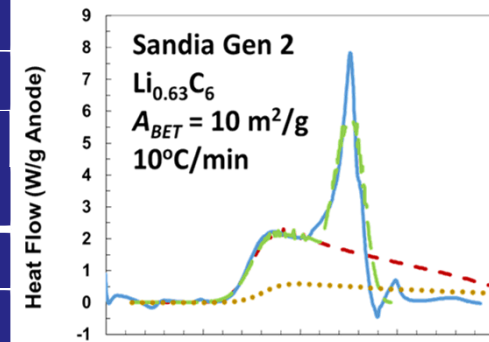
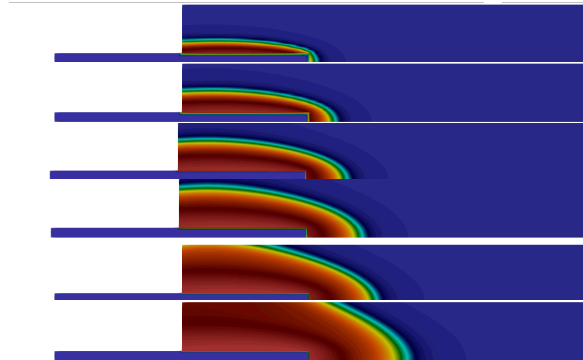
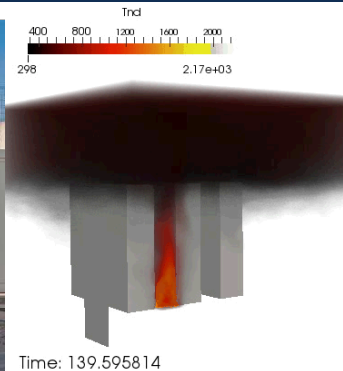


Thermal runaway in Li-ion battery packs



# Modeling the limits of thermal runaway in Li-ion packs and designing tests to measure those limits

John Hewson, Randy Shurtz

Summer Ferreira, Josh Lamb, Heather Barkholtz, Lorraine Torres-Castro

Sandia National Laboratories

JANNAF meeting  
December 4, 2017

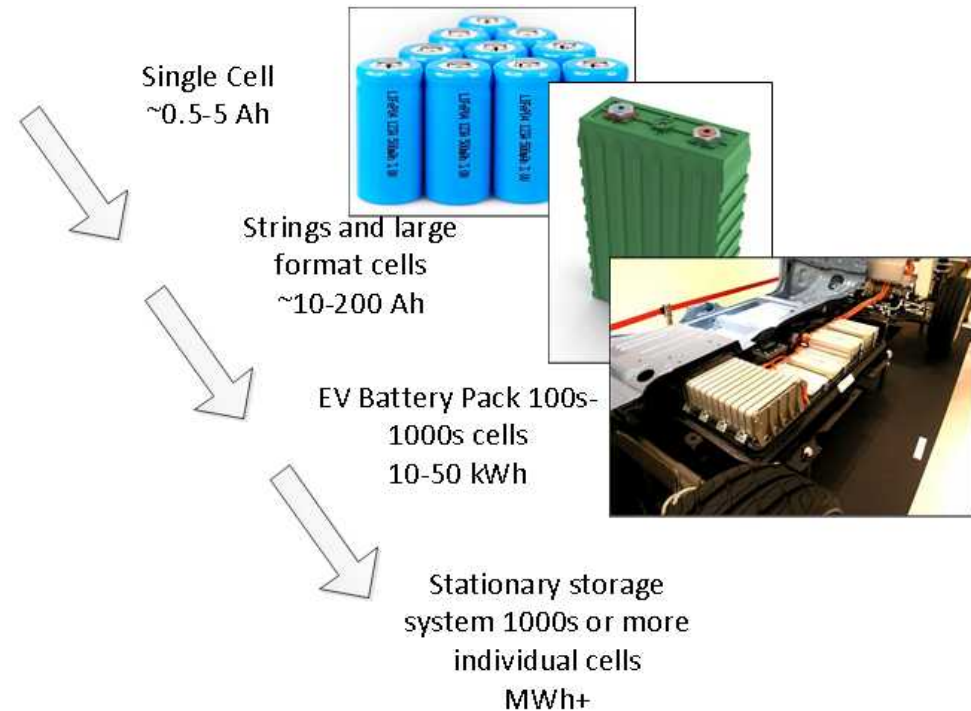
SAND2017-XXXXC



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# Validated reliability and safety is one of four critical challenges identified in 2013 Grid Energy Storage Strategic Plan

- Failure rates as low as 1 in several million,
- Potentially many cells used in energy storage.
- Moderate likelihood of 'something' going wrong,



[www.nissan.com](http://www.nissan.com)  
[www.internationalbattery.com](http://www.internationalbattery.com)  
[www.samsung.com](http://www.samsung.com)  
[www.saft.com](http://www.saft.com)

■ **How do we decrease the risk?**

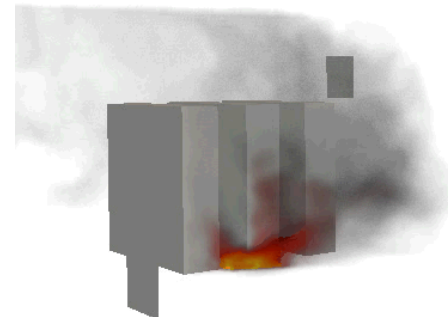
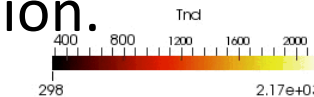
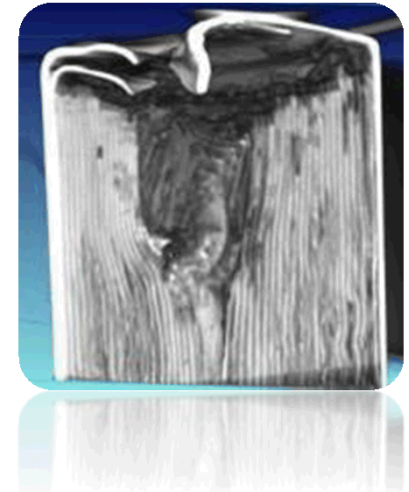
# Approaches to designing in safety

The current approach is to test our way into safety<sup>1</sup>

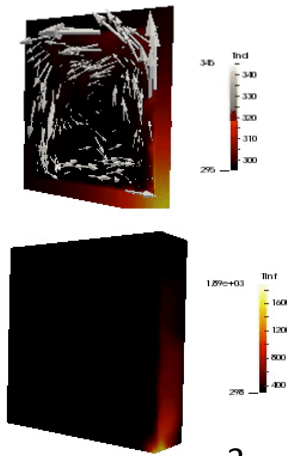
- Large system (>1MWh) testing is difficult and costly.

Consider supplementing testing with predictions of challenging scenarios and optimization of mitigation.

- Develop multi-physics models to predict failure mechanisms and identify mitigation.
- Build capabilities with small/medium scale measurements.
- Still requires some testing and validation.



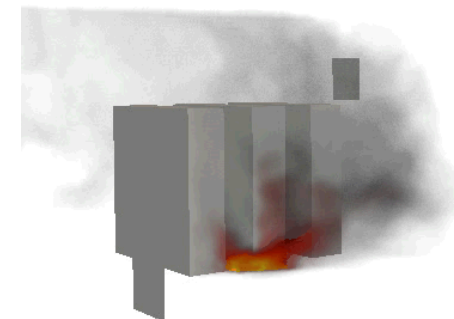
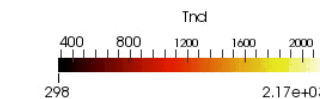
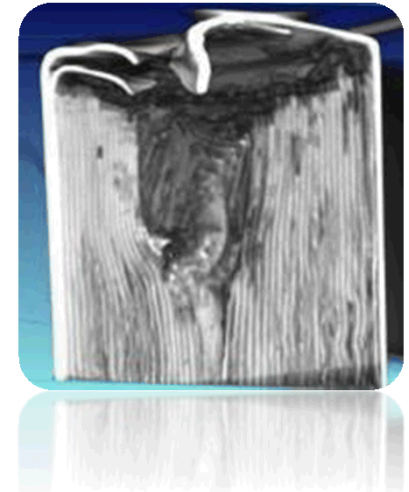
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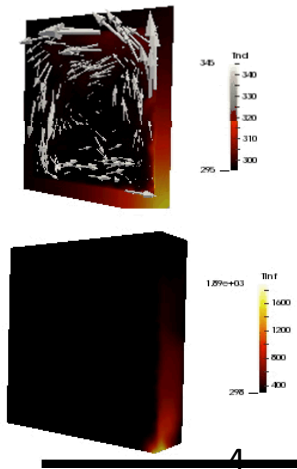
<sup>1</sup> 'Power Grid Energy Storage Testing Part 1.' Blume, P.; Lindenmuth, K.; Murray, J. EE – Evaluation Engineering. Nov. 2012.

