

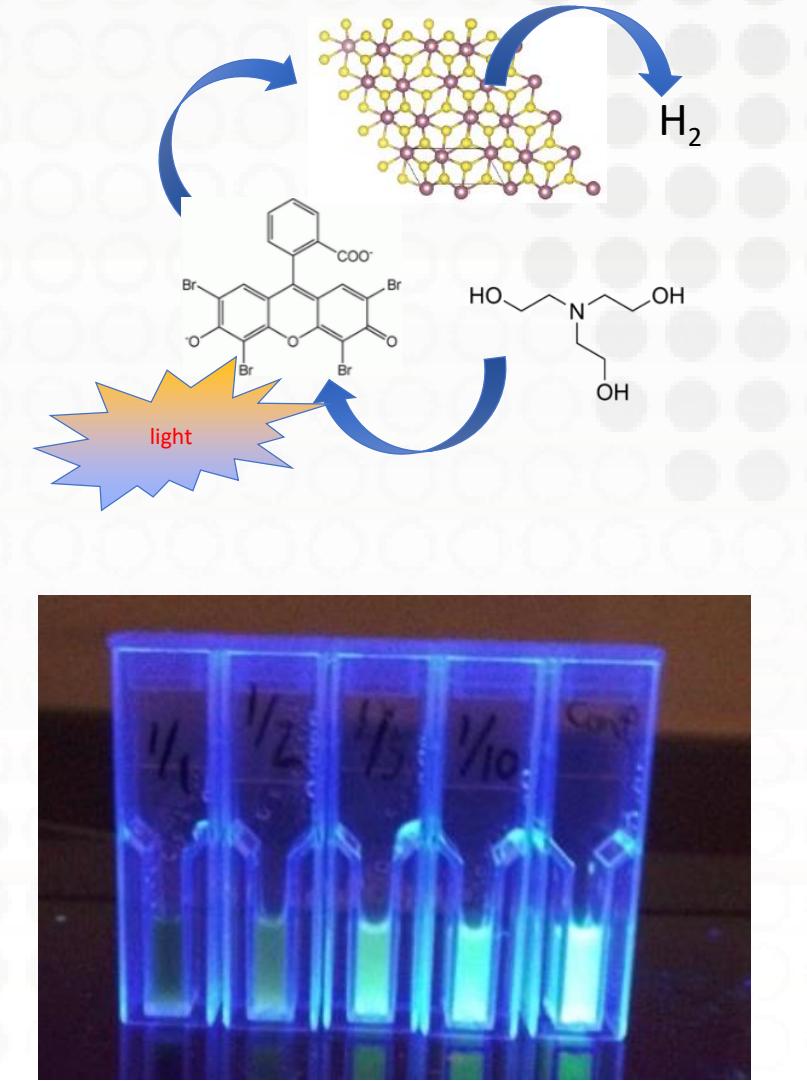
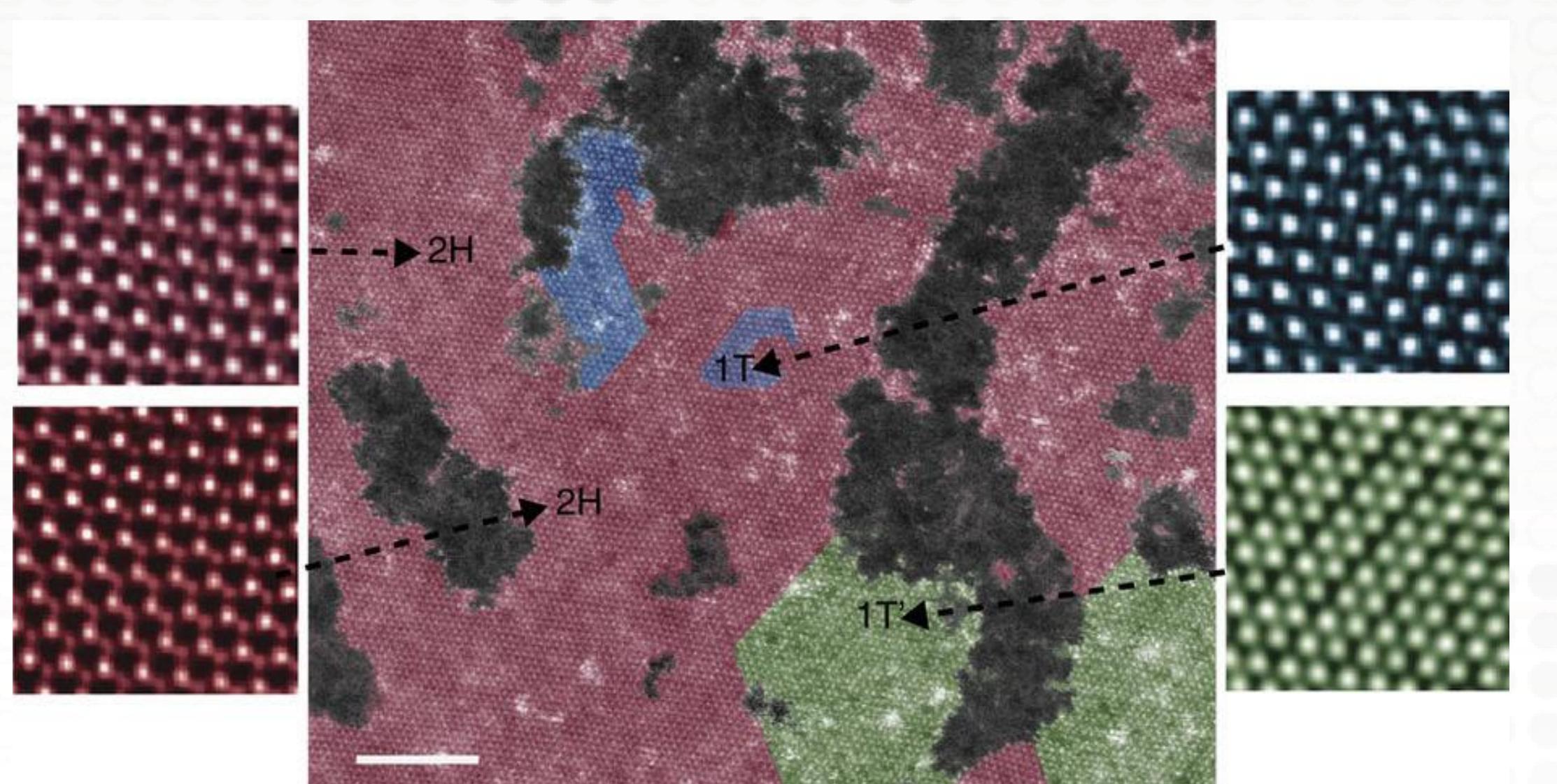
Self Assembly in Advanced Manufacturing

