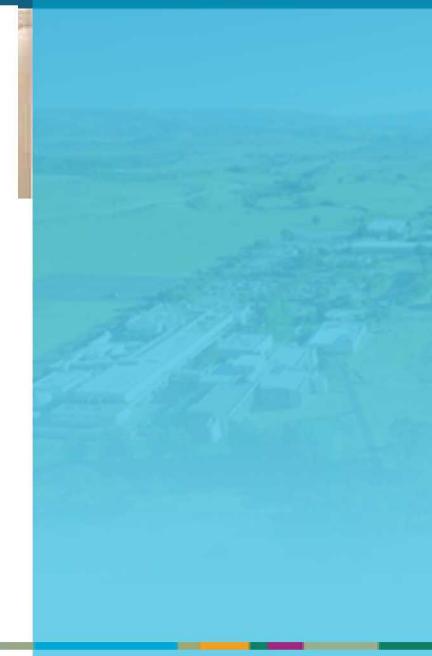


# Characterizing nano-scale helium bubbles within metals by electron energy loss spectroscopy

**Noelle R. Catarineu**, David B. Robinson, Norman C. Bartelt, Joshua D. Sugar, Xiaowang Zhou, Mark R. Homer, Warren L. York, Suzanne Vitale (SNL/CA)

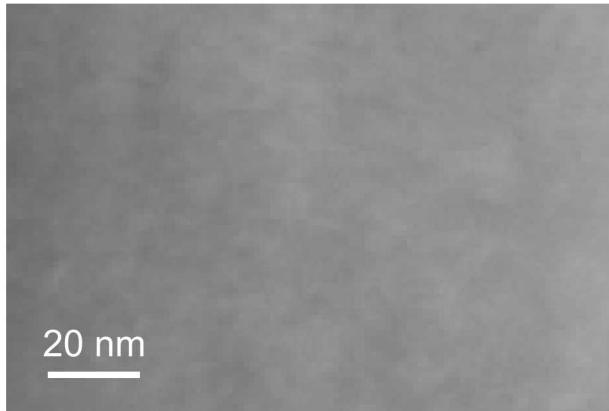
Caitlin A. Taylor, Clark S. Snow, Brittany R. Muntifering, Khalid M. Hattar (SNL/NM)

E. Lynn Bouknight, Kirk L. Shanahan (SRNL)

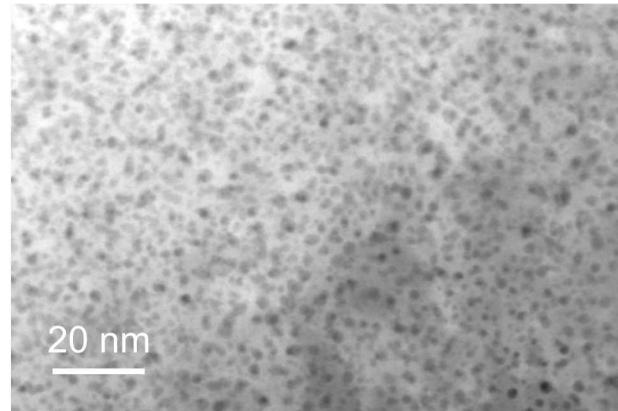


Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.

## Tritium decays within metals to form nanobubbles of helium-3



Pd-Ni never exposed to tritium

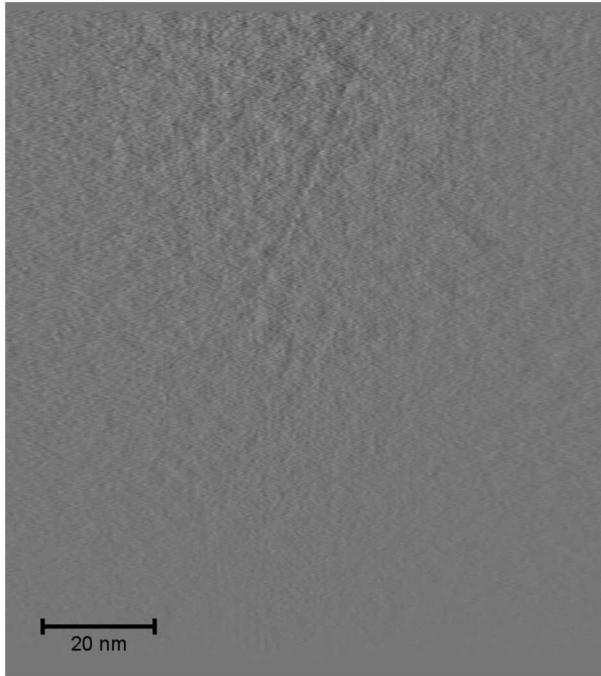


Pd-Ni stored under tritium for 4 years

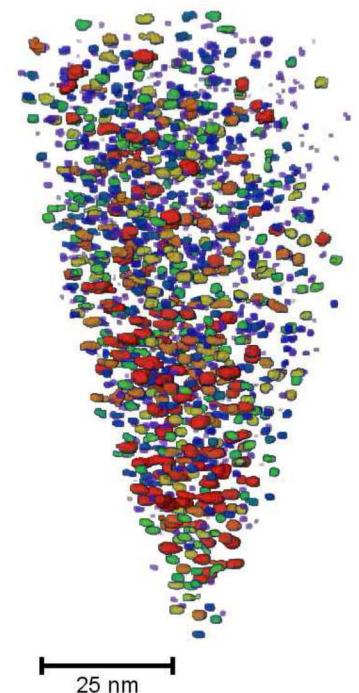
### Helium bubble formation in tritium-exposed metals:

1. Tritium radioactively decays to helium-3
2. The insoluble helium precipitates into bubbles
3. Growth of helium bubbles causes blistering and fracture of metals
4. Metal degradation leads to helium release and loss of structural integrity

