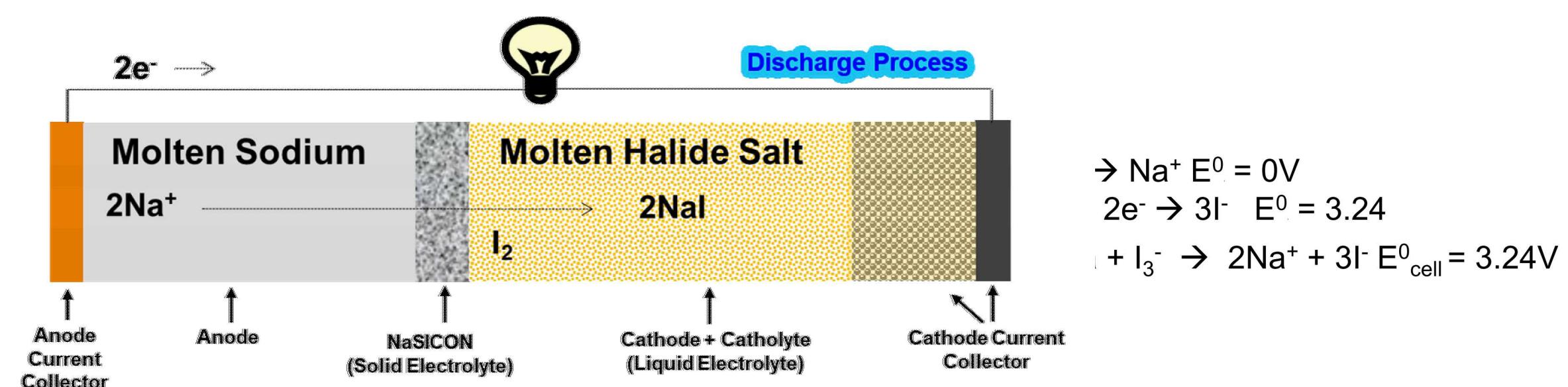




Solid State Separator Development for Sodium-Based Batteries

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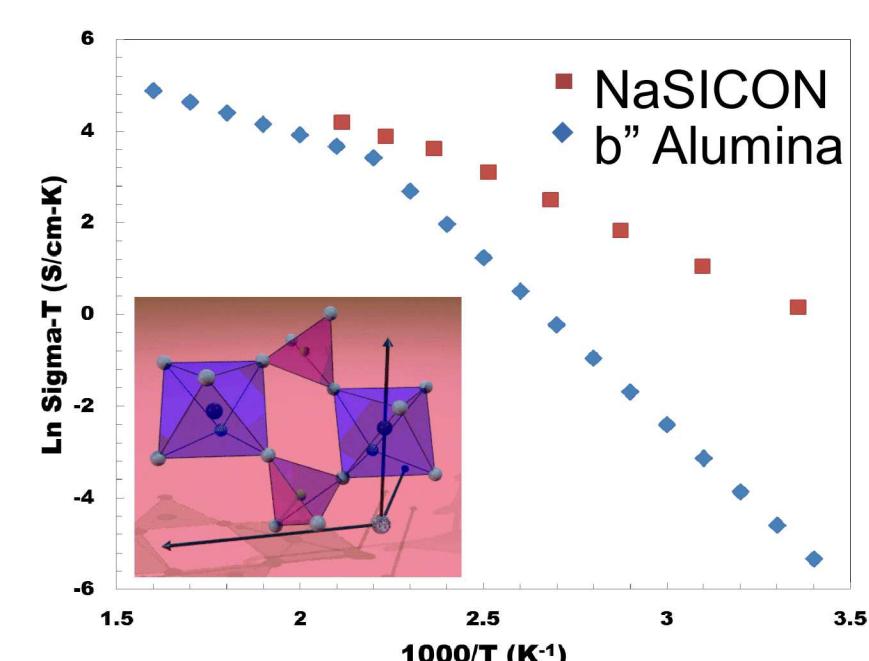
Objective: We aim to create zero-crossover solid state separators for safe, low cost, long cycle-life, low temperature ($<150^{\circ}\text{C}$) grid-scale sodium-based batteries.



