

Effect of Particle Size on Oxy-Fuel Combustion Rates of Pulverized Coal

Manfred Geier, Ethan Hecht, and Chris Shaddix

Combustion Research Facility
Sandia National Laboratories
Livermore, CA 94550

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Approach and Research Goals

- Perform experiments in Sandia's optical entrained flow reactor for project coals, measuring char combustion temperature/particle size statistics in oxyfuel environments
- Apply SKIPPY (= Surface Kinetics in Porous Particles) to understand contribution of boundary layer reactions and CO_2 and H_2O gasification reactions to measured char particle temperatures for different O_2 concentrations and particle sizes
- Derive appropriately detailed, CFD-friendly char combustion model that accounts for vagaries of oxyfuel combustion (char gasification reactions; boundary layer conversion of CO)

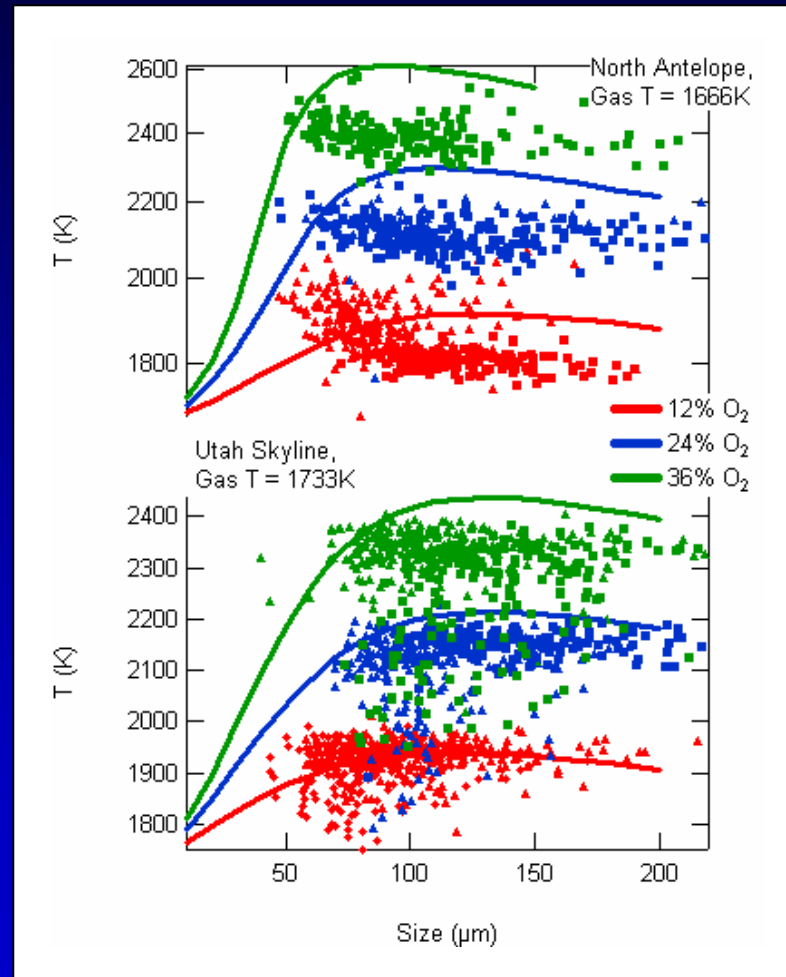


Status as of Last Review (Jan. 2010)

- Char combustion temperature/particle size statistics collected in oxyfuel environments with $12 \text{ vol-\%} < [\text{O}_2] < 36 \text{ vol-\%}$
- Single-film modeling of data with n th-order Arrhenius oxidation model shows overprediction of particle T with increasing O_2 levels
- SKIPPY simulations with base oxidation model shows even larger overprediction with increasing O_2 levels, unless substantially decrease assumed surface area density
- Conclusions:
 - CO_2 gasification reaction may need to be included (in both modeling approaches)
 - measurements should be made on prepared char particles to delineate effect of devolatilization environment on effective char surface area (i.e. reactivity)



