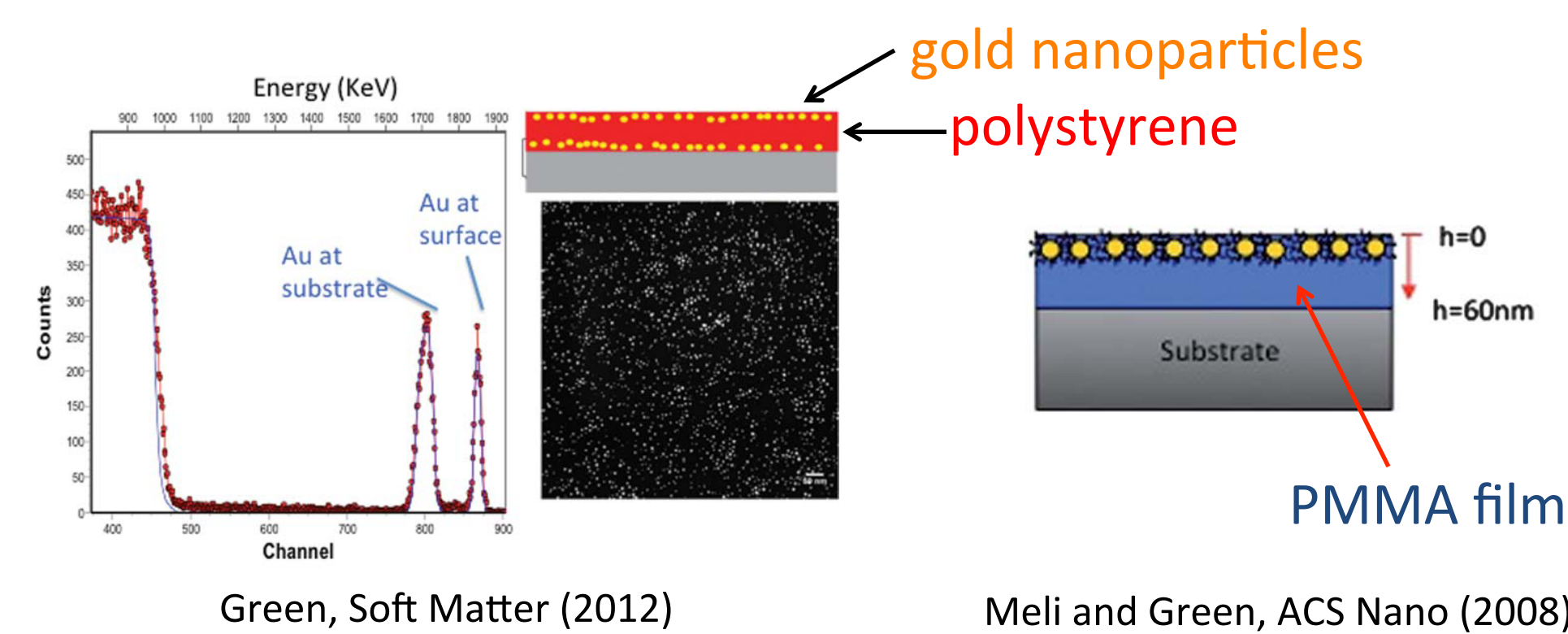