Stanley S. Chou¹, Bryan Kaehr², Kent Pfeifer³, Paul Clem⁴, Vincent Tung⁵, Adam Hecht⁶

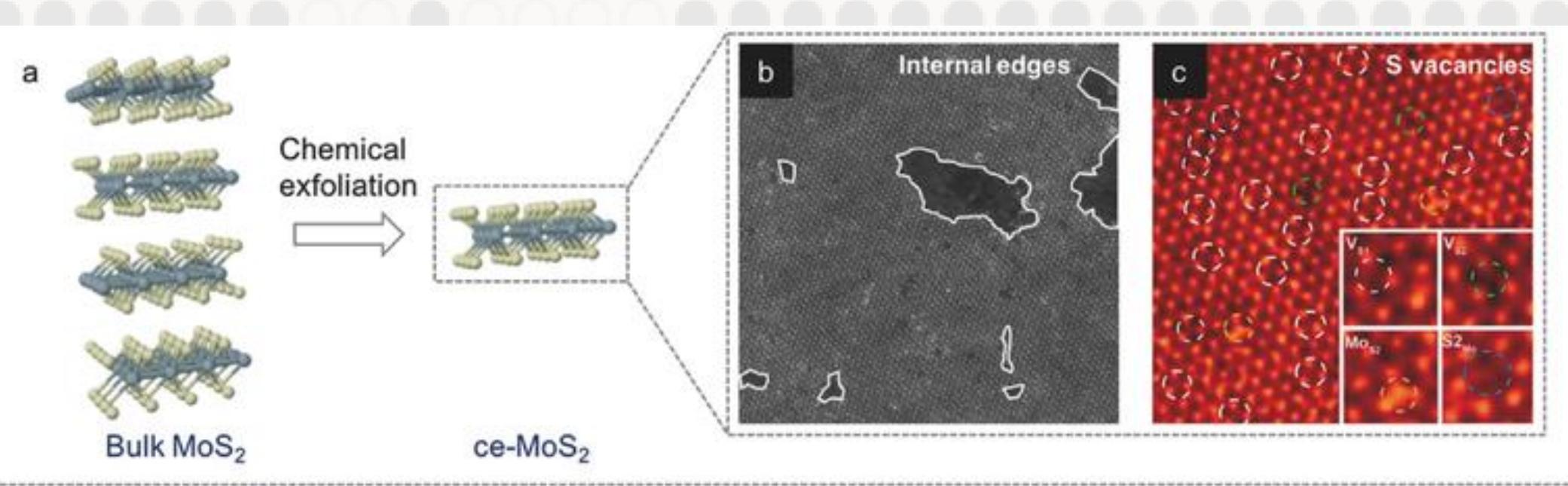
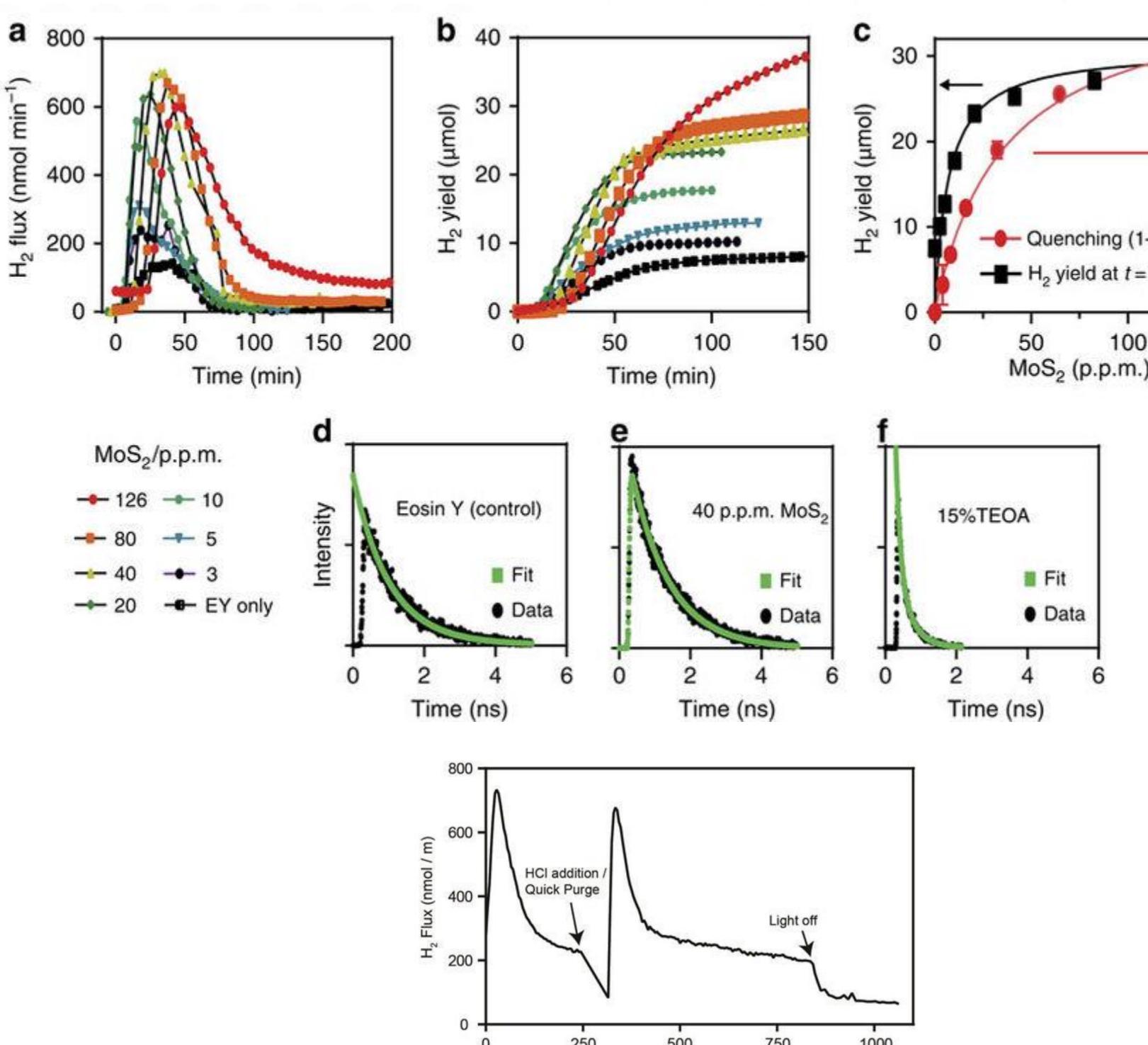
¹ Electronic, Optical and Nano Materials, Sandia National Laboratories Albuquerque, NM, USA. ² Advanced Materials Laboratory, Sandia National Laboratories Albuquerque, NM, USA. ³ Nano and Micro Sensors, Sandia National Laboratories, Albuquerque, NM, USA. ⁴ Electrical Sciences & Experiments, Sandia National Laboratories, Albuquerque, NM, USA. ⁵ Dept. of Materials Science and Engineering, University of California, Merced, CA, USA. ⁶ Dept. of Nuclear Engineering, University of New Mexico, Albuquerque, NM, USA.

