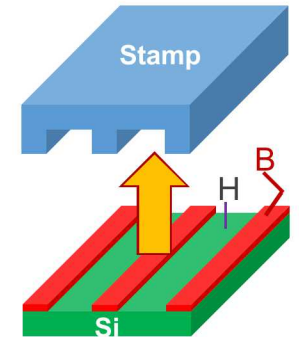
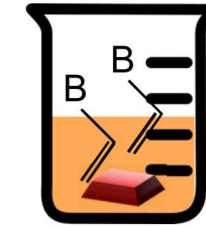
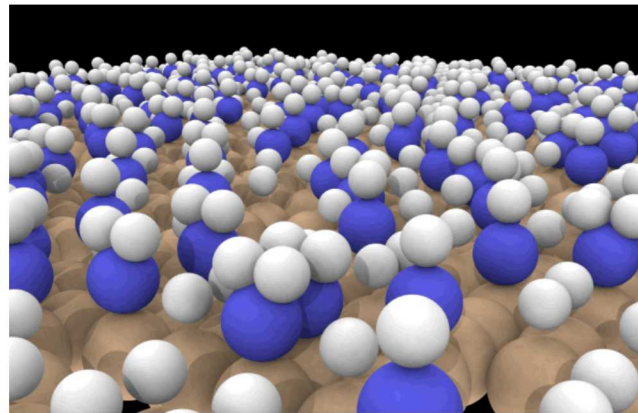
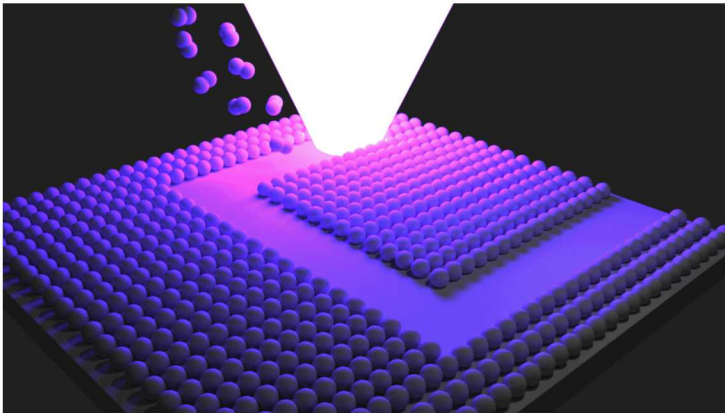


FAIR DEAL GC Thrust 4: Application Platform

FAIR DEAL GC Thrust 4: Application Platform

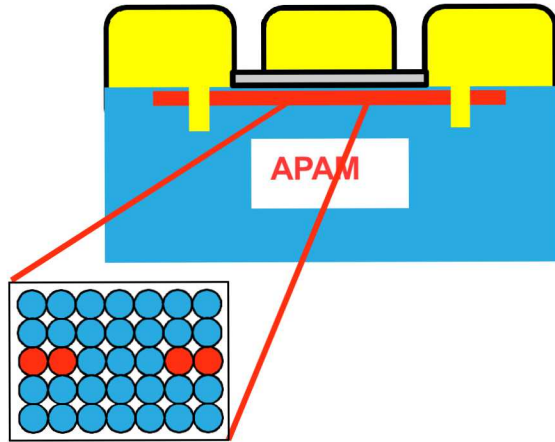


FAIR DEAL GC Thrust 4: Application Platform

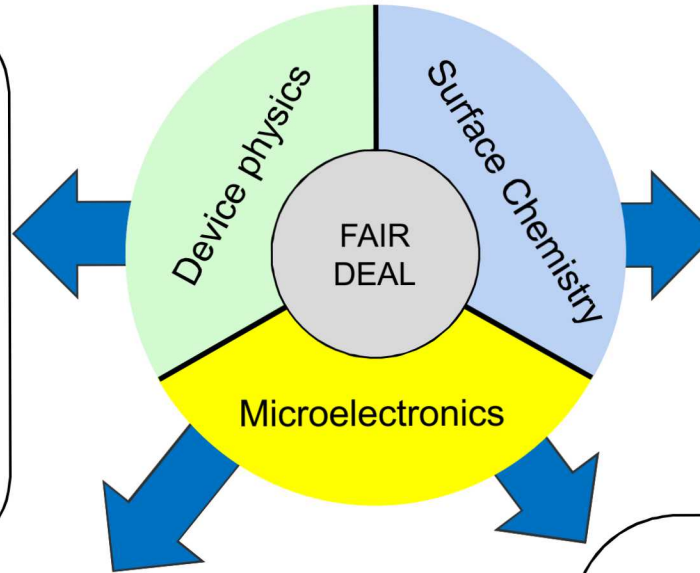
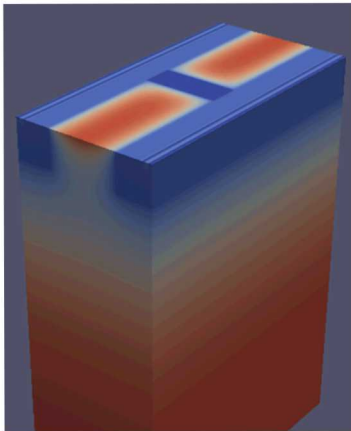
George T. Wang (Lead), Ezra Bussmann, Robert Butera,
Aaron Katzenmeyer

Digital electronics at the atomic limit (DEAL)

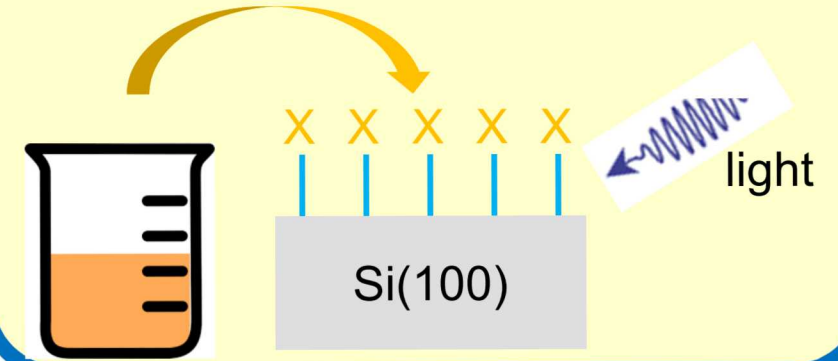
Thrust 1: APAM-enabled Devices



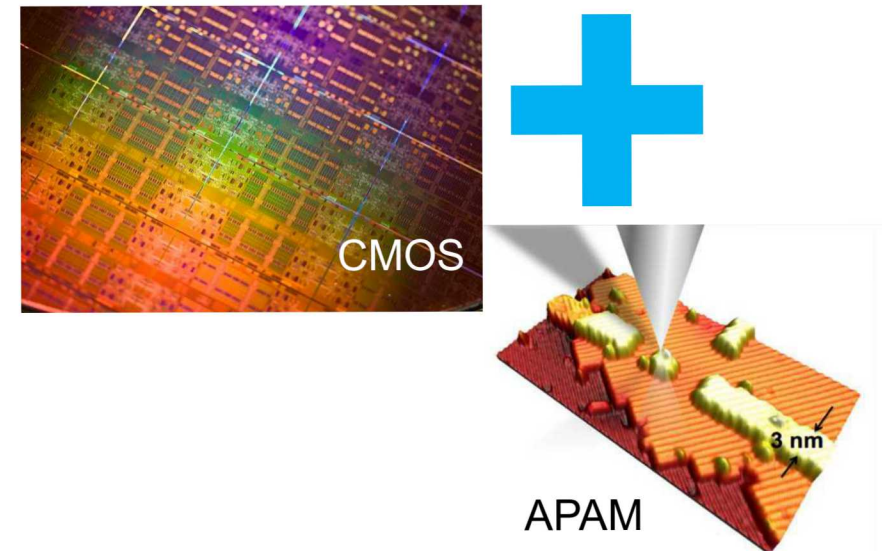
Thrust 2: APAM Modeling



Thrust 4: Application Platform

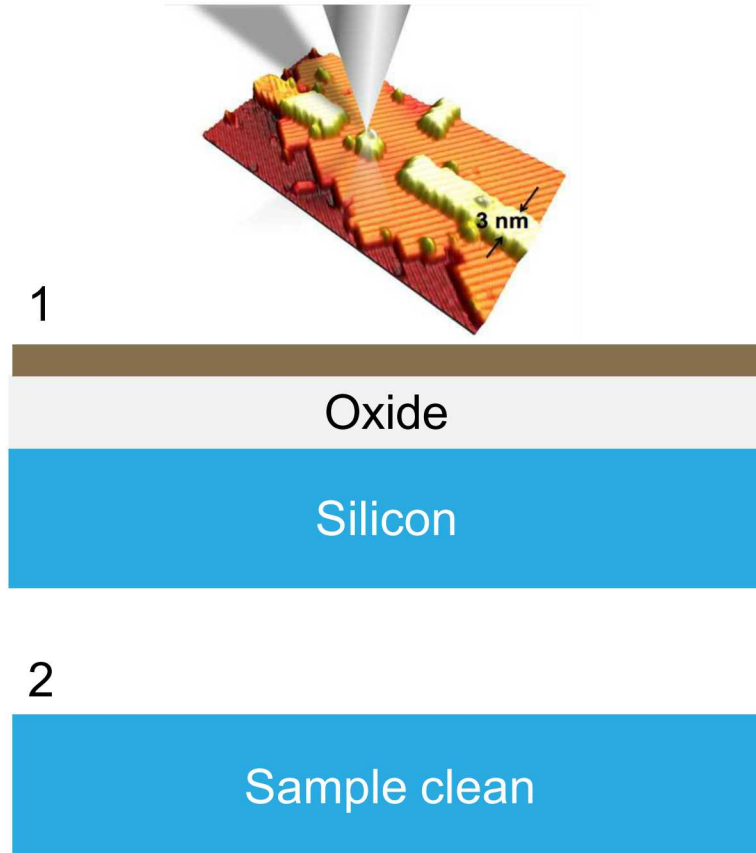


Thrust 3: CMOS Integration

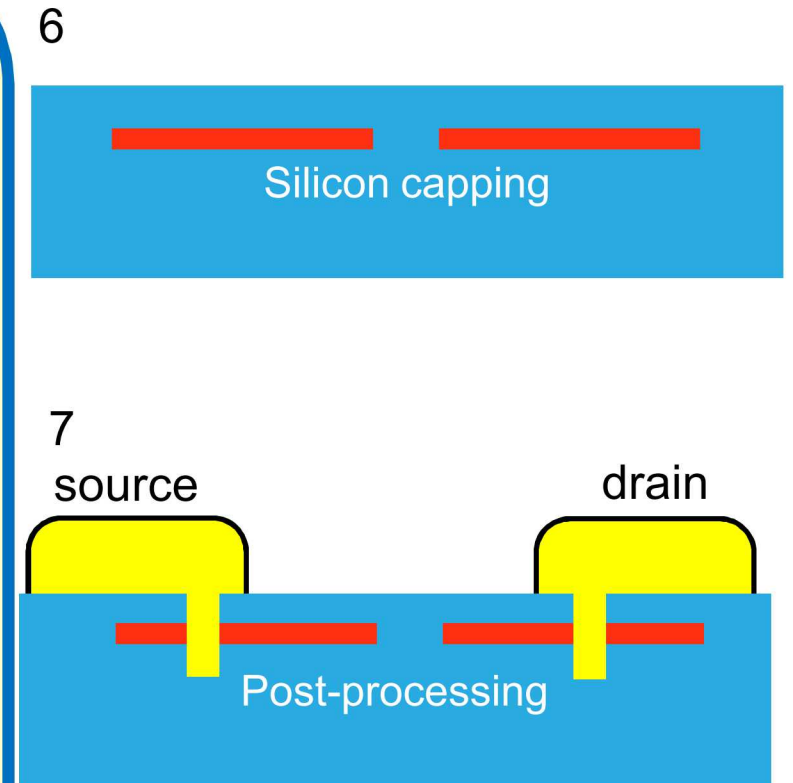
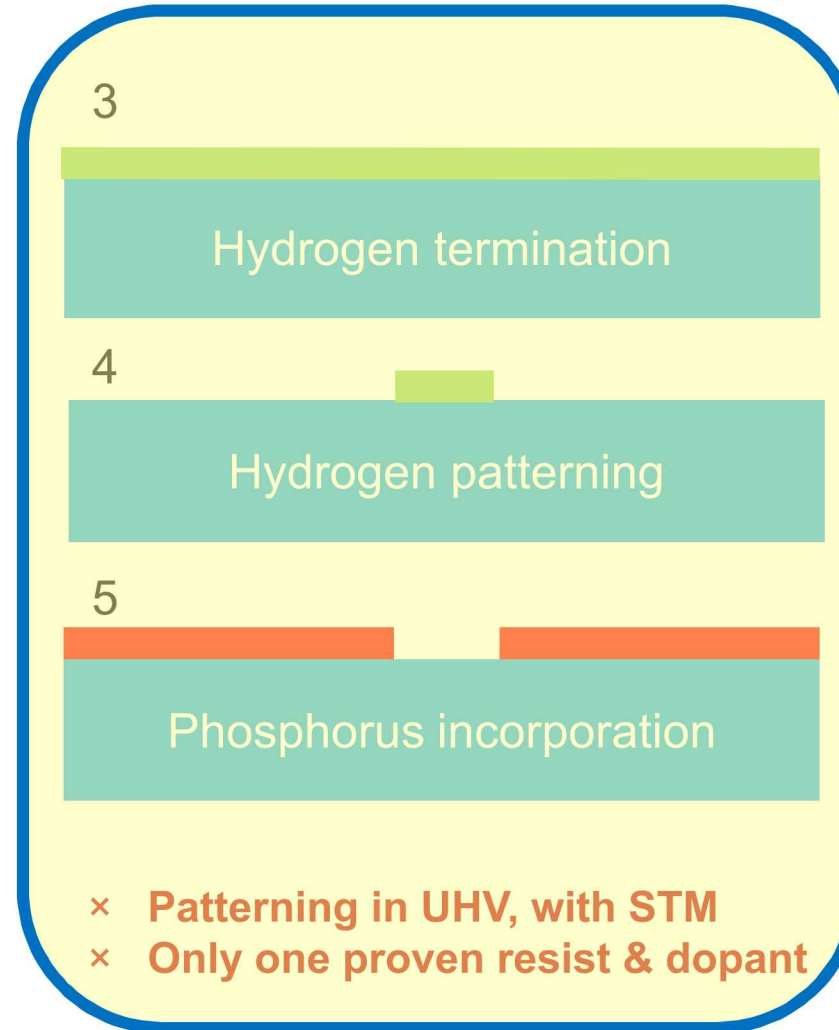


