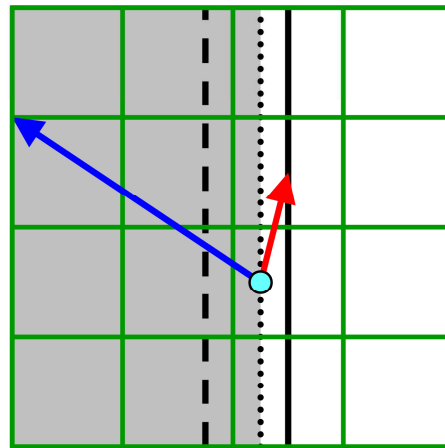
- Gas force on object, gas energy transfer to object
- Can tolerate stochastic noise out in bulk gas



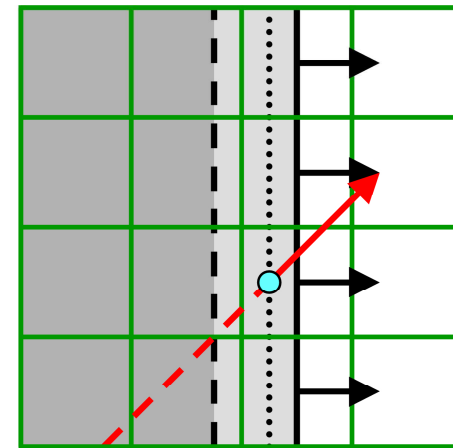
Rigorous Algorithm



mesh frame
incident molecule



object frame
ordinary reflection



mesh frame
reflected molecule

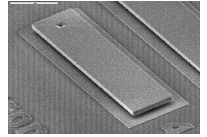
Reflection occurs where molecule and object paths intersect

- Molecule & object move with fixed velocities over time step
- Intersection point and remaining time are determined

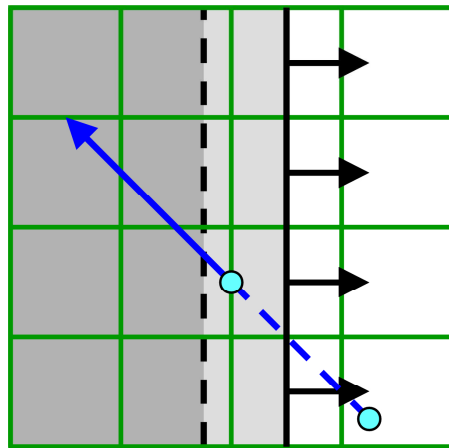
Molecule reflection is performed in object reference frame

- Object velocity is subtracted from molecule velocity
- Reflected molecule velocity is selected (diffuse, specular)
- Object velocity is added to reflected molecule velocity

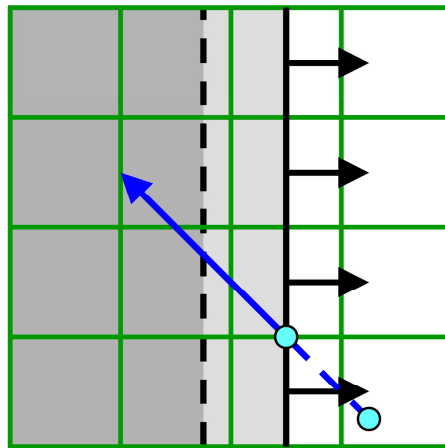
Molecule travels for remainder of time step



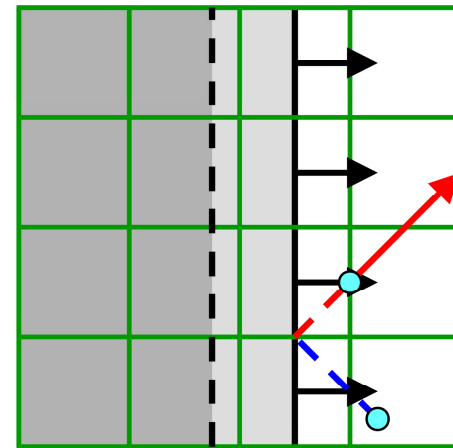
Approximate Algorithm



object & molecule
move, overlap



intersect path with
new object position



molecule reflects
from new object

Reflection occurs from envelope of object over time step

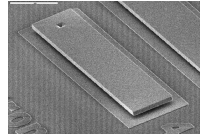
- Advancing/receding: reflect from new/old object position
- Ensures that reflected molecule remains outside object

