

1 Growth response of environmental bacteria under 2 exposure to nitramines from CO₂-capture

3 Cathrine Brecke Gundersen^{a*}, Tom Andersen^b, Rolf D. Vogt^a, Steven D. Allison^{c,d}

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5 ^aDepartment of Chemistry, University of Oslo, P.O. Box 1033, Blindern, NO-0315 Oslo,
6 Norway

7 ^bDepartment of Biosciences, University of Oslo, P.O. Box 1066, Blindern, NO-0316 Oslo,
8 Norway

9 ^cDepartment of Ecology and Evolutionary Biology, University of California, Irvine, California
10 92697, USA

11 ^dDepartment of Earth System Science, University of California, Irvine, California 92697, USA

12 [†]Currently at Norwegian Institute for Water Research, Gaustadaléen 21, 0349 Oslo, Norway

13 *Corresponding author, phone: (0047)41680804; e-mail: cbg@niva.no

14 **Abstract**

15 Nitramines are potentially carcinogenic by-products of amines used in post-combustion
16 CO₂-capture. The influence of monoethanol (MEA)-, monomethyl (MMA)-, and
17 dimethyl (DMA)-nitramines on the growth of environmental strains of bacteria,
18 *Pseudomonas fluorescens* (*P. fluorescens*) and *Rhodococcus spp.* (*R. spp.*), was investigated in the
19 laboratory. Additionally, the persistence of the nitramines in the presence of bacteria was
20 determined. Growth of *R. spp.* was found to be sensitive to MMA-nitramine (EC₅₀ = 157
21 mg L⁻¹), while *P. fluorescens* growth was insensitive to all nitramines tested. Moreover, *P.*
22 *fluorescens* was capable of degrading 8-10% of the nitramines during the 33 h experiments.
23 Results from this study provide insight into important processes of bacterial response to
24 nitramines that merit further investigation considering the ongoing implementation of
25 CO₂ capture technology.

26 **Keywords:** Biodegradation; CO₂ capture; Ecotoxicity; LC-MS; Nitramine

1. Introduction

Technology of CO₂ capture offers the opportunity to reduce greenhouse gas emissions from existing large-scale point sources. Most climate models rely on global-scale implementation of the technology to limit global warming to 2 °C (and especially 1.5 °C) (IPCC, 2014). Currently, the most feasible way of capturing CO₂ is using amines post-combustion (Rochelle, 2009; Wang et al., 2011). However, several potentially carcinogenic nitramines may form from the amines used in post-combustion CO₂ capture. Formation occurs in such a way that direct introduction to the nearby environment is inevitable (Nielsen et al., 2012). A thorough risk assessment, constituting both a prediction of final exposure levels and a detailed investigation of the toxicity of the relevant nitramines, is lacking (Chen et al., 2018).

From the “benchmark” amine solvent, monoethanolamine, the following three nitramines can form: MEA-, MMA-, and DMA-nitramines. As these nitramines are small and polar ($M \approx 106.8 \text{ g mol}^{-1}$ and $S_w \approx 176 \text{ g L}^{-1}$) they are thought to partition readily into the aqueous phase. However, a preceding study found that MEA- and DMA-nitramines preferentially bind to soils rich in organic matter (Gundersen et al., 2017a). Surface soil horizons are typically rich in organic matter and high in biological activity, especially in boreal ecosystems. Brakstad et al. (2018) investigated the biodegradability of MEA-, MMA-, and DMA-nitramines. They found none of them to be readily biodegradable by the standards of the OECD Guideline 301, which requires 60% decay. The apparent low biodegradability may be caused by toxic effects from the nitramines.

Studies on the potential ecotoxic effect of relevant nitramines are summarized in Table 1.

The most sensitive response was found in our preceding study, where a natural

oligotrophic lake-water bacterial community showed an estimated half effective concentration (EC₅₀) of 10 mg L⁻¹ MEA-nitramine (Gundersen et al., 2014). Other studies focusing on species from higher trophic levels (e.g. phytoplankton or larvae) report higher EC₅₀ values, ranging from 47 to > 2000 mg L⁻¹ (Table 1). For algae, a growth assay showed an EC₅₀ of 591 mg L⁻¹ for DMA-nitramine (Coutris et al., 2015).

