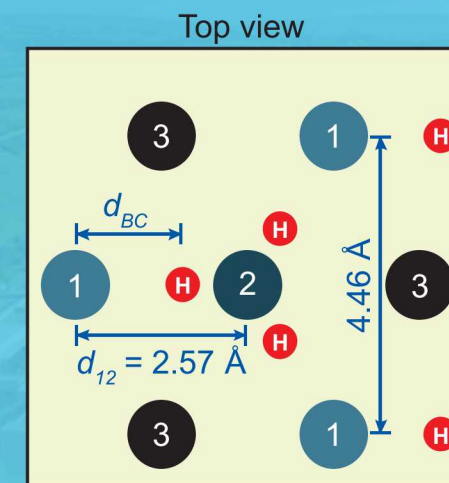
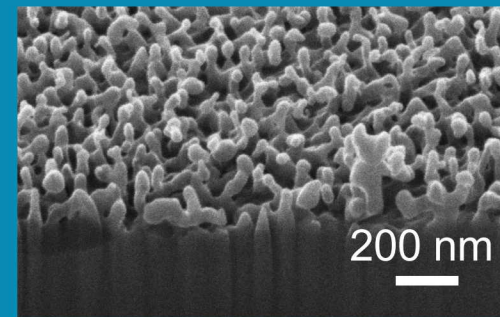


Helium and hydrogen interactions with tungsten surfaces



PRESENTED BY

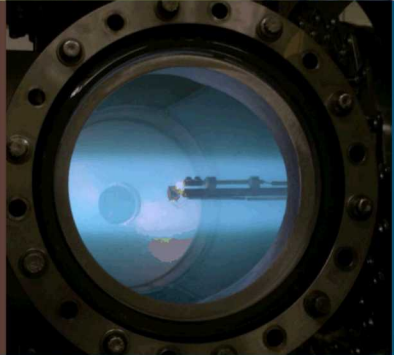
Tim Wong | October 23, 2019

Josh A. Whaley and Robert D. Kolasinski

Energy Innovation Department



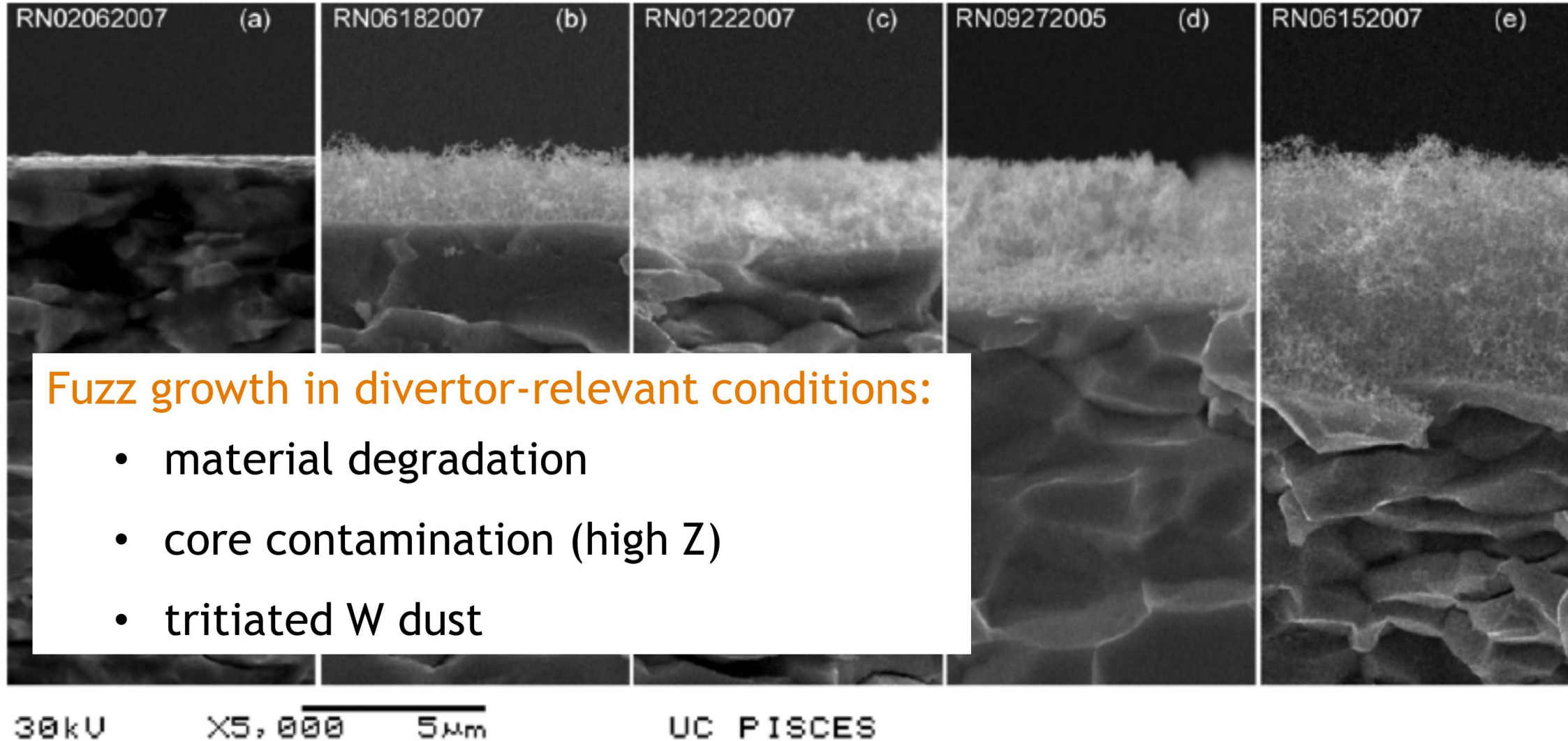
Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.



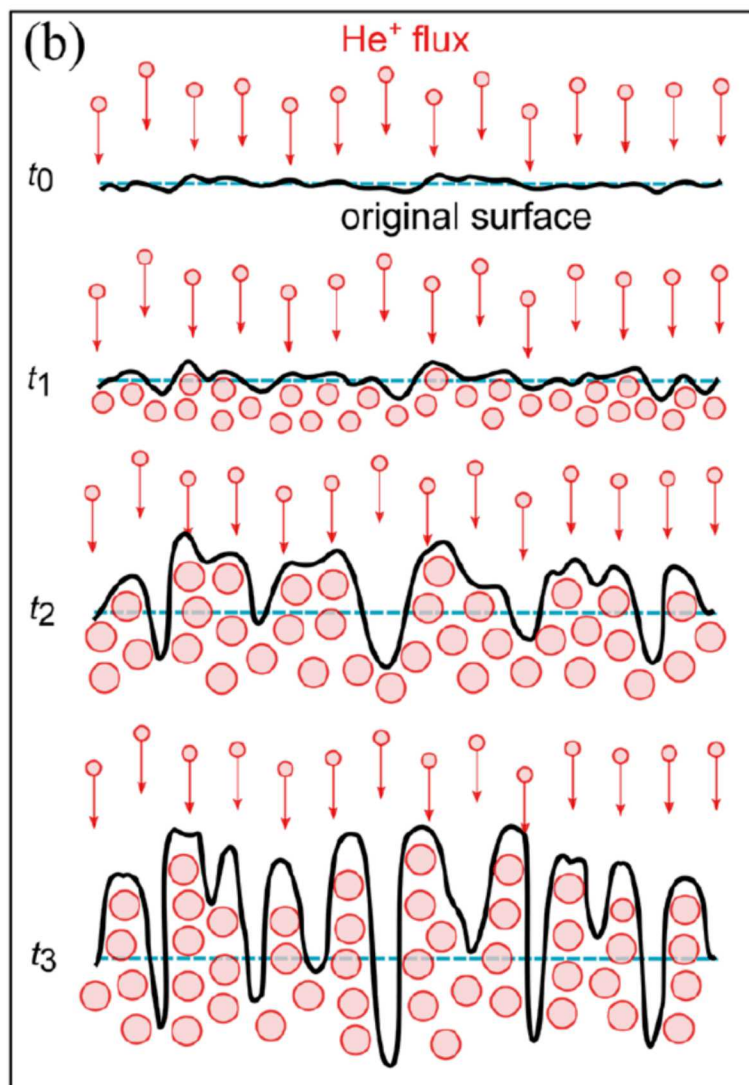
He plasma interactions with W surfaces

Effect of TDS on He-induced W nanostructure

Helium-induced “fuzz” on W surfaces



He-induced nanostructure growth on W surfaces



Dasgupta et al., Nucl. Fusion (2019).

