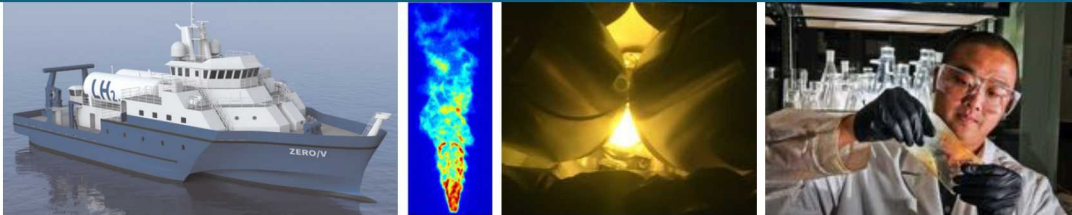




Hydrogen and Fuel Cells (H₂FC) Program

Enabling Hydrogen Technologies with Science and Engineering

*CSU – East Bay
Oct. 11, 2019*



PRESENTED BY

Jon Zimmerman and Mark Allendorf

jjzimmer@sandia.gov



Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525. SAND2019-6463 PE



2 | Sandia Brings Value through Research in Materials and Safety

Sandia provides deep, quantitative understanding and a scientific basis for...

Materials – for hydrogen production, storage and utilization

Safety – risk analysis and the creation of risk-informed standards

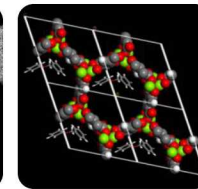
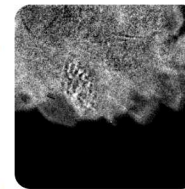
Hydrogen Storage Materials and Solutions

Hydrogen Production from Renewables



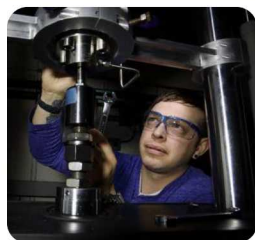
Advanced water-splitting materials and technologies for large-scale H₂ production

Discovering the behavior and performance of solid storage materials

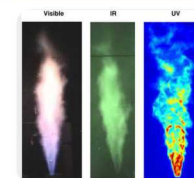


Hydrogen Fueling Infrastructure

Hydrogen-Materials Compatibility



Low-cost, hydrogen-compatible materials and the science basis for their qualification



H₂ release behavior and risk assessment to define the safety envelope for storage & delivery

Technologies for Non-Vehicle Applications

Marine and rail applications

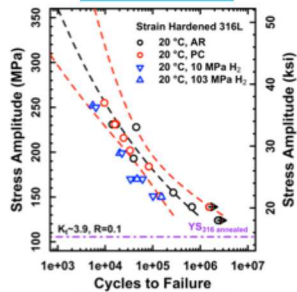




3 Developing the Technical Basis for Safety, Codes & Standards (SCS)

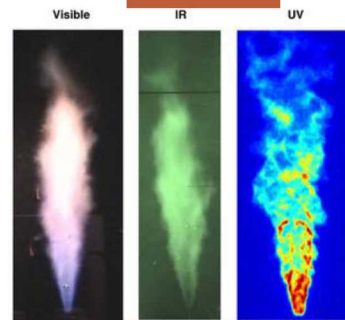
➤ Eliminate deployment barriers by informing codes and standards.

Hydrogen-Materials Compatibility



Understand material behavior

Hydrogen Behavior



Develop and validate scientific models

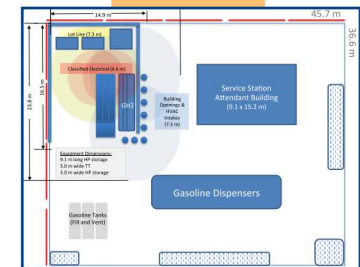
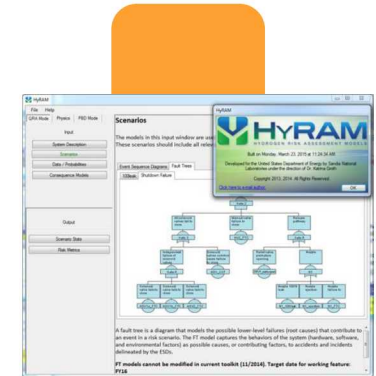
Quantitative Risk Assessment



Specific Component Type	Severity	Frequency	Units	Leak Size Description
Piping				
Pipe Diameter (d) < 50 mm	Rupture	1.00E-06	Per Meter Year	100% cross sectional area
d < 50 mm	Major	5.00E-06	Per Meter Year	1" (25 mm)
50 mm ≤ d < 149 mm	Rupture	5.00E-07	Per Meter Year	100% cross sectional area
50 mm ≤ d < 149 mm	Medium	2.00E-06	Per Meter Year	1" (25 mm)
150 mm ≤ d < 299 mm	Rupture	2.00E-07	Per Meter Year	100% cross sectional area
150 mm ≤ d < 299 mm	Major	4.00E-07	Per Meter Year	1/3 of pipe diameter
150 mm ≤ d < 299 mm	Minor	7.00E-07	Per Meter Year	1" (25 mm)
300 mm ≤ d < 499 mm	Rupture	7.00E-08	Per Meter Year	100% cross sectional area
300 mm ≤ d < 499 mm	Major	2.00E-07	Per Meter Year	1/3 of pipe diameter
300 mm ≤ d < 499 mm	Medium	4.00E-07	Per Meter Year	2" (50 mm)
300 mm ≤ d < 499 mm	Minor	5.00E-07	Per Meter Year	1" (25 mm)
500 mm ≤ d < 1000 mm	Rupture	2.00E-08	Per Meter Year	100% cross sectional area
500 mm ≤ d < 1000 mm	Major	1.00E-07	Per Meter Year	1/3 of pipe diameter
500 mm ≤ d < 1000 mm	Minor	2.00E-07	Per Meter Year	2" (50 mm)
500 mm ≤ d < 1000 mm	Very Small	4.00E-07	Per Meter Year	1" (25 mm)

Apply quantitative risk assessment techniques

Integration

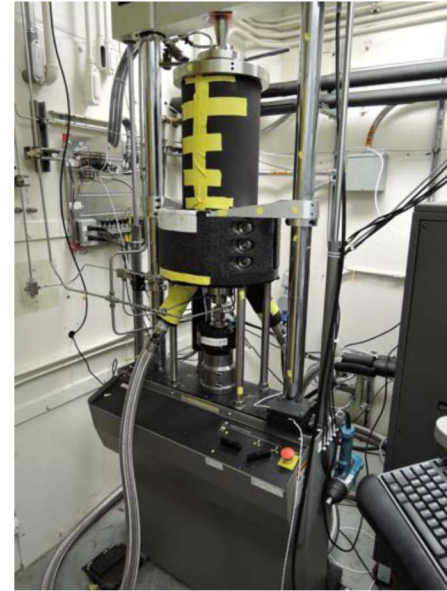
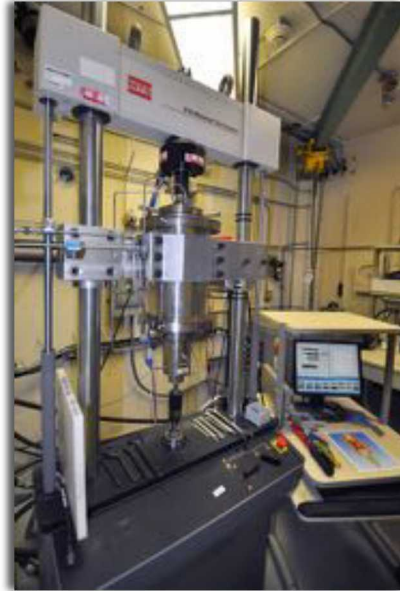
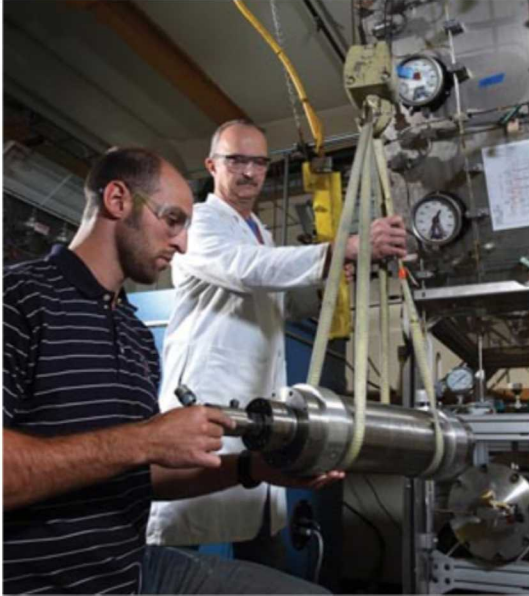


Develop integrated methods and algorithms



4 Hydrogen Effects on Materials Laboratory

This lab houses specialized equipment for evaluating materials in high-pressure gaseous hydrogen.