The aim of this study was to explore the bacterial response to MEA-, MMA-, and DMA-nitramines exposure in pure cultures of environmental strains of *P. fluorescens* and *R. spp.* The two bacteria were selected for their high environmental relevance. They are both abundant in soils and water, and they represent the two major groups of bacteria based on cell wall structure (Gram staining). Moreover, other studies have found strains of *P. fluorescens* and *R. spp.* capable of degrading a range of different types of contaminants (Agarry & Solomon, 2008; Martinkova et al., 2009), including the cyclic nitramine explosive, known as RDX (Coleman et al., 1998). Bacterial growth was used as a response parameter. The potential for nitramine biodegradation was also assessed by determining concentrations before and after the experiments.

Table 1: Summary of the available chronic and acute ecotoxic response expressed as the half effective concentration (EC₅₀, mg L⁻¹) or no observed concentration (NOEC, mg L⁻¹) of MEA-, MMA-, and DMA-nitramine. n.a. denotes not available.

	Test	MEA-nitramine	MMA-nitramine	DMA-nitramine
Acute	*Phytoplankton growth (Brakstad et al., 2011)	2535	754	>2000
	*Vertabrate growth (Brakstad et al., 2011)	1623	3314	2500

	Oyster larval development (Coutris et al., 2015)	107	n.a.	47
	Copepod mortality (Coutris et al., 2015)	NOEC \approx 100	n.a.	NOEC \approx 100
	Turbot mortality (Coutris et al., 2015)	NOEC \approx 100	n.a.	NOEC \approx 100
	Turbot growth (Coutris et al., 2015)	NOEC \approx 100	n.a.	NOEC \approx 100
	Algal growth (Coutris et al., 2015)	NOEC > 100	n.a.	591
	Bacterial community, aerob respiration (Gundersen et al., 2014)	4-8	n.a.	n.a.
Chronic	Macroalgae germling growth (Coutris et al., 2015)	NOEC = 100	n.a.	NOEC < 100
	Copepod reproduction (Coutris et al., 2015)	108	n.a.	70
	Turbot DNA damage (Coutris et al., 2015)	Massive, NOEC < 1	n.a.	157

68 *Test results are from publically available reports.

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2. Materials and Methods

2.1 Nitramine standard material

Standard material of 3-nitro-oxazolidon-2-one (MEA-nitramine precursor), MMA-, and DMA-nitramines at a purity of > 99 % was provided from the Norwegian University of Life sciences (NMBU), Ås, Norway (Antonsen et al., 2016).

2.2 Bacterial strains and sub-culturing procedure

Pure cultures of the environmental bacteria, *P. fluorescens* and *R. spp.*, were provided from the University of California, Irvine, CA, USA, and were previously isolated from grassland leaf litter (Loma Ridge, CA, USA) as described by Mouginot et al. (2014).

Prior to every growth experiment, a two-step sub-culturing was performed. From pure colonies grown on lysogeny broth (LB) agar plates a loopful of bacteria was inoculated into 10/15 mL fresh liquid LB (pre-buffered capsules, Fischer Scientific, USA) in 50 mL capped Erlenmeyer flasks and incubated at 28 °C and shaken at 150 rpm. When stationary phase had been reached (determined by optical density), the cultures were diluted 1000-fold using fresh liquid LB. The second time stationary phase was reached the 1000-fold dilution was repeated, producing the ready culture to be used in the growth experiments.

2.3 Bacterial growth inhibition test

The bacterial growth inhibition test was conducted in accordance with the OECD Guideline Test no. 201. The bacteria were grown under the same favourable conditions described in section 2.2 while being exposed to 40, 60, 80 or 100 mg L⁻¹ of MEA-, MMA-, or DMA-nitramines. Negative control with tetracycline (50 mg L⁻¹) and blank control containing inoculum and liquid LB were included. Samples and controls were incubated

in triplicates, except for MEA-nitramine assays that employed duplicates. The experiment was run until stationary growth phase had been reached ($t \approx 33$ h for *P. fluorescens* and $t \approx 43$ h for *R. spp.*). Growth was quantified by cell turbidity measurements of optical density at $\lambda = 600$ nm (OD_{600nm}) using a spectrophotometer (BioTek Synergy H4, Winooski, VT, USA). The repeatability of the sample replicates was $\approx 8\%$, except for *P. fluorescens* exposed to 100 mg L^{-1} MMA-nitramine at 12% .

At the start- and end of the experiment, sample aliquots (0.5 mL) were taken from samples and controls and kept frozen (-18°C) until analysis of nitramine determination.

