



SAND2019-7195PE

# An Introduction to Concentrating Solar Technologies at Sandia National Labs

**Andrea Ambrosini**

Seminar, 2 November 2018  
New Mexico State University

# Sandia National Labs



- Sandia is one of 17 National Laboratories



# Our Workforce: ~12,000 employees



~10,600 Regular employees  
~2,000 Temporary employees,  
students & postdoctoral appointees

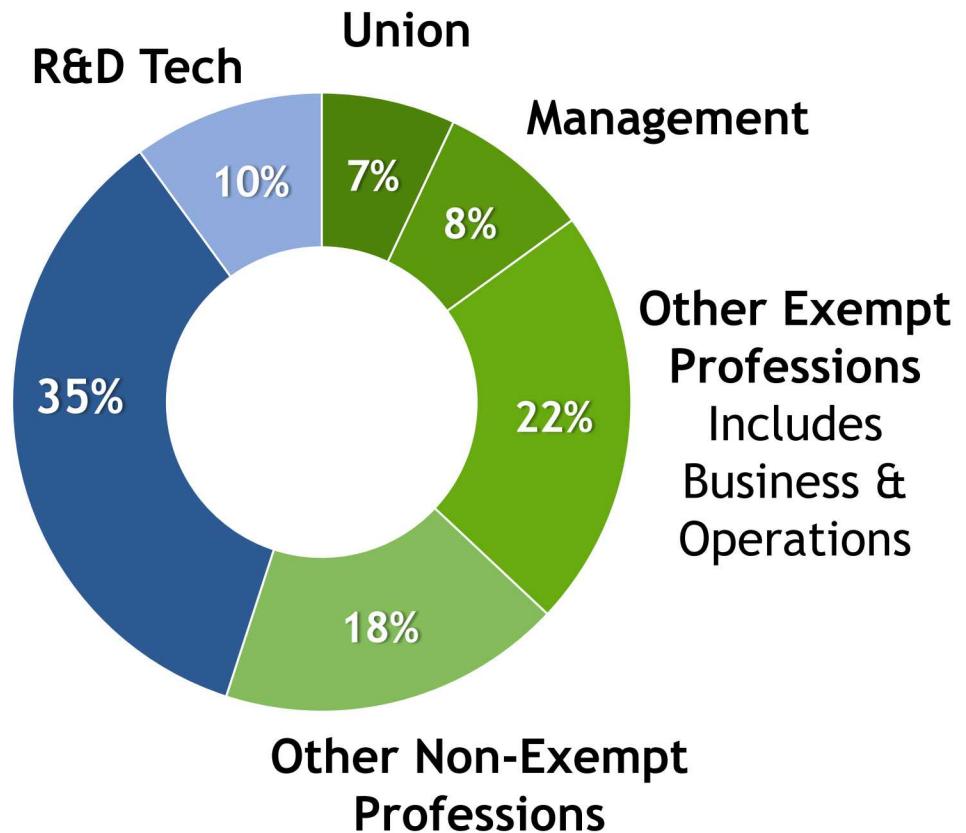
## New Mexico Site:

Workforce: ~9,600  
R&D employees: ~4,800  
(R&D Staff & Technologists)

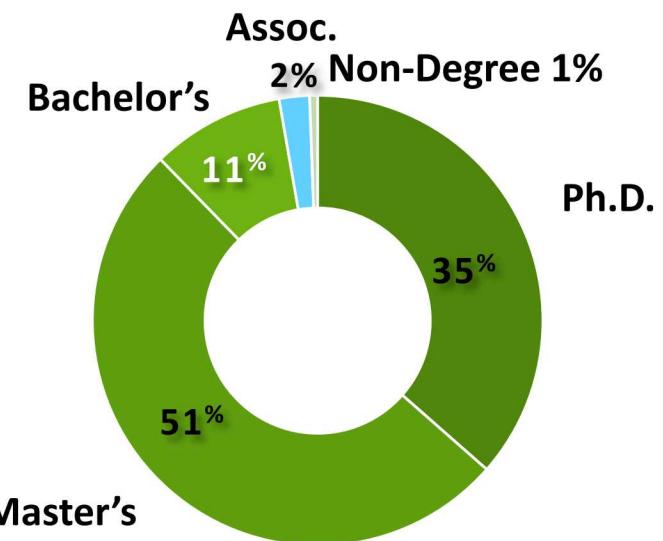
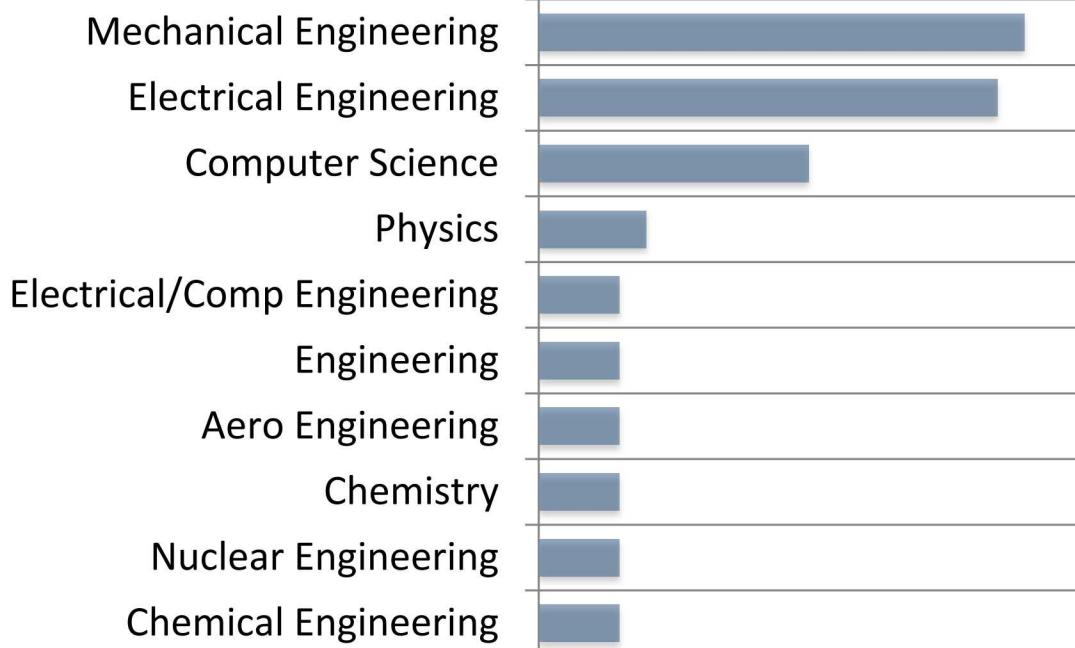
## California Site:

Workforce : ~1,000  
R&D employees: ~600  
(R&D Staff & Technologists)

## R&D Staff



# R&D by Discipline & Degree



Top 10 job descriptions shown, Regular exempt non-management employees only

Data as of June 30 end of Q3 of the FY17

# National Security Mission Areas



## *Nuclear Deterrence*



## *National Security Programs*



## *Energy & Homeland Security*



## *Defense Nuclear Non-Proliferation*

- Transportation and energy systems
- Renewable systems and energy infrastructure
- Homeland Security
- Cyber and Infrastructure Security

# Concentrating Solar Power (CSP)



- Intro to CSP
- National Solar Thermal Test Facility
- Materials in CSP
- CSP Research at Sandia

# 7 | Problem Statement

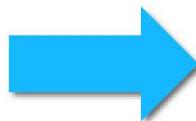


- Current renewable energy sources are intermittent
  - Causes curtailment or negative pricing during mid-day
  - Cannot meet peak demand in evenings
- Available energy storage options for solar PV & wind
  - Large-scale battery storage is expensive
    - \$0.20/kWhe - \$1.00/kWhe
  - Compressed air and pumped hydro - geography and/or resource limited





- Renewable energy technology with reliable, efficient, and inexpensive energy storage

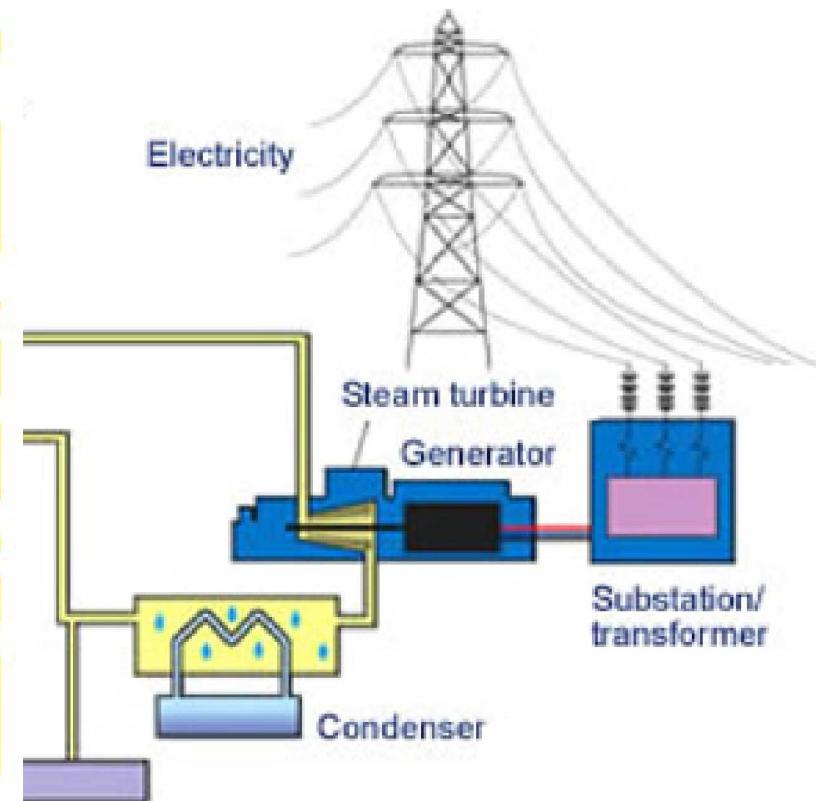
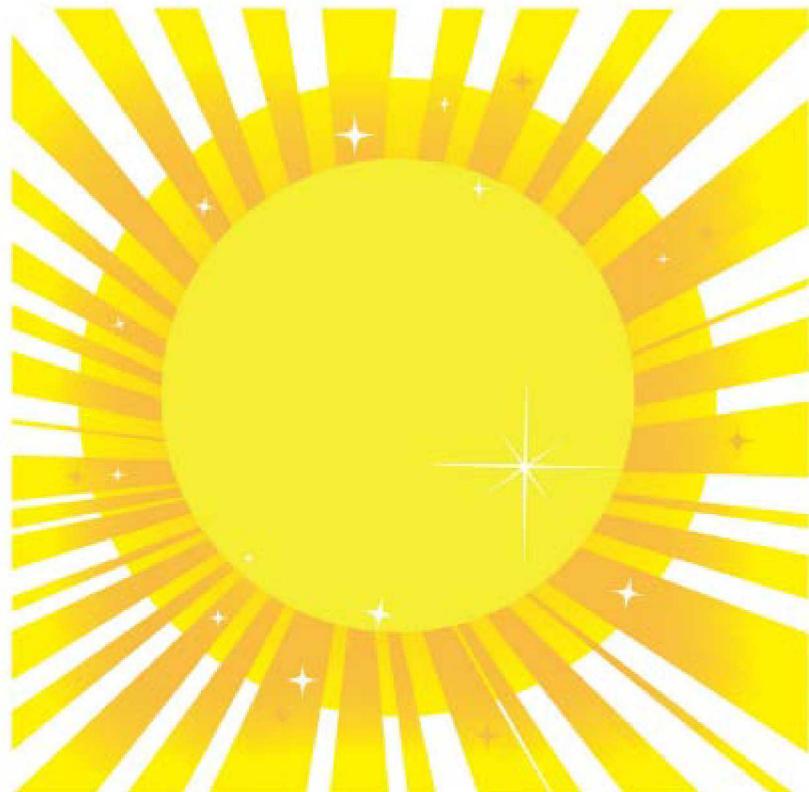


Concentrating solar power (CSP) with thermal energy storage



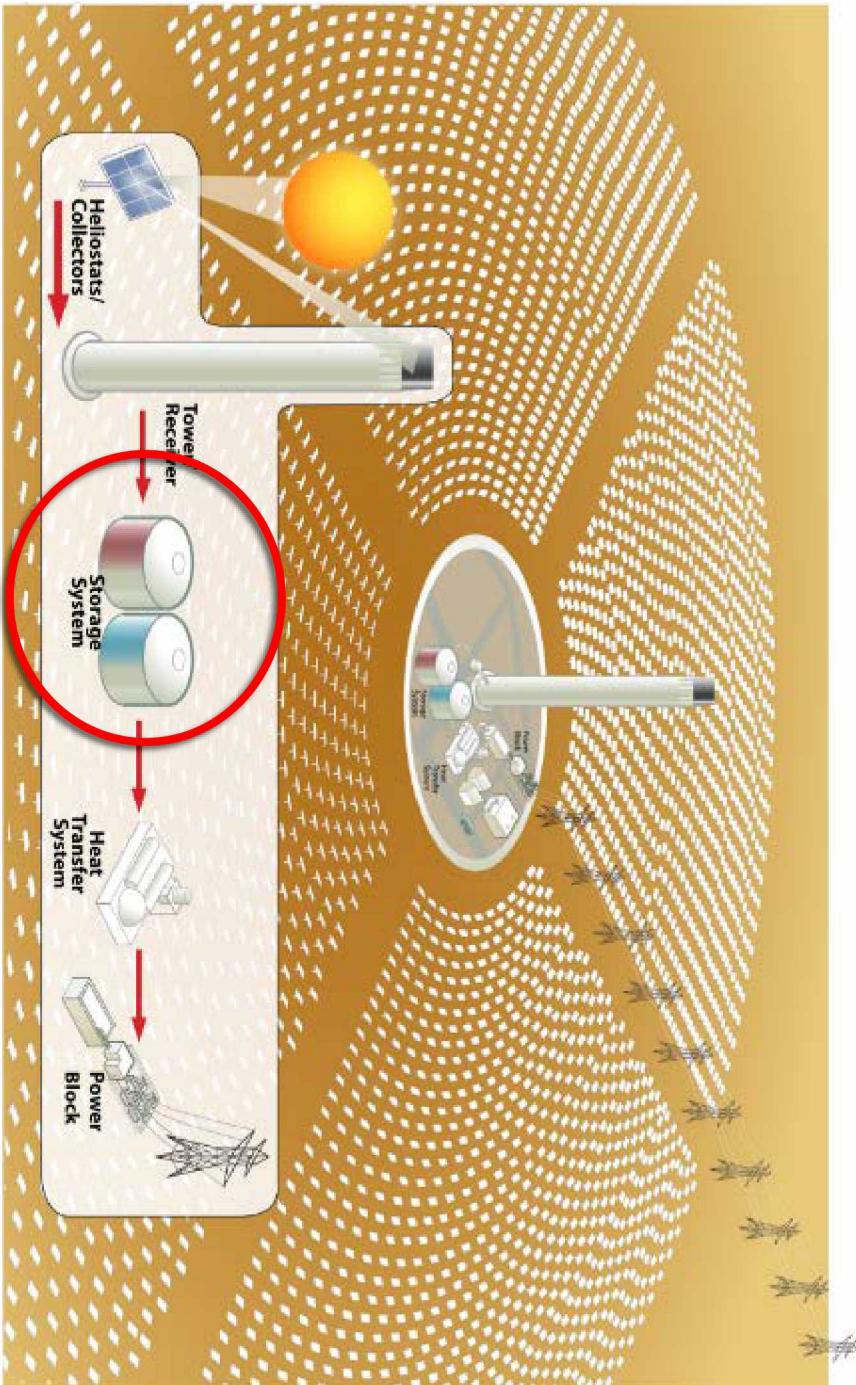
# What is Concentrating Solar Power (CSP)?

Conventional power plants burn fossil fuels (e.g., coal, natural gas) or use radioactive decay (nuclear power) to generate heat for the power cycle



Coal-Fired Power Plant

# CSP and Thermal Energy Storage

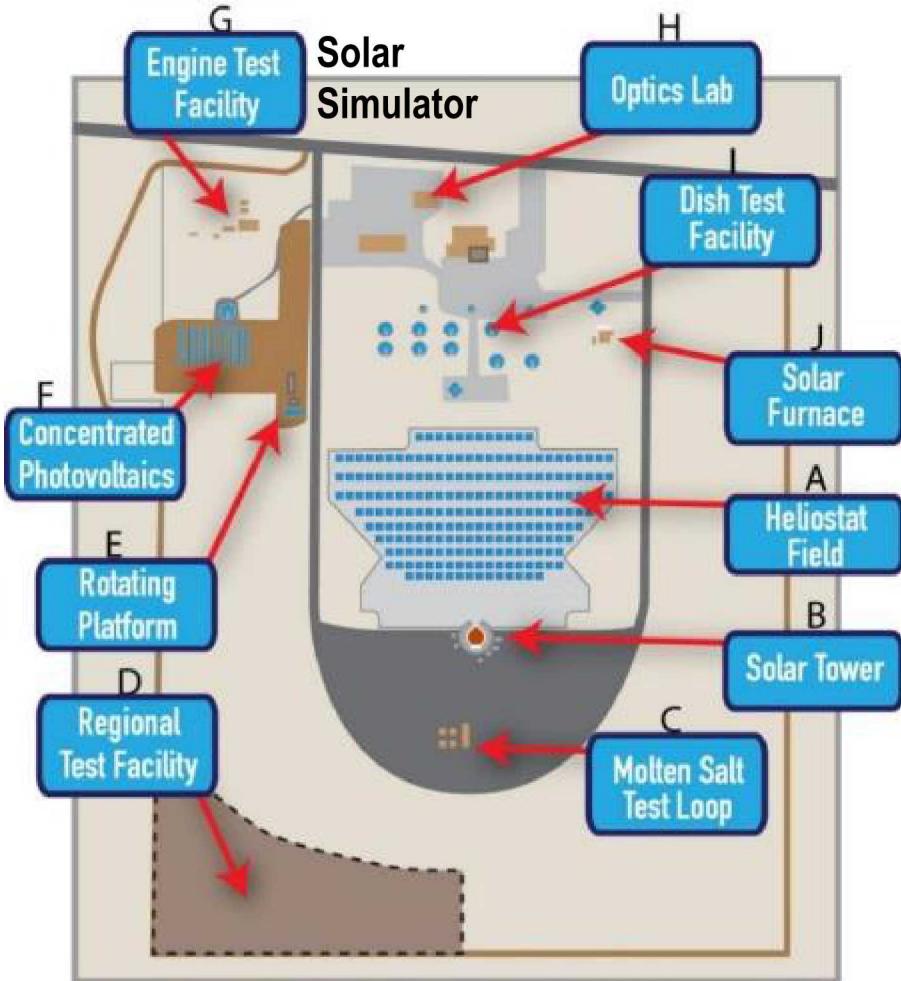


- Concentrating solar power uses mirrors to concentrate the sun's energy onto a receiver to provide heat to spin a turbine/generator to produce electricity
- Hot fluid can be stored as thermal energy efficiently and inexpensively for on-demand electricity production when the sun is not shining

