

Multivalent Prototyping and its Value for 5-5-5

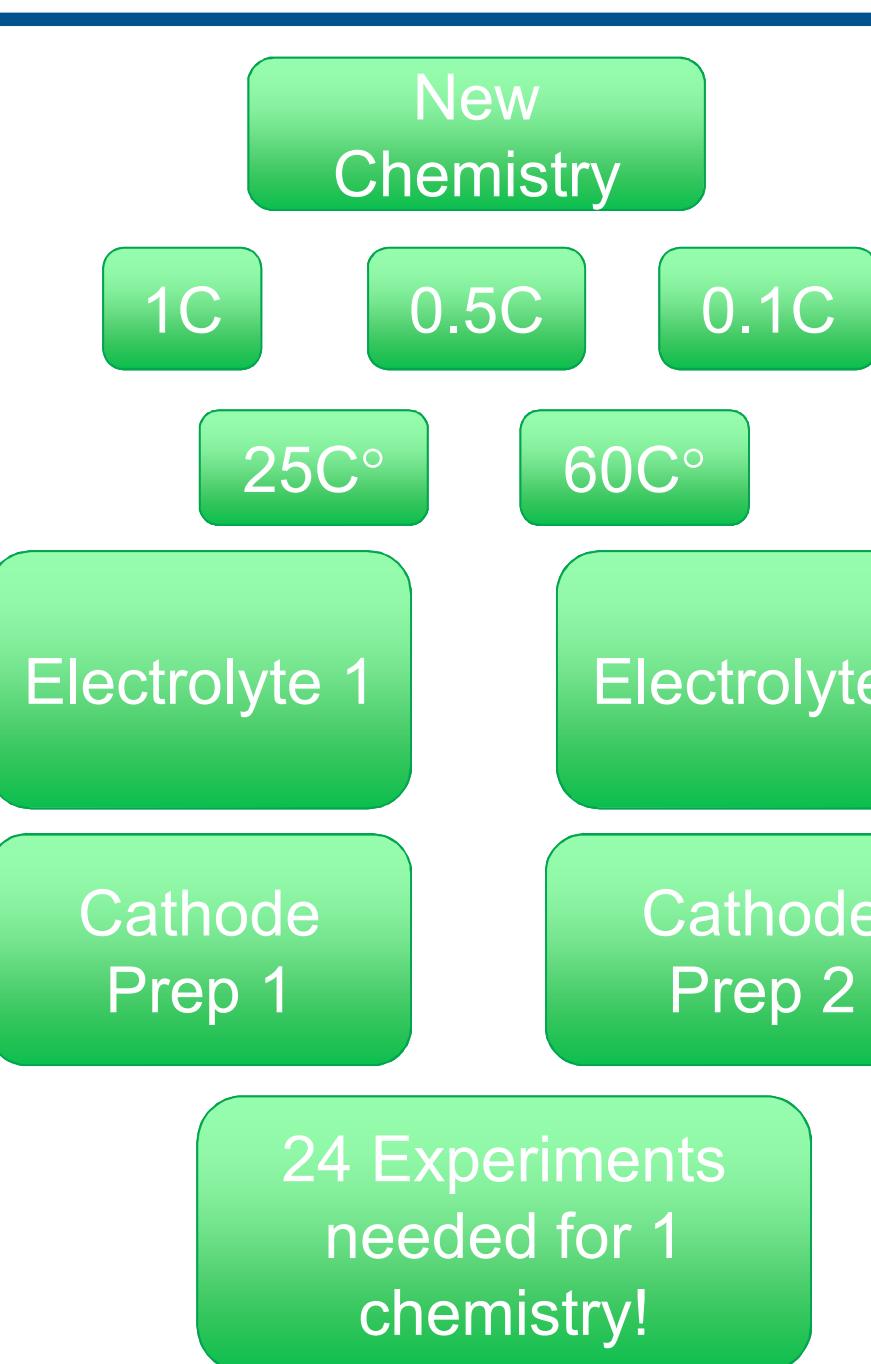
SAND2014-16003

¹Brian Perdue, ¹Nathan Hahn, ²Victor Duffort, ³Chen Liao, ¹Mark Rodriguez, ⁴Xiaoyin Chen, ⁵Sun Kim, ³Jack Vaughney, ²Linda Nazar, ⁴Johannes Schwank, ⁵Venkat Srinivasan, ¹Kevin Zavadil, ¹Chris Apblett

