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## **Plant Root Exudates Increase Methane Emissions through Direct and Indirect Pathways**

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## **Abstract**

The largest natural source of methane ( $\text{CH}_4$ ) to the atmosphere is wetlands, which produce 20% to 50% of total global emissions. Vascular plants play a key role regulating wetland  $\text{CH}_4$  emissions through multiple mechanisms. They often contain aerenchymatous tissues which act as a diffusive pathway for  $\text{CH}_4$  to travel from the anoxic soil to the atmosphere and for  $\text{O}_2$  to diffuse into the soil and enable methanotrophy. Plants also exude carbon from their roots which stimulates microbial activity and fuels methanogenesis. This study investigated these mechanisms in a laboratory experiment utilizing rootboxes containing either *Carex aquatilis* plants, silicone tubes that simulated aerenchymatous gas transfer, or only soil as a control.  $\text{CH}_4$  emissions were over 50 times greater from planted boxes than from control boxes or simulated plants, indicating that the physical transport pathway of aerenchyma was of little importance when not paired with other effects of plant biology. Plants were exposed to  $^{13}\text{CO}_2$  at two time-points and subsequent enrichment of root tissue, rhizosphere soil, and emitted  $\text{CH}_4$  was used in an isotope mixing model to determine the proportion of plant-derived versus soil-derived carbon supporting methanogenesis. Results showed that carbon exuded by plants was converted to  $\text{CH}_4$  but also that planted boxes emitted 28 times more soil-derived carbon than the other experimental treatments. At the end of the experiment, emissions of excess soil-derived carbon from planted boxes exceeded the emission of plant-derived carbon. This result signifies that plants and root exudates altered the soil chemical environment, increased microbial metabolism, and/or changed the microbial community such that microbial utilization of soil carbon was increased (e.g. microbial priming) and/or oxidation of soil-derived  $\text{CH}_4$  was decreased (e.g., by microbial competition for oxygen).

## **Introduction**

Methane ( $\text{CH}_4$ ) is a potent greenhouse gas responsible for 15-19% of total greenhouse gas radiative forcing, second only to carbon dioxide ( $\text{CO}_2$ ) (Intergovernmental Panel on Climate Change 2014). The largest natural source of  $\text{CH}_4$  to the atmosphere is wetlands, which produce between 20% and 50% of total (anthropogenic and natural) emissions (Ciais et al. 2013). Over half of global wetland area is in the Boreal region (Aselmann and Crutzen 1989) where temperatures are rising at a faster-than-average rate (Intergovernmental Panel on Climate Change 2014), and where wetland  $\text{CH}_4$  emissions are increasing (Gedney et al. 2004; Zhang et al. 2017). Wetlands generate  $\text{CH}_4$  because the saturated soils are anoxic and the anaerobic microbial metabolic pathways that break down organic carbon terminate in methanogenesis. The amount of  $\text{CH}_4$  emitted is controlled by factors such as vegetation type (Fritz et

al. 2011), season (Wang and Han 2005), temperature and precipitation (Hodson et al. 2011), and soil types (Kayranli et al. 2009). Understanding how these variables affect wetland CH<sub>4</sub> production and emissions is important because these variables are often sensitive to climate and environmental conditions; as the climate warms and environmental conditions change, the response of these variables could alter wetland CH<sub>4</sub> emissions, creating feedback loops (Gedney et al. 2004; Zhang et al. 2017). One such factor that both influences CH<sub>4</sub> emissions and in turn is influenced by climate warming is the growth of wetland plants (Kayranli et al. 2009). Modelling studies have identified the influence of plants on wetland CH<sub>4</sub> emissions as a prioritized area of needed study, even before climate effects are considered (Riley et al. 2011).

As global CO<sub>2</sub> concentrations and temperatures increase, boreal plants will respond with changed growth patterns and higher productivity (Forkel et al. 2016). Plant growth can influence CH<sub>4</sub> emissions in two ways. First, vascular wetland plants (e.g., sedges, shrubs, grasses) often contain aerenchymatous tissues that facilitate diffusion of gases between the soil and the atmosphere (Armstrong 1971). The hollow aerenchyma run from plant roots up into the leaves, increasing the total surface area of diffusional contact between the atmosphere and soil, and extending that contact area into saturated soil that would otherwise not be exposed to the atmosphere. Aerenchyma allow CH<sub>4</sub> to travel from the anoxic soil to the atmosphere, reducing CH<sub>4</sub> diffusion through the soil system, including through the oxic groundwater layer near the water-table surface and unsaturated soil where oxidation of CH<sub>4</sub> into CO<sub>2</sub> can occur (Shannon and White 1994; Popp et al. 2000; Fritz et al. 2011). The ability of CH<sub>4</sub> to diffuse through aerenchyma and bypass oxic water and unsaturated soil can decrease CH<sub>4</sub> oxidation, but only if the roots themselves are not surrounded by oxic groundwater. The soil surrounding roots (rhizosphere) can become oxygenated because aerenchymatous tissues allow diffusion in both directions, acting as a pathway for oxygen to enter the rhizosphere where it can preclude methanogenesis and facilitate CH<sub>4</sub> oxidation (i.e., methanotrophy) (Fritz et al. 2011). In highly reduced environments such as wetland soils, the presence of oxygen is a more important factor in CH<sub>4</sub> production than redox state (Fetzer and Conrad 1993). As boreal plants grow larger, which is predicted to occur with increasing CO<sub>2</sub> levels and temperatures (Idso et al. 1987; Jonasson et al. 1996), their root network will grow (Kummerow and Ellis 1984) and connect more of the soil to the aerenchymatous gas transport pathway.

Second, vascular plants contribute carbon to soil in the form of root exudates and leaf litter. It has been hypothesized that root exudates, which include low molecular weight sugars and amino acids, are more readily utilized by microbes than existing soil carbon and thereby fuel microbial activity

resulting in methanogenesis (Ström et al. 2003; Ström and Christensen 2007; Picek et al. 2007; Chanton et al. 2008; Kayranli et al. 2009). Methanogens can only use acetate or a combination of  $H_2$  and  $CO_2$  to actually form  $CH_4$ , but root exudates may be broken down by other microbes to form the substrates of methanogenesis. Throughout this paper, we refer to the eventual conversion of exudate-derived carbon to  $CH_4$  as being a “direct” effect of exudates, even though the microbial processing takes more than a single step. However, this causation is hard to prove in the complex plant-soil system. Root exudation increases with plant productivity (Weigel et al. 2005) as do  $CH_4$  emissions (Ström et al. 2003), but there are other factors correlated with plant productivity that could influence  $CH_4$  production and emissions, for example increased aerenchyma transport and warmer temperatures (microbial  $CH_4$  production increases with temperature (Yvon-Durocher et al. 2014)). Multiple studies have shown that at least some plant-derived carbon is utilized by microbes and emitted as  $CH_4$  (Megonigal et al. 1999; Ström et al. 2003; Trinder et al. 2008; Dorodnikov et al. 2011). However, these studies do not quantify to what extent root exudates increase total emissions, or if they instead replace soil carbon as the carbon source. In one case the portion of emitted  $CH_4$  that was plant derived was extremely small (Dorodnikov et al. 2011), and in another neither microbial biomass nor metabolic activity was correlated with the rate of root exudation (Trinder et al. 2008). One soil incubation study showed that addition of root-exudate analogs to peat soil increased  $CH_4$  production (Girkin et al. 2018a), but this effect was highly dependent on the composition of the exudate analogs (Girkin et al. 2018b). Because that approach could not replicate the delivery rate and complete carbon composition of root exudates from a living plant, nor can it account for the complex microbial community dynamics that exist in the rhizosphere.

It should be noted that the effects of aerenchyma transport and root exudation do not exist in isolation. For example, if root exudates provide carbon that is more readily used by the microbial community than the native soil carbon, exudates could stimulate increased microbial activity. An increase in aerobic microbial respiration would increase oxygen demand in the rhizosphere, which could decrease methanotrophy depending on total oxygen demand relative to supply from aerenchyma (Segers and Leffelaar 2001) and methanotrophs’ affinity for oxygen (Whalen 2005). If the rhizosphere consumes oxygen as quickly as it is delivered, then root exudation could create an anoxic pathway for  $CH_4$  to diffuse to root aerenchyma, where  $CH_4$  would quickly travel to the atmosphere. By decreasing methanotrophy in the rhizosphere, root exudation would allow the aerenchyma to transport more  $CH_4$  than if exudates were not present.

In addition to directly fueling methanogenic pathways, root exudates can also stimulate microbial priming of soil carbon (e.g., Basiliko et al. 2012). Priming is when the introduction of carbon in

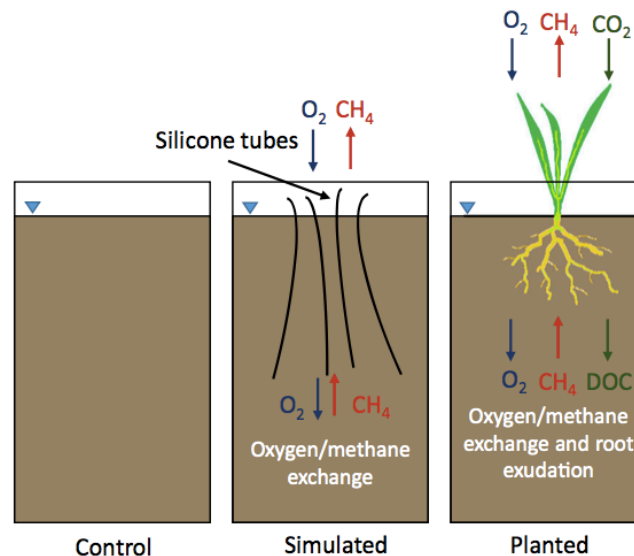
the form of simple sugars and acids can stimulate a microbial population into breaking down soil organic matter. One explanation for priming is the microbial nitrogen mining (N-mining) hypothesis, which assumes that microbes oxidize the carbon-rich but nutrient-poor exudates for energy, but then must process soil organic matter to extract nitrogen (and potentially other nutrients) for biomass production (Craine et al. 2007). From a plant evolutionary perspective, encouraging N-mining with root exudation would be beneficial in nutrient-poor soils (Carvalhais et al. 2011), such as those in peat bogs. Priming has been observed in peat soils using a mass balance approach in soil incubation experiments (Hamer and Marschner 2002; Basiliko et al. 2012; Ye et al. 2015) and by analyzing dissolved organic matter characteristics in a boreal peatland field experiment that compared planted plots to plots with plants removed (Robroek et al. 2016). However, other soil incubation experiments have failed to find evidence for priming in laboratory incubations of peat soils (Girkin et al. 2018a; Girkin et al. 2018b). While not in peat soils, one field study on a river-bank found evidence that the addition of water-soluble organic carbon decreased processing of solid-phase hydrophobic carbon, which the authors interpreted as contradicting the idea of priming (Graham et al. 2017). However, that study was conducted in a mineral sediment with far lower carbon content than peat soils, so the chemical environment and characteristics of carbon which control bioavailability were different. It is possible that priming occurs only in certain conditions.