3 High resolution STEM and tomography have improved imaging of bubbles



Tomographic reconstruction of Pd-Ni



He bubbles in Pd-Ni

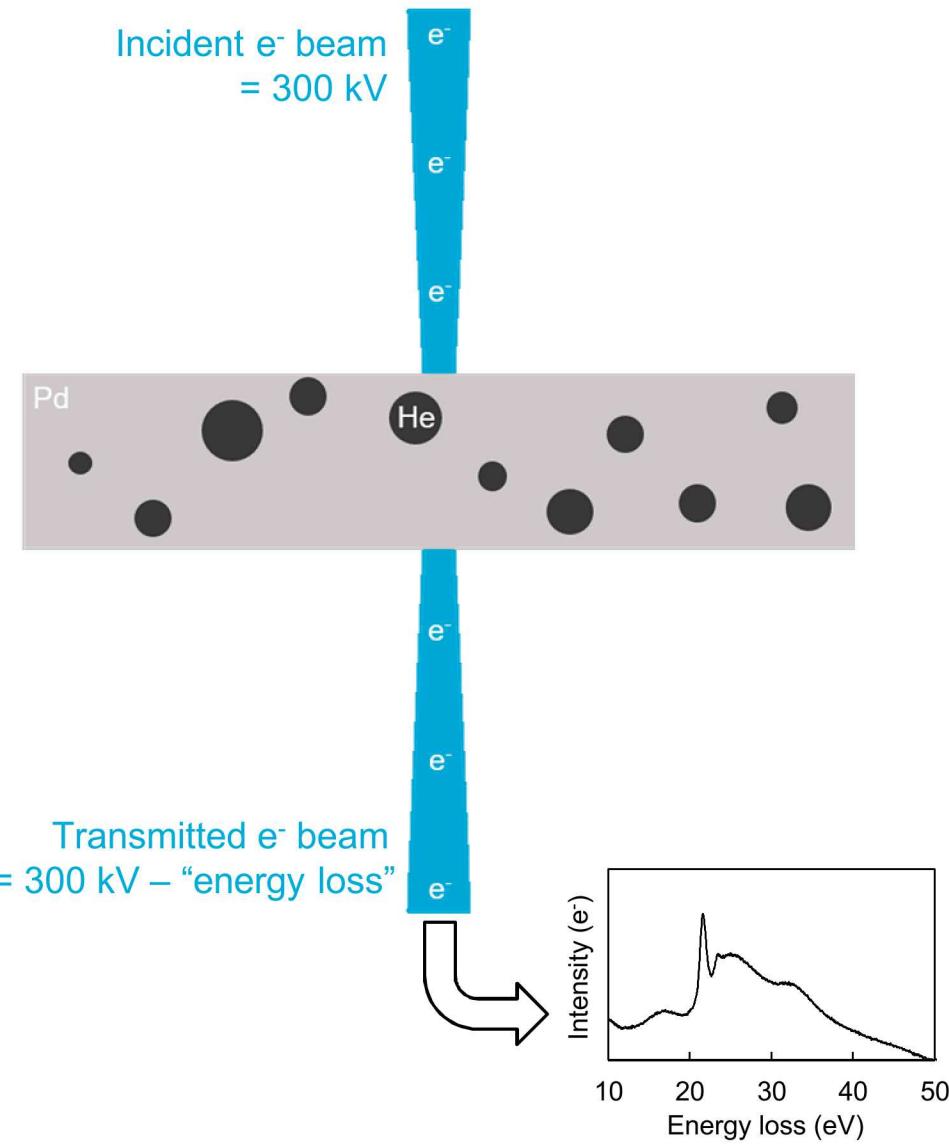
HR-STEM and electron tomography reveal:

- Total number of bubbles
- Bubble spatial distribution in 3D
- Bubble volumes, but...

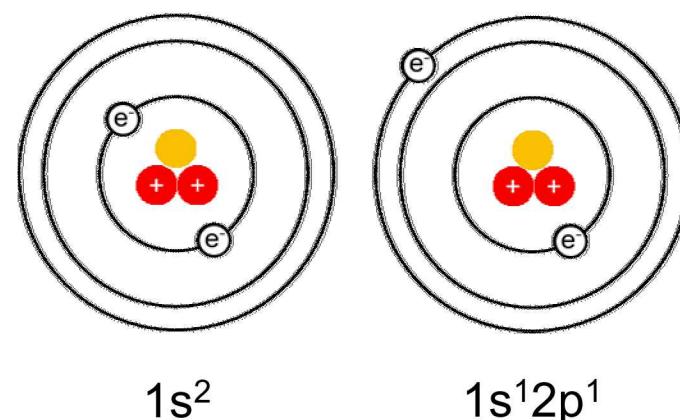
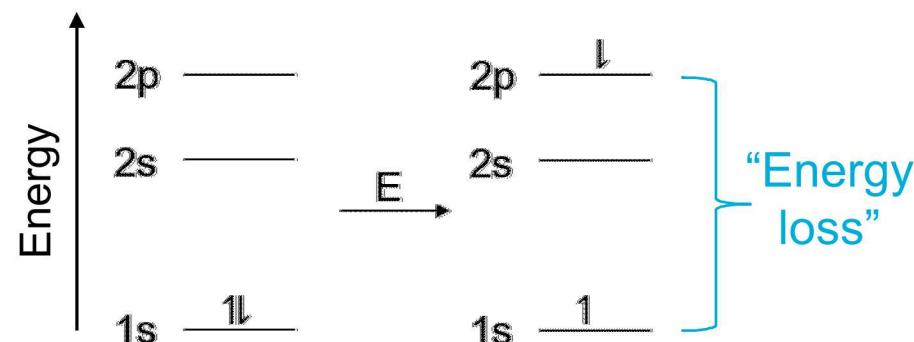
**How much helium is in the bubbles? What is the internal bubble pressure?**

# Electron energy loss spectroscopy provides elemental composition

## EELS measurement in a TEM



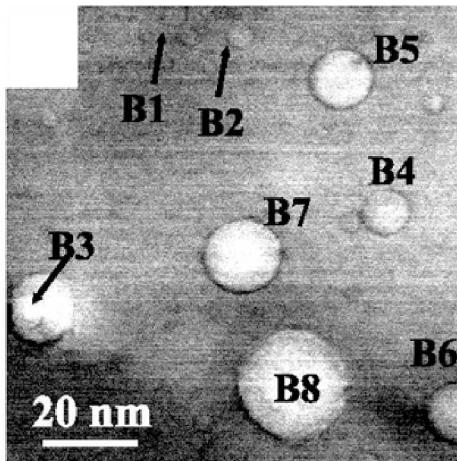
## $^3\text{He}$ 1s $\rightarrow$ 2p electronic transition



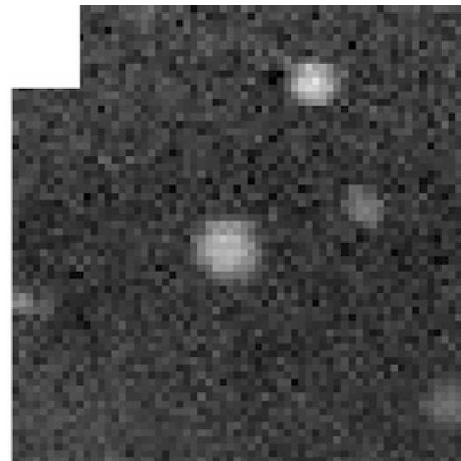
## Helium detection by EELS can be achieved for bubbles >10 nm

Previous work by others indicates difficulty in detecting helium within nanobubbles:

- Nano-scale bubble diameters → Low intensity He K-edge
- High internal bubble pressures → Overlap of He K-edge with metal plasmon peaks