Key Separator Properties:

- Selective, high ionic conductivity at reduced temperature ($<150^{\circ}\text{C}$)
- Chemical compatibility (molten Na, molten halide salts, strong base)
- Mechanical robustness
- Low cost, scalable production

Based on its high Na-ionic conductivity ($>10^{-3}$ S/cm at 25°C) and established chemical compatibility, NaSiCON ceramics ($\text{Na}_3\text{Zr}_2\text{PSi}_2\text{O}_{12}$) are good candidates for development.

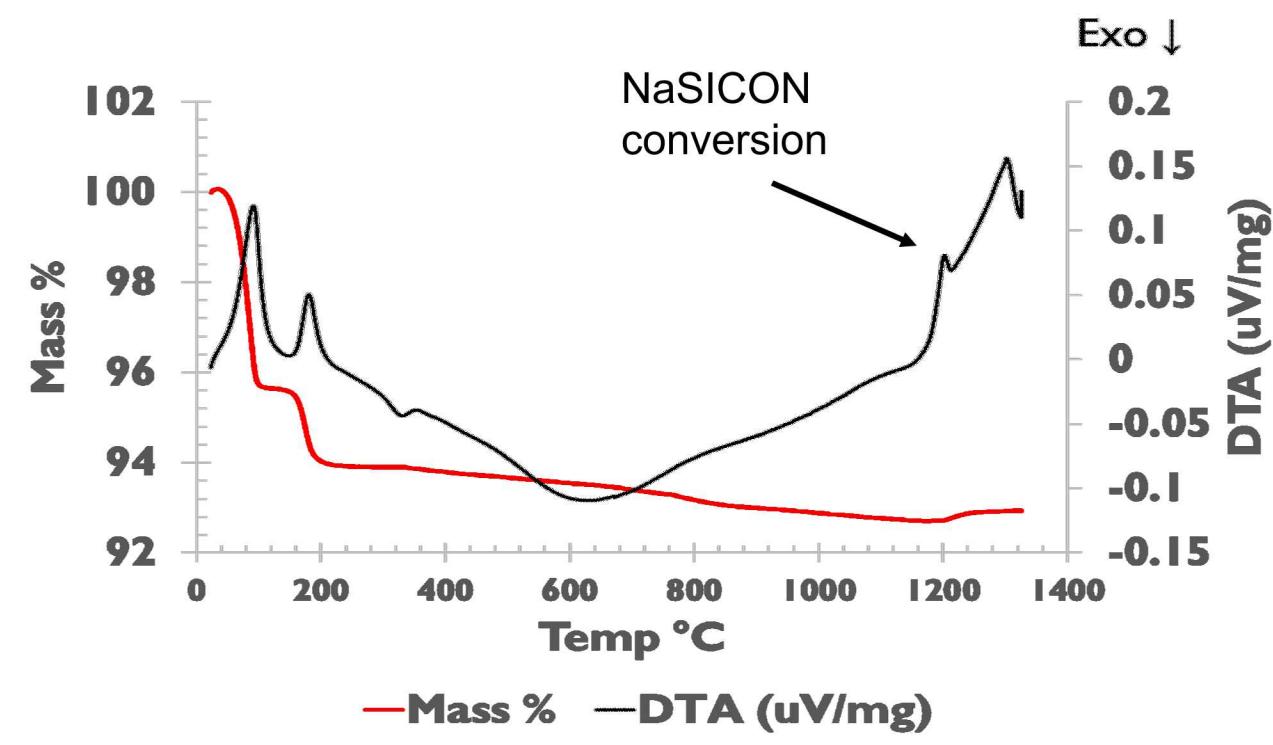


Thermal Analyses of Solid State Reaction

Goal: Inform low cost synthesis of high density, highly conductive NaSiCON separators



