

1 Growth response of environmental bacteria under

2 exposure to nitramines from CO₂-capture

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Abstract

15 Nitramines are potentially carcinogenic by-products of amines used in post-combustion
16 CO₂-capture. The influence of monoethanol (MEA)-, monomethyl (MMA)-, and
17 dimethyl (DMA)-nitramines on the growth of environmental strains of bacteria,
18 *Pseudomonas fluorescens* (*P. fluorescens*) and *Rhodococcus spp.* (*R. spp.*), was investigated in the
19 laboratory. Additionally, the persistence of the nitramines in the presence of bacteria was
20 determined. Growth of *R. spp.* was found to be sensitive to MMA-nitramine (EC₅₀ = 157
21 mg L⁻¹), while *P. fluorescens* growth was insensitive to all nitramines tested. Moreover, *P.*
22 *fluorescens* was capable of degrading 8-10% of the nitramines during the 33 h experiments.
23 Results from this study provide insight into important processes of bacterial response to
24 nitramines that merit further investigation considering the ongoing implementation of
25 CO₂ capture technology.

26 **Keywords:** Biodegradation; CO₂ capture; Ecotoxicity; LC-MS; Nitramine

27 **1. Introduction**

28 Technology of CO₂ capture offers the opportunity to reduce greenhouse gas emissions
29 from existing large-scale point sources. Most climate models rely on global-scale
30 implementation of the technology to limit global warming to 2 °C (and especially 1.5 °C)
31 (IPCC, 2014). Currently, the most feasible way of capturing CO₂ is using amines post-
32 combustion (Rochelle, 2009; Wang et al., 2011). However, several potentially carcinogenic
33 nitramines may form from the amines used in post-combustion CO₂ capture. Formation
34 occurs in such a way that direct introduction to the nearby environment is inevitable
35 (Nielsen et al., 2012). A thorough risk assessment, constituting both a prediction of final
36 exposure levels and a detailed investigation of the toxicity of the relevant nitramines, is
37 lacking (Chen et al., 2018).

38 From the “benchmark” amine solvent, monoethanolamine, the following three nitramines
39 can form: MEA-, MMA-, and DMA-nitramines. As these nitramines are small and polar
40 ($M \approx 106.8 \text{ g mol}^{-1}$ and $S_w \approx 176 \text{ g L}^{-1}$) they are thought to partition readily into the
41 aqueous phase. However, a preceding study found that MEA- and DMA-nitramines
42 preferentially bind to soils rich in organic matter (Gundersen et al., 2017a). Surface soil
43 horizons are typically rich in organic matter and high in biological activity, especially in
44 boreal ecosystems. Brakstad et al. (2018) investigated the biodegradability of MEA-,
45 MMA-, and DMA-nitramines. They found none of them to be readily biodegradable by
46 the standards of the OECD Guideline 301, which requires 60% decay. The apparent low
47 biodegradability may be caused by toxic effects from the nitramines.
48 Studies on the potential ecotoxic effect of relevant nitramines are summarized in Table 1.
49 The most sensitive response was found in our preceding study, where a natural

50 oligotrophic lake-water bacterial community showed an estimated half effective
51 concentration (EC_{50}) of 10 mg L^{-1} MEA-nitramine (Gundersen et al., 2014). Other studies
52 focusing on species from higher trophic levels (e.g. phytoplankton or larvae) report higher
53 EC_{50} values, ranging from 47 to $> 2000\text{ mg L}^{-1}$ (Table 1). For algae, a growth assay
54 showed an EC_{50} of 591 mg L^{-1} for DMA-nitramine (Courtris et al., 2015).

55 The aim of this study was to explore the bacterial response to MEA-, MMA-, and DMA-
56 nitramines exposure in pure cultures of environmental strains of *P. fluorescens* and *R. spp.*

57 The two bacteria were selected for their high environmental relevance. They are both
58 abundant in soils and water, and they represent the two major groups of bacteria based on
59 cell wall structure (Gram staining). Moreover, other studies have found strains of *P.*
60 *fluorescens* and *R. spp.* capable of degrading a range of different types of contaminants
61 (Agarry & Solomon, 2008; Martinkova et al., 2009), including the cyclic nitramine
62 explosive, known as RDX (Coleman et al., 1998). Bacterial growth was used as a response
63 parameter. The potential for nitramine biodegradation was also assessed by determining
64 concentrations before and after the experiments.

