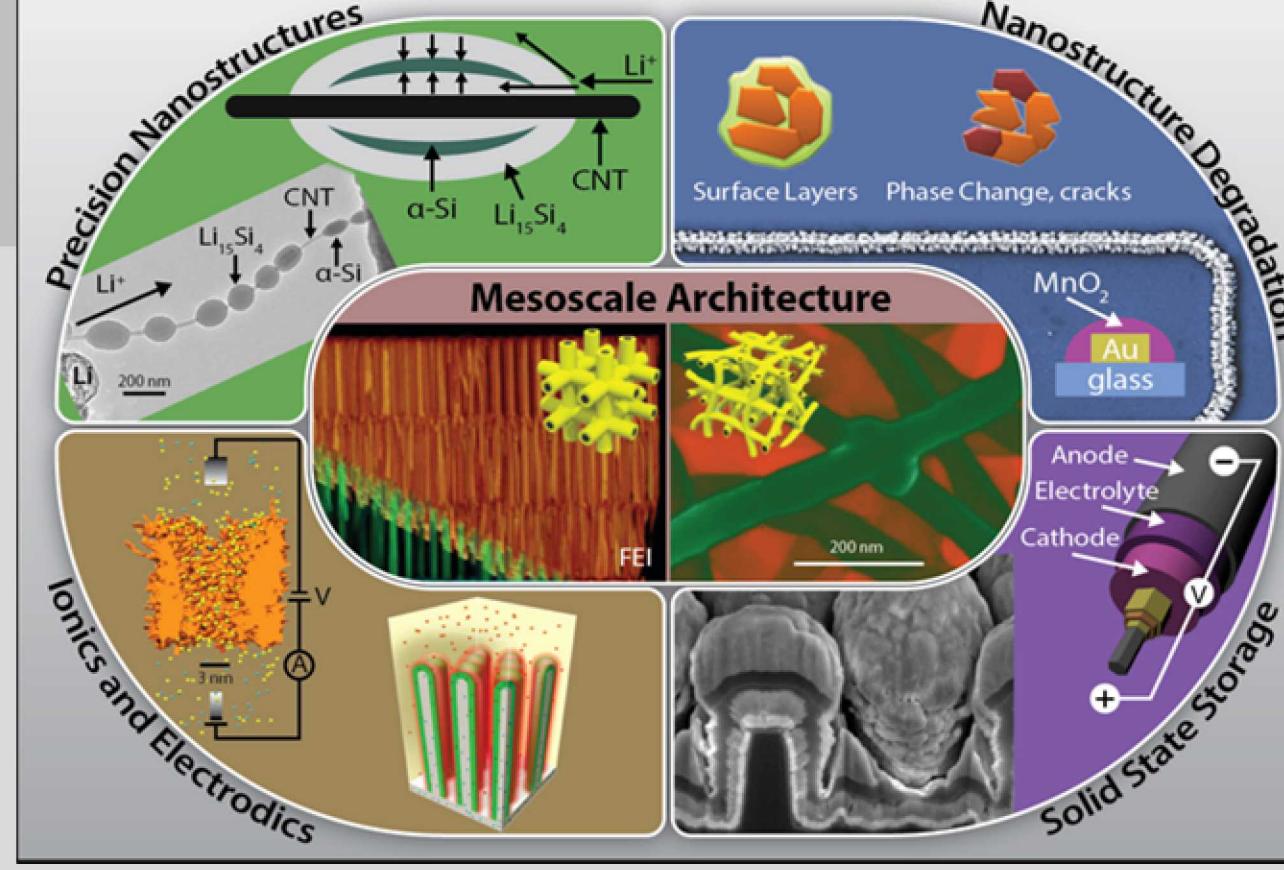


Electrochemical Theory Modeling of Battery Interfaces at Equilibrium and at Overpotentials

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RESEARCH

Use DFT calculations to predict voltage gradient at electrode/electrolyte interface

