

Kinetic simulation of breakdown in gaps with and without dielectric particles

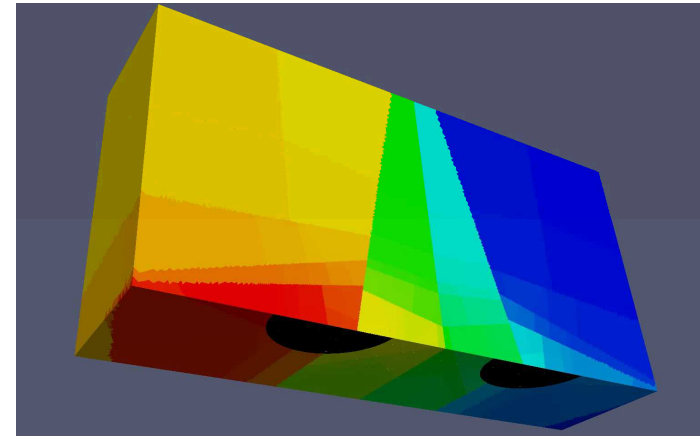
Chris Moore, Andrew Fierro, Roy Jorgenson, Ashish Jindal, Paul Clem, and Matthew Hopkins



Sandia National Laboratories is a multi-mission laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000. SAND NO. 2011-XXXXP

Aleph Overview

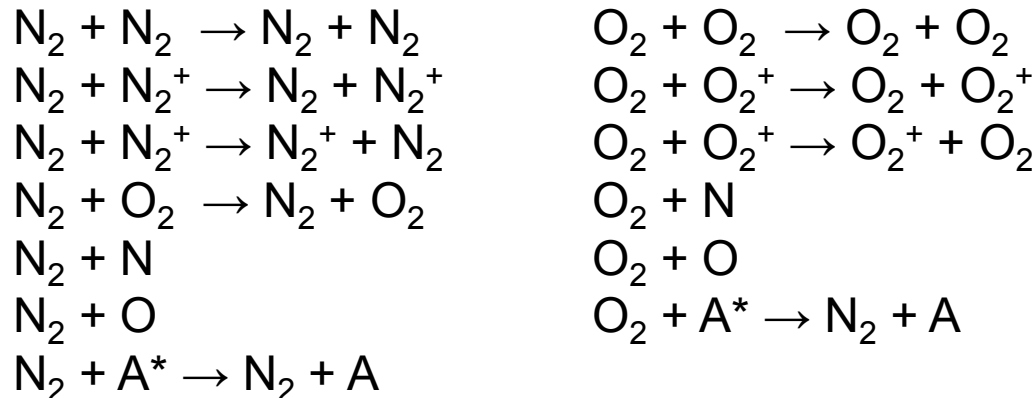
- Electrostatic Particle-In-Cell
 - Kinetic model of non-equilibrium plasma
 - 1, 2, or 3D unstructured FEM (CAD-compatible)
 - Accounts for relative permittivity of materials
- Massively parallel (scales up to ~60K procs)
 - Dynamic load balancing
- Surface physics models:
 - Fowler-Nordheim and thermionic e^- emission, sputtering, surface charging, auger-neutralization, SEE, photoemission, sublimation
- Direct Simulation Monte Carlo Collision physics:
 - Simulate all species as simulation particles with variable particle weights
 - Can simulate evolution of neutral gas densities (important at low pressures)
 - Elastic, charge exchange, chemistry (dissociation, exchange, etc.), excited states (w/ radiative decay & self-absorption), ionization, Coulomb collisions (Nanbu model)



Domain Decomposition

Air Chemistry: Model

- Lab cross sections vs. energy (linearly interpolate between data)
 - Experimental uncertainties of at **least** 20%
- Include elastic, charge exchange, and quenching heavy-heavy interactions. Assume N_2 and O_2 are dominant species.
 - Important for late-time thermalization of streamer channel & radiative emission intensity from excited states



- 3-body recombination and attachment included as rate eqns.

Air Chemistry: Model

- Include e- N_2 , O_2 , N, O, N_2^+ , and O_2^+ interactions
 - Elastic collisions: Preferentially forward scattered [1]
 - Ionization:
 - Single: $N_2 \rightarrow N_2^+$ and $O_2 \rightarrow O_2^+$
 - Double ionization: $N_2 \rightarrow N_2^{++}$ and $O_2 \rightarrow O_2^{++}$
 - Dissociative ionization: $N_2 \rightarrow N + N^+$ and $O_2 \rightarrow O + O^+$
 - Dissociation: $N_2 \rightarrow 2N$ and $O_2 \rightarrow 2O$
 - Attachment: $O_2 \rightarrow O + O^-$
 - Recombination: $N_2^+ \rightarrow 2N$ and $O_2^+ \rightarrow 2O$
 - Excitation of N_2 and O_2 vib. & rot. states (do not track them)
 - Electronic excitation:
 - 15 N_2 excited states
 - 6 O_2 excited states
 - 6 N excited states
 - 17 O excited states

Air Chemistry: Model

- Include e⁻, N₂, O₂, N, O, N₂⁺, and O₂⁺ interactions

- Elastic collisions: Preferentially forward scatter

- Ionization:

- Single: N₂ → N₂⁺ and O₂ → O₂⁺
- Double ionization: N₂ → N₂⁺⁺ and O₂ → O₂⁺⁺
- Dissociative ionization: N₂ → N + N⁺ and O₂ → O + O⁺

- Dissociation: N₂ → 2N and O₂ → 2O

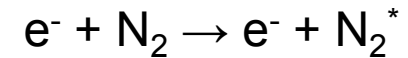
- Attachment: O₂ → O + O⁻

- Recombination: N₂⁺ → 2N and O₂⁺ → 2O

- Excitation of N₂ and O₂ vib. & rot. states (do not)

- Electronic excitation:

- 15 N₂ excited states
- 6 O₂ excited states
- 6 N excited states
- 17 O excited states



N ₂ [*]	E (eV)
A ³ Σ _u ⁺ , v=0-4	6.17
A ³ Σ _u ⁺ , v=5-9	7.00
B ³ Π _g	7.35
W ³ Δ _u	7.36
A ³ Σ _u ⁺ , v=10-v _{max}	7.80
B' ³ Σ _u ⁻	8.16
a' ¹ Σ _u ⁻	8.40
a ¹ Π _g	8.55
w ¹ Δ _u	8.89
C ³ Π _u	11.03
E ³ Σ _g ⁺	11.88
a'' ¹ Σ _g ⁺	12.26
b ¹ Π _u	12.50
c' ₄ ¹ Σ _u ⁺	12.85
b' ¹ Σ _u ⁺	12.94