65 **Table 1: Summary of the available chronic and acute ecotoxic response expressed as the half
66 effective concentration (EC_{50} , mg L^{-1}) or no observed concentration (NOEC, mg L^{-1}) of MEA-,**

67 **MMA-, and DMA-nitramine. n.a. denotes not available.**

	Test	MEA-nitramine	MMA-nitramine	DMA-nitramine
Acute	*Phytoplankton growth (Brakstad et al., 2011)	2535	754	>2000
	*Vertabrate growth (Brakstad et al., 2011)	1623	3314	2500

Chronic	Oyster larval development (Coutris et al., 2015)	107	n.a.	47
	Copepod mortality (Coutris et al., 2015)	NOEC \leq 100	n.a.	NOEC \leq 100
	Turbot mortality (Coutris et al., 2015)	NOEC \leq 100	n.a.	NOEC \leq 100
	Turbot growth (Coutris et al., 2015)	NOEC \leq 100	n.a.	NOEC \leq 100
	Algal growth (Coutris et al., 2015)	NOEC > 100	n.a.	591
	Bacterial community, aerob respiration (Gundersen et al., 2014)	4-8	n.a.	n.a.
	Macroalgae germling growth (Coutris et al., 2015)	NOEC = 100	n.a.	NOEC < 100
	Copepod reproduction (Coutris et al., 2015)	108	n.a.	70
	Turbot DNA damage (Coutris et al., 2015)	Massive, NOEC < 1	n.a.	157

68 *Test results are from publically available reports.

69

70 **2. Materials and Methods**

71 **2.1 Nitramine standard material**

72 Standard material of 3-nitro-oxazolidon-2-one (MEA-nitramine precursor), MMA-, and
73 DMA-nitramines at a purity of > 99 % was provided from the Norwegian University of
74 Life sciences (NMBU), Ås, Norway (Antonsen et al., 2016).

75 **2.2 Bacterial strains and sub-culturing procedure**

76 Pure cultures of the environmental bacteria, *P. fluorescens* and *R. spp.*, were provided from
77 the University of California, Irvine, CA, USA, and were previously isolated from grassland
78 leaf litter (Loma Ridge, CA, USA) as described by Mouginot et al. (2014).

79 Prior to every growth experiment, a two-step sub-culturing was performed. From pure
80 colonies grown on lysogeny broth (LB) agar plates a loopful of bacteria was inoculated
81 into 10/15 mL fresh liquid LB (pre-buffered capsules, Fischer Scientific, USA) in 50 mL
82 capped Erlenmeyer flasks and incubated at 28 °C and shaken at 150 rpm. When stationary
83 phase had been reached (determined by optical density), the cultures were diluted 1000-
84 fold using fresh liquid LB. The second time stationary phase was reached the 1000-fold
85 dilution was repeated, producing the ready culture to be used in the growth experiments.

86 **2.3 Bacterial growth inhibition test**

87 The bacterial growth inhibition test was conducted in accordance with the OECD
88 Guideline Test no. 201. The bacteria were grown under the same favourable conditions
89 described in section 2.2 while being exposed to 40, 60, 80 or 100 mg L⁻¹ of MEA-, MMA-,
90 or DMA-nitramines. Negative control with tetracycline (50 mg L⁻¹) and blank control
91 containing inoculum and liquid LB were included. Samples and controls were incubated

92 in triplicates, except for MEA-nitramine assays that employed duplicates. The experiment
93 was run until stationary growth phase had been reached ($t \leq 33$ h for *P. fluorescens* and $t \leq$
94 43 h for *R. spp.*). Growth was quantified by cell turbidity measurements of optical density
95 at $\lambda = 600$ nm (OD_{600nm}) using a spectrophotometer (BioTek Synergy H4, Winooski, VT,
96 USA). The repeatability of the sample replicates was $\leq 8\%$, except for *P. fluorescens*
97 exposed to 100 mg L⁻¹ MMA-nitramine at 12%.

