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## 5 **Plant Root Exudates Increase Methane Emissions through Direct and** 6 **Indirect Pathways**

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16

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35      **Abstract**

36      The largest natural source of methane ( $\text{CH}_4$ ) to the atmosphere is wetlands, which produce 20%  
37      to 50% of total global emissions. Vascular plants play a key role regulating wetland  $\text{CH}_4$  emissions  
38      through multiple mechanisms. They often contain aerenchymatous tissues which act as a diffusive  
39      pathway for  $\text{CH}_4$  to travel from the anoxic soil to the atmosphere and for  $\text{O}_2$  to diffuse into the soil and  
40      enable methanotrophy. Plants also exude carbon from their roots which stimulates microbial activity  
41      and fuels methanogenesis. This study investigated these mechanisms in a laboratory experiment  
42      utilizing rootboxes containing either *Carex aquatilis* plants, silicone tubes that simulated  
43      aerenchymatous gas transfer, or only soil as a control.  $\text{CH}_4$  emissions were over 50 times greater from  
44      planted boxes than from control boxes or simulated plants, indicating that the physical transport  
45      pathway of aerenchyma was of little importance when not paired with other effects of plant biology.  
46      Plants were exposed to  $^{13}\text{CO}_2$  at two time-points and subsequent enrichment of root tissue, rhizosphere  
47      soil, and emitted  $\text{CH}_4$  was used in an isotope mixing model to determine the proportion of plant-derived  
48      versus soil-derived carbon supporting methanogenesis. Results showed that carbon exuded by plants  
49      was converted to  $\text{CH}_4$  but also that planted boxes emitted 28 times more soil-derived carbon than the  
50      other experimental treatments. At the end of the experiment, emissions of excess soil-derived carbon  
51      from planted boxes exceeded the emission of plant-derived carbon. This result signifies that plants and  
52      root exudates altered the soil chemical environment, increased microbial metabolism, and/or changed  
53      the microbial community such that microbial utilization of soil carbon was increased (e.g. microbial  
54      priming) and/or oxidation of soil-derived  $\text{CH}_4$  was decreased (e.g., by microbial competition for oxygen).

55

56      **Introduction**

57      Methane ( $\text{CH}_4$ ) is a potent greenhouse gas responsible for 15-19% of total greenhouse gas  
58      radiative forcing, second only to carbon dioxide ( $\text{CO}_2$ ) (Intergovernmental Panel on Climate Change  
59      2014). The largest natural source of  $\text{CH}_4$  to the atmosphere is wetlands, which produce between 20%  
60      and 50% of total (anthropogenic and natural) emissions (Ciais et al. 2013). Over half of global wetland  
61      area is in the Boreal region (Aselmann and Crutzen 1989) where temperatures are rising at a faster-than-  
62      average rate (Intergovernmental Panel on Climate Change 2014), and where wetland  $\text{CH}_4$  emissions are  
63      increasing (Gedney et al. 2004; Zhang et al. 2017). Wetlands generate  $\text{CH}_4$  because the saturated soils  
64      are anoxic and the anaerobic microbial metabolic pathways that break down organic carbon terminate  
65      in methanogenesis. The amount of  $\text{CH}_4$  emitted is controlled by factors such as vegetation type (Fritz et

66 al. 2011), season (Wang and Han 2005), temperature and precipitation (Hodson et al. 2011), and soil  
67 types (Kayranli et al. 2009). Understanding how these variables affect wetland CH<sub>4</sub> production and  
68 emissions is important because these variables are often sensitive to climate and environmental  
69 conditions; as the climate warms and environmental conditions change, the response of these variables  
70 could alter wetland CH<sub>4</sub> emissions, creating feedback loops (Gedney et al. 2004; Zhang et al. 2017). One  
71 such factor that both influences CH<sub>4</sub> emissions and in turn is influenced by climate warming is the  
72 growth of wetland plants (Kayranli et al. 2009). Modelling studies have identified the influence of plants  
73 on wetland CH<sub>4</sub> emissions as a prioritized area of needed study, even before climate effects are  
74 considered (Riley et al. 2011).

75 As global CO<sub>2</sub> concentrations and temperatures increase, boreal plants will respond with  
76 changed growth patterns and higher productivity (Forkel et al. 2016). Plant growth can influence CH<sub>4</sub>  
77 emissions in two ways. First, vascular wetland plants (e.g., sedges, shrubs, grasses) often contain  
78 aerenchymatous tissues that facilitate diffusion of gases between the soil and the atmosphere  
79 (Armstrong 1971). The hollow aerenchyma run from plant roots up into the leaves, increasing the total  
80 surface area of diffusional contact between the atmosphere and soil, and extending that contact area  
81 into saturated soil that would otherwise not be exposed to the atmosphere. Aerenchyma allow CH<sub>4</sub> to  
82 travel from the anoxic soil to the atmosphere, reducing CH<sub>4</sub> diffusion through the soil system, including  
83 through the oxic groundwater layer near the water-table surface and unsaturated soil where oxidation  
84 of CH<sub>4</sub> into CO<sub>2</sub> can occur (Shannon and White 1994; Popp et al. 2000; Fritz et al. 2011). The ability of  
85 CH<sub>4</sub> to diffuse through aerenchyma and bypass oxic water and unsaturated soil can decrease CH<sub>4</sub>  
86 oxidation, but only if the roots themselves are not surrounded by oxic groundwater. The soil  
87 surrounding roots (rhizosphere) can become oxygenated because aerenchymatous tissues allow  
88 diffusion in both directions, acting as a pathway for oxygen to enter the rhizosphere where it can  
89 preclude methanogenesis and facilitate CH<sub>4</sub> oxidation (i.e., methanotrophy) (Fritz et al. 2011). In highly  
90 reduced environments such as wetland soils, the presence of oxygen is a more important factor in CH<sub>4</sub>  
91 production than redox state (Fetzer and Conrad 1993). As boreal plants grow larger, which is predicted  
92 to occur with increasing CO<sub>2</sub> levels and temperatures (Idso et al. 1987; Jonasson et al. 1996), their root  
93 network will grow (Kummerow and Ellis 1984) and connect more of the soil to the aerenchymatous gas  
94 transport pathway.

95 Second, vascular plants contribute carbon to soil in the form of root exudates and leaf litter. It  
96 has been hypothesized that root exudates, which include low molecular weight sugars and amino acids,  
97 are more readily utilized by microbes than existing soil carbon and thereby fuel microbial activity

98 resulting in methanogenesis (Ström et al. 2003; Ström and Christensen 2007; Picek et al. 2007; Chanton  
99 et al. 2008; Kayranli et al. 2009). Methanogens can only use acetate or a combination of H<sub>2</sub> and CO<sub>2</sub> to  
100 actually form CH<sub>4</sub>, but root exudates may be broken down by other microbes to form the substrates of  
101 methanogenesis. Throughout this paper, we refer to the eventual conversion of exudate-derived carbon  
102 to CH<sub>4</sub> as being a “direct” effect of exudates, even though the microbial processing takes more than a  
103 single step. However, this causation is hard to prove in the complex plant-soil system. Root exudation  
104 increases with plant productivity (Weigel et al. 2005) as do CH<sub>4</sub> emissions (Ström et al. 2003), but there  
105 are other factors correlated with plant productivity that could influence CH<sub>4</sub> production and emissions,  
106 for example increased aerenchyma transport and warmer temperatures (microbial CH<sub>4</sub> production  
107 increases with temperature (Yvon-Durocher et al. 2014)). Multiple studies have shown that at least  
108 some plant-derived carbon is utilized by microbes and emitted as CH<sub>4</sub> (Megonigal et al. 1999; Ström et  
109 al. 2003; Trinder et al. 2008; Dorodnikov et al. 2011). However, these studies do not quantify to what  
110 extent root exudates increase total emissions, or if they instead replace soil carbon as the carbon  
111 source. In one case the portion of emitted CH<sub>4</sub> that was plant derived was extremely small (Dorodnikov  
112 et al. 2011), and in another neither microbial biomass nor metabolic activity was correlated with the  
113 rate of root exudation (Trinder et al. 2008). One soil incubation study showed that addition of root-  
114 exudate analogs to peat soil increased CH<sub>4</sub> production (Girkin et al. 2018a), but this effect was highly  
115 dependent on the composition of the exudate analogs (Girkin et al. 2018b). Because that approach  
116 could not replicate the delivery rate and complete carbon composition of root exudates from a living  
117 plant, nor can it account for the complex microbial community dynamics that exist in the rhizosphere.

118 It should be noted that the effects of aerenchyma transport and root exudation do not exist in  
119 isolation. For example, if root exudates provide carbon that is more readily used by the microbial  
120 community than the native soil carbon, exudates could stimulate increased microbial activity. An  
121 increase in aerobic microbial respiration would increase oxygen demand in the rhizosphere, which could  
122 decrease methanotrophy depending on total oxygen demand relative to supply from aerenchyma  
123 (Segers and Leffelaar 2001) and methanotrophs’ affinity for oxygen (Whalen 2005). If the rhizosphere  
124 consumes oxygen as quickly as it is delivered, then root exudation could create an anoxic pathway for  
125 CH<sub>4</sub> to diffuse to root aerenchyma, where CH<sub>4</sub> would quickly travel to the atmosphere. By decreasing  
126 methanotrophy in the rhizosphere, root exudation would allow the aerenchyma to transport more CH<sub>4</sub>  
127 than if exudates were not present.

128 In addition to directly fueling methanogenic pathways, root exudates can also stimulate  
129 microbial priming of soil carbon (e.g., Basliko et al. 2012). Priming is when the introduction of carbon in

130 the form of simple sugars and acids can stimulate a microbial population into breaking down soil organic  
131 matter. One explanation for priming is the microbial nitrogen mining (N-mining) hypothesis, which  
132 assumes that microbes oxidize the carbon-rich but nutrient-poor exudates for energy, but then must  
133 process soil organic matter to extract nitrogen (and potentially other nutrients) for biomass production  
134 (Craine et al. 2007). From a plant evolutionary perspective, encouraging N-mining with root exudation  
135 would be beneficial in nutrient-poor soils (Carvalhais et al. 2011), such as those in peat bogs. Priming has  
136 been observed in peat soils using a mass balance approach in soil incubation experiments (Hamer and  
137 Marschner 2002; Basiliko et al. 2012; Ye et al. 2015) and by analyzing dissolved organic matter  
138 characteristics in a boreal peatland field experiment that compared planted plots to plots with plants  
139 removed (Robroek et al. 2016). However, other soil incubation experiments have failed to find evidence  
140 for priming in laboratory incubations of peat soils (Girkin et al. 2018a; Girkin et al. 2018b). While not in  
141 peat soils, one field study on a river-bank found evidence that the addition of water-soluble organic  
142 carbon decreased processing of solid-phase hydrophobic carbon, which the authors interpreted as  
143 contradicting the idea of priming (Graham et al. 2017). However, that study was conducted in a mineral  
144 sediment with far lower carbon content than peat soils, so the chemical environment and characteristics  
145 of carbon which control bioavailability were different. It is possible that priming occurs only in certain  
146 conditions.

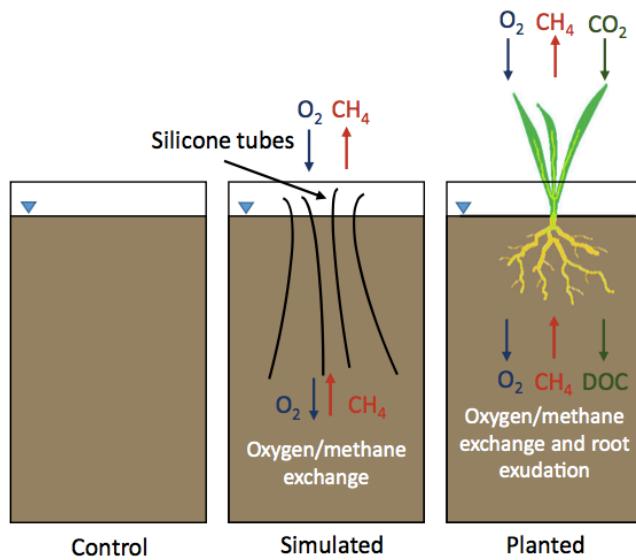
147 Our study was a combined investigation of the multiple mechanisms by which plants enhance  
148 CH<sub>4</sub> emissions, utilizing a vascular plant (*Carex aquatilis*) and peat soil typical of the boreal region. *Carex*  
149 *aquatilis* has been studied before and found to increase wetland CH<sub>4</sub> emissions (Schimel 1995). Our  
150 study used two avenues of investigation. The first approach sought to isolate the effect of aerenchyma  
151 transport by comparing soil oxygenation and CH<sub>4</sub> emissions from real *Carex* plants to that from hollow,  
152 gas permeable tubes that mimicked the aerenchyma transport capacity of the studied plants. The  
153 second approach sought to clarify the extent to which root exudates fuel CH<sub>4</sub> production by tracing the  
154 flow of isotopically labeled CO<sub>2</sub> fixed by the *Carex* plants, delivered to the microbial community, and  
155 emitted as CH<sub>4</sub>. Together, these approaches allowed us to examine the excess CH<sub>4</sub> emitted in the  
156 presence of *Carex* and attribute this excess to transport alone, microbial utilization of exudates, and/or  
157 indirect effects such as microbial priming and reduced methanotrophy.

