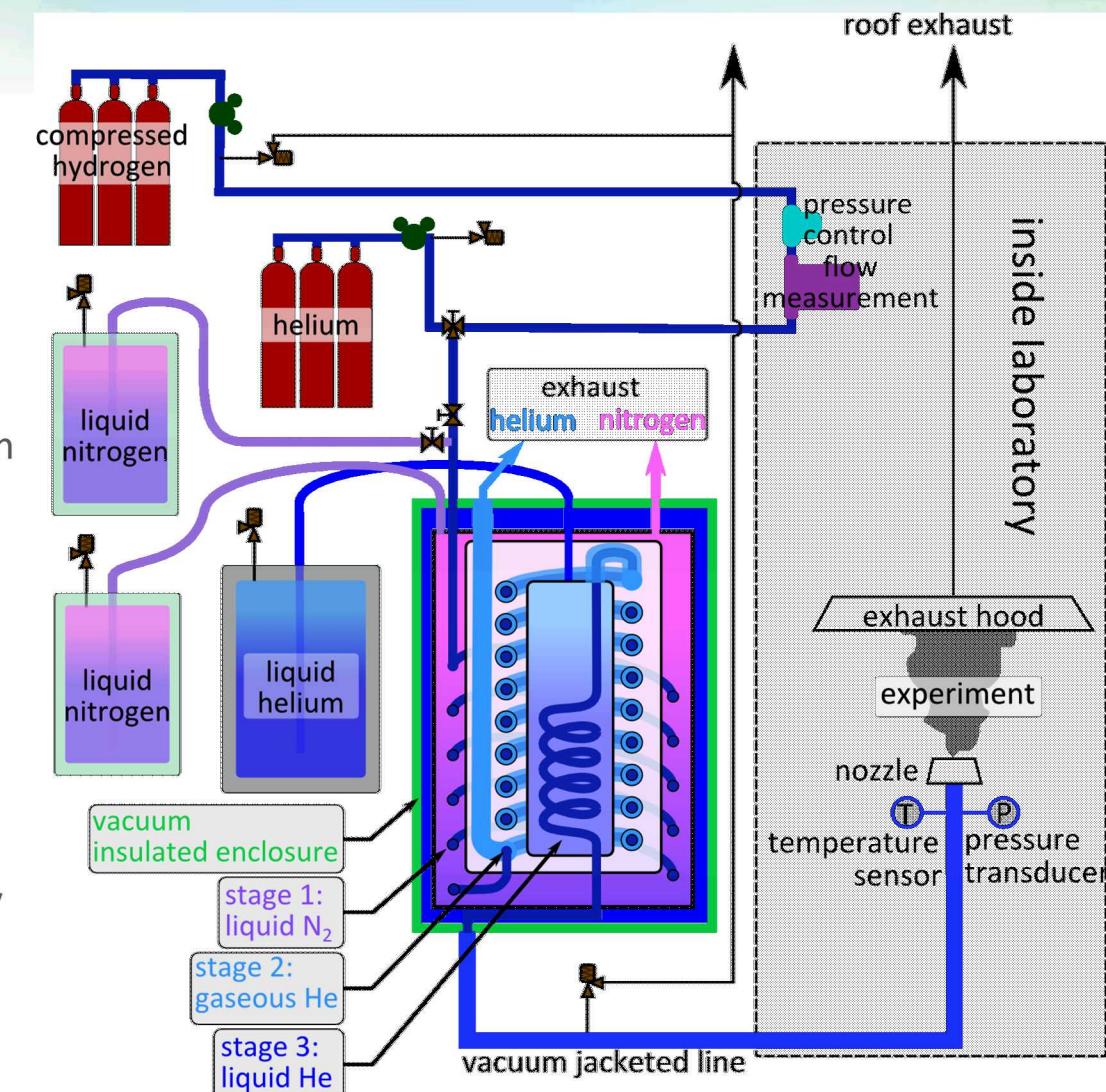


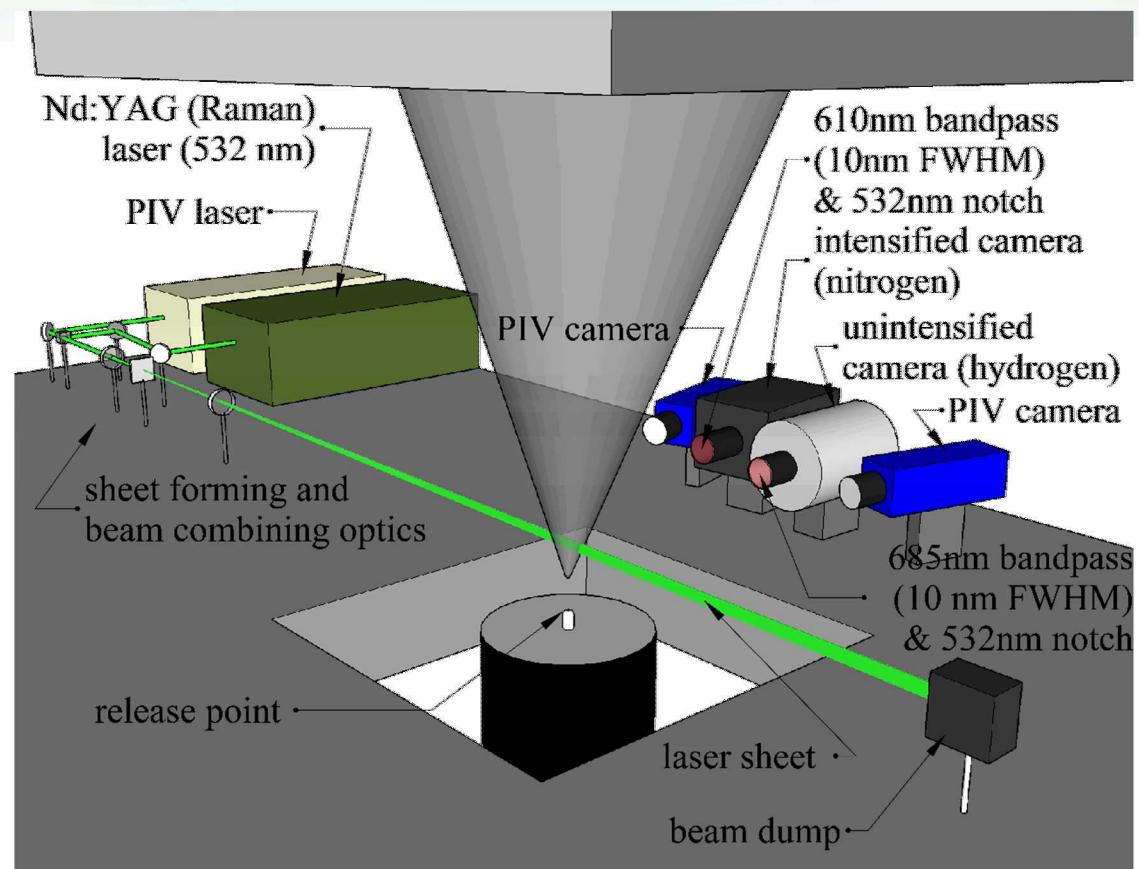
Where we left off (2016 Research Priorities Workshop)

- Experimental platform developed at Sandia to release cryogenic hydrogen through approximately 1 mm orifices at up to 10 bar
- Ignition and flame properties of cryogenic hydrogen measured
 - Cryogenic ignition distance collapses onto the same correlation for room temperature releases (scaled by effective diameter)
 - Flame length of cryogenic hydrogen scales the same as for room temperature hydrogen (square root of the Reynolds number)
 - Radiant fraction for cryogenic hydrogen flame also collapses when scaled appropriately (product of flame residence time, Plank-mean absorption coefficient, and adiabatic flame temperature to the 4th power)
- We were trying to determine a method for measuring cryogenic hydrogen concentration