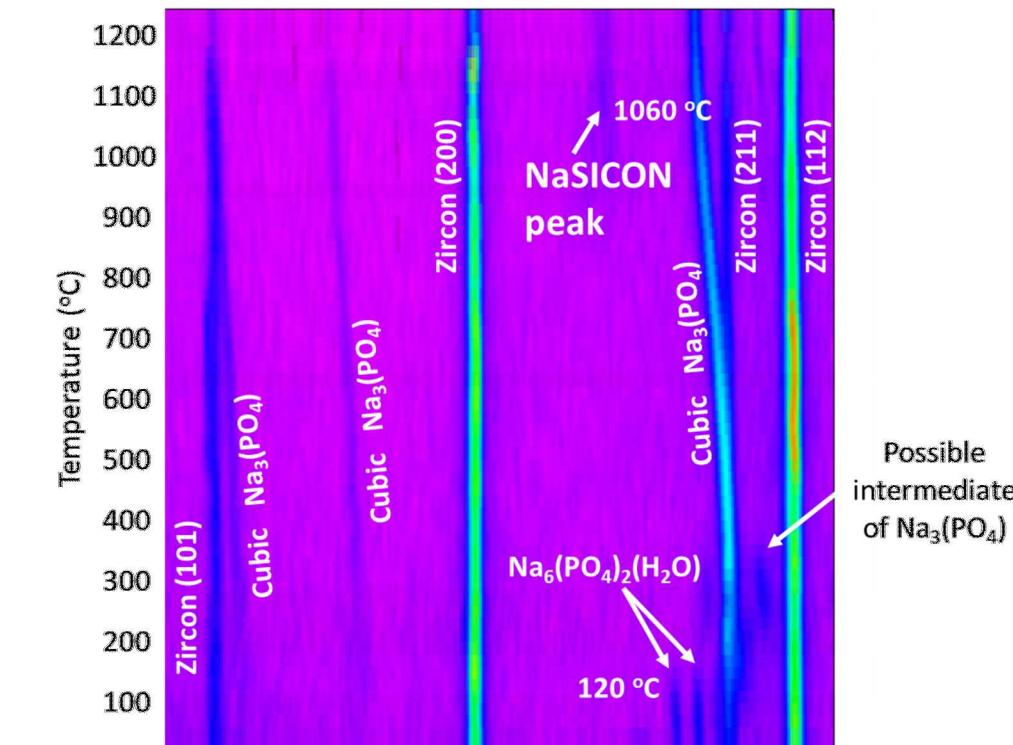
Differential Thermal Analysis and Thermogravimetric Analysis



- DTA/TGA show water removed from precursor powder by $\sim 250^{\circ}\text{C}$.
- NaSiCON conversion reaction evident between $1150\text{--}1230^{\circ}\text{C}$.
- Sintering above 1230°C \rightarrow poor ceramic integrity (melting?)



