

# **Technoeconomics for Sunshine to Petrol:**

## **Approach, Results, Challenges and Lessons Learned**

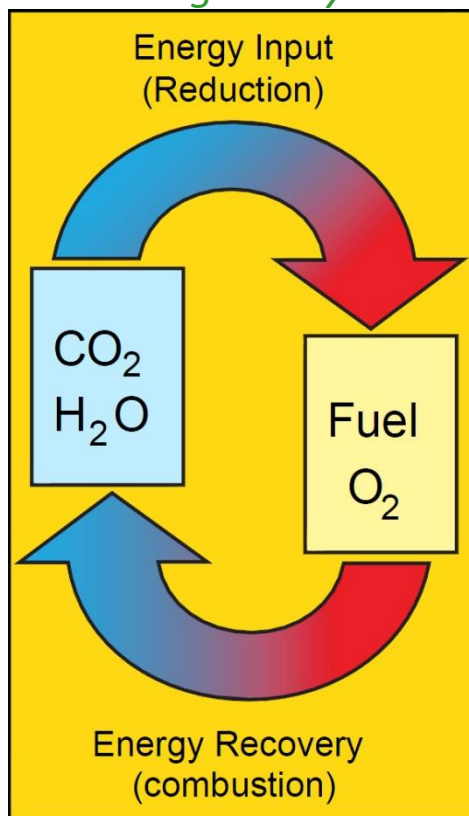
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Albuquerque, NM 87123

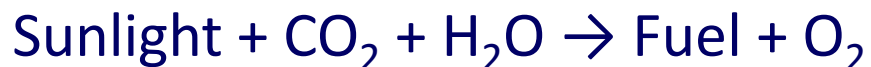
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## Closing the Cycle



For now and for transportation fuels, liquid hydrocarbons are the “Gold Standard”

Vision: Directly apply a solar thermal energy source to effectively reverse combustion and “energize” CO<sub>2</sub> and H<sub>2</sub>O into hydrocarbon form in a process analogous to the one that produces bio- and fossil fuels but at higher solar to fuel efficiency.



# Big Picture: A Question of Impact



- If we are wildly successful, does it matter?
- What are limits and limiting factors?
- Can we draw conclusions, set targets, etc.?
- Factors to consider
  - Problem definition – what is the scale?
  - Resources: CO<sub>2</sub>, H<sub>2</sub>O, Solar, Land, Materials
  - Economics
  - Timeframe



# Meeting 100% of transportation demand with solar fuel is plausible

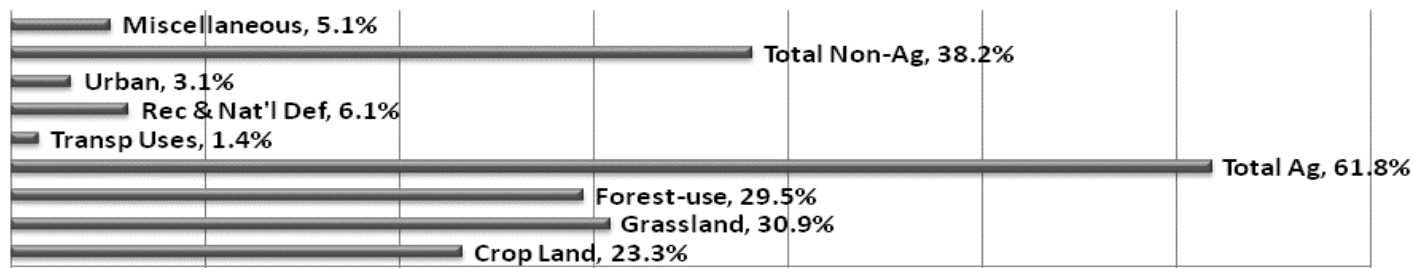
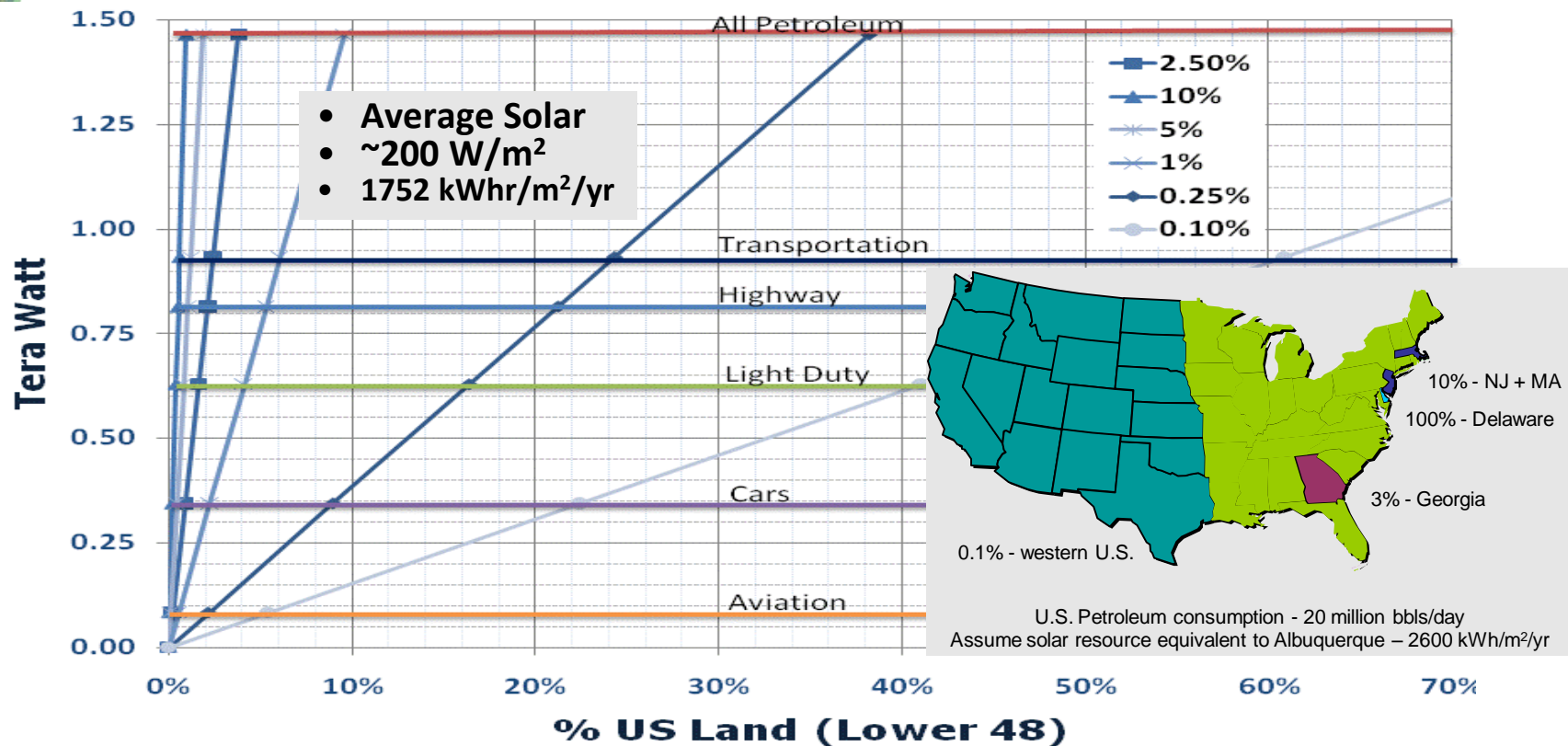


- High solar to fuel efficiency (>10%) is absolutely required.
  - Cost
  - Scale (land, materials of construction)
- Water, CO<sub>2</sub> are not limiting –
  - Water consumption/cost relatively low
  - High impact opportunity for CO<sub>2</sub>.
- Consistent with other human activities occurring over multiple decades.

Ellen B. Stechel and James E. Miller "Re-energizing CO<sub>2</sub> to fuels with the sun: Issues of efficiency, scale, and economics" in press J. CO<sub>2</sub> Util. (2013), <http://dx.doi.org/10.1016/j.jcou.2013.03.008>.



# Efficiency defines collector area

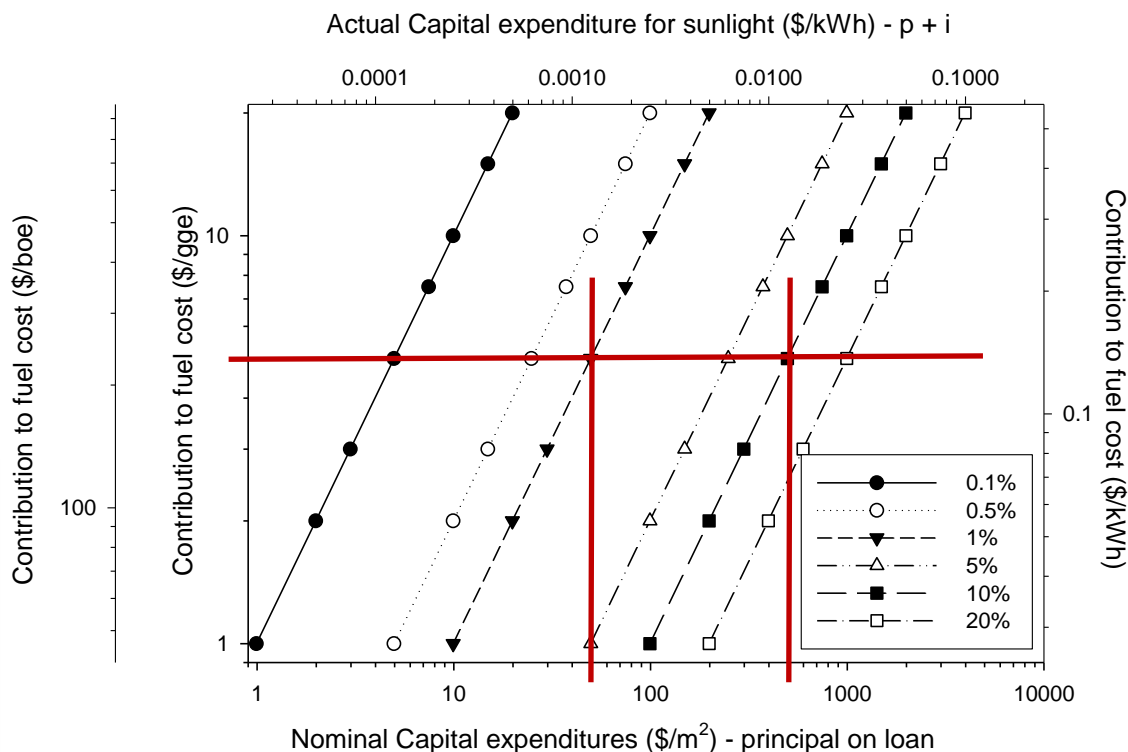


# Efficiency drives Costs



For capital cost < \$5/gge, expenditure can be no more than:

**\$50/m<sup>2</sup> for 1% Eff. (solar to fuel)**  
**\$500/m<sup>2</sup> for 10% Eff. (solar to fuel)**



## Benchmarks:

\$1/m<sup>2</sup> ≈ \$4,000/acre  
 6" Concrete slab ≈ \$20/m<sup>2</sup>  
 PE Greenhouse ≈ \$80/m<sup>2</sup>  
 Parabolic dish ~ \$300/m<sup>2</sup>

E.B. Stechel and J.E. Miller "Re-energizing CO<sub>2</sub> to fuels with the sun: Issues of efficiency, scale, and economics" Journal of CO<sub>2</sub> Utilization  
<http://dx.doi.org/10.1016/j.jcou.2013.03.008>.

