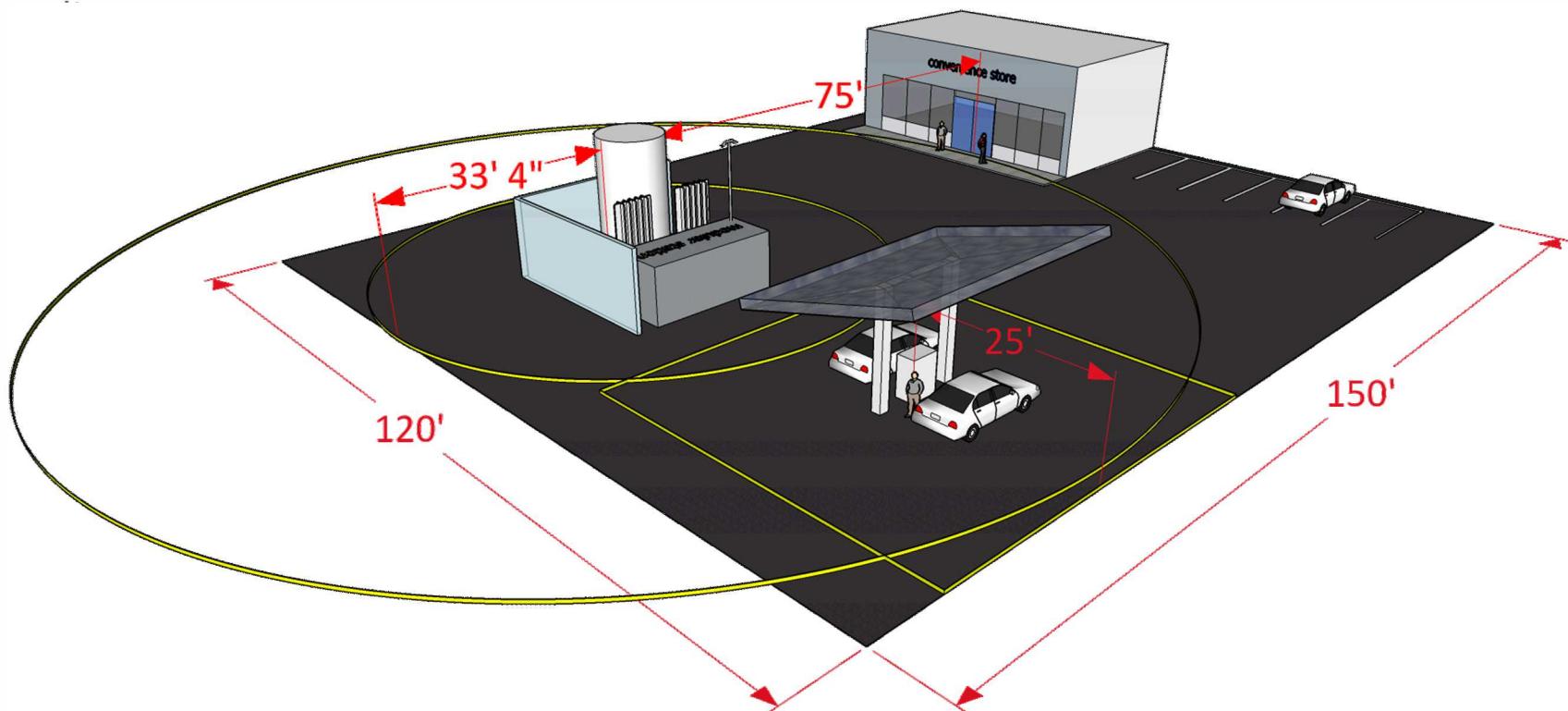


Hydrogen behavior R&D for Safety, Codes and Standards at Sandia National Labs

Ethan S. Hecht

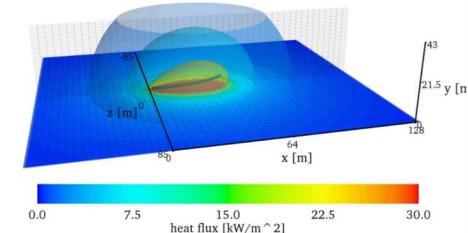
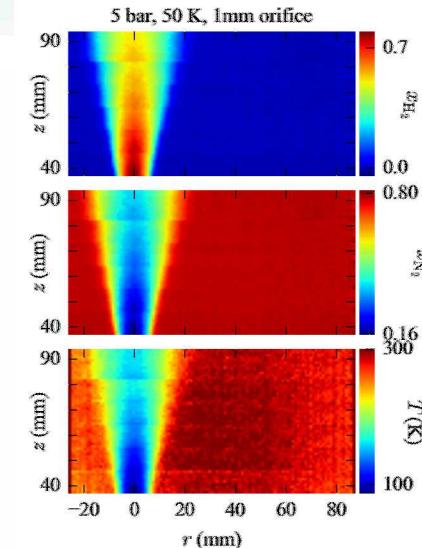
Current separation distances for liquid hydrogen are based on consensus, not science

- Higher energy density of liquid hydrogen over compressed H₂ (and lack of pipelines) make this technology viable for larger fueling stations (logistically and economically)
- Even with credits for insulation and fire-rated barrier wall 75 ft. offset to building intakes and parking make footprint large
- Previous work by our group led to science-based, reduced, gaseous H₂ separation



