



Pd-Catalyzed Oxidative Homocoupling of Pyrazole Boronic Esters to Access Bipyrazoles for Metal–Organic Frameworks

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Ever Velasquez,[§] Julia Oktawiec,[◇] Jonathan B. Lefton,[¶] Tomce
Runcevski,[¶] Jeffrey R. Long,^{§,◇,‡*} Ji-Woong Lee^{⊥*}

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[◇]Department of Chemistry, University of California, Berkeley, California 94720, United States

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[⊥] Department of Chemistry & Nano-Science Center, University of Copenhagen, Copenhagen 2100, Denmark

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Acknowledgements



Prof. Jiwoong Lee



Martin Juhl



Gul Barg Hadaf



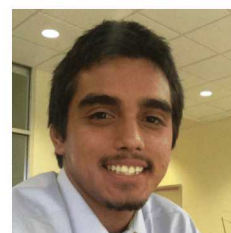
Dasol Hwang



Prof. Jeffrey Long



Julia Oktawiec



Ever Velazquez



Prof. Tomče Runčevski



Jonathan Lefton



Mercedes Taylor

Center for Gas Separations
Relevant to Clean Energy Technologies



U.S. DEPARTMENT OF
ENERGY

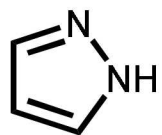


VILLUM FONDEN



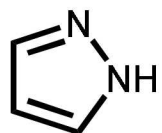
novo
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Pyrazole as a coordinating group

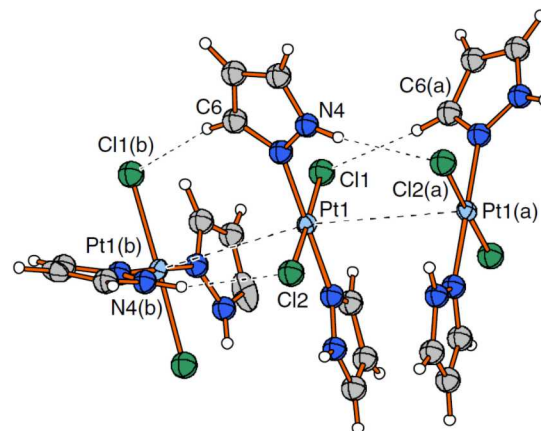


1*H*-Pyrazole

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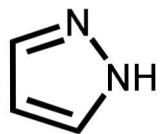


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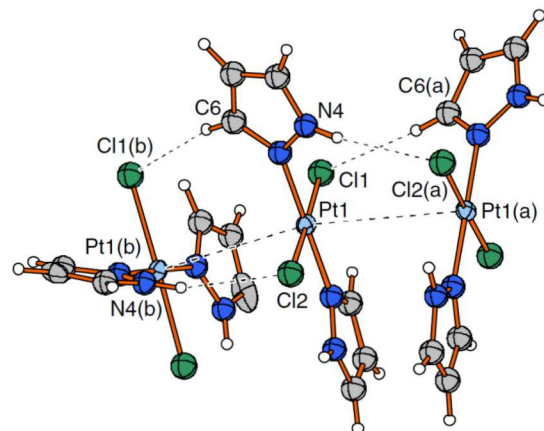


Inorg. Chim. Acta **359** (2006) 320-326.

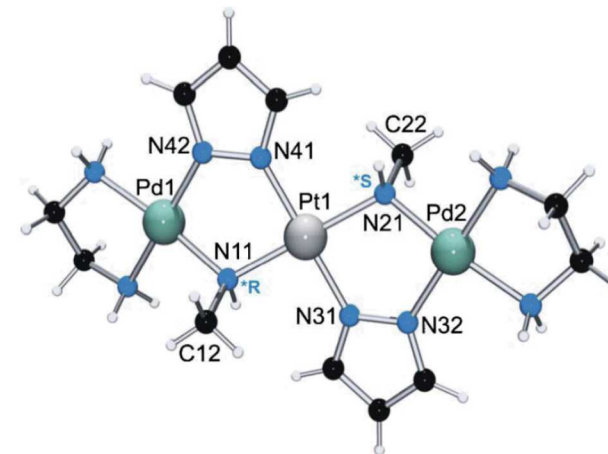
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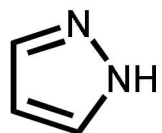


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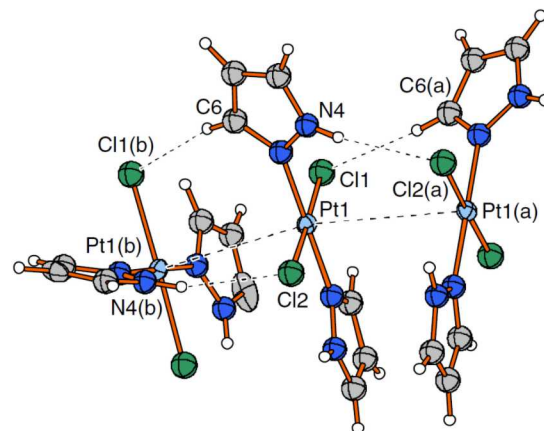


Dalton Trans **40** (2011) 10316-10318.

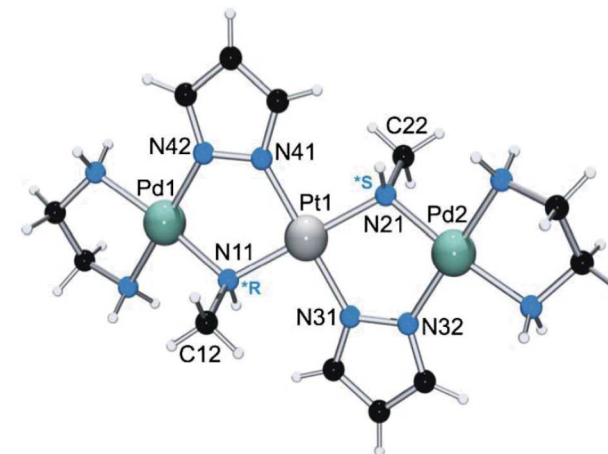
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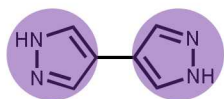
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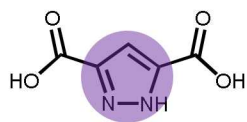
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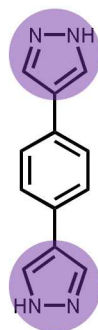
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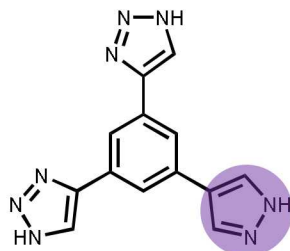
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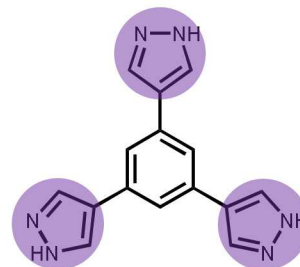
J Inorg. Organomet. Polym. **22** (2012) 1370-1376