Limitations of APAM state of the art

Application Platform



× Remove oxide: $T > 800^{\circ}\text{C}$



× Limit diffusion: $T < 450^{\circ}\text{C}$

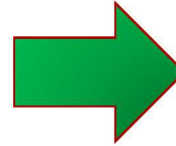
× Cryogenic operation only

Problems span **surface chemistry**, device physics, microelectronics

Current APAM : STM/UHV *Learning Platform*



- **Only one lithography pathway: Slow** STM done in extreme environment (**UHV**)
- **Only one** resist (H) + dopant (PH₃) combination in practice -> **limited toolkit & devices**



APAM “*Application Platform*”

Rapid lithography

New atomic resists and dopants

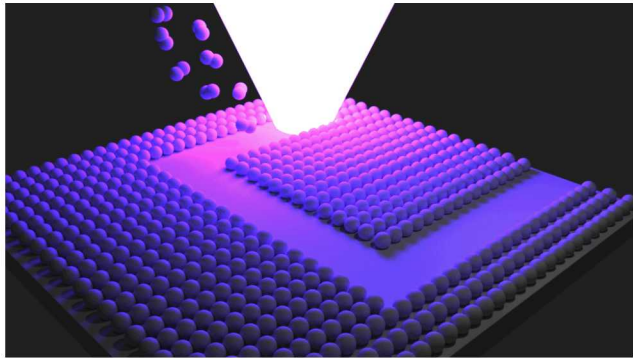
RT, Non-UHV, processes and chemistries

Discovery science to enable new principles & toolkits for atomic scale processes

Thrust 4: APAM Application Platform

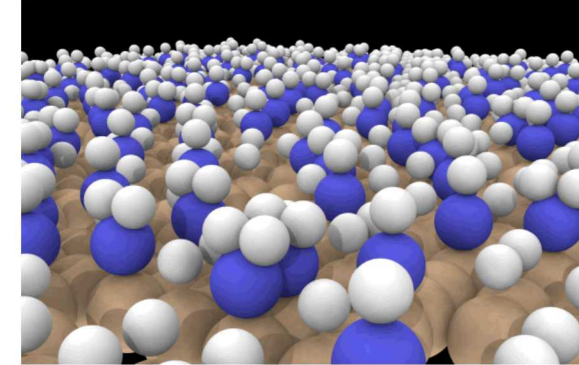
1. Photolithography

Scalable, light-based patterning of atomic resists



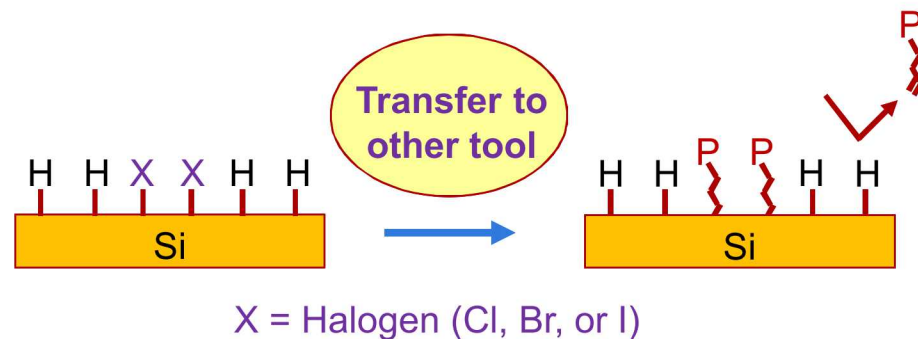
2. Acceptor Doping

p-type APAM processes



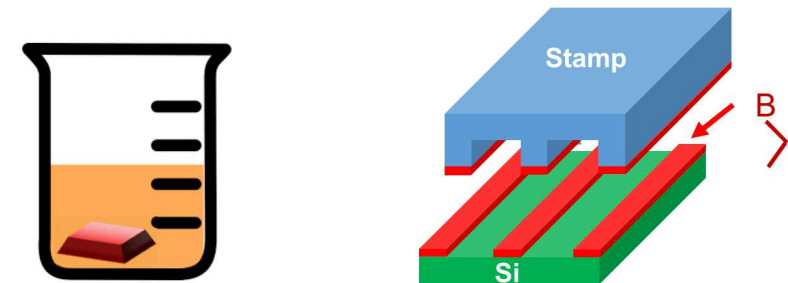
3. Alternative Resists

Halogen resists for pattern preservation

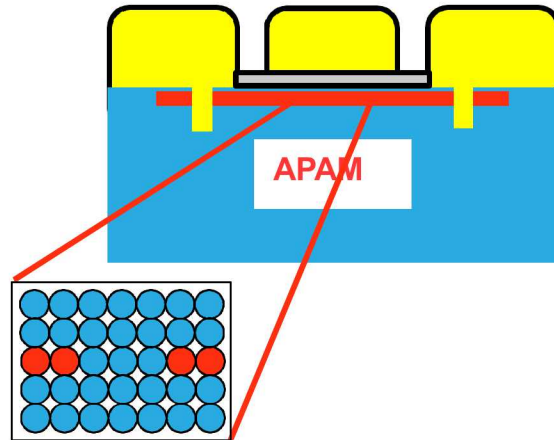


4. Wet Chemistries

Robust and scalable wet surface chemistries for resists and doping



Application Platform - Ties to Exemplars



APAM-MOS:

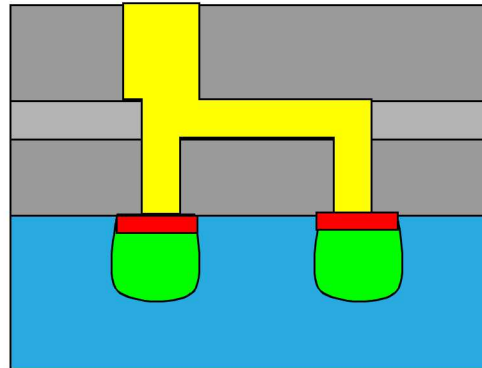
Need:

- Photolithography
- Acceptor doping:
complementary transistors

Interconnects:

Need:

Acceptor doping:
opposite polarity possible

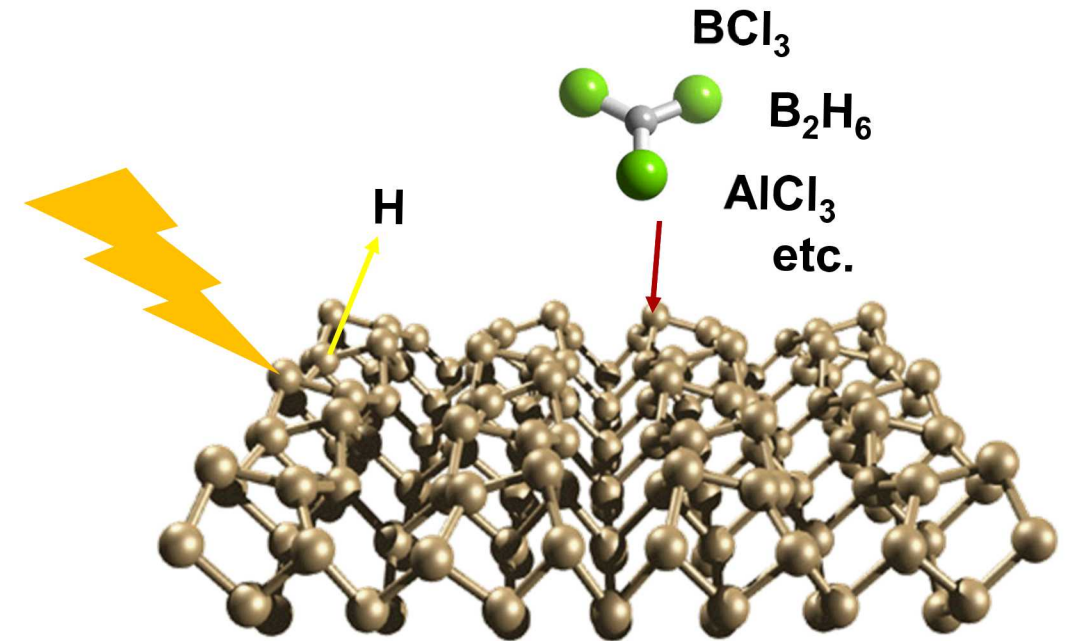


Fingerprint:

Need:

- Photolithography: need
large scale patterning
- Acceptor doping
- Wet Chemistry

Why is this hard?



Task	Challenge
Photolithography	Desorb H without damaging Si
Acceptor Doping	Discover new surface chemistries that give desired electrical properties
Alternative Resists	Unknown stability and reactivity
Wet Chemistry	Discover new surface chemistries that give desired electrical properties

Application Platform – Team Members

Thrust 4 Lead: George Wang

Photolithography: Aaron Katzenmeyer

Acceptor Doping: Ezra Bussmann

Alternative Resists: Robert Butera

Wet Chemistry: George Wang

Surface Science

George Wang
Esther Frederick (PD)
Shashank Misra
Ezra Bussmann
Scott Schmucker
Fabian Pena (PD)
Jeff Ivie

David Wheeler
Tim Lambert
Igor Kolesnichenko (PD)
Robert Butera (UMD)
Kevin Dwyer (UMD, PD)
Andrew Teplyakov (U. Del)
Alex Shestopalov (U. Roch)

Modeling

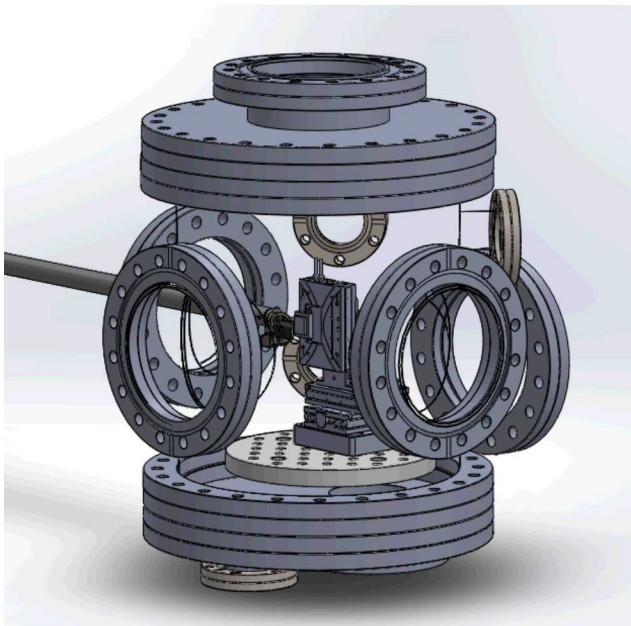
Andrew Baczewski
Quinn Campbell (PD)
Peter Schultz

Fab & Measure

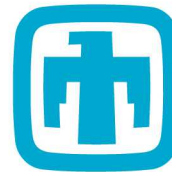
Fab
Dan Ward
DeAnna Campbell
Mark Gunter
Philip Gamache
Electrical
Lisa Tracy
Tzu-Ming Lu
Albert Grine

