



Overview of Sandia's Hydrogen Safety, Codes and Standards Program Area

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- Hydrogen Fuel Cell Powered Electric Vehicles (FCEV)
 - Electric vehicles which use a hydrogen fuel cell to convert hydrogen to electricity
- Fuel Cell
 - “Battery with a gas tank”
 - Electrochemical device to convert chemical energy to electrical energy
- Fueling
 - High pressure hydrogen (5,000 or 10,000psi)
 - Fueling Time is 3-5 min

More information:

[University of Washington Program](#)

[US Department of Energy Program](#)

[Sandia Safety, Codes and Standards](#)



Common uses of hydrogen:

- 9 million tons annual in US (56 Bkg global)
- Oil Refinery
 - 60% of current hydrogen use
 - “Sweetens” and removes sulfur
 - 21 Million Fuel Cell Vehicles (FCVs) equivalent
- Food – partially-hydrogenated fats
- Electrical power equipment coolant
- Ultra-clean heating ovens – metals, silicon wafers, etc.

Not yet common use:

- Personal vehicles, private residences
- 12,000 hydrogen stations would put hydrogen within two miles of 70 percent of the U.S. population



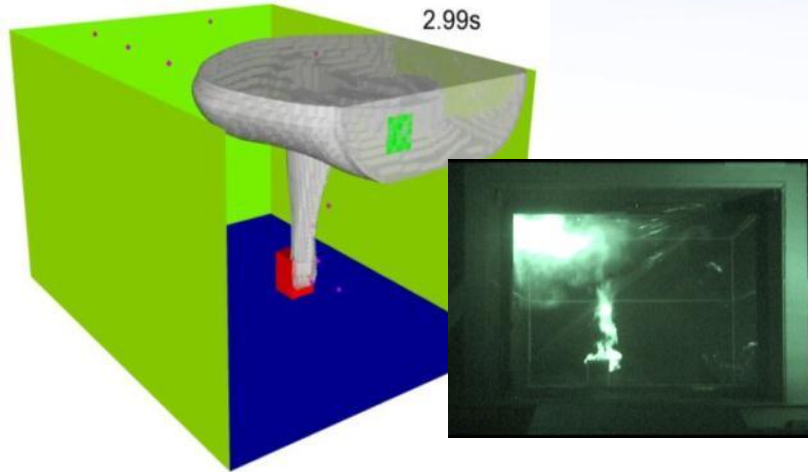


- Laboratory Mission Area: Energy, Climate & Infrastructure Security
 - Transportation Energy Center
 - Combustion and Technology Branch
 - Hydrogen & Combustion Technology Program
 - Hydrogen Safety, Codes and Standards R&D

Hydrogen Safety, Codes and Standards R&D Goals:

- Eliminate barriers to access and availability of safety data and information.
- Enable market development with consistent regulations, codes and standards.
- Provide technical data for creation and revision of regulations, codes and standards.

Hydrogen behavior



Simulation and experimental validation of release during indoor refueling

H₂ effects in materials, components, and systems

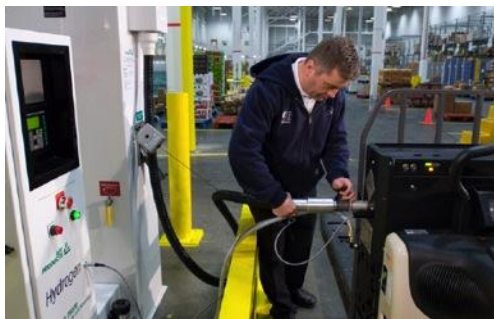


Technical Reference

Table of Contents			
Introduction	Material composition	Code	Section
Introduction		8125	(2/05)
Basic Carbon Ferrous Steels			
C-Mn Alloys	Fe-C-Mn	1300	(6/07)
Low-Alloy Ferrous Steels			
Quenched & Tempered Steels	Fe-C-Mn	1011	(12/05)
Cr-Mn Alloys	Fe-Mn-Cr-Mn	1012	(12/05)
Fe-Cr-Mn Alloys	Fe-Mn-Cr-Mn	1013	(12/05)
High-Alloy Ferrous Steels			
High-Strength Steels	Fe-Mn-Cr-Ni	1001	(10/06)
Fe-Cr	Fe-Mn-Cr-Ni	1002	(10/06)
Ferritic Stainless Steels	Fe-18Cr	1003	(10/06)
Duplex Stainless Steels	Fe-22Cr-5Ni-Mo	1004	(8/05)
Semi-Austenitic Stainless Steels	Fe-15Cr-7Ni	1005	(3/08)
Martensitic Stainless Steels	Fe-Cr-Ni	1006	(3/08)
Precipitation-Strengthened Heat Treatable	Fe-Cr	1007	(6/05)
Austenitic Steels			
300-Series Stainless Alloys	Fe-19Cr-10Ni	1008	(5/05)
Type 304 & 304L	Fe-18Cr-10Ni-Mo	1009	(3/05)
Type 316 & 316L	Fe-16Cr-10Ni-Mo	1010	(10/05)
Type 321 & 321H	Fe-18Cr-10Ni-Ti	1011	(10/05)