Advanced manufacturing techniques are convenient tools for translating dispersed colloids and molecules into larger patterns and structures. Thus, can be viewed as a creative platform for materials chemistry and self-assembly, one that merges confined reactions, microfluidics, and intermolecular forces with broad length scales for the emergence of intermolecular control and self-organization in large areas.

Self-Assembly and Droplet Templating



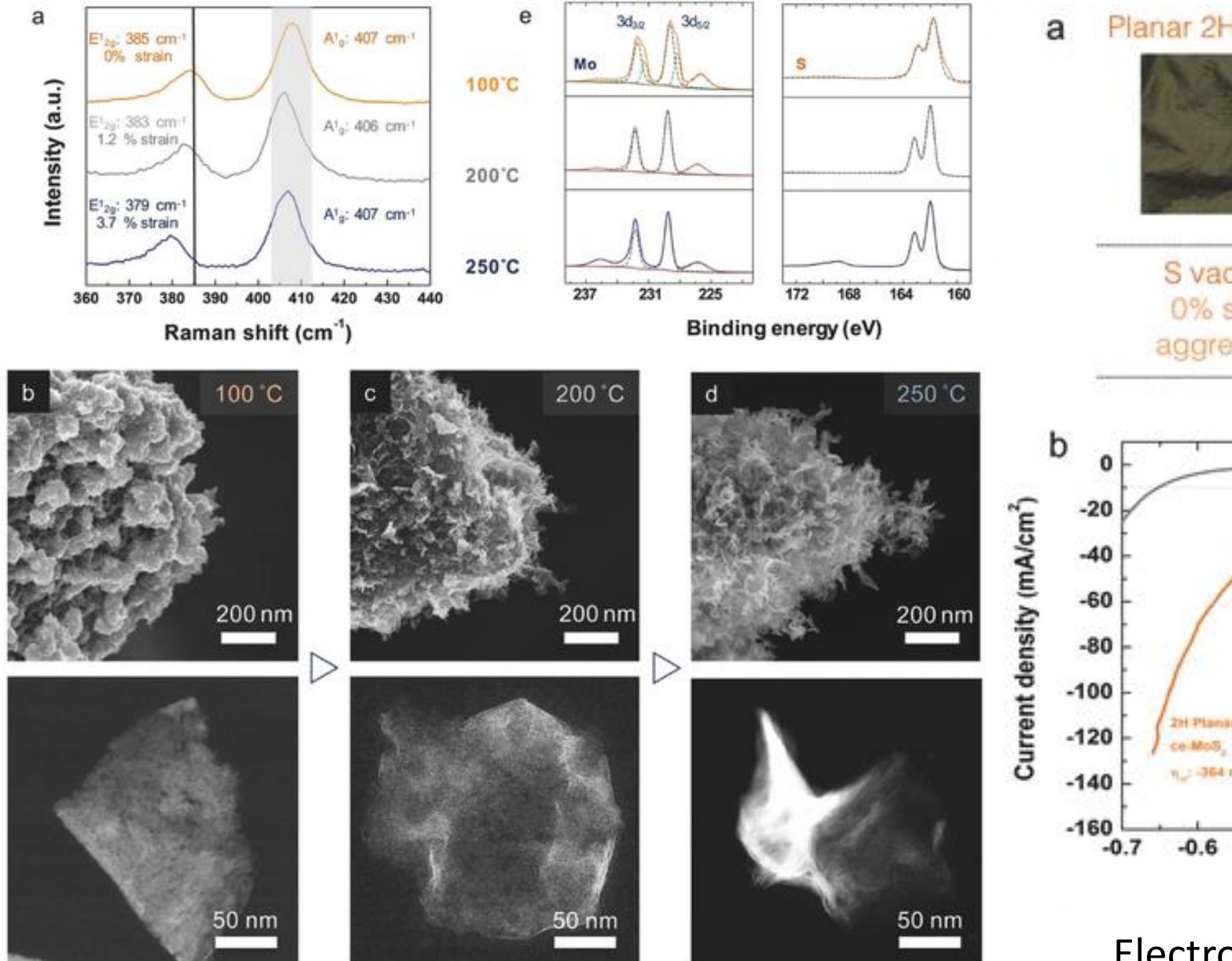
Solution self-assembly strategies applied to "ink" formulation. Exfoliated MoS₂ assembled with eosin Y to form charge transfer pathways between colloid and molecule. On Right: (a) Real time H₂ flux at incremental MoS₂ concentrations. Diminished reactions are indicative of proton exhaustion, and can be restarted with acid addition. (b) Corresponding cumulative H₂ yield (For a and b note key bottom left). (c) Incremental concentration of MoS₂ and correlative overlay of cumulative H₂ yield (black) with fluorescence quenching of dye (red) in solution. (d) fluorescence decay of eosin Y, (e) reaction saturated with eosin Y and MoS₂ showing similar decay, indicative of static-quenching and ground-state complex, and (f) collisional quenching between TEOA and eosin Y, resulting in second order decay and attenuated lifetime. The same strategy can be applied to other colloids, an example being transparent, thermally insulating coatings we are developing for ARPA-E.



