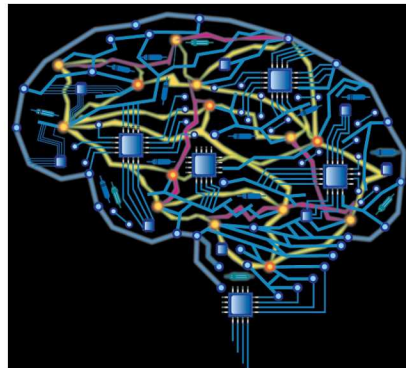
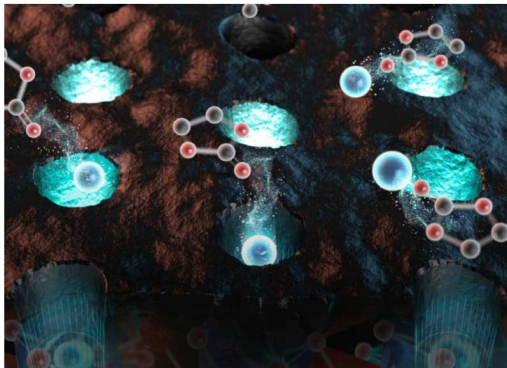




Electrochemical ion insertion for energy storage and computing



PRESENTED BY

Yiyang Li

The views expressed in the presentation do not necessarily represent the views of the U.S. Department of Energy or the United States Government.



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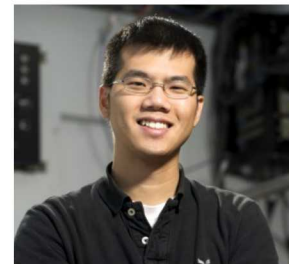
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W. Chueh



J. Lim

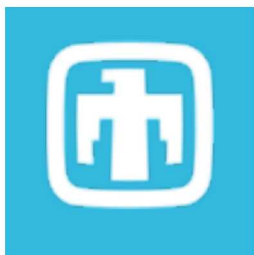
MIT

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Dan Cogswell



M. Bazant



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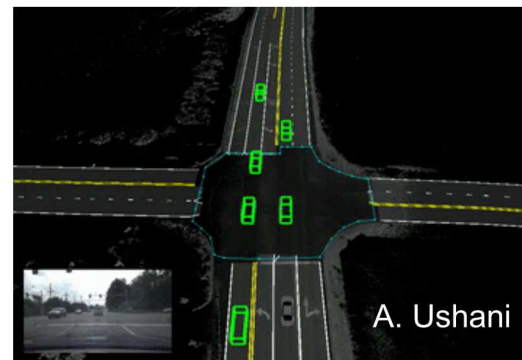
Ion insertion for energy storage and computing

Autonomous electric vehicles



Source: GM

Neural Network

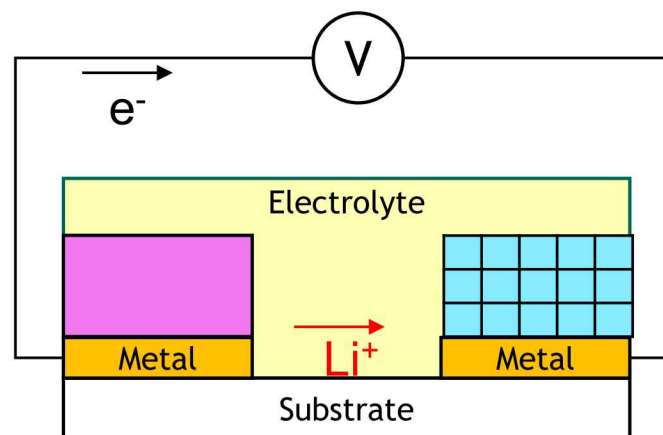


Batteries



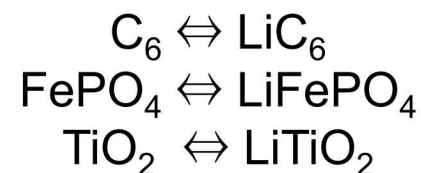
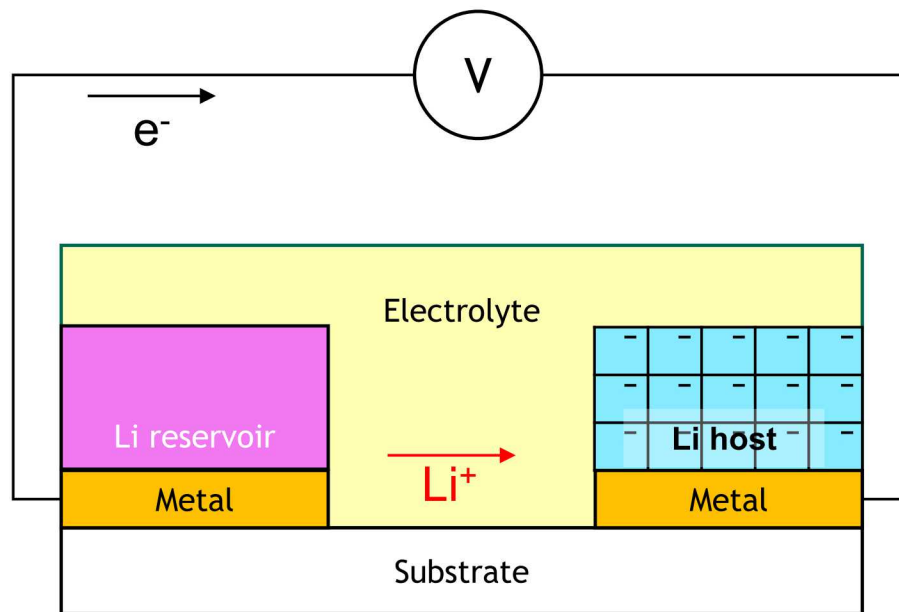
Panasonic

Ion insertion





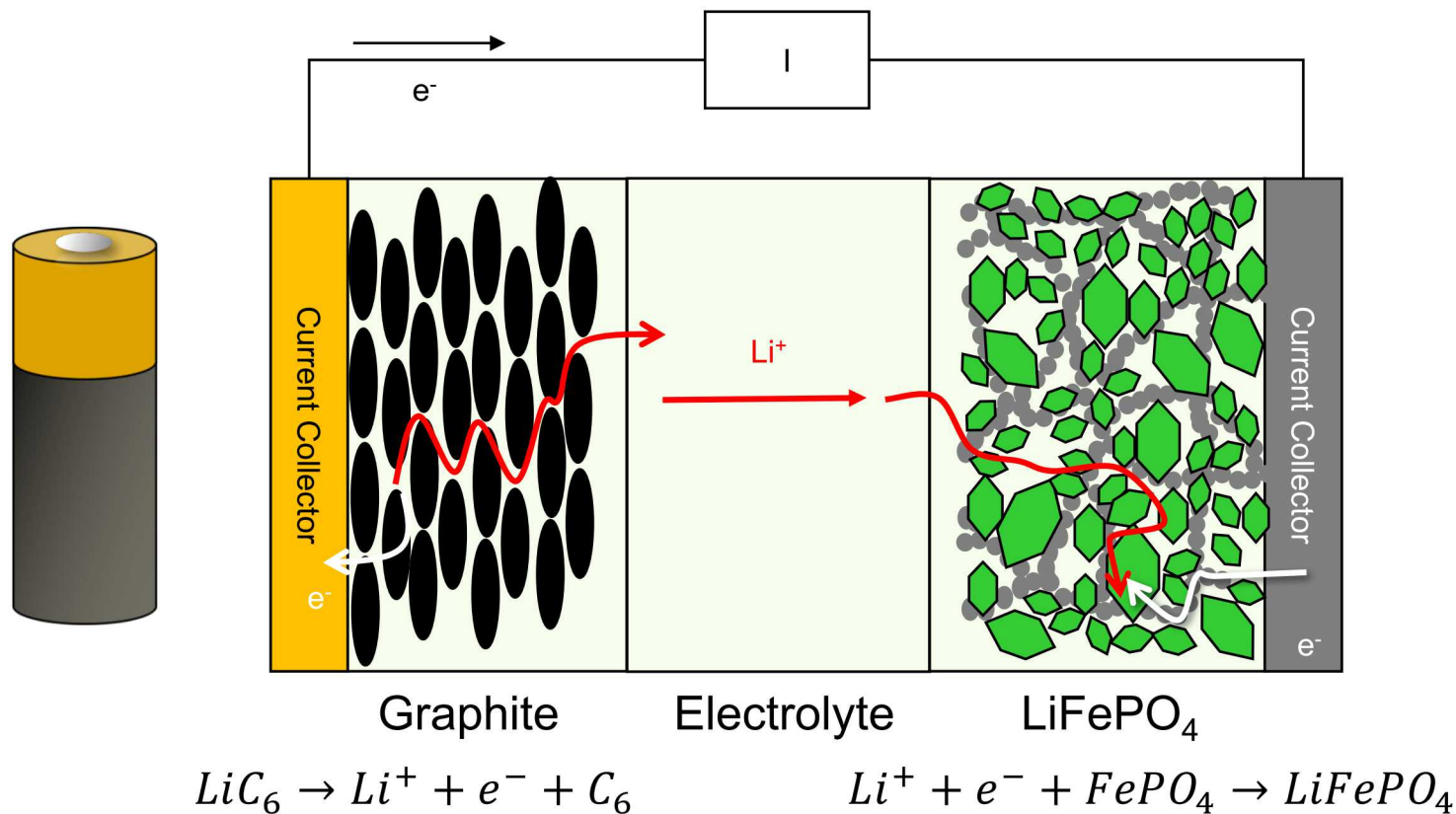
Electrochemical ion insertion controls materials chemistry



- 1) Using electronic controls, we alter the material composition of the Li host, conducting solid-state chemistry using current and voltage
- 2) The ions balance the electron charge, so a large amount of electrons can be moved without an electrostatic voltage



Ion insertion in a Li-ion battery



$$\Delta G = -320 \frac{\text{kJ}}{\text{mol}}$$

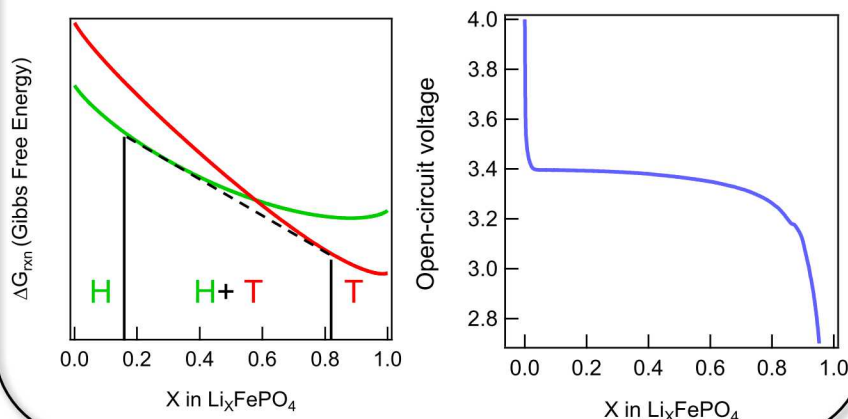
$$V = -\frac{\Delta G}{nFe} = 3.4 \text{ V}$$

Discharge: inserting Li into Li_xFePO_4
Charge: removing Li from Li_xFePO_4



Phase separation in Li_xFePO_4

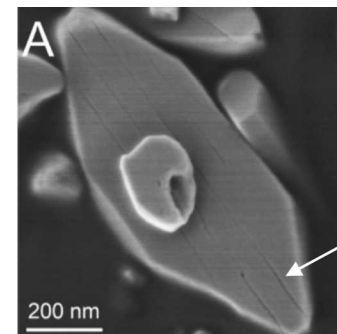
Large miscibility gap



Consequences of phase separation

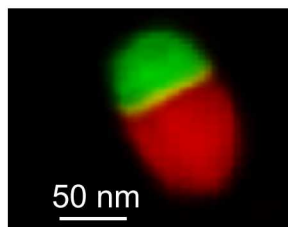
Mechanical strain

Yu et al. *Nano Lett.* (2015)