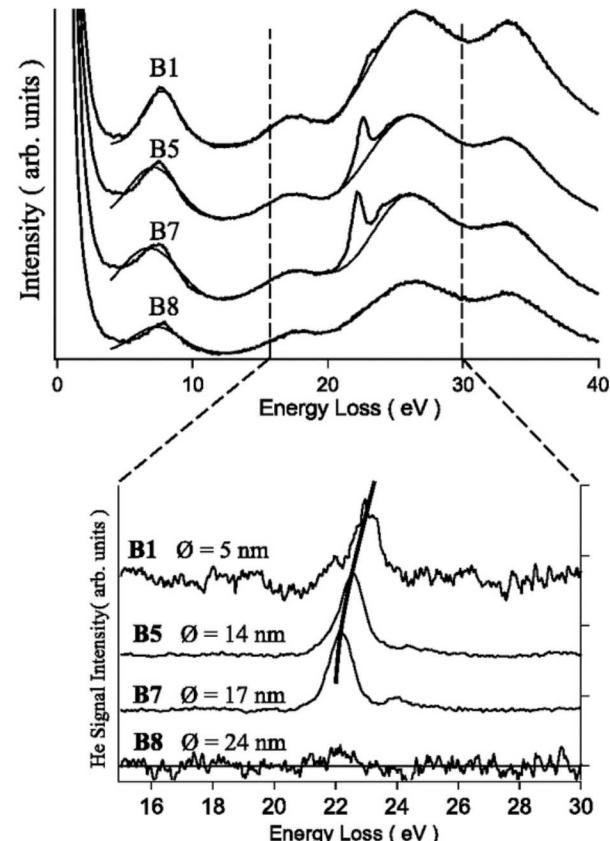


TEM image of He bubbles



Map of He K-edge

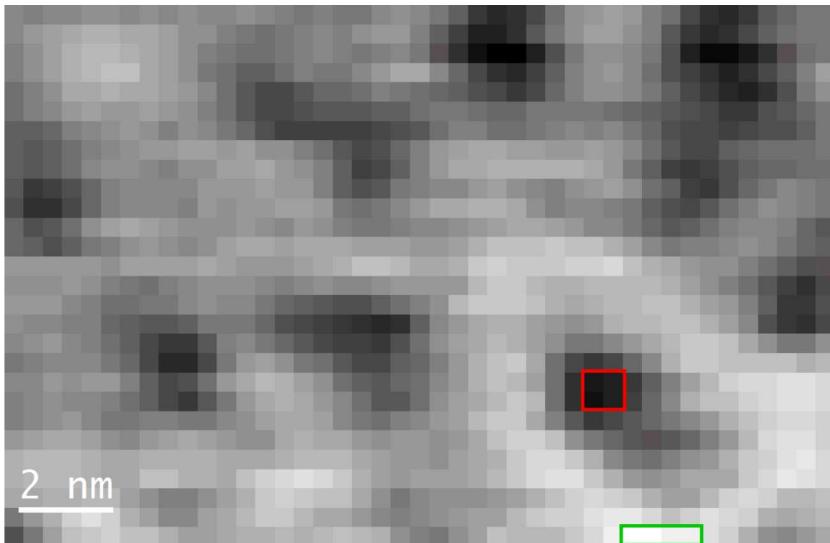
Energy loss spectra of various bubbles



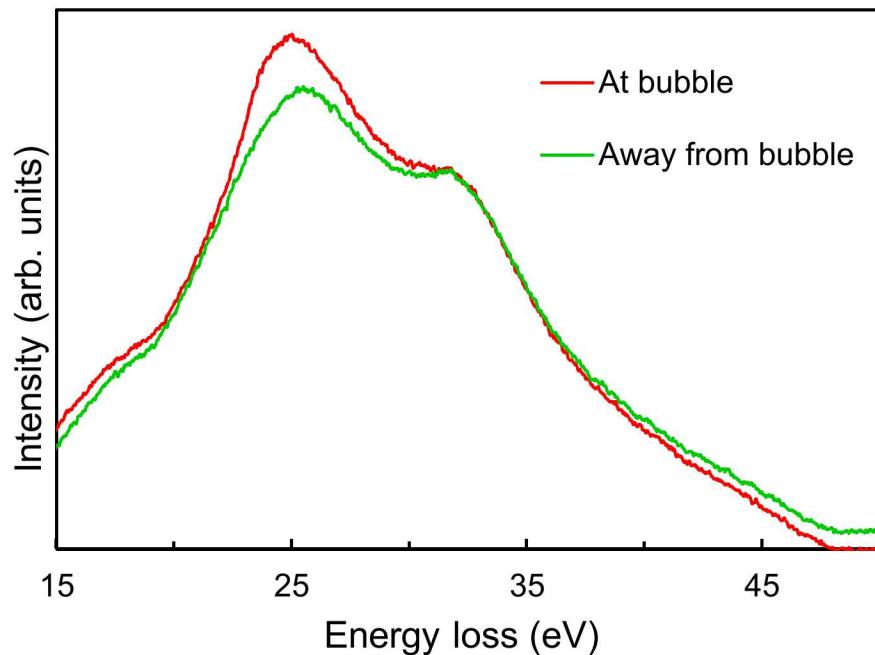
Spectra after subtraction of metal plasmon peaks

