

# Experimental demonstration of a 5kW<sub>th</sub> granular-flow reactor for solar thermochemical energy storage with aluminum-doped calcium manganite particles

Andrew J. Schrader,<sup>1</sup> Garrett L. Schieber,<sup>1</sup> Andrea Ambrosini<sup>2</sup>, and Peter G. Loutzenhiser<sup>1\*</sup>

<sup>1</sup>George W. Woodruff School of Mechanical Engineering, Georgia Institute of Technology, Atlanta, Georgia USA, 30332-0405

<sup>2</sup> Concentrating Solar Technologies, Sandia National Laboratories, Albuquerque, NM 87185

\*Corresponding Author: Peter G. Loutzenhiser, Email: peter.loutzenhiser@me.gatech.edu, Phone: +1. 404.894.3012

## Abstract

A two-step cycle was considered for solar thermochemical energy storage based on aluminum-doped calcium manganite reduction/oxidation reactions for direct integration into Air Brayton cycles. The two steps encompassed 1) the storage of concentrated solar direct irradiation via the thermal reduction of aluminum-doped calcium manganite and 2) the delivery of heat to an Air-Brayton cycle via re-oxidation of oxygen-deficient aluminum-doped calcium manganite. The re-oxidized aluminum-doped calcium manganite was fed back to the first step to complete the cycle. A 5 kW<sub>th</sub> solar thermochemical reactor operating under vacuum was fabricated and tested to examine the first cycle reduction step. Reactor operating conditions and high-flux solar simulator control were tuned for continuous reactor operation with particle temperatures > 1073 K. Continuous operation was achieved using intermittent, dense granular flows. A maximum absorption efficiency of 64.7% was demonstrated, accounting for both sensible and chemical heat storage.

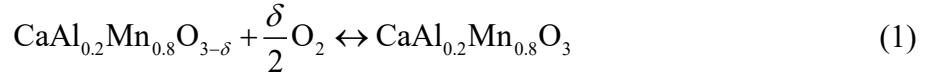
**Keywords:** *solar thermochemical reactor, dense granular flow, perovskite oxides, agglomeration, calcium manganites, thermochemical energy storage*

## 1. Introduction

Thermochemical energy storage (TCES) has been considered in two-step solar thermochemical cycles based on the reduction/oxidation (redox) of metal oxide particles for direct integration into an Air Brayton cycle to produce electricity [1-14]. The direct irradiation of particles is ideal for achieving high temperature operation ( $> 1073$  K) in solar thermochemical reactors [15]. The particles function simultaneously as both the heat transfer and TCES media, removing irreversible heat loss between different media [16]. This process requires novel solar receivers/reactors designed for the bulk transport behavior of granular media to direct irradiation of the TCES media. Granular media were observed to exhibit traits of solids (densely-packed particle structures), gases (dilute particle collisions), or liquids (dense particle flows) during bulk transport based on the relative impact of frictional and collisional effects [17-19]. Solid granular structures were achieved within solar receiver and solar thermochemical reactor designs with rotary kilns [20-22] and packed beds [23-26]. Dilute granular transport was achieved within solar receivers and solar thermochemical reactors using falling particle curtains [27-32], fluidized beds [33-35], and low-density flows along steep inclined planes [36-38]. Dense, granular flows have been achieved experimentally within solar receivers/reactors using slip-stick planes [39] gravity-fed tubes [40] and along inclined spirals [41, 42]. A 5 kW<sub>th</sub> Solar Thermochemical Inclined Granular-Flow Reactor (STInGR) was designed, fabricated, modeled, and tested to characterize the reduction reaction of the directly-irradiated oxide particles using dense granular flows. TCES was modeled using directly-irradiated, thin ( $< 1$  mm) dense, granular flows of redox-active metal oxide particles [16, 41] via a combination of mass flow rate, slope inclination angles, and radiative conditions. Dense, granular flows are governed by the complex frictional and collisional interactions of particles, which are impacted by the presence of particle agglomeration at high temperatures.

A suite of experiments was performed in STInGR using spray-dried CaAl<sub>0.2</sub>Mn<sub>0.8</sub>O<sub>3</sub> (CAM28) granular powders to achieve continuous operation with bulk particle temperatures  $> 1073$  K. CAM28 crystallizes in the cubic perovskite structure and demonstrates continuous reduction over a wide range of temperature and O<sub>2</sub> partial pressure [7]. This makes it a promising candidate for high-temperature thermal storage, for example, coupled to an Air Brayton cycle. The average diameter of the CAM28 particles was  $\bar{d}_p = 45$   $\mu$ m, determined from digital microscopy (Hirox

KH-3000 with ImagePro Plus image analysis software). The CAM28 was developed and optimized for two-step solar TCES cycles operating at  $\sim$ 1100 °C [7], represented as:



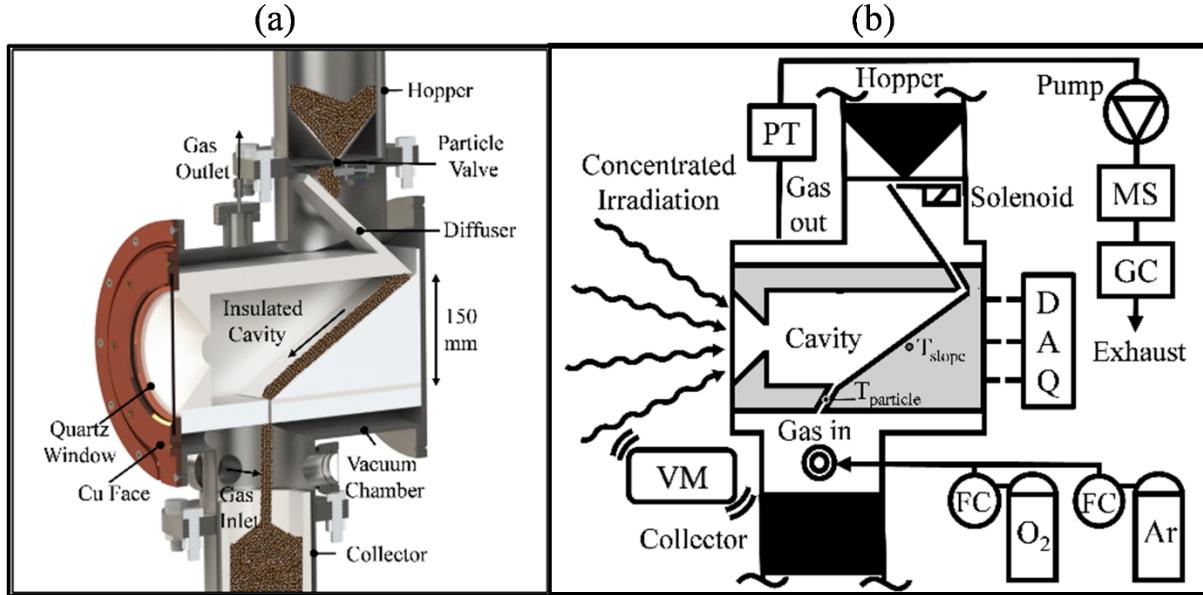
where  $\delta$  is the oxygen vacancy concentration (deviation from stoichiometry). Over the course of the experiments, particle agglomeration of the CAM28 occurred. Particle agglomeration has also been observed in chemical looping combustion reactors using spray-dried calcium manganite particles [43] and solar reactors and receivers with various energy storage materials [20, 36, 40, 44-48]. During agglomeration, individual particles retain their original shape, size, and properties but are held together by binding mechanisms, which include the formation of solid bridges from local sintering or chemical reactions between individual particles [49, 50]. The likelihood of agglomeration is amplified by the presence of fine particles [49]. The agglomeration results in deviations in granular flow behavior that must be addressed in the reactor design and/or operating conditions.

CAM28 agglomerate formation was investigated, and x-ray diffractometry (XRD) was employed to identify phase impurities contributing to agglomeration. The reactor design and operating conditions were optimized to overcome the agglomerate formation for continuous operation.

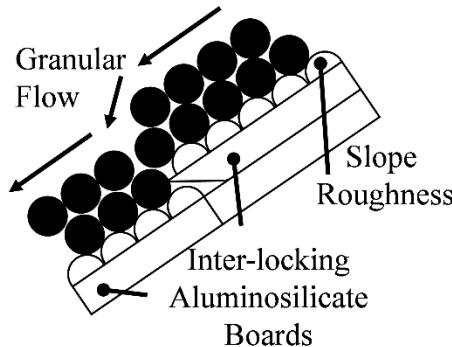
## 2. Experimental Setup

The STInGR system is schematically depicted in Figure 1 for (a) a Solidworks rendering of the reactor design and (b) a process flow diagram of the reactor with concentrated irradiation delivered by a high-flux solar simulator (HFSS) [51]. The cavity was fabricated using high-strength aluminosilicate boards (Zircar Zirconia M35 buster-type insulation). The boards were machined to form a 50 mm conical aperture with 45° acceptance angle, rectangular cavity ( $150 \times 150 \times 100 \text{ mm}^3$ ), a directly-irradiated slope, and a granular flow channel. Directly-irradiated cavity surfaces were coated with a reflective, high-purity alumina coating (Aremco PyroPaint AL). An 80 mm wide granular flow channel was produced from multiple, interlocking aluminosilicate boards to accommodate thermal expansion during high temperature operation. The boards were coated with high-purity alumina to develop a uniform surface roughness. The surface roughness was tuned to promote a dense granular flow at room temperature [16] and to reduce frictional wear

of the aluminosilicate board [52]. The assembled granular flow channel resulted in a directly-irradiated stair-step granular flow path along the inclined plane to relieve thermal stresses during high temperature operation, shown in Figure 2.



**Figure 1.** Design of the 5 kW<sub>th</sub> solar thermochemical inclined granular-flow reactor with (a) a Solidworks rendering of the final design and (b) a process flow diagram of the reactor within the high-flux solar simulator



**Figure 2.** Schematic of granular flow channel composed of interlocking aluminosilicate boards to accommodate thermal expansion and producing a stair-step granular flow behavior

The reactor cavity was positioned in a stainless steel case ( $190 \times 215 \times 405$  mm $^3$ ) within the vacuum chamber to ensure stability during high temperature operation. The vacuum chamber was fabricated from a stainless steel tube (NPS 10 Schedule 5) and fitted with steel flanges and

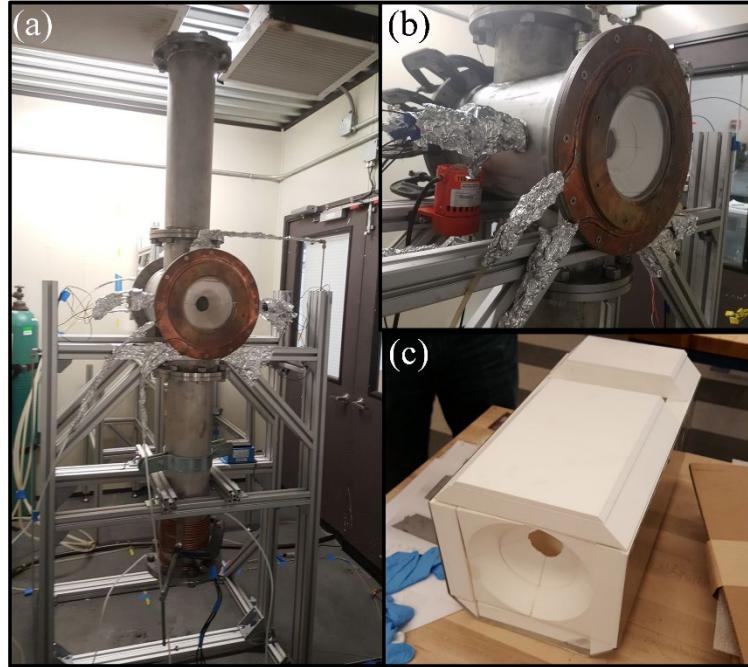
feedthroughs for instrumentation during experimentation. The vacuum chamber was sealed with custom mica gaskets using a steel back plate, vertical cylindrical stainless steel hopper and collector assemblies (NPS 6 Schedule 5), and a water-cooled Cu face. A 5 mm thick, 140 mm diameter quartz window was housed within the water-cooled Cu face assembly and sealed using custom mica gaskets.