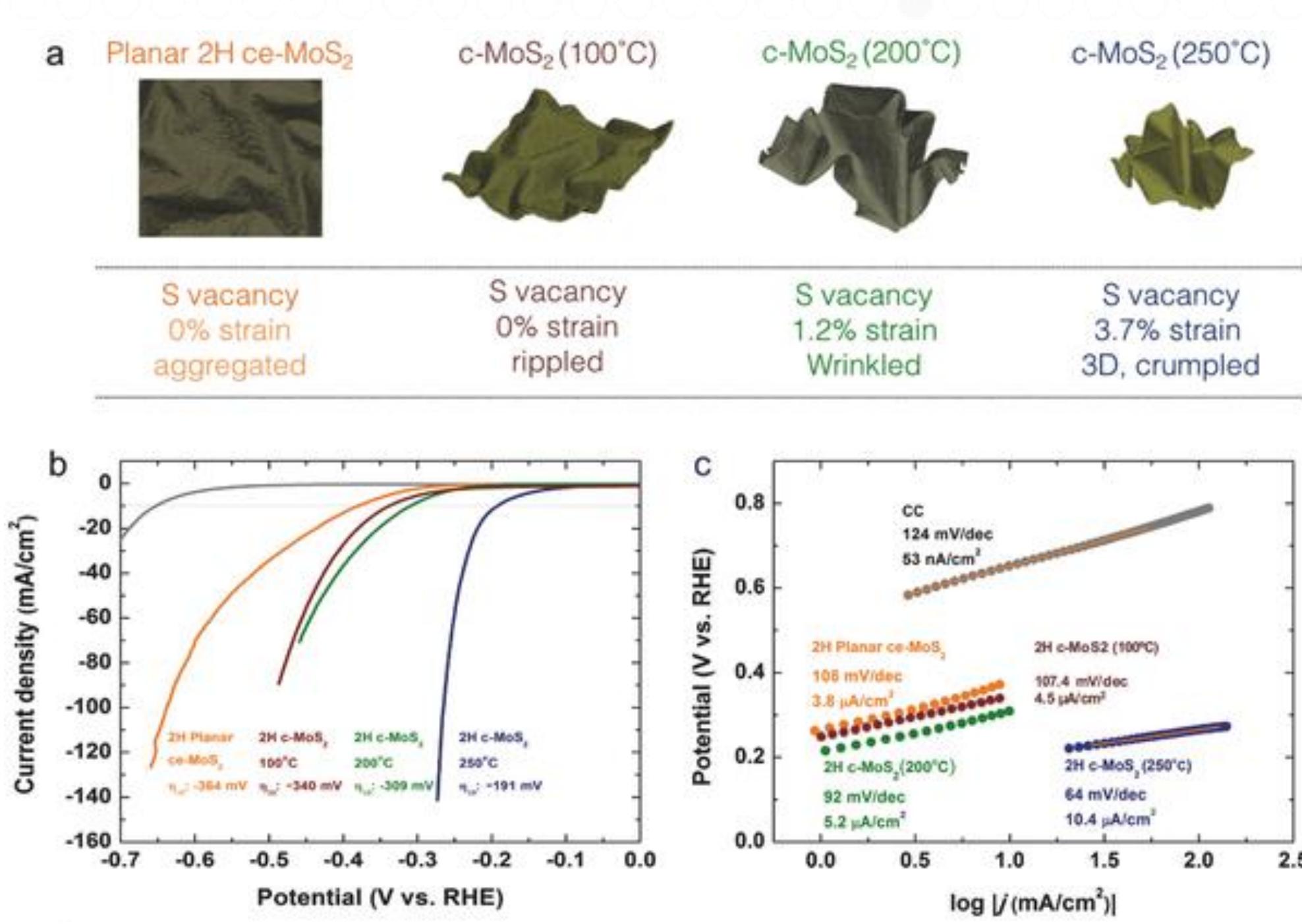
ce-MoS₂ with internal edges & S vacancies