Application Platform – Capabilities

**Stand-alone
photolithography system**

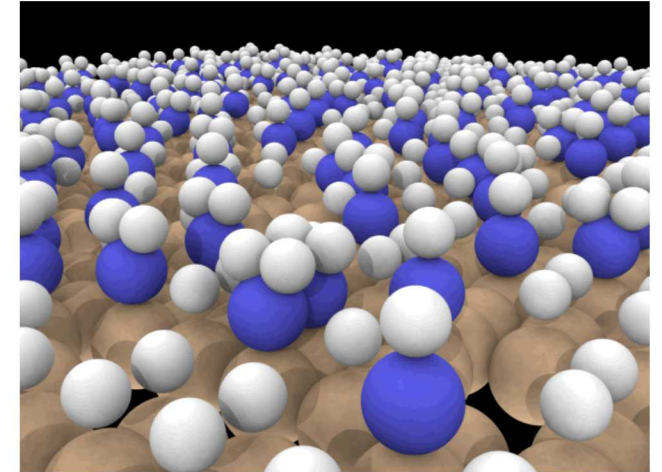


**Acceptor doping –
experimental
infrastructure in place
(Sandia, UMD, Zyvex)**



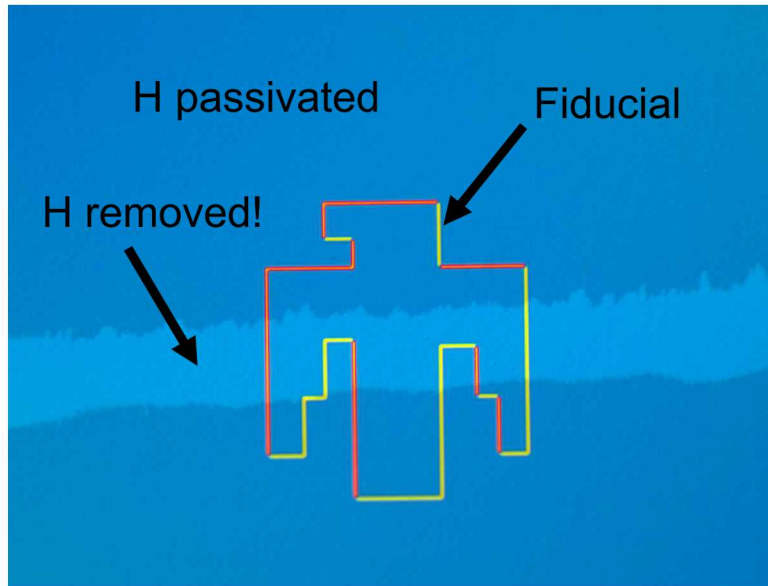
**Sandia
National
Laboratories**

**Chemistry modeling
toolkit**

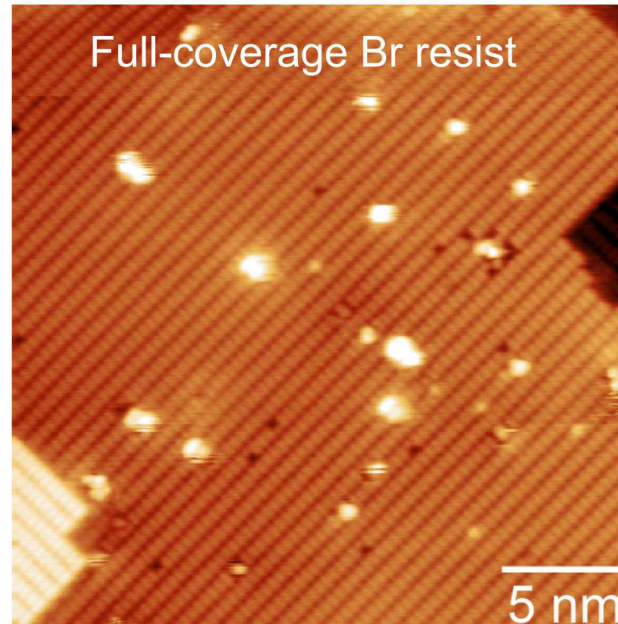


Application Platform – Discoveries

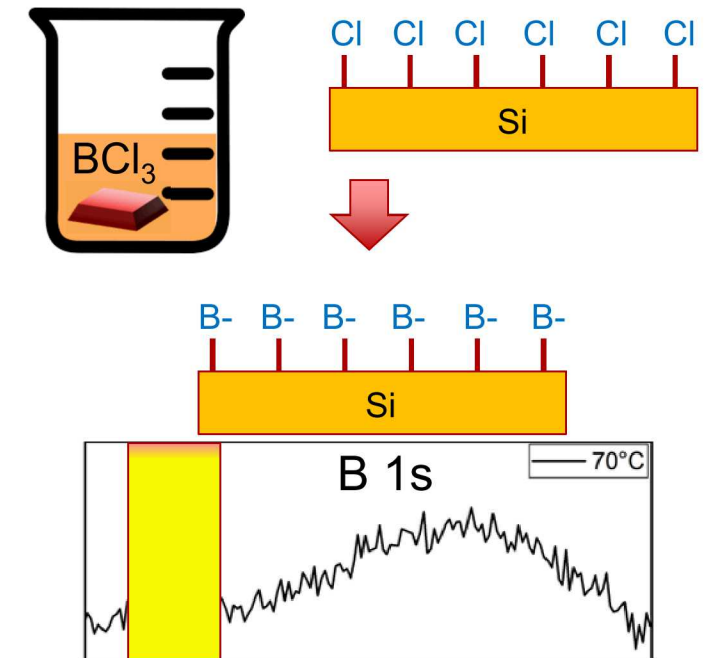
Photolithography demo: H-resist desorption by laser



New halogen (Cl, Br, I) atomic resists and stability determinations



Wet chemical pathway for direct dopant attachment



Application Platform: Metrics and external engagement

Written

- White paper submitted (wet chemistry)
- Three journal articles in preparation (Baczewski, Butera, Tepylakov)
- Technical Advance in preparation (photolithography)

Spoken

- Invited talks: DOE Workshop on Atom-by-Atom fabrication (2018, Oak Ridge NL)
- Contributed talks: NM AVS (Best talk award), APS March (2019), Int'l Conf. Electron, Ion, & Photon Beam Tech & Nanofab (2019)...

Collaborations

- Laboratory for Physical Sciences (LPS)
- University of Delaware (Teplyakov)
- University of Rochester (Shestopalov)
- Zyvex Labs



Application Platform – current risk assessment

Successful demonstration!

Strategy refocused

Critical Path

Photo-lithography

Alt. Resists & Dopants

Wet Chemistry

Acceptor Doping

A

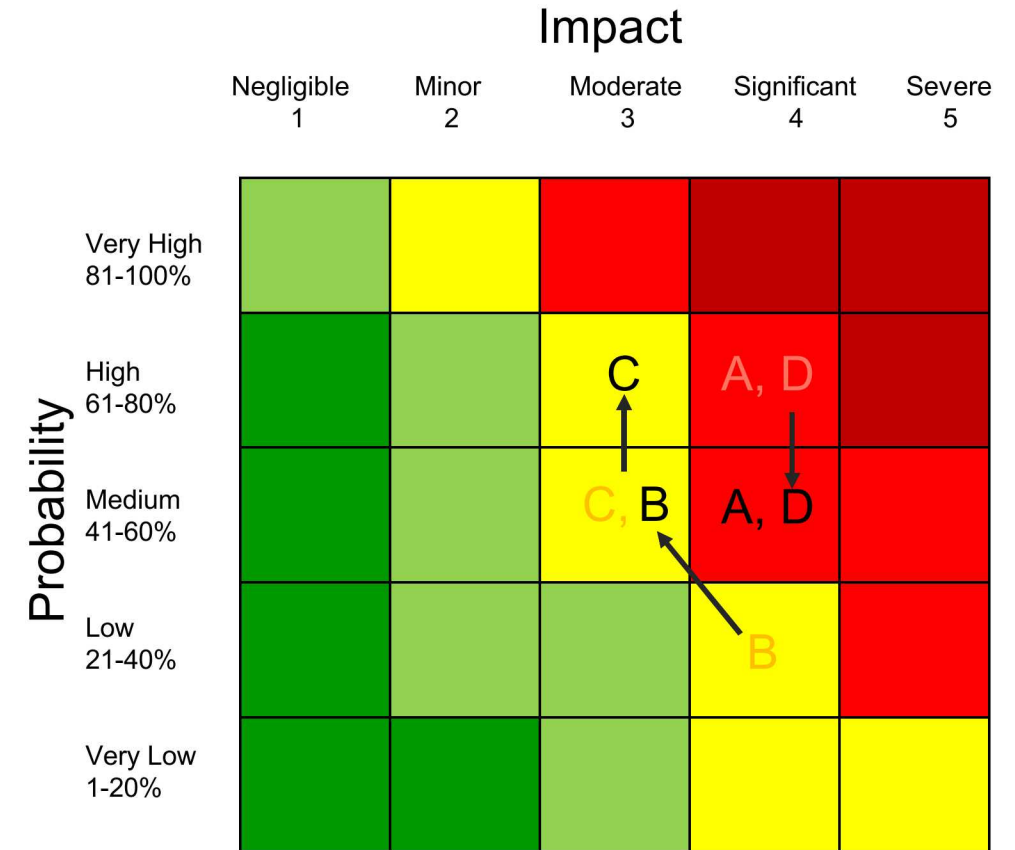
B

C

D

Question answered, merge with Wet Chemistry

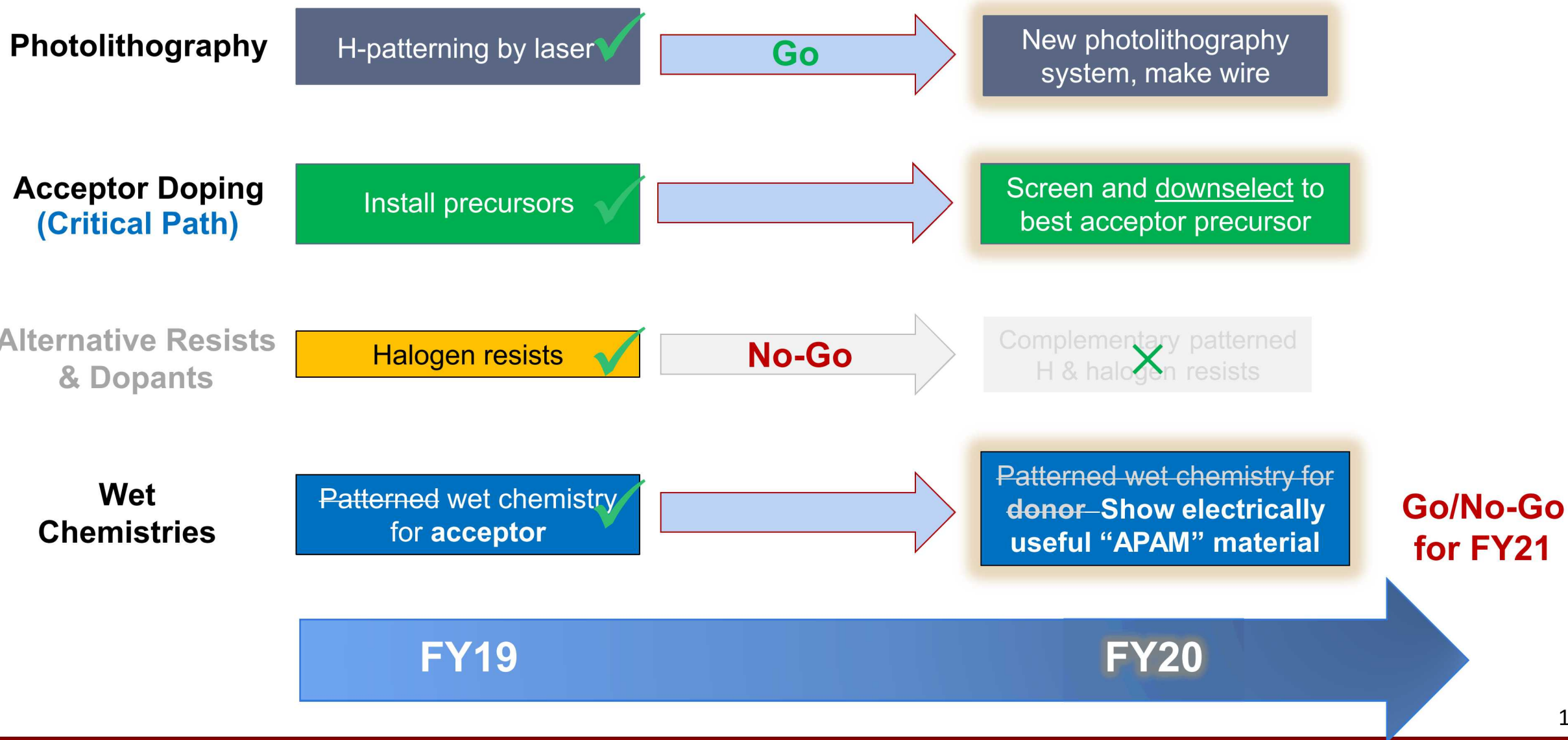
Additional precursors lowers risk



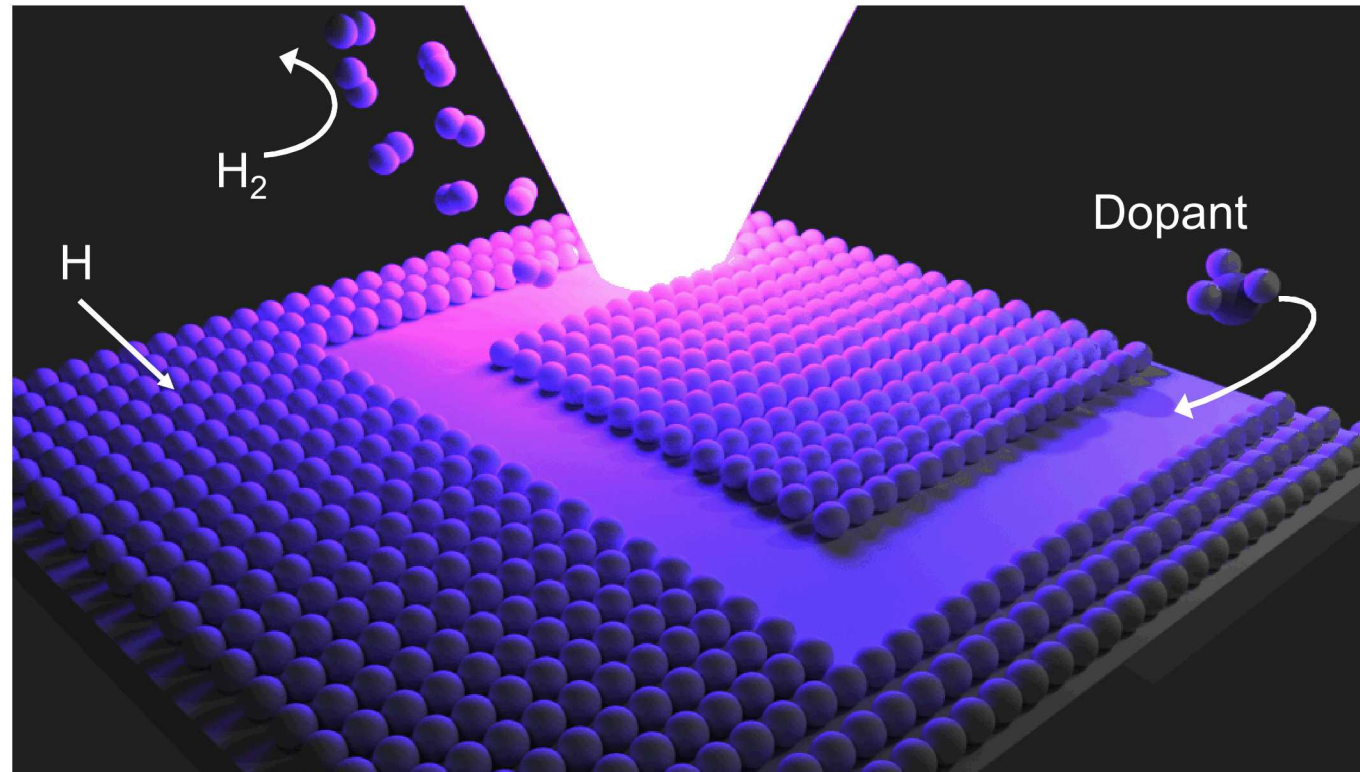
Color key:

- Dark Green – Minimum Risk
- Light Green – Low Risk
- Yellow – Moderate Risk
- Red – High Risk
- Dark Red – Extreme Risk