Phase separation

Single battery particle

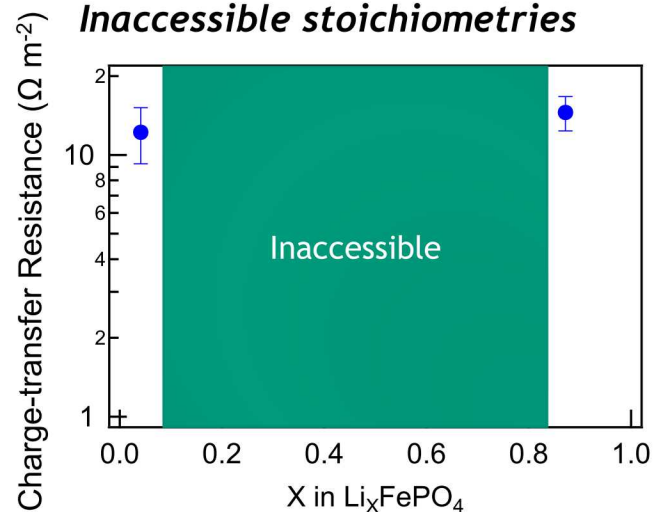


$X=1$  $X=0$

2% lattice mismatch
>1 GPa stress

Y. Li et al. *Nature Mater.* **13**, 1149-56 (2014)

Inaccessible stoichiometries

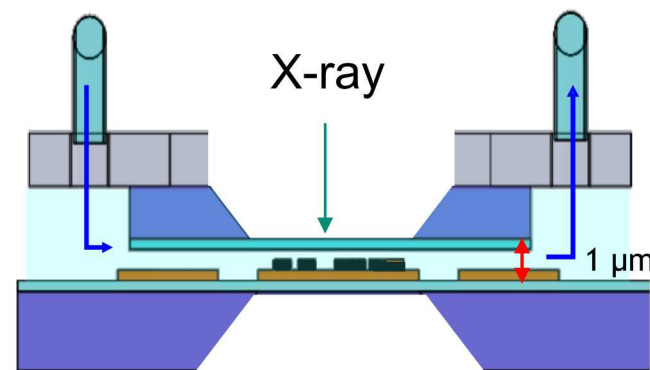
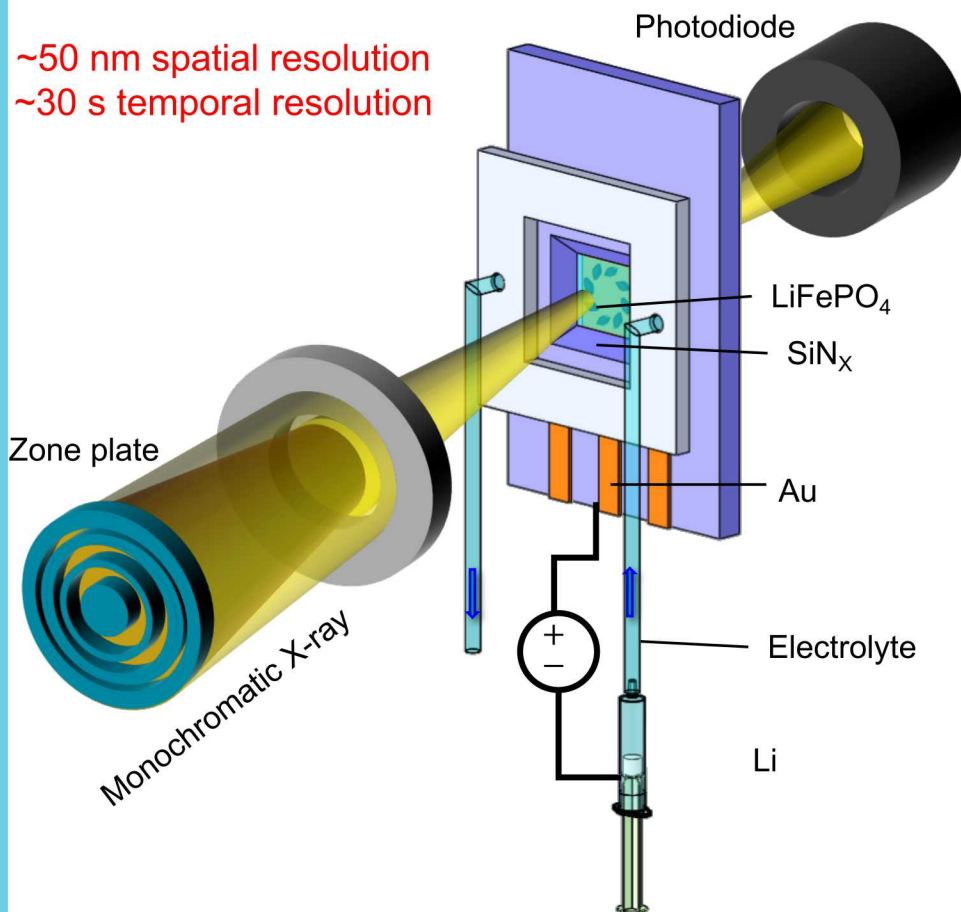


Goal: to understand the kinetics of phase separation and to suppress it

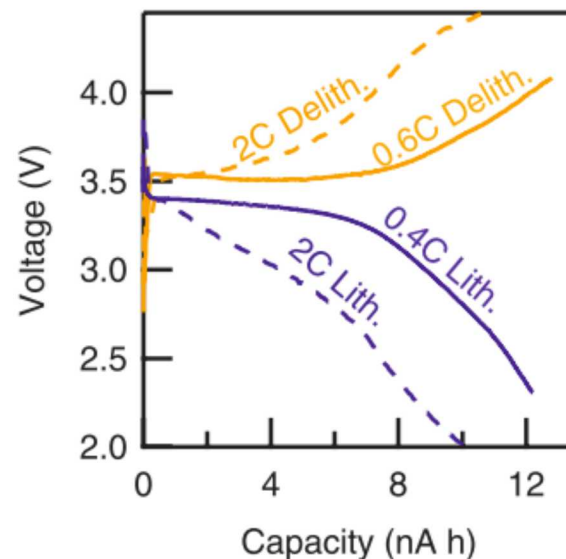


Visualizing battery (dis)charge using in situ X-ray microscopy

Tracking lithium insertion within individual battery particle building blocks



Robust electrochemical cycling

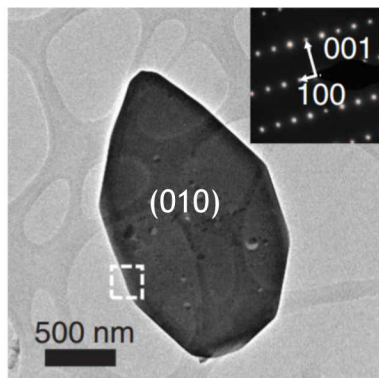


J. Lim*, Y. Li*, et al. *Science*, **353**, 566-571 (2016)

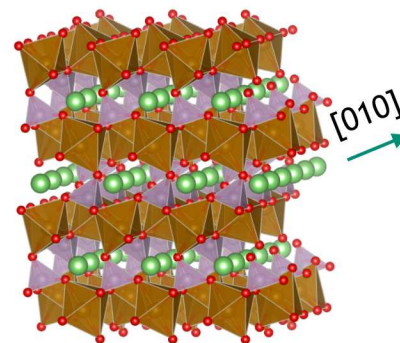
*equal contribution authors



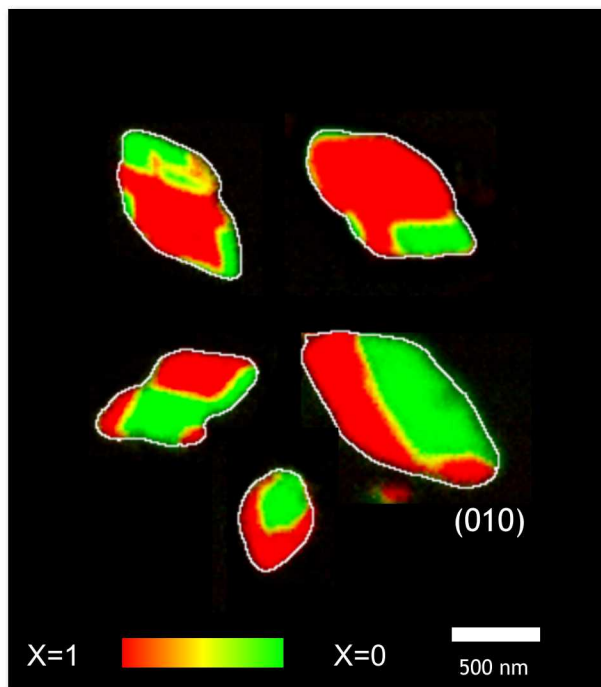
Tracking lithium insertion within particles *in situ*



100-nm thick single crystalline particles



One dimensional lithium conductor

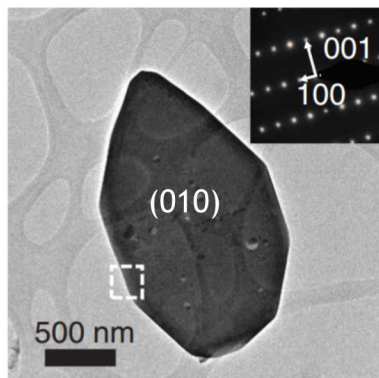


Equilibrium & slow rates

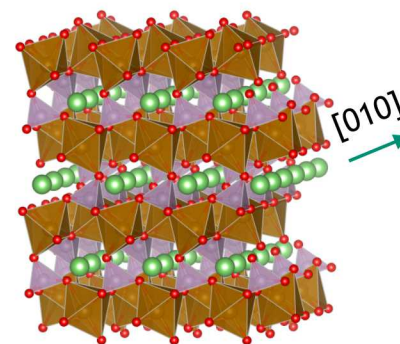
J. Lim*, Y. Li*, et al. *Science*, **353**, 566-571 (2016)
*equal contribution authors



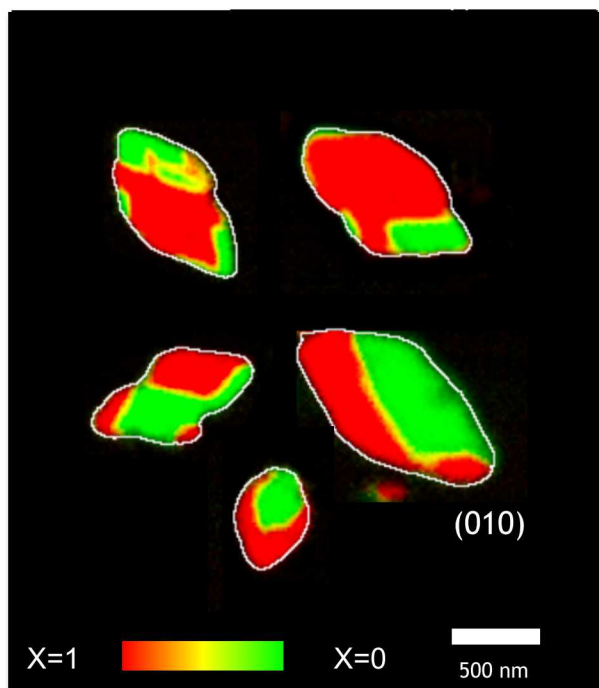
Tracking lithium insertion within particles *in situ*



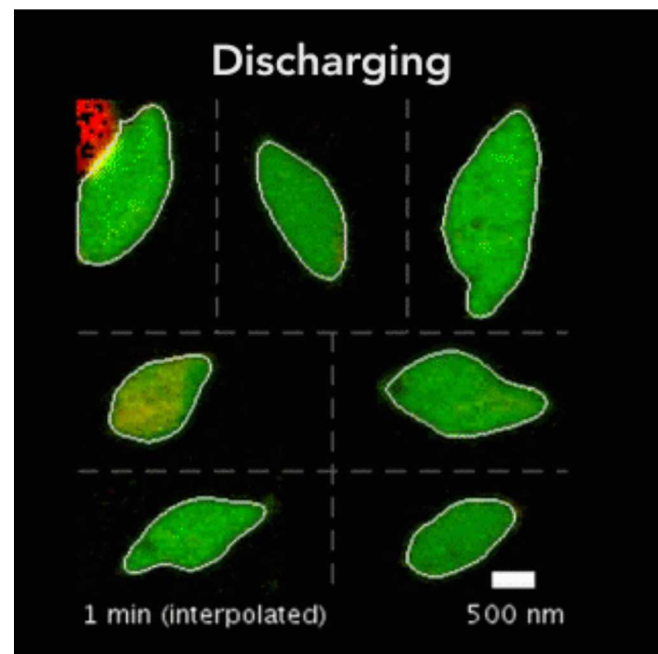
100-nm thick single crystalline particles



One dimensional lithium conductor



Equilibrium & slow rates

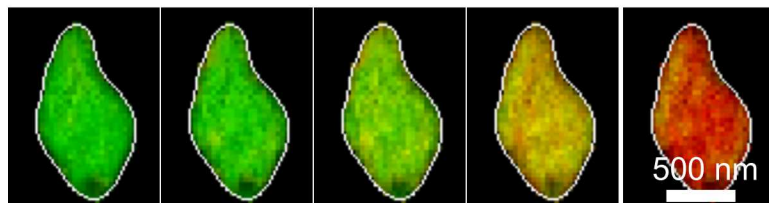


30 min lithiation



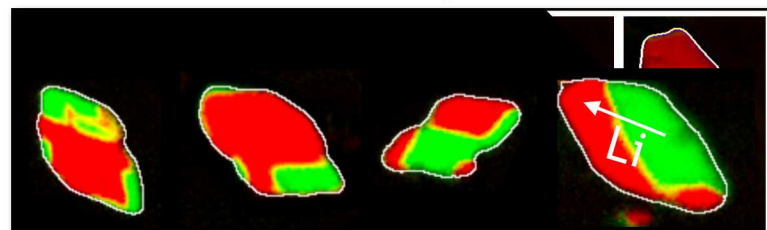
How do we prevent phase separation?