98 At the start- and end of the experiment, sample aliquots (0.5 mL) were taken from
99 samples and controls and kept frozen (-18 °C) until analysis of nitramine determination.

100 **2.4 Determination of nitramines**

101 The nitramines were determined using liquid chromatography-mass spectrometry (LC-MS)
102 consisting of a Dionex Ultimate 3000 RS LC and a triple quadrupole TSQ Vantage TMMS
103 equipped with heated electrospray ionization (Thermo Scientific, USA). The method used
104 for MEA- and DMA-nitramine, with the exception of a few modifications, is described in
105 Gundersen et al. (2017b). Modifications were required to deal with the severe matrix
106 effects caused by the liquid LB and the bacterial lysate, and consisted of decreasing the
107 injection volume (from 20 to 0.5 μ L), increasing the analysis time (from 10 to 15 min),
108 and for DMA-nitramine decreasing the water content in the mobile phase (from 90% to
109 80%). The method used for MMA-nitramine was similar to the one used for MEA-
110 nitramine, but with the following specific settings: Monitored MS/MS transitions (*m/z*)
111 were 75.1 → 46.0/60.0 for quantification and qualification, respectively, optimized
112 selected reaction monitoring collision energy was 35 a.u. and the S-lens set to 20 a.u., the
113 water content in the mobile phase was 95%, and the injection volume was 2.0 μ L.

114 Prior to analysis, thawed samples were passed through 0.2 µm filters (regenerated
115 cellulose, Chromacol, Thermo scientific, USA) to remove bacterial cells, and diluted 20-
116 fold using Type II water (>1 MΩ cm at 25 °C). No loss of nitramine to the filter material
117 was detected.

118 Matrix-matched five-point external calibration was used, and provided good linearity ($r^2 =$
119 0.995). The repeatability of sample triplicate readings was satisfactory ($\pm 10\%$ for MEA-
120 nitramine, $\pm 18\%$ for MMA-nitramine, and $\pm 7\%$ for DMA-nitramine).

121 **2.5 Data assessment**

122 R language and environment for statistical computing and graphics (R Core Team, 2016)
123 was used to analyse and illustrate the bacterial growth with the packages *gprof* (Kahm et al.,
124 2010), *drfit* (Ranke, 2016), and *Hmisc* (Harrell Jr, 2016). The integral of the growth curve,
125 including the lag- and the exponential phases, was used for the dose-response calculations.

126 The per cent inhibition (%I_i) was calculated for each treatment concentration as follows:

$$127 \%I_i = \frac{(X_C - X_T)}{X_C} * 100 \quad (1)$$

128 Where

129 X_C is the mean value of the blank controls

130 X_T is the mean value of the treatment replicates

131 **3. Results**

132 **3.1 Bacterial growth response**

133 In Figures 1A-F the growth curves of *P. fluorescens* (left) and *R. spp* (right) during exposure
134 to MEA- (top), MMA- (middle), and DMA-nitramines (bottom) are presented along with
135 corresponding dose-response plots.