Thrust 4 Roadmap - FY19 Progress to FY20 Plan



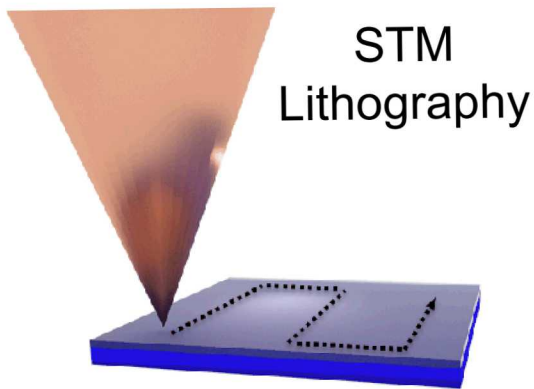
Task 1 – Photolithography



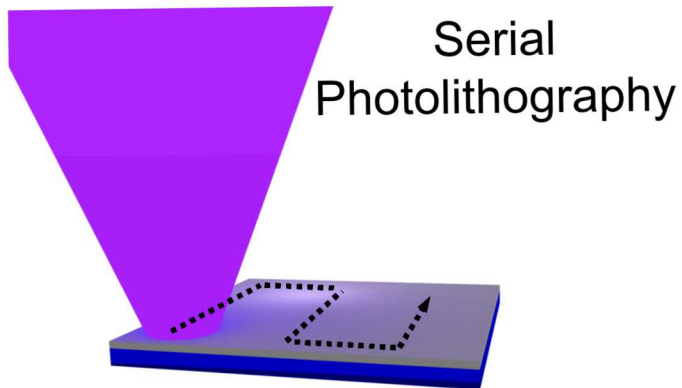
*Expedites **progress** throughout GC*
*Enables **basic science** and innovation*

Photolithography vs STM Patterning

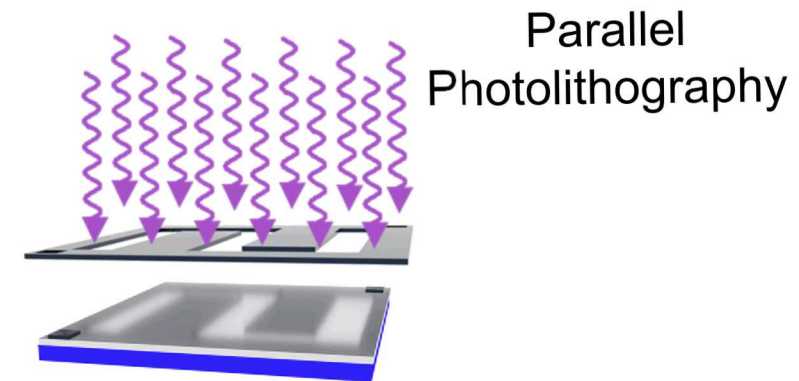
Patterning Method	Atomic Resolution	Large Areas	Rapid
Photolithography	x	✓	✓
STM	✓	x	✗



1 device / day



10 devices / day

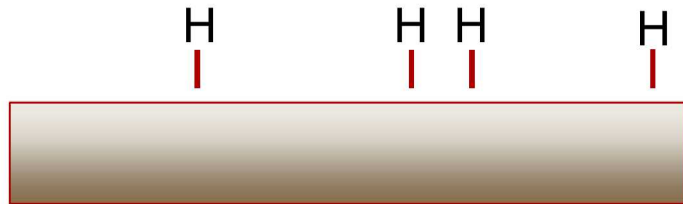


10^9 devices / day

Photolithography - Risks

A. Physical mechanism of H photodesorption not understood
Conflicting literature

B. Quality of device as good as STM?



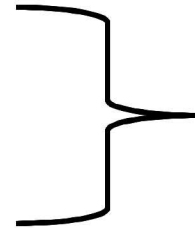
C. Silicon surface damaged by process
Catastrophic to fabrication



		Impact				
		Negligible 1	Minor 2	Moderate 3	Significant 4	Severe 5
Probability	Very High 81-100%			A		
	High 61-80%			↓		
	Medium 41-60%			A,B		
	Low 21-40%					C
	Very Low 1-20%					↓ C

Numerical model for photothermal heating developed

Distilled into a simple analytic expression

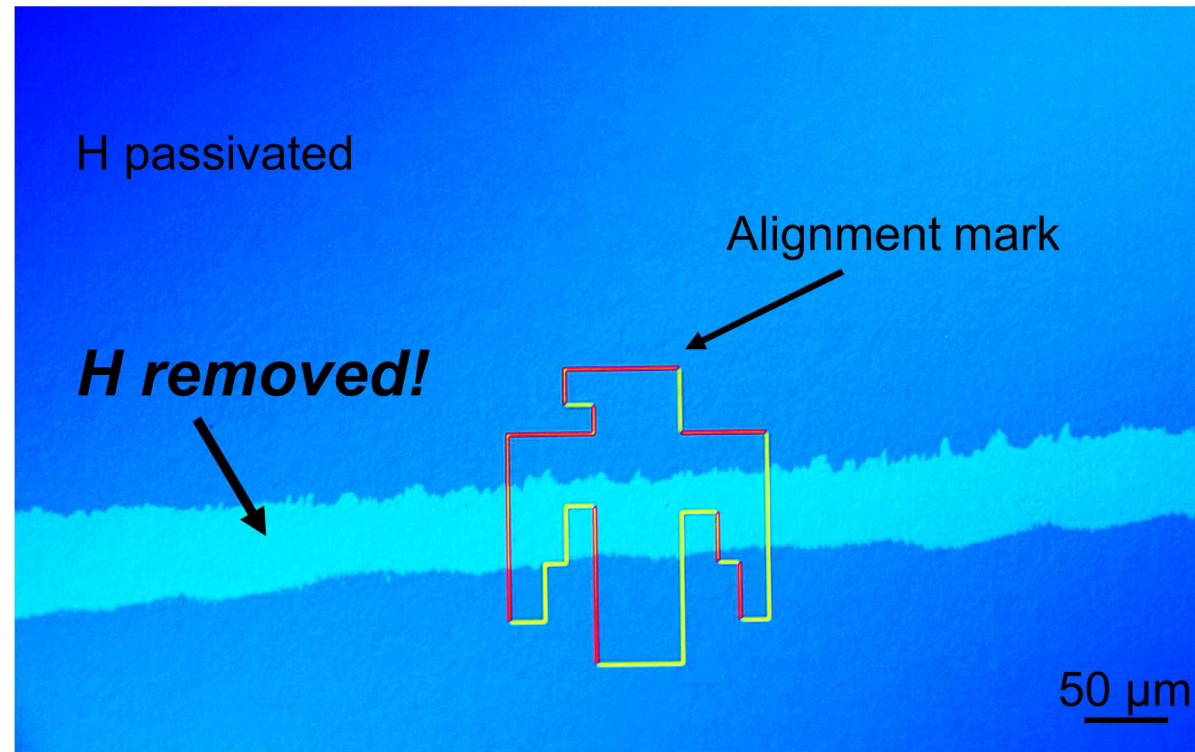


**Rapidly predicts
experimental observation!**

$$T_{peak} \approx \frac{2I [1 - R]}{\bar{\kappa}} \sqrt{\frac{D\Delta}{\pi}} + \mathcal{O} \left(\frac{1}{\alpha \sqrt{D\Delta}} \right)$$

FY19 – Progress (Experimental)

Demonstrated H photolithography in UHV!

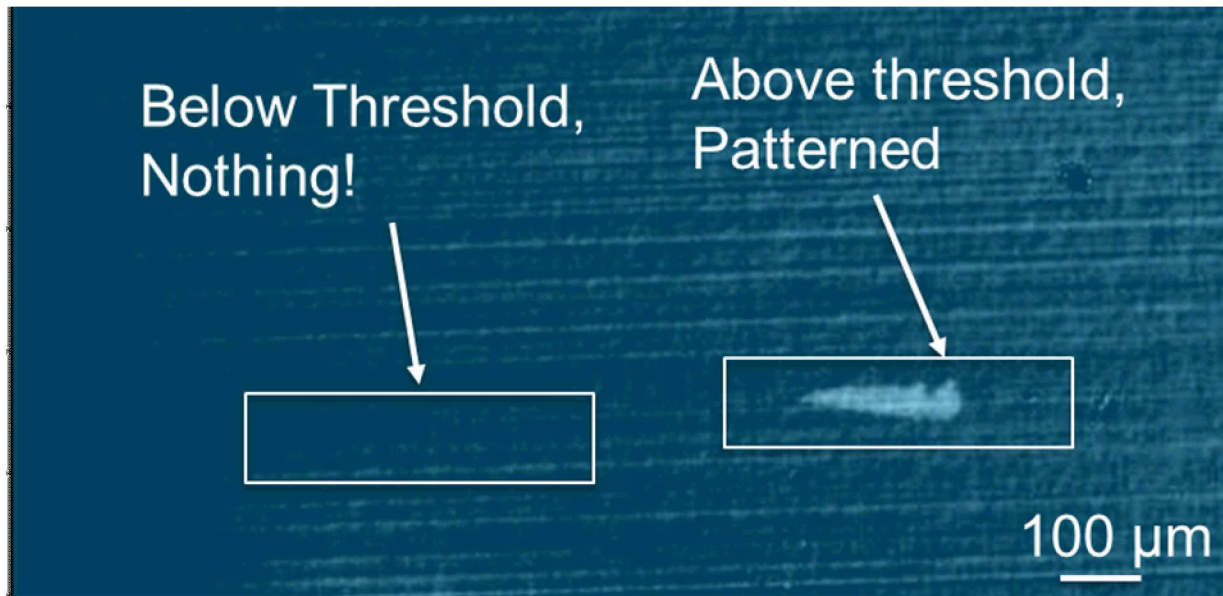


Interference contrast in optical microscope

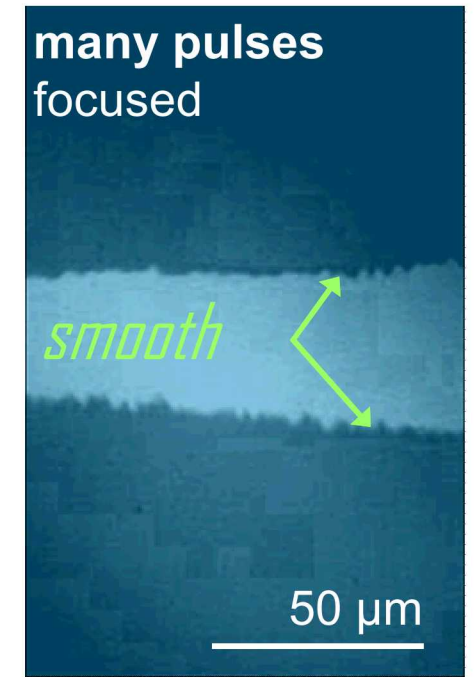
FY19 – Progress (Experimental)

Observed new resist behaviors

Self developing, digital resist

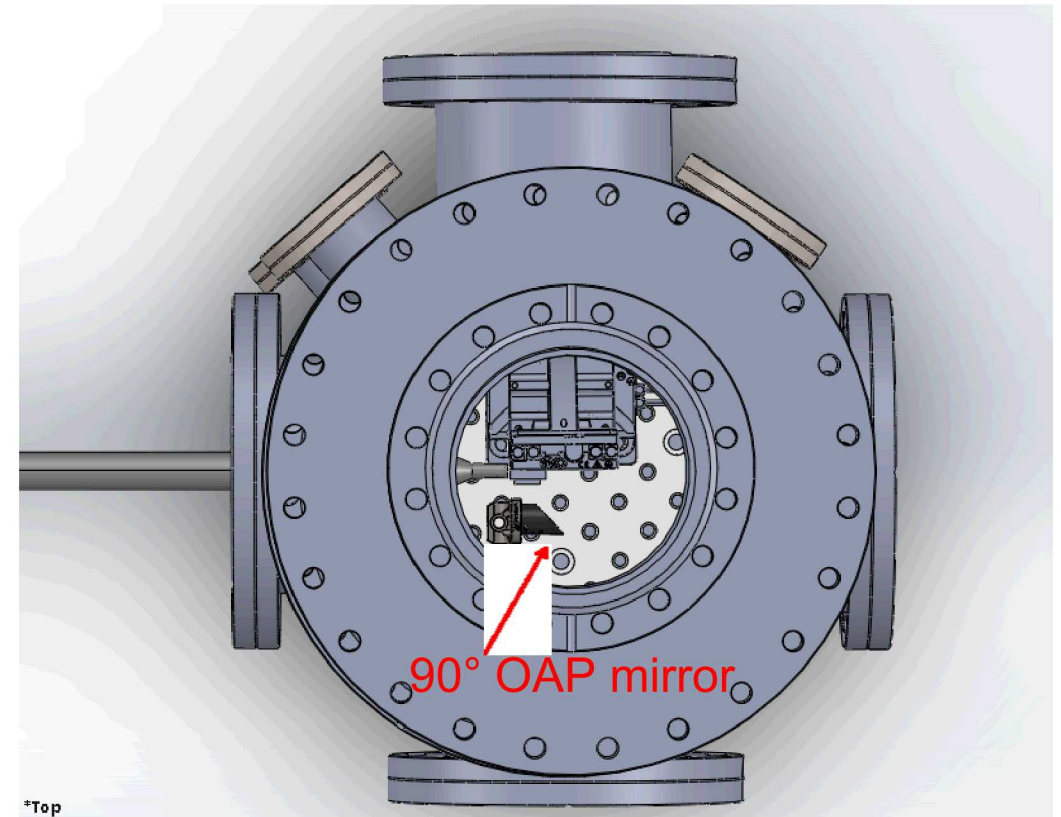


Reduced line-edge-roughness w/out overexposure

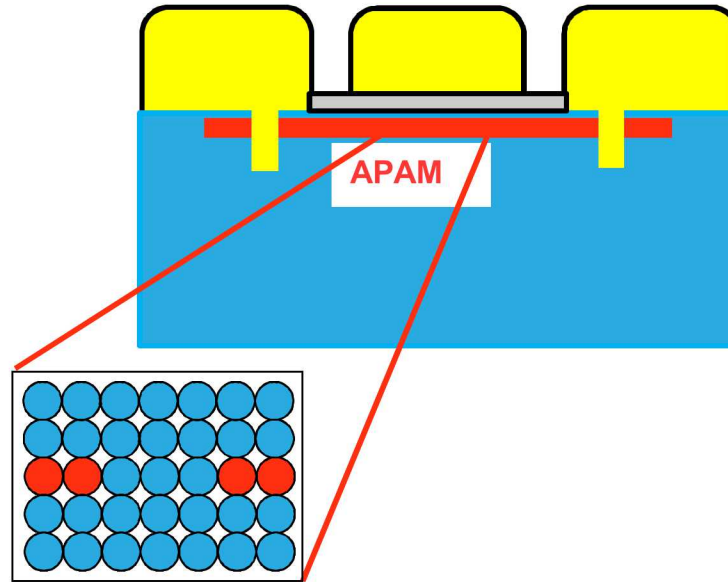


FY20 – Outlook and Improvements

- Next: Demonstrate doping
- New photolithography system
 - Improves accuracy and resolution
- VUV laser ($\lambda \approx 160$ nm)
 - Cleave Si-H bond w/out heating