Variable Temperature X-Ray Diffraction

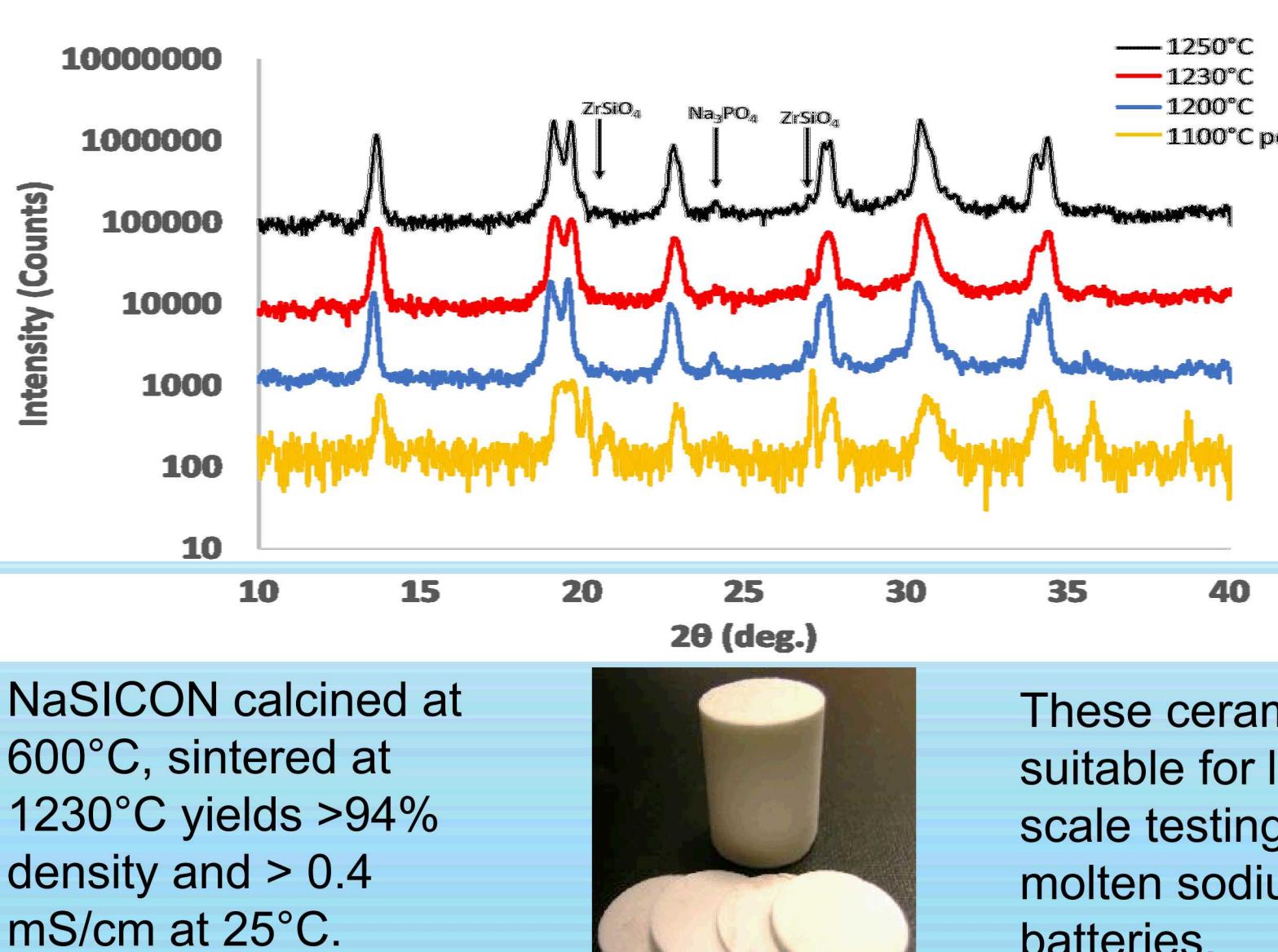
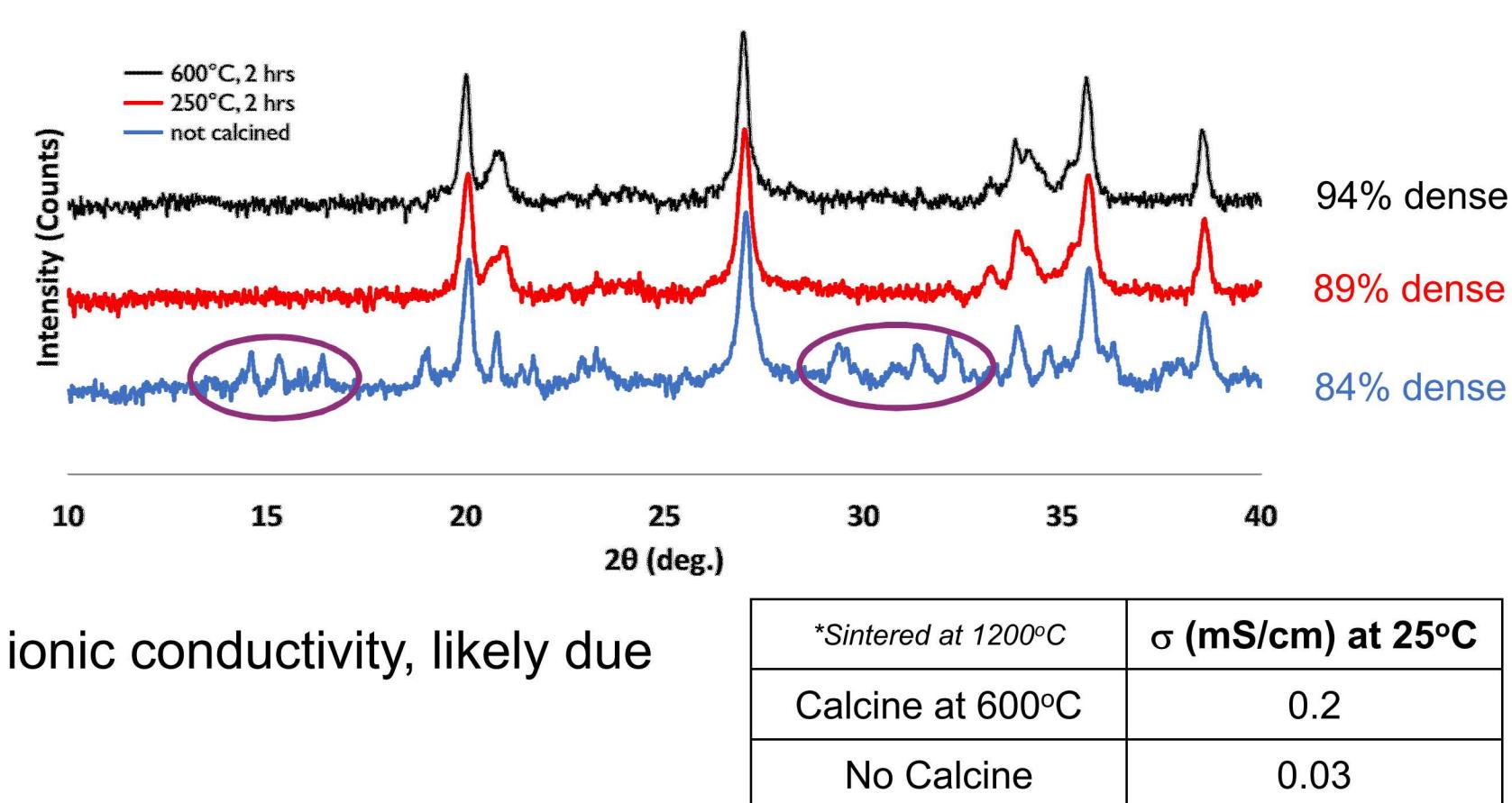


