

**Eldorado High School
Presentation – May 7 and 8, 2008
prepared by**

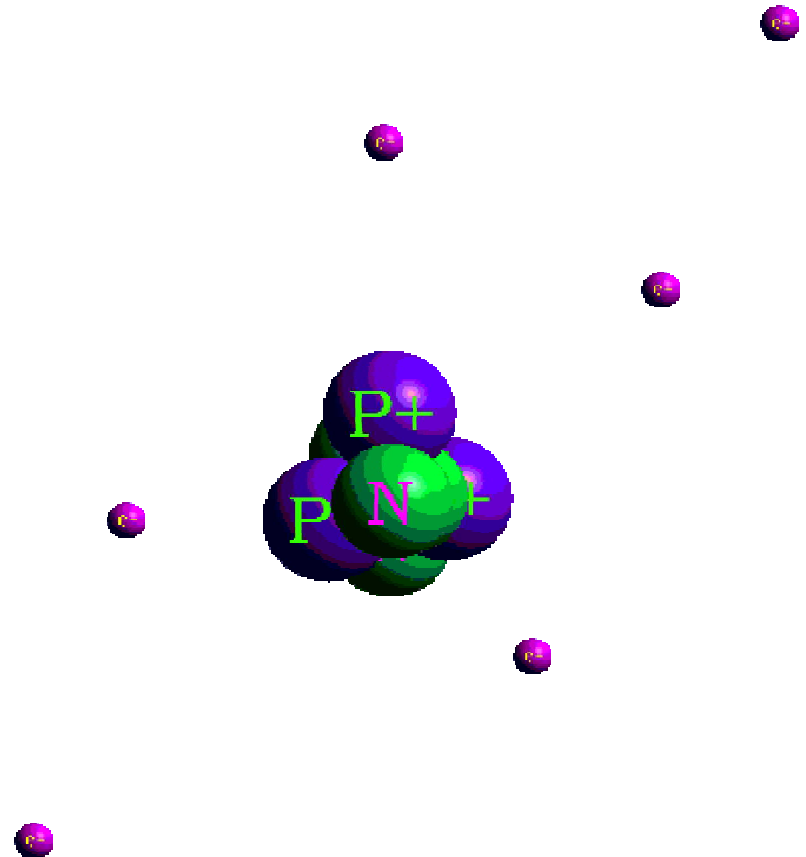
**Bob Miltenberger, CHP
Cynthia Kajder**

Atomic Structure

Protons (positive) 

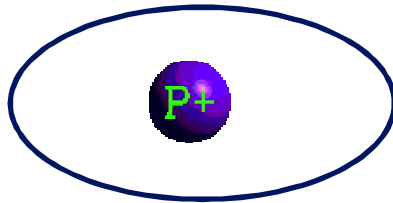
Neutrons (neutral) 

Electrons (negative) 



Stable vs. Unstable Atoms

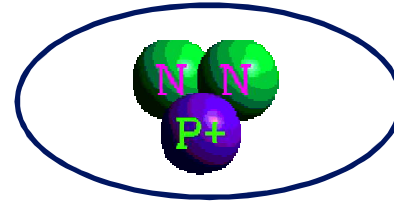
If there are too many or too few neutrons for a given number of protons, the nucleus will not be stable



hydrogen
(protium)

STABLE

“Non-Radioactive”



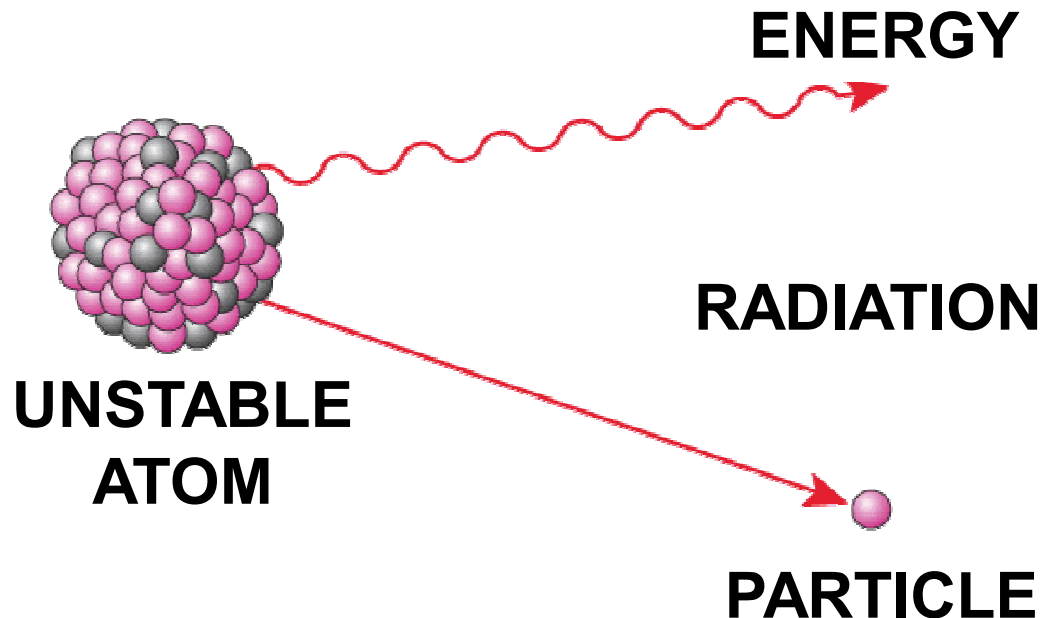
hydrogen
(tritium)

UNSTABLE

“Radioactive”

Radiation

- **Energy** released from unstable atoms and some devices in the form of rays or particles
- Can be either ionizing or non-ionizing



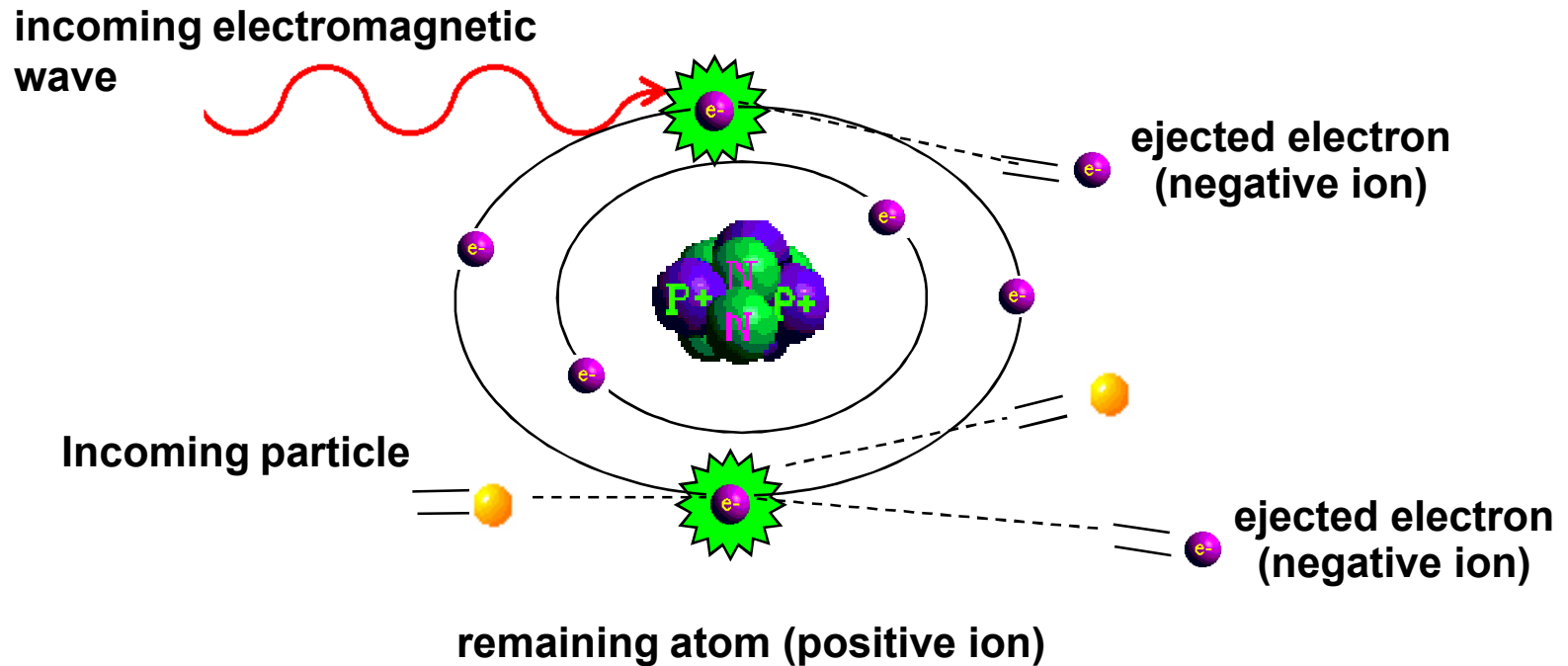
Non-ionizing Radiation

- Radiation that does not have enough energy to ionize atoms with which it interacts
- Examples:
 - radio waves
 - infrared radiation
 - visible light
 - radar waves
 - microwaves