Assumptions: gge = 36 kWh, solar resource = 2365 kWh/m<sup>2</sup>/yr, favorable financing (5% interest, 30 years)

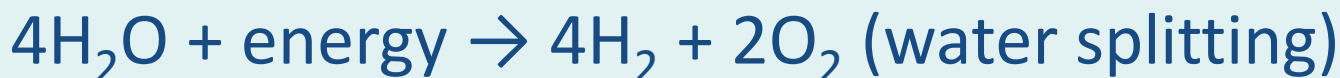
There is such a thing as too low an efficiency,  
even for "inexpensive materials"

# *The Technology*

Capitalize on decades of Synfuel technology, e.g.



**Build on these two reactions:**



WS and CDS are linked by the Water Gas Shift reaction



Only necessary to conduct one of the two to accomplish the chemistry

# Establish Metrics – Current Technology



Technology	Solar to electricity <sup>b</sup>	Electricity to hydrogen <sup>c</sup>	Solar to hydrogen
Photovoltaic/AE <sup>a</sup>	15%	73%	11%
Molten Salt Tower/AE	18%	73%	13%
Dish Stirling/AE	24%	73%	18%

<sup>a</sup>Alkaline electrolysis

<sup>b</sup>These are estimated annual average efficiency values

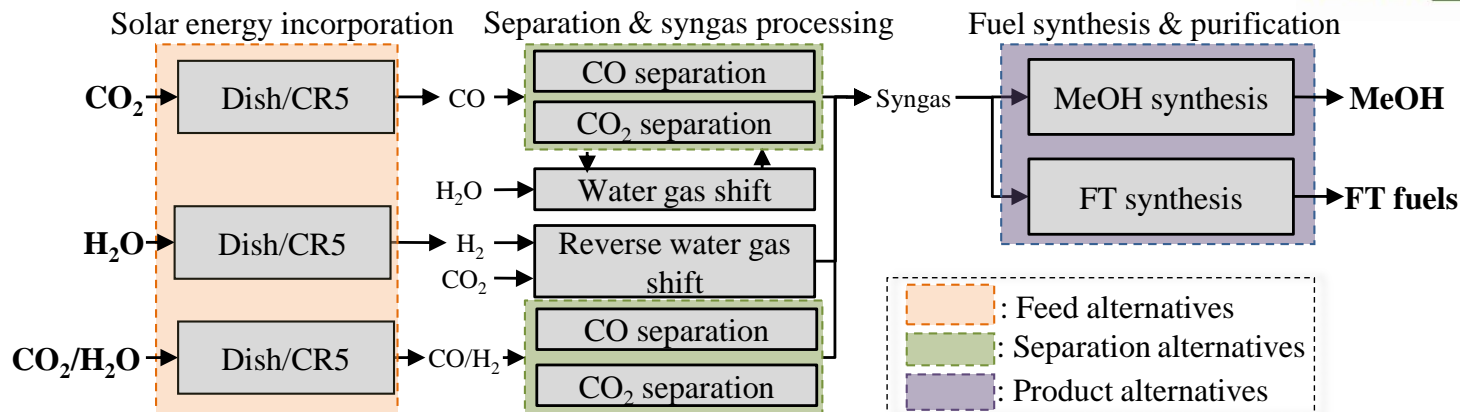
<sup>c</sup>This performance value is at the high end for an alkaline electrolyzer, is dependent on system operation as well as size, and is based on the higher heating value (HHV) of the hydrogen product.

*Literature and our own early work suggests 40-50% H<sub>2</sub> to Fuel possible*

- (1) *Electrical to Fuel; Mignard and Pritchard Trans IChemE, Part A, September 2006.*
- (2) *H<sub>2</sub> + utilities to Methanol; Henao, Maravelias, Miller and Kemp, presented @ FOCAPD 2009.*



# Multiple Pathways Considered Assume - 20% Solar to CO/H<sub>2</sub>



	Pathway	Energizing Reaction	CO <sub>2</sub> /CO separation	Fuel
A	CO <sub>2</sub>	CO <sub>2</sub>	CO <sub>2</sub>	MeOH
B	H <sub>2</sub> O	H <sub>2</sub> O	-	MeOH
C	Mixed	CO <sub>2</sub> / H <sub>2</sub> O	CO <sub>2</sub>	MeOH
D	Mixed FT	CO <sub>2</sub> / H <sub>2</sub> O	CO <sub>2</sub>	FT fuels
E	CO <sub>2</sub> –CO Sep.	CO <sub>2</sub>	CO	MeOH
F	Mixed CO Sep.	CO <sub>2</sub> / H <sub>2</sub> O	CO	MeOH

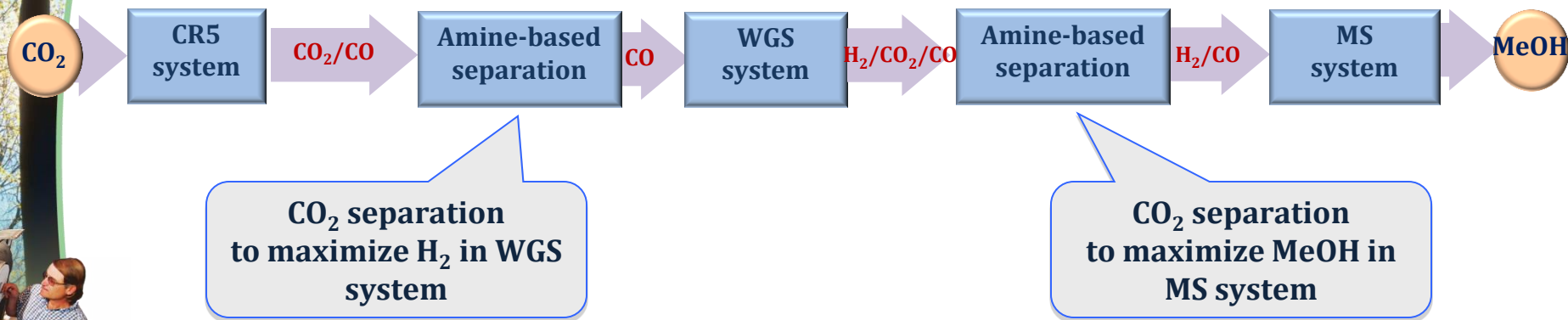
University of Wisconsin – Dr. Christos Maravelias' group / / Sandia

# Process Synthesis



## CO<sub>2</sub> pathway for MeOH production

### Basic configuration



Slides 9-27:

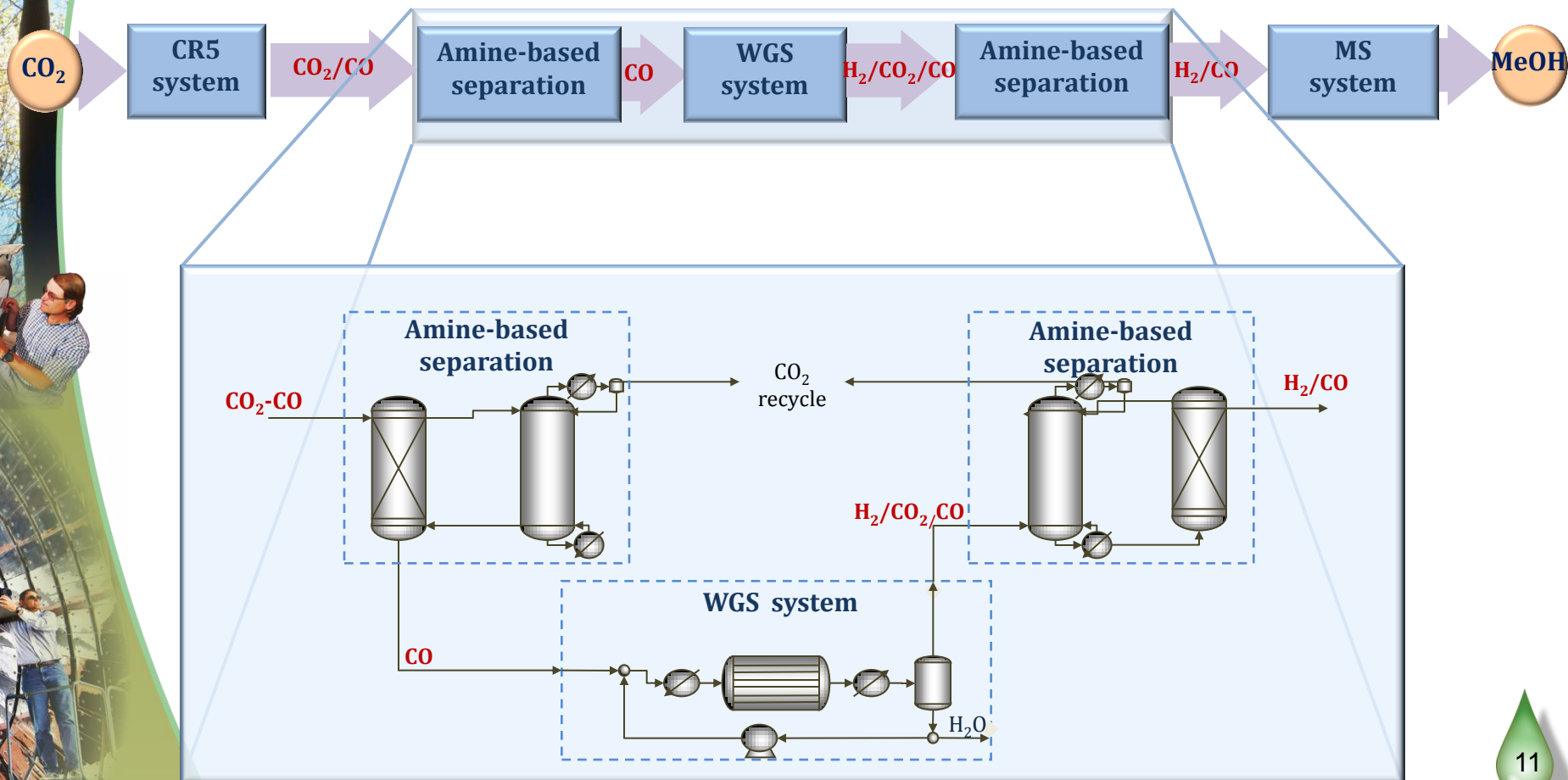
Jiyong Kim, Carlos A. Henao, Terry A. Johnson, Daniel E. Dedrick, James E. Miller, Ellen B. Stechel and Christos T. Maravelias "Methanol production from CO<sub>2</sub> using solar-thermal energy: process development and techno-economic analysis" *Energy Environ. Sci.*, 2011, 4, 3122.

