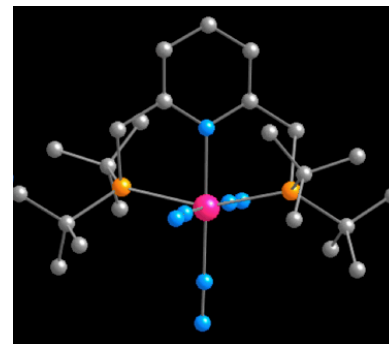
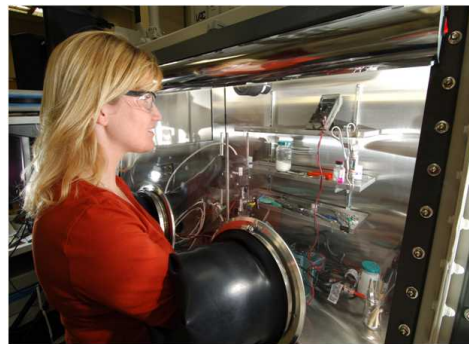


Exceptional service in the national interest



Advanced Power Sources R&D

Power Source Technology Group

Nick Hudak
Chris Applett
Department Manager: Tom Wunsch



Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

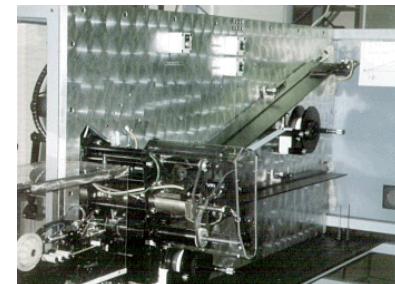
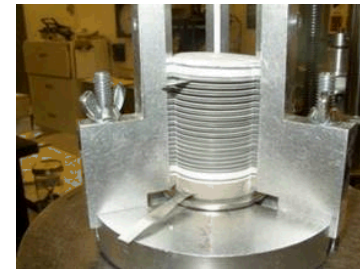
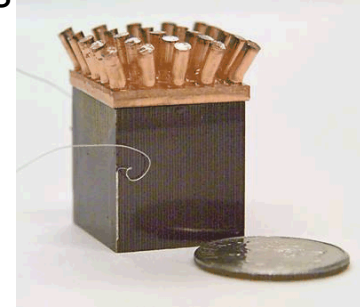
Program Areas

- DOE/National Nuclear Security Administration
 - Reserve battery technology for weapon system power
 - Primary battery technology for flight test (telemetry) power
- DOE/Office of Electricity Delivery & Energy Reliability
 - Materials, applied R&D, systems, characterization & testing, analysis
 - Batteries for grid-scale energy storage
- DOE/Office of Vehicle Technologies
 - Battery safety research and abuse testing
 - Materials R&D: Inactive materials (separator, electrolyte)
- LDRD (internal investments)
 - Modeling and characterization
 - Novel energy storage concepts and materials
 - Degradation mechanisms



PSTG Capabilities

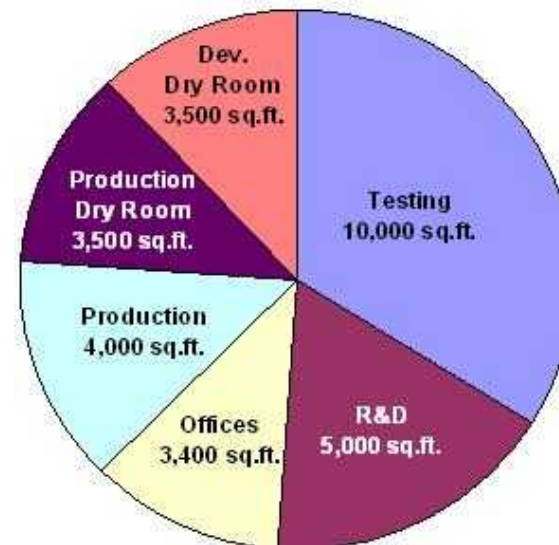
- Thermal battery and Li-ion cell prototyping facilities; small lot manufacturing of high-reliability thermal batteries
- Design and development of a wide variety of batteries, capacitors, and thermoelectric power supplies
- Unique battery abuse testing facilities; tester design and support for battery performance characterization and product acceptance
- Development of advanced electricity storage and power engineering technologies for electric supply
- High-fidelity modeling and simulation
- Fundamental electrochemical studies and development of improved battery materials
- More than 20 scientists and engineers, many with decades of experience in power source engineering



PSTG Research, Development, Test and Production Facilities

Two distinct work environments:

- **Building 894**
 - Three Dry Rooms
 - Production
 - Development
 - Research, cell prototyping
 - Chemistry Labs
 - Battery/Cell performance testing
 - Production
 - Header assembly
 - Battery assembly
 - Inspection (leak and x-ray)
 - Machine Shop
 - Records Room
 - Offices
- **Building 905**
 - Battery Abuse Testing Laboratory



Building 894
29,400 sq.ft.

Grid-Scale Energy Storage

Office of Electricity Program Area



Goal: Seek breakthroughs in materials, components and cell designs for next generation technologies.