2.4 Determination of nitramines

The nitramines were determined using liquid chromatography-mass spectrometry (LC-MS) consisting of a Dionex Ultimate 3000 RS LC and a triple quadrupole TSQ Vantage TMMS equipped with heated electrospray ionization (Thermo Scientific, USA). The method used for MEA- and DMA-nitramine, with the exception of a few modifications, is described in Gundersen et al. (2017b). Modifications were required to deal with the severe matrix effects caused by the liquid LB and the bacterial lysate, and consisted of decreasing the injection volume (from 20 to $0.5 \mu\text{L}$), increasing the analysis time (from 10 to 15 min), and for DMA-nitramine decreasing the water content in the mobile phase (from 90% to 80%). The method used for MMA-nitramine was similar to the one used for MEA-nitramine, but with the following specific settings: Monitored MS/MS transitions (m/z) were $75.1 \rightarrow 46.0/60.0$ for quantification and qualification, respectively, optimized selected reaction monitoring collision energy was 35 a.u. and the S-lens set to 20 a.u, the water content in the mobile phase was 95%, and the injection volume was $2.0 \mu\text{L}$.

Prior to analysis, thawed samples were passed through 0.2 µm filters (regenerated cellulose, Chromacol, Thermo scientific, USA) to remove bacterial cells, and diluted 20-fold using Type II water (>1 MΩ cm at 25 °C). No loss of nitramine to the filter material was detected.

Matrix-matched five-point external calibration was used, and provided good linearity ($r^2 \geq 0.995$). The repeatability of sample triplicate readings was satisfactory ($\leq 10\%$ for MEA-nitramine, $\leq 18\%$ for MMA-nitramine, and $\leq 7\%$ for DMA-nitramine).

2.5 Data assessment

R language and environment for statistical computing and graphics (R Core Team, 2016) was used to analyse and illustrate the bacterial growth with the packages *grofit* (Kahm et al., 2010), *drfit* (Ranke, 2016), and *Hmisc* (Harrell Jr, 2016). The integral of the growth curve, including the lag- and the exponential phases, was used for the dose-response calculations.

The per cent inhibition (%I_i) was calculated for each treatment concentration as follows:

$$\%I_i = \frac{(X_C - X_T)}{X_C} * 100 \quad (1)$$

Where

X_C is the mean value of the blank controls

X_T is the mean value of the treatment replicates

3. Results

3.1 Bacterial growth response

In Figures 1A-F the growth curves of *P. fluorescens* (left) and *R. spp* (right) during exposure to MEA- (top), MMA- (middle), and DMA-nitramines (bottom) are presented along with corresponding dose-response plots.