# The National Solar Thermal Test Facility



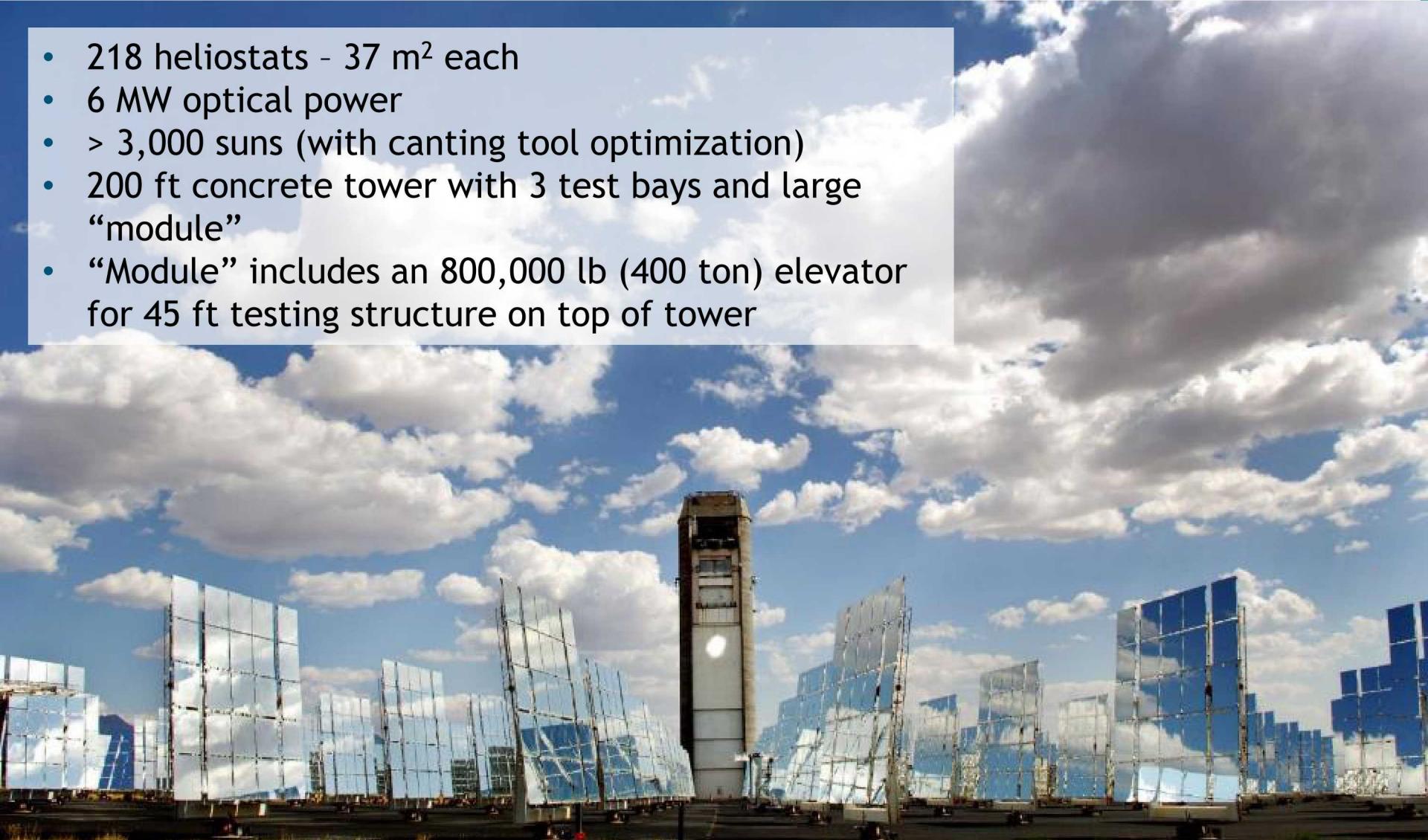
NSTTF is a DOE Designated User Facility



# Solar Tower



- 218 heliostats - 37 m<sup>2</sup> each
- 6 MW optical power
- > 3,000 suns (with canting tool optimization)
- 200 ft concrete tower with 3 test bays and large “module”
- “Module” includes an 800,000 lb (400 ton) elevator for 45 ft testing structure on top of tower



# Solar Furnace



- 16 kW Solar Furnace
- Peak flux ~600 W/cm<sup>2</sup> (6000 suns)
- 5 cm spot size

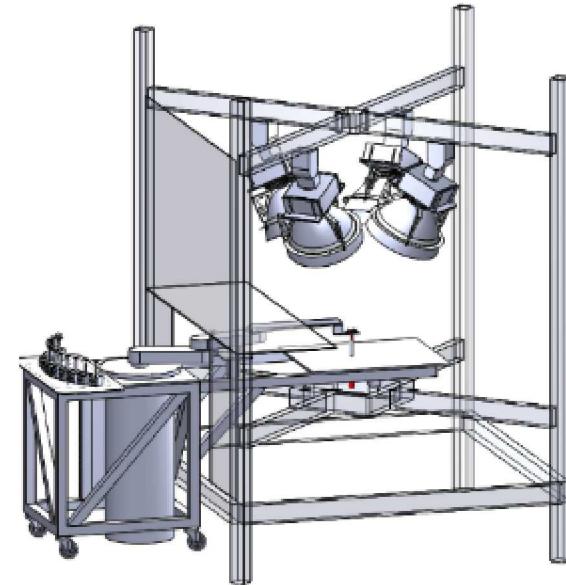
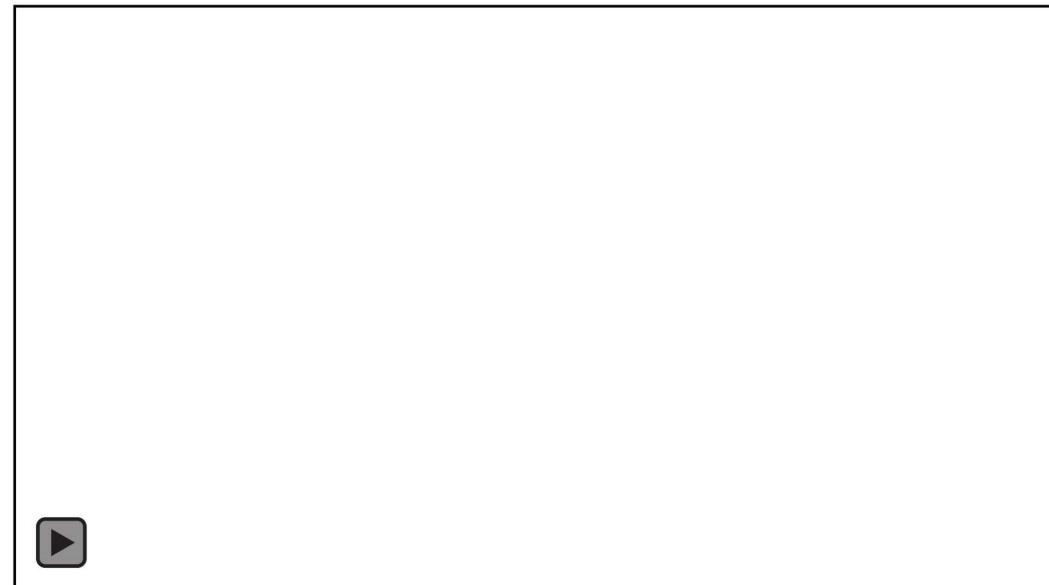




# Solar Simulator



- High-Flux Solar Simulator with Automated Sample Handling and Exposure System (ASHES)
  - Four 1.8 kW lamps
    - $7.2 \text{ kW}_{\text{electric}}, 6.2 \text{ kW}_{\text{radiative}}$
  - 1100 kW/m<sup>2</sup> peak flux over 1 inch spot size



# Materials in CSP



( or why you should be nice to your local materials chemist)

At its most basic, “*everything comes down to materials.*”

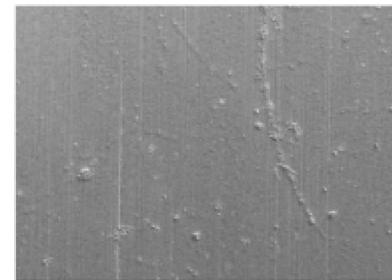
- Construction materials (alloys, ceramics, concrete)
- Coatings (absorptive, reflective, protective)
- Heat storage
- Corrosion
- Thermal conduction/expansion
- Heliostats
- Thermal stability
- Etcetera, etcetera, etcetera...



3M™ Solar Mirror Film 1100



Left:  $\text{FeCo}_2\text{O}_4$  coating



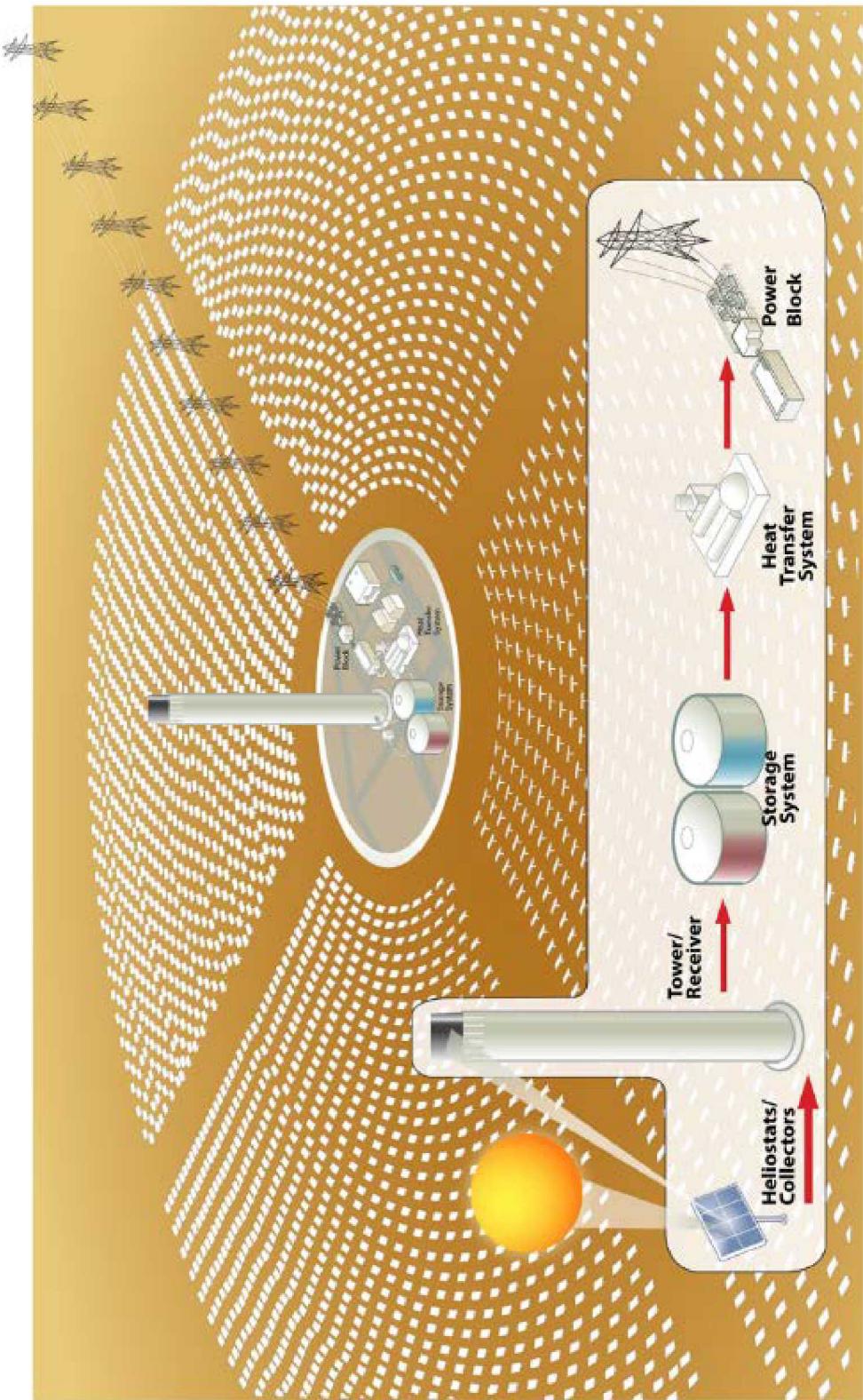
Right: SEM of  $\text{FeCo}_2\text{O}_4$  surface (Scale bar = 100  $\mu\text{m}$ )



Ceramic particle storage



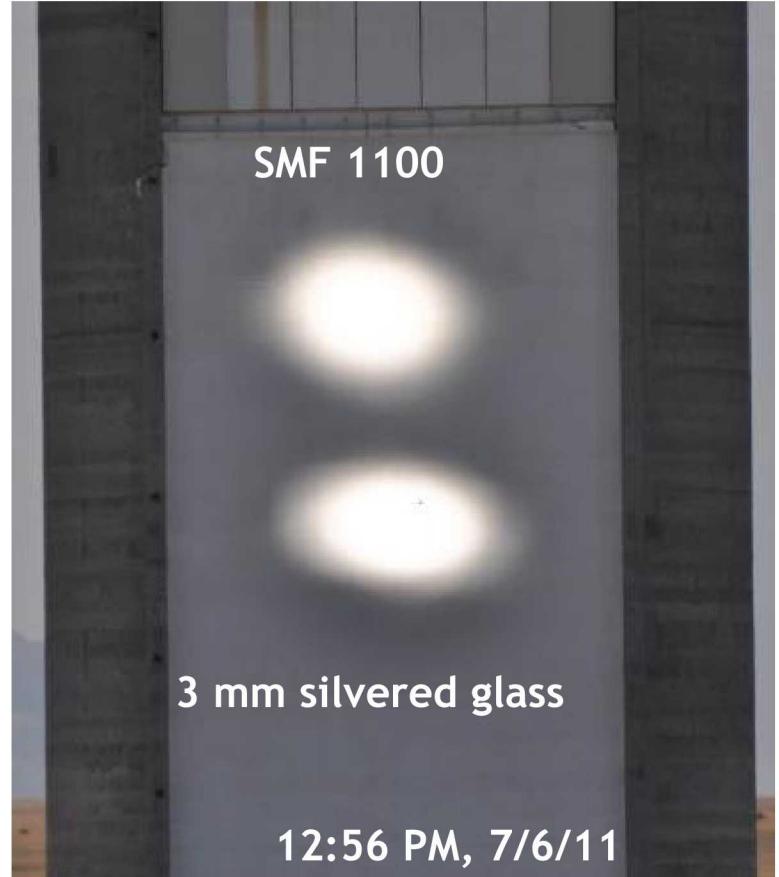
External tubular receiver



# Advanced Reflective Materials



Heliostat with 3M™ Solar Mirror  
Film 1100

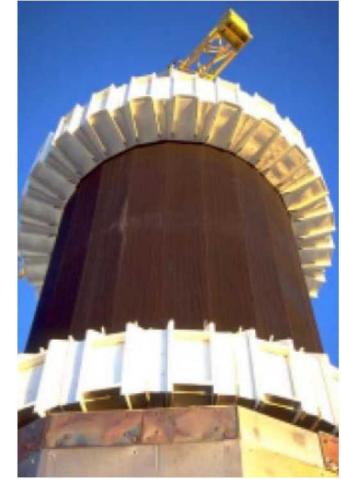


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# High-Temperature Receivers



Cavity receiver



External tubular receiver

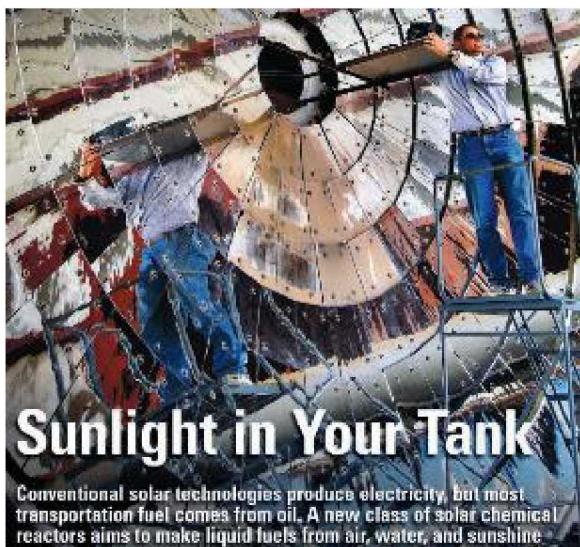
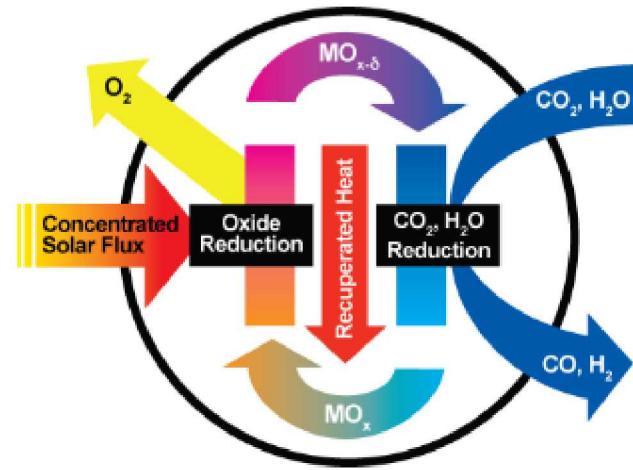
- Maximize solar absorptance and minimize heat loss (selective absorber coatings, geometry, concentration ratio)
- Need materials that operate at high temperature ( $>650$  °C) and are durable in air



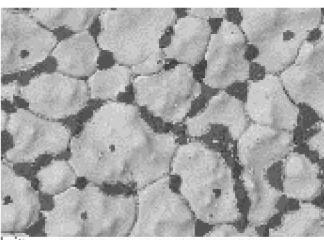
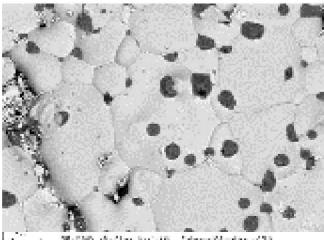
Solar absorptant coating



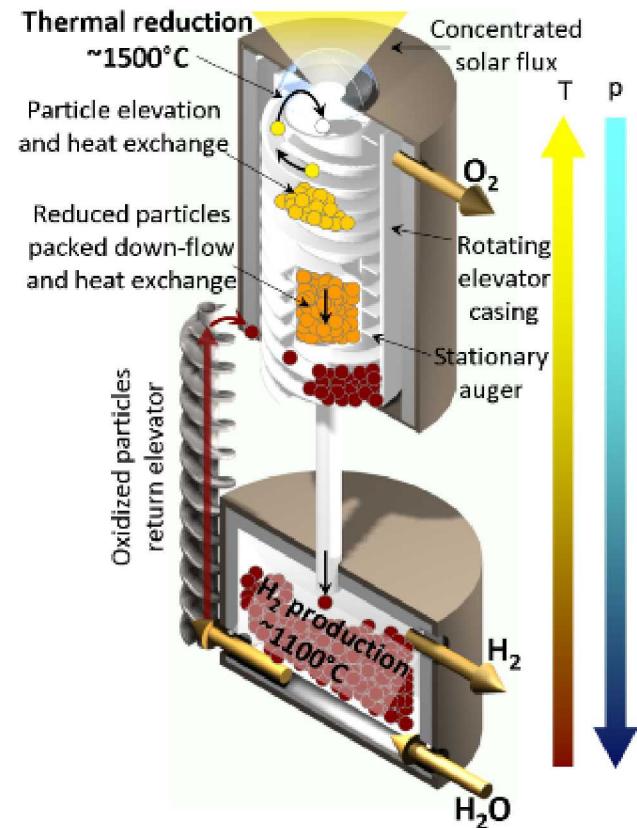
Creating hydrogen and liquid fuels with concentrated sunlight



As-sintered



Post CO<sub>2</sub> TGA

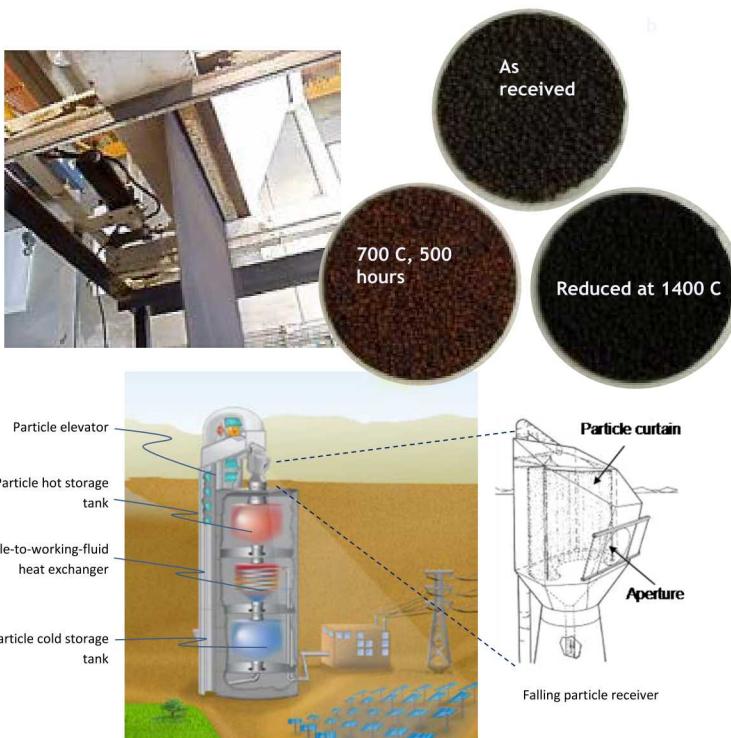


Ermanoski et al.