158

159 **Materials and Methods**

160 ***Experimental Materials and Conditions***

161 We grew the wetland sedge *Carex aquatilis* from plugs purchased from a local nursery (Plants of  
 162 the Wild, Tekoa, WA) for 10 weeks in 5 L (5cm x 20cm x 50cm) rootboxes filled with peat collected from  
 163 a thermokarst bog in the Bonanza Creek Experimental Forest near Fairbanks, Alaska (Neumann et al.  
 164 2015). Peat from that bog is high in organic content and nutrient limited, with a prior soil core study  
 165 finding mean loss on ignition of 80% with standard deviation 13% and a mean C:N ratio of 48 with  
 166 standard deviation 20 (Manies et al. 2017). In addition, we had unplanted control boxes filled only with  
 167 peat and boxes with silicone tubes inserted into the peat instead of plants (Fig. 1). We refer to the  
 168 silicone-tube boxes as the “simulated” plant treatment since they were simulating the gas-transfer  
 169 effects of aerenchymatous plant tissues without adding any root exudate carbon, following the method  
 170 of (King et al. 1998). The silicone tubes (1.47 mm inner diameter, 0.23 mm wall thickness) had an open  
 171 top 4 cm above the peat surface and were tied off at the bottom (approximately 20cm below peat  
 172 surface) to prevent water from entering the tube. This design allowed gases to diffuse through the gas-  
 173 permeable silicone into and out of the soil. We placed four tubes in each box. This number was chosen  
 174 so that we could space the tubes widely enough that each tube’s effect could be detected in spatial  
 175 measurements of oxygen (described under “Oxygen Optode Technology”). Turner et al. (manuscript  
 176 currently under review) conducted a field experiment using similar simulated plants and optodes in the  
 177 same thermokarst bog from which peat for this study was collected. Though the aims of that experiment  
 178 were different, the parallel methodologies allow it to provide useful context for our results.



179  
 180 **Fig. 1** Three box types used in experiment. Control boxes had only peat soil. Simulated-plant boxes had soil with silicone tubes  
 181 that allowed O<sub>2</sub> to diffuse down into the soil and CH<sub>4</sub> to diffuse up into the atmosphere. Planted boxes had similar ability to  
 182 transport gas as simulated-plant boxes, but plants additionally performed photosynthesis and released dissolved organic carbon  
 183 from their roots

184        We kept all rootboxes at a 30-degree angle during the experiment to encourage roots to grow  
185 along the face of the rootbox, which allowed for sample collection across the rhizosphere and  
186 visualization of oxygen concentrations. Oxygen concentrations were measured using an optical oxygen  
187 sensor (i.e., an optode) (Larsen et al. 2011) that was applied to the face of a subset of boxes (described  
188 under “Oxygen Optode Technology”). Boxes without optodes had opaque front panels instead. Boxes  
189 with optodes have opaque covers that were kept on to prevent light from reaching the soil.

190        We conducted two different experiments. The first experiment consisted of 16 planted boxes  
191 that were all outfitted with optodes and focused solely on assessing dissolved oxygen concentrations in  
192 the rhizosphere. The second experiment consisted of 11 planted boxes, 3 control boxes, and 3  
193 simulated-plant boxes, and it involved a wider range of measurements aimed at identifying the  
194 mechanisms by which plants influence CH<sub>4</sub> emissions. All the boxes with simulated plants were outfitted  
195 with optodes in the second experiment.

196        In both experiments, we watered boxes as needed to keep the peat surface continuously  
197 submerged under approximately 1 cm of water. The irrigation solution was de-ionized water with trace  
198 amounts of nutrients added to simulate rainwater. The irrigation solution recipe is given in SI table S1.

199        We kept rhizoboxes in a pair of growth chambers and randomly re-assigned box position within  
200 and between the growth chambers twice per week to avoid any effects of spatial variability in growth  
201 conditions. We controlled environmental conditions within the chambers with growth lights and air  
202 conditioners set to imitate central Alaska summer growth conditions with 18 hours of daylight, daytime  
203 temperatures of 18 °C, and nighttime temperatures of 10 °C. We recorded temperature, relative  
204 humidity, and photosynthetically active radiation (PAR) (SI Figures S1 and S2) with sensors placed at the  
205 height of the tallest plants. During the first experiment, the median temperatures recorded with these  
206 sensors were 18.6 °C in one chamber and 20.6 °C in the other. In the second experiment the median  
207 temperatures recorded were 20.0 °C in one chamber and 20.5 °C in the other. We believe there was a  
208 vertical temperature gradient in the growth chambers because the air conditioners were located  
209 underneath the rhizoboxes while the sensors were directly under the growth lights.

210 ***Oxygen Optode Technology***

211        The planar optical oxygen sensors (i.e., optodes) used in this experiment were modified from  
212 Larsen et al. (2011). The sensors were made from two fluorescent dyes, an indicator dye and an antenna  
213 dye. When the antenna dye was excited by a photon it used part of the energy to fluoresce green and  
214 also passed energy to the indicator dye. The indicator dye was reversibly quenched in the presence of

215 oxygen, but when not quenched fluoresced red when excited. The ratio of red to green fluorescence is  
216 the foundation of the technology.

217 We made the optodes by airbrushing a mixture of Pt(II) meso-Tetra(pentafluorophenyl)porphine  
218 (the indicator dye) and 10-(2-Benzothiazolyl)-2,3,6,7-tetrahydro-1,1,7,7-tetramethyl-1H,5H,11H-  
219 (1)benzopyropyrano(6,7-8-I,j)quinolizin-11-one (the antenna dye) suspended in acetone onto 3/6 inch  
220 thick polycarbonate sheets. Once the dye mixture had dried, we applied a graphite coating by pouring a  
221 mixture of graphite and silicone dissolved in hexane onto the optode and tilting the optode so that the  
222 surface was evenly coated. The graphite coating provides enough opacity to prevent soil and roots  
223 behind the optode from interfering with the optical measurements.

224 We calibrated the optodes by attaching them to boxes filled with water that was bubbled with  
225 nitrogen gas and air to create a range of dissolved oxygen (DO) concentrations (as determined by a  
226 calibrated oxygen probe (InsituG Model 3100)), and photographing the optode as we would during  
227 experimentation. We took both calibration and experimental photographs by placing the box in a dark  
228 chamber and illuminating it with a blue (447.5 nm wavelength) LED positioned at a 45 degree angle to  
229 the optode, and taking a picture with a digital SRL camera (Canon EOS RebelXS) which had its near-IR  
230 filter removed and was fitted with a yellow filter (Edmunds Optics, OG-530 Long Pass Filter). The camera  
231 was controlled by the Look@RGB software. The red to green ratio (R) in each pixel was used to calculate  
232 the oxygen concentration (C) according to Equation 1, using the calibration parameters  $K_{sv}$ ,  $R_0$ , and  $\alpha$ , as  
233 described in Larsen et al. (2011).

$$(1) C = \frac{R_0 - R}{K_{sv} * (R - R_0 * \alpha)}$$

235

### 236 ***Non-Destructive Measurements — Second Experiment***

237 In addition to optode measurements, in the second experiment we measured plant height and  
238 fluxes of  $\text{CH}_4$  and  $\text{CO}_2$ . Gas fluxes were measured using one of two greenhouse gas analyzers (Los Gatos  
239 Research Ultraportable Greenhouse Gas Analyzer (LGR) and Picarro G2201-I CRDS) attached to an  
240 opaque hood that was strapped over the box of interest. The hood was 4.9 L in volume and enveloped  
241 both the entire plant and the entire soil surface area of the box. Headspace concentration was recorded  
242 by the instrument once per second for 5 to 10 minutes. The LGR recorded only total concentration of  
243  $\text{CH}_4$  and  $\text{CO}_2$  while the Picarro recorded  $\delta^{13}\text{C}$  of both gases as well. We applied a linear regression to the  
244 change in concentration over time to determine the flux rate.

245 The  $\delta^{13}\text{C}$  measurements for emitted  $\text{CH}_4$  had a non-trivial amount of noise, especially at low  
246 concentrations, making it unreliable to use a single starting  $\delta^{13}\text{C}$  value for  $\text{CH}_4$  in the chamber headspace

247 from which to calculate the  $\delta^{13}\text{C}$  of emitted  $\text{CH}_4$ . Instead, we calculated the isotopic ratio of emitted gas  
248 by taking every possible combination of measurements during the flux and using the difference between  
249 the concentrations and  $\delta^{13}\text{C}$  of  $\text{CH}_4$  in the headspace at any two points in time to calculate the  $\delta^{13}\text{C}$  of  
250 emitted  $\text{CH}_4$ . That method produced tens of thousands of results per flux, from which we used the  
251 median.

252 We only used flux data if the root mean square error (RMSE) of a linear regression was less than  
253 0.5 parts per million  $\text{CH}_4$  concentration per hour, which was equal to a flux rate of  $3.9 \text{ mg m}^{-2} \text{ d}^{-1}$  from  
254 our boxes which had a surface area of  $0.01 \text{ m}^2$ . We only used isotope data if the RMSE of the flux rate  
255 was less than 0.5 ppm, and the  $R^2$  value of the flux rate was greater than 0.9. The additional quality  
256 control check was needed for isotope data because at very low mass flux rates the RMSE can be low in  
257 absolute terms, but still comprise a significant portion of the total variation. We need to include low flux  
258 rates (even those with low  $R^2$  values) so as to not bias the data towards large fluxes, but the isotope  
259 calculations produced unreliable values when the  $R^2$  was low, regardless of the flux rate; they spanned  
260 an extreme range including unrealistically high and low values. All  $\delta^{13}\text{C}$  results were referenced to  
261 Vienna Pee Dee Belemnite and reported in delta notation:

$$\delta^{13}\text{C} = \left( \frac{R_{\text{sample}}}{R_{\text{standard}}} - 1 \right) * 1000$$

262  
263 Where  $R_{\text{sample}}$  is the  $^{13}\text{C}$  to  $^{12}\text{C}$  ratio of the measured sample and  $R_{\text{standard}}$  is the  $^{13}\text{C}$  to  $^{12}\text{C}$  ratio of Vienne  
264 Pee Dee Belemnite (0.0112372).

265  
266 ***Stable Isotope Labeling and Destructive Sampling — Second experiment***  
267 In weeks 5 and 10 of the second experiment, we exposed 4 randomly selected plants to  $^{13}\text{CO}_2$  by  
268 placing a clear 6.5 L hood made of extruded acrylic on each rhizobox and injecting 15 mL of 99 atom %  
269  $^{13}\text{CO}_2$  (Sigma Aldrich) into the headspace every hour for 14 hours each day over a period of five  
270 consecutive days. This technique was based on previously described methods (Lu and Conrad 2005). We  
271 took flux measurements at least once per day per plant during and after both labeling events with a  
272 Picarro G2201-I. During each day's 14-hour labelling period we took fluxes with the clear fluxing hood  
273 still in place, immediately before the time for an injection of  $^{13}\text{CO}_2$ . After the entire labelling event was  
274 completed, we used the same fluxing protocol as the routine flux measurements.

275 We destructively sampled boxes 3-5 days after isotopic labeling ended. To harvest a box, we  
276 attached a spill-guard to the top of the box to prevent water from pouring out, placed the box in an

277 anaerobic glove bag purged three times with nitrogen gas, laid the box on its back and removed the  
278 front panel to expose the root-soil system. We took all samples directly from this exposed surface. For  
279 all three box types (planted, control and simulated) we collected root and soil samples at depths of  
280 approximately 5 cm, 20 cm, and 35 cm. At each depth, we took three samples, one in the center and  
281 one 6 cm from either edge of the box.

282 For planted boxes, we cut root sections approximately 8cm in length from each location and  
283 placed them in centrifuge tubes filled with phosphate buffered solution (PBS), which we capped and  
284 removed from the glove bag. We sonicated the sealed centrifuge tubes with the root samples for 10  
285 minutes then quickly moved the root to a new PBS-filled container. Both containers were then  
286 immediately frozen at -20 C. All the soil that fell off the root during sonication was considered  
287 rhizosphere soil.

288 For unplanted control boxes, we sampled approximately 1mL cubes of soil from the 9 standard  
289 locations. We immediately placed samples into 2 mL collection tubes with 1 mL of PBS (to keep the  
290 sample storage as similar as possible to the planted box procedure) and placed the tubes on dry ice in  
291 the anoxic glovebag. We consider these soil samples as 'bulk soil.'

292 For simulated-plant boxes, at the 5 and 20cm depths where silicone tubes were present, we  
293 sampled a thin layer of peat immediately adjacent to the silicone tubes. The tubes did not extend to the  
294 35 cm depth, so we sampled 1mL cubes of bulk soil at that depth. We immediately placed samples into  
295 phosphate buffered solution (PBS) and stored them on dry ice in the anoxic glove bag.

296 All samples from all the boxes were then stored at -20 C for up to 33 days before being shipped  
297 on dry ice to the Environmental Molecular Sciences Laboratory (EMSL) where they were stored at -80 C  
298 until analysis.

#### 299 ***Porewater Collection and Analysis — Second experiment***

300 We collected porewater from all boxes during the second week of the second experiment,  
301 which we considered an initial time point, and immediately prior to the second labelling event in week 8  
302 from boxes that were not harvested after the first labeling event. We collected 10-16 mL of porewater  
303 from a depth of 20cm below peat surface at the center of each box using a syringe attached to  
304 PushPoint porewater sampler (MHE Products) with a mesh filter (10 $\mu$ m pore size) over the inlet. We  
305 then filtered the water (0.2  $\mu$ m, nylon membrane syringe tip filter) and used a syringe needle to inject it  
306 into previously prepared 60 mL serum vials capped with a butyl rubber stopper. We prepared the vials  
307 before sampling by evacuating and flushing them with 99.999% nitrogen gas three times, leaving the  
308 vials at atmospheric pressure. We pre-acidified the vials with 200  $\mu$ L of phosphoric acid to minimize

309 microbial activity. After injecting porewater into the vials, we left them for at least one week to  
310 equilibrate before we transferred 15mL of headspace gas into a 12mL exetainer vial (Labco) using a  
311 syringe and needle.