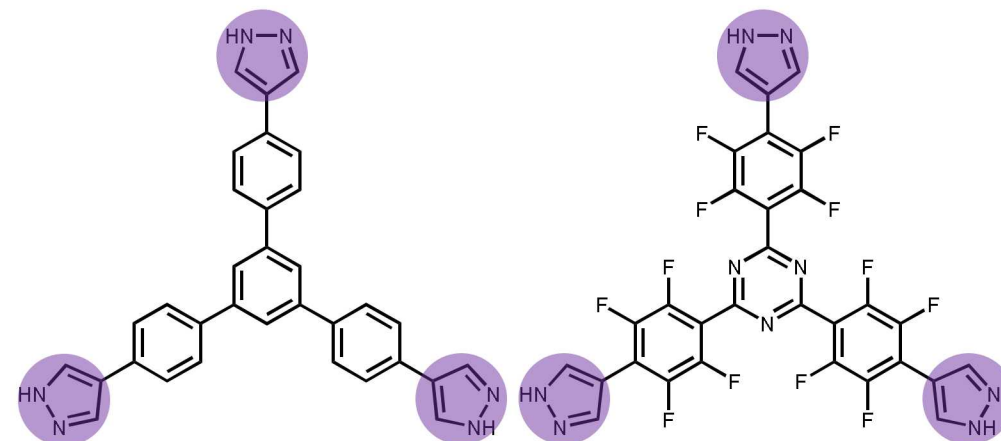
J Am. Chem. Soc. **130** (2008) 7847-7850



J. Am. Chem. Soc. **138** (2016) 7161-7170



Chem. Sci. **2** (2011) 1311-1319



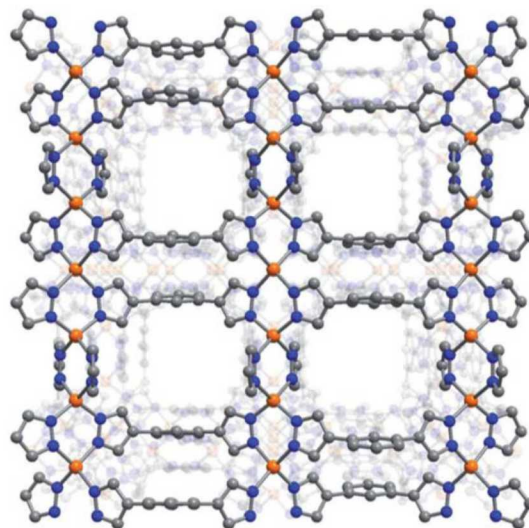
CrystEngComm **17** (2015) 4992-5001

J. Am. Chem. Soc. **140** (2018) 6014-6026

Pyrazolate-based metal–organic frameworks

Metal–organic frameworks:

- 3-dimensional materials
- Metal ions + organic linkers
- Crystalline
- Porous
- Tunable



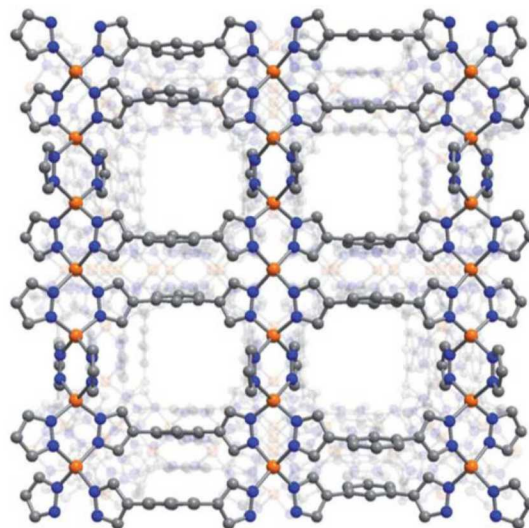
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- ✓ Gas adsorption
- ✓ Separations
- ✓ Catalysis
- ✓ Conductivity
- ✓ Magnetism



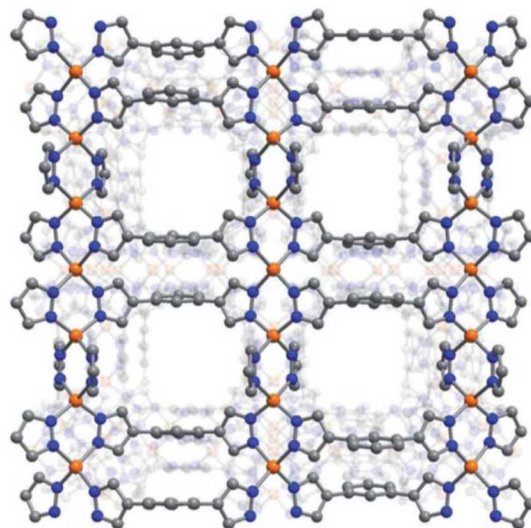
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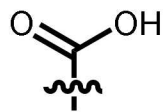
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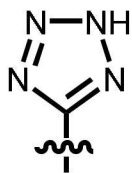


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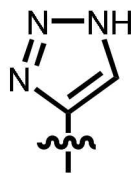
Organic linkers: carboxylate and azolate coordinating groups



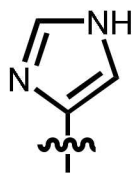
pKa: 4.2
(benzoic acid)



pKa: 4.9



pKa: 13.9



pKa: 18.6



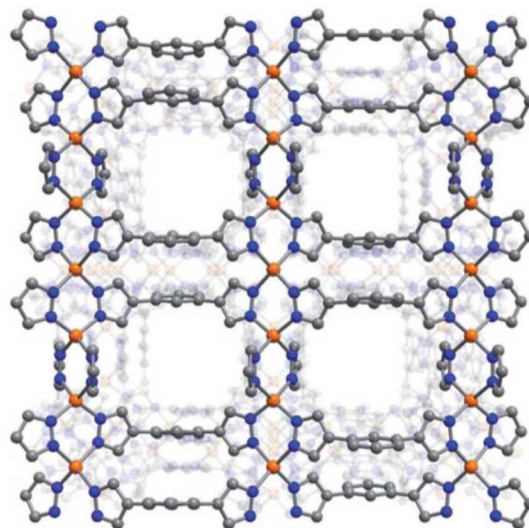
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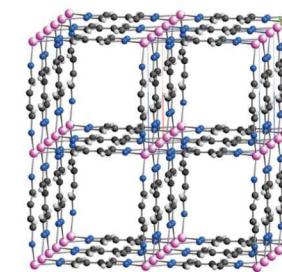
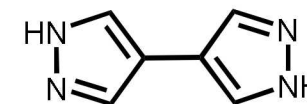
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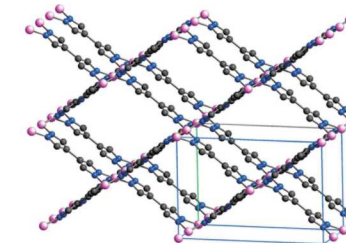
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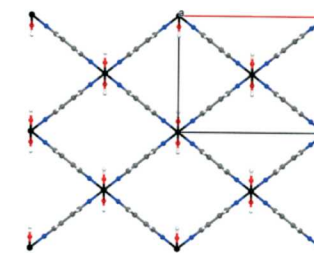
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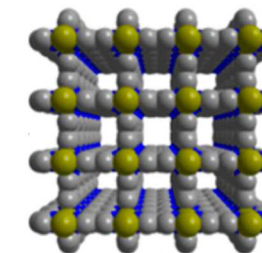
Inorg. Chem. **51** (2012) 5235–5245



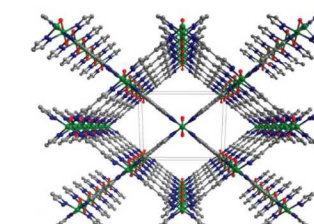
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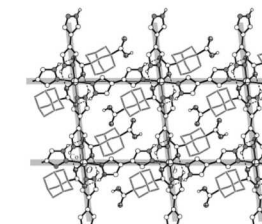
CrystEngComm **17** (2015) 313–322