Single to few layered ce-MoS₂ with internal edges & S vacancies

3D c-MoS₂ with internal edges & strained S vacancies



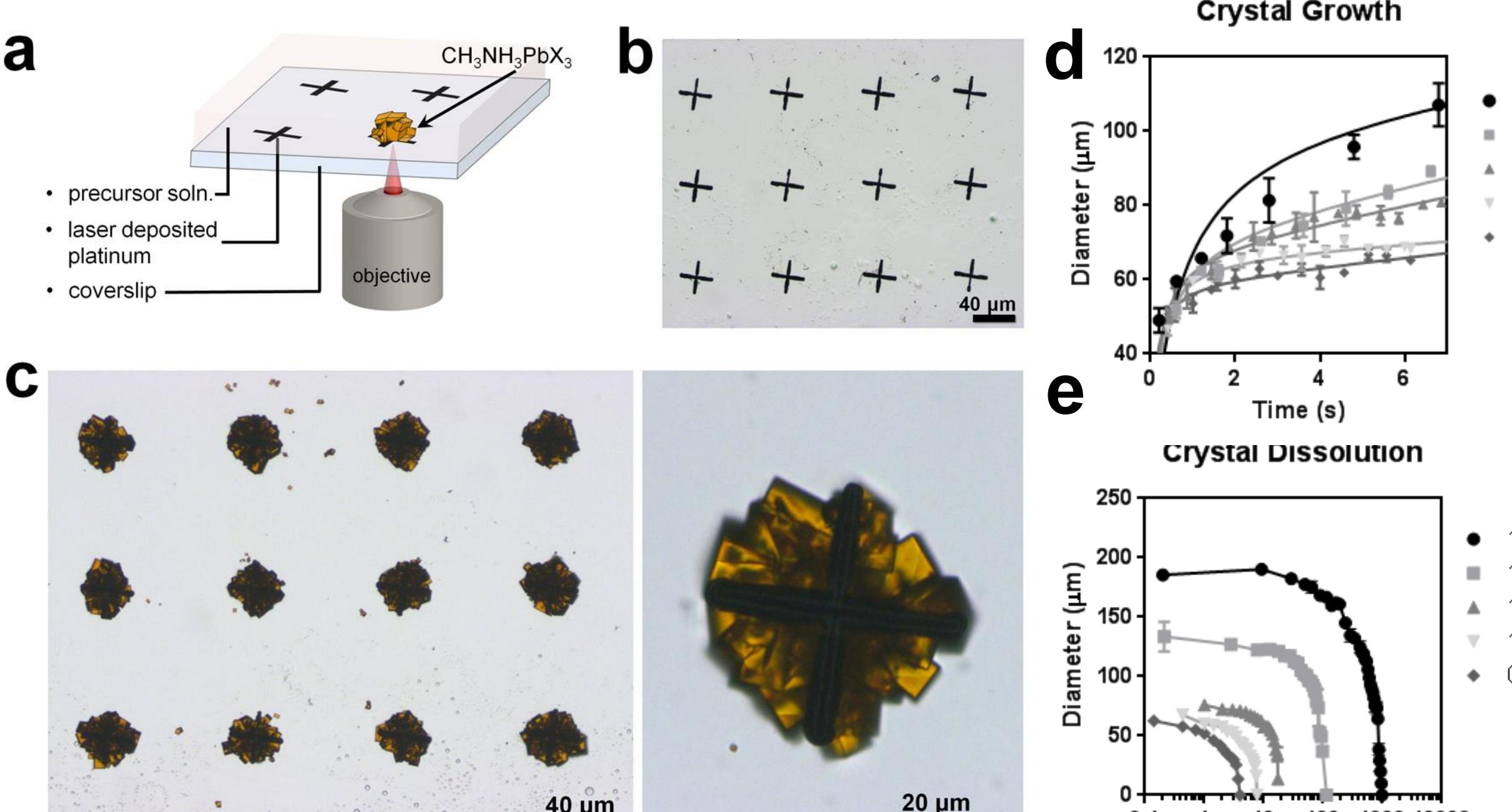
Surface tension and hydrodynamics can be designed to control wetting and patterning; further, each droplet is a reactor, capable of templating colloids and molecules inside. Shown on left, edges and atomic defects of (a) MoS₂ characterized by (b) TEM and (c) HAADF. (d) Charged, MoS₂ droplets undergo stages of (i) electrostatically induced separation, (ii) fission, and (iii) capillarity-induced crumpling into 3D MoS₂ structures. This brings about a high degree of strain in individual MoS₂ sheets, bringing about a semiconductor to metal transition in folded creases. The strained creases form the basis of the conductive molecular wire-network that electronically connect the catalytically dormant defects to enable orders of magnitude of increase in catalytic site density.



