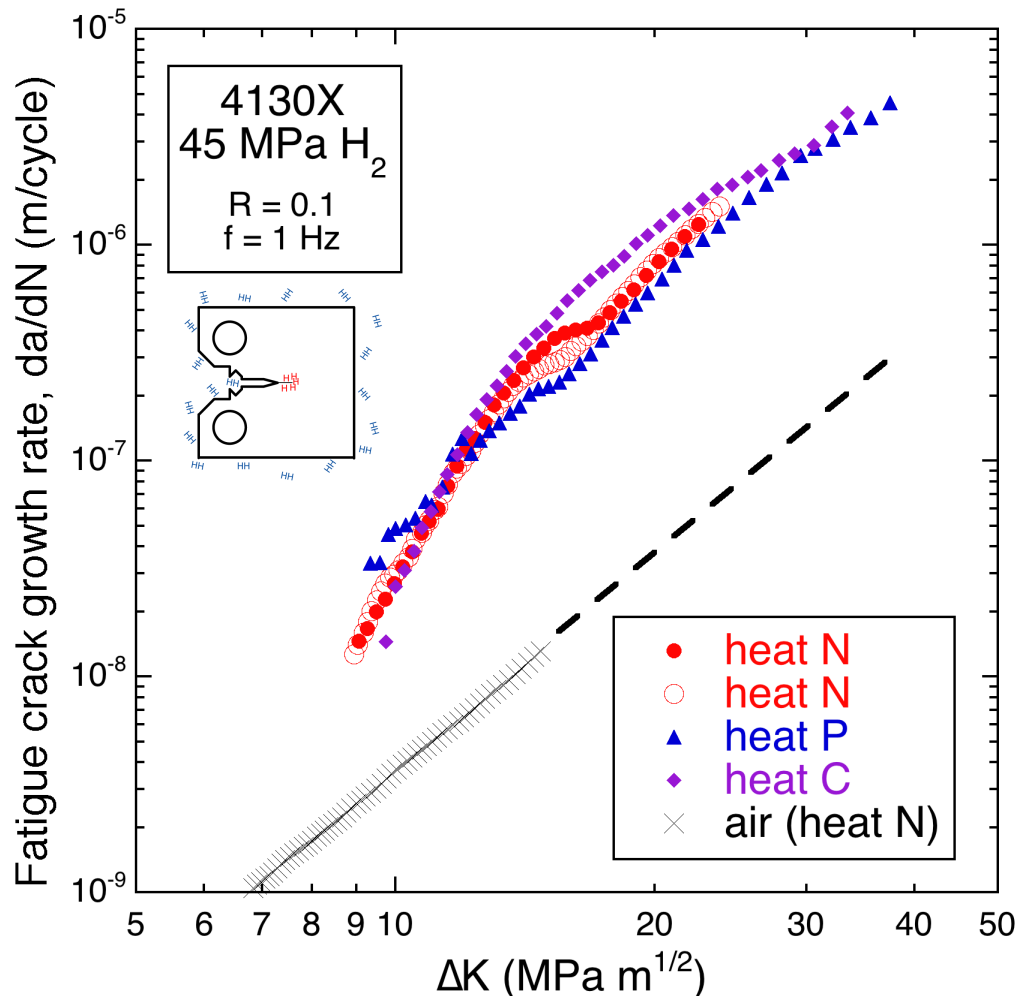


Dispelling Myths about Gaseous Hydrogen Environmental Fracture and Fatigue

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TMS, March 2018
Phoenix AZ

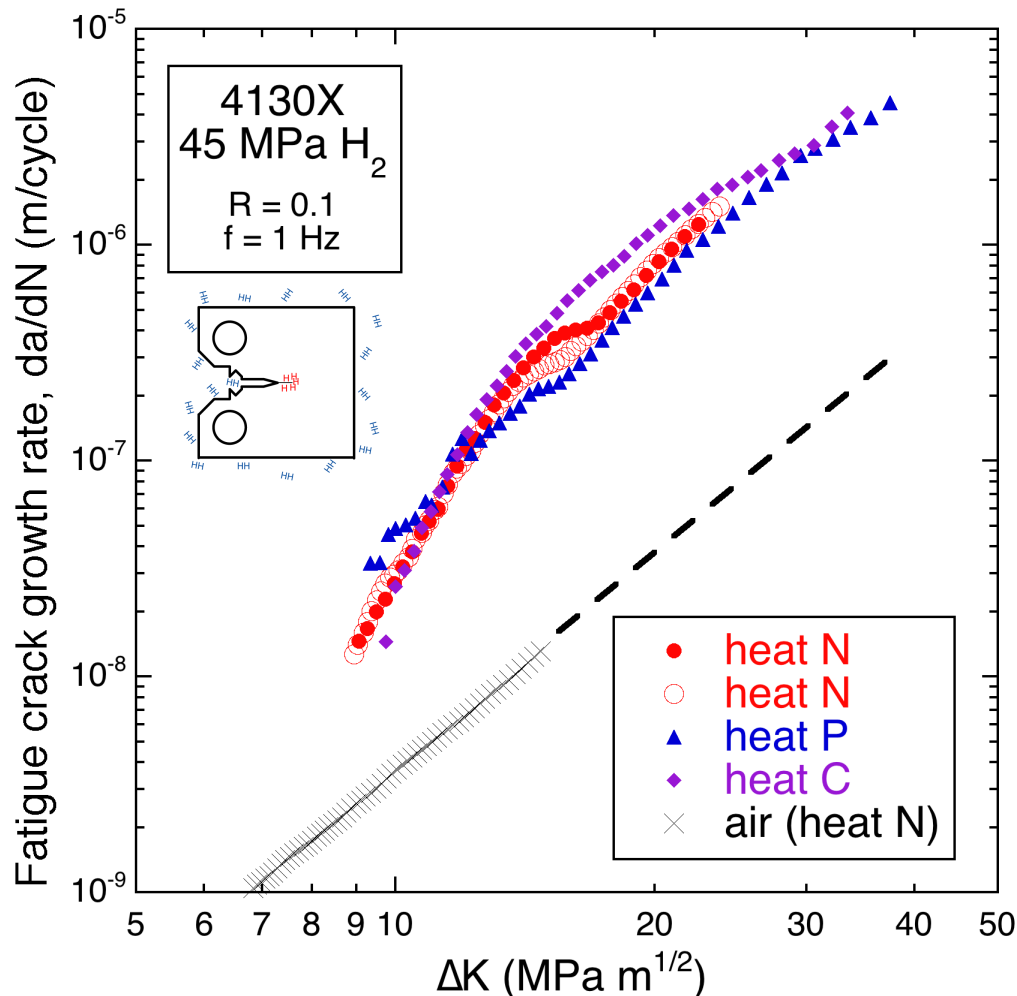
What are the requirements to use a given material in gaseous H₂ service?



- Fatigue crack growth rate is accelerated by 10X in H₂ compared to air
- Is this material safe to use in gaseous hydrogen?
 - Yes – No – Maybe

Laboratory gas cylinders are made of this material

What are the requirements to use a given material in gaseous H₂ service?



Materials requirements depend on the application and the design

- Gas cylinders are generally made from relatively low strength steels
- Wall stresses are relatively low
- Manufacturing defects are well characterized

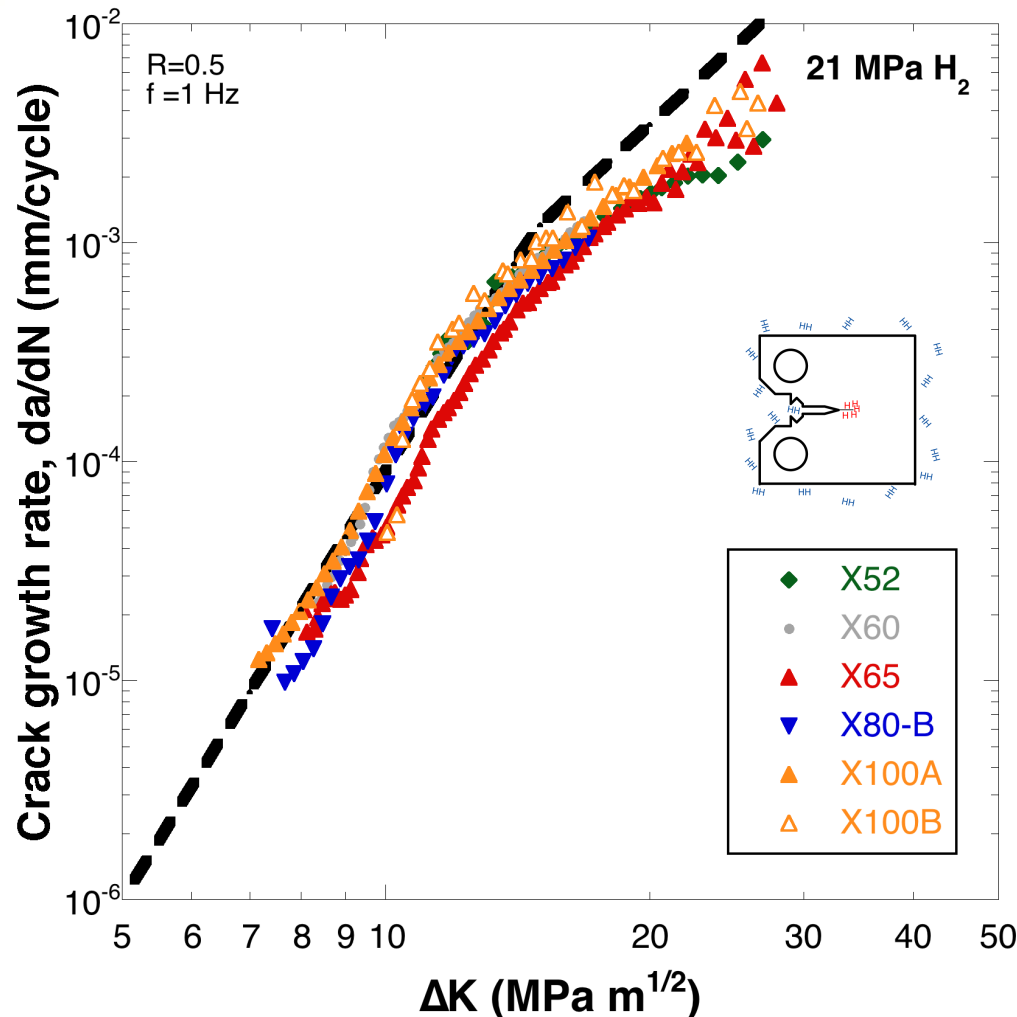


Outline

- **Materials**
 - The effects of H₂ depend on microstructure of ferritic steels
 - Austenitic stainless steels can be immune to H₂
 - Aluminum alloys are immune to H₂
- **Environment**
 - Stainless steels are sensitive to H₂ at low temperature
 - Low pressure H₂ has negligible effect on performance
 - Impurities can mitigate effects of H₂
- **Mechanics**
 - High-strength alloys have low fracture resistance in H₂
 - Suitability for H₂ service can be assessed with tensile tests

