

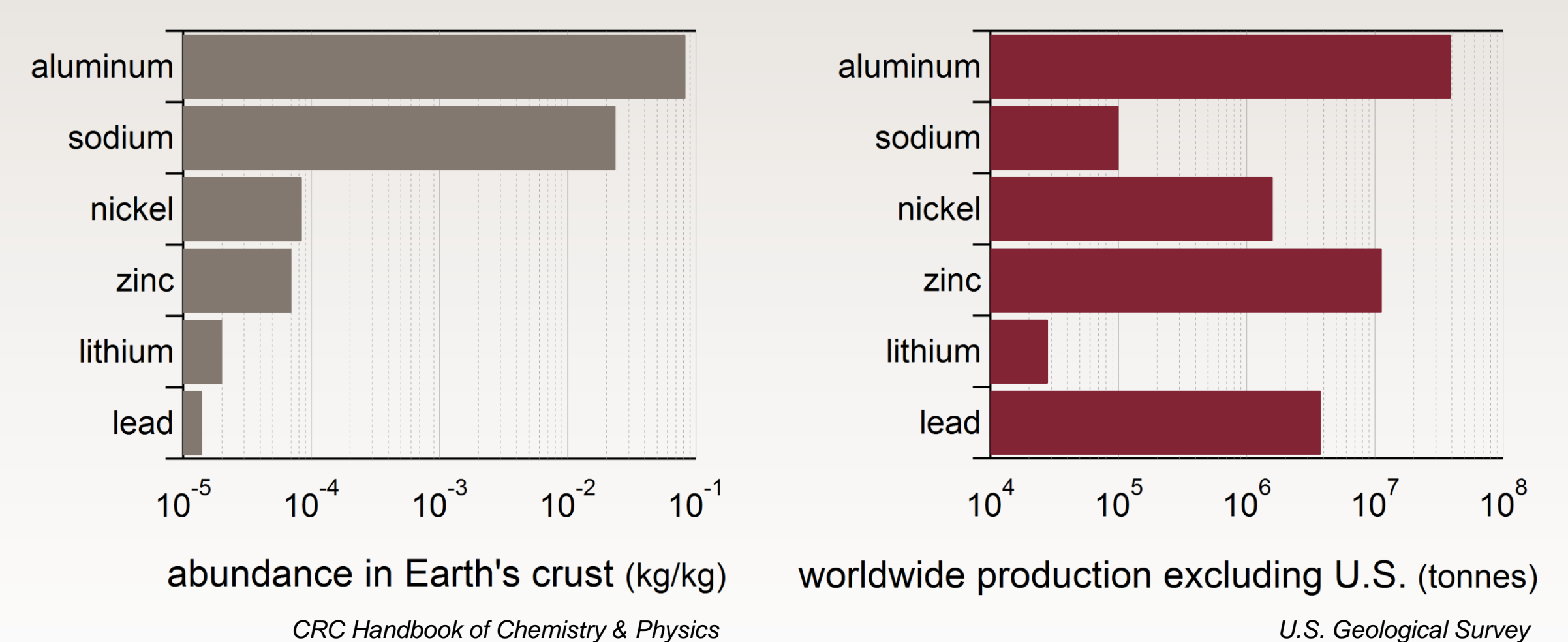
Chloride-Insertion Electrodes for Rechargeable Aluminum Batteries

Nicholas Hudak, Advanced Power Sources R&D (Org. 2546)
Early Career LDRD Program, New Start Project for CY12