Previous Single-Film Modeling Results

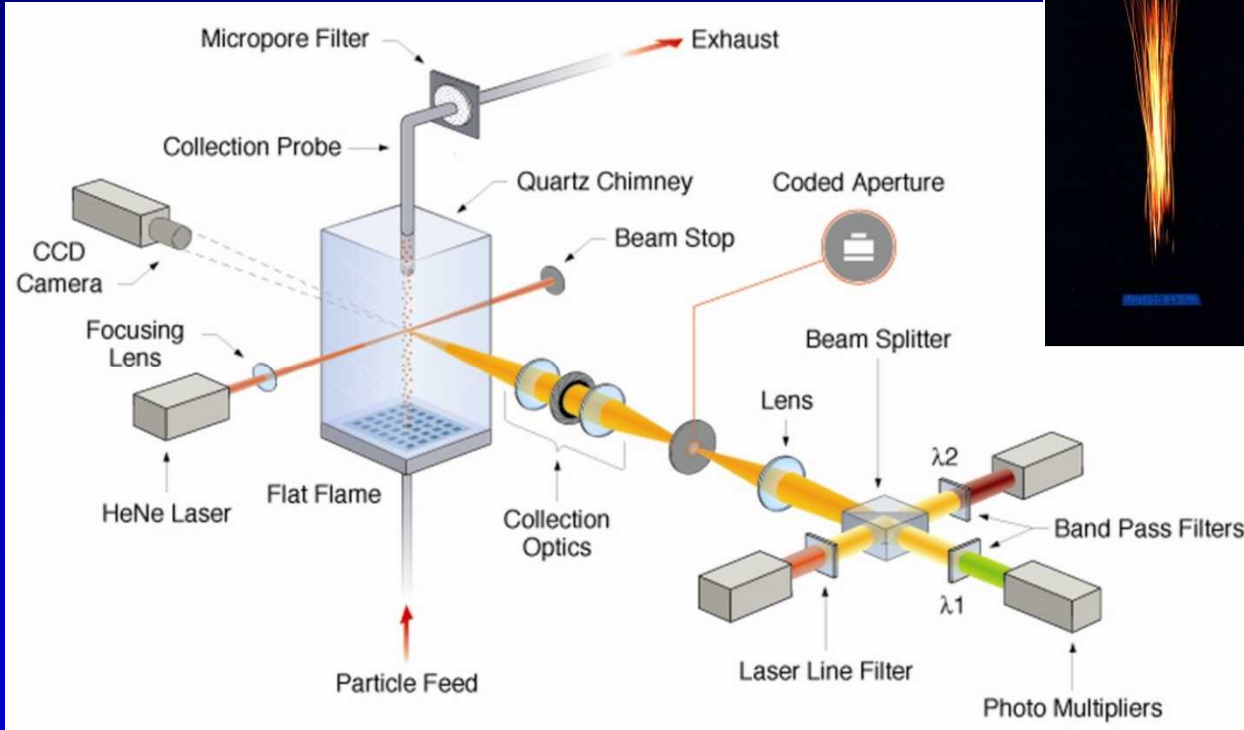


New Work

- Char combustion temperature/particle size statistics collected in N_2 environments with $12 \text{ vol-\%} < [\text{O}_2] < 36 \text{ vol-\%}$, so models have baseline dataset with minimal gasification reactions to calibrate oxidation model
- Char particle combustion data collected for partial set of conditions using char particles prepared in drop tube reactor
- CO_2 gasification reaction, and then steam gasification reaction, added to SKIPPY model simulations (16 vol-% H_2O in experimental furnace environment)
- Single-film, n th-order Arrhenius char combustion model fit to experimental data for different values of n

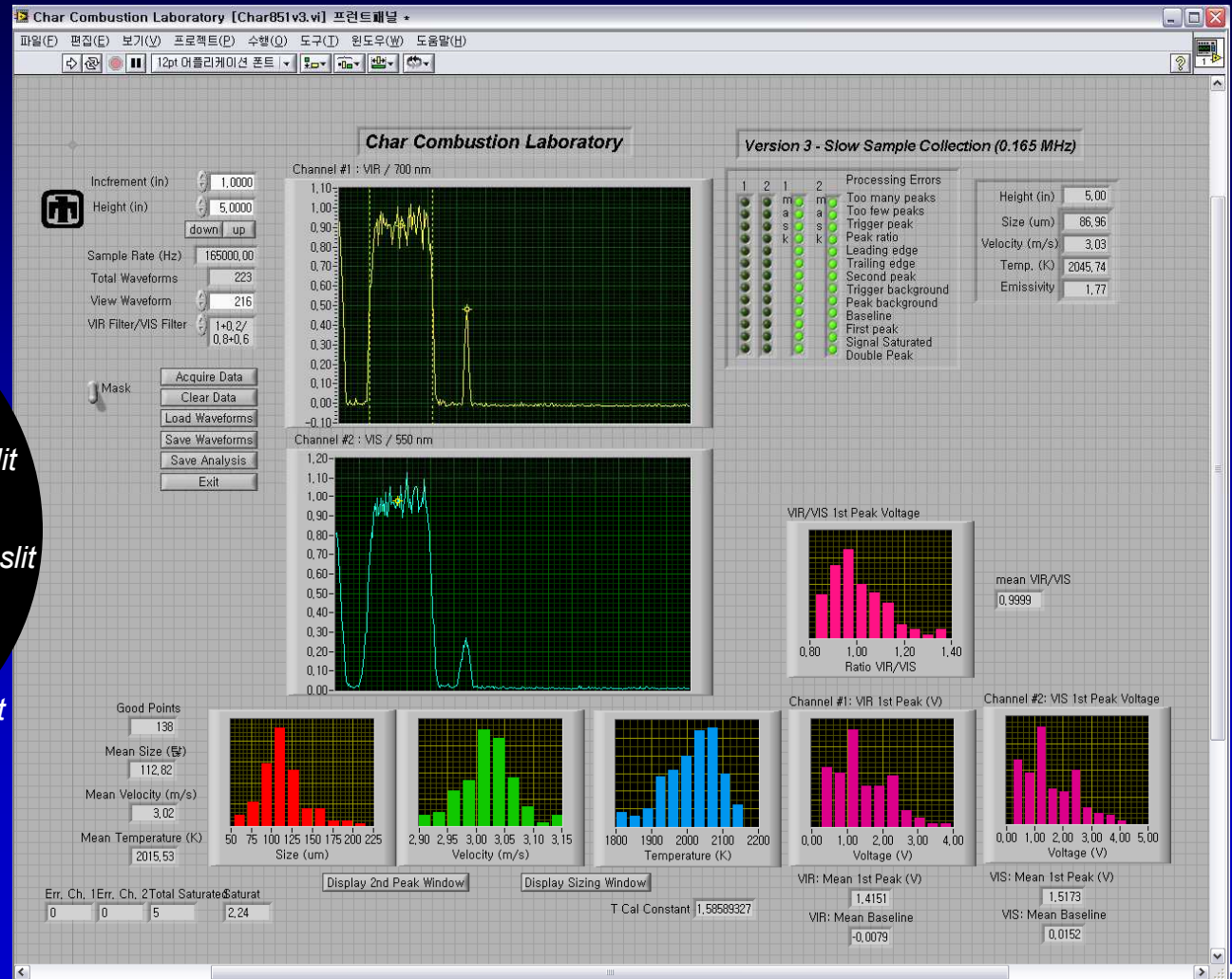
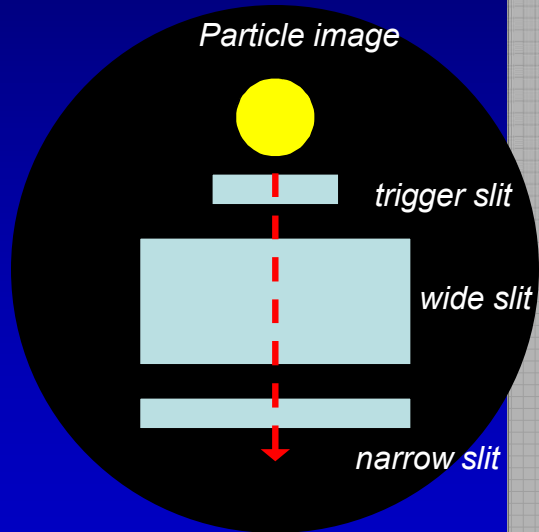