Air Chemistry: Model

- Include e- N₂, O₂, N, O, N₂⁺, and O₂⁺ interactions

- Elastic collisions: Preferentially forward scattered [1]

- Ionization:

- Single: N₂ → N₂⁺ and O₂ → O₂⁺

- Double ionization: N₂ → N₂⁺⁺ and O₂ → O₂⁺⁺

- Dissociative ionization: N₂ → N + N⁺ and O₂ → O + O⁺

- Dissociation: N₂ → 2N and O₂ → 2O

- Attachment: O₂ → O + O⁻

- Recombination: N₂⁺ → 2N and O₂⁺ → 2O

- Excitation of N₂ and O₂ vib. & rot. states (do not track them)

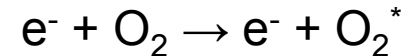
- Electronic excitation:

- 15 N₂ excited states

- 6 O₂ excited states

- 6 N excited states

- 17 O excited states



O ₂ [*]	E (eV)
a ¹ Δ _g	0.98
b ¹ Σ _g	1.63
c ¹ Σ _u ⁺	4.34
B ³ Σ _u	6.12
"Longest band"	9.97
"Second band"	10.29

Air Chemistry: Model

- Include e- N₂, O₂, N, O, N₂⁺, and O₂⁺ interactions

- Elastic collisions: Preferentially forward scattered [1]

- Ionization:

- Single: N₂ → N₂⁺ and O₂ → O₂⁺

- Double ionization: N₂ → N₂⁺⁺ and O₂ → O₂⁺⁺

- Dissociative ionization: N₂ → N + N⁺ and O₂ → O + O⁺

- Dissociation: N₂ → 2N and O₂ → 2O

- Attachment: O₂ → O + O⁻

- Recombination: N₂⁺ → 2N and O₂⁺ → 2O

- Excitation of N₂ and O₂ vib. & rot. states (do not track them)

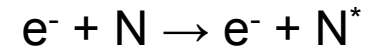
- Electronic excitation:

- 15 N₂ excited states

- 6 O₂ excited states

- 6 N excited states

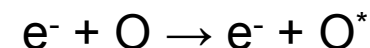
- 17 O excited states



N*	E (eV)
2s ² 2p ³ 2D ^o	2.38
2s ² 2p ³ 2P ^o	3.58
2s ² 2p ² (³ P)3s 4P	10.33
2s ² 2p ² (³ P)3s 2P	10.68
2s2p ⁴	10.98
2s ² 2p ² (¹ D)3s 2D	12.35

Air Chemistry: Model

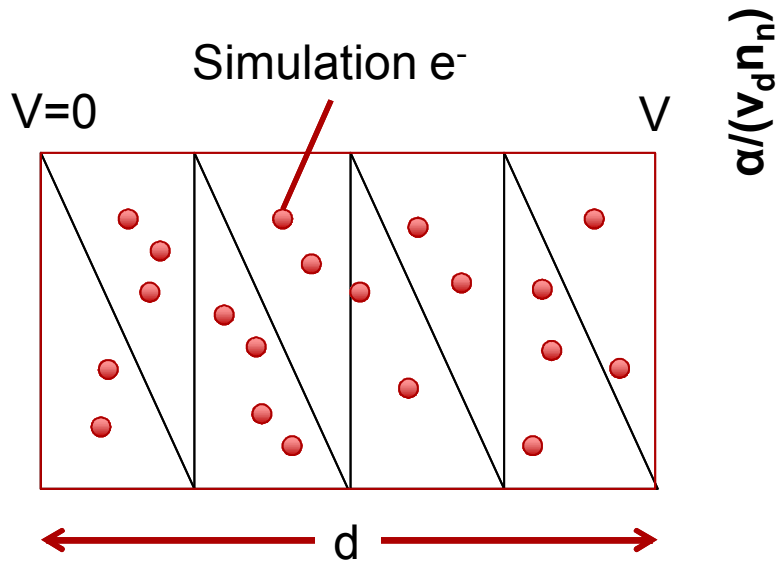
- Include e^- N_2 , O_2 , N , O , N_2^+ , and O_2^+ interactions
 - Elastic collisions: Preferentially forward scattering
 - Ionization:
 - Single: $N_2 \rightarrow N_2^+$ and $O_2 \rightarrow O_2^+$
 - Double ionization: $N_2 \rightarrow N_2^{++}$ and $O_2 \rightarrow O_2^{++}$
 - Dissociative ionization: $N_2 \rightarrow N + N^+$ and $O_2 \rightarrow O + O^+$
 - Dissociation: $N_2 \rightarrow 2N$ and $O_2 \rightarrow 2O$
 - Attachment: $O_2 \rightarrow O + O^-$
 - Recombination: $N_2^+ \rightarrow 2N$ and $O_2^+ \rightarrow 2O$
 - Excitation of N_2 and O_2 vib. & rot. states (do not include electronic excitation)
 - Electronic excitation:
 - 15 N_2 excited states
 - 6 O_2 excited states
 - 6 N excited states
 - 17 O excited states