 - Entrained moisture was scattering too much light for previously used Rayleigh scattering diagnostic
 - Wanted to maintain measurement in 2D of instantaneous concentration fields to measure statistical distribution of concentration (optical diagnostic)



Developed and implemented a planar Raman imaging diagnostic

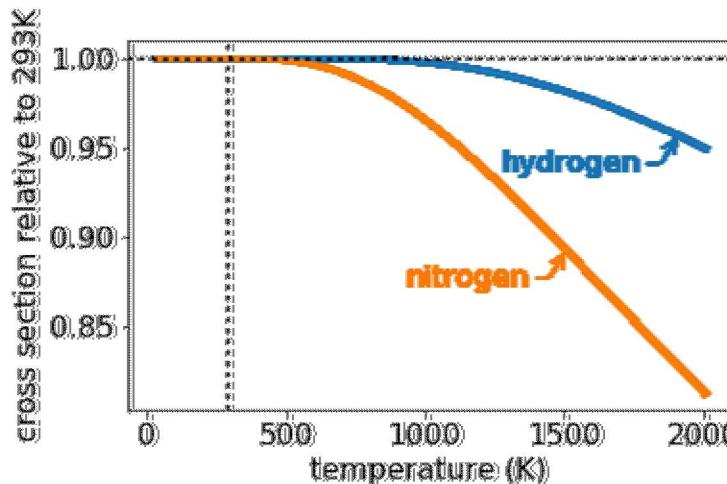
- Large Raman shift enables higher optical density filters to remove unwanted Mie scatter (off of entrained moisture)
 - 10 nm FWHM bandpass filters at wavelengths of interest
 - OD of 12 @ all wavelengths
 - OD of 18 @ 532 nm
- Simultaneously imaged scatter off of moisture to perform particle imaging velocimetry (PIV)



H₂: shift of 4161 cm⁻¹ (532nm → 683 nm, 355nm → 416 nm)
N₂: shift of 2331 cm⁻¹ (532nm → 607 nm, 355nm → 387 nm)