Experiments use a combustion-driven optical entrained flow reactor

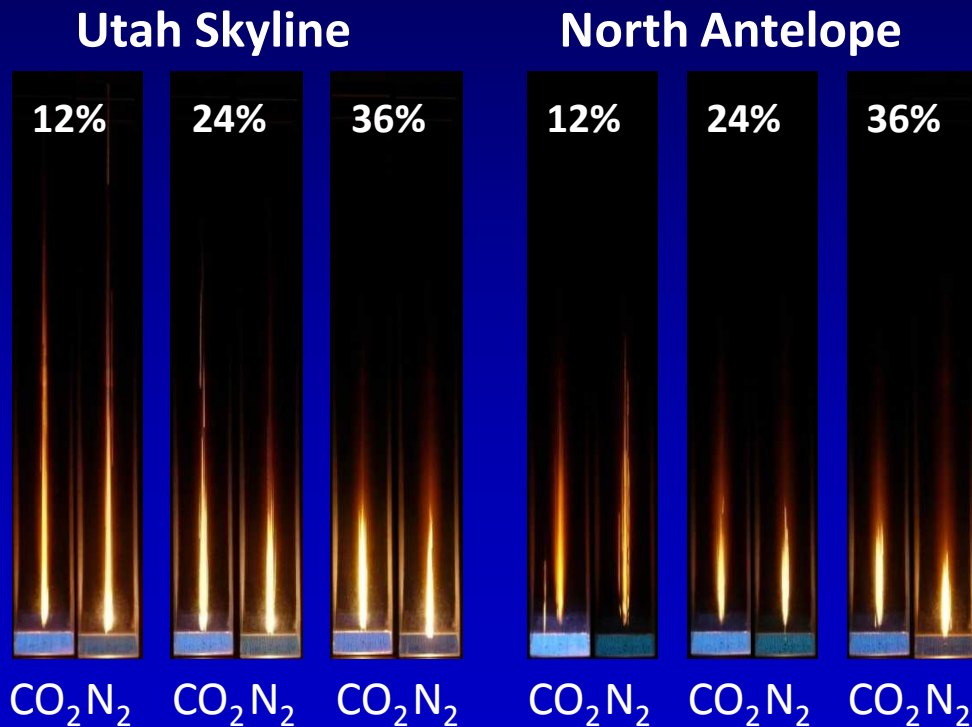


- 1 atm
- compact, diffusion-flamelet burner
- coal or char particles introduced along centerline
- quartz chimney
- coded-aperture, 2-color pyrometry diagnostic for char size, T , and velocity
- laser-triggered ICCD for single particle imaging

Particle-sizing pyrometry in the laminar entrained flow reactor

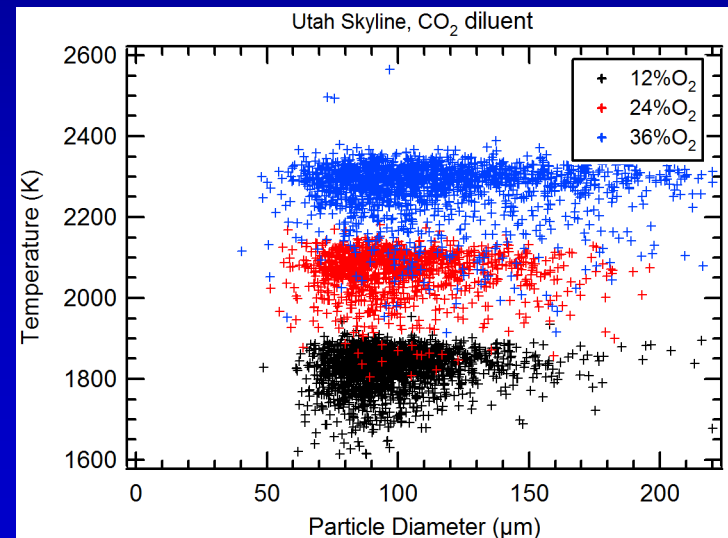
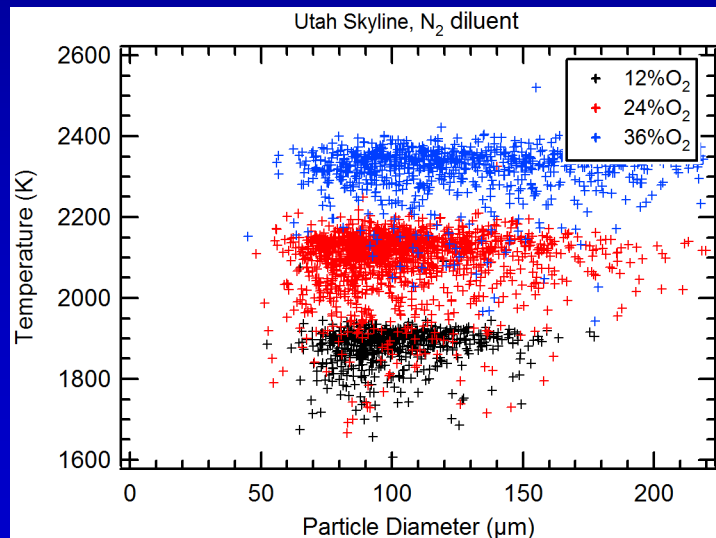
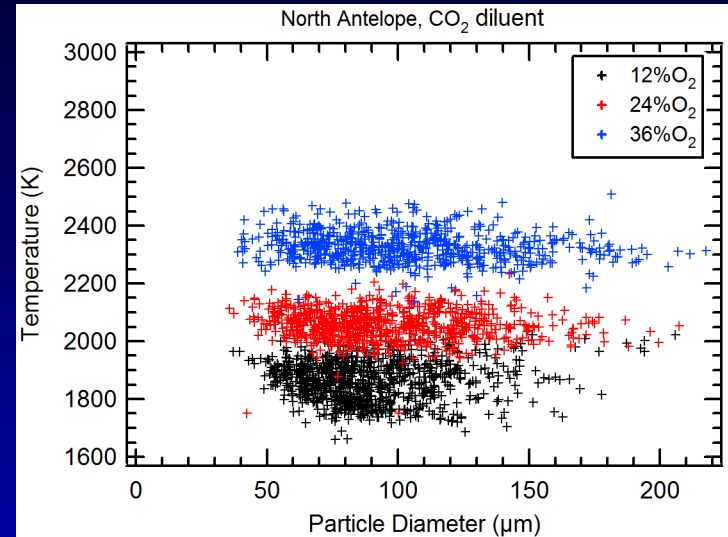
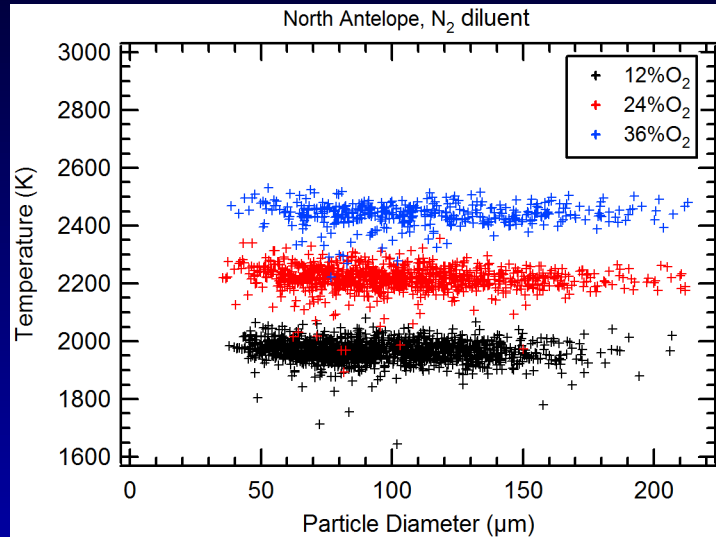