Ionization

The process of removing electrons from neutral atoms

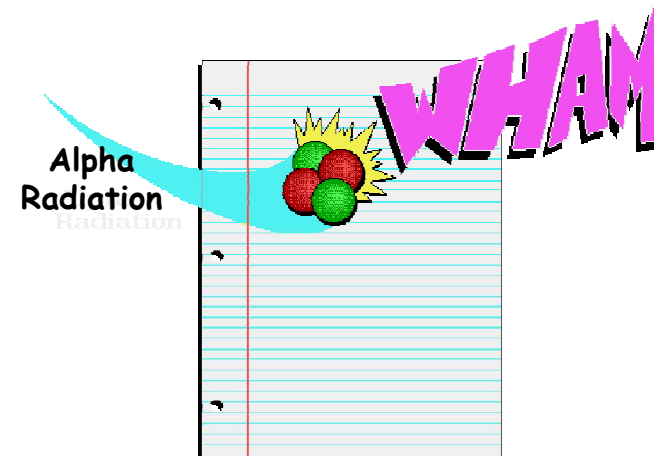
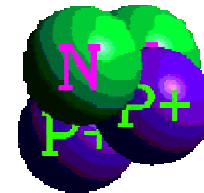


Ionizing Radiation

- Radiation that possesses enough energy to cause ionization in the atoms with which it interacts
- Released from unstable atoms and some devices in the form of rays or particles
 - alpha α
 - beta β
 - gamma/x-ray γ
 - neutron n

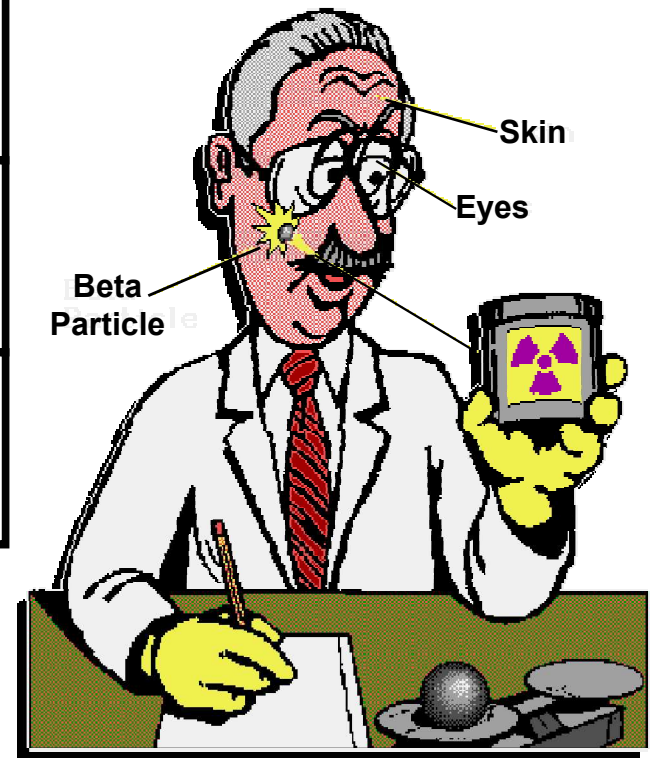
Alpha Radiation (α)

Characteristics	Particle, Large Mass, +2 Charge
Range	Very Short 1 - 2" in air
Shielding	Paper Outer layer of skin
Hazards	Internal



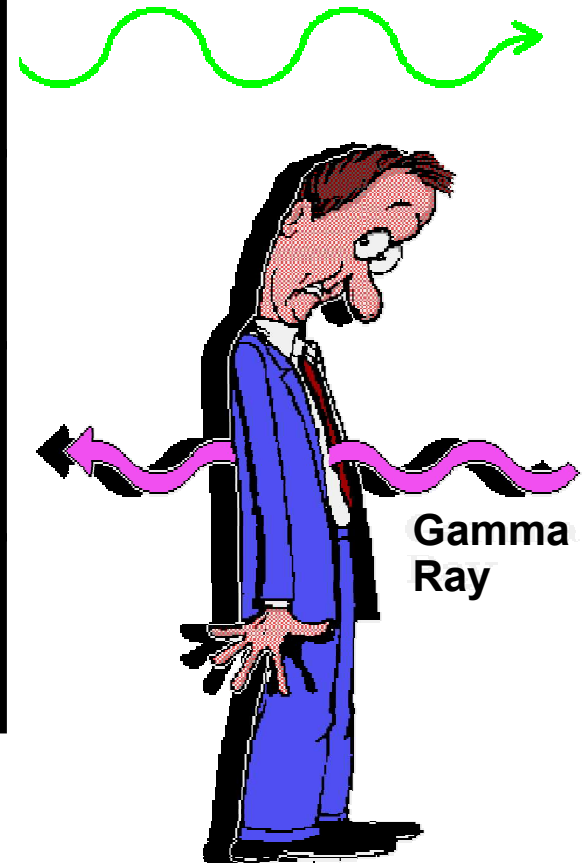
Beta Radiation (β)

Characteristics	Particle, Small Mass, -1 Charge
Range	12ft / MeV in air
Shielding	Plastic, glass, aluminum, wood
Hazards	Internal and the skin and eyes



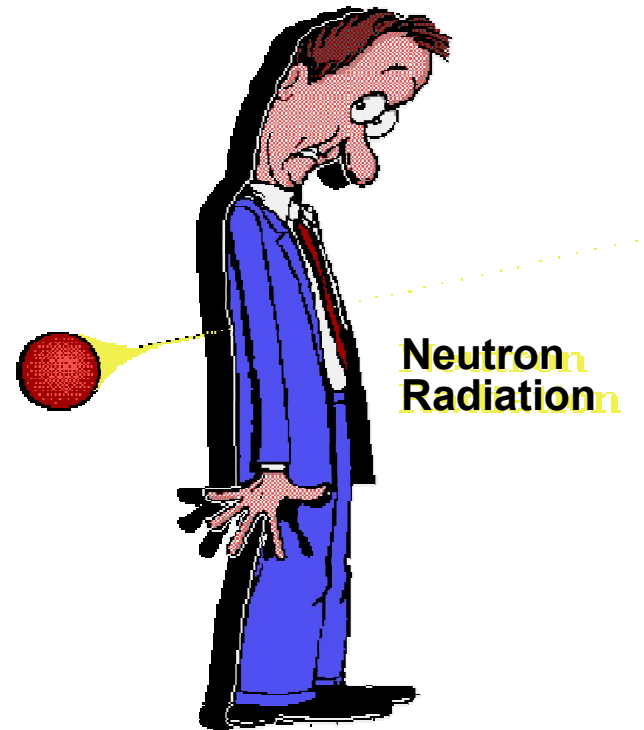
Gamma Rays (γ) and X-Rays

Characteristics	No mass, no charge electromagnetic
Range	Hundreds of feet in air
Shielding	Lead, Steel Concrete
Hazards	External Source Whole Body Penetrating



Neutron Radiation (η)

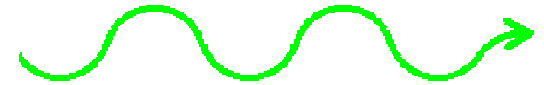
Characteristics	Particle with no charge
Range	Hundreds of feet in air
Shielding	Hydrogenous material - water, polyethylene
Hazards	External Source Whole Body Penetrating