Quantification of Raman signals

- Signal is proportional to number density of molecules
- We use the ideal gas law to relate temperature and mole fraction to number density
 - $\frac{n_{total}\Sigma x}{V} = \frac{P_{total}\Sigma x}{RT}$
 - other equation of state could be used but may not have analytical solution
- Cross-section dependence matters for high-T (flames), but not low-T (cryogenic)



measured values $\begin{cases} I_{H_2} \\ I_{N_2} \end{cases}$ $\begin{cases} I_0 \\ I_0 \end{cases}$ calibration constants

Eq. 1: $\frac{I_{H_2}}{I_0} = k_{H_2} \frac{x_{H_2}}{T} \leftarrow \begin{array}{l} \text{unknown 1} \\ \text{unknown 2} \end{array}$

Eq. 2: $\frac{I_{N_2}}{I_0} = k_{N_2} \frac{x_{N_2}}{T} \leftarrow \begin{array}{l} \text{unknown 3} \\ \text{based on the} \\ \text{composition of air} \end{array}$

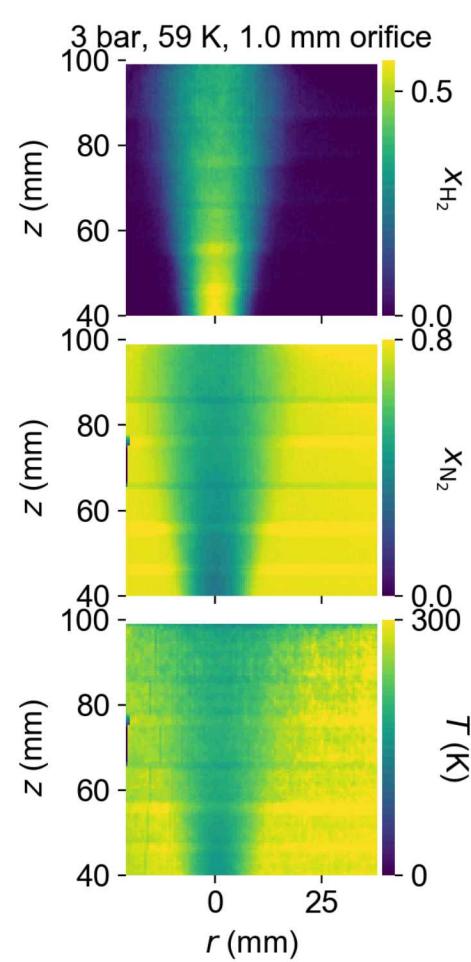
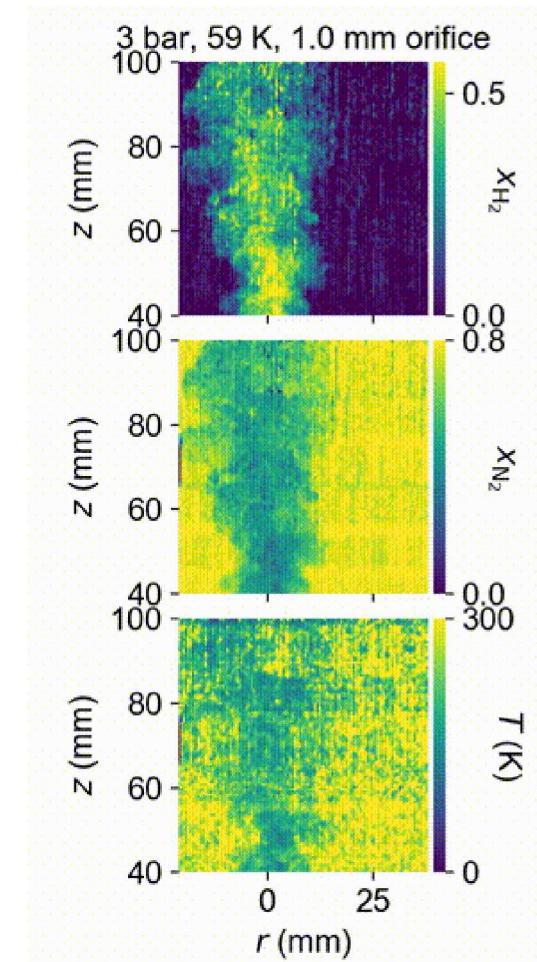
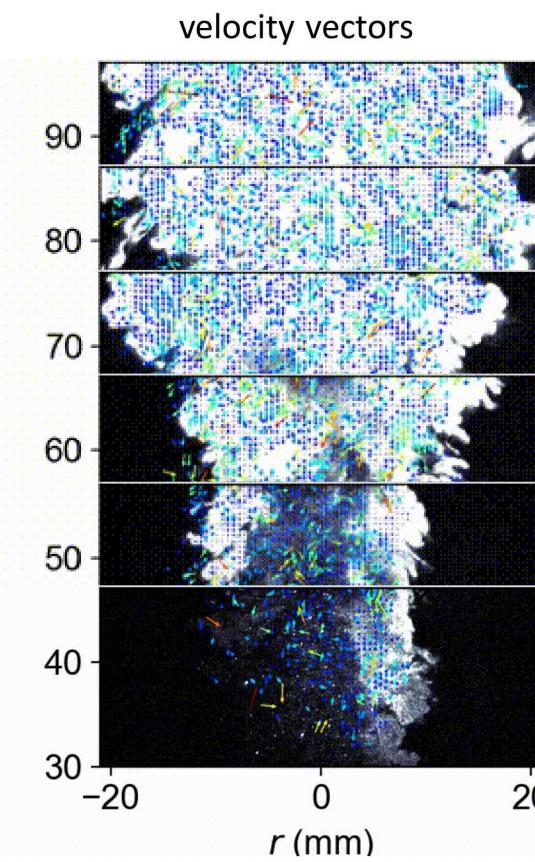
Eq. 3: $1 = x_{H_2} + 1.28x_{N_2}$