A constant-speed coal feeder ensured steady introduction of pre-sieved coal

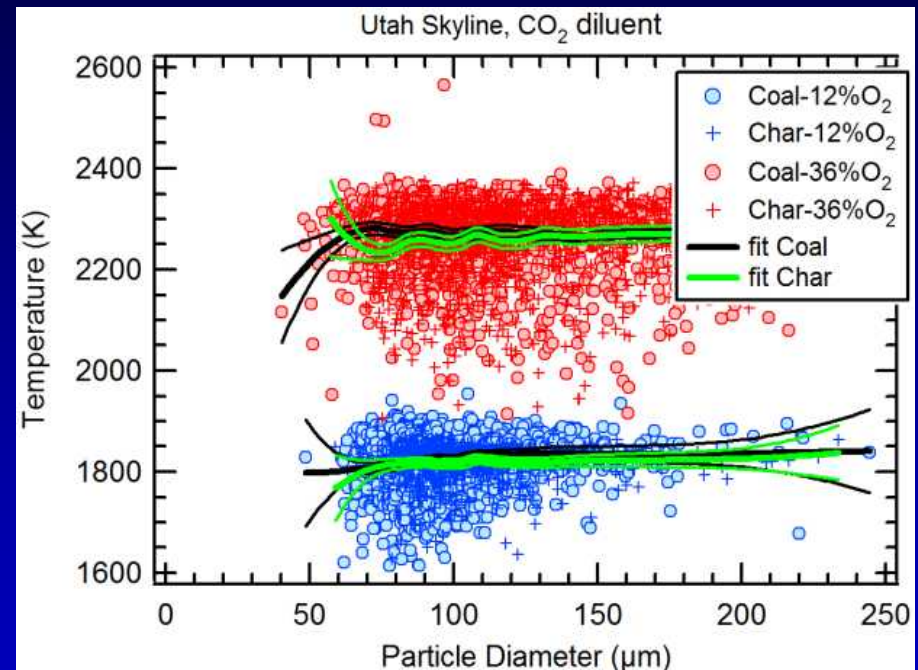
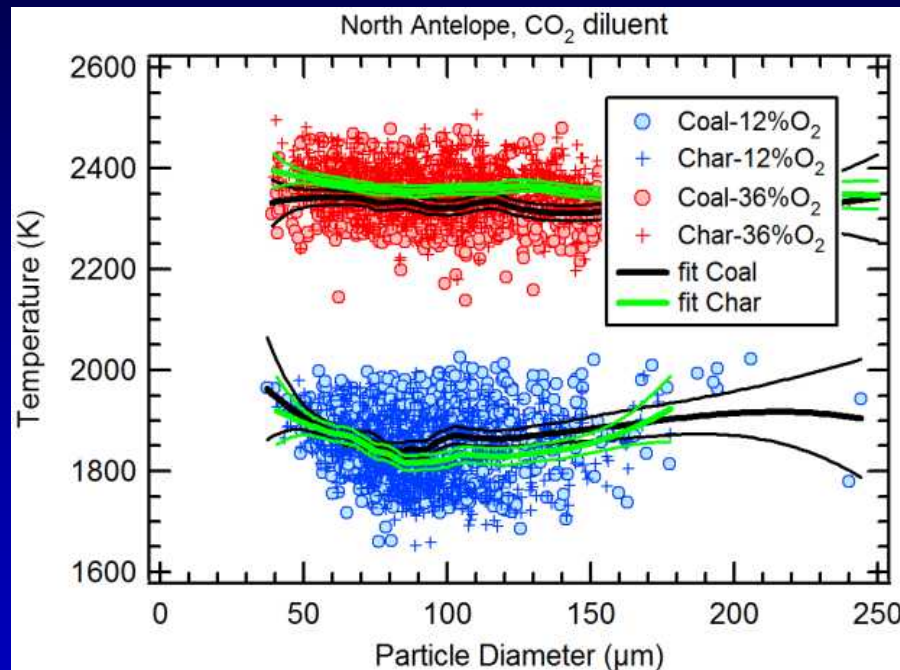


- 2 Coal ranks
Bituminous (Utah Skyline)
Subbituminous (North Antelope)
- 75 – 106 μm size fraction
- 3 Oxygen concentrations
12, 24, and 36 vol-%
- ~ 1700 K furnace temperature
- 16% moisture
- 60 slpm overall product flow rate
- < 1g/hr coal feed rate
- N_2 or CO_2 diluent gas