## He detection by EELS of tritiated sample with 1-2 nm bubbles

Dark field image of tritiated  $\text{Pd}_{95}\text{Ni}_5$



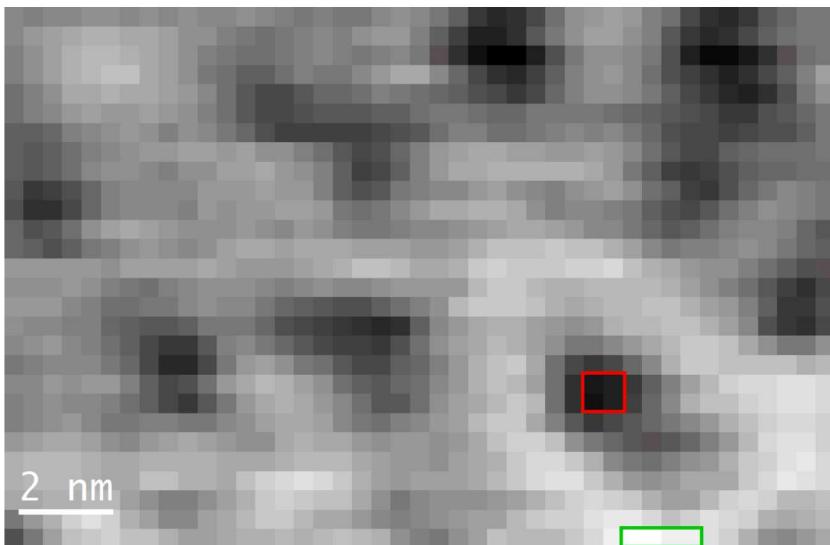
EELS of Tritiated Pd-Ni



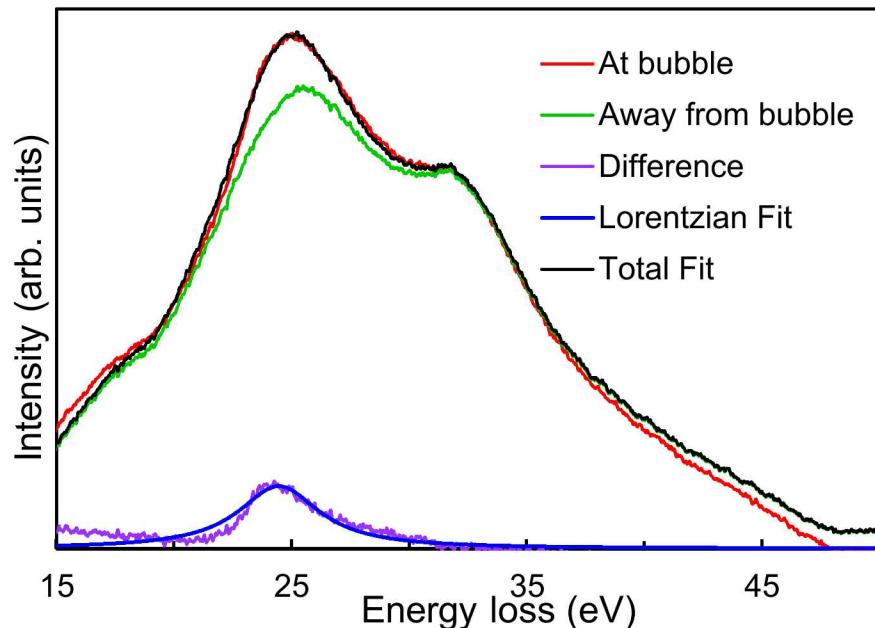
- $\text{Pd}_{95}\text{Ni}_5$  sample aged for 4 years under tritium contains bubbles roughly 2 nm in diameter
- Clear increase in intensity at 25 eV in bubble region compared to  $\text{Pd}_{95}\text{Ni}_5$  background measured away from bubbles
- Helium K-edge may overlap with Pd plasmon

## Difference in bubble and background spectra was fit by Lorentzian

Dark field image of tritiated  $\text{Pd}_{95}\text{Ni}_5$



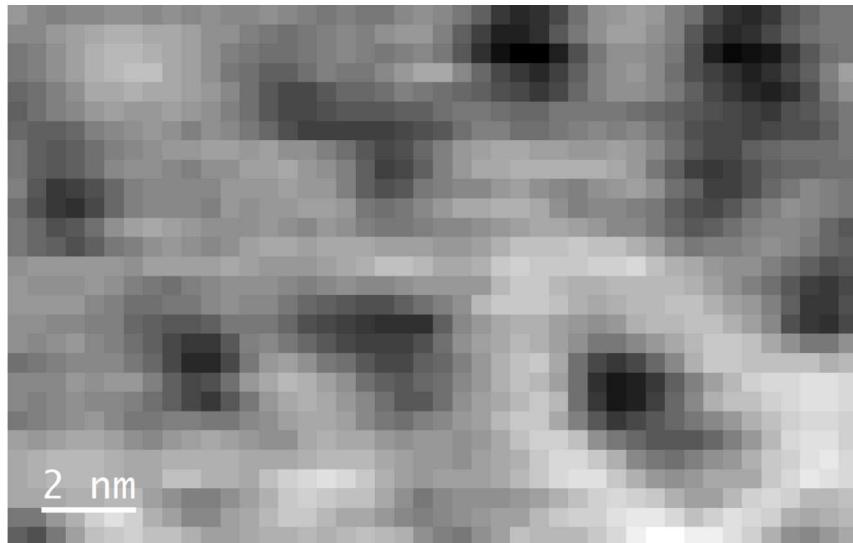
Least squares fit of bubble spectrum



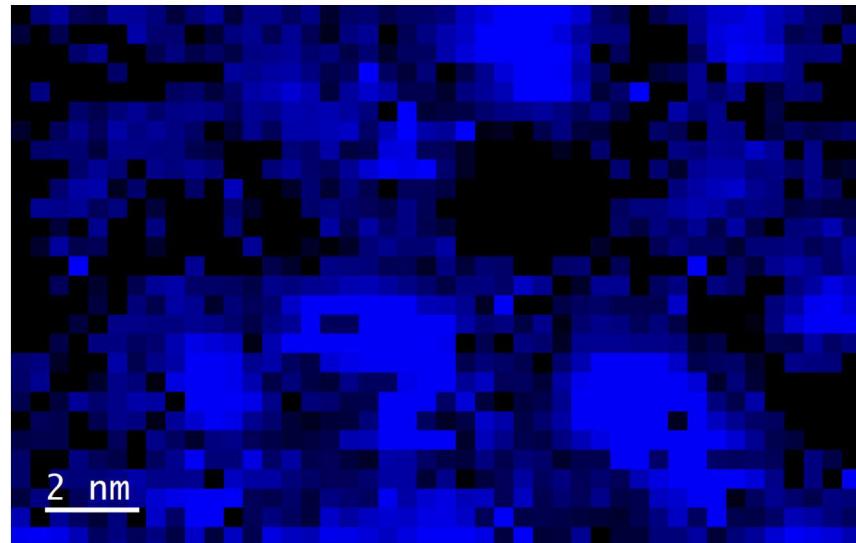
- Bubble spectrum after subtraction of Pd-Ni background was well fit by a Lorentzian function
- Non-linear least squares fit with Lorentzian function and Pd-Ni background matches spectrum measured at bubble
- Lorentzian function represents additional intensity due to helium

## Lorentzian function represents helium and maps to bubble regions

Dark field image of tritiated  $\text{Pd}_{95}\text{Ni}_5$



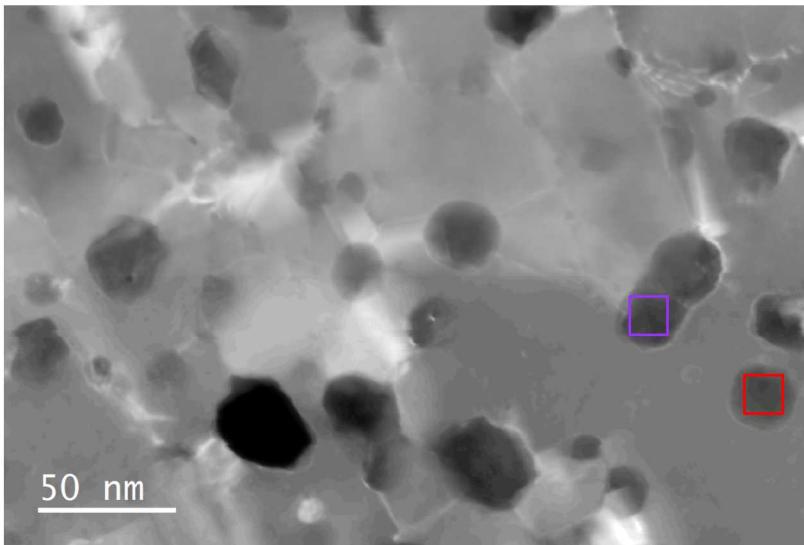
Lorentzian amplitude over background fit