Units of Measure

- **Radiation \longrightarrow Energy**

Roentgen, RAD, REM

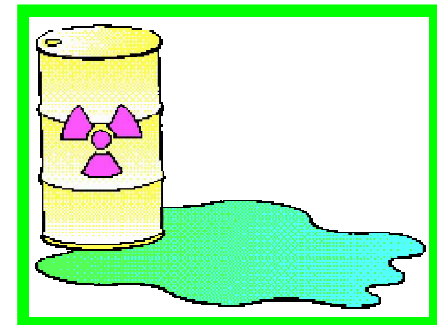


- **Radioactivity \longrightarrow Rate**

dpm, Curie

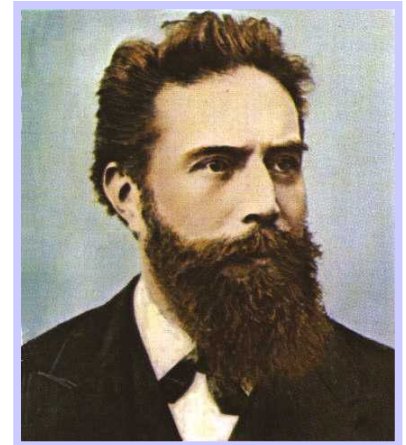


- **Contamination → Spread**

$$\frac{\text{Radioactivity}}{\text{Area or volume}}$$


Roentgen (R)

- Unit for measuring exposure
- Defined only for ionization in air
- Applies only to gamma and x-rays
- Not related to biological effects



Wilhelm Roentgen
1845 -1923
Discovered X-rays



RAD (Radiation Absorbed Dose)

- Unit for measuring absorbed dose in any material
- Applies to all types of radiation
- Does not take into account the potential effect that different types of radiation have on the body

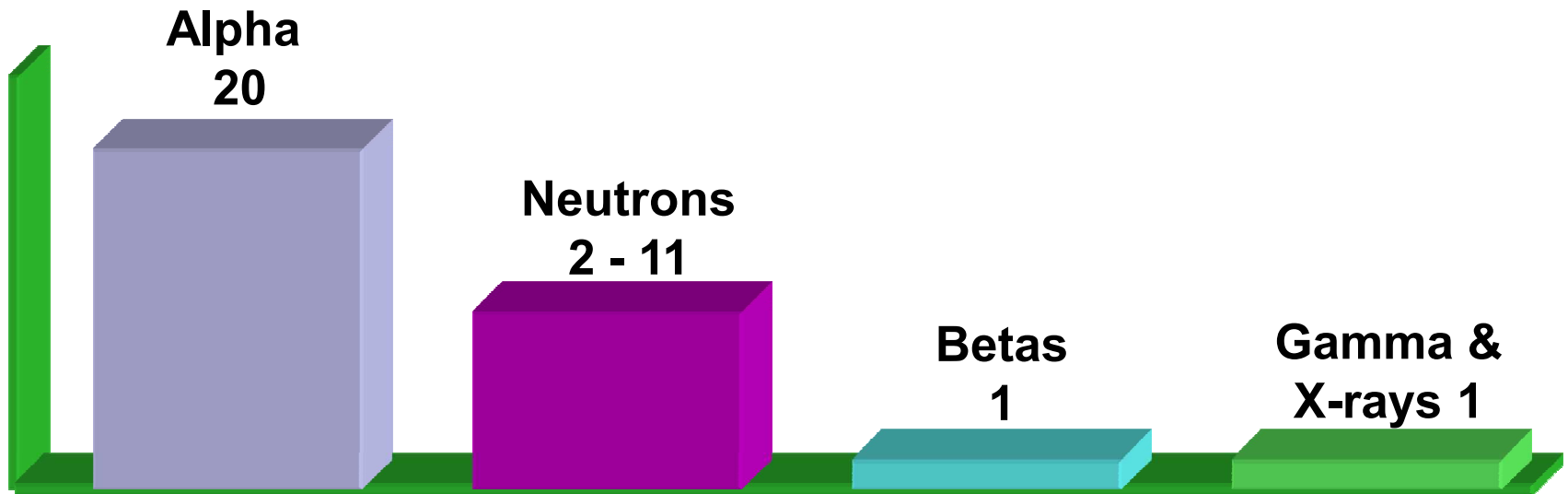
REM (Roentgen Equivalent Man)

Takes into account the energy absorbed (dose) and the biological effect on the body due to the different types of radiation

Quality Factor (QF)

The QF is used as a multiplier to reflect the relative amount of biological damage caused by the same amount of energy deposited in cells by the different types of ionizing radiation.

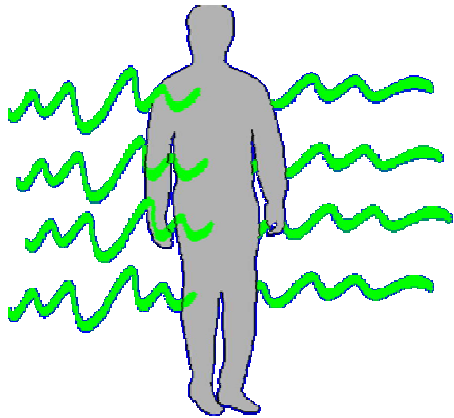
$$\text{rad} \times \text{QF} = \text{rem}$$



Dose vs. Dose Rate

- Dose rate is the *rate* at which you receive the dose.
- Dose rate = dose divided by time (rem/hr, mrem/hr).
- Dose is the *amount* of radiation you receive.

$$\text{Dose} = \text{Dose Rate} \times \text{Time}$$



200 mrem/hr

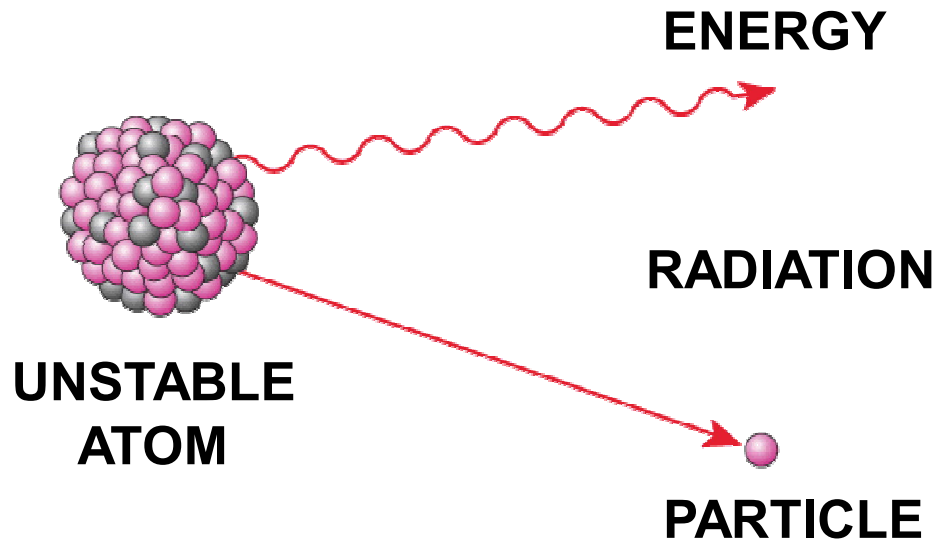
Question: How much dose would this individual receive in 2 hours?

Answer: _____

Measuring Radioactivity

A measure of the number of disintegrations radioactive material undergoes in a certain period of time.

We measure the rate of decay which will lead us to the quantity of radioactive material present.