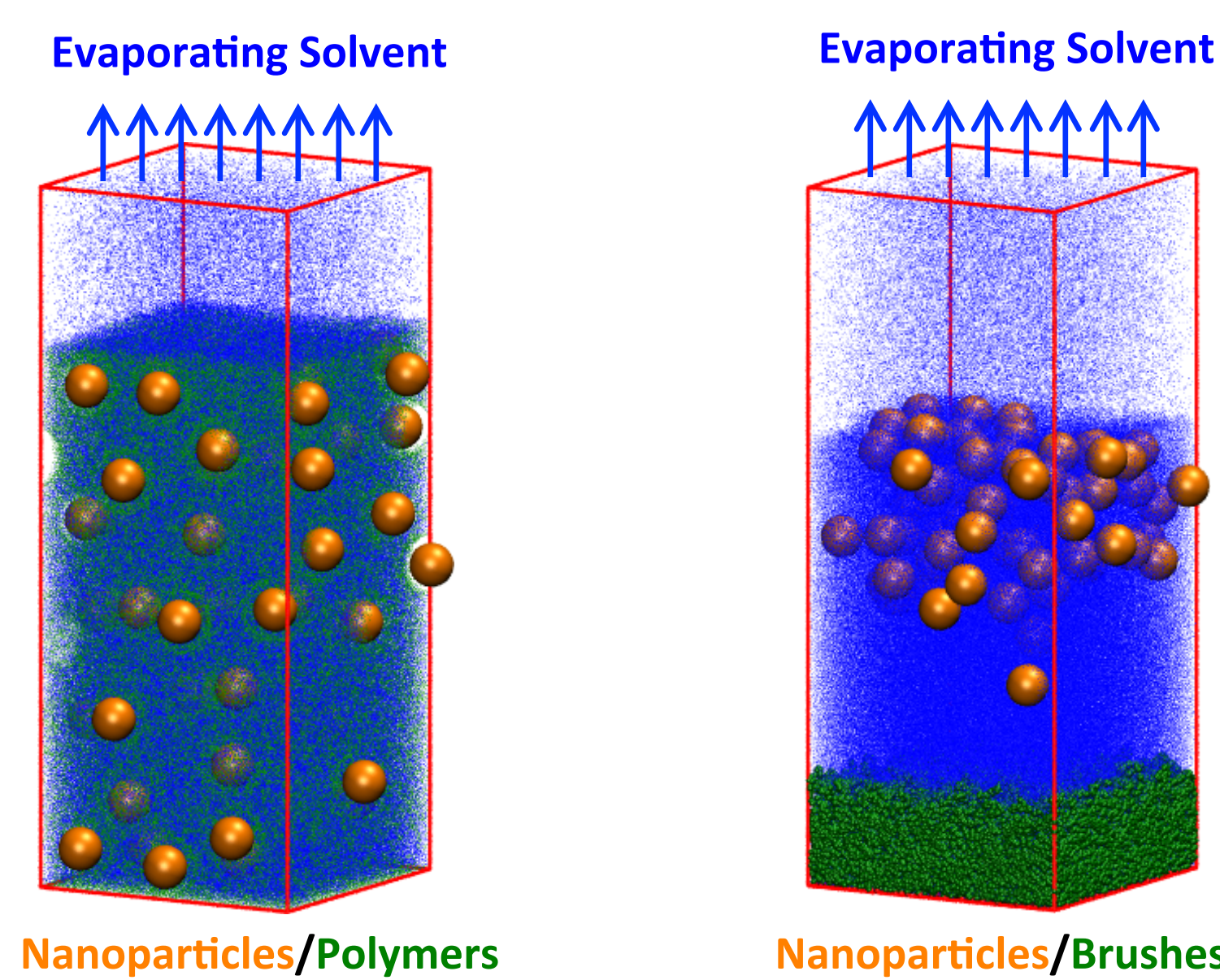