Concentrated irradiation from the HFSS was introduced through the quartz window, and reduced operating pressures were maintained with an oil-free rotary vane pump (Edwards nXDS10i oil-free scroll pump). The pressure was monitored with a transducer (PT, Omega PXM209) and convection gauge (Instrutech Series 101) during the experiments, ranging from 80 to 110 mbar. An Ar flow was introduced during experiments via mass flow controllers (FC, Bronkhorst F-201CV-20K-MAD-33-V) to maintain low O<sub>2</sub> partial pressures and high  $\delta$ . Gas flows were introduced and removed via gas vacuum feedthroughs (Kurt J. Lesker F0275X4SWG). Gas compositions were temporally monitored using coupled gas chromatography (GC, Agilent 490 Micro GC with Molsieve and PoraPLOT Q columns) and mass spectrometry (MS, OmniStar ThermoStar GSD320 Gas Analysis System). Particles were loaded into an upper hopper assembly [53, 54] prior to experimentation. The mass flow rate was controlled with an orifice plate and solenoid valve [55, 56]. Particles were introduced into the granular flow channel using a diffuser plate, designed to reduce the particle kinetic energy and promote uniform flow depth and width. A dense, granular flow was achieved within a custom flow channel along the inclined plane and was directly irradiated by concentrated irradiation. A vibratory motor (VM, Vibco SPR-21) was attached to the structural frame to prevent agglomerate formation along the directly-irradiated flow channel during experimentation.

Sheathed, N-type thermocouples (Omega SNNXL-020U-8.5-SHX) were placed at two locations within the reactor cavity: 1) ~ 6–7 mm below the aluminosilicate board to measure cavity/granular flow channel temperature and 2) across the granular flow outlet to measure the average granular flow temperature. An array of N-type thermocouples was positioned at the flow outlet. The array was designed to be fully immersed in the exiting granular flow and shielded by an aluminosilicate overhang to prevent direct irradiation of the probes. Unsheathed, J-Type thermocouples (Omega 5TC-GG-J-20-6) were positioned within the reactor to measure lower temperature reactor cavity components and seals. Thermocouples penetrated the vacuum chamber via thermocouple vacuum feedthroughs (Kurt J. Lesker custom N-type XTEMP-FT and

TFT5JY00003). An infrared thermometer (ETEKCITY LaserGrip 1080 infrared thermometer) was used to determine the quartz window seal and the hopper surface temperature.

The fully fabricated STInGR is shown in Figure 3 for (a) the STInGR sealed with upper hopper and lower collector assemblies and mounted within the support frame, (b) the water-cooled Cu face, and (c) the assembled aluminosilicate cavity within the stainless steel sleeve.

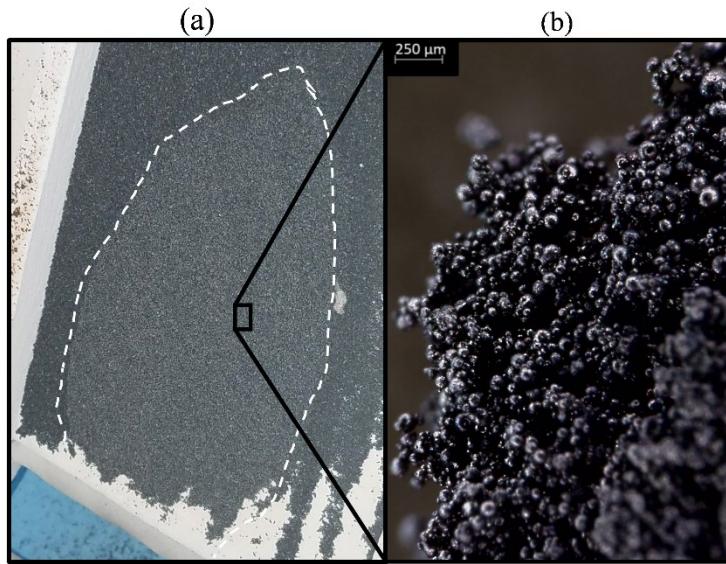


**Figure 3.** Fully-fabricated 5 kW<sub>th</sub> solar thermochemical inclined granular flow reactor (a) front-view with upper hopper and lower collector assemblies mounted in the support frame, (b) closer, side-view, and (c) assembled aluminosilicate cavity within stainless steel sleeve.

### 3. Agglomeration of spray-dried calcium manganite particles

Preliminary experimentation was performed with dense, granular flows of spray-dried CAM28 for various slope inclination angles ( $\theta_{\text{slope}}$ ) and radiative inputs from the HFSS. Agglomerate formation was observed for sustained particle flow temperatures  $>1150$  K at the flow outlet. The agglomerate structures are shown in Figure 4 (a) along the flow outlet bounded by the dashed line and (b) from an optical microscopy (Leica DVM 6 Optical Microscope) image taken after experimentation. The agglomerates formed along the inclined plane near the flow outlet, diverting flow around the structure and growing upwards along the flow channel over the course of the experiment. The agglomerate remained in place along the incline, as the gravitational shear was insufficient to overcome the inter-particle binding mechanisms. The optical microscopy image

