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Sandwich Layering in Binary Nanoparticle Films and Effect of Size Ratio on Stratification Behavior

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Abstract

Hypothesis: Stratification or self-segregation of multicomponent particle mixtures during drying is an important phenomenon to understand for the development of single-step deposition processes for complex coatings. We hypothesize that varying the ratio of particle Peclet numbers will lead to different types of stratification behavior. Experiments: Binary colloidal films of polystyrene and silica were prepared by evaporative film formation, and stratification of nanoparticles of different size ratio (7.70 – 1.25) was studied using microbeam small-angle X-ray scattering (SAXS). Findings: SAXS spectra showed noticeable variations at different film depths, consistent with stratification. These results are quantified to obtain vertical composition profiles. We observe “sandwich”-type layered structures at different nanoparticle size ratios, which to our knowledge have not been previously observed experimentally or predicted by theory. For example, for films of larger particle size ratios (7.70~4.84), large particles are enriched at the film top and bottom, leading to a large-small-large or “LSL” behavior; while within films of smaller particle size ratio (2.79~1.25), small particles are enriched at the top and bottom of the film (small-large-small or “SLS” structures). The enrichment of particles at the top persists over several hundred particle layers and is not just a single monolayer pinned to the upper surface.

Keywords: film formation; evaporative assembly; drying; stratification; colloidal films

Introduction

Multicomponent films prepared from drying colloidal and nanoparticle suspensions have a wide range of applications. Multifunctional coatings are in demand for various purposes; for example, corrosion-resistant coatings for metal substrates and antibacterial coatings for home furnishings, where only the surface of the film requires special functionality. Other advanced applications of multicomponent films include 3D printing[1], batteries for energy storage[2-3], anti-reflective coatings,[4] electronic devices[5-6], and medical diagnostic tools[7]. Traditionally, multicomponent coatings with vertical structures are achieved by a multi-step deposition technique, which can be costly and time-consuming. Therefore, it is desirable to develop a single-step method where a multifunctional film can be prepared by casting and drying colloidal suspensions on substrates[8].

The film formation process can be characterized by four stages, with three transformations between them[9-11]. The first transformation occurs as solvent evaporates and particles become close-packed[12]. This is followed by particle deformation, where forces arising from dry sintering, capillary effects, or wet sintering cause close-packed spherical particles to deform into a space-filling polyhedrons.[12-14] Finally, interdiffusion between particles forms a continuous film[13-14]. This work focuses on the first stage of drying, where particles distribute themselves during solvent evaporation. It is in this stage where vertical stratification may occur.

During the drying process, Routh and Russel[12-14] showed that particle distributions are affected by the Peclet number, which is defined as the rate of evaporation divided by the rate of diffusion and given as

$$Pe = \frac{6\pi\eta RHE}{kT} \tag{1}$$

where η is the solvent viscosity, R is the particle radius, H is the initial film thickness, E is the rate of evaporation, k is Boltzmann's constant, and T is the temperature. When $Pe << 1$, diffusion is fast enough that particles distribute themselves evenly throughout the film; when $Pe >> 1$, evaporation dominates, and particles will be trapped at the upper surface as drying occurs.

Trueman *et al.*[15] first predicted that, when a binary colloidal system has two types of particles with Pe straddling unity, i.e. $Pe_1 < 1$ and $Pe_2 > 1$ (in this work, we denote Pe_1 and Pe_2 as the Peclet numbers for small and large particles, respectively), particles may distribute themselves vertically during drying. In this case, large particles, which have a larger Pe, are predicted to be enriched at the air-film interface, while small particles will diffuse uniformly within the film, i.e. stratification occurs.[15] Experimental AFM studies using a binary system with polymeric particles have been performed to verify this model[16]. The experimental results were in general agreement with predicted trends, but only qualitatively. Additionally, as AFM studies are only able to probe the top surface of the films, these experiments did not provide insight into vertical concentration profiles within the films.

Other experimental studies have shown the importance of particle interactions to the stratification behavior. For example, Nikiforow et al.[8] studied systems with charged and neutral particles and found that neutral particles were preferentially enriched on the top of films. Routh and co-workers modified their diffusive model to incorporate the effects of particle interaction and determined that attractions between large particles as well as repulsion between small particles, regardless of the interaction between large and small

particles, can facilitate the stratification along with hydrodynamic effects[17]. Cheng and Grest[18] performed large-scale molecular dynamics simulations on dispersions containing large polymeric particles and small nanoparticles, and they demonstrated that stronger interactions between large polymeric particles and the solvent could result in a nanoparticle-enriched interface, white stronger interaction between small nanoparticles and the solvent could result in a polymer-enriched interface, under conditions of fast evaporation ($Pe >> 1$ for both particles).