- physics for fuzz growth still not well understood

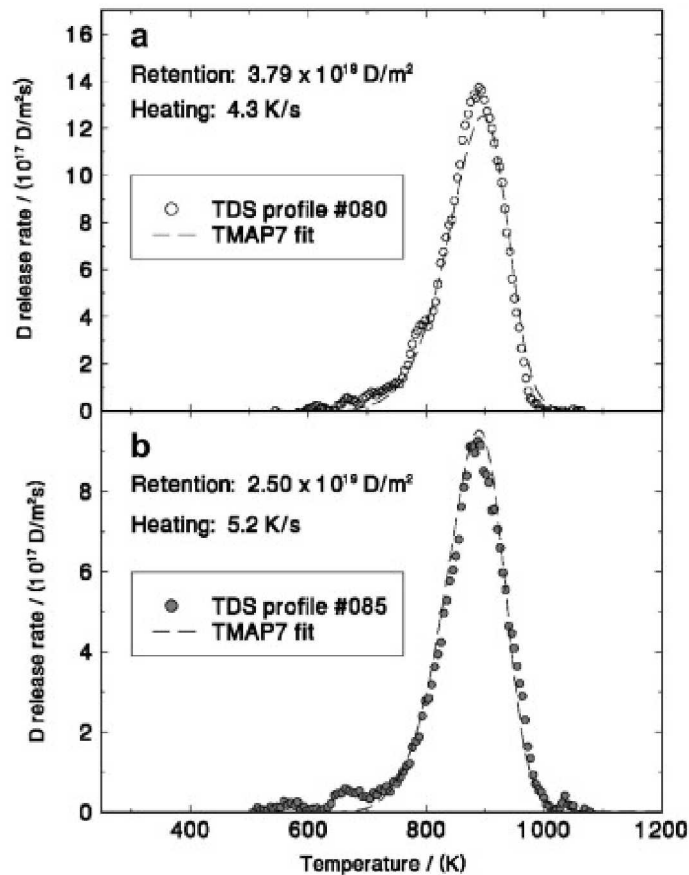
- general steps:

1. He implanted into W
2. forms stable pairs of He
3. coalesce into He bubbles
4. He bubbles + surface stresses + W adatom diffusion → nanostructure

thermal desorption spectroscopy (TDS) can investigate He-W interactions and trapping

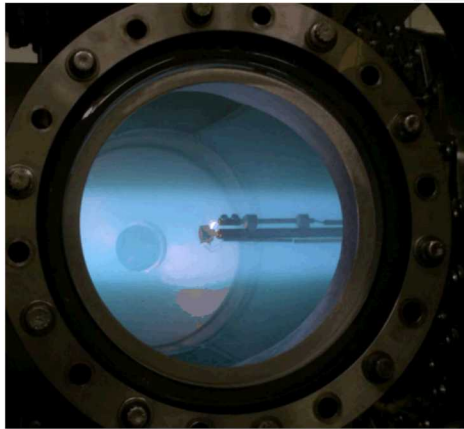
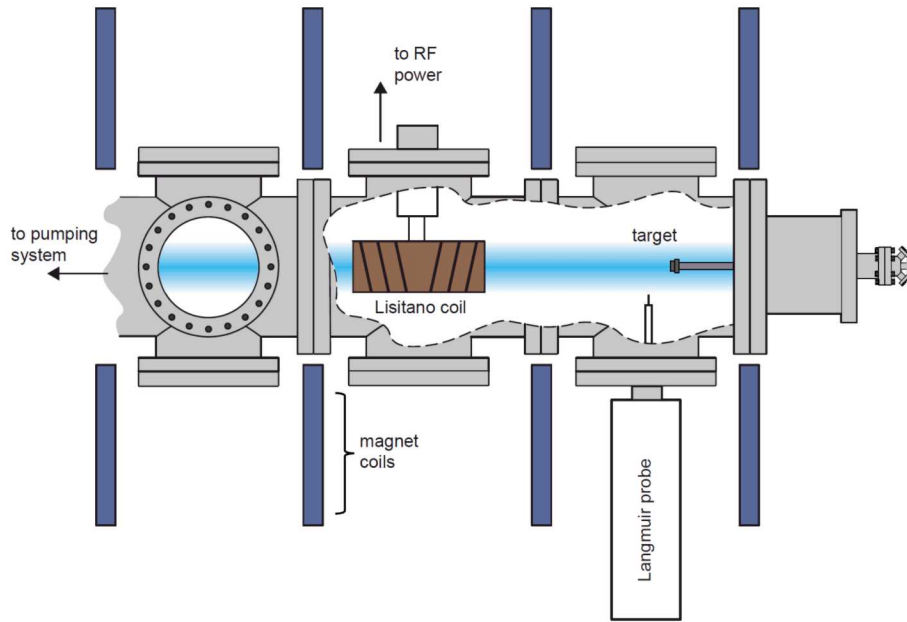
TDS to investigate trapping energies

TDS has been successfully used to determine D binding energies in W



1. Can TDS modeling be done in the same way for He-exposed W?
2. What happens to the nanostructure during TDS (>1800 K needed to fully desorb He)?

Plasma exposure and conditions



W sample exposed to He plasma

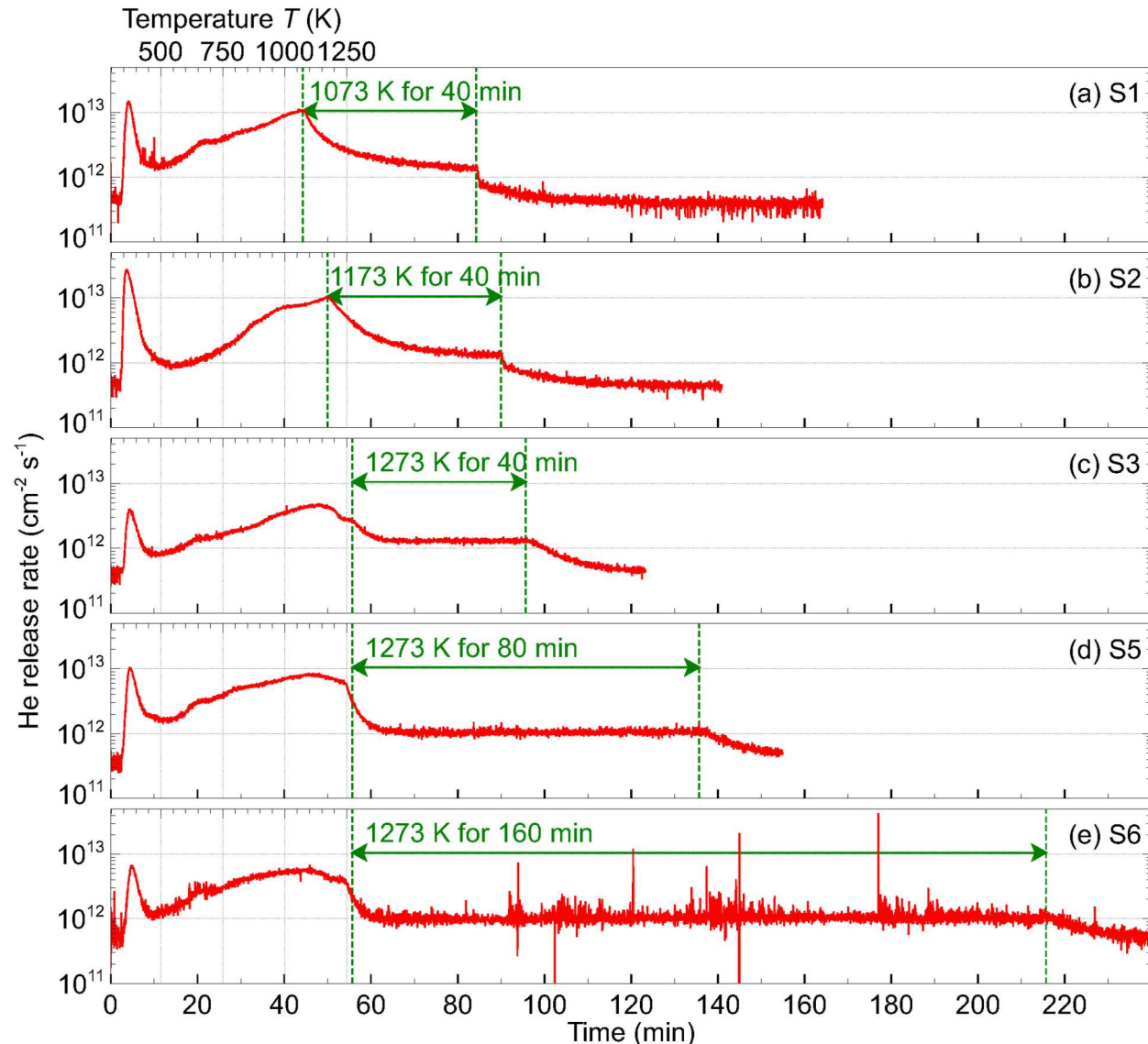
Plasma conditions

ion flux	Γ_{ion}	$= 4.0 \times 10^{16} \text{ cm}^{-2} \text{ s}^{-1}$
electron temp.	T_e	$= 11 \text{ eV}$
plasma potential	V_p	$= 40 \text{ V}$
plasma density	n_p	$= 2.0 \times 10^{10} \text{ cm}^{-3}$

