

S

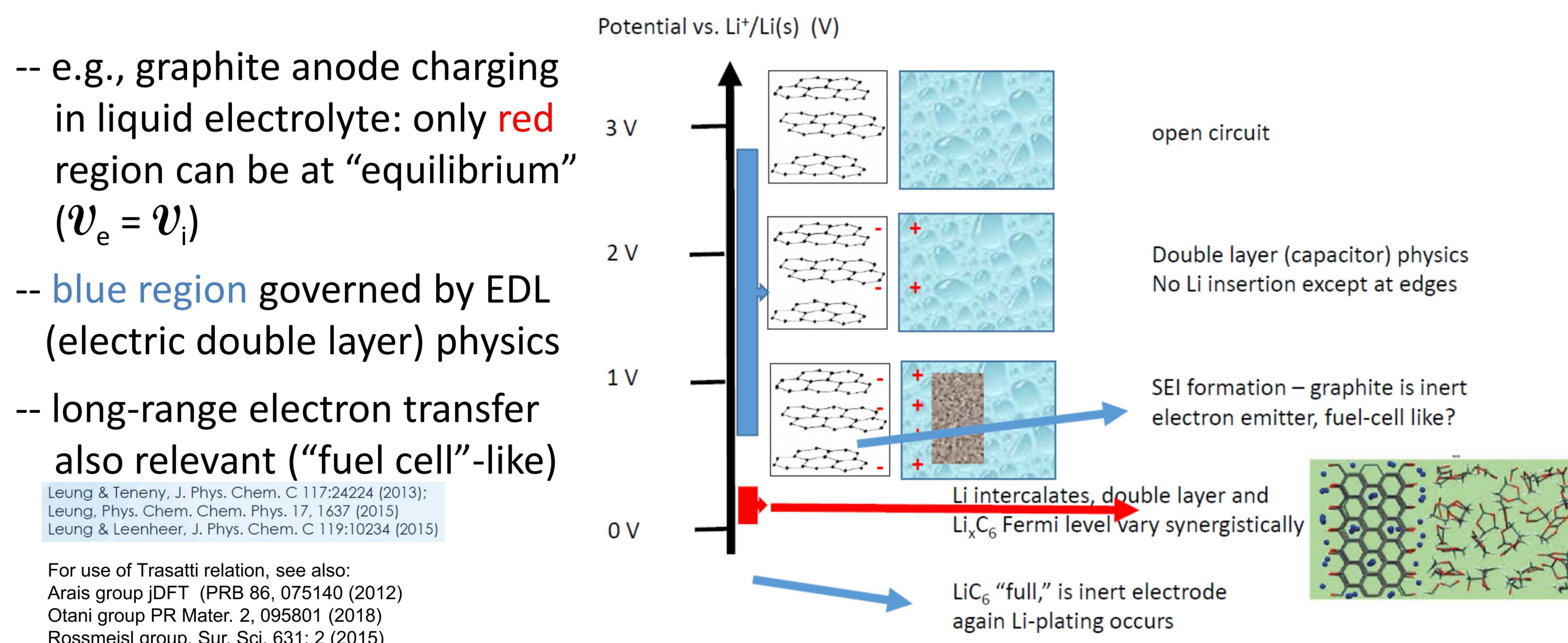
SU contribution

Leung's Perspectives (hopefully not too pedantic!)