If object is slow w.r.t. molecules, approximate → rigorous

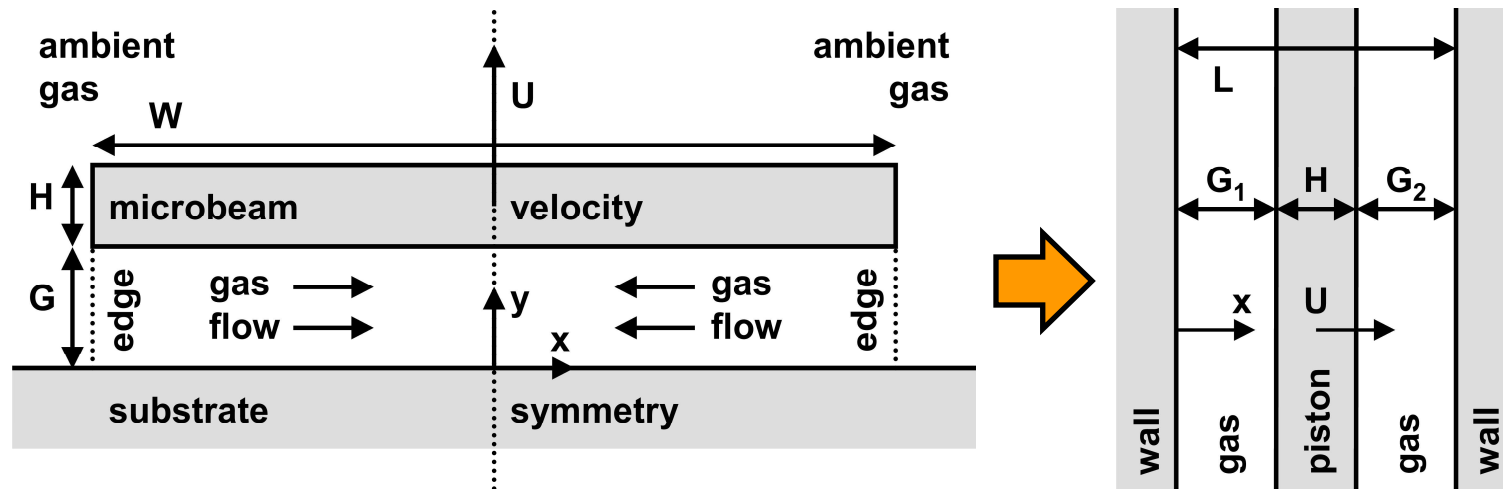
- Molecule's reflected velocity is same as rigorous algorithm

Advantages and disadvantages

- Faster than rigorous since reflection point is not calculated
- Positions of molecules near surface approximated



Simplified Geometry



Initially, implement/assess algorithms for simple geometry

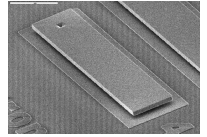
- Need less computational resources, verification is easier

Replace 2D microbeam cross section with 1D piston

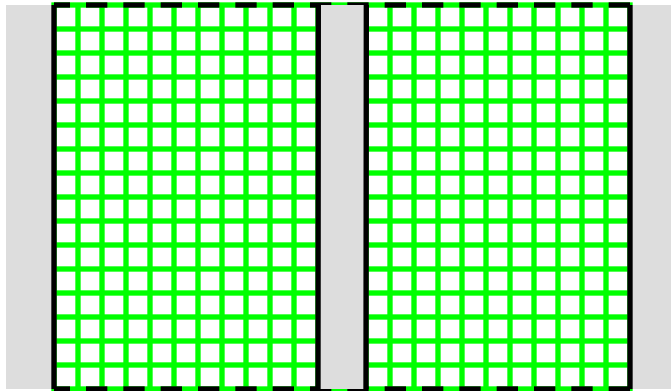
- Piston has opening and closing gaps like microbeam
- Molecule-surface intersection point is easy to determine
- Know early (wave) & long-time behavior for some cases

Prescribe piston position analytically for all time

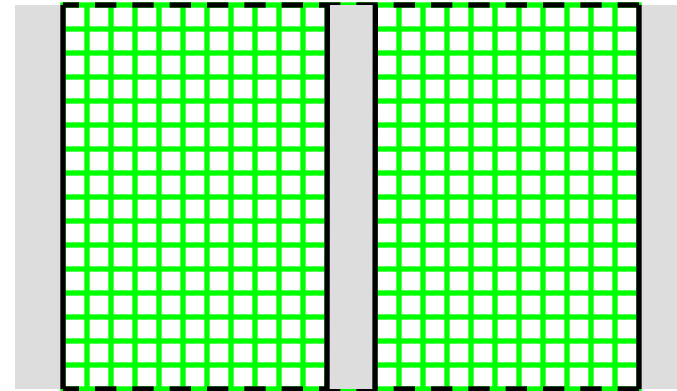
- Piston velocity is constant over each time step: average
- Could couple piston motion to gas force