Jiyong Kim, Terry A. Johnson, James E. Miller, Ellen B. Stechel and Christos T. Maravelias "Fuel production from CO<sub>2</sub> using solar-thermal energy: system level analysis" *Energy Environ. Sci.*, 2012, 5, 8417.

# Process Synthesis

## CO<sub>2</sub> pathway for MeOH production

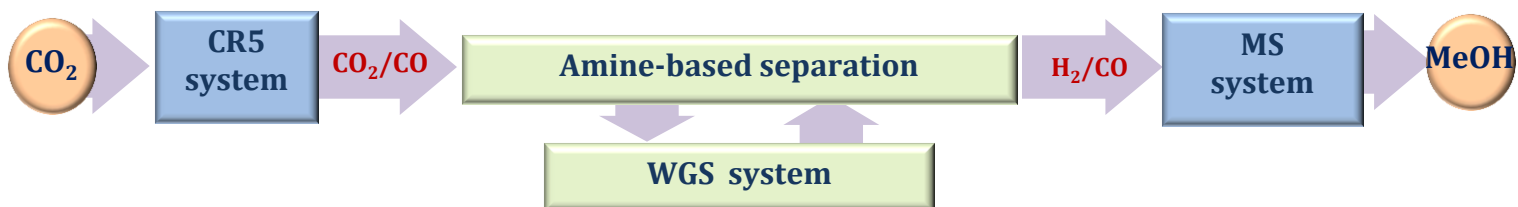
### Basic configuration



# Process Synthesis

CO<sub>2</sub> pathway for MeOH production

Improved amine-based separation network

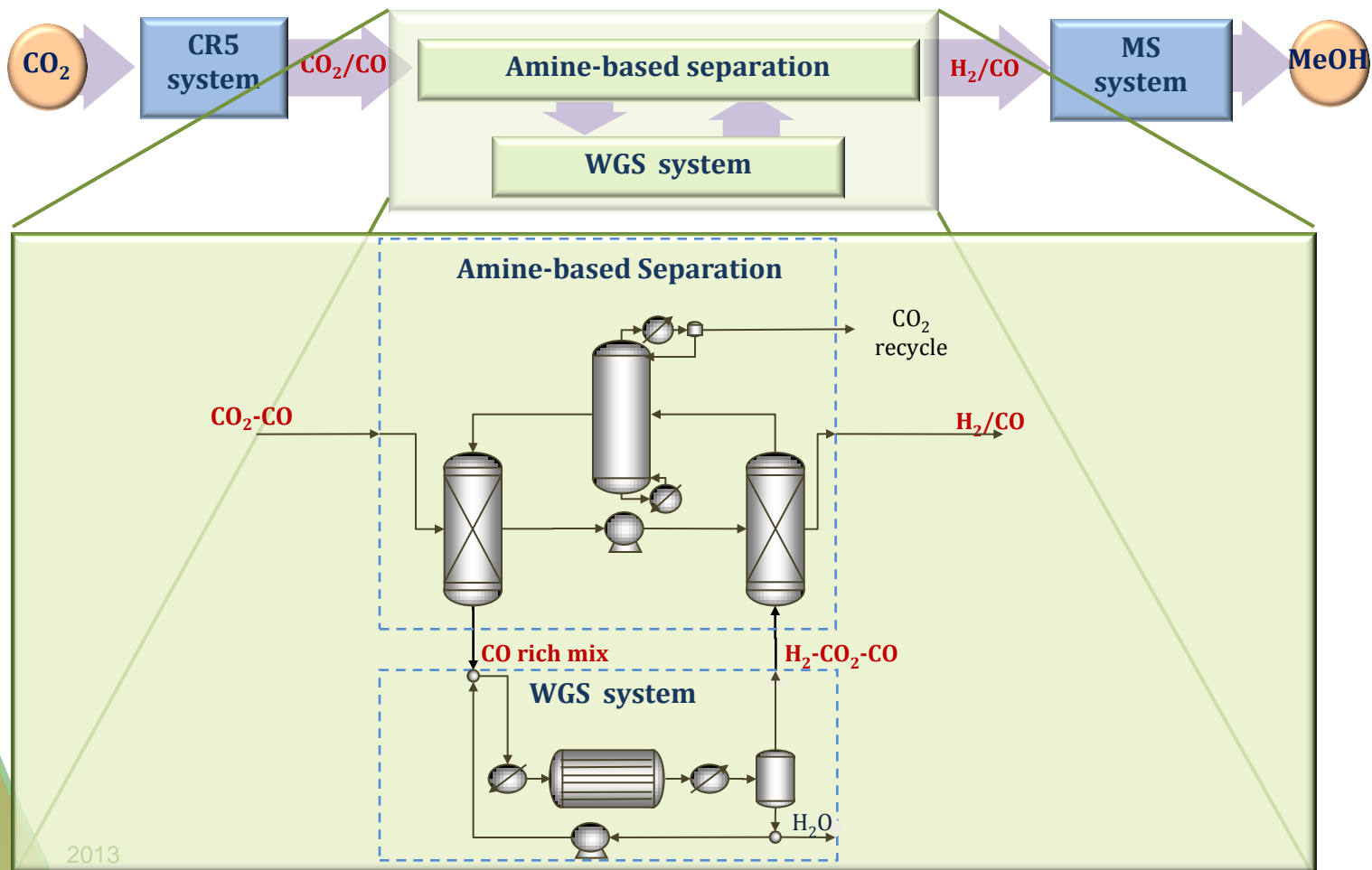




# Process Synthesis

## CO<sub>2</sub> pathway for MeOH production

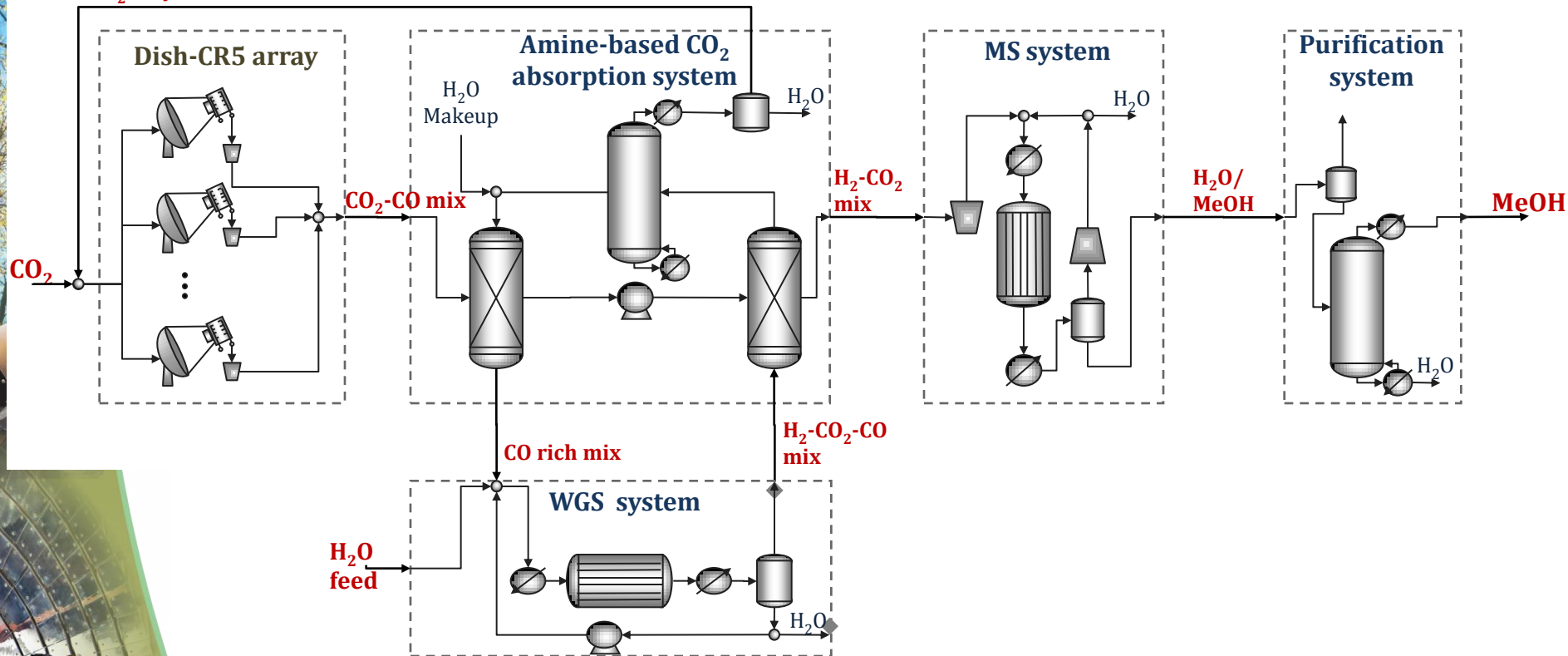
### Improved amine-based separation network