Sandia's Approach: coordinated activities that facilitate deployment of hydrogen technologies

- Hydrogen Behavior
 - Develop and validate scientific models to accurately predict hazards and harm from liquid releases, flames, etc.
- Quantitative Risk Assessment, tools R&D
 - Develop integrated methods and algorithms enabling consistent, traceable, and rigorous QRA (Quantitative Risk Assessment) for H₂ facilities and vehicles
- Enable Hydrogen Infrastructure through Science-based Codes and Standards
 - Apply QRA and behavior models to real problems in hydrogen infrastructure and emerging technology



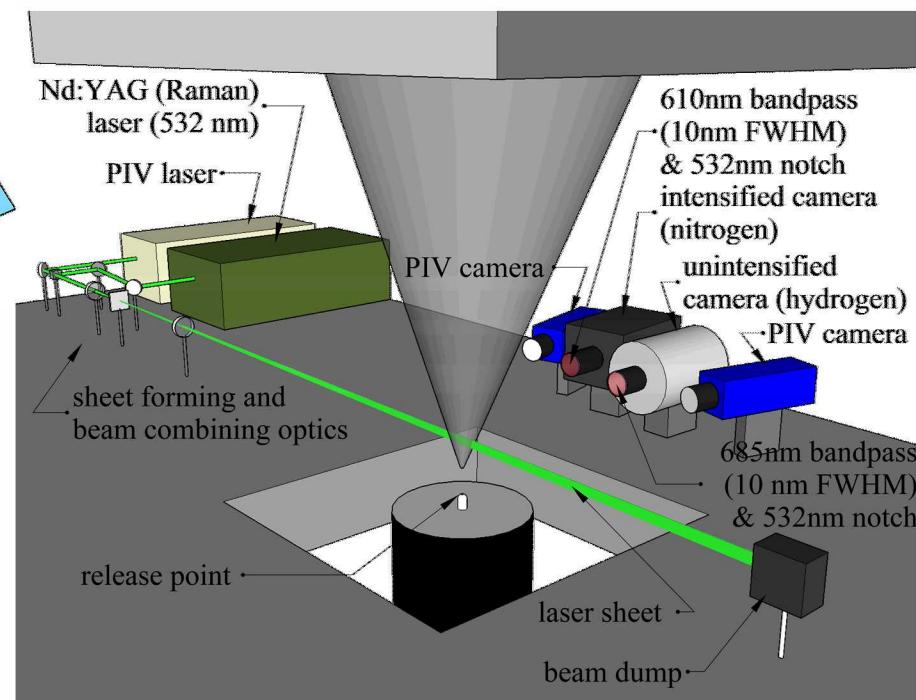
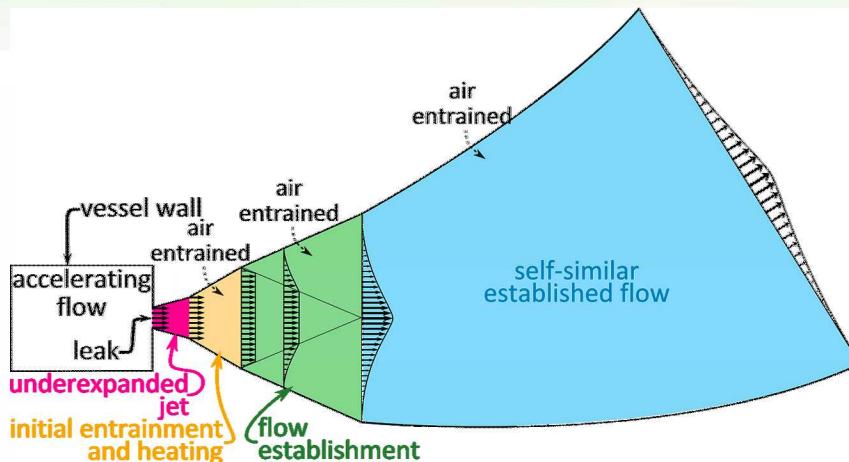
Need to enable *predictive* modeling across H₂'s range of use (cryogenic models lack validation)