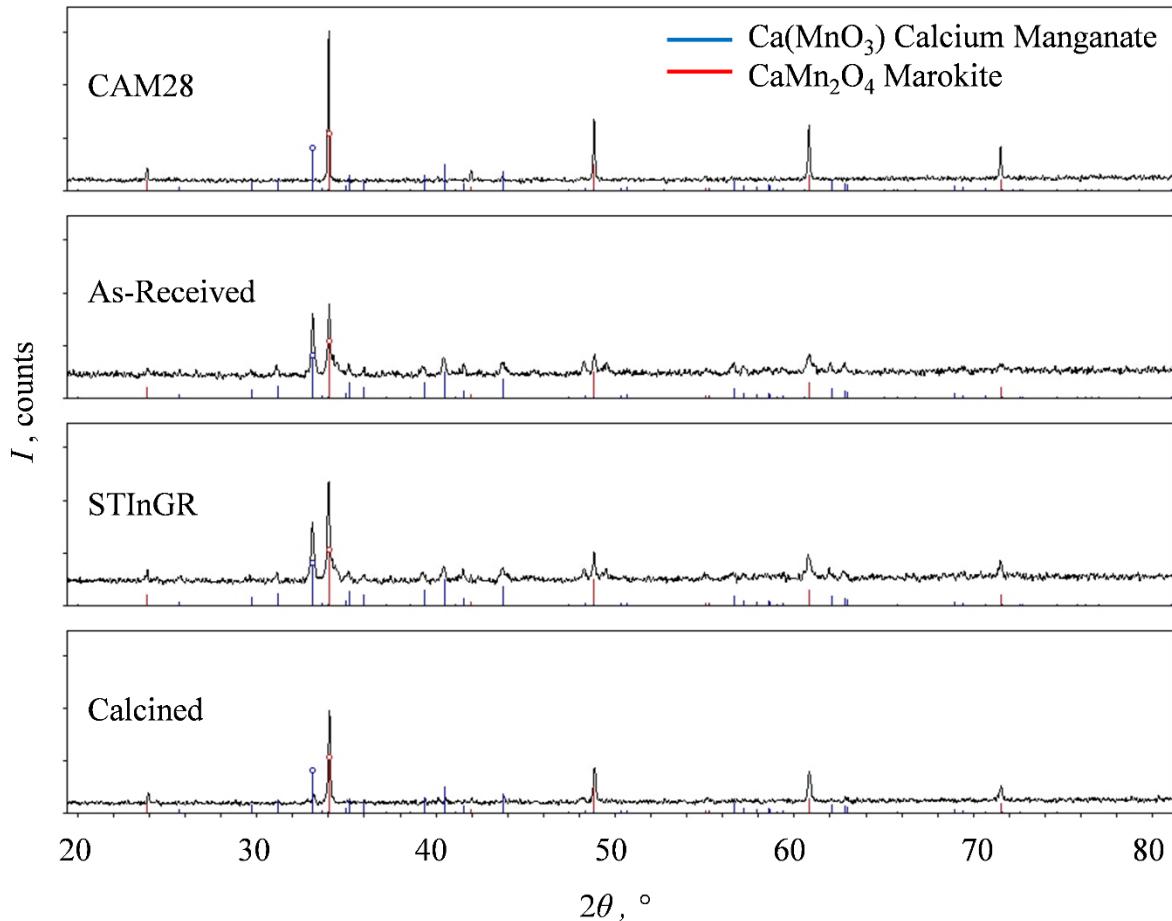
(Figure 4b) showed signs of the formation of dendritic agglomeration [57, 58] , where the agglomerate was formed from weakly-bonded individual particles. The agglomerate did not adhere to the flow channel and was easily removed from the slope [36]; however, it retained a stable inter-particle agglomerate structure. Upon removal from the slope the agglomerates were found to break apart easily, returning to the original state with no appreciable change in particle size.



**Figure 4.** Agglomerate structures of spray-dried calcium manganite particles pictured (a) along the outlet of the granular flow channel bounded by the dashed line and (b) from an optical microscopy image of a structure removed from the reactor.

Powder X-ray diffraction (XRD, PANalytical X’Pert PRO Alpha-1 diffractometer with  $2\theta = 10 - 90^\circ$ , step width  $0.17^\circ$ ) was performed on the CAM28 particles and analyzed using MDI JADE 9 (Materials Data Inc. Jade+, Livermore, CA) powder diffraction analysis software and Powder Diffraction File 4+ reference database to identify phases present in the CAM28 formulation. XRD patterns are shown in Figure 5 for single-phase CAM28 synthesized as described in prior work [7], compared to spray-dried CAM28 particles in the as-received condition, cycled through multiple STInGR reduction/oxidation cycles, and additionally calcined in an air furnace (Sento Tech Corp. ST-1600C-445) at 1473 K for 8 h. Single-phase CAM28 crystallizes in the cubic perovskite structure (PDF# 01-079-6765). The XRD patterns of the as-received spray-dried and STInGR-cycled particles revealed a secondary phase corresponding to the  $\text{CaMn}_2\text{O}_4$  spinel

structure (PDF# 4-015-3975). A related phase,  $\text{Ca}_2\text{MnO}_4$ , has been previously observed as an impurity in spray-dried calcium manganite particles, resulting in significant agglomeration for temperatures of 1173 to 1273 K [59]. Additional calcination of the spray-dried particles in air at 1473 K for 8 h, resulted in a decrease of the impurity spinel phase. As the Mn is in a lower oxidation state in  $\text{CaMn}_2\text{O}_4$  than in the CAM28 (+3 and +4, respectively), it is thought that the additional calcination acts to oxidize the sample.



**Figure 5.** X-ray diffraction patterns of spray-dried, aluminum-doped calcium manganite granular powder samples in the as-received condition, cycled through multiple reactor (STInGR) reduction / oxidation cycles, and calcined in air at 1473 K for 8 hours. Single-phase  $\text{CaAl}_{0.2}\text{Mn}_{0.8}\text{O}_{3-\delta}$  powder synthesized in the lab is shown for comparison. The peaks have been indexed to cubic perovskite  $\text{CaMnO}_3$  (red) and  $\text{CaMn}_2\text{O}_4$  spinel structure (blue).