# Process Simulation

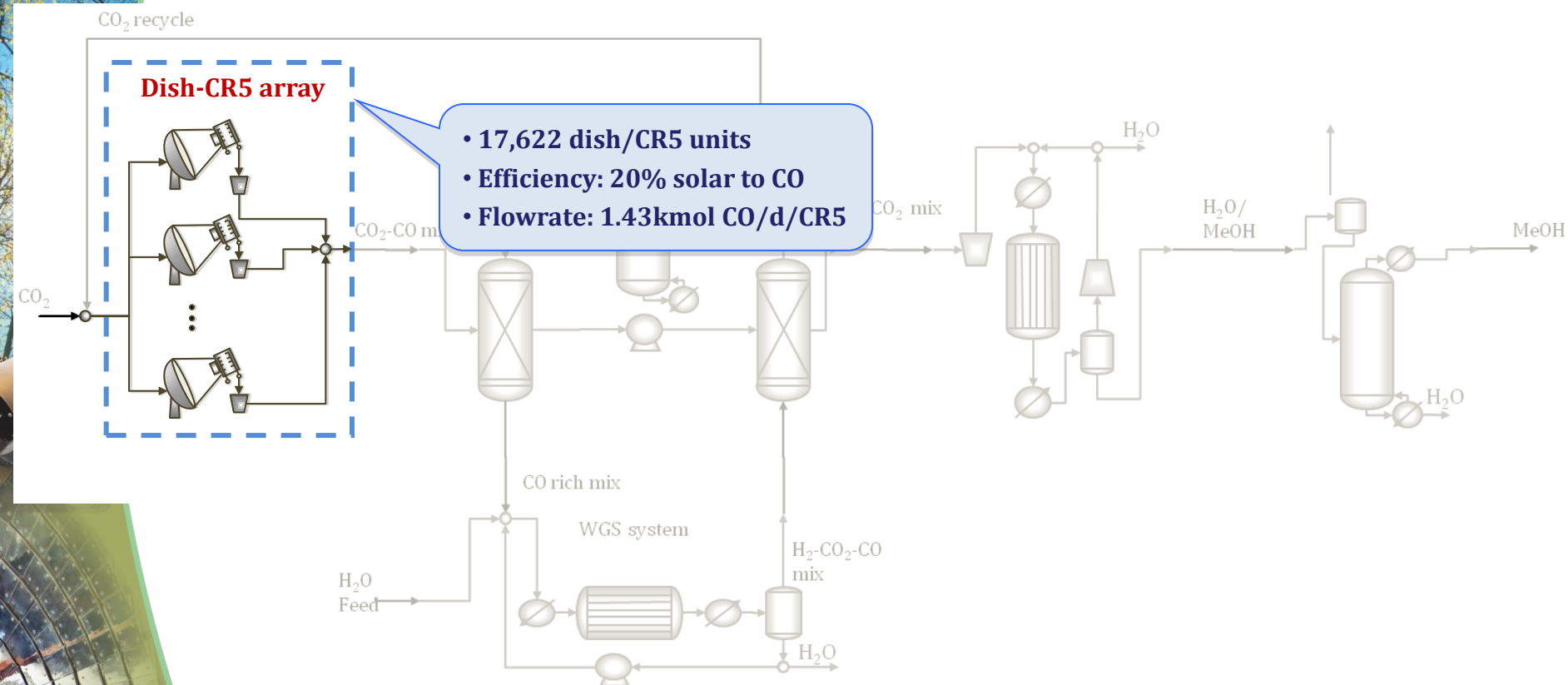
Process flow diagram (PFD) of CO<sub>2</sub> pathway for MeOH production

CO<sub>2</sub> recycle



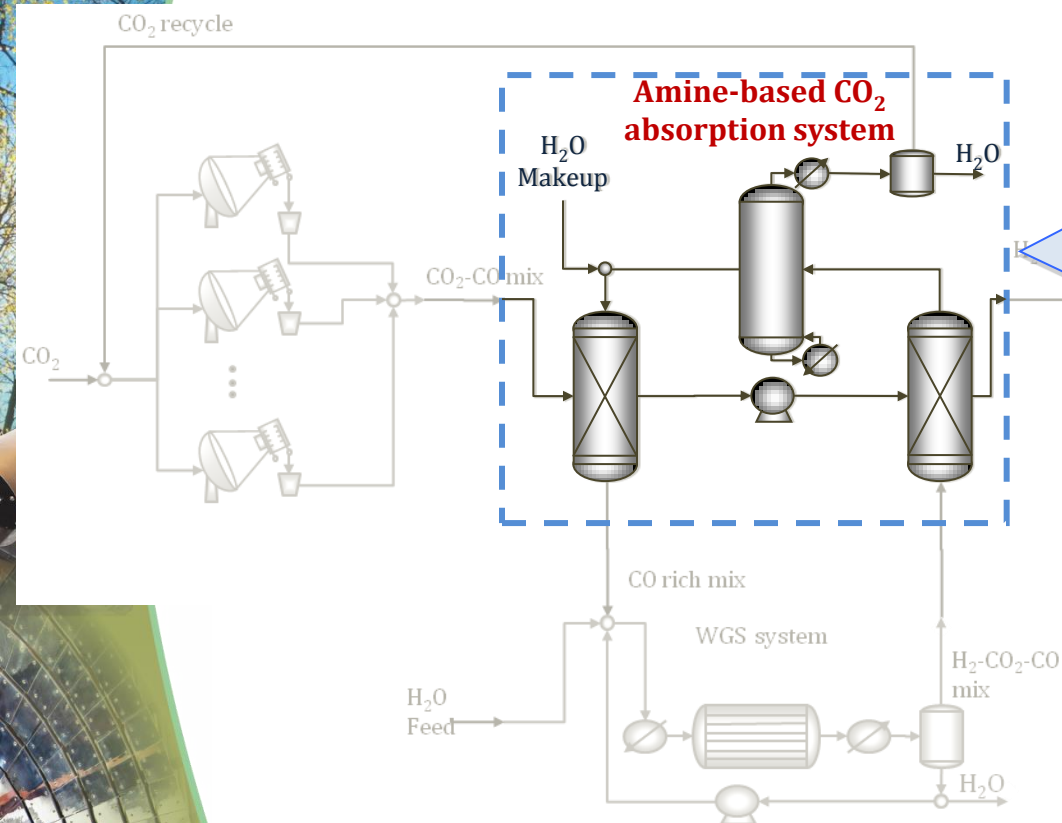
# Process Simulation

## Process flow diagram (PFD) of CO<sub>2</sub> pathway for MeOH production



# Process Simulation

## Process flow diagram (PFD) of CO<sub>2</sub> pathway for MeOH production

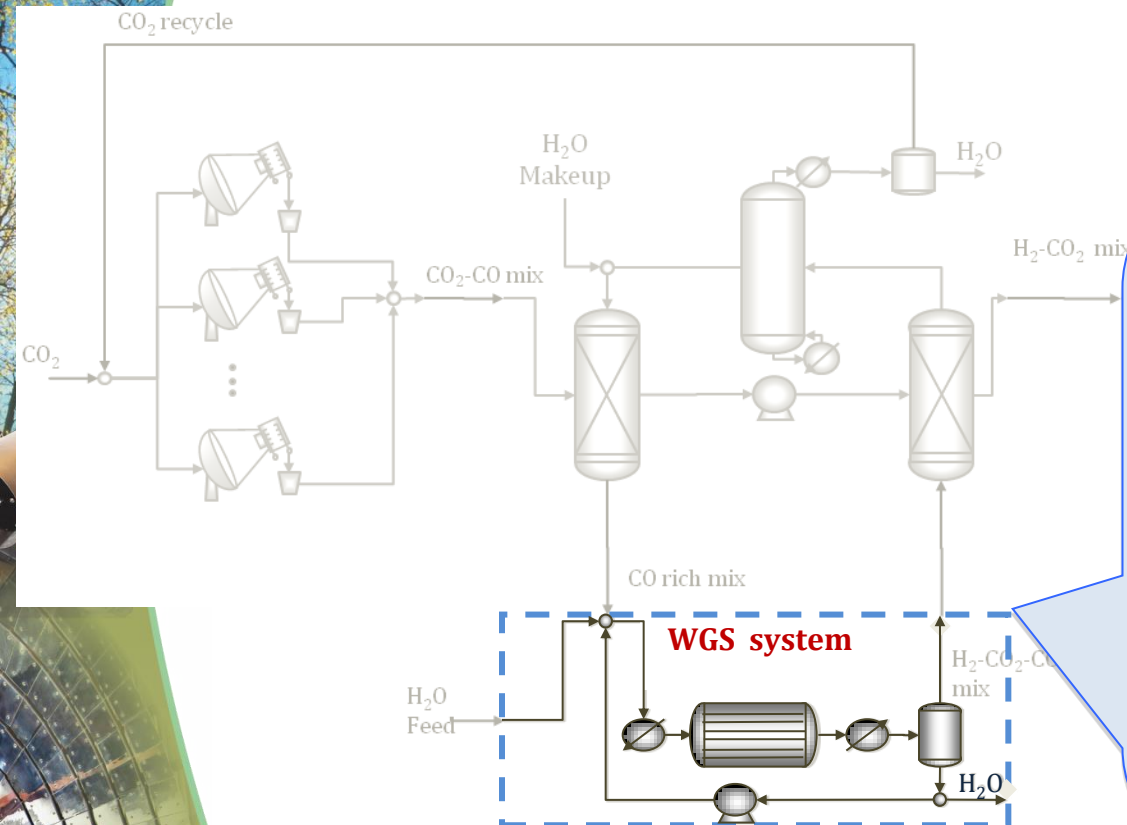


- MEA concentration: 25 wt%
- Operating conditions
  - 1<sup>st</sup> Absorber: 42°C, 1.1 atm
  - 2<sup>nd</sup> Absorber: 46°C, 20 atm
  - Regenerator: 42~72°C, 1.2 atm
  - Reboiler duty: 5,111 kW
  - Condenser duty: -4,901 kW
- Optimized variables: T, P, CO<sub>2</sub> loading



# Process Simulation

## Process flow diagram (PFD) of CO<sub>2</sub> pathway for MeOH production



- **WGS reactor:**



- **Catalyst : Cu/ZnO/Al<sub>2</sub>O<sub>3</sub>**

- **Kinetics**

$$r_{\text{WGS}} = \frac{k_4 \cdot P_{\text{CO}_2} \cdot \left( 1 - \frac{1}{K_{\text{WGS}}} \cdot \frac{P_{\text{H}_2\text{O}} \cdot P_{\text{CO}}}{P_{\text{H}_2}^3 \cdot P_{\text{CO}_2}} \right)}{\left( 1 + k_3 \cdot \frac{P_{\text{H}_2\text{O}}}{P_{\text{H}_2}} + k_1 \cdot \sqrt{P_{\text{H}_2}} + k_2 \cdot P_{\text{H}_2\text{O}} \right)}$$

$$k_1 = 0.499 \cdot e^{\frac{17197}{RT}}, k_2 = 6.62E-11 \cdot e^{\frac{124119}{RT}},$$

$$k_3 = 3453.38, k_4 = 1.22E10 \cdot e^{\frac{-94765}{RT}}, K_{\text{WGS}} = 10^{\left[ \frac{-2073}{T} + 2.029 \right]}$$