The effects of H₂ depend on the microstructure of ferritic steels

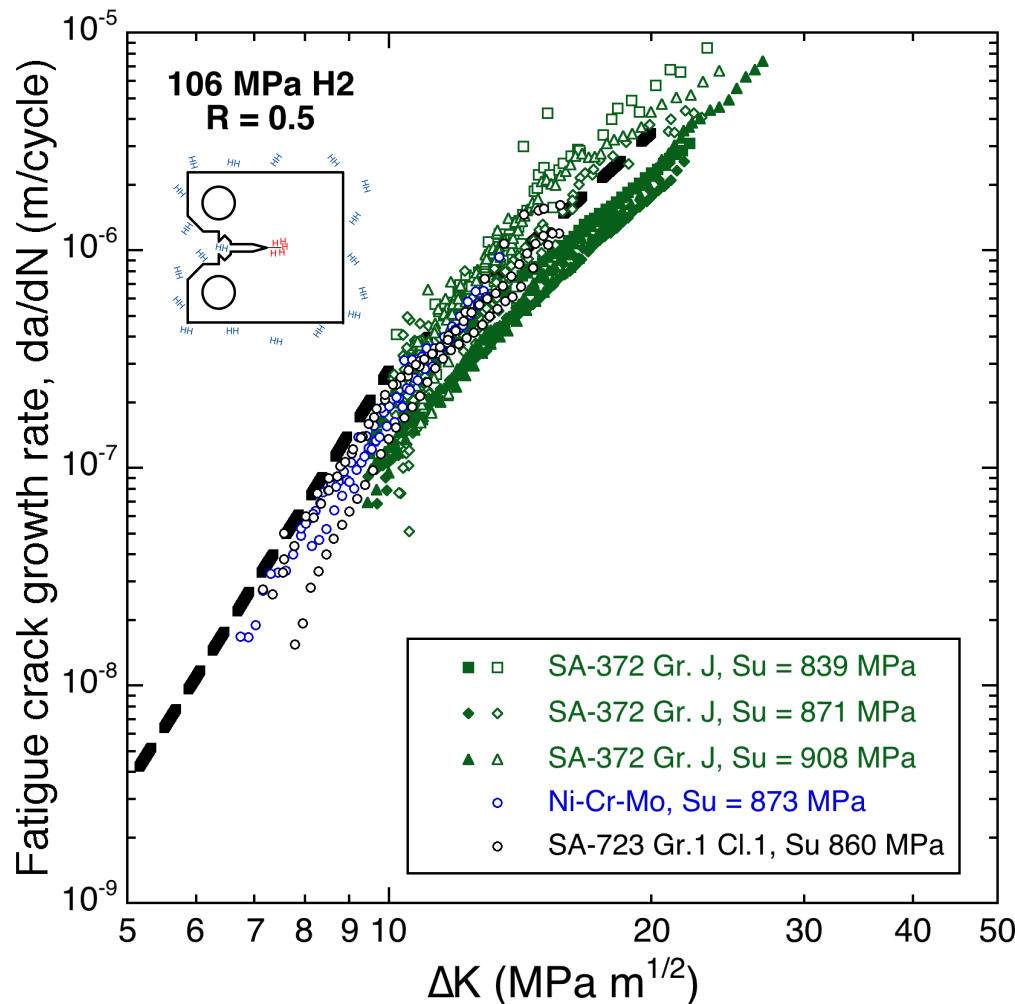
more myth than fact



- Pipeline (carbon) steels show nominally the same fatigue crack growth behavior in gaseous hydrogen
- These steels represent:
 - Wide range of strength
 - Wide range of microstructure

The effects of H₂ depend on the microstructure of ferritic steels

more myth than fact

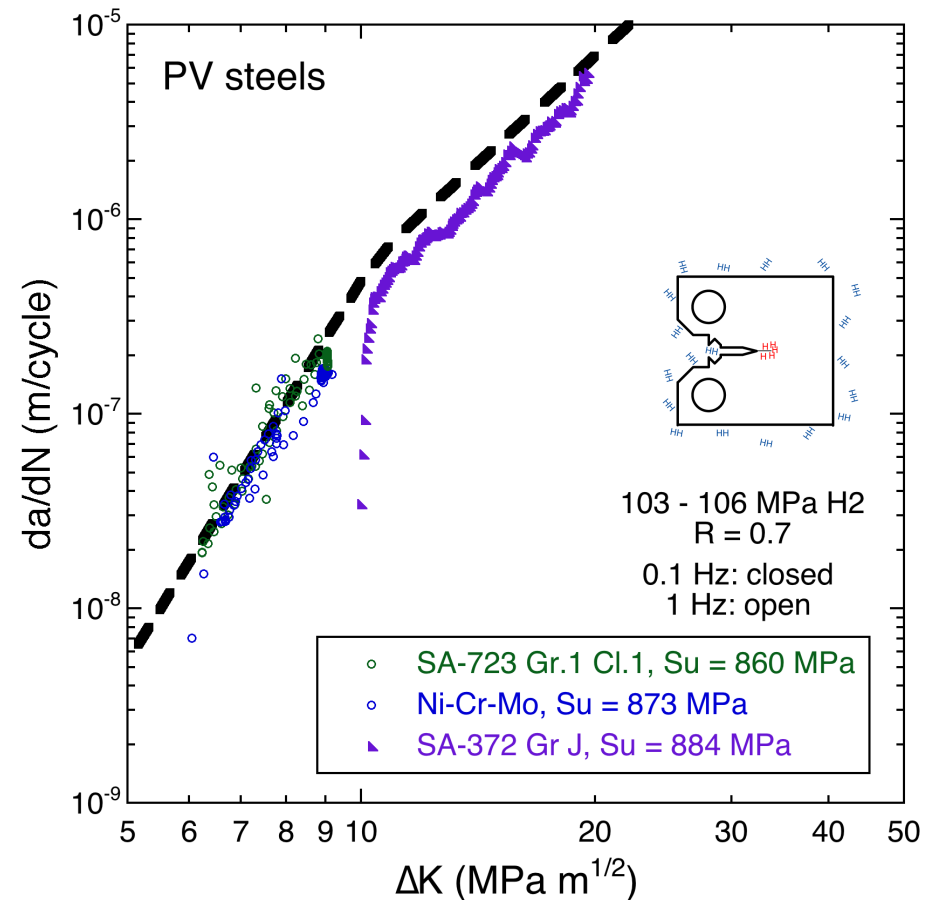
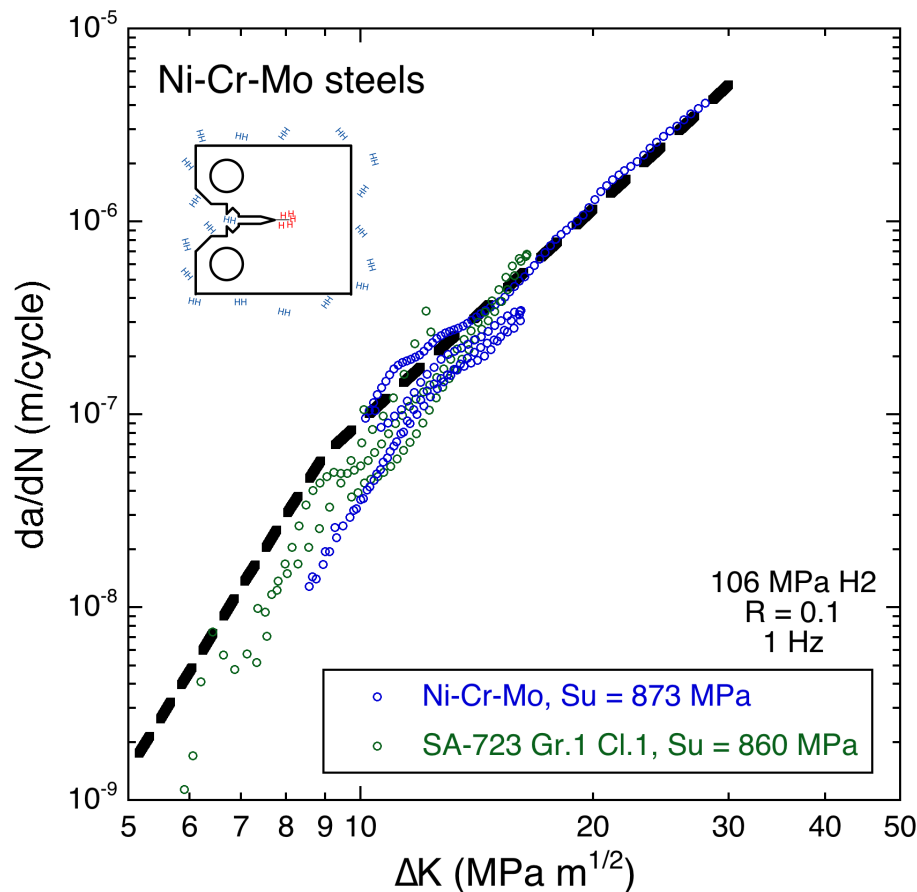