- **Fracture, fatigue and tensile testing** in high-pressure gaseous hydrogen at controlled temperature
- **Constant-displacement crack growth** testing in gas
- **Thermal pre-charging** to produce specific H₂ content
- **Pressure cycling** of metals and non-metals in controlled pressure and temperature

B.P. Somerday *et al*, *International Journal of Hydrogen Energy* (2017)

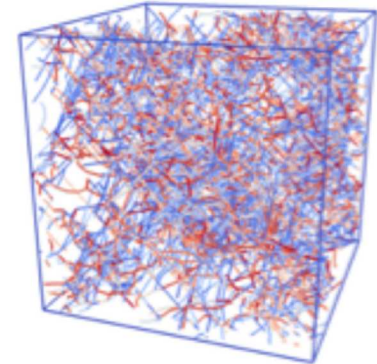
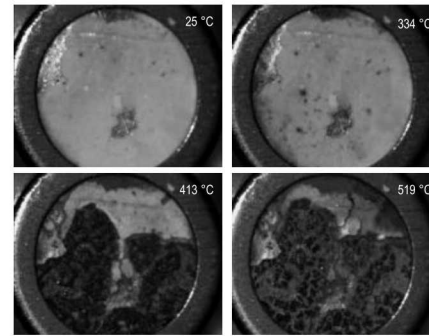
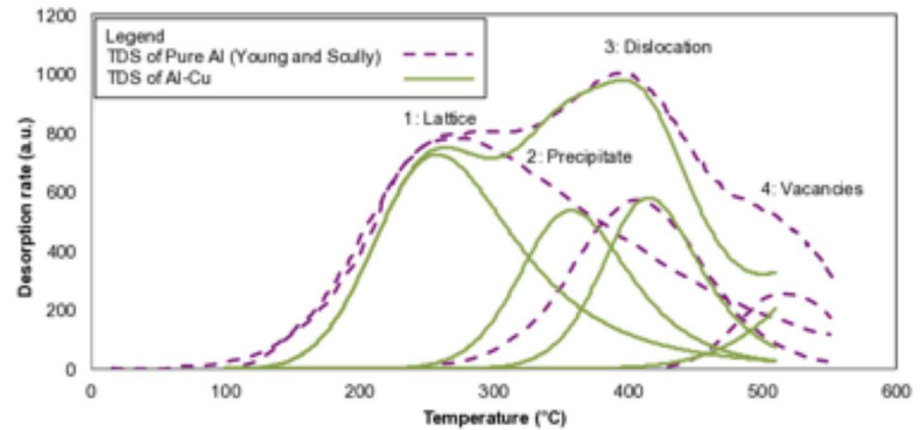
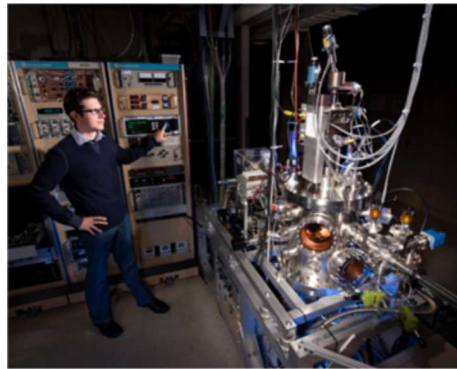


5 Scientific Capabilities for Hydrogen and Materials Research

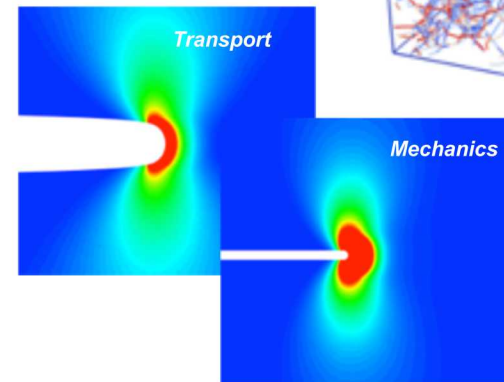
Hydrogen Transport and Trapping Laboratory: measuring diffusion of hydrogen in and through materials

Hydrogen Surface-Interactions Laboratory:

measuring adsorption and desorption of hydrogen at a material's surface



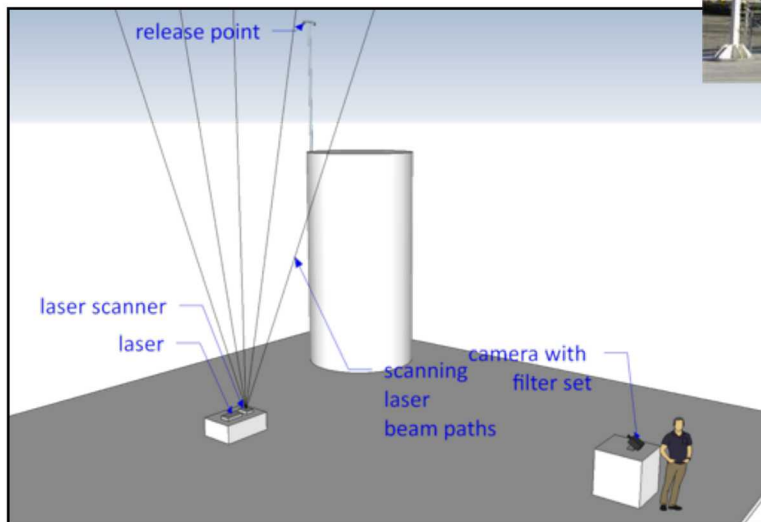
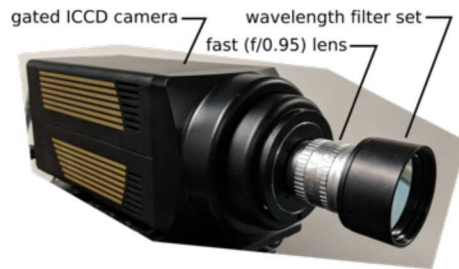
Computational Materials Science: models for studying the mechanisms of hydrogen-assisted behavior, and the evolution of microstructural features (e.g. dislocations, grain boundaries).





6 Diagnostics for Large-Scale Releases

- Standards committees needs validation of our physical models on large-scale gaseous and liquid hydrogen releases.
- Developing diagnostic to measure quantitative characteristics (i.e. temperature, velocity, concentration) of large-scale releases under realistic environmental conditions (e.g. wind).





7 Tunnel Safety for Urban Use of Fuel Cell Electric Vehicles

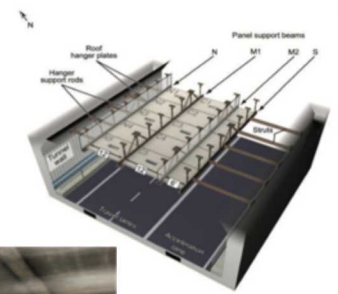
➤ Success of FCEVs requires assessing whether hydrogen vehicle use in tunnels presents challenges that differ from other fuels.