Inorg. Chem. **56** (2017) 12337–12347

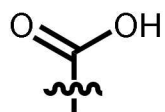


Dalton Trans. **47** (2018) 8779–8786

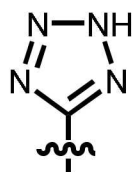


Acta Cryst. **C69** (2013) 232–236

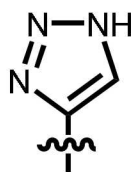
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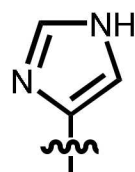
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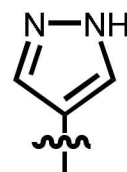
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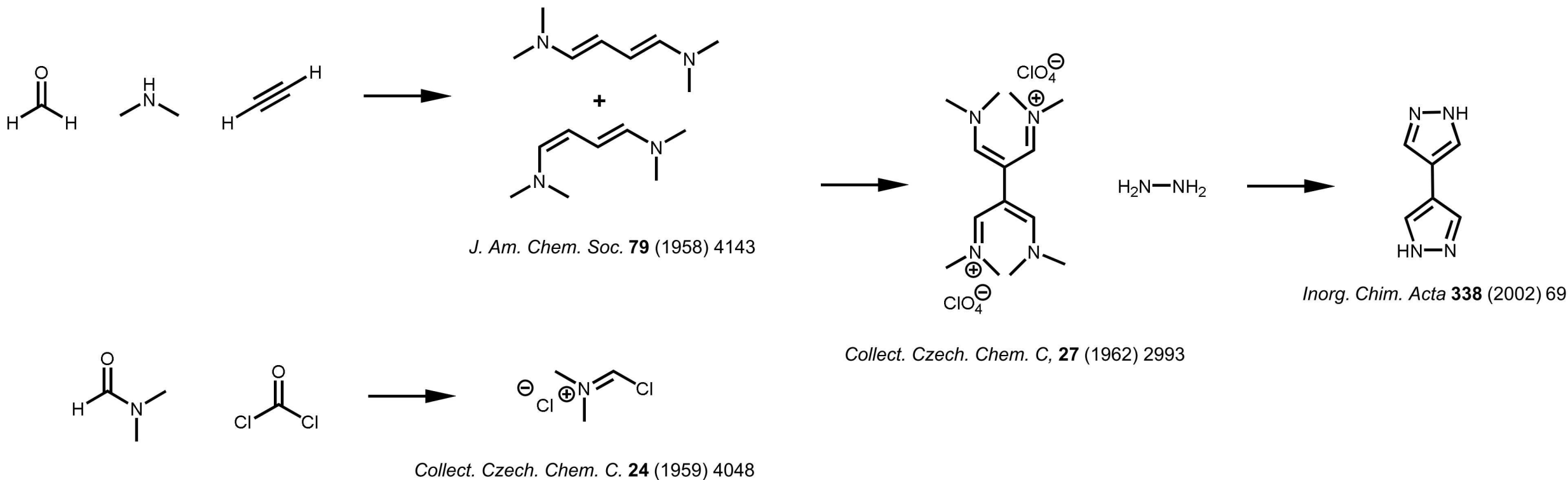
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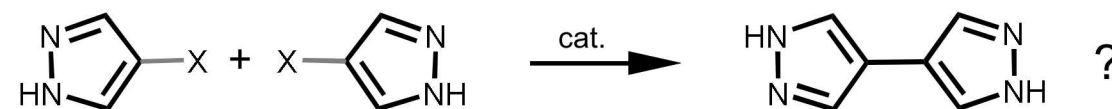
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- Shorter synthesis developed by Domasevitch and coworkers in 2002



Bipyrazole via homocoupling?



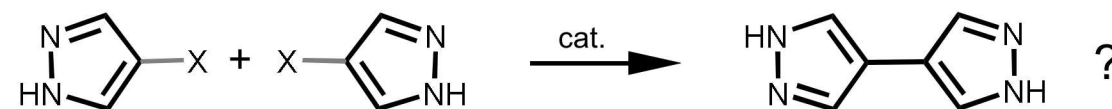
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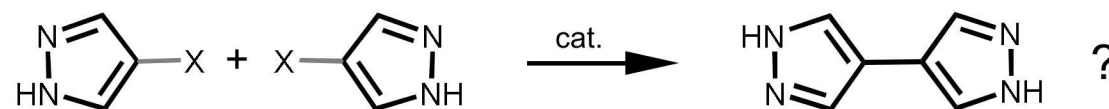
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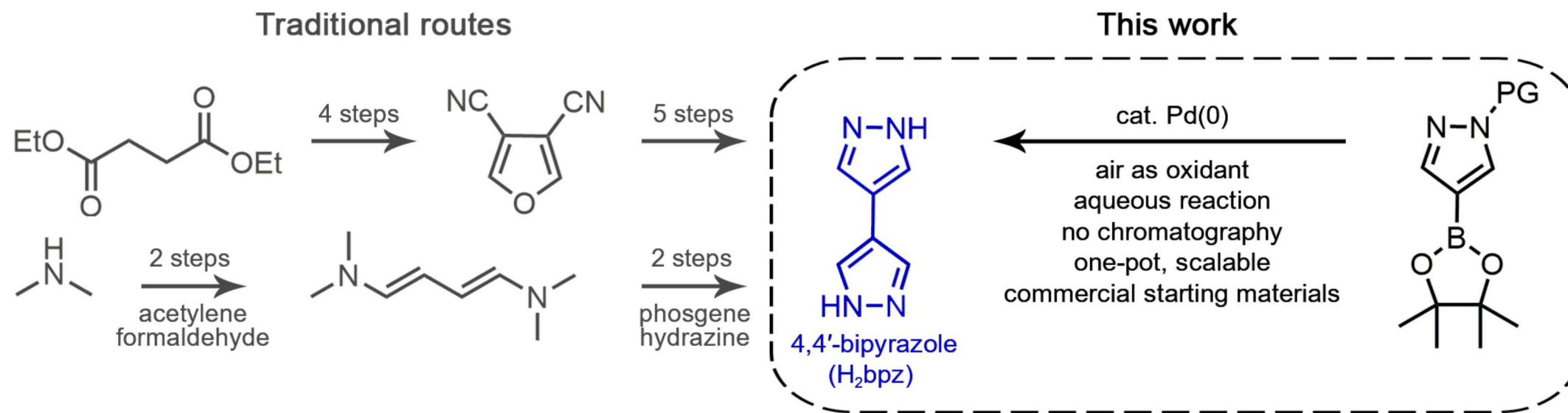
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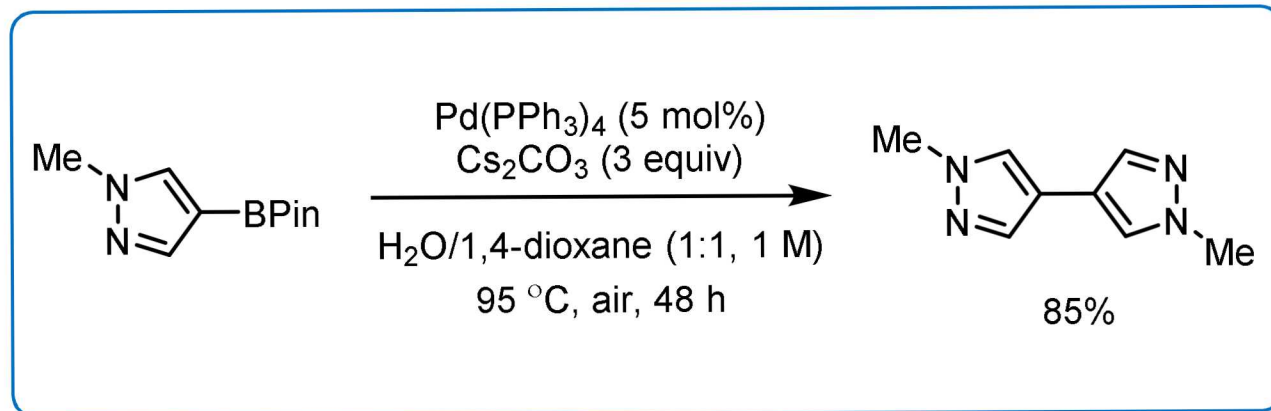