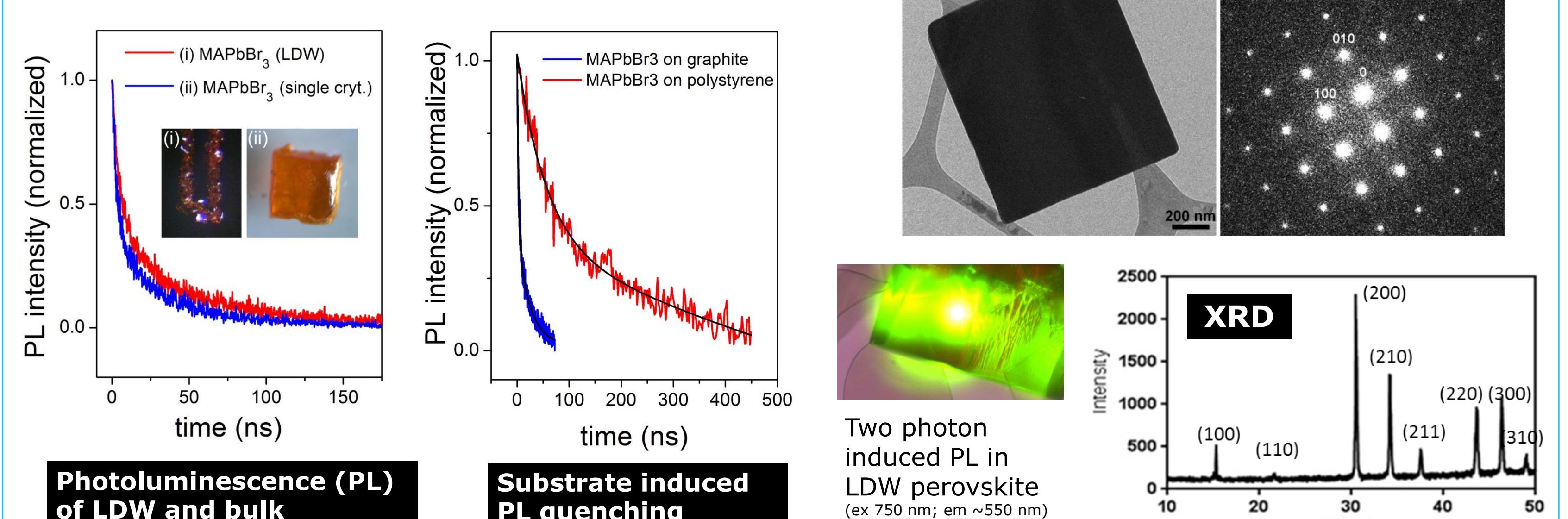
Electrochemical HER data courtesy of Prof. L. Li, KAUST