Our study was a combined investigation of the multiple mechanisms by which plants enhance CH<sub>4</sub> emissions, utilizing a vascular plant (*Carex aquatilis*) and peat soil typical of the boreal region. *Carex aquatilis* has been studied before and found to increase wetland CH<sub>4</sub> emissions (Schimel 1995). Our study used two avenues of investigation. The first approach sought to isolate the effect of aerenchyma transport by comparing soil oxygenation and CH<sub>4</sub> emissions from real *Carex* plants to that from hollow, gas permeable tubes that mimicked the aerenchyma transport capacity of the studied plants. The second approach sought to clarify the extent to which root exudates fuel CH<sub>4</sub> production by tracing the flow of isotopically labeled CO<sub>2</sub> fixed by the *Carex* plants, delivered to the microbial community, and emitted as CH<sub>4</sub>. Together, these approaches allowed us to examine the excess CH<sub>4</sub> emitted in the presence of *Carex* and attribute this excess to transport alone, microbial utilization of exudates, and/or indirect effects such as microbial priming and reduced methanotrophy.

## **Materials and Methods**

### ***Experimental Materials and Conditions***

We grew the wetland sedge *Carex aquatilis* from plugs purchased from a local nursery (Plants of the Wild, Tekoa, WA) for 10 weeks in 5 L (5cm x 20cm x 50cm) rootboxes filled with peat collected from a thermokarst bog in the Bonanza Creek Experimental Forest near Fairbanks, Alaska (Neumann et al. 2015). Peat from that bog is high in organic content and nutrient limited, with a prior soil core study finding mean loss on ignition of 80% with standard deviation 13% and a mean C:N ratio of 48 with standard deviation 20 (Manies et al. 2017). In addition, we had unplanted control boxes filled only with peat and boxes with silicone tubes inserted into the peat instead of plants (Fig. 1). We refer to the silicone-tube boxes as the “simulated” plant treatment since they were simulating the gas-transfer effects of aerenchymatous plant tissues without adding any root exudate carbon, following the method of (King et al. 1998). The silicone tubes (1.47 mm inner diameter, 0.23 mm wall thickness) had an open top 4 cm above the peat surface and were tied off at the bottom (approximately 20cm below peat surface) to prevent water from entering the tube. This design allowed gases to diffuse through the gas-permeable silicone into and out of the soil. We placed four tubes in each box. This number was chosen so that we could space the tubes widely enough that each tube’s effect could be detected in spatial measurements of oxygen (described under “Oxygen Optode Technology”). Turner et al. (manuscript currently under review) conducted a field experiment using similar simulated plants and optodes in the same thermokarst bog from which peat for this study was collected. Though the aims of that experiment were different, the parallel methodologies allow it to provide useful context for our results.



**Fig. 1** Three box types used in experiment. Control boxes had only peat soil. Simulated-plant boxes had soil with silicone tubes that allowed O<sub>2</sub> to diffuse down into the soil and CH<sub>4</sub> to diffuse up into the atmosphere. Planted boxes had similar ability to transport gas as simulated-plant boxes, but plants additionally performed photosynthesis and released dissolved organic carbon from their roots

We kept all rootboxes at a 30-degree angle during the experiment to encourage roots to grow along the face of the rootbox, which allowed for sample collection across the rhizosphere and visualization of oxygen concentrations. Oxygen concentrations were measured using an optical oxygen sensor (i.e., an optode) (Larsen et al. 2011) that was applied to the face of a subset of boxes (described under “Oxygen Optode Technology”). Boxes without optodes had opaque front panels instead. Boxes with optodes had opaque covers that were kept on to prevent light from reaching the soil.

We conducted two different experiments. The first experiment consisted of 16 planted boxes that were all outfitted with optodes and focused solely on assessing dissolved oxygen concentrations in the rhizosphere. The second experiment consisted of 11 planted boxes, 3 control boxes, and 3 simulated-plant boxes, and it involved a wider range of measurements aimed at identifying the mechanisms by which plants influence CH<sub>4</sub> emissions. All the boxes with simulated plants were outfitted with optodes in the second experiment.

In both experiments, we watered boxes as needed to keep the peat surface continuously submerged under approximately 1 cm of water. The irrigation solution was de-ionized water with trace amounts of nutrients added to simulate rainwater. The irrigation solution recipe is given in SI table S1.

We kept rhizoboxes in a pair of growth chambers and randomly re-assigned box position within and between the growth chambers twice per week to avoid any effects of spatial variability in growth conditions. We controlled environmental conditions within the chambers with growth lights and air conditioners set to imitate central Alaska summer growth conditions with 18 hours of daylight, daytime temperatures of 18 °C, and nighttime temperatures of 10 °C. We recorded temperature, relative humidity, and photosynthetically active radiation (PAR) (SI Figures S1 and S2) with sensors placed at the height of the tallest plants. During the first experiment, the median temperatures recorded with these sensors were 18.6 °C in one chamber and 20.6 °C in the other. In the second experiment the median temperatures recorded were 20.0 °C in one chamber and 20.5 °C in the other. We believe there was a vertical temperature gradient in the growth chambers because the air conditioners were located underneath the rhizoboxes while the sensors were directly under the growth lights.

### ***Oxygen Optode Technology***

The planar optical oxygen sensors (i.e., optodes) used in this experiment were modified from Larsen et al. (2011). The sensors were made from two fluorescent dyes, an indicator dye and an antenna dye. When the antenna dye was excited by a photon it used part of the energy to fluoresce green and also passed energy to the indicator dye. The indicator dye was reversibly quenched in the presence of

oxygen, but when not quenched fluoresced red when excited. The ratio of red to green fluorescence is the foundation of the technology.

We made the optodes by airbrushing a mixture of Pt(II) meso-Tetra(pentafluorophenyl)porphine (the indicator dye) and 10-(2-Benzothiazolyl)-2,3,6,7-tetrahydro-1,1,7,7-tetramethyl-1H,5H,11H-(1)benzopyrroprano(6,7-8-I,j)quinolizin-11-one (the antenna dye) suspended in acetone onto 3/6 inch thick polycarbonate sheets. Once the dye mixture had dried, we applied a graphite coating by pouring a mixture of graphite and silicone dissolved in hexane onto the optode and tilting the optode so that the surface was evenly coated. The graphite coating provides enough opacity to prevent soil and roots behind the optode from interfering with the optical measurements.

We calibrated the optodes by attaching them to boxes filled with water that was bubbled with nitrogen gas and air to create a range of dissolved oxygen (DO) concentrations (as determined by a calibrated oxygen probe (InsiteIG Model 3100)), and photographing the optode as we would during experimentation. We took both calibration and experimental photographs by placing the box in a dark chamber and illuminating it with a blue (447.5 nm wavelength) LED positioned at a 45 degree angle to the optode, and taking a picture with a digital SRL camera (Canon EOS RebelXS) which had its near-IR filter removed and was fitted with a yellow filter (Edmunds Optics, OG-530 Long Pass Filter). The camera was controlled by the Look@RGB software. The red to green ratio (R) in each pixel was used to calculate the oxygen concentration (C) according to Equation 1, using the calibration parameters  $K_{SV}$ ,  $R_0$ , and  $\alpha$ , as described in Larsen et al. (2011).

$$(1) \ C = \frac{R_0 - R}{K_{SV} * (R - R_0 * \alpha)}$$

### ***Non-Destructive Measurements — Second Experiment***

In addition to optode measurements, in the second experiment we measured plant height and fluxes of CH<sub>4</sub> and CO<sub>2</sub>. Gas fluxes were measured using one of two greenhouse gas analyzers (Los Gatos Research Ultraportable Greenhouse Gas Analyzer (LGR) and Picarro G2201-I CRDS) attached to an opaque hood that was strapped over the box of interest. The hood was 4.9 L in volume and enveloped both the entire plant and the entire soil surface area of the box. Headspace concentration was recorded by the instrument once per second for 5 to 10 minutes. The LGR recorded only total concentration of CH<sub>4</sub> and CO<sub>2</sub> while the Picarro recorded  $\delta^{13}\text{C}$  of both gases as well. We applied a linear regression to the change in concentration over time to determine the flux rate.

The  $\delta^{13}\text{C}$  measurements for emitted CH<sub>4</sub> had a non-trivial amount of noise, especially at low concentrations, making it unreliable to use a single starting  $\delta^{13}\text{C}$  value for CH<sub>4</sub> in the chamber headspace

from which to calculate the  $\delta^{13}\text{C}$  of emitted  $\text{CH}_4$ . Instead, we calculated the isotopic ratio of emitted gas by taking every possible combination of measurements during the flux and using the difference between the concentrations and  $\delta^{13}\text{C}$  of  $\text{CH}_4$  in the headspace at any two points in time to calculate the  $\delta^{13}\text{C}$  of emitted  $\text{CH}_4$ . That method produced tens of thousands of results per flux, from which we used the median.