Motivation

- Evaporating solvent is a widely used technique to disperse nanoparticles in a polymer matrix
- How to control the distribution of nanoparticles?
 - Interactions (nanoparticle/nanoparticle, nanoparticle/solvent, nanoparticle/polymer)
 - Particle size/shape/coating/concentration
 - State of polymers (solution/brush)



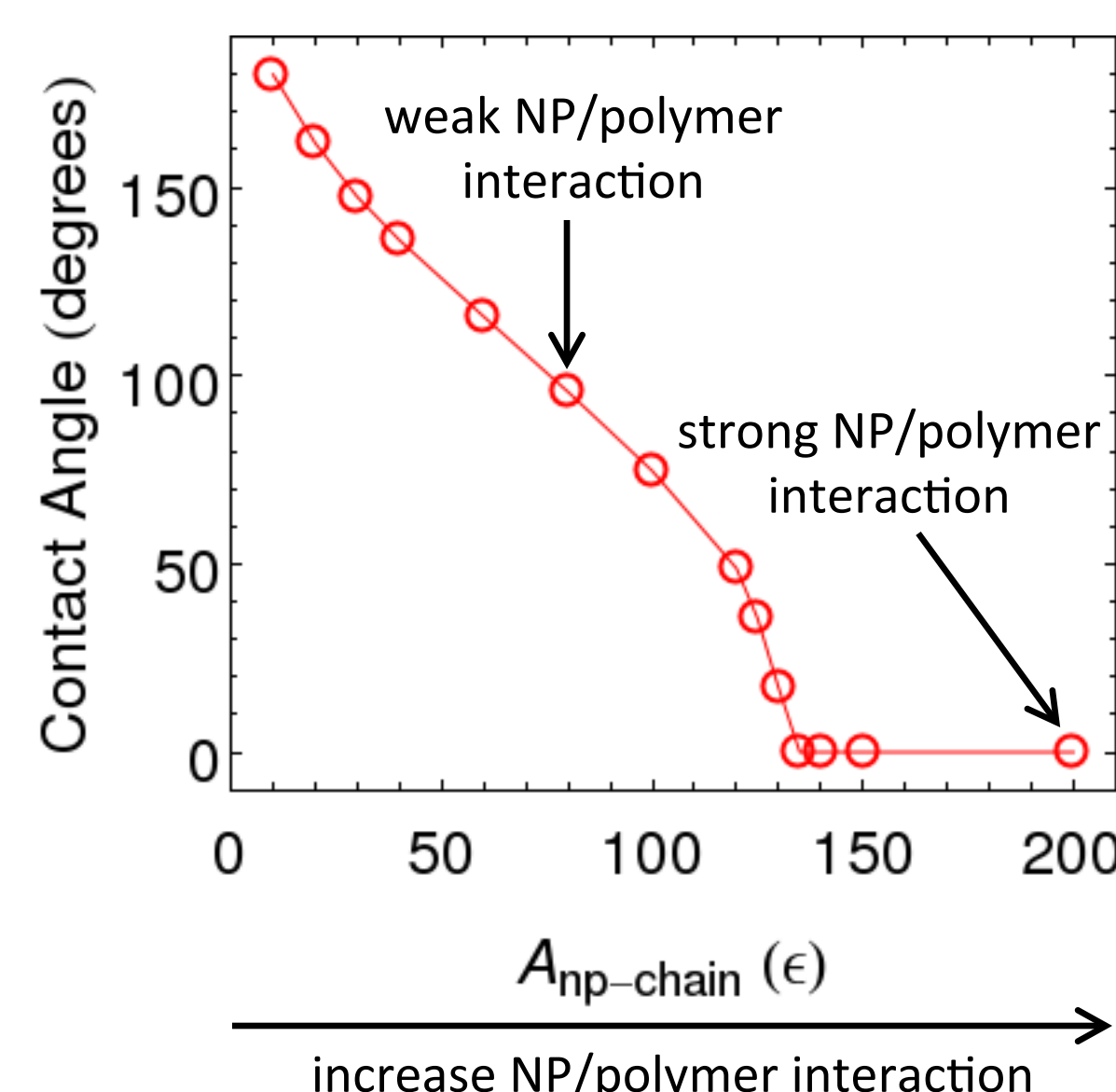
Method – Large Scale Molecular Dynamics



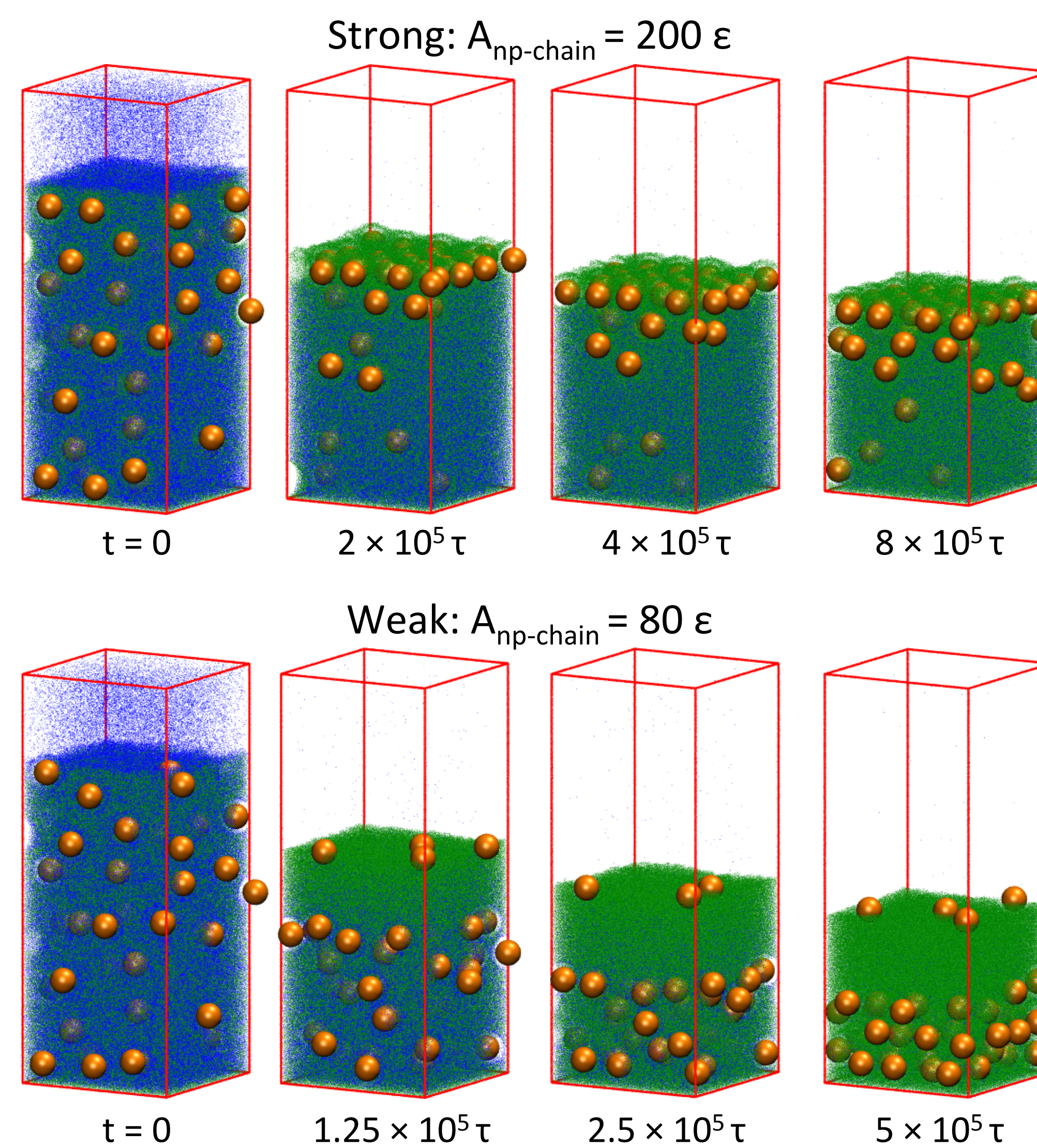
- Mixtures with dispersed polymers: Lennard-Jones solvent (~ 3 million atoms) + 100-bead polymer chains (~ 3 million beads) + nanoparticles (200, $D=20\sigma$)
- Mixtures with grafted polymers: LJ solvent (~ 1 million atoms) + brushes (100-bead chains, grafting density $0.01 \sim 0.24\sigma^{-2}$ over $100\sigma \times 100\sigma$ plane) + nanoparticles ($20 \sim 100$, $D=10\sigma$)
- Integrated LJ potential for interactions involving nanoparticles and bead-spring model for polymer chains
- Strong nanoparticle/solvent interaction \rightarrow solvated nanoparticles
- Vary nanoparticle/polymer interaction
- Remove vapor of solvent \rightarrow evaporation
- Monitor nanoparticle distribution in polymer/brush matrix

Contact Angle

- Strong (weak) nanoparticle/polymer interaction \rightarrow small (large) contact angle of nanoparticles on pure polymer melts