Particle-sizing pyrometry data (coal feed)



Particle-sizing pyrometry data comparison for raw coal and prepared char feed

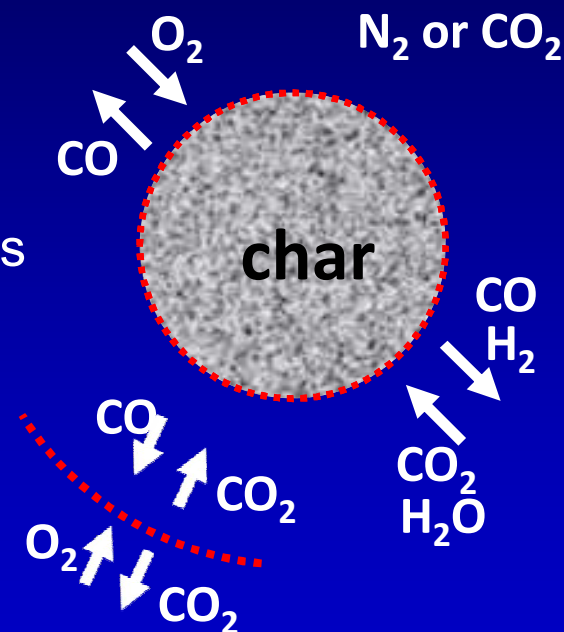


- Chars prepared by feeding 75–150 μm size-classified coal into drop tube reactor operating at 1200 deg C with 3% O₂ in N₂; 0.25 s res. time
- Chars collected in water-cooled, gas-quench sampling probe, separated in cyclone, then sieved to 75–106 μm for combustion experiments

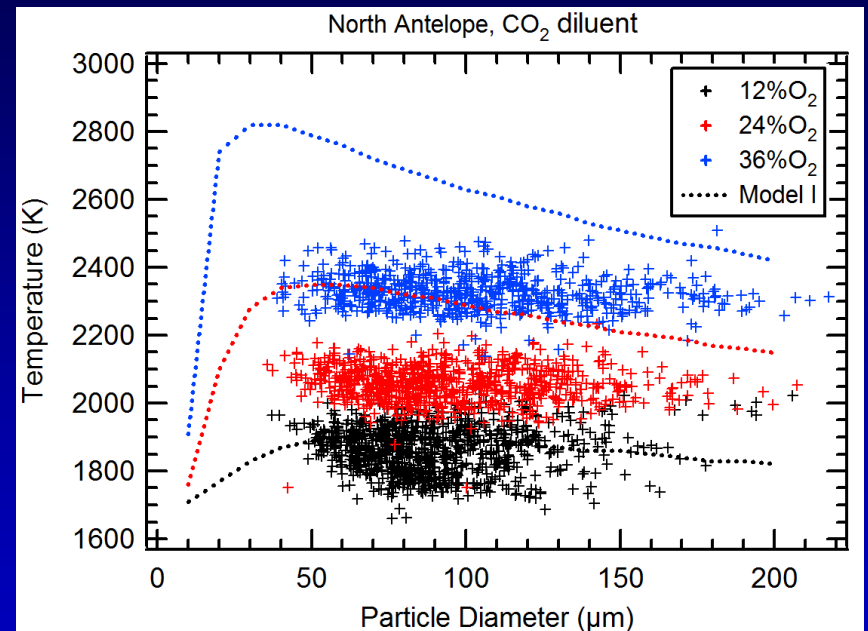
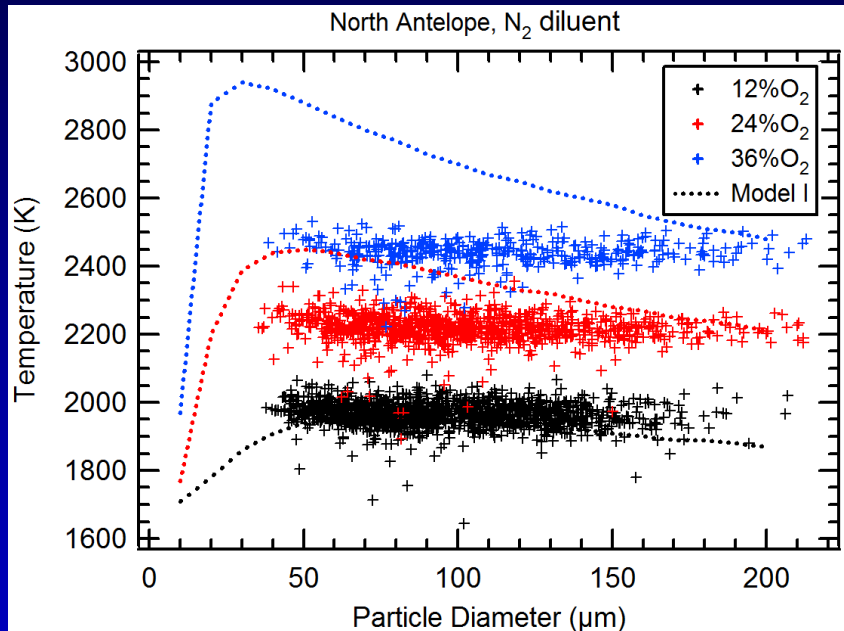
SKIPPY calculates temperature and species profiles for steady-state burning of porous char particles

- SKIPPY = Surface Kinetics in Porous Particles
- Utilizes CHEMKIN II surface kinetic approach
- Our simulations:
 - GRI-MECH 3.0 gas-phase kinetics
 - typical char properties, experimental conditions

Reaction	A (g/cm ² s)	E (kJ/mol)
Heterogeneous oxidation:		
(R1) C_s + O ₂ => CO + O_s	3.3E+15	167.4
(R2) O_s + 2C(b) => CO + C_s	1.0E+08	0.
(R3) C_s + O ₂ => O _{2_s} + C(b)	9.5E+13	142.3
(R4) O _{2_s} + 2C(b) => C_s + CO ₂	1.0E+08	0.
CO₂ gasification reaction:		
(R5) C_s + CO ₂ => CO + O_s + C(b)	3.60E+15	251.0
Steam gasification reaction:		
(R6) C_s + H ₂ O => H ₂ + O_s + C(b)	4.35E+14	222.8