Mechanical load-frame used to characterize H₂ effects in materials

Quantitative Risk Assessments



Quantitative Risk Assessment helps establish requirements for hydrogen installations

C&S development support



Regulations Codes and Standards Advocacy

Risk-Informed Approach

Use validated simulations, field data and expert input to determine risk through quantitative risk assessment.

Release Probability

- Permeation
- Buoyant creeping flow
- Turbulent jet
- Volumetric rupture

Field Data Input

- incident data,
- environmental/human factors,
- system design/mitigation

Ignition Probability

- Ignition mechanism
- Mixture ignitability
- Ignition delay/location
- Sustained light-up

Hazard Probability

- Flame radiation
- Pressure wave (deflagration/detonation)
- O₂ dilution/depletion

Informed Input

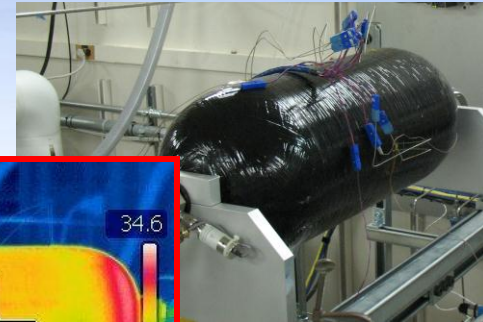
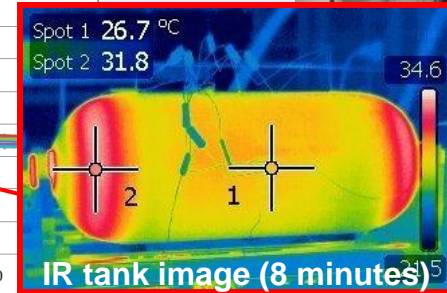
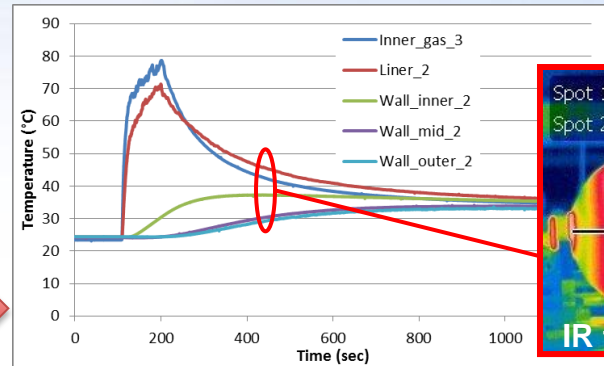
Code development groups, industry, regulators and code enforcers

Harm Probability

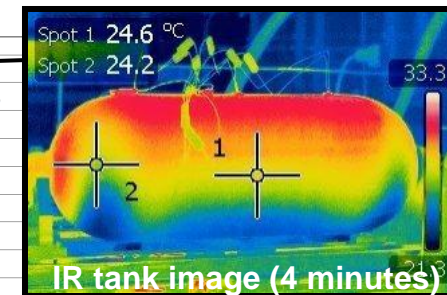
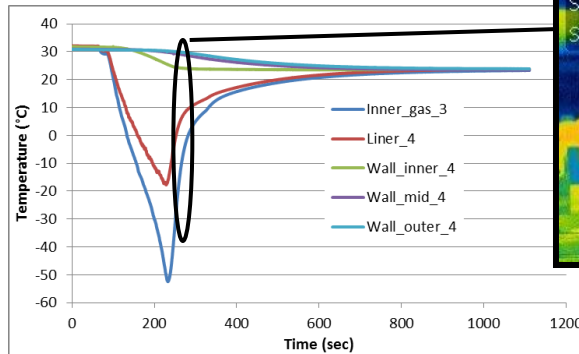
- Burns
- Lung damage
- Shrapnel wounds
- Building collapse