# Sandia Research in Thermal Energy Storage



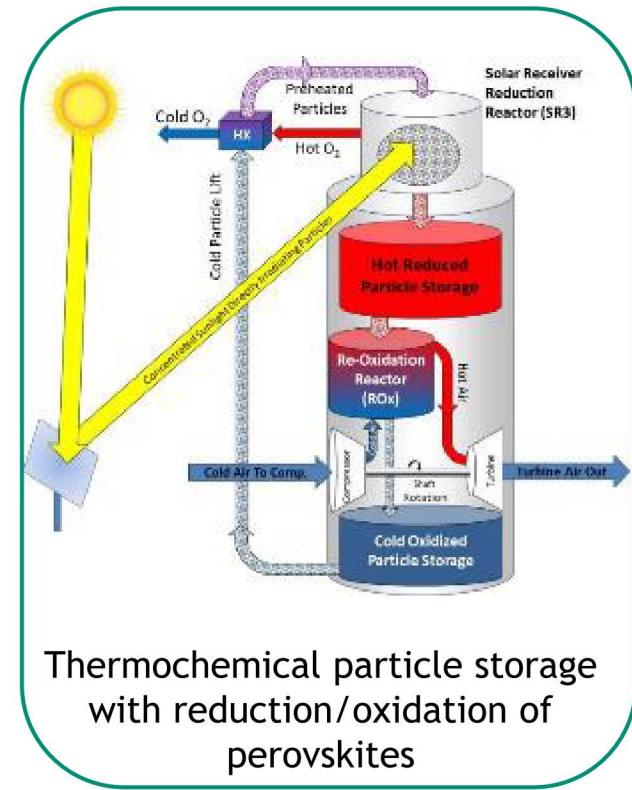
Corrosion studies in molten salt up to 700 C in “salt pots”



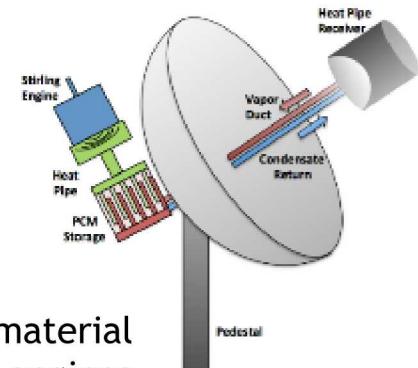
Ceramic particle storage and heating with falling particle receiver



Component testing with molten-salt test



Thermochemical particle storage with reduction/oxidation of perovskites



Latent phase-change material storage in dish engines



# High Performance Reduction/Oxidation Metal Oxides for Thermochemical Energy Storage (PROMOTES)

**Andrea Ambrosini, James Miller, Sean Babiniec, Peter Loutzenhiser, Ellen Stechel, Sheldon Jeter, Hany Al-Ansary**

**Seminar, 2 November 2018**  
**New Mexico State University**

# Thermochemical Energy Storage for CSP

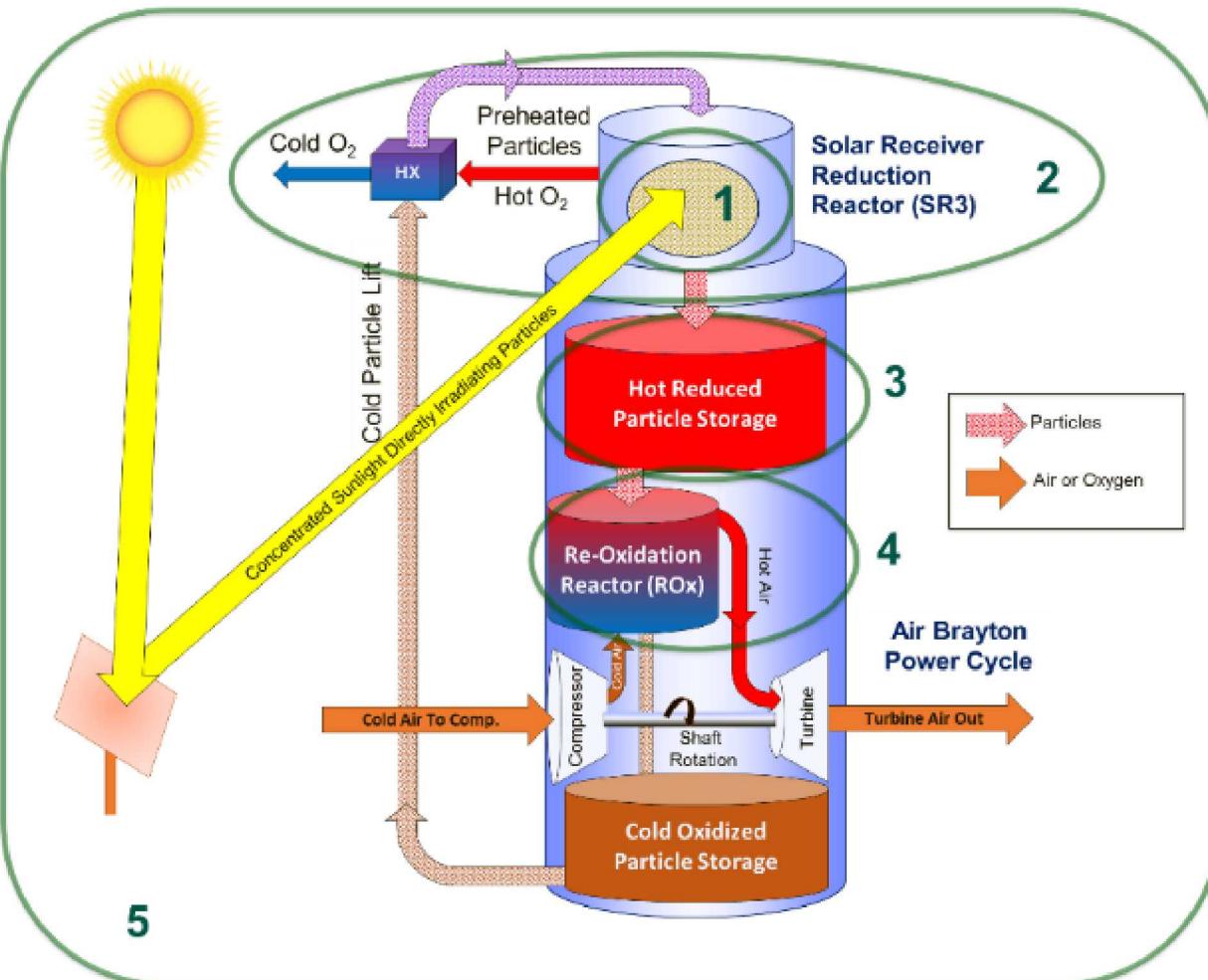


- Next-generation (Gen3) CSP calls for operation at  $T > 700 \text{ } ^\circ\text{C}$
- Current molten-salt storage systems are limited
  - Sensible-only storage, low storage densities
  - Salt decomposition limits turbine operating temperatures
- Thermochemical energy storage (TCES) offers many advantages
  - Ability store both sensible and redox reaction enthalpy, resulting in high storage densities:

$$\Delta H_{\text{tot}} = \Delta H_{\text{rxn}} + C_p \Delta T$$

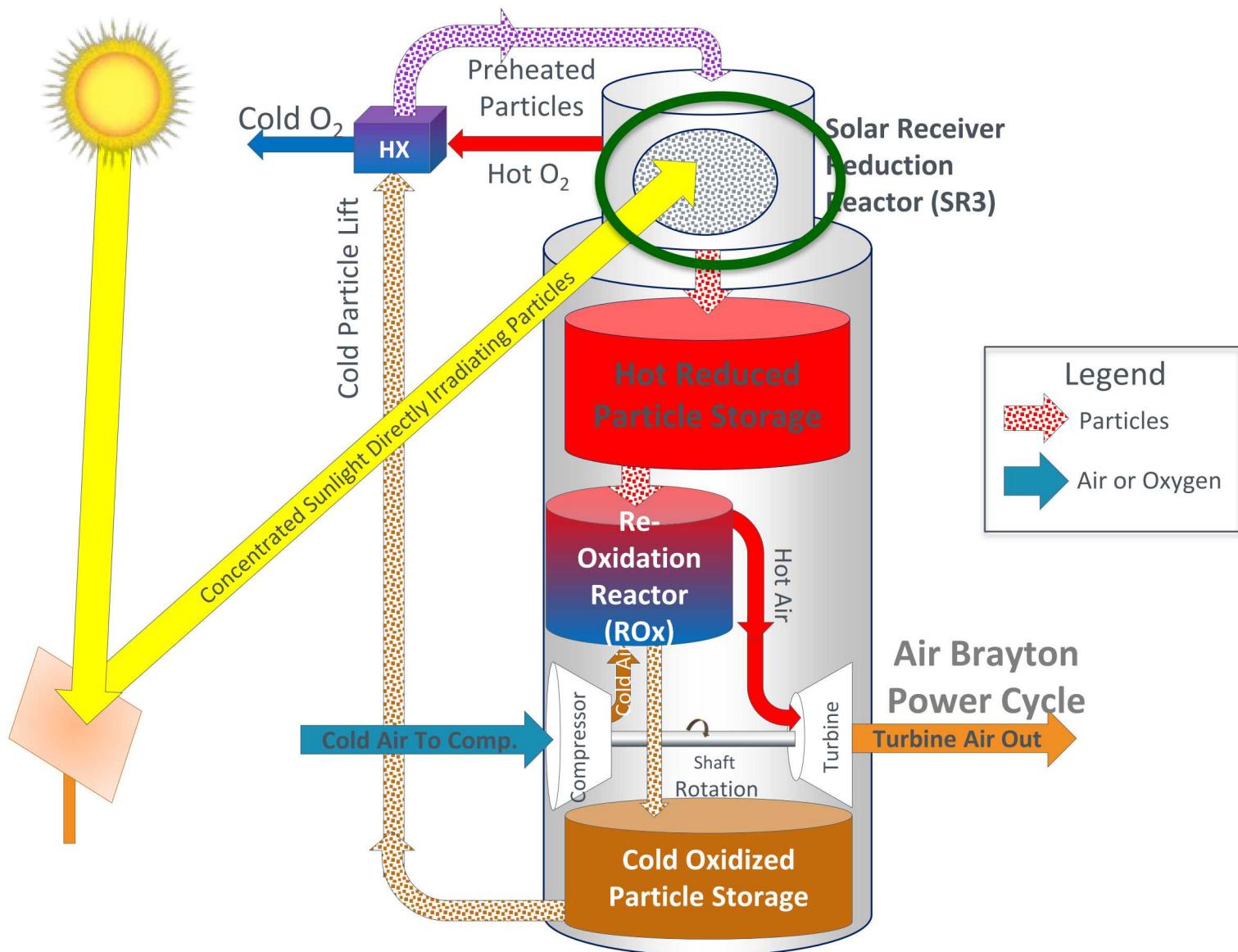
- Increased storage temperatures can enable the use of high-efficiency power cycles
  - Air Brayton inlet  $T \geq 1200 \text{ } ^\circ\text{C}$
- Direct irradiation of thermal storage media

# High Performance Reduction/Oxidation Metal Oxides for Thermochemical Energy Storage

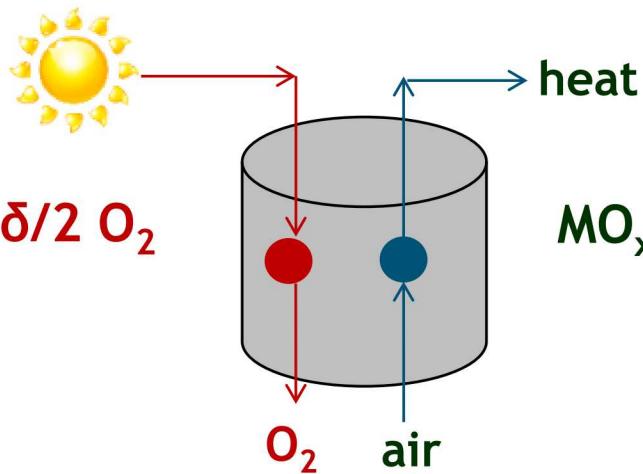


1. Materials Enabled Innovation ( $\Delta H_{\text{total}} \geq 1500 \text{ kJ/kg}$ )
2. Solar Receiver Reduction Reactor
3. Particle Storage at  $T > 1000 \text{ }^{\circ}\text{C}$
4. Pressurized oxidation reactor. Air acts as reactant and heat transfer fluid. Open cycle - no gas storage
5. Systems and technoeconomics to predict cost and performance and to guide development efforts

# I. Materials



Metal oxides are ideal materials for storage in high temperature cycles



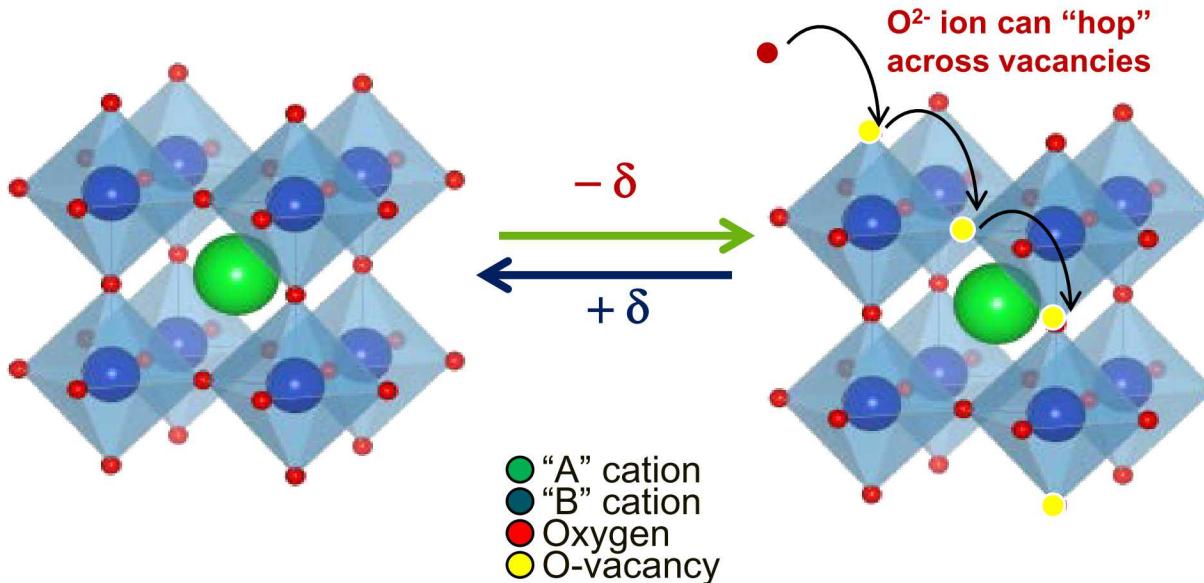
### Advantages of Metal Oxides (MO):

- Open or closed configurations
- Air can act as both the reactant and heat transfer fluid
- Environmentally benign
- No catalyst necessary
- No compression required for storage
- Amenable to multiple scales and temperature ranges

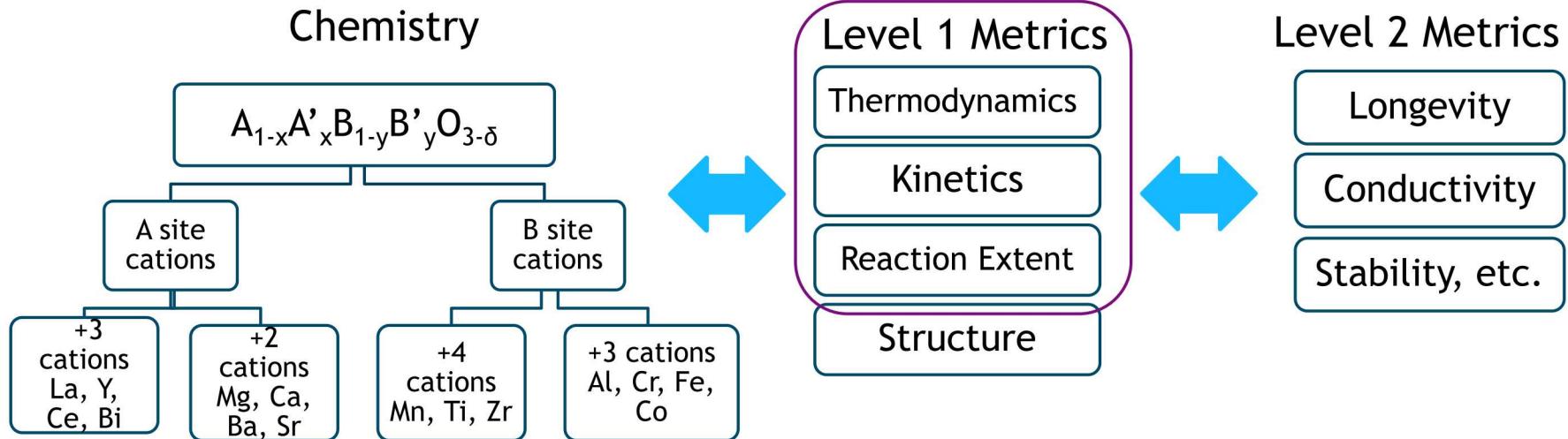
# Mixed Ionic-Electronic Conducting (MIEC) Perovskites