Plasma exposure (same for all samples):

- (1) 6 hours in plasma
- (2) held at 1123 K
- (3) biased at -50 V (ion energy of 90 V)

Thermal desorption spectroscopy



Ramp rate of 17.5 K s^{-1}

- consistent desorption peaks near $T=373 \text{ K}$ and 1123 K for all samples
- 1123 K peak observed in literature for similar He-plasma exposures
- spikes in (e) are indication of bursting of near-surface He bubbles

Helium ion microscopy

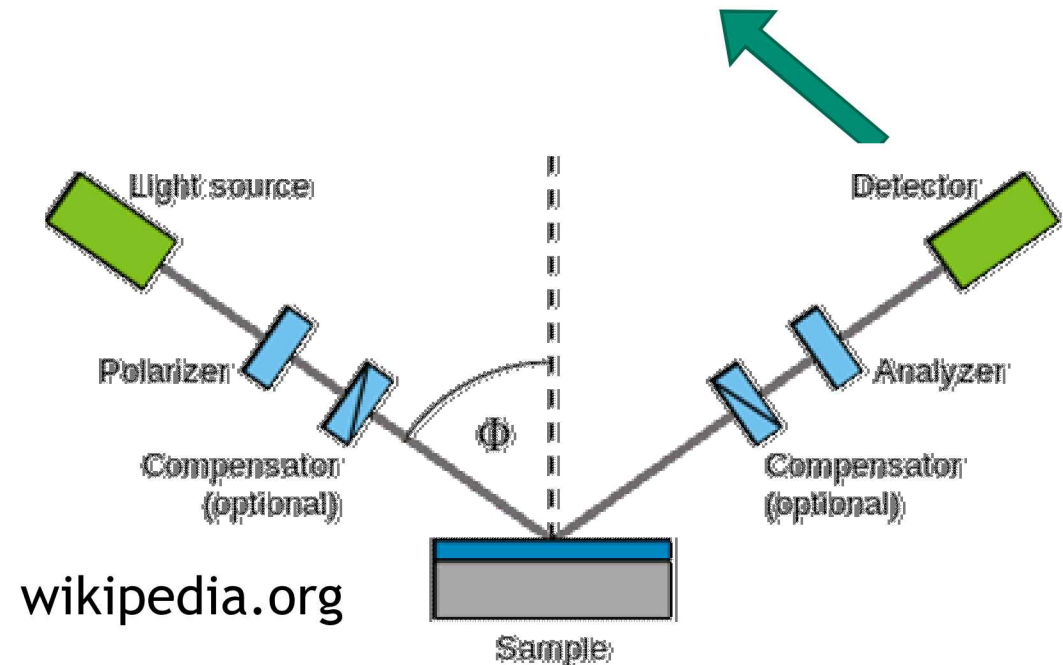
- better depth of field, resolution, and contrast than SEM (but \$\$)
- performed at UC Berkeley BNC



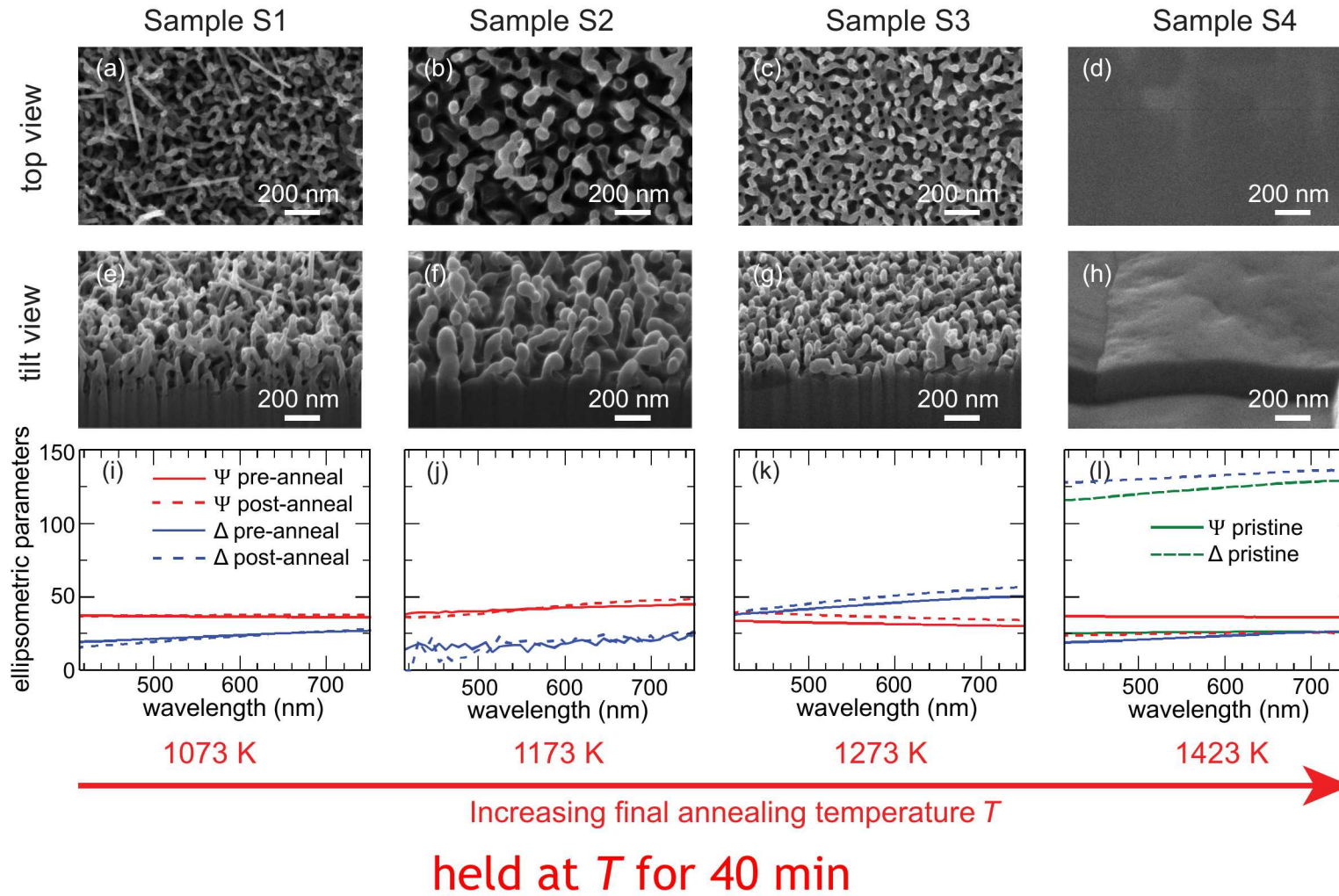
Spectroscopic ellipsometry

- can be performed in-situ and in real time
- non-destructive & contactless
- sensitivity to nm-sized changes

Ellipsometric parameters: Ψ , Δ

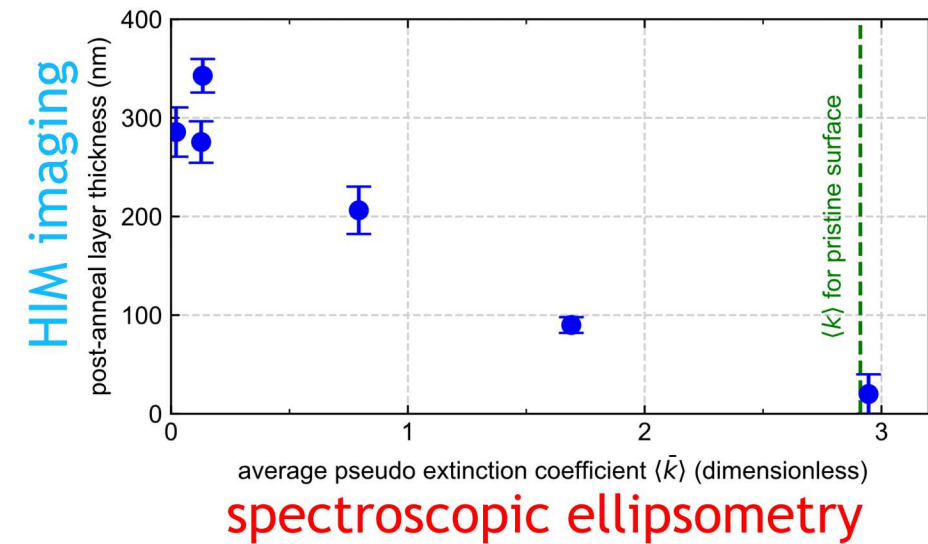


Effect of annealing temperature on surface morphology



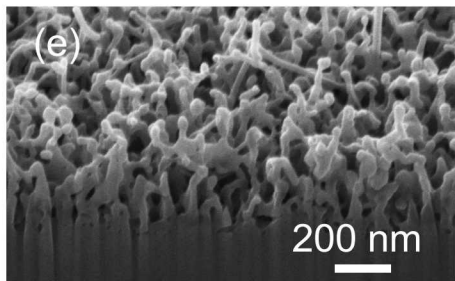
changes to nanostructure
starting around 1173 K