Task 2 - Acceptor Doping



Current state:

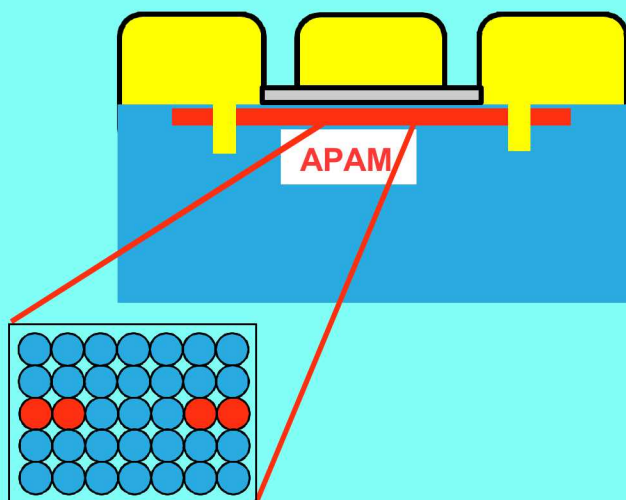
- n-doping only
- Via phosphine (PH_3) & H resist
(gas phase in vacuum)

For APAM MOS & other exemplars:

- **NEED: acceptor/p-type doping
(gas phase in vacuum)**

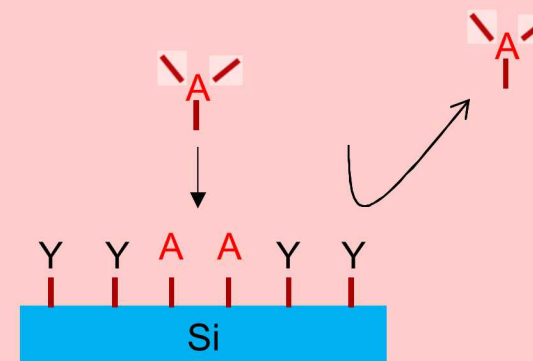
Challenge: satisfy two sets of criteria

Electronic Performance



- **APAM Doping >> conductive than Si**
Extreme dopant density, $\geq 10^{14} \text{ cm}^{-2}$
- **Integrate in device process**
Contact, minimal dopant diffusion

APAM chemistry



- **Selective templated surface chemistry**
e.g. contrast H >1000:1, T < 500C

success!

* Acceptor doped APAM material meeting all criteria would be world 1st

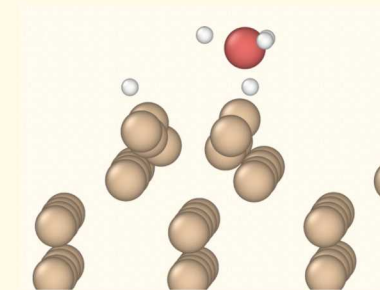
Overview of our approach to R&D problem

1. Identify candidate chemistries & precursors

Acceptor	Hydride	Halide	Organic	Amine
Boron	B_2H_6 $B_{10}H_{14}$	BCl_3 BF_3 BBr_3	Trimethylboron	Ruled out – poor sticking, attacks H resist
Aluminum	AlH_3	$AlCl_3$	Trimethylaluminum Triethylaluminum	
Gallium		$GaCl_3$	Triethylgallium	
...	

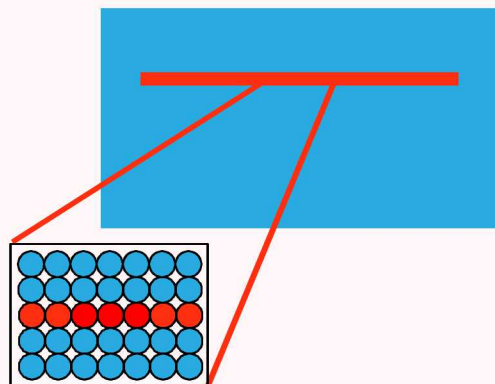
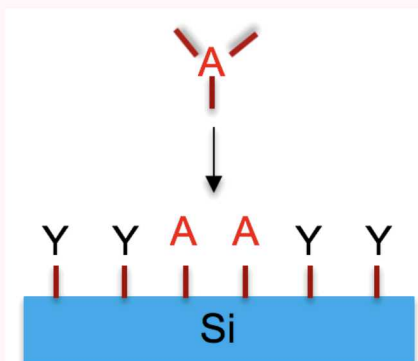
2. Prescreen (literature+models)

-Is a chemistry/recipe likely to work?

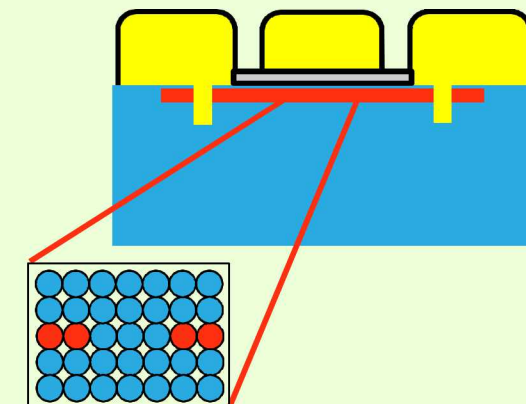


3. Lab tests & recipe development

- APAM surface chemistry test
- Materials physical & chemical tests



4. Fab integration - device process



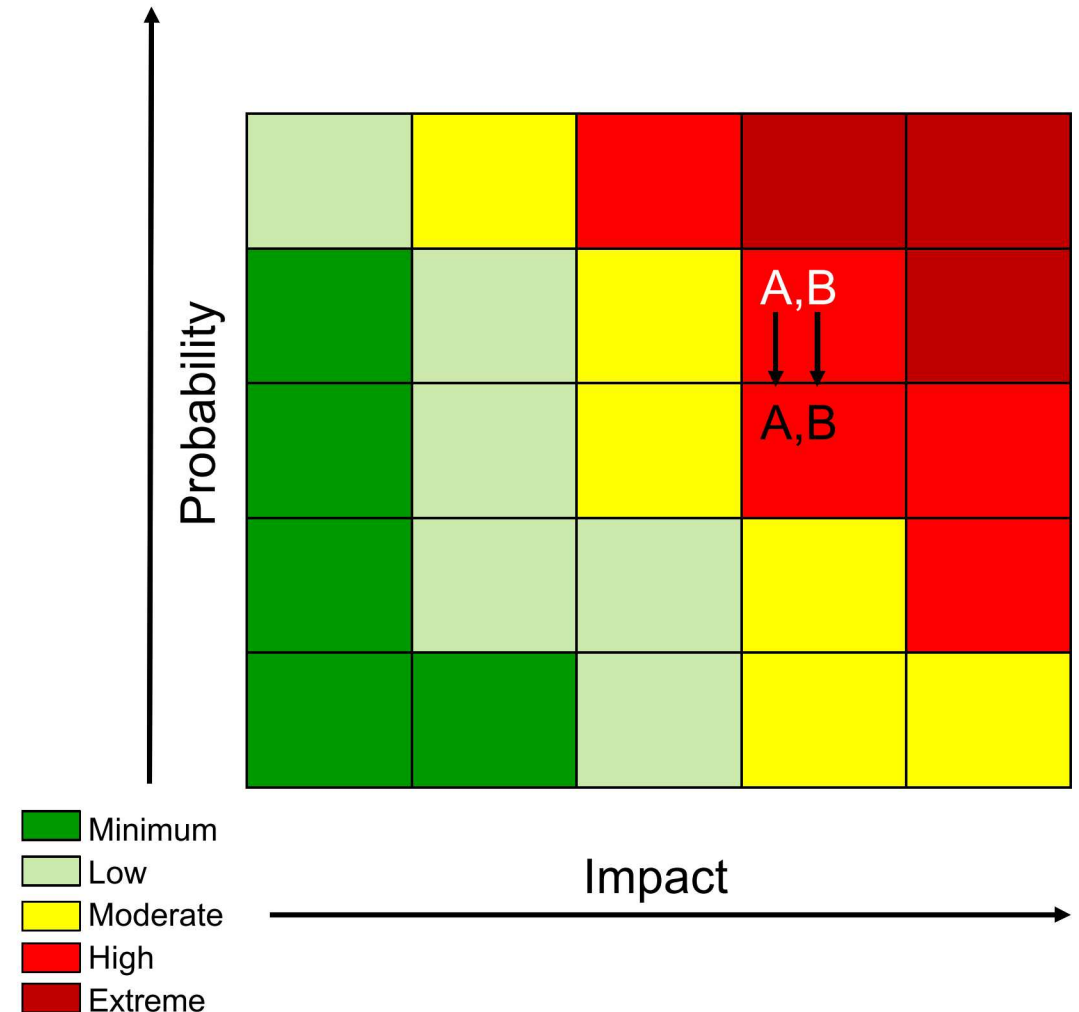
Risks, impact to project, mitigation

Risks

- A. Uncertain difficulty to implement lab tests
→ potential delays, time/cost
- B. Unable to discover method meeting all requirements
→ cost/risk pushed to other tasks

Mitigations

1. Engage capable partners in surface chemistry tests
2. Use chemical kinetic models to guide experiment



Progress & current status

1. Identify candidate chemistries & precursors

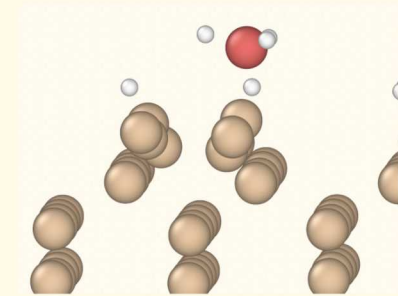
Acceptor	Hydride	Halide	Organic	Amine
Boron	B_2H_6 $B_{10}H_{14}$	BCl_3 BF_3 BBr_3	Trimethylboron	Ruled out – poor sticking, attacks H resist
Aluminum	AlH_3	$AlCl_3$	Trimethylaluminum Triethylaluminum	
Gallium		$GaCl_3$	Triethylgallium	
...	

 Sandia  U. Maryland  Zyvex Labs Inc

☐ We engaged partners to mitigate cost/time/risk

2. Prescreen (literature+models)

-Is a given chemistry likely to work?

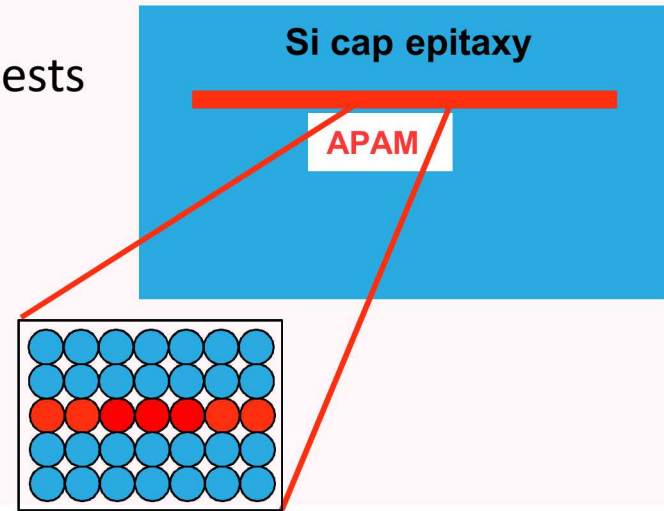


$$BH_3 E_A = -1.90 \text{ eV}$$

- ☐ We built models now guiding decisions
- ☐ Density functional theory - reaction paths, energetics, and barriers
- ☐ See postdoc Quinn Campbell's poster

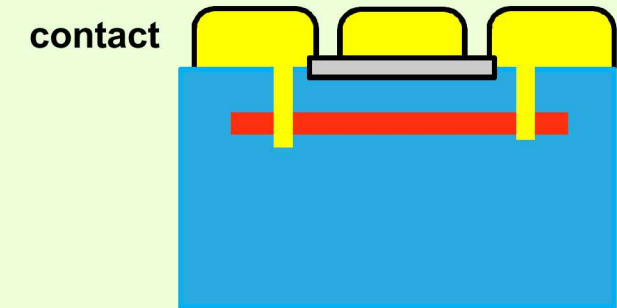
3. Lab tests & recipe development

- APAM surface chemistry tests
- Materials physical & chemical tests
- Electronic properties tests



☐ Lab tests & recipe development in-progress

4. Fab integration - device process



- ☐ ID'd challenges, e.g. electrical contact
- ☐ Solutions in-progress

FY20 plan

FY 19

FY20 Q1/Q2

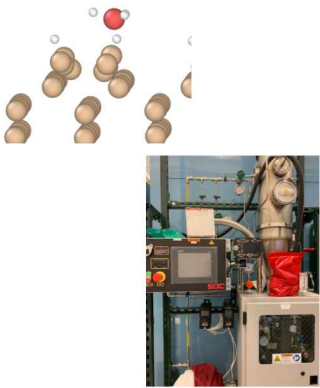
Q2

Q3

FY20 Q4

Lab tests & recipe development

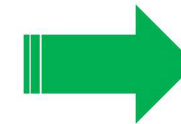
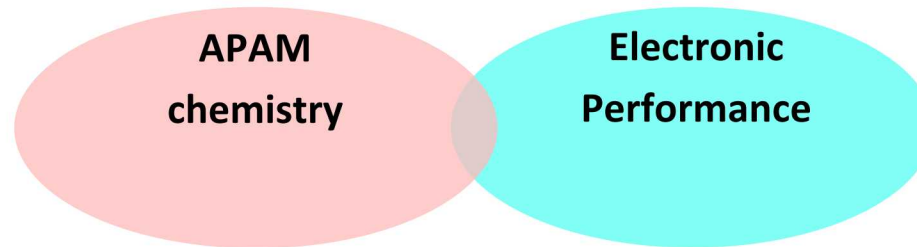
Labs set-up
& model built