#### 4. Reactor Demonstration Results and Discussion

Experimental conditions were optimized for continuous flow that avoided agglomeration. STInGR performance was evaluated based upon the average outlet granular flow temperature, evolved  $\text{O}_2$  (*i.e.*, calculated from the evolved  $\text{O}_2$  flow), and absorption efficiency, given as:

$$\eta_{\text{abs}} = \frac{\dot{m}_{\text{flow}} \cdot (C_p [\bar{T}_{\text{particle}} - \bar{T}_{\text{hopper}}] + \Delta H_{\text{rxn,0} \rightarrow \delta})}{\dot{Q}_{\text{HFSS}}} \cdot 100 \quad (1)$$

where  $\dot{m}_{\text{flow}}$  is the granular flow mass flow rate;  $C_p$  is the specific heat capacity of the aluminum-doped calcium manganite;  $T_{\text{particle}}$  is the outlet granular flow temperature;  $T_{\text{hopper}}$  is the hopper surface temperature;  $\Delta H_{\text{rxn,0} \rightarrow \delta}$  is the non-isothermal reduction enthalpy assuming stoichiometric deviation of  $0 \rightarrow \delta$  based on evolved O<sub>2</sub> measurements of pure CAM28 [8, 60]; and  $\dot{Q}_{\text{HFSS}}$  is concentrated radiative heat input delivered by HFSS lamps within the STInGR aperture at the focal plane. The specific heat of CAM28 was predicted with the Dulong-Petit law [7, 16, 61, 62] as 0.918 kJ/kg-K. This was accurately correlated to measured specific heats from differential scanning calorimetry (Netszch STA 409 C/CD) for pre-reduced CAM28 (1273 K) from 473 – 1273 K using the ASTM E 1269 standard. A comparison between measured and modeled specific heats is available in the Supplemental Material (Supplemental Fig. A). The experimental conditions during steady-state operation for two replicate experiments are given in Table 1. The cavity depth ( $l_{\text{cavity}}$ ) and granular flow width ( $w$ ) were optimized to avoid the formation of hot spots from individual lamps that resulted in agglomerates.  $\theta_{\text{slope}}$  was selected to increase shear within the granular flow to inhibit agglomeration.  $T_{\text{hopper}}$  was measured at the completion of HFSS operation while particles continued to flow through the reactor cavity. The  $\dot{m}_{\text{flow}}$  was determined from the total mass loss in the hopper and the total particle flow time.  $\dot{Q}_{\text{HFSS}} = 4.43 \pm 0.03 \text{ kW}_{\text{th}}$  was determined from coupled calorimetry and flux map measurements at steady-state for the same lamp power inputs after the completion of the experiments. The five outer HFSS lamps delivered  $\dot{Q}_{\text{HFSS}}$  to the reactor to reduce localized hot spots that lead to agglomerates. The spatial radiative fluxes were measured with a CCD camera for a Lambertian surface mounted in the focal plane. The pixel intensities were correlated to the spatial radiative heat fluxes [51, 63].

**Table 1.** Steady-state experimental conditions for reactor operation in the high-flux solar simulator

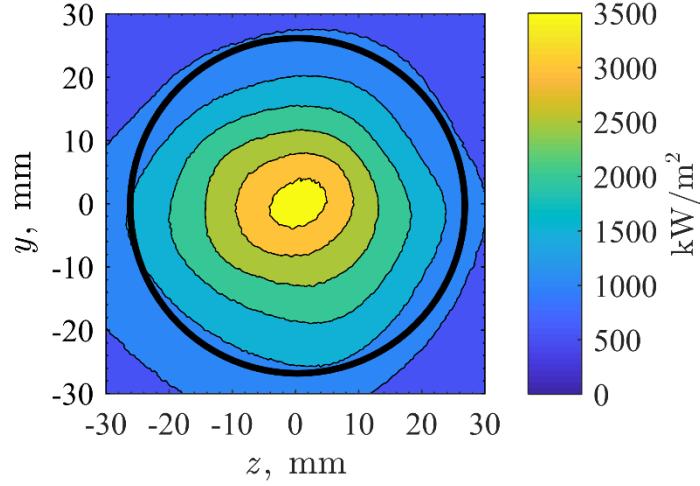
Parameters	Values
$l_{\text{cavity}}$	100 mm
$w$	80 mm
$\theta_{\text{slope}}$	35°
Vibratory motor	On
$\dot{V}_{\text{Ar}}$	0.5 L <sub>N</sub> /min

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$T_{\text{hopper}}$	297 – 300 K
$\dot{m}_{\text{flow}}$	222 - 224 g/min
$\dot{Q}_{\text{HFSS}}$	$4.43 \pm 0.03 \text{ kW}_{\text{th}}$

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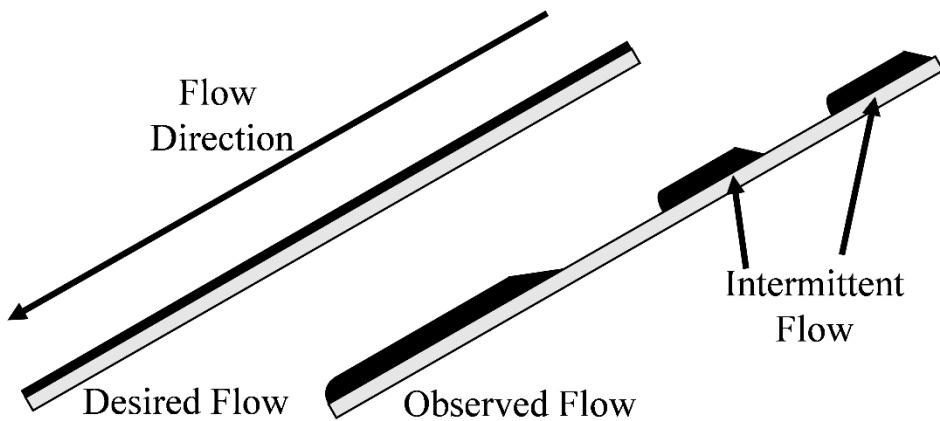
The spatial radiative heat fluxes in the focal plane are given in Figure 6, with an overlaid 50 mm circle that denotes the STInGR aperture. Peak fluxes up to  $3700 \text{ kW/m}^2$  were measured at the center of the aperture, and the fluxes decreased to between  $1200$  and  $1600 \text{ kW/m}^2$  at the aperture edges. Higher radiative flux concentrations were measured at the lower portion of the aperture area, as the lower three lamps were operated at higher powers during experimentation.