Francis et al.[19] first reported stratification phenomenon in drying binary mixtures of polymeric and ceramic particles. However, they focused on systems prepared at relatively high temperature, i.e. under fast evaporation. In addition, relatively large particle sizes were used, resulting in Pe_1 of 120 and Pe_2 of 460. Therefore, it is likely that the stratification behavior they observe is due to a different mechanism than that proposed by the diffusive model of Trueman et al. Francis and co-workers created films with small particles enriched at the interface and large particles enriched within the body of the film, which can be described as inverted stratification. They proposed, as a possible mechanism for the inverted stratification, that the existence of some very small ceramic nanoparticles might be able to diffuse into the interstitial spaces among latex particles. This might also be attributed to the fact that hydrodynamic effects become less significant when both particles have very high Peclet numbers ($Pe >> 1$).

Many recent theoretical studies predict an inverted stratification structure, where small particles are enriched at the top surface and large particles were pushed to the bottom, though most of these systems used relatively fast evaporation rates, i.e. both particles have Peclet

numbers >> 1. Fortini et al.[20] performed simulations of drying binary colloidal mixtures with Pe_1 of 75 and Pe_2 of 525. Complementary experiments were also performed with Pe_1 of 14 and Pe_2 of 100. Both simulations and experiments found that the air-film interface was enriched with small particles, and the body of film was enriched with large particles. They proposed that an osmotic pressure gradient would push large particles downward to the bottom of the film, which might be account for the inverted stratification. Later Fortini and Sear extended their conclusions to ternary mixtures and polydisperse colloidal systems.[21] Panagiotopoulos and co-workers[22] performed implicit-solvent molecular dynamic simulations with similar systems and conditions as Fortini et al. Inverted stratification was also observed in their results. They proposed that the difference in chemical potential gradients of particles outweighs the effects of the counteraction of friction coefficient on migration velocities of particles. Moreover, their results suggested that this inverted stratification not only occurs under fast evaporation but also moderate evaporation, i.e. $Pe_1 \approx 1$ and $Pe_2 >> 1$, although to a lesser extent. They found that large particle size ratios sped up and enhanced the establishment of the stratified layers. Additional work of Panagiotopoulos has focused on the stratification of drying polymer-polymer and colloid-polymer binary mixtures, and the results showed good agreement with their previous work on binary colloidal mixtures.[23] Zhou et al.[24] also predicted inverted stratification in binary mixtures. Their simulation results rely on an asymmetric cross interaction between large and small particles to obtain this result. Interestingly, they found that the condition for this inverted stratification to occur only depends on the particle size ratio, *Pe* for the small particles, and the volume fraction of small particles. The condition for inverted stratification to occur is given as,

$$\alpha^2(1+Pe_1)\Phi_{01} > 1 \quad (2)$$

where α is the size ratio between large and small particle and Φ_{01} is the initial volume fraction for small particles. Recently, Sear and Warren[25] used the Asakura-Oosawa polymer model to suggest a mechanism of the inverted stratification. Their model took the effect of solvent backflow into account and should not significantly overestimate the effects of diffusiophoretic drift, a concern with some earlier studies.[20-24] Finally, very recent studies of Sear[26] and by Cheng and co-workers[27] include an explicit solvent in molecular dynamics simulations. Their results show both small-on-top and large-on-top regimes of stratification but suggest that the implicit solvent approach used by Zhou et al. overestimates small-on-top aggregation. Their results, together with the work of Sear and Warren,[25] lead to a condition for small-on-top stratification of $Pe_s\phi_s \gtrsim C$, where C is found to be 0.64 by Sear,[26] ~ 2 by Sear and Warren,[25] and ~ 1 by Cheng and co-workers.[27]

Concurrent with these advanced theory and simulations, recent experimental studies have demonstrated different types of stratification behavior. Previous experimental studies in our group[28] utilizing AFM have shown that both the particle size ratio and processing factors are important for the distribution of particles in drying colloidal mixtures. Humidity, for example, can affect the evaporation rate, and thus the Peclet numbers. In one case, good humidity control allowed particles to have Peclet numbers straddling unity and to have the largest number of large particles migrate to the top surface, which was in general agreement with Routh's predictions. Results also suggested that, while Pe_1 and Pe_2 were set to straddle unity, a larger particle size ratio would decrease the extent of stratification; i.e., result in more small particles on top. We proposed that large particle size ratios might induce depletion attractions,

causing aggregation of large particles which then migrate to the bottom of the film. Similar mechanisms have been discussed by others[16, 19]. However, AFM characterization only probes the surface of the film over a small area, making it difficult to draw quantitative conclusions. Similarly, Makepeace et al.[29] recently used AFM to perform surface characterization and study the effects of particle size ratio and concentration of small particles on stratification. They found qualitative agreement in experiments and simulations with the theory proposed by Zhou et al.[24] Cryo-SEM was applied to obtain cross-sectional images as well. Nevertheless, their experiments presented the same limitation as our previous AFM studies, and their implicit-solvent simulations still overestimated the degree of stratification.