- Pressure vessel steels (low alloy Cr-Mo and Ni-Cr-Mo steels) show same FCGR behavior in gaseous H₂ as pipeline steels
- Dashed line represents closed form expression for both pipeline and pressure vessel steels

$$\frac{da}{dN} = C \left[\frac{1 + C_H R}{1 - R} \right] \Delta K^m \left(\frac{f}{f_0} \right)^{1/2}$$

The effects of H₂ depend on the microstructure of ferritic steels

more myth than fact



Consistent FCGR behavior for wide range of carbon steels

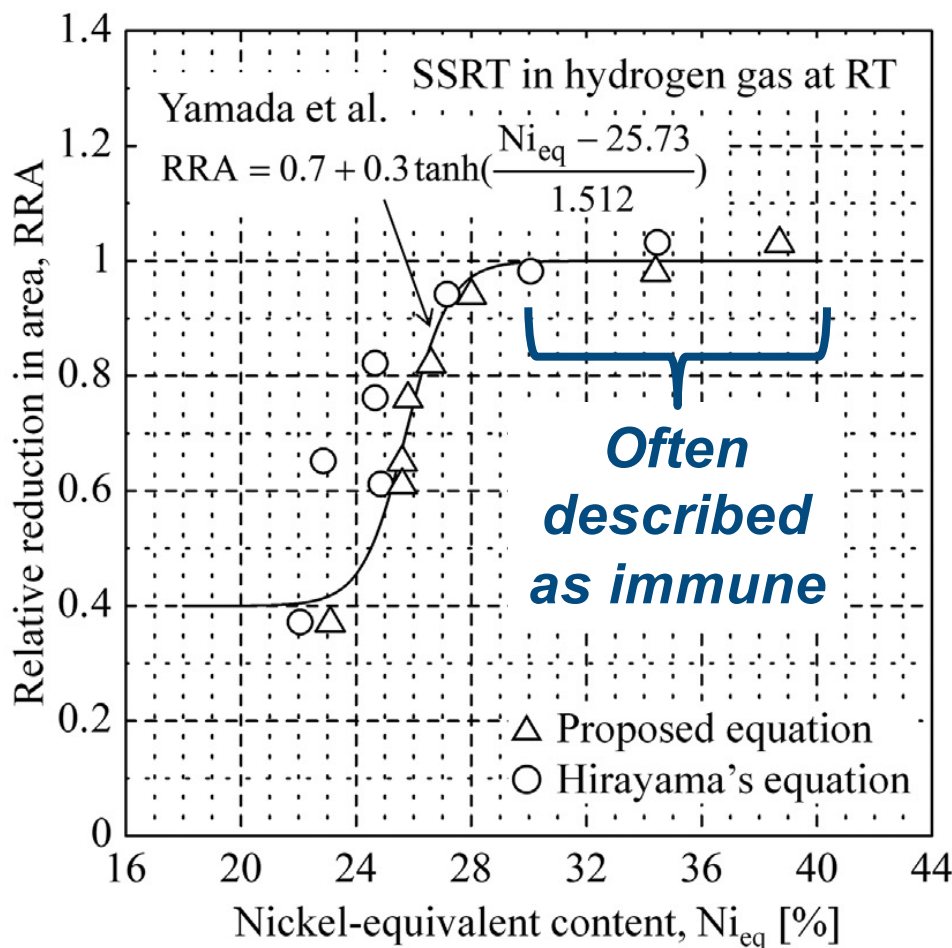
Austenitic stainless steels can be immune to effects of H₂

misleading

- Some alloy compositions show little degradation of tensile ductility (RA) in hydrogen

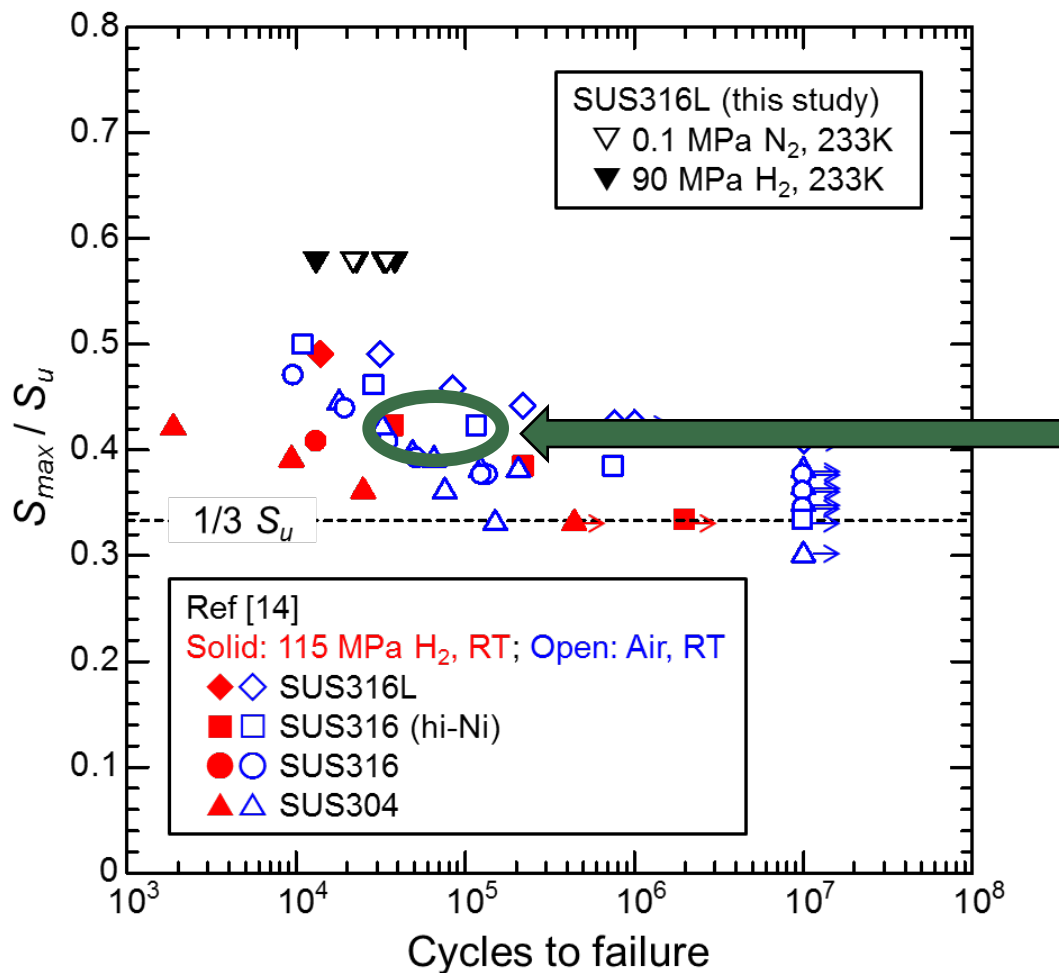
– Suggesting “immunity”

- Represents narrow compositional space *not* representative of common alloy composition
- Trend does *not* necessarily translate to other properties



Austenitic stainless steels can be immune to effects of H₂

misleading



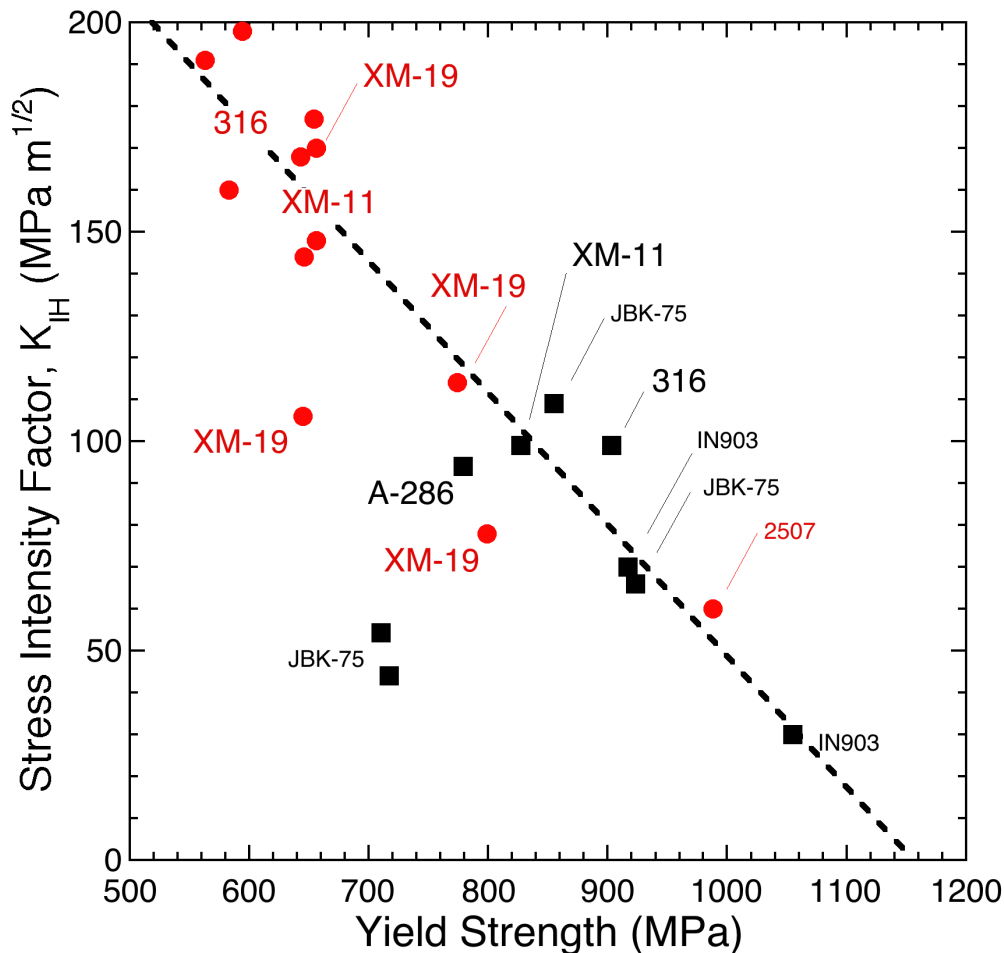
- Fatigue life data generally show similar trends with tensile data
- However, fatigue life data of high nickel alloys show *decrease* in H₂, even with high nickel content

Hydrogen affects low-cycle fatigue even of high-Ni alloys

From: San Marchi et al, ASME PVP-2018 conf.
 Ref 14: Nakamura et al JSME M&M2017 conf.