**Figure 6.** Measured radiative heat flux map in  $\text{kW/m}^2$  at the focal plane of the high flux solar simulator, with the position of the reactor aperture represented by the black circle

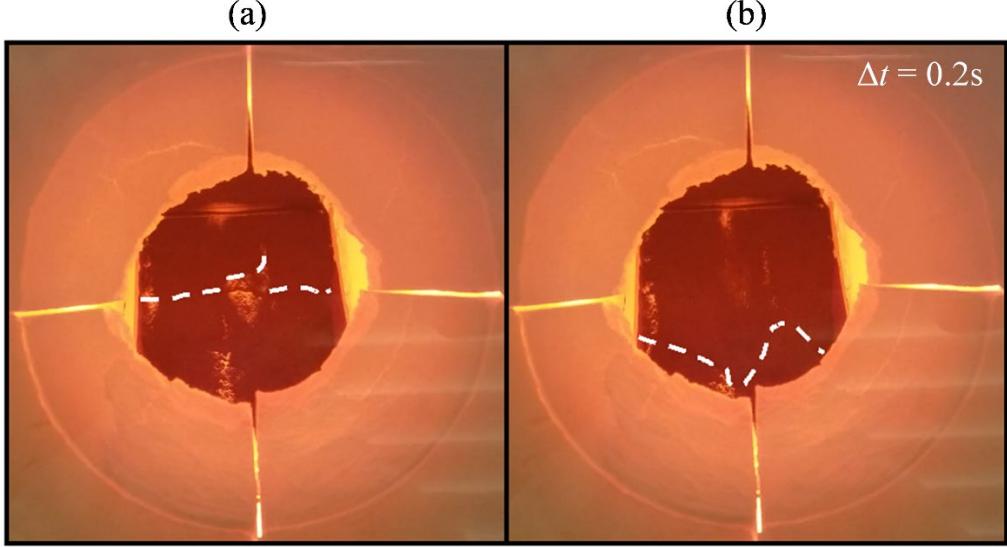
#### 4.1 Intermittent, dense granular flows

Agglomeration was avoided for the replicate experiments; however, increased inter-particle cohesion at high temperatures was observed. The increased cohesion produced intermittent, dense granular flows. Comparison between the desired, steady, granular flow [16, 19, 64] and the observed, intermittent granular flow during high-temperature operation is shown in Figure 7. The intermittent granular flow layer was thicker than expected compared to ambient temperatures, which resulted in direct exposure of the granular flow channel to concentrated irradiation during the experimental runs.



**Figure 7.** Schematic of desired, steady, dense granular flow compared to the observed, intermittent, dense granular flows during high temperature operation with the spray-dried, aluminum-doped calcium manganite granular powder

Two images of the intermittent, dense, granular flow viewed through the quartz window and aperture immediately following continuous reactor operation are displayed in Figure 8, normal to the aperture for (a) an initial frame and (b) a second frame 0.2 s later with the flow front indicated by a dashed line. The intermittent, dense granular flow was not evenly distributed across the flow channel as is shown by the non-uniform flow front and the exposed alumina within the flow. The intermittent flow was caused by a particle heap formed at the top of the incline as particles were introduced into the cavity. The heap formation was not observed at room temperature flow experiments [16], indicative of increased cohesion between particles at higher temperatures. The heap built-up until gravitational shear forces broke the heap and a dense, granular flow wave was developed along the incline. No particle agglomeration was observed along the incline. However, slight agglomeration was observed along the sidewalls near the particle outlet due to the increased frictional effects from the combined incline and sidewall interactions. When the HFSS was turned off, the slight agglomeration was dissipated by the continuous particle flow as the reactor cooled, indicating agglomeration formation was dependent on temperature. Hence, by adjusting parameters such as mass flow rate, inclination angle, or concentrated irradiation levels minimize the formation of agglomerates.