In this work, we prepared nanoparticle films with polystyrene and silica particles of different sizes and studied how stratification was affected by different nanoparticle size ratios. As discussed below, our results show behavior beyond the “small-on-top” and “large-on-top” structures predicted by theory, yielding small-large-small and large-small-large layering behavior. While Francis et al. observed particles can enrich at both interfaces and have relatively small amount at the center of one-component films with control of evaporation, sedimentation and diffusion forces in computation and experiments[30], “sandwich” structures with one type of particles enrich the top and bottom and another enrich the center of films have not been seen or predicted before. Probing the depth-resolved particle composition is enabled through the use of a novel microbeam small-angle X-ray scattering (SAXS) approach that we have described in a previous paper,[31] which can give more quantitative results of the particle volume fractions throughout the film thickness. These experiments involve conducting a vertical scan throughout the film and require a narrowly-

focused beam with high intensity. Furthermore, SAXS is an attractive approach for studying drying of colloidal mixtures, since it provides good vertical resolution with corresponding quantification of nanoscale packing, and is relatively non-destructive. Although beam damage may occur with polymeric samples if they are exposed to synchrotron radiation for long times, in our experiments we find that modest acquisition times at any given vertical position in the film do not generate beam damage in dried colloidal films. Dingenouts and Ballauff used SAXS techniques to explore structure changes of latex films upon drying and their results indicated a liquid-like short range order instead of crystallinity within the dried films.[32-33] Goehring et al. performed dip-coating to make colloidal films and conducted SAXS experiments to follow variation of scattering patterns at different times. Similarly, they found the lack of long range order in the final films, while short range order presented for a small period at the liquid-solid transition. Additionally, their results could follow volume fraction variations and capture structure changes in transitions between stages during drying. The results also indicated no significant far-field flow was present to affect distribution of the number of particles at different positions horizontally.[34] We believe our work can be used to guide future work on modelling the dynamics of multicomponent film formation by evaporative assembly, and can be inspiring for industrial applications to better design systems for making multifunctional films with desired compositions.

Experiments

Materials

In the experiments, polystyrene (PS) particles in aqueous suspensions were obtained from Invitrogen, and silica nanoparticles (Ludox) were obtained from Sigma-Aldrich and Grace.

The specification of samples and measured particle diameters are listed in Table 1. The particle radii were determined by performing at least three 5-minute measurements, at 25 °C, pH = 9 (approximate to silica suspension storage pH value) with a NanoBrook Omni dynamic light scattering instrument (Brookhaven Instruments), which has a wavelength of 640 nm. In the resulting size distributions, minor scattering peaks potentially from aggregates, precursors and dust were omitted to get accurate particle sizes. Sample names refer to the material type and the nominal radii in nm. With the sizes and densities, the Stokes settling velocity of particles U_0 can be calculated as,

$$U_0 = \frac{2R^2 \Delta\rho g}{9\mu} \tag{3}$$

where R is the radii of particle, $\Delta\rho$ is the difference between the density of particle and of solvent, g is the gravitational constant and μ is the solvent viscosity. For all the particles used in this paper, the U_0 are small ($\leq 1.63 \times 10^{-9}\ m/s$) and can be neglected on the time scales of the drying process. If large aggregates formed during the drying process, we would expect these to have a faster sedimentation rate, ,which might affect the final particle However, the absence of aggregation was confirmed in SAXS spectra, as no dramatic increase of intensity at low q was observed, which would be expected if there were any large aggregates formed.

Content	**Name**	**Radius (nm)**	**Peclet Number**
Polystyrene	PS-48	47.93	10.35
Polystyrene	PS-21	21.42	4.62
Silica	Silica-17	17.16	3.71
Silica	Silica-10	9.91	2.14

Silica	Silica-8	8.41	1.82
Silica	Silica-6	6.23	1.34

Table 1. The specification of samples, particle radius and their Peclet numbers.