312 We used a Shimadzu GC-FID 2014 to measure concentration of CH<sub>4</sub> in the exetainer vial and  
313 calculated the porewater concentrations using Henry's Law and the mass of water in the original vial.  
314 We calibrated the GC measurements using standards of various concentrations (MESA Specialty Gases &  
315 Equipment). The standards purchased from MESA were a 100 ppm CH<sub>4</sub> standard, a 10.2 ppm CH<sub>4</sub>  
316 standard, and a 50% CH<sub>4</sub> standard which we diluted with N<sub>2</sub> to form an array of concentrations.

317 ***Isotope Ratio Mass Spectroscopy – Second experiment***

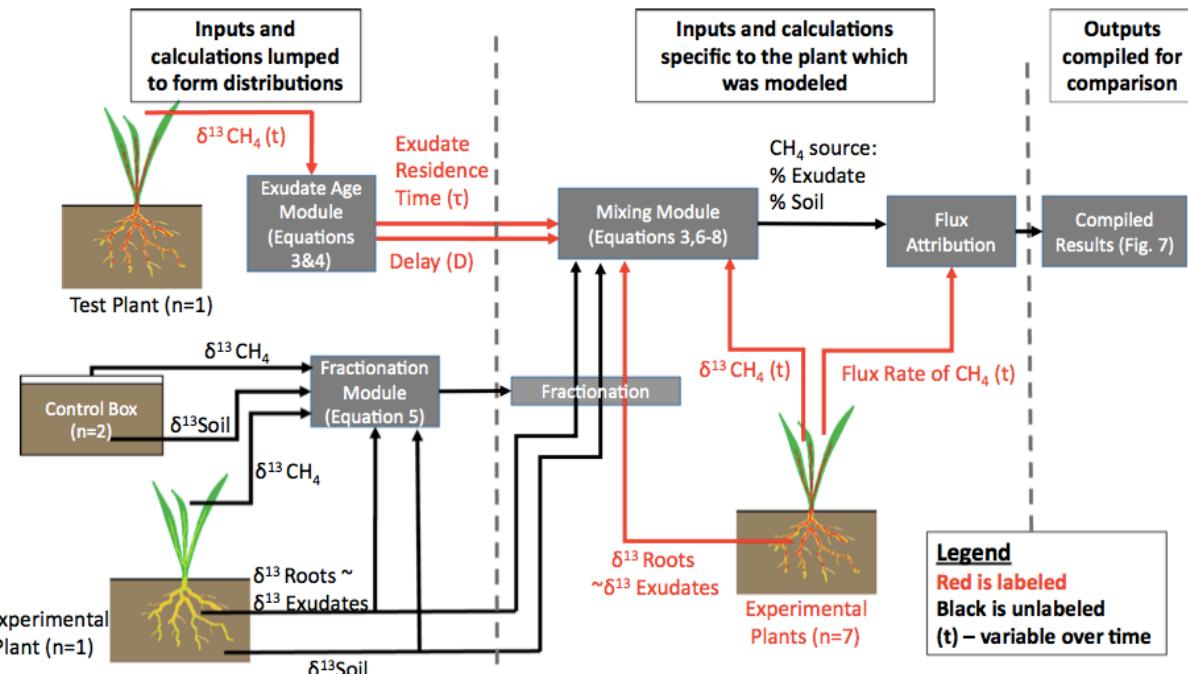
318 At EMSL, we used an a Costech Analytical Technologies, Inc. elemental analyzer (EA) coupled to  
319 a Thermo Scientific Delta V Plus isotope ratio mass spectrometer (IRMS) to perform isotopic analysis on  
320 root, rhizosphere and bulk soil samples. All soil samples (both bulk and rhizosphere) were removed from  
321 -80 °C storage and immediately lyophilized using a VirTis Benchtop K lyophilizer. We lyophilized 21 of 27  
322 root samples in the same manner, but the other 6 root samples were first analyzed at room  
323 temperature by laser ablation mass spectroscopy for data to be published in a different manuscript  
324 before being lyophilized. The laser ablation removed some mass from the surface of the root, but we  
325 assumed that it was a low enough percentage of the root that the average isotopic composition of the  
326 root was unaffected.

327 After lyophilization, samples were stored at room temperature. We sub-sampled the lyophilized  
328 rhizosphere samples into tin capsules (4 by 6 mm, part number 41070, Costech Analytical Technologies,  
329 Inc.) using 200-300 µg of sample per replicate with a minimum of triplicate analytical replicates. We cut  
330 lyophilized root samples into 1 to 3 mm thick cross sections using a razor and loaded the cross sections  
331 into tin capsules as described above. The EA combustion reactor was loaded with cobaltic oxide and  
332 chromium oxide catalyst and maintained at 1,020 °C while the reduction reactor was loaded with copper  
333 catalyst and maintained at 650 °C. We used two in-house glutamic acid standards that were themselves  
334 calibrated against USGS 40 and USGS 41 standards ( $\delta^{13}\text{C} = -26.39 \text{ ‰ VPBD}$  and  $37.63 \text{ ‰ VPDB}$   
335 respectively) and applied a two-point, slope intercept correction to the data (Coplen et al. 2006). In  
336 addition, we used an in-house acetanilide standard as a check on the isotope measurement accuracy.

337 ***Isotope Mixing Model – Second experiment***

338 We used the IRMS and CH<sub>4</sub> flux data to construct a mixing model to determine what portion of  
339 the CH<sub>4</sub> from each individual flux measurement was derived from soil carbon and what portion was  
340 derived from root exudates (Fig. 2). The model consisted of three modules: the exudate age module, the

341 fractionation module, and the mixing module. The exudate age and fractionation modules (described in  
 342 detail below) generated parameter distributions which were fed into the mixing module for the final  
 343 calculations that indicated what percentage of each  $\text{CH}_4$  flux what derived from soil versus exudate  
 344 carbon.



345

346 **Fig. 2** The isotope mixing model. In the first section, inputs from multiple boxes were combined to calculate distributions of  
 347 parameters for the model. Data a labeled test plant was used to calculate exudate residence times in the exudate age module,  
 348 while only unlabeled plants and control boxes were used to calculate isotopic fractionation effects. In the second section, each  
 349 flux was individually modeled using a distribution of parameters generated in the first step using a Monte-Carlo approach. The  
 350 mixing module used fractionated soil and root data to determine the percentage of each flux that came from soil, labeled  
 351 exudates, or unlabeled exudates, then in the flux attribution stage those percentages were multiplied by the total amount of  $\text{CH}_4$   
 352 emitted. In the final section of the model all of the fluxes were compiled to establish a range of results, as presented in Fig. 7. Any  
 353 variable given with (t) after it varied over time, while other variables do not

354 The exudate age module calculated the portion of total exudates in the rhizosphere that would  
 355 contain elevated  $\delta^{13}\text{C}$  at specific time points after labeling, given the residence time of exudates in the  
 356 soil ( $\tau$ ) and the delay (D) between the end of labeling and maximum  $^{13}\text{C}$  enrichment of emitted  $\text{CH}_4$ . The  
 357 module was run in two separate contexts: once on an isotopically labeled test plant independent of the  
 358 full model to determine parameter values (as described below), and again on experimental plants using  
 359 the fitted parameter values. The module modeled exudates in the soil as a continuously stirred tank  
 360 reactor (CSTR), and assumed the isotopic composition of emitted  $\text{CH}_4$  at any time point was linearly  
 361 correlated with the isotopic composition of exudates in the soil. The isotopic composition of the soil  
 362 carbon was a constant. This approach required that the  $^{13}\text{C}$  content of the exudates did not affect the  
 363 percentage of  $\text{CH}_4$  derived from those exudates. While various factors can affect the degree to which

364 microbes discriminate against  $^{13}\text{C}$  *within* a carbon source (Lehmeier et al. 2016), we could not identify  
365 any study which showed isotopic enrichment affecting the type of carbon compounds utilized by  
366 microbes. Therefore, we believe that we are justified in our assumption that increasing enrichment of  
367 exudates after labeling did not have a measureable impact on the portion of  $\text{CH}_4$  emissions that were  
368 soil-derived versus exudate-derived.

369 The general equation for a CSTR is given in Equation 2, and calculates the concentration ( $C(t)$ ) of  
370 a species at a given time ( $t$ ) based on residence time ( $\tau$ ), the initial concentration of the species ( $C_0$ ) and  
371 its concentration after infinite time ( $C_\infty$ ), which is equal to input concentration. In the case of our model,  
372 we defined isotopically labeled exudates as a “concentration” of one and unlabeled exudates as a  
373 “concentration” of zero, assuming that exudates maintained a consistent enrichment during the period  
374 over which they were labeled. Therefore, we can write Equations 3 and 4 to represent our system in  
375 which the “concentration” at a given time ( $C(t)$  in Equation 2) was the percentage of exudates in the  
376 rhizosphere that were labelled ( $E_L(t)$ ). The percentage labeled ( $E_L(t)$ ) was defined in Equation 7 as the  
377 portion of total emissions derived from labeled exudates ( $P_{LE}(t)$ ) divided by the portion of total  
378 emissions derived from both labeled exudates ( $P_{LE}(t)$ ) and unlabeled exudates ( $P_{UE}$ ). Because our  
379 labelling was a pulse input that ended after five days, the isotopic enrichment of emitted  $\text{CH}_4$  reached a  
380 maximum and then began to decline at some point after the 5-day long labeling period. Equations 3 and  
381 4 are both modified versions of Equation 2. Equation 3 calculates an input of labeled exudates to a  
382 previously unlabeled system and was used prior to the peak, while Equation 4 calculates the loss of  
383 labeled exudates from a previously labeled system and was used after the peak. Because we harvested  
384 the experimental plants before the peak, only Equation 3 was used on experimental plants. Equation 4  
385 was only used on the test plant to determine parameter values. The delay ( $D$ ) was the length of time, in  
386 days, between when labeling ended (i.e., after the 5-day long labeling period) and when emitted  $\text{CH}_4$   
387 reached maximum enrichment. In all cases, time ( $t$ ) zero was defined as the beginning of labeling. We  
388 calculated the portion of exudates labeled at the peak ( $C_{\text{peak}}$ ) using Equation 3 at  $t$  equal to length of  
389 labeling (five days) plus the delay. We used  $C_{\text{peak}}$  as the starting percentage of exudates that are labeled  
390 for time points beyond the peak (Equation 4).

391

392 Table 1, equations used in the isotope mixing model. Variables used:  $C_0$  is concentration at time zero, and no units are assigned  
 393 since it is not actually used in the model (Equation 2 is a generic form).  $C(t)$  is concentration at time  $t$ , where  $t$  is in days.  $C_\infty$  is  
 394 concentration at time infinity.  $\tau$  is residence time of exudates in days.  $P_{LE}$  is the percent of carbon used for methanogenesis  
 395 that is derived from labeled exudates.  $P_{UE}$  is the percent of carbon derived from unlabeled exudates.  $P_s$  is the percent of carbon  
 396 from soil.  $F$  is the fractionation between source carbon and emitted  $CH_4$ , in units  $\delta^{13}C\text{‰}$ .  $M_U$  is the  $^{13}C$  content of  $CH_4$  emitted  
 397 from unlabeled plants, in units  $\delta^{13}C\text{‰}$ .  $M_L$  is the enrichment of  $CH_4$  emitted from labeled plants, in units  $\delta^{13}C\text{‰}$ .

Equation #	Modules utilizing	Description	Equation
2	n/a	Generic CSTR	$C(t) = C_\infty + (C_0 - C_\infty)e^{-\frac{t}{\tau}}$
3	Exudate age Valid from $t=D$ to $t = 5 + D$	CSTR pre-peak (modified Eqn. 2) Initial condition: unlabeled exudates = 0 Input: labeled exudates = 1	$E_L(t) = 1 + (0 - 1)e^{-(t-D)/\tau}$
4	Exudate age Valid from $t = 5+D$ to $t = \infty$	CSTR post-peak (modified Eqn. 2) Initial condition: Proportion of labeled exudates calculated in Eqn. 3 at peak ( $t = 5+D$ ) = $C_{peak}$ Input: unlabeled exudates = 0	$E_L(t) = 0 + (C_{peak} - 0)e^{-(t-D-5)/\tau}$
5	Fractionation	Definition of fractionation	$F = M_U - S_{UR}$
6	Mixing	$\delta^{13}CH_4$ is linear combination of source $\delta^{13}$	$P_{LE}(t) * [F + S_{LR}] + P_{UE}(t) * [F + S_{UR}] + P_s * [F + S_S] = M_L(t)$
7	Mixing	Ratio of old to new exudates	$E_L(t) = \frac{P_{LE}(t)}{P_{LE}(t) + P_{UE}(t)}$
8	Mixing	All carbon sources add up to 100%	$P_{LE}(t) + P_{UE}(t) + P_s = 1$

398

399 The exudate age module was used in conjunction with data from a test plant to determine what  
 400 residence times and delay periods should be used as inputs to the model. A variety of residence times  
 401 and delays were used to generate theoretical enrichment curves, which we graphed against the test  
 402 plant data (SI Figure S3). We assigned a likelihood to each of these curves proportional to the inverse of  
 403 the sum of squares of the residuals, and those likelihoods were used to generate distributions from  
 404 which we sampled in the model. The median residence time and delay used are reported in Table 3  
 405 along with upper and lower quartiles.