- **Feed ratio: CO/CO<sub>2</sub>/H<sub>2</sub> = 0.34/0.05/0.61**

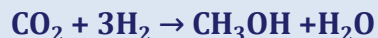
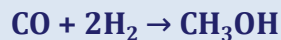
- **Reactor inlet condition: 280°C , 21.2 atm**

- **Optimized variables: T, P , feed ratio.**

# Process Simulation

## Process flow diagram (PFD) of CO<sub>2</sub> pathway for MeOH production

- **MS reactor:**



- **Catalyst: Cu/ZnO/Al<sub>2</sub>O<sub>3</sub>**

- **Kinetics**

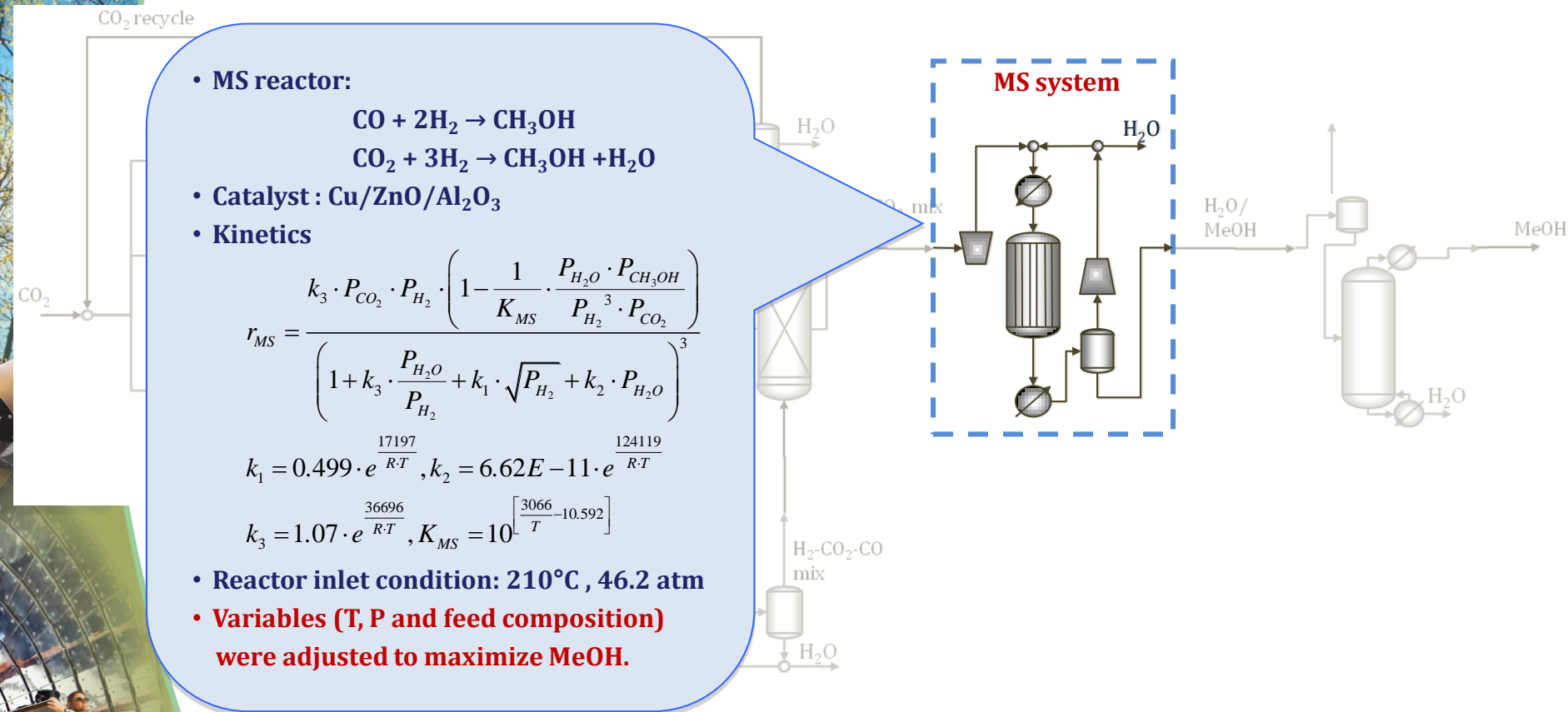
$$r_{MS} = \frac{k_3 \cdot P_{\text{CO}_2} \cdot P_{\text{H}_2} \cdot \left( 1 - \frac{1}{K_{MS}} \cdot \frac{P_{\text{H}_2\text{O}} \cdot P_{\text{CH}_3\text{OH}}}{P_{\text{H}_2}^3 \cdot P_{\text{CO}_2}} \right)}{\left( 1 + k_3 \cdot \frac{P_{\text{H}_2\text{O}}}{P_{\text{H}_2}} + k_1 \cdot \sqrt{P_{\text{H}_2}} + k_2 \cdot P_{\text{H}_2\text{O}} \right)^3}$$

$$k_1 = 0.499 \cdot e^{\frac{17197}{RT}}, k_2 = 6.62E-11 \cdot e^{\frac{124119}{RT}}$$

$$k_3 = 1.07 \cdot e^{\frac{36696}{RT}}, K_{MS} = 10^{\left[ \frac{3066}{T} - 10.592 \right]}$$

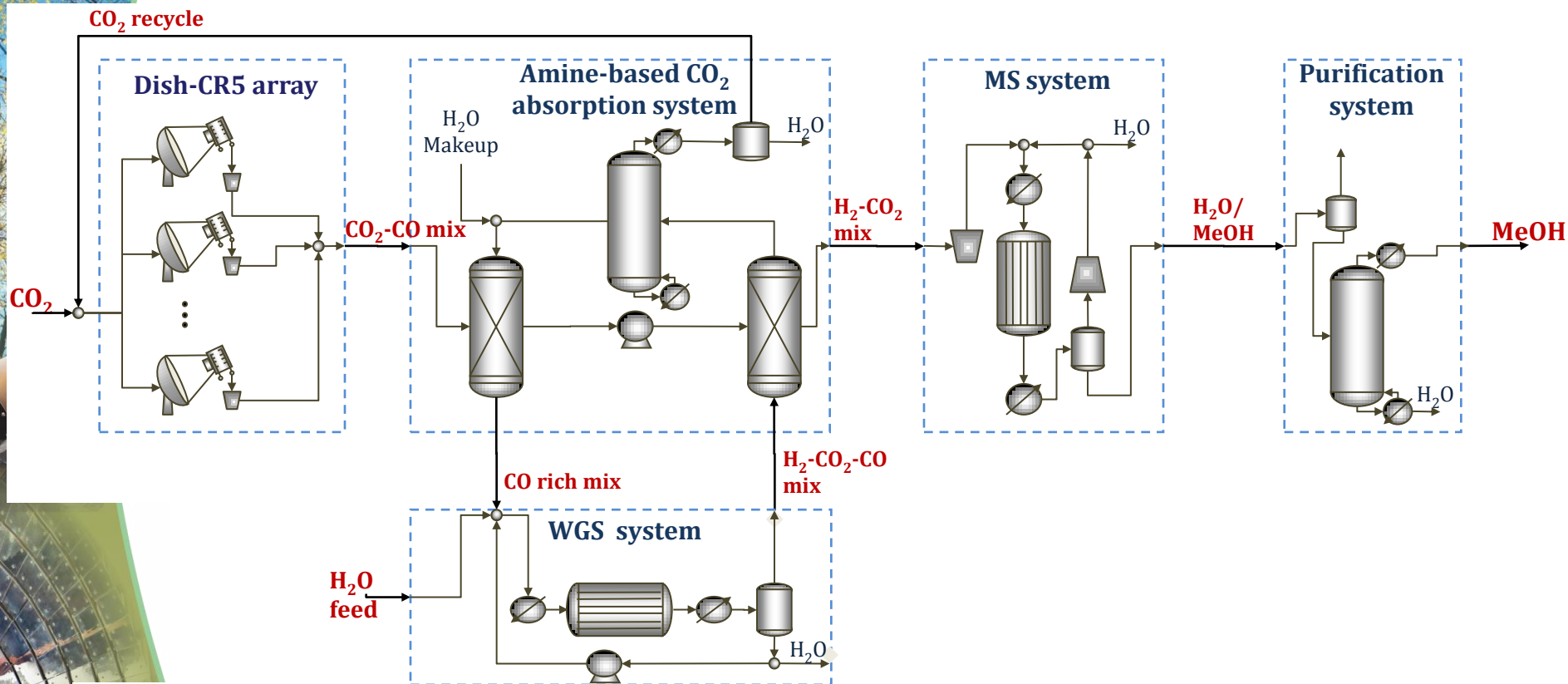
- **Reactor inlet condition: 210°C, 46.2 atm**