Sample Preparation

To prepare colloidal films, polystyrene and silica suspensions were first diluted to 3% v/v with distilled water at pH = 9. To make pure polystyrene or pure silica films, the diluted suspensions were cast and dried; to make binary mixture films, equal volumes of diluted polystyrene and silica were mixed, and suspensions of 1.5% v/v of polystyrene and 1.5% v/v of silica were cast and dried. Each suspension was dried on a deep-well projection slide from Diatec AS, which has a diameter of 23.5 mm and a depth of 1.5 mm. The suspensions were then placed into an environmental chamber (Caron 6010). The temperature was set at 40 °C, which is below the glass transition temperature of PS, and the humidity was set at 60%. The films were dried overnight. An illustration of the projection slides, pictures of dried films and a similar experimental process was available in a previous paper.[28]

The Peclet numbers can be calculated for this drying condition, using a solvent viscosity of 6.5143×10^{-4} Pa·s and density of 0.9922 g/cm^{-3} at 40 °C, an initial film thickness was 1.5 mm, and a rate of evaporation of 0.0786 g/h. The rate of evaporation was determined by averaging three measurements of the rate of evaporation of pure water in the same slide wells, and can be converted into the rate that the solvent front moves downward, which was 5.0734×10^{-8} m/s. The Peclet numbers for the particles are listed in Table 1.

Small Angle X-ray Scattering (SAXS)

SAXS experiments were performed at the Complex Materials Scattering (CMS, 11-BM)

beamline at the National Synchrotron Light Source II (NSLS-II), Brookhaven National Laboratory (BNL). In the experiment, an x-ray micro-beam with a full-width half-maximum (vertical direction) of 26.6 μm was generated using a two-slit system. Samples were measured under vacuum to reduce background. Scattering intensity data were collected on a Dectris Pilatus 2M detector, covering a wave vector range of 0.003 – 5 $Å^{-1}$. SAXS data were acquired across the film at vertical spacings of 10 μm to extract volume fractions at each height of the film. Scattering intensity of each height was collected for films made with pure polystyrene, pure silica and the binary mixtures. Each scan took 30 s or 120 s to achieve good resolution.

For each vertical scan, some images at the very beginning or end of the scan exhibit low signal, as the X-ray beam is not fully intersecting the sample; these data were excluded from analysis. Scattering intensity was normalized by acquisition time. Background subtraction was not required due to the in-vacuum measurements (which eliminates scattering from windows and air) and the low detector background. In the plots presented below, film depths are scaled and normalized on the total film thickness, so that the scaled depth ranges from 0 to 1, where 1 represents the top of the film (vacuum interface) and 0 represents the bottom (substrate interface).

Results

By performing SAXS on mixed PS/silica films, we obtained scattering patterns at different depths. In Fig. 1, data from representative depths are plotted. Fig. 1 also lists the total film thickness for each sample. Scattered intensity is plotted in arbitrary units; in the main plots, intensity has been shifted