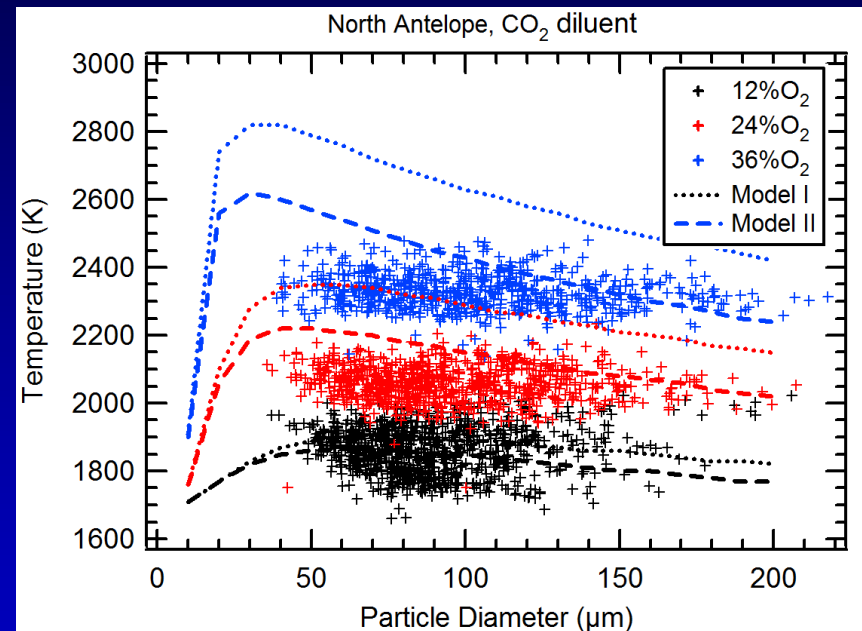
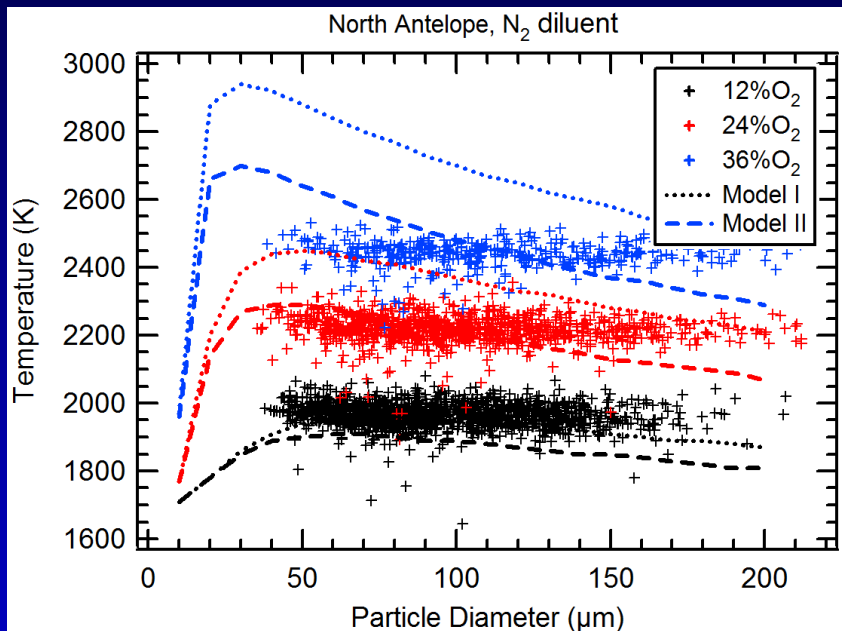
SKIPPY predicted and measured single particle temperatures



Use constant assumed surface area density: 10 m²/g

- Model I: char oxidation only

SKIPPY predicted and measured single particle temperatures

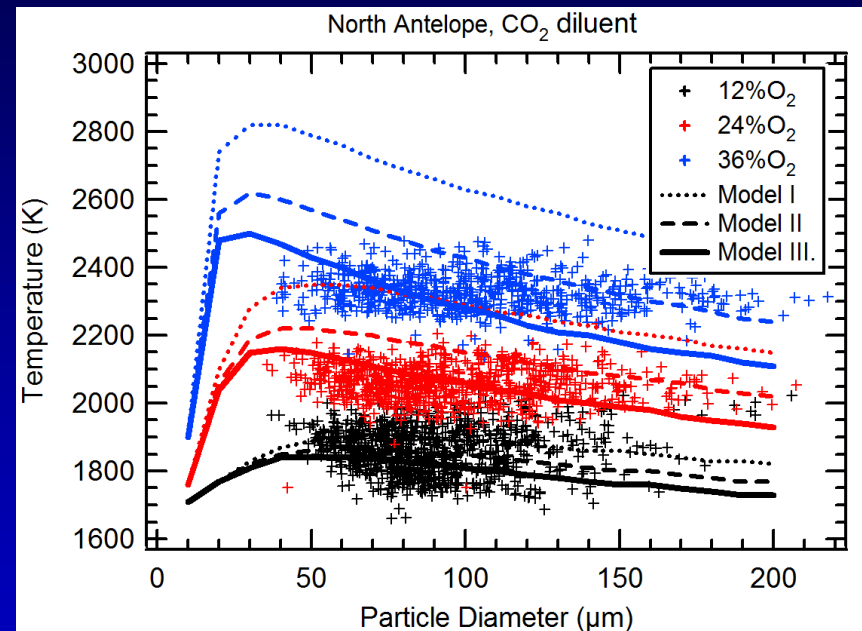
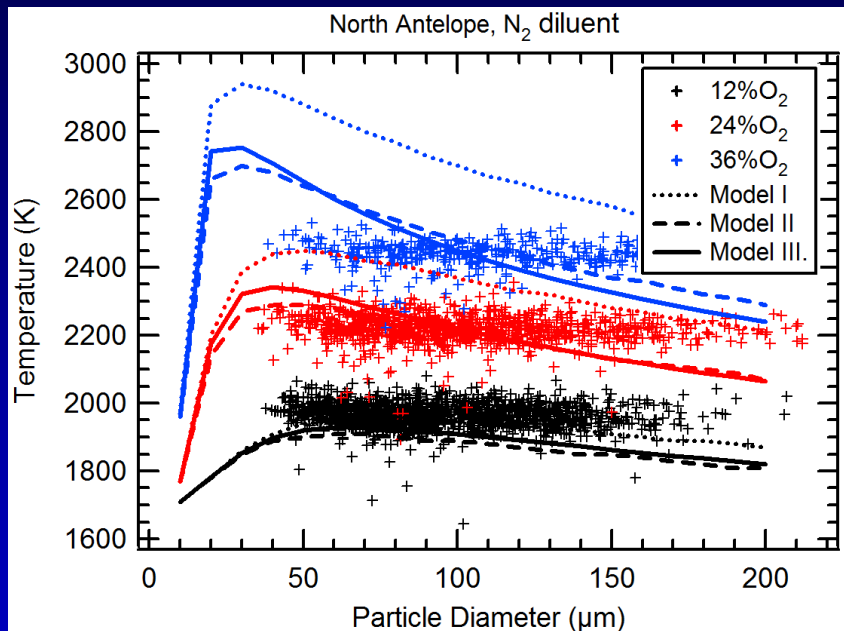