406 We calculated separate fractionation factors for planted and control boxes. The fractionation  
 407 module calculated the apparent isotope fractionation associated with microbial conversion of either

408 soil-derived or root-derived carbon into  $\text{CH}_4$  and with subsequent emission of  $\text{CH}_4$ . The module  
409 subtracted the  $\delta^{13}\text{C}$  of  $\text{CH}_4$  emitted from unlabeled plants ( $M_U$ , fluxes from all plants prior to labeling  
410 were used) from that of unlabeled roots ( $S_{UR}$ ), and the  $\delta^{13}\text{C}$  of  $\text{CH}_4$  emitted from control boxes and  
411 simulated plants from that of unlabeled soil ( $S_{US}$ ) to calculate the isotopic fractionation of carbon (F).  
412 This calculation was done using all emissions from each type of box (planted or control) compared  
413 against all source carbon measurements from that type. This was done because only one flux  
414 measurement from the unlabeled plant which was harvested passed quality control, so we included  
415 fluxes from other plants prior to their harvest. The narrow range of  $\delta^{13}\text{C}$  for unlabeled fluxes justified  
416 this grouping (Fig. 5A). A distribution was generated from all fractionation factors calculated, and values  
417 used in the model were randomly sampled from that distribution.

418 The mixing module used output from the fractionation and exudate age modules to solve a  
419 system of equations that determined the proportion of emitted  $\text{CH}_4$  derived from soil ( $P_S$ ), isotopically  
420 labeled root exudates ( $P_{LE}(t)$ ), and unlabeled root exudates present in the soil prior to labeling ( $P_{UE}(t)$ ) as  
421 a function of time. In Equation 6, the isotopic enrichment of the emitted  $\text{CH}_4$  ( $M_L(t)$ , data from each flux  
422 individually) had to be a linear combination of the  $\delta^{13}\text{C}$  of the three fractionated sources: labeled roots  
423 ( $S_{LR}$ , data taken from only the box from which the  $M_L$  measurement was taken), unlabeled roots ( $S_{UR}$ ,  
424 median of a distribution) and unlabeled soil ( $S_S$ , median of a distribution). We used medians for  $S_{UR}$  and  
425  $S_S$  instead of a Monte-Carlo sampling approach because the distributions generated were very small and  
426 during model development we determined that simply using the median would reduce the amount of  
427 computation needed without influencing the results. Equation 7 established the ratio of the labeled and  
428 unlabeled exudates, according to the output of the exudate age module for that time point. Equation 8  
429 made the sum of all carbon sources equal to the full amount of the  $\text{CH}_4$  emitted.

430 In the model we used 30 randomly selected values from each of the four parameter  
431 distributions we generated (enrichment of exudates ( $S_{LR}$ ), fractionation factor (F), residence time ( $\tau$ ), and  
432 delay (D) between start of labeling and when the label was emitted as  $^{13}\text{CH}_4$ ). This resulted in 810,000  
433 results for each of the 21 fluxes used. The model (written in MATLAB 2018b) and all necessary functions  
434 and data are attached as a .zip file in the electronic supplementary information.

435 ***Statistical analysis***

436 We made all statistical comparisons using either the Wilcoxon rank sum test or the Kruskall-  
437 Wallace test, depending on whether we were comparing two groups or more than two groups of data.  
438 Both tests are non-parametric methods to determine whether two sets of measurements are likely to be  
439 from the same distribution or not.

440

441 **Results**

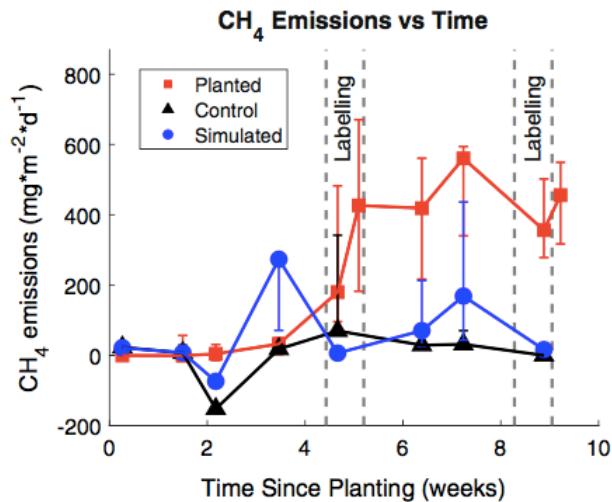
442 All results below are from the second experiment, except for optode images from planted  
 443 boxes. Oxygen data for planted boxes were collected in the first experiment. Growth chamber  
 444 conditions, plant height and CH<sub>4</sub> emission data from the first experiment are available in the Online  
 445 Resources (Figure S1 and Tables S3 and S5).

446

447 **Plant Height and Gas Emissions**

448 Plants grew steadily from a median height of 9.1 cm (n=11) to 32.9 cm (n=6) through the entire  
 449 period measured (between weeks 1 and 10 of the experiment). We report plant height and emissions  
 450 data for each rootbox in Online Resources tables S4 and S6, respectively.

451 Planted boxes emitted little CH<sub>4</sub> at first; the median emission flux was less than 25 mg m<sup>-2</sup> d<sup>-1</sup> for  
 452 each of the first four weeks of measurement (Fig. 3) (n ≥ 5 per week). In week five, emissions from  
 453 planted boxes increased slightly, with a median CH<sub>4</sub> flux rate of 98 mg m<sup>-2</sup> d<sup>-1</sup> and one box emitting up to  
 454 516 mg m<sup>-2</sup> d<sup>-1</sup>. For the remainder of the experiment, planted boxes had high CH<sub>4</sub> emissions with weekly  
 455 medians ranging from 370 to 580 mg m<sup>-2</sup> d<sup>-1</sup> (n ≥ 4 per week). A Kruskal-Wallace test showed that  
 456 emissions from planted boxes during the second half of the experiment (weeks six to ten) were  
 457 significantly greater than those from the first half (p < 0.05).



458

459 **Fig. 3** Medians for weekly CH<sub>4</sub> emissions from the three experimental treatments with error bars indicating upper and lower  
 460 quartiles. Vertical dashed lines show days when isotopic labelling began and ended. Isotopic labeling lasted 5 days. Methane  
 461 emissions from planted boxes increased over the course of the experiment, while those from control boxes and boxes with  
 462 simulated plants did not increase. The number of fluxes per week differed because of variability in the number of fluxes that  
 463 passed quality control and an increase in the number of flux measurements that were made during labeling. For planted boxes,

464 weekly  $n=2-15$ , median  $n=7.5$ . For control boxes, weekly  $n=1-3$ , median  $n=2.5$ . For simulated boxes, weekly  $n=1-3$ , median  $n=1.5$ .  
465 Negative fluxes for control and simulated plant treatments after week 2 are each from a single box

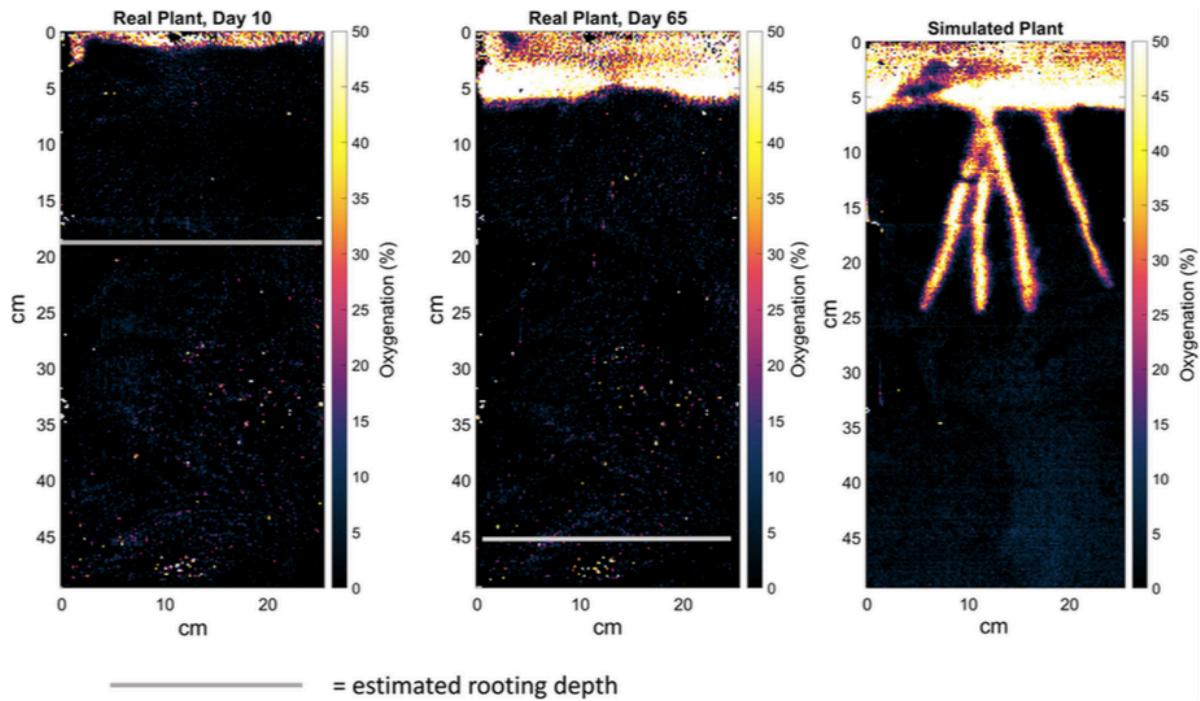
466 Control boxes emitted little CH<sub>4</sub> throughout the experiment (n=19 measurements), with weekly  
467 median emissions never exceeding 70 mg m<sup>-2</sup> d<sup>-1</sup>. Boxes with simulated plants typically emitted little CH<sub>4</sub>  
468 (n =16), though on three of the nine weeks of measurements they emitted larger than usual amounts,  
469 most notably in week 4 when the median flux was 273 mg m<sup>-2</sup> d<sup>-1</sup>. The differences between emissions of  
470 control boxes and simulated boxes were not significantly different from each other or from the  
471 emissions of planted boxes during the first half of the experiment ( $p > 0.05$ ), but emissions from both  
472 control boxes and simulated boxes were significantly lower than emissions from planted boxes during  
473 the second half of the experiment ( $p < 0.05$ ). These data indicate that emissions were related to plant  
474 growth, but not directly correlated; plants switched between a lower-emission state and a higher-  
475 emission state quickly (i.e., within two weeks), while their height increased in a nearly linear fashion.

476 All three box types had high variation in CO<sub>2</sub> fluxes, including periods of both uptake and  
477 emission (Online Resources Table S6 and Figure S4). As with CH<sub>4</sub> emissions, the Planted boxes did emit  
478 more CO<sub>2</sub> during the second half of the experiment than the other box types. Because the fluxes were  
479 taken in the dark, an increase in emission of CO<sub>2</sub> is consistent with increased plant and/or microbial  
480 respiration.

#### 481 ***Optode Imaging***

482 The planted boxes equipped with optodes during the first experiment never had measurable  
483 concentrations of oxygen in the soil. Representative oxygen-concentration images are shown in the two  
484 left panels of Fig. 4, and all 113 images (16 plants imaged weekly over 9 weeks, with some harvested at  
485 intermediate points) are in the Online Resources (Figures S5-S20). In contrast, the simulated plants in  
486 the second experiment had significant amounts of oxygen surrounding the inserted silicone tubes  
487 throughout the entire experiment (Fig. 4). The 27 simulated plant optode images (three boxes imaged  
488 weekly over nine weeks) are also in the Online Resources (Figures S21-S23).

489 It is worth acknowledging that we did not directly measure the gas-transport capacity of the  
490 plants or the silicone tubes, and so cannot claim that the simulated plants transported more or less  
491 oxygen than the real plants. Rather, what we can say is that the real plants did not allow what oxygen  
492 they did transport to build up in the soil.



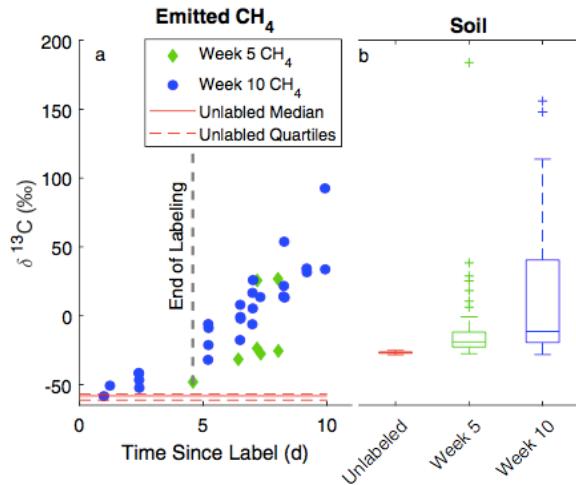
493

494 **Fig. 4** Examples of optode data showing that real plants never oxygenated the rhizosphere, while simulated plants had  
495 measurable oxygen concentrations surrounding the tubes throughout the experiment

496

#### 497 **Methane, Root, and Rhizosphere Isotope Enrichment**

498 During both labeling events (weeks 5 and 10 of the experiment),  $\text{CH}_4$  emitted from the boxes  
499 steadily became more isotopically enriched over a five-day period between the end of labeling and  
500 harvest (Fig. 5). During the week 5 labeling event fewer flux measurements were made than during the  
501 week ten event, and the  $\text{CH}_4$  emissions were lower which meant that fewer flux measurements passed  
502 the isotope quality control protocol. During the week 10 labeling event, the final  $\text{CH}_4$  from each box  
503 before harvest had  $\delta^{13}\text{C}$  greater than +30‰, having increased from a pre-label median of -58‰. This  
504 strong enrichment signal shows that plant-derived carbon was converted to  $\text{CH}_4$  within days of being  
505 photosynthesized.