# A range of risks to energy storage systems

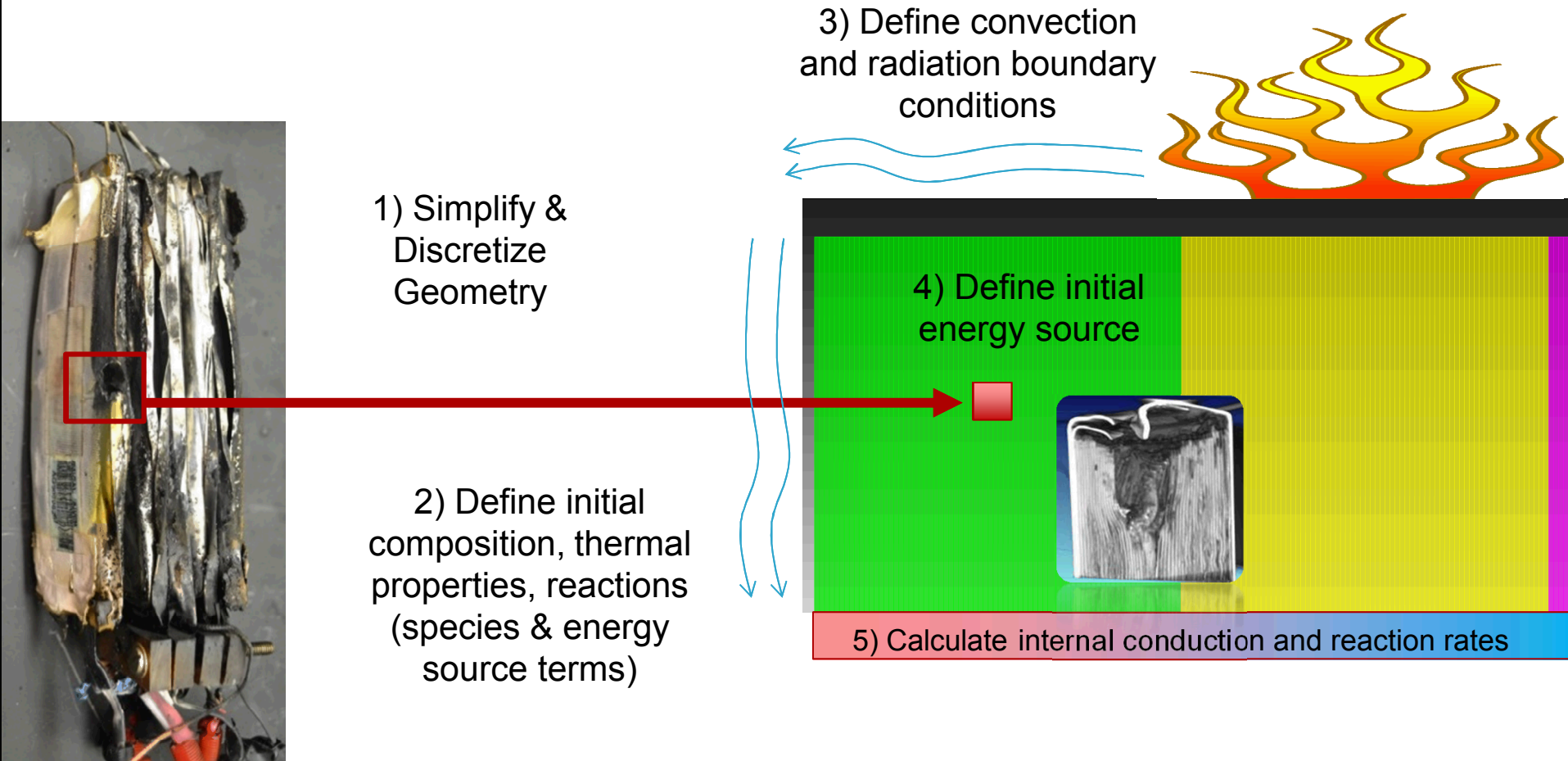
- Internal short circuit
  - Soft short—gradual discharge.
  - Hard short—sudden discharge.
- Mechanical damage,
- External heating or internal overheating.
- Overcharge.



Time: 46.683046



# Models for thermal runaway processes



# Models Need Parameters

- Preliminary chemistry model from literature
  - Based on Dahn group from 2000, 2001
  - Derived from calorimetry data (ARC and DSC)
  - Needs to be recalibrated—not expressed in fundamental cell characteristics.
- Empirical chemical reactions
 

- SEI decomposition  $2 \text{ROCO}_2\text{Li} \rightarrow \text{Li}_2\text{CO}_3 + \text{prod}$
  - Cathode-electrolyte  $\text{CoO}_2 + \text{C}_3\text{H}_4\text{O}_3 \rightarrow \frac{1}{3}\text{Co}_3\text{O}_4 + \text{prod}$
  - Electrolyte-salt  $\text{C}_3\text{H}_4\text{O}_3 + \text{LiPF}_6 \rightarrow \text{prod}$
  - Anode-electrolyte  $\text{C}_6\text{Li} + \text{C}_3\text{H}_4\text{O}_3 \rightarrow \text{Li}_2\text{CO}_3 + \text{prod}$
- This model form has been utilized repeatedly for oven-test runaway prediction.

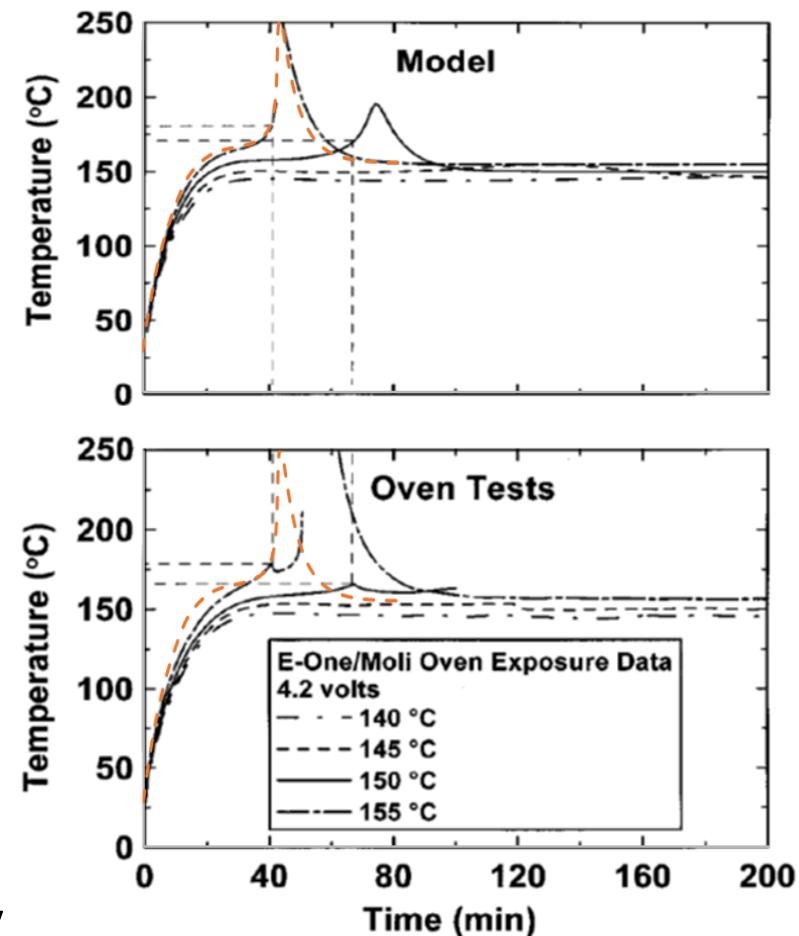


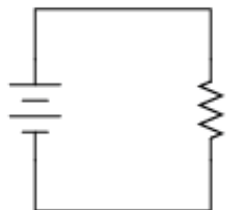
Figure 2. A comparison of oven exposure test results to model predictions: (top) model predictions and (bottom) oven test results for 18650 E-One/Moli Energy cells charged to 4.2 V.



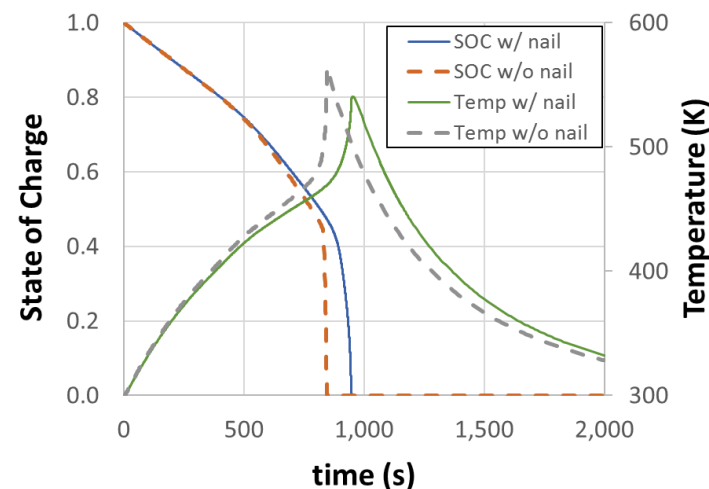
# Thermal and electrochemical reactants are same

- Discharge reaction:  $C_6Li + CoO_2 \rightarrow C_6 + LiCoO_2$
- Charged materials are reactive.

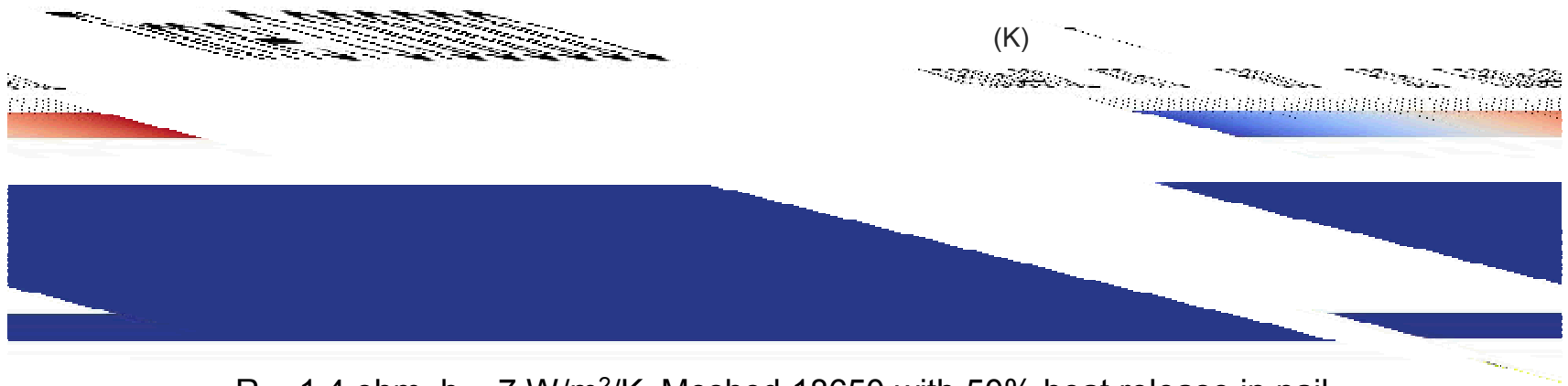
- SEI decomposition  $2 ROCO_2Li \rightarrow Li_2CO_3 + \text{prod}$
- Cathode-electrolyte  $CoO_2 + C_3H_4O_3 \rightarrow \frac{1}{3}Co_3O_4 + \text{prod}$
- Electrolyte-salt  $C_3H_4O_3 + LiPF_6 \rightarrow \text{prod}$
- Anode-electrolyte  $C_6Li + C_3H_4O_3 \rightarrow Li_2CO_3 + \text{prod}$