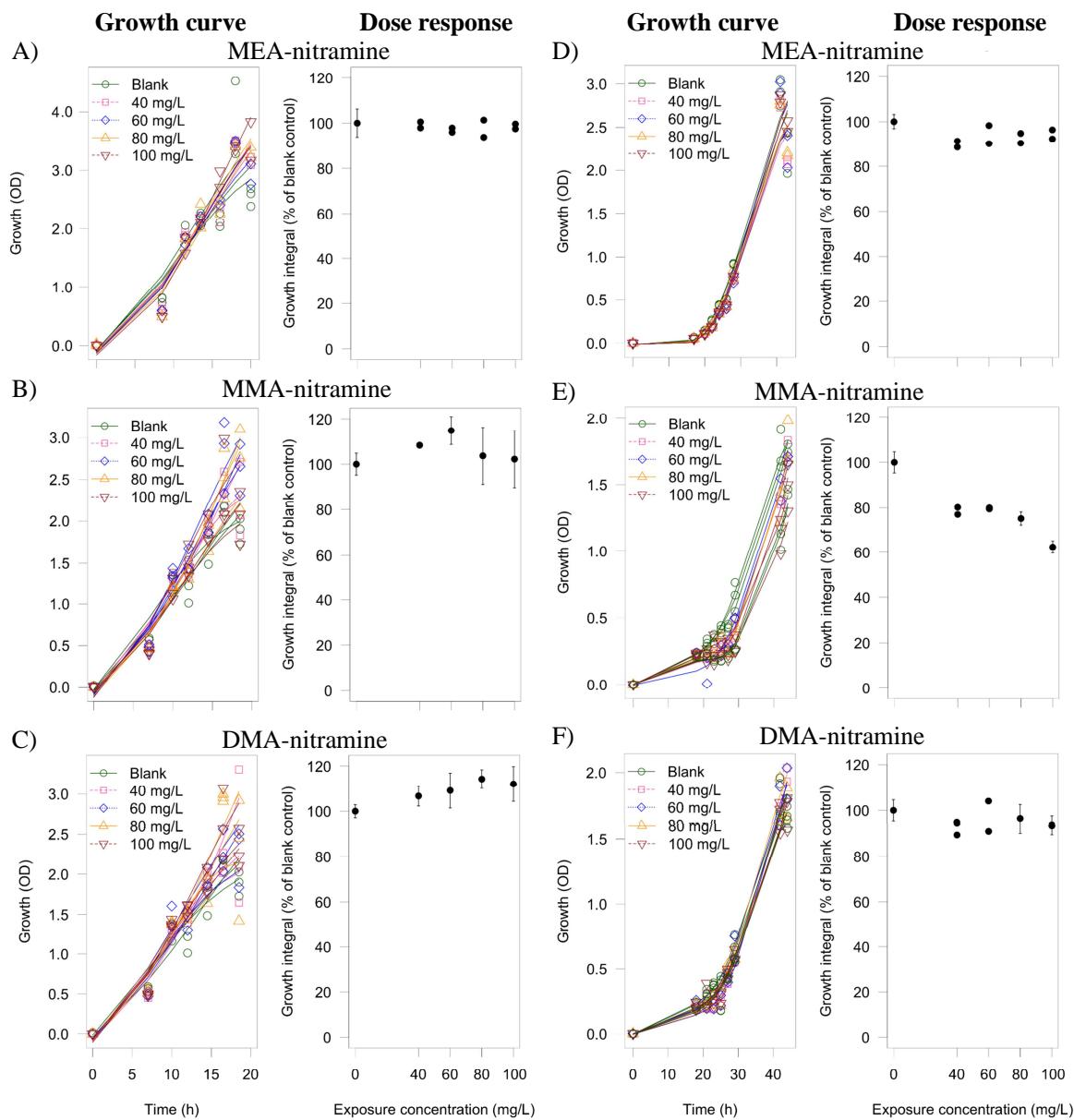
136 Remarkably, *R. spp* growth was significantly reduced by as much as 40% in the presence
137 of MMA-nitramine compared to the blank control ($p \leq 0.05$, ANOVA). Moreover, the
138 magnitude of the reduced growth was linearly correlated with the nitramine exposure level
139 (Figure 1 E: $r^2 = 0.79$, $p \leq 0.05$). The EC₅₀ was estimated at 157 mg L⁻¹. No such effect
140 was observed for *R. spp*. growth following exposure to MEA- or DMA-nitramines ($p >$
141 0.05). The ecotoxicity of MMA-nitramine has previously only been tested on
142 phytoplankton and vertebrate growth (Table 1). These studies found that phytoplankton
143 growth was more sensitive to MMA-nitramine than to the other two nitramines, with an
144 EC₅₀ of 754 mg L⁻¹ (Brakstad et al., 2011). Growth of *P. fluorescens* was unaffected by
145 exposure to MEA-, MMA-, and DMA-nitramine ($p \leq 0.05$, ANOVA).