$$\left\{ \begin{array}{l} x_{H_2} = \frac{I_{H_2}}{k_{H_2} \left(\frac{I_{H_2}}{k_{H_2}} + \frac{1.28I_{N_2}}{k_{N_2}} \right)} \\ x_{N_2} = \frac{I_{N_2}/I_0}{k_{N_2} \left(\frac{I_{H_2}}{k_{H_2}} + \frac{1.28I_{N_2}}{k_{N_2}} \right)} \\ T = \frac{1}{\frac{I_{H_2}}{k_{H_2}} + \frac{1.28I_{N_2}}{k_{N_2}}} \end{array} \right.$$

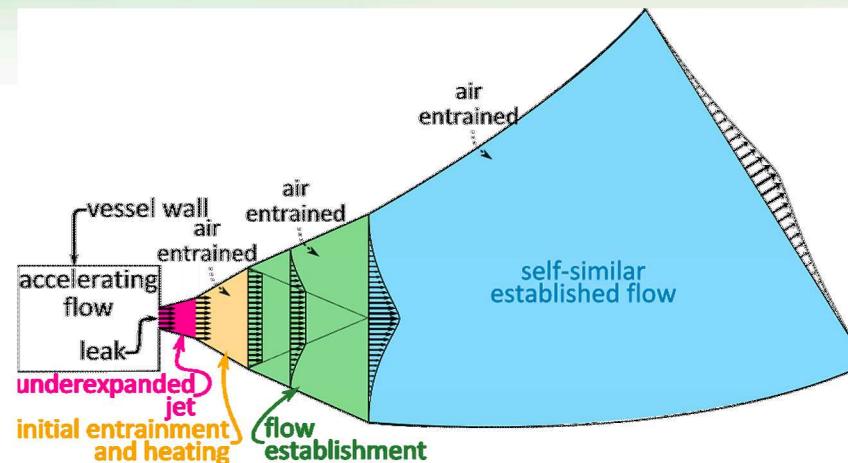
Raman has been used in a lab-scale campaign to measure releases from ≈ 1 mm orifices

T_{noz} [K]	P_{noz} [bar _{abs}]	d [mm]	T_{throat} [K]	n_{hts}
58	2	1	43.5	4
56	3	1	41.9	4
53	4	1	39.6	4
50	5	1	37.4	5
61	2	1.25	45.7	6
51	2.5	1.25	38.2	2
51	3	1.25	38.2	6
55	3.5	1.25	41.2	3
54	4	1.25	40.4	2
43	4	1	32.1	2
59	3	1	44.2	6
56	3.5	1	41.9	1
80	3	1	60.3	5