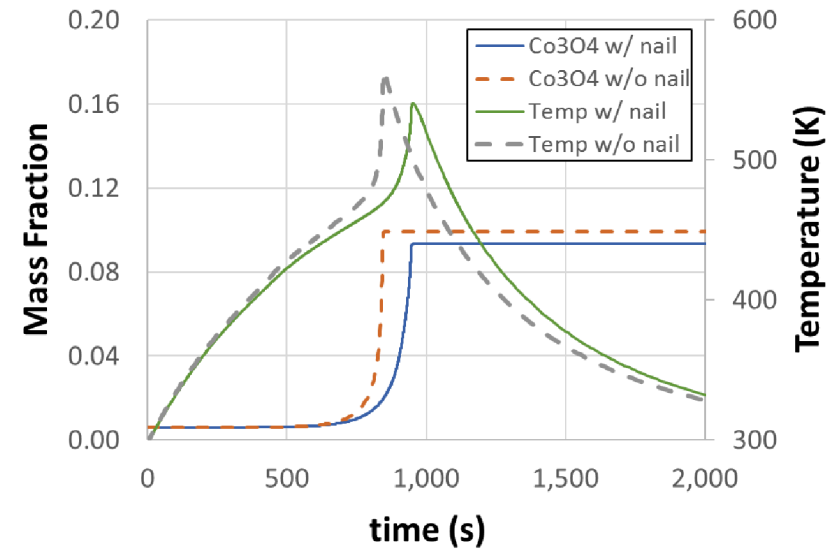
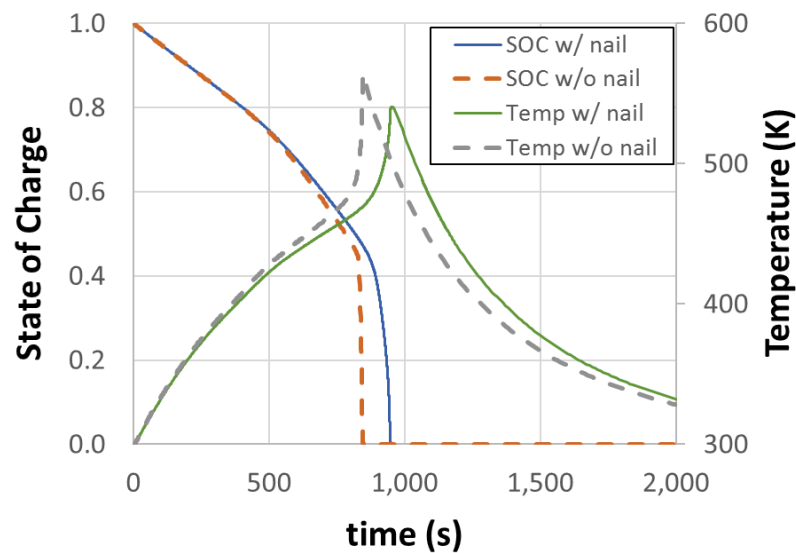
Simulated short circuit  
plus thermal runaway



# Short-circuit induced runaway in meshed 18650 with nail



$R = 1.4 \text{ ohm}$ ,  $h = 7 \text{ W/m}^2/\text{K}$ , Meshed 18650 with 50% heat release in nail



Effects of inhomogeneity increase as scale increases beyond the lumped-capacitance regime.

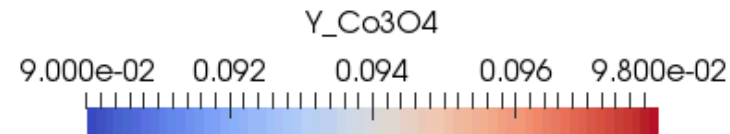


# Relative importance of short-circuit versus thermal reactions

$R = 1.4 \text{ ohm}$ ,  $h = 7 \text{ W/m}^2/\text{K}$ , Meshed 18650 with 50% heat release in nail

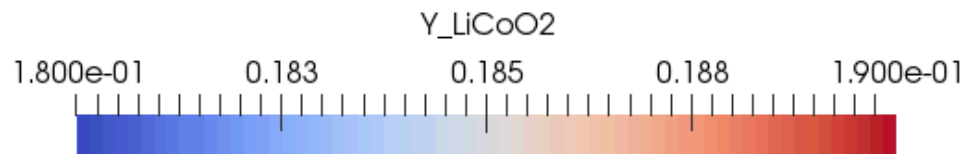
Time: 1004.759876

## Thermal Reaction Cathode Product



Time: 1004.759876

## Short Circuit Cathode Product

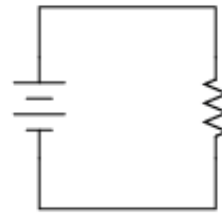


# Energetics of thermal runaway in high-temperature environments and under internal short-circuits

## Simulated oven test



high  
temp  
environ

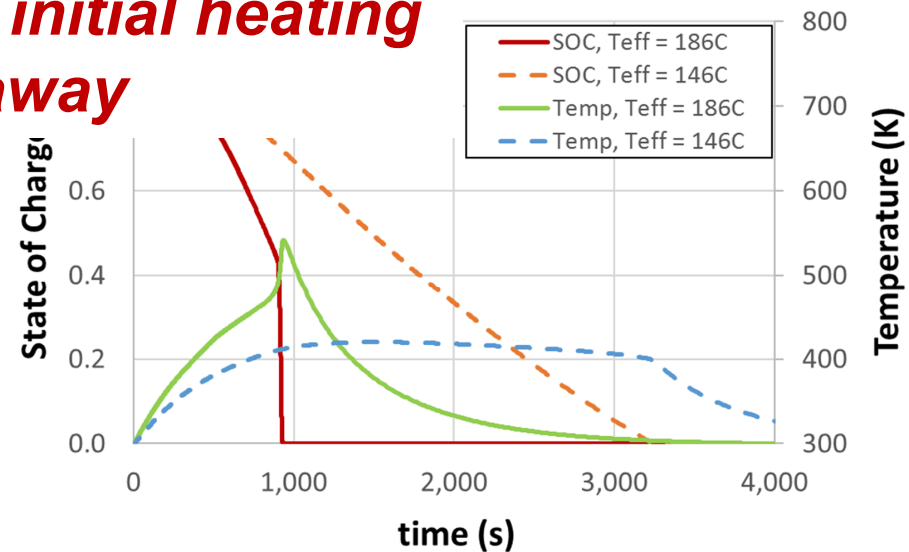
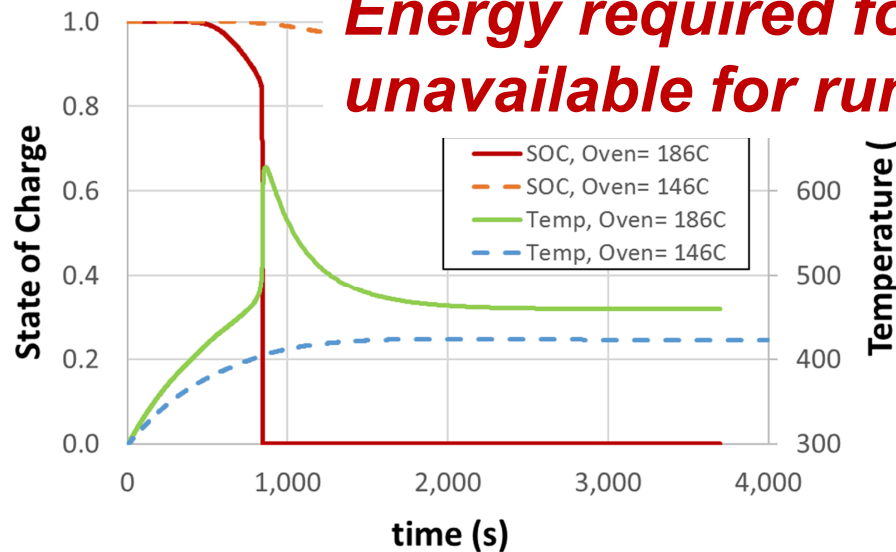


