

# GOLDSCHMIDT

BOSTON 2018 | AUGUST 12-17

## Molecular Modeling of Interaction between Kerogen and Pore Fluid

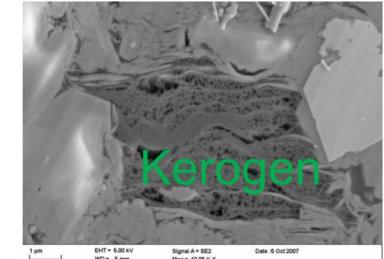
**Tuan A. Ho, Yifeng Wang, Louise J. Criscenti**  
Sandia National Laboratories



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# Introduction

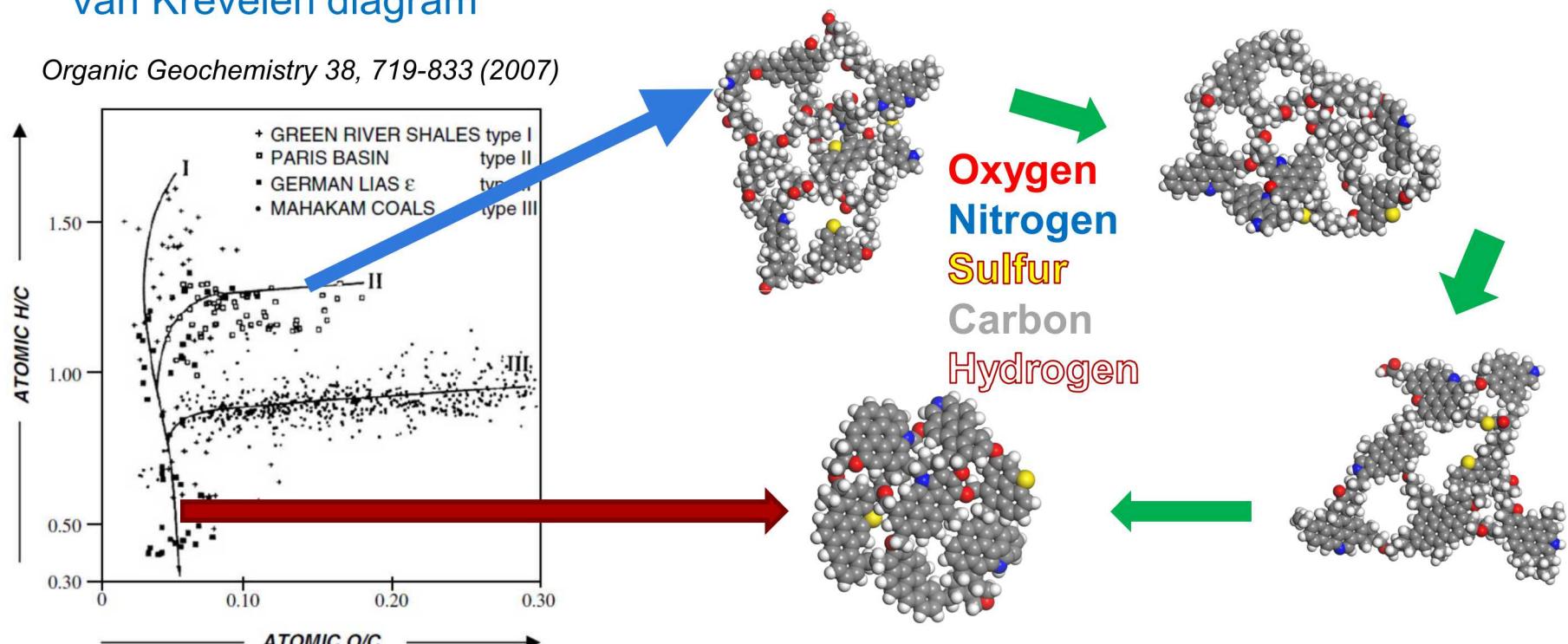
## Kerogen



- Insoluble organic matter found in sedimentary rocks (geochemistry)
- Cracks into petroleum products (kerogen maturation, petroleum generation)

### Van Krevelen diagram

Organic Geochemistry 38, 719-833 (2007)



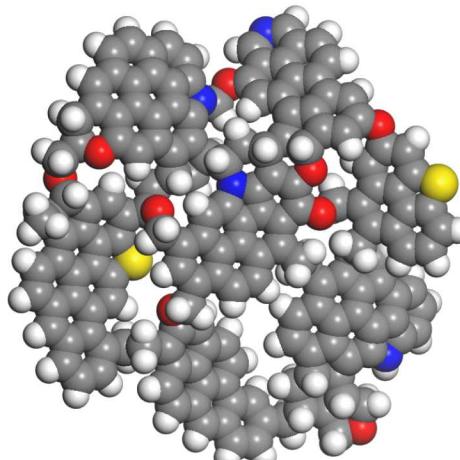
- Hosts pore space responsible for petroleum storage and transport

# Overmature Kerogen

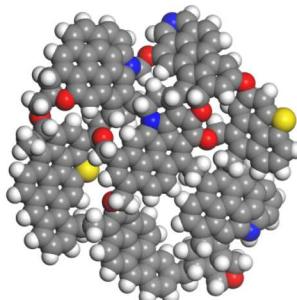
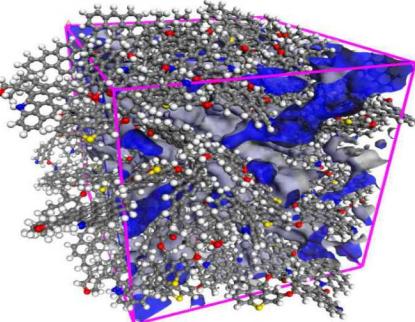
- Use the available elemental analysis and functional data from XPS and  $^{13}\text{C}$  NMR
  - (*Energy & Fuels, 2007, 21, 1548-1561*)

Oxygen  
 Nitrogen  
 Sulfur  
 Carbon  
 Hydrogen

	II-D	
	analytical data	model unit
H/C	0.56	0.58
O/C	0.047	0.051
N/C	0.021	0.023
S/C	0.01	0.011
% of aromatic carbon from XPS(a) or NMR(b)	72(a), 80(b)	79
avg. number of C atoms per aromatic cluster	20	19.9
fraction of aromatic carbons with attachments ( $\text{sp}^3$ C, N, S, O)	0.24	0.28