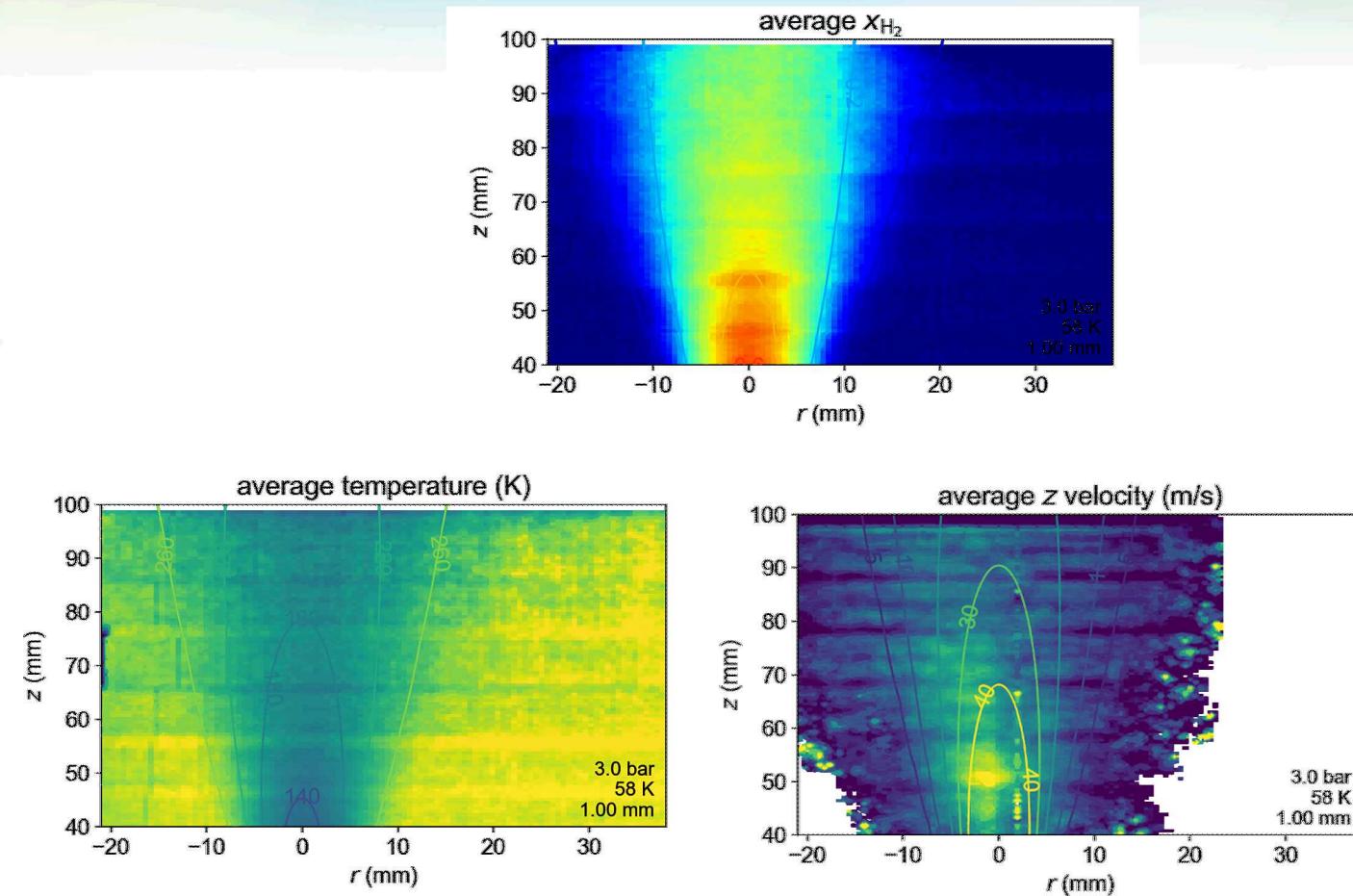
With PIV ↓



ColdPLUME model shows good agreement with the data

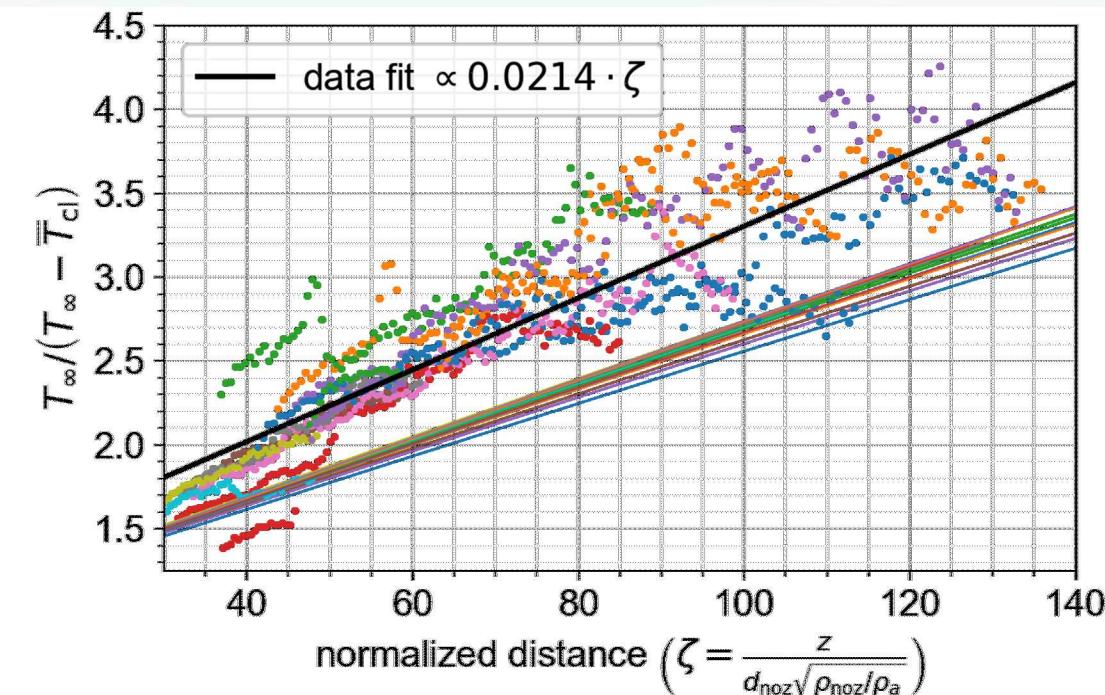
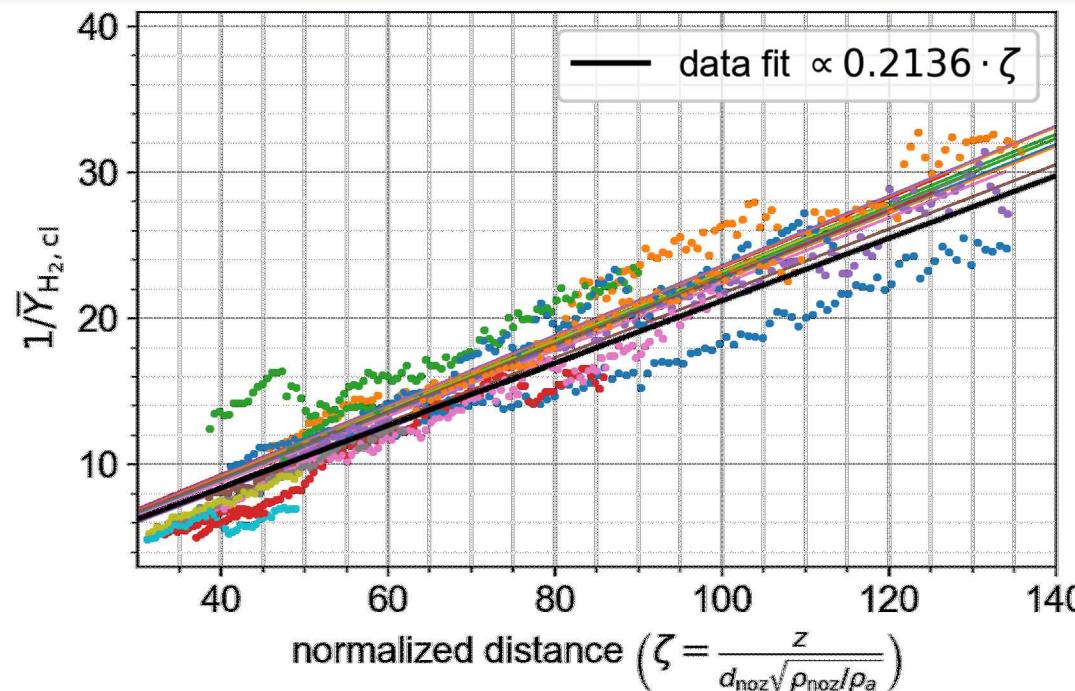


- Experimental results shown by shading and thick, dashed lines
- ColdPLUME model results are thin, solid lines