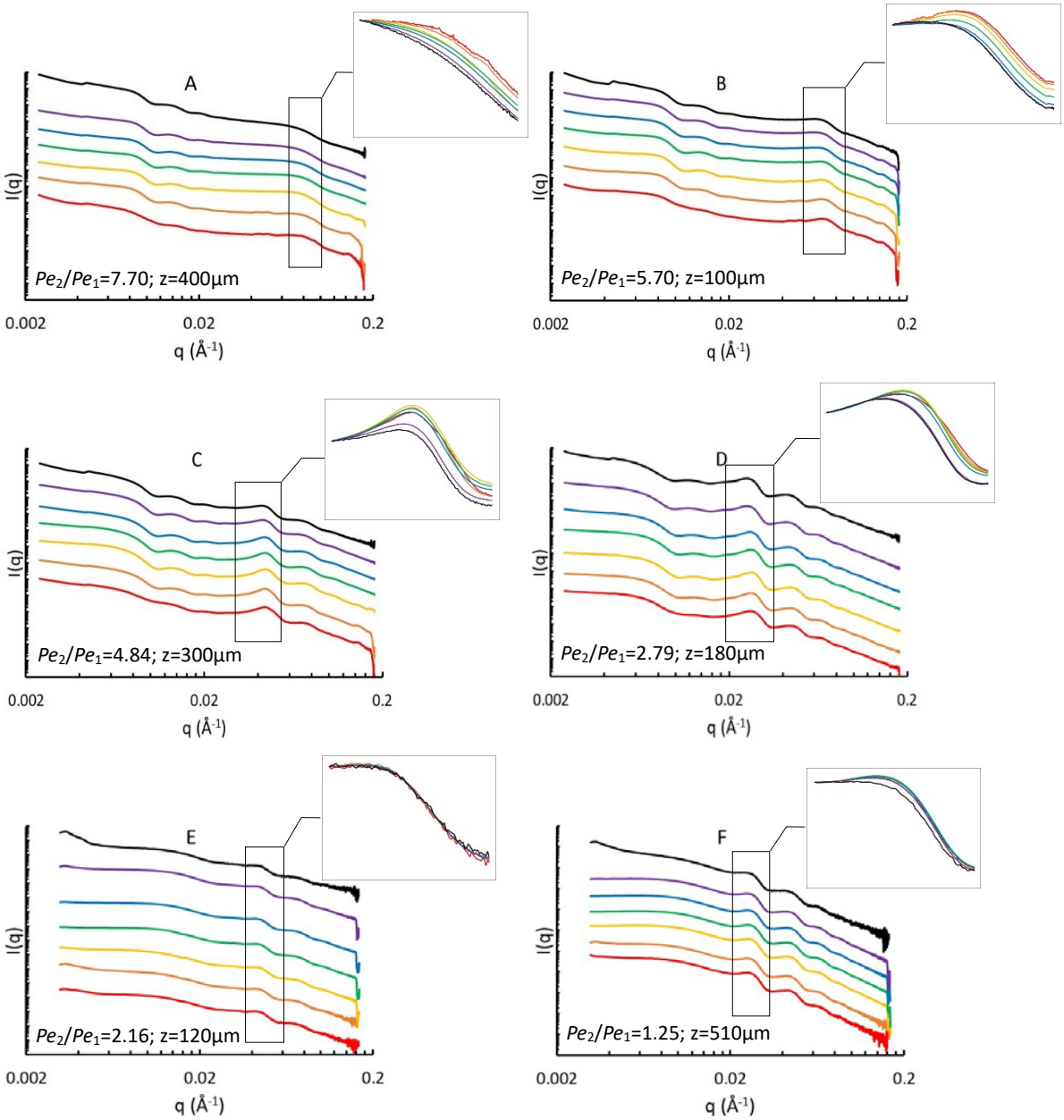

Fig. 1 Small-angle X-ray scattering at selected vertical positions of mixture films made with (A) PS-48 and Silica-6 (Pe_2/Pe_1=7.70); (B) PS-48 and Silica-8 (Pe_2/Pe_1=5.70); (C) PS-48 and Silica-10 (Pe_2/Pe_1=4.84); (D) PS-48 and Silica-17 (Pe_2/Pe_1=2.79); (E) PS-21 and Silica-10 (Pe_2/Pe_1=2.16); and (F) PS-21 and Silica-17 (Pe_2/Pe_1=1.25). Intensity and scattering vector are plotted in log-log scales for both main and inset graphs. Intensities are in arbitrary units and curves are shifted for clarity. From bottom to top, curves represent selected scaled film depths from 0.0 to 1.0. Comparisons of peak shapes at q ranges assigned as silica primary peaks are given.

vertically to avoid overlap; in the insets, intensities are again shifted vertically to highlight any observed differences in peak intensity and position at different depths in the films. Based on the sizes of silica and PS particles and scattering from pure silica and PS films

(Supplementary Information), peaks corresponding to silica and PS in the mixed films were identified. Scattered intensities for different films in the *q* ranges assigned for primary peaks of silica particles are compared in the insets of Fig. 1, since silica scatters X-rays more strongly than polystyrene. Variations at different depths are clearly seen. In Fig. 1A and 1B, obvious differences of silica scattering peaks can be seen for large Peclet number ratios, i.e. nanoparticle size ratios of 7.70 and 5.70. Silica peak heights decreased along the normalized film depth 0.0 to 1.0. In Fig. 1C and 1D, for intermediate Peclet number ratios 4.84 and 2.79, the silica peak heights first increased from the bottoms towards the centers, and then decreased from the centers of the films towards the top surfaces. In Fig. 1E and 1F, most peaks remain unchanged at different depths for small Peclet number ratios 2.16 and 1.25, while the peak height at depth 1.0 in Fig. 1F appeared smaller than the others.

The comparisons of silica scattering peaks show evidence for stratification within the mixture films. Typically, stronger scattering peaks for silica indicate a larger volume fraction of silica particles in particular regions of the film. For comparison, insets of the regions corresponding to PS particles are provided in the Supplementary Information (Fig. S1), but trends are not as clear in these data, due to weaker scattering from the PS particles, compared to the silica particles.

In order to quantify these results, we applied a linear combination analysis to the data, assuming the scattered intensity of the binary mixtures came from the weighted summation of scattering from pure polystyrene and silica. Details are given in supplemental information. The silica volume fraction at different scaled film depths for binary mixture films are plotted in Fig. 2. The results of our data analysis are consistent with the qualitative discussion of the

data in Fig. 1 in most cases; that is, in general our data fitting yields high volume fractions of silica for scaled film depths that display a more intense silica peak (Fig. 1). There are a few exceptions; the degree of stratification from data analysis for the sample with Pe_2/Pe_1=1.25 (Fig. 2F) is more pronounced than may be expected from looking at the peaks in these spectra, which do not show a very dramatic change throughout the film (Fig. 1F).