Optimization of Pd-catalyzed homocoupling



- ✓ Catalyst
- ✓ Base
- ✓ Solvent system
- ✓ Atmosphere
- ✓ Temperature
- ✓ Duration
- ✓ Stoichiometry
- ✓ Work-up

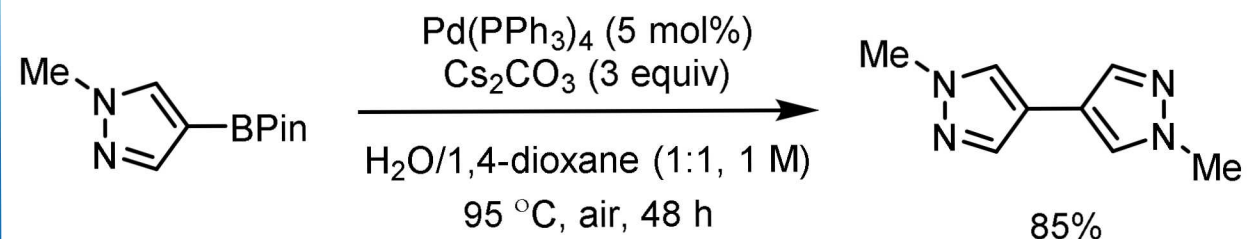
Optimized conditions



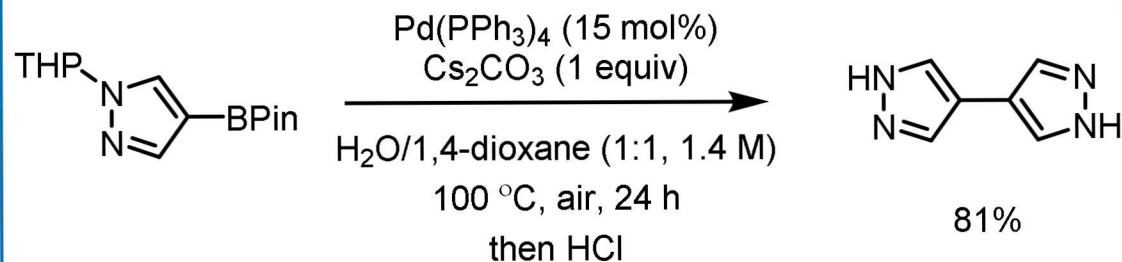
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Improved protecting group

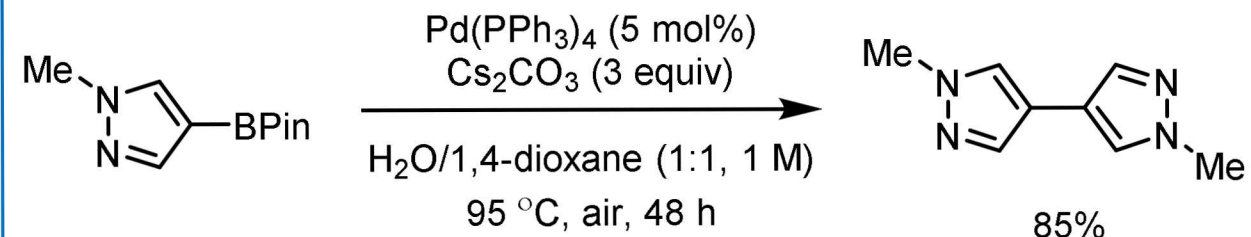


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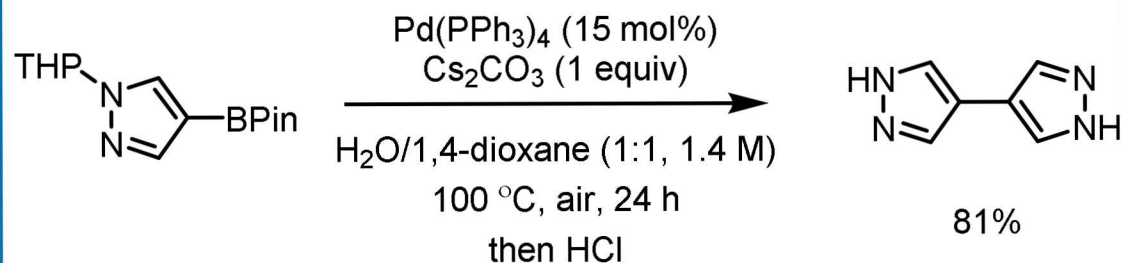


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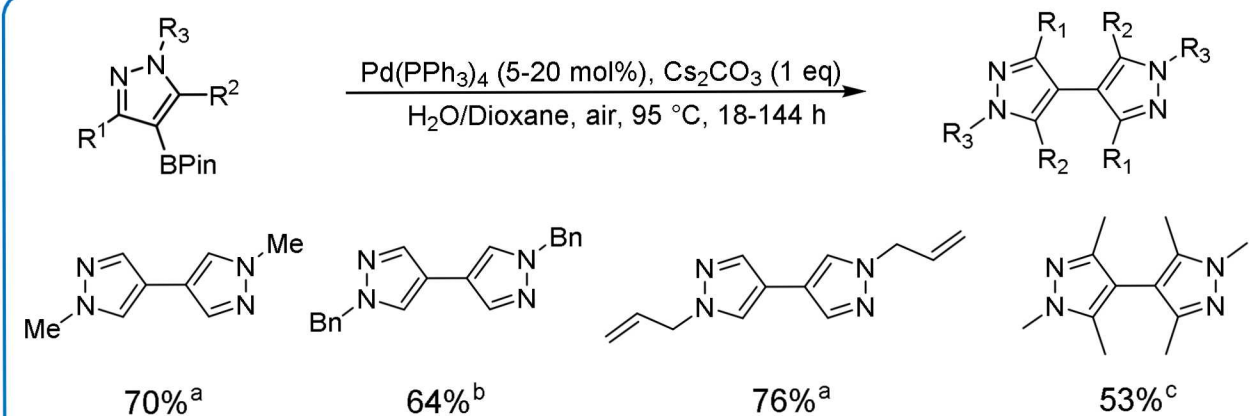
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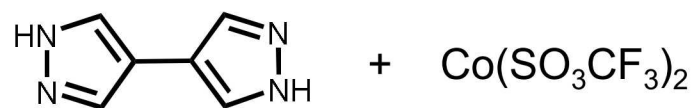
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Substituent variation