Use constant assumed surface area density: 10 m²/g

- Model I: char oxidation only
- Model II: char oxidation + steam gasification



SKIPPY predicted and measured single particle temperatures



Use constant assumed surface area density: 10 m²/g

- Model I: char oxidation only
- Model II: char oxidation + steam gasification
- Model III: char oxidation + steam gasification + CO₂ gasification



Single-film model prediction of particle temperature as function of particle size

- Use single-film n th-order Arrhenius kinetic rate expression for oxidation

$$k_s = A \exp(-E/RT_p) \quad q = k_s p_{O_2,s}^n$$

- Solve instantaneous energy balance for a reacting particle:

$$\frac{d_p \rho_p C_{v_p}}{6} \frac{dT_p}{dz} = -\varepsilon \sigma (T_p^4 - T_w^4) - \frac{2\lambda}{d_p} \left[\frac{\kappa/2}{e^{\kappa/2} - 1} \right] (T_p - T_g) + q \Delta h$$

- Determine reaction heat release using Tognotti CO₂/CO production ratio

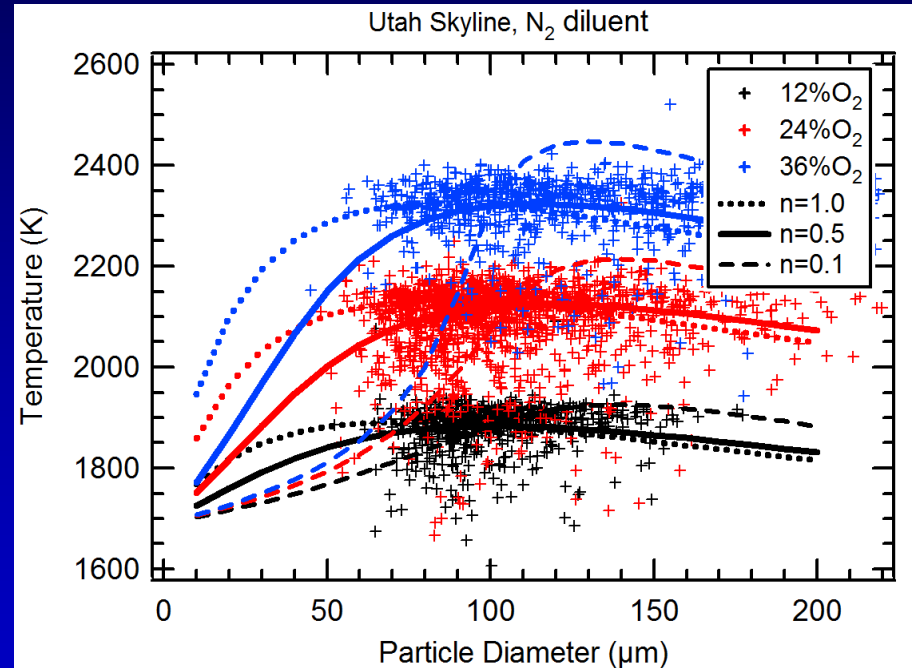
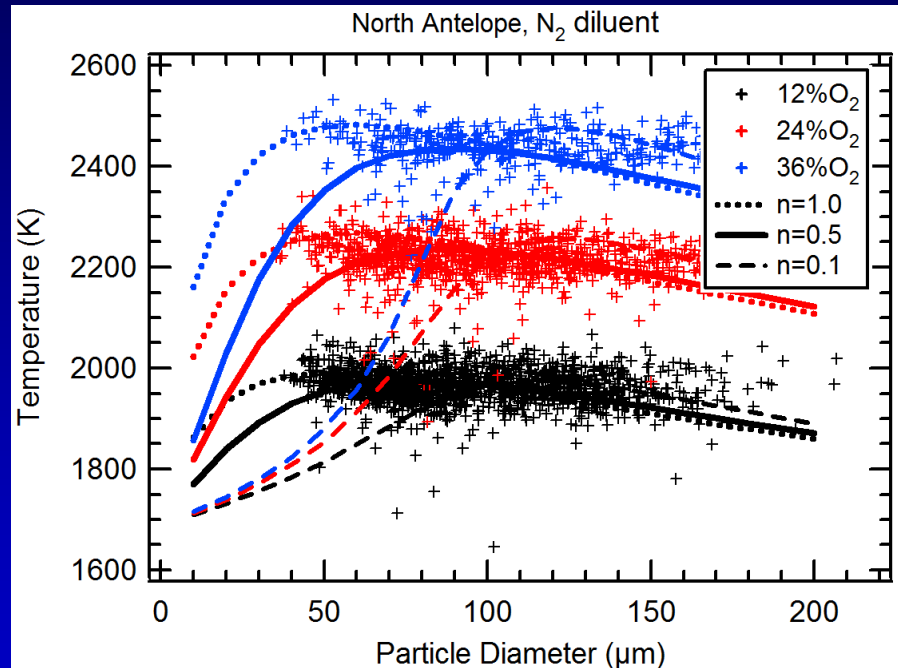
$$\text{CO}_2/\text{CO} = 0.02 p_{O_2,s}^{0.21} \exp(3070/T_p)$$