## Simulated short circuit analogy with oven temperature

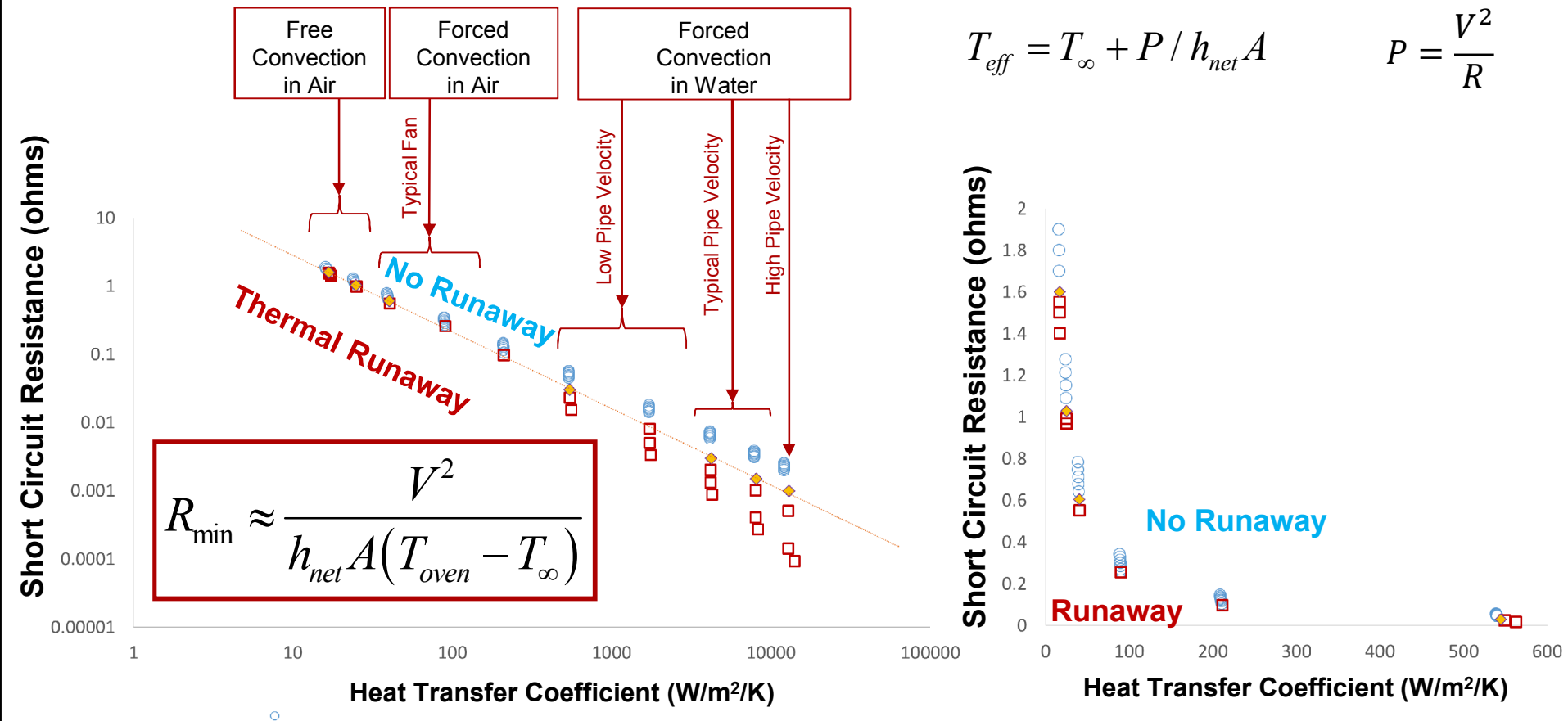
$$P = \frac{V^2}{R}$$

$$T_{eff} = T_{\infty} + P / h_{net} A$$

**Energy required for initial heating  
unavailable for runaway**

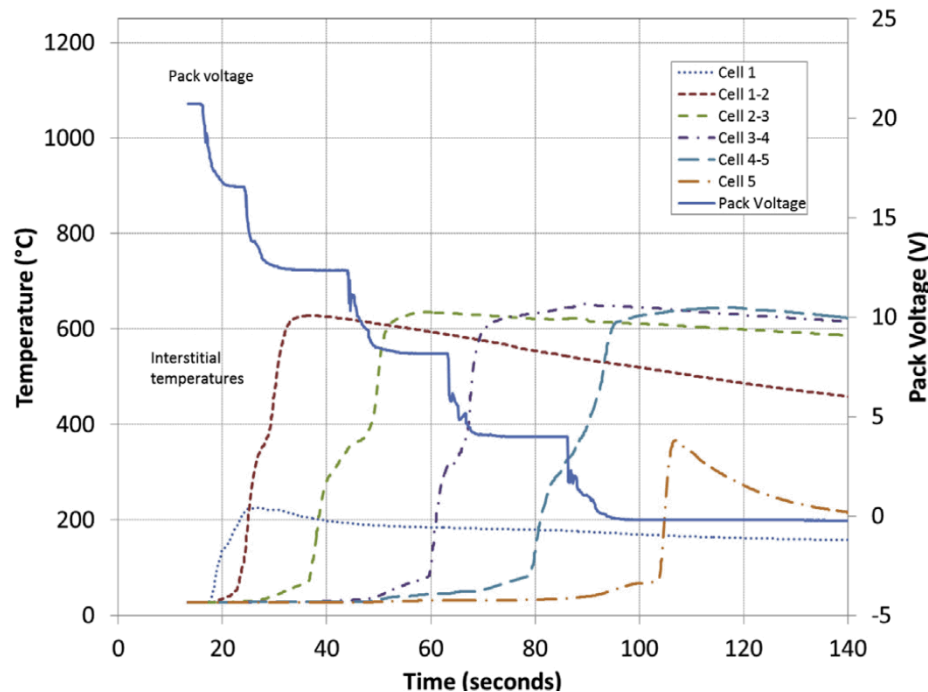


# How Much Cooling to Suppress Runaway with Internal Short Circuit?

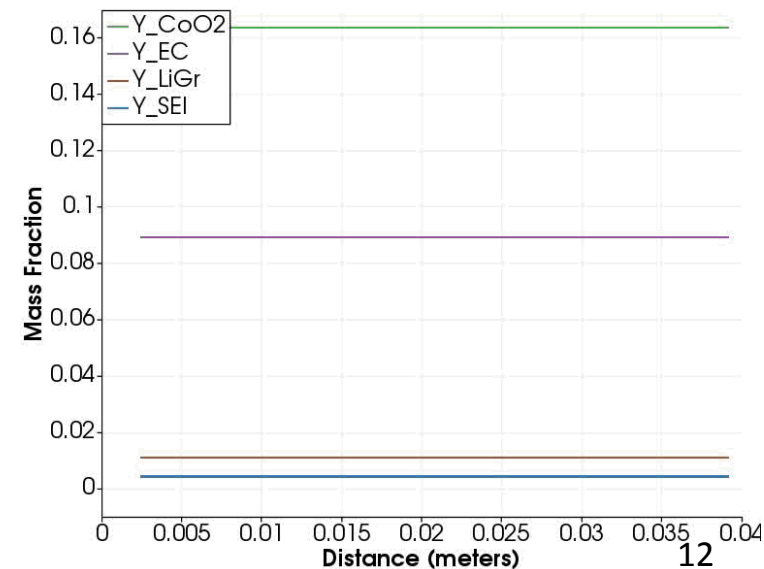
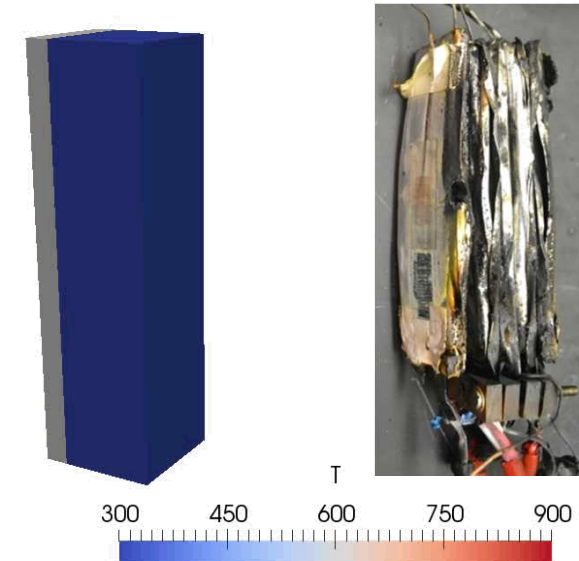


- Models can be used to estimate cooling requirements
  - Simulation shows homogeneous heating of 18650 cells (varying short resistance and cooling)
  - Internal temperature variation will be worse for large format systems and localized shorts

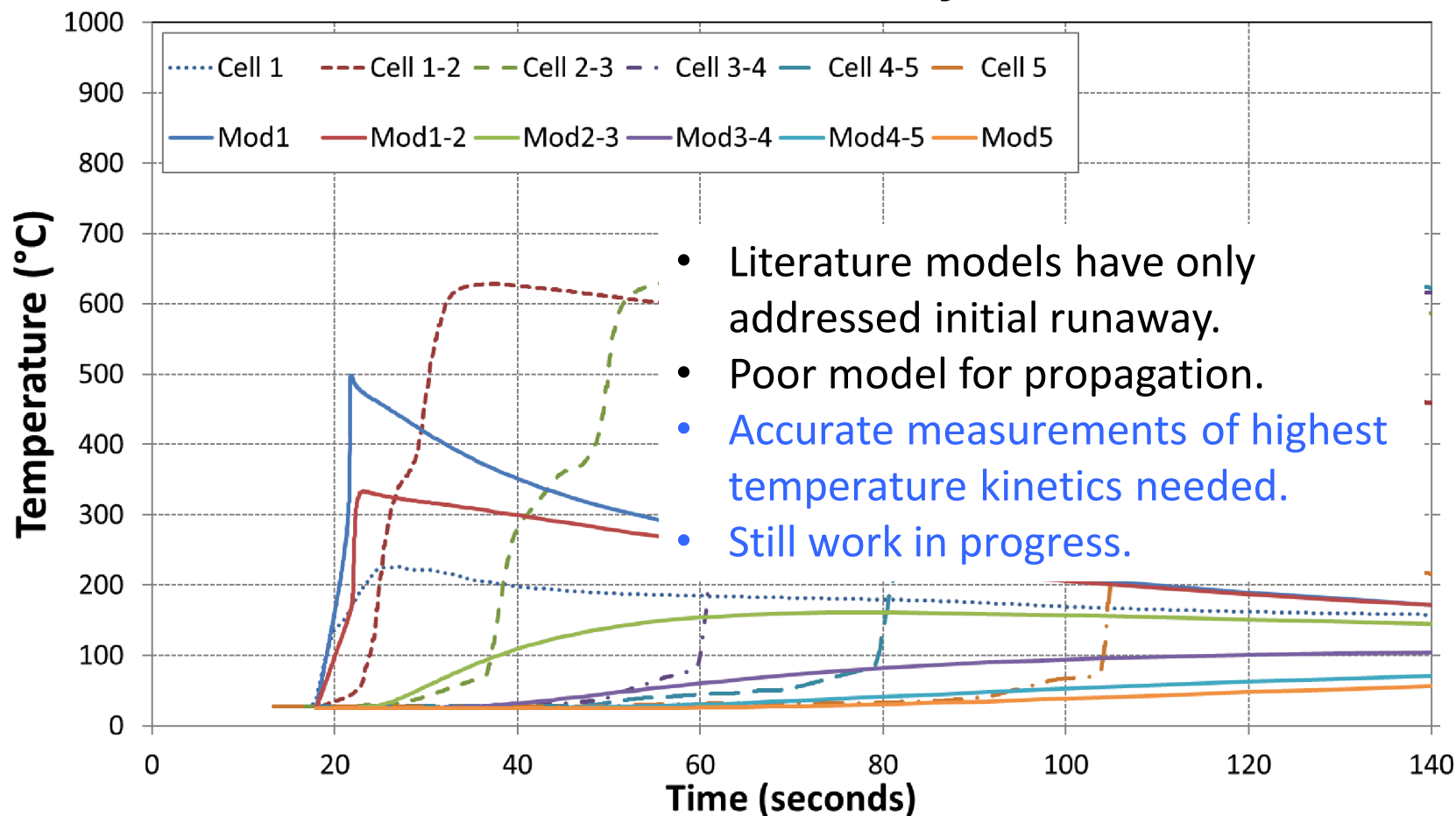
# Cascading Propagation Observed in Li-Ion Packs