Solid solution



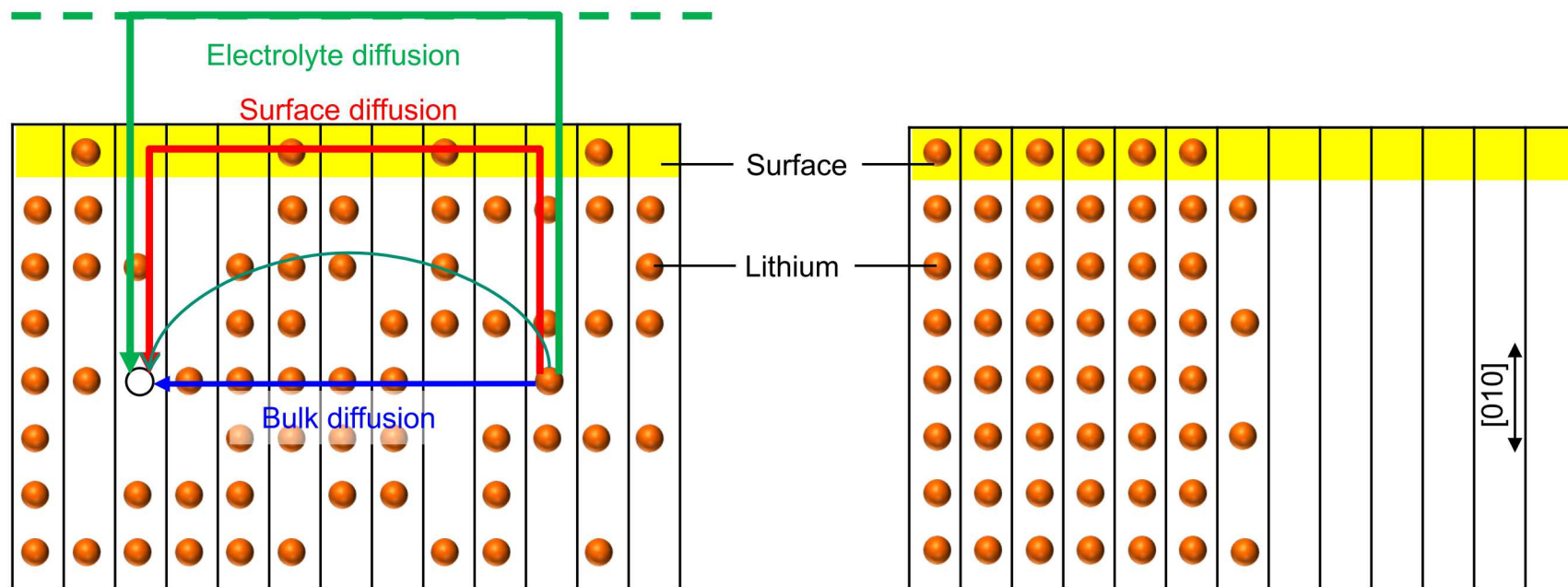
30 min lithiation

Phase separation



Open-circuit

Lithium must diffuse along the non-conducting directions for phase separation



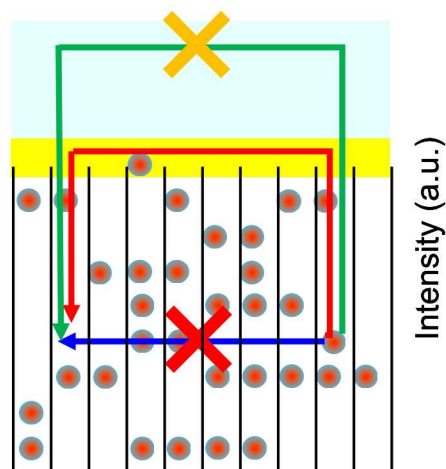
How do we prevent lithium from migrating between channels?



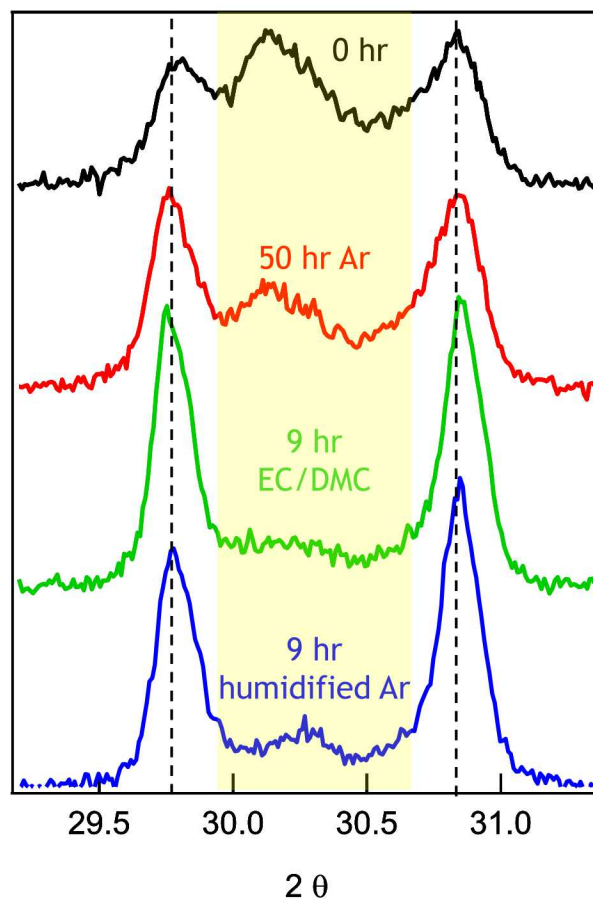
Phase separation without electrolyte

Solid solution $\text{Li}_{0.5}\text{FePO}_4$ prepared under
30 min discharge

LFP Solid solution FP



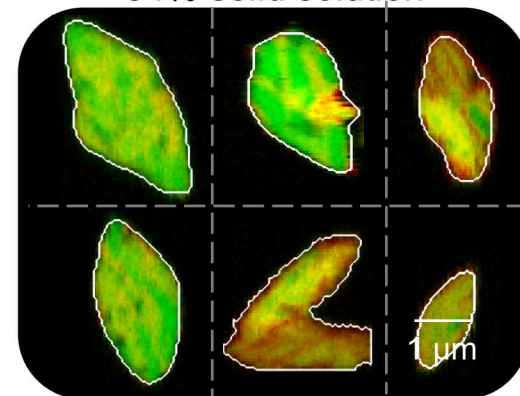
Intensity (a.u.)



Li_xFePO_4
X=1 X=0

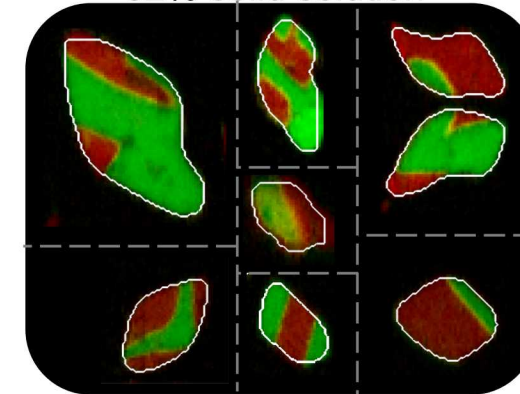
100 h in Ar

84% solid solution



100 h in air

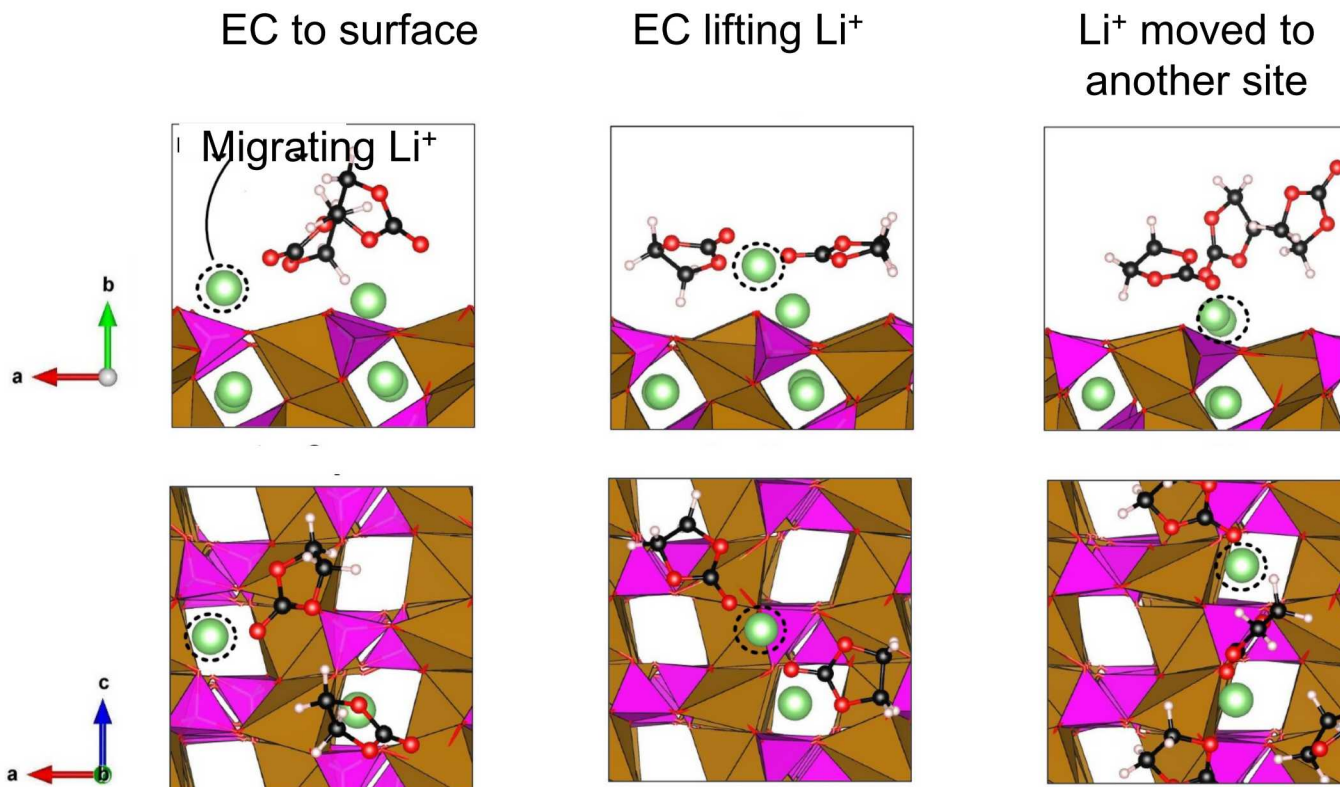
32% solid solution



Fluid molecules enhance phase separation rate.

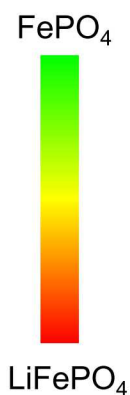
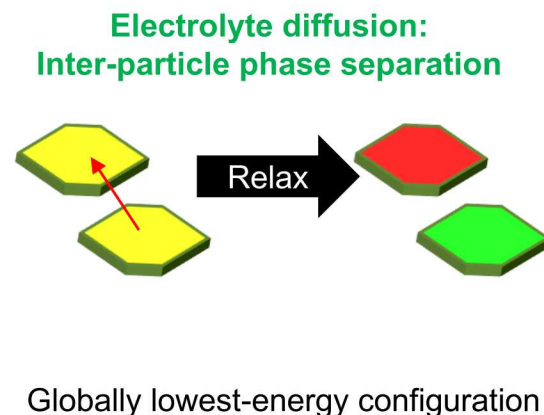
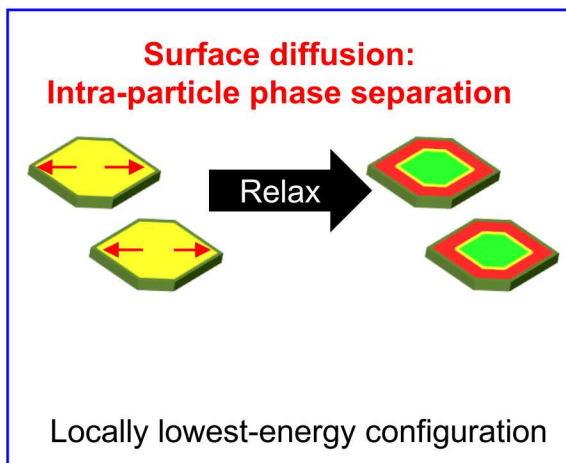
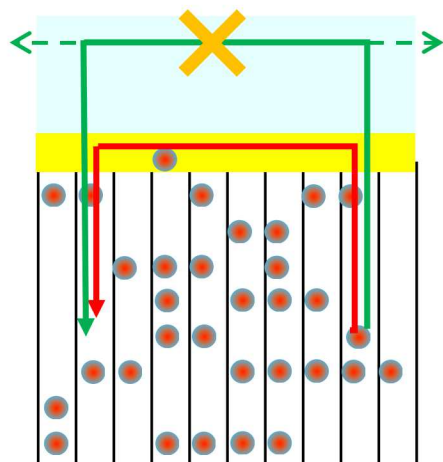