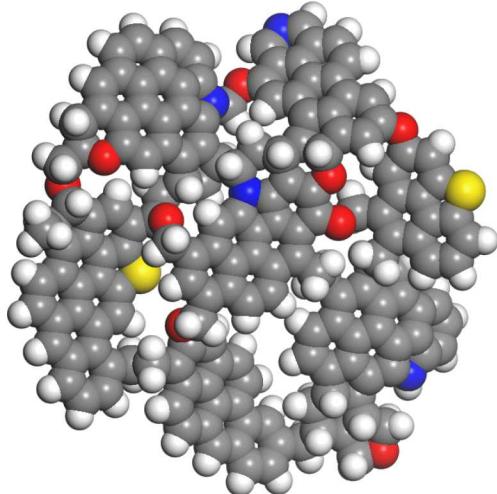
# Molecular Modeling of Interaction between Kerogen and Pore Fluid



## Outline:

1. **Nanostructure of kerogen**
2.  **$\text{CH}_4/\text{CO}_2$  adsorption onto kerogen**
3. **Kerogen swelling**

# Formation of condensed kerogen



24 Kerogens in  $10 \times 10 \times 10 \text{ nm}^3$  box, 1000K

NVT

9 snapshots

NPT,  
100at,  
900K to  
300K

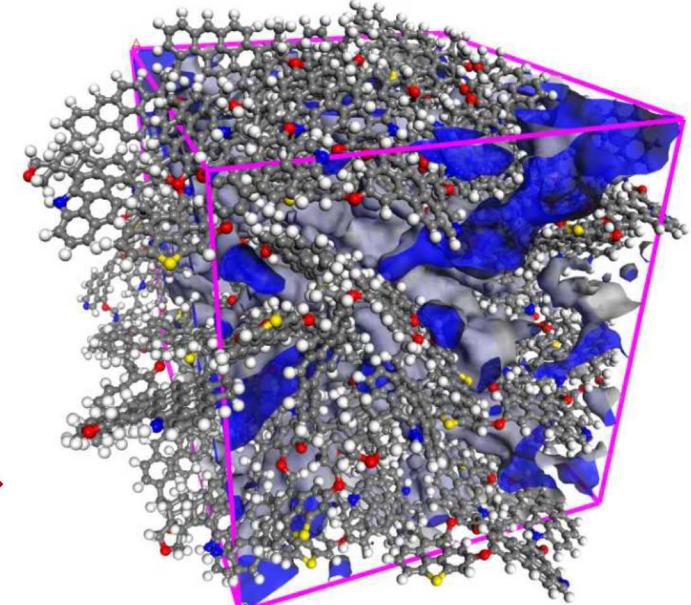
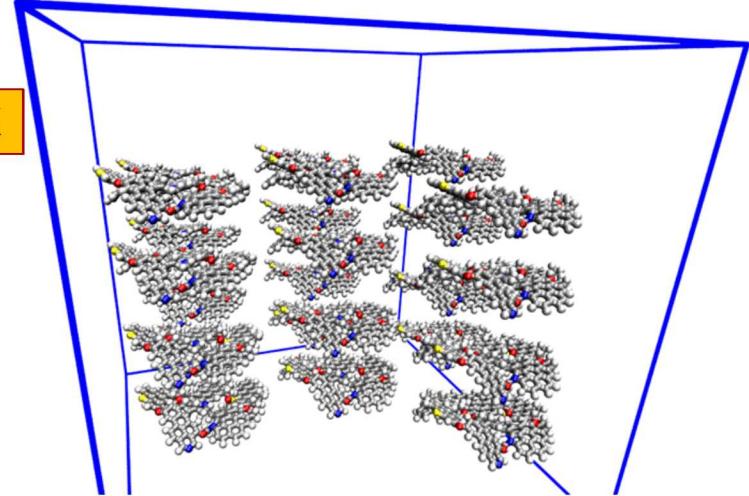