Laser-Directed Writing of Perovskite Crystals

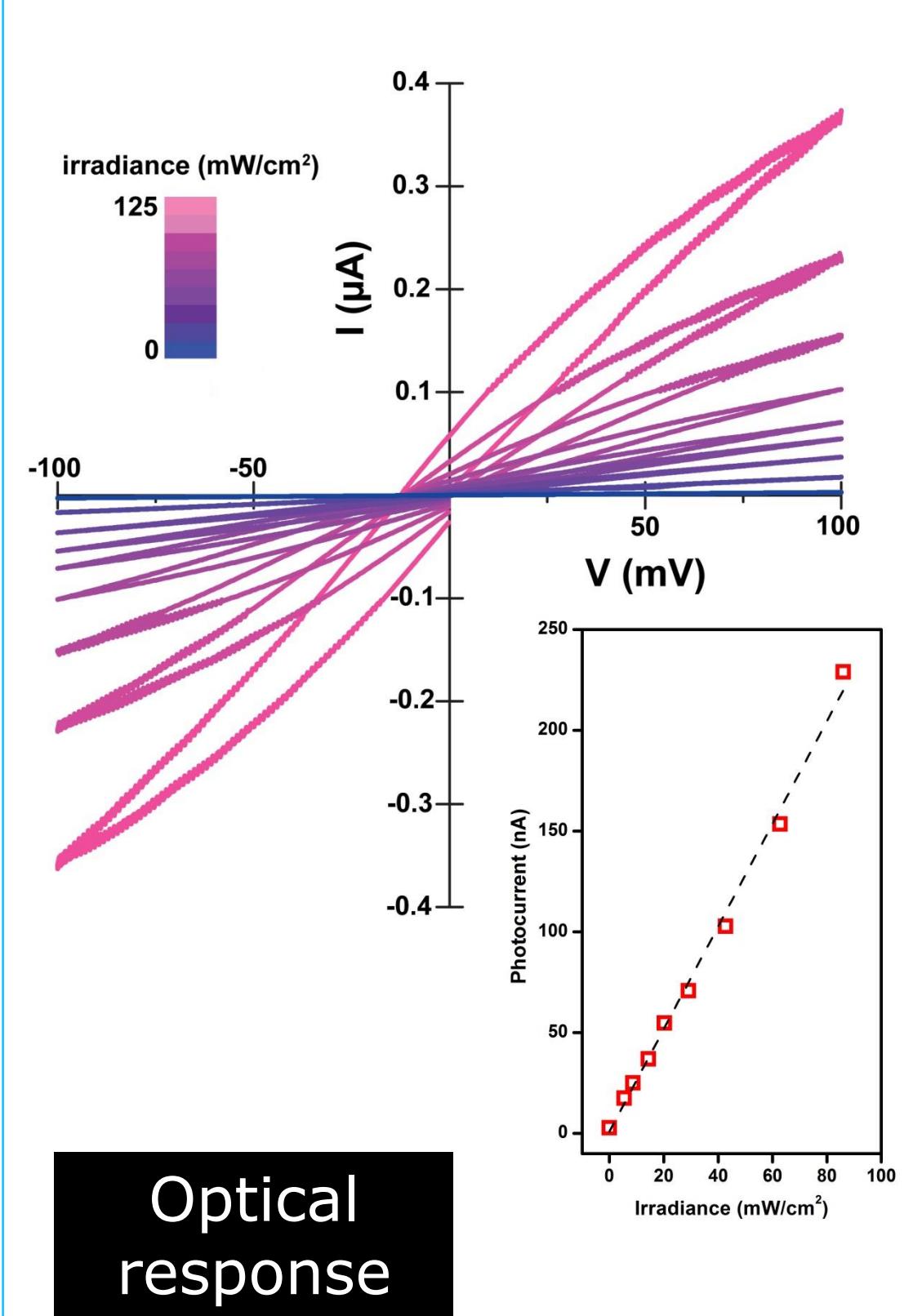
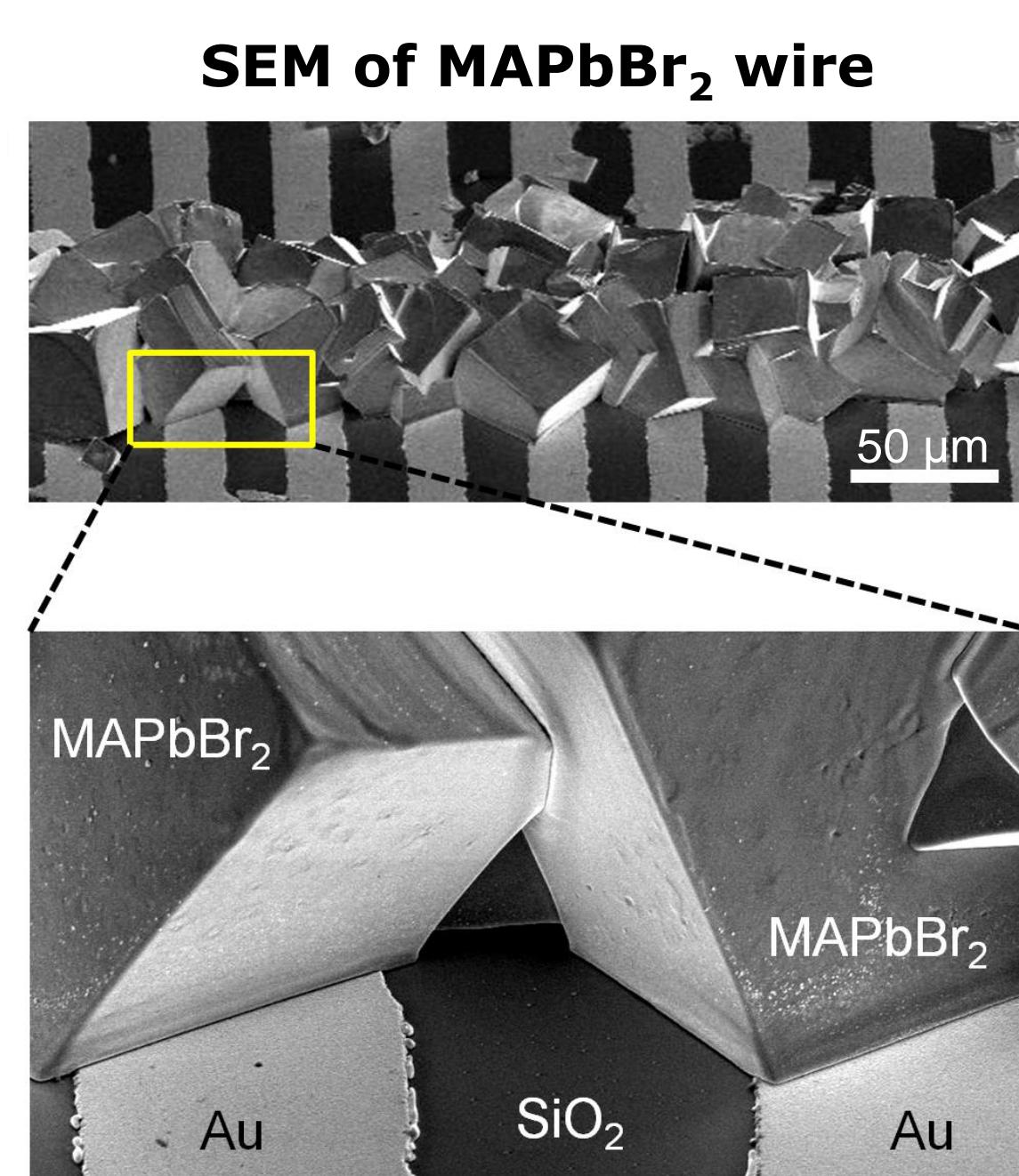
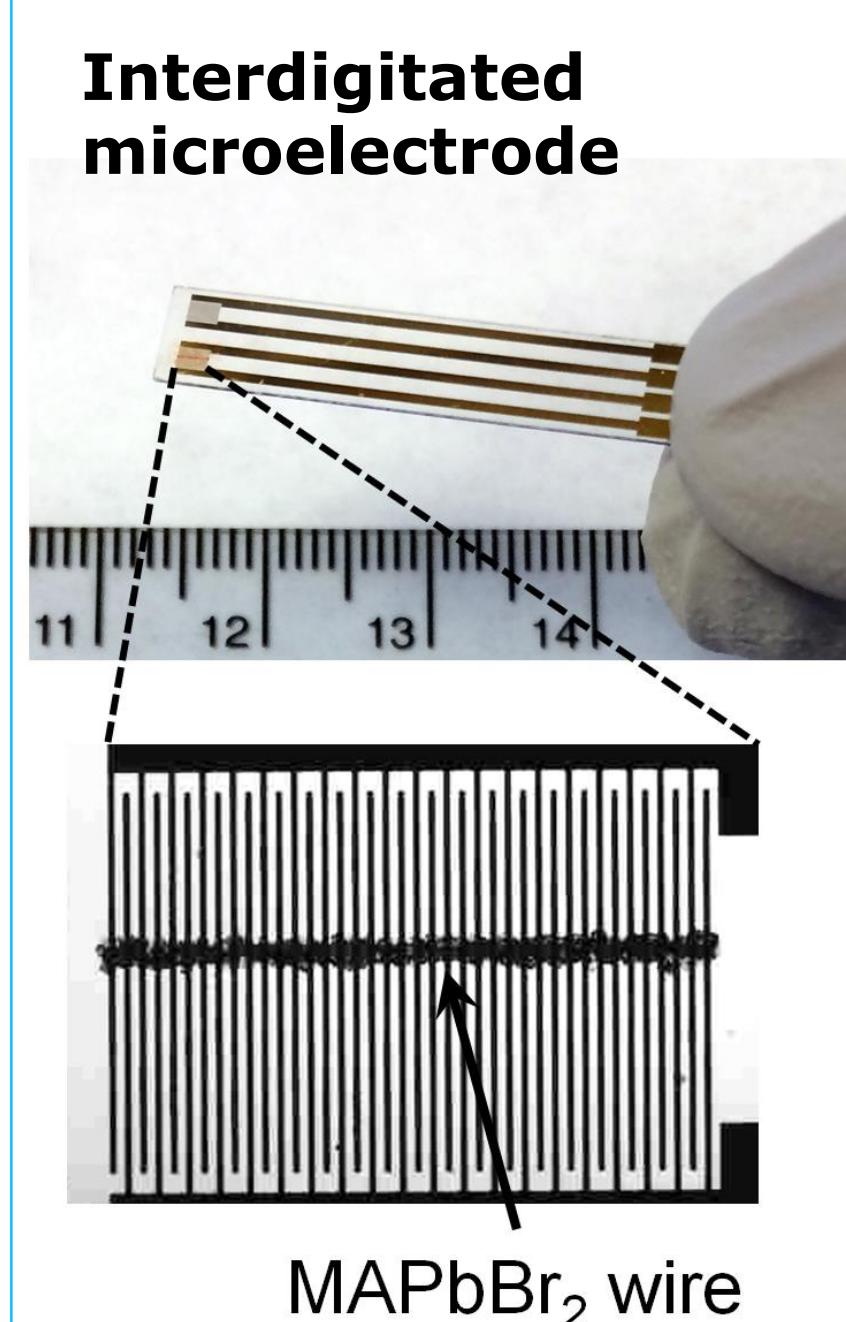
Laser directed perovskite crystallization, as enabled by localized heating. (a) Schematic of the experiment. Platinum crosses are patterned onto glass and immersed in precursor solution. The backside of the Pt metal is heated to induce MAPbBr₃ crystallization. (b) Laser patterned platinum crosses. (c) An array of MAPbBr₃ crystals following laser heating. This site-specific heating enabled a detailed understanding of the dependence of crystal growth (d) and dissolution rates (e) on precursor concentration.