146 The EC₅₀ value obtained for *R. spp*. growth by exposure to MMA-nitramine was two
147 orders of magnitude higher than the EC₅₀ presented for the natural lake-water bacterial
148 community exposed to MEA-nitramine (Gundersen et al., 2014). The reason for this
149 difference may in part be due to the different growth conditions across the two studies: *R.*
150 *spp* was grown here on rich medium, whereas the bacterial community was grown on low-
151 nutrient medium similar to natural lake water (Gundersen et al., 2014). In the literature,
152 several studies have found reduced toxic response from bacteria grown under optimal
153 conditions as compared to the same type of bacteria grown under conditions mimicking

154 their natural habitat, e.g. see Czechowska and van der Meer (2011). Extended periods of
155 exponential growth, such as observed here in this laboratory study, are not likely to occur
156 in natural habitats. Additional factors that likely contribute to the observed difference in
157 EC₅₀ include the different cell densities of the samples and the different strains of bacteria
158 used in the two studies. The cause of *R. spp.* insensitivity to MEA- and DMA-nitramine is
159 not known.

160 **3.2 Nitramine stability**

161 When exposed to *P. fluorescens*, a significant decay of all three nitramines was observed
162 (two-tailed t-test, $p \leq 0.05$). This was not the case during growth of *R. spp.* The average
163 decay caused by *P. fluorescens* was found to be $8 \pm 5\%$, $9 \pm 14\%$, and $10 \pm 4\%$ for MEA-,
164 MMA-, and DMA-nitramine, respectively, and to be independent of initial nitramine
165 concentration. (The relatively high uncertainty associated with MMA-nitramine is
166 attributed to the overall poorer analytical signal for this nitramine.) In another study,
167 Brakstad et al. (2018) found MEA-nitramine biodegradation of 27% over 28 days, and
168 extending the experiment to 56 days resulted in almost complete loss of the nitramine.
169 Considering the shorter duration of the experiments presented here (33 h for *P. fluorescens*
170 and 43 h for *R. spp.*), the rate of nitramine decay by *P. fluorescens* was one order of
171 magnitude higher than the decay obtained by Brakstad et al. (2018) over 28 days. With the
172 assumption of continued exponential growth of *P. fluorescens*, 60% decay of the nitramines
173 could be accomplished within 8 days. However, testing this assumption would require
174 another experimental setup that allows for continued growth without reaching the
175 stationary phase.



178 **Figure 1: Growth curves and dose-response curves for *P. fluorescens* (left side column: A, B, and**
 179 ***C*) and *R. spp.* (right side column: D, E, and F) exposed to the three nitramines (from top: MEA-,**
 180 **MMA-, and DMA-nitramine). Growth curves show sample replicate OD_{600nm} readings by time (h).**
 181 **Dose-response curves were calculated from the integral of the growth curves and average values**
 182 **are provided with one standard deviation error bars (n = 2 for MEA-nitramine and n=3 for MMA-**
 183 **and DMA-nitramines).**

184 **4. Concluding remarks**

185 Environmental exposure to MEA-, MMA-, and DMA-nitramines is expected from use of
186 the “benchmark” CO₂ capture amine solvent. These nitramines have been tested for
187 ecotoxic effects on environmentally relevant *P. fluorescens* and *R. spp.* strains. Growth of *R.*
188 *spp.* was inhibited by MMA-nitramine with an estimated EC₅₀ of 157 mg L⁻¹. No such
189 effect was observed for *R. spp.* exposed to MEA- or DMA-nitramine. *P. fluorescens* was
190 insensitive to all three nitramines. On the other hand, *P. fluorescens* was capable of
191 degrading all the nitramines at rates of 8 - 10% during the 33 h experiment. Considering
192 the large number of CO₂ capture plants needed to significantly reduce anthropogenic CO₂
193 emissions, future studies should explore responses of additional bacterial strains and
194 communities to the potentially carcinogenic nitramines, both with regards to ecotoxicity-
195 and biodegradation potential.

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202

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