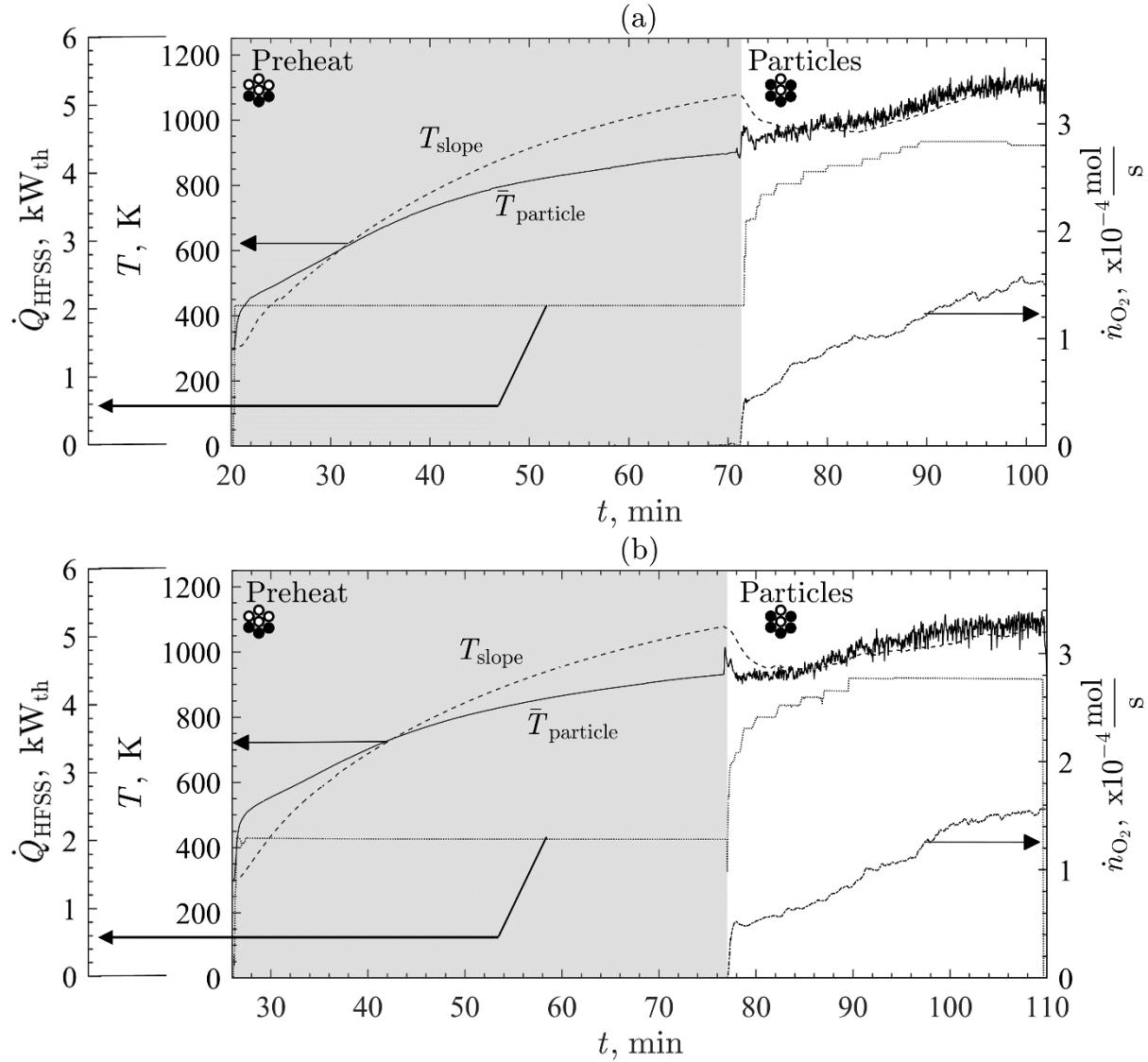


**Figure 8.** Images of intermittent, dense, granular flow of spray-dried aluminum-doped calcium manganite viewed through the quartz window and aperture immediately after high-flux solar simulator operation for (a) an initial frame and (b) a second frame 0.2 seconds later with the flow front indicated by a dashed line.

#### 4.2 Continuous operation

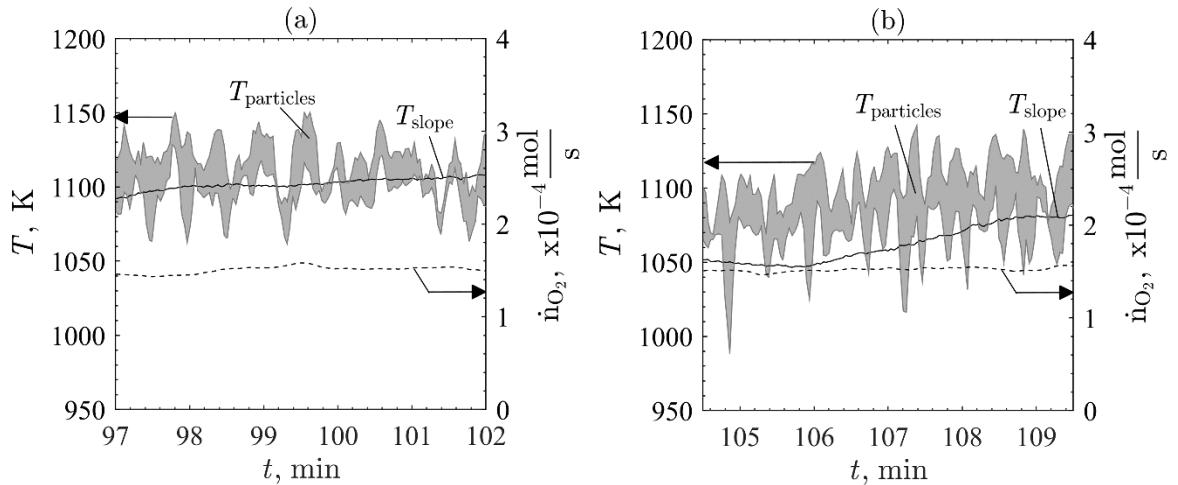
Two experiments were performed under similar conditions to assess reactor performance. Prior to experimentation, the total pressure was reduced to  $< 80$  mbar. For the continuous experiments, a steady pressure of 55-60 mbar was achieved using the vacuum pump. A 0.5 L<sub>N</sub>/min (where L<sub>N</sub> denotes liters at normal conditions of 273 K and 1 bar) flow of Ar was introduced to further reduce O<sub>2</sub> partial pressure ( $p_{O_2}$ ). The experiment commenced with the introduction of concentrated irradiation. The temporal slope temperature ( $T_{slope}$ ), average particle outlet temperature ( $\bar{T}_{particle}$ ), evolved O<sub>2</sub> ( $\dot{n}_{O_2}$ ), and  $\dot{Q}_{HFSS}$  are shown in Figure 9 for both replicated continuous experiments. The cavity was preheated for  $\sim 50$  min using the three lower lamps until  $T_{slope} \rightarrow 1073$  K. Once  $T_{slope} = 1073$  K, particles were introduced to the cavity, and the VM was engaged to prevent agglomerate formation.  $\dot{Q}_{HFSS}$  was increased after the particles were introduced by igniting two additional lamps and gradually increasing the power to the five lamps until near steady state was observed. The introduction of particles at  $t = 70$  and 77 min resulted in a rapid decrease in  $T_{slope}$  and a rapid increase in  $\bar{T}_{particle}$  and  $\dot{n}_{O_2}$  as concentrated irradiation was stored sensibly and

chemically in the flowing CAM28 particles.  $\bar{T}_{\text{particle}} > T_{\text{slope}}$  was expected as the dense granular flow intercepted direct irradiation from the HFSS lamps and transferred the absorbed radiation downstream towards the outlet thermocouple array. The fluctuating measurements of  $\bar{T}_{\text{particle}}$  were due to intermittent granular flows striking the outlet thermocouple array. All the lamps were turned off after  $\sim 5$  min of steady operation.



**Figure 9.** Experimental summaries of solar thermochemical inclined granular flow reactor experiments in the high-flux solar simulator including temporal measurements of outlet particle and slope (dash) temperatures, power delivered to a 50 mm aperture at the focal plane with representative lamp combinations (dotted), and evolved O<sub>2</sub> (dashed-dotted), for both preheat and particle flow operation regimes for (a) first and (b) second replicate experiments.