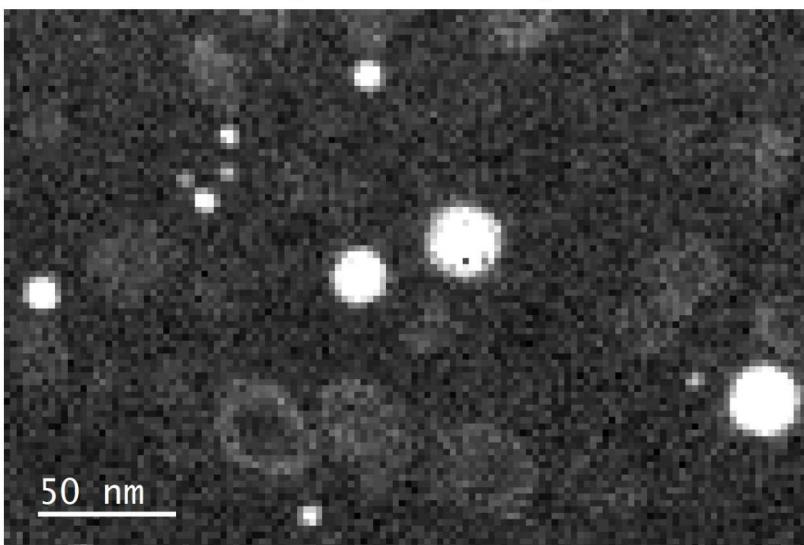
- Amplitude of Lorentzian function represents concentration of helium
- Positions of bubbles correspond with increase in amplitude of Lorentzian

## He bubbles and voids have similar 25 eV Pd plasmon peaks

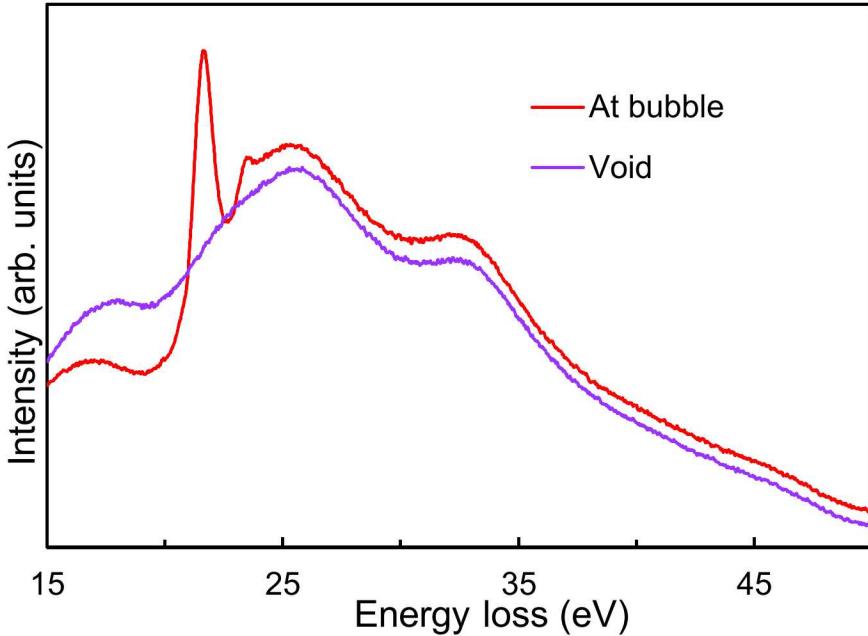
Dark field image of tritiated  $\text{Pd}_{95}\text{Ni}_5$



Map of He K-edge



EELS of Pd implanted with He



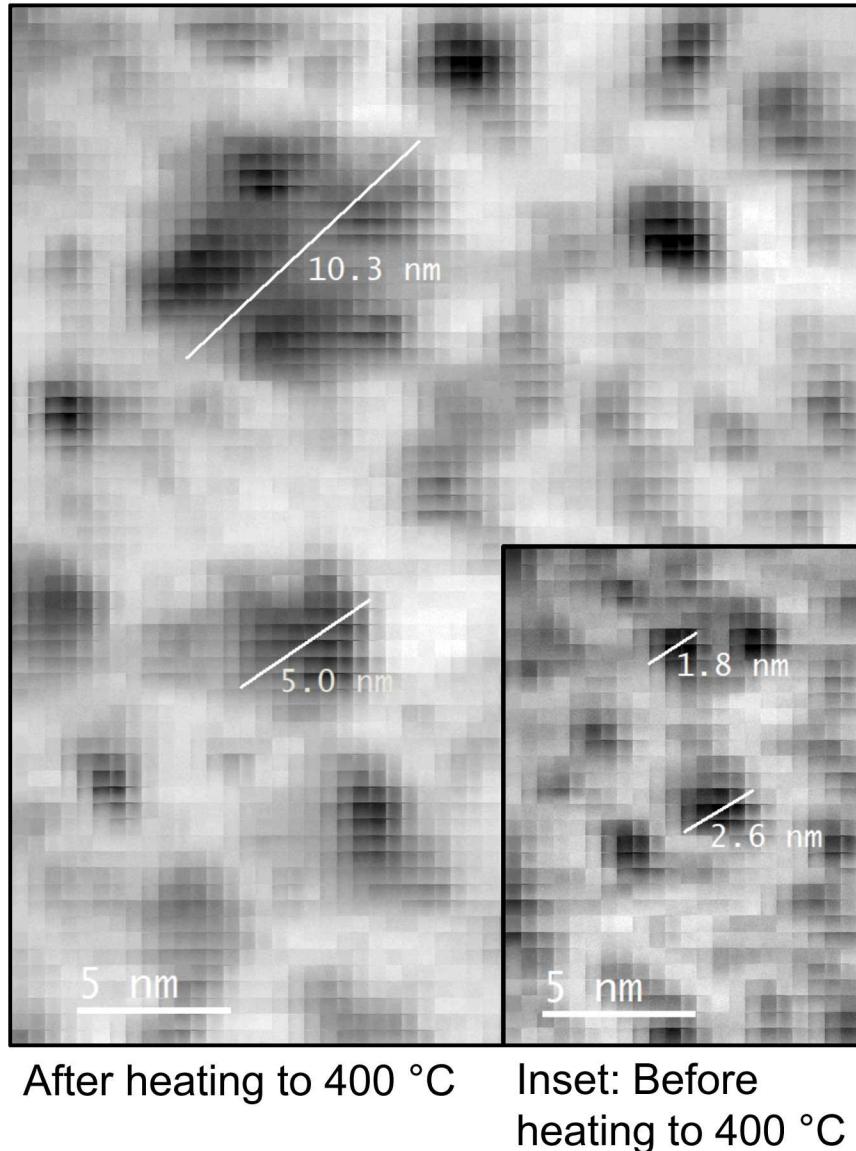
- Pd foil was implanted with  $1 \times 10^{17} \text{ He}^+/\text{cm}^2$  and annealed at  $600^\circ\text{C}$  for 2 hours
- Large bubbles formed with clear He K-edges
- Pd plasmon peaks at 25 eV are similar for He bubbles and voids