We only used flux data if the root mean square error (RMSE) of a linear regression was less than 0.5 parts per million  $\text{CH}_4$  concentration per hour, which was equal to a flux rate of  $3.9 \text{ mg m}^{-2} \text{ d}^{-1}$  from our boxes which had a surface area of  $0.01 \text{ m}^2$ . We only used isotope data if the RMSE of the flux rate was less than 0.5 ppm, and the  $R^2$  value of the flux rate was greater than 0.9. The additional quality control check was needed for isotope data because at very low mass flux rates the RMSE can be low in absolute terms, but still comprise a significant portion of the total variation. We need to include low flux rates (even those with low  $R^2$  values) so as to not bias the data towards large fluxes, but the isotope calculations produced unreliable values when the  $R^2$  was low, regardless of the flux rate; they spanned an extreme range including unrealistically high and low values. All  $\delta^{13}\text{C}$  results were referenced to Vienna Pee Dee Belemnite and reported in delta notation:

$$\delta^{13}\text{C} = \left( \frac{R_{\text{sample}}}{R_{\text{standard}}} - 1 \right) * 1000$$

Where  $R_{\text{sample}}$  is the  $^{13}\text{C}$  to  $^{12}\text{C}$  ratio of the measured sample and  $R_{\text{standard}}$  is the  $^{13}\text{C}$  to  $^{12}\text{C}$  ratio of Vienne Pee Dee Belemnite (0.0112372).

### ***Stable Isotope Labeling and Destructive Sampling — Second experiment***

In weeks 5 and 10 of the second experiment, we exposed 4 randomly selected plants to  $^{13}\text{CO}_2$  by placing a clear 6.5 L hood made of extruded acrylic on each rhizobox and injecting 15 mL of 99 atom %  $^{13}\text{CO}_2$  (Sigma Aldrich) into the headspace every hour for 14 hours each day over a period of five consecutive days. This technique was based on previously described methods (Lu and Conrad 2005). We took flux measurements at least once per day per plant during and after both labeling events with a Picarro G2201-I. During each day's 14-hour labelling period we took fluxes with the clear fluxing hood still in place, immediately before the time for an injection of  $^{13}\text{CO}_2$ . After the entire labelling event was completed, we used the same fluxing protocol as the routine flux measurements.

We destructively sampled boxes 3-5 days after isotopic labeling ended. To harvest a box, we attached a spill-guard to the top of the box to prevent water from pouring out, placed the box in an

anaerobic glove bag purged three times with nitrogen gas, laid the box on its back and removed the front panel to expose the root-soil system. We took all samples directly from this exposed surface. For all three box types (planted, control and simulated) we collected root and soil samples at depths of approximately 5 cm, 20 cm, and 35 cm. At each depth, we took three samples, one in the center and one 6 cm from either edge of the box.

For planted boxes, we cut root sections approximately 8cm in length from each location and placed them in centrifuge tubes filled with phosphate buffered solution (PBS), which we capped and removed from the glove bag. We sonicated the sealed centrifuge tubes with the root samples for 10 minutes then quickly moved the root to a new PBS-filled container. Both containers were then immediately frozen at -20 C. All the soil that fell off the root during sonication was considered rhizosphere soil.

For unplanted control boxes, we sampled approximately 1mL cubes of soil from the 9 standard locations. We immediately placed samples into 2 mL collection tubes with 1 mL of PBS (to keep the sample storage as similar as possible to the planted box procedure) and placed the tubes on dry ice in the anoxic glovebag. We consider these soil samples as 'bulk soil.'

For simulated-plant boxes, at the 5 and 20cm depths where silicone tubes were present, we sampled a thin layer of peat immediately adjacent to the silicone tubes. The tubes did not extend to the 35 cm depth, so we sampled 1mL cubes of bulk soil at that depth. We immediately placed samples into phosphate buffered solution (PBS) and stored them on dry ice in the anoxic glove bag.

All samples from all the boxes were then stored at -20 C for up to 33 days before being shipped on dry ice to the Environmental Molecular Sciences Laboratory (EMSL) where they were stored at -80 C until analysis.

### ***Porewater Collection and Analysis — Second experiment***

We collected porewater from all boxes during the second week of the second experiment, which we considered an initial time point, and immediately prior to the second labelling event in week 8 from boxes that were not harvested after the first labeling event. We collected 10-16 mL of porewater from a depth of 20cm below peat surface at the center of each box using a syringe attached to PushPoint porewater sampler (MHE Products) with a mesh filter (10µm pore size) over the inlet. We then filtered the water (0.2 µm, nylon membrane syringe tip filter) and used a syringe needle to inject it into previously prepared 60 mL serum vials capped with a butyl rubber stopper. We prepared the vials before sampling by evacuating and flushing them with 99.999% nitrogen gas three times, leaving the vials at atmospheric pressure. We pre-acidified the vials with 200 µL of phosphoric acid to minimize

microbial activity. After injecting porewater into the vials, we left them for at least one week to equilibrate before we transferred 15mL of headspace gas into a 12mL exetainer vial (Labco) using a syringe and needle.

We used a Shimadzu GC-FID 2014 to measure concentration of CH<sub>4</sub> in the exetainer vial and calculated the porewater concentrations using Henry's Law and the mass of water in the original vial. We calibrated the GC measurements using standards of various concentrations (MESA Specialty Gases & Equipment). The standards purchased from MESA were a 100 ppm CH<sub>4</sub> standard, a 10.2 ppm CH<sub>4</sub> standard, and a 50% CH<sub>4</sub> standard which we diluted with N<sub>2</sub> to form an array of concentrations.

#### ***Isotope Ratio Mass Spectroscopy – Second experiment***

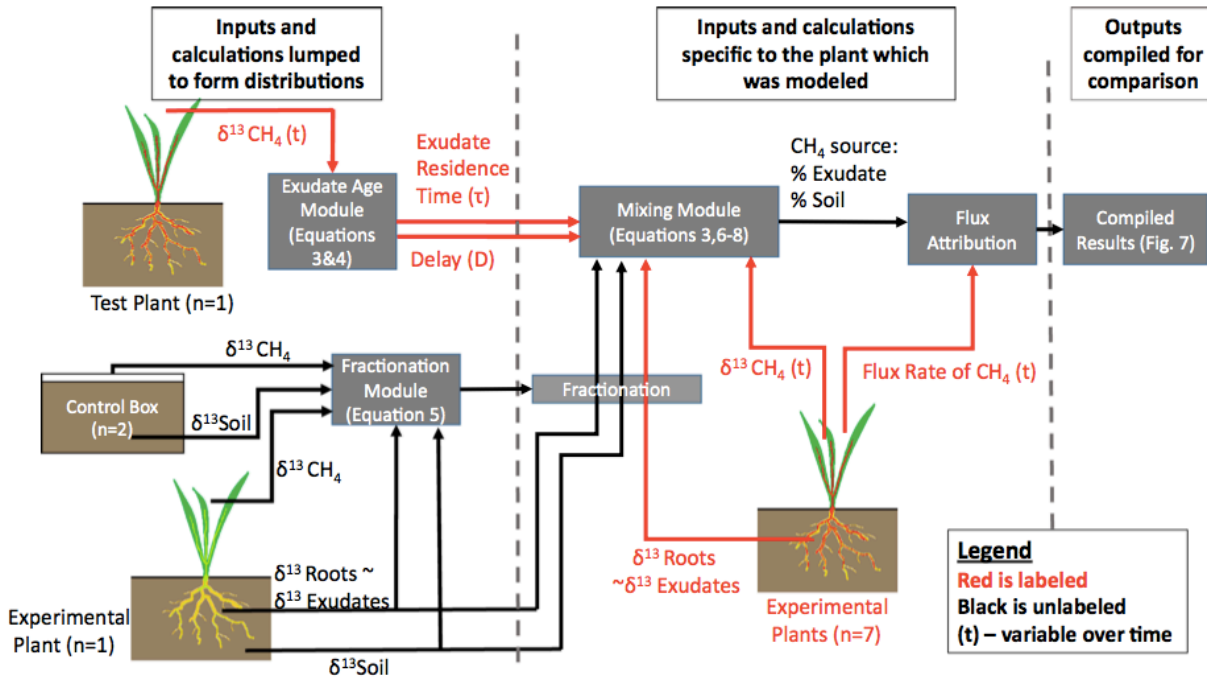
At EMSL, we used an a Costech Analytical Technologies, Inc. elemental analyzer (EA) coupled to a Thermo Scientific Delta V Plus isotope ratio mass spectrometer (IRMS) to perform isotopic analysis on root, rhizosphere and bulk soil samples. All soil samples (both bulk and rhizosphere) were removed from -80 C storage and immediately lyophilized using a VirTis Benchtop K lyophilizer. We lyophilized 21 of 27 root samples in the same manner, but the other 6 root samples were first analyzed at room temperature by laser ablation mass spectroscopy for data to be published in a different manuscript before being lyophilized. The laser ablation removed some mass from the surface of the root, but we assumed that it was a low enough percentage of the root that the average isotopic composition of the root was unaffected.

After lyophilization, samples were stored at room temperature. We sub-sampled the lyophilized rhizosphere samples into tin capsules (4 by 6 mm, part number 41070, Costech Analytical Technologies, Inc.) using 200-300 µg of sample per replicate with a minimum of triplicate analytical replicates. We cut lyophilized root samples into 1 to 3 mm thick cross sections using a razor and loaded the cross sections into tin capsules as described above. The EA combustion reactor was loaded with cobaltic oxide and chromium oxide catalyst and maintained at 1,020 °C while the reduction reactor was loaded with copper catalyst and maintained at 650 °C. We used two in-house glutamic acid standards that were themselves calibrated against USGS 40 and USGS 41 standards ( $\delta^{13}\text{C} = -26.39 \text{ ‰ VPDB}$  and  $37.63 \text{ ‰ VPDB}$  respectively) and applied a two-point, slope intercept correction to the data (Coplen et al. 2006). In addition, we used an in-house acetanilide standard as a check on the isotope measurement accuracy.