Austenitic stainless steels can be immune to effects of H₂

misleading

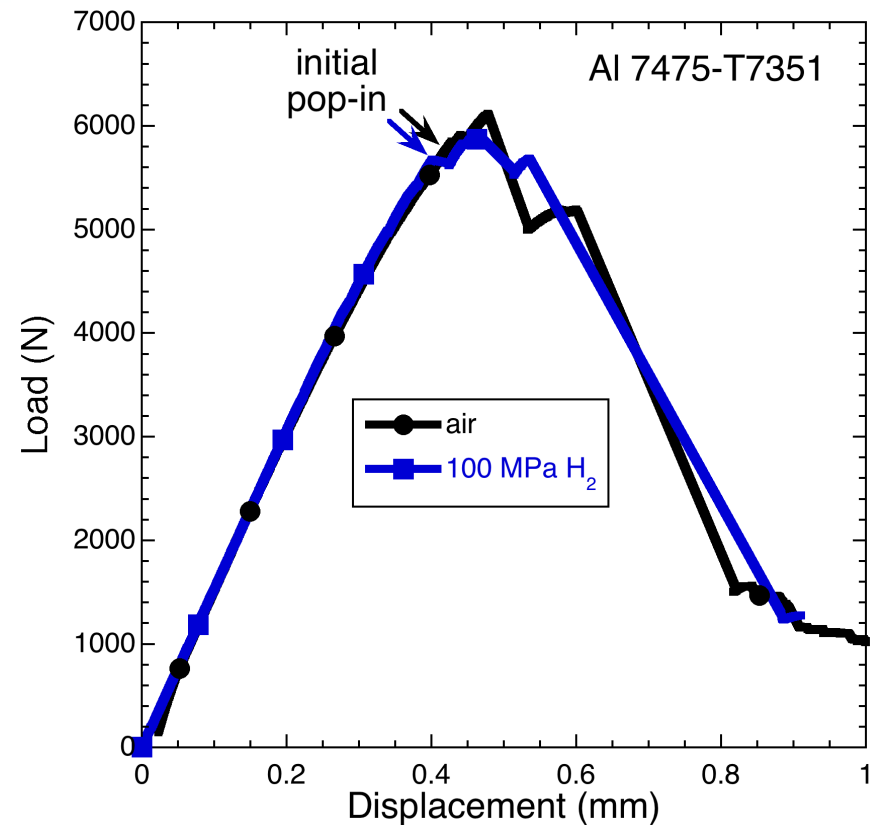
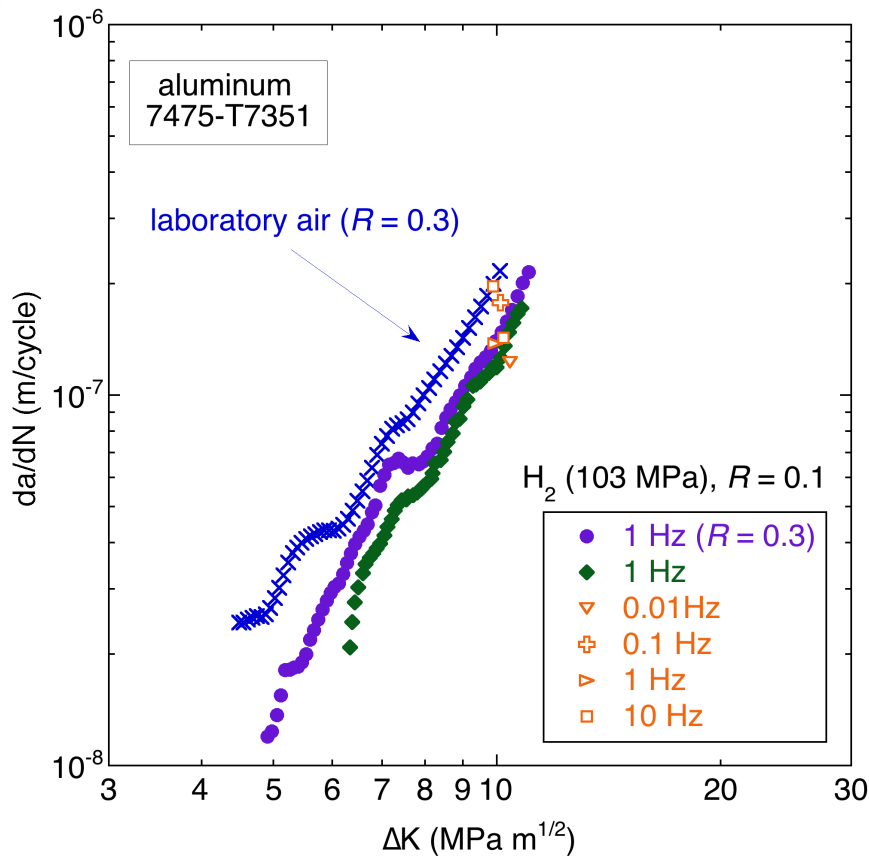


- “Immunity” to H₂ is often attributed to alloy stability
- However, fracture data show
 - SS alloys are not immune to H₂
 - Stable alloys also show strong effect of H₂

Both high-Ni alloys and stable austenitic stainless steels show significant effects of H₂