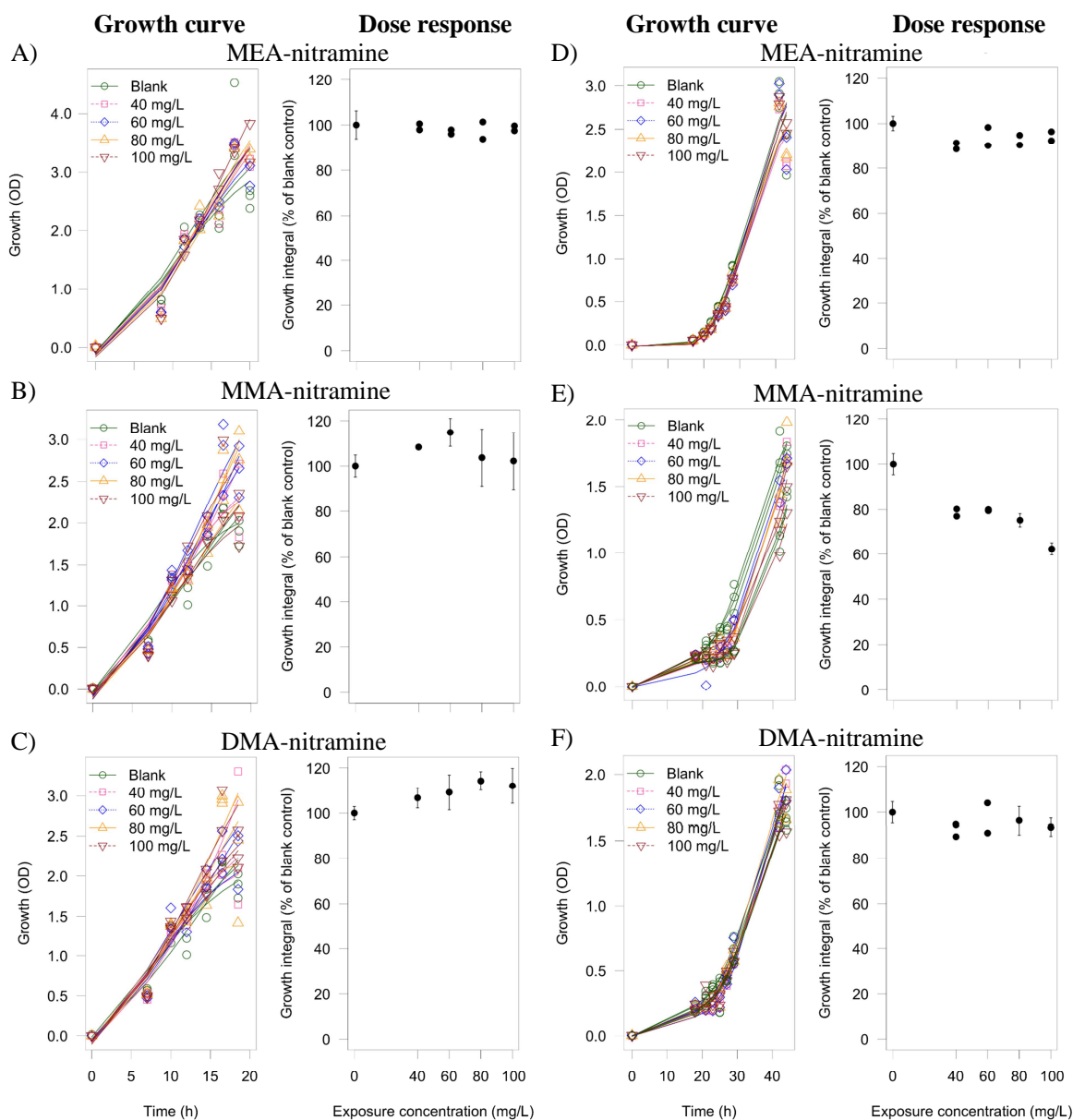
Remarkably, *R. spp* growth was significantly reduced by as much as 40% in the presence of MMA-nitramine compared to the blank control ($p \leq 0.05$, ANOVA). Moreover, the magnitude of the reduced growth was linearly correlated with the nitramine exposure level (Figure 1 E: $r^2 = 0.79$, $p \leq 0.05$). The EC_{50} was estimated at 157 mg L^{-1} . No such effect was observed for *R. spp* growth following exposure to MEA- or DMA-nitramines ($p > 0.05$). The ecotoxicity of MMA-nitramine has previously only been tested on phytoplankton and vertebrate growth (Table 1). These studies found that phytoplankton growth was more sensitive to MMA-nitramine than to the other two nitramines, with an EC_{50} of 754 mg L^{-1} (Brakstad et al., 2011). Growth of *P. fluorescens* was unaffected by exposure to MEA-, MMA-, and DMA-nitramine ($p \leq 0.05$, ANOVA).

The EC_{50} value obtained for *R. spp* growth by exposure to MMA-nitramine was two orders of magnitude higher than the EC_{50} presented for the natural lake-water bacterial community exposed to MEA-nitramine (Gundersen et al., 2014). The reason for this difference may in part be due to the different growth conditions across the two studies: *R. spp* was grown here on rich medium, whereas the bacterial community was grown on low-nutrient medium similar to natural lake water (Gundersen et al., 2014). In the literature, several studies have found reduced toxic response from bacteria grown under optimal conditions as compared to the same type of bacteria grown under conditions mimicking

their natural habitat, e.g. see Czechowska and van der Meer (2011). Extended periods of exponential growth, such as observed here in this laboratory study, are not likely to occur in natural habitats. Additional factors that likely contribute to the observed difference in EC_{50} include the different cell densities of the samples and the different strains of bacteria used in the two studies. The cause of *R. spp.* insensitivity to MEA- and DMA-nitramine is not known.