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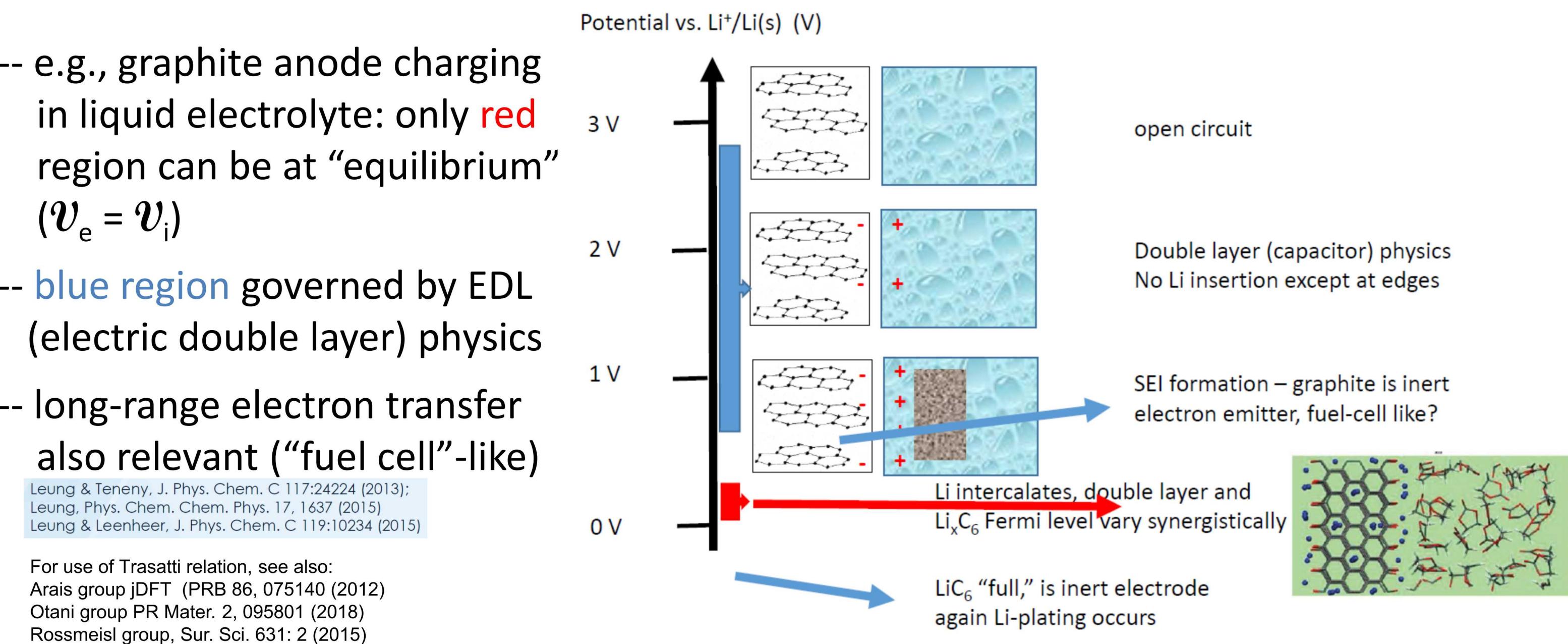
SU contribution

Leung's Perspectives (hopefully not too pedantic!)