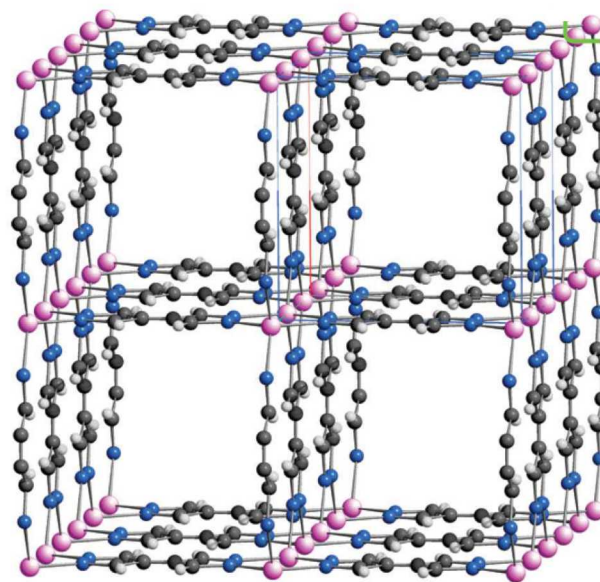
Application to metal–organic framework synthesis



Co(bpz)

Langmuir surface area

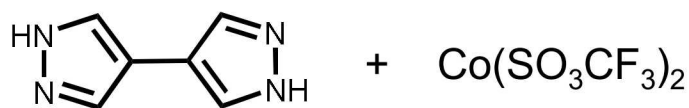
Lit value	1057 m ² /g
This work	1064 m ² /g



Zn(bpz)

Inorg. Chem. **51** (2012) 5235–5245

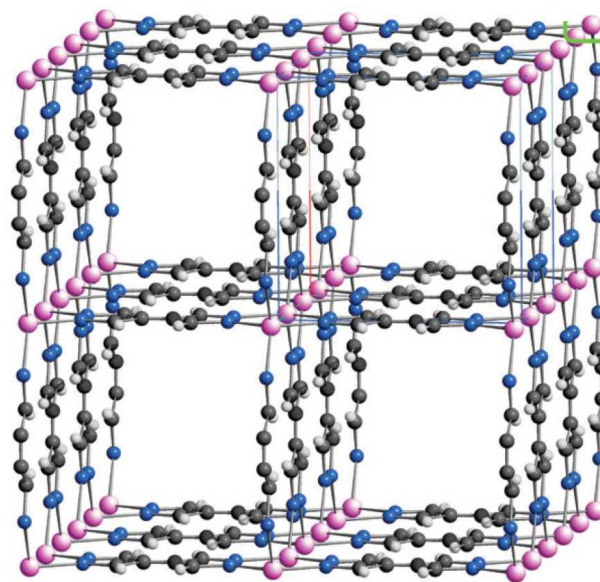
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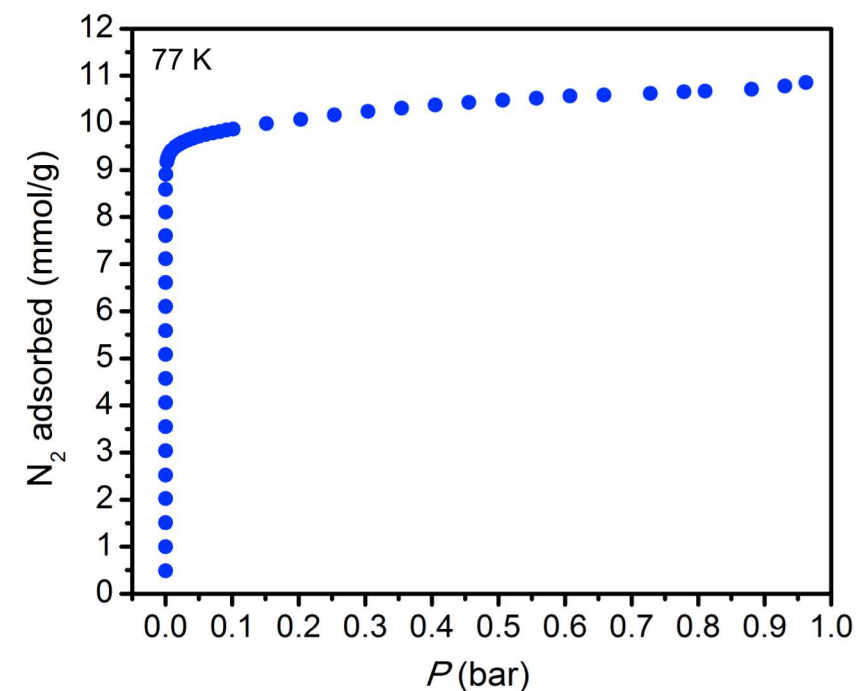
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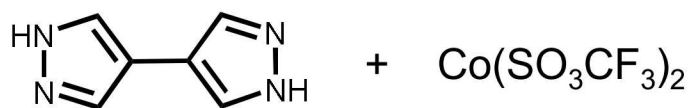


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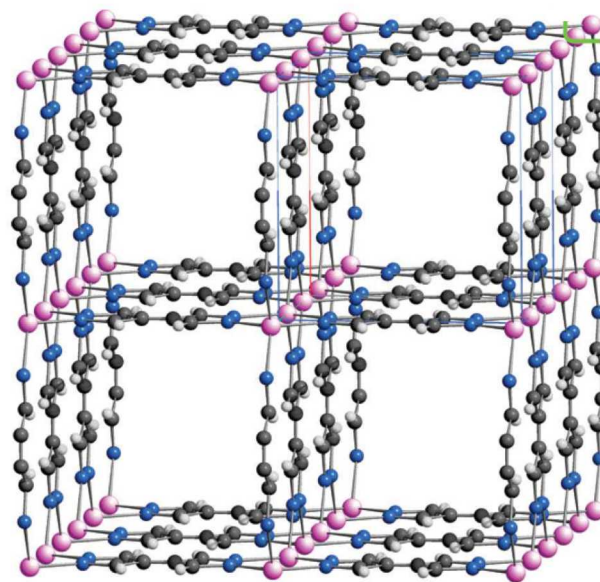
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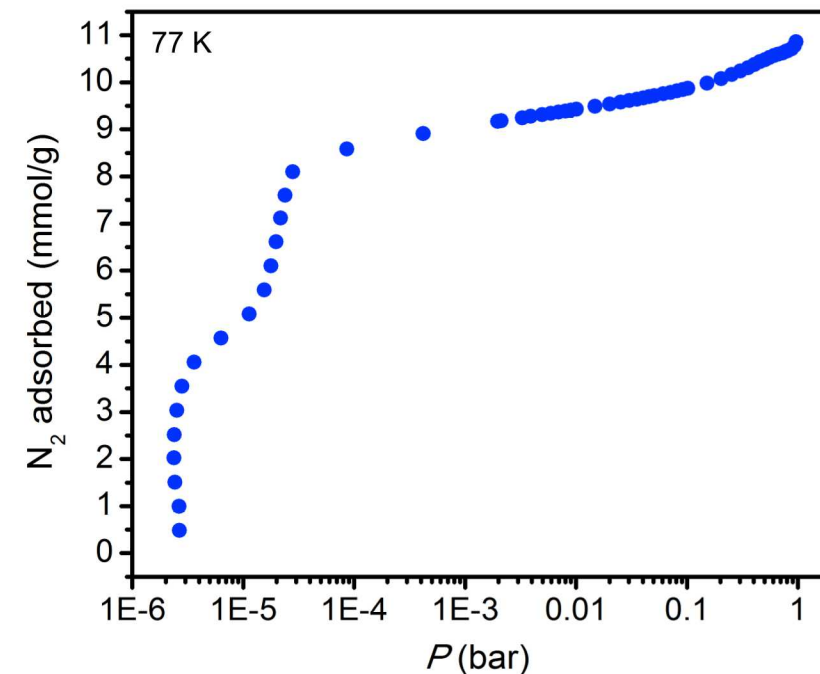
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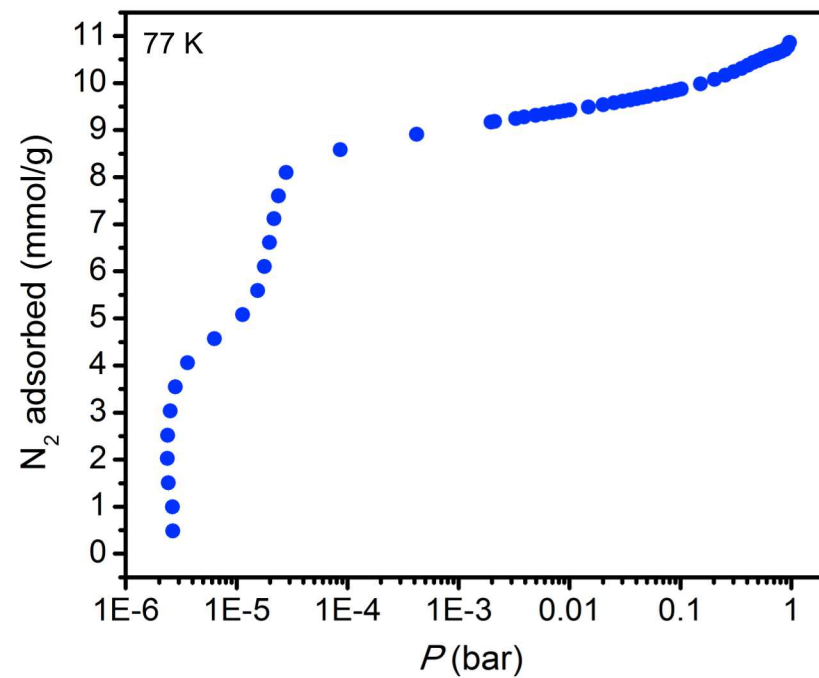