#### ***Isotope Mixing Model – Second experiment***

We used the IRMS and CH<sub>4</sub> flux data to construct a mixing model to determine what portion of the CH<sub>4</sub> from each individual flux measurement was derived from soil carbon and what portion was derived from root exudates (Fig. 2). The model consisted of three modules: the exudate age module, the

fractionation module, and the mixing module. The exudate age and fractionation modules (described in detail below) generated parameter distributions which were fed into the mixing module for the final calculations that indicated what percentage of each  $\text{CH}_4$  flux was derived from soil versus exudate carbon.



**Fig. 2** The isotope mixing model. In the first section, inputs from multiple boxes were combined to calculate distributions of parameters for the model. Data a labeled test plant was used to calculate exudate residence times in the exudate age module, while only unlabeled plants and control boxes were used to calculate isotopic fractionation effects. In the second section, each flux was individually modeled using a distribution of parameters generated in the first step using a Monte-Carlo approach. The mixing module used fractionated soil and root data to determine the percentage of each flux that came from soil, labeled exudates, or unlabeled exudates, then in the flux attribution stage those percentages were multiplied by the total amount of  $\text{CH}_4$  emitted. In the final section of the model all of the fluxes were compiled to establish a range of results, as presented in Fig. 7. Any variable given with (t) after it varied over time, while other variables do not

The exudate age module calculated the portion of total exudates in the rhizosphere that would contain elevated  $\delta^{13}\text{C}$  at specific time points after labeling, given the residence time of exudates in the soil ( $\tau$ ) and the delay (D) between the end of labeling and maximum  $^{13}\text{C}$  enrichment of emitted  $\text{CH}_4$ . The module was run in two separate contexts: once on an isotopically labeled test plant independent of the full model to determine parameter values (as described below), and again on experimental plants using the fitted parameter values. The module modeled exudates in the soil as a continuously stirred tank reactor (CSTR), and assumed the isotopic composition of emitted  $\text{CH}_4$  at any time point was linearly correlated with the isotopic composition of exudates in the soil. The isotopic composition of the soil carbon was a constant. This approach required that the  $^{13}\text{C}$  content of the exudates did not affect the percentage of  $\text{CH}_4$  derived from those exudates. While various factors can affect the degree to which

microbes discriminate against  $^{13}\text{C}$  within a carbon source (Lehmeier et al. 2016), we could not identify any study which showed isotopic enrichment affecting the type of carbon compounds utilized by microbes. Therefore, we believe that we are justified in our assumption that increasing enrichment of exudates after labeling did not have a measureable impact on the portion of  $\text{CH}_4$  emissions that were soil-derived versus exudate-derived.

The general equation for a CSTR is given in Equation 2, and calculates the concentration ( $C(t)$ ) of a species at a given time ( $t$ ) based on residence time ( $\tau$ ), the initial concentration of the species ( $C_0$ ) and its concentration after infinite time ( $C_\infty$ ), which is equal to input concentration. In the case of our model, we defined isotopically labeled exudates as a “concentration” of one and unlabeled exudates as a “concentration” of zero, assuming that exudates maintained a consistent enrichment during the period over which they were labeled. Therefore, we can write Equations 3 and 4 to represent our system in which the “concentration” at a given time ( $C(t)$  in Equation 2) was the percentage of exudates in the rhizosphere that were labelled ( $E_L(t)$ ). The percentage labeled ( $E_L(t)$ ) was defined in Equation 7 as the portion of total emissions derived from labeled exudates ( $P_{LE}(t)$ ) divided by the portion of total emissions derived from both labeled exudates ( $P_{LE}(t)$ ) and unlabeled exudates ( $P_{UE}$ ). Because our labelling was a pulse input that ended after five days, the isotopic enrichment of emitted  $\text{CH}_4$  reached a maximum and then began to decline at some point after the 5-day long labeling period. Equations 3 and 4 are both modified versions of Equation 2. Equation 3 calculates an input of labeled exudates to a previously unlabeled system and was used prior to the peak, while Equation 4 calculates the loss of labeled exudates from a previously labeled system and was used after the peak. Because we harvested the experimental plants before the peak, only Equation 3 was used on experimental plants. Equation 4 was only used on the test plant to determine parameter values. The delay ( $D$ ) was the length of time, in days, between when labeling ended (i.e., after the 5-day long labeling period) and when emitted  $\text{CH}_4$  reached maximum enrichment. In all cases, time ( $t$ ) zero was defined as the beginning of labeling. We calculated the portion of exudates labeled at the peak ( $C_{\text{peak}}$ ) using Equation 3 at  $t$  equal to length of labeling (five days) plus the delay. We used  $C_{\text{peak}}$  as the starting percentage of exudates that are labeled for time points beyond the peak (Equation 4).

Table 1, equations used in the isotope mixing model. Variables used:  $C_0$  is concentration at time zero, and no units are assigned since it is not actually used in the model (Equation 2 is a generic form).  $C(t)$  is concentration at time  $t$ , where  $t$  is in days.  $C_\infty$  is concentration at time infinity.  $\tau$  is residence time of exudates in days.  $P_{LE}$  is the percent of carbon used for methanogenesis that is derived from labeled exudates.  $P_{UE}$  is the percent of carbon derived from unlabeled exudates.  $P_s$  is the percent of carbon from soil.  $F$  is the fractionation between source carbon and emitted  $CH_4$ , in units  $\delta^{13}C$  ‰.  $M_U$  is the  $^{13}C$  content of  $CH_4$  emitted from unlabeled plants, in units  $\delta^{13}C$  ‰.  $M_L$  is the enrichment of  $CH_4$  emitted from labeled plants, in units  $\delta^{13}C$  ‰.

Equation #	Modules utilizing	Description	Equation
2	n/a	Generic CSTR	$C(t) = C_\infty + (C_0 - C_\infty)e^{-\frac{t}{\tau}}$
3	Exudate age Valid from $t=D$ to $t = 5 + D$	CSTR pre-peak (modified Eqn. 2) Initial condition: unlabeled exudates = 0 Input: labeled exudates = 1	$E_L(t) = 1 + (0 - 1)e^{-(t-D)/\tau}$
4	Exudate age Valid from $t = 5+D$ to $t = \infty$	CSTR post-peak (modified Eqn. 2) Initial condition: Proportion of labeled exudates calculated in Eqn. 3 at peak ( $t = 5+D$ ) = $C_{peak}$ Input: unlabeled exudates = 0	$E_L(t) = 0 + (C_{peak} - 0)e^{-(t-D-5)/\tau}$
5	Fractionation	Definition of fractionation	$F = M_U - S_{UR}$
6	Mixing	$\delta^{13}CH_4$ is linear combination of source $\delta^{13}$	$P_{LE}(t) * [F + S_{LR}] + P_{UE}(t) * [F + S_{UR}] + P_s * [F + S_s] = M_L(t)$
7	Mixing	Ratio of old to new exudates	$E_L(t) = \frac{P_{LE}(t)}{P_{LE}(t) + P_{UE}(t)}$
8	Mixing	All carbon sources add up to 100%	$P_{LE}(t) + P_{UE}(t) + P_s = 1$

The exudate age module was used in conjunction with data from a test plant to determine what residence times and delay periods should be used as inputs to the model. A variety of residence times and delays were used to generate theoretical enrichment curves, which we graphed against the test plant data (SI Figure S3). We assigned a likelihood to each of these curves proportional to the inverse of the sum of squares of the residuals, and those likelihoods were used to generate distributions from which we sampled in the model. The median residence time and delay used are reported in Table 3 along with upper and lower quartiles.

We calculated separate fractionation factors for planted and control boxes. The fractionation module calculated the apparent isotope fractionation associated with microbial conversion of either

soil-derived or root-derived carbon into  $\text{CH}_4$  and with subsequent emission of  $\text{CH}_4$ . The module subtracted the  $\delta^{13}\text{C}$  of  $\text{CH}_4$  emitted from unlabeled plants ( $M_U$ , fluxes from all plants prior to labeling were used) from that of unlabeled roots ( $S_{UR}$ ), and the  $\delta^{13}\text{C}$  of  $\text{CH}_4$  emitted from control boxes and simulated plants from that of unlabeled soil ( $S_{US}$ ) to calculate the isotopic fractionation of carbon ( $F$ ). This calculation was done using all emissions from each type of box (planted or control) compared against all source carbon measurements from that type. This was done because only one flux measurement from the unlabeled plant which was harvested passed quality control, so we included fluxes from other plants prior to their harvest. The narrow range of  $\delta^{13}\text{C}$  for unlabeled fluxes justified this grouping (Fig. 5A). A distribution was generated from all fractionation factors calculated, and values used in the model were randomly sampled from that distribution.

The mixing module used output from the fractionation and exudate age modules to solve a system of equations that determined the proportion of emitted  $\text{CH}_4$  derived from soil ( $P_S$ ), isotopically labeled root exudates ( $P_{LE}(t)$ ), and unlabeled root exudates present in the soil prior to labeling ( $P_{UE}(t)$ ) as a function of time. In Equation 6, the isotopic enrichment of the emitted  $\text{CH}_4$  ( $M_L(t)$ , data from each flux individually) had to be a linear combination of the  $\delta^{13}\text{C}$  of the three fractionated sources: labeled roots ( $S_{LR}$ , data taken from only the box from which the  $M_L$  measurement was taken), unlabeled roots ( $S_{UR}$ , median of a distribution) and unlabeled soil ( $S_S$ , median of a distribution). We used medians for  $S_{UR}$  and  $S_S$  instead of a Monte-Carlo sampling approach because the distributions generated were very small and during model development we determined that simply using the median would reduce the amount of computation needed without influencing the results. Equation 7 established the ratio of the labeled and unlabeled exudates, according to the output of the exudate age module for that time point. Equation 8 made the sum of all carbon sources equal to the full amount of the  $\text{CH}_4$  emitted.

In the model we used 30 randomly selected values from each of the four parameter distributions we generated (enrichment of exudates ( $S_{LR}$ ), fractionation factor ( $F$ ), residence time ( $\tau$ ), and delay ( $D$ ) between start of labeling and when the label was emitted as  $^{13}\text{CH}_4$ ). This resulted in 810,000 results for each of the 21 fluxes used. The model (written in MATLAB 2018b) and all necessary functions and data are attached as a .zip file in the electronic supplementary information.