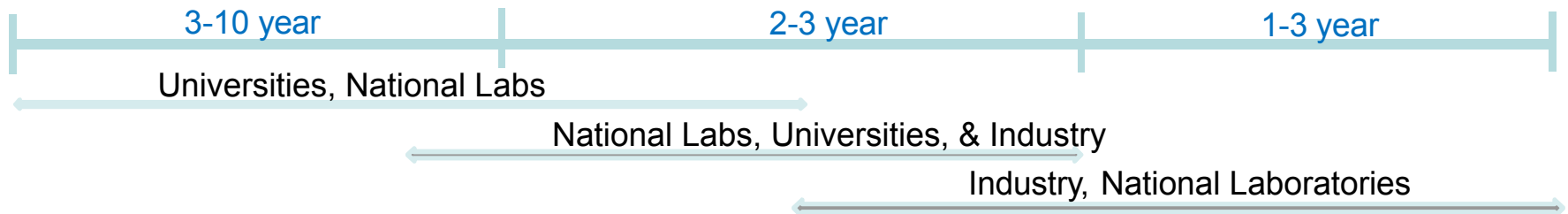
- ☐ Novel electrodes, materials, and membranes;
- ☐ New electrolyte and new redox chemistry;
- ☐ Next gen. redox flow batteries;
- ☐ Next gen. low cost Na batteries;
- ☐ State-of-the-art diagnostic/characterization.

Goal: Improve key technologies for mid term applications in power management and other specialized applications.

- ☐ Advanced lead-carbon batteries;
- ☐ Low cost Li-ion batteries;
- ☐ Long life redox flow batteries;
- ☐ Electrochemical modeling

Goal: Test and validate grid relevant performance of components and systems:

- ☐ Components and systems;
- ☐ Protocols and standards;
- ☐ Performance and cost;
- ☐ Grid viability and scalability;
- ☐ Non-electrochemical options



Cross-cutting activities:

Analysis, Modeling, and Integration

- ☐ Impact of renewable and storage on grid;
- ☐ Storage needs for different applications;
- ☐ System analysis and storage options;
- ☐ Cost and performance of different technologies;

Battery Abuse Testing

Office of Vehicle Technologies Program Area

Full range of material, cell, module and pack level thermal characterization, electrical, and mechanical abuse testing.



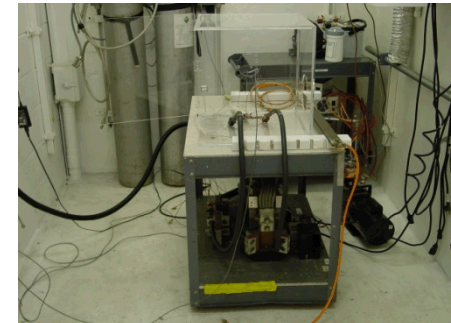
Over Discharge



Over Temperature



Overcharge

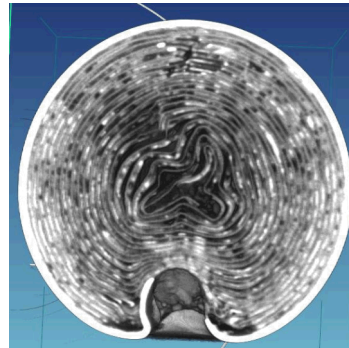


Short circuit

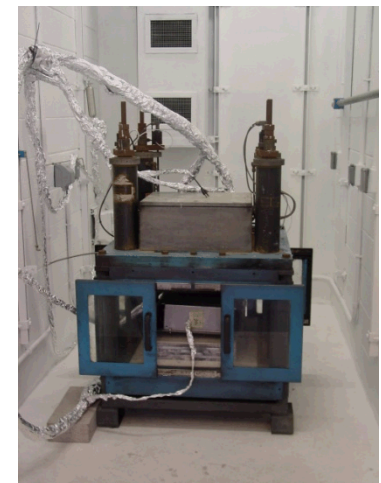
Cell Prototyping



CT X-ray



**Thermal Stability
(Accelerating Rate Calorimetry)**



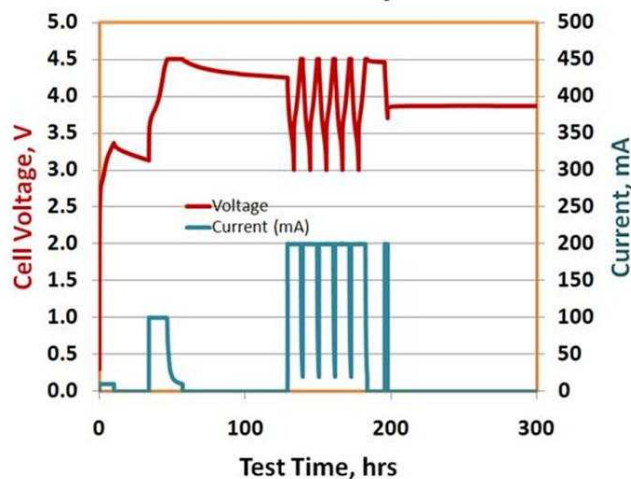
Crush

SNL-Built, Rechargeable 18650 Cells with SNL Coated Electrodes

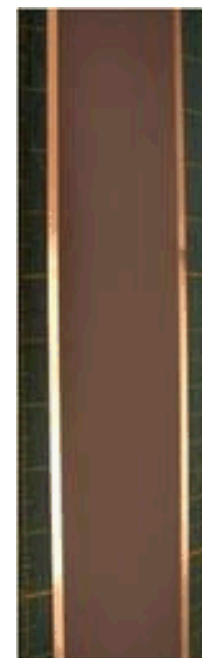
- Coated both anode and cathodes with in-house facility
- Anode is Conoco Phillips carbon
- High voltage cathodes:
 - $\text{LiNi}_{0.4}\text{Mn}_{0.3}\text{Co}_{0.3}\text{O}_2$ (4.3 -4.5 V)
 - $\text{LiMn}_{1.5}\text{Ni}_{0.5}\text{O}_4$ (>4.8 V)
- Fabrication of 18650 cells and testing
- Thermal abuse facility
 - ARC run on 18650 cell