Piston Motion



Impulsively starts and stops



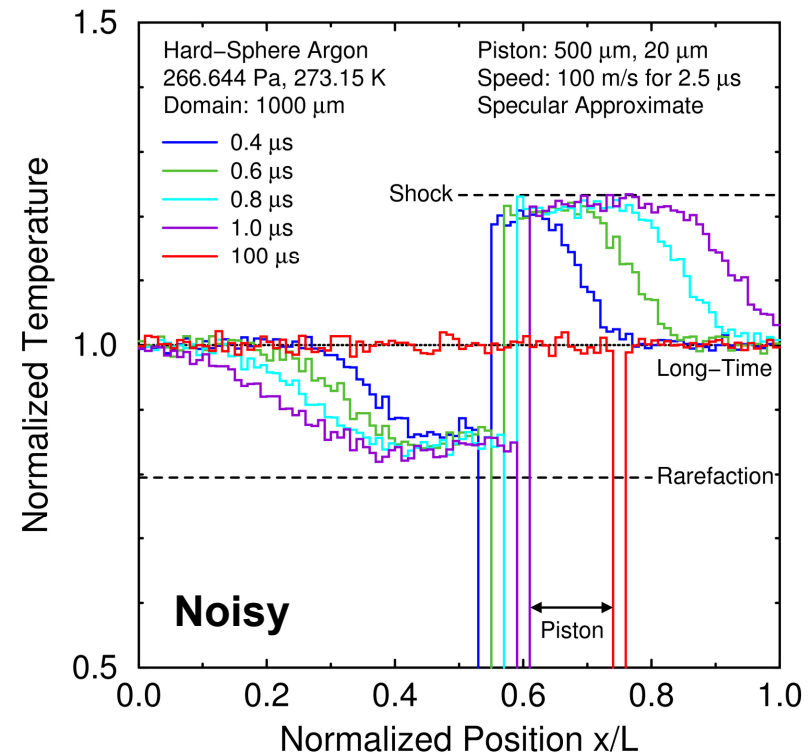
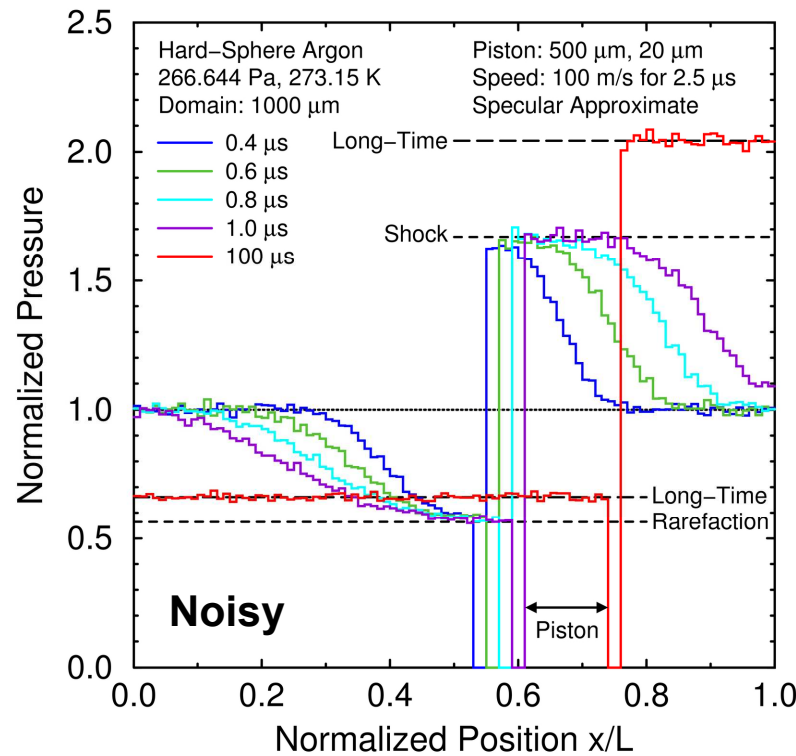
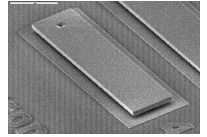
Sinusoidally oscillates continually

Simulation conditions

- Gas is hard-sphere argon at 266.644 Pa and 273.15 K
- Domain is 1000 μm , cells are 10 μm (100 cells)
- End walls are motionless, diffuse at 273.15 K
- Piston is 20 μm (2 cells), specular or diffuse at 273.15 K
- Time step is 1 ns, no averaging over multiple time steps
- Molecules per cell: 10^4 (“noisy”) or 10^5 (“smooth”, most)
- **Case 1:** starts at 50%, moves at 100 or 10 m/s, stops at 75%
- **Case 2:** centered, 1 MHz, velocity amplitude 100 or 10 m/s
- Denote 100 m/s as “fast” and 10 m/s as “slow”