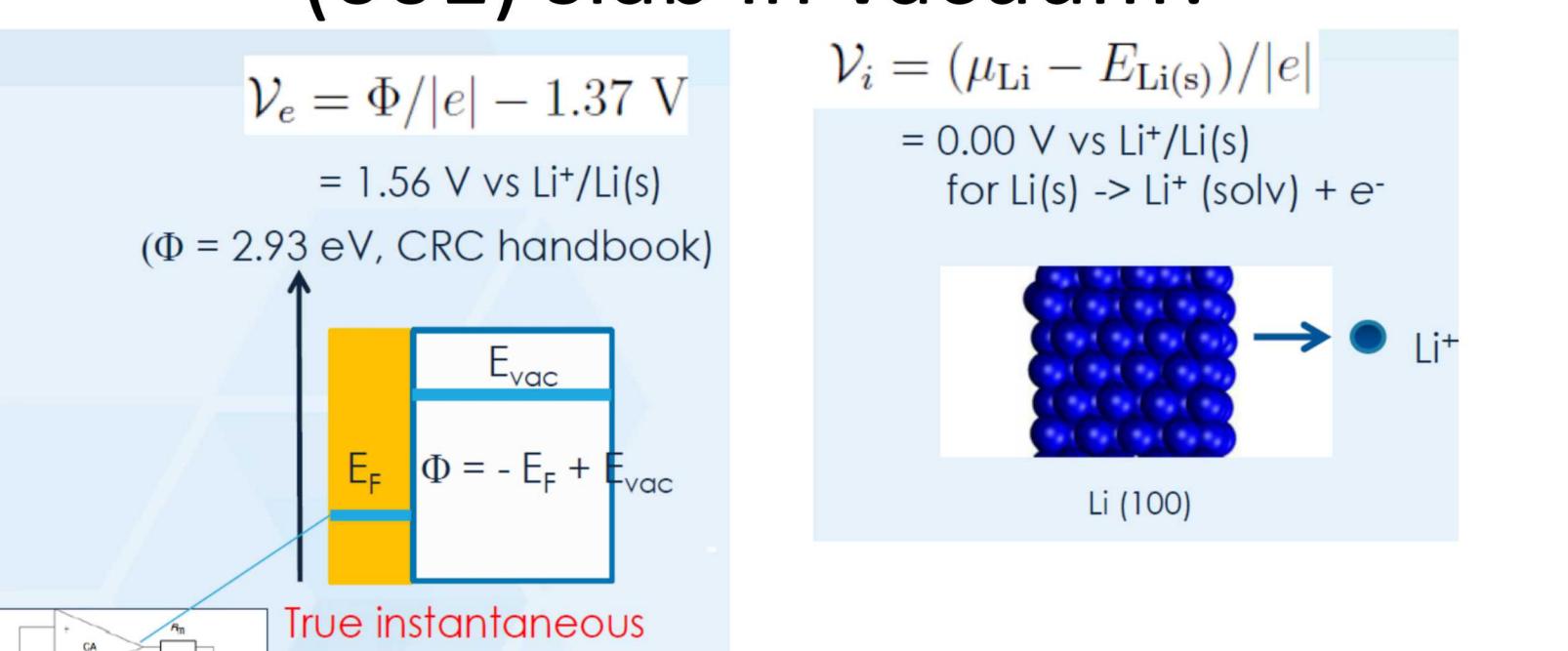
1. Most DFT battery calculations with interfaces in simulation cells are out of electrochemical equilibrium (i.e., are at some overpotential) conditions unless care is taken!
2. Need to take that "care" – otherwise, hard to compare with measurements (e.g., KPFM voltage drop at interface) if DFT calculation has confusion about voltage conditions
3. By "non-equilibrium" I mean the electronic voltage, \mathcal{V}_e , does not match ionic voltage, \mathcal{V}_i

Two Definitions of Voltages: \mathcal{V}_e and \mathcal{V}_i

- Unlike other devices, battery electrodes can emit e^- & insert Li atoms – hence 2 "voltages"
- \mathcal{V}_i is familiar to battery theorists: $\mathcal{V}_i = -\frac{\mu_{\text{Li}}^{\text{cathode}}(x) - \mu_{\text{Li}}^{\text{anode}}}{zF}$; at interfaces, doesn't govern \mathcal{V}_e !
- \mathcal{V}_e is "true" or instantaneous voltage reported in fuel cell/supercapacitor modeling work
 - within DFT, most readily calculated via $\mathcal{V}_e = (\text{absolute work function}) - 4.44 \text{ V (vs. S.H.E)}$ [S. Trasatti, *Pure & Appl. Chem.* 58:959 (1986) – cited 700+ times, has a Wikipedia page!]
 - seldom reported in battery DFT work because of historical reasons
 - > most battery DFT modeling work are single phase, no interface in simulation cell
 - > without interfaces, absolute electrostatic potential and work functions not well defined [de Leeuw, Perram, *Smith Proc. Royal Soc. London* 373: 27 (1980)]
 - but in DFT cells with interfaces, can calculate \mathcal{V}_e via Trasatti relation (-1.37 V vs. $\text{Li}^+/\text{Li}(s)$)
 - critical for NEES3 goal of elucidating spatial voltage profile during charge/discharge