Relative errors in the silica volume fractions derived from the fits estimated from the coefficient of determination, R^2 were less than 5%, indicating good agreement between the data and model fit. However, the R^2 values are based on fits of a large number of data points (700-800 values of $I(q)$ at each film position); thus, the uncertainty derived from the goodness-of-fit may be an underestimate of the error. Thus, to provide a more conservative estimate of the uncertainty, the error bars shown in Fig. 2 are based on the average uncertainty in I for each film position, which we estimate based on the standard deviation of the 2D data along an arc at a particular q, err_{std}, using a 95% confidence interval (e.g. 1.96 $err_{std}/n^{1/2}$, where n is the number of data points over which err_{std} is calculated.

Our results on the type of stratification behavior are summarized in Table 2, along with the film thickness. When the size ratios are large (Fig. 2A and 2C, with Pe_2/Pe_1=7.70 and 4.84, respectively), one can see the stratification where large particles are driven upward, but also enrich the bottom, with small particles concentrated at the center. This is an interesting type of stratification behavior which has not been observed in previous studies of film stratification. We refer to this new type of stratification behavior as an “LSL” (large-small-large) structure.

Many earlier experimental and theoretical studies showed or implied only two types of

stratification behavior: small-on-top or large-on-top structures. In some cases reported in previous literature, interfacial effects resulted in a small-on-top structure with a few large particles pinned to the very top surface.[22-23] However, in our experiments, each data point arises from the average scattering of a thickness of 10 μm within the sample. This is approximately 100-1000 times of the dimension of nanoparticles. Thus, the first few data points of the top and bottom of the films represents several hundred layers of particles, and we are confident that the trends we observe are not due solely to particle interactions with the interface or the lower substrate. In Fig. 2A and 2C, there are at least two or more data points at both the top and bottom surfaces where the volume fraction of small particles is lower than the average throughout the film, indicated by the dashed line. Thus, the layers of large particles at the top and bottom of these films are ≥ 20 μm thick, and the layer of large particles at the tops of these films cannot be attributed to a few large particles, or even a monolayer of large particles, pinned at the upper surface.

Pe_2/Pe_1	**Stratification Behavior**	**Overall Film Thickness (μm)**
7.70	LSL	400
5.70	--	100
4.84	LSL	300
2.79	--	180
2.16	SLS	120
1.25	SLS	510

Table 2. Summary of Peclet number ratios and type of stratification behavior.

Another film with a large particle size ratio but a relatively small total thickness (Pe_2/Pe_1=5.70, shown in Fig. 2B), is more difficult to classify. There is clearly a segregation of small and large particles in this film, but the small thickness of this sample results in a lower resolution in the SAXS experiments (i.e., fewer data points throughout the film). The data point for the topmost 10 μm of the film is very different than the data point immediately beneath it. Higher resolution of the microbeam experiments, particularly near the upper and lower surfaces of this film, are needed to classify the stratification behavior in this case.

When the size ratios are small (Pe_2/Pe_1=2.16 and 1.25, Fig. 2E and 2F, respectively), we can see the stratification with small particles enriched at the top and the bottom of films ("SLS," or small-large-small structure). The sample shown in Fig. 2D, with Pe_2/Pe_1=2.79, also seems to show SLS behavior, although as in Fig. 2B, the data point corresponding to the topmost 10 μm of the film is very different than the layer immediately beneath it, and higher resolution of the composition in the vertical direction would be desirable.

Among the "LSL" films, the film with the largest size ratio, 7.70, appears to have the smallest degree of stratification. Although this does not agree with our expectations from diffusive theories of stratification, it is consistent with trends seen in our earlier AFM studies. The difference in the degree of stratification is small among "SLS" films. This may be because the differences in size ratio are not as large.