Impulsive Fast Specular Approximate



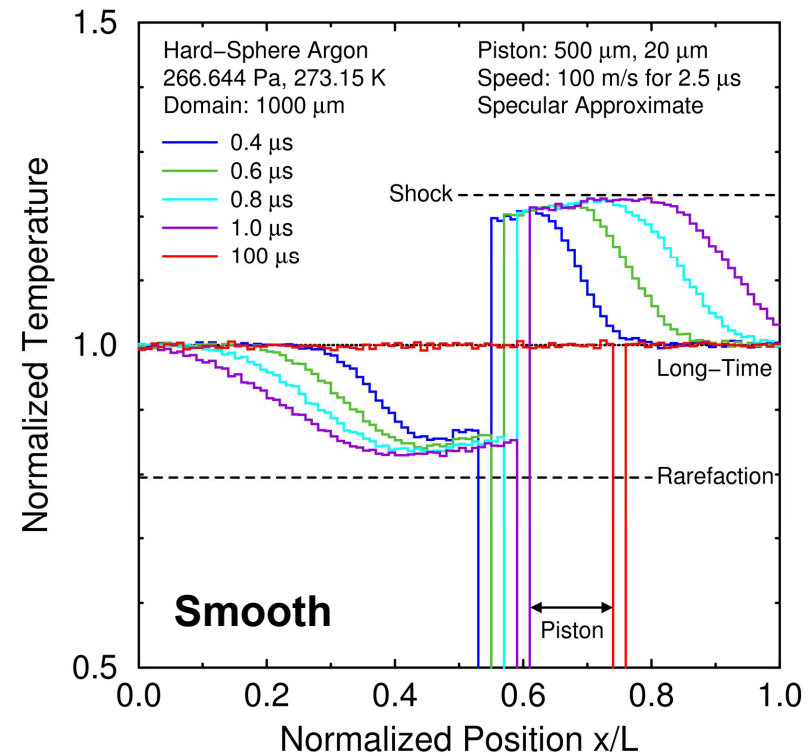
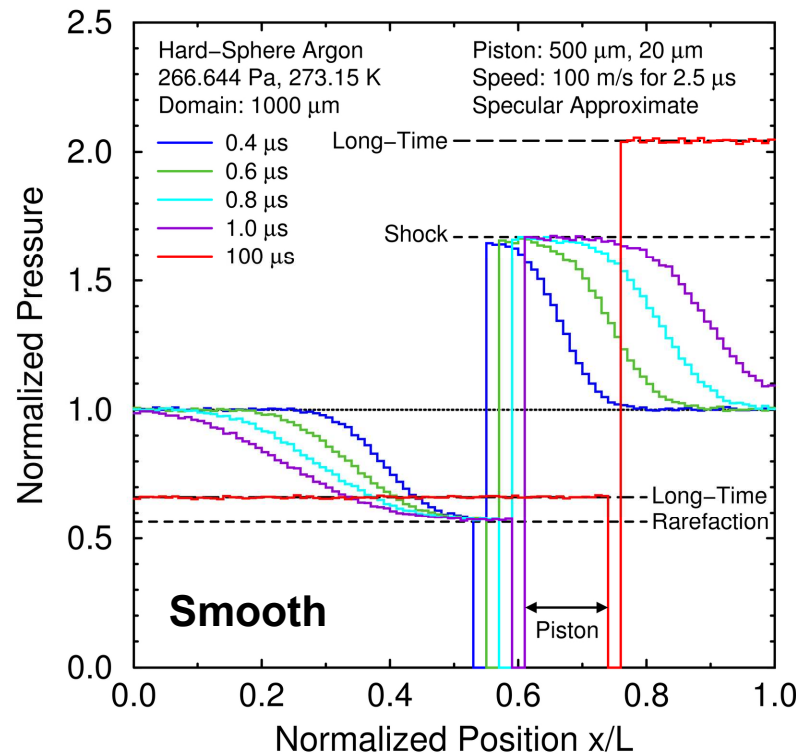
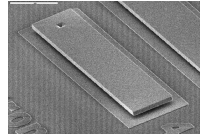
Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Noise is significant with 10^4 molecules per cell



Impulsive Fast Specular Approximate



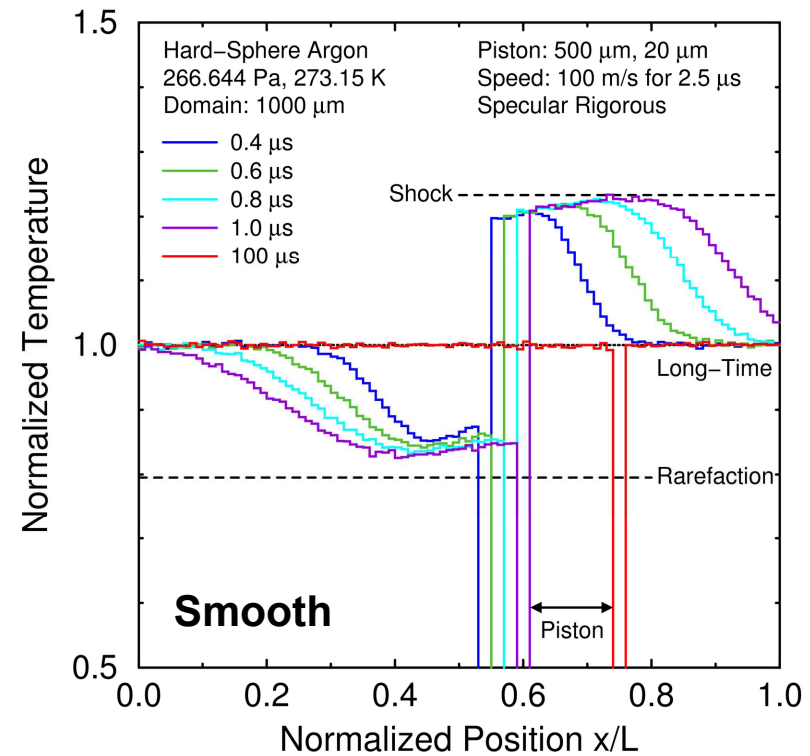
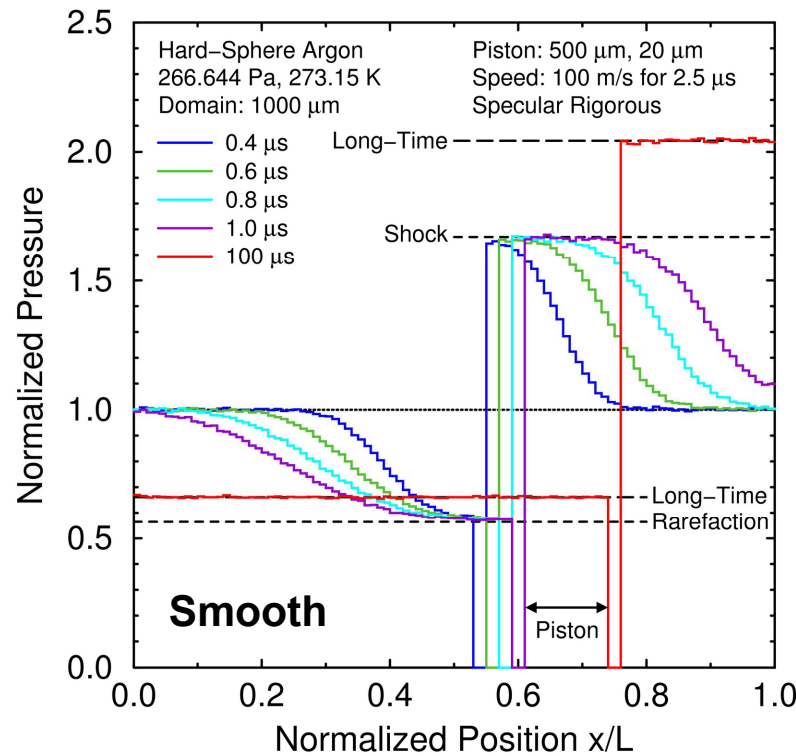
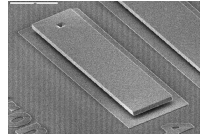
Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Noise is insignificant with 10^5 molecules per cell