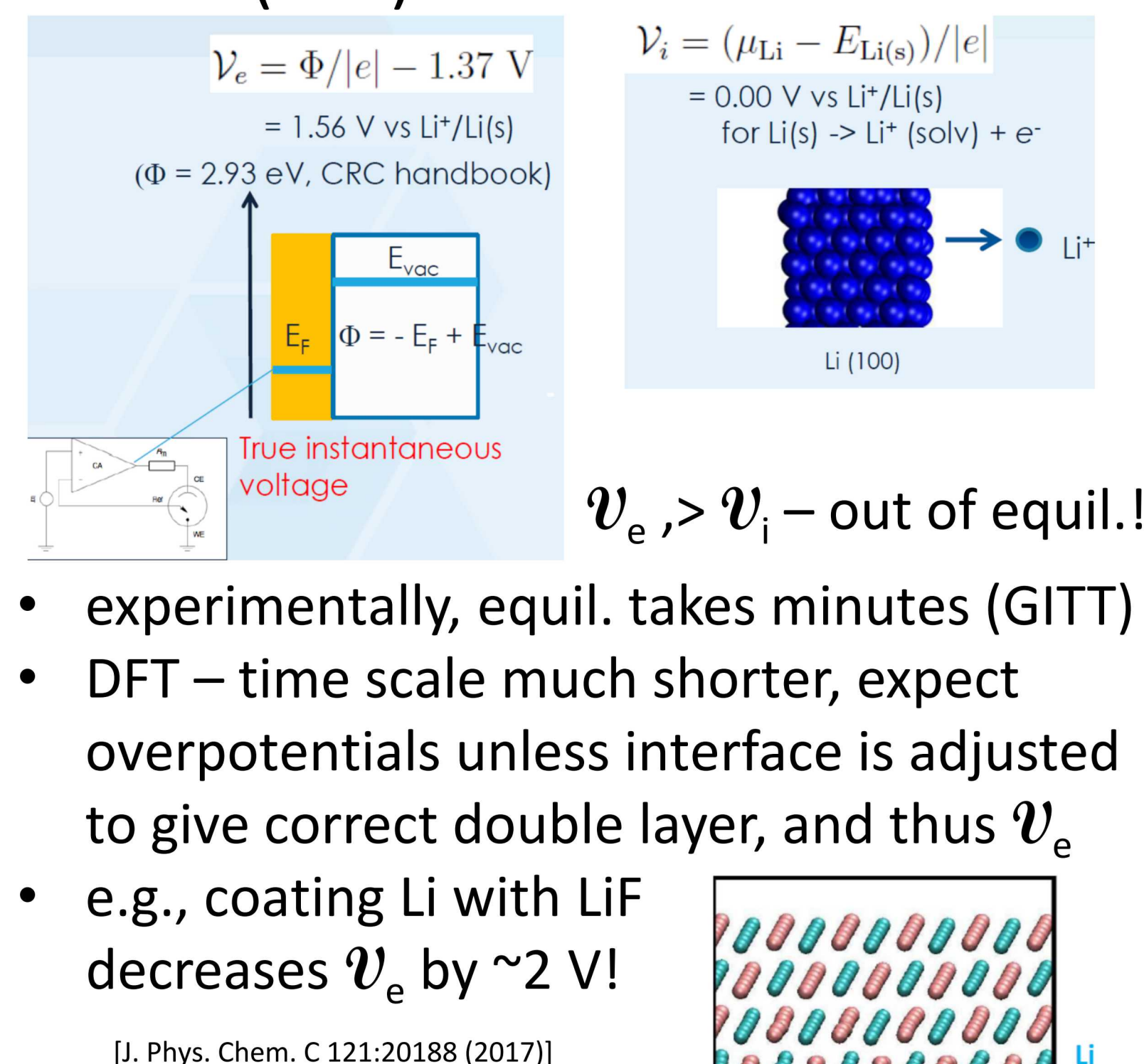
1. Most DFT battery calculations with interfaces in simulation cells are out of electrochemical equilibrium (i.e., are at some overpotential) conditions unless care is taken!
2. Need to take that "care" – otherwise, hard to compare with measurements (e.g., KPFM voltage drop at interface) if DFT calculation has confusion about voltage conditions
3. By "non-equilibrium" I mean the electronic voltage, \mathcal{V}_e , does not match ionic voltage, \mathcal{V}_i