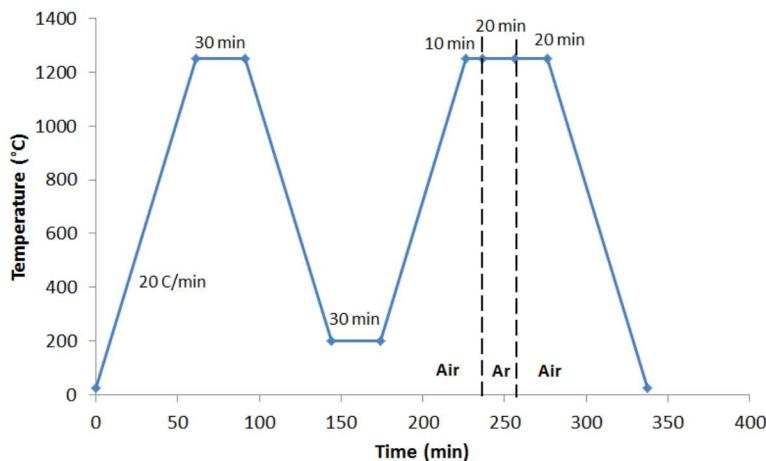
- No crystallographic phase change occurs during redox
- Vacancies facilitate oxide ion transport
- Electronic conductivity balances ionic conduction
- Redox activity continuous over variety of T and  $\text{pO}_2$
- Can tune properties by doping on both A and B sites



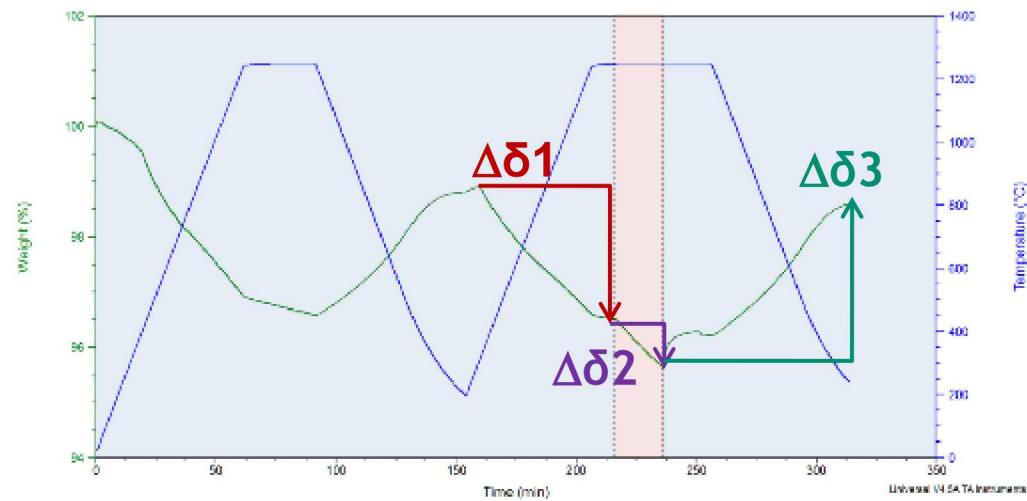
# Systematic Approach and Metrics



## Preliminary Screening – Thermogravimetric Analysis

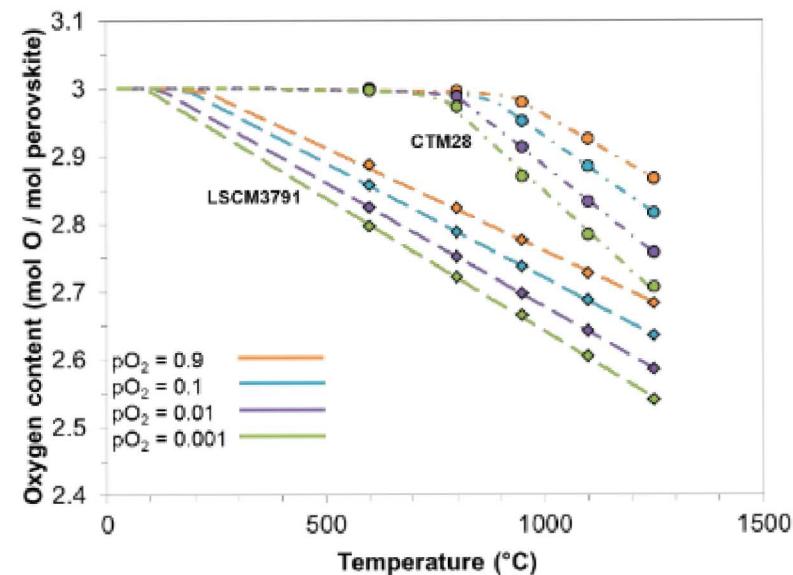
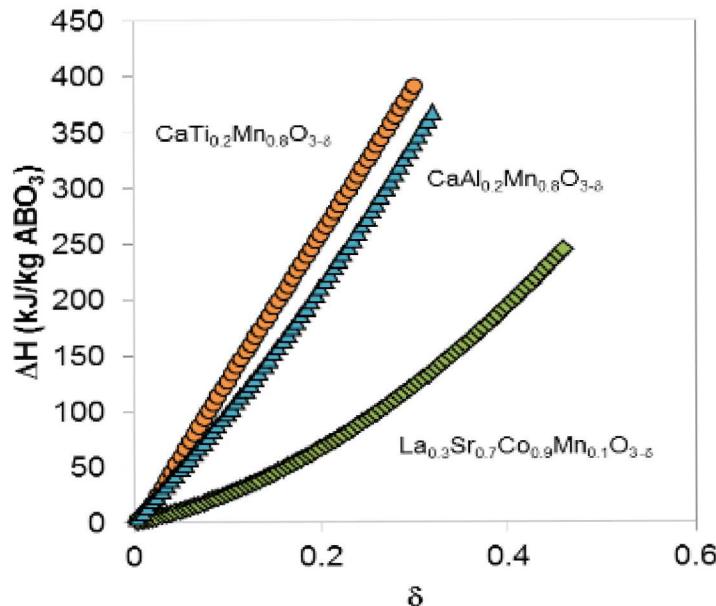


Pre-screen method



Example TGA

# Reaction Enthalpy



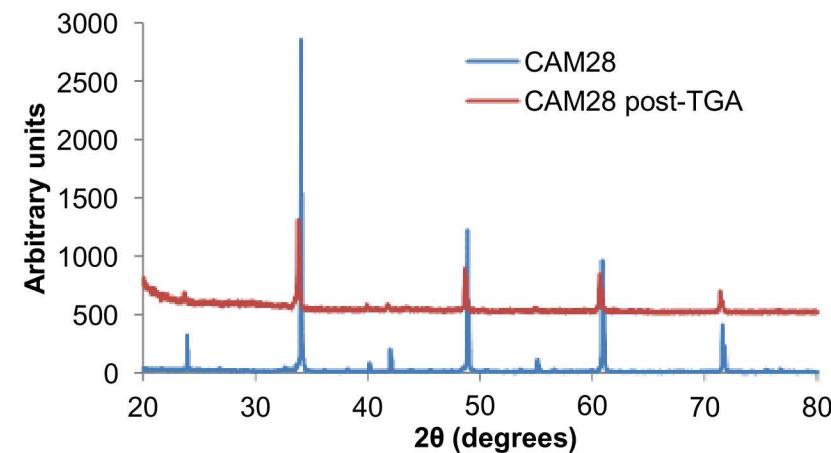
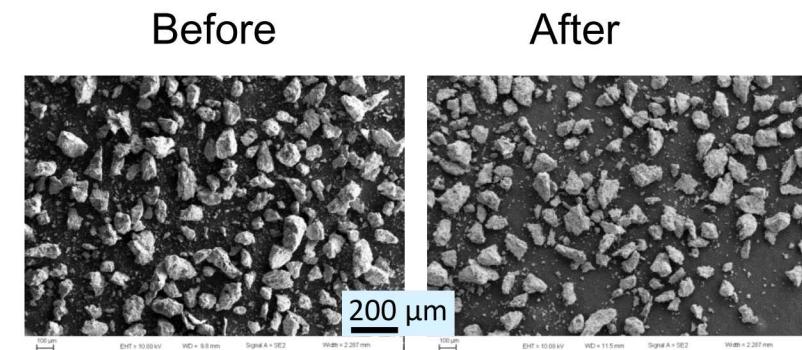
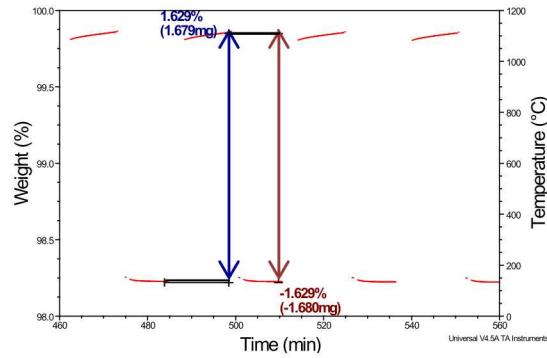
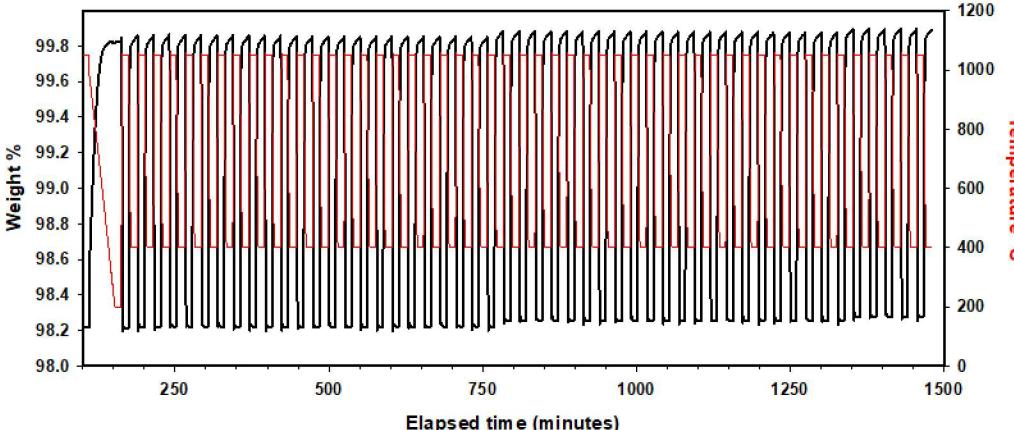
$$\Delta H_{\text{tot}} = \Delta H_{\text{rxn}} + C_p \Delta T$$

| Candidate material | Mol weight (g/mol) | $T_{\text{red}}$ Onset (°C) | Max $\delta$ | $\Delta H_{\text{rxn}}$ (kJ/kg) (at $\delta_{\text{max}}$ ) | $C_p$ (kJ/kg-K) | $\Delta H_{\text{tot}}$ (kJ/kg) |
|--------------------|--------------------|-----------------------------|--------------|---|-----------------|---------------------------------|
| LSCM3791           | 209.5              | 343                         | 0.461        | 242   | *0.595          | 837                             |
| CTM28              | 141.6              | 901                         | 0.293        | 393   | *0.881          | 1274                            |
| CAM28              | 135.8              | 759                         | 0.322        | 371   | *0.910          | 1281                            |

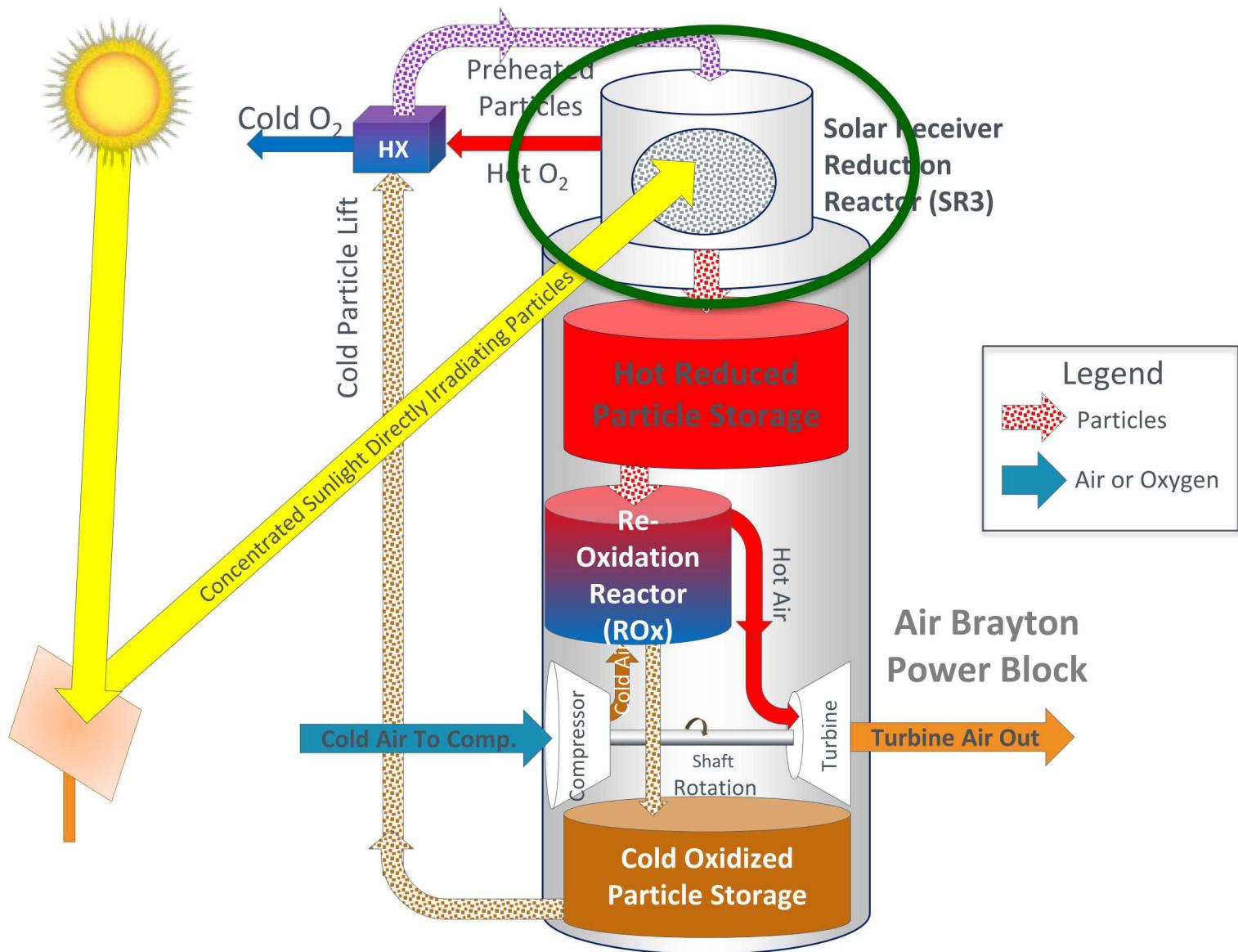
# CaAl<sub>0.2</sub>Mn<sub>0.8</sub>O<sub>3</sub> (CAM28) Cyclic Behavior



- Multi-cycle TGA used to measure cyclic repeatability and stability
  - Particles heated to 1000 °C, cooled to 400 °C for 100 cycles
- Scanning electron microscopy shows no change in morphology
- XRD shows no change in crystal phase



## 2. Solar Receiver Reduction Reactor (SR3)



# Solar Thermochemical Inclined Granular-Flow Reactor (STINGR)



- 5 kW<sub>th</sub> scale reactor designed, manufactured, tested in the Solar Fuels and Technologies Laboratory at Georgia Tech using a Xe arc-lamp high flux solar simulator
- Reactor goals:
  - Direct irradiation of energy storage media
  - Continuous, on-sun operation
  - Matched incident concentrated solar power to rate of energy storage
- Reactor goals achieved using directly irradiated, thin, dense granular flows of reactive CAM28 particles
- Reactor operated at reduced pressure using quartz-glass window, promote CAM28 reduction



Solar Fuels and Technologies Laboratory High Flux Solar Simulator at Georgia Tech

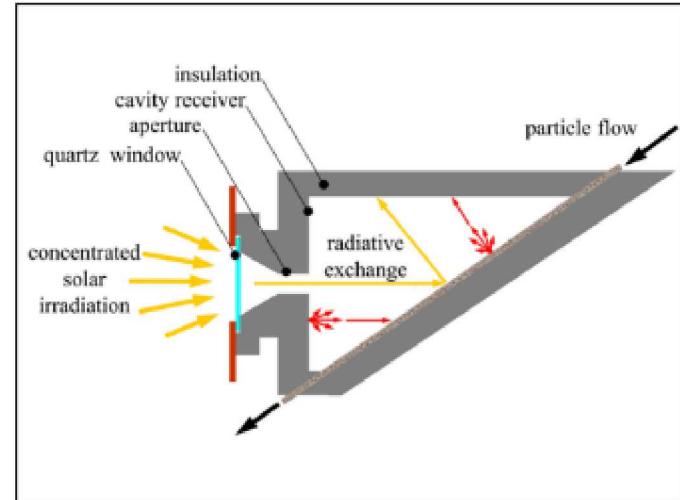
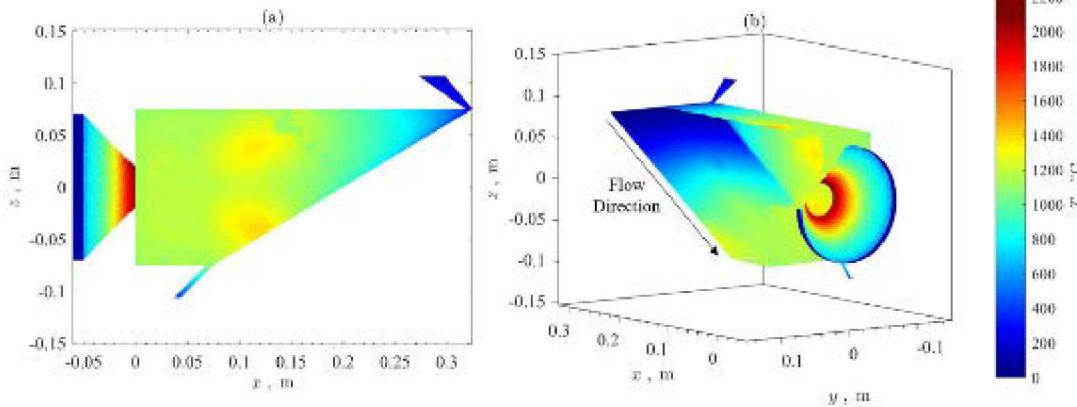