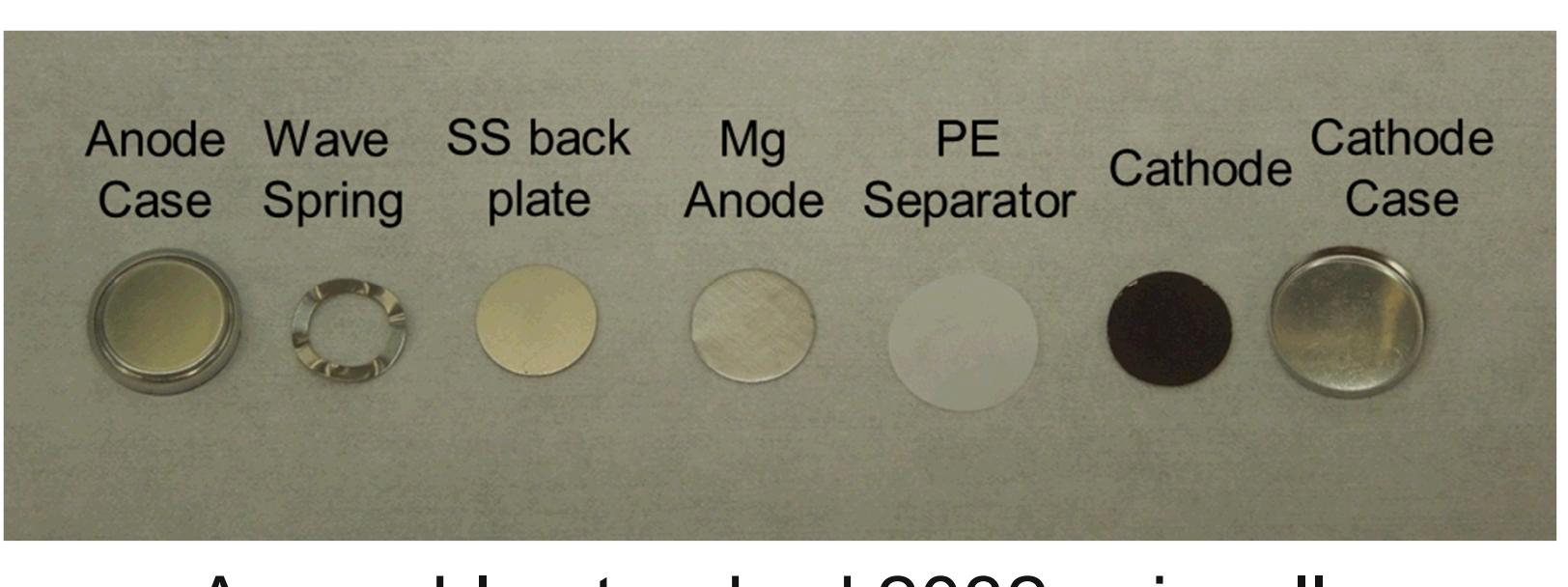
¹Sandia National Lab; ²University of Waterloo; ³Argonne National Laboratory; ⁴University of Michigan; ⁵Lawrence Berkeley National Laboratory

Motivation

Cell evaluation requires full assemblies. Many factors can affect the performance of the cell. This can lead to a lot of experiments. Thus, a fast, stable platform is needed to evaluate permutations.



Approach



Assemble standard 2032 coincells

This affords high throughput with great control and reproducibility!