Aluminum alloys are immune to effects of H₂

seemingly true but misleading

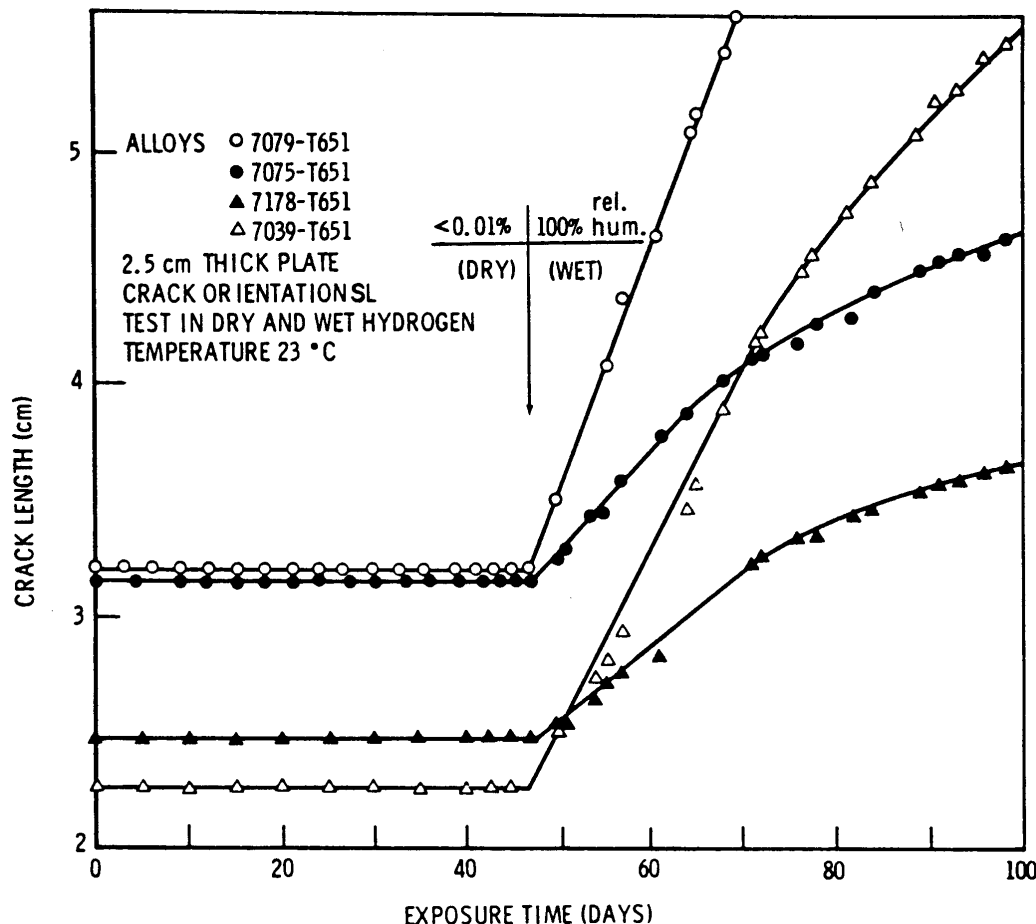


Fatigue crack growth rates in H₂ are less than in air

Fracture resistance is nominally the same in H₂ and in air

Aluminum alloys are immune to effects of H₂

seemingly true but misleading



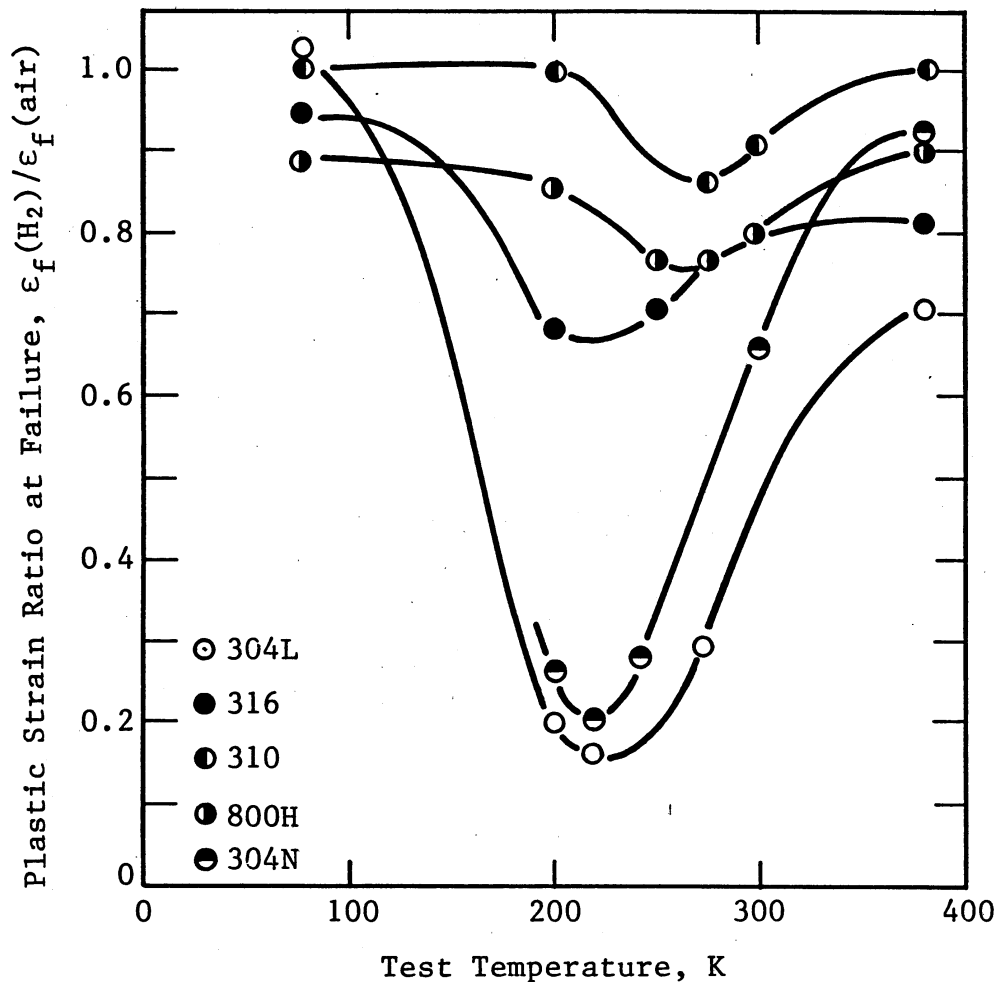
From: Speidel, Hydrogen Embrittlement and Stress Corrosion Cracking, 1984

- Seminal experiments show clear difference of fracture resistance in dry H₂ and wet H₂
- There are no data that show fracture and fatigue of Al alloys are affected by dry H₂

Moisture in H₂ can promote environment-assisted cracking

Austenitic stainless steels are sensitive to H₂ at low temperature

misleading

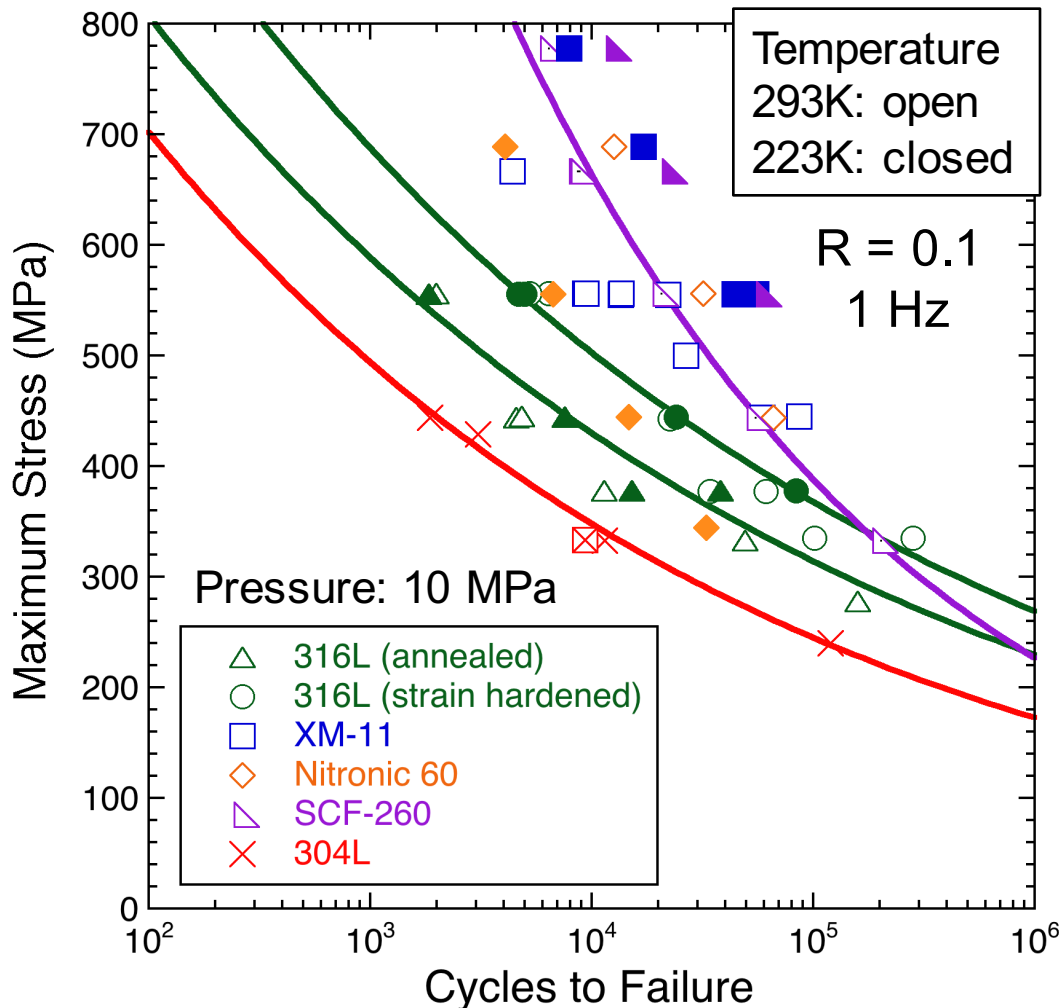


- Tensile ductility can be significantly degraded in hydrogen at low temperature
- However, tensile ductility does not represent relevant failure modes, such as subcritical cracking and fatigue

From: Caskey, Hydrogen Compatibility Handbook for Stainless Steels, 1983

Austenitic stainless steels are sensitive to H₂ at low temperature

misleading



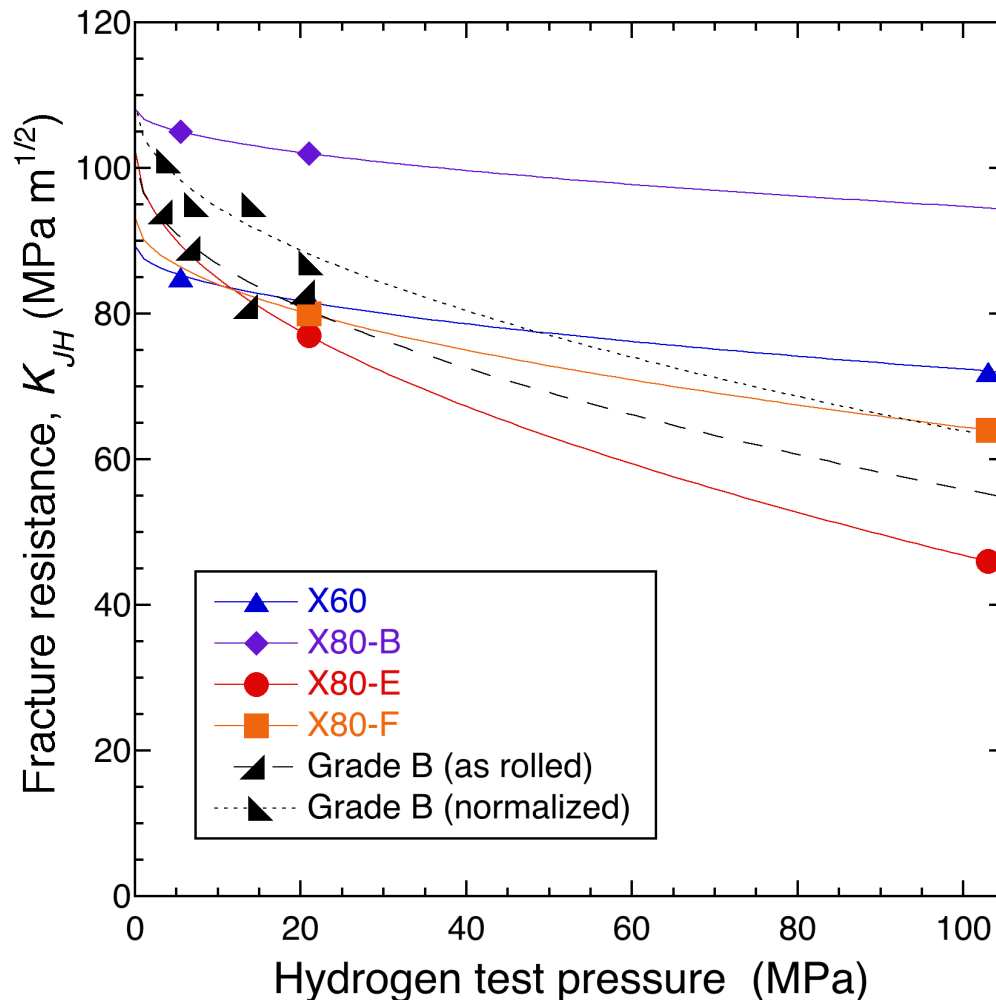
- Fatigue life at low temperature is similar or greater than at RT
- Fracture resistance (not shown) is modestly reduced at low temperature but remains large (>100 MPa m^{1/2})