Diagram of solar thermochemical inclined granular-flow reactor

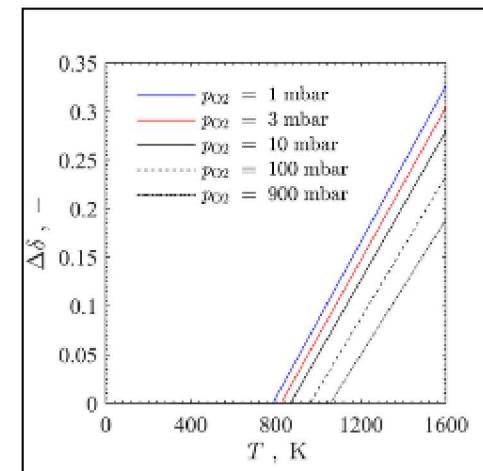
# STINGR Heat and Mass Transfer Modeling



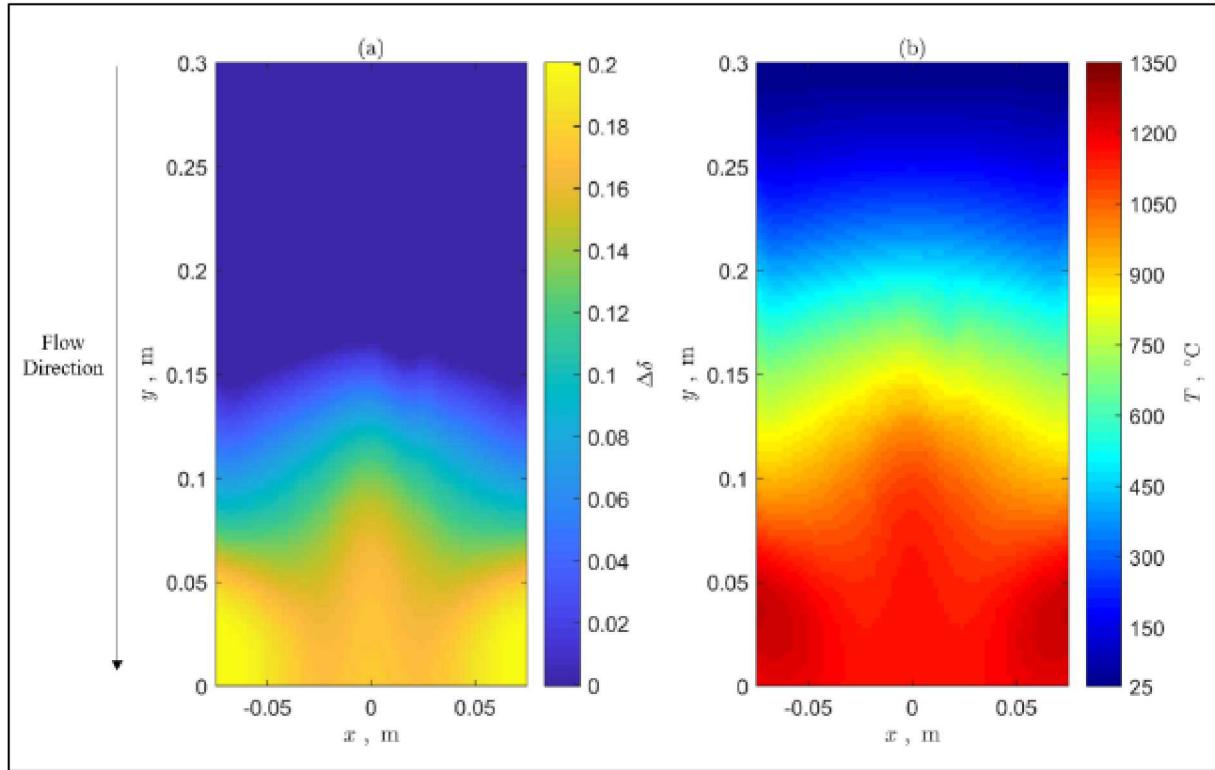
- Combined heat and mass transfer model developed to gain further insight into performance of STINGR
  - Energy storage, transport of dense granular flow of CAM28 particles
  - Radiative exchange, absorption incident radiation from HFSS in reactive dense, granular flow
- Coupled, radiation models
  - Directional, radiative input - Monte Carlo Ray Tracing
  - Inter-cavity radiative exchange - Discretized radiative transport equation
  - Optically thick dense, granular particle flow - Rosseland diffusion approximation
- CAM28 reduction assumed thermodynamically limited, characterized from prior Van't Hoff equilibrium analysis



Temperature contours for steady-state model as viewed from the side (left) and above (right)



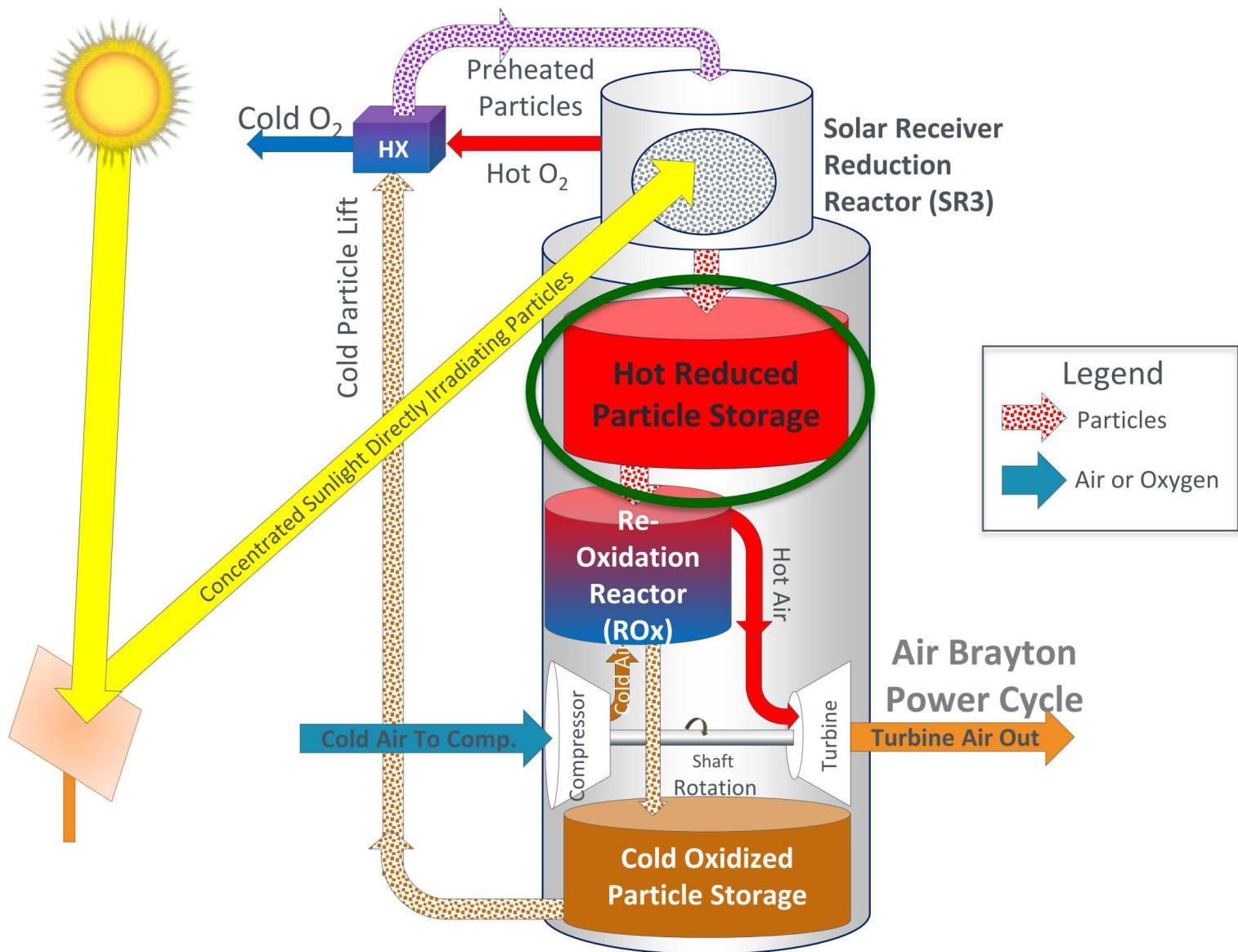
Fitted response surface for CAM28 stoichiometric deviation as a function of T and  $pO_2$



Contours of CAM28 deviation from stoichiometry (left) and CAM28 temperature (right) for dense, granular particle flow within the solar thermochemical inclined granular-flow reactor

| Performance Parameters             | Value     |
|------------------------------------|-----------|
| $\bar{T}_{\text{particle,outlet}}$ | 1173.6 °C |
| $\Delta\delta_{\text{outlet}}$     | 0.183     |
| $\eta$                             | 0.701     |

### 3. Hot Particle Storage



# Hot Particle Storage

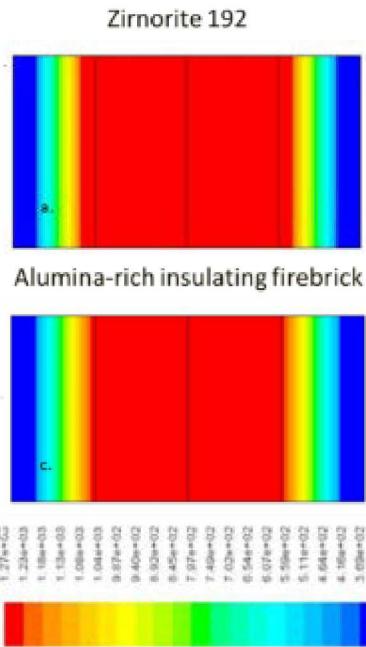


## Design, cost, & thermal analysis of inert atmosphere hot particle storage bin

|  | Internal bin temperature |           |
|--|--------------------------|-----------|
|  | 1000°C                   | 1350°C    |
| Temperature range in IFB (°C)              | 817-1000                 | 1100-1350 |
| Temperature range in PC (°C)               | 162-817                  | 209-1100  |
| Temperature range in EB (°C)               | 63-162                   | 74-209    |
| Temperature range in RC (°C)               | 45-63                    | 51-74     |
| Rate of heat loss (kW)                     | 111                      | 152       |
| Heat loss to nitrogen (GJ)                 | 2.0                      | 2.7       |
| Total energy loss over storage period (GJ) | 5.2                      | 4.4       |
| Percentage loss of energy content          | 0.12%                    | 0.18%     |



Bin wall construction



Chemical compatibility of insulating materials with MIECs:  
Zr-rich liners offer improved chemical resistance with thermal performance similar to conventional alumina firebrick

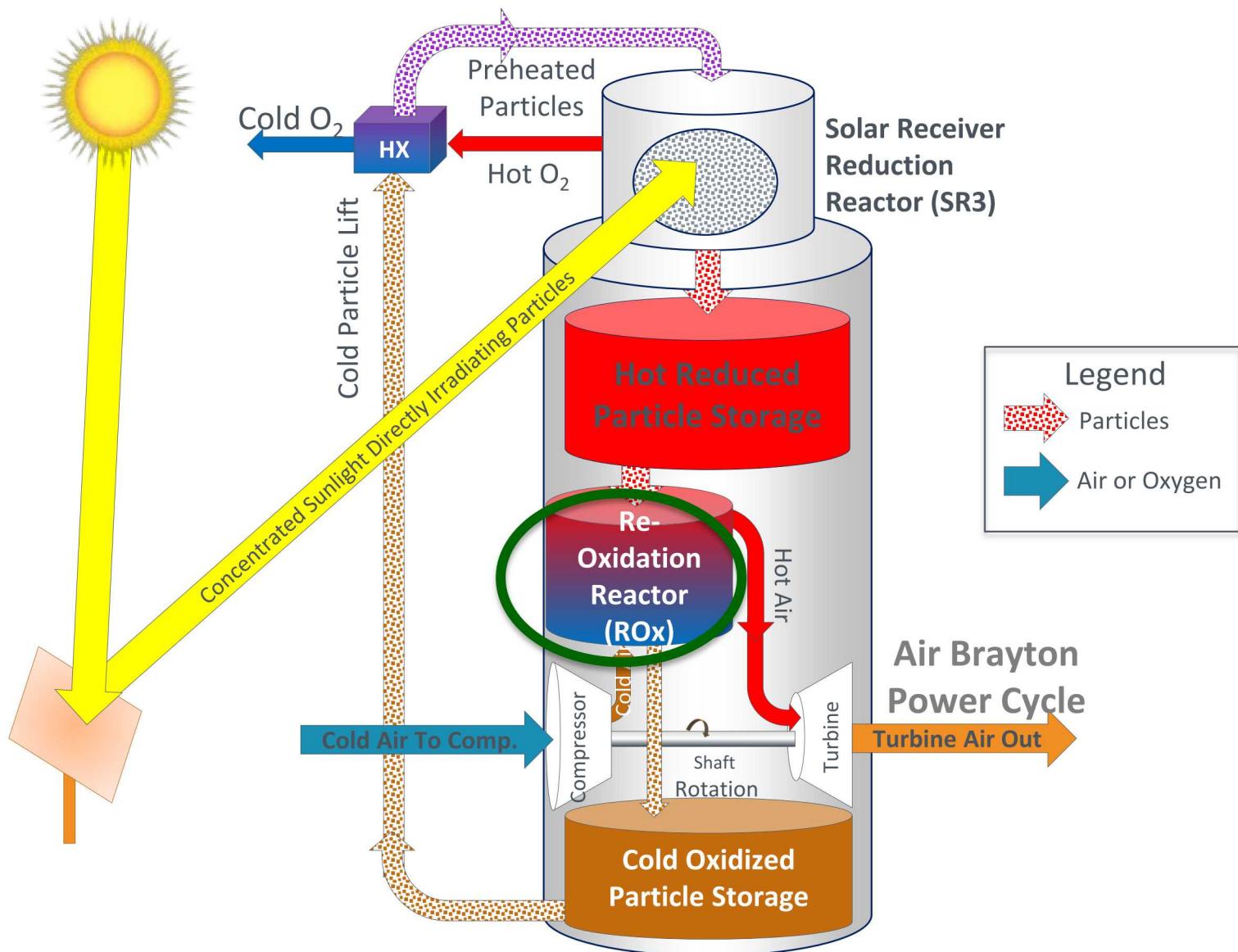
|  | SRI HF-IB 1260 | ZIRMUL | Zirnorite 699 | Zirnorite 192 | Silicon Carbide |
|--|----------------|--------|---------------|---------------|-----------------|
| $\text{Fe}_2\text{O}_3$                      | R              | R      | NR            | NR            | NR              |
| $\text{La}_{0.8}\text{Sr}_{0.2}\text{CoO}_3$ | R              | R      | NR            | NR            | NR              |
| $\text{SrFe}_{0.5}\text{Co}_{0.5}\text{O}_x$ | R              | R      | I             | I             | I               |
| $\text{CaO}$                                 | R              | -      | -             | NR            | NR              |
| $\text{MgO}$                                 | R              | -      | -             | NR            | I               |
| $\text{CaAl}_{0.2}\text{Mn}_{0.8}\text{O}_3$ | R              | -      | -             | NR            | NR              |
| $\text{CaTi}_{0.2}\text{Mn}_{0.8}\text{O}_3$ | R              | -      | -             | NR            | NR              |

R = Reactive; NR = Non-reactive; I = Inconclusive; - = Not tested



Characterizing oxidation resistance of duct materials

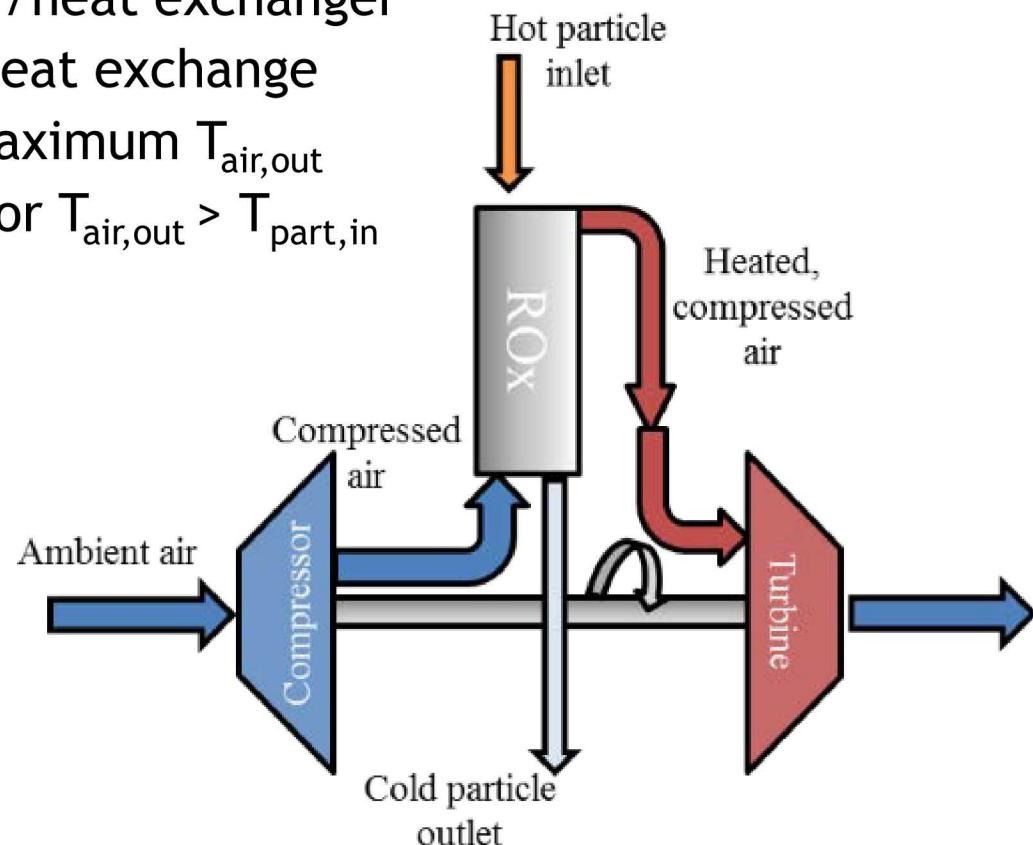
## 4. ReOxidation Reactor (ROx)