Formation Cycle for Sandia-Built 18650
Cell at room temperature

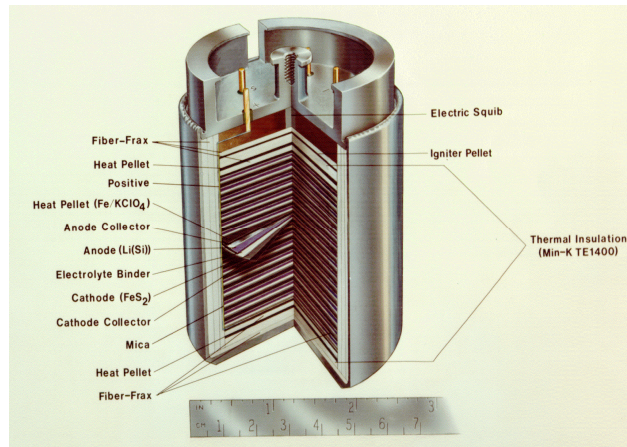


Cathode

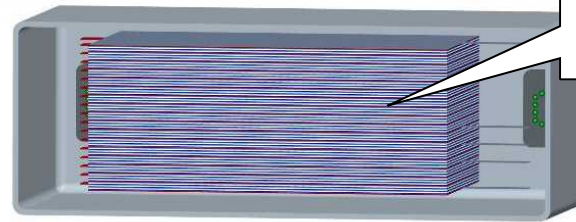


CP Anode

Advanced Thermal Battery Technology



Traditional Design



Each layer will be much thinner than traditional devices

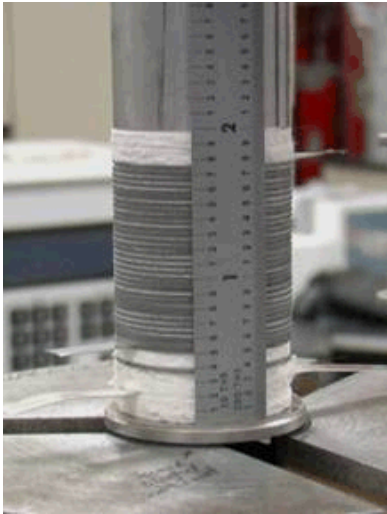
Based on thin film coating technology, enabled by new binder technology

Much more volume efficient

- Improved Electrical Performance
- More Robust
- Lower Cost
- Enhanced Safety
- Enable New Form Factors
- Design for Commercial Manufacture
- Enable Commercial Applications

Advanced Thermal Battery Architecture Project

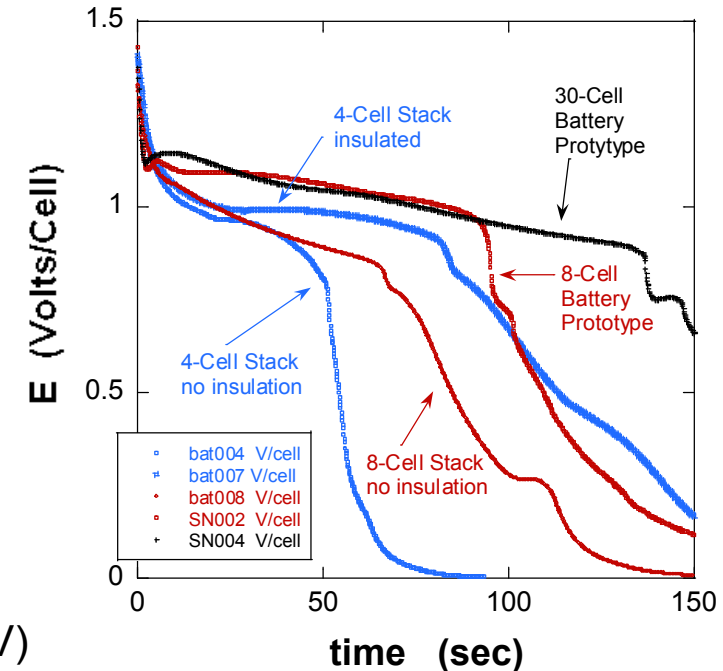
Thin Film Thermal Battery (TFTB) Prototypes



8-cell stack
1.25 inch electrode diameter

- Five TFTB prototypes fabricated and tested (1-10V, 4-36V)
- 2X improvement in energy density using pellet heat source (82% of stack)
- No gas pressure, no electrolyte leakage, no short circuits
- Pulse Power below requirements ($\sim 0.3X$)

Provisional patent application filed 12/09



36-Volt
Prototype

NNSA-Funded Ambient Temperature Lithium Battery Development

Significant lithium battery work started at SNL in the 1970's

- Li/SO₂ fielded in multiple applications
- Li/SO₂ continues to be used in JTA TM applications today
- Li/MnO₂ used in CMS applications

Li/SOCl₂ work started in late 1980's

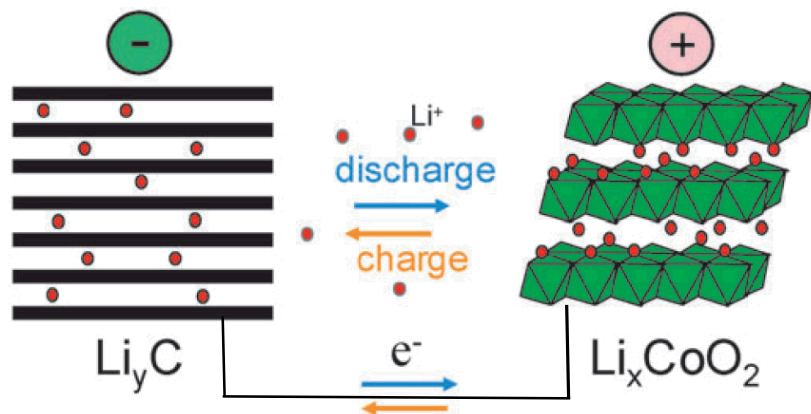
- Needed to meet -55°C requirements
- Sandia designed family of "D" cells
- Project terminated in late 1990's
- No funding to continue characterization of cells