Surface
chemistry
tests

Material
Physicochemical
test

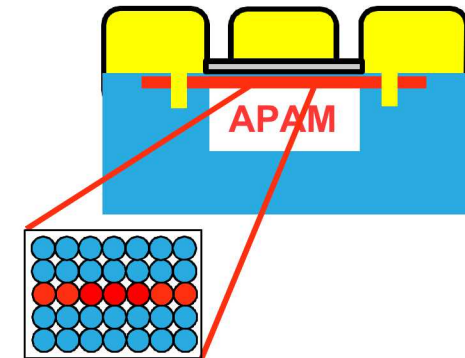
Material
Electronic
test



Downselect

amongst
chemistries-
hydride,
halide,
alkyl

**APAM
p-doped device
e.g. wire**



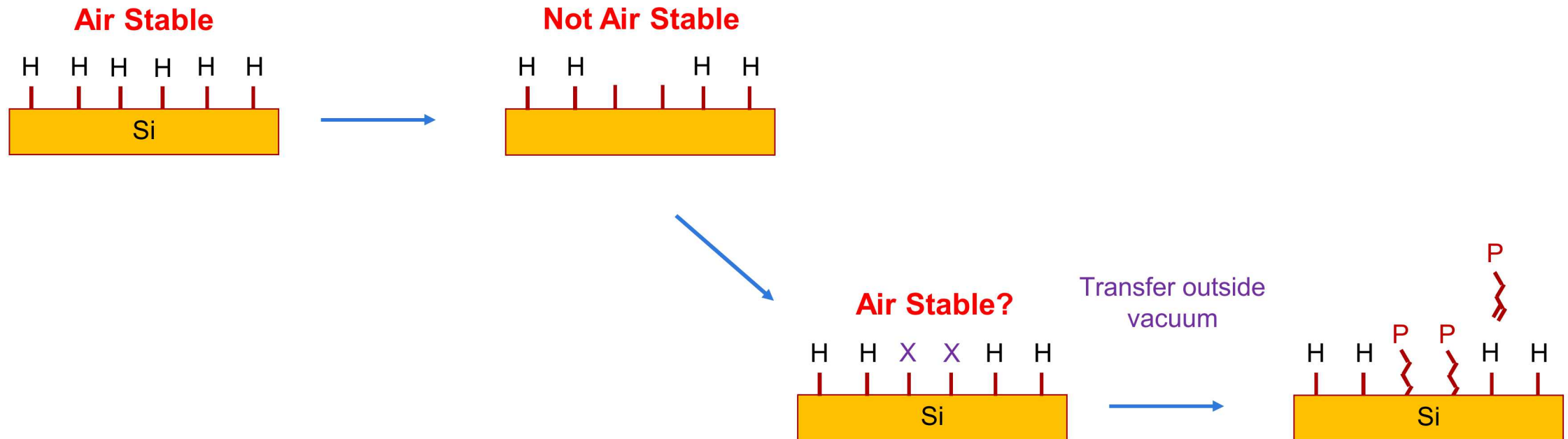
Fab integration

ID integration
challenges

Engineer
solutions

Task 3 - Alternative Resists

GOAL: Can APAM patterns survive outside of UHV
to transfer between different processing tools?



QUESTION: Are halogen resists stable in air?

Risks

Goal: Alternative and complementary resist and dopant chemistries

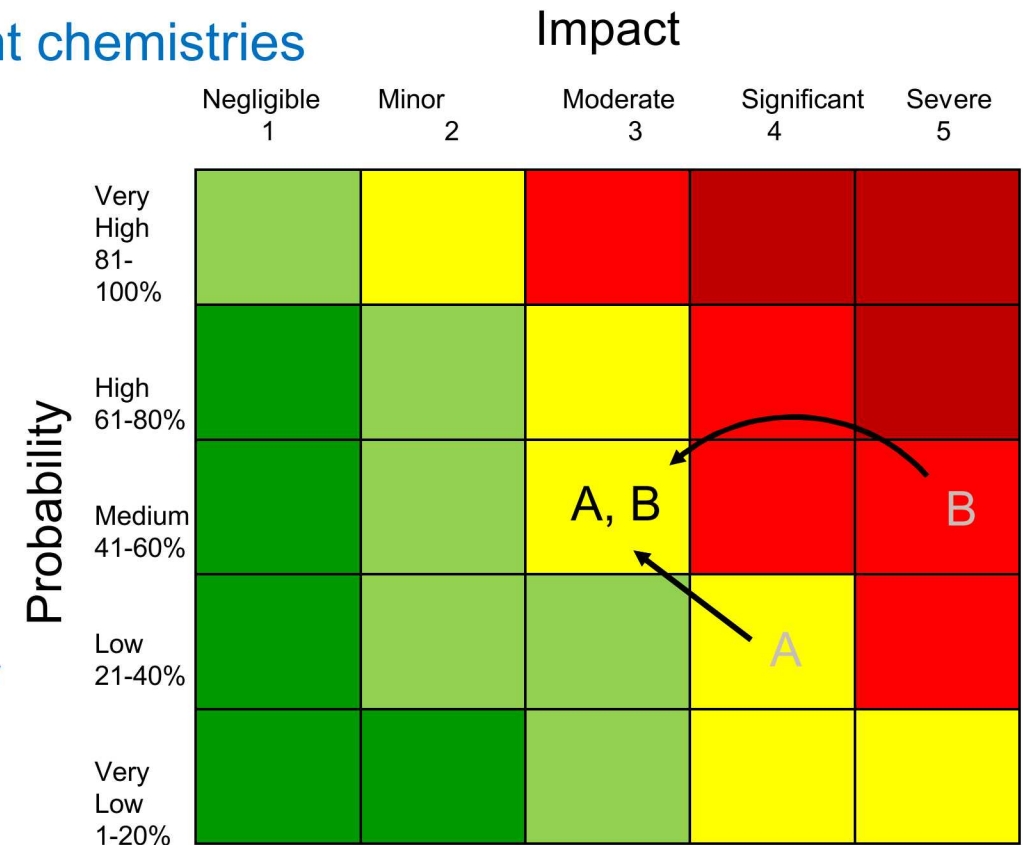
Challenges/Risks:

A. Halogen resist not stable outside UHV

a. **Stable in N₂**

B. No selective dopant chemistry pathway

a. **Wet chemical selective chemistry pathway**

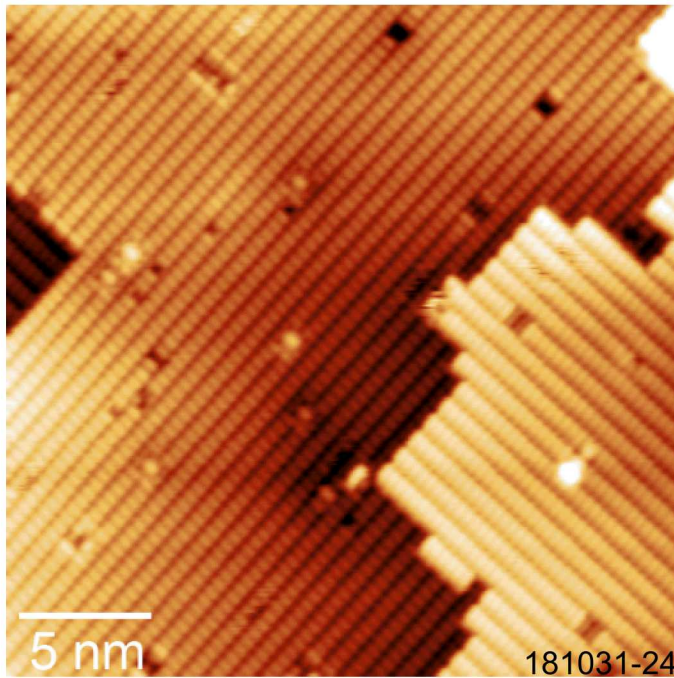


Color key:

- Dark Green – Minimum Risk
- Light Green – Low Risk
- Yellow – Moderate Risk
- Red – High Risk
- Dark Red – Extreme Risk

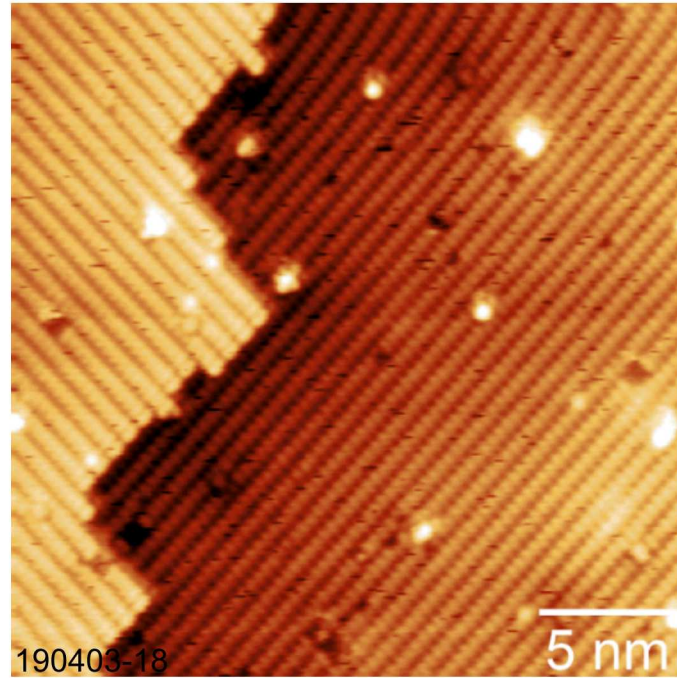
FY19 – Alternative Resists (Cl, Br, and I)

Full Cl Passivation



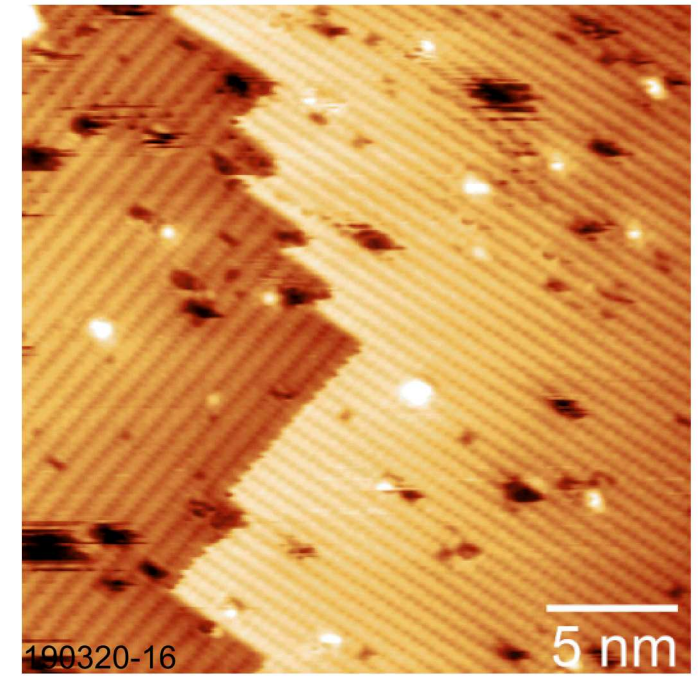
-1.7V, 0.7nA

Full Br Passivation



-1.7V, 0.4nA

Full I Passivation

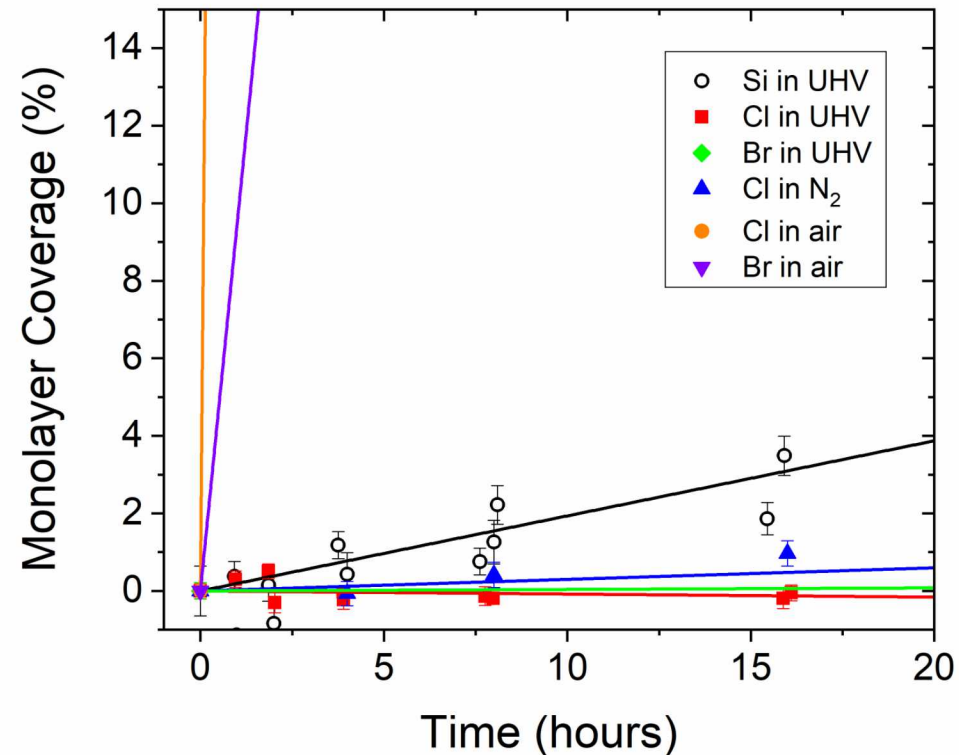


-1.7V, 0.5nA

First observation complete iodine passivation

FY19 – Stability of Halogen Resists

Surface Coverage of Contaminants (O, C)



- **All** halogen resists contaminated by air exposure
- **All** halogen resists stable in N₂ for hours