### **Statistical analysis**

We made all statistical comparisons using either the Wilcoxon rank sum test or the Kruskal-Wallis test, depending on whether we were comparing two groups or more than two groups of data. Both tests are non-parametric methods to determine whether two sets of measurements are likely to be from the same distribution or not.

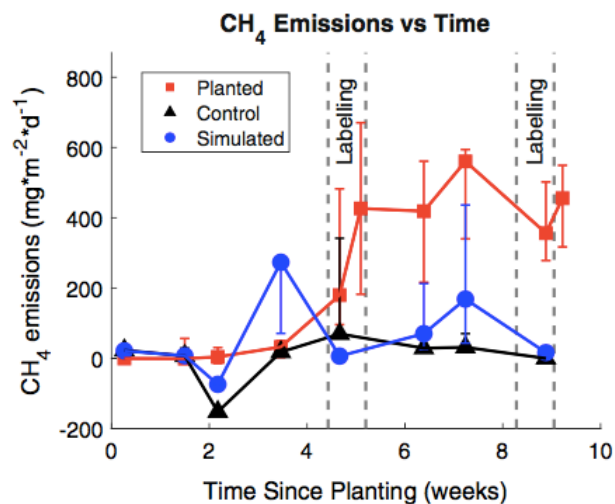
## Results

All results below are from the second experiment, except for optode images from planted boxes. Oxygen data for planted boxes were collected in the first experiment. Growth chamber conditions, plant height and CH<sub>4</sub> emission data from the first experiment are available in the Online Resources (Figure S1 and Tables S3 and S5).

### Plant Height and Gas Emissions

Plants grew steadily from a median height of 9.1 cm (n=11) to 32.9 cm (n=6) through the entire period measured (between weeks 1 and 10 of the experiment). We report plant height and emissions data for each rootbox in Online Resources tables S4 and S6, respectively.

Planted boxes emitted little CH<sub>4</sub> at first; the median emission flux was less than 25 mg m<sup>-2</sup> d<sup>-1</sup> for each of the first four weeks of measurement (Fig. 3) (n ≥ 5 per week). In week five, emissions from planted boxes increased slightly, with a median CH<sub>4</sub> flux rate of 98 mg m<sup>-2</sup> d<sup>-1</sup> and one box emitting up to 516 mg m<sup>-2</sup> d<sup>-1</sup>. For the remainder of the experiment, planted boxes had high CH<sub>4</sub> emissions with weekly medians ranging from 370 to 580 mg m<sup>-2</sup> d<sup>-1</sup> (n ≥ 4 per week). A Kruskal-Wallis test showed that emissions from planted boxes during the second half of the experiment (weeks six to ten) were significantly greater than those from the first half (p < 0.05).



**Fig. 3** Medians for weekly CH<sub>4</sub> emissions from the three experimental treatments with error bars indicating upper and lower quartiles. Vertical dashed lines show days when isotopic labelling began and ended. Isotopic labeling lasted 5 days. Methane emissions from planted boxes increased over the course of the experiment, while those from control boxes and boxes with simulated plants did not increase. The number of fluxes per week differed because of variability in the number of fluxes that passed quality control and an increase in the number of flux measurements that were made during labeling. For planted boxes,

weekly  $n=2-15$ , median  $n=7.5$ . For control boxes, weekly  $n=1-3$ , median  $n=2.5$ . For simulated boxes, weekly  $n=1-3$ , median  $n=1.5$ . Negative fluxes for control and simulated plant treatments after week 2 are each from a single box

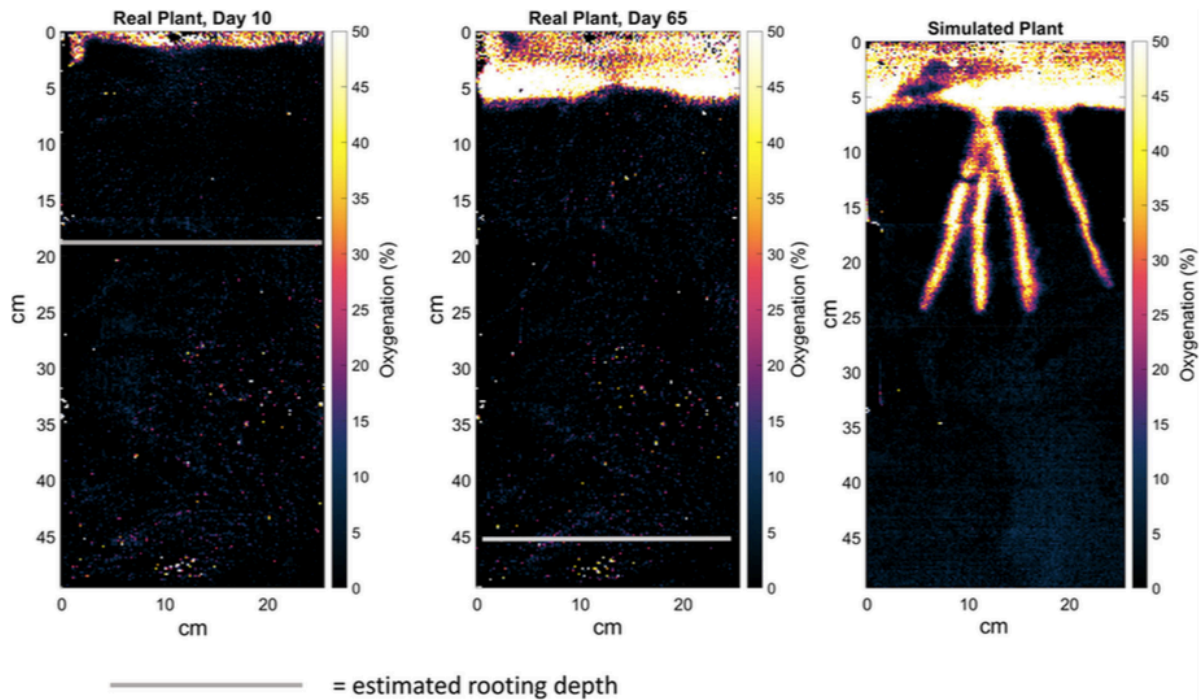
Control boxes emitted little  $\text{CH}_4$  throughout the experiment ( $n=19$  measurements), with weekly median emissions never exceeding  $70 \text{ mg m}^{-2} \text{ d}^{-1}$ . Boxes with simulated plants typically emitted little  $\text{CH}_4$  ( $n=16$ ), though on three of the nine weeks of measurements they emitted larger than usual amounts, most notably in week 4 when the median flux was  $273 \text{ mg m}^{-2} \text{ d}^{-1}$ . The differences between emissions of control boxes and simulated boxes were not significantly different from each other or from the emissions of planted boxes during the first half of the experiment ( $p > 0.05$ ), but emissions from both control boxes and simulated boxes were significantly lower than emissions from planted boxes during the second half of the experiment ( $p < 0.05$ ). These data indicate that emissions were related to plant growth, but not directly correlated; plants switched between a lower-emission state and a higher-emission state quickly (i.e., within two weeks), while their height increased in a nearly linear fashion.

All three box types had high variation in  $\text{CO}_2$  fluxes, including periods of both uptake and emission (Online Resources Table S6 and Figure S4). As with  $\text{CH}_4$  emissions, the Planted boxes did emit more  $\text{CO}_2$  during the second half of the experiment than the other box types. Because the fluxes were taken in the dark, an increase in emission of  $\text{CO}_2$  is consistent with increased plant and/or microbial respiration.

#### **Optode Imaging**

The planted boxes equipped with optodes during the first experiment never had measurable concentrations of oxygen in the soil. Representative oxygen-concentration images are shown in the two left panels of Fig. 4, and all 113 images (16 plants imaged weekly over 9 weeks, with some harvested at intermediate points) are in the Online Resources (Figures S5-S20). In contrast, the simulated plants in the second experiment had significant amounts of oxygen surrounding the inserted silicone tubes throughout the entire experiment (Fig. 4). The 27 simulated plant optode images (three boxes imaged weekly over nine weeks) are also in the Online Resources (Figures S21-S23).

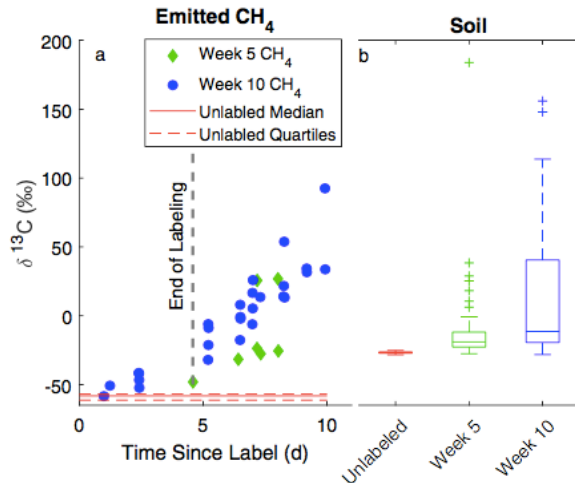
It is worth acknowledging that we did not directly measure the gas-transport capacity of the plants or the silicone tubes, and so cannot claim that the simulated plants transported more or less oxygen than the real plants. Rather, what we can say is that the real plants did not allow what oxygen they did transport to build up in the soil.



**Fig. 4** Examples of optode data showing that real plants never oxygenated the rhizosphere, while simulated plants had measurable oxygen concentrations surrounding the tubes throughout the experiment

### **Methane, Root, and Rhizosphere Isotope Enrichment**

During both labeling events (weeks 5 and 10 of the experiment), CH<sub>4</sub> emitted from the boxes steadily became more isotopically enriched over a five-day period between the end of labeling and harvest (Fig. 5). During the week 5 labeling event fewer flux measurements were made than during the week ten event, and the CH<sub>4</sub> emissions were lower which meant that fewer flux measurements passed the isotope quality control protocol. During the week 10 labeling event, the final CH<sub>4</sub> from each box before harvest had  $\delta^{13}\text{C}$  greater than +30‰, having increased from a pre-label median of -58‰. This strong enrichment signal shows that plant-derived carbon was converted to CH<sub>4</sub> within days of being photosynthesized.