Issue: Cryogenic H₂ releases have been outdoors and/or instrumented with low fidelity sensors (space and time), with experimental uncertainty too high for model validation

- FY18 goals:
 - Complete analysis of lab-scale experimental data and validation of ColdPLUME model - complete
 - FY18 milestone: Develop a diagnostic and measure the plume from a liquid hydrogen truck depressurization in at least 2 dimensions

➤ Enable the simulation of critical scenarios and provide the science for revisions to the 2022 edition of NFPA 2

Lab-scale cryogenic hydrogen dispersion analysis and model validation



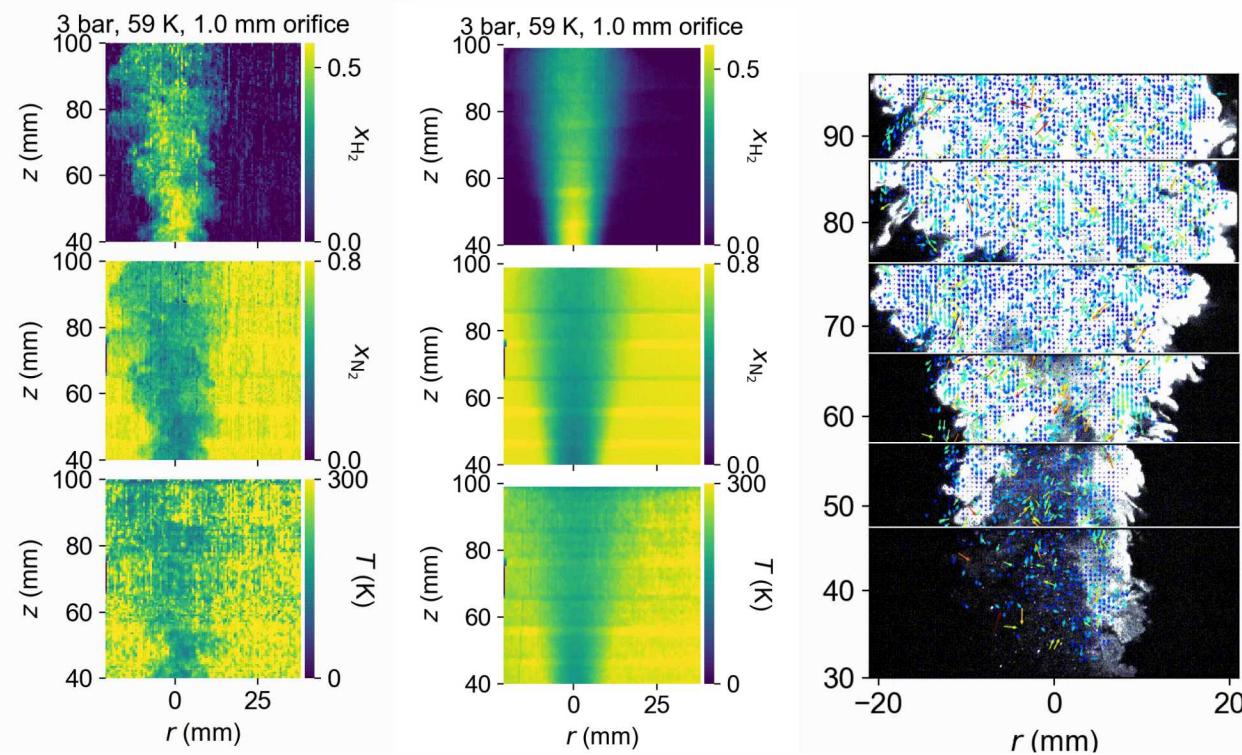
Measuring all independent model parameters:

- ✓ T - temperature
- ✓ x - mole fraction
- ✓ v - velocity
- ✓ B - halfwidth (velocity, concentration, temperature)

Have completed experiments with variations in temperature, pressure and nozzle size

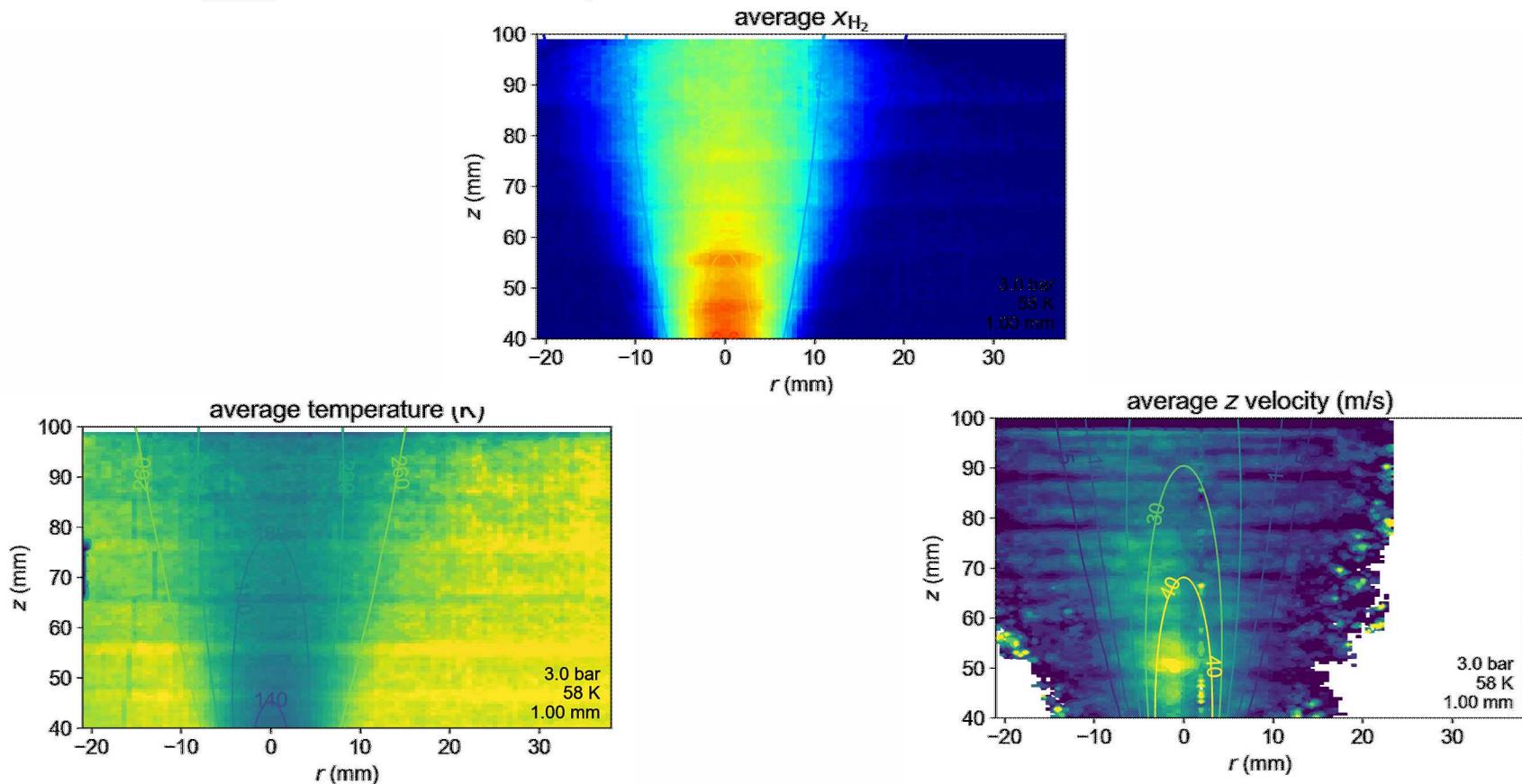
T_{noz} [K]	P_{noz} [bar _{abs}]	d [mm]	T_{throat} [K]	n_{hts}
58	2	1	43.5	4
56	3	1	41.9	4
53	4	1	39.6	4
50	5	1	37.4	5
61	2	1.25	45.7	6
51	2.5	1.25	38.2	2
51	3	1.25	38.2	6
55	3.5	1.25	41.2	3
54	4	1.25	40.4	2
43	4	1	32.1	2
59	3	1	44.2	6
56	3.5	1	41.9	1
80	3	1	60.3	5