Molecular dynamics shows solvent-assisted surface diffusion

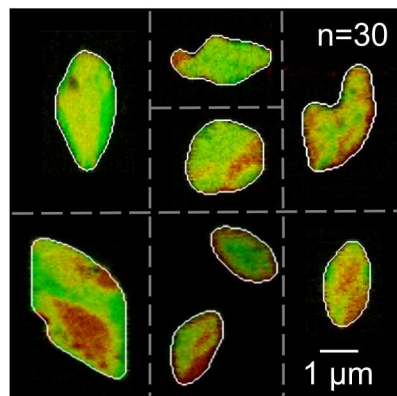




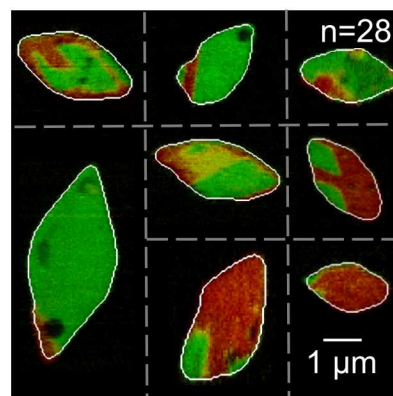
Phase separation with electrolyte



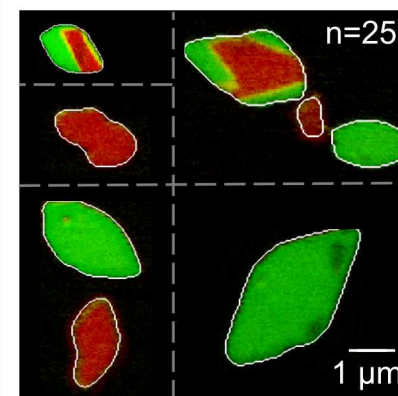
No electrolyte exposure
Mosaic: 0.19 ± 0.14



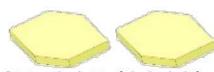
8 hr electrolyte
Mosaic: 0.29 ± 0.16



500 hr electrolyte
Mosaic: 0.62 ± 0.20

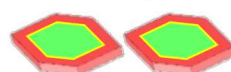


Solid solution



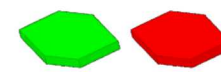
Relax

Phase separation



Relax

Mosaic

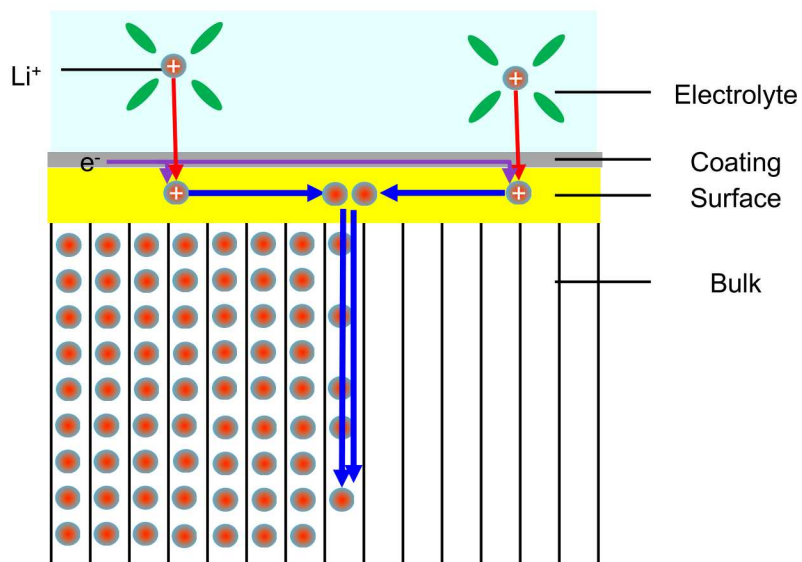


Y. Li, et al. *Nature Mater.*, **17**, 915 (2018)



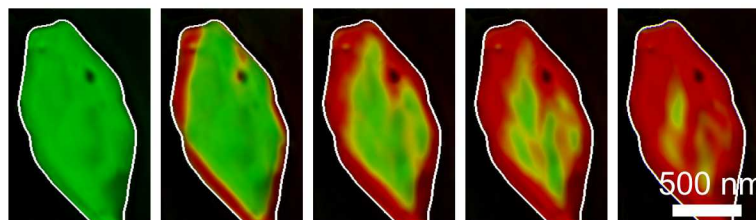
Surface diffusion controls phase separation

Diffusion is faster than lithiation



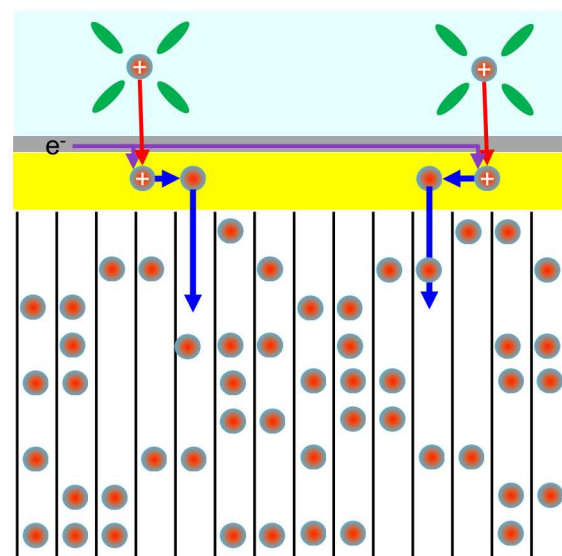
Phase separation

$Da < 1$



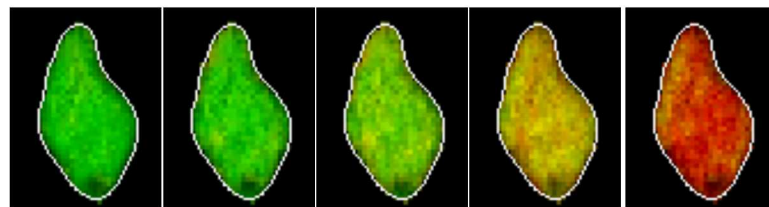
30 min lithiation

Diffusion is slower than lithiation



Solid solution

$Da > 1$



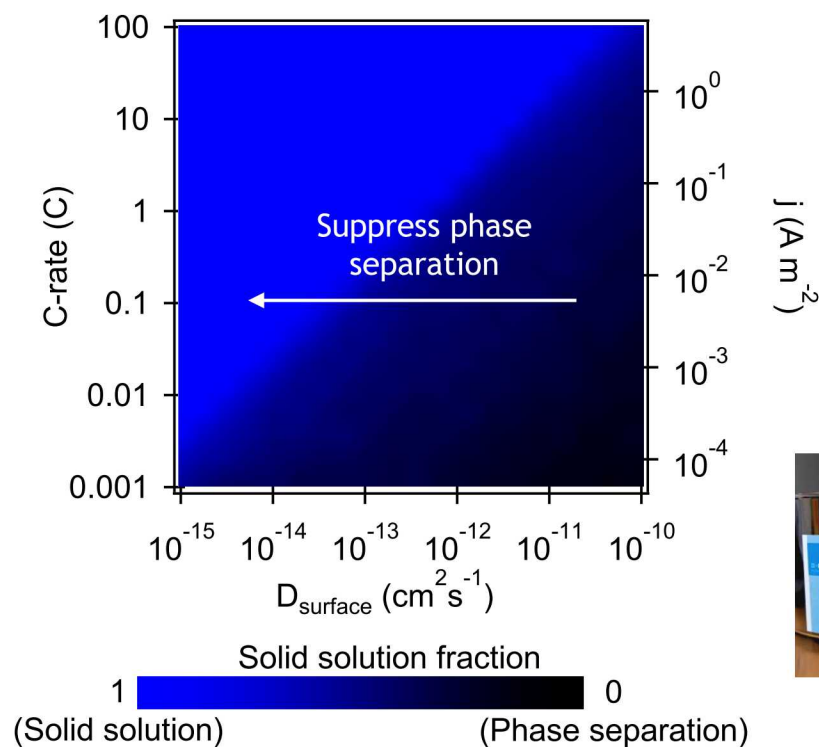
7 hr lithiation



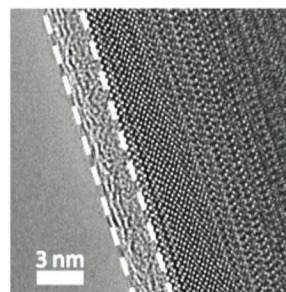
Surface diffusion controls phase separation

Cahn-Hilliard: $\frac{\partial X}{\partial t} = \nabla \left[\frac{D}{k_B T} X(1-X) \nabla \mu_{Li} \right] + j(X, \mu_{Li})$

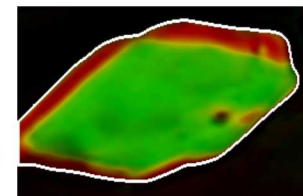
$$\mu_{Li} = \Omega(X)(1-X) + k_B T \ln \frac{X}{1-X} - \kappa \nabla^2 X + \dots$$



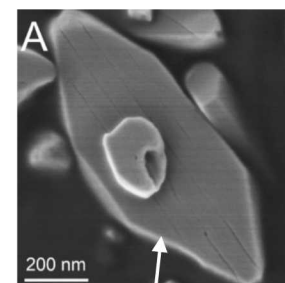
Coating



Phase boundary:
2-4% lattice mismatch
>1 GPa stress



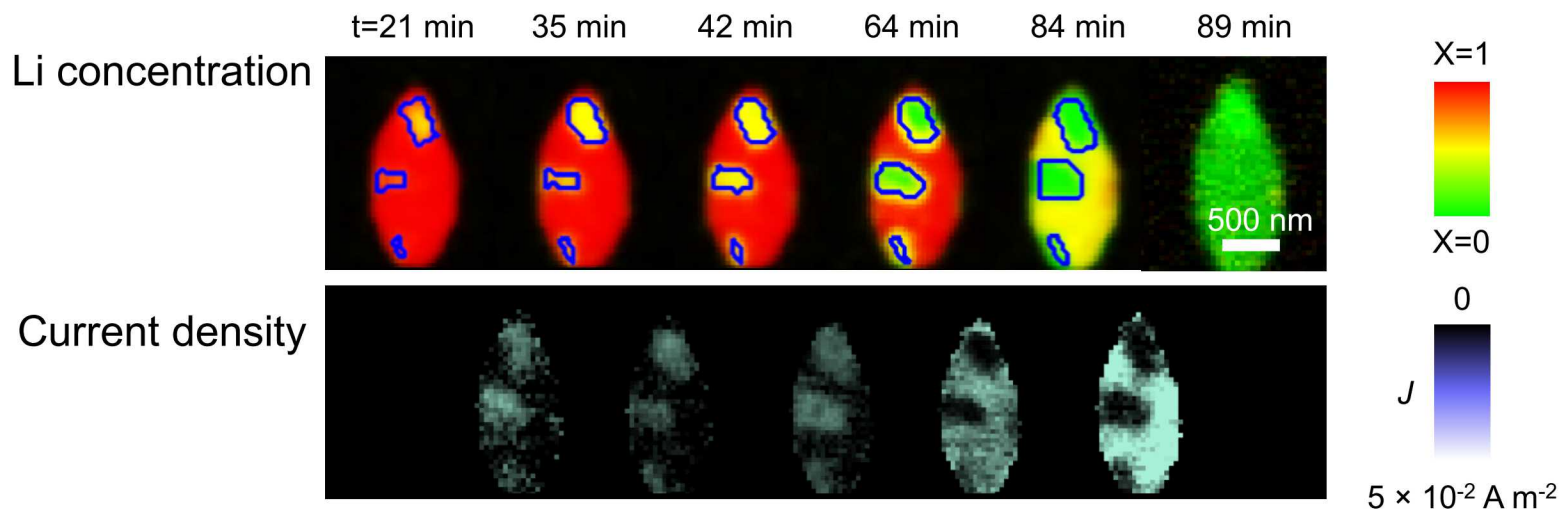
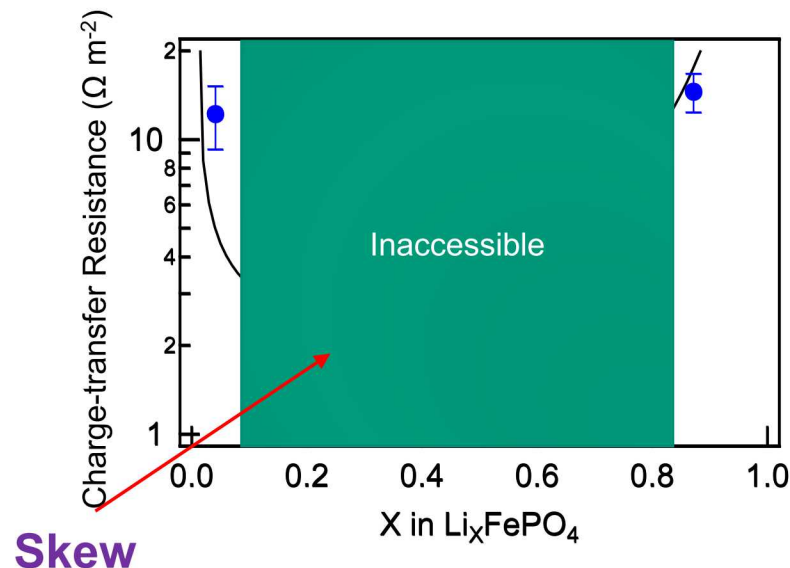
Yu et al. *Nano Lett.* (2015)