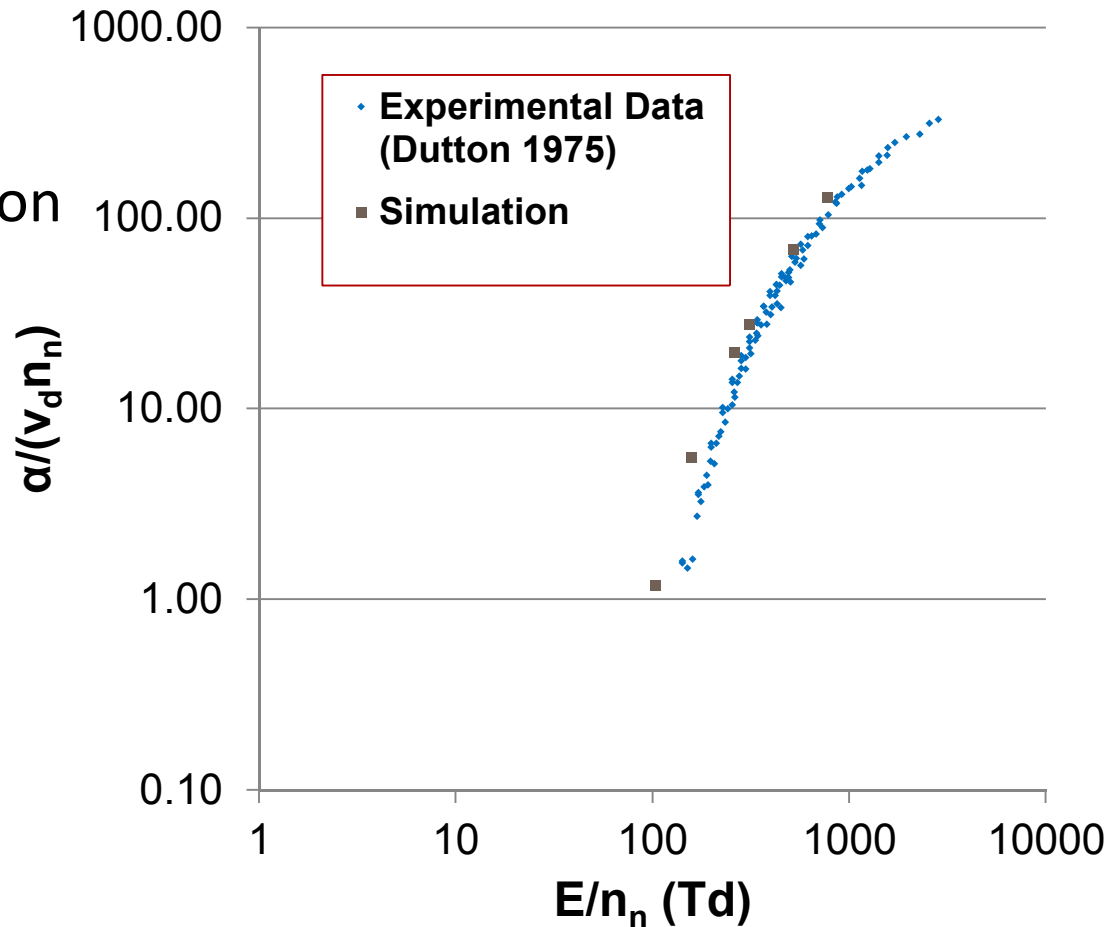
O^*	E (eV)
$2s^2 2p^4 \ ^1D$	1.97
$2s^2 2p^4 \ ^1S$	4.19
$2s^2 2p^3 (4S^o) 3s \ ^5S^o$	9.15
$2s^2 2p^3 (4S^o) 3s \ ^3S^o$	9.52
$2s^2 2p^3 (4S^o) 3p \ ^5P$	10.74
$2s^2 2p^3 (4S^o) 3s \ ^3P$	10.99
$2s^2 2p^3 (4S^o) 4s \ ^3S^o$	11.93
$2s^2 2p^3 (4S^o) 3d \ ^3D^o$	12.09
$2s^2 2p^3 (4S^o) 4p \ ^3P$	12.36
$2s^2 2p^3 (2D^o) 3s \ ^3D^o$	12.54
$2s^2 2p^3 (4S^o) 4d \ ^3D^o$	12.76
$2s^2 2p^3 (4S^o) 5d \ ^3D^o$	13.07
$2s^2 2p^3 (4S^o) 6d \ ^3D^o$	13.24
$2s^2 2p^3 (4S^o) 7d \ ^3D^o$	13.34
$2s^2 2p^3 (2P^o) 3s \ ^3P^o$	14.12
$2s 2p^5 \ ^3P^o$	15.66
$2s^2 2p^3 (2D^o) 4d \ ^3P^o$	16.11

Air Chemistry: Swarm data

- Compare experimental and simulated electron drift velocity and ionization coefficient vs. E/n



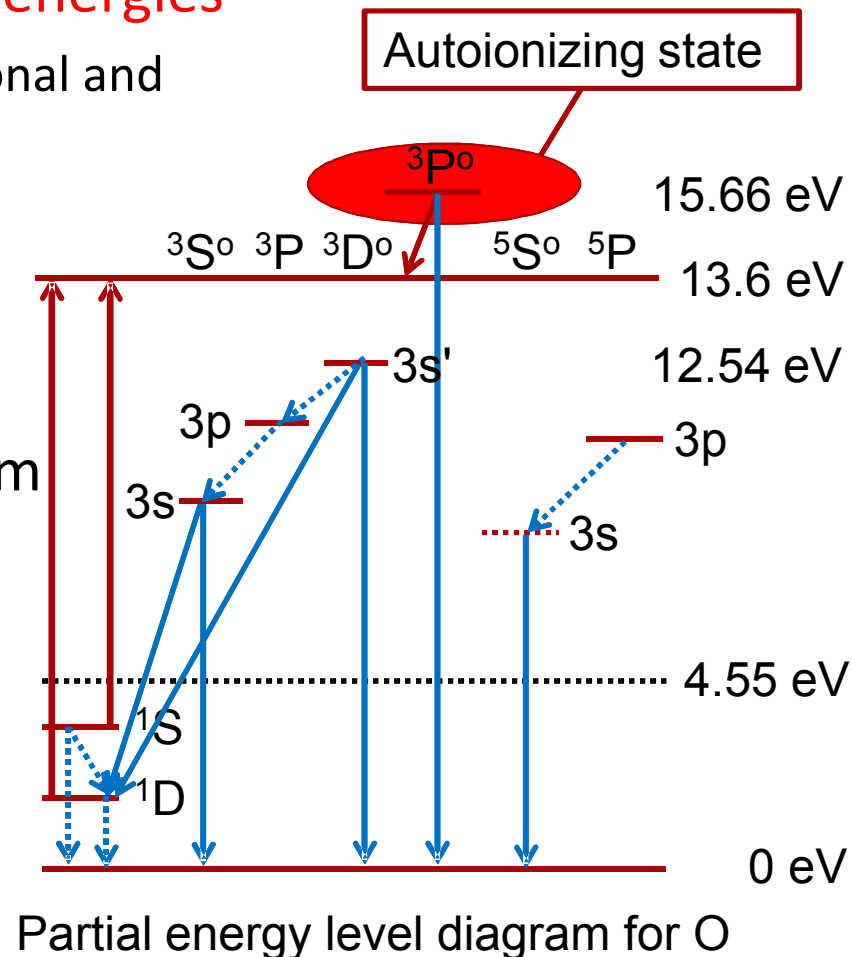
- Electrons undergo collisions with background air and gain energy in const. E-field