LEP work started in 2000

- Changed direction to investigate use of commercial cells
- Used smaller cells to improve safety and reliability
- Project terminated in 2006
- No funding to continue characterization of cells
- One battery pack continued on real time test

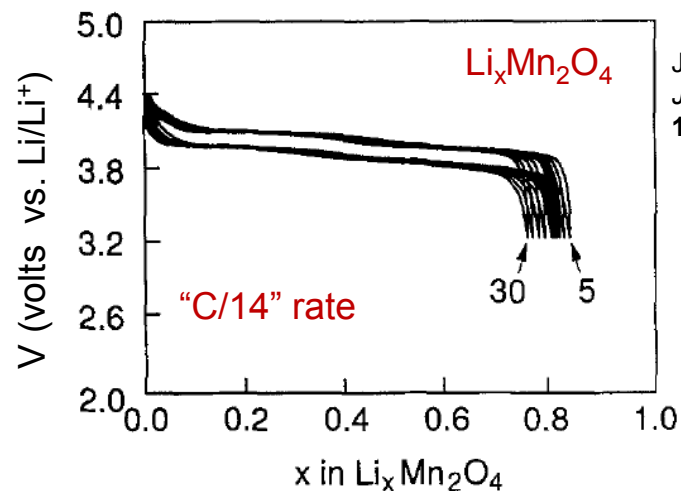


Portable Power: Lithium-ion Batteries

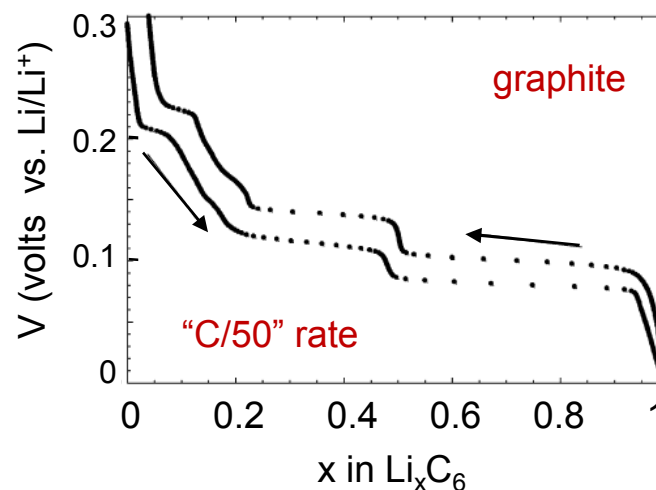


M. R. Palacín, *Chemical Society Reviews* **38**: 2565 (2009)

Typical galvanostatic voltage profiles



J. M. Tarascon et al.,
J. Electrochemical Society
138: 2859 (1991)



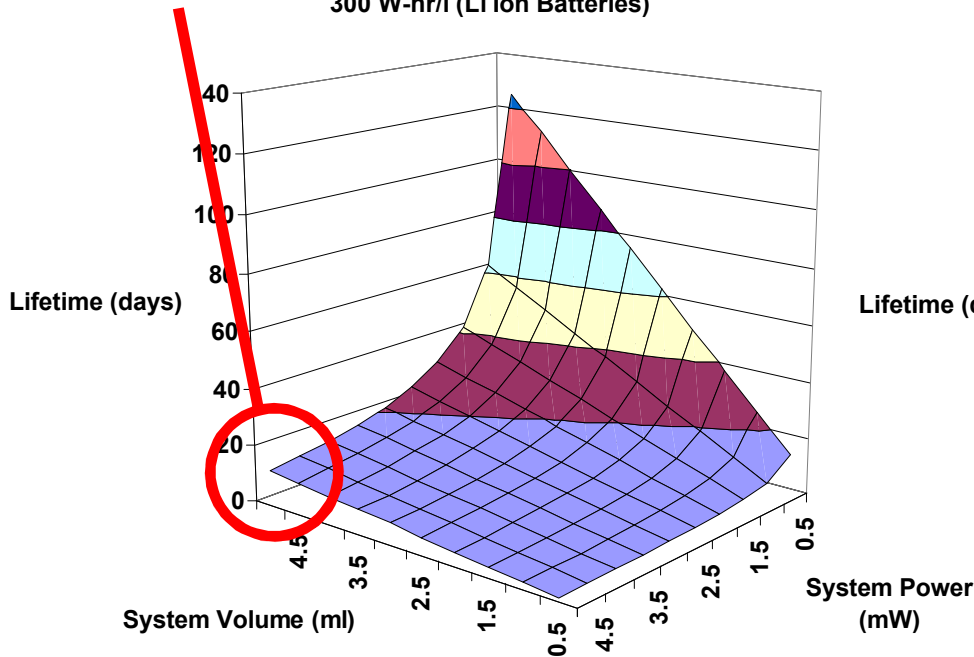
Y. Reynier et al.,
J. Power Sources
119-121: 850 (2003)

electrode	voltage (V)	capacity (mAh/g)
Li_xCoO_2	3.7	140
$\text{Li}_x\text{Mn}_2\text{O}_4$	4	120
graphite Li_xC_6	0.1	370