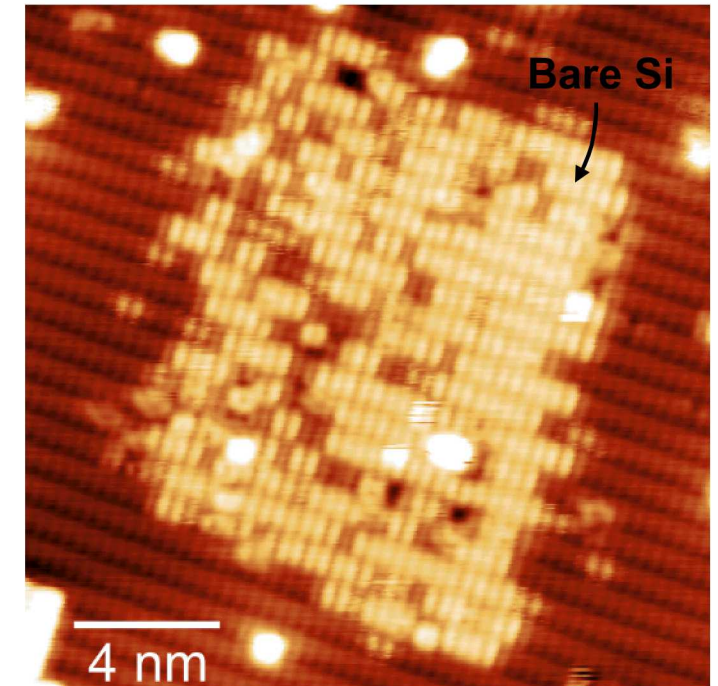
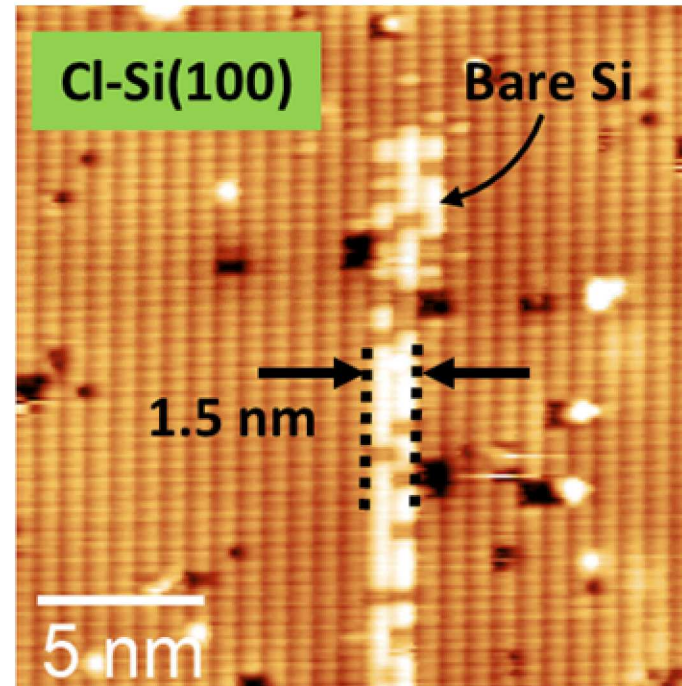
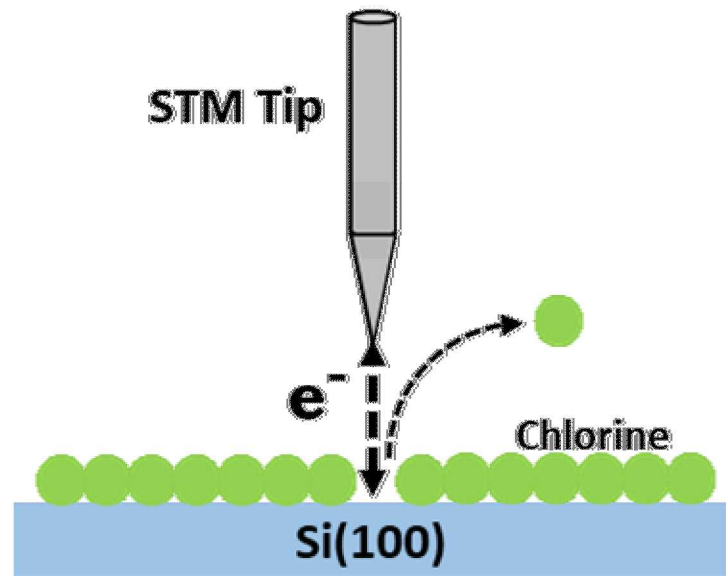
FY20 – Impact & Outlook

- Fabricated Cl, Br, and I resists
 - First observation of complete iodine passivation
- Determined stability of halogen resists:
 - ***Not stable*** in air, but...
 - ***stable*** in N₂ → *Suitable for wet chemistry pathway*
- Patternable halogen resists in UHV: Cl, Br, I

FY20: Merge complimentary resists with Wet Chemistry Task

- ***Re-allocate resources to acceptor doping task.***

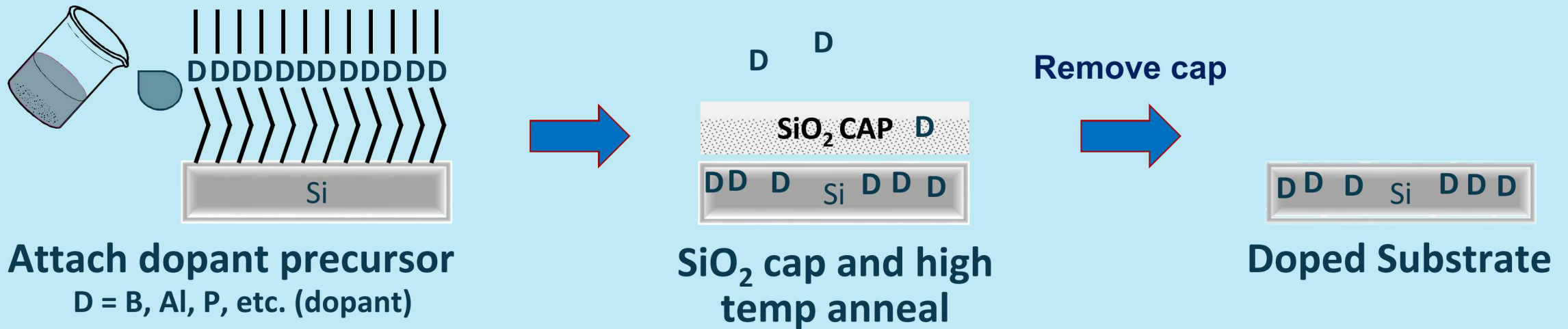
Alternative Resists for Acceptor Doping



Patternable halogen resist expands available acceptor chemistry

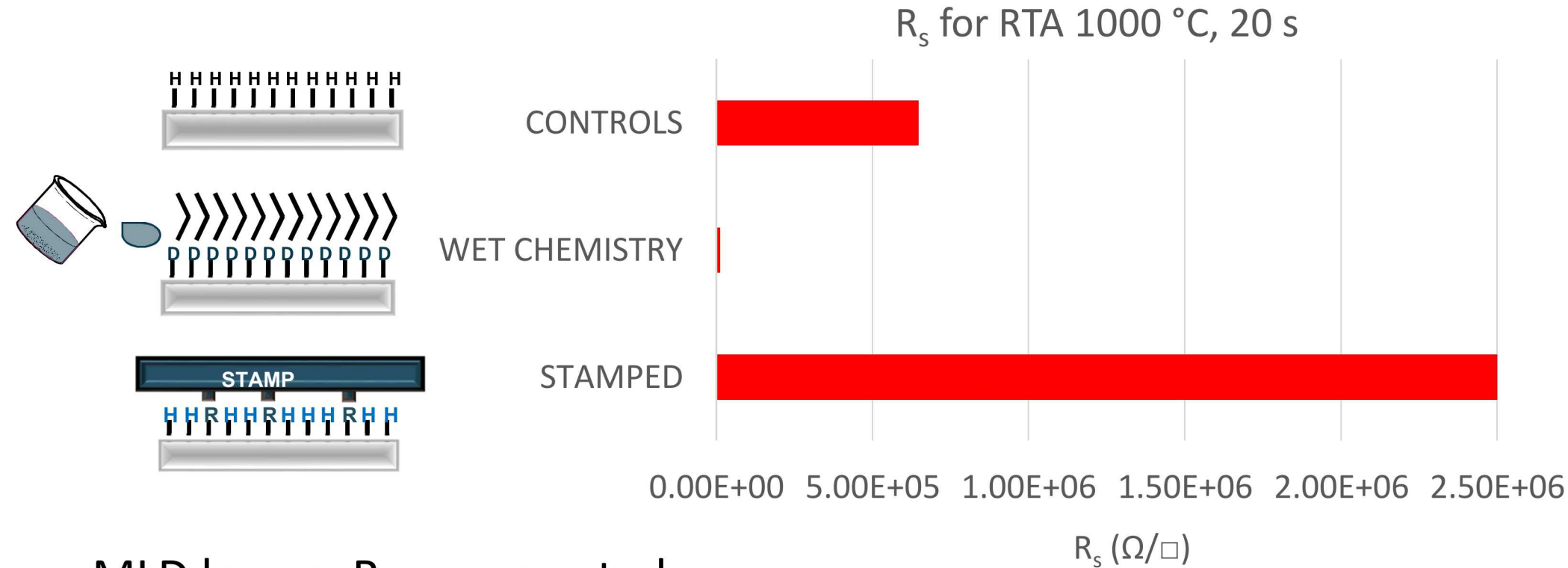
Task 4 - Wet Chemistry

Goal - Develop a scalable wet-chemistry method to dope Si(100) with high active dopant concentration & desirable electronic properties.



“Monolayer Doping (MLD)” (Javey, 2007) allows for a large variety of dopant attachment chemistries via organic functionalization of the silicon surface.

FY19 - Hines/Javey MLD Chemistry



- MLD lowers R_s , as expected
- Stamping made R_s worse than undoped control
- Lowest achievable experimental $R_s = 1.3E+04 \Omega/\square$, Javey reported $R_s = 2.0E+03 \Omega/\square$
- MLD doesn't meet "APAM-like" target goal: $<500 \Omega/\square$

Filling in the Gaps

	Current APAM	Current MLD
Electronic Properties	Green	Yellow
Patterning	Green	Red
Process Temperature	Yellow	Red
Scalable	Red	Green

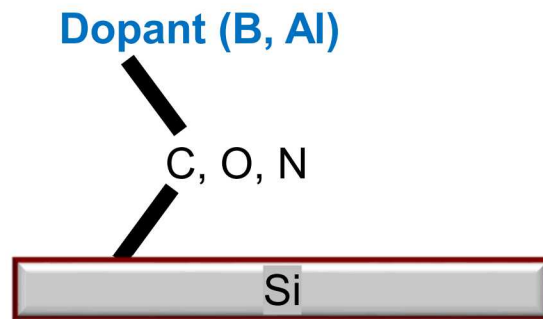
Goal – Discover a scalable, low temperature process to dope Si(100) to achieve material with “APAM-like” properties (e.g. very high doping).

Impact

Problem: Current SOTA MLD chemistry has not produced “APAM-like” doping levels

Hypothesis: Si-C bonds require high temperature annealing to break, resulting in **diffusion-limited doping** (i.e. current fab processes) instead of **desired surface delta doping**

Current MLD chemistry

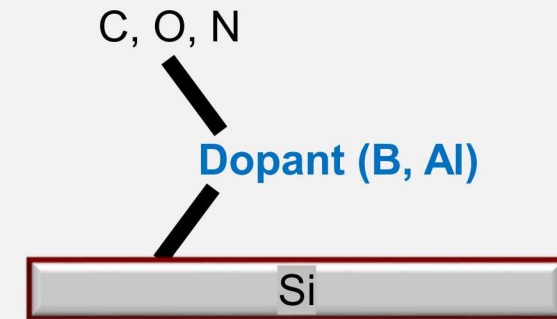


Indirect dopant attachment (e.g. Si-C-B) limits doping to solid solubility

FY20



Desired MLD chemistry

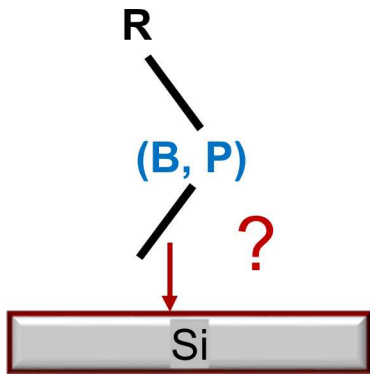


New strategy: Direct Si-dopant bond

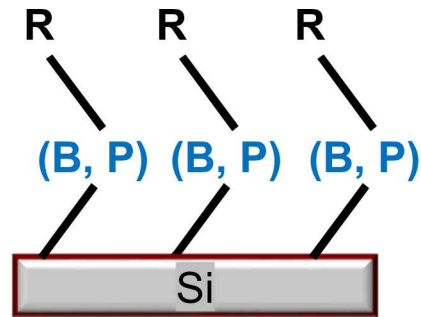
But: Little precedent for direct Si-dopant bond formation and stability on Si(100) surface

FY20 New approach: challenges

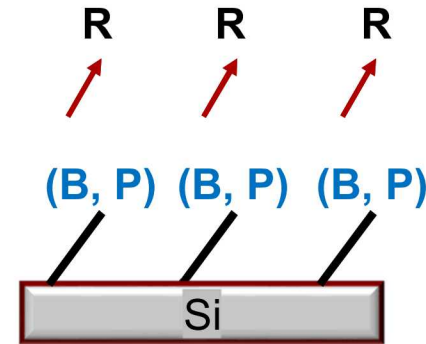
Discover direct
dopant-Si
wet chemistries



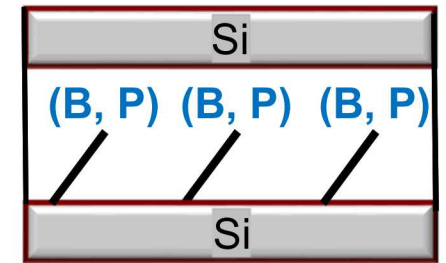
Optimize for high
density dopant
attachment