$T_{\text{particle}}$ ,  $T_{\text{slope}}$  and  $\dot{n}_{\text{O}_2}$  are given as functions of time for steady-state operation in Figure 10 for two experiments approaching steady state. The range of  $T_{\text{particle}}$  between temporal max/min readings are indicated by the grayed region. Steady state was achieved for the final 2 min of operation for the second experiment (Figure 11b) as  $T_{\text{slope}}$  approached a steady isotherm. Rapid increases in  $T_{\text{particle}}$  were due to the intermittent, dense granular flow striking a portion of the thermocouple array.  $T_{\text{particle}} > T_{\text{slope}}$  for particles in contact with the outlet thermocouple array. The wide range of  $T_{\text{particle}}$  when in contact with an intermittent flow was indicative of flow deviations across the granular flow width. The  $\dot{n}_{\text{O}_2}$  remained constant during steady-state operation independent of the intermittent granular flow as dispersion through the reactor cavity damped variations in the  $\text{O}_2$  evolving from the granular flow.



**Figure 10.** Temporal measurements of the temperature range of the slope temperature, particle temperature range (gray region), and evolved  $\text{O}_2$  (dashed) for near steady-state operating conditions during high-flux solar simulator experimentation for (a) first and (b) second replicate experiments

The reactor performance metrics at steady-state are shown in Table 2 for  $\bar{T}_{\text{particle}}$ ,  $\Delta\delta$ , and  $\eta_{\text{total}}$  for each experiment. A repeatable  $\bar{T}_{\text{particle}}$  from 1112- 1123 K was measured during contact between the intermittent granular flow and the outlet thermocouple array. No agglomerates were formed along the directly irradiated flow channel. A bulk  $\Delta\delta$  was between 0.0110 and 0.0113 from both experiments, determined from evolved  $\text{O}_2$  measurements. The low  $\Delta\delta$  was due to a combination of reduced residence times of thin, granular flows along the directly irradiated portion of the flow channel and impurity phase(s) in the TCES material.

The impact of impurity phase(s) on reduction potential was investigated using thermogravimetric analysis (TGA, Netszch STA 449 F3 Jupiter) of granular powder samples with reduction cycles with 10% O<sub>2</sub>/Ar flow at 1473 K and oxidation cycles with 80% O<sub>2</sub>/Ar flow at 773 K. As-received, STInGR-cycled, and re-calcined granular powder samples all achieved a nearly identical mass loss/gain of 1.6%, which is significantly less than the 2.2% mass loss/gain predicted for single-phase CAM28 under similar cycling conditions. The smaller mass change is due to the presence of the spinel phase, which is not redox-active. A repeatable  $\eta_{\text{abs}}$  from 63.4% to 64.7% was achieved for STInGR experiments 1 and 2, respectively. Lower  $\bar{T}_{\text{particle}}$  and  $\eta_{\text{total}}$  were observed for the second experiment despite the reduced  $\dot{m}_{\text{flow}}$ . This was likely due to a slightly lower  $\dot{Q}_{\text{HFSS}}$  in experiment 2. A low  $\eta_{\text{abs}}$  for both experiments was a result of the reflection losses from the quartz window, the water-cooled Cu plate, and the restricted flow width.

**Table 2.** Average and standard deviation of steady-state reactor performance parameters measured during replicate experiments

Experiment	1	2
$\bar{T}_{\text{particle}}$	$1123 \pm 21 \text{ K}$	$1112 \pm 22 \text{ K}$
$\Delta\delta$	$0.0110 \pm 0.0002$	$0.0113 \pm 0.0002$
$\eta_{\text{abs}}, \%$	$64.7 \pm 1.7\%$	$63.4 \pm 1.8 \%$

#### 4. Summary and Conclusions

The 5 kWth solar thermochemical inclined granular-flow reactor was tested in a high flux solar simulator using spray-dried aluminum-doped calcium manganite particles as a heat transfer and thermochemical energy storage medium. The reactor was designed to efficiently store concentrated solar irradiation sensibly and chemically in dense, granular flows of particles for continuous on-sun operation. Particle agglomeration was investigated, and x-ray diffractometry was used to identify potential impurity phases that promoted agglomeration and decreased thermal reduction performance. The reactor and high-flux solar simulator operating conditions were tuned to mitigate severe inter-particle agglomeration observed in the spray-dried aluminum-doped calcium manganite particles for temperatures exceeding 1150 K. The reactor was operated continuously using dense granular flows of particles with bulk granular flow temperatures greater than 1073 K, thermochemical energy storage, and a maximum absorption efficiency of 64.7%.

The results from the experimental campaign provide valuable insights into the performance of granular flow reactors that operate continuously. While inter-particle agglomeration introduced a significant challenge, the initial experimental results showed that the operating conditions are tunable for overcoming agglomeration formations.

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## Nomenclature

$C_p$  specific heat

$d$  diameter

$\Delta H$  enthalpy

$l$  length

$m$  mass

$n$  moles

$P$  pressure

$Q$  thermal energy

$T$  temperature

$V$  volume

$w$  flow width

## Greek Letters

$\delta$	non-stoichiometric oxygen vacancy concentration
$\Delta$	change in property
$\eta$	efficiency
$\theta$	angle

## Subscripts

abs	absorption
cavity	reactor cavity
flow	particle flow
$O_2$	oxygen gas
Ar	argon gas
particle	particles
hopper	hopper surface
rxn	reaction
slope	inclined slope

## Superscripts/Accents

.	time rate of change
-	average

## Abbreviations

CAM28	$CaAl_{0.2}Mn_{0.8}O_{3-\delta}$
DAQ	data acquisition
GC	gas chromatograph
HFSS	high-flux solar simulator
MS	mass spectrometer
PT	pressure transducer
STInGR	Solar Thermochemical Inclined Granular Flow Reactor
VM	vibratory motor

XRD x-ray diffractometry

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