Cracks



Solid solution allows us to access intermediate compositions

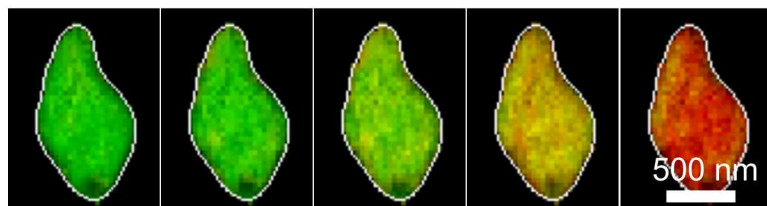


J. Lim*, Y. Li*, et al. *Science*, **353**, 566-571 (2016)

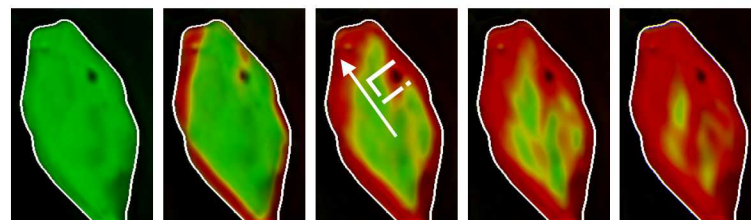


Phase separation in Li_xFePO_4

LiFePO_4 undergoes phase separation thermodynamically, but the phase separation is suppressed upon fast charge and discharge



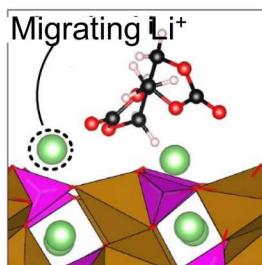
30 min lithiation



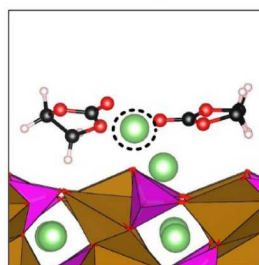
7 hour lithiation

Surface diffusion and fluid molecules facilitate phase separation

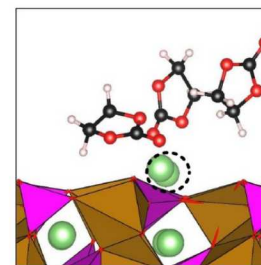
EC to surface



EC lifting Li^+



Li^+ moved to another site





Artificial neural networks and deep learning

Computers are fast and efficient at task-specific programming

$$\frac{dx}{dt} = x^2$$

```
x=1;
dt = 0.01;
for i = 1:1000 {
    dxdt = x.^2;
    xnew = dt*dxdt;
    x = xnew; }
end
```

Computers struggle when there are no clear instructions for the task

Which one of these images is a cat?

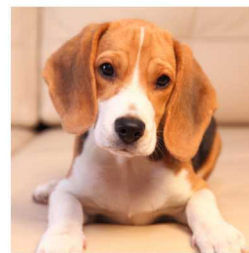
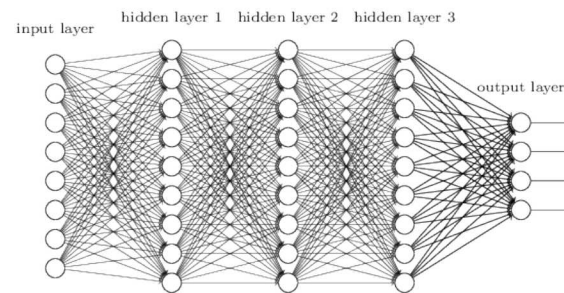


Image recognition
Autonomous driving
Natural language processing

Artificial neural networks: use training examples and error backpropagation to find the matrix weights that correctly maps the input $\mathbf{x}$ onto the desired output $\mathbf{y}$

$$\begin{bmatrix} y_1 \\ \vdots \\ y_m \end{bmatrix} = \begin{bmatrix} w_{1,1} & \cdots & w_{1,n} \\ \vdots & \ddots & \vdots \\ w_{m,1} & \cdots & w_{m,n} \end{bmatrix} \begin{bmatrix} x_1 \\ \vdots \\ x_n \end{bmatrix}$$

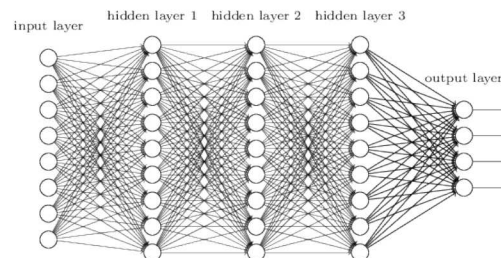
Artificial neural networks are power intensive





Implementing a neural network

$$\begin{bmatrix} y_1 \\ \vdots \\ y_m \end{bmatrix} = \begin{bmatrix} w_{1,1} & \cdots & w_{1,n} \\ \vdots & \ddots & \vdots \\ w_{m,1} & \cdots & w_{m,n} \end{bmatrix} \begin{bmatrix} x_1 \\ \vdots \\ x_n \end{bmatrix}$$

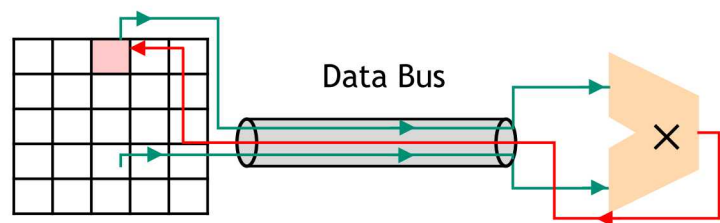


Von Neumann Digital

Separate logic and memory structures

SRAM to store the weights

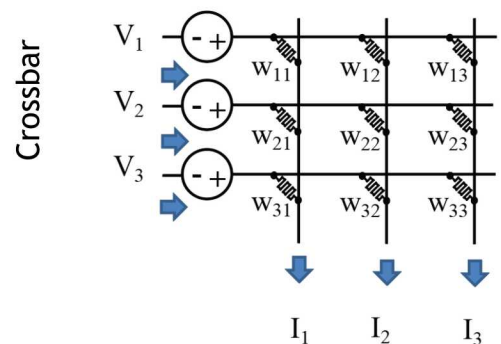
Arithmetic logic unit for multiplication



Uses established CMOS technology
Data bus results in latency and power

In-memory Analog

Use non-volatile memory



$$I_1 = V_1 W_{11} + V_2 W_{21} + V_3 W_{31}$$

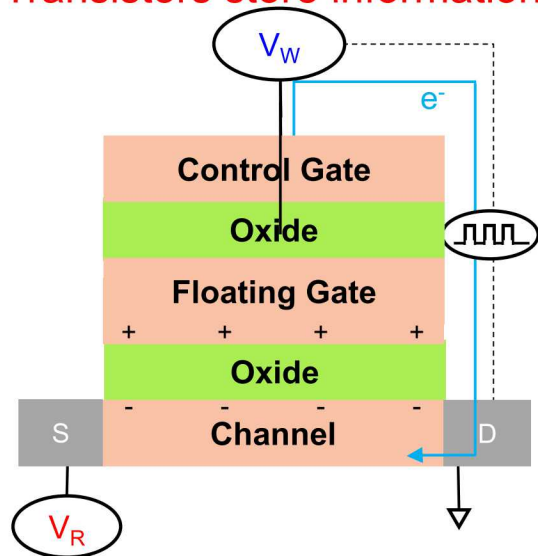
Simultaneous logic and memory
3 orders of magnitude less power

Challenge: find an appropriate non-volatile memory whose conductance can be tuned in a linear and predictable manner



Retention time of transistor-based memory

Transistors store information by moving charge between the gate and the channel



$$\tau_{\text{retention}} = C_{\text{gate}} R_{\text{OFF}}$$

$$C_{\text{gate}} = \epsilon_0 \epsilon_r \frac{A}{t_{\text{ox}}} \approx 10^{-6} \text{ F cm}^{-2} = 10^{-16} \text{ F} \quad (A = 100 \text{ nm}^2)$$

Option 1: transistor switch: $R_{\text{OFF}} \sim 10^{15} \Omega$

$\tau_{\text{retention}} < 1 \text{ s}$ (Dynamic RAM)

S. Ambrogio, G. Burr, et al. *Nature*, **558**, 60 (2018)

Option 2: floating gate oxide switch, $R_{\text{OFF}} \sim \infty$

$\tau_{\text{retention}} > 10 \text{ yr}$ (Flash memory)

10V, high power, limited endurance (10^5)

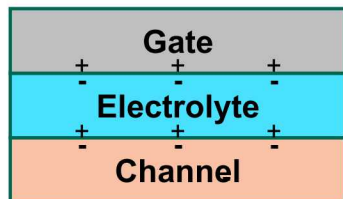
Improving retention by increasing C_{gate} ? $C = \frac{\Delta Q}{\Delta V}$

Electrochemical double-layer transistors

Electrons stored at the **interface**

$$C_{\text{gate}} \sim 10^{-4} \text{ F cm}^{-2}$$

$$t_{\text{ox}} < 1 \text{ nm}$$



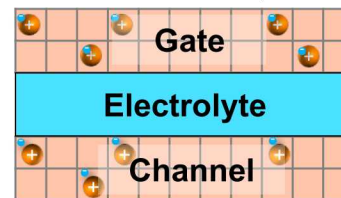
P. Gkoupidenis, G. Malliaras, *Adv Mater*, **27**, 7176 (2015)

Ion insertion transistors

Electrons (and ions) are stored in the **bulk** of the material (pseudo-capacitor)

Electron/ion neutral ambipolar pair enables ΔQ without electrostatic charging

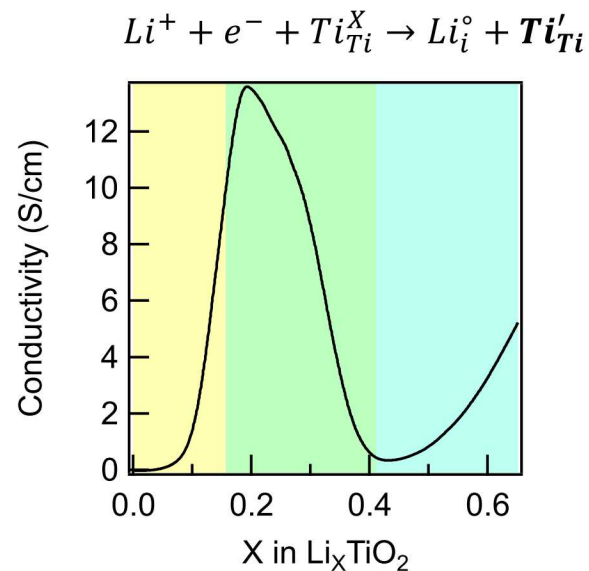
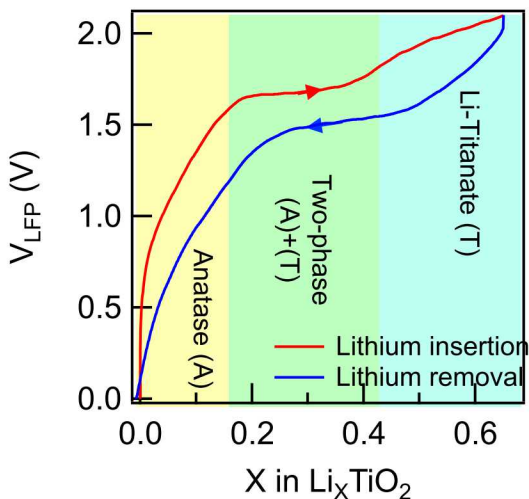
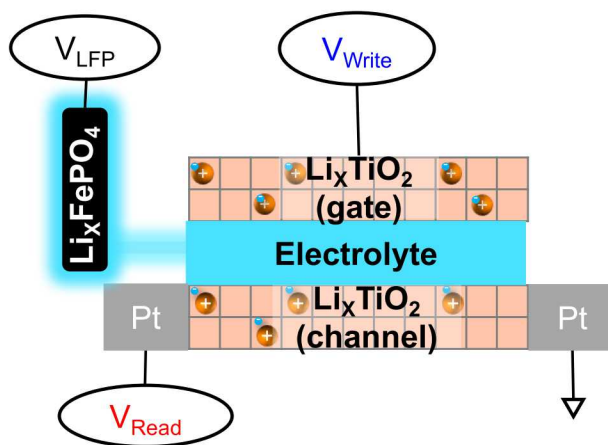
$$C_{\text{gate}} \sim 10^{-1} \text{ F cm}^{-2} \text{ (100-nm film)}$$



E. Fuller, *Adv Mater*, **29**, 1604310 (2017)
Y. van de Burgt, *Nat. Mater.*, **16**, 414 (2017)



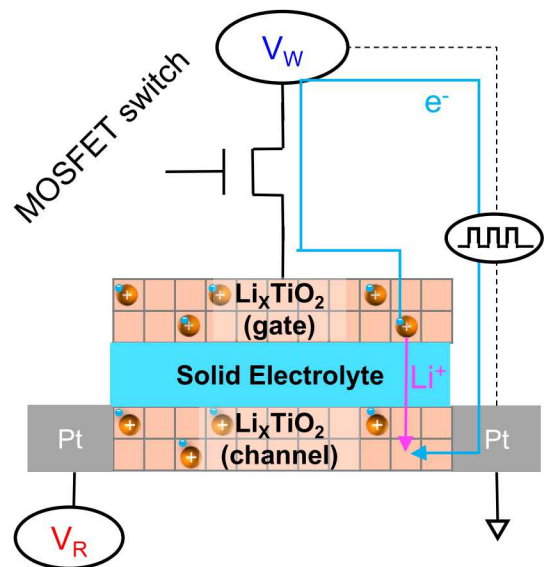
Lithium insertion into Li_xTiO_2 (anatase)