Energy Fuels 2015, 29, 91-105

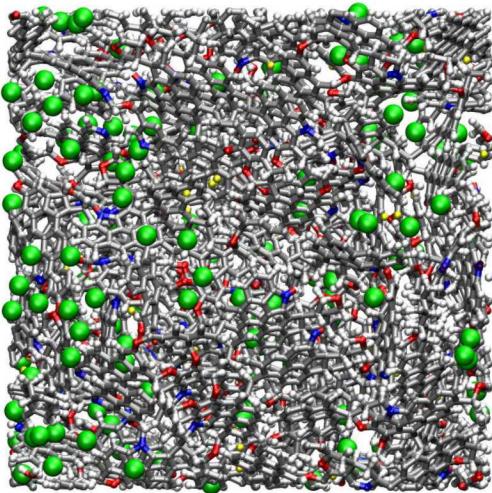
300K and 100atm

NPT,  
1atm,  
300K

9 samples at  
300K and 1atm



# Characterization

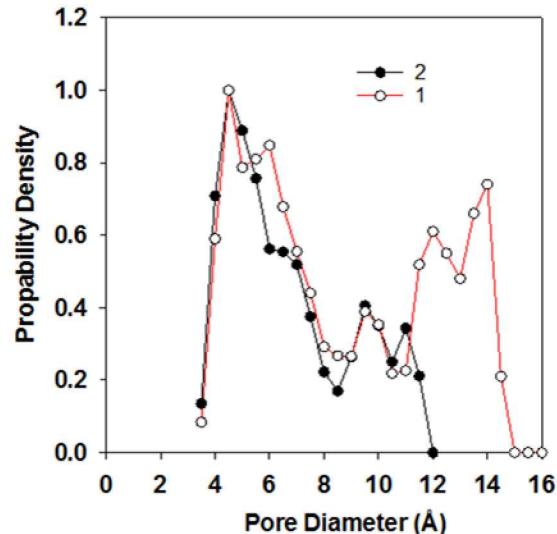


## Density

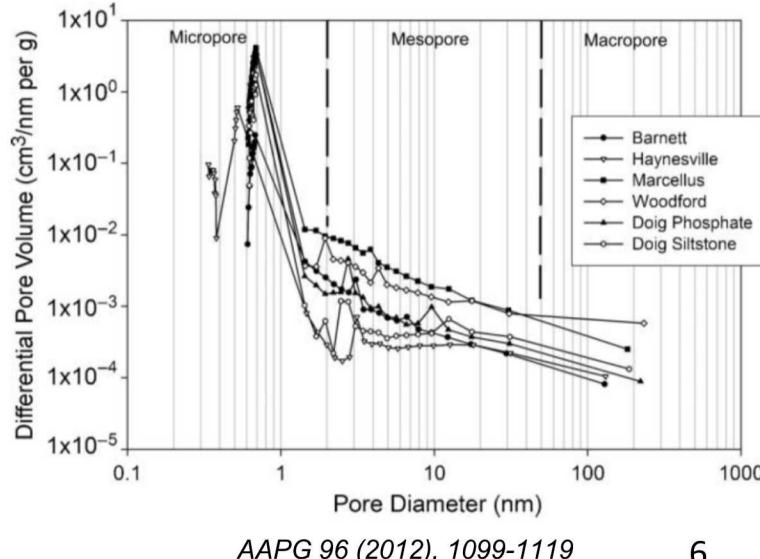
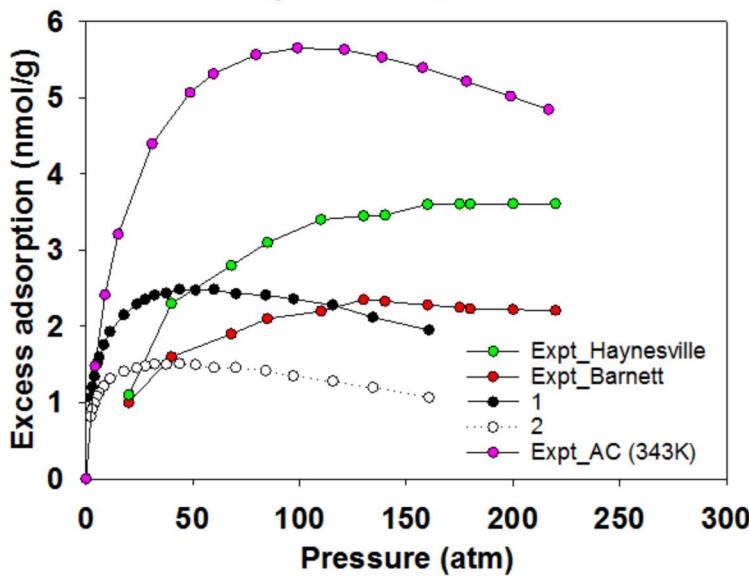
Average :  $1.22 \pm 0.04 \text{ g/cm}^3$   
 Experiment:  $1.28 \pm 0.3 \text{ g/cm}^3$

Stankiewicz A, et al. (2015) Kerogen density revisited – lessons from the Duvernay Shale. In: Paper URTeC 2157904 at the Unconventional Resources Technology Conference, San Antonio, Texas, July 2015

## Pore size distribution

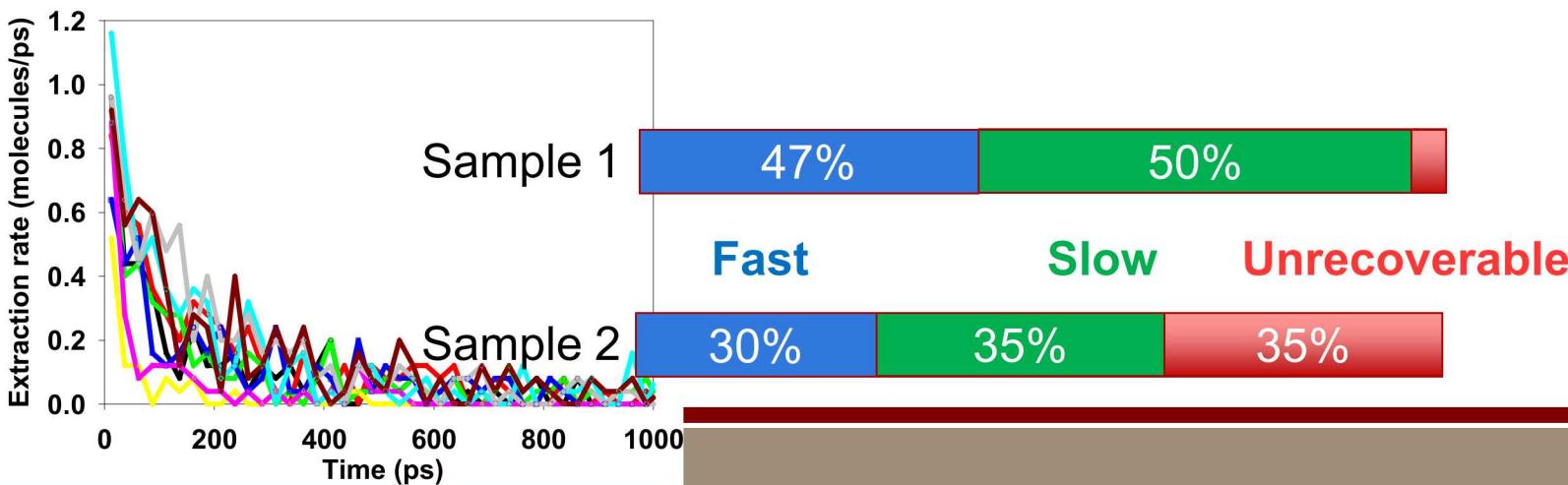
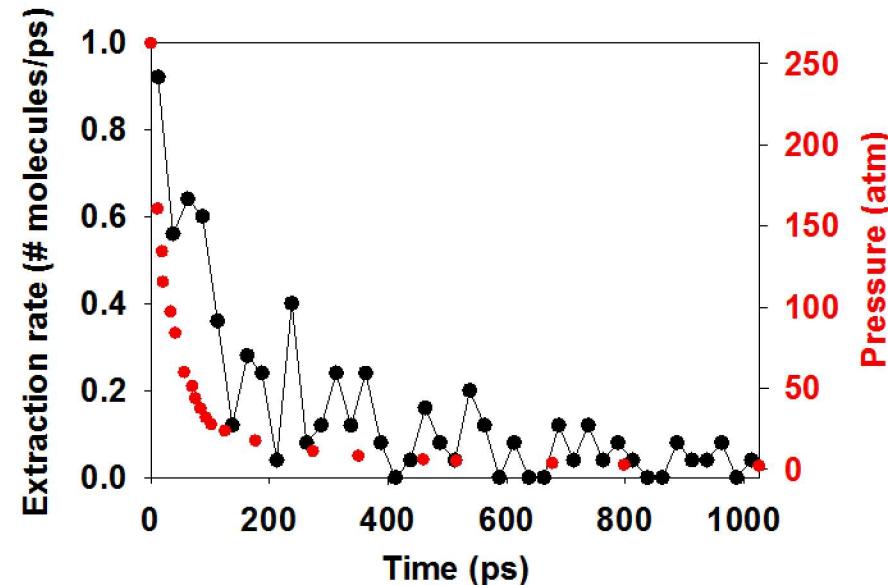
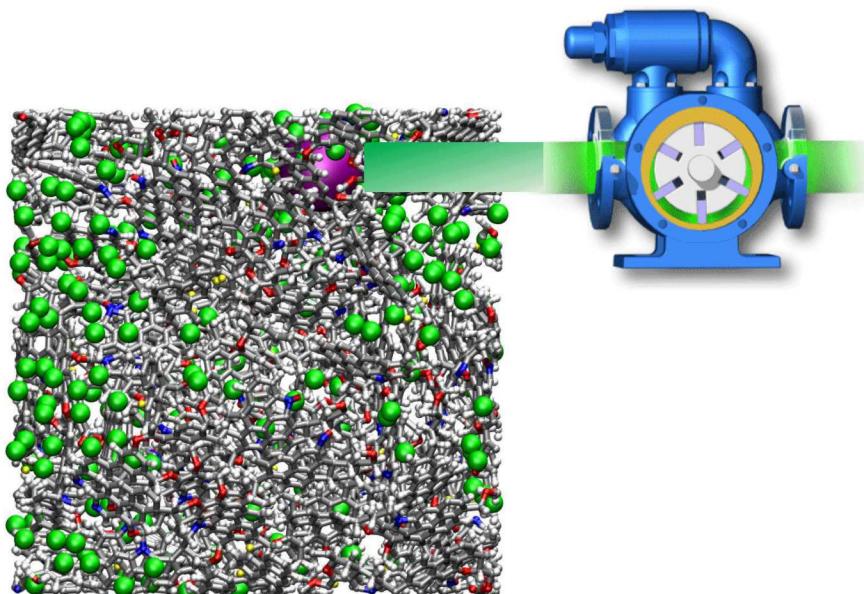