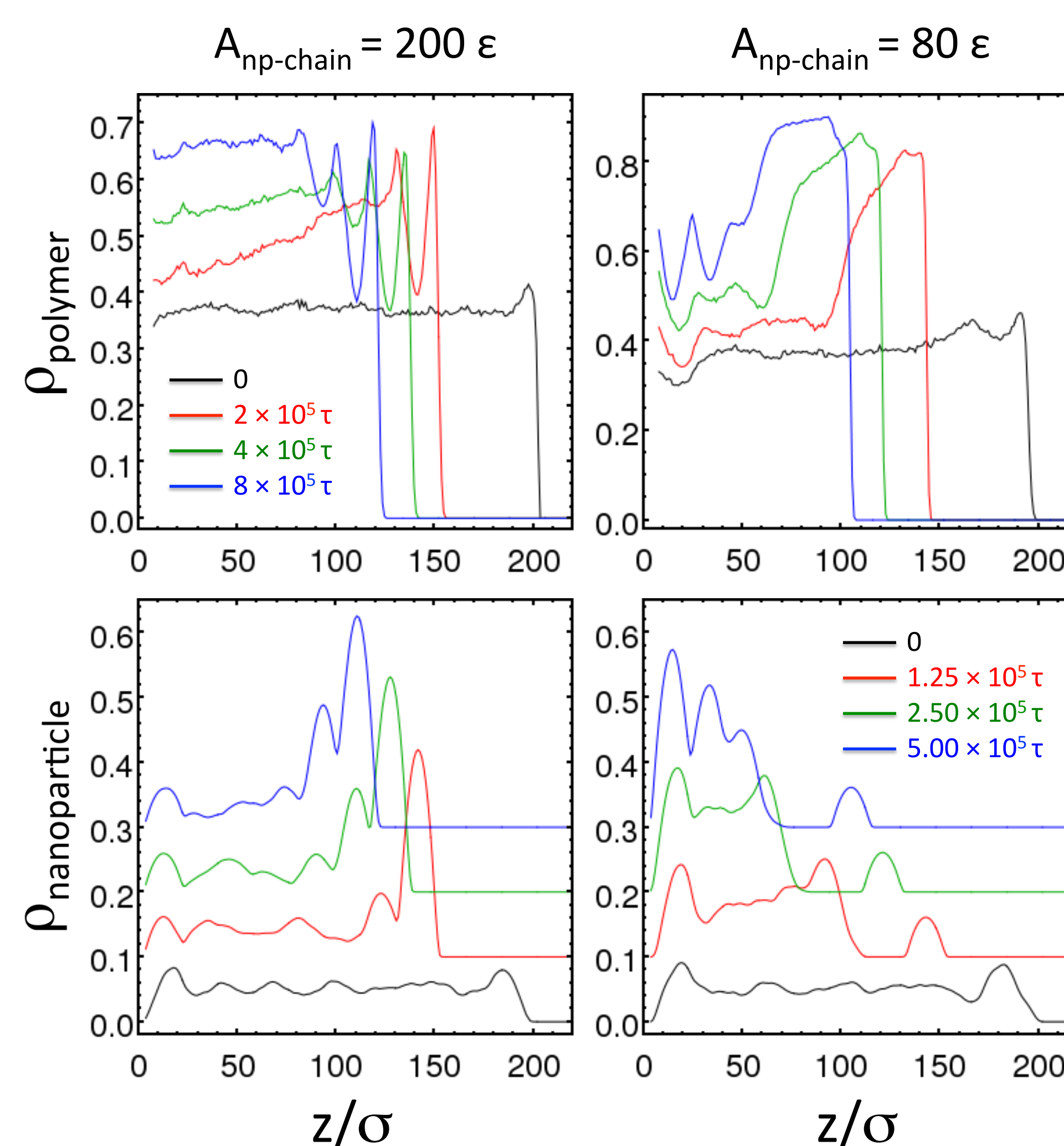


Mixtures with Dispersed Polymers: Strong vs. Weak Nanoparticle/Polymer Interaction