Lifetime and Size are Key for Low Power

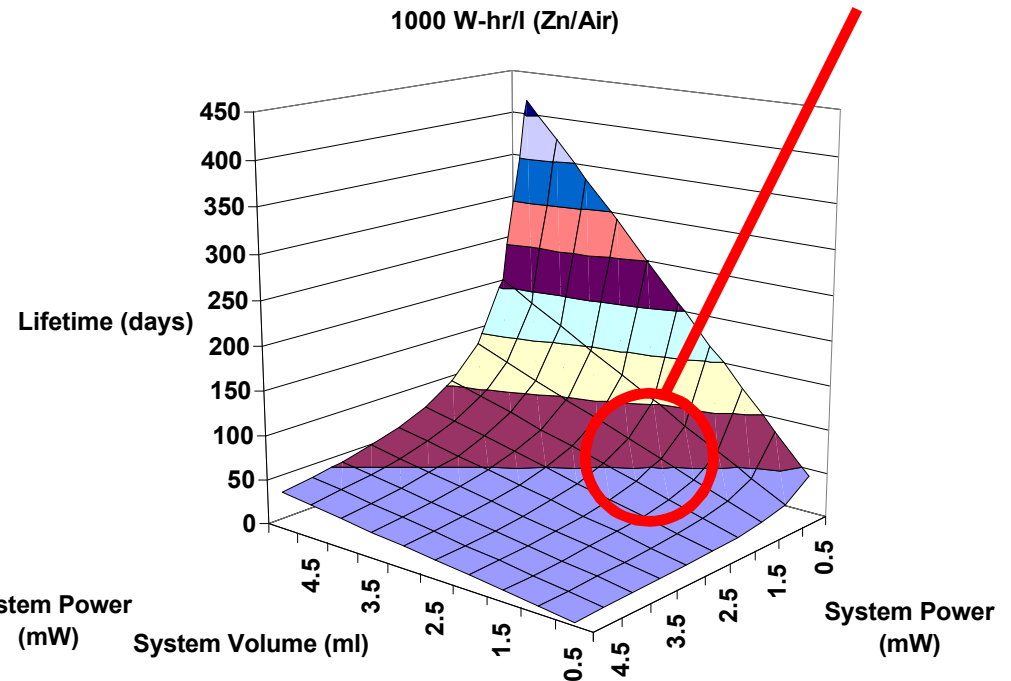
5 ml system, 5mW draw, 10 days

300 W-hr/l (Li Ion Batteries)



2 ml system, 1mW draw, 50 days

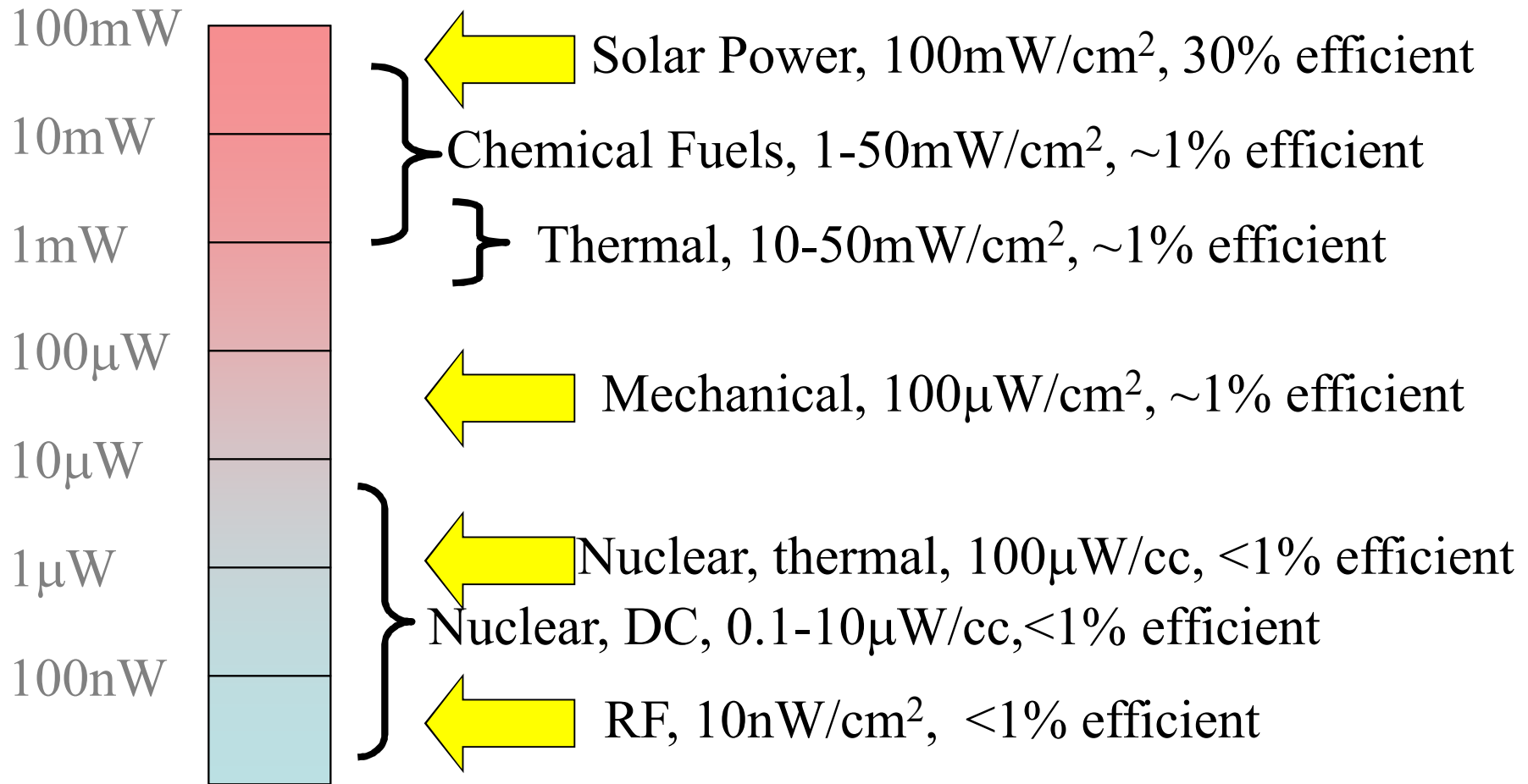
1000 W-hr/l (Zn/Air)