- $dT_p/dz = 0$



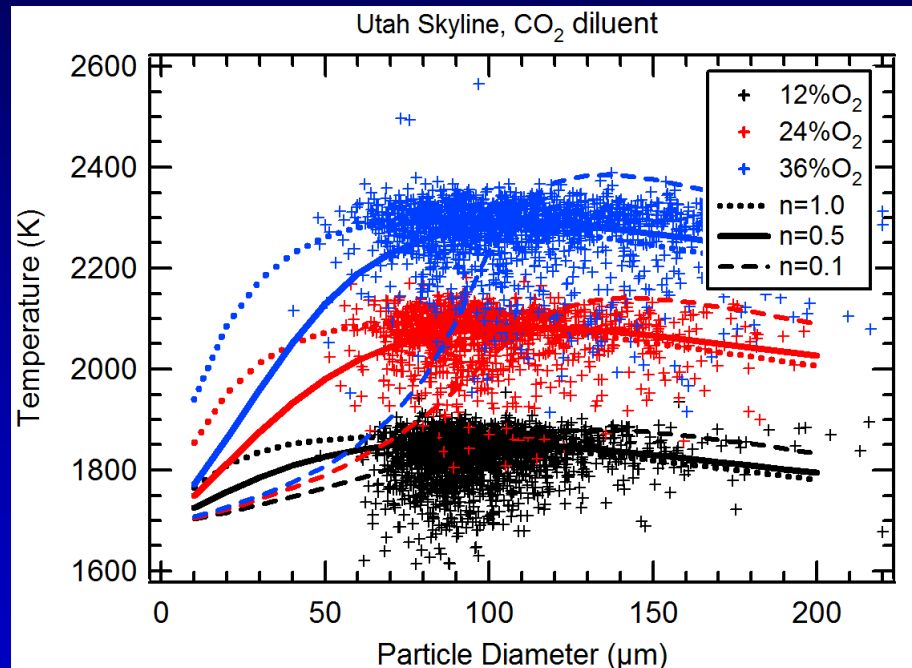
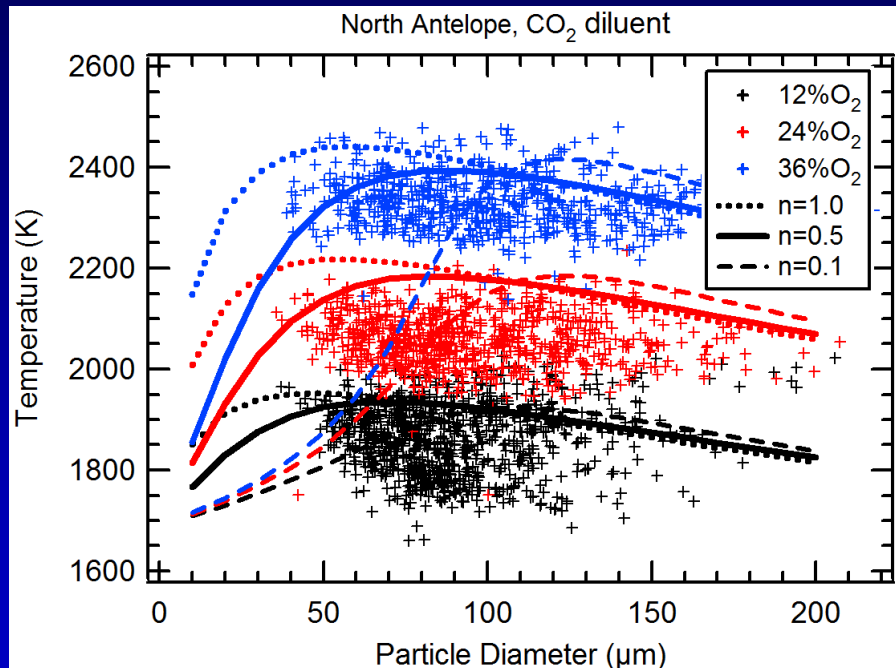
Single-film oxidation simulation results

N_2 diluent



Single-film oxidation simulation results

CO₂ diluent



Single-film oxidation simulation results

*n*th-order fit parameters

Fixed <i>n</i> =	North Antelope:	
	A (kmol/m ² ·s)	E (kJ/mol)
1	0.44	0.0
0.5	0.67	36.5
0.1	1.70	76.3
Fixed <i>n</i> =	Utah Skyline:	
	A (kmol/m ² ·s)	E (kJ/mol)
1	0.39	12.0
0.5	1.05	51.8
0.1	2.31	83.5

Non-physical fit parameters (e.g. $E_a \ll 80$ kJ/mol)
suggest important physics missing from model; danger
in extrapolating beyond fit conditions



Conclusions

- North Antelope coal shows small sensitivity of reactivity to devolatilization of isolated particles in flow reactor with different O₂ levels (whereas Utah Skyline is insensitive) – char combustion studies with preformed char particles recommended (in general)
- SKIPPY simulations show both steam and CO₂ gasification reactions need to be included to get acceptable agreement in measured and calculated burning rate (i.e. char particle temperature) for both conventional and oxyfuel combustion conditions
- Single-film simulations with n th-order Arrhenius oxidation show decent agreement with data as long as $n = 0.5$ or greater. Fits for N₂ diluent data give high prediction for North Antelope data during oxyfuel combustion. Activation energies for these fits with large n are very low.



Next Steps

- Investigate effect of assumed surface area density in SKIPPY on variation of predicted char particle temperatures as function of particle size
- Implement single-film model that includes char oxidation and gasification reactions – compare optimized predictions against data
- Develop recommendations for best CFD-friendly way to model char combustion rates of North Antelope and Utah Skyline coals during oxyfuel combustion



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Thank you!