QRA

Test #	Initial pressure (psi)	Final pressure (psi)	Fill time (sec)
1	20	1500	60
2	20	1500	90
3	20	1500	150
4	20	1500	300
5	20	1000	90
6	20	2000	90
7	145	2000	90
8	290	2000	90
9	725	2000	90

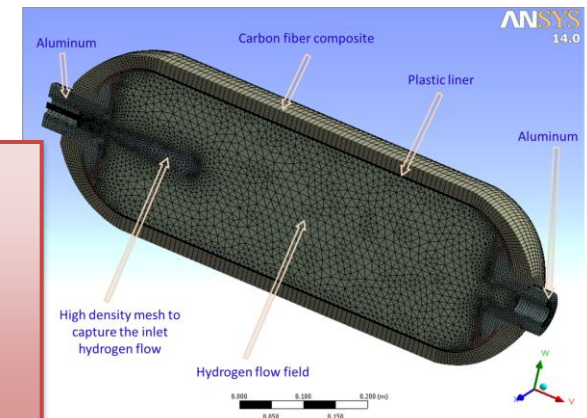


Test #	Initial pressure (psi)	Final pressure (psi)	Release rate (g/sec)
1	1300	20	1.5
2	1300	20	0.75
3	1300	20	1.0
4	1300	20	1.9
5	1700	20	0.4
6	1700	20	1.5
7	1700	20	0.75



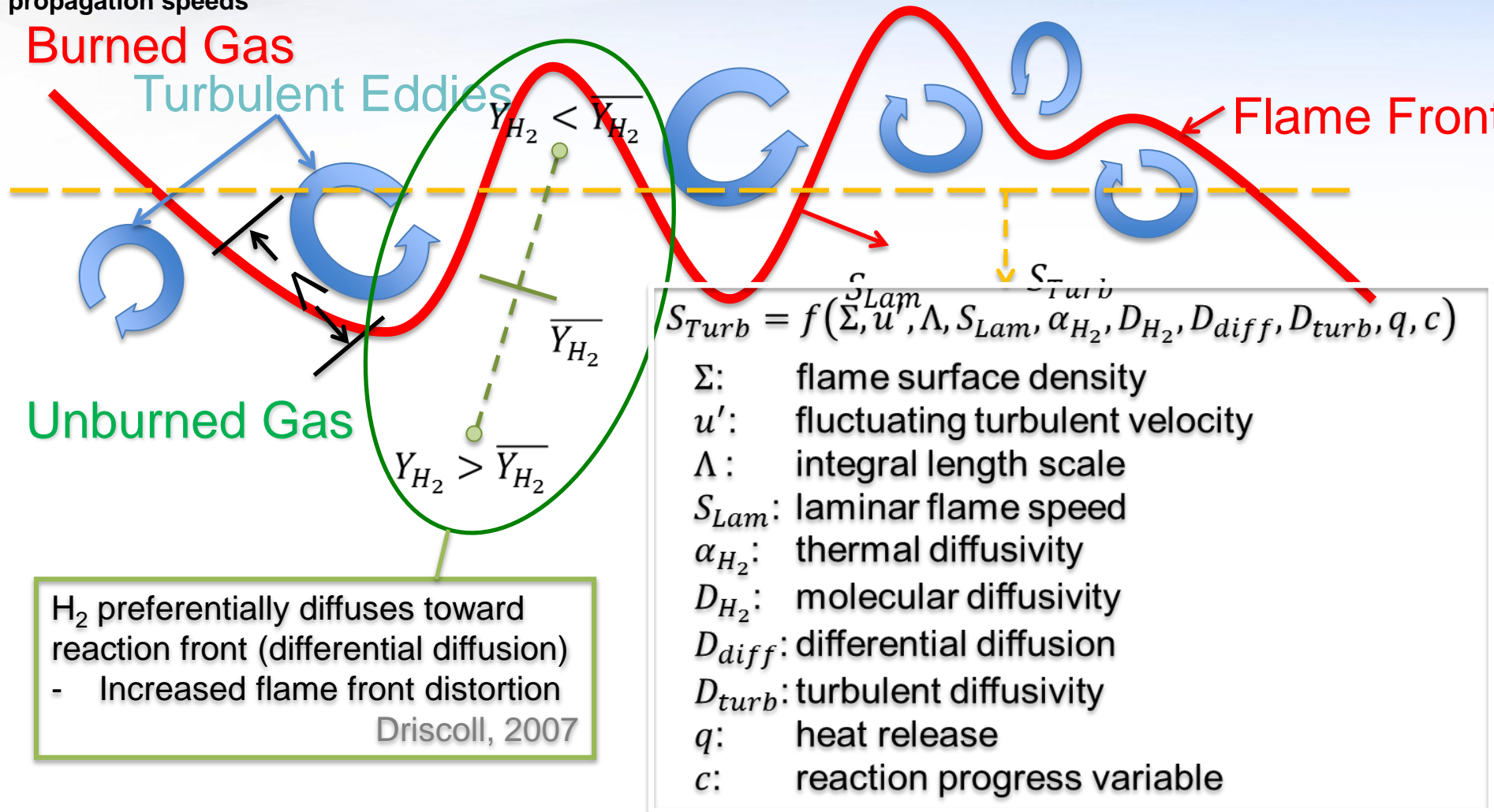
Model Results:

- ✓ Model completed, currently in validation
- ✓ Goal: results available for SAE Interface Group (J2601) discussion; Sept 2012
- ✓ Goal: Comparison with other research and industrial datasets in support of 2015 vehicle deployment



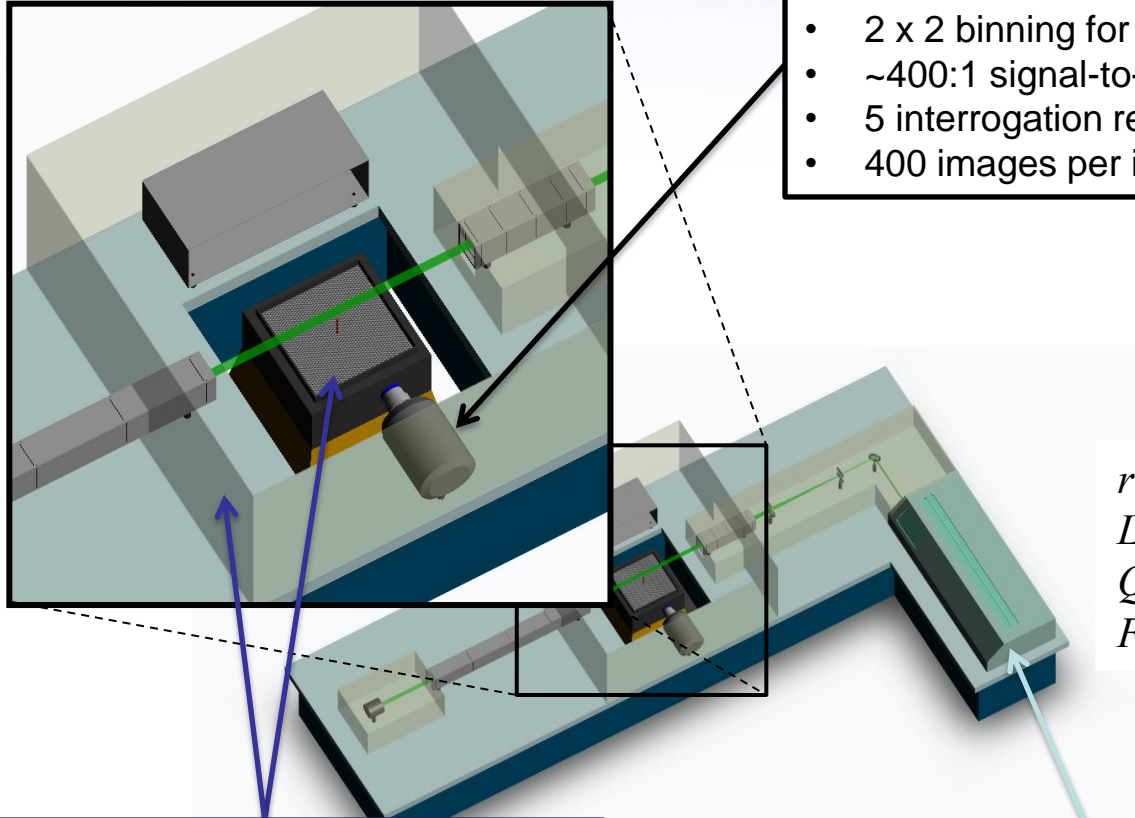
Improve Mixture Measurement

Flamelet models can be used to predict these turbulent flame front propagation speeds



Turbulent Jet Experiment

Scalar field of a momentum driven, turbulent H₂ jet was examined via high-resolution Planar Rayleigh Scatter Imaging (PLRS).