Convergence required
 $\Delta t < 5 \times 10^{-14} \text{ s} \ll v_{\text{col}}^{-1} < \omega_{\text{pe}}^{-1}$
 $\Delta x * E < 1 \text{ V} \ll \lambda_{\text{MFP}}$

Photons: Spontaneous Emission

- Model excited states of N_2 , O_2 , N, and O due to electron impact
- Assume photon interaction cross sections vary little over rotational and vibrational energies
 - Group all excited molecules into rotational and vibrational ground state ($v=0$, $j=0$).
- Allow spontaneous emission to multiple lower states based on Einstein-A coefficients
- Emit multiple sim. photons in random directions per emitting particle
 - Reduces numerical streamer branching due to sim. noise in particle count
- Include electron de-excitation (quenching) and ionization from metastable excited states



Photons: Neutral Interactions

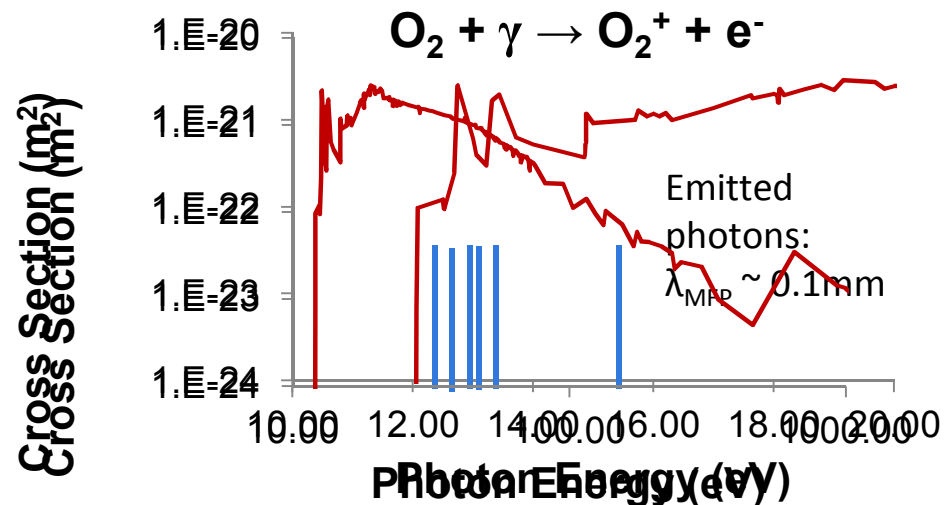
- Photons are simulation particles
 - During push: Velocity is constant, but spatial position updated
 - During collide: Photon can interact with any other simulation particle in the same element (DSMC) provided the input deck specifies a cross section for the interaction

- λ -dependent photon interactions

- Photon absorption based on Einstein-B coefficients
- Photo-ionization of O_2 and N_2
- Photo-dissociation of N_2 & O_2
- Photo-emission from surfaces

- Current results do not include self-absorption!

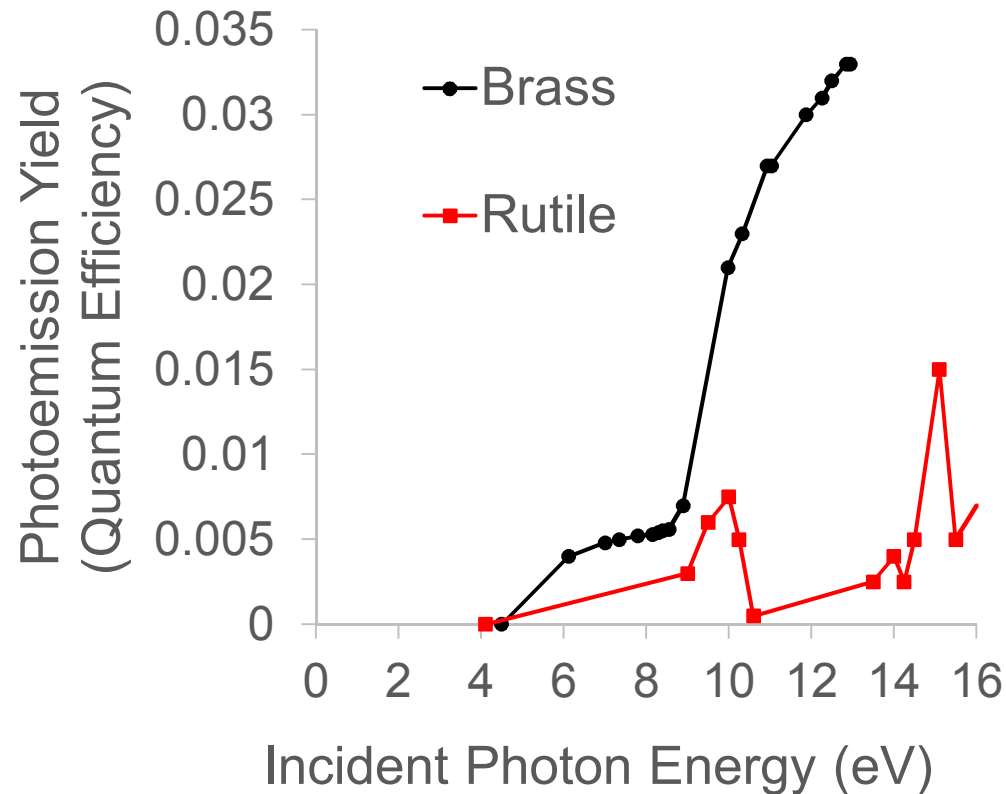
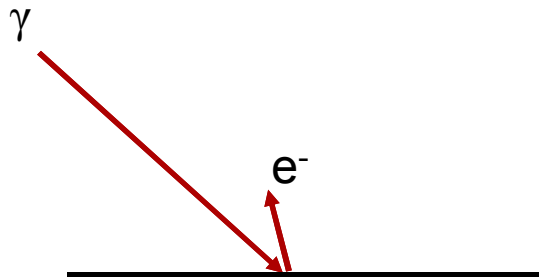
- Need to track vib. & rot. states