## CH<sub>4</sub> Adsorption

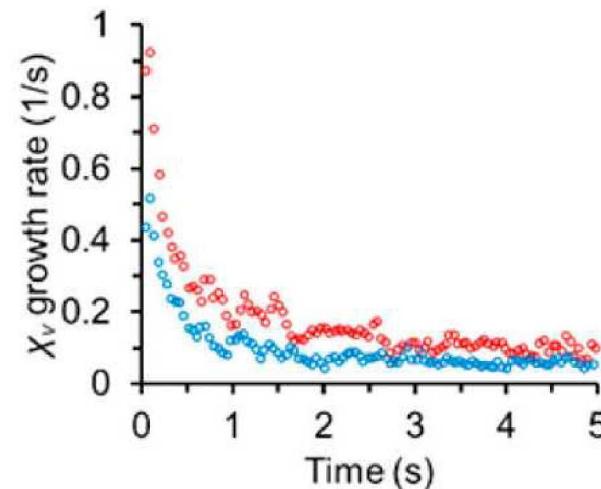
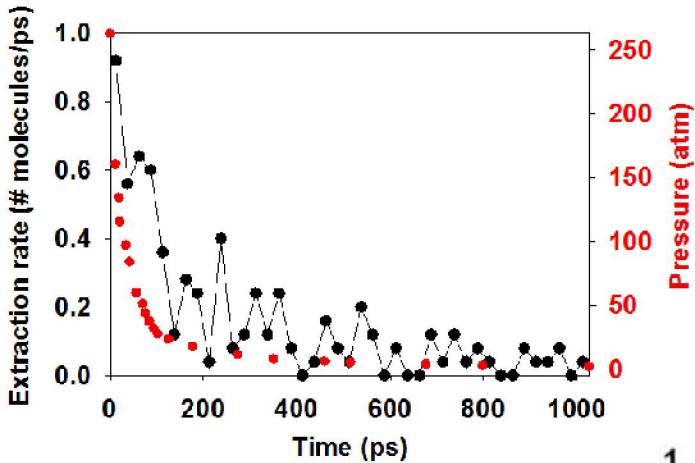


# Methane extraction from kerogen

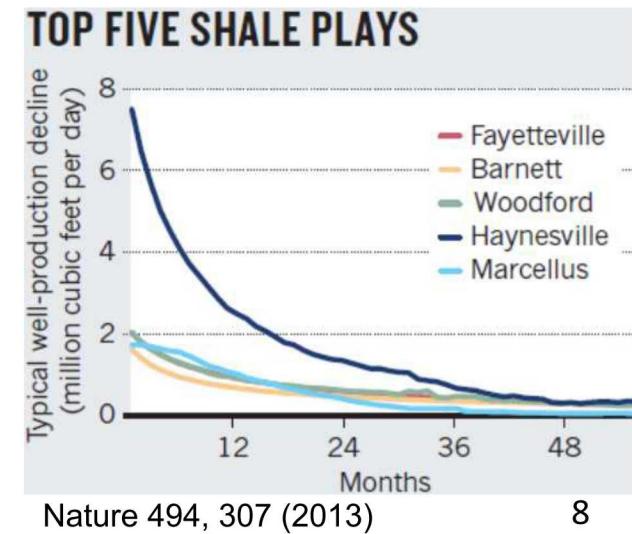
(Scientific Reports 6, 28053, 2017)