506

507 **Fig. 5** Both panels share the y-axis. (A) Isotopic enrichment of  $\text{CH}_4$  emissions over time since labeling for four planted boxes  
 508 during each labeling event. Red lines are median and interquartile range of unlabeled emissions. During both labeling events  
 509 emitted  $\text{CH}_4$  became substantially enriched. (B) Box and whisker plot of soil carbon isotopic enrichment. The unlabeled soil  
 510 includes both rhizosphere and bulk soil ( $n=68$ ) while the soil data from the Week 5 ( $n=43$ ) and Week 10 ( $n=69$ ) plant harvests is  
 511 all rhizosphere soil. The rhizosphere became substantially more enriched after the week 10 harvest

512 Roots from both labeling events were highly enriched (Table 2), and there was not a statistically  
 513 significant difference in root enrichment between the two labeling events ( $p > 0.05$ ), but roots from  
 514 labeled plants were significantly more enriched than those from unlabeled plants ( $p < 0.05$ ). Unlabeled  
 515 peat from the control boxes and simulated-plant boxes, and rhizosphere soil from unlabeled plants all  
 516 had statistically similar isotopic signatures ( $p > 0.05$ ), and thus we treated them collectively as  
 517 'unlabeled peat.' The unlabeled peat had a significantly lower  $^{13}\text{C}$  content ( $p < 0.01$ ) than rhizosphere  
 518 soil collected after the first and second labeling events, and rhizosphere soil collected after the first  
 519 labeling events had a significantly lower  $^{13}\text{C}$  content ( $p < 0.01$ ) than rhizosphere soil collected after the  
 520 second labeling event, as shown in Fig. 5. The first labeling event enriched carbon in the rhizosphere  
 521 from  $-26.6\text{\textperthousand}$  (the median  $\delta^{13}\text{C}$  of unlabeled peat, mean  $-26.8\text{\textperthousand}$ ) to a median of  $-18.9\text{\textperthousand}$  (mean  $-8.6\text{\textperthousand}$ ),  
 522 while the second labelling event enriched the rhizosphere even more effectively, bringing the median  
 523 enrichment up to  $-11.6\text{\textperthousand}$  (mean  $+30.8\text{\textperthousand}$ ) (Fig. 5B). The means had higher values than the medians for  
 524 both labeling events because the data were clustered low, with tails skewed high.

525

526      *Table 2, isotopic composition of roots from unlabeled plants and plants labeled during both events. Roots from both labeling*  
 527      *events are significantly different ( $p < 0.05$ ) than those from unlabeled plants, but data from the two labeling events are not*  
 528      *different from each other ( $p > 0.05$ ).*

Isotopic Composition of Roots	Unlabeled	Week 5 Labeling	Week 10 Labeling
Lower Quartile (‰)	-34.9	341.9	104.5
Median (‰)	-33.1	630.4	431.6
Upper Quartile (‰)	-32.4	1178.5	1981.0
Number of samples	9	53	38

529      **Porewater Dissolved Gases**

530      None of the treatments had a statistically significant change in dissolved  $\text{CH}_4$  concentration  
 531      between the initial porewater sampling event (week 2) and the final porewater sampling event (week 8)  
 532      (Table 3). Due to the small sample size for control boxes and boxes with simulated plants ( $n=3$  for each)  
 533      the statistical power of the Wilcoxon test used was low and it would have been unlikely to detect a small  
 534      change. Nevertheless, multiplying the largest change in concentration recorded in any box by the  
 535      volume of the box results in an increase of only 1.6 mg  $\text{CH}_4$  over the course of the experiment. If this  
 536      amount of  $\text{CH}_4$  were emitted, it would have increased emission by  $3.5 \text{ mg m}^{-2} \text{ d}^{-1}$ , a trivial amount  
 537      relative to the magnitude of the fluxes recorded. For comparison, control boxes emitted on the order of  
 538      one to ten  $\text{mg m}^{-2} \text{ d}^{-1}$ , while planted boxes emitted over  $200 \text{ mg m}^{-2} \text{ d}^{-1}$  during the second half of the  
 539      experiment (Fig. 3).

540      Given these results, for all boxes we could equate emitted  $\text{CH}_4$  with the net generation of  $\text{CH}_4$   
 541      (production minus oxidation) because a negligible amount of  $\text{CH}_4$  was stored within the soil during the  
 542      experiment.

543      *Table 3, porewater  $\text{CH}_4$  concentration changes over the course of the experiment show that there was negligible buildup of  $\text{CH}_4$*   
 544      *in the soil in any of the box types.*

Box Type	$\Delta \text{CH}_4$ Concentration Over Six Weeks ( $\mu\text{M}$ )		Increase in Flux if all $\text{CH}_4$ Were Emitted ( $\text{mg m}^{-2} \text{ d}^{-1}$ )		# of boxes
	Range	Median	Range	Median	
Control	-2 to 20	1	-0.3 to 3.5	0.2	3
Simulated	-1 to 10	0	-1.3 to 0.2	-0.7	3
Planted	-14 to 16	6	-2.4 to 2.7	1.1	6

545

546      **Isotope Mixing Model**

547      The results of the fractionation and exudate age modules which produced parameter  
 548      distributions are shown in the bottom half of Table 4. The fractionation effects in both planted and

549 unplanted boxes were similar, and within the range of values previously reported in the literature.  
 550 Previously conducted studies have examined the residence time of all root-derived carbon (Gaudinski et  
 551 al. 2000; Rasse et al. 2005), instead of separating exudate carbon from carbon that was sloughed off of  
 552 the root itself or part of a dead root. Here, we did separate exudates from other types of root-derived  
 553 carbon. Because of this difference, previous studies have reported values on the time-scale of several  
 554 years (Gaudinski et al. 2000) as opposed to the timescale of a couple months that we measured for root  
 555 exudates. Similarly, our delay period is not a variable that other studies have estimated. The residence  
 556 time of carbon in plants has been measured, but that includes carbon that is incorporated into plant  
 557 tissues and, as with residence time in the soil, can stretch into years as opposed to the scale of days  
 558 measured here (Gaudinski et al. 2000).

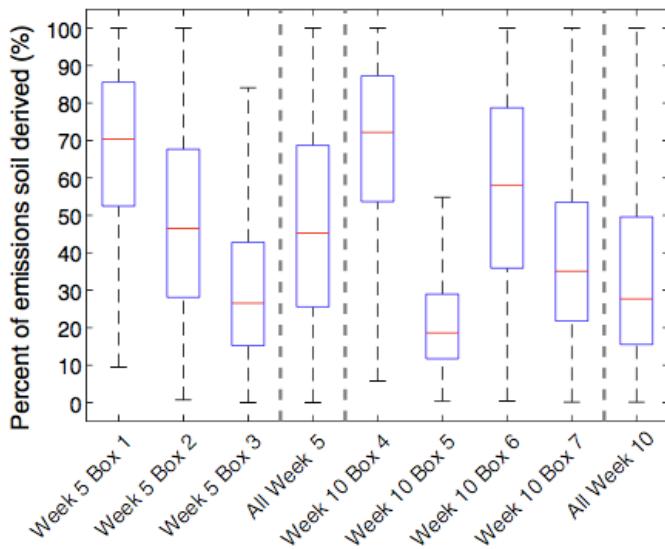
559  
 560 *Table 4, isotopic composition of various carbon pools, along with results of the fractionation module and exudate age module of*  
 561 *the mixing model. The fractionation module calculated fractionation factors based on measured  $\delta^{13}\text{C}$  values in unlabeled boxes.*  
 562 *The mixing module calculated the residence times of exudates in the soil and the delay between the end of labeling and peak*  
 563 *enrichment based on data from a test plant. The medians of all values calculated or measured in this study are reported with*  
 564 *interquartile ranges in parenthesis. The ranges for the literature values are the full range of values reported in those studies.*

Measured Values (‰)	Control Boxes	Number of control boxes/ samples	Planted Boxes	Number of planted boxes/ samples	Literature values
$\delta^{13}\text{C}$ of unlabeled soil ( $S_{\text{US}}$ )	-26.7 (-27.1 to -26.5)	3/29	-26.5 (-26.8 to -26.2)	1/6	-28 to - 24.5 <sup>a</sup>
$\delta^{13}\text{C}$ of unlabeled roots ( $S_{\text{UE}}$ )	n/a	n/a	-33.1 (-34.9 to -32.4)	1/9	-22 to -34 <sup>b</sup>
$\delta^{13}\text{C}$ of emitted unlabeled methane ( $M_{\text{U}}$ )	-58.2 (-61.3 to -56.7)	3/6	-61.0 (-62.3 to -57.8)	7/12	-70.3 to - 54.9 <sup>c</sup>
Calculated Values	Control Boxes and Simulated Plants	Number of results for CB & SP	Planted Boxes	Number of results for planted boxes	Literature values
Net apparent fractionation of $\text{CH}_4$ emission (F) (‰)	-31.6 (-34.4 to -29.9)	621	-27.1 (-29.6 to -24.2)	90	-75 to +14 <sup>d</sup>
Residence times of exudates in soil ( $\tau$ ) (days)	n/a	n/a	61.1 (44.7 to 77.3)	780	n/a
Delay between end of labeling and peak enrichment (D) (days)	n/a	n/a	2.6 (1.7 to 3.5)	780	n/a

565 a. (Chanton et al. 1992; Galand et al. 2010; Krohn et al. 2017)

566 b. (Bender, 1971; Nielsen et al., 2017; O'Leary, 1988)

567 c. (Chanton et al. 1992; Popp Trevor J. et al. 1999)  
 568 d. (Games et al. 1978; Gelwicks et al. 1994; Botz et al. 1996; Whiticar 1999; Valentine et al. 2004; Londry et  
 569 al. 2008; Feisthauer et al. 2011; Neumann et al. 2015) Net fractionation calculated by adding fractionation  
 570 effects of methanogenesis and methanotrophy reported in the various publications cited.

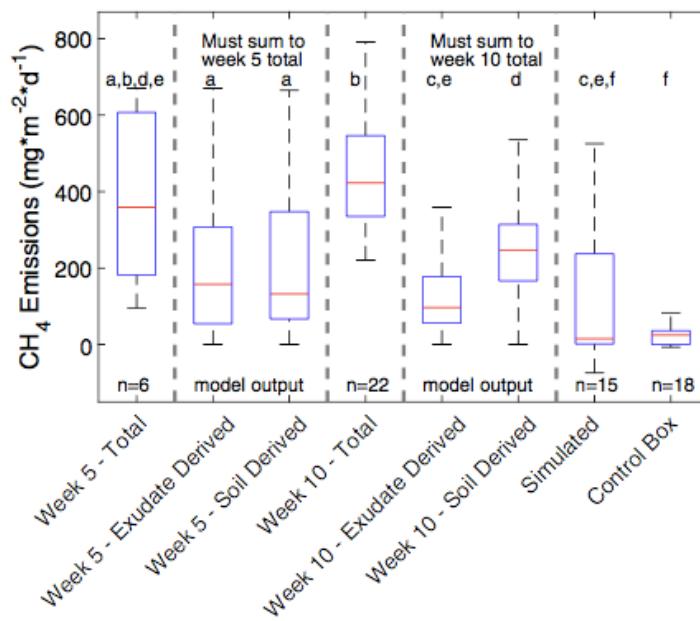


571  
 572 **Fig. 6** Results from isotope mixing model show the percentage of  $\text{CH}_4$  emitted from each box that was derived from root  
 573 exudates, with the remainder derived from soil carbon. There was no significant difference between the two labeling events

574 Both soil and root exudates were responsible for a large portion of total  $\text{CH}_4$  emissions in both  
 575 the first and second labeling events (Fig. 6). Note that only three boxes are presented from the week 5  
 576 labeling event. This is because one box harvested during that week did not produce any fluxes which  
 577 passed isotopic quality control, and therefore could not be used in the model. The median exudate-  
 578 derived portion of emitted  $\text{CH}_4$  from each box ranged from 22 to 68% in the week 5 labeling event with  
 579 an overall median for that labeling event of 43%. During the week 10 labeling event, the median  
 580 exudate-derived portion of  $\text{CH}_4$  emitted from each box ranged from 18 to 69%, with an overall median  
 581 for the labeling event of 25%. In both cases, all of the  $\text{CH}_4$  that was not exudate-derived was soil-  
 582 derived. In order to compare these values to the amount of  $\text{CH}_4$  emitted from simulated plants and  
 583 control boxes, we multiplied the percentage of  $\text{CH}_4$  derived from each source during each flux by the  
 584 total flux of  $\text{CH}_4$ . This converted the percentages given in Fig. 6 into the flux units shown in Fig. 7.

585 During both events, planted boxes emitted significantly more ( $p < 0.05$ ) soil-derived  $\text{CH}_4$  than  
 586 the simulated plants or control boxes. The median values for soil-derived  $\text{CH}_4$  emissions were  $359 \text{ mg m}^{-2} \text{ d}^{-1}$  for week five,  $423 \text{ mg m}^{-2} \text{ d}^{-1}$  for week ten,  $15 \text{ mg m}^{-2} \text{ d}^{-1}$  for simulated-plant boxes, and  $25 \text{ mg m}^{-2} \text{ d}^{-1}$   
 588 for control boxes. During week 10 the planted boxes emitted significantly more ( $p < 0.05$ ) soil-derived

589 CH<sub>4</sub> than exudate-derived CH<sub>4</sub>, but during week 5 the amount of CH<sub>4</sub> emitted from the two carbon  
 590 sources was not significantly different ( $p > 0.05$ ).