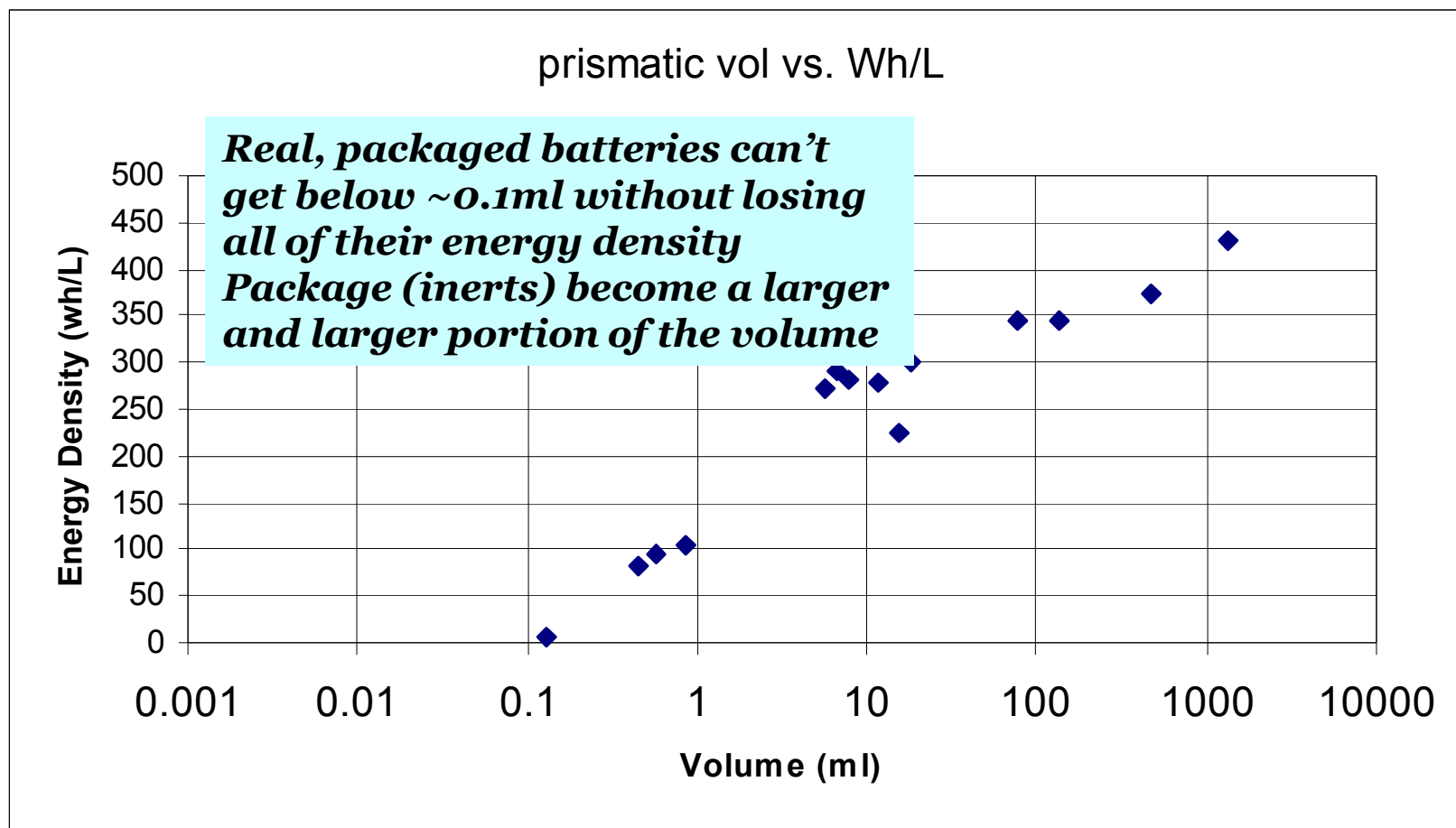
If we want to beat batteries with low power, we have to be small, and we have to last a long time

Lifetime constrained by size forces a consideration of harvesting technologies (battery chemical energy is a fixed quantity)