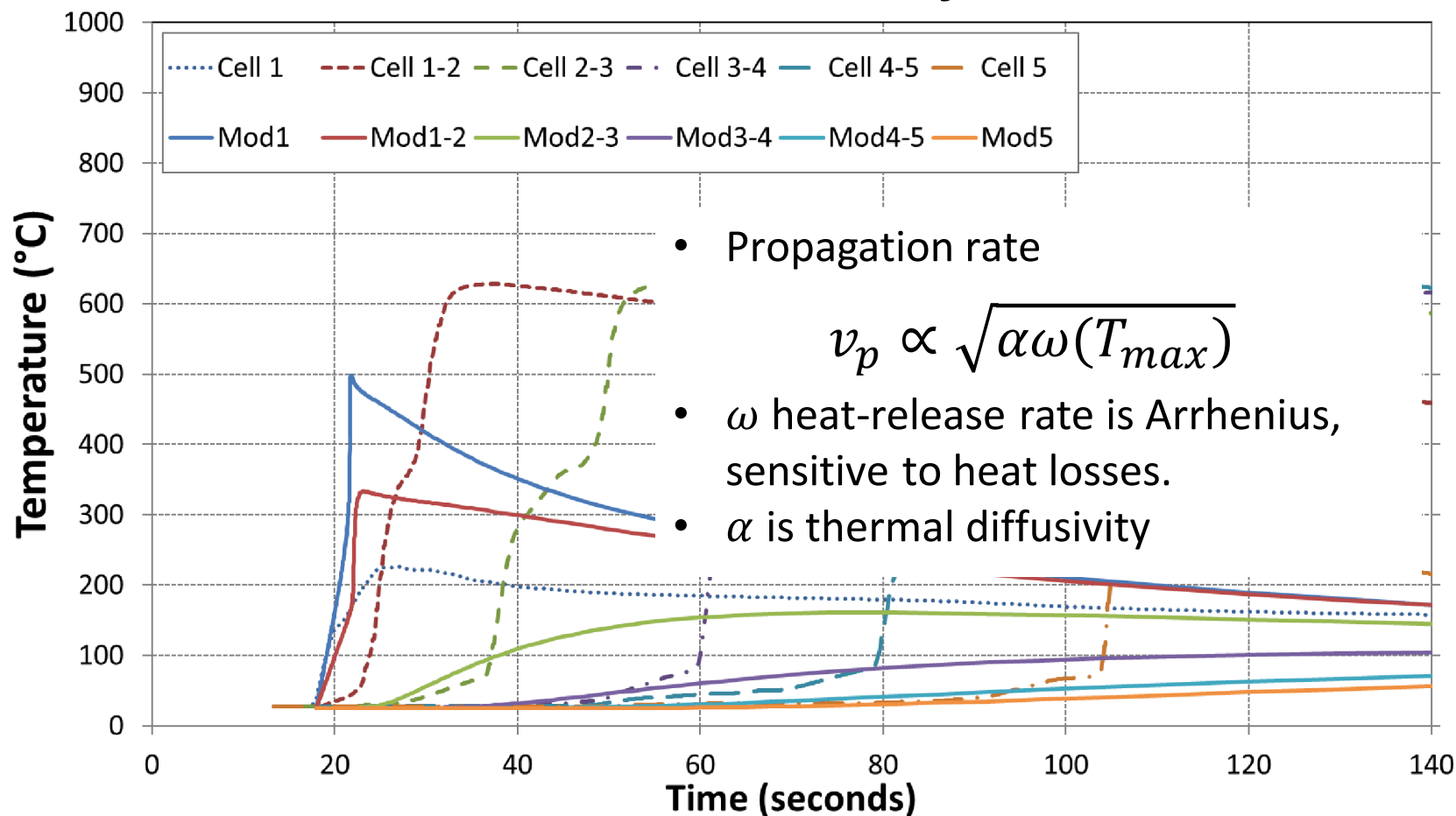
- Experimental propagation in 5 stacked pouch cells at Sandia
- Investigating effects of
  - State of charge
  - Intermediate layers
  - System geometry
- Good pack-scale model validation cases



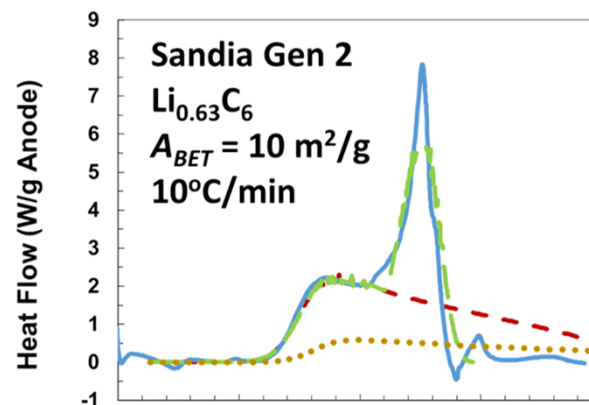
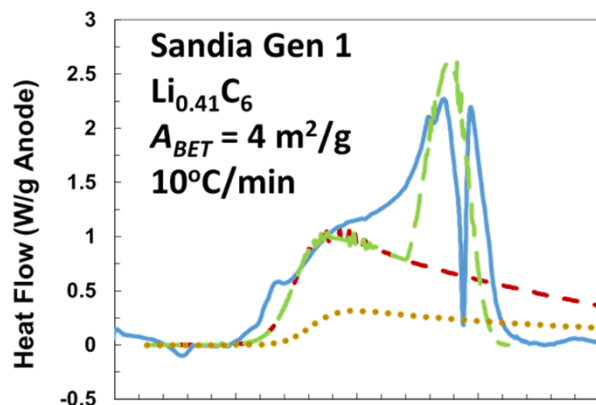
## Literature Chemistry Model



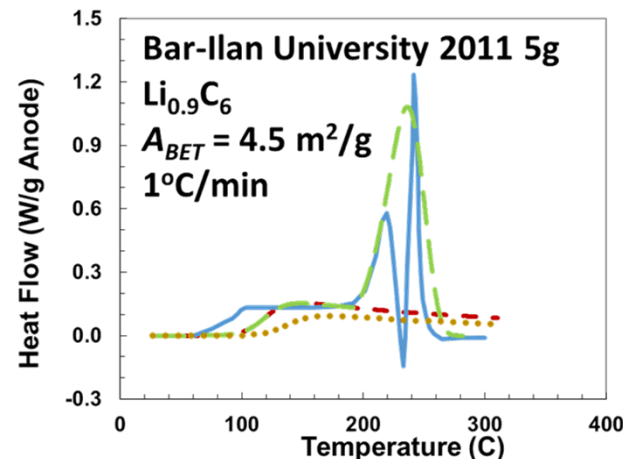
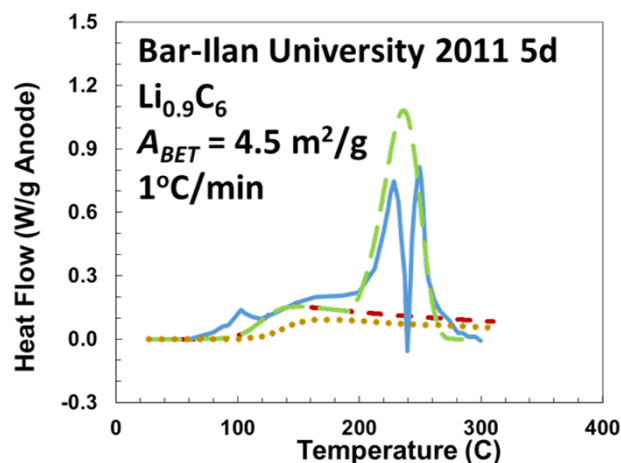
## Literature Chemistry Model