Fatigue life is not limited by low temperature performance



Low pressure H₂ has negligible effect on performance of steels

myth

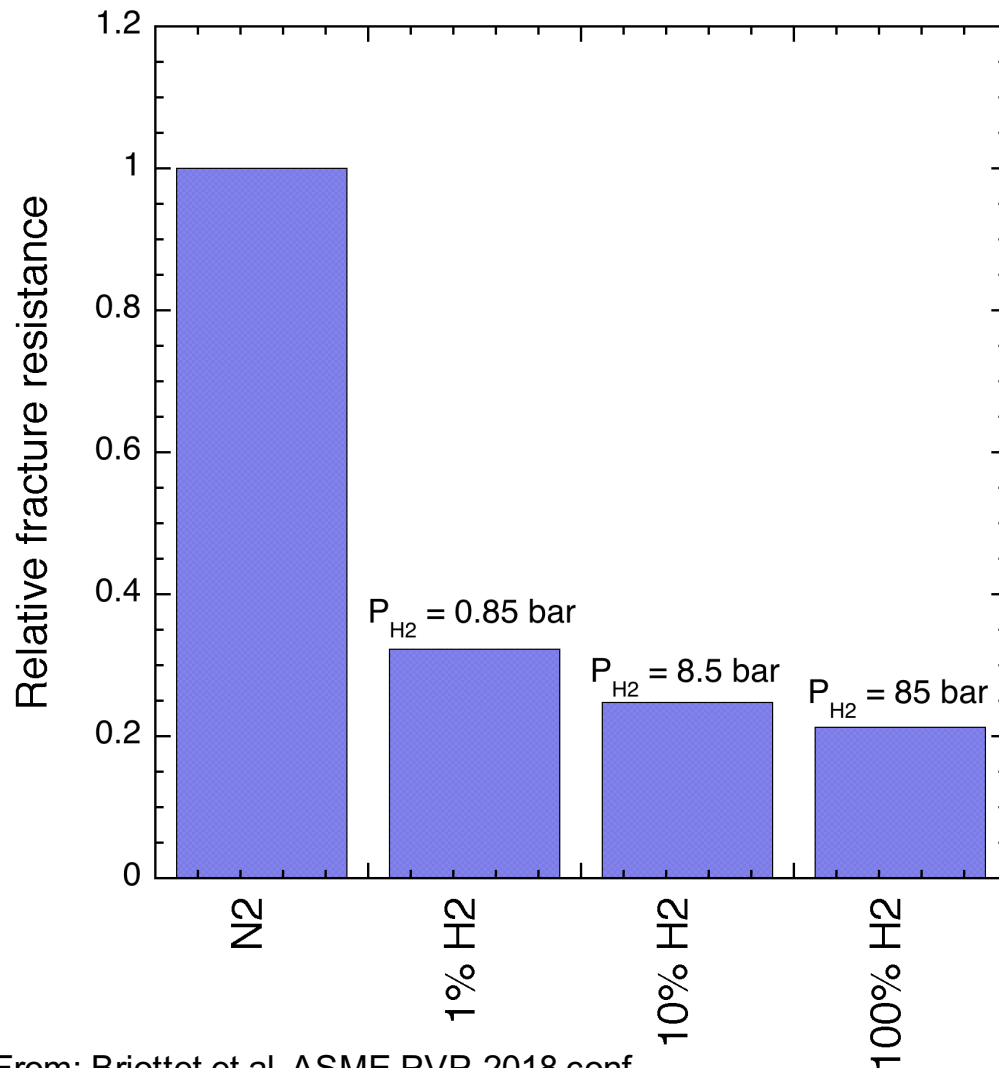


- Data sets that evaluate effect of pressure on fracture are relatively limited
- Available data suggest fracture depends on pressure
- Fracture resistance at low pressure appears to be significantly lower than in air



Low pressure H₂ has negligible effect on performance of steels

myth

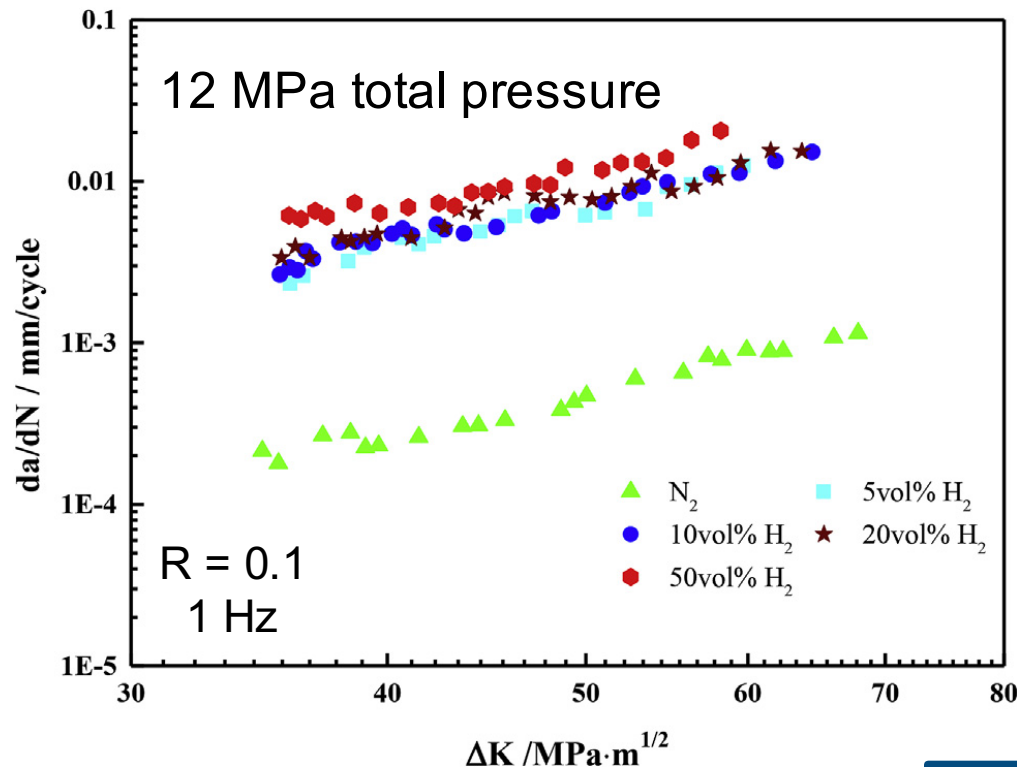


- Measurements in gaseous mixtures of H₂ and N₂ show substantial reduction of fracture resistance for low H₂ partial pressure
- Only modest change in fracture resistance for $P_{H_2} > 1$ bar

<1 bar of H₂ reduces fracture resistance

Low pressure H₂ has negligible effect on performance of steels

myth



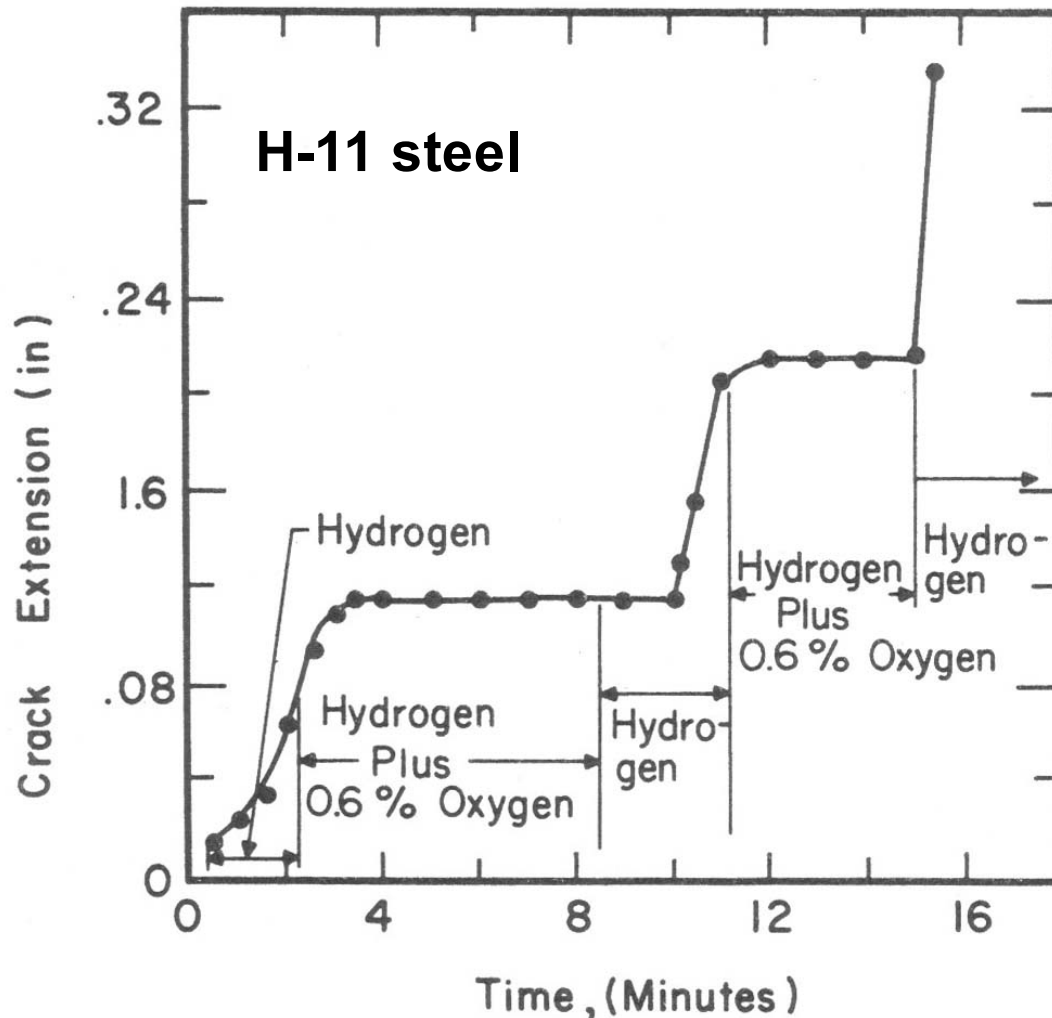
- Measurements in gaseous mixtures of H₂ and N₂ show acceleration of fatigue crack growth rate with 5% H₂
 - But little additional acceleration with higher H₂ content