591  
 592 *Fig. 7 Box and whisker plots showing the total measured fluxes from planted boxes (Week 5 – Total and Week 10 – Total), the*  
 593 *portion of that total which was calculated to be from each carbon source (Soil Derived and Exudate Derived), and measured*  
 594 *emissions from the two unplanted box types (Simulated Plants and Control Boxes), which were entirely soil-derived. Outliers are*  
 595 *not shown. Letters indicate groups that are statistically similar ( $p > 0.05$ ). Any two boxes which do not share a letter are*  
 596 *significantly different ( $p < 0.05$ ) from each other.*

## 597 Discussion

### 598 *Biological and physical mechanisms for plants to increase CH<sub>4</sub> emission*

599 The planted boxes' greater emission of CH<sub>4</sub> than control boxes (Fig. 3) confirmed that plants can  
 600 increase CH<sub>4</sub> emissions, as has been shown previously (Shannon and White 1994; Joabsson et al. 1999;  
 601 Popp et al. 2000; Whalen 2005), but by itself does not explain the mechanism. The use of simulated  
 602 plants allowed for a direct comparison between a transport-alone scenario, where CH<sub>4</sub> and oxygen are  
 603 transported without plant use, and real plants, which were biologically active and also transported  
 604 gases. Gas transport in the simulated plants was clearly extensive, as demonstrated by oxygenation of  
 605 the soil (Fig. 5), but CH<sub>4</sub> emissions were not significantly greater than those from the control boxes (Figs.  
 606 3 & 7). This finding is consistent with what other studies using similar methodology have found. King et  
 607 al. (King et al. 1998) compared four treatments in a boreal wetland: natural plots with sedges and moss,  
 608 plots with the sedges removed, plots with the moss removed, and plots with sedges removed and  
 609 silicone tubes inserted. Turner et al. (under review) similarly utilized natural, sedge-removal, and  
 610 simulated-plant plots in a field study. Both experiments found that the natural plots had high CH<sub>4</sub>

611 emissions, while the sedge-removal and simulated-plant plots both had similar, low emissions (King et  
612 al. 1998, Turner et al. (under review)). The moss-removal plots in the King et al. (1998) study emitted as  
613 much CH<sub>4</sub> as the natural plots, indicating that it is the vascular plants which made the difference. Our  
614 results reinforce that gas transport alone does not increase emissions.

615 Plants do more than simply transport gas. They consume some oxygen within their root tissues  
616 (Armstrong 1971), and they also release root exudates which can affect oxygen dynamics in the soil by  
617 stimulating microbial respiration (Popp et al. 2000; Ding et al. 2004). The real plants in our experiment  
618 did not oxygenate the rhizosphere (Fig. 4) but they did greatly increase CH<sub>4</sub> emissions relative to the  
619 control boxes (Fig 4). The fact that the real and simulated plants were different in both regards indicates  
620 that the biological effects of the plants were key, and gas transport only increased emissions in  
621 conjunction with other root impacts. The importance of biological factors is further supported by the  
622 fact that the plants switched between a lower-emission state and a higher-emission state quickly (i.e.,  
623 within two weeks), while their height increased in a nearly linear fashion. This switch indicates that the  
624 emissions increase caused by the plants was not due to an effect that scales linearly with plant size, such  
625 as the gas transport capacity of aerenchyma. Instead, there must have been some change in the plants  
626 or rhizosphere which drove the increase in emissions. Examples of variables which could have changed  
627 to cause such an increase include the rate of root exudation, composition of root exudates (Girkin et al.  
628 2018b), or composition of the microbial community, which is dependent (among other things) on plant  
629 growth stage (Houlden et al. 2008).

630 Root exudates can increase CH<sub>4</sub> emissions through two effects: either increasing CH<sub>4</sub> production  
631 (Ström et al. 2003; Ström and Christensen 2007; Picek et al. 2007; Chanton et al. 2008; Kayranli et al.  
632 2009) or decreasing CH<sub>4</sub> oxidation (Popp et al. 2000). These effects can take place through multiple  
633 mechanisms including direct processing of root exudates into substrates for methanogenesis,  
634 stimulating the growth of heterotrophic microbes which compete with methanotrophs for electron  
635 acceptors such as oxygen, and/or triggering microbial priming which can include changes to the  
636 composition and size of the microbial community as well as changes to the soil chemical environment.  
637 All three of these mechanisms could have occurred in this study.

638

639 *Root exudate conversion to CH<sub>4</sub>*

640 Direct conversion of root exudates into CH<sub>4</sub> is the most straight-forward mechanism by which  
641 exudates can increase CH<sub>4</sub> emissions. The sharp rise in isotopic enrichment of CH<sub>4</sub> after labeling (Fig. 5A)  
642 showed that root exudates fueled methanogenesis, and the isotope model results (Fig. 7) show that

643 exudates were used in conjunction with soil carbon. Exudates did not diminish or replace use of soil  
644 carbon. The fact that the two labelling events produced different rhizosphere enrichments (Fig. 5B) from  
645 similarly enriched roots (Table 2) implies that the amount of exudates being emitted from the roots  
646 increased between the two labeling events, consistent with previous research showing root exudation is  
647 correlated with plant productivity (Weigel et al. 2005). The use of exudates to ultimately fuel CH<sub>4</sub>  
648 production provides an explanation for why CH<sub>4</sub> emissions are often correlated with primary  
649 productivity (Whiting and Chanton 1992).

650 However, the isotope model showed that while the emitted CH<sub>4</sub> was much more isotopically  
651 enriched than unlabeled emissions, it was not as enriched as it would have been if exudates were the  
652 primary carbon source (Fig. 7). The model attributed a large portion of emissions from planted boxes to  
653 soil-derived carbon. During week 10, more of the CH<sub>4</sub> was soil-derived than exudate-derived. In fact, the  
654 model showed that much more soil-derived CH<sub>4</sub> was emitted from planted boxes than from control or  
655 simulated boxes, so utilization of root exudates alone cannot fully explain the greater CH<sub>4</sub> emissions  
656 from planted boxes relative to the other treatments. There are two potential ways that plants could  
657 have increased emissions of soil-derived CH<sub>4</sub> in this experiment. They could have reduced  
658 methanotrophy, or they could have stimulated increased production of soil-derived CH<sub>4</sub>.

659 *Plants May Have Reduced Oxidation of Soil-Derived CH<sub>4</sub>*

660 Exudates may have increased emissions by decreasing methanotrophy through increased  
661 competition for oxygen (Lenzewski et al. 2018). This effect is a synergy between root exudation and the  
662 gas transport effects of aerenchyma. There is significant existing literature showing that plant transport  
663 of CH<sub>4</sub> is substantial and can increase total emissions (Shannon and White 1994; Joabsson et al. 1999;  
664 Popp et al. 2000; Whalen 2005). Other studies have found that real plants can oxygenate the soil and  
665 oxidize CH<sub>4</sub> to CO<sub>2</sub> (Schipper and Reddy 1996; Fritz et al. 2011; Lenzewski et al. 2018), and in some cases  
666 soil oxygenation is extensive enough that oxidation reduces CH<sub>4</sub> emissions to zero (Fritz et al. 2011). Gas  
667 transport by plants can therefore increase CH<sub>4</sub> oxidation when it creates an oxic zone around roots, or  
668 decrease CH<sub>4</sub> oxidation when there is no oxic zone around roots and the aerenchyma allow CH<sub>4</sub> to  
669 diffuse from the anoxic rhizosphere directly to the atmosphere. Whether the rhizosphere is oxygenated  
670 or not depends on the balance between oxygen supply and demand, which is dictated by the biology of  
671 the plant species in question, as well as by carbon bioavailability and microbial ecology of the soil. The  
672 lack of rhizosphere oxygen in our system (Fig. 4) is consistent with a field study at the site where  
673 materials for this experiment were sourced, which utilized field-deployed optodes to study belowground  
674 oxygen dynamics (Turner et al., under review), and with previous work showing that *Carex aquatilis*

675 transport gases less than other aerenchymatous sedges (Schimel 1995). We did not directly test rates of  
676 methanotrophy in this study, but the lack of oxygen in the rhizosphere of the real plants and readily  
677 available oxygen surrounding tubes of simulated plants indicates that real plants likely reduced  
678 methanotrophy relative to simulated plants. Reducing methanotrophy can take multiple forms however.  
679 The utilization of oxygen can take place either within roots or in the soil, and competition for oxygen can  
680 occur either between microbial species or between metabolisms within facultative methanotrophs.

681 More oxygen is utilized along the transport pathway in plants than in abiotic tubes because the  
682 roots themselves need oxygen for respiration (Armstrong 1971). If this oxygen demand is strong enough  
683 relative to the transport capacity of the aerenchyma, the roots could effectively become selective gas-  
684 transport pathways where  $\text{CH}_4$  can diffuse to the atmosphere but  $\text{O}_2$  cannot diffuse to the soil. There is  
685 also another mechanism for increasing oxygen utilization: increasing demand in the soil through root  
686 exudation of carbon which stimulates aerobic microbial activity (Mueller et al. 2016; Lenzewski et al.  
687 2018). The isotopic enrichment of  $\text{CH}_4$  after labeling indicates that microbes did utilize carbon derived  
688 from root exudates. While this result indicates anaerobic metabolism, any available oxygen would have  
689 allowed microbes to also aerobically metabolize root exudates, thus competing with methanotrophy for  
690 limited oxygen.

691 Note that we have framed this discussion as competition for oxygen between methanotrophy  
692 and other metabolisms. While most species of methanotrophs are obligate methanotrophs (Conrad  
693 2009), and therefore would be competing for oxygen against other aerobic microbes, there do exist  
694 facultative methanotrophs which can utilize multiple carbon compounds (Dedysh et al. 2005). The list of  
695 other carbon compounds which these facultative methanotrophs can utilize is limited (Dedysh et al.  
696 2005; Crombie and Murrell 2014), but when available, the microbes preferentially use the non-methane  
697 compounds (Theisen et al. 2005; Wieczorek et al. 2011). It has also been suggested that organic acids  
698 may inhibit methanotrophy through toxic as well as competitive effects (Wieczorek et al. 2011). In the  
699 oxygen-limited rhizosphere of this study, it is possible that the addition of root exudates triggered  
700 facultative methanotrophs to utilize non-methane substrates and therefore oxidize less  $\text{CH}_4$ , in addition  
701 to the inter-species competition for oxygen which likely occurred.

702 However, we reason that the combination of reduced methanotrophy and utilization of  
703 exudates was not enough to account for the full increase in  $\text{CH}_4$  emissions caused by plants relative to  
704 the other treatments. A study at the field site from which peat for this experiment was obtained directly  
705 investigated  $\text{CH}_4$  oxidation by measuring  $\text{CH}_4$  emissions under anaerobic conditions, achieved by flushing  
706 the fluxing chamber headspace with  $\text{N}_2$  (Turner et al., under review). That study found that a maximum

707 of 36% of produced  $\text{CH}_4$  was oxidized prior to emission in plots without vascular plants, regardless of  
708 whether they had simulated plants or not (Turner et al., under review). Assuming similar rates of  
709 methanotrophy in our unplanted boxes, the simulated plants and control boxes would have produced a  
710 median of 31 and 38  $\text{mg CH}_4 \text{ m}^{-2} \text{ d}^{-1}$  respectively, still an order of magnitude less than the  $237 \text{ mg m}^{-2} \text{ d}^{-1}$   
711 of soil-derived  $\text{CH}_4$  that was emitted from planted boxes (median of all data). This 88% excess soil-  
712 derived  $\text{CH}_4$  emitted by planted boxes relative to other treatments could only have been the result of  
713 roots triggering increased use of soil carbon.

714

715 *Root Effects on Utilization of Soil Carbon*

716 There are two mechanisms by which roots may enhance anaerobic soil carbon processing: the  
717 creation of mixed-redox zones or microbial priming. The mixed-redox mechanism is a result of  
718 transitions in either time or space between oxic and anoxic environments (Canfield 1994; Aller 1998;  
719 Chanton et al. 2008). In these transitional regions, oxygen may be quickly utilized and so not be present,  
720 (as was the case in the rhizosphere in this study (Fig. 4)), but still have the effect of raising the overall  
721 redox state of the location and facilitating the creation of alternate electron acceptors (Keiluweit et al.  
722 2016). If these additional electron acceptors were used to fully mineralize soil carbon to  $\text{CO}_2$  then we  
723 would see no increase in  $\text{CH}_4$  production. However, it has also been found that in mixed-redox zones,  
724 oxygen and other thermodynamically favorable electron acceptors are used to partially oxidize large  
725 molecules, and the resulting smaller molecules are further processed anaerobically (Chanton et al. 2008;  
726 Corbett et al. 2015). If such partial processing occurred in the rhizosphere of our study, then it could  
727 explain some of the increased microbial conversion of soil carbon to  $\text{CH}_4$ .

728 However, oxygen visualization around the simulated plants demonstrate that partial processing  
729 of soil organic matter was likely only a minor effect in this study. Mixed redox zones could have existed  
730 around the oxygenated soil surrounding the simulated plants (Fig. 4). The fact that  $\text{CH}_4$  emissions from  
731 simulated-plant boxes were not significantly greater than control boxes (Figs. 3 & 7), paired with  
732 evidence from the field investigation that rates of methanotrophy were similar between unplanted field  
733 plots with and without simulated plants (Turner et al., under review), indicates that any mixed-redox  
734 zone surrounding the simulated plants did not substantially increase  $\text{CH}_4$  production. Lacking an  
735 explanation for why the mixed-redox zones would have been larger or more influential around real roots  
736 than around simulated plants, we conclude that priming is the mechanism that better explains the  
737 enhanced  $\text{CH}_4$  emissions.

738 Priming is a broad term encompassing any process by which the addition of a different carbon  
739 source (e.g., root exudates or leaf litter) increases microbial utilization of soil carbon. Priming can  
740 happen by stimulating an increase in microbial biomass, a change in composition of the microbial  
741 community, and/or a change in what metabolisms are active within the microbial community (Craine et  
742 al. 2007; Kuzyakov 2010; Ruirui et al. 2014). Priming has been observed in methanogenic systems  
743 before, such as one incubation experiment which found that the addition of rice straw to peat soils  
744 greatly increased soil-derived CH<sub>4</sub> production (Ye et al. 2015). In our experiment, priming provides a  
745 straightforward explanation of why real plants emitted such a large amount of soil-derived CH<sub>4</sub> relative  
746 to the simulated plants (Figs. 3 & 7). Notably, during the week 10 labeling event, more CH<sub>4</sub> was soil-  
747 derived than exudate-derived (Figs. 6 & 7), indicating a notable priming effect. During the week 5 event  
748 there was still priming, but the soil-derived and exudate-derived portions were about equal. Based on  
749 rhizosphere enrichment data (Fig. 5), we know that there was more root exudation happening during  
750 week 10. Together, the data indicate that more root exudation led to a greater portion of the total  
751 emissions being soil-derived (Figs. 6 & 7).