Fire Technology
 © 2019 Springer Science + Business Media, LLC, part of Springer Nature
 Manufactured in The United States
<https://doi.org/10.1007/s10694-019-00910-z>

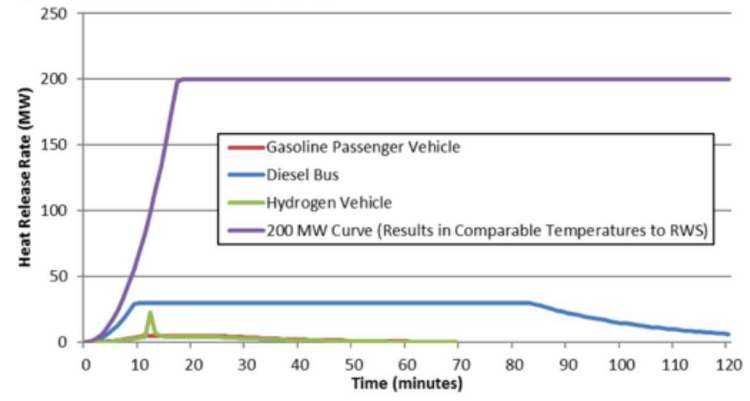
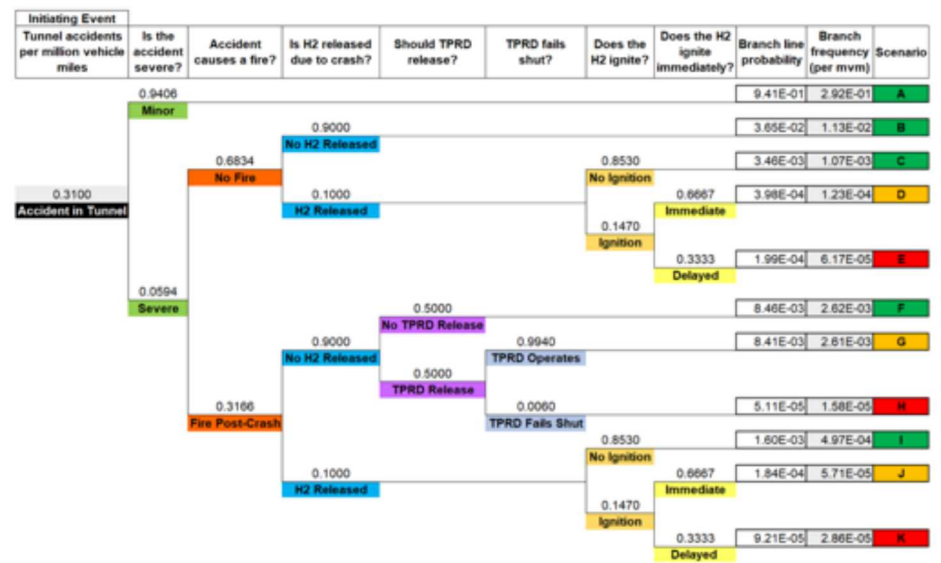
Risk Assessment of Hydrogen Fuel Cell Electric Vehicles in Tunnels

Brian D. Ehrhart, Dusty M. Brooks, Alice B. Muna and Chris B. LaFleur*, Sandia National Laboratories, Albuquerque, NM 871 USA

Received: 23 August 2018/Accepted: 14 Sep 2018



Scenario of concern for Boston AHJ: overturned vehicle with TPRD pointed toward tunnel ceiling



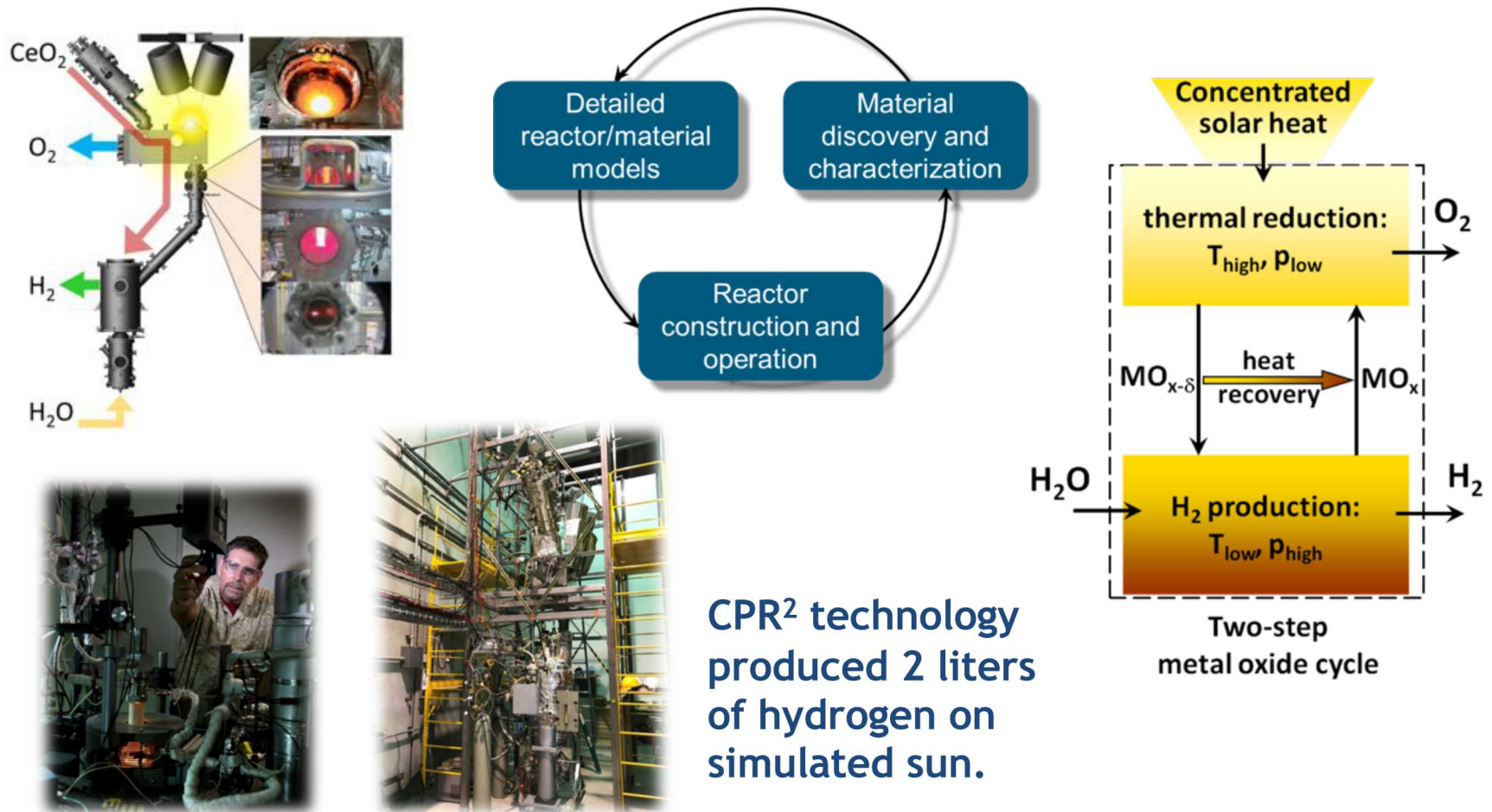
Ehrhart, Brian D., Muna, Alice B., LaFleur, Chris, "Risk Assessment of Hydrogen Fuel Cell Vehicles in Tunnels," Accepted by *Fire Technology*.

FCEV releases less energy than a typical Heavy Goods Vehicle fire



8 Solar Thermo-Chemical Hydrogen Production (STCH)

- Clean and sustainable hydrogen production at large-scale is needed to realize emissions reductions over steam methane reforming (SMR).
- Solar heat can thermo-chemically induce water-splitting in some materials.