➤ Model accurately simulates mole fraction, temperature, and velocity -- can be used as a predictive tool

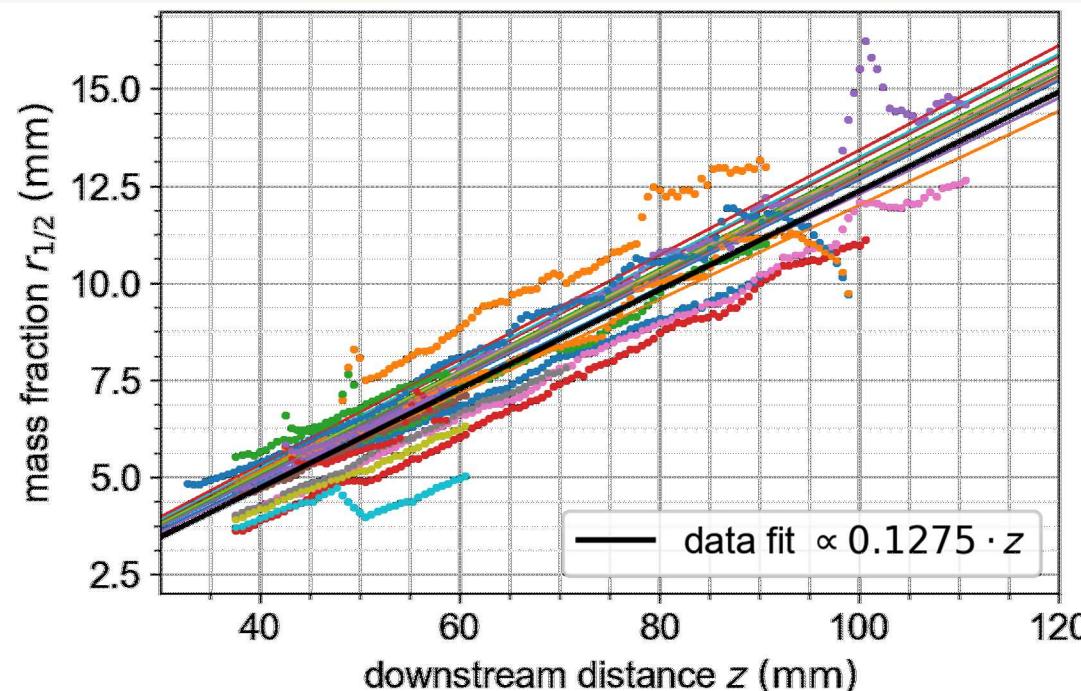
The centerline mass fraction and temperature decay linearly when plotted against the normalized distance