From: Meng et al, *IJ Hydrogen Energy* **42** (2017) 7404.

Small amounts of hydrogen can have substantial effect on fatigue and fracture

Impurities can mitigate the effects of H₂

true for some impurities



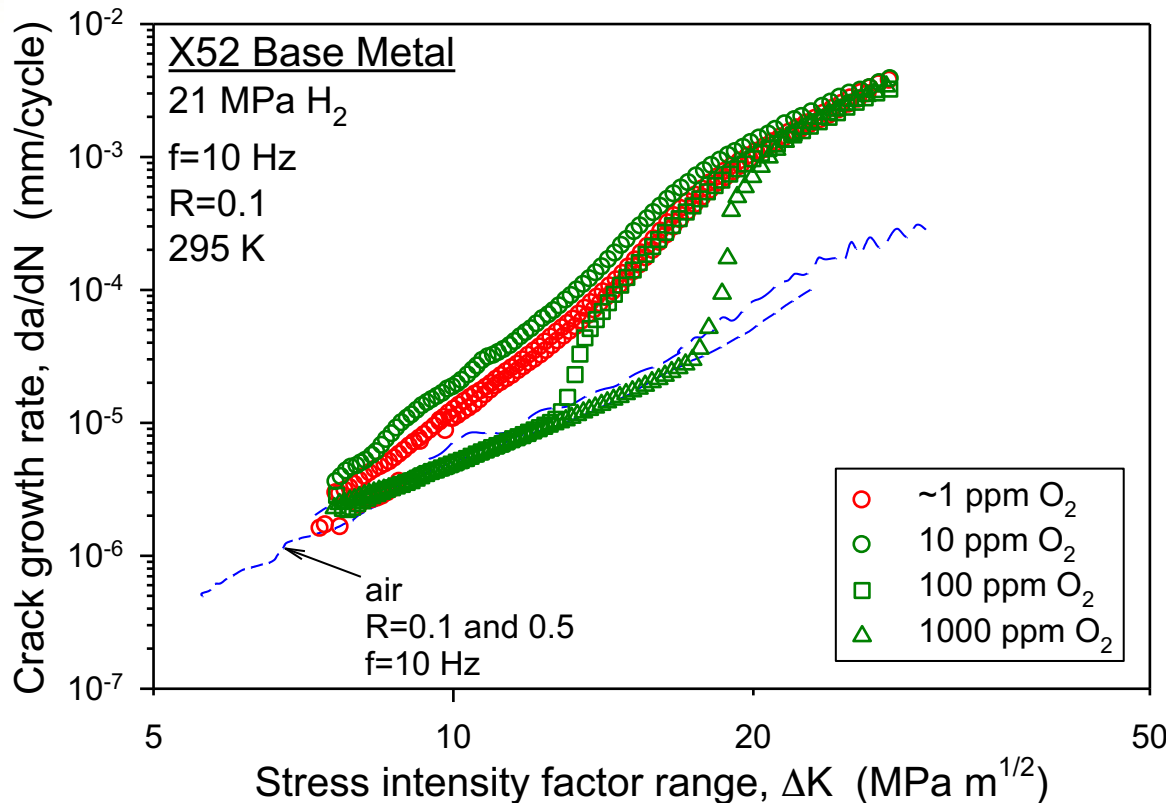
From: H.H. Johnson, Fundamental Aspects of Stress Corrosion Cracking, 1967

- Seminal experiments show clear effect of O₂ on mitigation of H₂-assisted crack growth
- Early experiments, however, did not quantify effects on meaningful properties and relevant alloys

Oxygen can have significant effect on crack extension in H₂

Impurities can mitigate the effects of H₂

true for some impurities



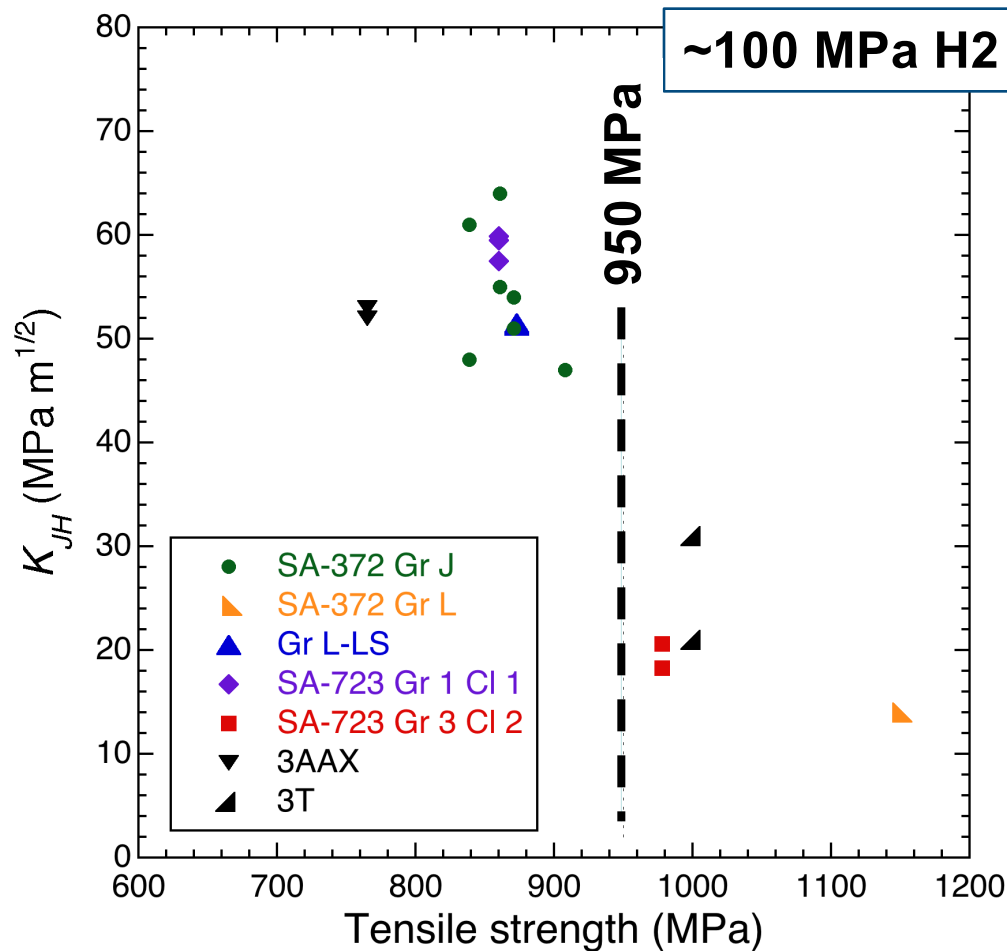
- Oxygen mitigates H₂-accelerated fatigue crack growth rates at low ΔK
- Attributed to oxygen diffusion to new crack surfaces

From: Somerday et al, *Acta Mater* **61** (2013) 6153.

Impurity content in H₂ can have substantial effects on both measurements and in-service performance

High-strength steels have low fracture resistance in gaseous H₂

generally true

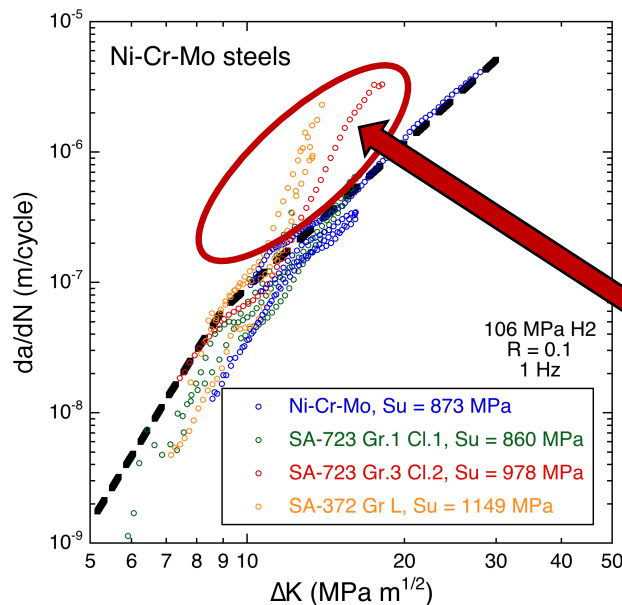


- Pressure vessel steels display low fracture resistance in H₂ when tensile strength >950 MPa
 - Cr-Mo and Ni-Cr-Mo quench and tempered low alloy steels
 - Low fracture resistance will affect fatigue performance at high ΔK