9 Early-Market Demonstration of Hydrogen Fuel Cell Technology

Hydrogen Fuel Cell Mobile Light Tower



Hydrogen Fuel Cell Generator for Maritime/Ports



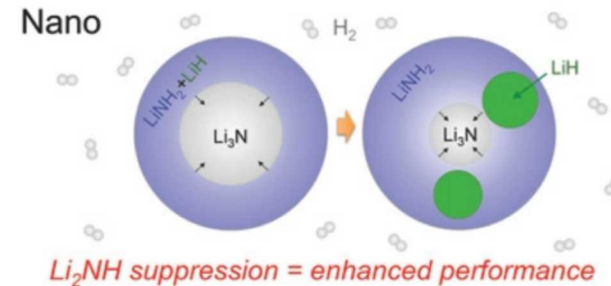
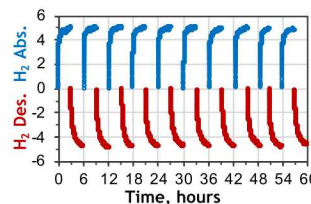
Zero Emission Maritime Vessels



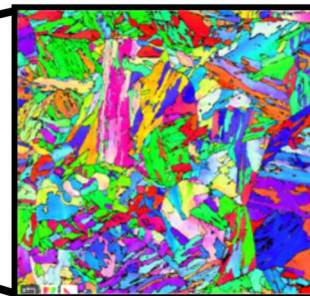
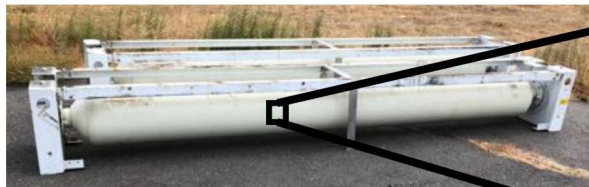


10 Advances in Materials for Hydrogen Fuel Cell Technologies

- Hydrogen **production**, **storage** and **utilization** rely on low-cost, high-performing materials.
- Nano-structuring for hydrogen storage materials



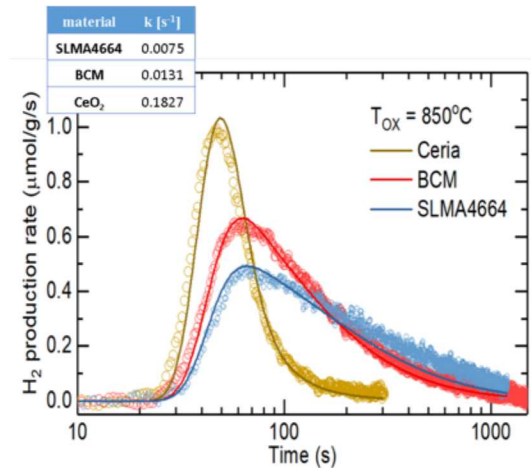
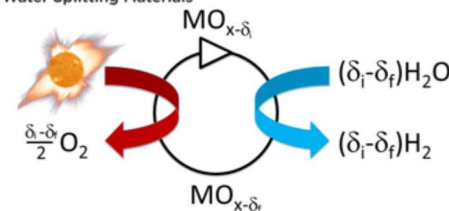
- Alternative microstructures improve fracture toughness of storage tank steels



- New materials for solar thermo-chemical hydrogen production



HydroGEN
Advanced Water Splitting Materials



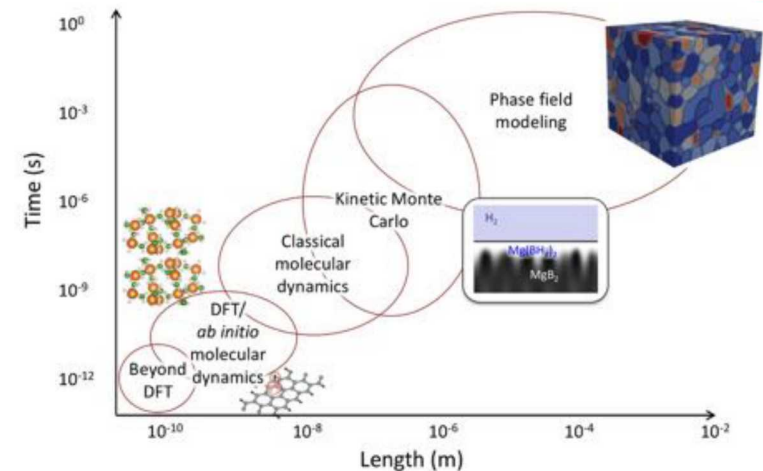
HyMARC objective: accelerate discovery of breakthrough storage materials by providing capabilities and foundational understanding



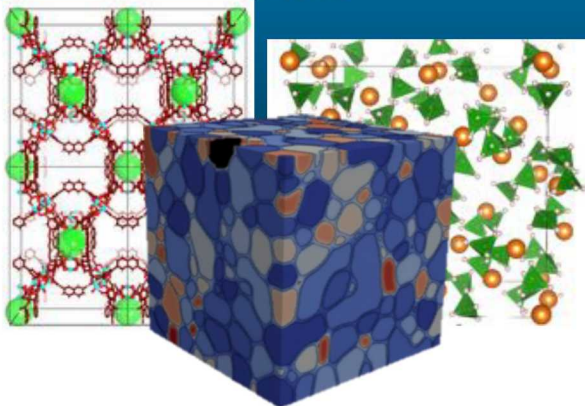
Foundational understanding of phenomena governing thermodynamics and kinetics limiting the development of solid-state hydrogen storage materials

HyMARC will deliver **community tools and capabilities:**

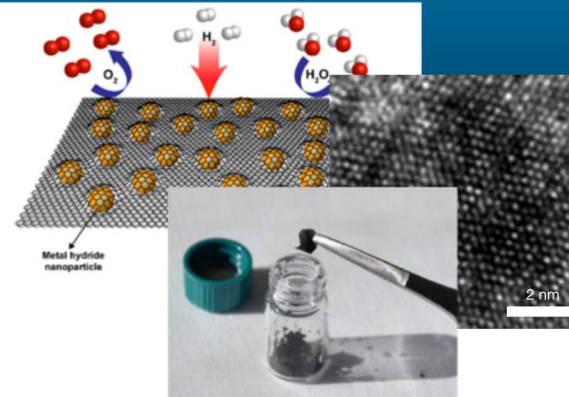
- **Computational models and databases** for high-throughput materials screening
- **New characterization tools and methods** (surface, bulk, soft X-ray, synchrotron)
- **Tailorable synthetic platforms** for probing nanoscale phenomena