Remove
ligands &
contaminants



Capping &
electrical
characterization

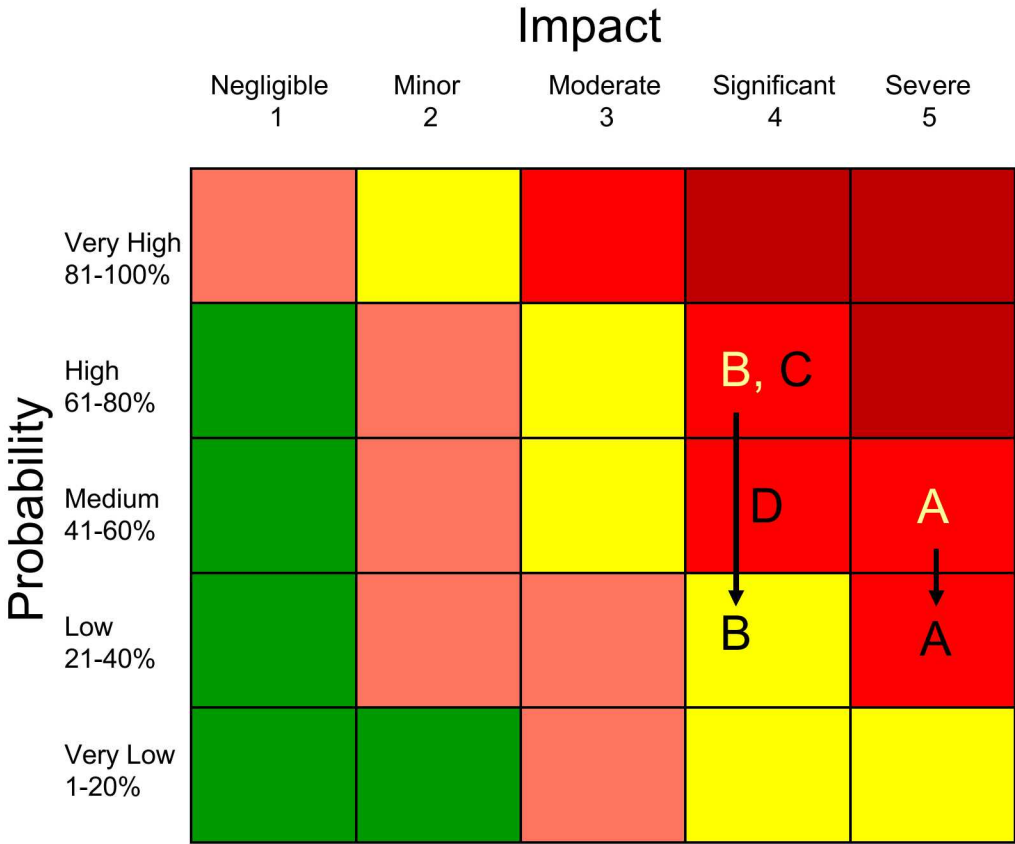


Risks Assessment

Goal: Scalable wet chemical processes for dopant attachment resulting in APAM-like electrical properties

Challenges/Risks:

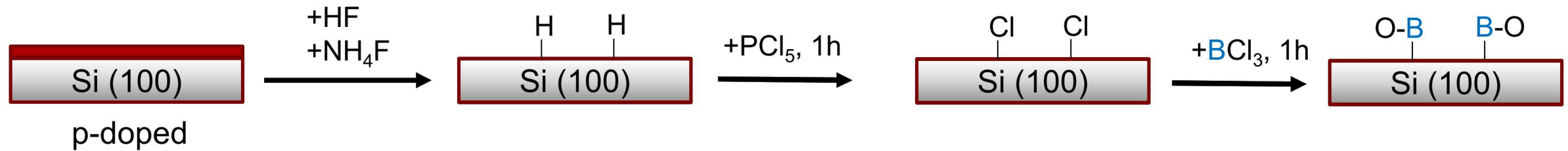
- A. No stable direct dopant attachment chemistries
- B. Low yield of dopants attached
- C. Unable to remove ligands/impurities (C, O, etc.)
- D. No clean Si capping process



- Color key:
- Dark Green – Minimum Risk
 - Light Green – Low Risk
 - Yellow – Moderate Risk
 - Red – High Risk
 - Dark Red – Extreme Risk

Demonstration of wet-chemical Si-B formation

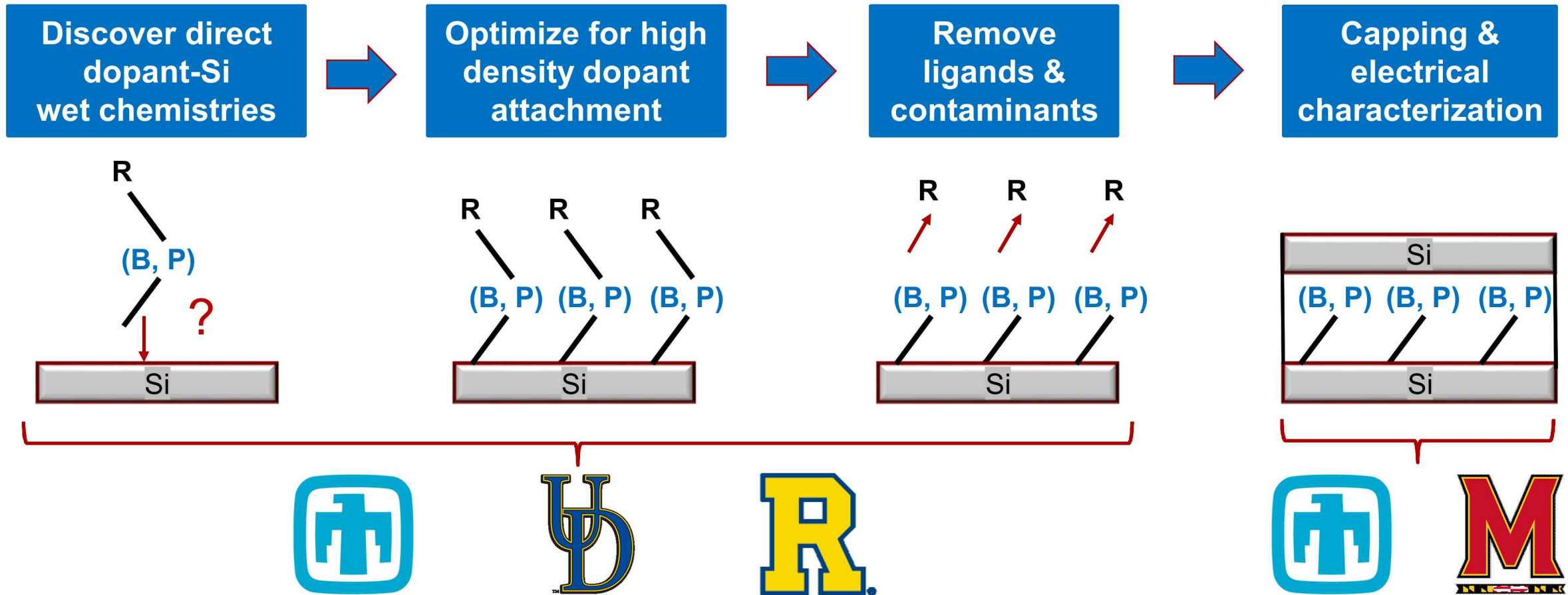
w/U. Delaware and UMD/LPS



- **Showned direct Si-B formation possible by wet chemistry**
- **B** incorporation with high $\sim 1\text{-}4 \times 10^{14} \text{ cm}^{-2}$ density
- Need to mitigate **high C & O levels**
- Selective to Cl-Si over H-Si – compatible with standard patterned H resist?
- **IMPACT/SYNERGIES** - Acceptor Doping Task: result leads to BCl_3 doping in UHV, other suitable precursors for UHV dosing may be discovered

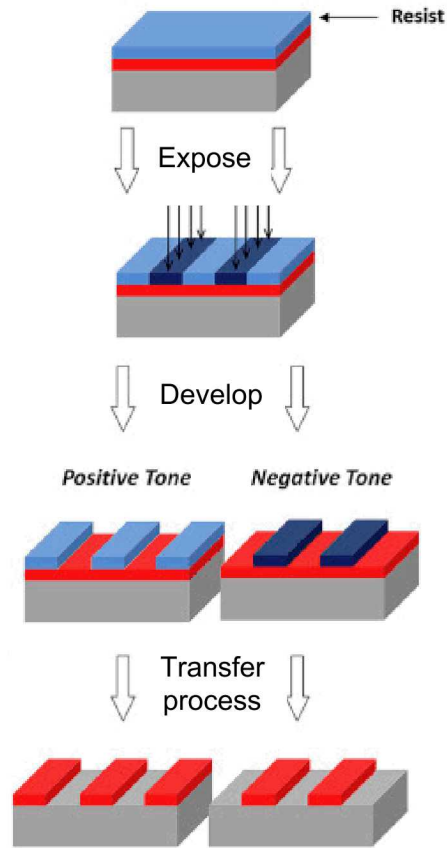
FY20: Leverage external collaborators

Many scientific and technical challenges to overcome!



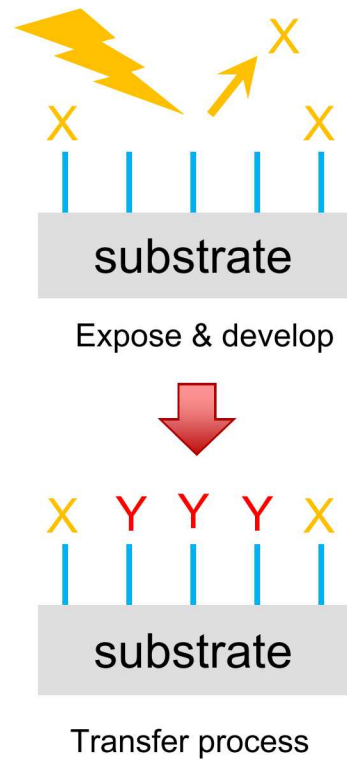
Backup & Previous EAB Slides

Atomic scale processing - opportunity

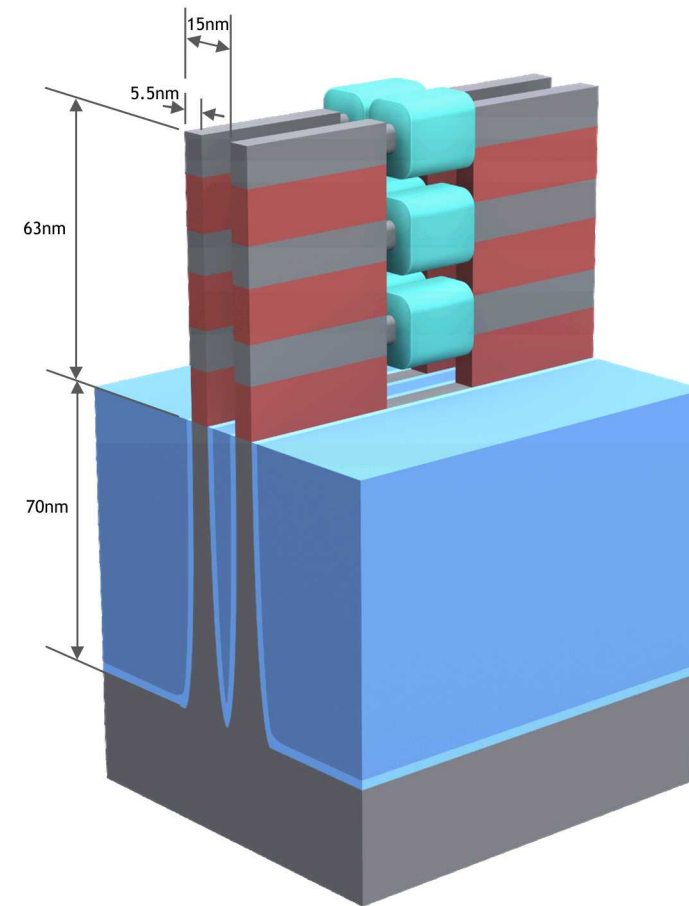


Traditional etch process
Scaling limitations

Unexplored, new phase space



Area selective chemistry
Obviate limitations?



Produces material that defies expectation

R. Arghavani estimates

Parameters	I_{ON} (mA/ μ m) @ $V_{GS}=V_{DS}=0.7V$
Si FinFET ($H_{FIN}=37nm$)	0.630
1 NW GAA	0.286
2 NW GAA	0.525
3 NW GAA	0.576
Si FinFET ($H_{FIN}=54nm$)	0.690

Compare to APAM:
over 2 mA/ μ m
(cryogenic)!

Thrust 4 – milestones (backup slide)

- Photolithography Task (Lead: Katzenmeyer)
- Q1: Fabricate large area patterned sample via photodesorption of H with Hall density within 1/5 of a full-dose delta layer.
- Q2: Determine depassivation dose range and associated line edge roughness (via SMIM)
- Q3: Determine conductivity for 2 um wide photopatterned wire with new system
- Q4: Determine better mode for H photolithography (thermal vs. photodissociation) based on sheet resistance and line-edge roughness of P-doped regions

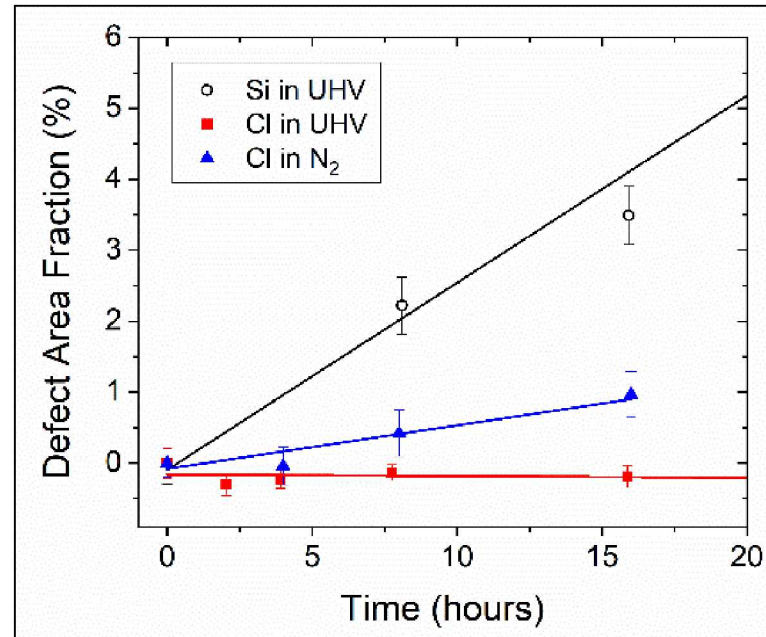
- Acceptor Doping (Lead: Bussmann)
- Q1: Performers demo APAM acceptor chemistries in-vacuum
- Q2: Demonstrate test chip for acceptor Hall bars with acceptor implant proxy, for both B and Al.
- Q3: Down-select to best acceptor precursor with $>5 \times 10^{13} \text{ cm}^{-2}$ carrier density and conductivity $> 0.1 \text{ mS/sq}$.
- Q4: Implement down-selected process at Sandia

- Wet Chemistry Task (Lead: Wang)
- Q1: Show wet chemical acceptor and donor doping chemistries on H- and Cl- terminated Si
- Q2: N/A
- Q3: Go/no-go: Determine if wet chemistry doping processes can produce $>5 \times 10^{13} \text{ cm}^{-2}$ carrier density and conductivity $> 0.1 \text{ mS/sq}$.
- Q4: Implement down-selected process at Sandiaa

FY19 – Stability of Halogen Resists

Back up slide

Stability in N₂ (STM)



- All halogen resists stable in N₂ for many hours
- Halogen resists show oxidation and carbon contamination in air
 - *Br is most stable, comparable to H resist(?)*