Impulsive Fast Specular Rigorous



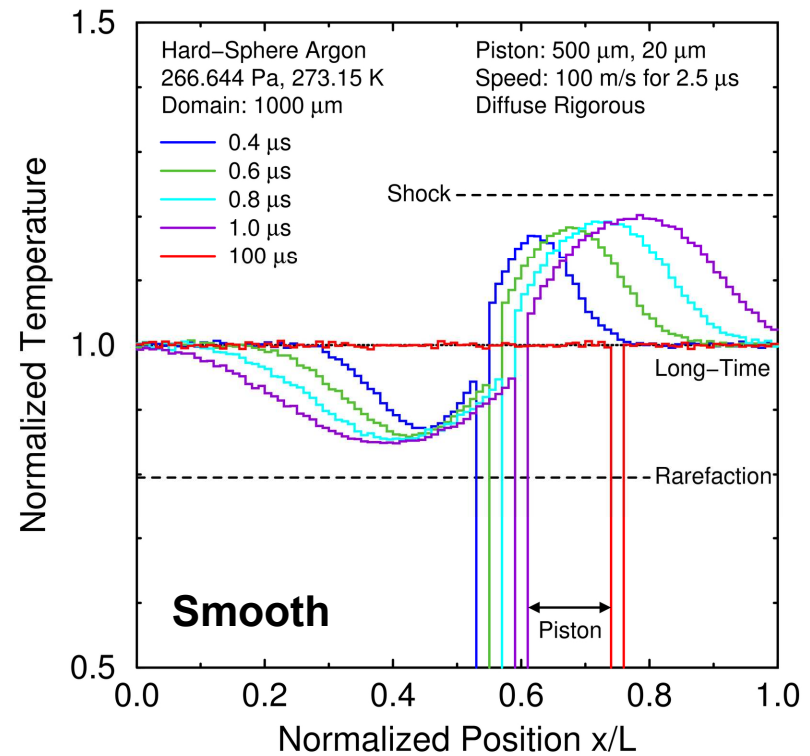
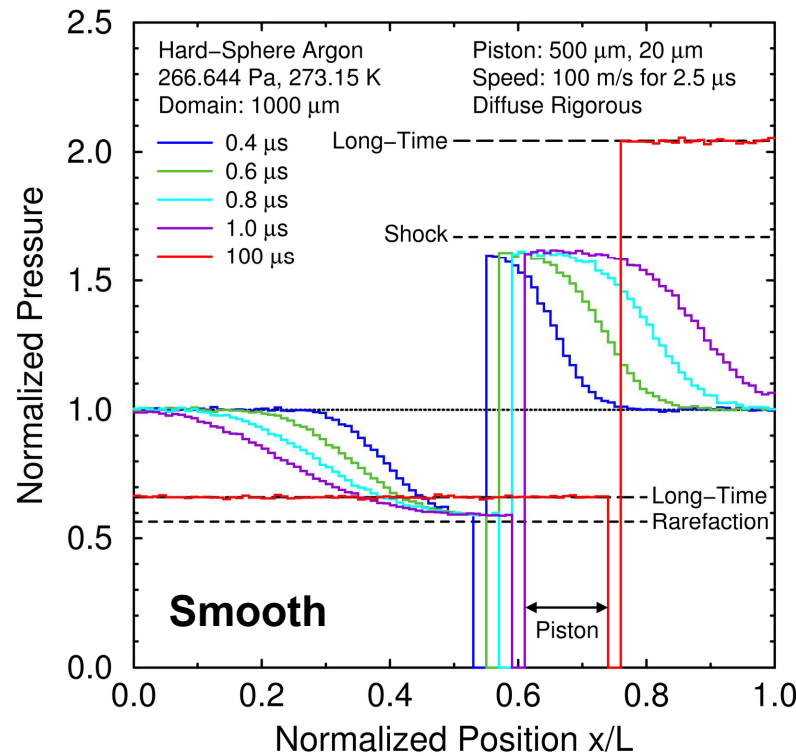
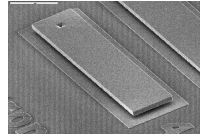
Excellent agreement with theoretical predictions

- Shock pressure and temperature on compression side
- Rarefaction pressure and temperature on expansion side
- Long-time temperatures are the wall temperature
- Long-time pressures correspond to new volumes