- **Variables (T, P and feed composition) were adjusted to maximize MeOH.**



# Process Simulation

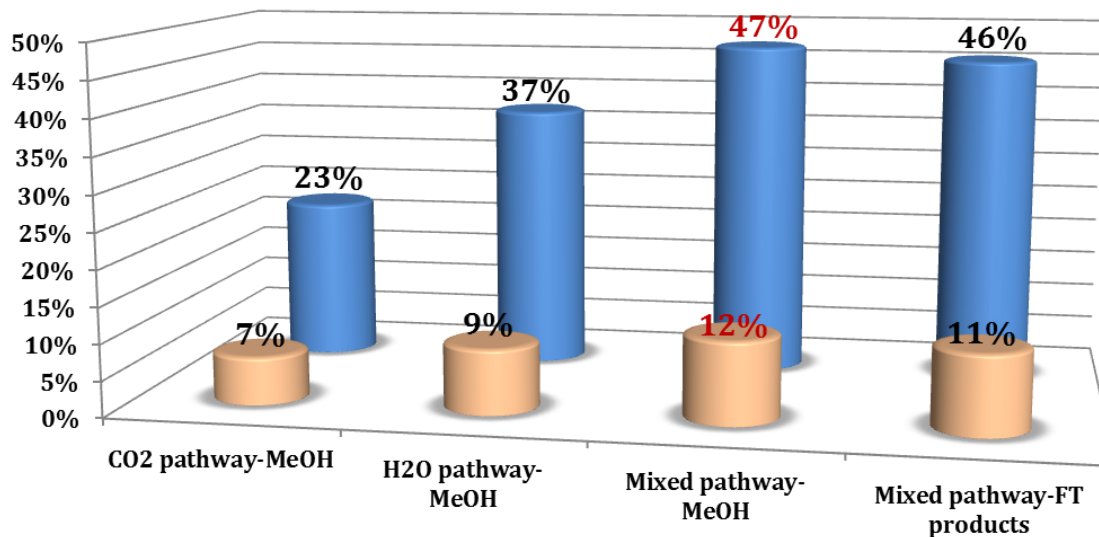
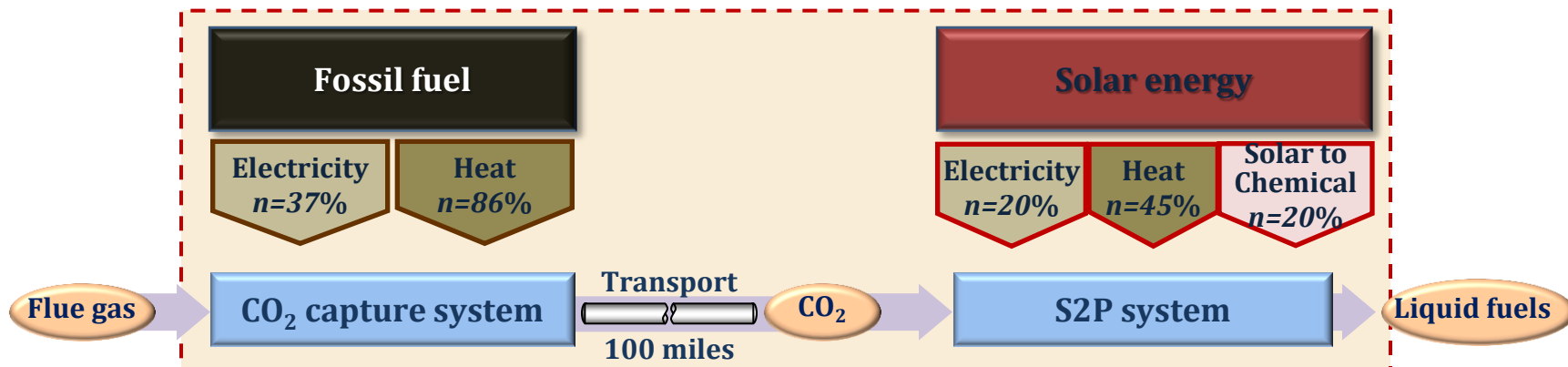
Process flow diagram (PFD) of CO<sub>2</sub> pathway for MeOH production



Mole flow (kmol/hr)	CO <sub>2</sub>	CO <sub>2</sub> -CO mix	CO rich mix	H <sub>2</sub> O feed	H <sub>2</sub> -CO <sub>2</sub> -CO mix	H <sub>2</sub> -CO <sub>2</sub> mix	H <sub>2</sub> O/MeOH	MeOH
CO <sub>2</sub>	431	3,246	1,055	-	790	48	54	-
CO	-	1,055	122	-	380	380	38	-
H <sub>2</sub>	-	-	-	-	675	674	-	-
H <sub>2</sub> O	-	-	15	673	-	-	338	1
MeOH	-	-	-	-	-	-	336	319

# Energy Efficiency

## Primary energy efficiency- from solar energy



■ Primary energy efficiency - from solar energy

■ Process energy efficiency [=product heating value/(process energy+chemical energy)]



# Economic Evaluation

Process simulation results: flows, T, P, heat duties

Sizing & costing data

Total capital  
investment  
(TCI)

=

Direct  
plant cost

+

Indirect  
plant cost

- ✓ Equipment cost
- ✓ Equipment setting
- ✓ Piping
- ✓ Civil work
- ✓ Steel & instrumentation
- ✓ Electrical systems

- ✓ Engineering & supervision
- ✓ G&A and Overheads
- ✓ Contract Fee
- ✓ Contingencies

# Economic Evaluation



Process simulation results: flows, T, P, heat duties

Sizing & costing data

Total capital  
investment  
(TCI)