752 If the priming effect is equally powerful in natural systems, it could have profound  
753 environmental impacts. A field study in a Boreal peatland, similar to the one upon which this study is  
754 based, found that vascular plants contributed to increased microbial processing of soil organic matter,  
755 but did not directly link that activity to CH<sub>4</sub> (Robroek et al. 2016). In our laboratory study we have shown  
756 that CH<sub>4</sub> was generated from soil carbon that was processed because of plants. Furthermore, because  
757 root exudation and wetland CH<sub>4</sub> production are correlated with plant productivity (Whiting and Chanton  
758 1992; Weigel et al. 2005), our findings imply that as plant productivity in temperate northern latitudes  
759 increases with climate change (Forkel et al. 2016), so will the conversion of soil carbon to CH<sub>4</sub>. It has long  
760 been known that CH<sub>4</sub> emissions are a positive feedback on climate-driven productivity changes  
761 (Megonigal and Schlesinger 1997), but wetlands have also been counted on as a carbon sink (Kayranli et  
762 al. 2009). Identifying peat soils as the source of the additionally emitted CH<sub>4</sub> implies that this ability to  
763 sequester carbon may not be as robust as believed. Northern peatlands contain 25% of the world's  
764 soil carbon (500 Gt C), despite only covering 2% of surface area (Yu 2012). For context, forecasts  
765 of global wetland CH<sub>4</sub> emissions by 2100 are 0.17 to 0.25 Gt G (Zhang et al. 2017), so the  
766 potential for peatland carbon to increase emissions is limited only by the rate at which it is  
767 converted. Further research into how priming could act as a positive climate feedback is warranted.

768        Another area of needed research highlighted by this study is an examination into why we found  
769        evidence of priming while other recent studies have failed to find evidence for it (Girkin et al. 2018a) or  
770        found evidence that root exudates protect soil carbon, which is the opposite of priming (Graham et al.  
771        2017). The data presented here are inadequate to explain why or when priming does or does not occur.  
772        One explanation could be that priming only occurs in certain circumstances depending on the nutritional  
773        needs of microbes (Craine et al. 2007), soil properties (Jones et al. 2003), or other environmental  
774        variables. It is also possible that the definition of priming used in each study causes discrepancies. For  
775        example, Graham et al. (2017) determined that in the presence of plants microbial processing of  
776        dissolved organic carbon (DOC) increased but processing of physically bound carbon decreased. This  
777        approach does not distinguish plant-derived DOC from soil-derived DOC, and so is answering a different  
778        question than that addressed in this study. In order to elucidate the mechanisms of priming we must  
779        understand what microbes are involved, and what types of soil-derived molecules are additionally  
780        utilized during priming.

781        **Conclusions**

782        In our experiment root exudates drove CH<sub>4</sub> production, but not through direct utilization alone.  
783        An isotope mixing model showed that a large portion of the increase in CH<sub>4</sub> emissions caused by plants  
784        was fueled by soil-derived carbon. Our data indicate root exudates increased the amount of soil-derived  
785        CH<sub>4</sub> that was emitted by increasing O<sub>2</sub> demand and thus reducing methanotrophy and/or increasing CH<sub>4</sub>  
786        production from soil-derived carbon by stimulating microbial priming. We did not measure  
787        methanotrophy directly, but its reduction in planted boxes relative to the simulated plants was  
788        supported by the lack of oxygen detected in the rhizosphere. The simulated plants we used for  
789        comparison did not emit significantly more CH<sub>4</sub> than control boxes and did have large oxygenated zones  
790        around their roots. The comparison between the real and simulated plants shows that transport of  
791        gases alone cannot increase CH<sub>4</sub> emissions without other biological activity. However, the amount of  
792        soil-derived CH<sub>4</sub> emitted from the planted boxes was an order of magnitude greater than what was likely  
793        ever produced in unplanted boxes, based on prior estimates of methanotrophy. Increased production of  
794        soil-derived CH<sub>4</sub> was the only explanation for the unattributed carbon that was emitted from planted  
795        boxes, which comprised over half of the total emissions from those boxes. That increase in production is  
796        best explained by microbial priming. An order of magnitude increase in conversion of soil carbon to CH<sub>4</sub>  
797        was driven by plant growth, which is projected to increase in the boreal region under forecasted climate  
798        conditions. The presence of such a large priming effect implies that increased plant productivity could  
799        potentially lead to increased conversion of soil carbon to CH<sub>4</sub> on climatically relevant scales.

800 **References**

801 Aller RC (1998) Mobile deltaic and continental shelf muds as suboxic, fluidized bed reactors. *Mar Chem*  
802 61:143–155 . doi: 10.1016/S0304-4203(98)00024-3

803 Armstrong W (1971) Radial Oxygen Losses from Intact Rice Roots as Affected by Distance from the  
804 Apex, Respiration and Waterlogging. *Physiol Plant* 25:192–197 . doi: 10.1111/j.1399-  
805 3054.1971.tb01427.x

806 Aselmann I, Crutzen PJ (1989) Global distribution of natural freshwater wetlands and rice paddies, their  
807 net primary productivity, seasonality and possible methane emissions. *J Atmospheric Chem*  
808 8:307–358 . doi: 10.1007/BF00052709

809 Basiliko N, Stewart H, Roulet NT, Moore TR (2012) Do Root Exudates Enhance Peat Decomposition?  
810 *Geomicrobiol J* 29:374–378 . doi: 10.1080/01490451.2011.568272

811 Botz R, Pokojski H-D, Schmitt M, Thomm M (1996) Carbon isotope fractionation during bacterial  
812 methanogenesis by CO<sub>2</sub> reduction. *Org Geochem* 25:255–262 . doi: 10.1016/S0146-  
813 6380(96)00129-5

814 Canfield DE (1994) Factors influencing organic carbon preservation in marine sediments. *Chem Geol*  
815 114:315–329 . doi: 10.1016/0009-2541(94)90061-2

816 Carvalhais LC, Dennis PG, Fedoseyenko D, Hajirezaei M-R, Borriss R, Wirén N von (2011) Root  
817 exudation of sugars, amino acids, and organic acids by maize as affected by nitrogen, phosphorus,  
818 potassium, and iron deficiency. *J Plant Nutr Soil Sci* 174:3–11 . doi: 10.1002/jpln.201000085

819 Chanton JP, Glaser PH, L. S. Chasar, D. J. Burdige, Hines M. E., D. I. Siegel, L. B. Tremblay, W. T.  
820 Cooper (2008) Radiocarbon evidence for the importance of surface vegetation on fermentation  
821 and methanogenesis in contrasting types of boreal peatlands. *Glob Biogeochem Cycles* 22: . doi:  
822 10.1029/2008GB003274

823 Chanton JP, Martens CS, Kelley CA, Crill PM, Showers WJ (1992) Methane transport mechanisms and  
824 isotopic fractionation in emergent macrophytes of an Alaskan tundra lake. *J Geophys Res* 97: .  
825 doi: 10.1029/90JD01542

826 Ciais P, Sabine C, Bala G, Bopp L, Brovkin V, Canadell J, Chhabra A, DeFries R, Galloway J, Heimann  
827 M, Jones C, Le Quéré C, Myneni RB, Piao S, Thornton P (2013) Carbon and Other  
828 Biogeochemical Cycles. In: Stocker TF, Qin D, Plattner G-K, Tignor M, Allen SK, Boschung J,  
829 Nauels A, Xia Y, Bex V, Midgley PM (eds) *Climate Change 2013: The Physical Science Basis.*  
830 Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel  
831 on Climate Change. Cambridge University Press, Cambridge, United Kingdom and New York,  
832 NY, USA, pp 465–570

833 Conrad R (2009) The global methane cycle: recent advances in understanding the microbial processes  
834 involved. *Environ Microbiol Rep* 1:285–292 . doi: 10.1111/j.1758-2229.2009.00038.x

835 Coplen TB, Brand WA, Gehre M, Gröning M, Meijer HAJ, Toman B, Verkouteren RM (2006) New  
836 Guidelines for  $\delta^{13}\text{C}$  Measurements. *Anal Chem* 78:2439–2441 . doi: 10.1021/ac052027c

837 Corbett JE, Tfaily MM, Burdige DJ, Glaser PH, Chanton JP (2015) The relative importance of  
838 methanogenesis in the decomposition of organic matter in northern peatlands. *J Geophys Res*  
839 *Biogeosciences* 120:280–293 . doi: 10.1002/2014JG002797

840 Craine JM, Morrow C, Fierer N (2007) Microbial nitrogen limitation increases decomposition. *Ecology*  
841 88:2105–2113 . doi: 10.1890/06-1847.1

842 Crombie AT, Murrell JC (2014) Trace-gas metabolic versatility of the facultative methanotroph  
843 *Methylocella silvestris* . *Nature* 510:148–151 . doi: 10.1038/nature13192

844 Dedysh SN, Knief C, Dunfield PF (2005) Methylocella Species Are Facultatively Methanotrophic. *J*  
845 *Bacteriol* 187:4665–4670 . doi: 10.1128/JB.187.13.4665-4670.2005

846 Ding W, Cai Z, Tsuruta H (2004) Summertime variation of methane oxidation in the rhizosphere of a  
847 Carex dominated freshwater marsh. *Atmos Environ* 38:4165–4173 . doi:  
848 10.1016/j.atmosenv.2004.04.022

849 Dorodnikov M, Knorr K-H, Kuzyakov Y, Wilmking M (2011) Plant-mediated CH<sub>4</sub> transport and  
850 contribution of photosynthates to methanogenesis at a boreal mire: a <sup>14</sup>C pulse-labeling study.  
851 *Biogeosciences* 8:2365–2375 . doi: 10.5194/bg-8-2365-2011

852 Feisthauer S, Vogt C, Modrzynski J, Szlenkier M, Krüger M, Siegert M, Richnow H-H (2011) Different  
853 types of methane monooxygenases produce similar carbon and hydrogen isotope fractionation  
854 patterns during methane oxidation. *Geochim Cosmochim Acta* 75:1173–1184 . doi:  
855 10.1016/j.gca.2010.12.006

856 Fetzer S, Conrad R (1993) Effect of redox potential on methanogenesis by *Methanosarcina barkeri*. *Arch*  
857 *Microbiol* 160:108–113 . doi: 10.1007/BF00288711

858 Forkel M, Carvalhais N, Rödenbeck C, Keeling R, Heimann M, Thonicke K, Zaehle S, Reichstein M  
859 (2016) Enhanced seasonal CO<sub>2</sub> exchange caused by amplified plant productivity in northern  
860 ecosystems. *Science* 351:696–699 . doi: 10.1126/science.aac4971

861 Fritz C, Pancotto VA, Elzenga JTM, Visser EJW, Grootjans AP, Pol A, Iturraspe R, Roelofs JGM,  
862 Smolders AJP (2011) Zero methane emission bogs: extreme rhizosphere oxygenation by cushion  
863 plants in Patagonia. *New Phytol* 190:398–408 . doi: 10.1111/j.1469-8137.2010.03604.x

864 Galand P, Yrjälä K, Conrad R (2010) Stable carbon isotope fractionation during methanogenesis in three  
865 boreal peatland ecosystems

866 Games LM, HayesRobert JM, Gunsalus P (1978) Methane-producing bacteria: natural fractionations of  
867 the stable carbon isotopes. *Geochim Cosmochim Acta* 42:1295–1297 . doi: 10.1016/0016-  
868 7037(78)90123-0

869 Gaudinski JB, Trumbore SE, Davidson EA, Zheng S (2000) Soil carbon cycling in a temperate forest:  
870 radiocarbon-based estimates of residence times, sequestration rates and partitioning of fluxes.  
871 *Biogeochemistry* 51:33–69 . doi: 10.1023/A:1006301010014

872 Gedney N, Cox PM, Huntingford C (2004) Climate feedback from wetland methane emissions. *Geophys*  
873 *Res Lett* 31:L20503 . doi: 10.1029/2004GL020919

874 Gelwicks JT, Risatti JB, Hayes JM (1994) Carbon isotope effects associated with aceticlastic  
875 methanogenesis. *Appl Environ Microbiol* 60:467–472

876 Girkin NT, Turner BL, Ostle N, Craigon J, Sjögersten S (2018a) Root exudate analogues accelerate CO<sub>2</sub>  
877 and CH<sub>4</sub> production in tropical peat. *Soil Biol Biochem* 117:48–55 . doi:  
878 10.1016/j.soilbio.2017.11.008

879 Girkin NT, Turner BL, Ostle N, Sjögersten S (2018b) Composition and concentration of root exudate  
880 analogues regulate greenhouse gas fluxes from tropical peat. *Soil Biol Biochem* 127:280–285 .  
881 doi: 10.1016/j.soilbio.2018.09.033

882 Graham EB, Tfaily MM, Crump AR, Goldman AE, Bramer LM, Arntzen E, Romero E, Resch CT,  
883 Kennedy DW, Stegen JC (2017) Carbon Inputs From Riparian Vegetation Limit Oxidation of  
884 Physically Bound Organic Carbon Via Biochemical and Thermodynamic Processes. *J Geophys  
885 Res Biogeosciences* 122:3188–3205 . doi: 10.1002/2017JG003967

886 Hamer U, Marschner B (2002) Priming effects of sugars, amino acids, organic acids and catechol on the  
887 mineralization of lignin and peat. *J Plant Nutr Soil Sci* 165:261–268 . doi: 10.1002/1522-  
888 2624(200206)165:3<261::AID-JPLN261>3.0.CO;2-1