# New model based on measureable quantities and thermodynamic material properties

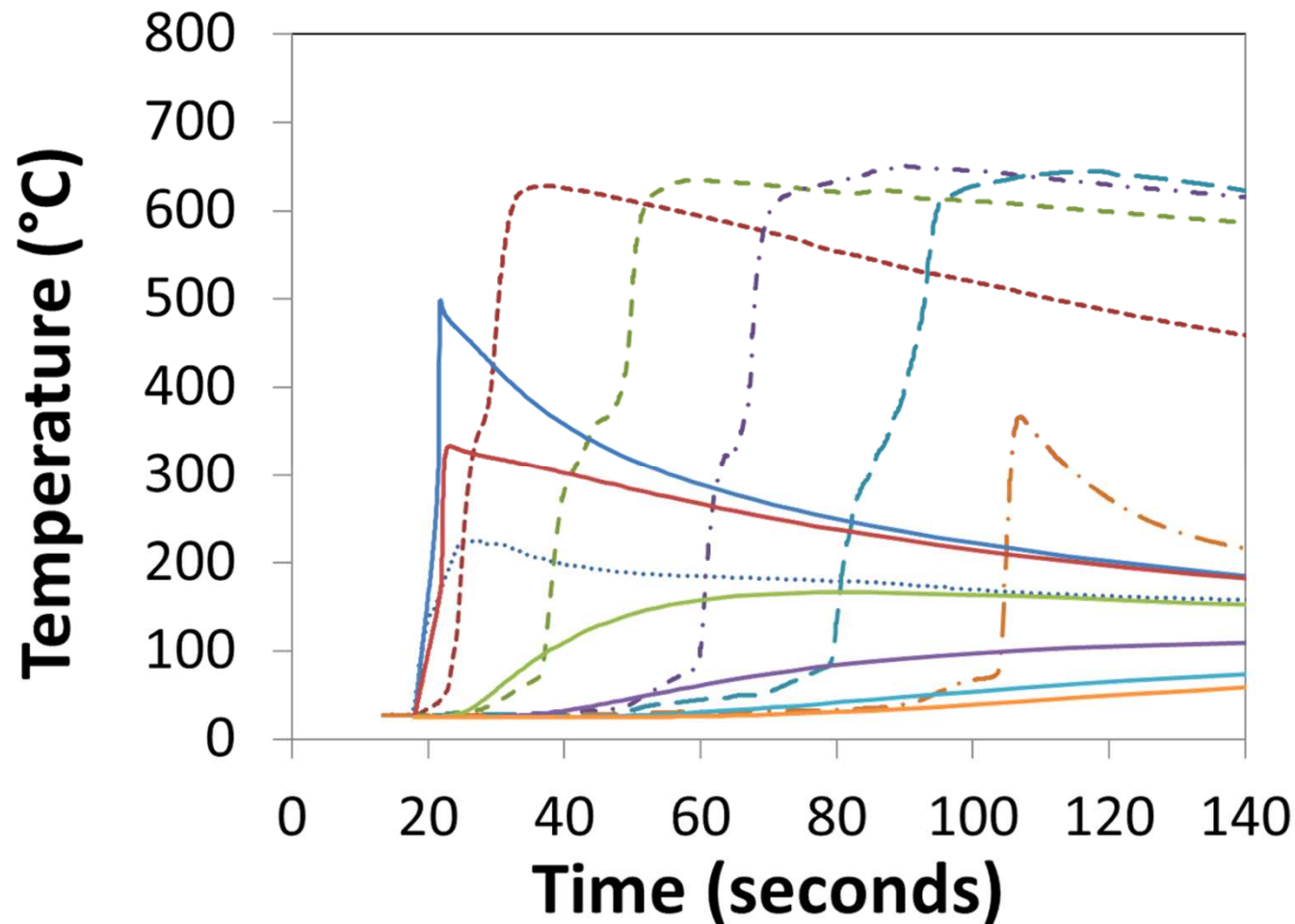


**Data**   **Dahn Model**   **Area-Scaled**   **Critical Thickness**

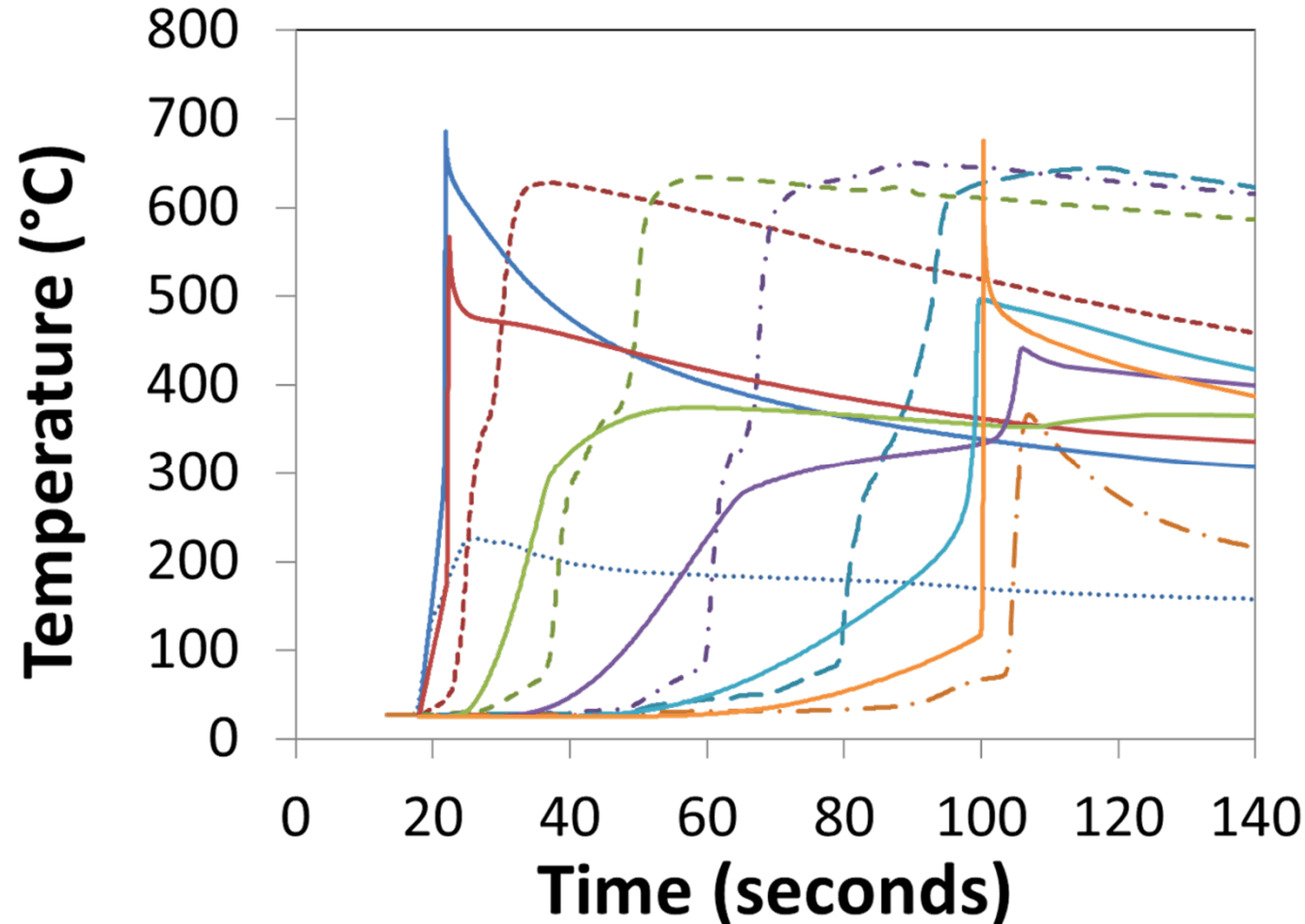




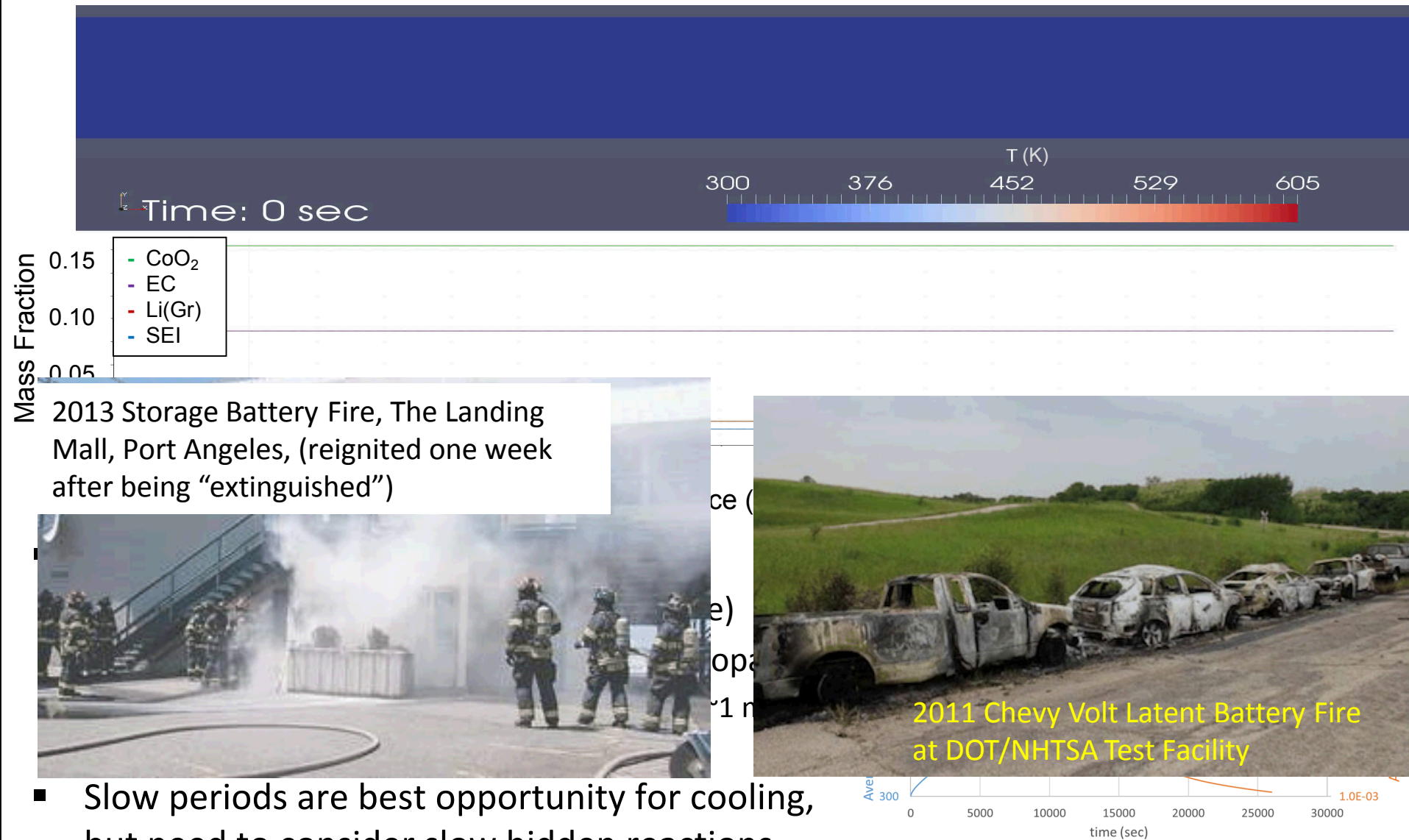
## Dahn Model (Hatchard et al. 2001)