How do we prototype



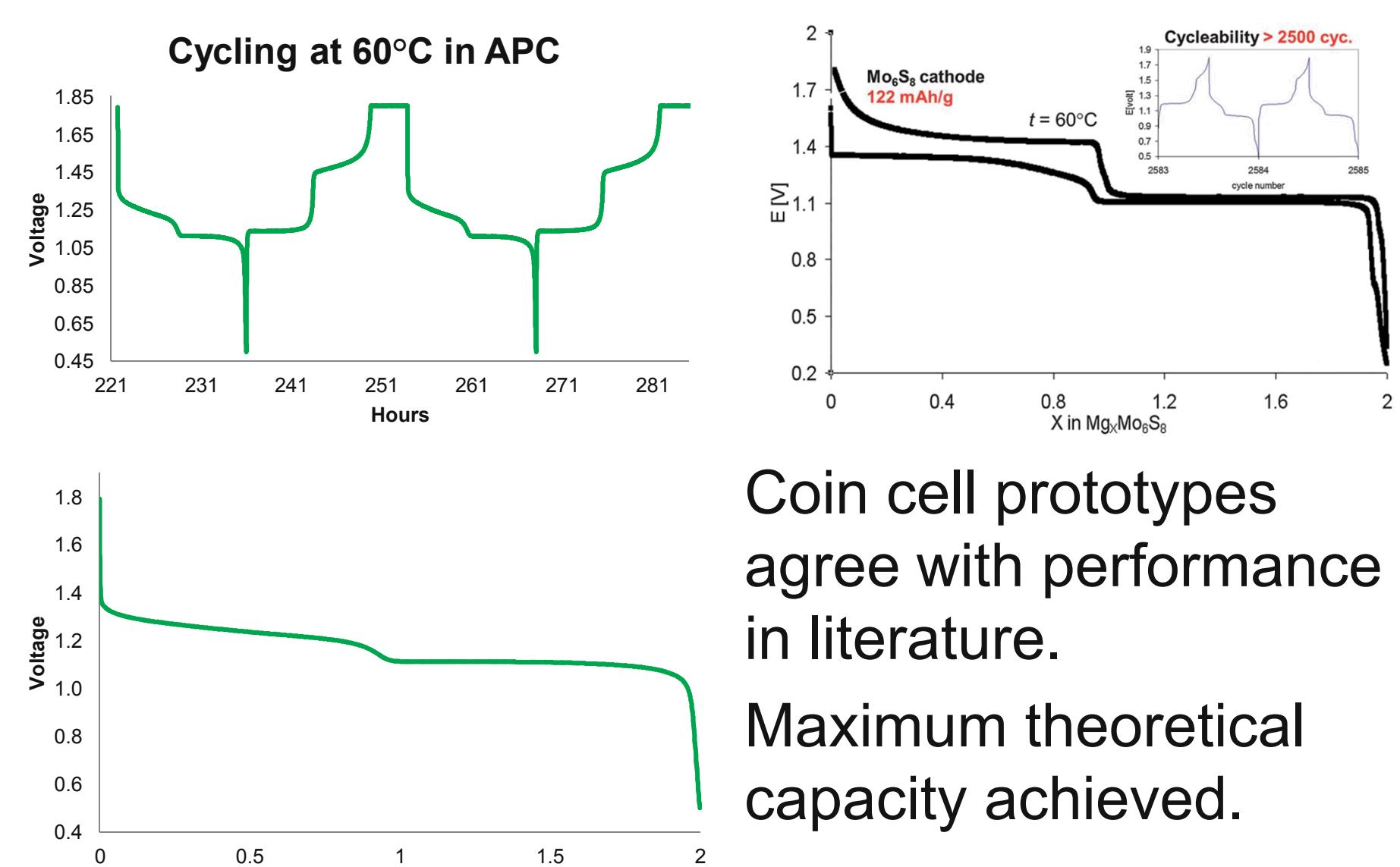
Place in temperature controlled environment