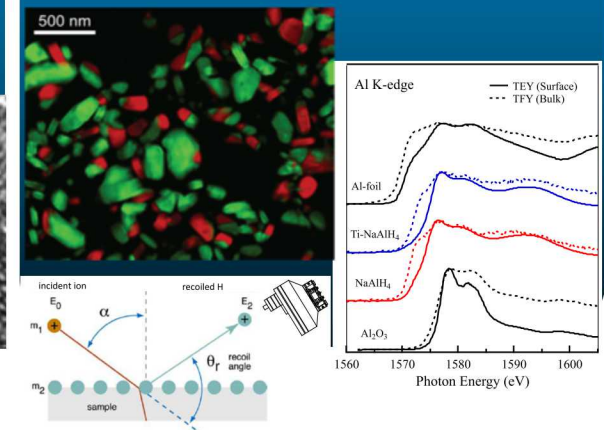
Theory, simulation, & data



Controlled synthesis



In situ characterization

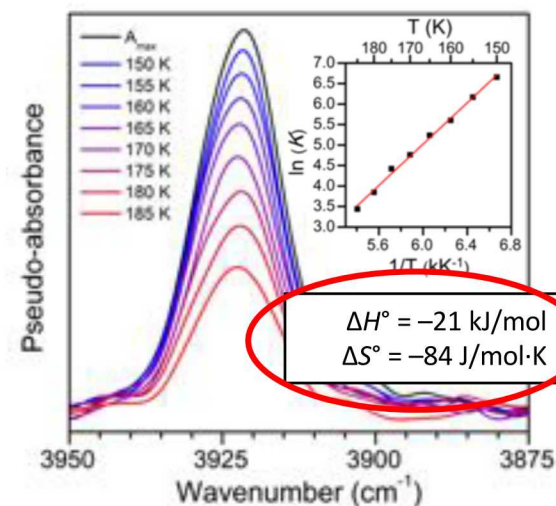
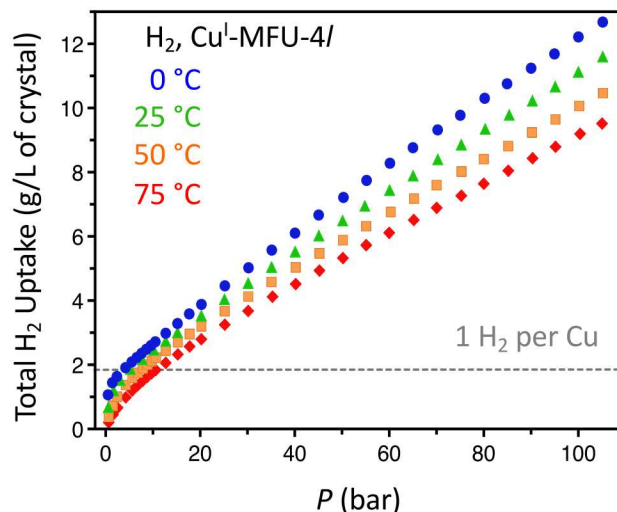
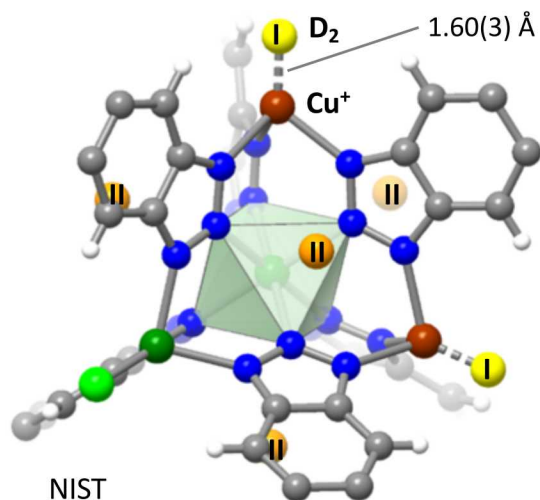


Recent HyMARC research indicates H₂ binding energies can be increased in MOFs



Copper and Vanadium MOFs have heats of adsorption 21 – 33 kJ mol⁻¹

Research of J. R. Long and coworkers

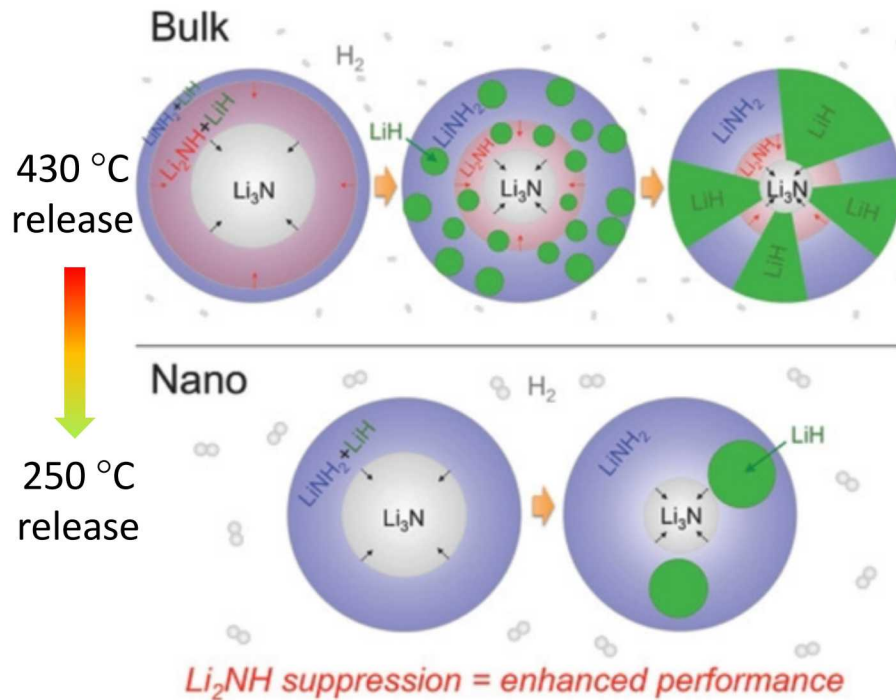


- **In situ powder neutron diffraction:** Extremely short Cu–D₂ distance observed in Cu^I-MFU-4l by neutron powder diffraction. Corroborates strong binding enthalpy and large red-shift of ν(H–H) observed from DRIFTS.
- **High-P adsorption:** Open Cu⁺ sites saturate at relatively low pressures. Volumetric usable capacity for Cu^I-MFU-4l surpasses Ni₂(m-dobdc) at 75 °C.
- **DRIFTS in V₂Cl_{2.8}(btdd):** VTIR confirms high enthalpy of adsorption. Enthalpy–entropy relation distinct from M₂(dobdc) family.

Nanoscale metal hydrides have faster H₂ uptake and release

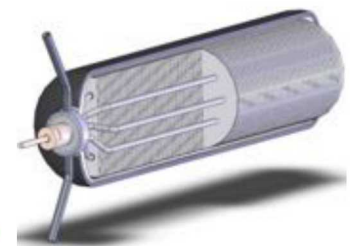
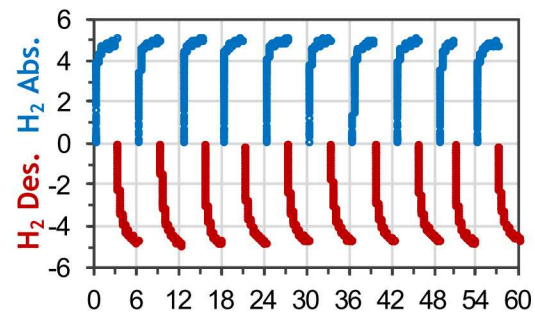


Metal Hydride Finite Element model reveals non-intuitive tradeoffs and benefits of using nanoscale metal hydrides in an operational H₂ storage tank



Design Parameters	bulk-Li ₃ N	KH-6nm-Li ₃ N
Reversible capacity, wt%	8.2	5.4
Thermal cond., W m ⁻¹ K ⁻¹	1.0	9.6
Density of hydride bed, kg m ⁻³	710	760
Total system mass, kg	312	252
Total hydride mass, kg	112	116
Tank outer diameter, m	0.46	0.45
Tank length, m	2.21	2.19
System volume, m ³	0.256	0.227
% 2025 Gravimetric Target	33	40
% 2025 Volumetric Target	55	62