# Production decline curve



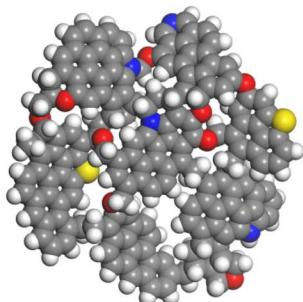
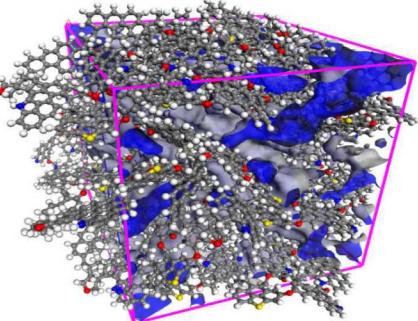
ACS Applied Nano Materials1,  
1332-1338 (2018)



Nature 494, 307 (2013)

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# Molecular Modeling of Interaction between Kerogen and Pore Fluid



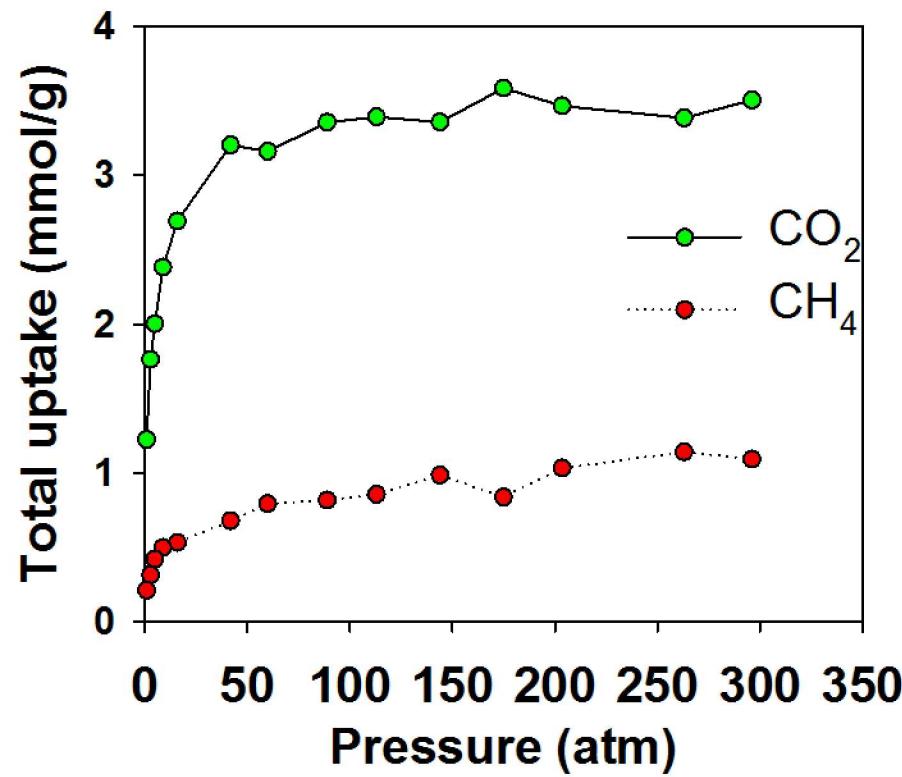
## Outline:

1. Nanostructure of kerogen
2.  $\text{CH}_4/\text{CO}_2$  adsorption onto kerogen
3. Kerogen swelling

# Differential retention and release of $\text{CO}_2$ and $\text{CH}_4$ in kerogen nanopores (Fuel 220, 1-7, 2018)

Implications for **gas enhanced recovery** and **carbon sequestration**

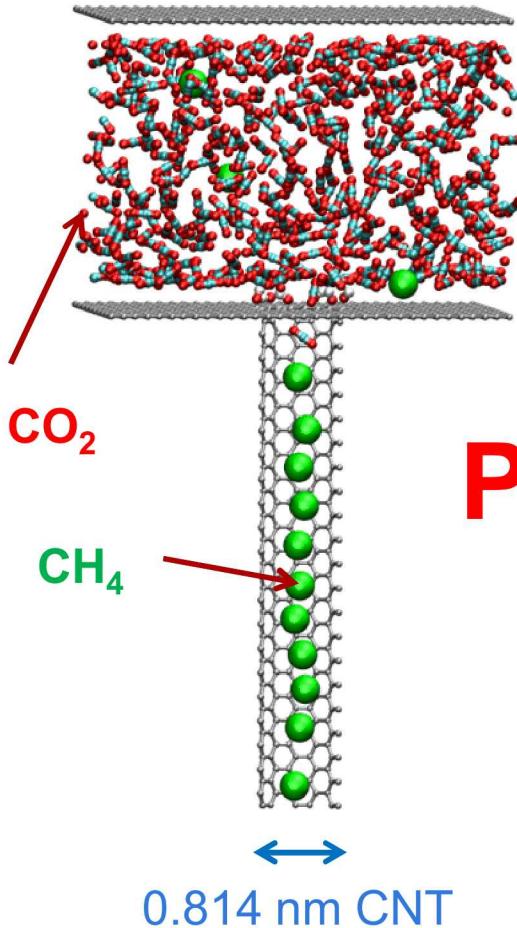
1:1 binary gas adsorption



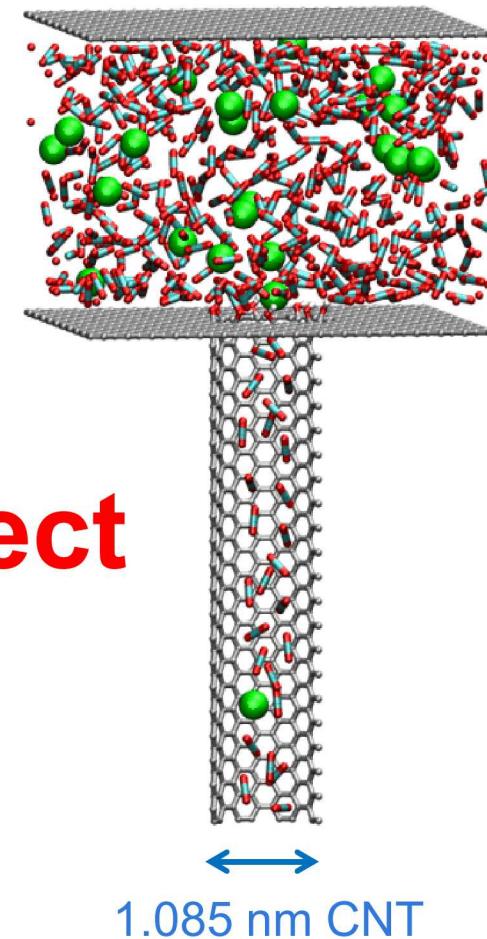
Kerogen preferentially retains  $\text{CO}_2$  over  $\text{CH}_4$