Control with tester

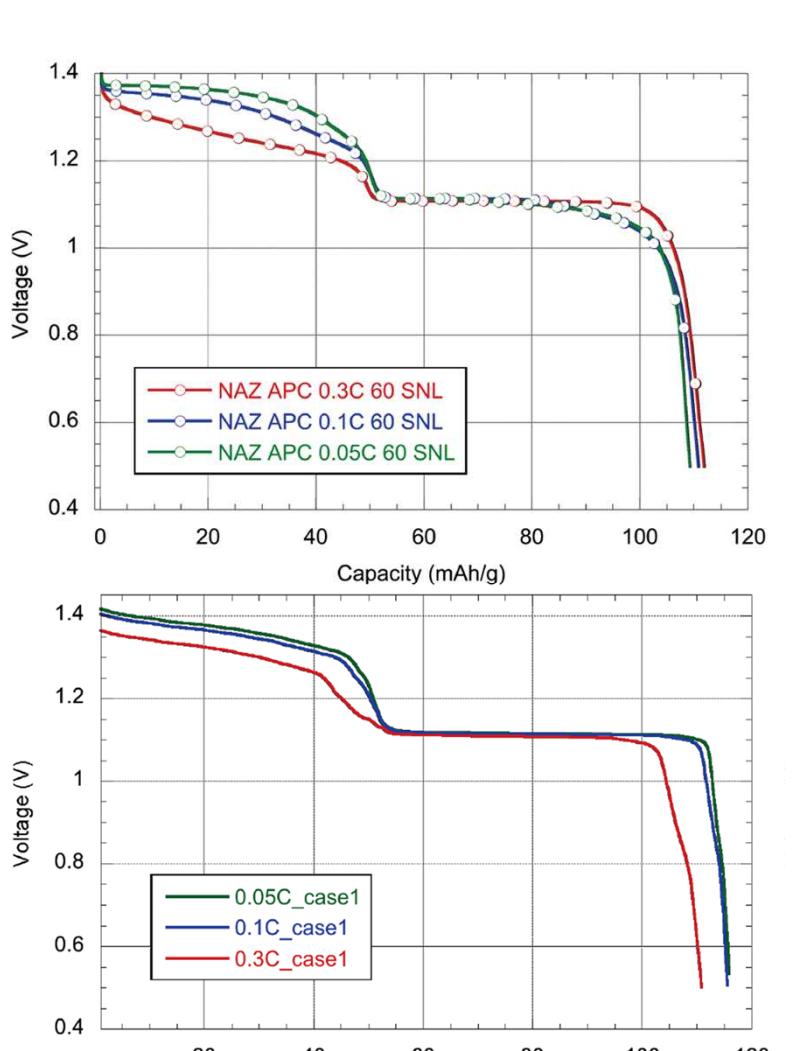
Major Accomplishments

Replication of Best in Class Literature

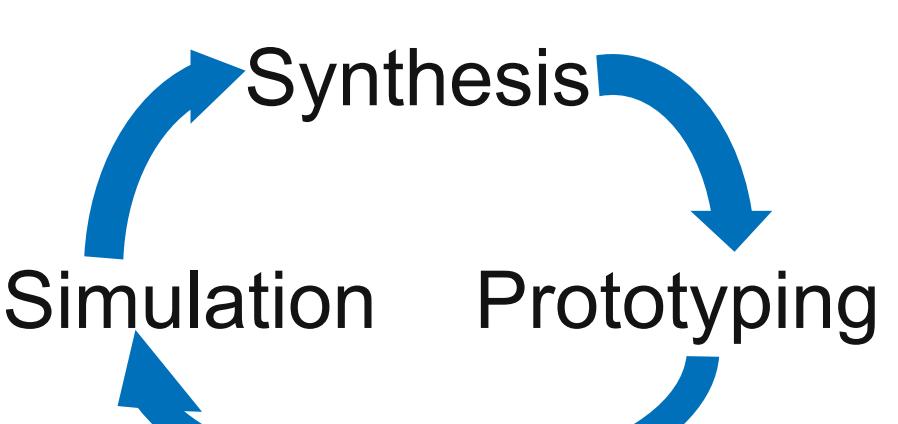


Coin cell prototypes agree with performance in literature. Maximum theoretical capacity achieved.

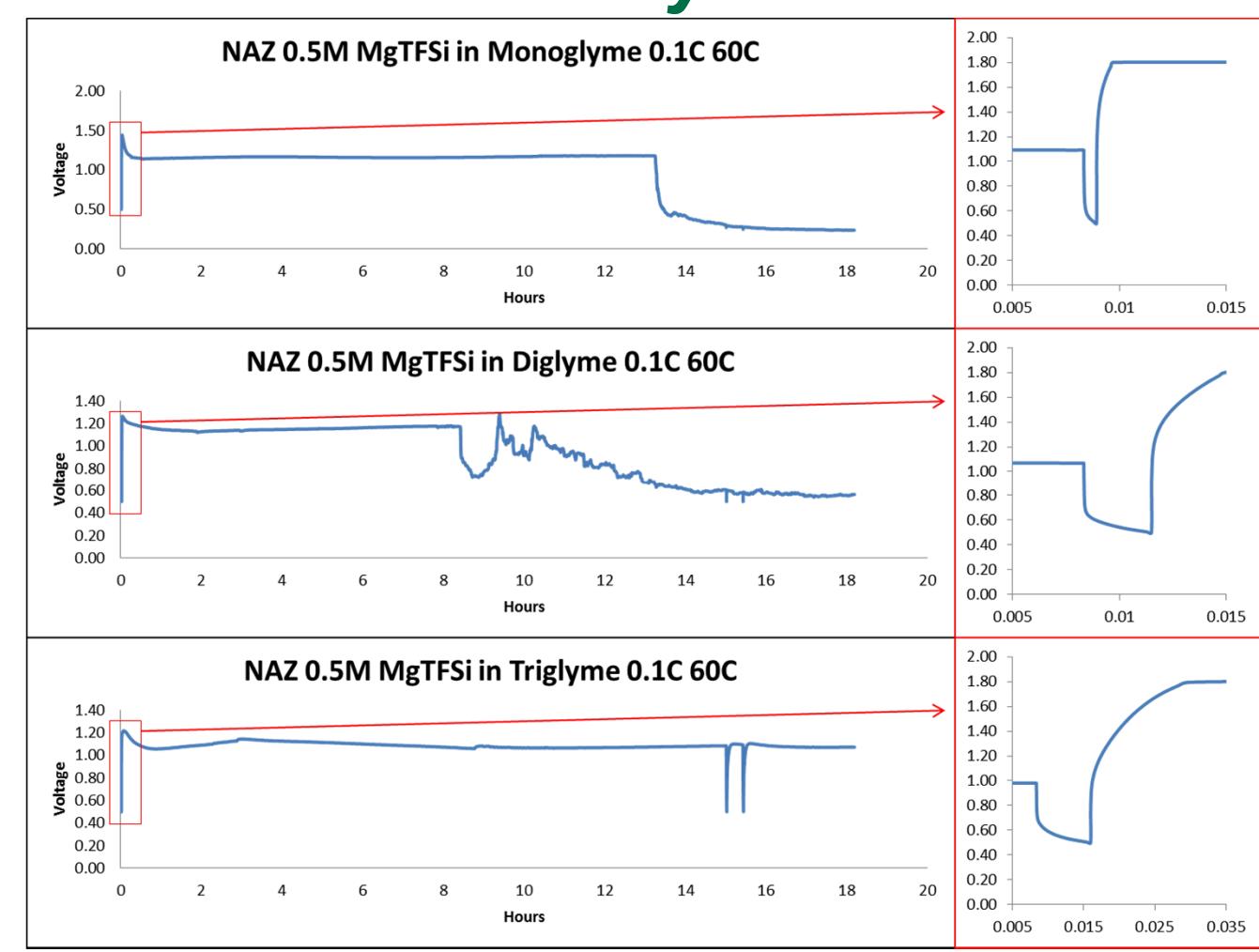
Preliminary Simulation Results



Initial results show good agreement. More collaboration needed to improve simulation.