## Helium bubbles coarsen significantly upon heating to 400 °C

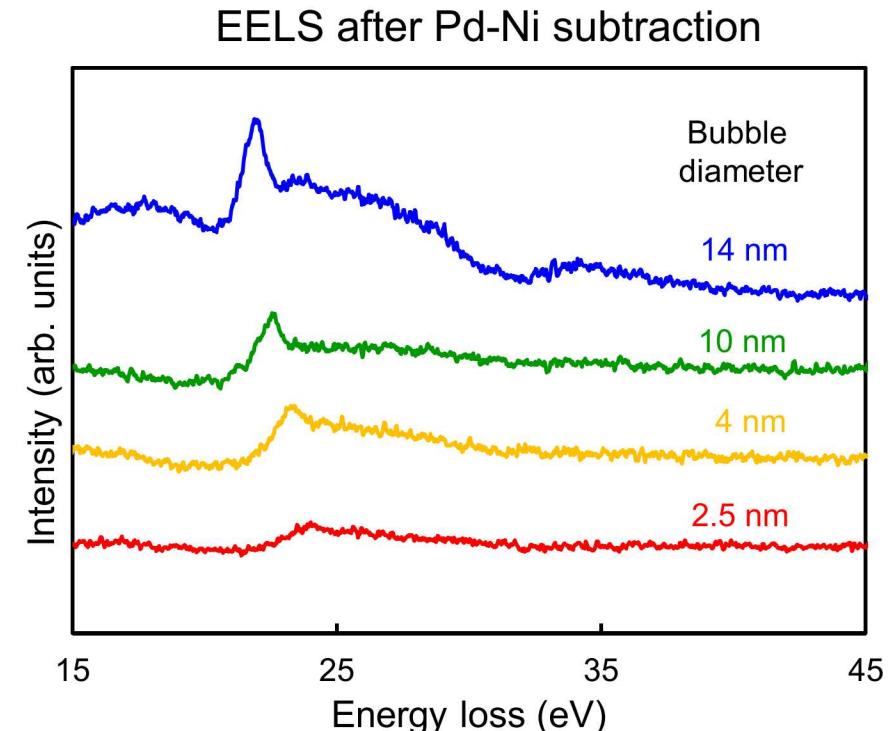
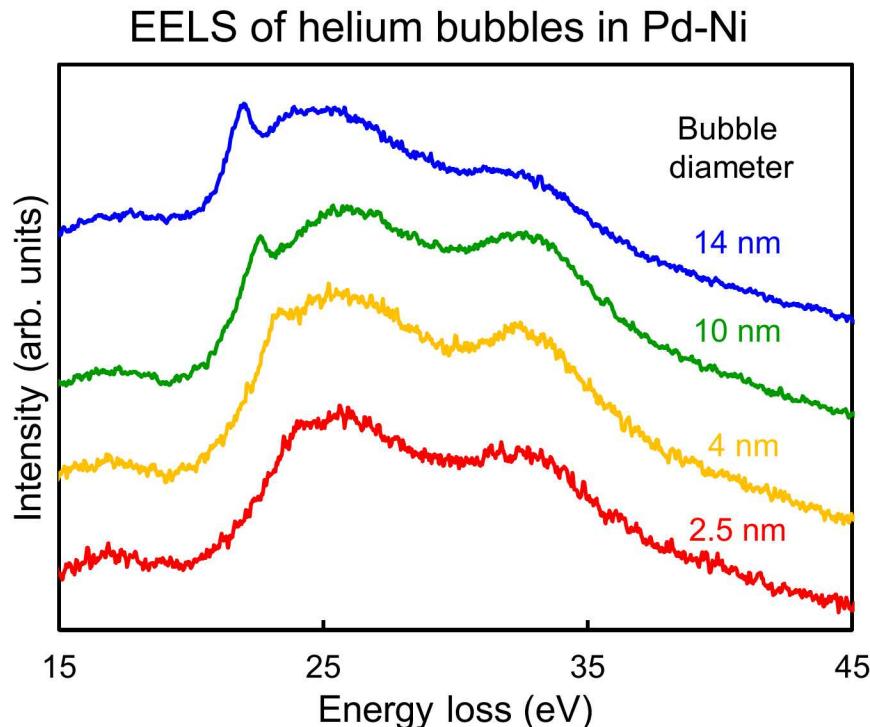
*In situ* heating procedure:

1. Heated to 300 °C  
→ No visible change
2. Heated to 400 °C  
→ Stark increase in bubble size  
→ Appearance of sharp EELS peaks at 22 to 24 eV
3. Cooled to 25 °C  
→ No visible change