Photons: Photoemission

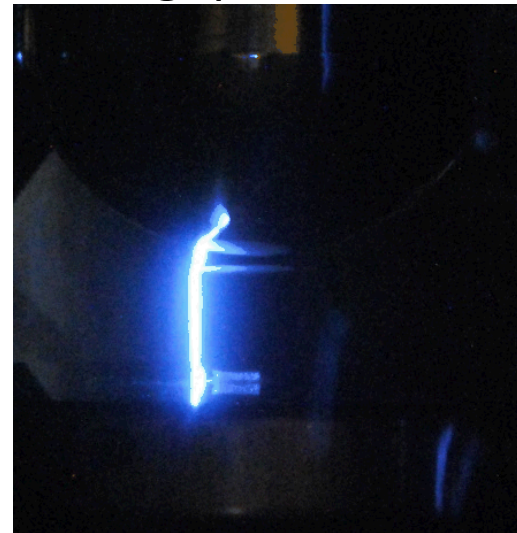
- Photons incident on the electrodes and dielectric have material specific electron emission yields

- Use experimental yields
- Dependent on the photon energy
- $E_{e^-} = E_{\gamma} - \phi_w$



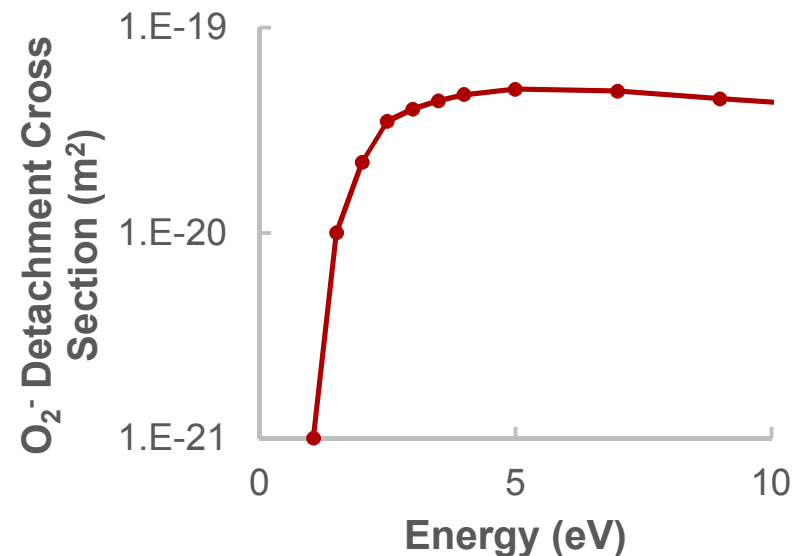
(Some) Open Questions

- How much does dielectric photoemission yield matter to streamer dynamics (velocity, attachment to dielectric, etc.)?
- 2D vs. 3D: sustainment/branching/etc. ?
- At \sim ion timescales (~ 100 ns) does the small gap fill with plasma allowing a streamer to propagate/start outside of the gap and travel down the dielectric surface?
- How important (to streamer dynamics) is accounting for vibrational and rotational states when spontaneously emitting photons?
- How important is self-absorption?



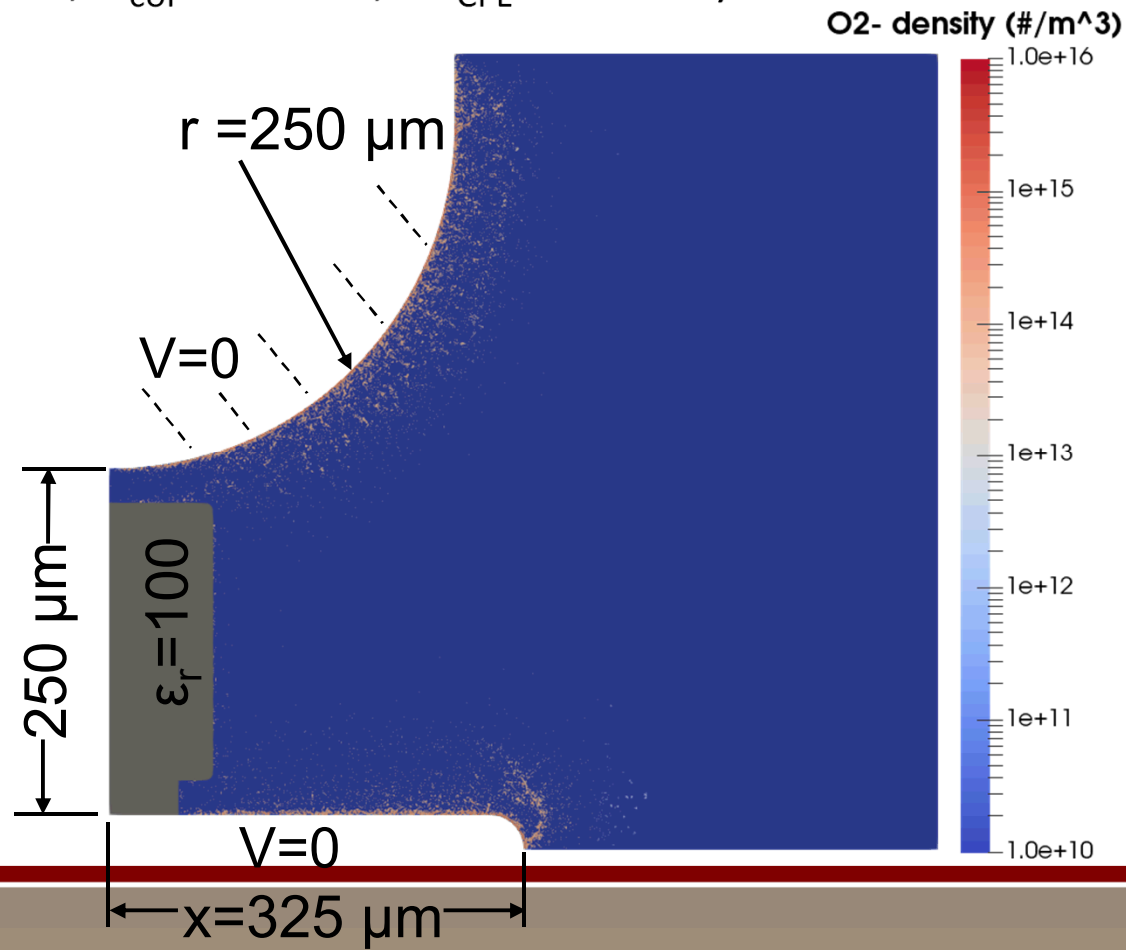
Initial Seed Plasma

- Desire a more physical initial seed plasma in order to explain variability in breakdown voltage/time
- In un-stressed state Cosmic Rays result in free electrons which then rapidly attach to O_2
- O_2^- long-lived at room temperature and no E-field
- Applied fields accelerate O_2^- resulting in sufficient energy for $O_2^- + M \rightarrow O_2 + e^- + M$
 - Source of initial e^- in regions of high fields (e.g., dielectric corners)