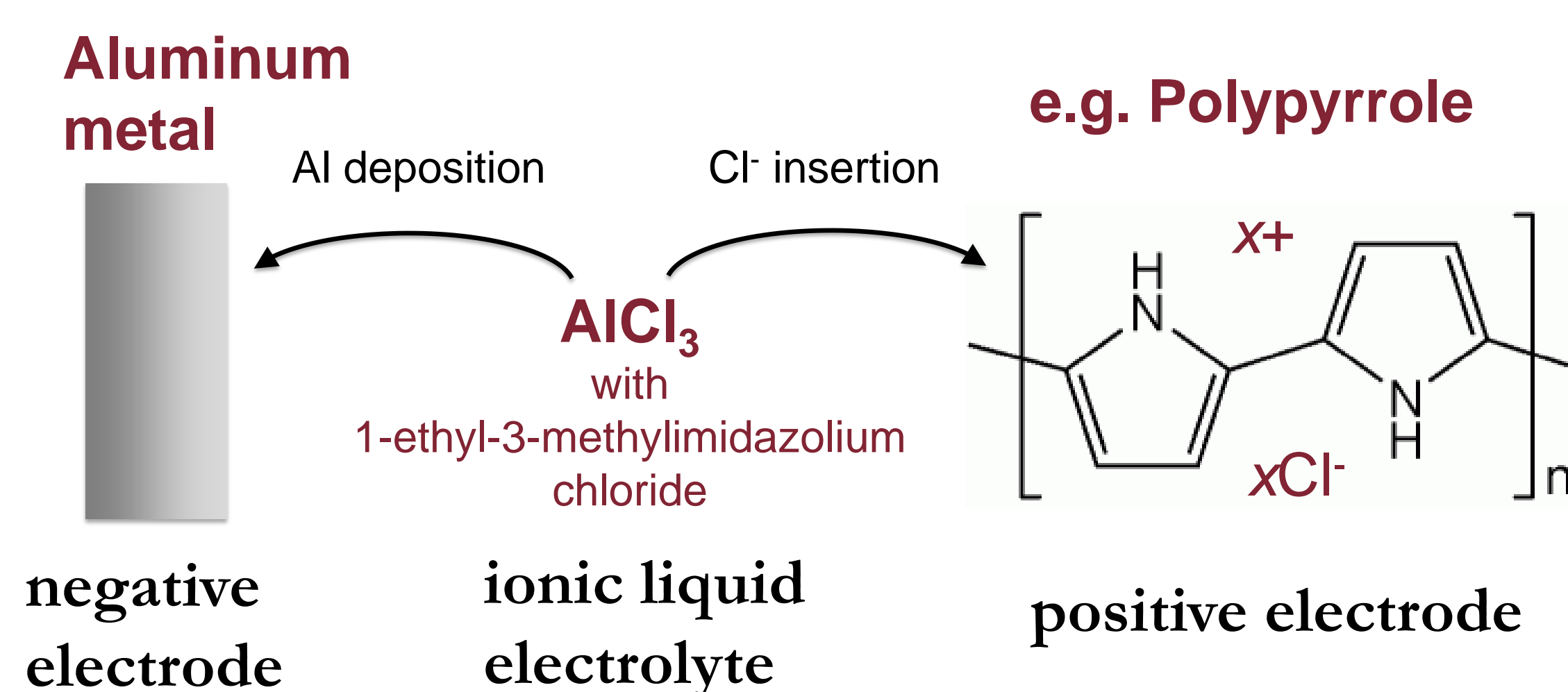
Problem and Motivation

- Demand for rechargeable battery systems continues to increase
 - Electric vehicles already commercially available
 - Stationary energy storage being introduced to electrical grid
- Aluminum can be the energy-carrying material (active electrode material) in rechargeable batteries
 - Competitive material in terms of abundance, cost, and charge-storage capacity
 - Alternative to lithium, zinc, etc.
- The use of ionic liquids enables room-temperature plating and stripping of Al^0 (the negative electrode in an aluminum battery)

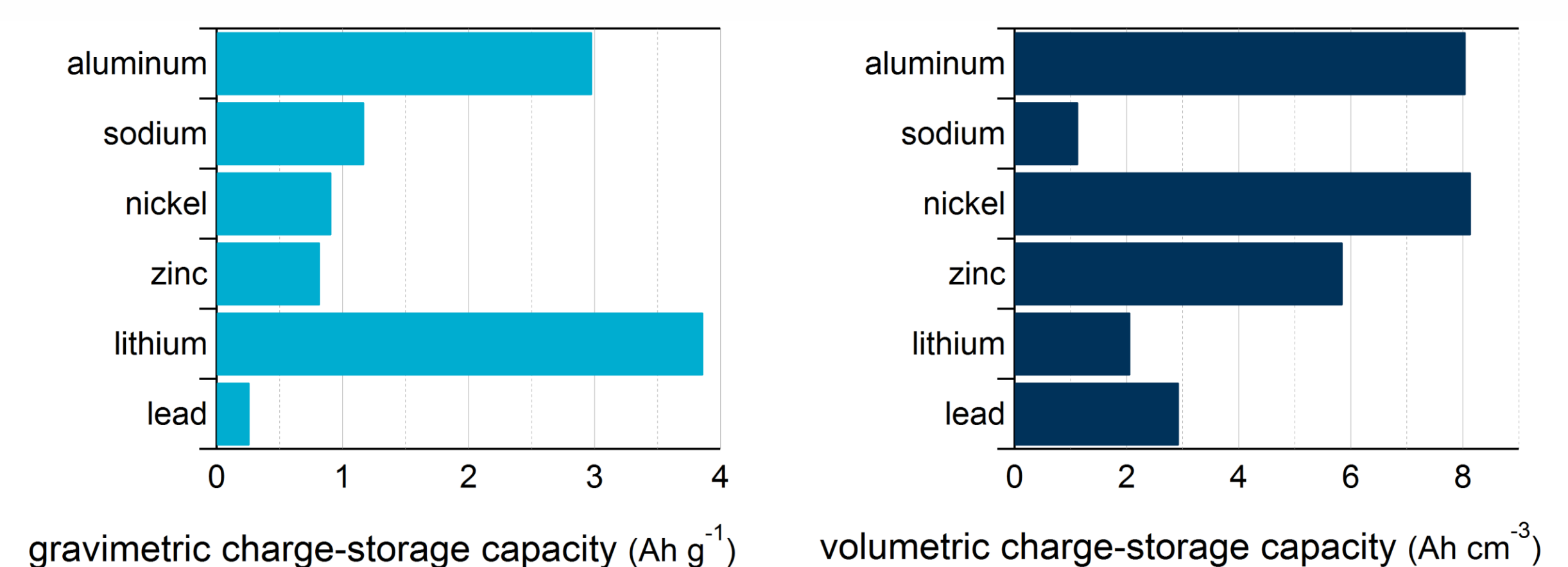
Aluminum is abundant and widely produced compared to other energy-storage metals.