fully recrystallized at 1423 K

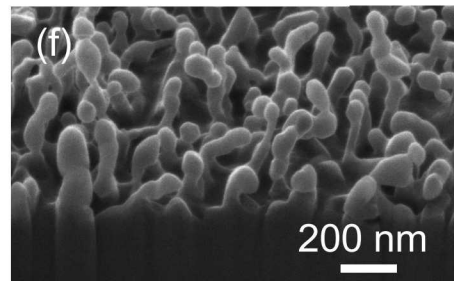


Conclusions for high-temperature annealing study

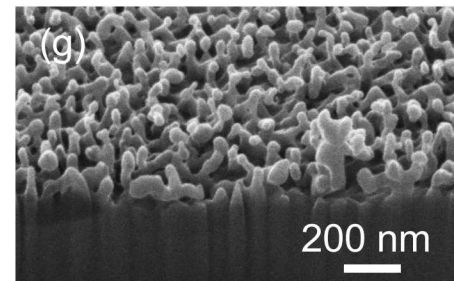
1. TDS can greatly alter surface morphology of W nanostructure
 - changes observed at 1173 K
 - full recrystallization at 1423 K after 40 min
2. modeling of TDS spectra should account for complications that may arise from changing surface morphology
3. ellipsometry can potentially be used to track surface recovery in-situ



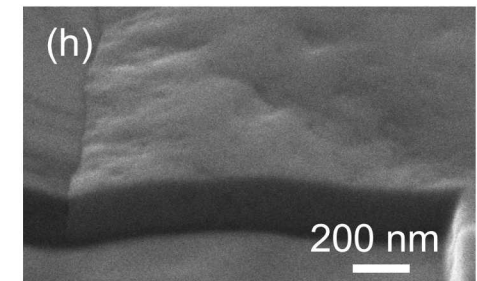
1073 K



1173 K

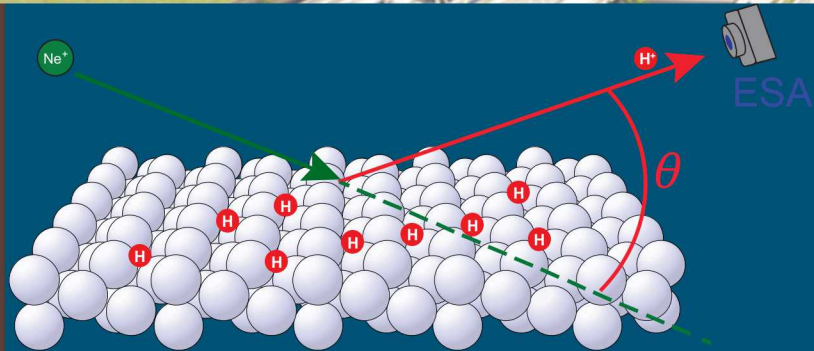


1273 K



1423 K

Increasing final annealing temperature T



Hydrogen adsorption on W surfaces

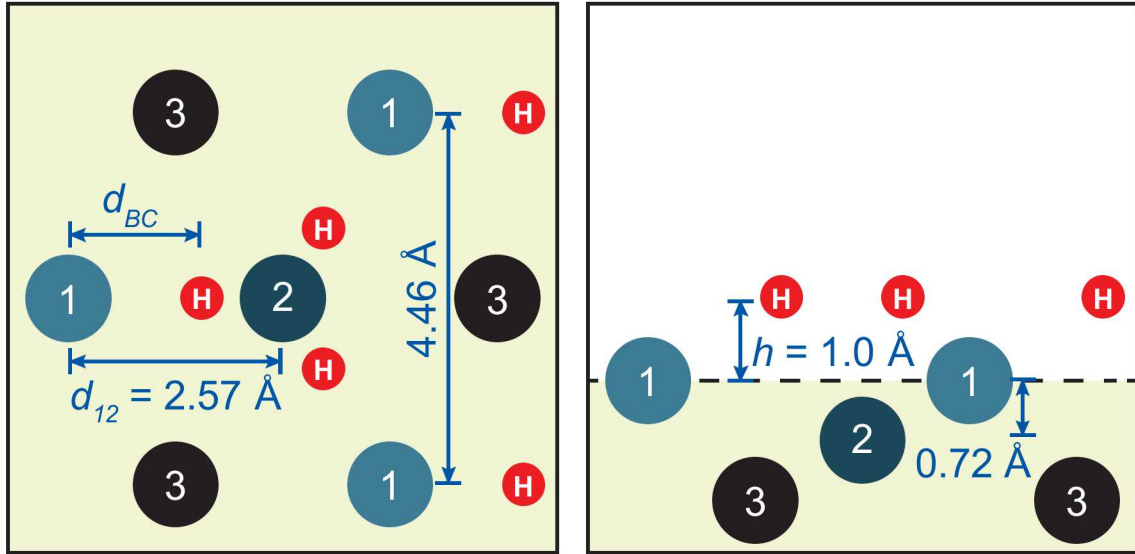
Characterizing the $W(111)+H(\text{ads})$ system

DFT predictions of H adsorption on W(111)

Bond-centered (BC) H adsorption on W(111) was predicted using DFT [1] by Z. Bergstrom, Brian Wirth's group at U. Tennessee

Top view

Side view $[0\ 1\bar{1}]$



$$1.49\text{ \AA} \leq d_{BC} \leq 1.69\text{ \AA} \quad 0.96\text{ \AA} \leq h \leq 1.12\text{ \AA}$$

Experimentally validate DFT results:

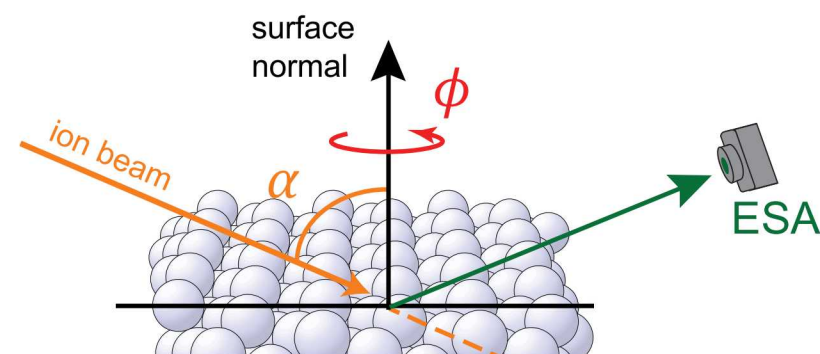
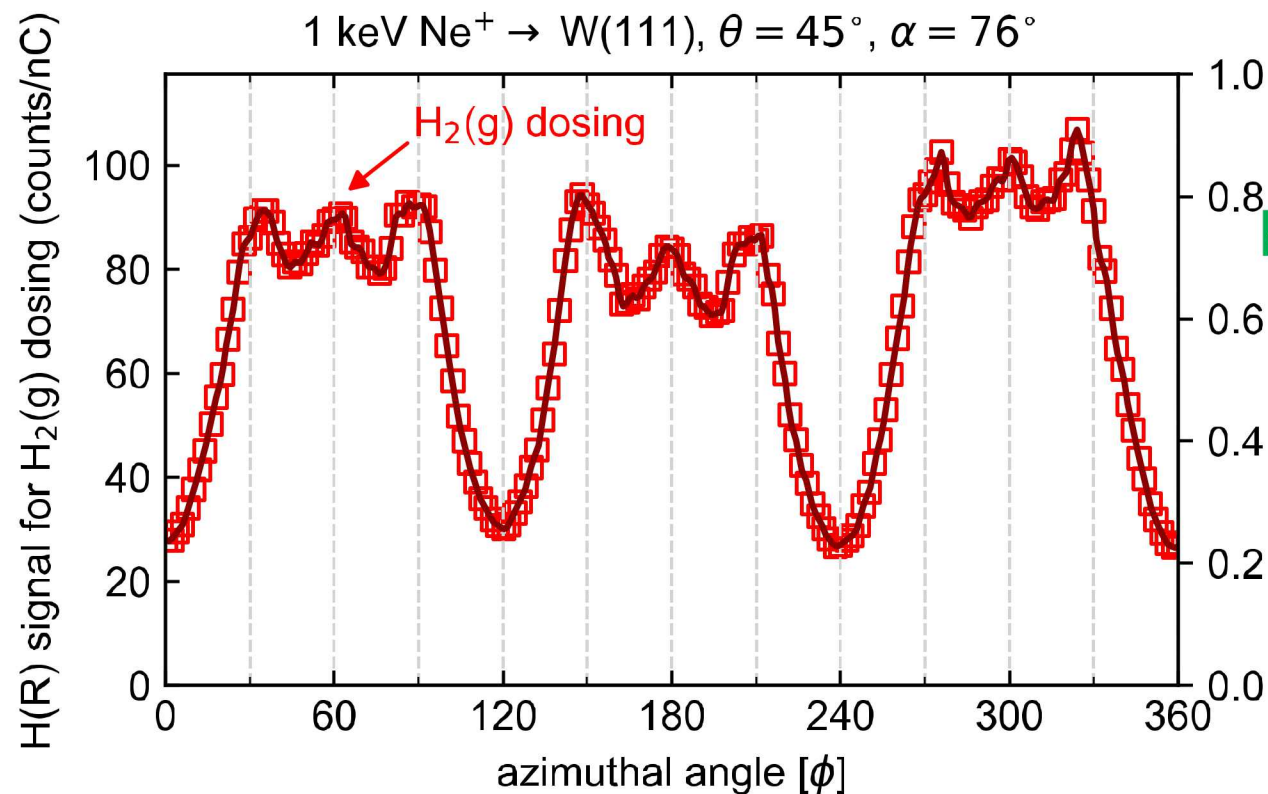
DFT provides inputs for larger scale models, e.g.