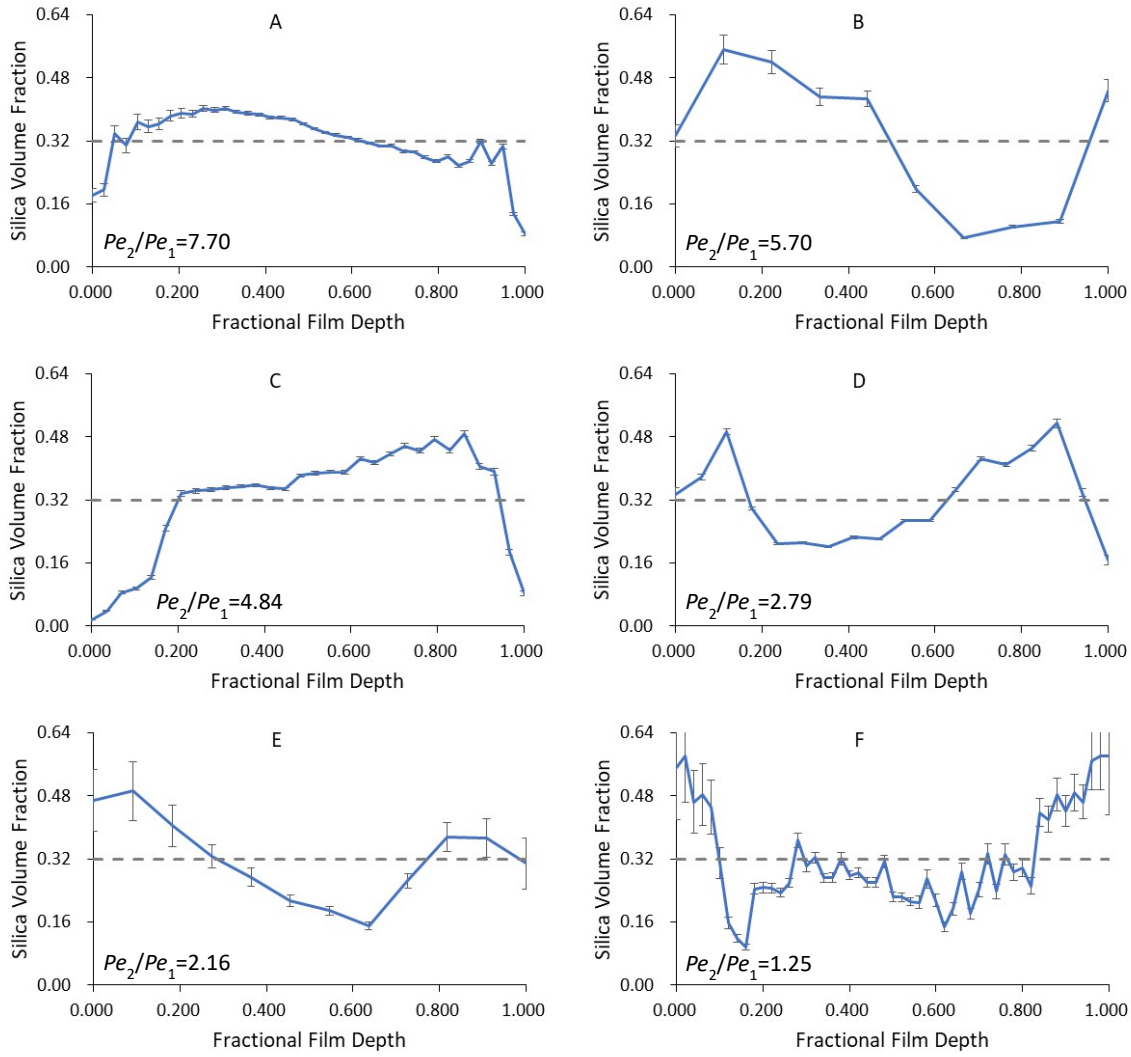

Fig. 2 Silica volume fractions at different fractional film depth of mixtures films made with (A) PS-48 and Silica-6 (Pe_2/Pe_1=7.70); (B) PS-48 and Silica-8 (Pe_2/Pe_1=5.70); (C) PS-48 and Silica-10 (Pe_2/Pe_1=4.84); (D) PS-48 and Silica-17 (Pe_2/Pe_1=2.79); (E) PS-21 and Silica-10 (Pe_2/Pe_1=2.16); and (F) PS-21 and Silica-17 (Pe_2/Pe_1=1.25). Grey dash lines represent the expected bulk volume fraction, 0.32, for silica particles in dried films.

In order to visualize how particles arranged and compare the difference in the extent of stratification among films, illustrations generated from results are made using MATLAB. GIMP software was used to blur the color change between layers to mimic smooth composition changes, which is essentially an interpolation (Fig. 3A-3F).