889 Hodson EL, Poulter B, Zimmermann NE, Prigent C, Kaplan JO (2011) The El Niño–Southern Oscillation  
890 and wetland methane interannual variability. *Geophys Res Lett* 38:L08810 . doi:  
891 10.1029/2011GL046861

892 Houlden A, Timms-Wilson TM, Day MJ, Bailey MJ (2008) Influence of plant developmental stage on  
893 microbial community structure and activity in the rhizosphere of three field crops. *FEMS  
894 Microbiol Ecol* 65:193–201 . doi: 10.1111/j.1574-6941.2008.00535.x

895 Idso SB, Kimball BA, Anderson MG, J.R. Mauney (1987) Effects of atmospheric CO<sub>2</sub> enrichment on  
896 plant growth: the interactive role of air temperature. *Agric Ecosyst Environ* 20:1–10 . doi:  
897 10.1016/0167-8809(87)90023-5

898 Intergovernmental Panel on Climate Change (ed) (2014) Climate Change 2013 - The Physical Science  
899 Basis: Working Group I Contribution to the Fifth Assessment Report of the Intergovernmental  
900 Panel on Climate Change. Cambridge University Press, Cambridge

901 Joabsson A, Christensen TR, Wallén B (1999) Vascular plant controls on methane emissions from  
902 northern peatforming wetlands. *Trends Ecol Evol* 14:385–388 . doi: 10.1016/S0169-  
903 5347(99)01649-3

904 Jonasson S, Lee JA, Callaghan TV, Havström M, Parsons AN (1996) Direct and Indirect Effects of  
905 Increasing Temperatures on Subarctic Ecosystems. *Ecol Bull* 180–191

906 Jones DL, Dennis PG, Owen AG, van Hees PAW (2003) Organic acid behavior in soils – misconceptions  
907 and knowledge gaps. *Plant Soil* 248:31–41 . doi: 10.1023/A:1022304332313

908 Kayranli B, Scholz M, Mustafa A, Hedmark Å (2009) Carbon Storage and Fluxes within Freshwater  
909 Wetlands: a Critical Review. *Wetlands* 30:111–124 . doi: 10.1007/s13157-009-0003-4

910 Keiluweit M, Nico PS, Kleber M, Fendorf S (2016) Are oxygen limitations under recognized regulators  
911 of organic carbon turnover in upland soils? *Biogeochemistry* 127:157–171 . doi: 10.1007/s10533-  
912 015-0180-6

913 King JY, Reeburgh WS, Regli SK (1998) Methane emission and transport by arctic sedges in Alaska:  
914 Results of a vegetation removal experiment. *J Geophys Res Atmospheres* 103:29083–29092 . doi:  
915 10.1029/98JD00052

916 Krohn J, Lozanovska I, Kuzyakov Y, Parvin S, Dorodnikov M (2017) CH<sub>4</sub> and CO<sub>2</sub> production below  
917 two contrasting peatland micro-relief forms: An inhibitor and δ<sup>13</sup>C study. *Sci Total Environ*  
918 586:142–151 . doi: 10.1016/j.scitotenv.2017.01.192

919 Kummerow J, Ellis BA (1984) Temperature effect on biomass production and root/shoot biomass ratios  
920 in two arctic sedges under controlled environmental conditions. *Can J Bot* 62:2150–2153 . doi:  
921 10.1139/b84-294

922 Kuzyakov Y (2010) Priming effects: Interactions between living and dead organic matter. *Soil Biol*  
923 *Biochem* 42:1363–1371 . doi: 10.1016/j.soilbio.2010.04.003

924 Larsen M, Borisov SM, Grunwald B, Klimant I, Glud RN (2011) A simple and inexpensive high  
925 resolution color ratiometric planar optode imaging approach: application to oxygen and pH  
926 sensing. *Limnol Oceanogr Methods* 9:348–360 . doi: 10.4319/lom.2011.9.348

927 Lehmeier CA, Ballantyne F, Min K, Billings SA (2016) Temperature-mediated changes in microbial  
928 carbon use efficiency and <sup>13</sup>C discrimination. doi: 10.5194/bg-13-3319-2016

929 Lenzewski N, Mueller P, Meier RJ, Liebsch G, Jensen K, Koop Jakobsen K (2018) Dynamics of oxygen  
930 and carbon dioxide in rhizospheres of *Lobelia dortmanna* – a planar optode study of belowground  
931 gas exchange between plants and sediment. *New Phytol* 218:131–141 . doi: 10.1111/nph.14973

932 Londry KL, Dawson KG, Grover HD, Summons RE, Bradley AS (2008) Stable carbon isotope  
933 fractionation between substrates and products of *Methanosarcina barkeri*. *Org Geochem* 39:608–  
934 621 . doi: 10.1016/j.orggeochem.2008.03.002

935 Lu Y, Conrad R (2005) In Situ Stable Isotope Probing of Methanogenic Archaea in the Rice Rhizosphere.  
936 *Science* 309:1088–1090 . doi: 10.1126/science.1113435

937 Manies KL, Fuller CC, Jones MC, Waldrop MP, McGeehin JP (2017) Soil data for a thermokarst bog and  
938 the surrounding permafrost plateau forest, located at Bonanza Creek Long Term Ecological  
939 Research Site, Interior Alaska. U.S. Geological Survey, Reston, VA

940 Megonigal JP, Schlesinger WH (1997) Enhanced CH<sub>4</sub> emission from a wetland soil exposed to Elevated  
941 CO<sub>2</sub>. *Biogeochemistry* 37:77–88 . doi: 10.1023/A:1005738102545

942 Megonigal JP, Whalen SC, Tissue DT, Bovard BD, Allen AS, Albert DB (1999) A Plant-Soil-  
943 Atmosphere Microcosm for Tracing Radiocarbon from Photosynthesis through Methanogenesis.  
944 *Soil Sci Soc Am J* 63:665–671 . doi: 10.2136/sssaj1999.03615995006300030033x

945 Mueller P, Jensen K, Megonigal JP (2016) Plants mediate soil organic matter decomposition in response  
946 to sea level rise. *Glob Change Biol* 22:404–414 . doi: 10.1111/gcb.13082

947 Neumann RB, Blazewicz SJ, Conaway CH, Turetsky MR, Waldrop MP (2015) Modeling CH<sub>4</sub> and CO<sub>2</sub>  
948 cycling using porewater stable isotopes in a thermokarst bog in Interior Alaska: results from three  
949 conceptual reaction networks. *Biogeochemistry* 127:57–87 . doi: 10.1007/s10533-015-0168-2

950 Picek T, Čížková H, Dušek J (2007) Greenhouse gas emissions from a constructed wetland—Plants as  
951 important sources of carbon. *Ecol Eng* 31:98–106 . doi: 10.1016/j.ecoleng.2007.06.008

952 Popp TJ, Chanton JP, Whiting GJ, Grant N (2000) Evaluation of methane oxidation in the rhizosphere of  
953 a Carex dominated fen in northcentral Alberta, Canada. *Biogeochemistry* 51:259–281 . doi:  
954 10.1023/A:1006452609284

955 Popp Trevor J., Chanton Jeffrey P., Whiting Gary J., Grant Nick (1999) Methane stable isotope  
956 distribution at a Carex dominated fen in north central Alberta. *Glob Biogeochem Cycles*  
957 13:1063–1077 . doi: 10.1029/1999GB900060

958 Rasse DP, Rumpel C, Dignac M-F (2005) Is soil carbon mostly root carbon? Mechanisms for a specific  
959 stabilisation. *Plant Soil* 269:341–356 . doi: 10.1007/s11104-004-0907-y

960 Riley WJ, Subin ZM, Lawrence DM, Swenson SC, Torn MS, Meng L, Mahowald NM, Hess P (2011)  
961 Barriers to predicting changes in global terrestrial methane fluxes: analyses using CLM4Me, a  
962 methane biogeochemistry model integrated in CESM. *Biogeosciences* 8:1925–1953 . doi:  
963 10.5194/bg-8-1925-2011

964 Robroek BJM, Albrecht RJH, Hamard S, Pulgarin A, Bragazza L, Buttler A, Jassey VE (2016) Peatland  
965 vascular plant functional types affect dissolved organic matter chemistry. *Plant Soil* 407:135–  
966 143 . doi: 10.1007/s11104-015-2710-3

967 Ruirui C, Mehmet S, Sergey B, Olga M, Klaus D, Xiangui L, Evgenia B, Yakov K (2014) Soil C and N  
968 availability determine the priming effect: microbial N mining and stoichiometric decomposition  
969 theories. *Glob Change Biol* 20:2356–2367 . doi: 10.1111/gcb.12475

970 Schimel JP (1995) Plant transport and methane production as controls on methane flux from arctic wet  
971 meadow tundra. *Biogeochemistry* 28:183–200 . doi: 10.1007/BF02186458

972 Schipper LA, Reddy KR (1996) Determination of Methane Oxidation in the Rhizosphere of *Sagittaria*  
973 *lancifolia* Using Methyl Fluoride. *Soil Sci Soc Am J* 60:611–616 . doi:  
974 10.2136/sssaj1996.03615995006000020039x

975 Segers R, Leffelaar PA (2001) Modeling methane fluxes in wetlands with gas-transporting plants: 1.  
976 Single-root scale. *J Geophys Res Atmospheres* 106:3511–3528 . doi: 10.1029/2000JD900484

977 Shannon RD, White JR (1994) A three-year study of controls on methane emissions from two Michigan  
978 peatlands. *Biogeochemistry* 27:35–60 . doi: 10.1007/BF00002570

979 Ström L, Christensen TR (2007) Below ground carbon turnover and greenhouse gas exchanges in a sub-  
980 arctic wetland. *Soil Biol Biochem* 39:1689–1698 . doi: 10.1016/j.soilbio.2007.01.019

981 Ström L, Ekberg A, Mastepanov M, Christensen TR (2003) The effect of vascular plants on carbon  
982 turnover and methane emissions from a tundra wetland. *Glob Change Biol* 9:1185–1192 . doi:  
983 10.1046/j.1365-2486.2003.00655.x

984 Theisen AR, Ali MH, Radajewski S, Dumont MG, Dunfield PF, McDonald IR, Dedysh SN, Miguez CB,  
985 Murrell JC (2005) Regulation of methane oxidation in the facultative methanotroph *Methylocella*  
986 *silvestris* BL2. *Mol Microbiol* 58:682–692 . doi: 10.1111/j.1365-2958.2005.04861.x

987 Trinder CJ, Artz RRE, Johnson D (2008) Contribution of plant photosynthate to soil respiration and  
988 dissolved organic carbon in a naturally recolonising cutover peatland. *Soil Biol Biochem*  
989 40:1622–1628 . doi: 10.1016/j.soilbio.2008.01.016

990 Valentine DL, Chidthaisong A, Rice A, Reeburgh WS, Tyler SC (2004) Carbon and hydrogen isotope  
991 fractionation by moderately thermophilic methanogens 11 Associate editor: N. E. Ostrom.  
992 *Geochim Cosmochim Acta* 68:1571–1590 . doi: 10.1016/j.gca.2003.10.012

993 Wang Z-P, Han X-G (2005) Diurnal variation in methane emissions in relation to plants and  
994 environmental variables in the Inner Mongolia marshes. *Atmos Environ* 39:6295–6305 . doi:  
995 10.1016/j.atmosenv.2005.07.010

996 Weigel HJ, Pacholski A, Burkart S, Helal M, Heinemeyer O, Kleikamp B, Manderscheid R, Fruhauf C,  
997 Hendrey GF, Lewin K, Nagy J (2005) Carbon Turnover in a Crop Rotation Under Free Air CO<sub>2</sub>  
998 Enrichment (FACE). *Pedosphere* 15:728–738

999 Whalen S c. (2005) Biogeochemistry of Methane Exchange between Natural Wetlands and the  
1000 Atmosphere. *Environ Eng Sci* 22:73–94 . doi: 10.1089/ees.2005.22.73

1001 Whiticar MJ (1999) Carbon and hydrogen isotope systematics of bacterial formation and oxidation of  
1002 methane. *Chem Geol* 161:291–314 . doi: 10.1016/S0009-2541(99)00092-3

1003 Whiting GJ, Chanton JP (1992) Plant-dependent CH<sub>4</sub> emission in a subarctic Canadian fen. *Glob*  
1004 *Biogeochem Cycles* 6:225–231 . doi: 10.1029/92GB00710

1005 Wieczorek AS, Drake HL, Kolb S (2011) Organic acids and ethanol inhibit the oxidation of methane by  
1006 mire methanotrophs. *FEMS Microbiol Ecol* 77:28–39 . doi: 10.1111/j.1574-6941.2011.01080.x

1007 Ye R, Doane TA, Morris J, Horwath WR (2015) The effect of rice straw on the priming of soil organic  
1008 matter and methane production in peat soils. *Soil Biol Biochem* 81:98–107 . doi:  
1009 10.1016/j.soilbio.2014.11.007

1010 Yu ZC (2012) Northern peatland carbon stocks and dynamics: a review. *Biogeosciences* 9:4071–4085 .  
1011 doi: <https://doi.org/10.5194/bg-9-4071-2012>

1012 Yvon-Durocher G, Allen AP, Bastviken D, Conrad R, Gudasz C, St-Pierre A, Thanh-Duc N, Giorgio PA  
1013 del (2014) Methane fluxes show consistent temperature dependence across microbial to  
1014 ecosystem scales. *Nature* 507:488–491 . doi: 10.1038/nature13164

1015 Zhang Z, Zimmermann NE, Stenke A, Li X, Hodson EL, Zhu G, Huang C, Poulter B (2017) Emerging  
1016 role of wetland methane emissions in driving 21st century climate change. *Proc Natl Acad Sci*  
1017 114:9647–9652 . doi: 10.1073/pnas.1618765114

1018