- interparticle potentials for MD
- hydrogen dissociation energies on surfaces

Low energy ion scattering (LEIS)

Model this signal with MD to identify H binding sites

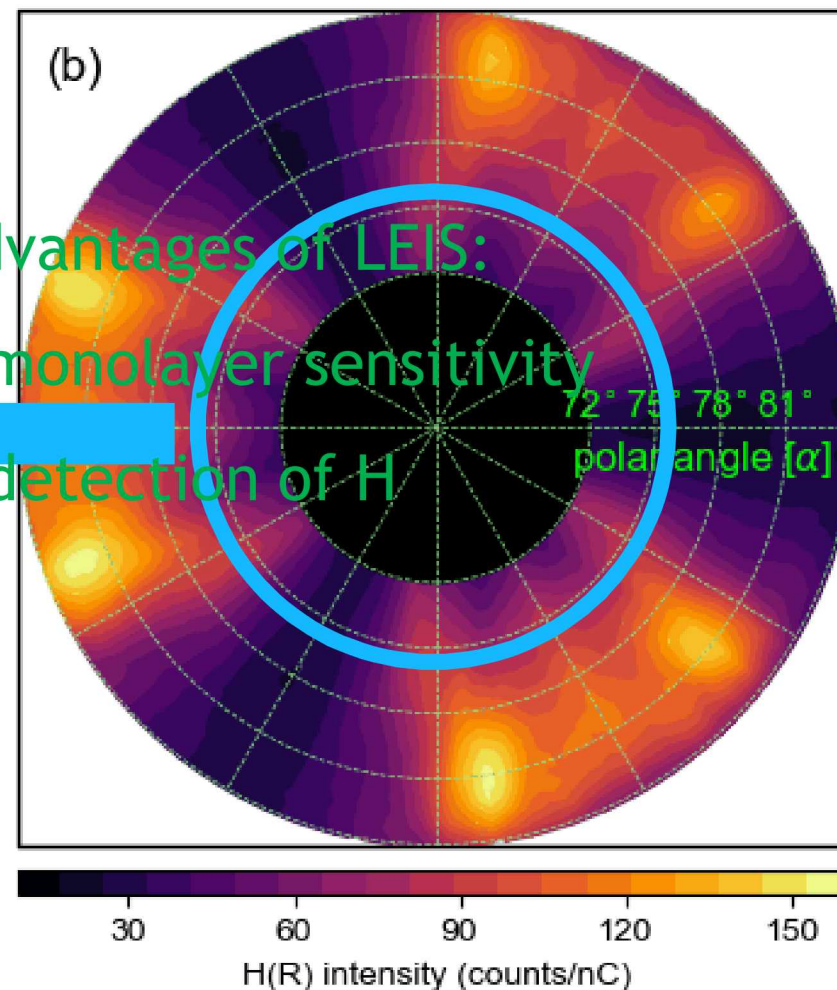
Scattering signal from a substrate



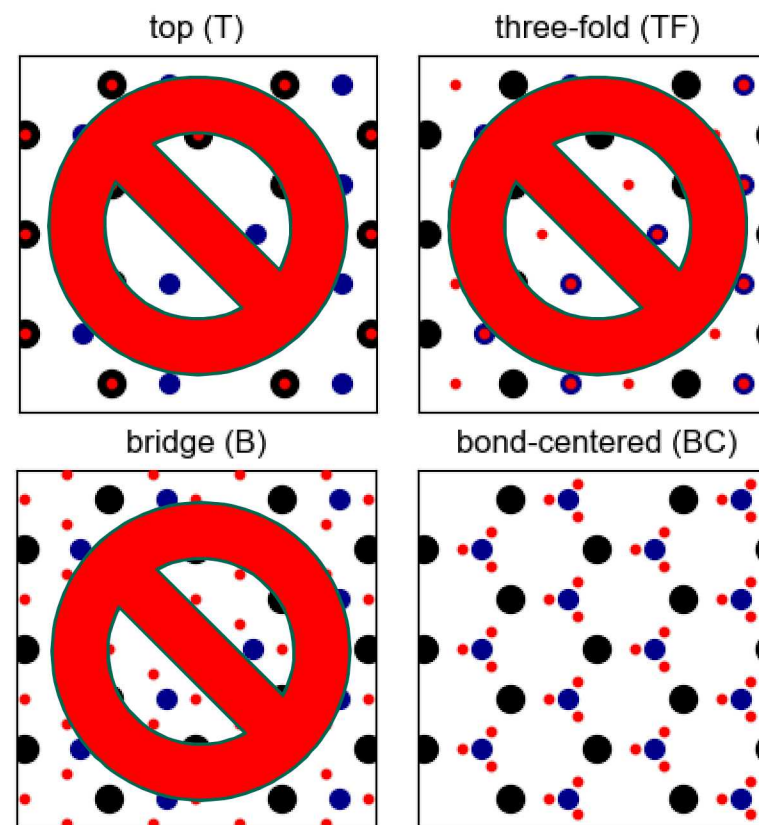
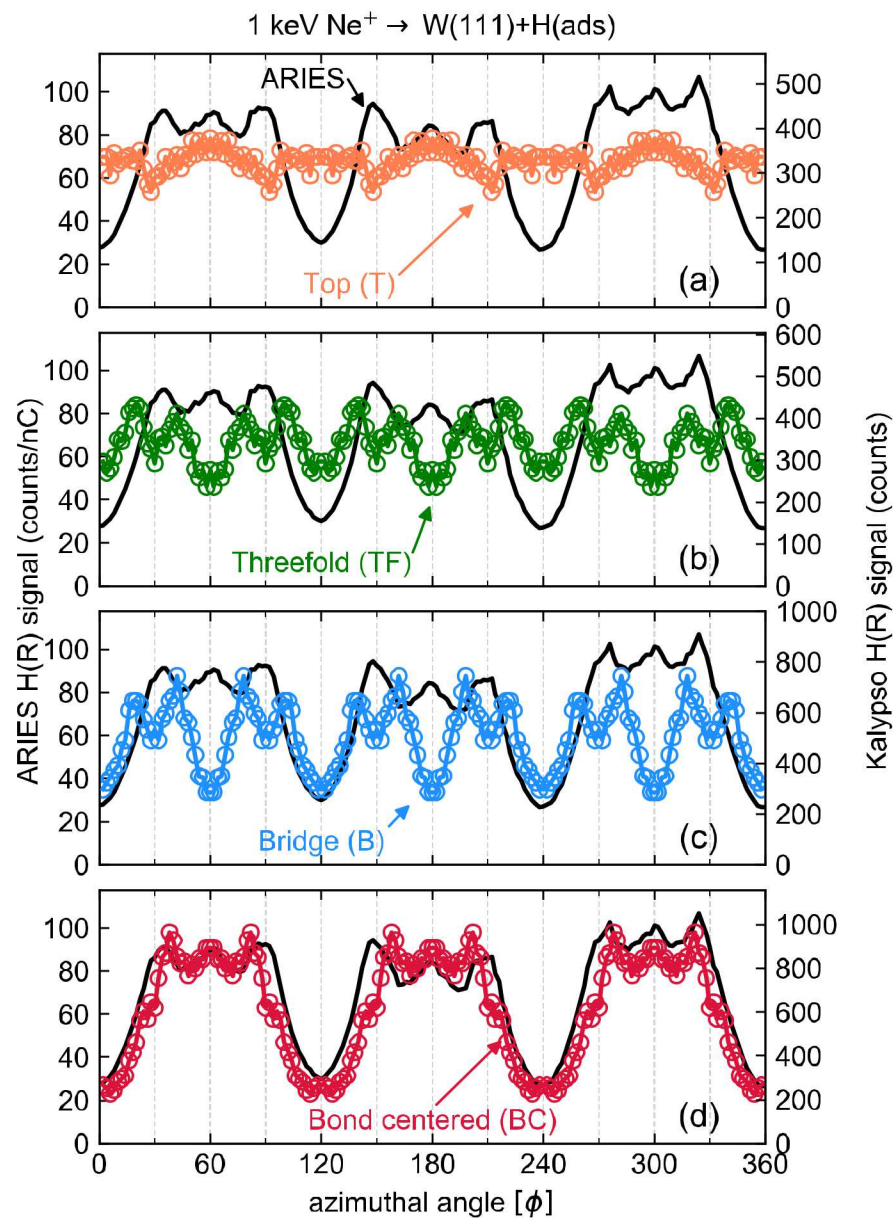
Recoil signal of H from ion channeling