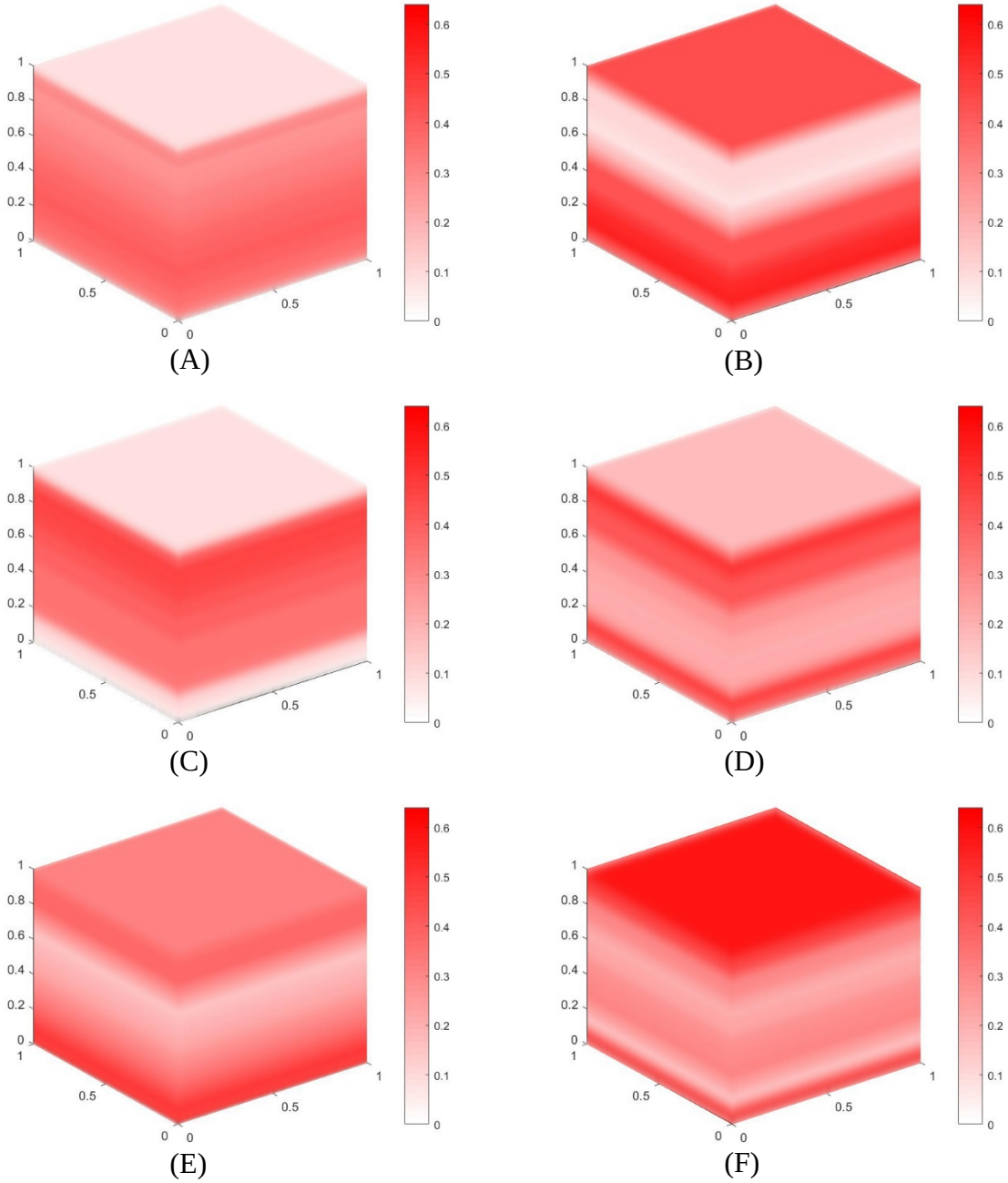

Fig. 3 Illustrations of film structures with nanoparticle size ratios: (A) 7.70; (B) 5.70; (C) 4.84; (D) 2.79; (E) 2.16; and (F) 1.25. Color bars indicate corresponding silica volume fraction, 0-0.64, (Darker red indicates higher silica volume fraction, and lighter red/white indicated lower silica volume fraction). Interpolation between data

points was made to simulate smooth concentration change.

Conclusions

We have successfully utilized SAXS experiments with an X-ray micro-beam as a novel method to study the depth-resolved structures of binary mixture films. SAXS patterns with small variations at different depths of films were captured and fit to get quantitative results on particle compositions. We observe two new types of stratification behavior, large-small-large or “LSL” film structures, which we tend to find at large nanoparticle size ratios; and small-large-small or “SLS” film structures, which is observed at small size ratios. These interesting “sandwich” structures have not been predicted by previous theories or observed in earlier experimental studies, although several experimental studies in the literature have focused mainly on characterization of the top surface of the films. It is important to note that we have kept the initial volume fraction of both particles fixed in these studies. Other types of stratification behavior (small-on-top, large-on-top) may be favored for small and large particle size ratios as the initial volume fraction is varied, so our results are not necessarily in conflict with earlier studies that show other structures. However, these experiments provide some insight that can be used to drive future theoretical developments in film drying, colloidal stratification, and evaporative assembly; and can guide design of improved single-step processes for production of multicomponent nanoparticle films.

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