STEM dark field images



## Larger bubble size corresponds with more distinct He K-edge



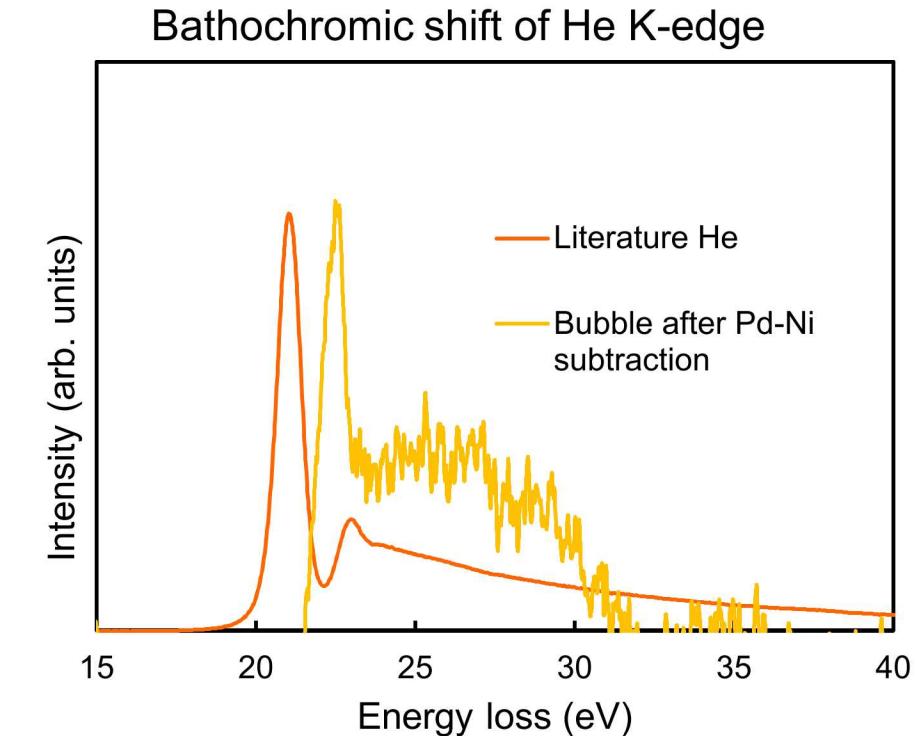
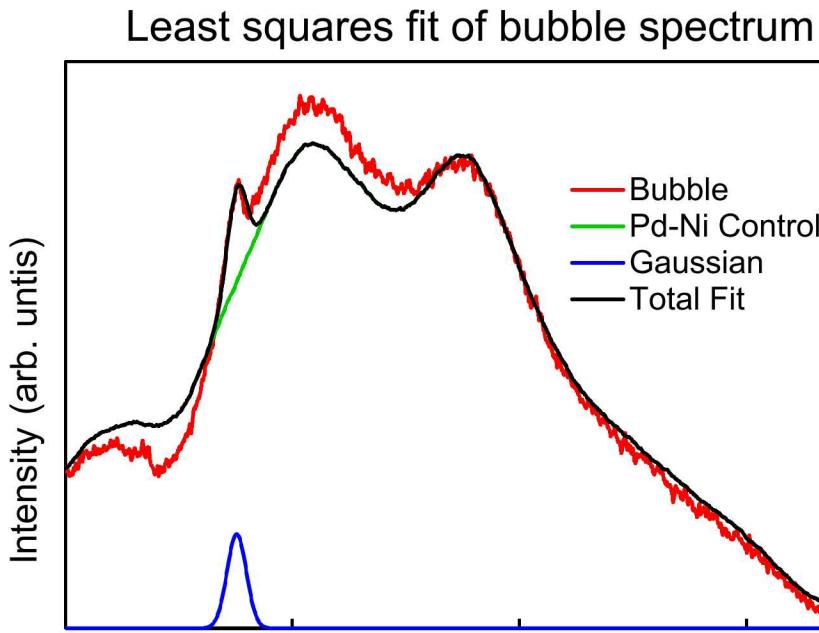
More helium atoms in electron beam path

→ Increase in intensity of helium K-edge

Lower helium excitation energy at higher pressure

→ Less overlap with palladium plasmon peaks

## Bubble spectra resemble a summation of Pd-Ni and helium K-edge



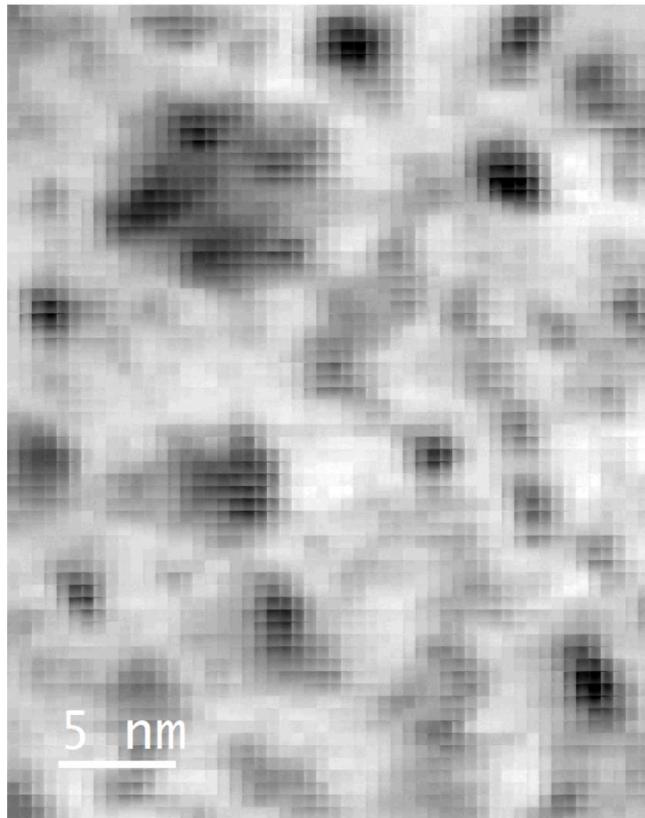
Bubble spectra are well-fit by:

- 1) Pd-Ni control never exposed to tritium +
- 2) Gaussian function representing helium

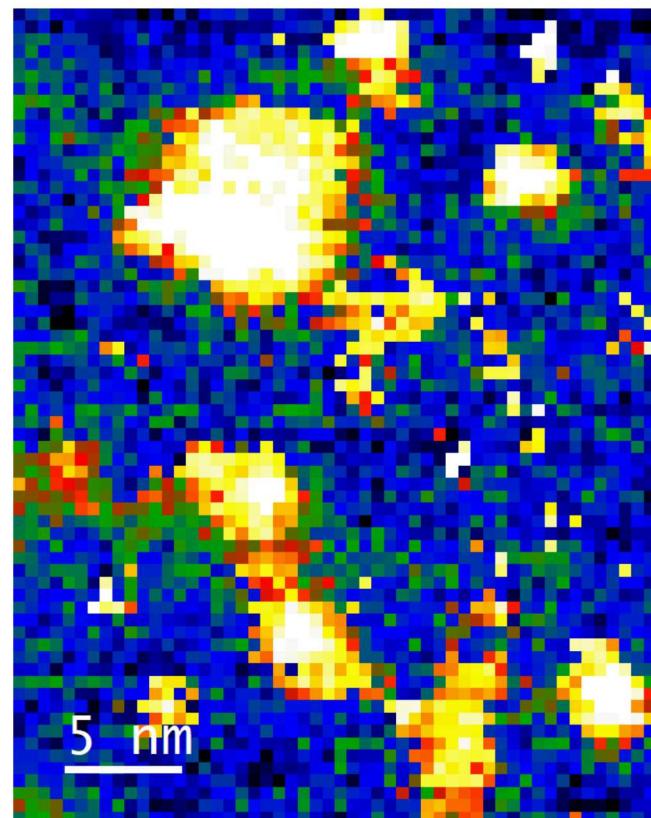
After subtraction of Pd-Ni, bubble spectrum resembles literature spectrum of helium