- $\text{Co}(\text{bpz})$ reproducibly demonstrates previously unobserved step in N_2 adsorption isotherm

Stepped isotherms in Co(bpz)

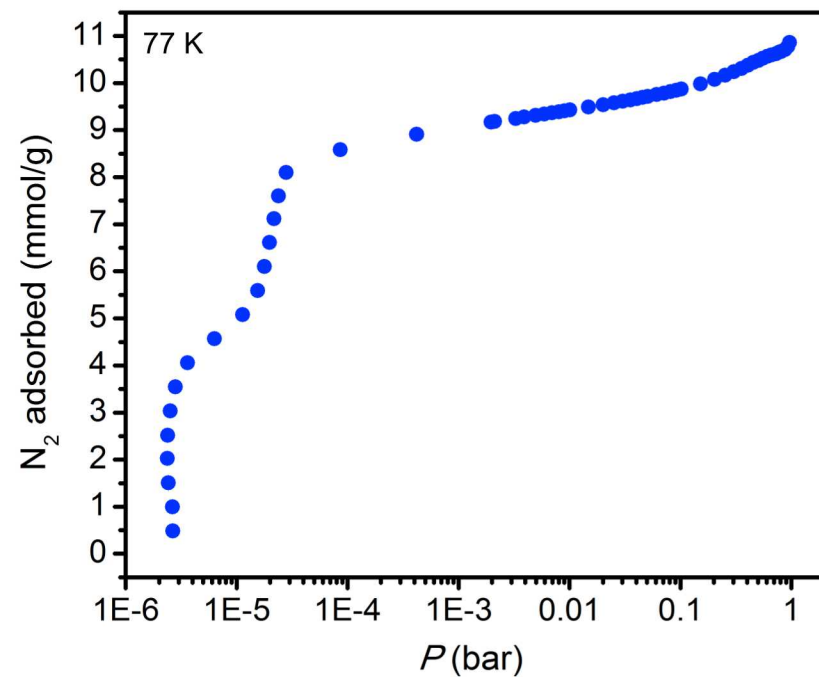


N₂ adsorption

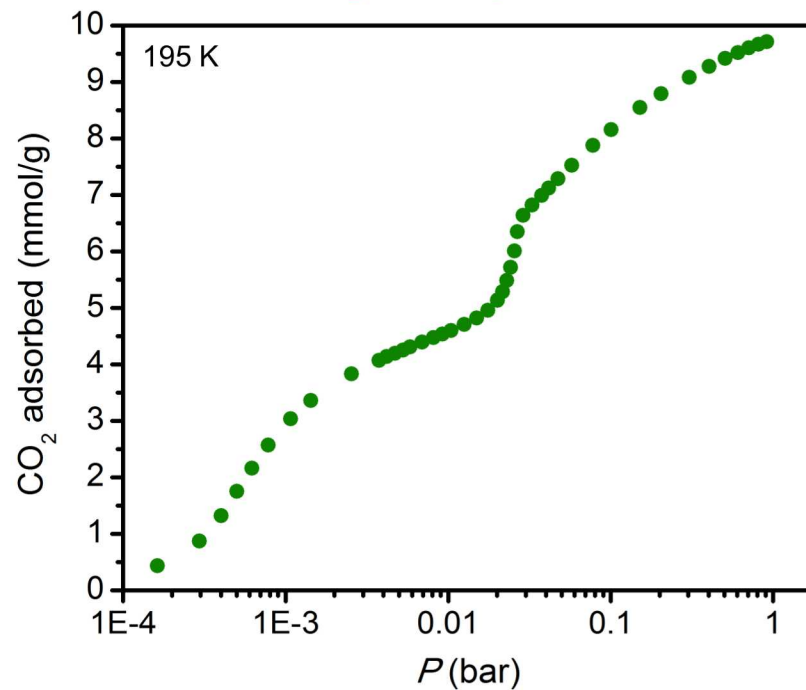


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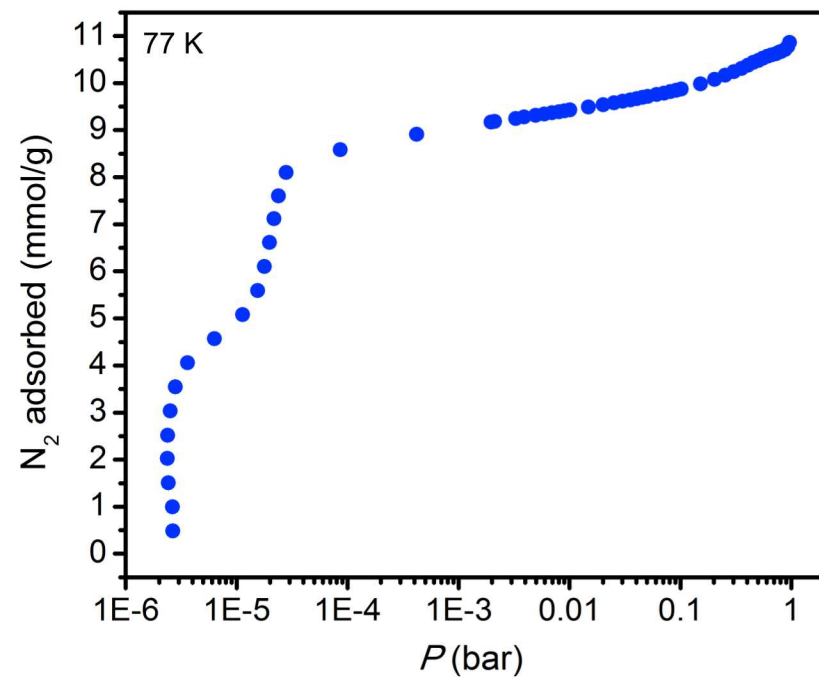