# Reoxidation Reactor (Rox) Operational Concept

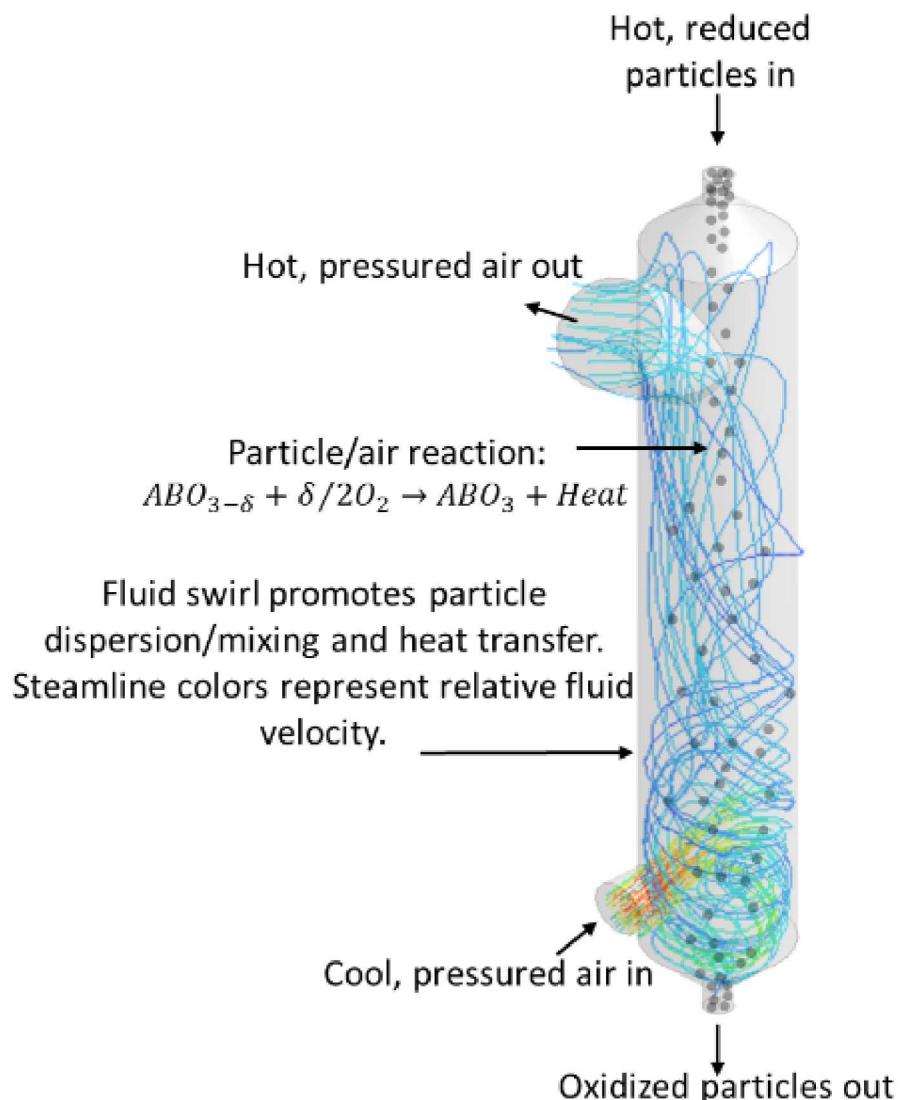


- ROx integrated into Air Brayton cycle
  - Replaces traditional combustor
- Air is working fluid and reactant stream ( $O_2$ )
  - Functions as a coupled reactor/heat exchanger
  - Direct-contact particle/fluid heat exchange
    - Counterflow operation for maximum  $T_{air,out}$
    - Exothermic reaction allows for  $T_{air,out} > T_{part,in}$
- Operating conditions
  - Pressure = 5-20 atm
  - Fluid inlet 300-400 °C
    - Isentropic compression
  - Particle inlet 1050-1150 °C
  - Fluid Tout  $\geq 1200$  °C



## Gravity-driven flow in vertical configuration appears most effective

- Counter-flow falling-particle design
- Flow pattern optimizes heat transfer and reaction kinetics
- Low pressure drop due to dispersed particles
- 3-D geometry developed based on constraints identified in the 1-D model and entrainment calculations
- Variations of this geometry analyzed using 3-D Fluent model
  - Outlet tube diameter and main tube diameter studied

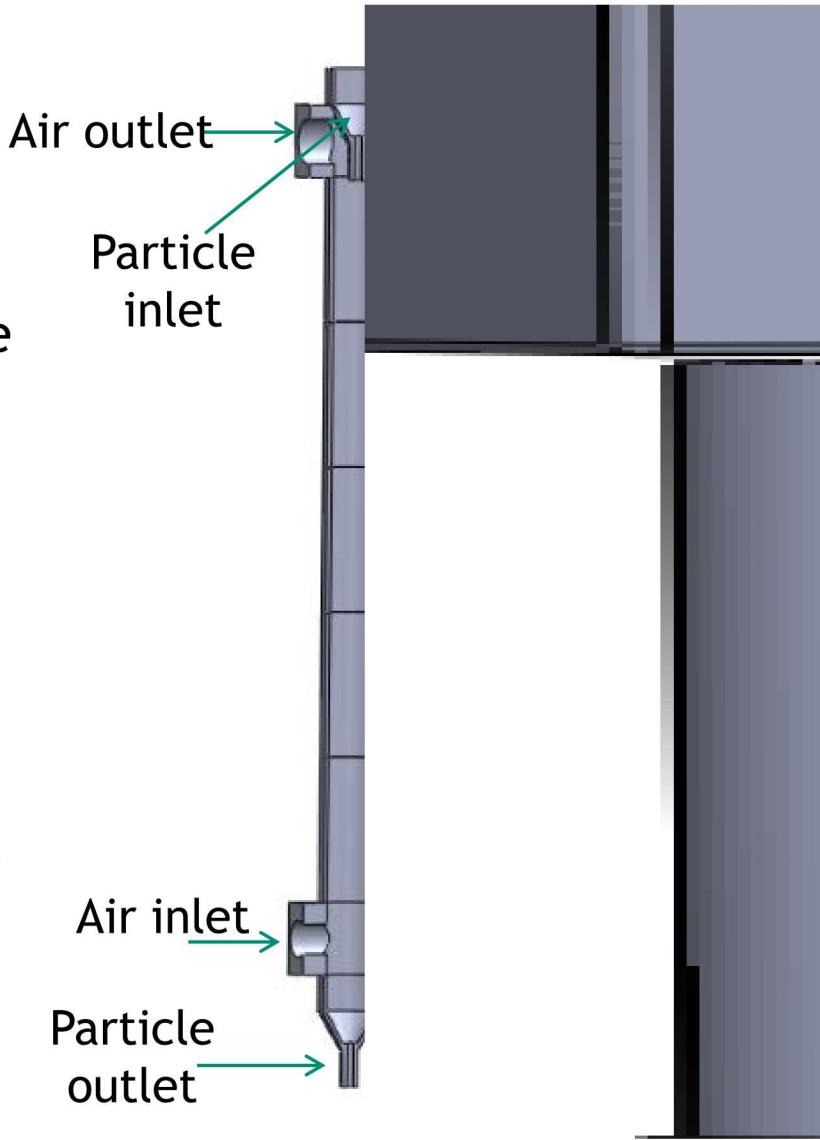


# Bench-scale 1 kW ROx Reactor

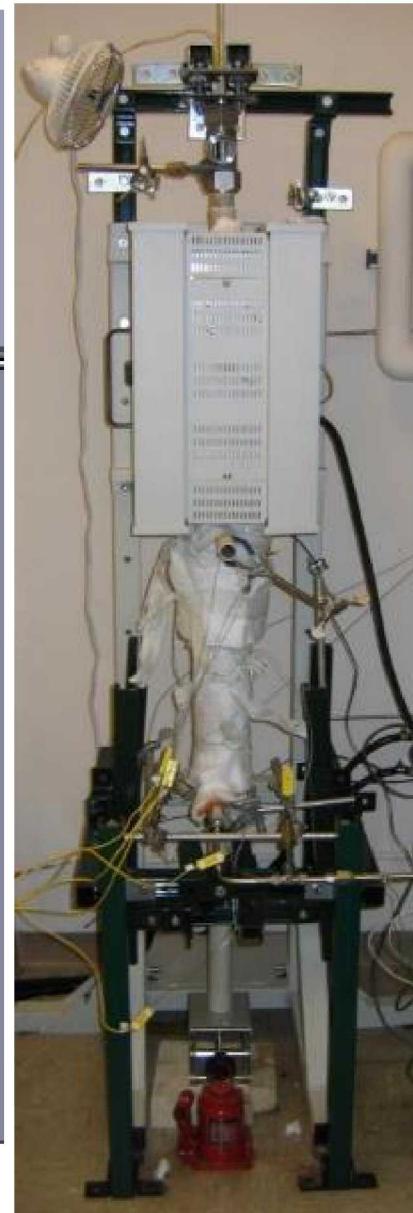


## Instrumentation:

- 1200 °C vertical tube furnace
- 1 particle reduction tube T/C
- 1 particle out T/C
- 3 fluid out T/C
- 1 “probe” T/C - insulation and exterior reactor wall
- Gas preheat up to 350 °C with T/C and control circuit
- DAQ unit for T/C data collection



Segmented reactor allows thermal expansion, ability to add/remove segments

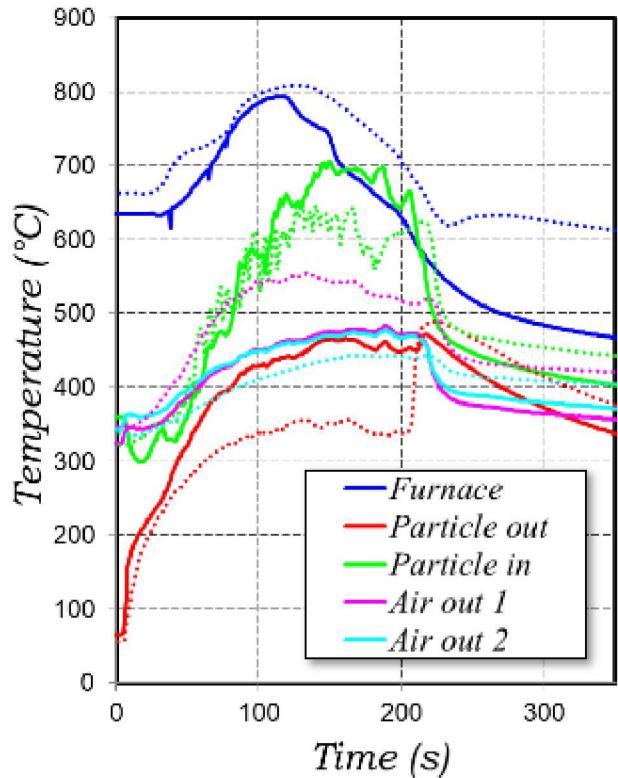


# Reactive particle testing



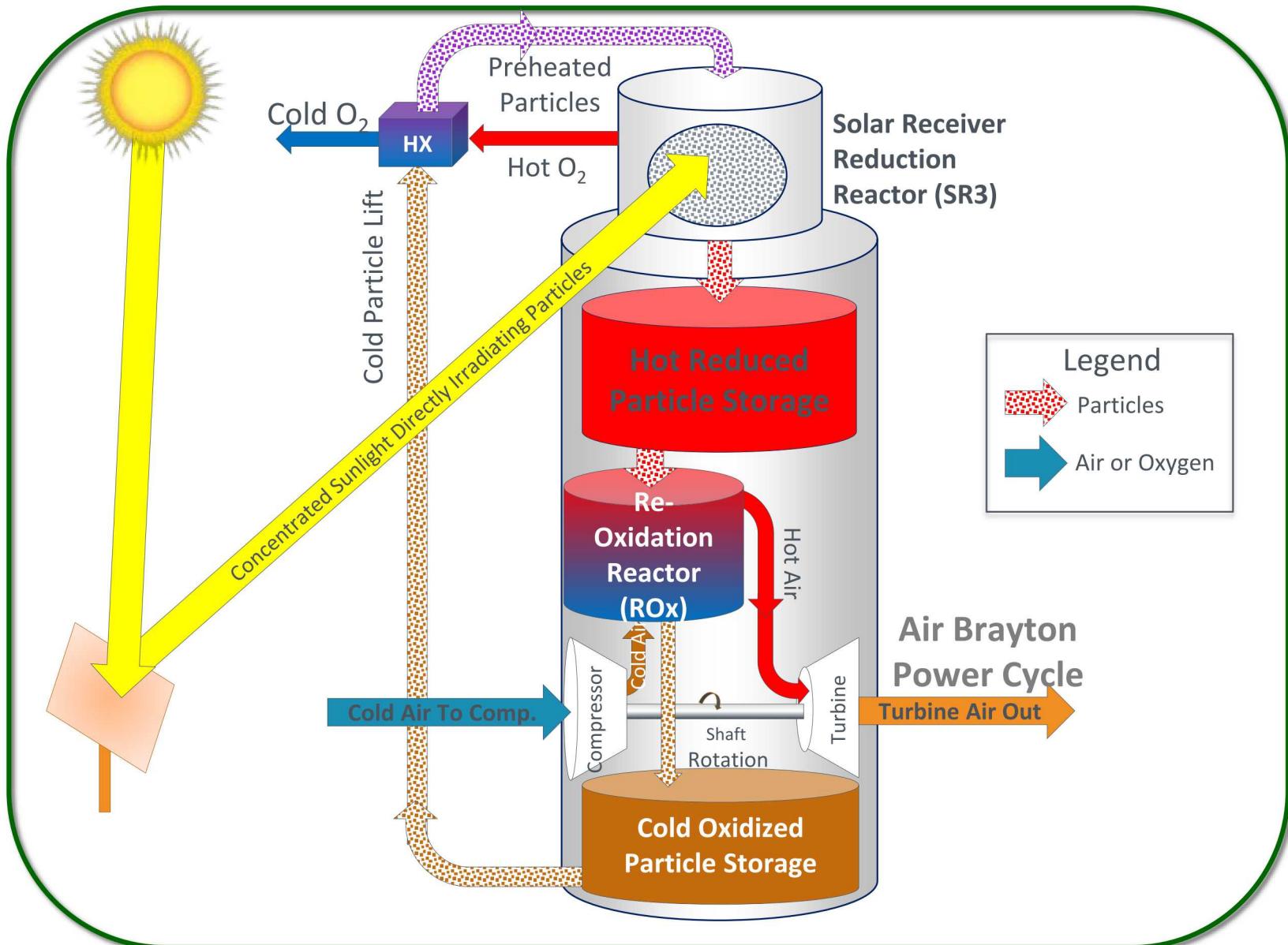
Reactive testing was performed using the  $\text{Ca}_{1-x}\text{Sr}_x\text{MnO}_{3-\delta}$  (CSM) material provided by Colorado School of Mines

- Model thermodynamics updated with published literature
- Flowability of the particles suffered at  $T > 800$  °C
  - 1000 °C runs exhibited clogging due to particle sintering
    - Attempted 3 runs, none were successful
  - Only 800 °C data was useful for model validation/calibration
    - Two runs completed with improved repeatability

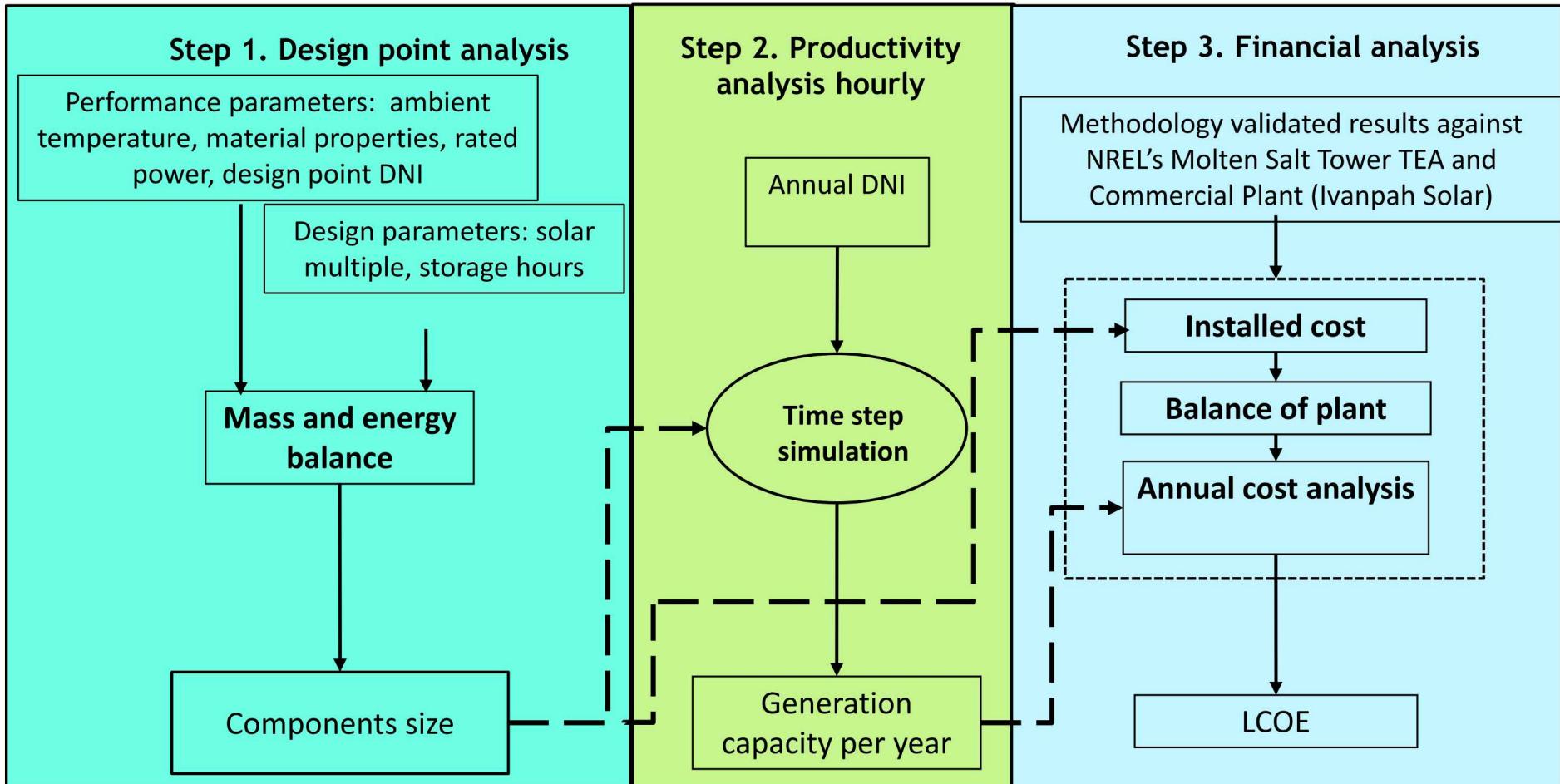


Data from these two runs used to further calibrate/validate the 1-D model

# 5. Technoeconomics and Systems



# Coupled Performance and Economic Analysis Methodology

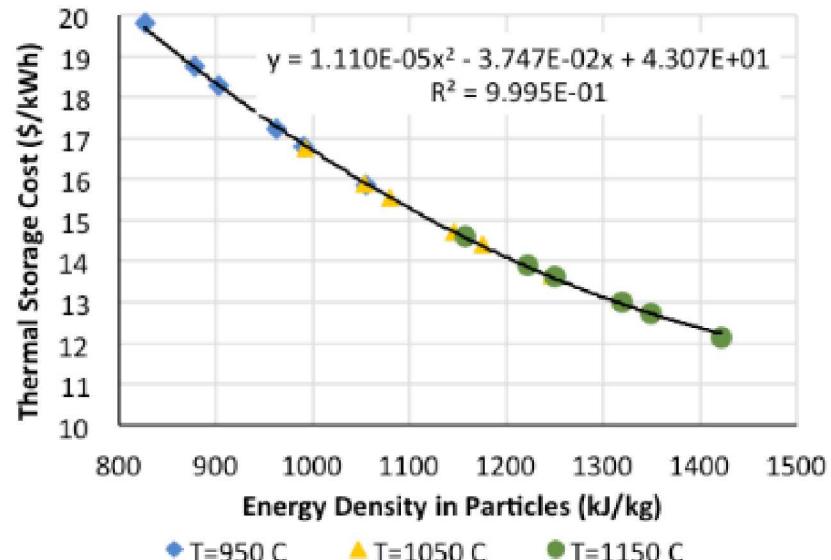


# Technoeconomic Modeling



TE and performance models at various scales are continually updated and refined as new data is available. Information shown incorporates data for CAM28 and assumes a scale of 111.7 MWe.