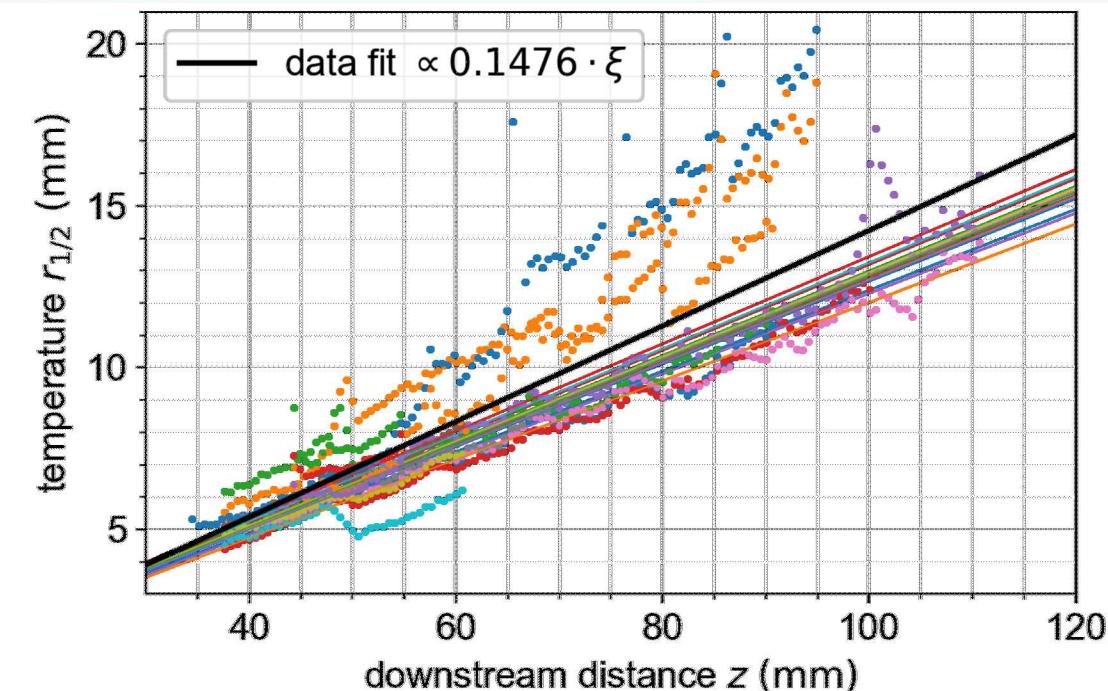
- Data shown by points, model by thin lines
- Literature inverse mass-fraction decay rate for warm releases: 0.21–0.271 – same as this data
- Model does not predict perfect collapse of the data

• 2 bar, 58 K, 1.00 mm	• 4 bar, 54 K, 1.25 mm
• 3 bar, 56 K, 1.00 mm	• 4 bar, 45 K, 1.25 mm
• 4 bar, 53 K, 1.00 mm	• 3.0 bar, 58 K, 1.00 mm
• 5 bar, 50 K, 1.00 mm	• 3.0 bar, 82 K, 1.00 mm
• 2 bar, 61 K, 1.25 mm	• 3.5 bar, 55 K, 1.00 mm
• 2.5 bar, 51 K, 1.25 mm	• 4.0 bar, 40 K, 1.00 mm
• 3 bar, 51 K, 1.25 mm	• 4.0 bar, 63 K, 1.00 mm
• 3.5 bar, 55 K, 1.25 mm	

With two-dimensional images, we can also ensure that the widths of the jets are behaving as expected