Small polarons increase the electronic conductivity

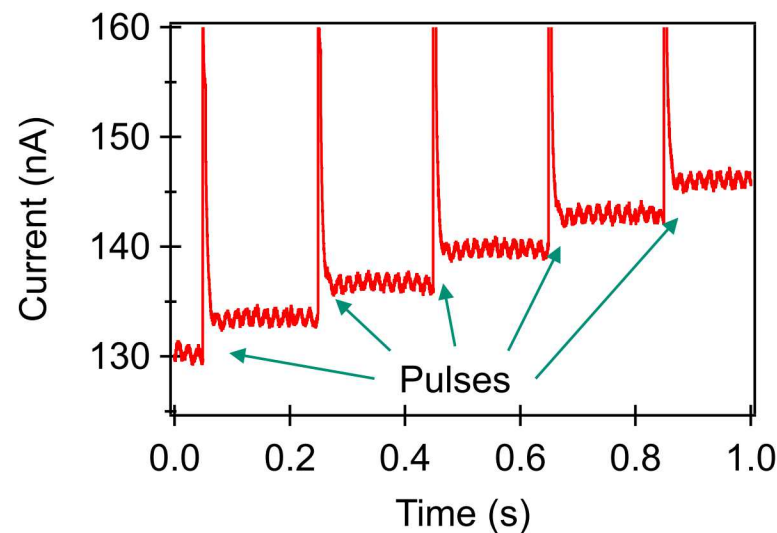
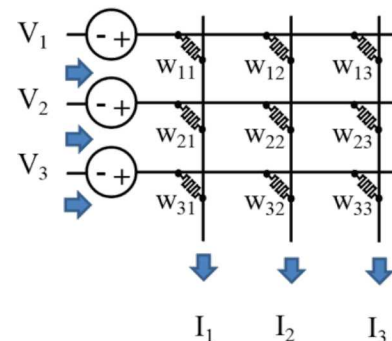


Ion insertion transistor using Li_xTiO_2 (anatase)



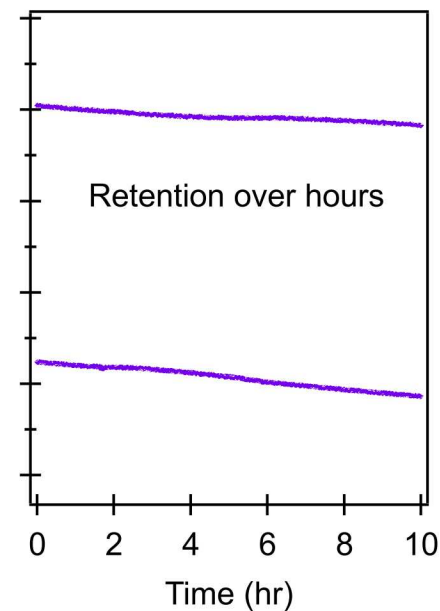
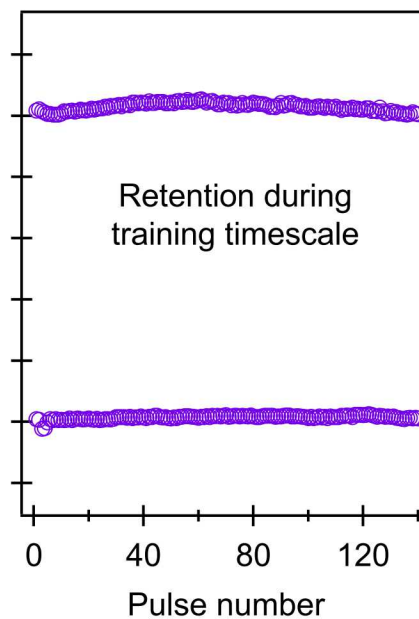
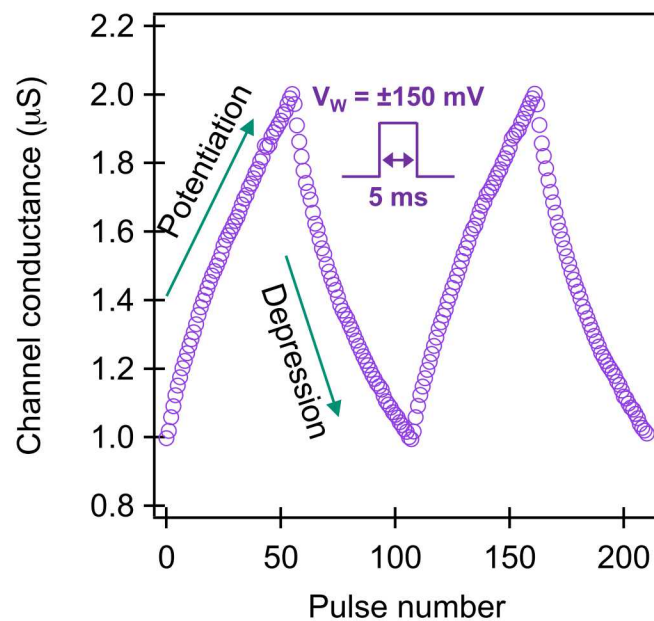
- Analog, non-volatile states
- Linear and symmetric programming
 - 150-mV write pulses

$$V_W = \pm 150 \text{ mV}$$





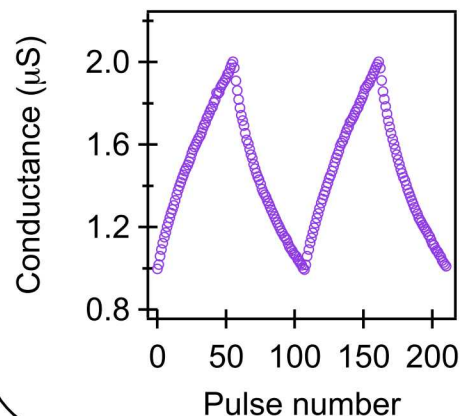
Ion insertion transistor using Li_xTiO_2 (anatase)



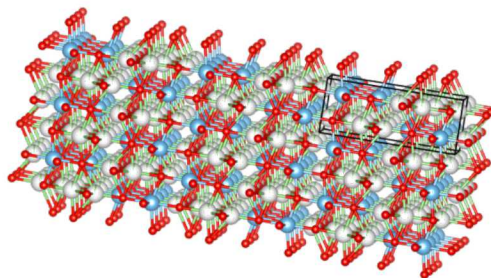


Linear programming and high accuracy

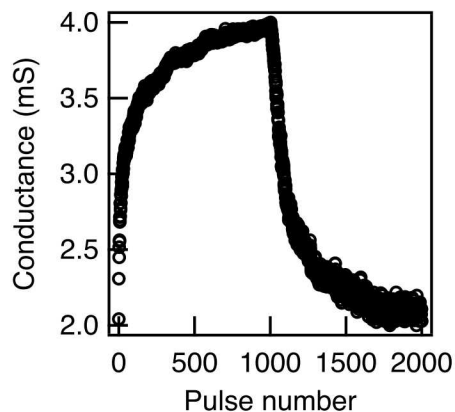
Li_xTiO_2 transistor
Voltage: ± 0.15 V



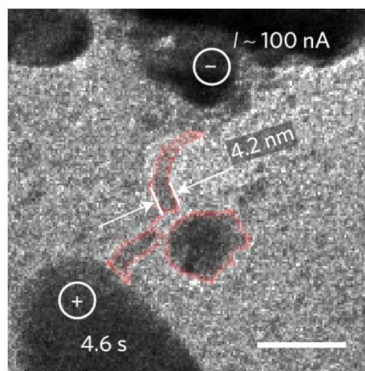
Ion insertion: stores information continuously as dopants in a crystal



TaO_x Memristor
Voltage: ± 1 V



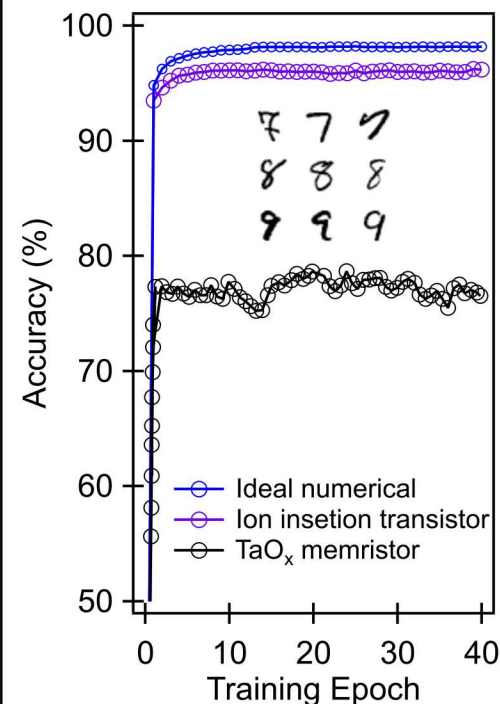
Memristors: stores information at filaments



R. Jacobs-Gedrim et al. *ICRC*, 2017

Wang et al. *Nat Mater.* 2017

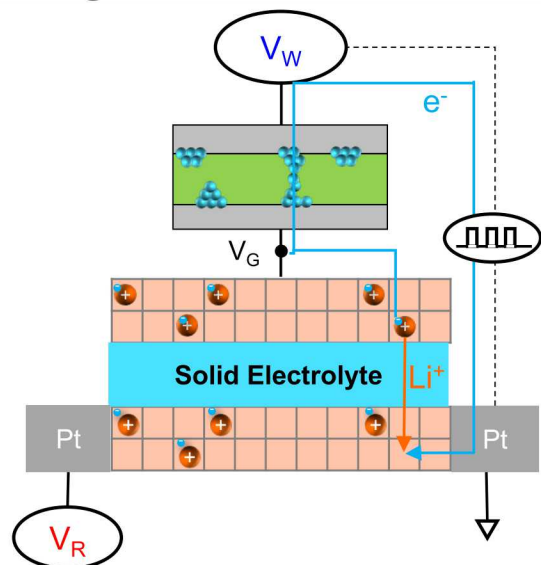
Linearity is essential to train an accurate neural network



Crossbar Simulator



Low-voltage, Si-free electrochemical memory



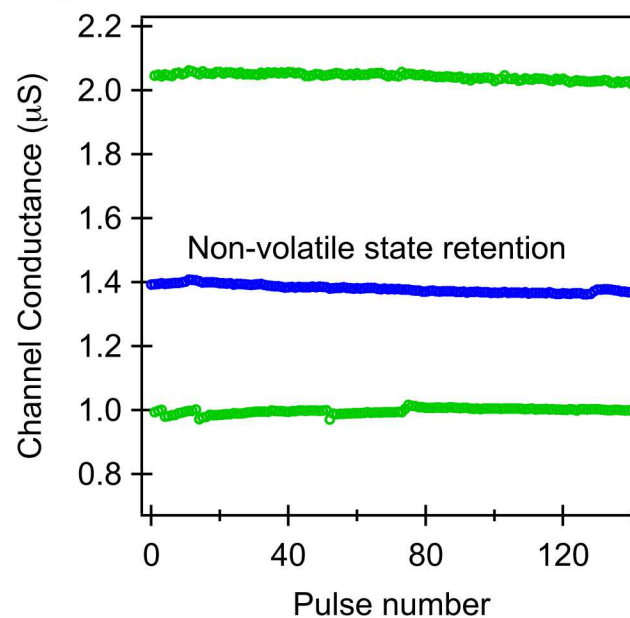
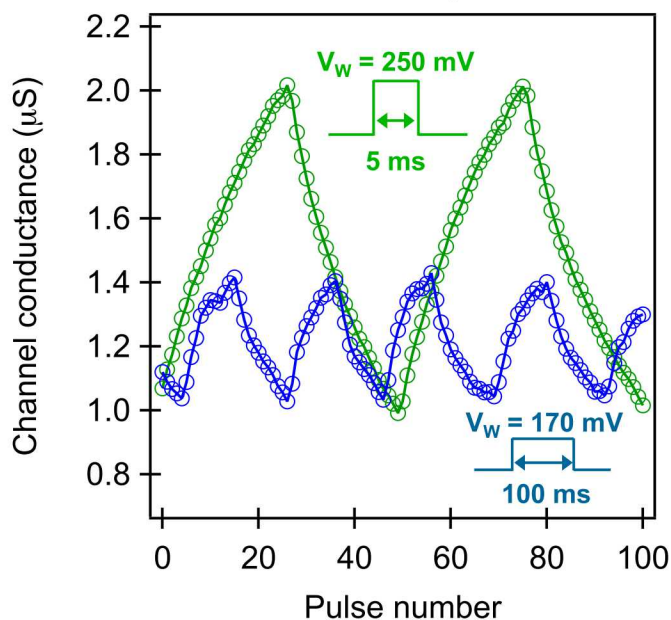
Diffusive memristor (Ag in SiO_x): high ON/OFF ratio