- VT-XRD shows conversion of Zircon and cubic $\text{Na}_3(\text{PO}_4)$ to NaSiCON starting near 1100°C
- Hydrate form of $\text{Na}_3(\text{PO}_4)$ up to 120°C , converts to cubic $\text{Na}_3(\text{PO}_4)$ at $\sim 300^{\circ}\text{C}$.

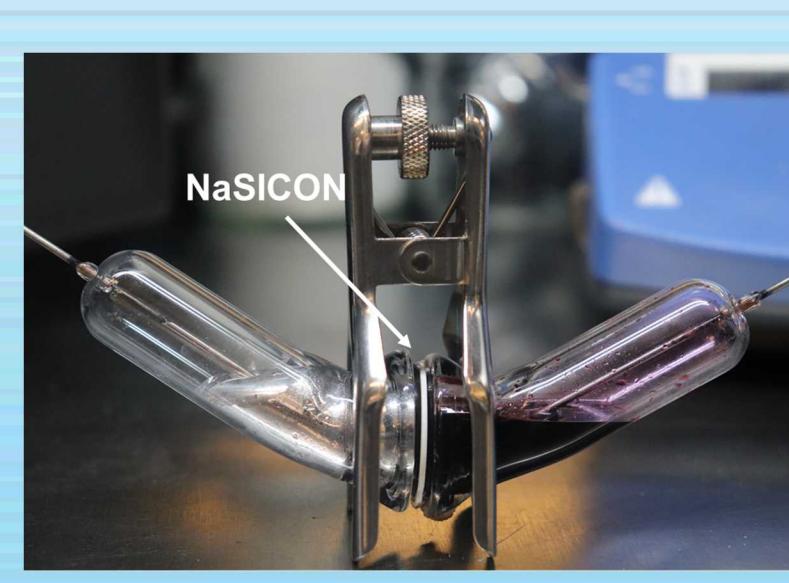
Why is this important?