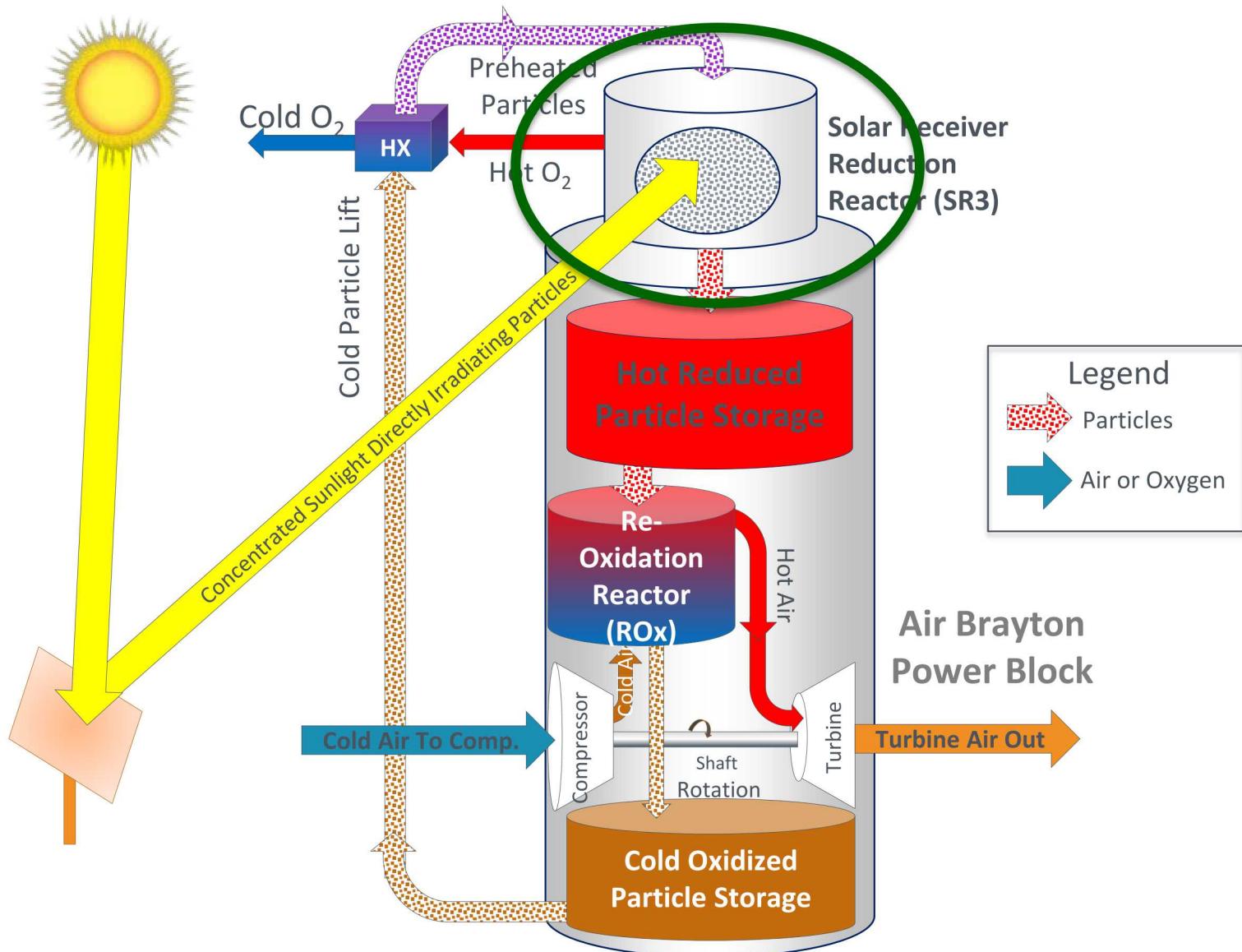
| Component List            | Cost           | % of Total |
|---------------------------|----------------|------------|
| SR3                       | \$ 31,990,464  | 8.3%       |
| Vacuum Pump               | \$ 26,597,883  | 6.9%       |
| Particles                 | \$ 11,123,973  | 2.9%       |
| Tower                     | \$ 10,967,142  | 2.8%       |
| Elevator                  | \$ 1,129,862   | 0.3%       |
| Heat Exchange             | \$ 1,865,733   | 0.5%       |
| Storage Hot               | \$ 3,593,935   | 0.9%       |
| Storage Lower Hopper      | \$ 2,355,678   | 0.6%       |
| Storage Upper Hopper      | \$ 1,247,124   | 0.3%       |
| ROx Reactor               | \$ 1,696,460   | 0.4%       |
| Controls                  | \$ 3,523,857   | 0.9%       |
| Solar Field               | \$ 68,403,311  | 17.7%      |
| Power Block               | \$ 93,583,548  | 24.3%      |
| Balance of Plant          | \$ 16,276,905  | 4.2%       |
| Contingency & Indirect    | \$ 64,519,742  | 16.7%      |
| Owner's Cost              | \$ 46,640,498  | 12.1%      |
| Multiple Components/Total | \$ 385,516,114 |            |



Storage cost as a function of energy density. Data points assume CAM 28 with the energy density varying as a function of the SR3 temperature and  $pO_2$ .

- Particle inventory sensitive to temperature of the incoming air, SR3 operation, and the fabrication factor
  - Particle cost estimated at  $\$8.50/kWh_{th}$  based on CAM28 reduced at  $1050^\circ C$  and  $200\text{ Pa}$   $pO_2$  ( $\delta=0.203$ ), residual particle heat =  $388^\circ C$  after ROx
- Storage volume scales with amount of particles, cost scales more slowly
  - Estimated at  $\$4.60/kW_{th}$

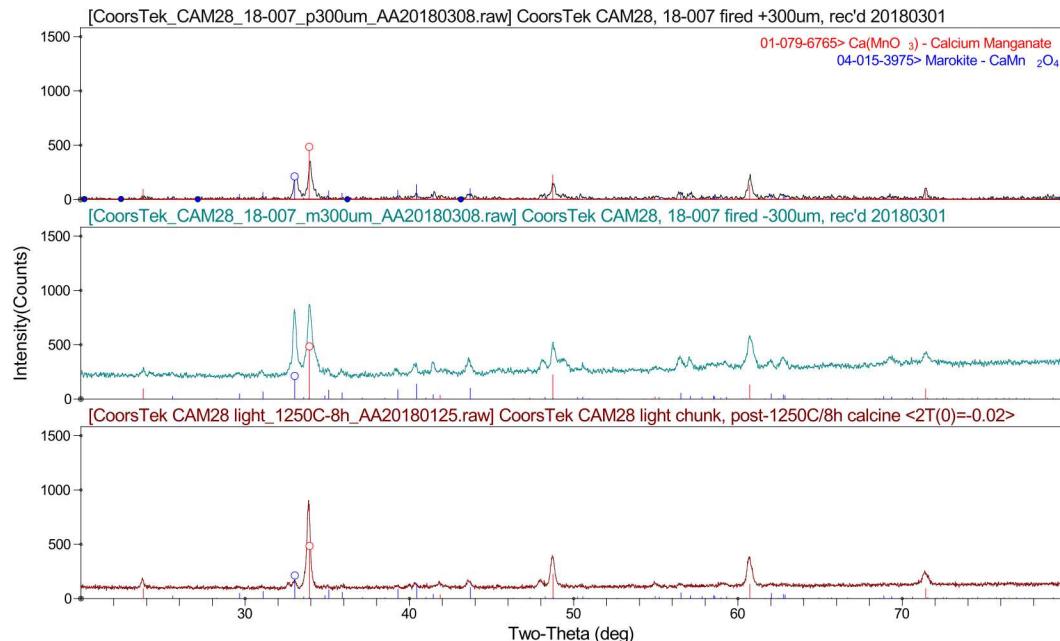
# SR3 Scale Up and On-Sun Testing



# Particle Scale-up



- Particle scale-up (250 kg) undertaken by CoorsTek
- Particles were spray-dried and calcined in air in static furnace
- Resulting particles were not single phase
  - Combination of spinel  $[\text{Ca}(\text{Al},\text{Mn})_2\text{O}_4 / \text{Ca}_2(\text{Al},\text{Mn})\text{O}_4]$  and perovskite phases
  - Spinel is not as redox active as the perovskite, impacts the redox efficiency of the bulk material
  - Hypothesis: not enough  $\text{O}_2$  reached the bottom of the reaction crucible resulting on oxygen-deficient spinel phase
- Exact effect of mixed phase on 100 kW on-sun test is uncertain

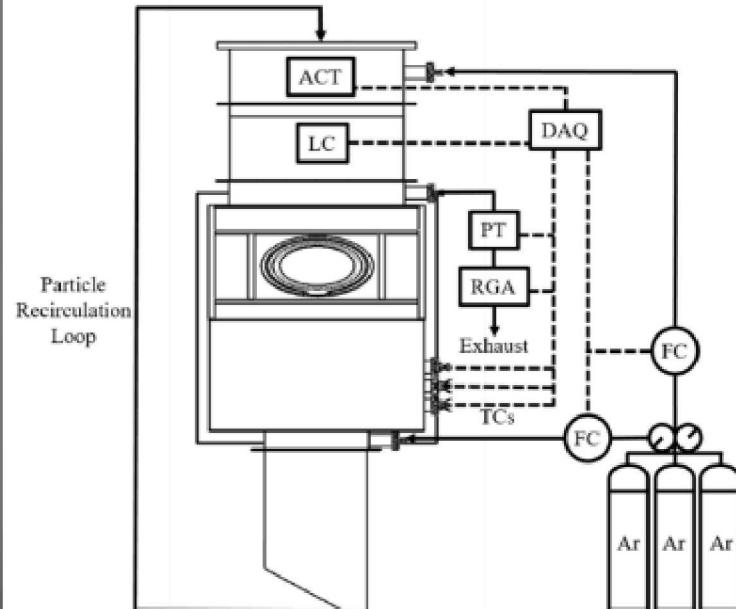
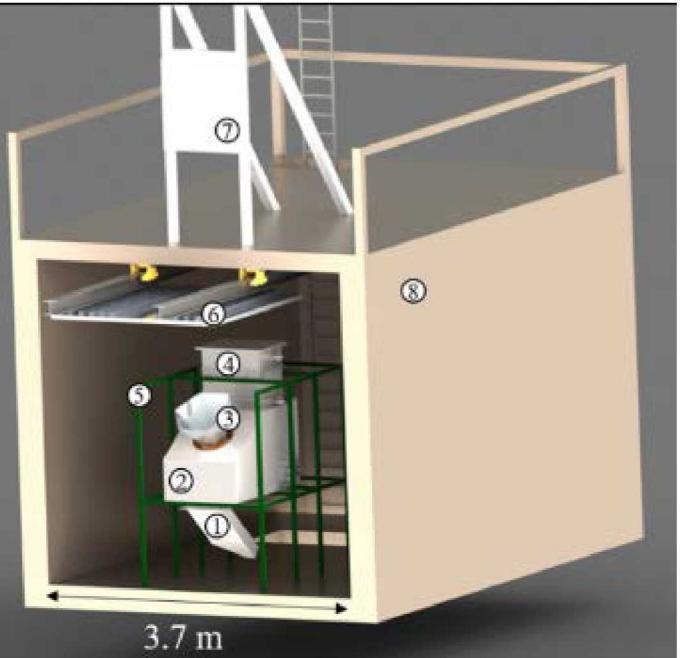


# Scale-up 100 kW<sub>th</sub> Reactor Design



- 100 kW<sub>th</sub> scaled-up reactor designed, to be tested at Riyadh Techno Valley Tower at King Saud University
- Due to scale, reactor operated in atmospheric pressure, inert Ar environment

- 1) Lower Collector
- 2) SR3 Module
- 3) Secondary Concentrator
- 4) Upper Hopper
- 5) Support Structure
- 6) Beam Down Mirror
- 7) Heliostat Positioning Target
- 8) Test Module

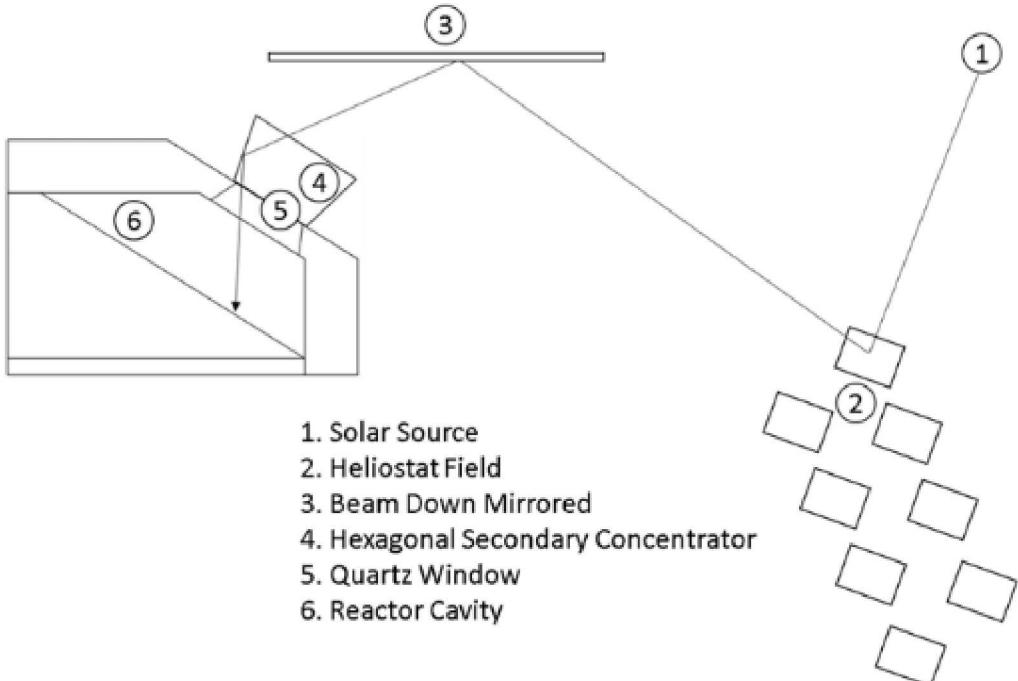


Solidworks rendering of 100 kW<sub>th</sub> design mounted reactor design atop Riyadh heliostat field (left) and experimental diagram (right) of 100 kW<sub>th</sub>

# Scale-up 100 kW<sub>th</sub> SR3 – Radiation Modeling



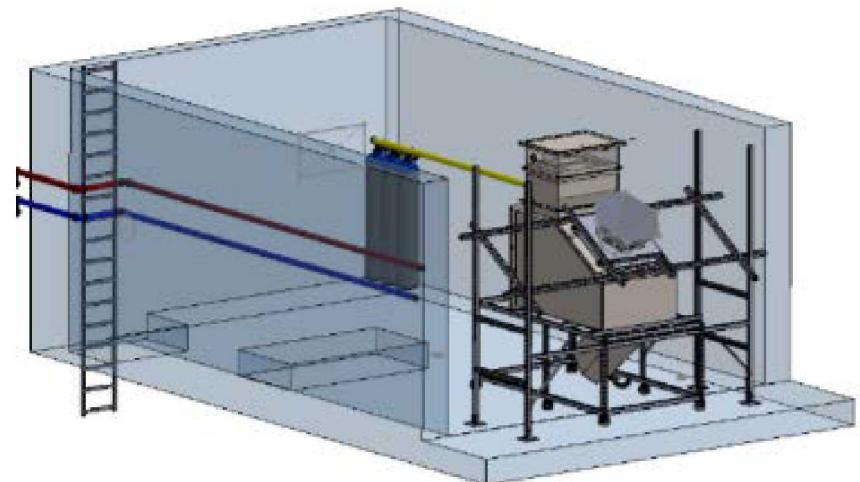
- Monte Carlo Ray Trace Model developed to capture radiative exchange from the heliostat field to the reactor cavity
- Model used to improve designs of mirrored subsystems and cavity geometry
- Model used to predict temporal, absorbed incident heat flux distributions upon inclined plane



Schematic of 100 kW<sub>th</sub> field including major components participating in radiative exchange



# Promotes Module Layout



# Summary



- We have discovered and characterized a family of redox active MIEC oxides , CXM, which exhibit total enthalpies > 1200 kJ/kg
  - Stable at high temperatures
  - Reproducibly cycled with little loss in performance
  - Comprised of earth abundant elements
  - To our knowledge, these materials outperform any reported oxide TCES material operating above 1000 °C
- The Solar Thermochemical Inclined Granular-Flow Reactor (STINGR) constructed at Georgia Tech and tested to validate models
- Designed storage bins and identified MIEC compatible liner materials
- A counter-flow falling-particle Re-oxidation (ROx) reactor was designed and constructed at lab scale
  - Multi-phase, thermo-fluid modeling accomplished in ANSYS Fluent
- Techno-economic modeling is underway, with constant refinement as new data is obtained
  - Current results show that the storage cost goal of \$15/kWhth is achievable
- 100 kW-scale SR3 reactor constructed and ready to test at RTV in Riyadh