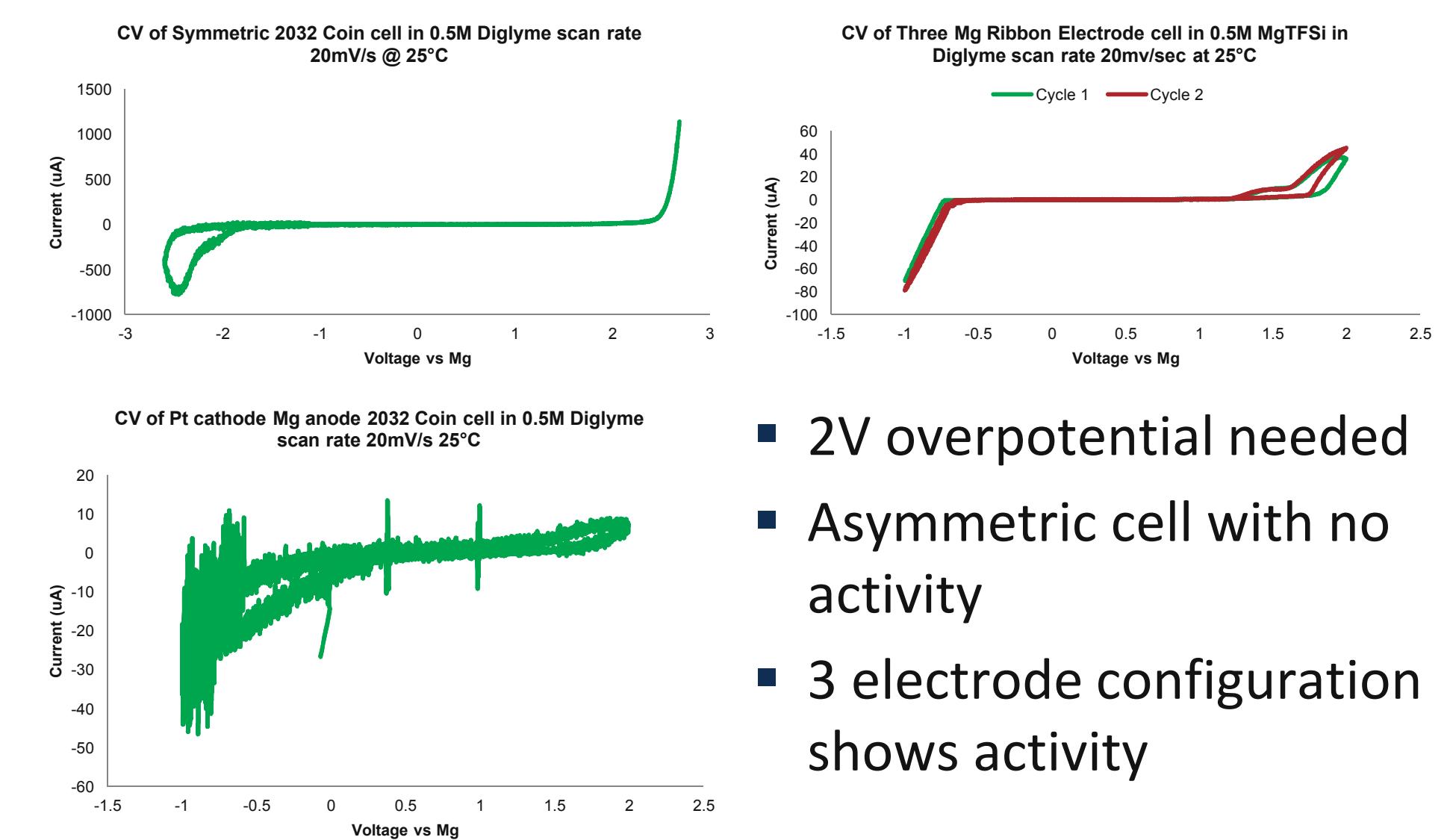


Evaluating the Performance of Current Electrolyte



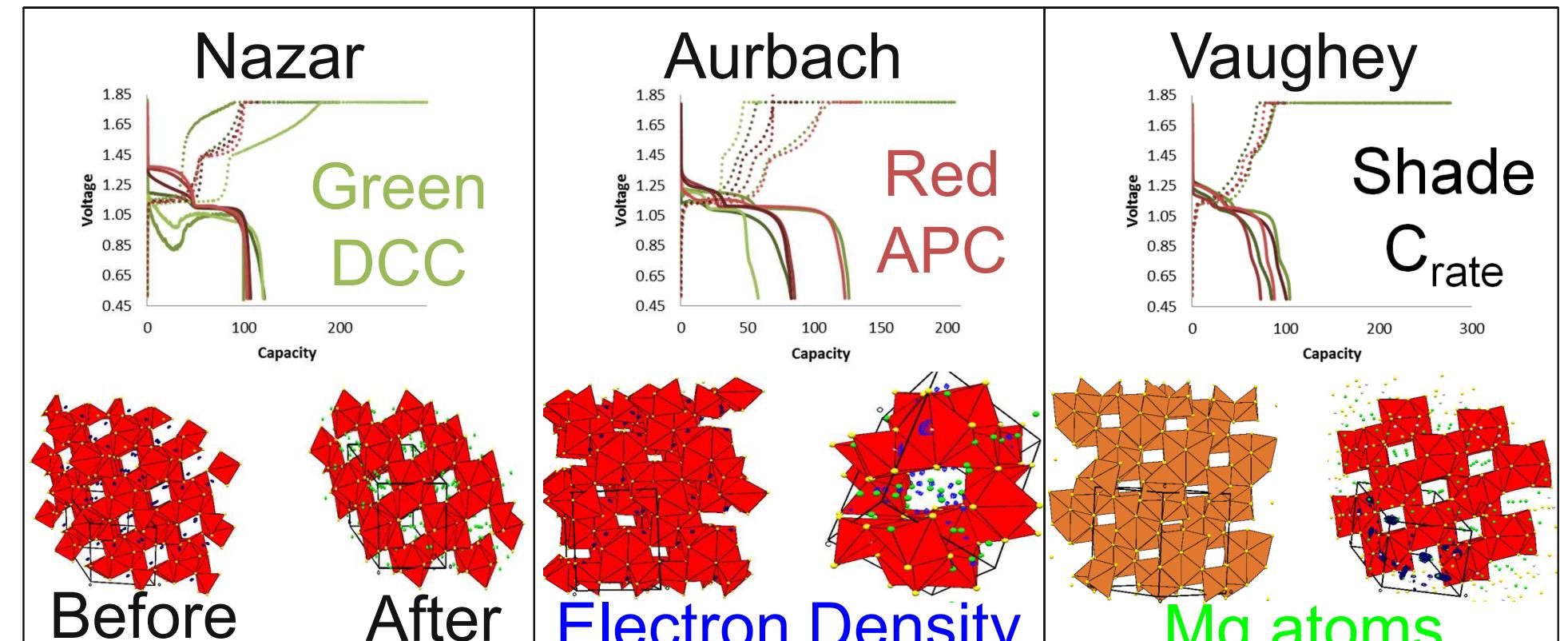
No performance with this electrolyte against a known working cathode