Effective NaSiCON Conversion/Densification

- XRD confirms that calcining precursor powder to at least 250°C eliminates sodium phosphate hydrates in precursor.
- Density measurements, though, show that higher calcining temperature (600°C) leads to still better sintered ceramic density.
- Calcining also results in improved ionic conductivity, likely due to improved density.



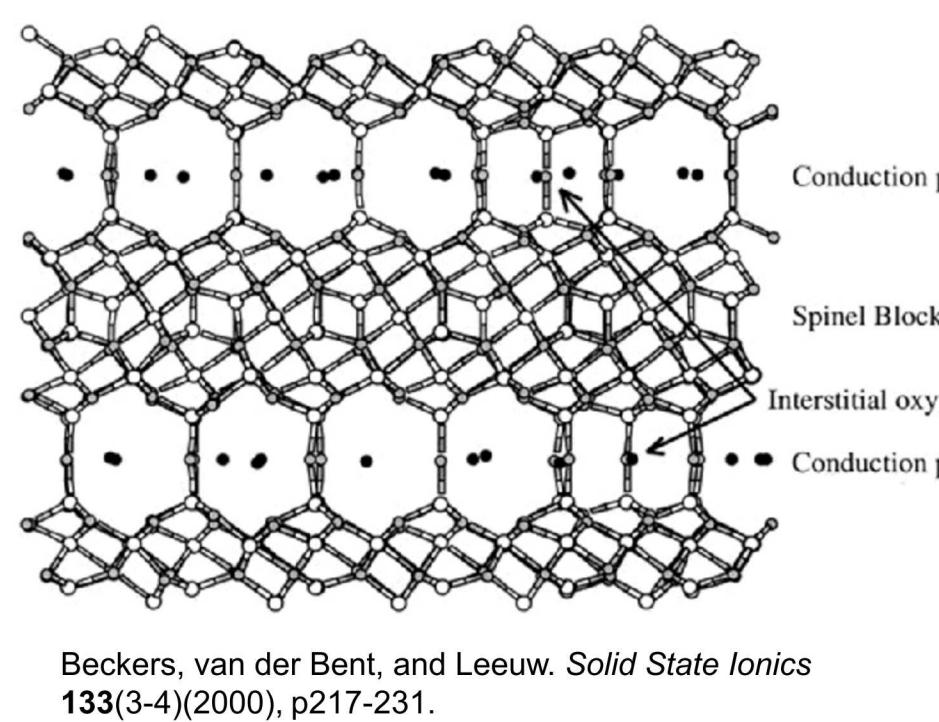
These ceramics are suitable for lab-scale testing of molten sodium batteries.



Alternative Separator Materials

Goal: Identify new, highly conductive, low cost sodium ion conductors.