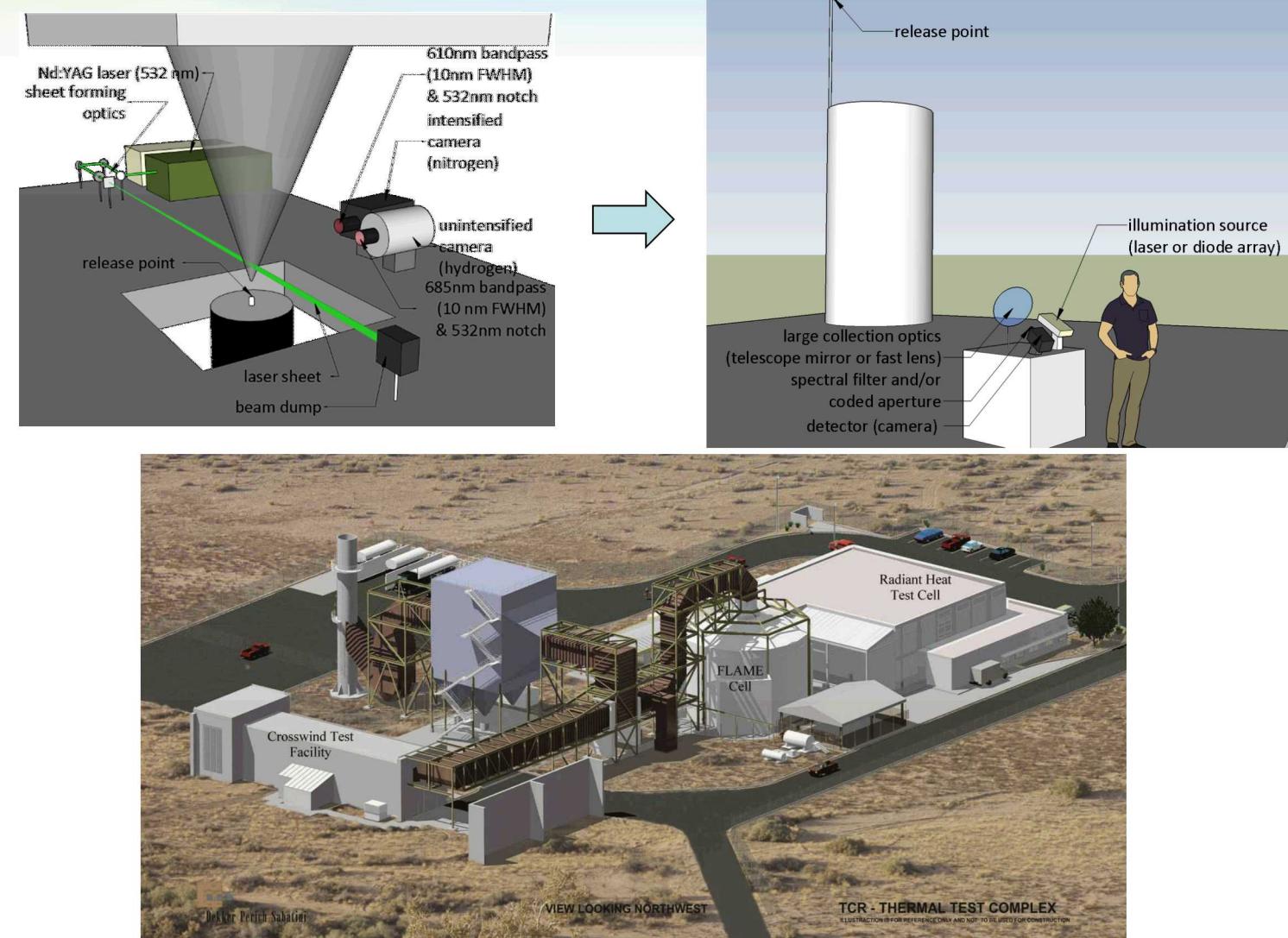
- Data shown by points, model by thin lines
- Literature mass-fraction half-width spreading rate (warm gases): 0.1–0.11 – similar to this data
- Both axes must have same units (some literature reports have incorrect axes definitions)



• 2 bar, 58 K, 1.00 mm	• 4 bar, 54 K, 1.25 mm
• 3 bar, 56 K, 1.00 mm	• 4 bar, 45 K, 1.25 mm
• 4 bar, 53 K, 1.00 mm	• 3.0 bar, 58 K, 1.00 mm
• 5 bar, 50 K, 1.00 mm	• 3.0 bar, 82 K, 1.00 mm
• 2 bar, 61 K, 1.25 mm	• 3.5 bar, 55 K, 1.00 mm
• 2.5 bar, 51 K, 1.25 mm	• 4.0 bar, 40 K, 1.00 mm
• 3 bar, 51 K, 1.25 mm	• 4.0 bar, 63 K, 1.00 mm
• 3.5 bar, 55 K, 1.25 mm	

Lab scale data is great for model validation but we are working to scale-up our Raman imaging techniques

- **Remaining priorities for LH₂:**
 - Relate concentration data to ignition distance
 - Study non-circular orifices
 - Characterization and modeling of
 - Interactions with ambient (i.e. wind)
 - Pooling
 - Evaporation from LH₂ pools
- Currently developing an imaging diagnostic for outdoor and large-scale experiments
 - Quantitative concentration measurements
 - 2- or 3-dimensions
 - Video frame rates
 - Portable
- Will apply diagnostic to normally occurring outdoor releases (e.g., venting after LH₂ fill)
- Dedicated validation experiments (pooling, cross-wind) at well-controlled facilities next fiscal year (FY19-20)



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