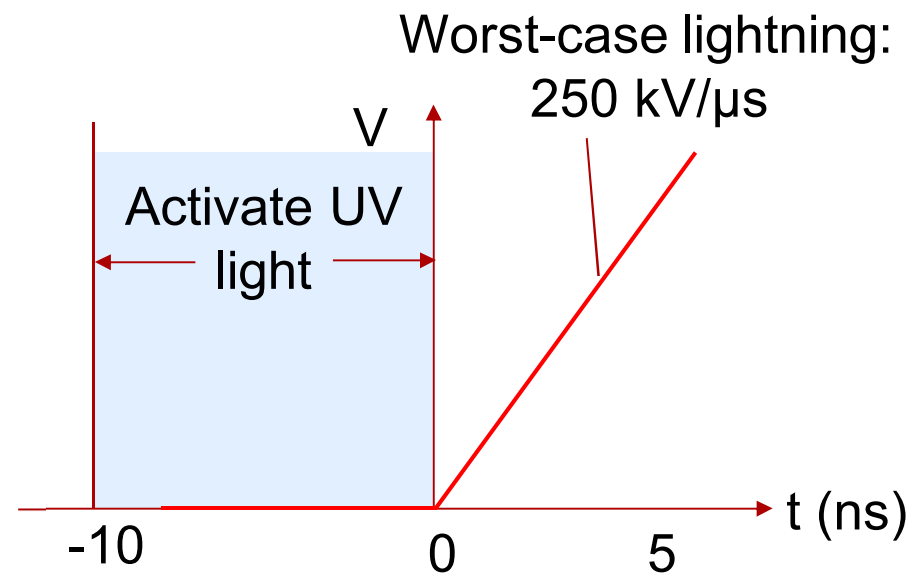
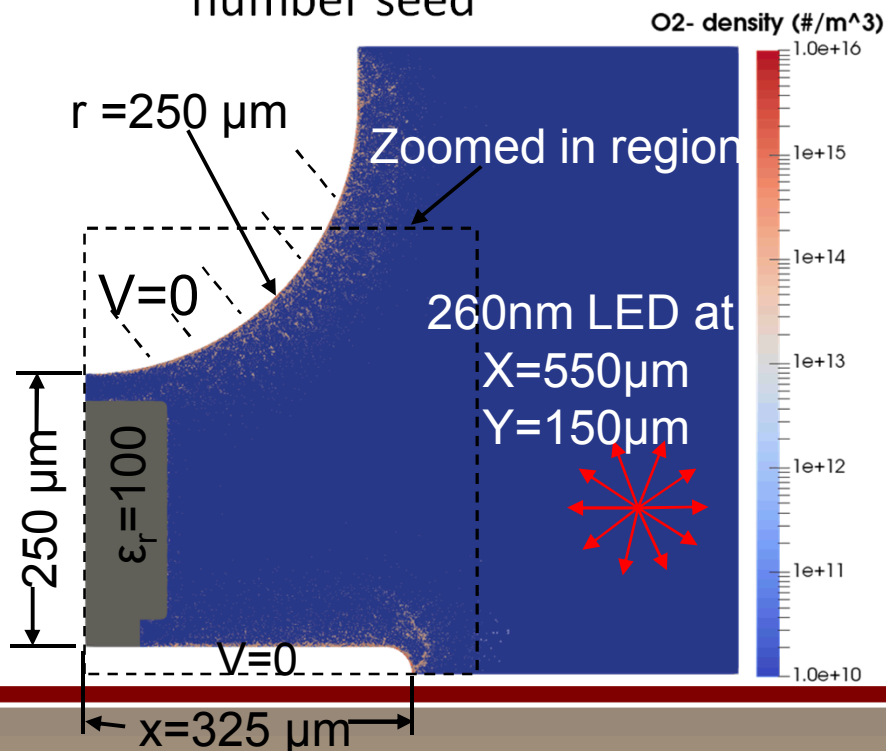
Simulation Domain

- 2D simulation of a 600 Torr, air-filled hemisphere-to-plane 250 μm gap with 200 μm TiO_2 ($\epsilon_r = 100$) cylinder on top of a 25 μm dielectric ($\epsilon_r = 3$) spacer between brass electrodes.
- $\Delta t = 5 \times 10^{-14} \text{ s}$ ($\omega_{pe}^{-1} \sim 10^{-12} \text{ s}$; $\nu_{col} \sim 10^{-12} \text{ s}$; $\Delta t_{CFL} \sim 10^{-13} \text{ s}$)
- “ Δx ” = 0.2 μm
($\lambda_D \sim 0.2 \mu\text{m}$ in streamer channel)
 - ~30 million elements
- Particle merger kept
~128 charged particles
per element