## Dahn + New Anode Model, Alternate Anode Parameters



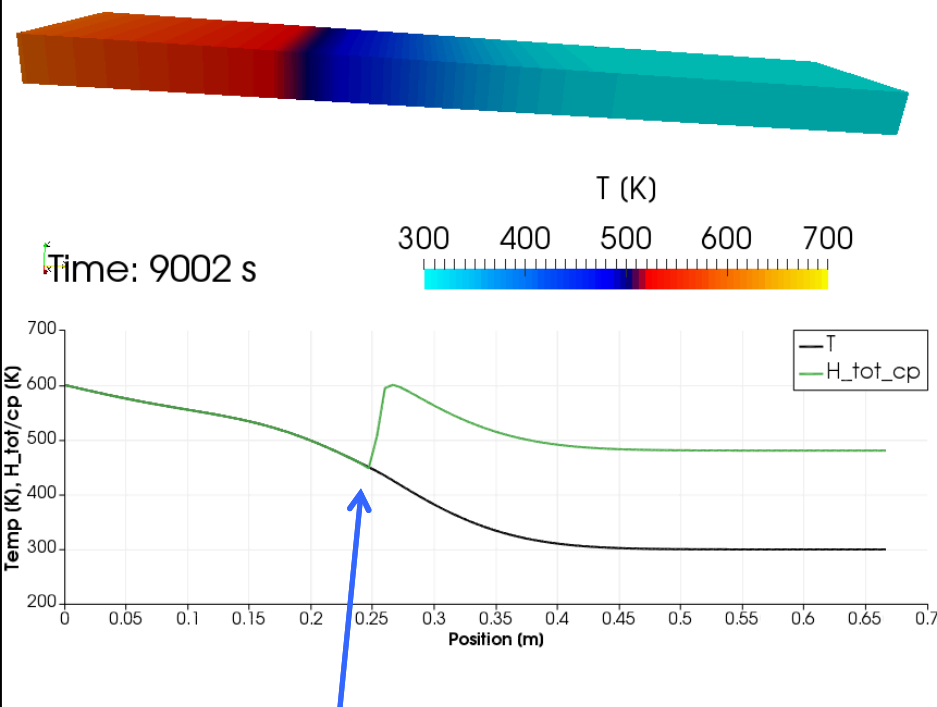
# Extend modeling to Large Scales – Pulsating propagation



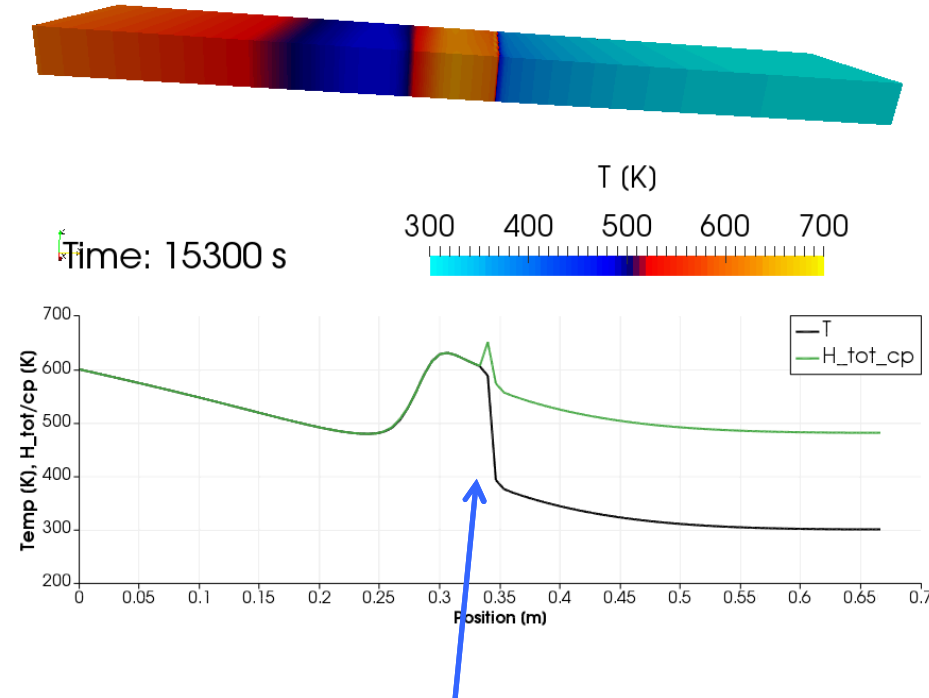
- Slow periods are best opportunity for cooling, but need to consider slow hidden reactions.

# The mechanism of pulsating propagation

- Heat released is conducted upstream of reaction front, increasing the total enthalpy (sum of sensible and chemical enthalpy)  $H_{TOT} = c_p T + Y_r \Delta H_r$
- Front propagates rapidly through preheated region with larger  $H_{TOT}$ .

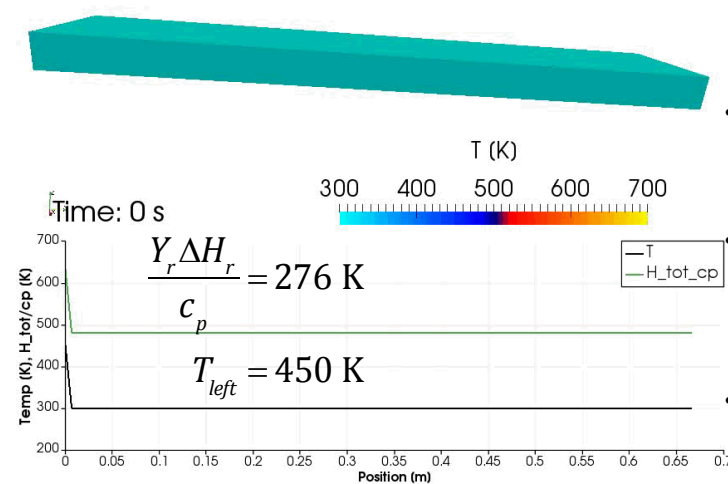
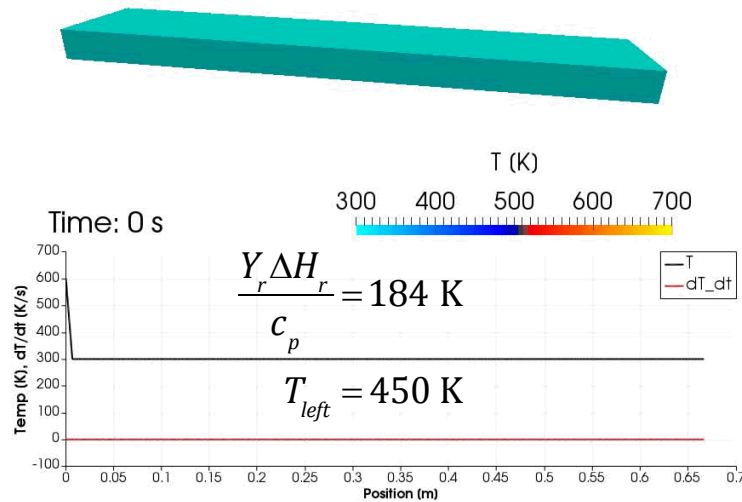


- Slow propagation (low Temp), but preheating mixture ahead of reaction front.

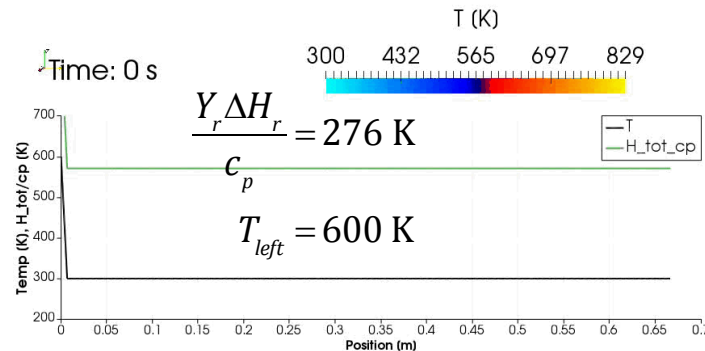
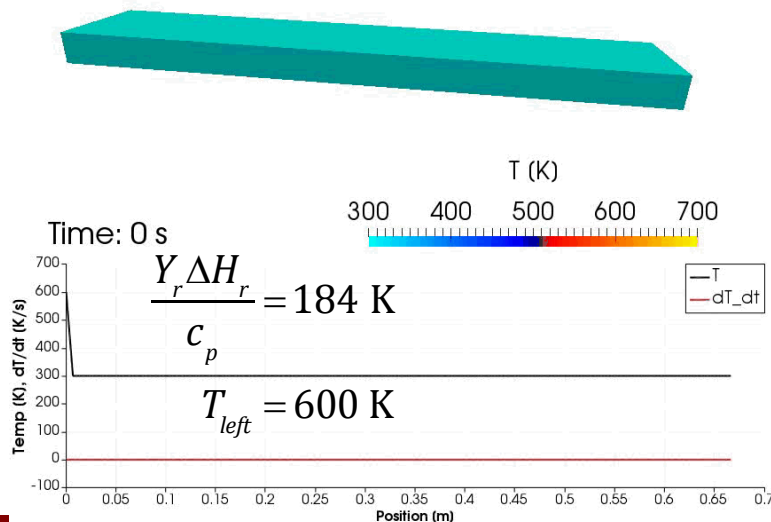


- Rapid propagation (high Temp), into preheated mixture.