**Fig. 5** Both panels share the y-axis. (A) Isotopic enrichment of  $\text{CH}_4$  emissions over time since labeling for four planted boxes during each labeling event. Red lines are median and interquartile range of unlabeled emissions. During both labeling events emitted  $\text{CH}_4$  became substantially enriched. (B) Box and whisker plot of soil carbon isotopic enrichment. The unlabeled soil includes both rhizosphere and bulk soil ( $n=68$ ) while the soil data from the Week 5 ( $n=43$ ) and Week 10 ( $n=69$ ) plant harvests is all rhizosphere soil. The rhizosphere became substantially more enriched after the week 10 harvest

Roots from both labeling events were highly enriched (Table 2), and there was not a statistically significant difference in root enrichment between the two labeling events ( $p > 0.05$ ), but roots from labeled plants were significantly more enriched than those from unlabeled plants ( $p < 0.05$ ). Unlabeled peat from the control boxes and simulated-plant boxes, and rhizosphere soil from unlabeled plants all had statistically similar isotopic signatures ( $p > 0.05$ ), and thus we treated them collectively as ‘unlabeled peat.’ The unlabeled peat had a significantly lower  $^{13}\text{C}$  content ( $p < 0.01$ ) than rhizosphere soil collected after the first and second labeling events, and rhizosphere soil collected after the first labeling events had a significantly lower  $^{13}\text{C}$  content ( $p < 0.01$ ) than rhizosphere soil collected after the second labeling event, as shown in Fig. 5. The first labeling event enriched carbon in the rhizosphere from -26.6‰ (the median  $\delta^{13}\text{C}$  of unlabeled peat, mean -26.8‰) to a median of -18.9‰ (mean -8.6‰), while the second labelling event enriched the rhizosphere even more effectively, bringing the median enrichment up to -11.6‰ (mean +30.8‰) (Fig. 5B). The means had higher values than the medians for both labeling events because the data were clustered low, with tails skewed high.

Table 2, isotopic composition of roots from unlabeled plants and plants labeled during both events. Roots from both labeling events are significantly different ( $p < 0.05$ ) than those from unlabeled plants, but data from the two labeling events are not different from each other ( $p > 0.05$ ).

Isotopic Composition of Roots	Unlabeled	Week 5 Labeling	Week 10 Labeling
Lower Quartile (‰)	-34.9	341.9	104.5
Median (‰)	-33.1	630.4	431.6
Upper Quartile(‰)	-32.4	1178.5	1981.0
Number of samples	9	53	38

### Porewater Dissolved Gases

None of the treatments had a statistically significant change in dissolved  $\text{CH}_4$  concentration between the initial porewater sampling event (week 2) and the final porewater sampling event (week 8) (Table 3). Due to the small sample size for control boxes and boxes with simulated plants ( $n=3$  for each) the statistical power of the Wilcoxon test used was low and it would have been unlikely to detect a small change. Nevertheless, multiplying the largest change in concentration recorded in any box by the volume of the box results in an increase of only 1.6 mg  $\text{CH}_4$  over the course of the experiment. If this amount of  $\text{CH}_4$  were emitted, it would have increased emission by 3.5 mg  $\text{m}^{-2} \text{d}^{-1}$ , a trivial amount relative to the magnitude of the fluxes recorded. For comparison, control boxes emitted on the order of one to ten mg  $\text{m}^{-2} \text{d}^{-1}$ , while planted boxes emitted over 200 mg  $\text{m}^{-2} \text{d}^{-1}$  during the second half of the experiment (Fig. 3).

Given these results, for all boxes we could equate emitted  $\text{CH}_4$  with the net generation of  $\text{CH}_4$  (production minus oxidation) because a negligible amount of  $\text{CH}_4$  was stored within the soil during the experiment.

Table 3, porewater  $\text{CH}_4$  concentration changes over the course of the experiment show that there was negligible buildup of  $\text{CH}_4$  in the soil in any of the box types.

Box Type	$\Delta \text{CH}_4$ Concentration Over Six Weeks ( $\mu\text{M}$ )		Increase in Flux if all $\text{CH}_4$ Were Emitted (mg $\text{m}^{-2} \text{d}^{-1}$ )		# of boxes
	Range	Median	Range	Median	
Control	-2 to 20	1	-0.3 to 3.5	0.2	3
Simulated	-1 to 10	0	-1.3 to 0.2	-0.7	3
Planted	-14 to 16	6	-2.4 to 2.7	1.1	6

### Isotope Mixing Model

The results of the fractionation and exudate age modules which produced parameter distributions are shown in the bottom half of Table 4. The fractionation effects in both planted and

unplanted boxes were similar, and within the range of values previously reported in the literature. Previously conducted studies have examined the residence time of all root-derived carbon (Gaudinski et al. 2000; Rasse et al. 2005), instead of separating exudate carbon from carbon that was sloughed off of the root itself or part of a dead root. Here, we did separate exudates from other types of root-derived carbon. Because of this difference, previous studies have reported values on the time-scale of several years (Gaudinski et al. 2000) as opposed to the timescale of a couple months that we measured for root exudates. Similarly, our delay period is not a variable that other studies have estimated. The residence time of carbon in plants has been measured, but that includes carbon that is incorporated into plant tissues and, as with residence time in the soil, can stretch into years as opposed to the scale of days measured here (Gaudinski et al. 2000).

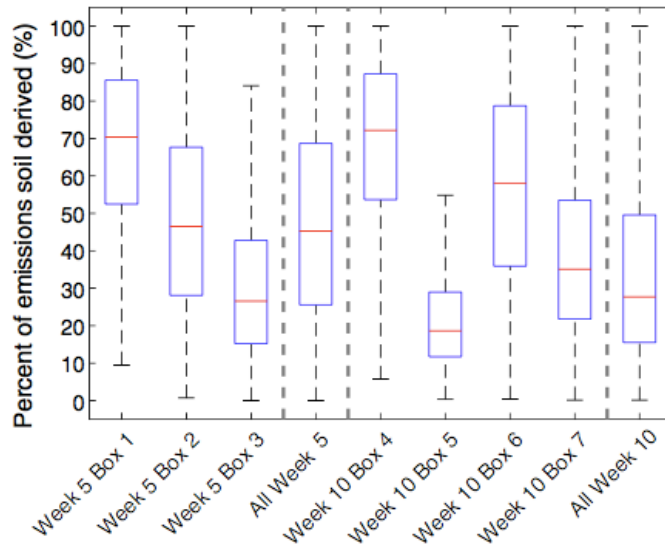
*Table 4, isotopic composition of various carbon pools, along with results of the fractionation module and exudate age module of the mixing model. The fractionation module calculated fractionation factors based on measured  $\delta^{13}\text{C}$  values in unlabeled boxes. The mixing module calculated the residence times of exudates in the soil and the delay between the end of labeling and peak enrichment based on data from a test plant. The medians of all values calculated or measured in this study are reported with interquartile ranges in parenthesis. The ranges for the literature values are the full range of values reported in those studies.*

<b>Measured Values (‰)</b>	<b>Control Boxes</b>	<b>Number of control boxes/samples</b>	<b>Planted Boxes</b>	<b>Number of planted boxes/samples</b>	<b>Literature values</b>
$\delta^{13}\text{C}$ of unlabeled soil ( $S_{\text{US}}$ )	-26.7 (-27.1 to -26.5)	3/29	-26.5 (-26.8 to -26.2)	1/6	-28 to -24.5 <sup>a</sup>
$\delta^{13}\text{C}$ of unlabeled roots ( $S_{\text{UE}}$ )	n/a	n/a	-33.1 (-34.9 to -32.4)	1/9	-22 to -34 <sup>b</sup>
$\delta^{13}\text{C}$ of emitted unlabeled methane ( $M_{\text{U}}$ )	-58.2 (-61.3 to -56.7)	3/6	-61.0 (-62.3 to -57.8)	7/12	-70.3 to -54.9 <sup>c</sup>
<b>Calculated Values</b>	<b>Control Boxes and Simulated Plants</b>	<b>Number of results for CB &amp; SP</b>	<b>Planted Boxes</b>	<b>Number of results for planted boxes</b>	<b>Literature values</b>
Net apparent fractionation of $\text{CH}_4$ emission (F) (‰)	-31.6 (-34.4 to -29.9)	621	-27.1 (-29.6 to -24.2)	90	-75 to +14 <sup>d</sup>
Residence times of exudates in soil ( $\tau$ ) (days)	n/a	n/a	61.1 (44.7 to 77.3)	780	n/a
Delay between end of labeling and peak enrichment (D) (days)	n/a	n/a	2.6 (1.7 to 3.5)	780	n/a

a. (Chanton et al. 1992; Galand et al. 2010; Krohn et al. 2017)

b. (Bender, 1971; Nielsen et al., 2017; O'Leary, 1988)

- c. (Chanton et al. 1992; Popp Trevor J. et al. 1999)
- d. (Games et al. 1978; Gelwicks et al. 1994; Botz et al. 1996; Whiticar 1999; Valentine et al. 2004; Londry et al. 2008; Feisthauer et al. 2011; Neumann et al. 2015) Net fractionation calculated by adding fractionation effects of methanogenesis and methanotrophy reported in the various publications cited.