Feed & utility data

Total  
expenses

=

Variable  
expense

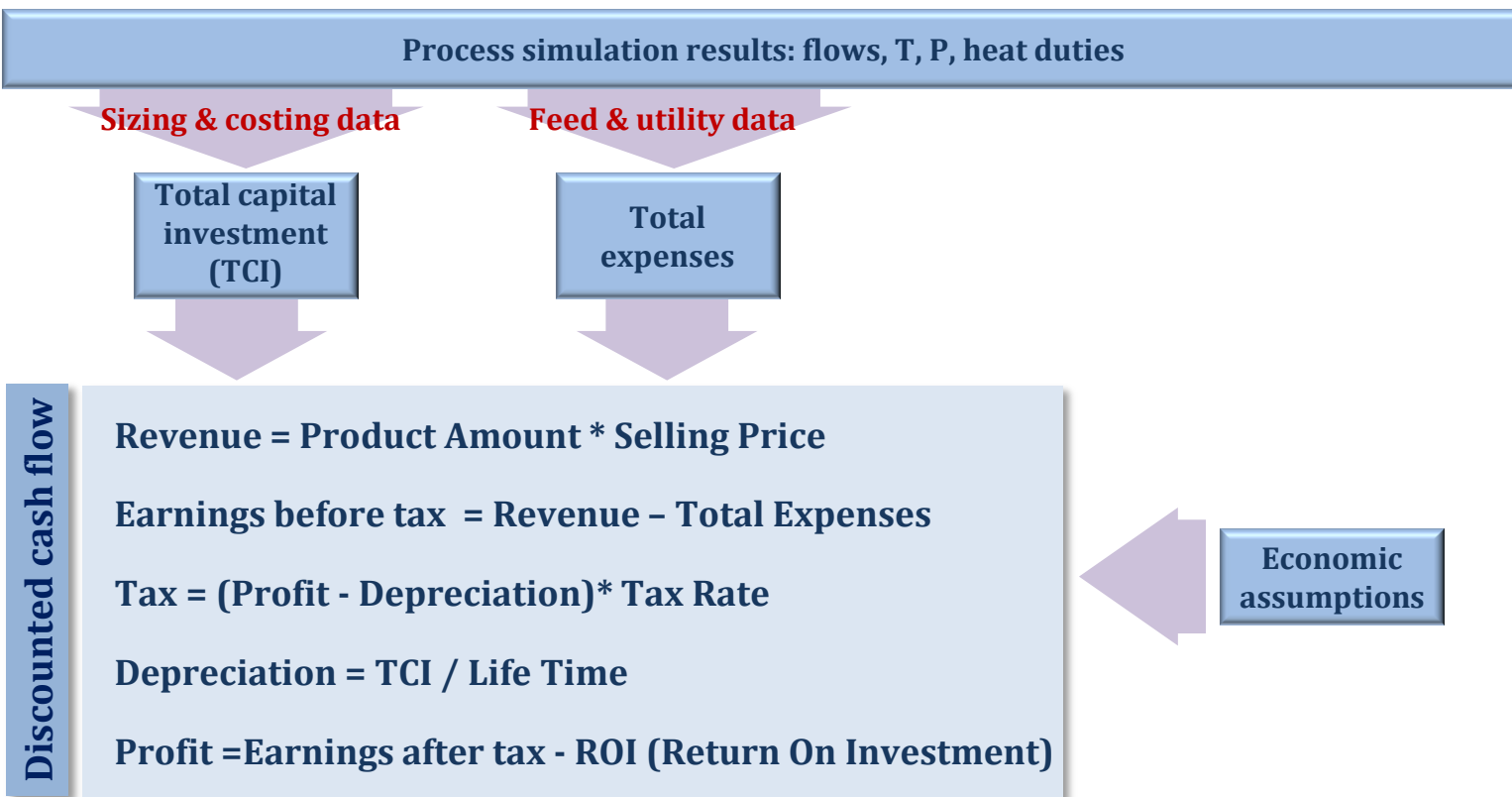
+

Fixed  
expense

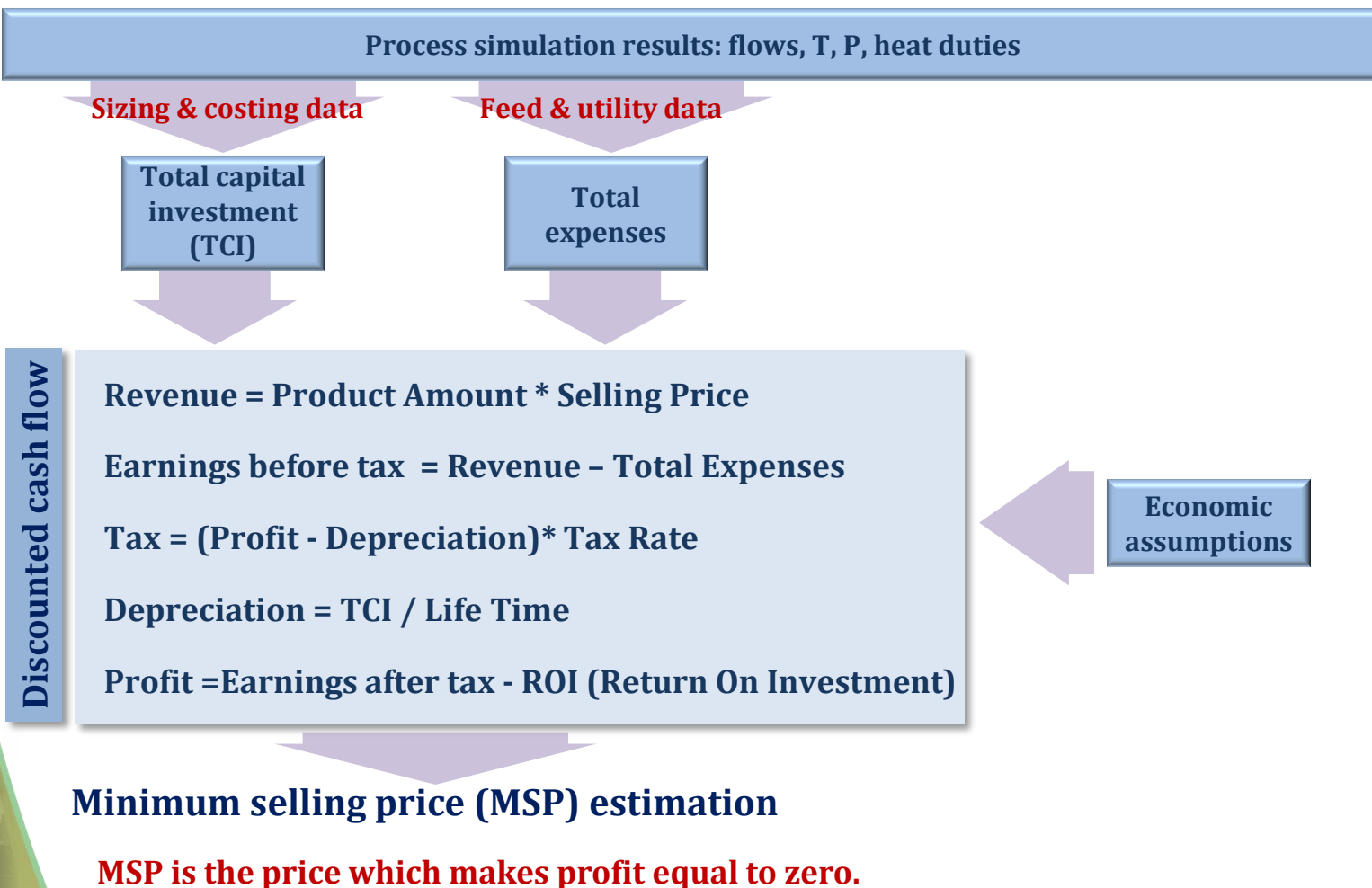
- ✓Utility
- ✓Raw material

- ✓Operating labor
- ✓Supervision labor
- ✓Operating supplies
- ✓Maintenance

# Economic Evaluation

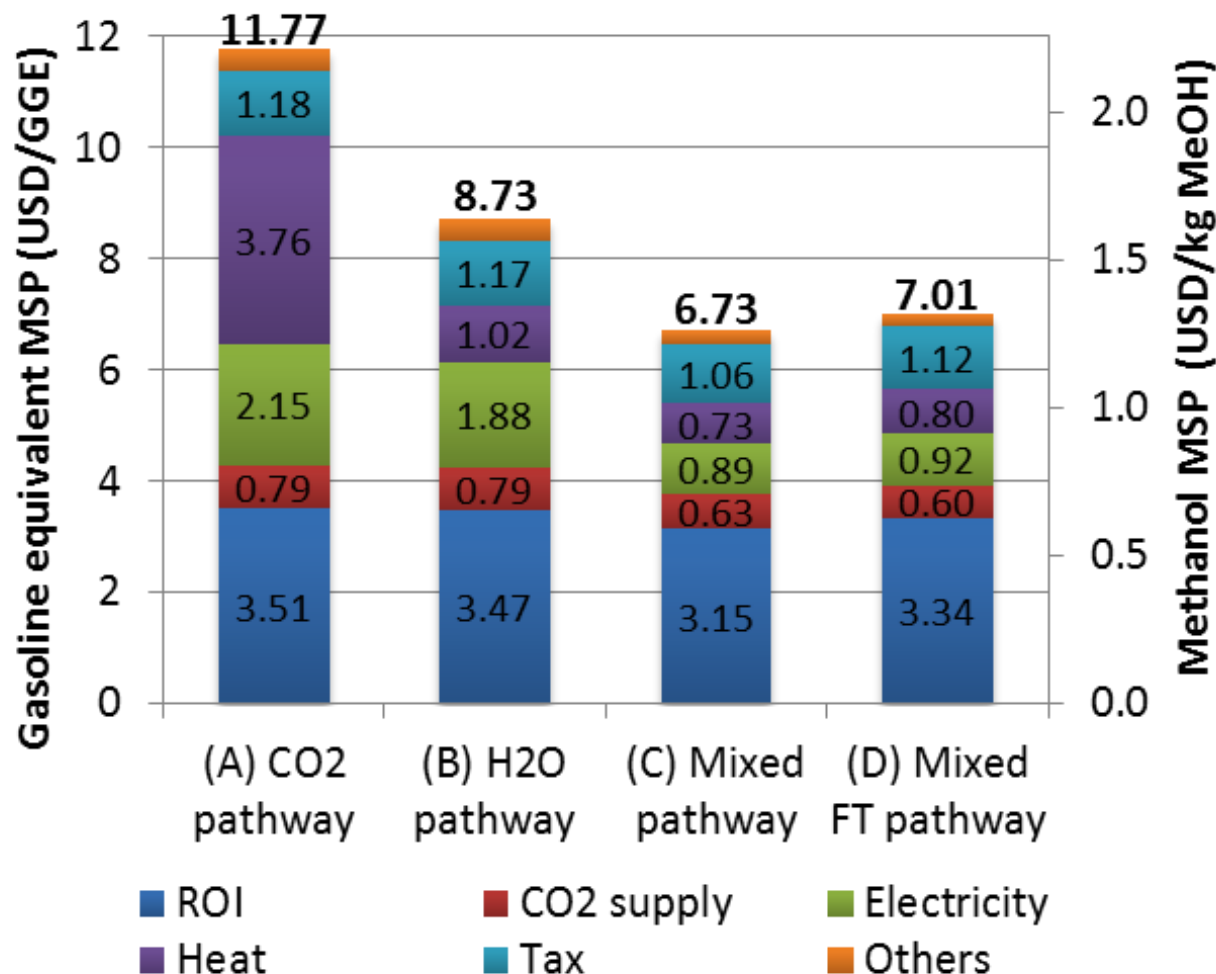


# Economic Evaluation



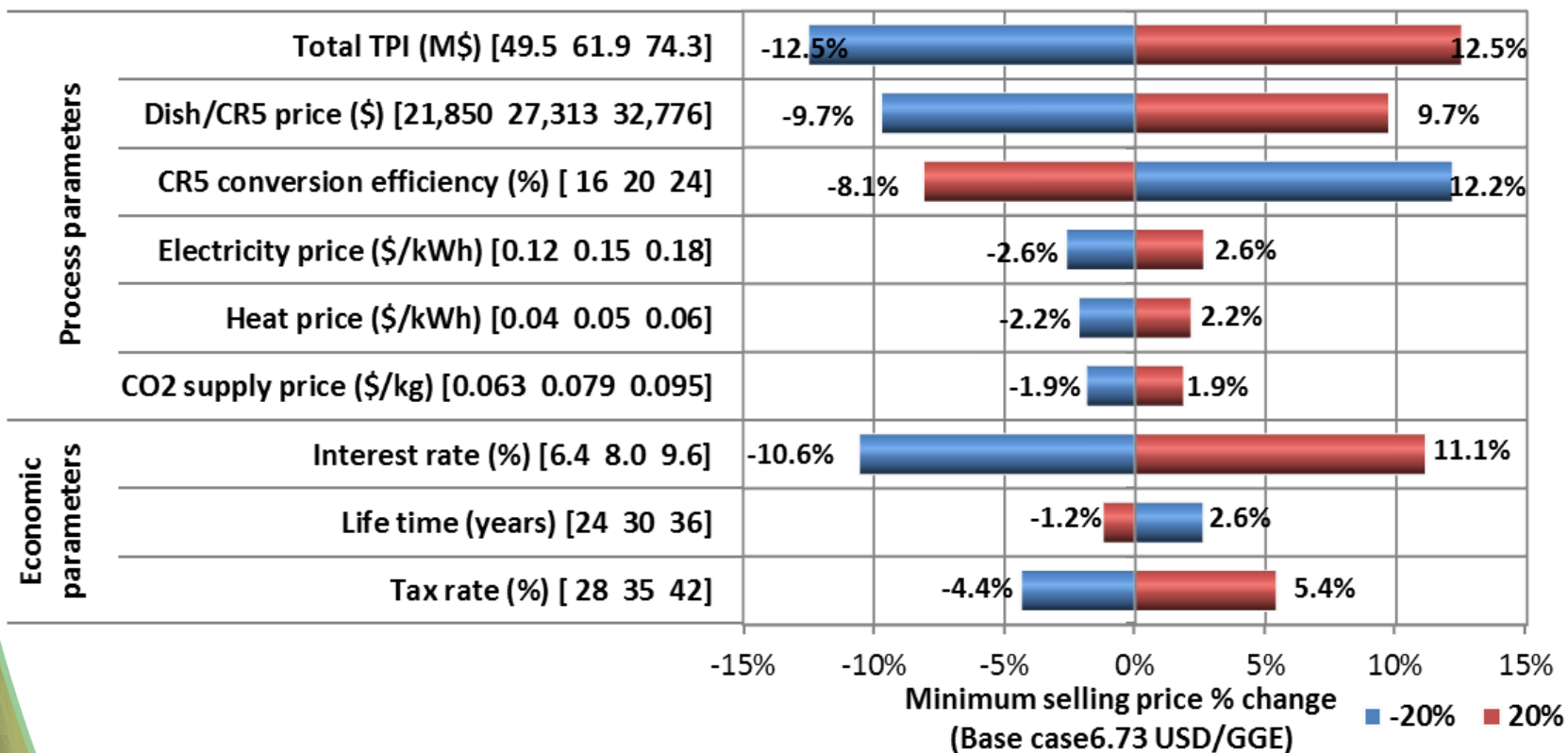


# Economic Evaluation: Minimum Selling Price



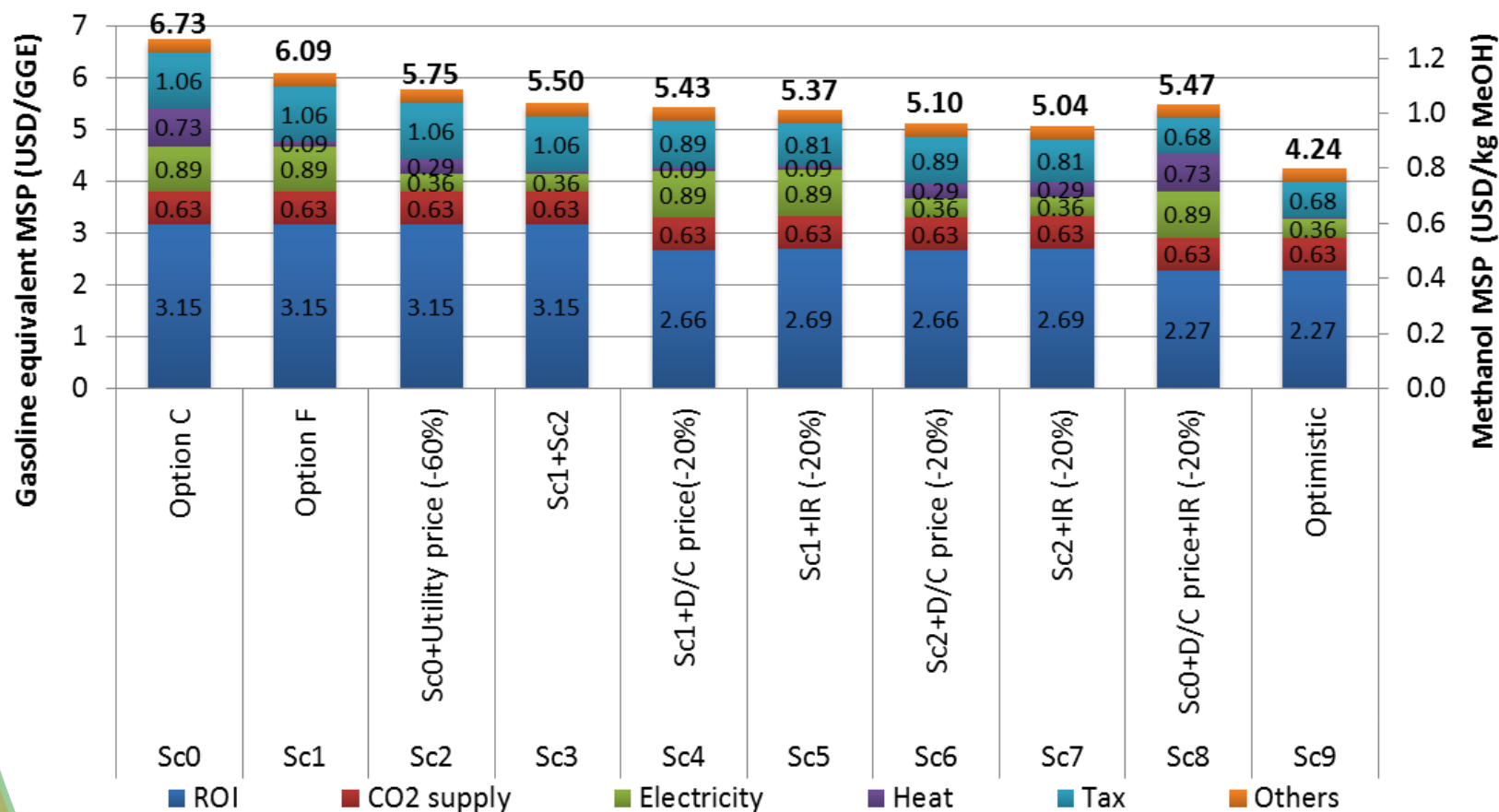
# Economic Evaluation: Sensitivity Analysis

## Mixed pathway to MeOH



# Economic Evaluation: Sensitivity Analysis

## Mixed pathway to MeOH



# Can 20% or better be achieved? Efficiency: Solar to Thermal



Resource efficiency = 95% for Daggett, CA ( $\text{DNI} > 300\text{W}/\text{m}^2$ )

Operational ~ 94%

Equip. Availability = 97%, Blocking&Shading = 98%, Wind Outage = 99%

Optical ~ 79%

Reflectivity = 93% (two reflections)

Dirt = 95%

Window = 95%

Tracking = 99%

Intercept = 95%

Receiver ~ 82%

Radiation = 82%

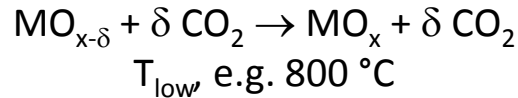
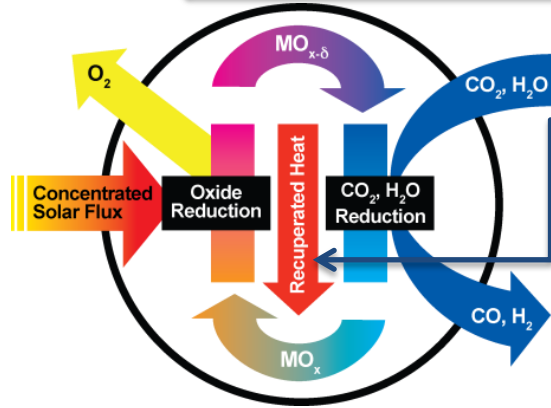
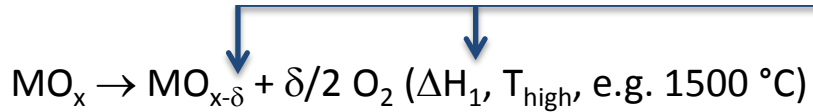
Conduction/Convection = 0 %

Solar-to  
heat:  
~58%

Solar to Fuel = 10%  
If Heat to Fuel = 17%



# 20%? Thermal to Chemical Efficiency



Efficiency is a function of:

Thermodynamics:  $\Delta H_1$  ( $T_{\text{high}}$  &  $T_{\text{low}}$ ),  $\delta$

Kinetics:  $\delta$

The reactor: recuperation effectiveness & pressures

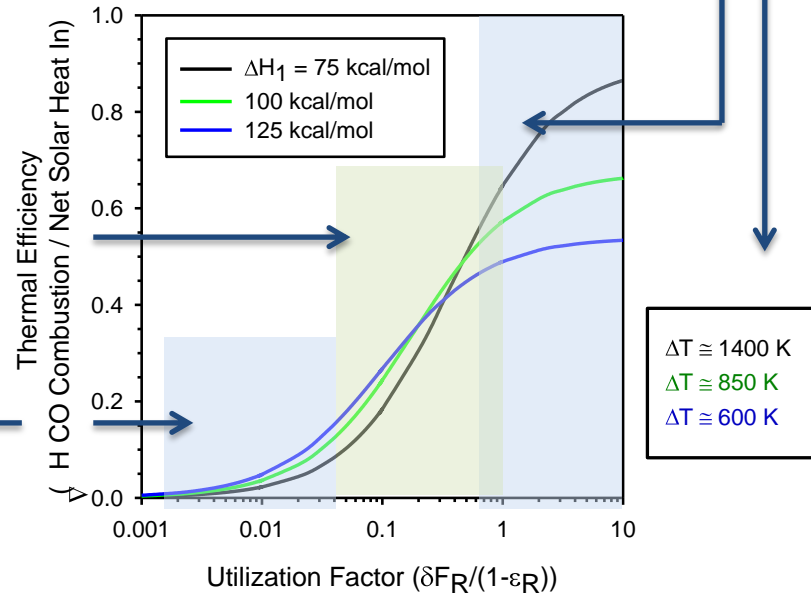
The maximum possible efficiency is limited by  $\Delta H_1$ .

High efficiency (small  $\Delta H_1$ )

corresponds to a large  $T_{\text{high}} - T_{\text{low}}$ .

The possible efficiency increases with degree of reaction ( $\delta$ ) and/or effectiveness of recuperation.

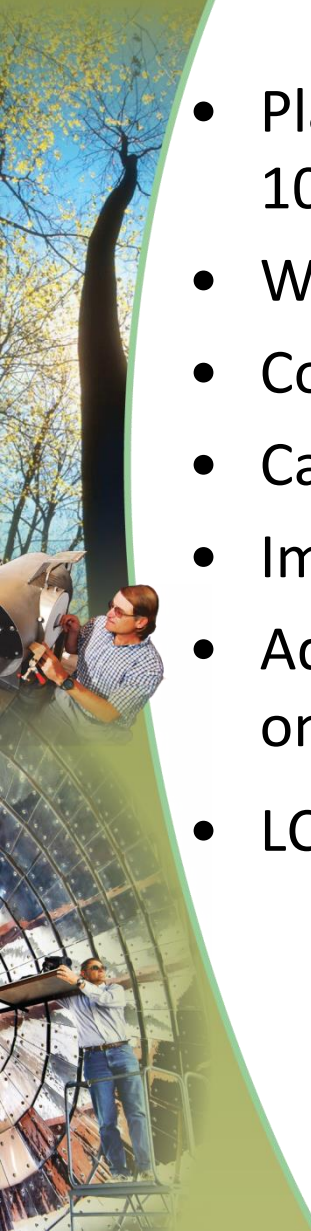
When utilization is low, sensible heat demand becomes a more dominant factor than  $\Delta H_1$ .



# Technical Summary



- Plausible route to Solar Fuels – if efficiency is high enough – 10% solar to hydrocarbon minimum.