Environmental Scavenging: Summary

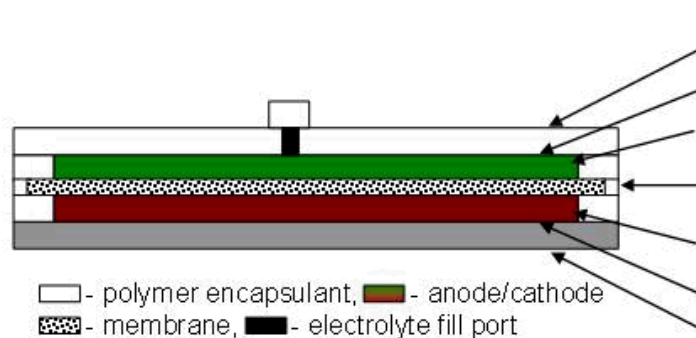
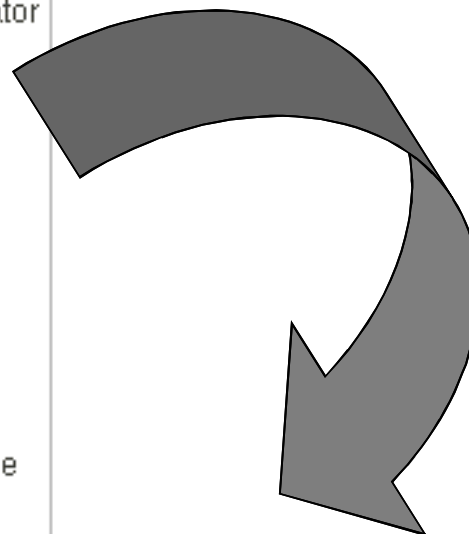
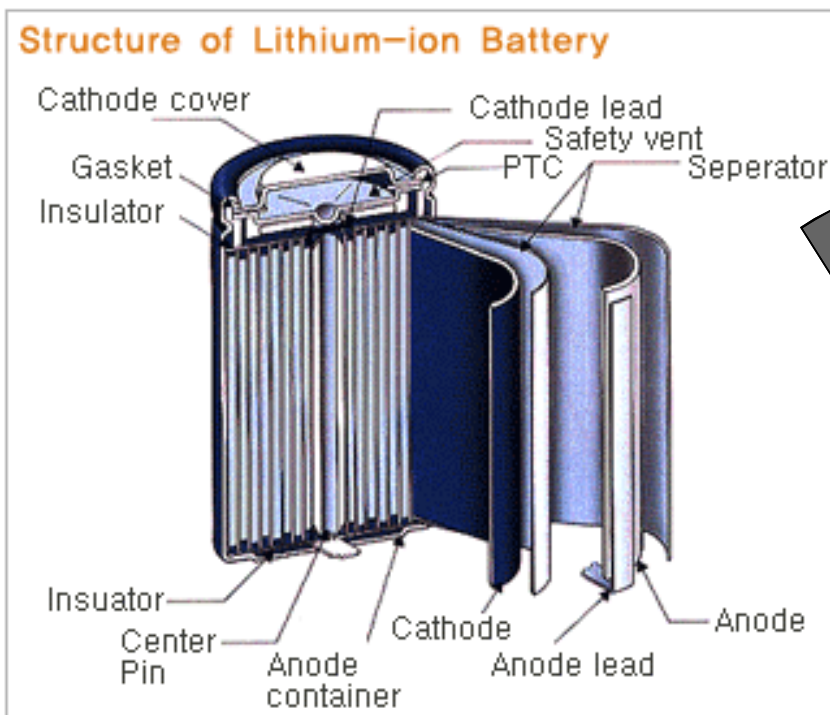


Energy Storage: Batteries



Data from Linden, Reddy, plot by E. Wang

Thinning Down Traditional Li-Ion Batteries



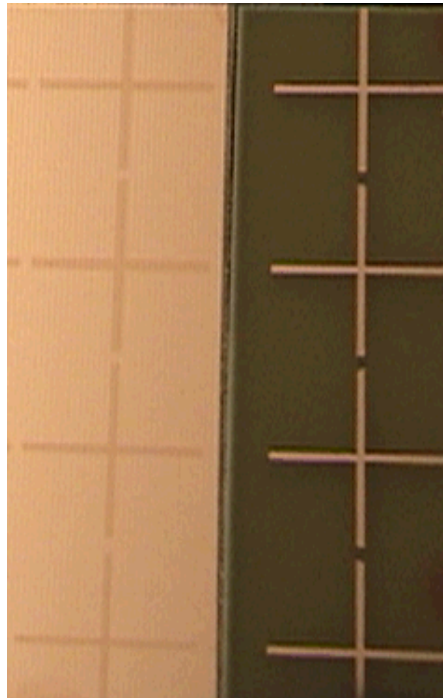
- Encapsulant (DW SU-8, UV-curable epoxy) ✓
- Current collector (DW gold nanoparticle ink) ✓
- Anode (DW graphite/carbon) ✓
- Separator (DW mesoporous SiO_2) ✓
- Cathode (DW LiFePO_4) ✓
- Current collector (DW gold nanoparticle ink) ✓
- Substrate (polyimide ✓, paper, low T polymers ✓)

Direct write enables 3D FSS, antennas, and microsystem interconnects

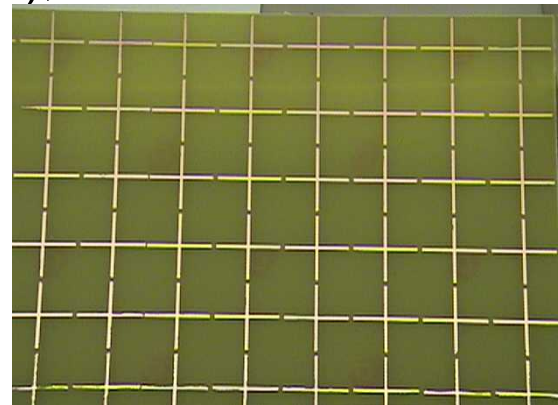
Aerosol jet on robotic x-y-z stage prints:

- etch resists on metallized polymer (subtractive), or
- silver ink on on polymers (additive),