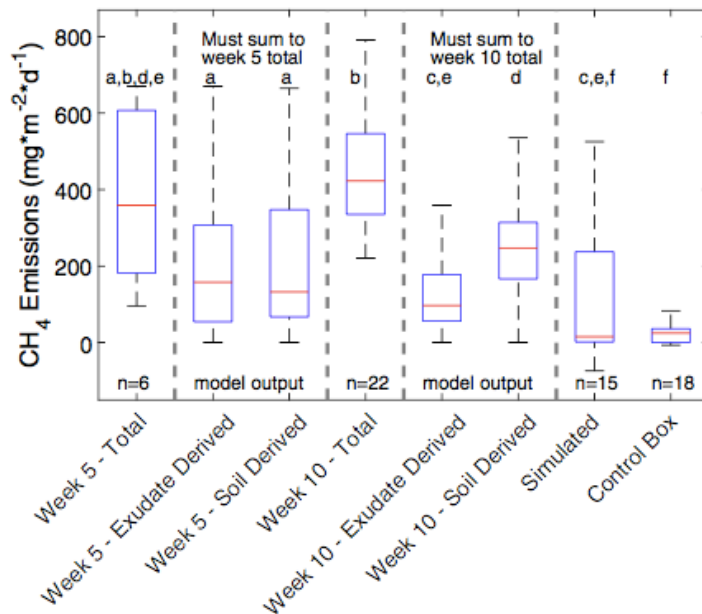


**Fig. 6** Results from isotope mixing model show the percentage of  $\text{CH}_4$  emitted from each box that was derived from root exudates, with the remainder derived from soil carbon. There was no significant difference between the two labeling events

Both soil and root exudates were responsible for a large portion of total  $\text{CH}_4$  emissions in both the first and second labeling events (Fig. 6). Note that only three boxes are presented from the week 5 labeling event. This is because one box harvested during that week did not produce any fluxes which passed isotopic quality control, and therefore could not be used in the model. The median exudate-derived portion of emitted  $\text{CH}_4$  from each box ranged from 22 to 68% in the week 5 labeling event with an overall median for that labeling event of 43%. During the week 10 labeling event, the median exudate-derived portion of  $\text{CH}_4$  emitted from each box ranged from 18 to 69%, with an overall median for the labeling event of 25%. In both cases, all of the  $\text{CH}_4$  that was not exudate-derived was soil-derived. In order to compare these values to the amount of  $\text{CH}_4$  emitted from simulated plants and control boxes, we multiplied the percentage of  $\text{CH}_4$  derived from each source during each flux by the total flux of  $\text{CH}_4$ . This converted the percentages given in Fig. 6 into the flux units shown in Fig. 7.

During both events, planted boxes emitted significantly more ( $p < 0.05$ ) soil-derived  $\text{CH}_4$  than the simulated plants or control boxes. The median values for soil-derived  $\text{CH}_4$  emissions were  $359 \text{ mg m}^{-2} \text{ d}^{-1}$  for week five,  $423 \text{ mg m}^{-2} \text{ d}^{-1}$  for week ten,  $15 \text{ mg m}^{-2} \text{ d}^{-1}$  for simulated-plant boxes, and  $25 \text{ mg m}^{-2} \text{ d}^{-1}$  for control boxes. During week 10 the planted boxes emitted significantly more ( $p < 0.05$ ) soil-derived

CH<sub>4</sub> than exudate-derived CH<sub>4</sub>, but during week 5 the amount of CH<sub>4</sub> emitted from the two carbon sources was not significantly different ( $p > 0.05$ ).



**Fig. 7** Box and whisker plots showing the total measured fluxes from planted boxes (Week 5 – Total and Week 10 – Total), the portion of that total which was calculated to be from each carbon source (Soil Derived and Exudate Derived), and measured emissions from the two unplanted box types (Simulated Plants and Control Boxes), which were entirely soil-derived. Outliers are not shown. Letters indicate groups that are statistically similar ( $p > 0.05$ ). Any two boxes which do not share a letter are significantly different ( $p < 0.05$ ) from each other.

# Discussion

## Biological and physical mechanisms for plants to increase CH<sub>4</sub> emission

The planted boxes' greater emission of CH<sub>4</sub> than control boxes (Fig. 3) confirmed that plants can increase CH<sub>4</sub> emissions, as has been shown previously (Shannon and White 1994; Joabsson et al. 1999; Popp et al. 2000; Whalen 2005), but by itself does not explain the mechanism. The use of simulated plants allowed for a direct comparison between a transport-alone scenario, where CH<sub>4</sub> and oxygen are transported without plant use, and real plants, which were biologically active and also transported gases. Gas transport in the simulated plants was clearly extensive, as demonstrated by oxygenation of the soil (Fig. 5), but CH<sub>4</sub> emissions were not significantly greater than those from the control boxes (Figs. 3 & 7). This finding is consistent with what other studies using similar methodology have found. King et al. (King et al. 1998) compared four treatments in a boreal wetland: natural plots with sedges and moss, plots with the sedges removed, plots with the moss removed, and plots with sedges removed and silicone tubes inserted. Turner et al. (under review) similarly utilized natural, sedge-removal, and simulated-plant plots in a field study. Both experiments found that the natural plots had high CH<sub>4</sub>

emissions, while the sedge-removal and simulated-plant plots both had similar, low emissions (King et al. 1998, Turner et al. (under review)). The moss-removal plots in the King et al. (1998) study emitted as much CH<sub>4</sub> as the natural plots, indicating that it is the vascular plants which made the difference. Our results reinforce that gas transport alone does not increase emissions.

Plants do more than simply transport gas. They consume some oxygen within their root tissues (Armstrong 1971), and they also release root exudates which can affect oxygen dynamics in the soil by stimulating microbial respiration (Popp et al. 2000; Ding et al. 2004). The real plants in our experiment did not oxygenate the rhizosphere (Fig. 4) but they did greatly increase CH<sub>4</sub> emissions relative to the control boxes (Fig 4). The fact that the real and simulated plants were different in both regards indicates that the biological effects of the plants were key, and gas transport only increased emissions in conjunction with other root impacts. The importance of biological factors is further supported by the fact that the plants switched between a lower-emission state and a higher-emission state quickly (i.e., within two weeks), while their height increased in a nearly linear fashion. This switch indicates that the emissions increase caused by the plants was not due to an effect that scales linearly with plant size, such as the gas transport capacity of aerenchyma. Instead, there must have been some change in the plants or rhizosphere which drove the increase in emissions. Examples of variables which could have changed to cause such an increase include the rate of root exudation, composition of root exudates (Girkin et al. 2018b), or composition of the microbial community, which is dependent (among other things) on plant growth stage (Houlden et al. 2008).

Root exudates can increase CH<sub>4</sub> emissions through two effects: either increasing CH<sub>4</sub> production (Ström et al. 2003; Ström and Christensen 2007; Picek et al. 2007; Chanton et al. 2008; Kayranli et al. 2009) or decreasing CH<sub>4</sub> oxidation (Popp et al. 2000). These effects can take place through multiple mechanisms including direct processing of root exudates into substrates for methanogenesis, stimulating the growth of heterotrophic microbes which compete with methanotrophs for electron acceptors such as oxygen, and/or triggering microbial priming which can include changes to the composition and size of the microbial community as well as changes to the soil chemical environment. All three of these mechanisms could have occurred in this study.

#### *Root exudate conversion to CH<sub>4</sub>*

Direct conversion of root exudates into CH<sub>4</sub> is the most straight-forward mechanism by which exudates can increase CH<sub>4</sub> emissions. The sharp rise in isotopic enrichment of CH<sub>4</sub> after labeling (Fig. 5A) showed that root exudates fueled methanogenesis, and the isotope model results (Fig. 7) show that

exudates were used in conjunction with soil carbon. Exudates did not diminish or replace use of soil carbon. The fact that the two labelling events produced different rhizosphere enrichments (Fig. 5B) from similarly enriched roots (Table 2) implies that the amount of exudates being emitted from the roots increased between the two labeling events, consistent with previous research showing root exudation is correlated with plant productivity (Weigel et al. 2005). The use of exudates to ultimately fuel CH<sub>4</sub> production provides an explanation for why CH<sub>4</sub> emissions are often correlated with primary productivity (Whiting and Chanton 1992).

However, the isotope model showed that while the emitted CH<sub>4</sub> was much more isotopically enriched than unlabeled emissions, it was not as enriched as it would have been if exudates were the primary carbon source (Fig. 7). The model attributed a large portion of emissions from planted boxes to soil-derived carbon. During week 10, more of the CH<sub>4</sub> was soil-derived than exudate-derived. In fact, the model showed that much more soil-derived CH<sub>4</sub> was emitted from planted boxes than from control or simulated boxes, so utilization of root exudates alone cannot fully explain the greater CH<sub>4</sub> emissions from planted boxes relative to the other treatments. There are two potential ways that plants could have increased emissions of soil-derived CH<sub>4</sub> in this experiment. They could have reduced methanotrophy, or they could have stimulated increased production of soil-derived CH<sub>4</sub>.

#### *Plants May Have Reduced Oxidation of Soil-Derived CH<sub>4</sub>*

Exudates may have increased emissions by decreasing methanotrophy through increased competition for oxygen (Lenzowski et al. 2018). This effect is a synergy between root exudation and the gas transport effects of aerenchyma. There is significant existing literature showing that plant transport of CH<sub>4</sub> is substantial and can increase total emissions (Shannon and White 1994; Joabsson et al. 1999; Popp et al. 2000; Whalen 2005). Other studies have found that real plants can oxygenate the soil and oxidize CH<sub>4</sub> to CO<sub>2</sub> (Schipper and Reddy 1996; Fritz et al. 2011; Lenzowski et al. 2018), and in some cases soil oxygenation is extensive enough that oxidation reduces CH<sub>4</sub> emissions to zero (Fritz et al. 2011). Gas transport by plants can therefore increase CH<sub>4</sub> oxidation when it creates an oxic zone around roots, or decrease CH<sub>4</sub> oxidation when there is no oxic zone around roots and the aerenchyma allow CH<sub>4</sub> to diffuse from the anoxic rhizosphere directly to the atmosphere. Whether the rhizosphere is oxygenated or not depends on the balance between oxygen supply and demand, which is dictated by the biology of the plant species in question, as well as by carbon bioavailability and microbial ecology of the soil. The lack of rhizosphere oxygen in our system (Fig. 4) is consistent with a field study at the site where materials for this experiment were sourced, which utilized field-deployed optodes to study belowground oxygen dynamics (Turner et al., under review), and with previous work showing that *Carex aquatilis*

transport gases less than other aerenchymatous sedges (Schimel 1995). We did not directly test rates of methanotrophy in this study, but the lack of oxygen in the rhizosphere of the real plants and readily available oxygen surrounding tubes of simulated plants indicates that real plants likely reduced methanotrophy relative to simulated plants. Reducing methanotrophy can take multiple forms however. The utilization of oxygen can take place either within roots or in the soil, and competition for oxygen can occur either between microbial species or between metabolisms within facultative methanotrophs.