3.2 Nitramine stability

When exposed to *P. fluorescens*, a significant decay of all three nitramines was observed (two-tailed t-test, $p \leq 0.05$). This was not the case during growth of *R. spp.* The average decay caused by *P. fluorescens* was found to be $8 \pm 5\%$, $9 \pm 14\%$, and $10 \pm 4\%$ for MEA-, MMA-, and DMA-nitramine, respectively, and to be independent of initial nitramine concentration. (The relatively high uncertainty associated with MMA-nitramine is attributed to the overall poorer analytical signal for this nitramine.) In another study, Brakstad et al. (2018) found MEA-nitramine biodegradation of 27% over 28 days, and extending the experiment to 56 days resulted in almost complete loss of the nitramine. Considering the shorter duration of the experiments presented here (33 h for *P. fluorescens* and 43 h for *R. spp.*), the rate of nitramine decay by *P. fluorescens* was one order of magnitude higher than the decay obtained by Brakstad et al. (2018) over 28 days. With the assumption of continued exponential growth of *P. fluorescens*, 60% decay of the nitramines could be accomplished within 8 days. However, testing this assumption would require another experimental setup that allows for continued growth without reaching the stationary phase.



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178 Figure 1: Growth curves and dose-response curves for *P. fluorescens* (left side column: A, B, and
 179 C) and *R. spp.* (right side column: D, E, and F) exposed to the three nitramines (from top: MEA-,
 180 MMA-, and DMA-nitramine). Growth curves show sample replicate OD_{600nm} readings by time (h).
 181 Dose-response curves were calculated from the integral of the growth curves and average values
 182 are provided with one standard deviation error bars (n = 2 for MEA-nitramine and n=3 for MMA-
 183 and DMA-nitramines).

4. Concluding remarks

Environmental exposure to MEA-, MMA-, and DMA-nitramines is expected from use of the “benchmark” CO₂ capture amine solvent. These nitramines have been tested for ecotoxic effects on environmentally relevant *P. fluorescens* and *R. spp.* strains. Growth of *R. spp.* was inhibited by MMA-nitramine with an estimated EC₅₀ of 157 mg L⁻¹. No such effect was observed for *R. spp.* exposed to MEA- or DMA-nitramine. *P. fluorescens* was insensitive to all three nitramines. On the other hand, *P. fluorescens* was capable of degrading all the nitramines at rates of 8 - 10% during the 33 h experiment. Considering the large number of CO₂ capture plants needed to significantly reduce anthropogenic CO₂ emissions, future studies should explore responses of additional bacterial strains and communities to the potentially carcinogenic nitramines, both with regards to ecotoxicity- and biodegradation potential.

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Declaration of interest: None