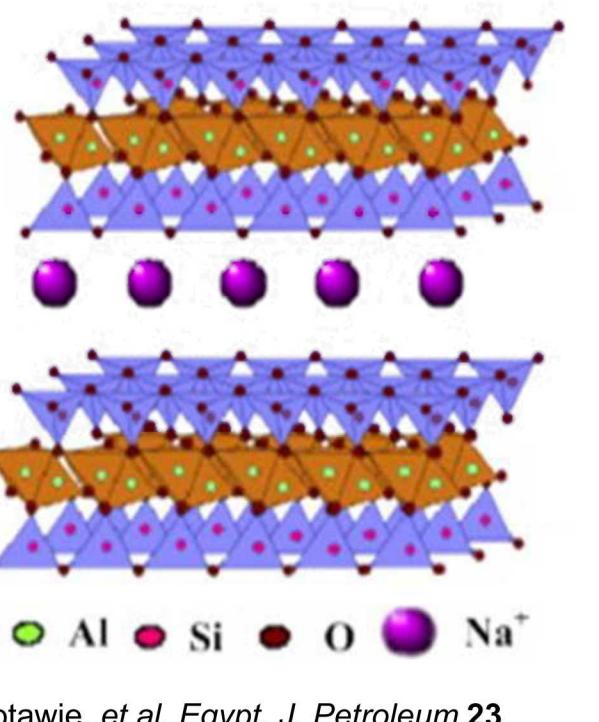
In the Na-ion conductor $\beta''\text{-Al}_2\text{O}_3$, Na^+ conduction follows ordered conduction planes.



Beckers, van der Bent, and Leeuw. *Solid State Ionics* 133(3-4)(2000), p217-231.

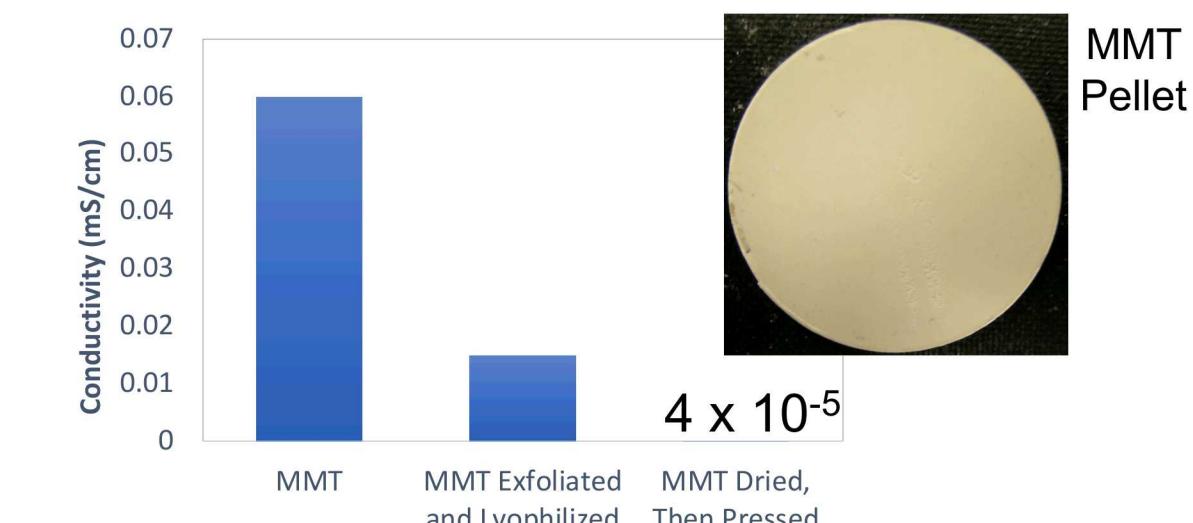
The ordered layers in low cost montmorillonite (MMT) clay create similar sodium-rich conduction planes.

Challenge: Can we utilize MMT to create a low cost sodium ion conductor?

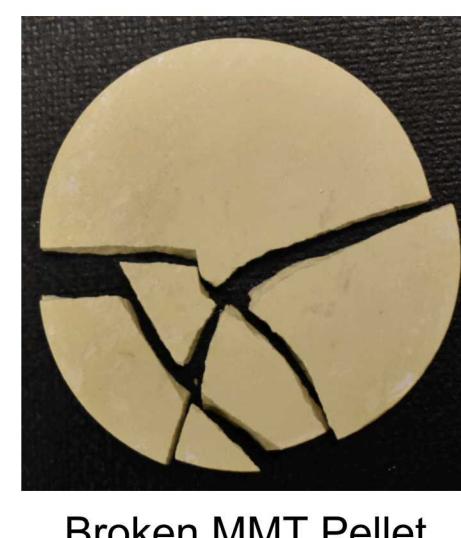


Motavie, et al. *Egypt. J. Petroleum* 23 (3)(2014), p331-338.