Radioactivity Units

Basic unit

- disintegration per minute (dpm)
- derived from the number of counts measured by instrument and the instrument efficiency

Larger unit

- Curie (Ci)
- $1 \text{ Ci} = 2.22 \times 10^{12} \text{ dpm}$
- $1 \text{ Ci} = 3.70 \times 10^{10} \text{ dps or Bq}$

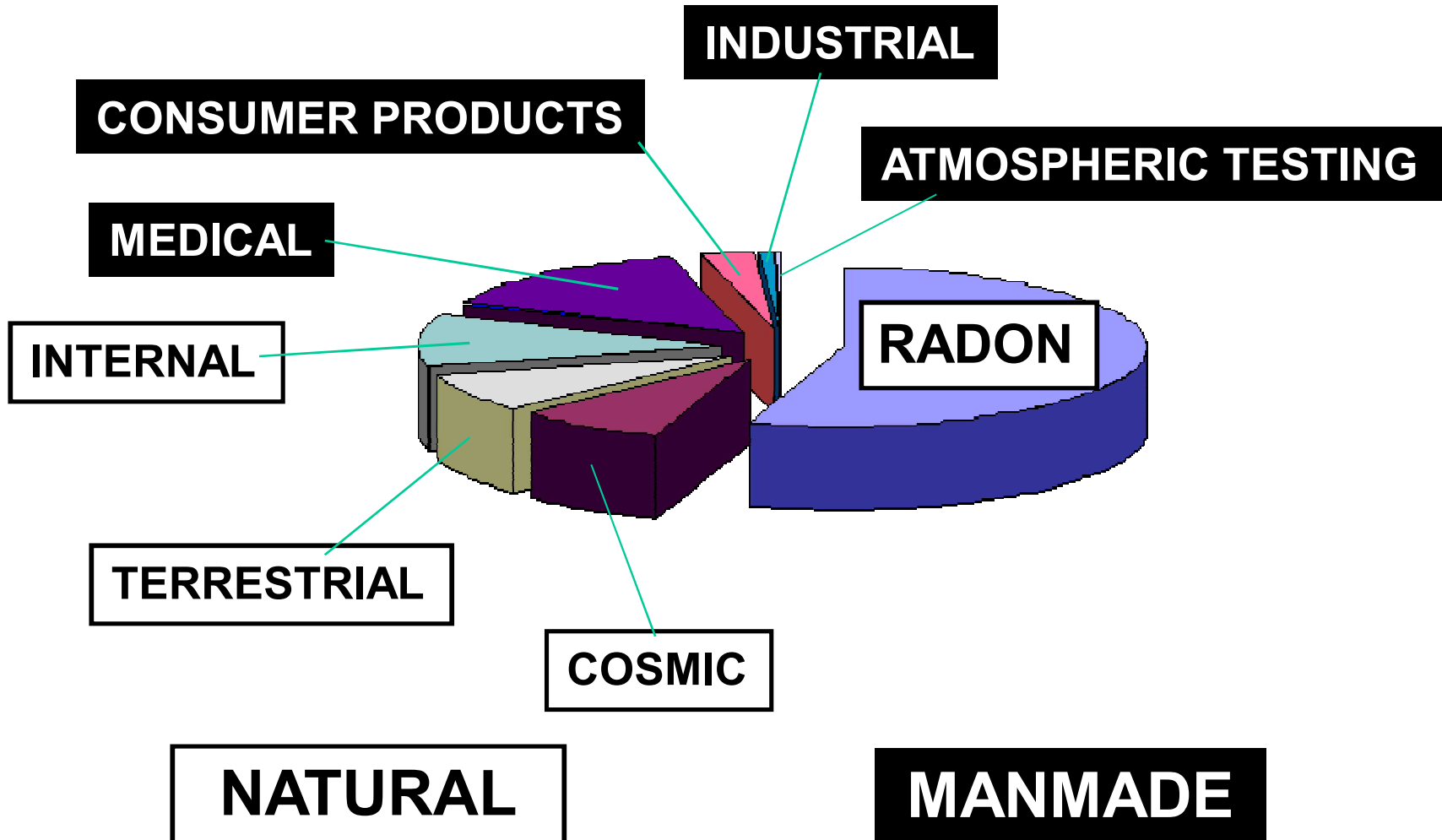


Marie Curie
1867 - 1934
Discovered
radium & polonium

Background Sources

- **Natural**
- **Manmade**
- **U.S. Average**

Background Sources of Ionizing Radiation

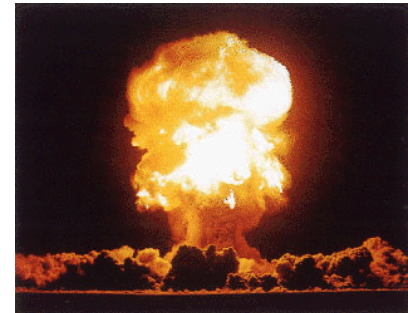


Natural Background Sources

SOURCE	AVG DOSE
COSMIC - sun & outer space	28 mrem/yr
TERRESTRIAL - Earth's crust	28 mrem/yr
INTERNAL - our own bodies	40 mrem/yr
RADON - Uranium in the Earth	200 mrem/yr

Man-Made Radiation Sources

SOURCE	AVG DOSE
Medical	54 mrem/yr
Consumer Products	10 mrem/yr
Industrial Uses	< 3 mrem/yr
Atmospheric Testing	< 1 mrem/yr





Radiation Doses

- Medical Procedures -



Radiation Therapy	600,000 mrem - tumor
CAT Scan	5,800 mrem - head 1,500 mrem - lower spine
Fluoroscope	5,000 mrem/min. - skin
Mammogram	400 mrem - breast 0.2 mrem (low-dose screen)
Dental X-Ray	55 - 65 mrem/shot - mouth
Chest X-Ray	10 mrem/shot - chest

Radiation Doses

- Consumer Products -

PRODUCT	AVG DOSE
Cigarettes (1.5 packs/day)	8,000 mrem/yr - lungs
Dental Porcelain	60 rem/yr - gums
Tinted Glasses	4 rem/yr - eyes
Building Materials	7 mrem/yr - whole-body
Radium Dial Watch	6 mrem/yr - whole-body
Smoke Detector	1 mrem/yr - whole-body

U.S. Average

The **U.S. average** for exposure to natural background and man-made sources is approximately **360 mrem per year**.

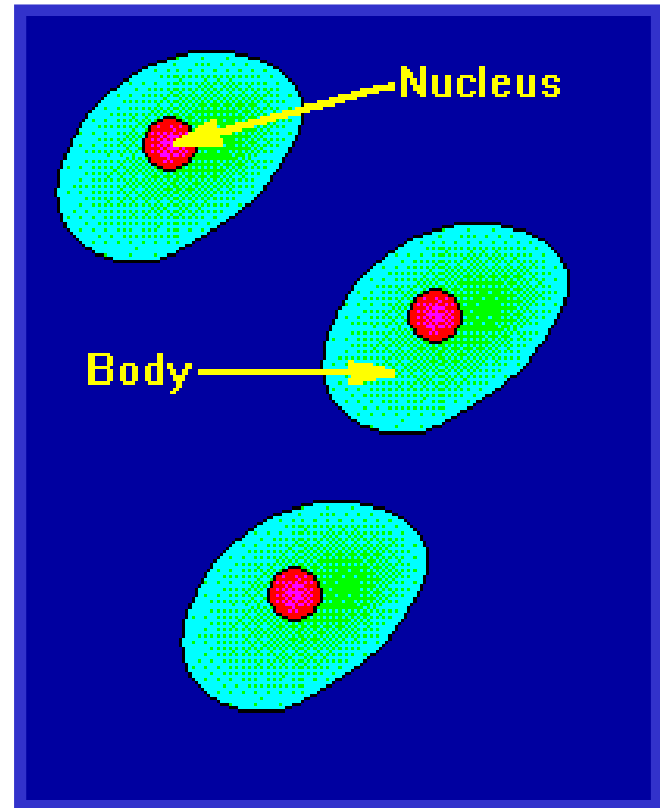


Radiation Effects

- **Cell Damage**
- **Cell Sensitivity**
- **Possible Effects on Cells**
- **Radiation Damage Factors**
- **Acute vs. Chronic**
- **Somatic vs. Heritable**