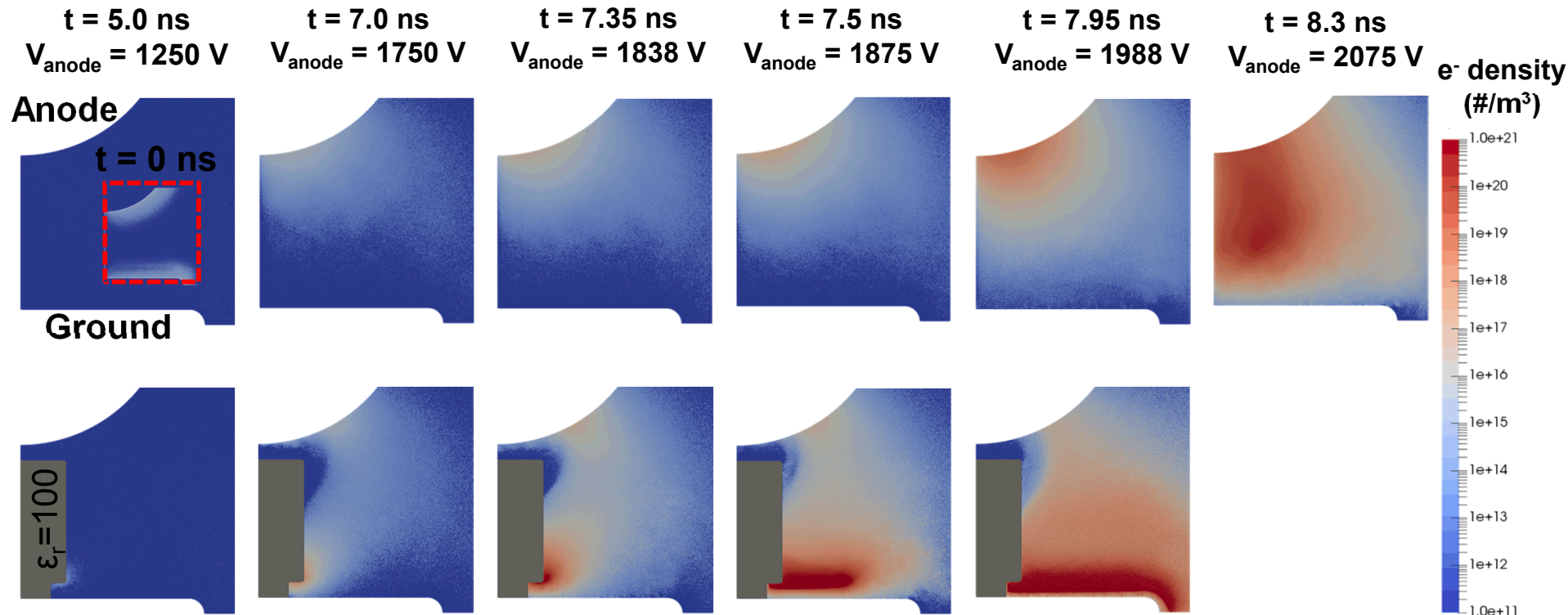


Generation of Initial O_2^-

- Activate isotropic, 260 nm UV LED light source (1.6 mW/cm^2 on axis) for 10 ns with no applied potential and then ramp anode voltage at $250 \text{ kV}/\mu\text{s}$
- Electrons diffuse through the background neutral gas and attach to O_2 through 3-body collisions
 - Gives initial density distribution for e^- & O_2^- which varies with random number seed



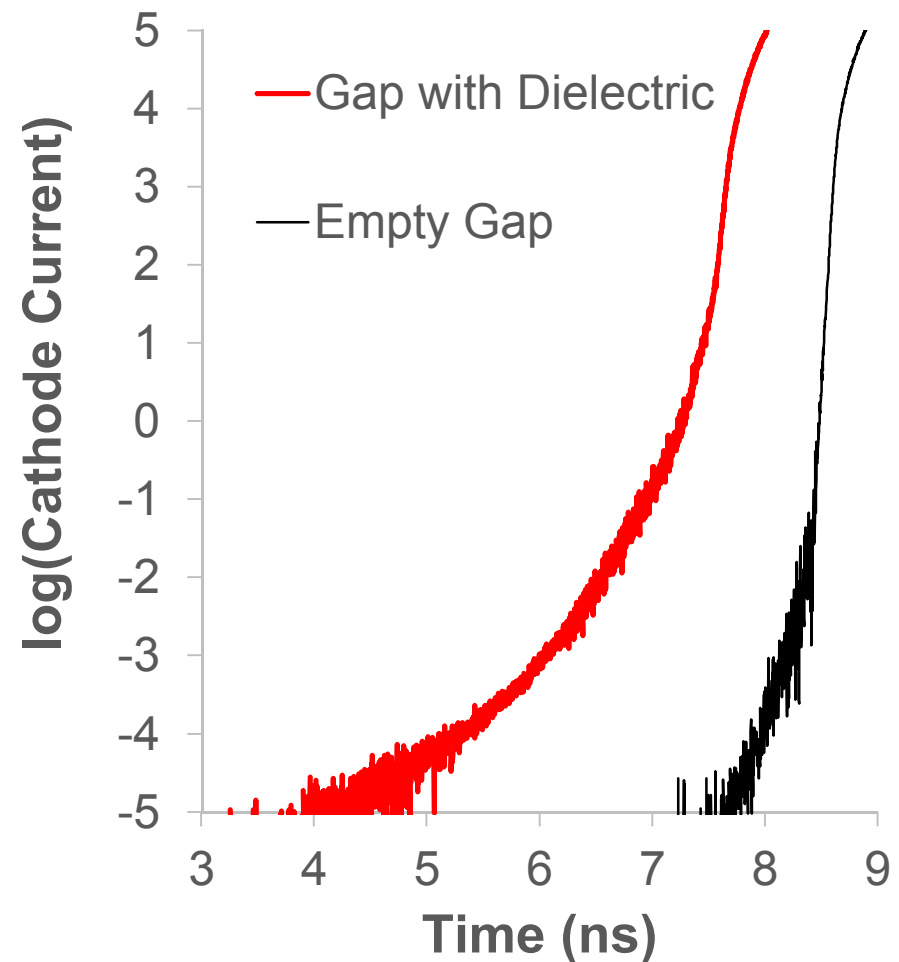
Results: e⁻ densities



- Initial photoemitted e⁻ swept out of the gap in the first several ns while the field is still too low to result in breakdown
- After ~5ns the field near the dielectric corner causes significant O₂⁻ detachment supplying e⁻ in the high-field (and thus high α) region
- In contrast, there is much less O₂⁻ detachment in the empty gap (with lower E)
- After 7ns significant e⁻ avalanche starts near cathode in the high-field region near the dielectric corner and leads to 2nd positive excitation of N₂.
- In the empty gap e⁻ densities are highest near the anode as would be expected from seed e⁻ avalanching back to the anode

Cathode Currents

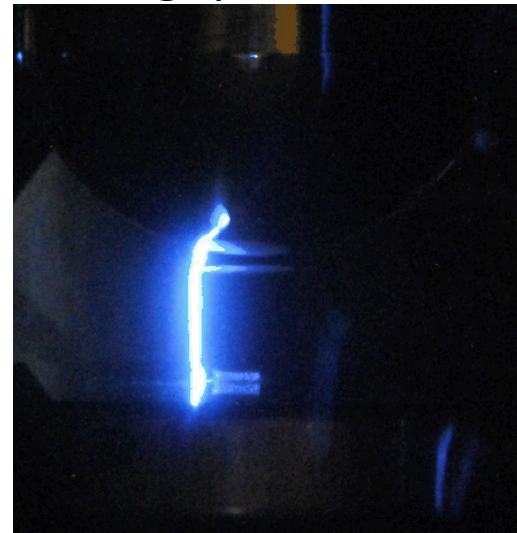
- The empty gap breaks down in ~ 8.5 ns via a positive streamer
 - Breakdown voltage ~ 2125 V
 - Significant variation in breakdown voltage for different random seeds
- Presence of dielectric particle causes gap to breakdown in under 8 ns
 - Breakdown voltage ~ 2000 V
 - Less variation in breakdown voltage when changing random seed