# Acknowledgments



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Travis Anderson



Andrew Schrader, Clayton Nguyen, Robert Gill, Hagen Bush



Nathan Johnson, Brandon Gorman, Mariana Lopez,  
Briana Lucero



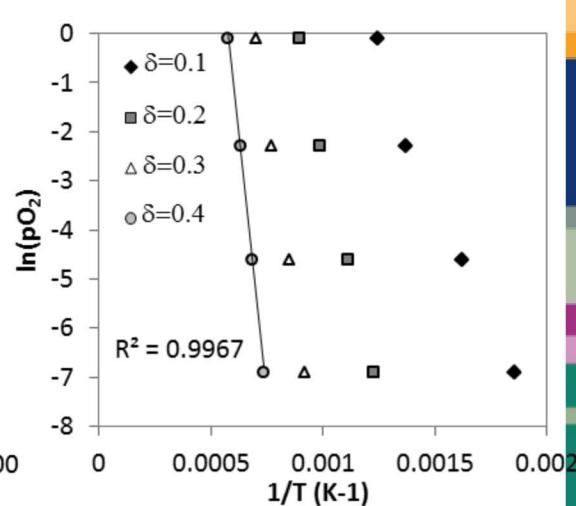
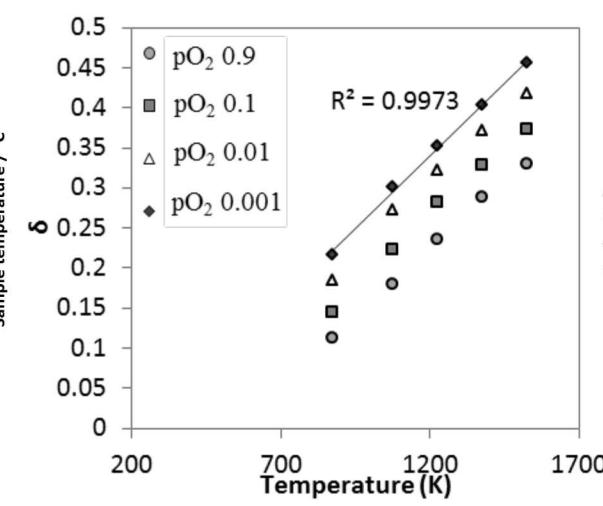
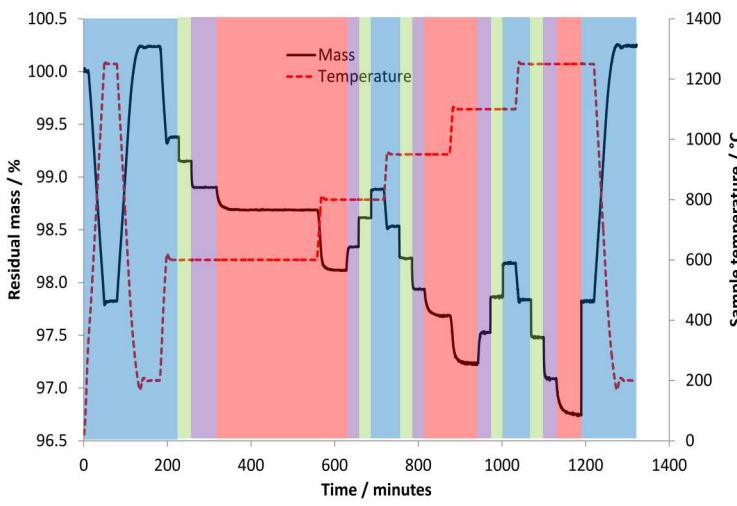
- Prof. Greg Jackson (Colorado School of Mines)
- Devin Clay (CoorsTek)



# High-resolution equilibrium TGA



- Used to estimate thermodynamic parameters
- Isothermal holds at 600, 800, 950, 1100, and 1250 °C;  $pO_2$  varied at each temperature and held until equilibrium
- Thermodynamic parameters extracted by van't Hoff approach:
  - $\ln(pO_2) = 2 \frac{-\Delta G_{rxn}}{RT} = 2 \left( \frac{1}{T} \cdot \frac{-\Delta H_{rxn}}{R} + \frac{\Delta S_{rxn}}{R} \right)$
  - Enthalpy determined by slope, entropy by intercept for each value of  $\delta$



$pO_2 = 0.9$     $pO_2 = 0.1$     $pO_2 = 0.01$     $pO_2 = 0.001$

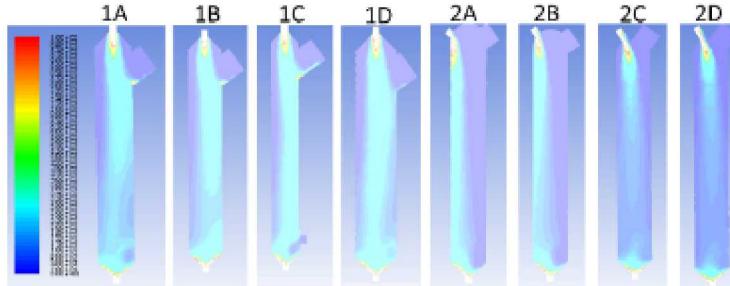
# 4. ROx: Design, Modeling, Demonstration

## Counter-flow falling-particle design

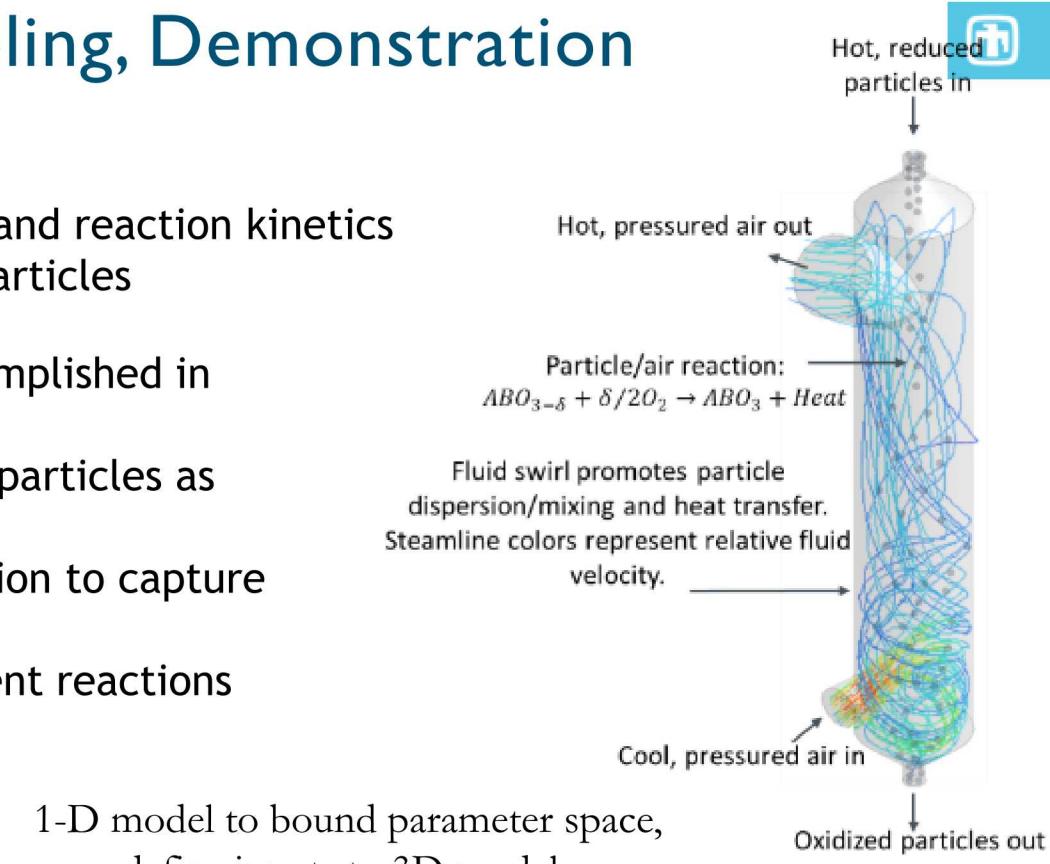
- Flow pattern optimizes heat transfer and reaction kinetics
- Low pressure drop due to dispersed particles

## Multi-phase, thermo-fluid modeling accomplished in ANSYS Fluent (right)

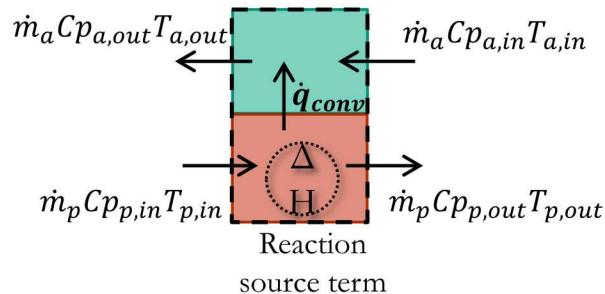
- Eulerian-Eulerian approach simulates particles as equivalent “fluid”
- Granular theory used for particle motion to capture particle-particle collisions
- Custom user-defined code to implement reactions



Fabricating a lab-scale (~2.5 kW) ROx demonstration unit - geometry optimized via Fluent modeling.



1-D model to bound parameter space, define inputs to 3D models

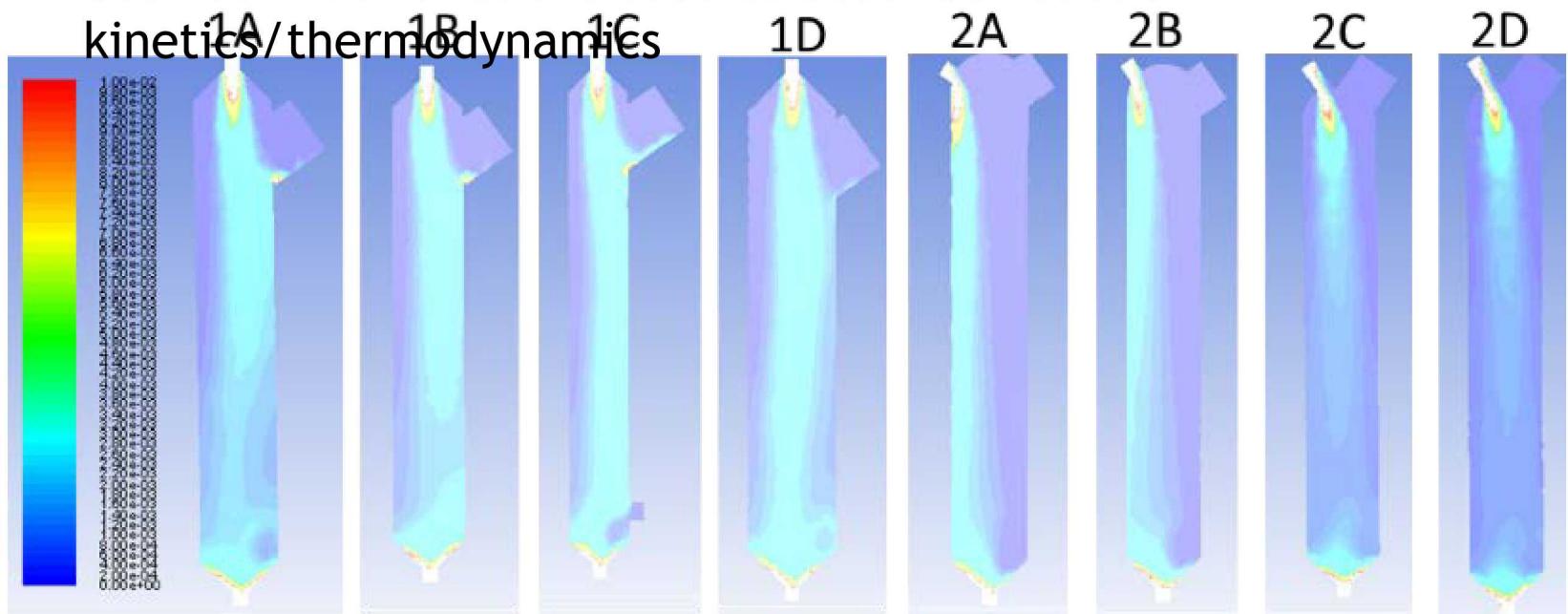


Functions of T:  $\rho_a, v_a, \mu_a, Cp_a, Cp_p, Re/Nu/Pr, h, \delta_{equil}$

# Multi-phase, thermo-fluid modeling using ANSYS Fluent applied to ROx design



- Eulerian-Eulerian approach simulates particles as equivalent “fluid”
- Granular theory implemented to describe particle motion and capture particle-particle collisions
- Multi-species approach used to quantify local oxygen consumption and particle conversion
- User-defined functions used to describe reaction kinetics/thermodynamics

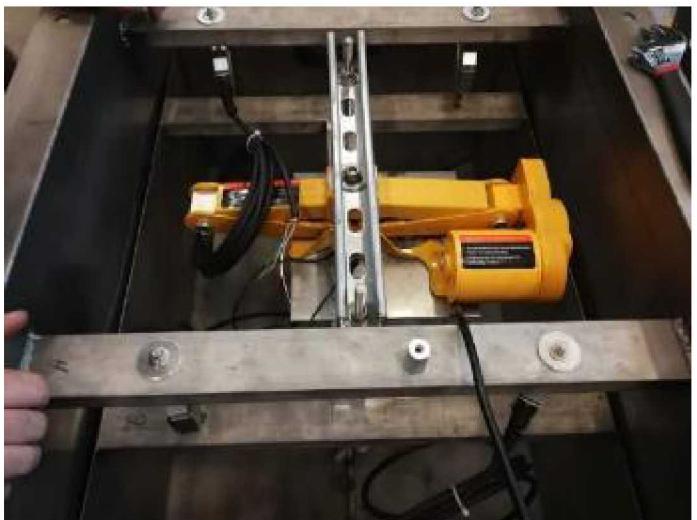


Steady-state particle volume fraction for each geometry

# Particle Lift and Flow Control



- The particles are lifted using an air conveyor
- The use of air ensures that the particles are fully oxidized and cooled prior to a new test run
- The particles are then dropped into the top hopper
- The outlet of the top hopper is controlled using a lift jack



# SR3 Test Unit



The SR3 Test unit (RIGHT) has been constructed and is under going dry run testing at Georgia Tech.

The reactor slope, and necessary insulation have been installed for the purpose of testing, seen below.



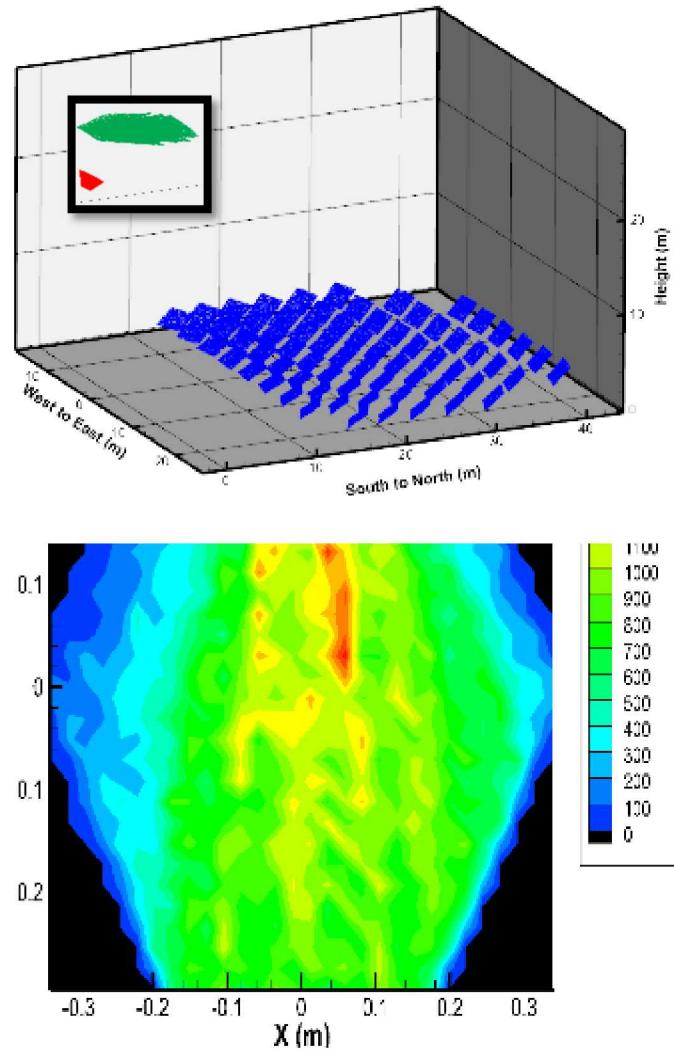
# SolTrace Modeling for the 100 kWth Reactor Design



SolTrace estimated the performance of the RTV field and informs the design parameter for the beam down mirror and the secondary concentrator.

The resulting flux maps are used to predict the irradiance into the SR3 and the expected heat loads on the beam down and secondary concentrator surfaces.

The raytracing data is being used in ANSYS to model the SR3.



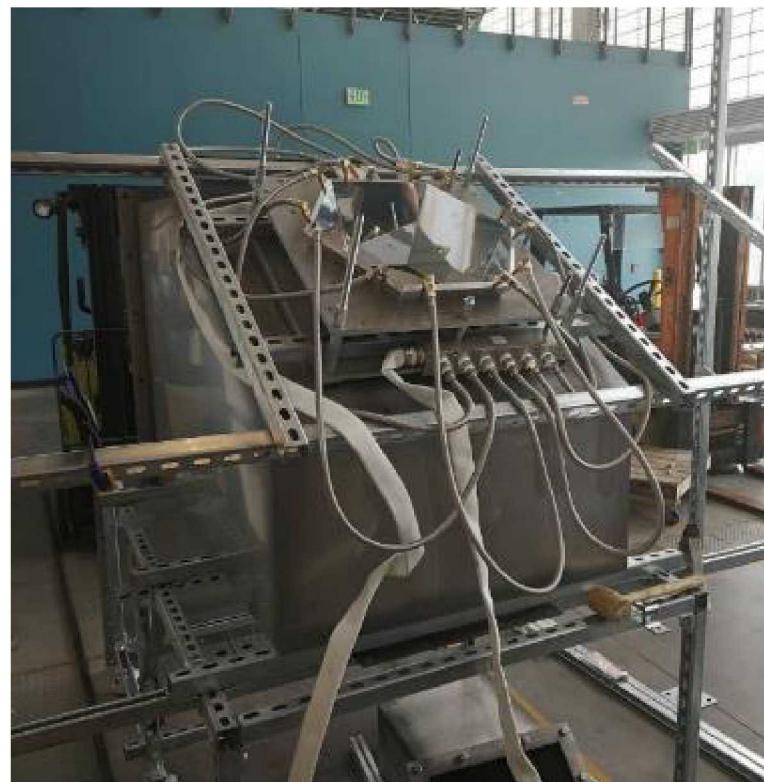
# Secondary Concentrator Design



The secondary is used to concentrate a 400 sun flux over 1 m x 1 m target to over 1000 suns in 0.3 m diameter aperture.

The secondary is made of 6 polished aluminum cold plates.

Each trapezoidal cold plate utilizes 1.5 GPM of water to keep the system adequately cooled.



# Beam Down Design



The beam down mirror is made of a series of smaller facets arranged into a 2.1 m x 2.1 m

Each facet is 0.3 m x 0.3 m and will be cooled using forced convection.

The small size and additional cooling mitigate thermal stresses to a manageable level.