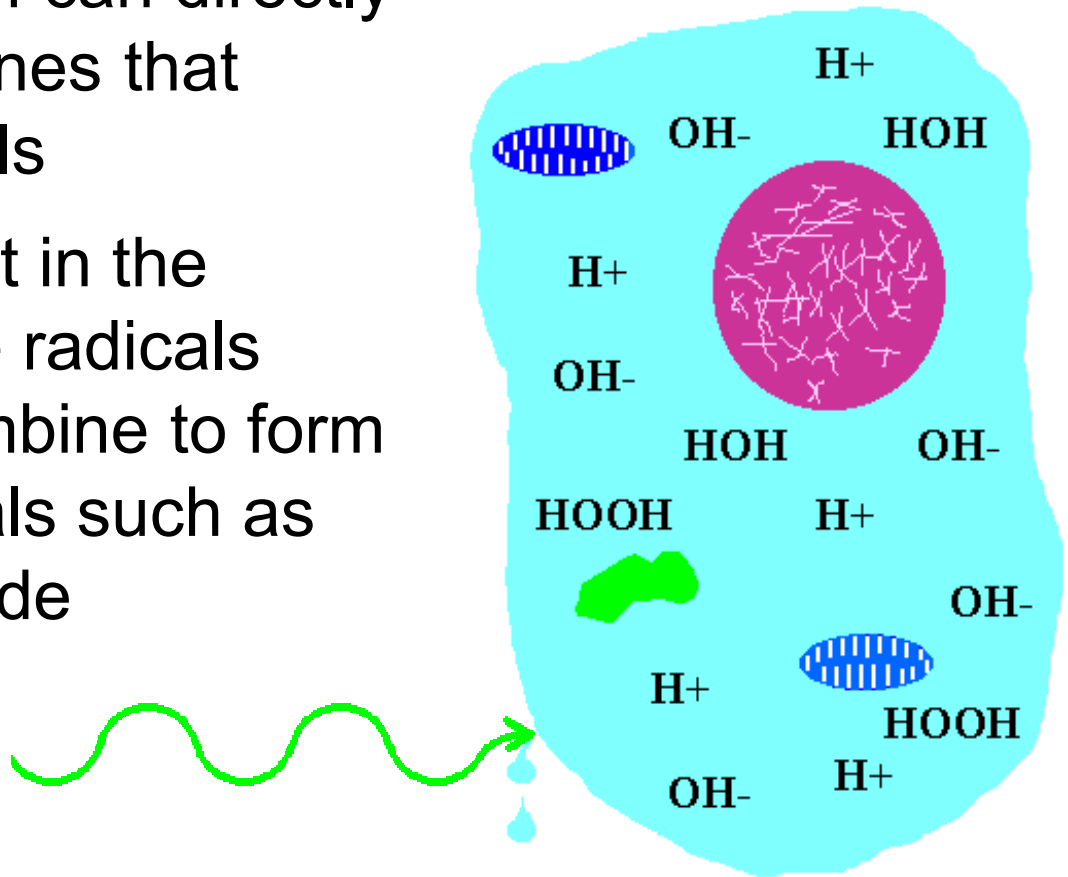
Cell Damage

The method by which radiation causes damage to human cells is by ionization of atoms in the cells. Any potential radiation damage begins with damage to atoms.



Cell Damage

- Ionizing radiation can directly rupture membranes that surround the cells
- Ionizations result in the formation of free radicals which can recombine to form harmful chemicals such as hydrogen peroxide

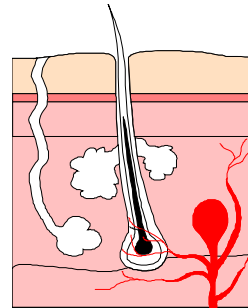
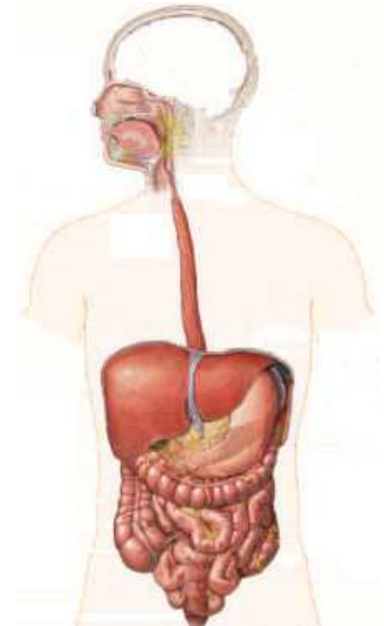
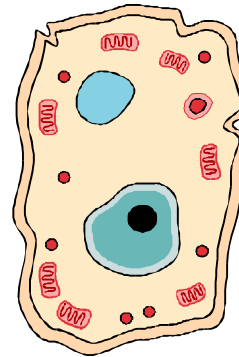


Highest Sensitivity

- Actively dividing cells
- Non-specialized cells

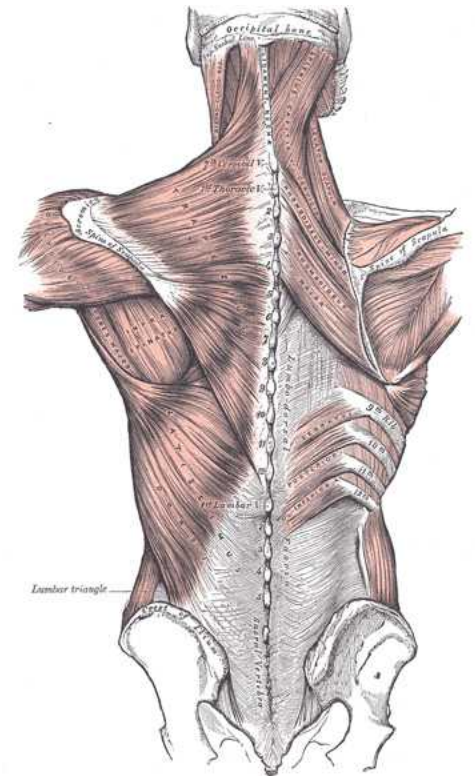
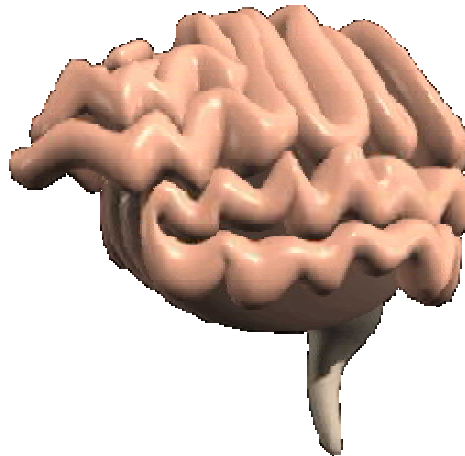
EXAMPLES:

- Blood forming cells
- Hair follicles
- Cells that form sperm
- Intestinal tract lining

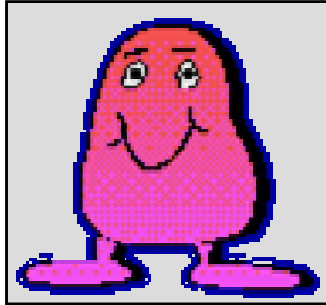


Lowest Sensitivity

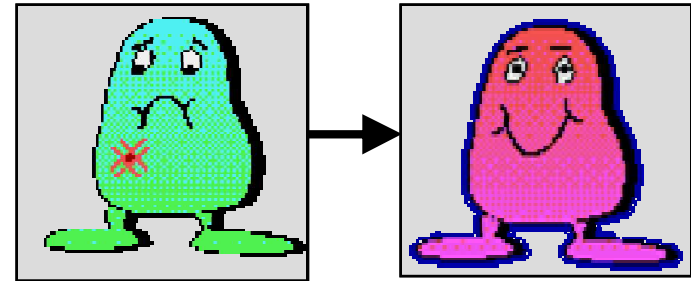
- Less actively dividing cells
- More specialized cells
- Brain cells
- Muscle cells