CO₂ adsorption

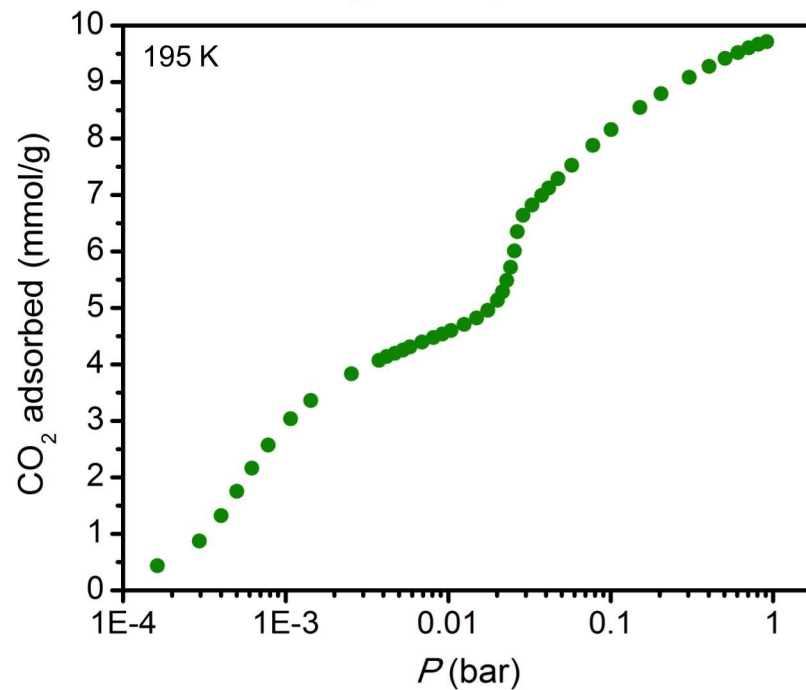


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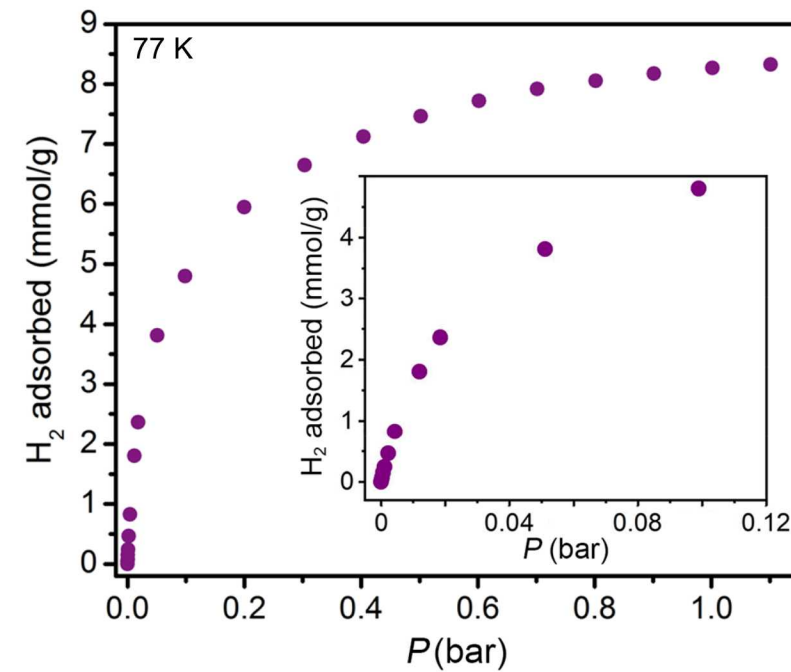
N₂ adsorption



CO₂ adsorption



H₂ adsorption

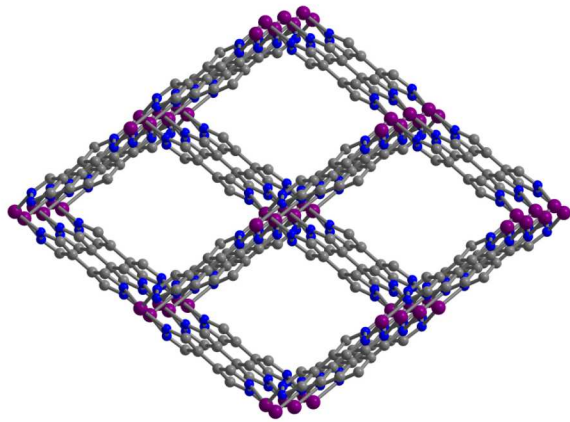


- N₂ and CO₂ isotherms contain steps while H₂ isotherm does not, under conditions tested

In situ diffraction studies for Co(bpz)



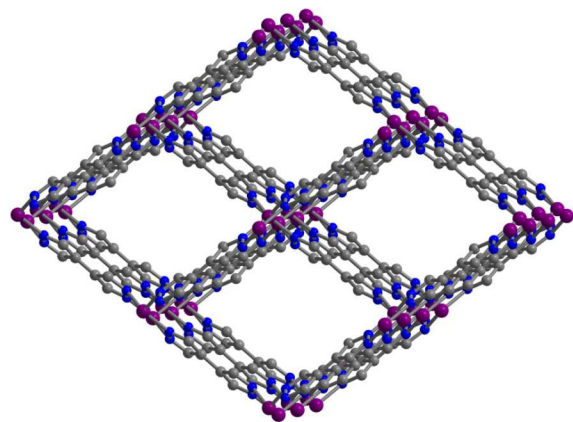
Advanced Photon Source,
Argonne National Lab



Co(bpz), 0 bar

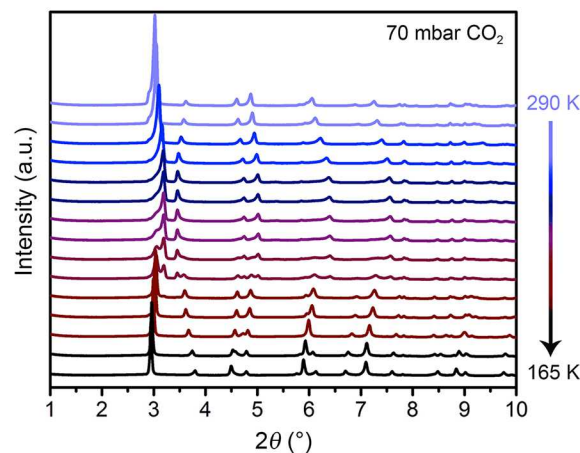
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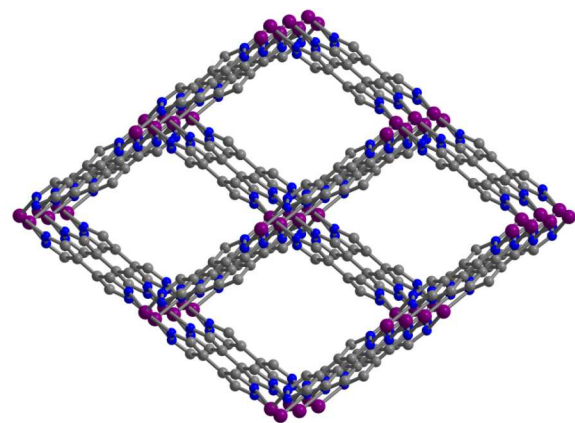
CO₂ dosing studies