# Parameter studies of propagation at large scales are possible with models

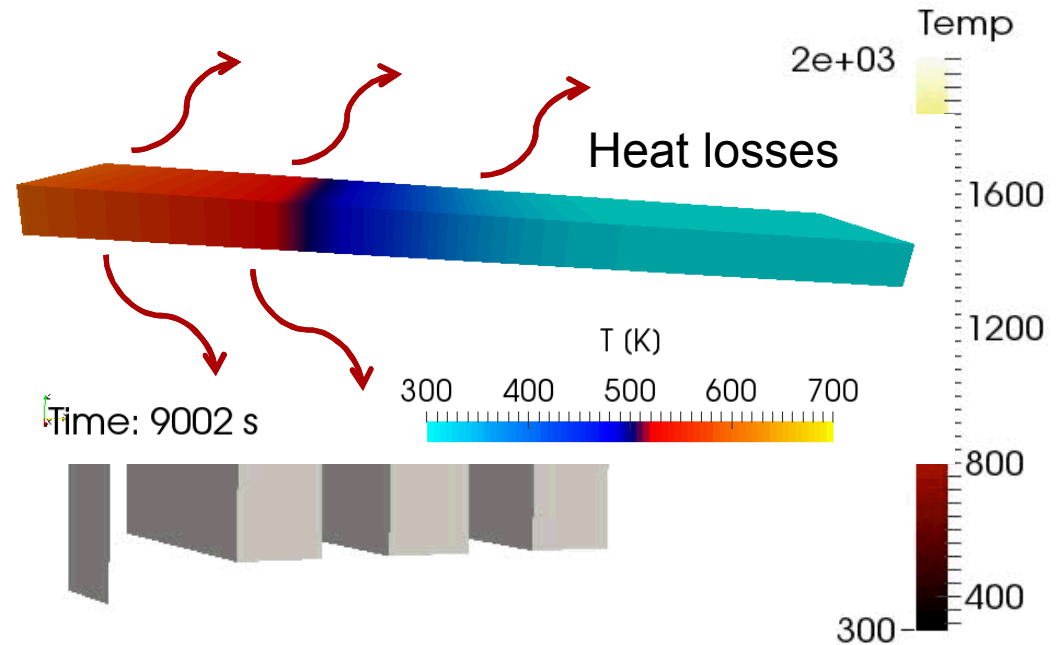


- Prediction and mitigation of cell-to-cell propagation is key to addressing risk.
- Single-step heat-release predictions with a range of heat release and boundary temps.
- Propagation across a large pack (80 cells here) exhibits pulsating instabilities.



# External risks to energy storage systems

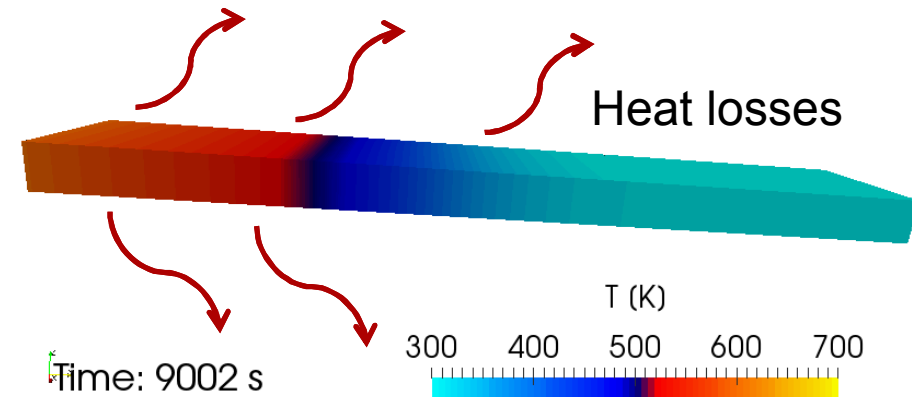
- External fires:
  - Relatively low heat flux.
  - Installation integrated suppression.



Time: 0.000000

# Looking forward

- Thermal management can be an effective approach to managing thermal runaway risks.
- Heat capacity can inhibit ignition and propagation.
- Moderate cooling can quench propagation, and sometimes ignition.
- Simulations allow exploration of trade space if the physics are known.
  - But physics still must be observed and measured.
- → **Ultimate goal: *Predict criteria for cascading failure to act as a design tool in developing mitigation strategies.***





- Thermal runaway is a risk and potential barrier to development and acceptance.
- Heat release rates are moderate relative to potential dissipation.
- Multi-physics thermal models can potentially identify critical ignition and propagation trends.
- Quality measurements are key to parameter identification.
- Recent progress
  - Development of thermal source terms.
  - Identification of thermal ignition criterion.
  - Cell-to-cell propagation along homogenized pack structures.

# Acknowledgements

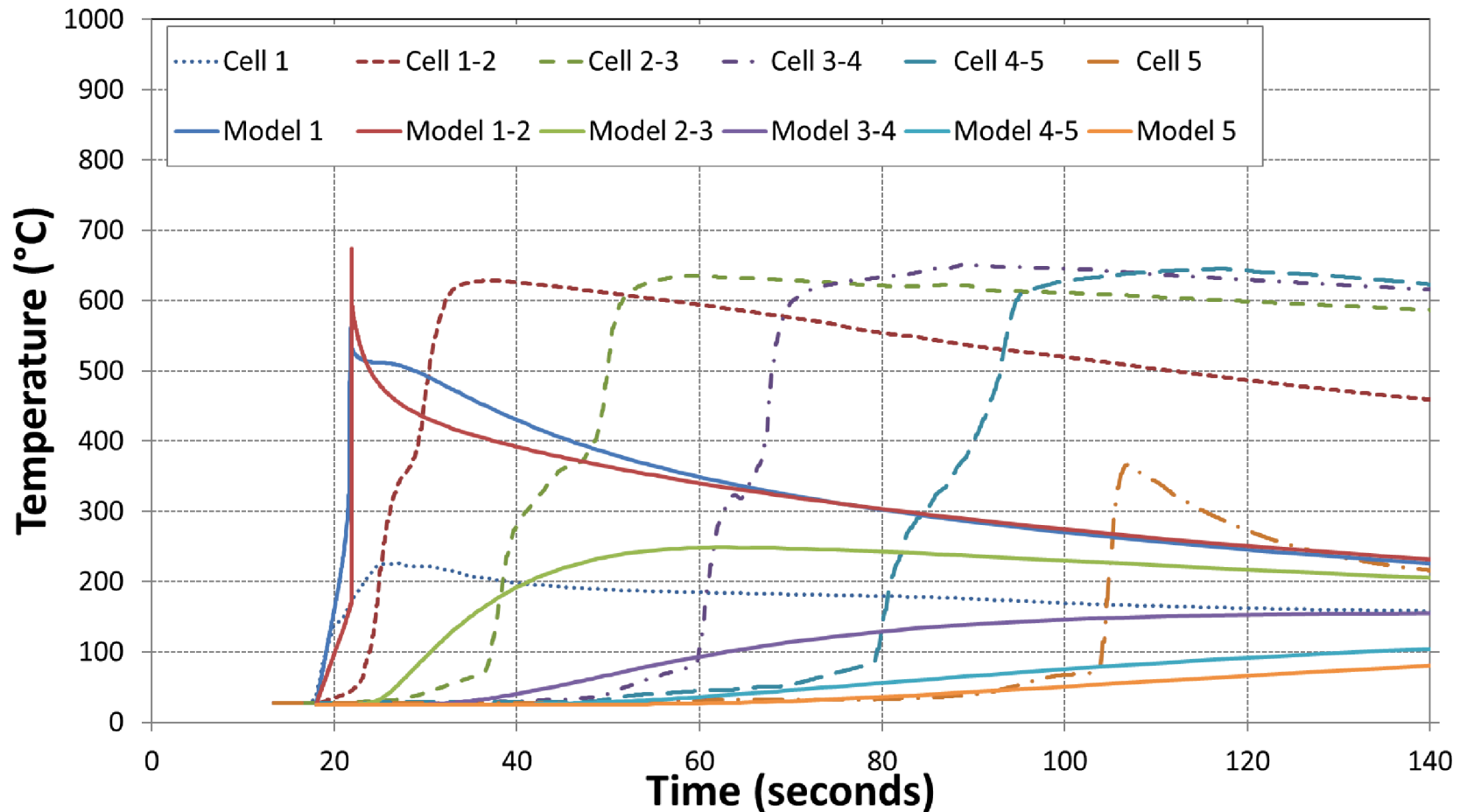
- Supported by Imre Gyuk and the OE Electrical Energy Storage Program.
- Collaborative discussions with Heather Barkholtz, Lorraine Castro, Summer Ferreira, Josh Lamb, Chris Orendorff and Dave Ingersoll have been instrumental in understanding the range of possible topics that might be addressed.

**THANK YOU**

**QUESTIONS:**  
**JOHN HEWSON**  
jchewso@sandia.gov



## Decrease high-temperature reaction rate by 2x again



- Propagation predictions will improve with fidelity of high-temperature chemistry