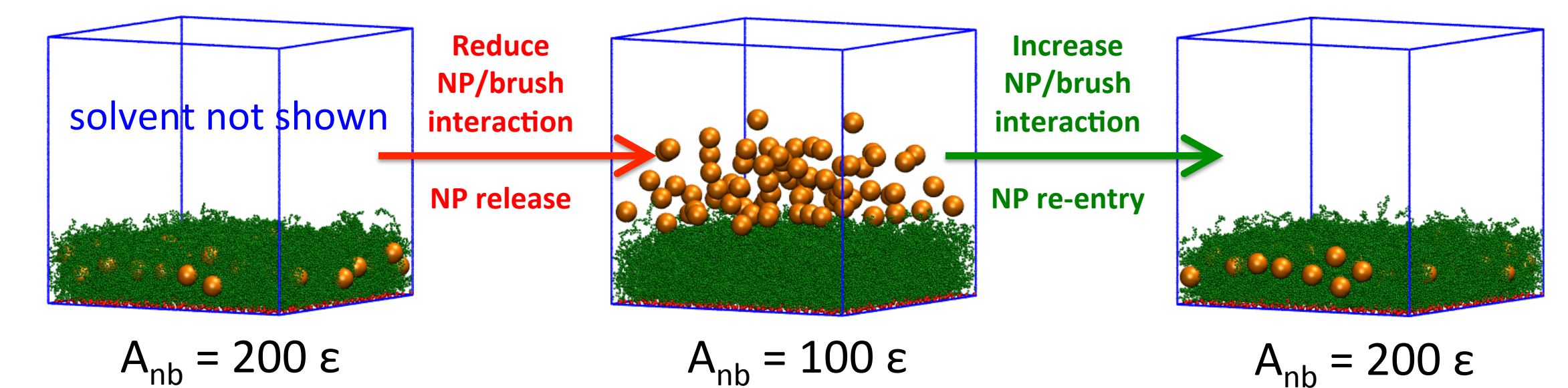
- During evaporation of solvent, density of polymer chains is enhanced at liquid/vapor interface in both cases \rightarrow a skin layer of polymers forms
- Nanoparticles are accumulated (depleted) in this skin layer of polymers for strong (weak) nanoparticle/polymer interaction
- Strong interaction: nanoparticles form layers near interface \rightarrow alternating density peaks of nanoparticles and polymers (see density plots below)
- Weak interaction: a few nanoparticles penetrate skin layer of polymers and straddle interface; most nanoparticles are trapped below skin layer and disorderly distributed

Evolution of Polymer/Nanoparticle Distribution



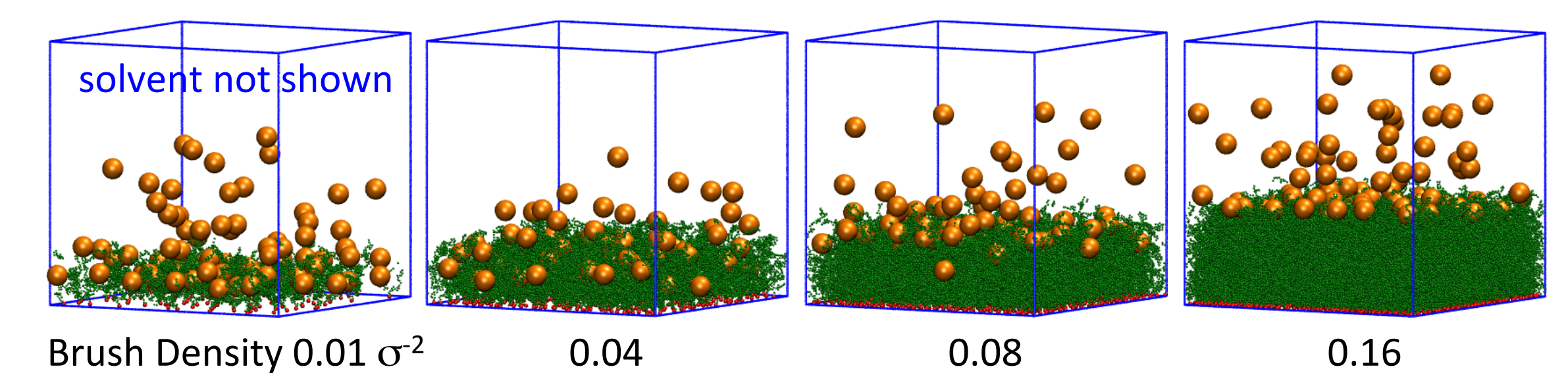
Equilibrate Nanoparticle/Brush/Solvent Mixtures

- Brush density $0.08\sigma^{-2}$
- Nanoparticle (2D) density $0.75\sigma^{-2}$
- Reversible nanoparticle capture/release \rightarrow equilibrated mixtures