Low-alloy steels with >950 MPa tensile strength are not recommended for H₂ service

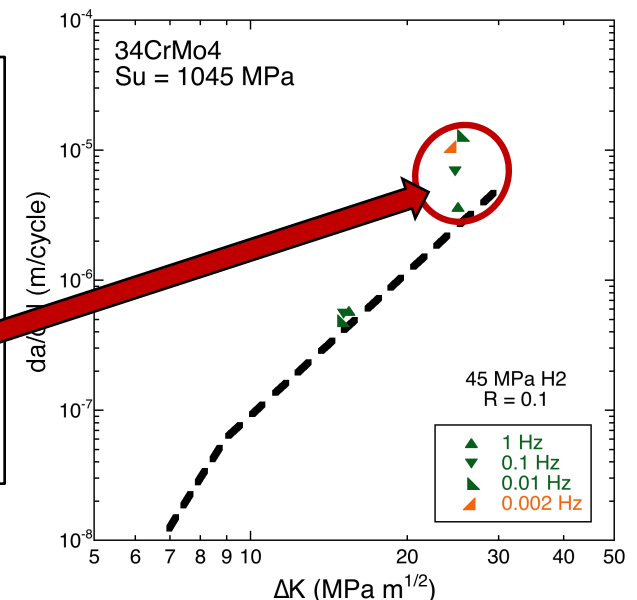
High-strength steels have low fracture resistance in gaseous H_2 *generally true* & *impacts fatigue*

- High-strength steels also show transition to accelerated crack growth (eg, stage III) related to baseline H_2 behavior
 - only observed in tests of high-strength steels: tensile strength > 950 MPa
 - Related to K_{max} in fatigue approaching fracture resistance



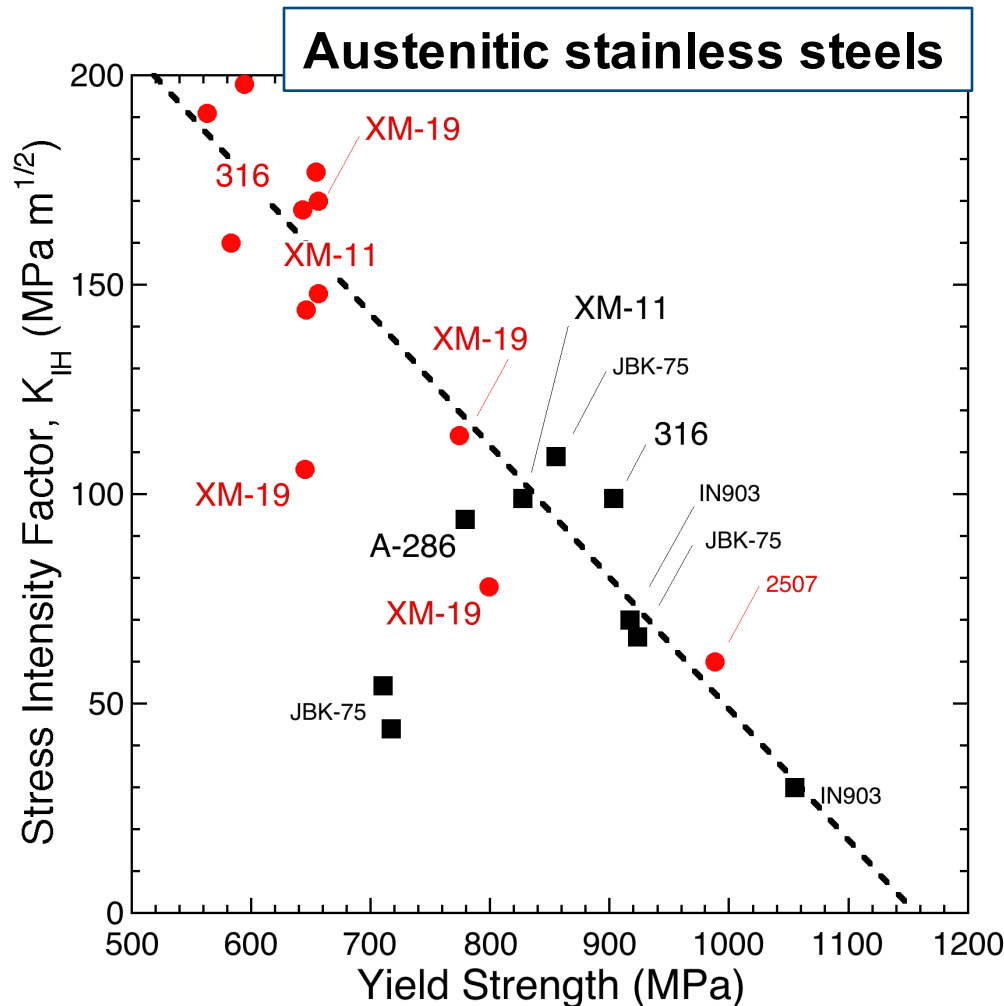
For PV steels with
 $S_u > 950$ MPa

- Accelerated fatigue crack growth rate is observed
- $K_{JH} \sim 20$ MPa $m^{1/2}$



High-strength steels have low fracture resistance in gaseous H₂

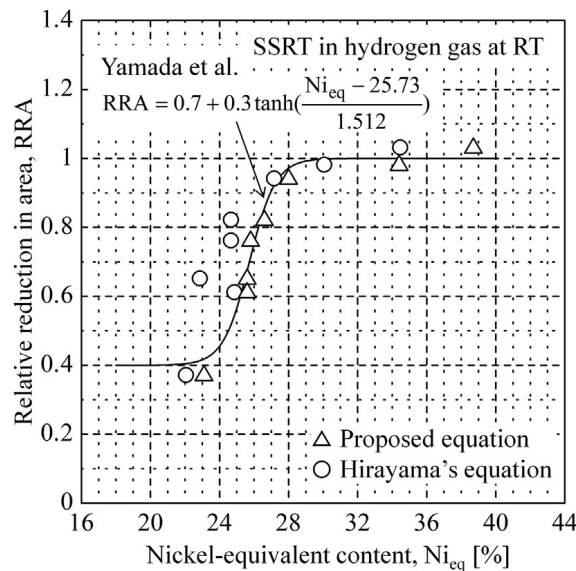
generally true



- For austenitic stainless steels, fracture resistance is a consistent function of yield strength
 - However, data is limited in gaseous H₂
- SS alloys with YS <700 MPa display high fracture resistance in H₂ environments

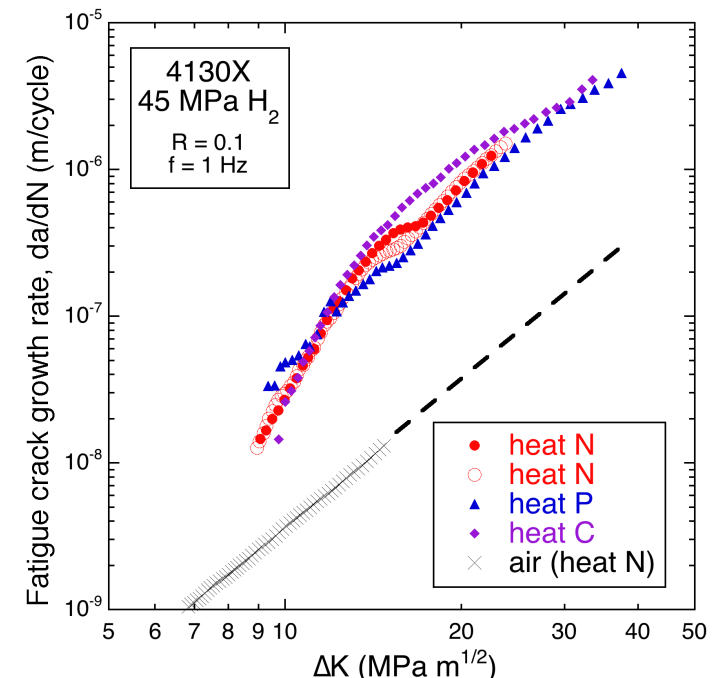
Austenitic stainless steels appear to maintain high fracture resistance (>50 MPa m^{1/2}) in H₂

Suitability for H₂ service can be assessed with slow strain rate tensile (SSRT) tests in H₂ **myth**



- Strength properties are generally not changed in H₂
- Tensile ductility may be reduced H₂, although reduction does not equate to lack of fitness for service

- Suitability for H₂ service depends on materials that express design properties that are sufficient for the required service environment
 - Usually fatigue and fracture properties



The role of impurities and mixed gas environments should be considered carefully

- Oxygen can mitigate effects of H₂ in ferritic steels
 - But sensitive to mechanical and environmental variables
 - Other passivating species can have similar effects
- Water can enhance effects of H₂ in aluminum alloys
 - Threshold content is not known
- Nitrogen is inert, natural gas may not be inert
 - NG has many impurities, which aid mitigation of H₂ effects
 - Pure methane is inert and does not change effects of H₂

Materials performance in hydrogen or mixed hydrogen-containing gas streams will depend sensitively on *materials*, *environmental* and *mechanical* variables

Materials requirements depend on the application and the design



Conclusions

- **Materials**

- H₂ effects are sensitive to microstructure *more myth than fact*
- Austenitic stainless steels can be immune *misleading*
- Aluminum alloys are immune *seemingly true but misleading*

- **Environment**

- Stainless steels are sensitive to H₂ at low temp *misleading*
- Low pressure H₂ has negligible effect on performance *myth*
- Impurities can mitigate effects of H₂ *true for some impurities*

- **Mechanics**

- High-strength alloys have low fracture resistance *generally true*
- Suitability for H₂ service can be assessed with tensile tests *myth*