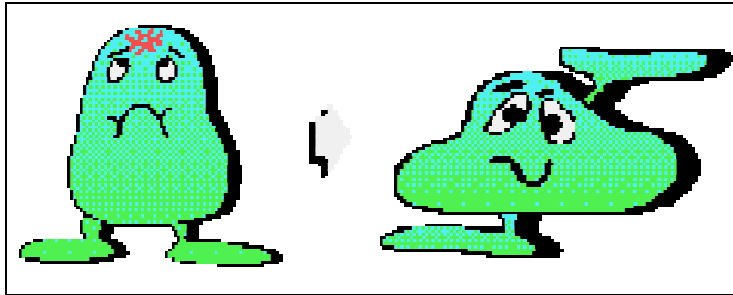
Possible Effects of Radiation on Cells



There is no damage



**Cells repair the damage
and operate normally**



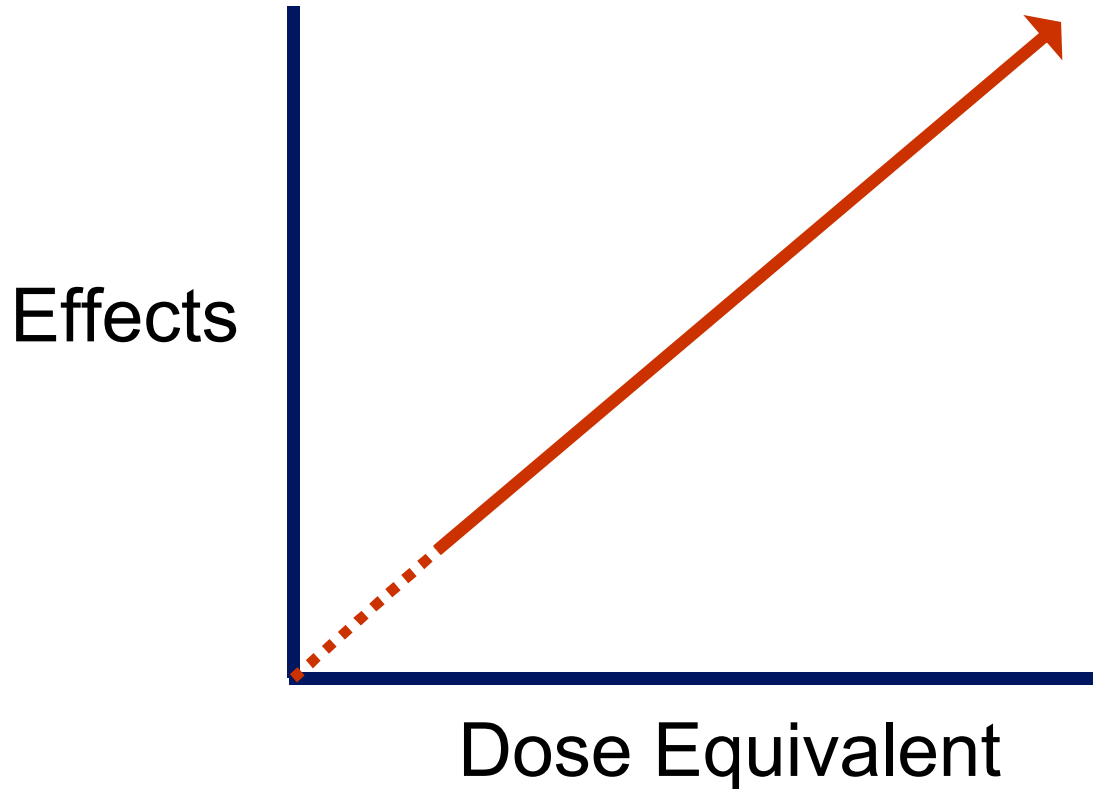
**Cells are damaged and
operate abnormally**



Cells die

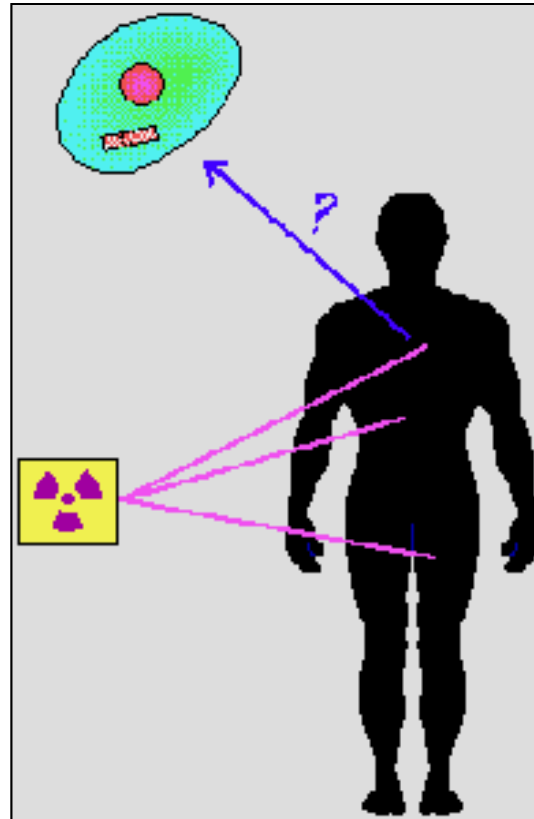
Total Dose

In general, the greater the dose, the greater the potential for biological effects.



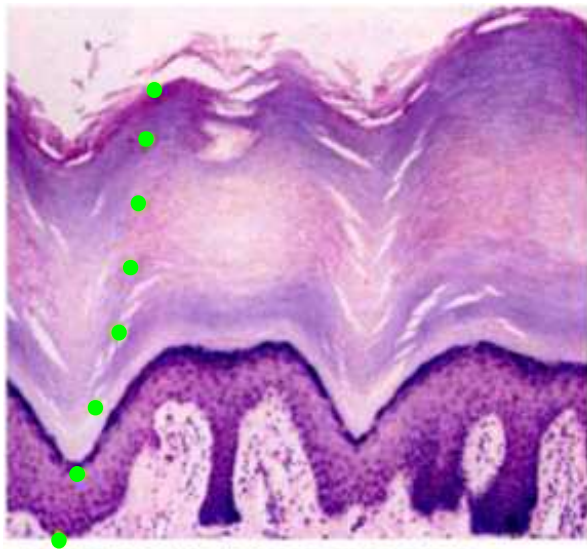
Dose Rate

The faster the dose is delivered, the less time the body has to repair itself.



Type of Radiation

For example, internally deposited alpha emitters are more damaging than x-rays or gamma rays for the same energy deposited.



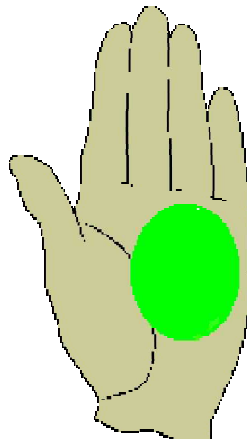
Specific ionization by
X-rays or gamma rays



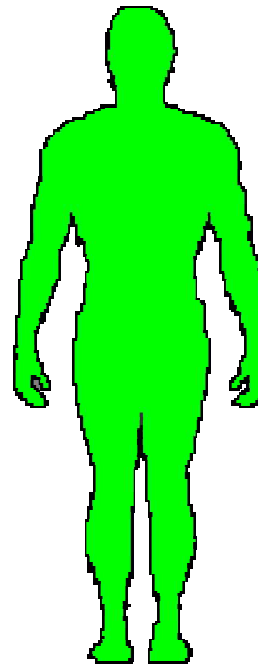
Specific ionization by
alpha particles

Area of Body Exposed

- In general, the larger the area of the body that receives a dose, the greater the biological effect.
- Extremities are less sensitive than blood forming and other critical organs.



vs.



Individual Sensitivity

- Age

The human body becomes less sensitive to ionizing radiation with increasing age; however, elderly people are more sensitive than middle-aged adults.

- Genetic make-up

Some individuals are more sensitive to environmental factors.

Acute vs. Chronic Dose

Potential biological effects depend on how much and how fast a radiation dose is received.

Radiation doses are grouped into:

- **Acute** - high dose of radiation received in a short period of time (seconds to days)
- **Chronic** - a small dose of radiation received over a long period of time (months to years)

Acute Exposure Effects

AVG DOSE	DAMAGE
2 - 5 rem	Annual Limit
25 - 50 rem	Slight Blood Changes
100 - 200 rem	Radiation Sickness
200 - 500 rem	Blood System Damaged
450 - 600 rem	LD 50-60
> 500 rem	Gastrointestinal Damage
> 5000 rem	Death Within 2-3 days

Effects of High-Level Acute Doses (Skin/Extremities)

Burns (Erythema)



Necrosis



Loss of fingers / limbs

Chronic Dose

A small dose of radiation received over a long period of time.

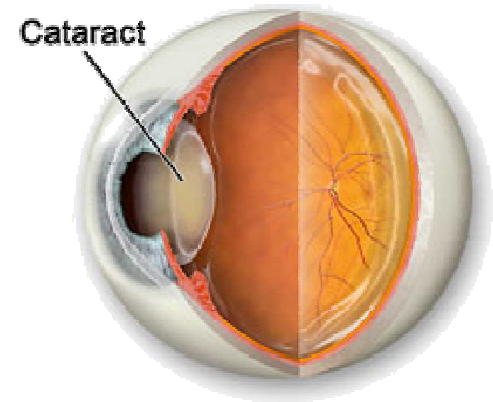
Typical examples are:

- The dose we receive from natural background
- The dose we receive from occupational exposure

Body is better equipped to tolerate chronic doses

Effects of Chronic Doses

- Increased risk of cataract formation (if over 450 rem)
- Increased risk of developing cancer



Cancer Risk

- Current rate of cancer death among Americans is about 20%.
- An individual who receives 25,000 millirem over a working life increases his/her risk of cancer by 1% to about 21%.
- The average annual dose to workers in nuclear industry is less than 100 millirem (work in nuclear power workers receive more than average) .