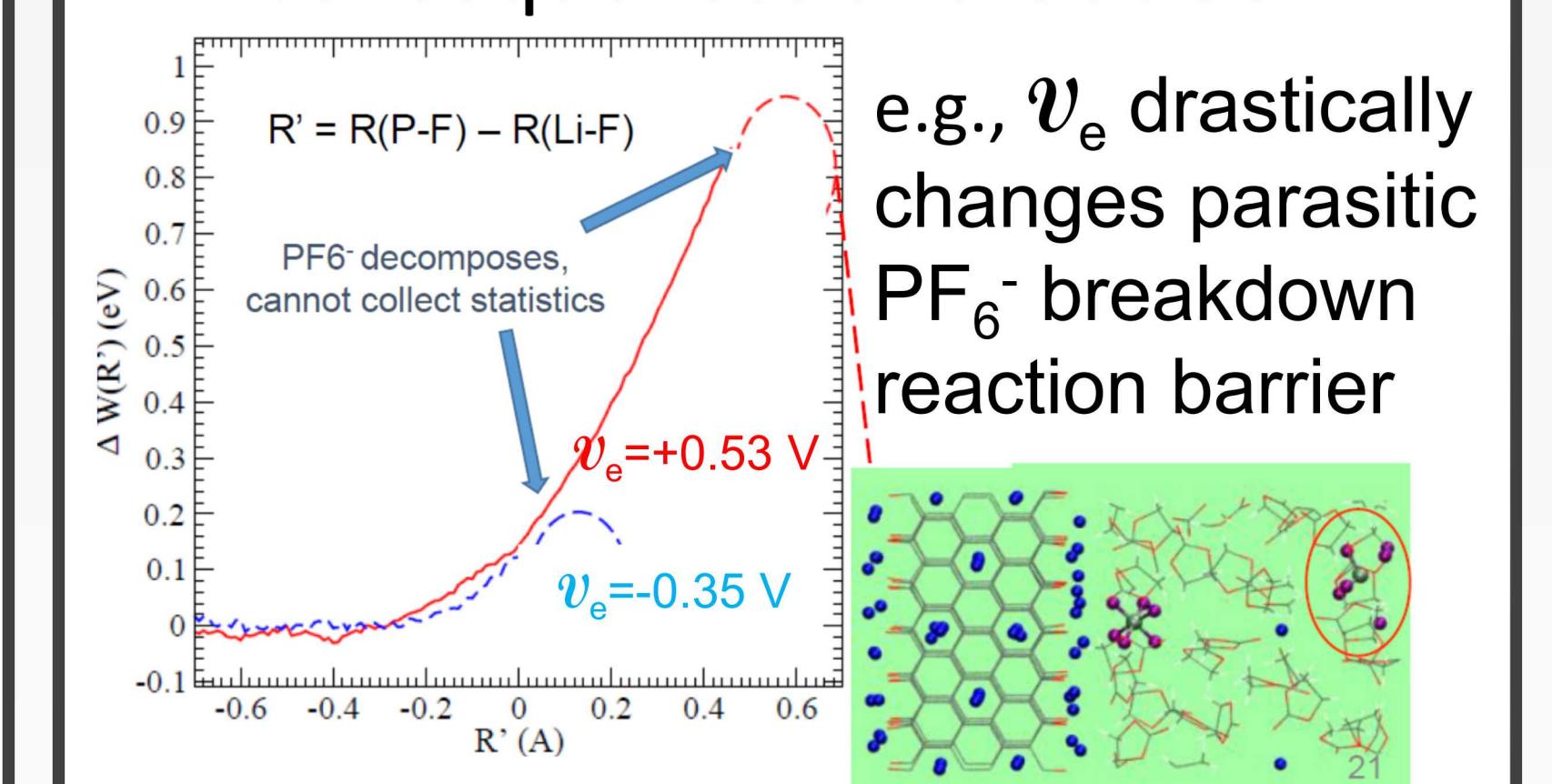


Quiz: what is "voltage" of lithium (001) slab in vacuum?



- experimentally, equil. takes minutes (GITT)
 - DFT – time scale much shorter, expect overpotentials unless interface is adjusted to give correct double layer, and thus \mathcal{V}_e
 - e.g., coating Li with LiF decreases \mathcal{V}_e by $\sim 2 \text{ V}$!
- [J. Phys. Chem. C 121:20188 (2017)]

Consequences and Outlook



- DFT study of voltage effect on Li^+ transport & degradation in all-solid-state battery in its infancy, need **DFT grand canonical Monte Carlo**
- Kinetics-Controlled Degradation Reactions at Crystalline LiPON/Li₂CoO₃ and Crystalline LiPON/Li-Metal Interfaces
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SCIENCE QUESTIONS

1. What is the potential landscape at the electrolyte interface?
2. What do the time scales for various transfer processes look like?
3. What are the chemical and physical reactions at all-solid electrolyte interfaces?

SIGNIFICANCE

A correct interpretation of DFT overpotential effects, will remove controversy in the battery modeling field. Correct voltage control, along with Microscopy (KPFM) measurement of voltage gradients at electrode/electrolyte interfaces on lithium ion transport and parasitic reactions. This will improve the lifetime and performance of batteries.

NATIONS & SYNERGY