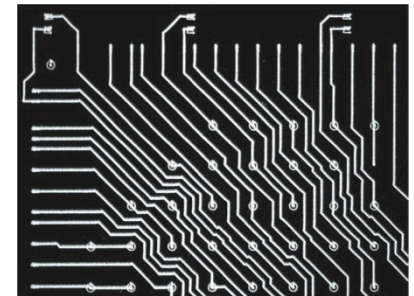
inkjetted
resist etched
Cu FSS



2D FSS

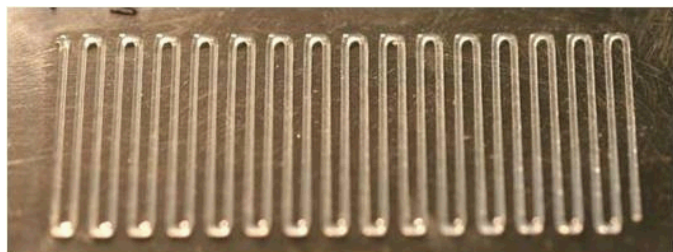
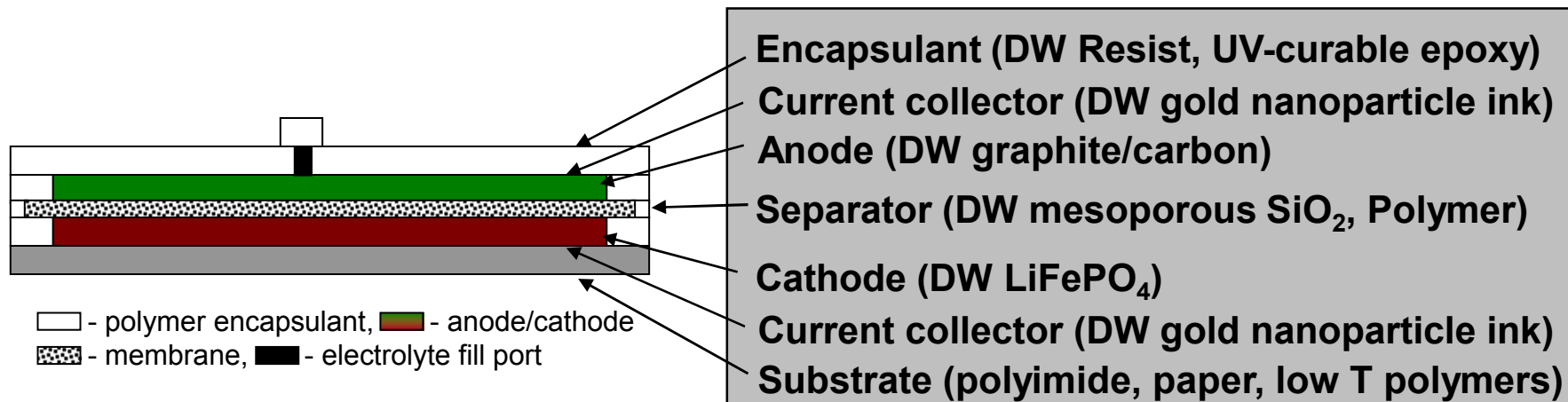


25 μm silver lines
processed at 150C:
polymer-compatible



3D FSS

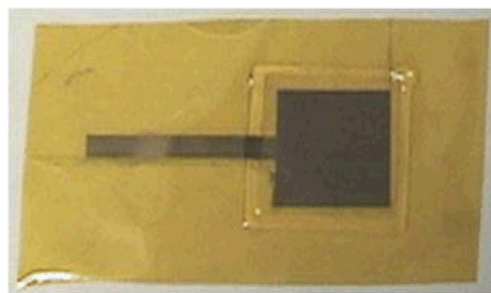
LiFePO₄ Printed Battery Using Direct Write (DW) Technology



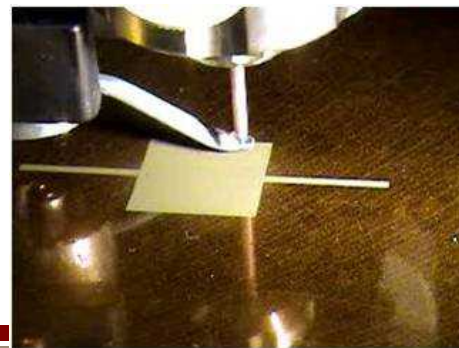
Direct written epoxy encapsulant



DW LiFePO₄ on current collector

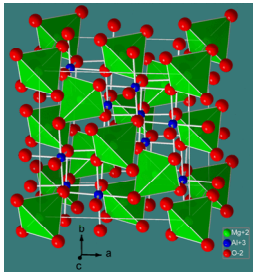


DW cathode +
encapsulant on
Kapton



DW Gold
on Kapton

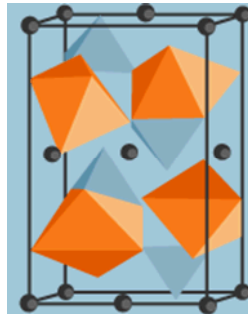
All Printed "Open Air" Cell



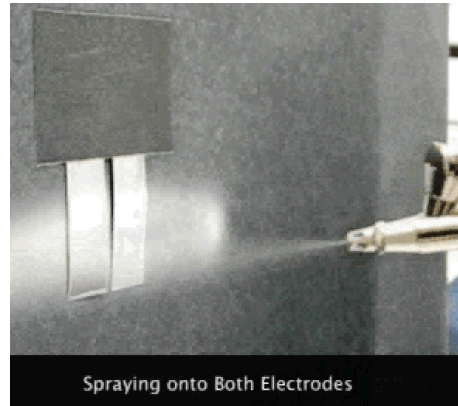
Spinel Anode
 $\text{Li}_{4/3}\text{Ti}_{5/3}\text{O}_4$

**Ionic Liquid
Electrolyte**

Li^+

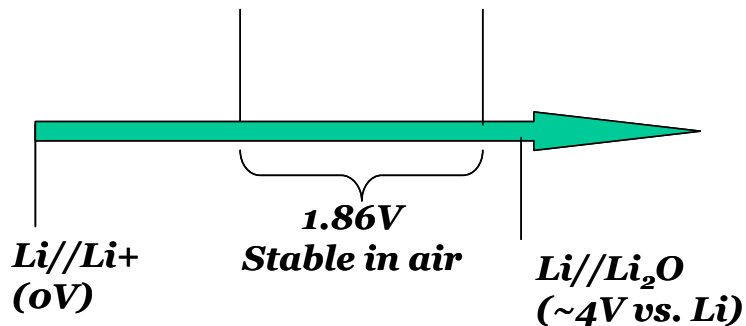


Olivine Cathode
 LiFePO_4



$\text{Li}_7\text{Ti}_5\text{O}_{12} // \text{Li}_4\text{Ti}_5\text{O}_{12}$
(~1.55V vs. Li)

$\text{FePO}_4 // \text{LiFePO}_4$
(~3.5V vs. Li)



- Allows printing in air
- Flexible anode/cathode allows bendable battery
- Good charge/discharge characteristics