## Map of K-edge indicates bubbles are filled with helium

Dark field image of annealed  $\text{Pd}_{95}\text{Ni}_5$



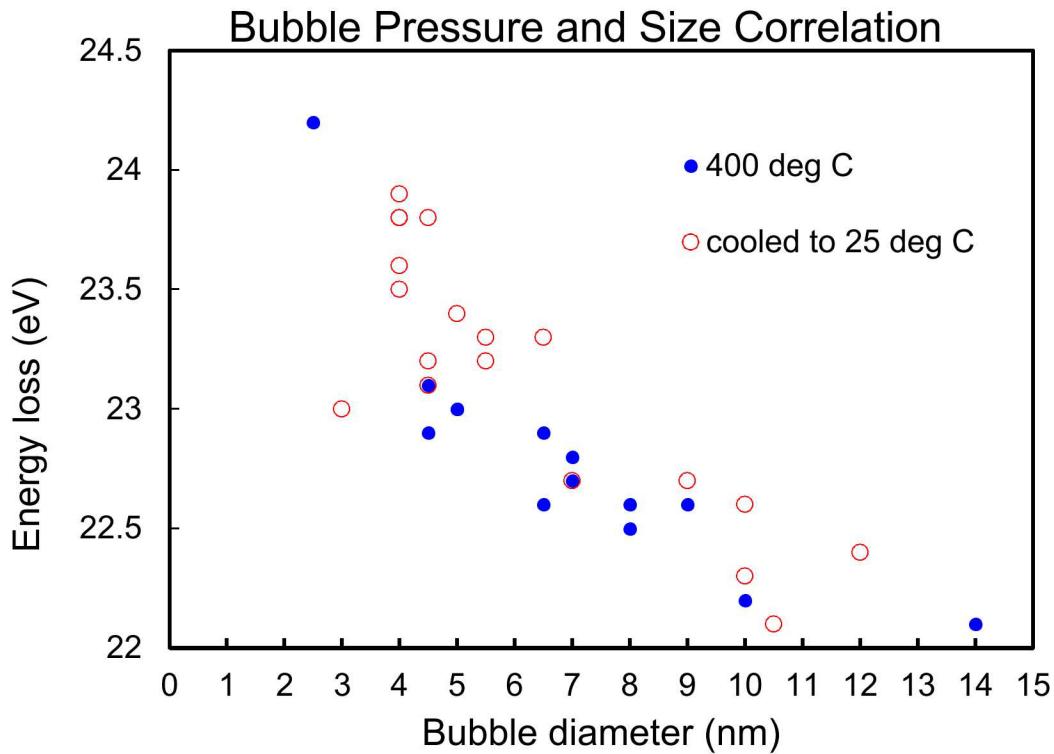
Amplitude of Gaussian corresponding to He



Map of helium K-edge reveals nearly all bubbles contain helium

*In situ* heating followed by EELS allowed detection of helium in a sample that grew bubbles only up to  $\sim 2$  nm

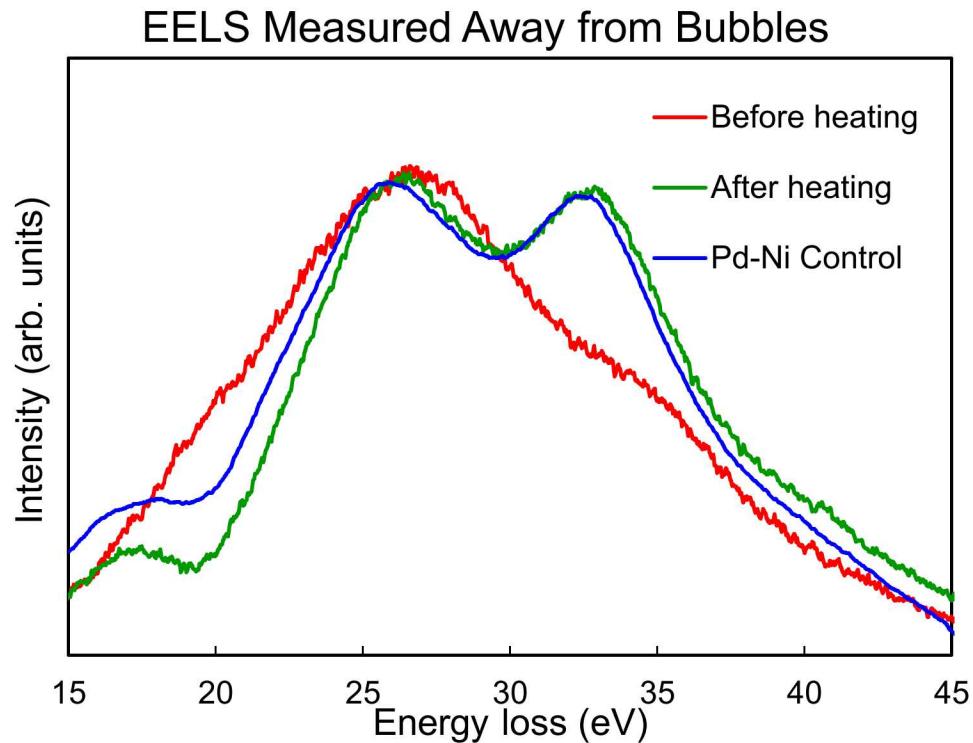
## Bubble size and helium pressure are strongly correlated



He K-edge of larger bubbles is at lower energies indicating lower internal pressure

Overlap of atomic orbitals of neighboring He atoms under high pressure increases helium  $1s \rightarrow 2p$  excitation

## Annealed Pd-Ni away from bubbles resembles Pd-Ni control



Post-heating spectra resemble Pd-Ni control; pre-heating spectra do not

Heating may liberate helium trapped in the bulk palladium lattice allowing it to migrate to bubbles

**Does helium remain in tritiated metals after several years?**

Yes, results indicate nearly all of the bubbles in tritiated  $Pd_{95}Ni_5$  contain helium

**How can we detect helium in <10 nm bubbles?**

Annealing to induce bubble coarsening followed by electron energy loss spectroscopy reveals the helium K-edge

**How can we measure the internal pressure and total helium content?**

Bubble size is inversely correlated with pressure; degree of bathochromic shift of helium K-edge indicates internal bubble pressure

**Is there helium in the metal lattice outside of bubbles?**

Changes in EELS intensity at 25 eV measured away from bubbles may result from overlap of helium K-edge with metal plasmon peak

## Acknowledgements

Sandia National Lab, CA team:



Dave Robinson

Norm Bartelt

Josh Sugar

Xiaowang Zhou

Warren York

Suzy Vitale

Sandia National Lab, NM collaborators:

Caitlin Taylor, Brittany Muntifering, Clark Snow, Khalid Hattar

Savannah River National Lab collaborators:

Lynn Bouknight, Kirk Shanahan

Sandia National Laboratories is a multimission laboratory managed and operated by National Technology & Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.

Work at Savannah River National Laboratory was performed under contract number DE-AC09-08SR22470 with the U.S. Department of Energy (DOE) Office of Environmental Management (EM).

# Figure Slide