Key advantages of LEIS:

- monolayer sensitivity
- detection of H



Experimental data compared to MD simulation (Kalypso [2])



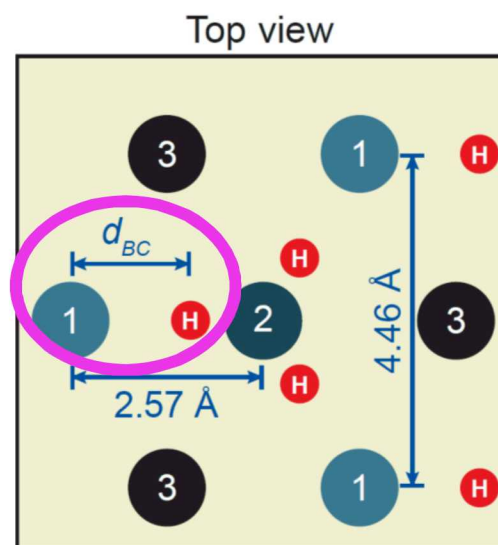
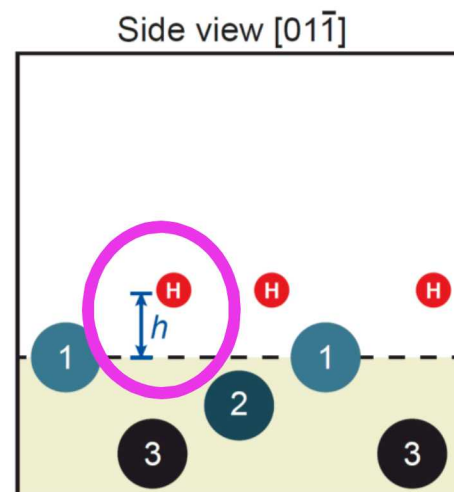
Constraining adsorbate height and position

$$h = 1.0 \pm 0.1 \text{ \AA}$$

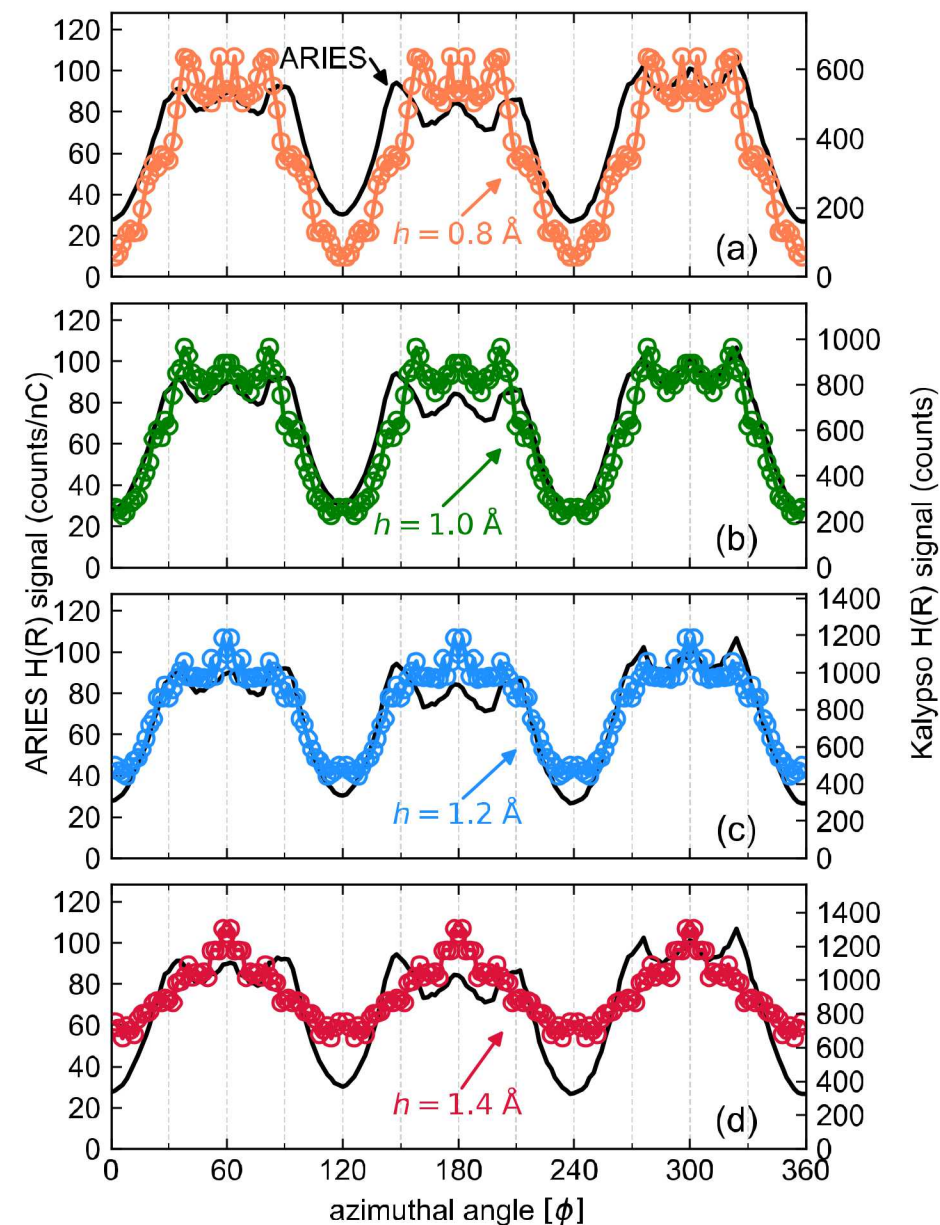
DFT prediction:
 $0.96 \text{ \AA} \leq h \leq 1.12 \text{ \AA}$