PIXIS 400B low noise CCD Camera

- 2 x 2 binning for 3.94 pix/mm resolution
- ~400:1 signal-to-noise
- 5 interrogation regions (37 x 125 mm²)
- 400 images per interrogation region

$$\begin{aligned} r_0 &= 0.95 \text{ mm} \\ L_{\text{pipe}} &= 250 \text{ mm} \\ Q &= 100 \text{ lit/min} \\ Fr_{\text{den}} &= 1170 \end{aligned}$$

Air co-flow & barriers to minimize impact of room currents

High power injection seeded ND: Yag laser (1 J/pulse, 532 nm)

Additional diagnostics include Particle Image velocimetry (PIV), Laser Doppler Velocimetry (LDV) and OH Laser Induced Fluorescence (LIF).



Flame Light-Up Imaging

Ignition w/ light-up

Rapid volumetric ignition kernel growth

Kernel forms into turbulent flame front

Flame front broadens

Front **overcomes** flow convection

Ignition w/o light-up

Slower kernel growth

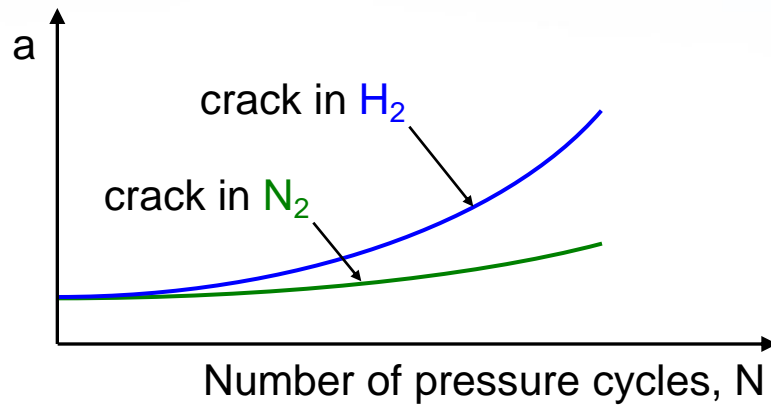
Thinner, less turbulent flame front

Front **overcome** by convection

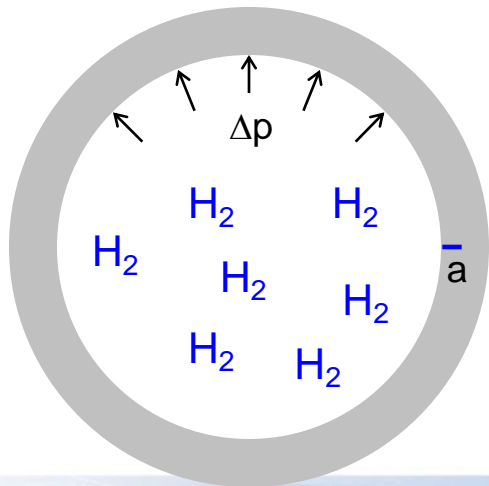
- ✓ Imaging observations indicate sustained flame light-up correlates with 1D flame front propagation speeds.

Safety qualification must consider hydrogen embrittlement of metal components

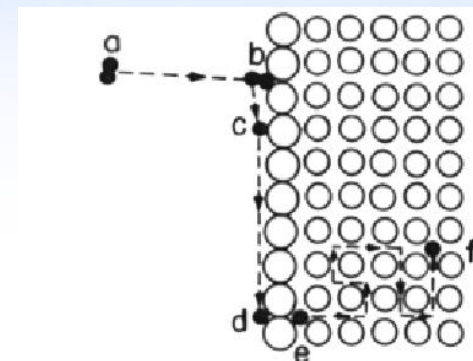
Hydrogen embrittlement accelerates fatigue cracking



$$N_f = N_{\text{initiation}} + N_{\text{growth}}$$



- Hydrogen is the smallest element
- Hydrogen 'interrupts' the metal structure
- Material becomes more brittle
- Hydrogen affects are different for different types or classes of metals
- Sandia has an enduring program in hydrogen effects in materials for the national defense mission area
- This expertise was used to provide assistance to this investigation



HYDROGEN MOLECULE
 HYDROGEN ATOM
 FERROUS ATOM

REACTION STEPS

- a → b Molecular physisorption
 b → c Dissociation
 c → d Adatom migration and chemisorption
 d → e Solution
 e → f Lattice diffusion

AC Transit Emeryville, CA Incident Summary



- Hydrogen release and subsequent jet fire from vent system
- No significant damage, injuries or fatalities

- Root Causes – improper material and poor quality control
- Contributing factor – inadequate communication of critical data
- Contributing factor – inadequate communication of design changes (canopy)
- Contributing factor – lack of sub-system isolation and inadequate communication of vulnerabilities during safety reviews

Questions?