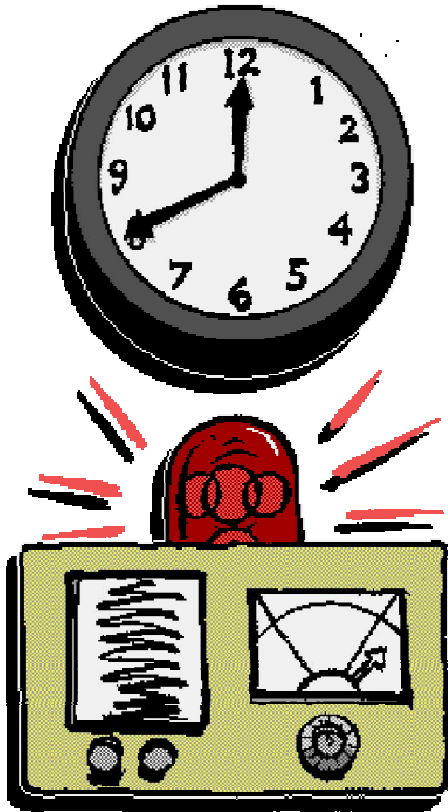
Comparison of Health Risks

Health Risk	Days Lost
Unmarried Male	3500
Tobacco User	2250
Unmarried Female	1600
Overweight Individual	777
Alcohol Consumer	365
Motor Vehicle Driver	207
100 mrem/yr for 70 yrs	10

Comparison of Occupational Risk

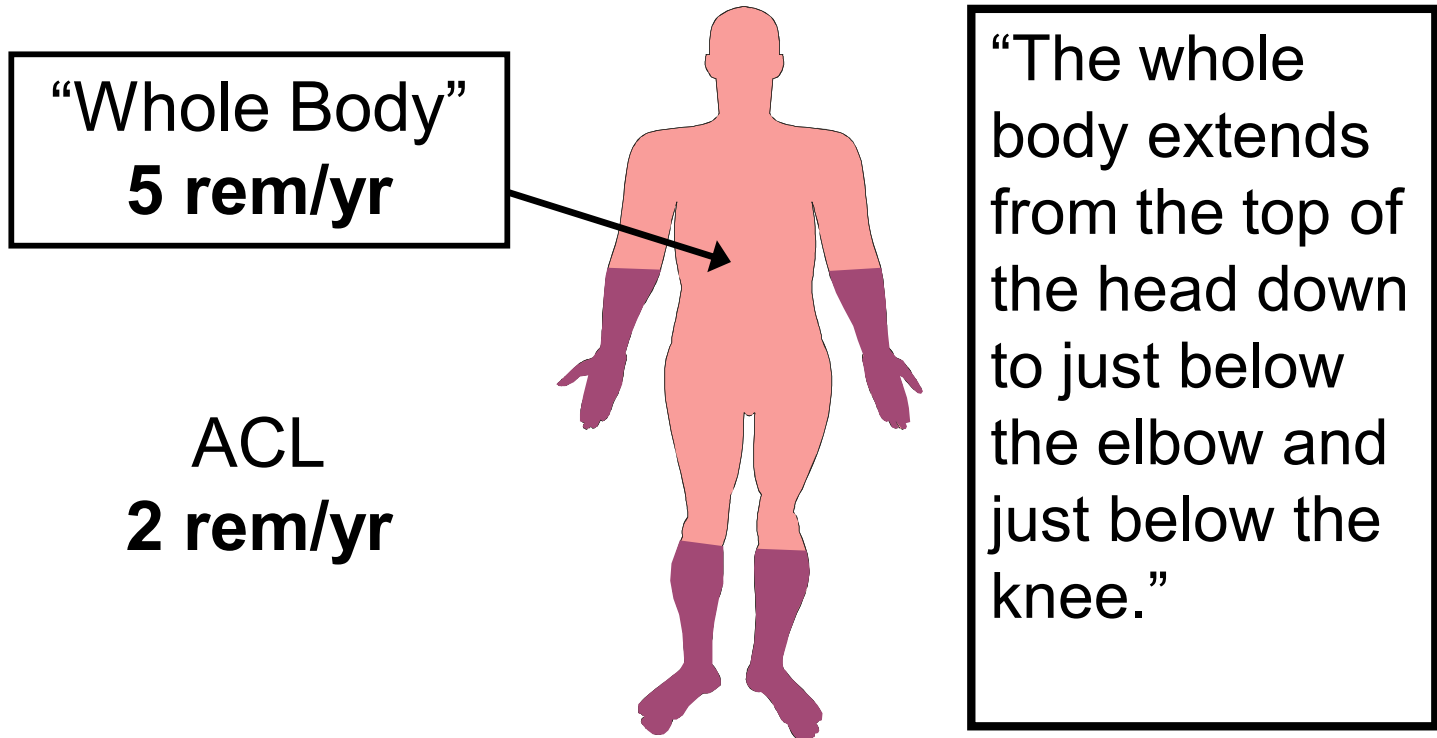
Industry	Days Lost
Coal Miner	328
Farmer	277
Transportation Worker	164
U.S. Average	74
Manufacturer	43
Radiological Worker	40
Trades Employee	30

Radiation Limits



- **Basis and Purposes for Dose Limits and Administrative Control Levels**
- **Dose Limits and Administrative Control Levels**
- **Worker Responsibilities**

DOE Whole Body Dose Equivalent Limit



NOTE: The whole body dose equivalent limit is based on the sum of internal and external dose.

Members of the Public

Member of the public - An individual who is not a general employee.

Public limited to 100 mrem in a year as a consequence of all routine activities.

As Low As Reasonably Achievable (ALARA)



Personal radiation exposure shall be maintained As Low As Reasonably Achievable.

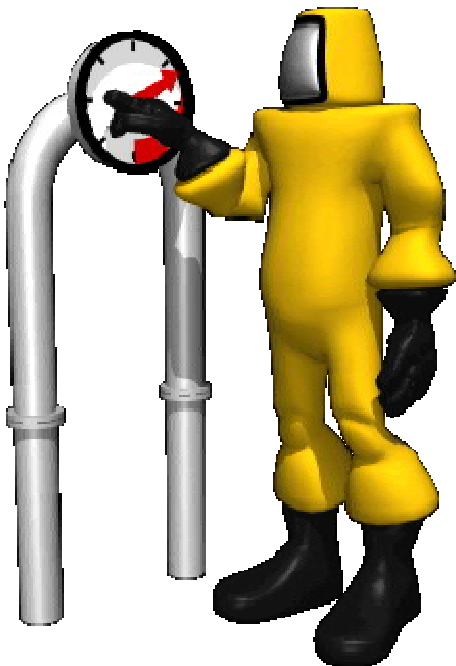
Radiation exposure to the work force and public shall be controlled such that:

- Radiation doses are well below regulatory limits.
- There is no radiation exposure without an overall benefit.