$$d_{BC} = 1.6 \pm 0.1 \text{ \AA}$$

DFT prediction:
 $1.49 \text{ \AA} \leq d_{BC} \leq 1.69 \text{ \AA}$



varying h

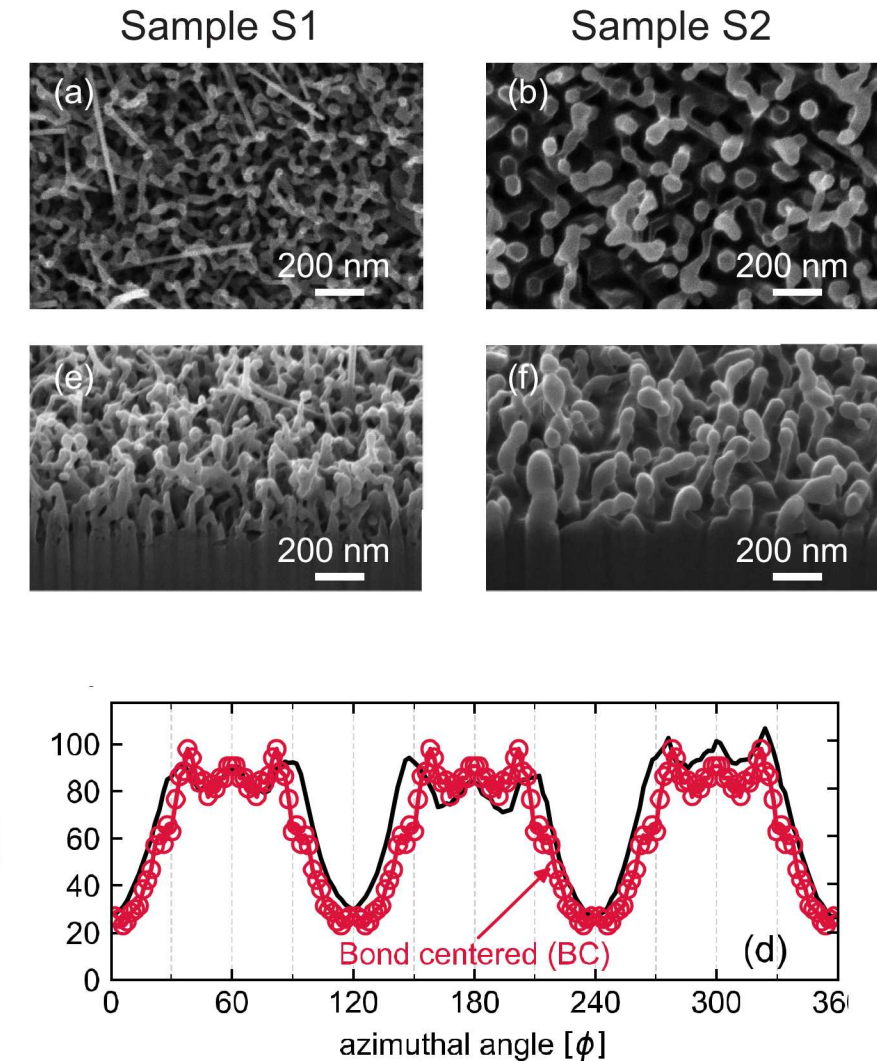


Summary

1. TDS of He-induced W nanostructure

- changes in morphology observed at 1173 K ($\ll 1800$ K for full He desorption)
- interpretation of TDS data needs to account for changes in surface nanostructure

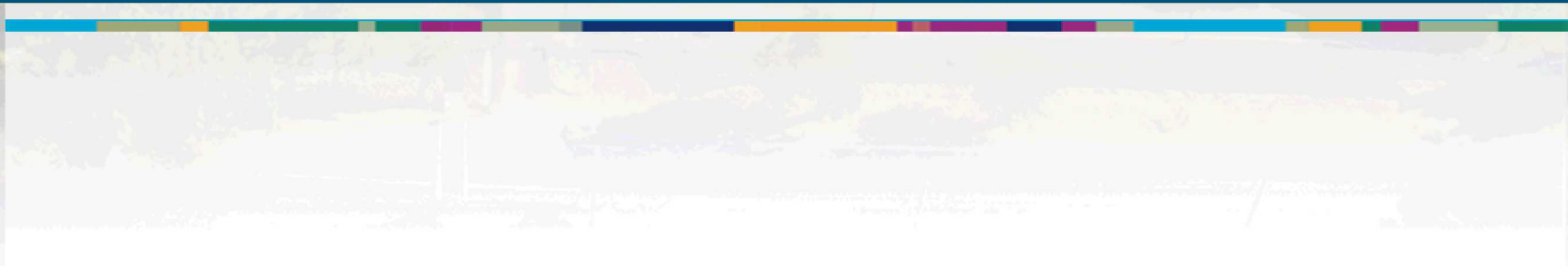
2. Validated bond-centered (BC) hydrogen binding for the W(111)+H(ads) system



Thank you for your attention!