# Pore specific effects on enhanced gas recovery

(Fuel 220, 1-7, 2018)



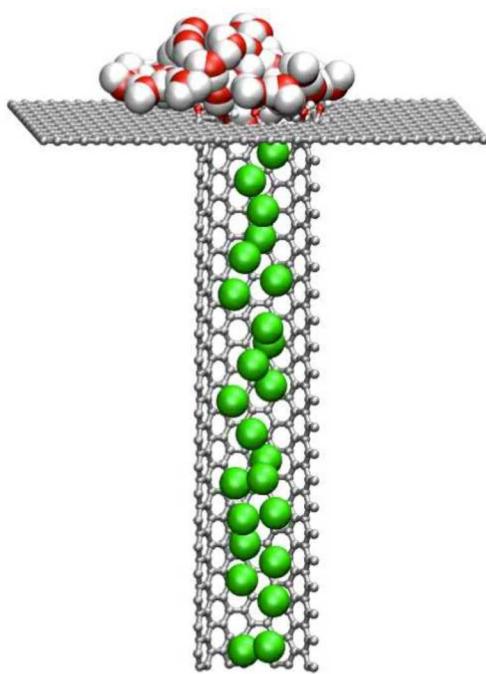
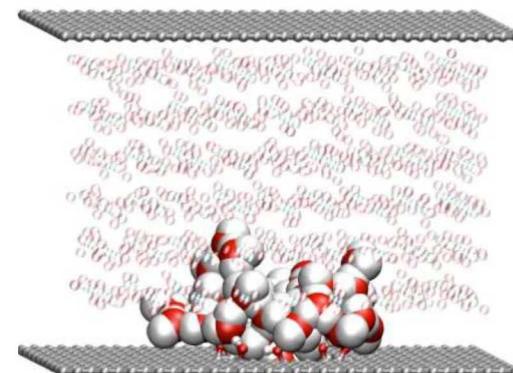
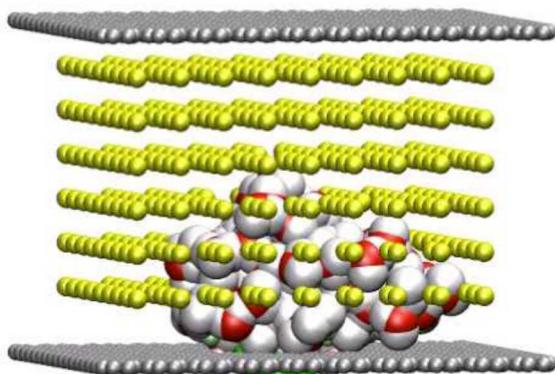
Pore is too small for the invasion of CO<sub>2</sub>



Pore is big enough for the invasion of CO<sub>2</sub>

# Pore specific effects on enhanced gas recovery

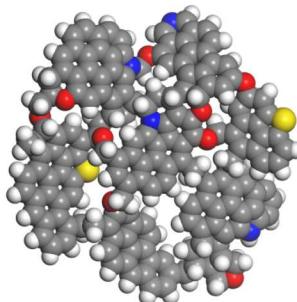
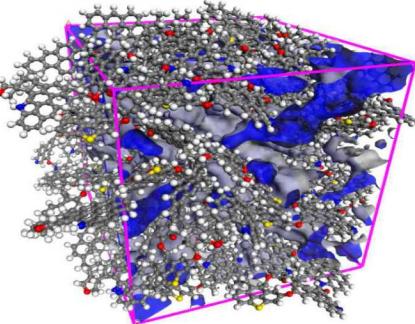
(Fuel 220, 1-7, 2018)



Assume that water drop  
blocks the pore entrance.

CO<sub>2</sub> invades through water and replaces  
CH<sub>4</sub> in the nanopore.