Two Definitions of Voltages: \mathcal{V}_e and \mathcal{V}_i

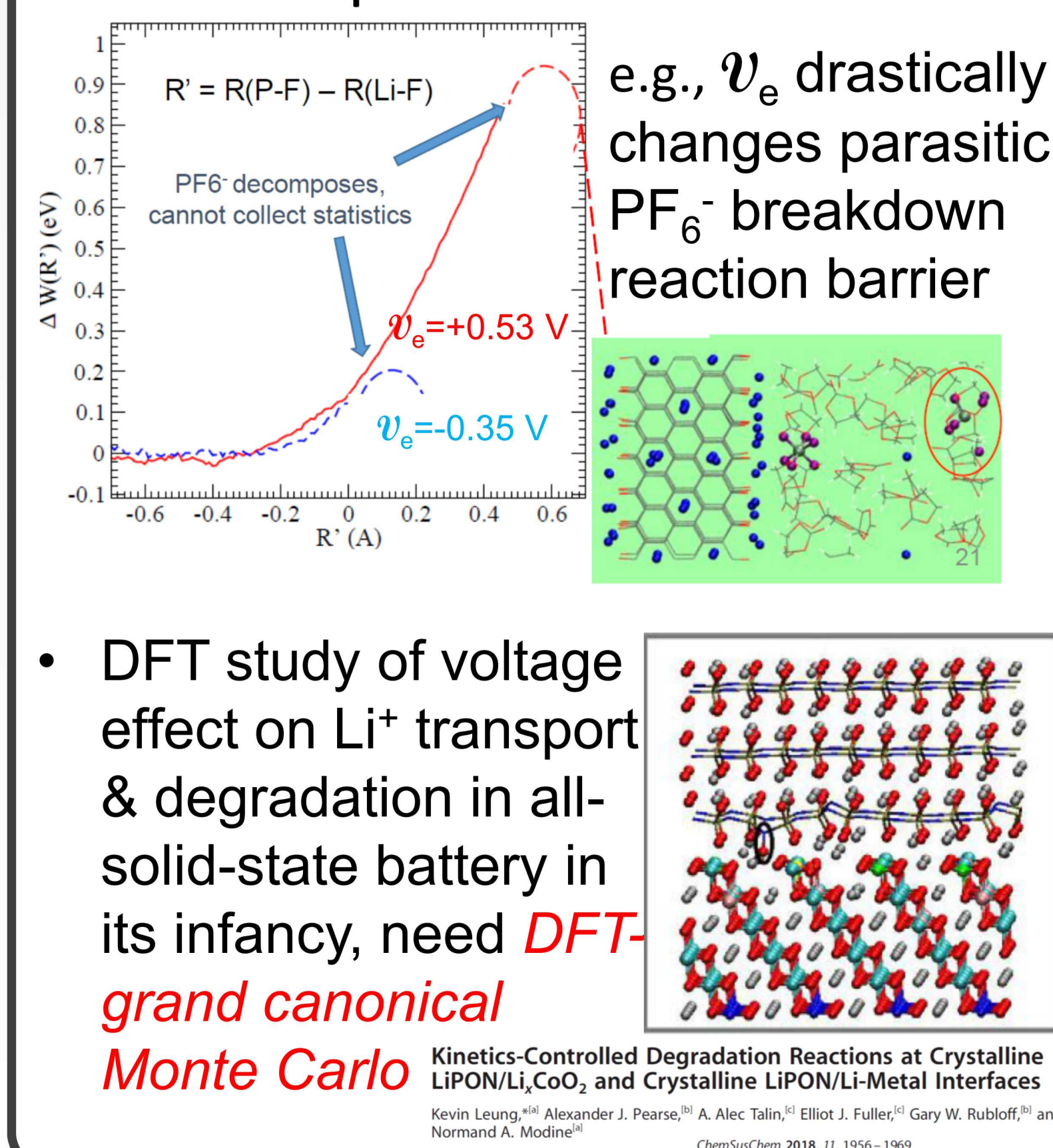
- Unlike other devices, battery electrodes can emit e^- & insert Li atoms – hence 2 "voltages"
- \mathcal{V}_i is familiar to battery theorists: $\mathcal{V}_i = -\frac{\mu_{\text{Li}}^{\text{cathode}}(x) - \mu_{\text{Li}}^{\text{anode}}}{zF}$; at interfaces, doesn't govern \mathcal{V}_e !
- \mathcal{V}_e is "true" or instantaneous voltage reported in fuel cell/supercapacitor modeling work
 - within DFT, most readily calculated via $\mathcal{V}_e = (\text{absolute work function}) - 4.44 \text{ V (vs. S.H.E)}$ [S. Trasatti, *Pure & Appl. Chem.* 58:959 (1986) – cited 700+ times, has a Wikipedia page!]
 - seldom reported in battery DFT work because of historical reasons
 - > most battery DFT modeling work are single phase, no interface in simulation cell
 - > without interfaces, absolute electrostatic potential and work functions not well defined [de Leeuw, Perram, Smith *Proc. Royal Soc. London* 373: 27 (1980)]
 - but in DFT cells with interfaces, can calculate \mathcal{V}_e via Trasatti relation (-1.37 V vs. Li⁺/Li(s))
 - critical for NEES3 goal of elucidating spatial voltage profile during charge/discharge



Quiz: what is "voltage" of lithium (001) slab in vacuum?



Consequences and Outlook



Acknowledgement

Sandia National Laboratories is a multimission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525. This paper describes objective technical results and analysis. Any subjective views or opinions that might be expressed in the paper do not necessarily represent the views of the U.S. Department of Energy or the United States Government.

RESEARCH

Use DFT calculations to calculate the voltage gradient at electrode/electrolyte interface

SCIENCE QU

1. What is the potential landscape at the electrode/electrolyte interface?
2. What do the time scales for charge transfer processes look like?
3. What are the chemical and electrochemical reactions at all-solid electrode/electrolyte interfaces?

SIGNIFICANCE

A correct interpretation of DFT results, including overpotential effects, will remove the controversy in the battery modeling community. Correct voltage control, alongside with Atomic Force Microscopy (KPFM) measurements, will help to understand the effect of voltage gradients at electrode/electrolyte interfaces on lithium ion transport and parasitic reactions. This understanding is critical for the lifetime and performance of solid-state batteries.