More oxygen is utilized along the transport pathway in plants than in abiotic tubes because the roots themselves need oxygen for respiration (Armstrong 1971). If this oxygen demand is strong enough relative to the transport capacity of the aerenchyma, the roots could effectively become selective gas-transport pathways where  $\text{CH}_4$  can diffuse to the atmosphere but  $\text{O}_2$  cannot diffuse to the soil. There is also another mechanism for increasing oxygen utilization: increasing demand in the soil through root exudation of carbon which stimulates aerobic microbial activity (Mueller et al. 2016; Lenzewski et al. 2018). The isotopic enrichment of  $\text{CH}_4$  after labeling indicates that microbes did utilize carbon derived from root exudates. While this result indicates anaerobic metabolism, any available oxygen would have allowed microbes to also aerobically metabolize root exudates, thus competing with methanotrophy for limited oxygen.

Note that we have framed this discussion as competition for oxygen between methanotrophy and other metabolisms. While most species of methanotrophs are obligate methanotrophs (Conrad 2009), and therefore would be competing for oxygen against other aerobic microbes, there do exist facultative methanotrophs which can utilize multiple carbon compounds (Dedysh et al. 2005). The list of other carbon compounds which these facultative methanotrophs can utilize is limited (Dedysh et al. 2005; Crombie and Murrell 2014), but when available, the microbes preferentially use the non-methane compounds (Theisen et al. 2005; Wieczorek et al. 2011). It has also been suggested that organic acids may inhibit methanotrophy through toxic as well as competitive effects (Wieczorek et al. 2011). In the oxygen-limited rhizosphere of this study, it is possible that the addition of root exudates triggered facultative methanotrophs to utilize non-methane substrates and therefore oxidize less  $\text{CH}_4$ , in addition to the inter-species competition for oxygen which likely occurred.

However, we reason that the combination of reduced methanotrophy and utilization of exudates was not enough to account for the full increase in  $\text{CH}_4$  emissions caused by plants relative to the other treatments. A study at the field site from which peat for this experiment was obtained directly investigated  $\text{CH}_4$  oxidation by measuring  $\text{CH}_4$  emissions under anaerobic conditions, achieved by flushing the fluxing chamber headspace with  $\text{N}_2$  (Turner et al., under review). That study found that a maximum

of 36% of produced CH<sub>4</sub> was oxidized prior to emission in plots without vascular plants, regardless of whether they had simulated plants or not (Turner et al., under review). Assuming similar rates of methanotrophy in our unplanted boxes, the simulated plants and control boxes would have produced a median of 31 and 38 mg CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup> respectively, still an order of magnitude less than the 237 mg m<sup>-2</sup> d<sup>-1</sup> of soil-derived CH<sub>4</sub> that was emitted from planted boxes (median of all data). This 88% excess soil-derived CH<sub>4</sub> emitted by planted boxes relative to other treatments could only have been the result of roots triggering increased use of soil carbon.

#### *Root Effects on Utilization of Soil Carbon*

There are two mechanisms by which roots may enhance anaerobic soil carbon processing: the creation of mixed-redox zones or microbial priming. The mixed-redox mechanism is a result of transitions in either time or space between oxic and anoxic environments (Canfield 1994; Aller 1998; Chanton et al. 2008). In these transitional regions, oxygen may be quickly utilized and so not be present, (as was the case in the rhizosphere in this study (Fig. 4)), but still have the effect of raising the overall redox state of the location and facilitating the creation of alternate electron acceptors (Keiluweit et al. 2016). If these additional electron acceptors were used to fully mineralize soil carbon to CO<sub>2</sub> then we would see no increase in CH<sub>4</sub> production. However, it has also been found that in mixed-redox zones, oxygen and other thermodynamically favorable electron acceptors are used to partially oxidize large molecules, and the resulting smaller molecules are further processed anaerobically (Chanton et al. 2008; Corbett et al. 2015). If such partial processing occurred in the rhizosphere of our study, then it could explain some of the increased microbial conversion of soil carbon to CH<sub>4</sub>.

However, oxygen visualization around the simulated plants demonstrate that partial processing of soil organic matter was likely only a minor effect in this study. Mixed redox zones could have existed around the oxygenated soil surrounding the simulated plants (Fig. 4). The fact that CH<sub>4</sub> emissions from simulated-plant boxes were not significantly greater than control boxes (Figs. 3 & 7), paired with evidence from the field investigation that rates of methanotrophy were similar between unplanted field plots with and without simulated plants (Turner et al., under review), indicates that any mixed-redox zone surrounding the simulated plants did not substantially increase CH<sub>4</sub> production. Lacking an explanation for why the mixed-redox zones would have been larger or more influential around real roots than around simulated plants, we conclude that priming is the mechanism that better explains the enhanced CH<sub>4</sub> emissions.

Priming is a broad term encompassing any process by which the addition of a different carbon source (e.g., root exudates or leaf litter) increases microbial utilization of soil carbon. Priming can happen by stimulating an increase in microbial biomass, a change in composition of the microbial community, and/or a change in what metabolisms are active within the microbial community (Craine et al. 2007; Kuzyakov 2010; Ruirui et al. 2014). Priming has been observed in methanogenic systems before, such as one incubation experiment which found that the addition of rice straw to peat soils greatly increased soil-derived CH<sub>4</sub> production (Ye et al. 2015). In our experiment, priming provides a straightforward explanation of why real plants emitted such a large amount of soil-derived CH<sub>4</sub> relative to the simulated plants (Figs. 3 & 7). Notably, during the week 10 labeling event, more CH<sub>4</sub> was soil-derived than exudate-derived (Figs. 6 & 7), indicating a notable priming effect. During the week 5 event there was still priming, but the soil-derived and exudate-derived portions were about equal. Based on rhizosphere enrichment data (Fig. 5), we know that there was more root exudation happening during week 10. Together, the data indicate that more root exudation led to a greater portion of the total emissions being soil-derived (Figs. 6 & 7).

If the priming effect is equally powerful in natural systems, it could have profound environmental impacts. A field study in a Boreal peatland, similar to the one upon which this study is based, found that vascular plants contributed to increased microbial processing of soil organic matter, but did not directly link that activity to CH<sub>4</sub> (Robroek et al. 2016). In our laboratory study we have shown that CH<sub>4</sub> was generated from soil carbon that was processed because of plants. Furthermore, because root exudation and wetland CH<sub>4</sub> production are correlated with plant productivity (Whiting and Chanton 1992; Weigel et al. 2005), our findings imply that as plant productivity in temperate northern latitudes increases with climate change (Forkel et al. 2016), so will the conversion of soil carbon to CH<sub>4</sub>. It has long been known that CH<sub>4</sub> emissions are a positive feedback on climate-driven productivity changes (Meronigal and Schlesinger 1997), but wetlands have also been counted on as a carbon sink (Kayranli et al. 2009). Identifying peat soils as the source of the additionally emitted CH<sub>4</sub> implies that this ability to sequester carbon may not be as robust as believed. Northern peatlands contain 25% of the world's soil carbon (500 Gt C), despite only covering 2% of surface area (Yu 2012). For context, forecasts of global wetland CH<sub>4</sub> emissions by 2100 are 0.17 to 0.25 Gt G (Zhang et al. 2017), so the potential for peatland carbon to increase emissions is limited only by the rate at which it is converted. Further research into how priming could act as a positive climate feedback is warranted.

Another area of needed research highlighted by this study is an examination into why we found evidence of priming while other recent studies have failed to find evidence for it (Girkin et al. 2018a) or found evidence that root exudates protect soil carbon, which is the opposite of priming (Graham et al. 2017). The data presented here are inadequate to explain why or when priming does or does not occur. One explanation could be that priming only occurs in certain circumstances depending on the nutritional needs of microbes (Craine et al. 2007), soil properties (Jones et al. 2003), or other environmental variables. It is also possible that the definition of priming used in each study causes discrepancies. For example, Graham et al. (2017) determined that in the presence of plants microbial processing of dissolved organic carbon (DOC) increased but processing of physically bound carbon decreased. This approach does not distinguish plant-derived DOC from soil-derived DOC, and so is answering a different question than that addressed in this study. In order to elucidate the mechanisms of priming we must understand what microbes are involved, and what types of soil-derived molecules are additionally utilized during priming.

## **Conclusions**

In our experiment root exudates drove CH<sub>4</sub> production, but not through direct utilization alone. An isotope mixing model showed that a large portion of the increase in CH<sub>4</sub> emissions caused by plants was fueled by soil-derived carbon. Our data indicate root exudates increased the amount of soil-derived CH<sub>4</sub> that was emitted by increasing O<sub>2</sub> demand and thus reducing methanotrophy and/or increasing CH<sub>4</sub> production from soil-derived carbon by stimulating microbial priming. We did not measure methanotrophy directly, but its reduction in planted boxes relative to the simulated plants was supported by the lack of oxygen detected in the rhizosphere. The simulated plants we used for comparison did not emit significantly more CH<sub>4</sub> than control boxes and did have large oxygenated zones around their roots. The comparison between the real and simulated plants shows that transport of gases alone cannot increase CH<sub>4</sub> emissions without other biological activity. However, the amount of soil-derived CH<sub>4</sub> emitted from the planted boxes was an order of magnitude greater than what was likely ever produced in unplanted boxes, based on prior estimates of methanotrophy. Increased production of soil-derived CH<sub>4</sub> was the only explanation for the unattributed carbon that was emitted from planted boxes, which comprised over half of the total emissions from those boxes. That increase in production is best explained by microbial priming. An order of magnitude increase in conversion of soil carbon to CH<sub>4</sub> was driven by plant growth, which is projected to increase in the boreal region under forecasted climate conditions. The presence of such a large priming effect implies that increased plant productivity could potentially lead to increased conversion of soil carbon to CH<sub>4</sub> on climatically relevant scales.

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