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# Kerogen swelling

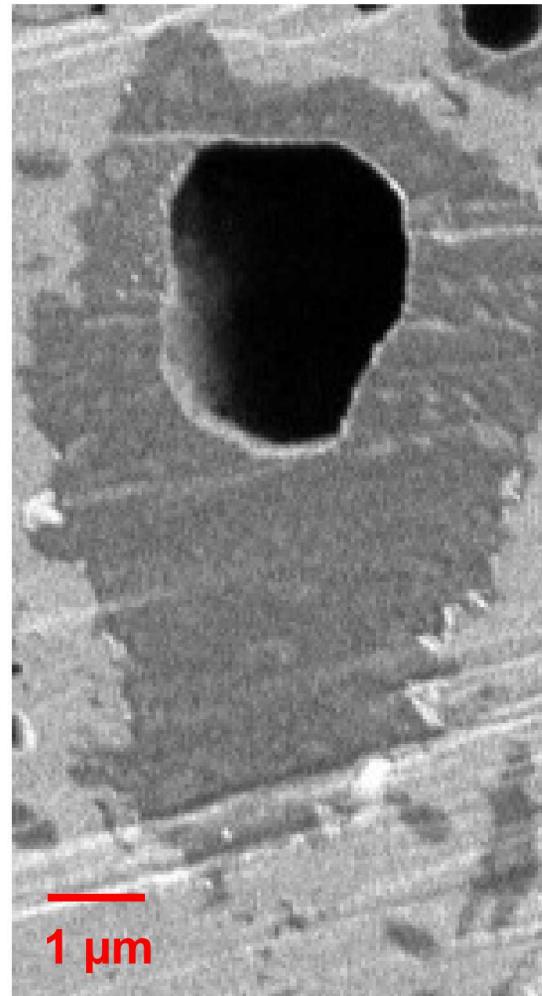
Rubber swelling in oil



From: **Drew Pomerantz**,  
Schlumberger



**Will kerogen swell  
upon gas adsorption?**



Intact shale with  
swollen kerogen

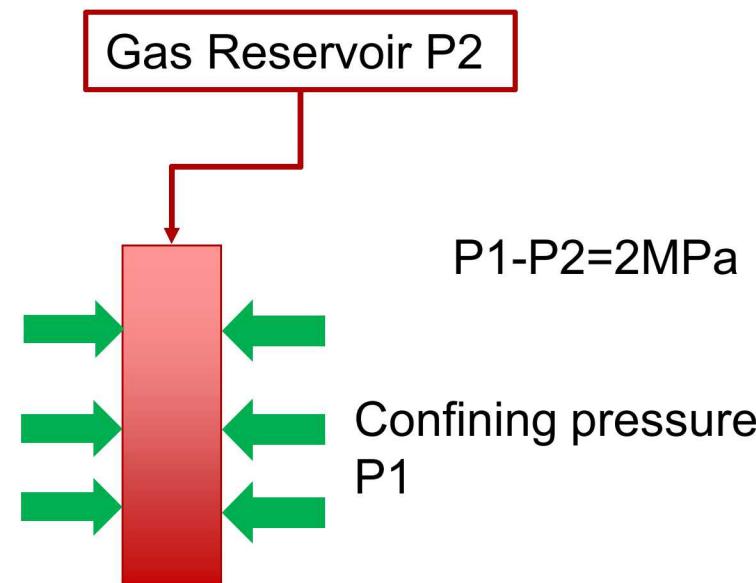


Bitumen-extracted shale  
with collapsed kerogen

# Chemo-mechanical coupling in kerogen gas adsorption (PCCP 20, 11390, 2018)

## Experimental setup

(J. Unconv. Oil Gas Resour., 2014)

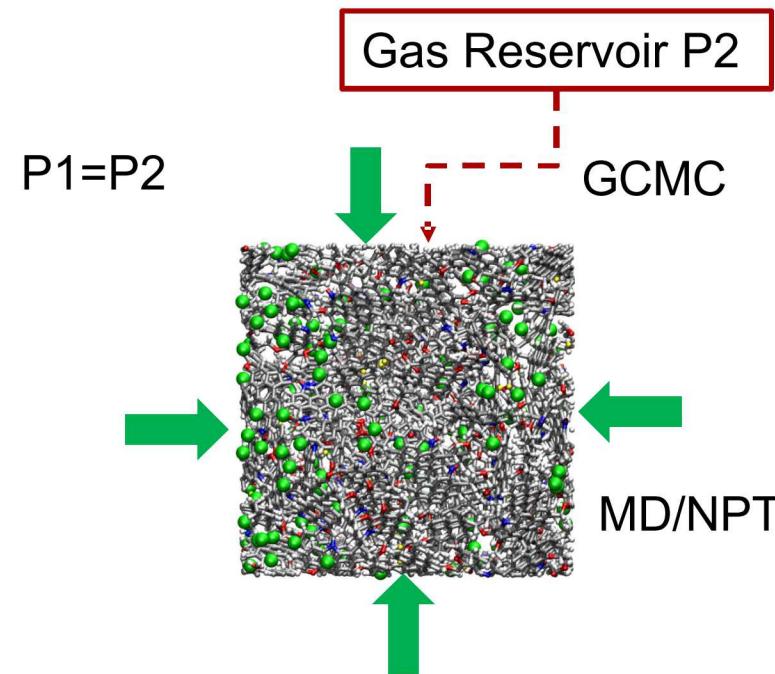


$$\text{Volumetric strain} = \frac{V - V_o}{V_o}$$

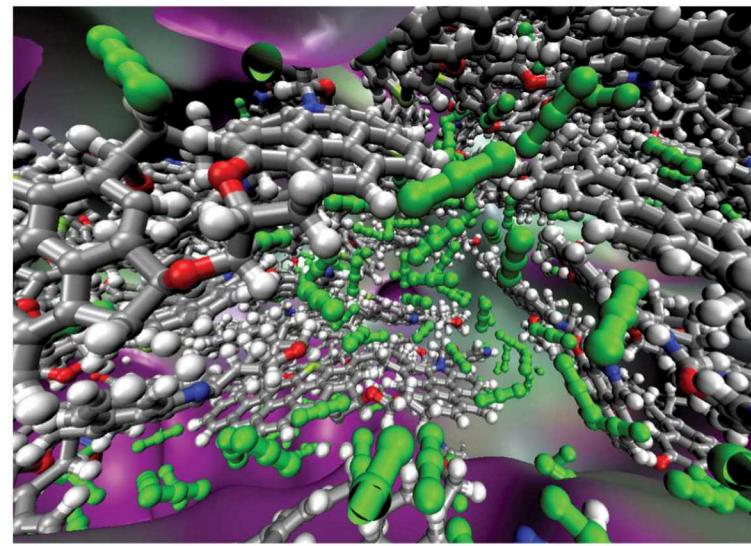
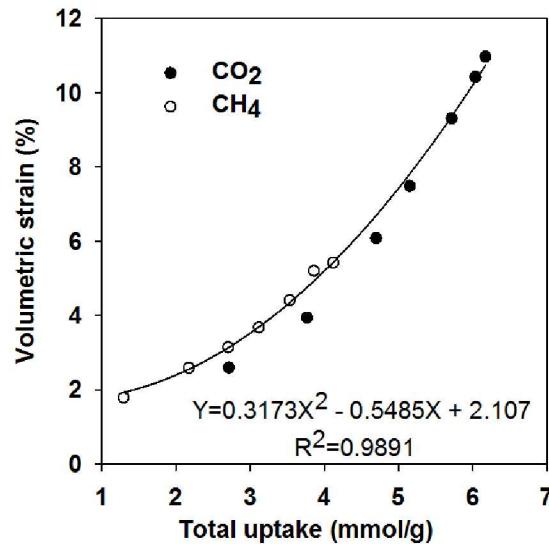
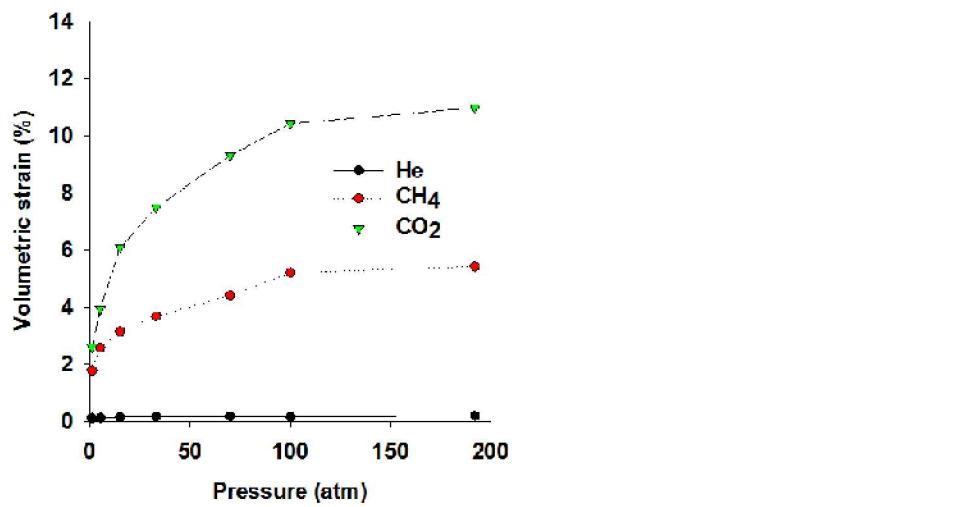
V: kerogen volume after gas adsorption