- Within the scale of other human endeavors
- Costs are “in the ballpark” of competitiveness.
- Capital cost is a primary driver
- Improvements are needed in separations
- Advances in thermochemistry required but plausible – focus on utilization (recuperation and reaction extent).
- LCA work to be published soon.

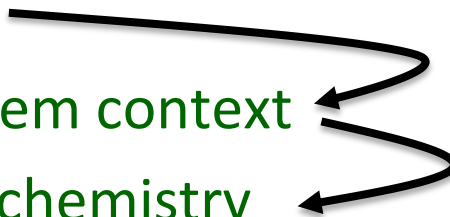




# Lessons Learned

# In an Ideal World

- The Vision
- The Big Picture
  - If we succeed, does it matter?
  - If so, under what circumstances
- The Technology
  - Syngas
  - Full system context
  - Thermochemistry



# Concurrent efforts are not uncommon

Reality is less linear

- The Vision
- The Big Picture
  - If we succeed, does it matter?
  - If so, under what circumstances
- The Technology
  - Syngas
  - Full system context
  - Thermochemistry



# Underappreciated



- The big picture matters:
  - End to End is critical – BOP matters
  - “efficiency can be low if cost is low” breaks down
  - Efficiency needs to be defined back to the source
- Important to state assumptions, work from a common basis
  - Necessary for good decision making
  - Avoid creating false expectations that undermine the enterprise in general
- There are disincentives for a research effort to undertake this type of work
  - more difficult as work becomes more collaborative, cross discipline, blurs the lines of basic and applied...

# Challenges



- What do we assume about the future?
  - “When will this be competitive” is a loaded question assuming no changes in the status quo.
  - Externalities, regulations ...
  - What do we assume about other technologies? - In our case “where does the CO<sub>2</sub> come from?”
- How do we compare across technologies
  - Energy is not energy is not energy ...
  - Biomass looks to capital investment per unit capacity – the feedstock solar fuels largely appear as capital
  - What metrics are important and how are they valued? water, land, security, CO<sub>2</sub> ... (\$, social costs, etc?)
- How do we use this tool to set targets, goals, and expectations without discouraging innovation?