Virtually same as approximate except for noise



Impulsive Fast Diffuse Rigorous



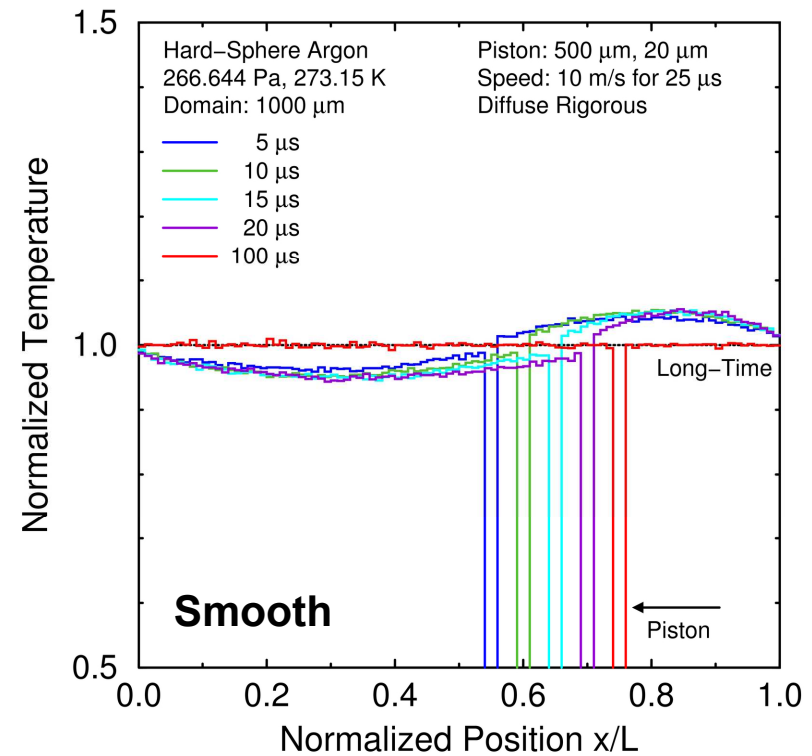
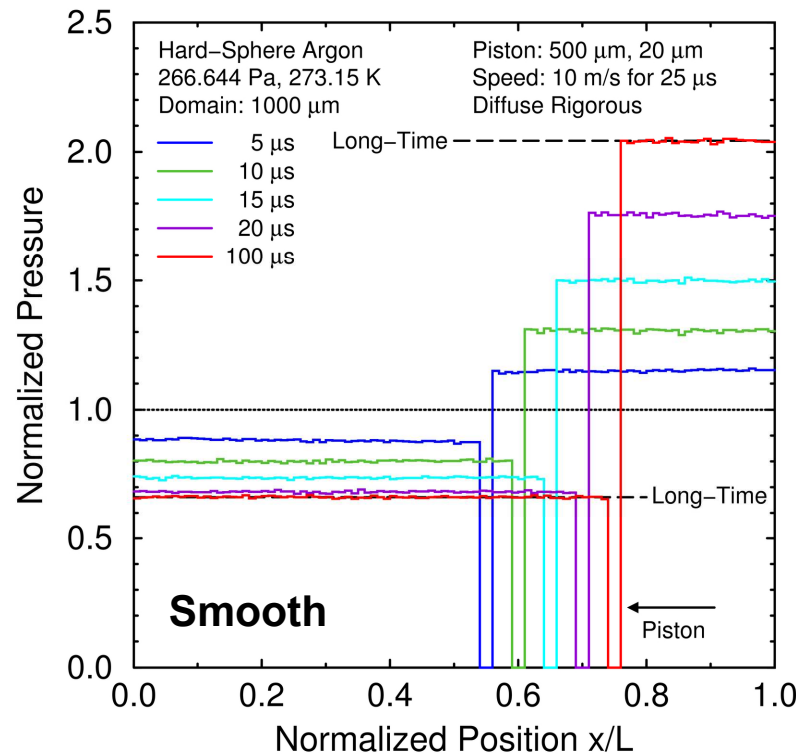
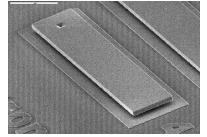
Good agreement with theoretical expectations

- Gas temperatures near piston are close to piston value
- Temperature jumps at piston and walls are evident
- Thermal boundary layers from piston weaken both waves
- Long-time pressures and temperatures are as expected