Ion insertion transistor: high charge density via bulk storage

Both: Low switching voltages

Wang et al. Nat. Mater. 2017

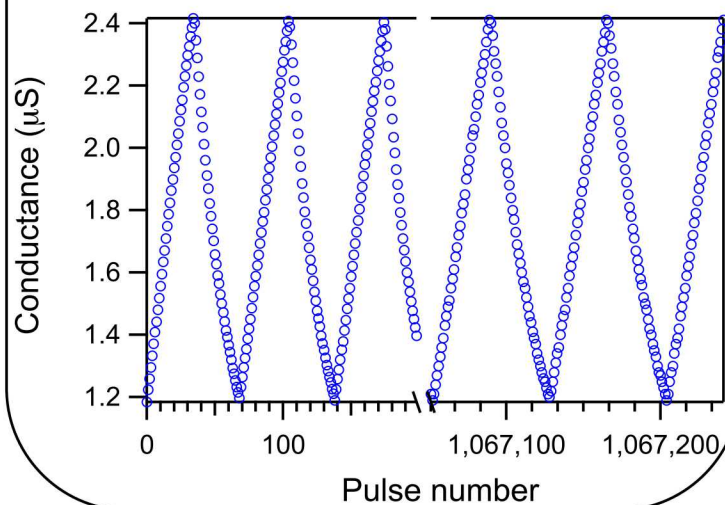
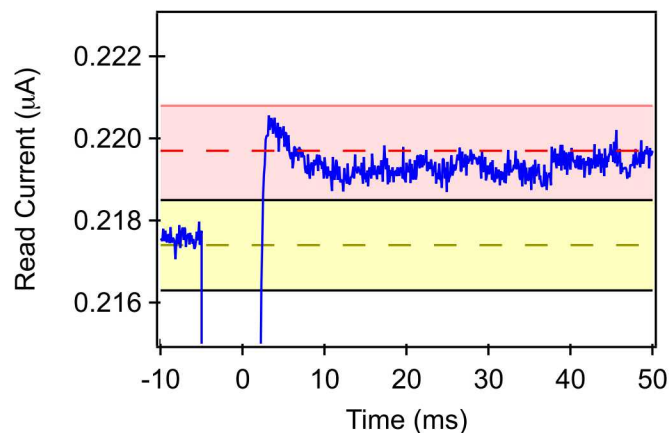
Non-volatile memory that switches at just 6 times the thermal voltage



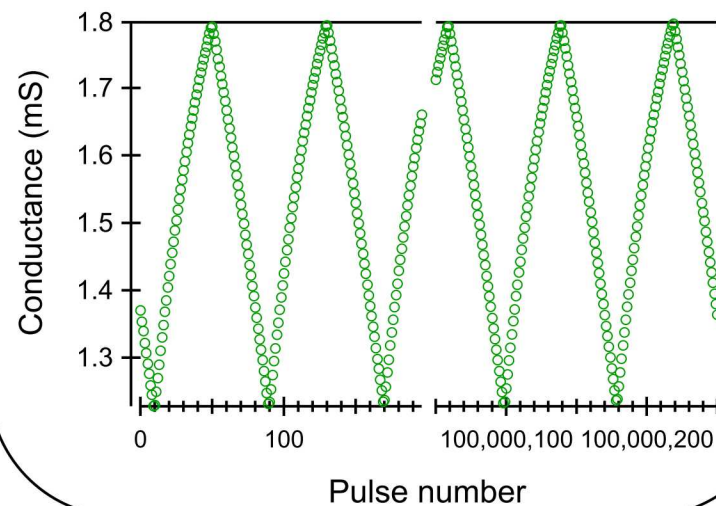
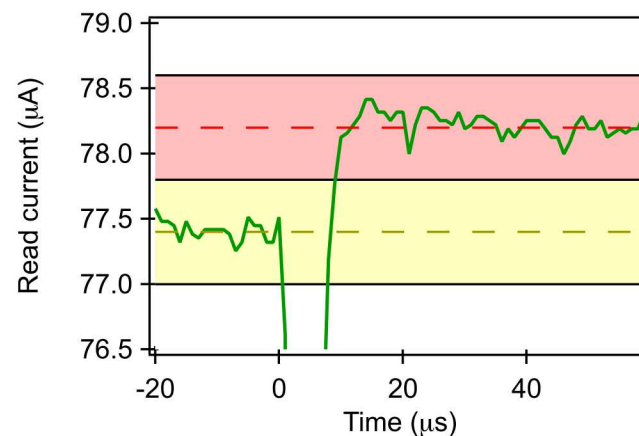


Switching speed and endurance

Inorganic (Li_xTiO_2) ~5 ms,
 10^6 read/write cycles



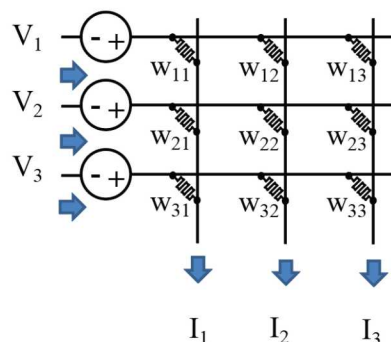
Organic (PEDOT/PSS) ~10 μs,
 10^8 read/write cycles



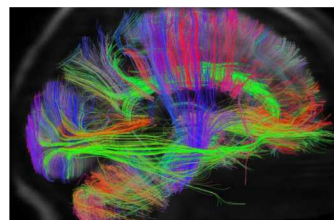


Ion insertion transistors for neuromorphic computing

Analog neuromorphic computing provides lower energy and more parallelism compared to digital computing.

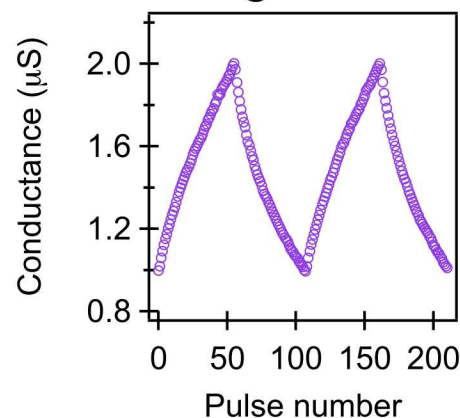
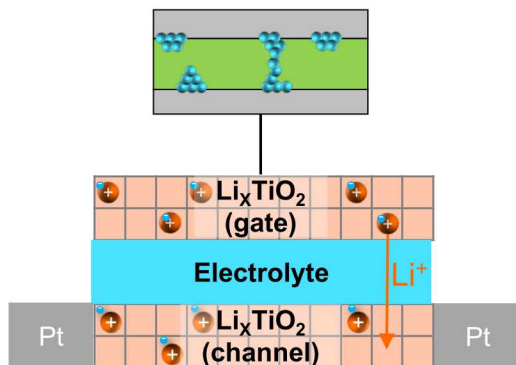


the brain ~10 Hz



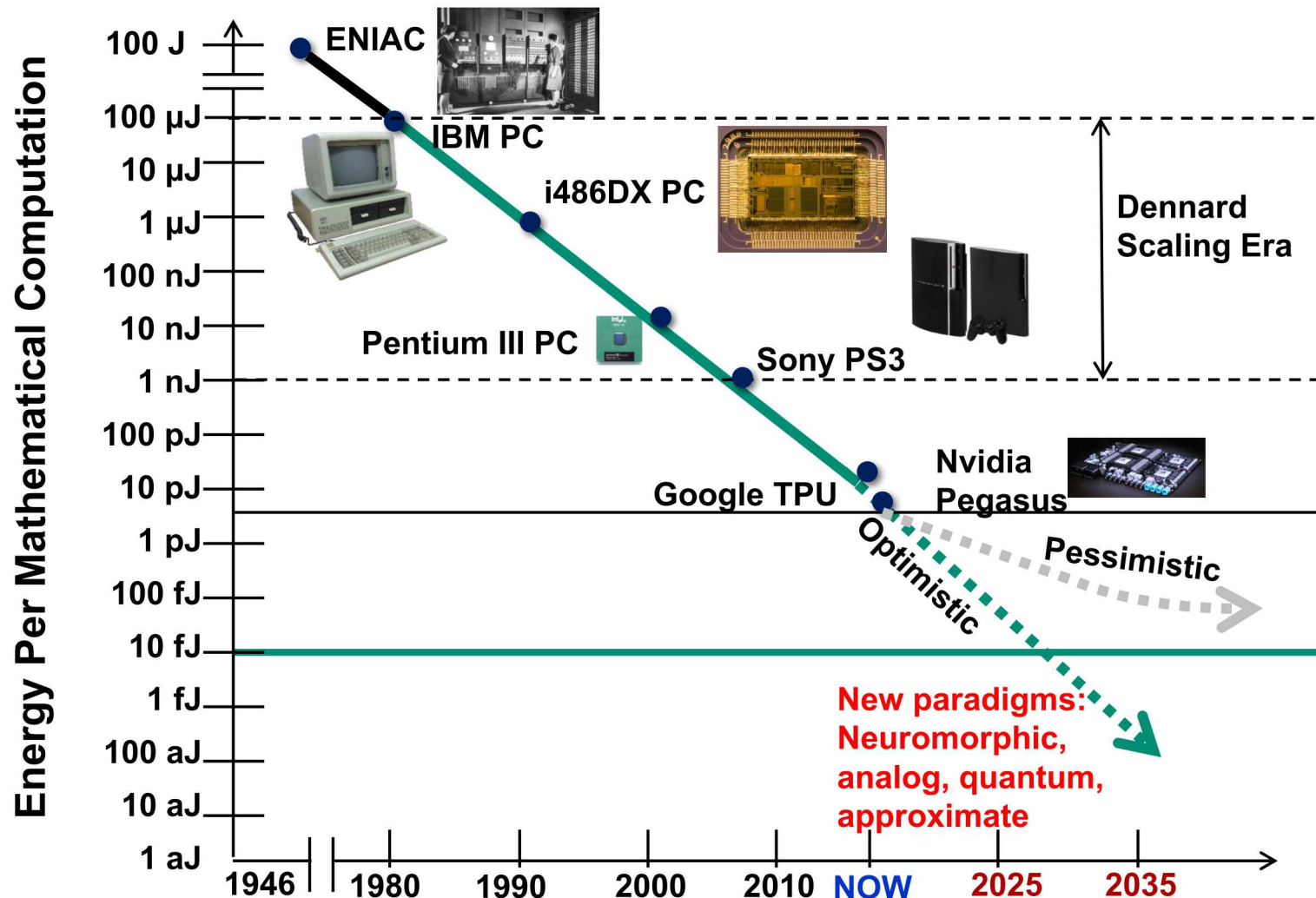
100 billion neurons
100 trillion synapses

Electrochemical ion insertion can be used to create highly linear, low voltage non-volatile analog transistors





Evolution of Computing Machinery





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Our team: Materials Physics, Livermore. Intern and postdoc positions available. Contact yiyli@sandia.gov or aatalin@sandia.gov



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Saiful Islam

Hungru Chen

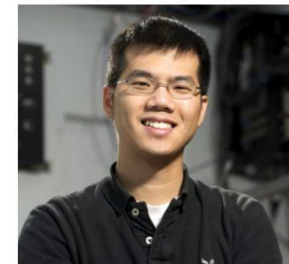
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Will Gent

Norman Jin



W. Chueh



J. Lim

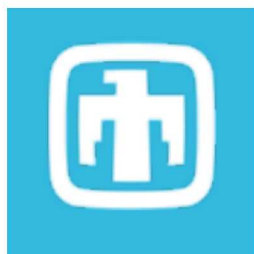
MIT

Martin Bazant

Dan Cogswell



M. Bazant



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Energy and computing

2000: Why is my computer so slow?



2018: Why is my phone's battery dead?