The positive electrode in the proposed battery is a conductive polymer that electrochemically inserts chloride-ion from AlCl_3 .

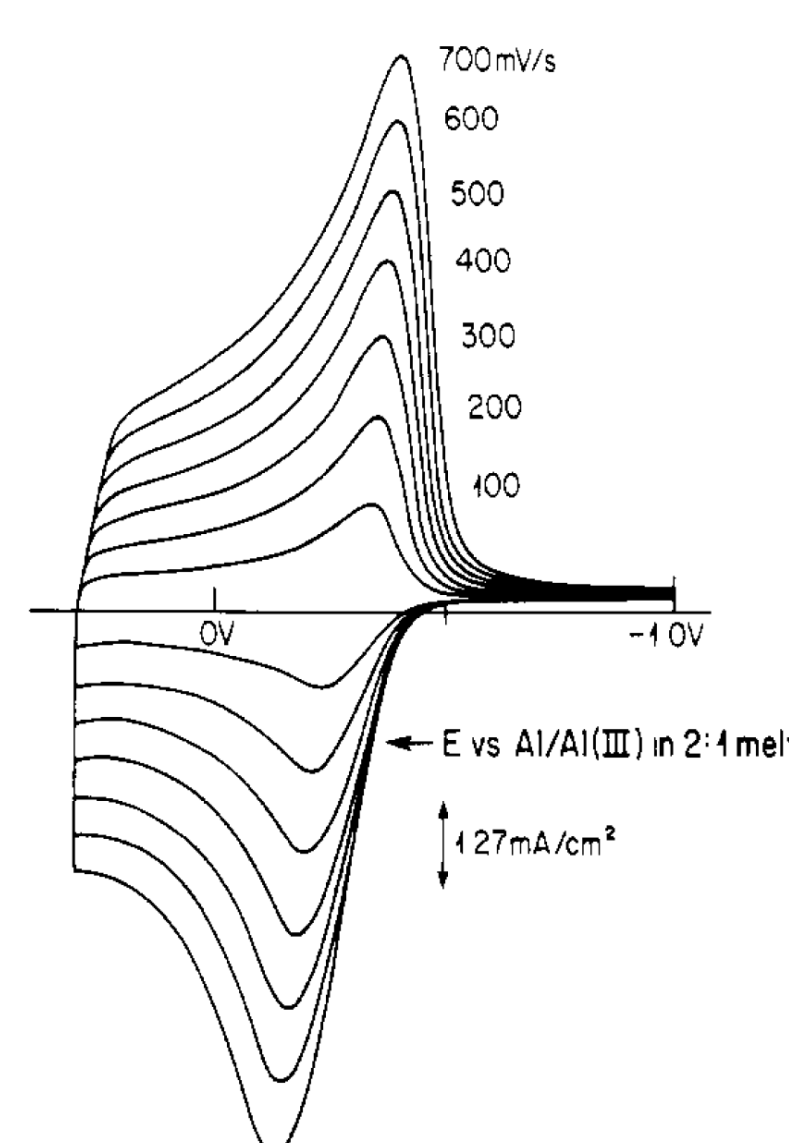


Aluminum's capacity for charge storage is competitive on gravimetric and volumetric scales.



Approach and Goals

- Demonstrate electrochemical synthesis of chloride-doped conductive polymers
 - e.g. electropolymerization of pyrrole or thiophene in AlCl_3 ionic liquid
 - effects of electrochemistry on polymer composition
- Demonstrate galvanostatic cycling of aluminum cells
 - conductive, conjugated polymer as positive electrode (via chloride insertion)
 - effects of polymer composition on cell performance
 - quantify charge-storage capacity, energy density, cycle life for comparison to other battery systems



Potential Impact

- May lead to development of rechargeable batteries made of inexpensive, domestically available materials
- Increased understanding of conductive polymers and electrochemistry in ionic liquids

Programmatic Interest

- Basic energy science
- Electricity delivery and energy reliability (grid-scale storage)
- Energy security
- Proposed batteries may be commercially relevant if performance proves to be competitive