Qualitatively similar to specular piston



Impulsive Slow Diffuse Rigorous



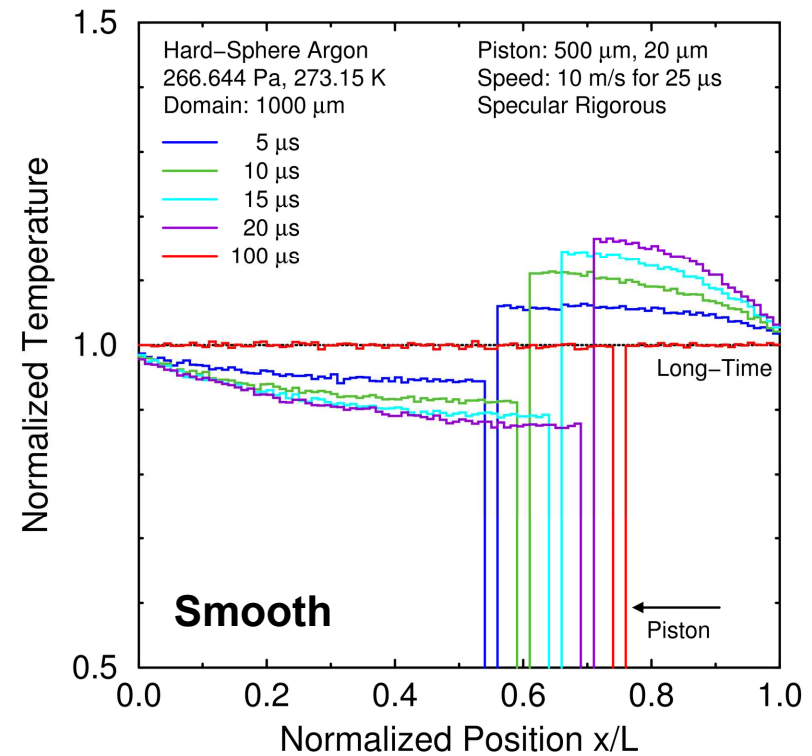
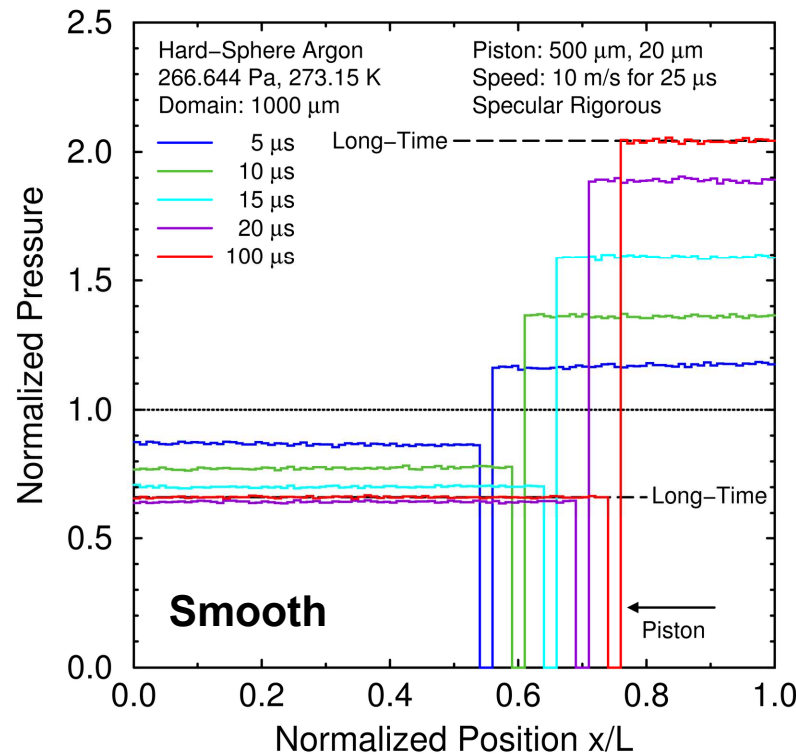
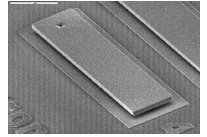
Good agreement with theoretical expectations

- Gas temperatures near piston are close to piston value
- Temperature jumps at piston and walls are small
- Pressures are spatially uniform, waves are negligible
- Long-time pressures and temperatures are as expected