B. C. Wood *Adv. Mater. Interfaces* (2017)
<https://doi.org/10.1002/admi.201600803>

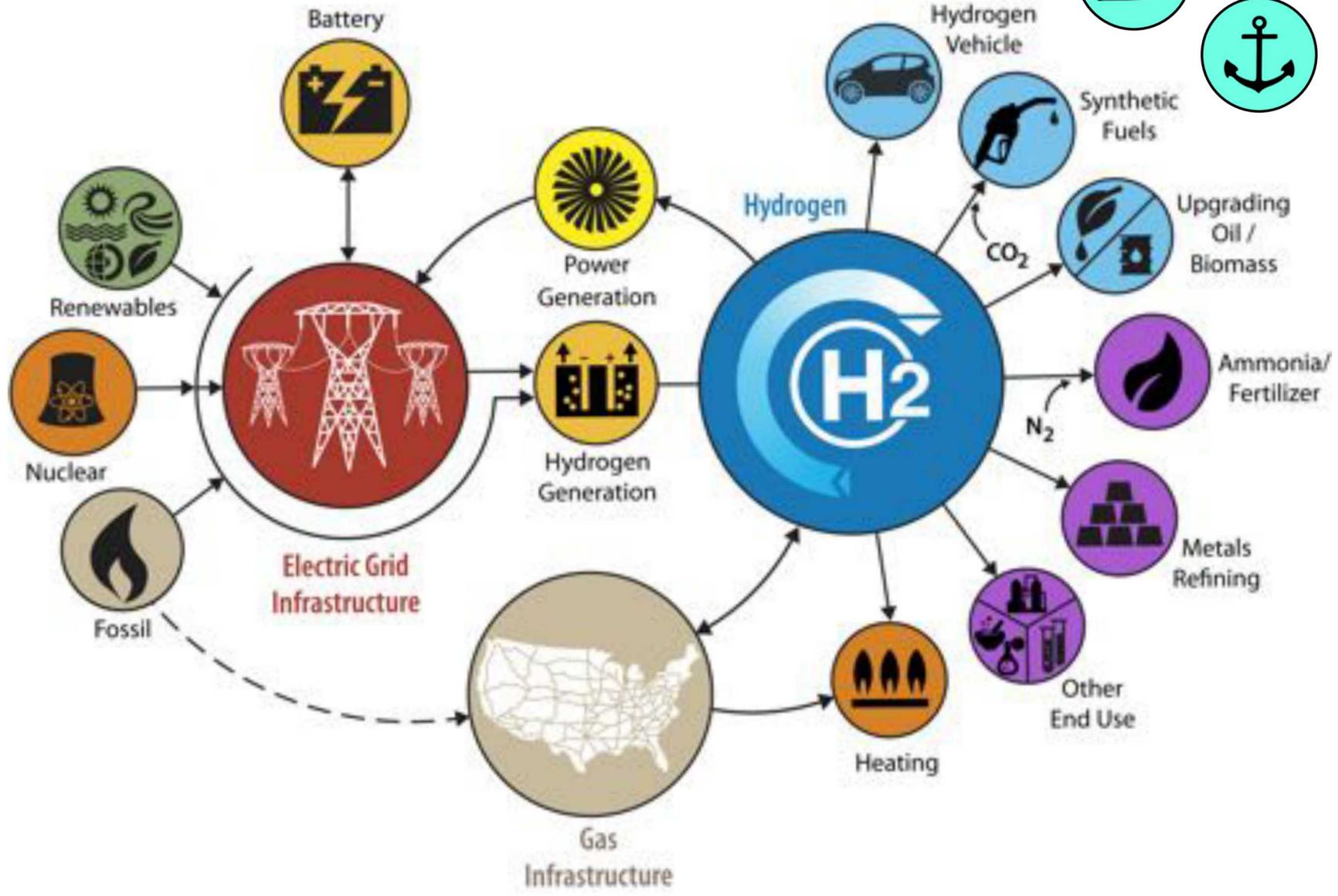


MHFE-SAH tank design

Customers and Partners



H₂@Scale: Hydrogen as an Energy Carrier to Enable...



Rail & Ports



Hydrogen Vehicle

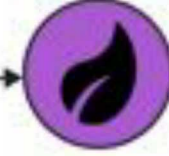


Synthetic Fuels

CO₂



Upgrading Oil / Biomass



Ammonia/ Fertilizer

N₂



Metals Refining



Other End Use



Heating



Gas Infrastructure



Electric Grid Infrastructure



Battery



Power Generation



Hydrogen Generation



Renewables



Nuclear



Fossil



Hydrogen Fueling Infrastructure Research and Station Technology (H2FIRST)



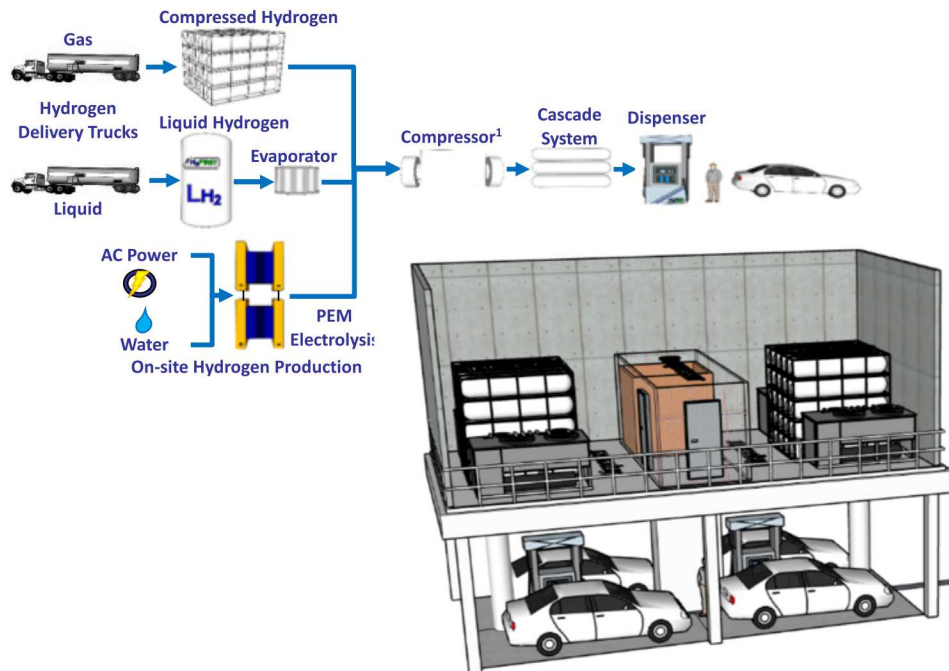
➤ Goal: Address challenges/obstacles to hydrogen fueling station deployment.

H2FIRST Reference Station Design Task
Project Deliverable 2-2
 Joseph Pratt
 Sandia National Laboratories
 Danny Terlip, Chris Ainscough, Jennifer Kurtz
 National Renewable Energy Laboratory
 Amgad Elgowainy
 Argonne National Laboratory

Comparison of conventional vs. modular hydrogen stations, and on-site production vs. delivery
 Ethan S. Hecht, Joseph Pratt
 Sandia National Laboratories

H2FIRST
 Hydrogen Fueling Infrastructure Research and Station Technology
 Hydrogen Reference Stations for Urban Sites
 Brian Ehrhart
 Gabriela Bran-Anleu, Ethan Hecht, Chris LaFleur, Alice Muna,
 Ethan Sena, Carl Rivkin (NREL), Joe Pratt

Building off of previous reference design projects to increase station capacity, reduce footprint and improve station details