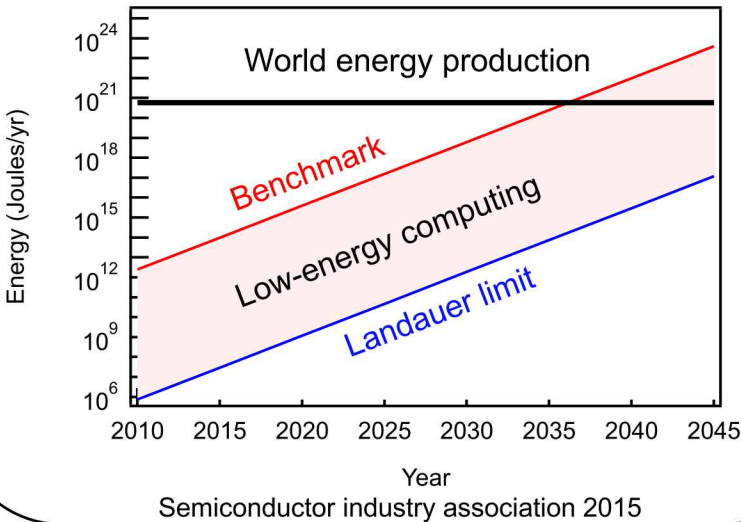


Autonomous Chevy Bolt (Electric)
Propulsion at 20mph: 5 kW
Computers & sensors: 2 kW



J. Gawron, *Env. Sci.Tech.* 2018

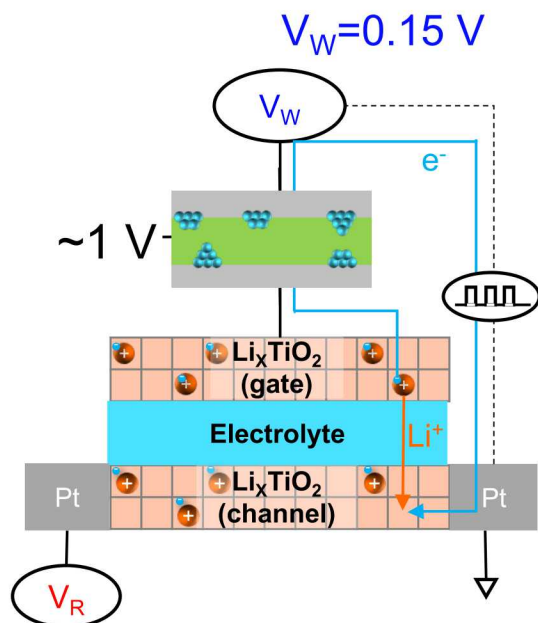
Computing already consumes over 10% of global electricity



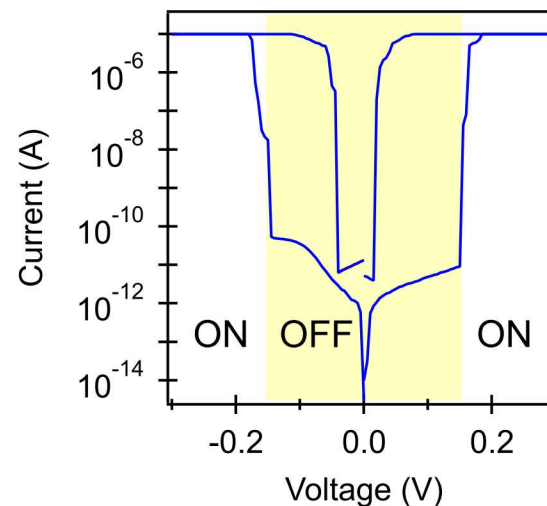
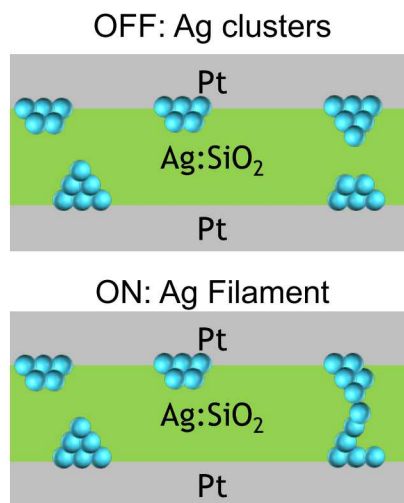


Non-volatile memory at the thermal limit

What is the minimum switching voltage for non-volatile memory?



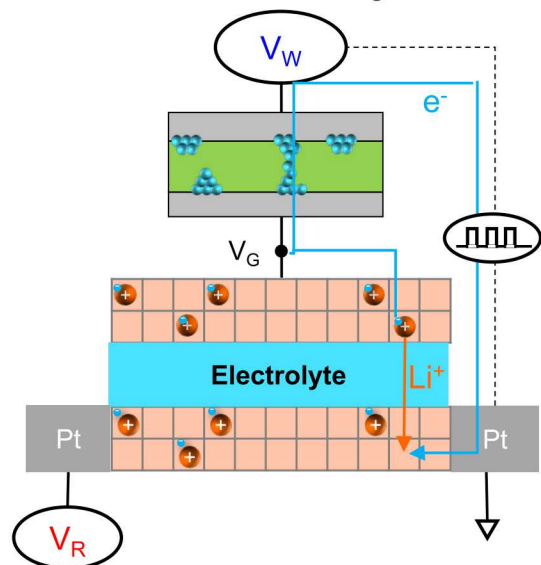
Diffusive memristor



Diffusive memristor has an extremely sharp ON/OFF ratio at just 150mV threshold



Non-volatile memory at the thermal limit

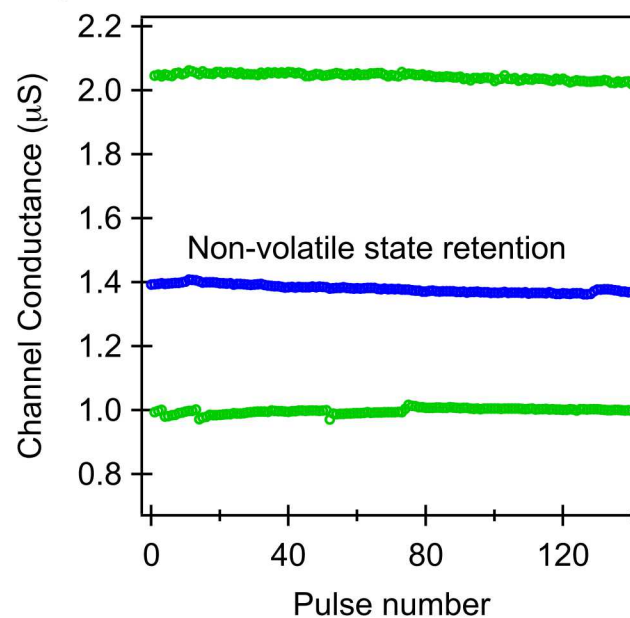
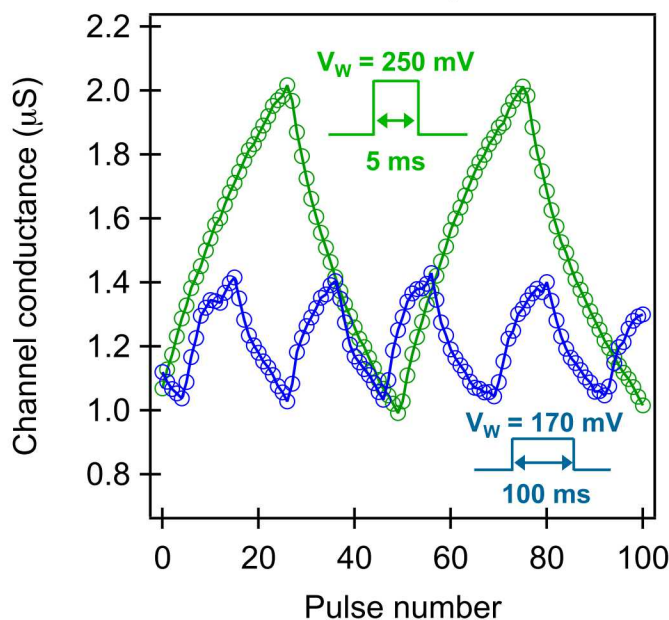


Diffusive memristor: high ON/OFF ratio

Ion insertion transistor: high charge density via bulk storage

Both: Low switching voltages

Non-volatile memory that switches at just 6 times the thermal voltage

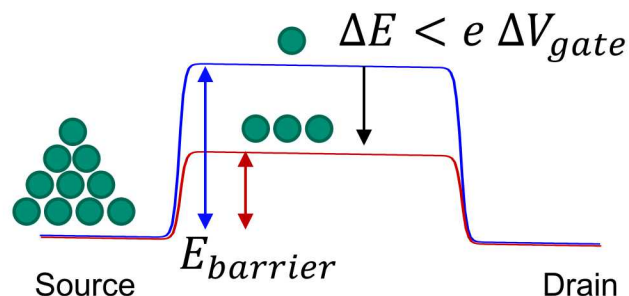




How is non-volatility attained at low switching voltages

High ON/OFF ratios with low

Field-effect transistor



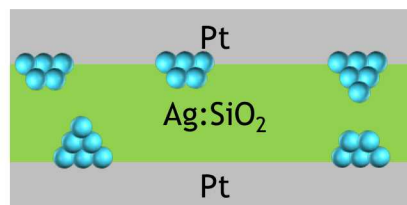
$$n \sim \exp\left(-\frac{E_{\text{barrier}}}{k_B T}\right)$$

The gate voltage lowers the energy barrier

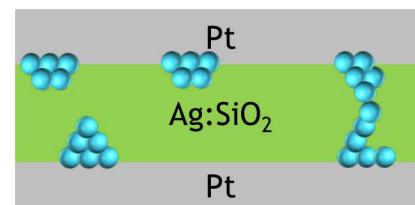
Transistor gain cannot exceed 60 mV/decade due to the Boltzmann electron energy distribution

Diffusive memristor

OFF (Ag clusters)



ON (Ag filament)



$$E_{\text{barrier}} \sim n_{\text{Ag}} (\mu_{\text{Ag,filament}} - \mu_{\text{Ag,clusters}})$$

Voltage modifies $\Delta\mu_{\text{Ag}}$

$$\Delta E_{\text{barrier}} > e\Delta V$$

Switching slope ~ 2 mV/decade, allowing for ~ 150 mV switching thresholds



Materials technologies for energy and climate challenges

Energy generation



Solar

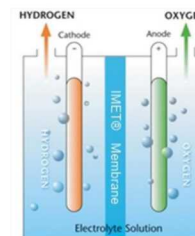


Wind

Energy storage



Batteries



Chemical fuels

Negative carbon emission

Energy efficiency



Solid-state
LED Lighting



Low-energy
computing

Resource management



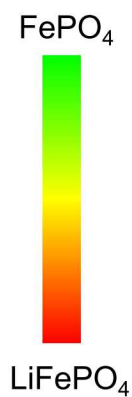
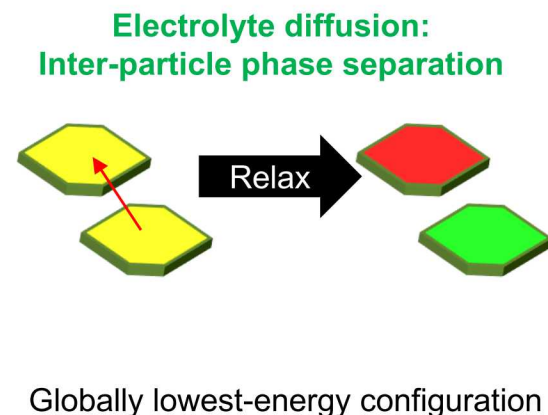
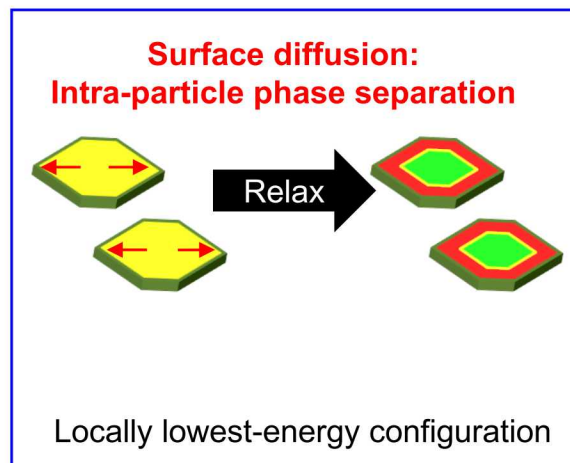
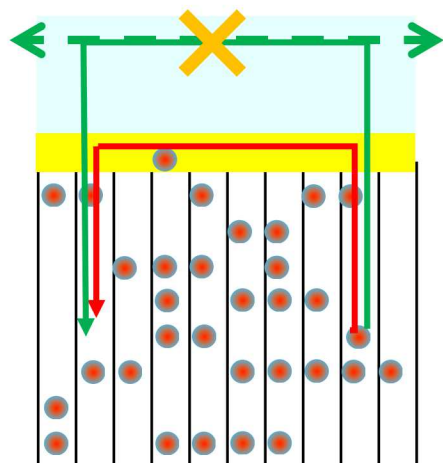
Mining



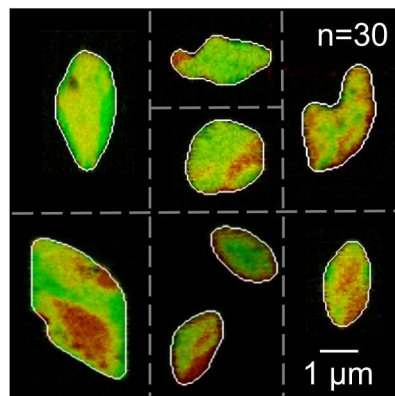
Recycling



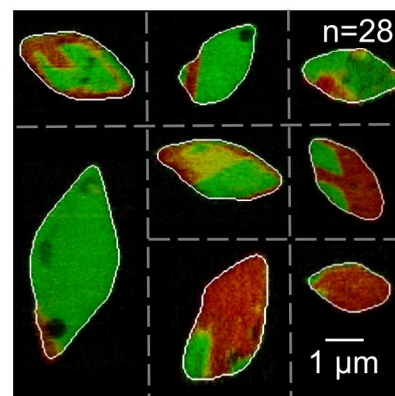
Phase separation with electrolyte



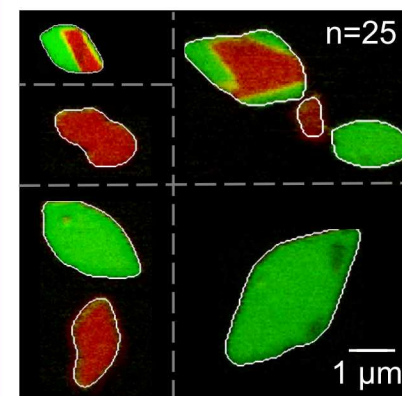
No electrolyte exposure
Mosaic: 0.19 ± 0.14



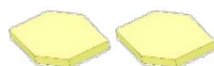
8 hr electrolyte
Mosaic: 0.29 ± 0.16



500 hr electrolyte
Mosaic: 0.62 ± 0.20



Solid solution



Relax

Phase separation



Relax

Mosaic