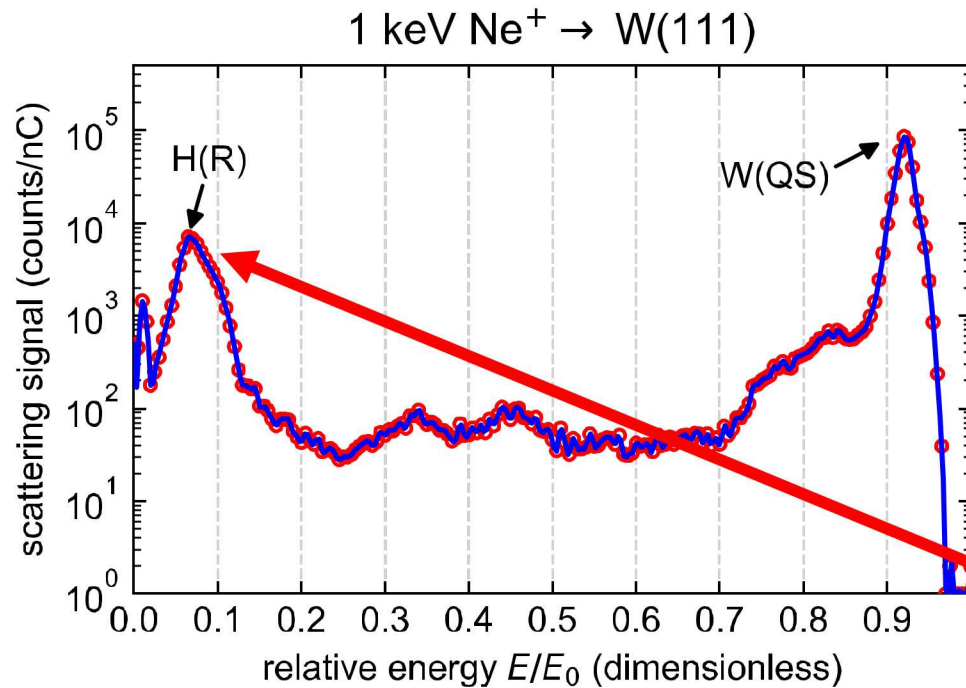


Extra slides

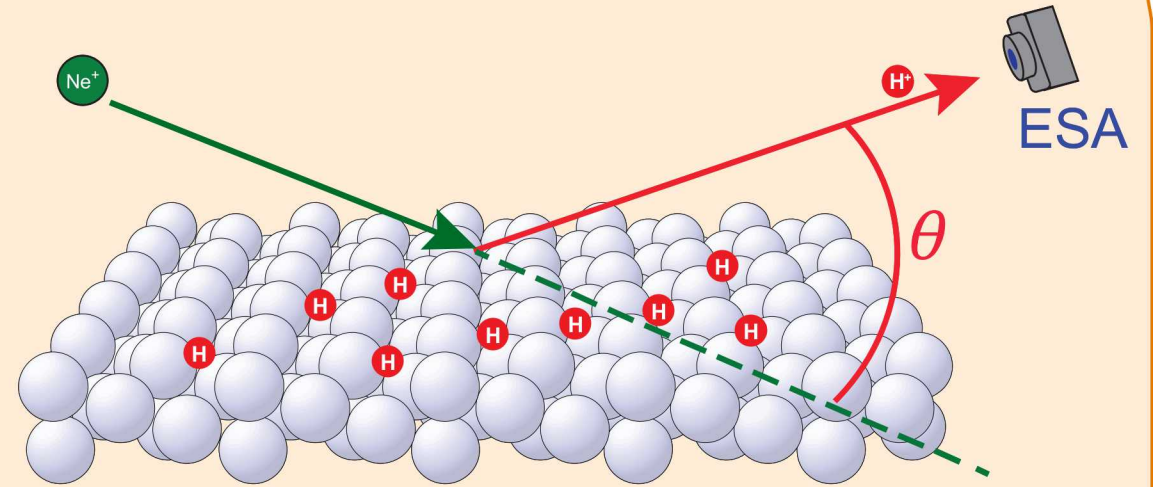


Direct recoil spectroscopy

One of the few surface science techniques with hydrogen sensitivity

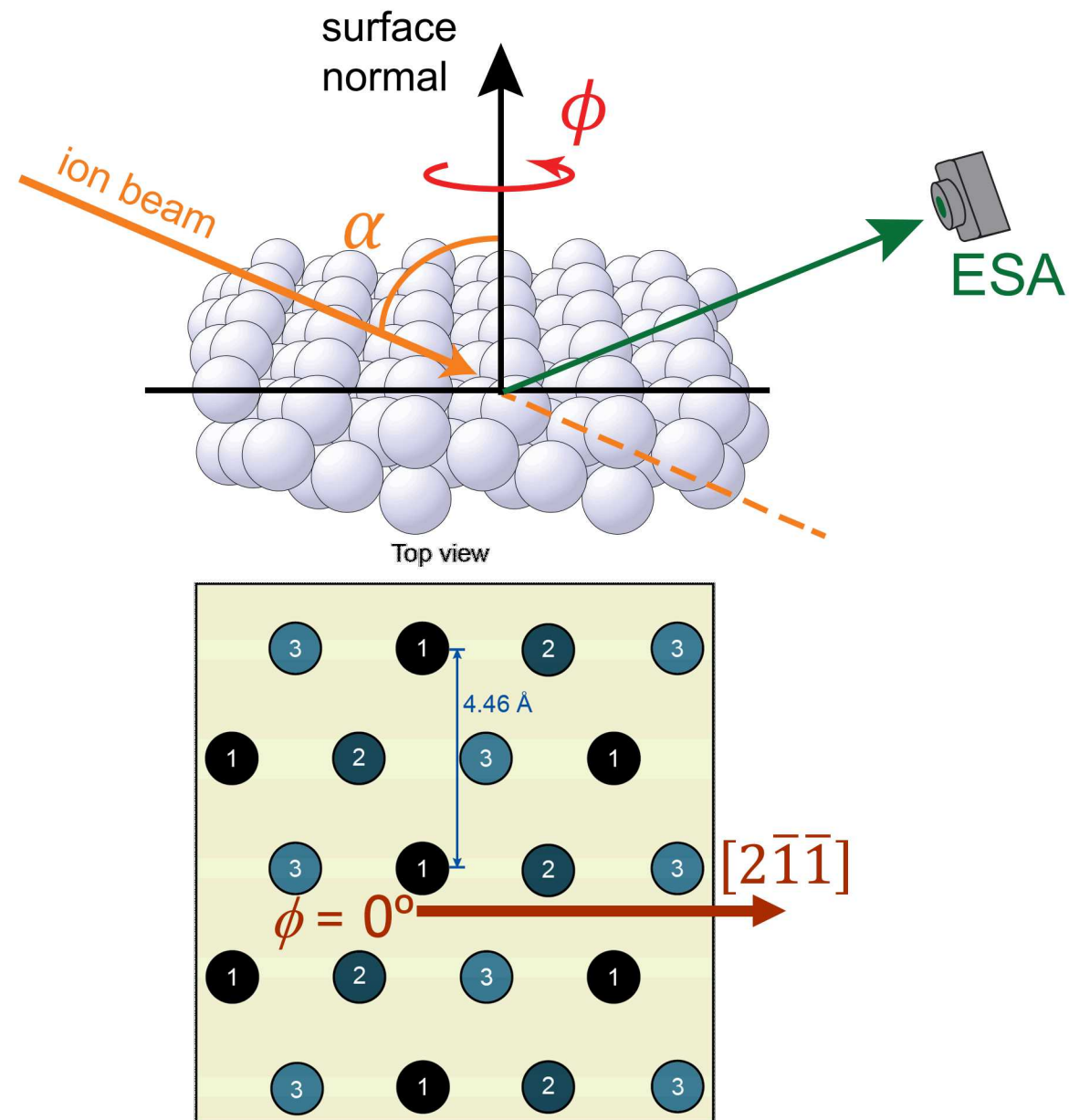
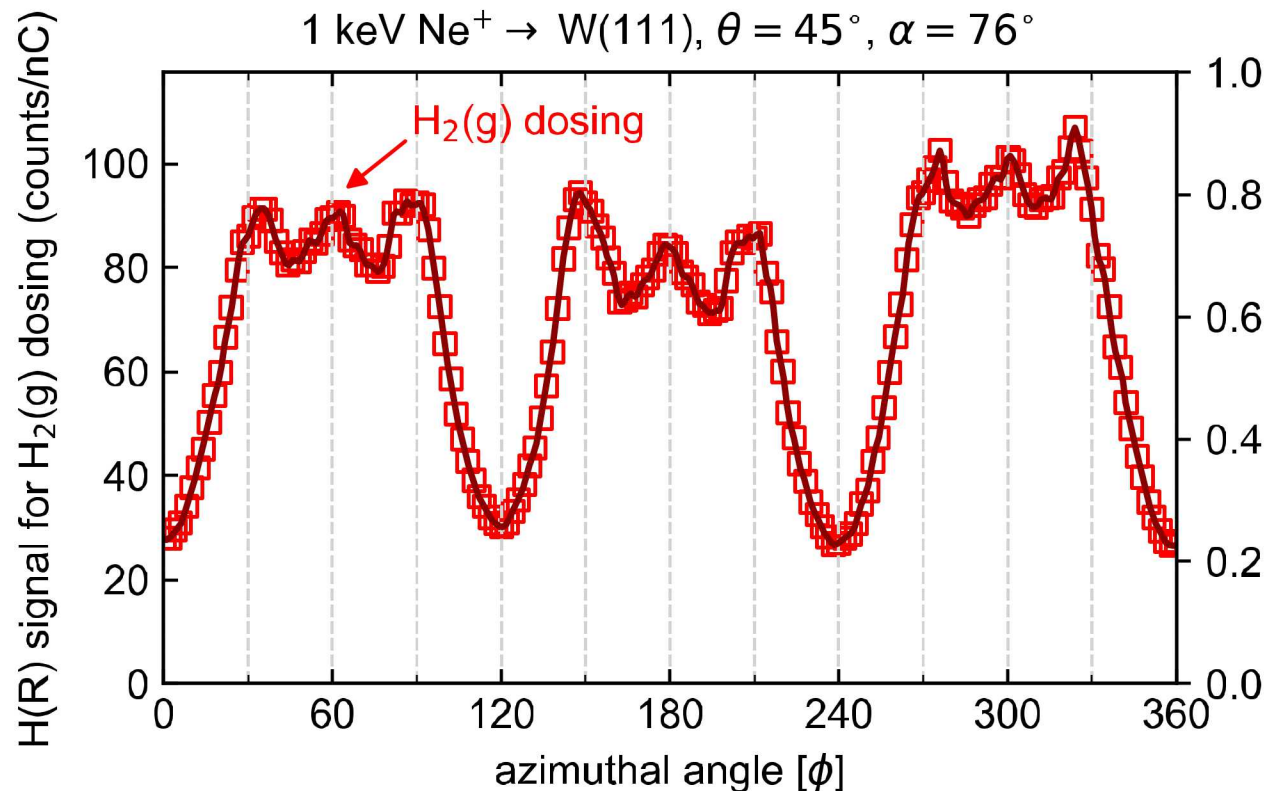


Direct recoil spectroscopy (DRS)



$$\frac{E}{E_0} = \frac{4\mu}{(1 + \mu)^2} \cos^2 \theta, \quad \mu = \frac{m_T}{m_P}$$

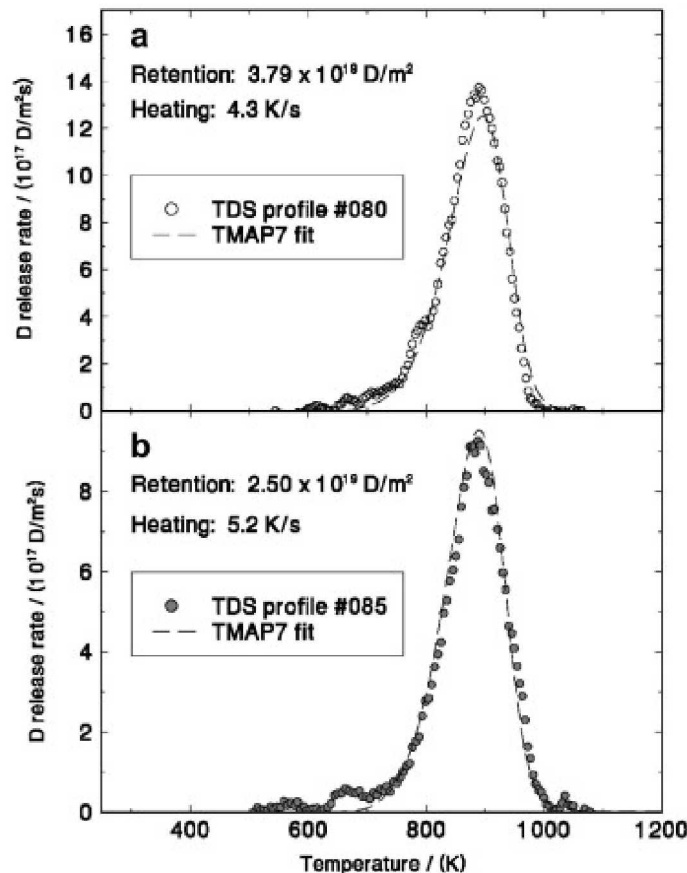
Obtaining H recoil measurement at a single α



TDS to investigate trapping energies

TDS has been successfully used to determine D binding energies in W

1. Can TDS modeling be done in the same way for He-exposed W?
2. What happens to the nanostructure during TDS (>1800 K needed to fully desorb He)?



up to 60% of retained He found in the fuzz