Coincell testing of Electrolyte



- 2V overpotential needed
- Asymmetric cell with no activity
- 3 electrode configuration shows activity

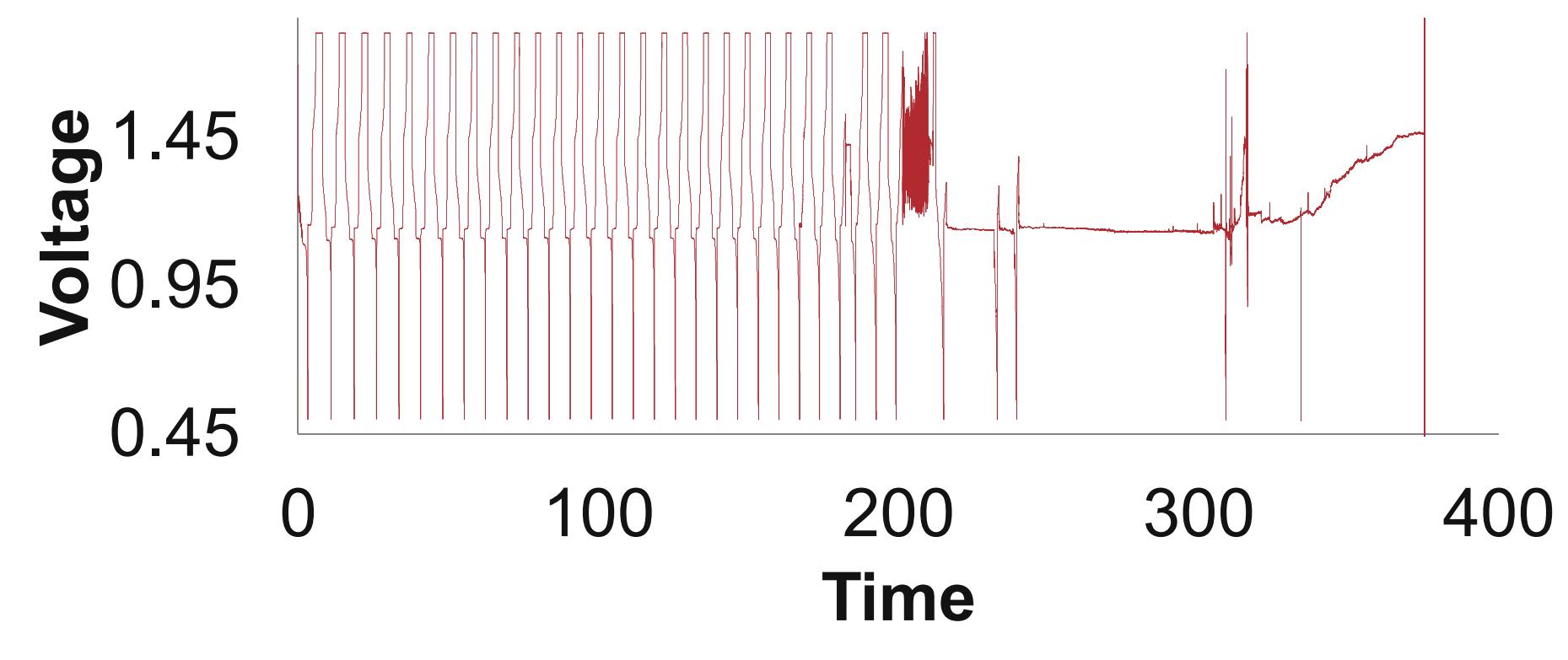
Better Understanding Through Collaboration