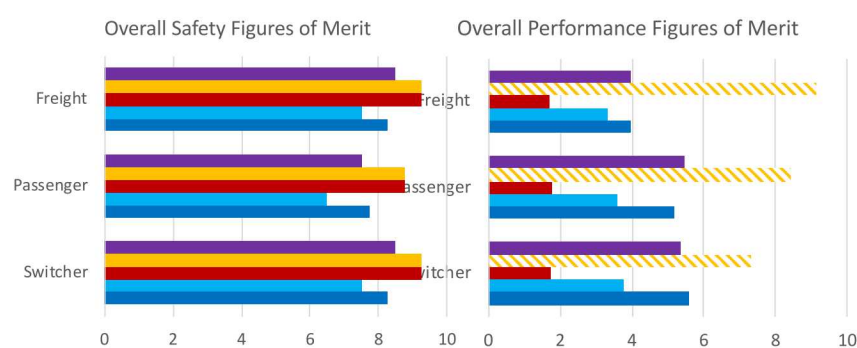
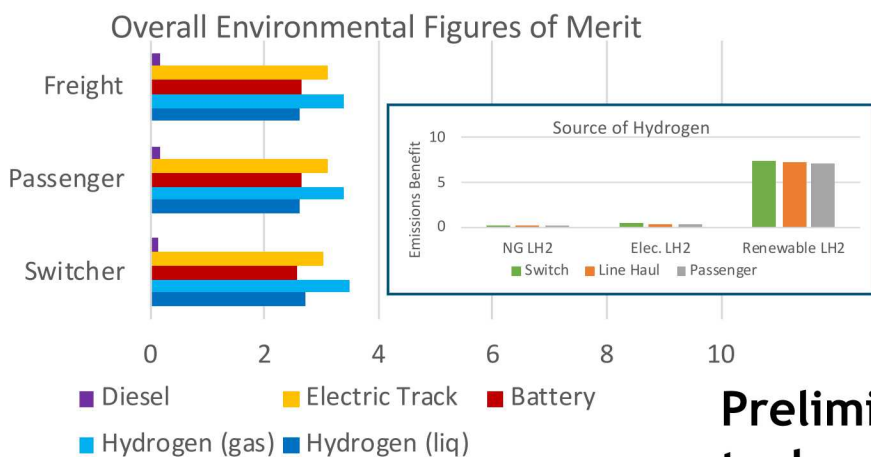
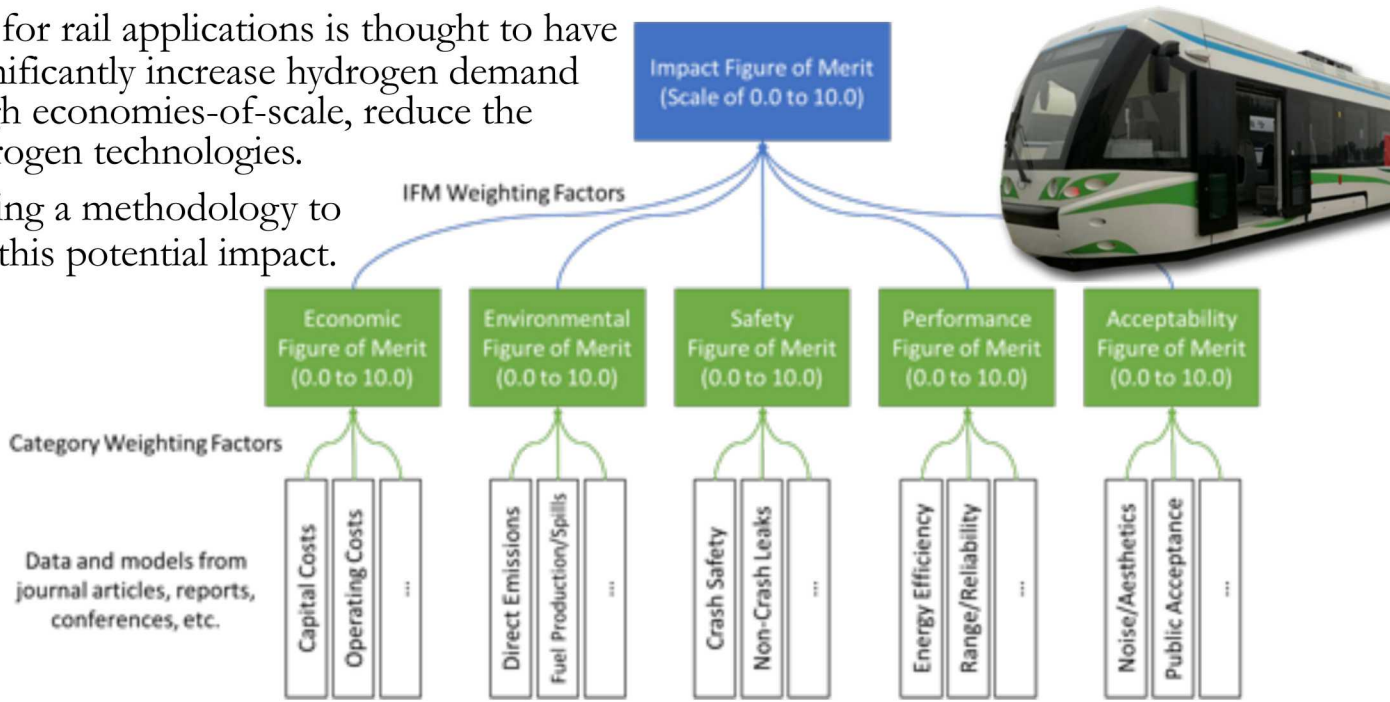
	Total Lot Area (ft ²)	Reduction from Base Case
Base Case Gas	17,640	--
New NFPA Separation Distances	16,240	7.9%
New Delivery Single Truck	14,500	17.8%
New Delivery Double Truck	15,875	10.0%
Gasoline Co-Location	21,000	-19.0% (Increase)
Underground Direct-Bury	15,400	12.7%
Underground Vault	13,720	22.2%
Rooftop Storage	16,000	9.3 %

	Total Lot Area (ft ²)	Reduction from Base Case
Base Case Liquid	21,250	--
New NFPA Separation Distances	18,252	14.1%
New Liquid Delivery	17,400	18.1%
Gasoline Co-Location	22,040	-3.7% (Increase)
Underground Direct-Bury	15,515	27.0%
Rooftop Storage	19,840	6.63 %

19 Assessing the Impact of Hydrogen for Rail (H₂@Rail)



- Use of hydrogen for rail applications is thought to have the potential to significantly increase hydrogen demand and thereby, through economies-of-scale, reduce the overall cost of hydrogen technologies.
- Sandia is developing a methodology to evaluate and verify this potential impact.



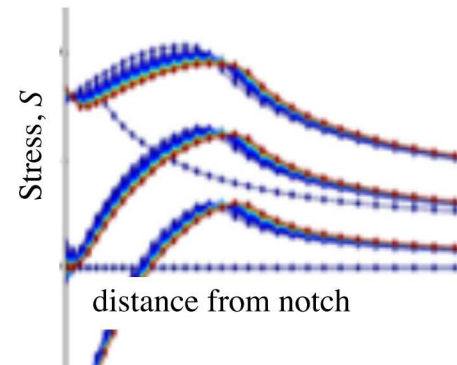
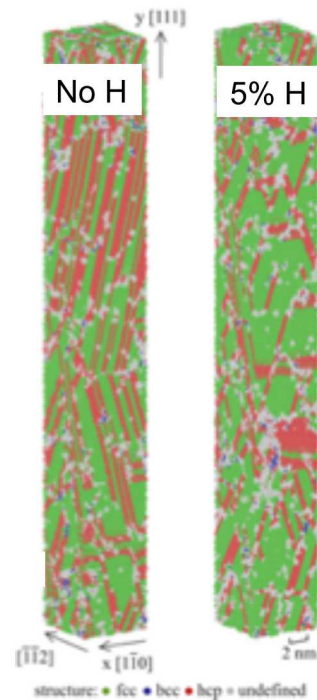
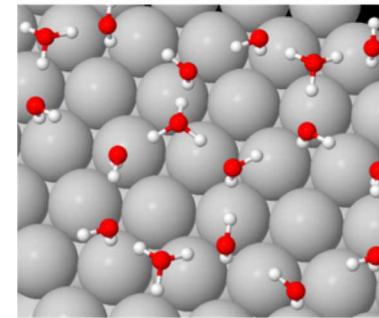
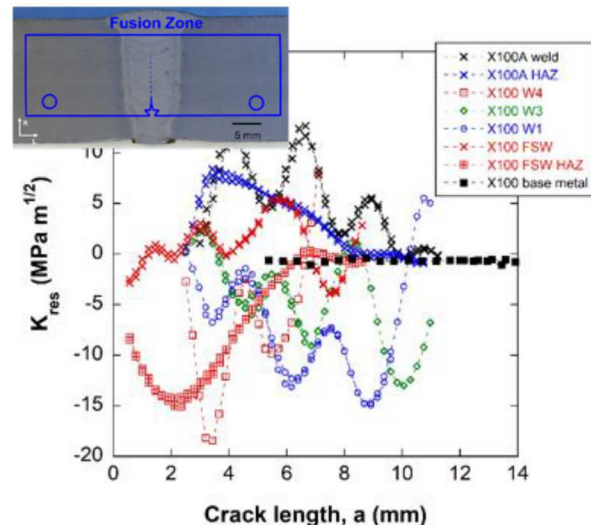
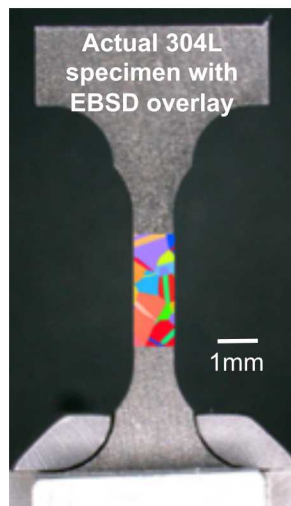
Preliminary results show trade-offs between all technologies → More refinement needed