V<sub>o</sub>: kerogen volume before gas adsorption

## Simulation: Hybrid MD/MC



# Chemo-mechanical coupling in kerogen gas adsorption (PCCP 20, 11390, 2018)



Highlighting shale gas research from the Geoscience Group, Sandia National Laboratories, NM, USA. This work was conducted by Dr Tuan Ho, thanks to funding granted to Dr Yifeng Wang by the DOE National Energy Technology Laboratory.

Chemo-mechanical coupling in kerogen gas adsorption/desorption

We use an integrated experimental and modeling approach to fundamentally understand the interaction of gas and fluid with kerogen and clay under reservoir conditions. Specifically, nanostructural properties of subsurface porous media, gas adsorption and release from the kerogen network, deformation of shale associated with adsorption and lithostatic stress, and wettability of inorganic and organic matter.

As featured in:

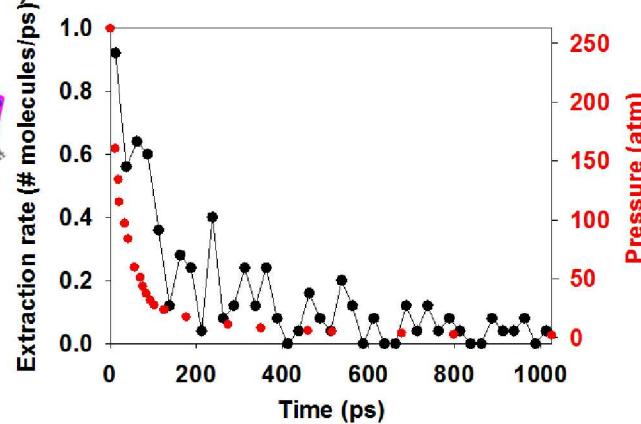
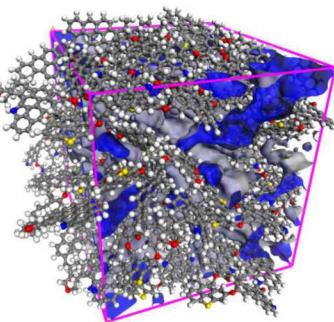


See Tuan Anh Ho et al.,  
*Phys. Chem. Chem. Phys.*,  
2018, 20, 12390.

# Summary

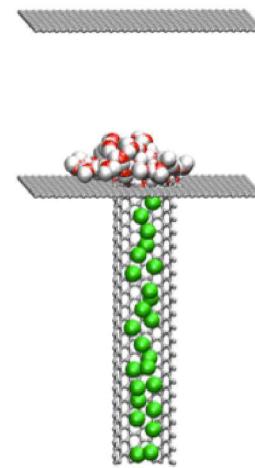
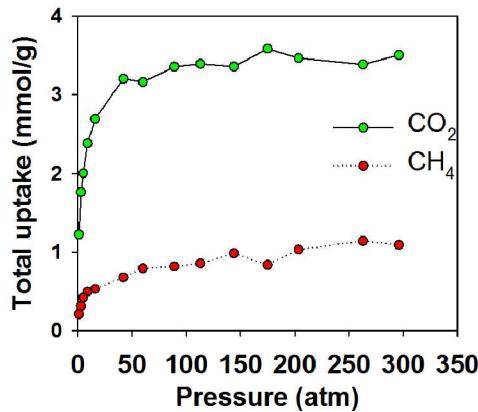
## 1. Nanostructure of kerogen

(Sci. Rep. 2016)



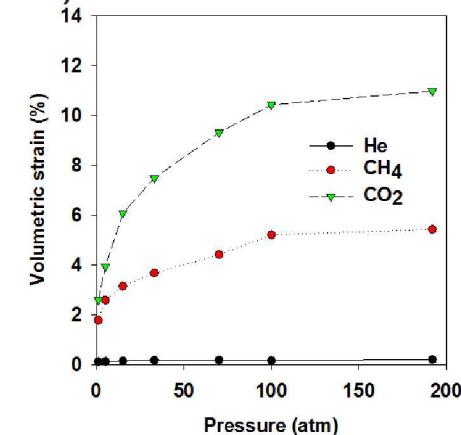
## 2. CH<sub>4</sub>/CO<sub>2</sub> adsorption onto kerogen

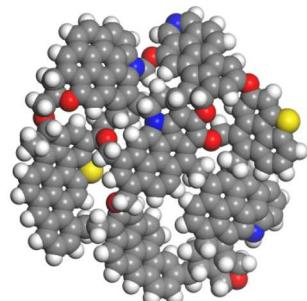
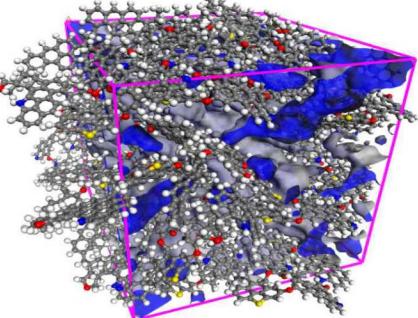
(Fuel 2018)



## 3. Kerogen swelling

(PCCP 2018)





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Acknowledgement: LDRD and NETL