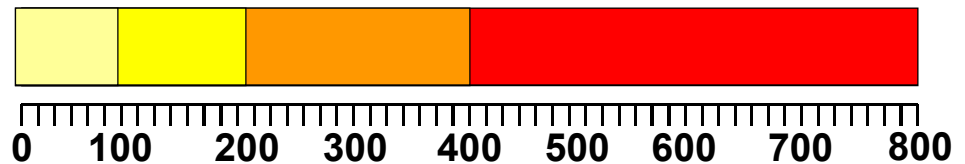
Time

Minimize time in a field of radiation

100 mrem/hour field



Time in Area = 8 hours

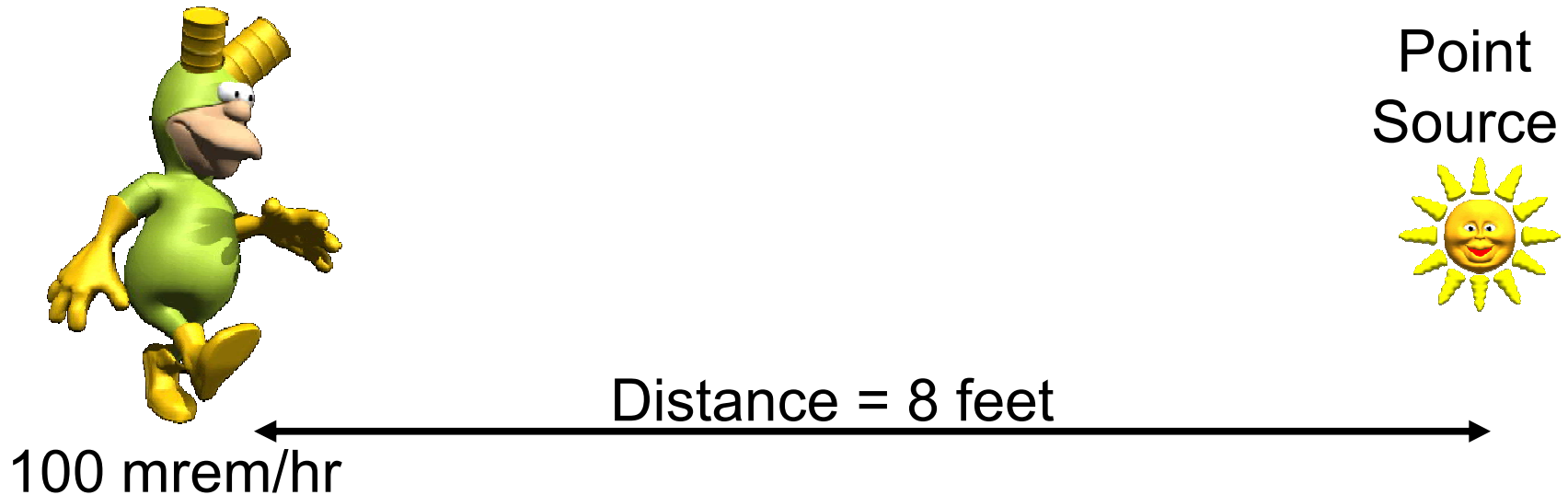


mrem received

Distance

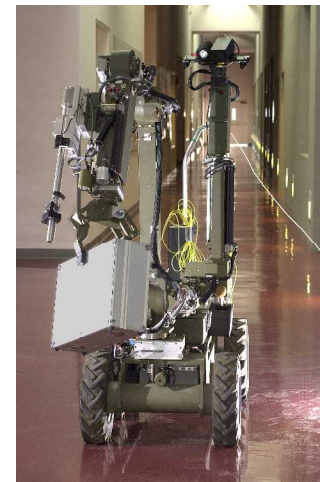
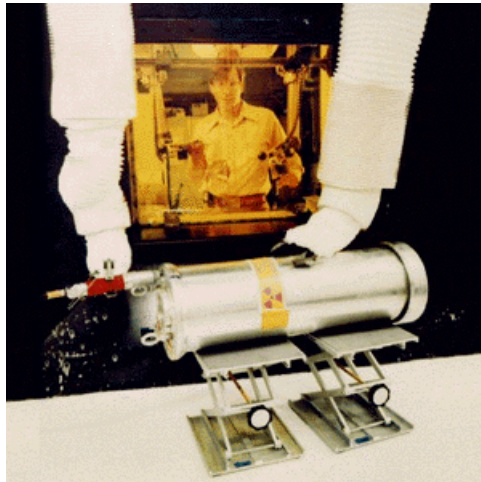
Maximize distance from a source of radiation

$$2 \times \text{Distance} = \text{Dose Rate} / 4$$

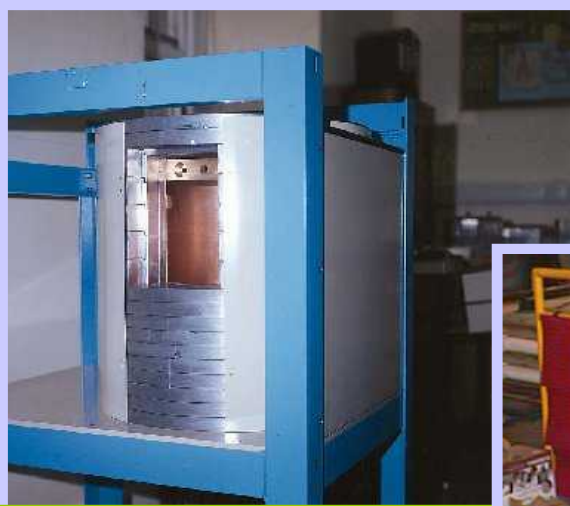


Methods for Increasing Distance

- ✓ Be familiar with radiological conditions in the area
- ✓ Move to lower dose rate areas during work delays
- ✓ Use remote handling devices when possible

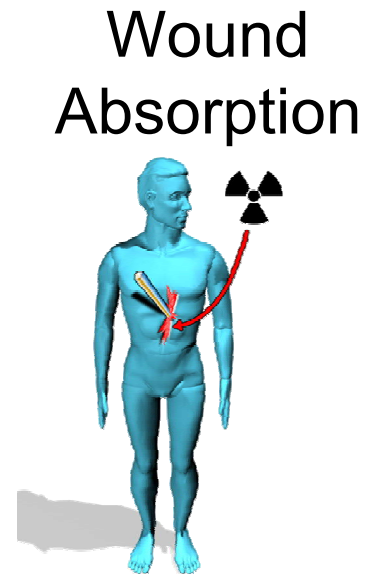
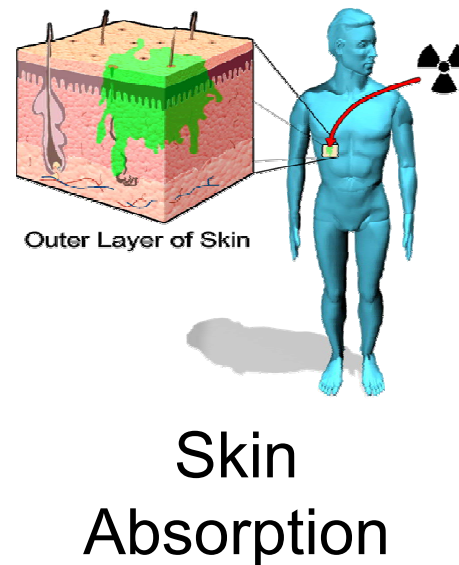
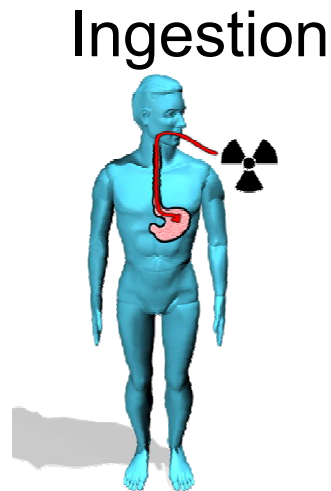
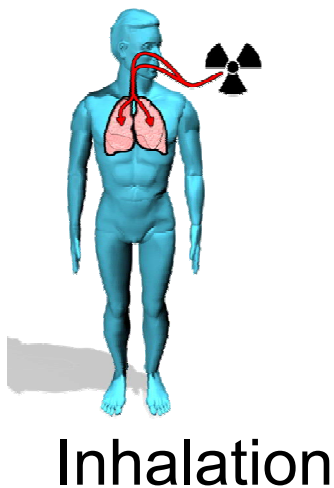


Shielding Examples



Internal Exposure Pathways

Internal dose results from radioactive material being taken into the body through:



Preventing Internal Exposure

- Engineering Controls
 - Fume Hood
 - Glove Bag
- Personnel Protective Equipment
 - Safety Glass
 - Lab Coats
 - Coveralls

Engineering Controls

- Glove Bag



- Glove Box



Personnel Monitoring Devices

Various types of dosimetry devices are used to measure personnel dose received from exposure to external sources of radiation:

**Whole-Body
Dosimeter**



**Extremity
Dosimeter**



**Electronic
Dosimeter**



**Pocket Ion
Chamber**



Internal Monitoring Methods

- Bioassay



- Whole-body counting



- Monitor air



Careers in New Mexico

- **National Laboratories (Sandia and Los Alamos)**
- **Waste Isolation Pilot Project (Carlsbad)**
- **Fuel Fabrication (Eunice)**
- **Mining (Grants)**
- **Government**
- **Universities (UNM and NMSU)**

Careers in New Mexico

- **Technician**

Education - H.S. Diploma

Salary \$30,000 to \$50,000

Demand exceeds availability

- **Professional**

**Education – B.S or M.S HP or
hard science**

Salary \$70,000 to mid \$150,000

Demand exceeds availability