C_{rate}, Temperature, and electrolyte were tested against the three cathode preps. Prototyping allowed for the rapid and complete characterization of these materials

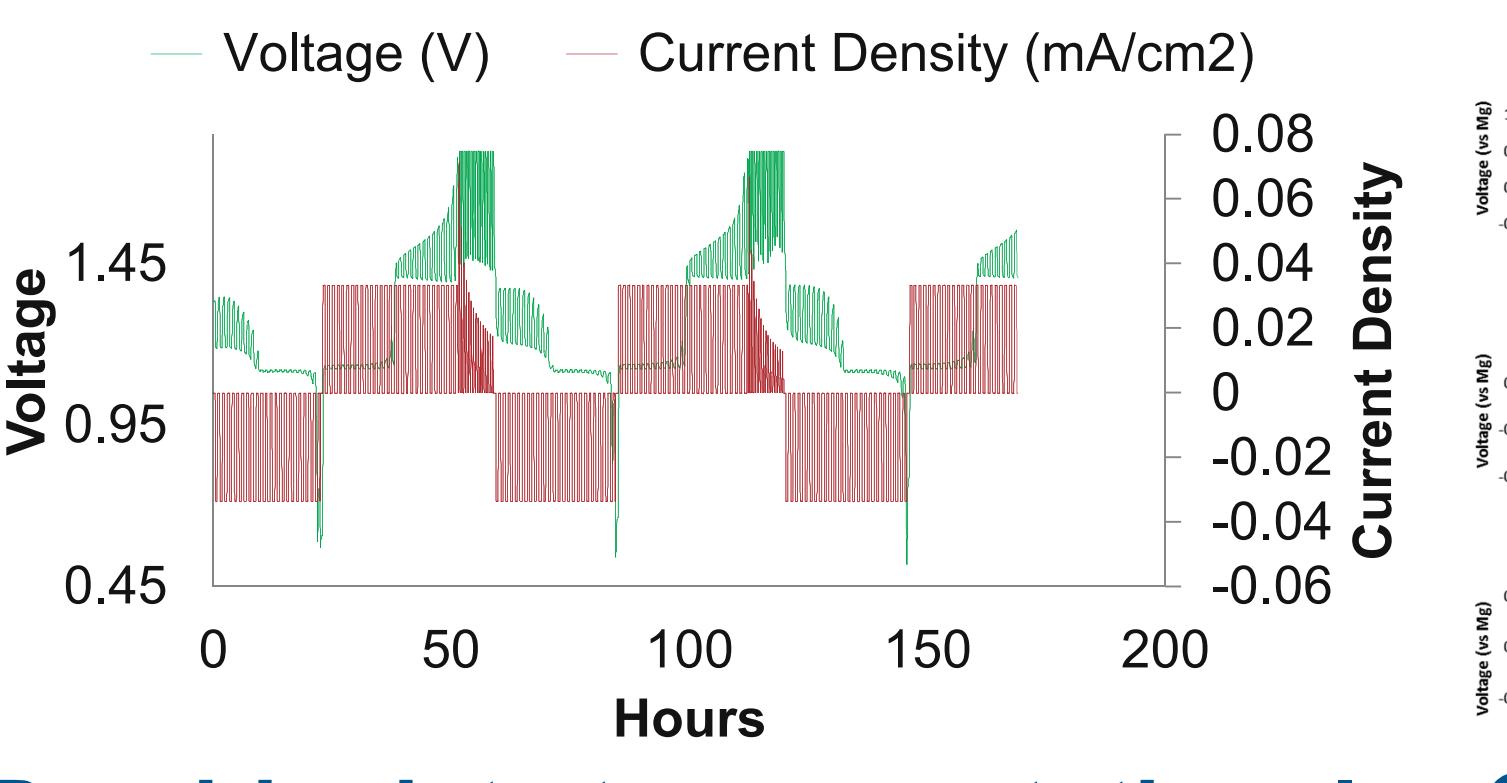
Future Directions and Impact

NAZ APC 0.3C 60



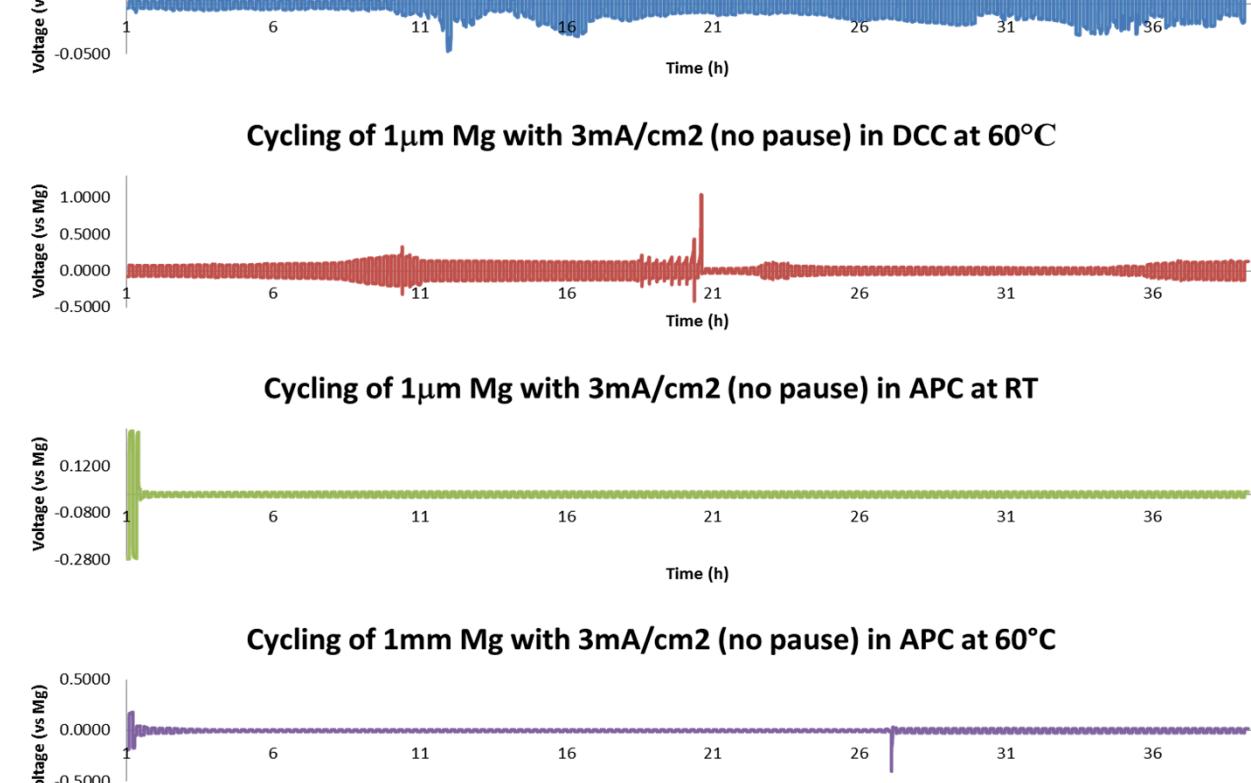
Investigate cell death

GITT of Nazar Cathode C/10 50% Duty Cycle



Provide data to computational groups to improve fit

Cycling of 1μm Mg with 3mA/cm² (no pause) in DCC at RT



Characterize electrolytes with Zavadil Group

Provide Feedback to synthetic and electrolyte groups to develop compatible chemistries for high voltage cathodes

Key Take-Away

The Mg²⁺/Mo₆S₈ system may not fulfill JCESR's aggressive requirements

Highlights the electrode/electrolyte compatibility challenge

Establishes the value of prototyping for meeting the 5-5-5 plan

Relevant publications:

Yoo, H. D.; Shterenberg, I.; Gofer, Y.; Gershinsky, G.; Pour, N.; Aurbach, D., Mg rechargeable batteries: an on-going challenge. *Energy & Environmental Science* **2013**, 6 (8), 2265-2279

Synthesis of Mo₆S₈ Cathode and its Subsequent Performance in a Mg ion Battery (Manuscript out)

Acknowledgements:

Tony Burrell
Mike Brumbach
Yosef Gofer
Doron Aurbach