- MMT can be pressed into pellets with excellent ionic conductivity!
- The layered structure of the clays plays a key role Na^+ mobility through the separator..
- H_2O content increases conductivity of composite.
- Clay pellets can be fragile, though!



Initial Composite Assessment

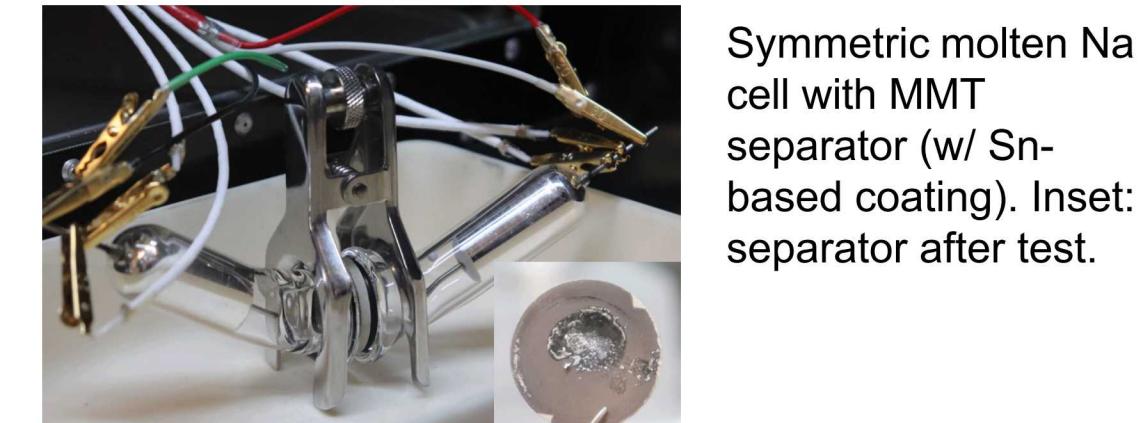


Can we integrate MMT into a composite with high conductivity and improved mechanical integrity?

	σ (S/cm)	Qualitative Mechanical Properties
MMT	1×10^{-4}	Brittle, breaks with little effort
MMT: PE-block-PEG (1:1)	9×10^{-11}	Soft, crumbly
MMT: PEG (1:1)	5×10^{-8}	Pliable, but crumbly
MMT: PEG:NaTFSI (1:1:0.05)	4×10^{-6}	Pliable, but crumbly
MMT pressed at 150°C	5×10^{-5}	Very fragile
MMT: HDPE (3% HDPE, 150°C)	6×10^{-5}	Rigid and stronger

- Doping with sodium trifluoromethanesulfonimide (NaTFSI) can recover some of the conductivity lost through addition of insulating polymer matrix.
- Low temperature, functional polymers (e.g., PEG), do not significantly improve pellet integrity and are not suitable for molten Na-batteries, but inform composite design.
- Heating MMT decreases conductivity (likely through water loss).
- Adding a small amount of HDPE significantly improves pellet integrity, without significant impact to conductivity, beyond thermal treatment.

- Alternatively, increasing pellet thickness and application of sputtered metallic surface coatings stabilize MMT pellets for battery testing.
- Initial tests show promise, but reveal high impedance across separator interfaces with Na.



Conclusions and Future Directions

- Thermal analyses informed key thermal processing that enabled dense, functionally conductive NaSiCON separators for lab-scale testing.
- Ionic conductivity in novel MMT separators depends on H_2O content and retention of layered clay structure.
- MMT composites show promise toward reducing pellet brittleness, but insulating polymer content or processing treatment (e.g., heating) can degrade performance.
- Initial tests indicate that MMT could prove useful as a novel ion conductor for Na-based batteries.
- Future work will expand these results toward improving NaSiCON ceramic conductivity through ionic doping and synthesis of clay-based or NaSiCON-based composites with high conductivity and robust mechanical properties.

Enhanced understanding of materials/properties relationships inform improved separator development!