Characterization of LDW perovskite materials

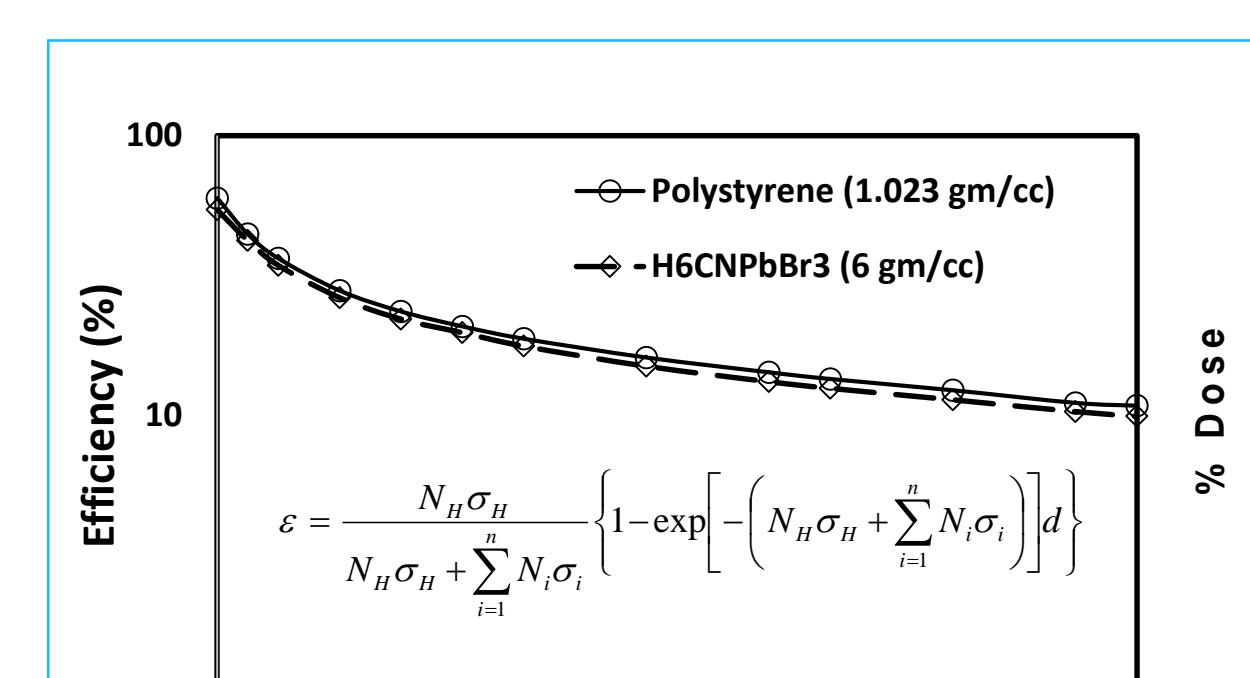


Microfabricated photodetector employing a LDW perovskite wire



Optical response

Relevant Publications.
 1. Chou, et al. 10.1038/ncomms9311
 2. Chen, et al. 10.1002/adma.201703863
 3. Chou, et al. 10.1021/acs.jpcllett.6b01557



This work was supported by the U.S. Department of Energy, Office of Basic Energy Sciences, Division of Materials Sciences and Engineering, the ARPA-E SHIELD program and the Laboratory Directed Research and Development program at Sandia National Laboratories, a multimission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC, a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA0003525.