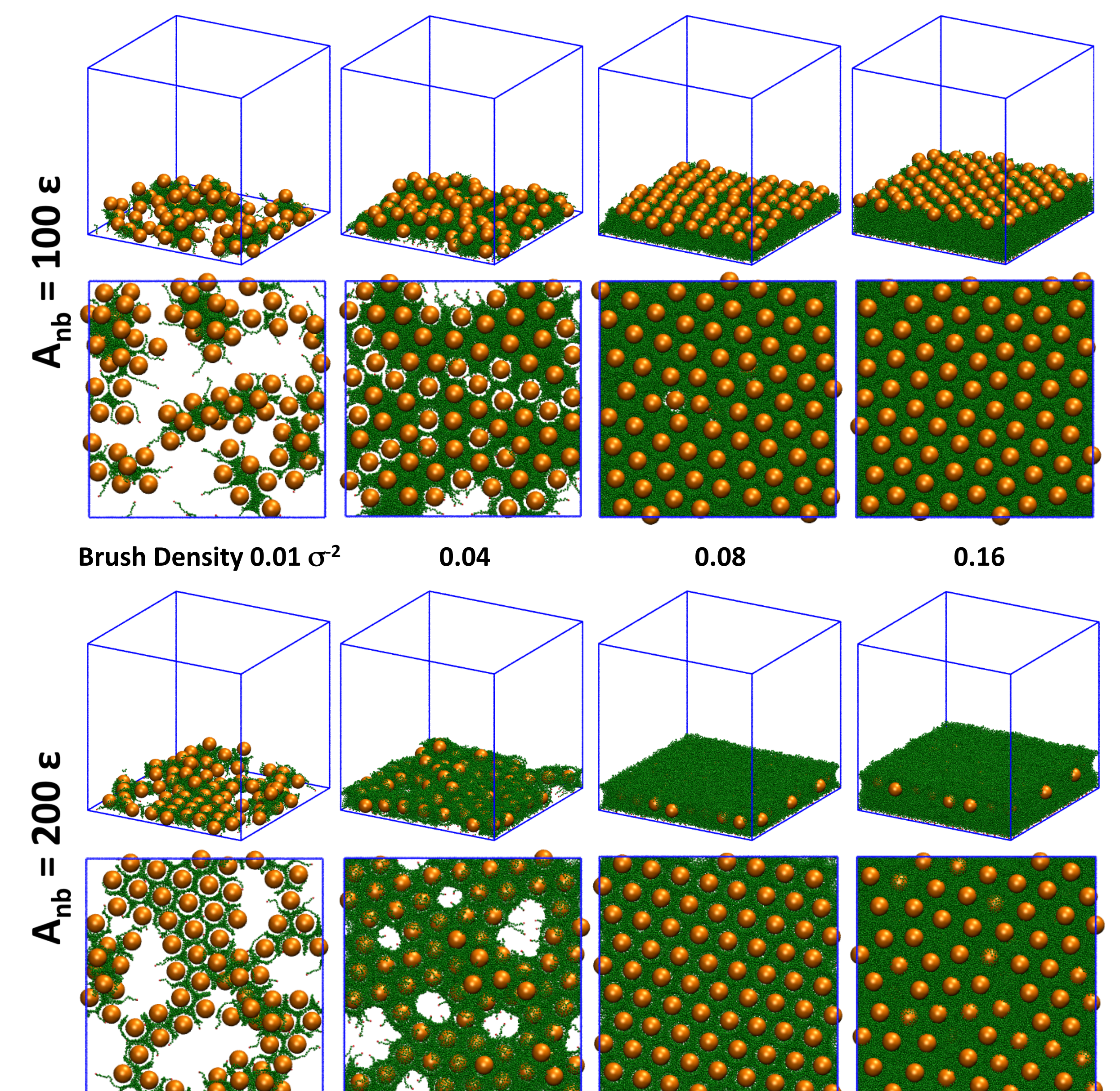


Nanoparticle Distribution with Varying Grafting Density

- Nanoparticle (2D) density $0.75\sigma^{-2}$
- $A_{nb} = 120\epsilon \rightarrow$ nanoparticles straddle surface of pure polymer melt
- Better nanoparticle adsorption at intermediate brush densities



Nanoparticle Organization after Evaporation of Solvent



- Better nanoparticle arrays are formed at:
 - high brush density with weak nanoparticle/brush interaction \rightarrow nanoparticles form one layer straddling brush surface
 - finely tuned intermediate brush density with strong nanoparticle/brush interaction \rightarrow brush accommodates just one layer of nanoparticles