203 References

- 204 Agarry, S. E. & Solomon, B. O. (2008). Kinetics of batch microbial degradation of phenols by
 205 indigenous *Pseudomonas* fluorescence. *International Journal of Environmental Science &*
 206 *Technology*, 5 (2): 223-232. doi: 10.1007/bf03326016.
- 207 Antonsen, S., Aursnes, M., Gallantree-Smith, H., Dye, C. & Stenstrøm, Y. (2016). Safe synthesis of
 208 alkylhydroxy and alkylamino nitramines. *Molecules*, 21 (12): 1738.
- 209 Brakstad, O. G., Hansen, B. H., Bonaunet, K. & Hansen, S.-H. (2011). *Ecotoxicity*. In Dye, C., Fjellsbø, L.
 210 M. B. & Dusinska, M. (eds). Nitramine analysis procedures development and screening
 211 toxicity study. Kjeller, Norway: Norwegian institute for air research (NILU).
- 212 Brakstad, O. G., Sørensen, L., Zahlsen, K., Bonaunet, K., Hyldbakk, A. & Booth, A. M. (2018).
 213 Biotransformation in water and soil of nitrosamines and nitramines potentially generated
 214 from amine-based CO₂ capture technology. *International Journal of Greenhouse Gas Control*,
 215 70: 157-163. doi: <https://doi.org/10.1016/j.ijggc.2018.01.021>.
- 216 Chen, X., Huang, G., An, C., Yao, Y. & Zhao, S. (2018). Emerging N-nitrosamines and N-nitramines
 217 from amine-based post-combustion CO₂ capture – A review. *Chemical Engineering Journal*,
 218 335: 921-935. doi: 10.1016/j.cej.2017.11.032.
- 219 Coleman, N. V., Nelson, D. R. & Duxbury, T. (1998). Aerobic biodegradation of hexahydro-1,3,5-
 220 trinitro-1,3,5-triazine (RDX) as a nitrogen source by a *Rhodococcus* sp., strain DN22. *Soil*
 221 *Biology and Biochemistry*, 30 (8): 1159-1167. doi: [https://doi.org/10.1016/S0038-](https://doi.org/10.1016/S0038-0717(97)00172-7)
 222 [0717\(97\)00172-7](https://doi.org/10.1016/S0038-0717(97)00172-7).
- 223 Coutris, C., Macken, A. L., Collins, A. R., El Yamani, N. & Brooks, S. J. (2015). Marine ecotoxicity of
 224 nitramines, transformation products of amine-based carbon capture technology. *Science of*
 225 *the total environment*, 527-528: 211-219. doi: 10.1016/j.scitotenv.2015.04.119.
- 226 Czechowska, K. & van der Meer, J. R. (2011). A flow cytometry based oligotrophic pollutant exposure
 227 test to detect bacterial growth inhibition and cell injury. *Environmental Science & Technology*,
 228 45 (13): 5820-5827. doi: 10.1021/es200591v.
- 229 Gundersen, C. B., Andersen, T., Lindahl, S., Linke, D. & Vogt, R. D. (2014). Bacterial response from
 230 exposure to selected aliphatic nitramines. *Energy Procedia*, 63: 791-800. doi:
 231 10.1016/j.egypro.2014.11.089.
- 232 Gundersen, C. B., Breedveld, G. D., Foseid, L. & Vogt, R. D. (2017a). Soil sorption of two nitramines
 233 derived from amine-based CO₂ capture. *Environmental Science: Processes & Impacts*, 19 (6):
 234 812-821. doi: 10.1039/c7em00131b.
- 235 Gundersen, C. B., Zhu, L., Lindahl, S., Wang, S., Wilson, S. R. & Lundanes, E. (2017b). LC-MS/MS
 236 Method for Simultaneous Determination of Monoethanol- and Dimethylnitramine in
 237 Aqueous Soil Extracts. *Chromatographia*, 80 (9): 1475-1481. doi: 10.1007/s10337-017-3355-6.
- 238 Harrell Jr, F. E. (2016). *Hmisc: Harrell Miscellaneous* (Version with contributions from Charles Dupont
 239 and many others). Available at: <https://CRAN.R-project.org/package=Hmisc>.
- 240 IPCC. (2014). *Climate Change 2014: Mitigation of Climate Change. Contribution from Working Group*
 241 *III to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*. In
 242 Edenhofer, O., R. Pichs-Madruga, Y. Sokona, E. Farahani, S. Kadner, K- Seyboth, A. Adler, I.
 243 Baum, S. Brunner, P. Eickmeier, B. Kriemann, J. Savolainen, S. Schlömer, C. von Stechow, T.
 244 Zwickel and J. C. Minx (eds.). (ed.). Cambridge, United Kingdom and New York, NY, USA.
- 245 Kahm, M., Hasenbrink, G., Lichtenberg-Fraté, H., Ludwig, J. & Kschischo, M. (2010). grofit: Fitting
 246 Biological Growth Curves with R. 2010, 33 (7): 21. doi: 10.18637/jss.v033.i07.
- 247 Martinkova, L., Uhnakova, B., Patek, M., Nesvera, J. & Kren, V. (2009). Biodegradation potential of
 248 the genus *Rhodococcus*. *Environment International*, 35 (1): 162-77. doi:
 249 10.1016/j.envint.2008.07.018.
- 250 Mougnot, C., Kawamura, R., Matulich, K. L., Berlemont, R., Allison, S. D., Amend, A. S. & Martiny, A. C.
 251 (2014). Elemental stoichiometry of Fungi and Bacteria strains from grassland leaf litter. *Soil*
 252 *Biology and Biochemistry*, 76: 278-285. doi: 10.1016/j.soilbio.2014.05.011.

253 Nielsen, C. J., Herrmann, H. & Weller, C. (2012). Atmospheric chemistry and environmental impact of
254 the use of amines in carbon capture and storage (CCS). *Chemical Society Reviews*, 41 (19):
255 6684-6704. doi: 10.1039/C2CS35059A.

256 R Core Team. (2016). *R: A language and environment for statistical computing*. Vienna, Austria: R
257 Foundation for Statistical Computing.

258 Ranke, J. (2016). *drfit: Dose-Response Data Evaluation*. *R package version 0.6.7*. Available at:
259 <https://CRAN.R-project.org/package=drfit>.

260 Rochelle, G. T. (2009). Amine scrubbing for CO₂ capture. *Science*, 325 (5948): 1652-1654.

261 Wang, M., Lawal, A., Stephenson, P., Sidders, J. & Ramshaw, C. (2011). Post-combustion CO₂ capture
262 with chemical absorption: A state-of-the-art review. *Chemical Engineering Research and*
263 *Design*, 89 (9): 1609-1624. doi: doi.org/10.1016/j.cherd.2010.11.005.

264