Temperature	225 K	197 K	165 K
Space Group	<i>C2/c</i>	<i>C2/c</i>	<i>C2/c</i>
a (Å)	14.927(4)	14.3231(16)	13.675(5)
b (Å)	9.695(3)	10.6736(14)	11.475(4)
c (Å)	7.0387(17)	7.1453(7)	7.227(2)
Volume (Å ³)	1013.1(5)	1088.0(2)	1128.6(6)

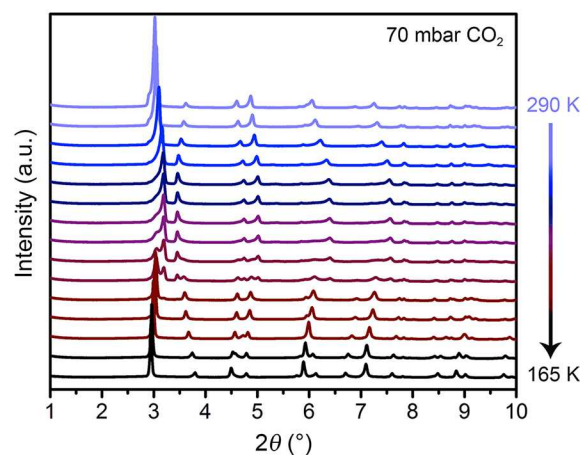
In situ diffraction studies for Co(bpz)

Advanced Photon Source,
Argonne National Lab



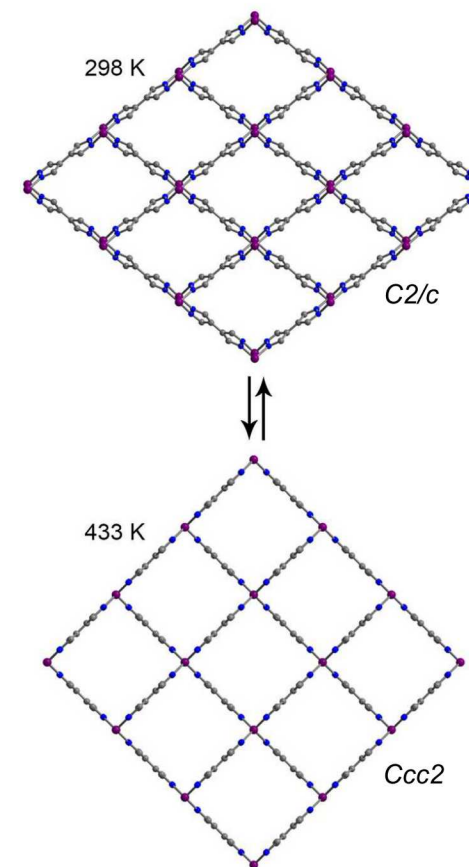
Co(bpz), 0 bar

CO₂ dosing studies

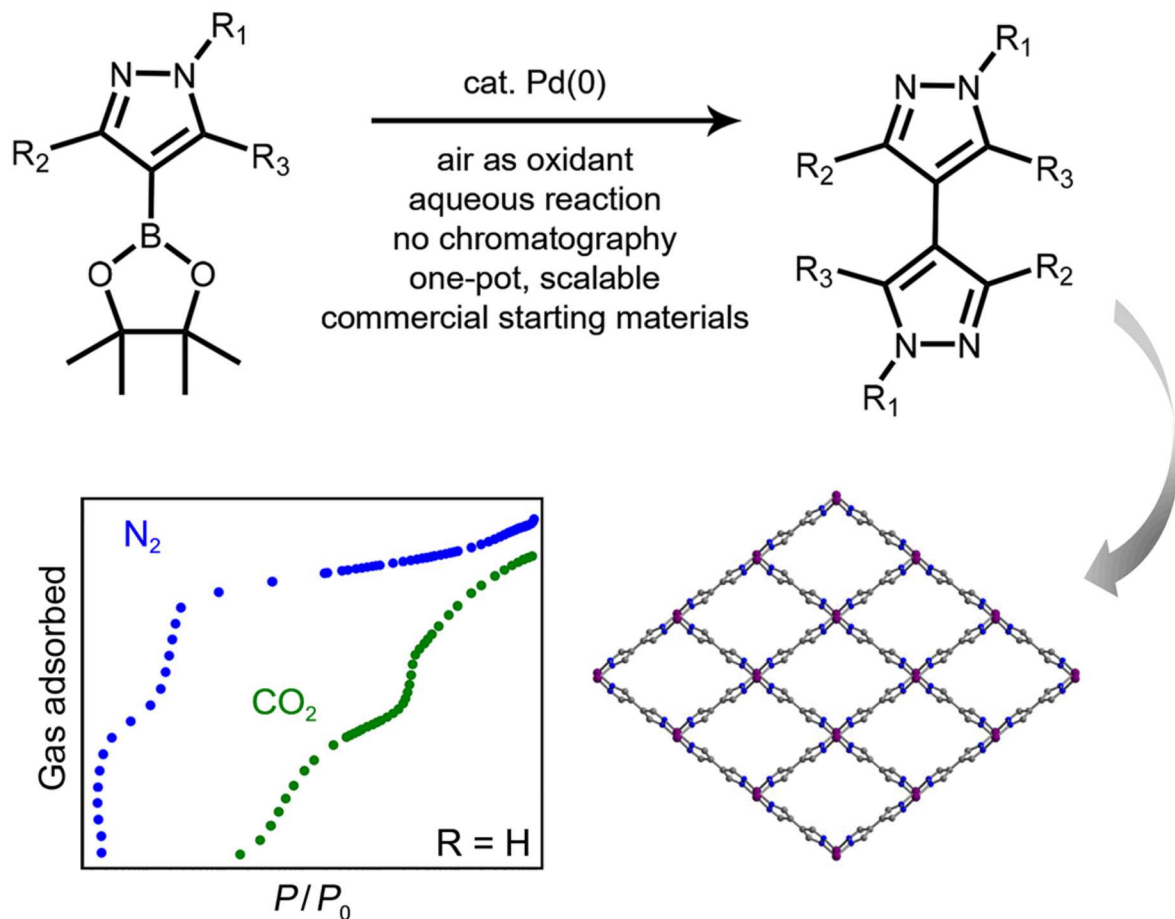


Temperature	225 K	197 K	165 K
Space Group	<i>C2/c</i>	<i>C2/c</i>	<i>C2/c</i>
a (Å)	14.927(4)	14.3231(16)	13.675(5)
b (Å)	9.695(3)	10.6736(14)	11.475(4)
c (Å)	7.0387(17)	7.1453(7)	7.227(2)
Volume (Å ³)	1013.1(5)	1088.0(2)	1128.6(6)

Variable-temperature studies



Conclusions and outlook



- First reported homocoupling of pyrazole boronic esters to access symmetrical bipyrazoles
- Optimized reaction conditions improve route to promising ligands for metal-organic frameworks
- Ligand synthesis enabled first crystal structure of Co(bpz)
- Co(bpz) found to possess adsorbate-dependent and temperature-induced structural flexibility
- Pd-catalyzed homocoupling will allow for more insights into and development of pyrazole-based metal-organic frameworks