Qualitatively different from fast piston



Impulsive Slow Specular Rigorous



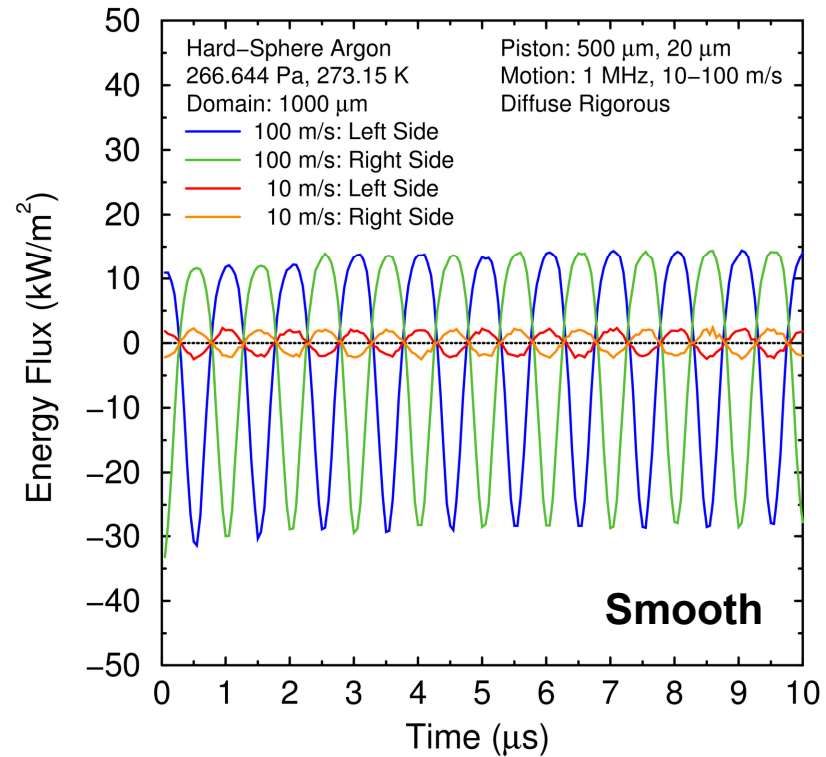
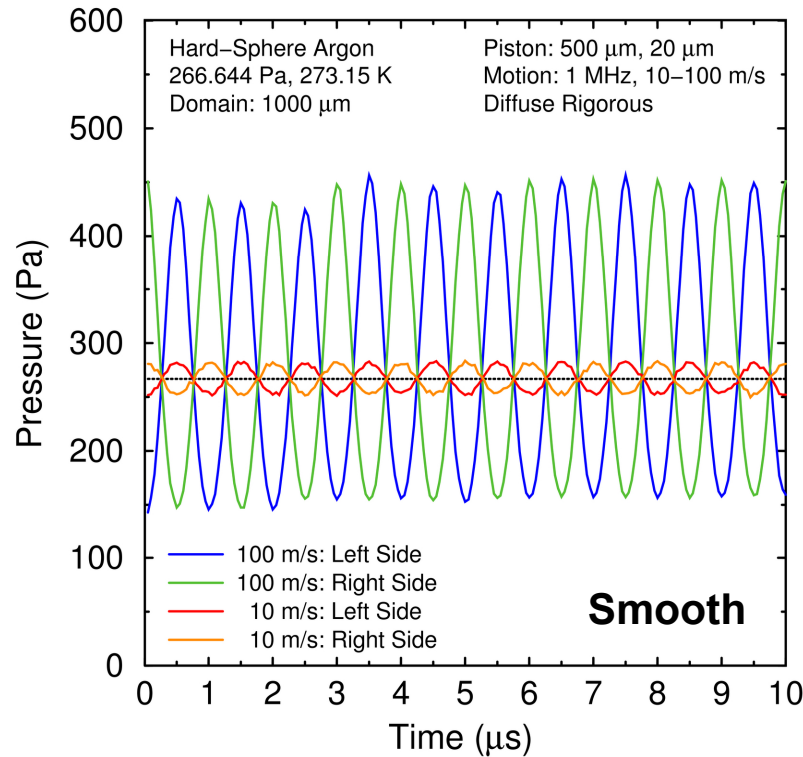
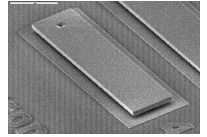
Good agreement with theoretical expectations

- Temperature gradients (heat fluxes) near piston are small
- Temperature jumps at walls are small
- Pressures are spatially uniform, waves are negligible
- Long-time pressures and temperatures are as expected

Qualitatively similar to diffuse piston



Oscillating Fast and Slow Diffuse



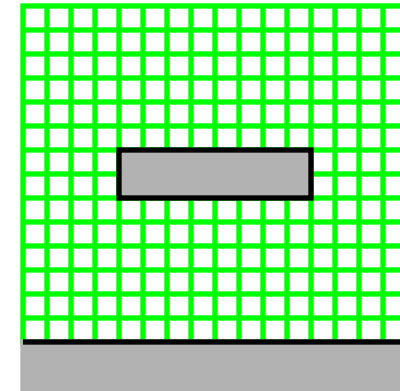
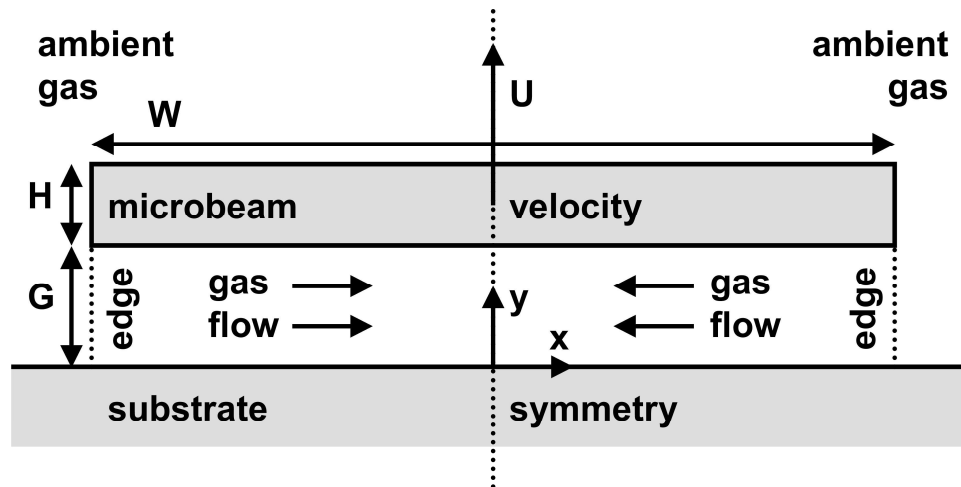
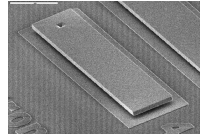
Force on and energy transfer to object are most important

- Slow has linear waves, advancing and receding are same
- Fast has nonlinear waves, advancing and receding differ
- Increasing velocity 10x does not increase response 10x
- Slight rise over time is attributed to net heating of gas

Energy flux is mainly work, not heat transfer



Conclusions and Future Efforts



Moving-boundary DSMC algorithms have been developed

- Implemented for 1D piston moving between end walls
- Rigorous and approximate results agree with theory

Future work will focus on 2D implementation

- Harder to find where object and molecule paths intersect
- Must maintain good scaling for massively parallel cases