With PIV ↓



➤ Measuring dispersion in two-dimensions

The ColdPLUME model shows good agreement with the data



- Model accurately simulates mole fraction, temperature, and velocity, therefore can be used as a predictive tool

How to measure large-scale LH₂ releases (vent stacks after LH₂ delivery, evaporation from LH₂ pool, etc.)

Ideal data for model validation: quantitative concentration, velocity, temperature in 3-dimensions + time at high resolution

- Sensors
 - Can be low cost and easy to implement
 - Require gas to flow over them (placed in flow, or suction, disturbs flow)
 - Point measurements (challenging to get spatial resolution)
 - Can be affected by environmental factors (not specific to H₂)
- Optical
 - Can provide high spatial and temporal resolution
 - Non-intrusive
 - H₂ is a challenging gas to measure optically



Decision: pursue optical techniques

Many optical techniques are not suitable for measuring large-scale LH₂ releases

- Schlieren – cannot distinguish between temperature and concentration caused density variations (not quantitative)
- Fluorescence – no fluorescing species in the flow or species that could be seeded into the flow
- Absorption – no strong absorption features, and complex detector/illumination scheme
- Rayleigh – cannot distinguish between temperature and concentration caused density differences, entrained moisture scatters too much light
- Raman – shown to work in a laboratory setting, enables quantification of temperature and composition in multiple dimensions

➤ Challenge is how to scale-up laboratory Raman setup for larger releases

We are currently evaluating hardware to enable the large-scale diagnostic

- Need large light collection area to capture the small number of photons emitted
 - Reflective optics (large telescope mirror)
 - Refractive optics (Fresnel lens)
- High-powered light source required to excite as many molecules as possible
 - High-power laser with volumetric illumination
 - High-repetition rate laser scanned across the area quickly
 - High-power diodes
- Effective background light suppression is key (both sunlight and illumination source that reflects off of condensed water vapor)
 - Time gating
 - Spectral gating

We are also planning well-controlled large-scale release experiments

Need experiments to characterize:

- Pooling
- Evaporation from LH₂ pools

Planning underway for experiments at Sandia (Albuquerque) facilities:

- Thermal test complex
 - Flame cell
 - Up to 3m diameter pool
 - 18.3 m dia. x 12.2 m high
 - Well characterized conditions for model validation
 - Crosswind test facility
 - Dispersion in controlled crosswind
 - Single-direction flow
 - Well-characterized ambient conditions
- Severe Accident Phenomena/Analysis (Surtsey)
 - 100 m³ pressure vessel with 6 levels of instrumentation ports



Summary:

- **Approach:**
 - Develop and validate scientific models to accurately predict hazards and harm from liquid releases, flames, etc. Generate validation data where it is lacking.
- **Accomplishments:**
 - Measured cryogenic hydrogen dispersion (concentration, temperature, and velocity in two-dimensions) at lab-scale (≈ 1 mm sized orifice)
 - Validated integral model of dispersion
 - Studied ignition distance and radiative heat flux from cryogenic hydrogen flames
- **Future Work:**
 - Develop diagnostic for large scale cryogenic hydrogen dispersion experiments
 - Perform experiments on outdoor, normal LH₂ vent stack releases
 - Perform experiments on well-controlled (i.e. indoor) large LH₂ releases

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