H-Mat: Science-Based Advancement of Materials for Hydrogen Technologies

- Elucidate the mechanisms of hydrogen-materials interactions to inform **science-based strategies to design the microstructure** of metals with improved resistance to hydrogen degradation

Tasks include:

- High-strength ferritic steel microstructures
- High-strength aluminum alloys
- Transferability of damage and crack nucleation
- Microstructure of austenitic stainless steels
- Materials for cryogenic hydrogen service
- Mechanisms of polymer degradation
- Multiscale modeling of polymer materials
- Hydrogen-resistant polymeric formulations



<https://www.energy.gov/eere/fuelcells/h-mat-hydrogen-materials-consortium>

H-Mat

Hydrogen
Materials
Compatibility
Consortium

Sandia National Laboratories

Pacific
Northwest
NATIONAL
LABORATORY

Argonne
NATIONAL LABORATORY

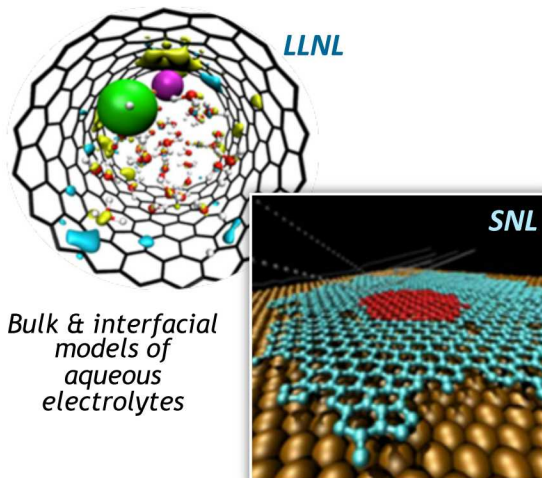
OAK RIDGE
National Laboratory

SRNL



➤ HydroGEN combines more than 80 unique, world-class capabilities (nodes) to foster cross-cutting innovation in materials to advance all emerging water-splitting pathways for hydrogen production.

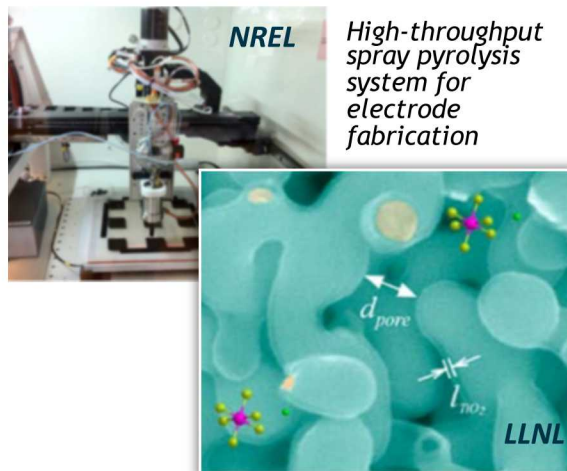
Materials Theory/Computation



Bulk & interfacial models of aqueous electrolytes

LAMMPS classic molecular dynamics modeling relevant to H₂O splitting

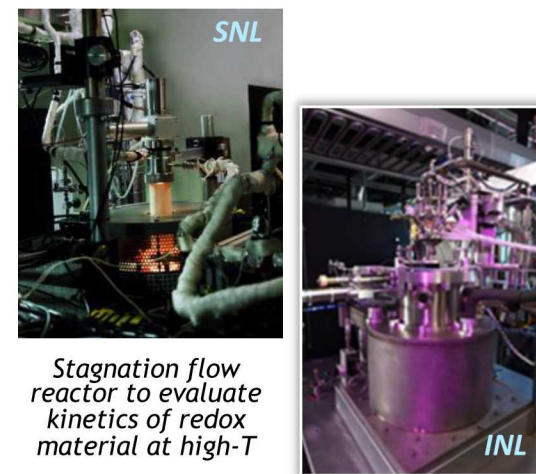
Advanced Materials Synthesis



High-throughput spray pyrolysis system for electrode fabrication

Conformal ultrathin TiO₂ ALD coating on bulk nanoporous gold

Characterization & Analytics



Stagnation flow reactor to evaluate kinetics of redox material at high-T

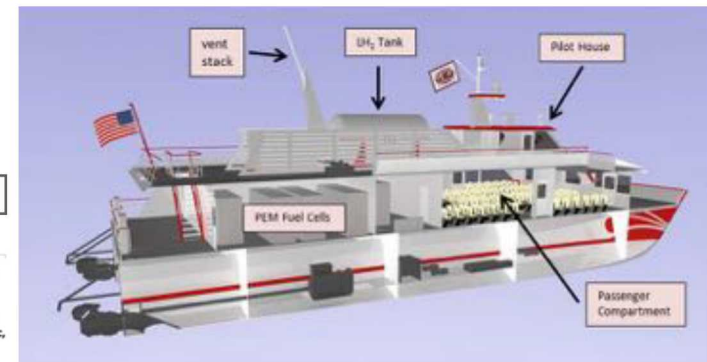
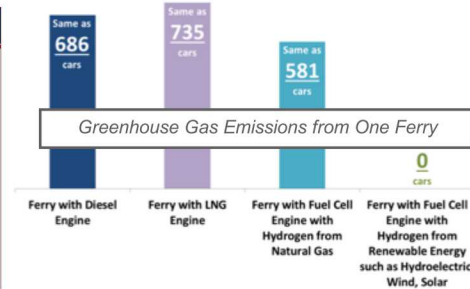
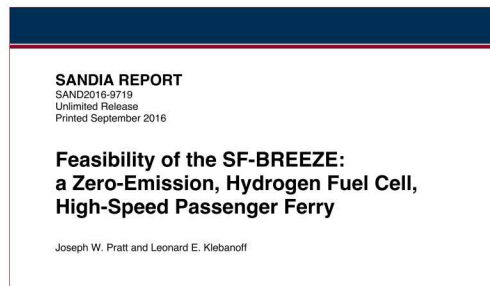
TAP reactor for extracting quantitative kinetic data

<https://www.h2awsm.org/>

Feasibility of Hydrogen Fuel Cell-Powered Maritime Vessels



- Are hydrogen powered marine vessels feasible considering technical, economic and regulatory factors?
- SF-BREEZE high-speed passenger ferry
 - Catamaran design has similar capacity to existing ferry service.
 - Analysis shows modest emissions reduction with SMR-produced hydrogen, zero emission with renewable hydrogen.



- Zero-V coastal research vessel
 - Feasibility was found for a vessel able to perform 14 Scripps science missions with a 2,400 nautical mile range.