(Some) Open Questions

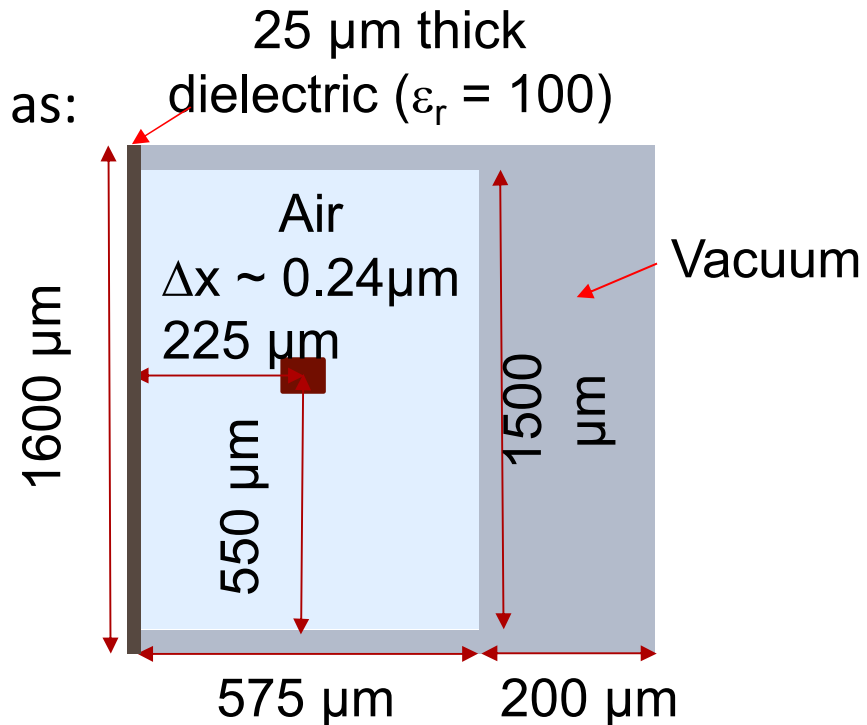
- How do initial seed plasma/electrons form and influence breakdown delay time?

- 2D vs. 3D: sustainment/branching/etc. ?
- At \sim ion timescales (~ 100 ns) does the small gap fill with plasma allowing a streamer to propagate/start outside of the gap and travel down the dielectric surface?
- How important (to streamer dynamics) is accounting for vibrational and rotational states when spontaneously emitting photons?
- How important is self-absorption?



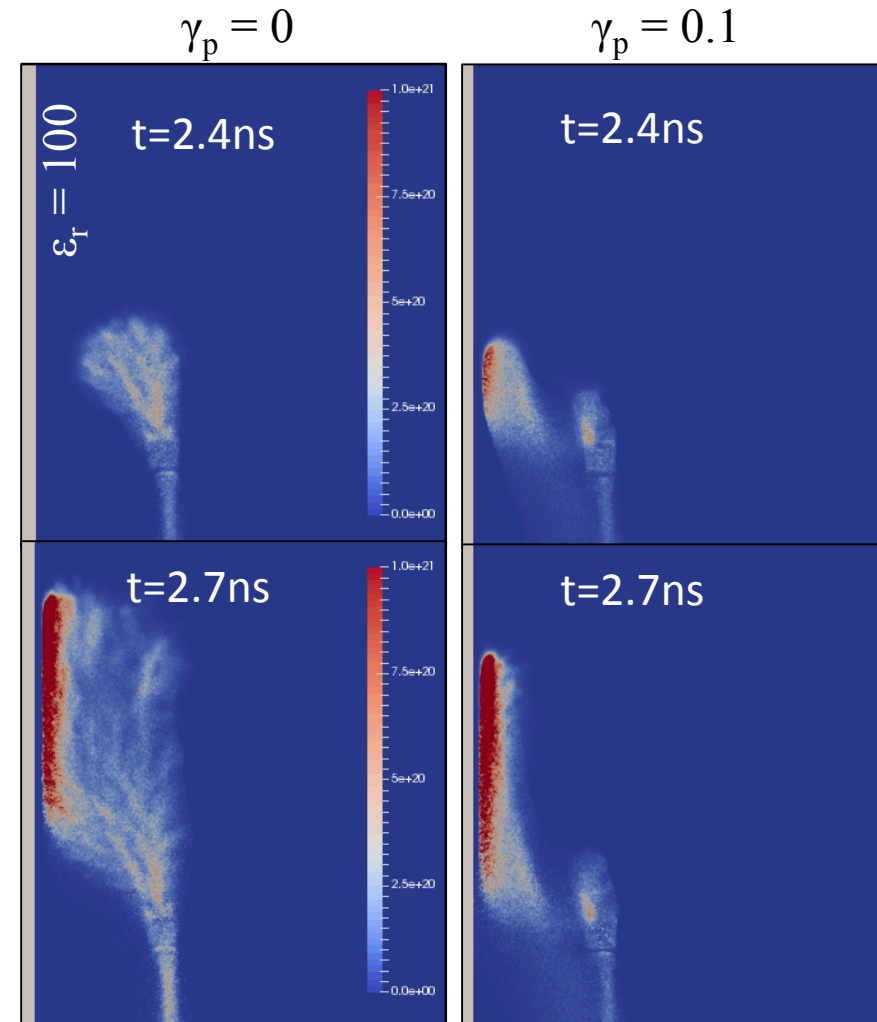
Domain: Photoemission Sensitivity

- 2D simulation of a 760 Torr 1.5 mm air filled gap with a 25 μm thick TiO_2 ($\epsilon_r = 100$) cylinder between electrodes
- Over-volted state (8 kV anode voltage) allows for rapid evolution of the streamer
- Initial seed plasma density modeled as:
 - $T_e = T_i = 1 \text{ eV}$ and $n_e = 10^{20} \text{ m}^{-3}$
 - 50 μm^2 square centered at 225 μm radially from dielectric surface and 550 μm axially from bottom vacuum interface



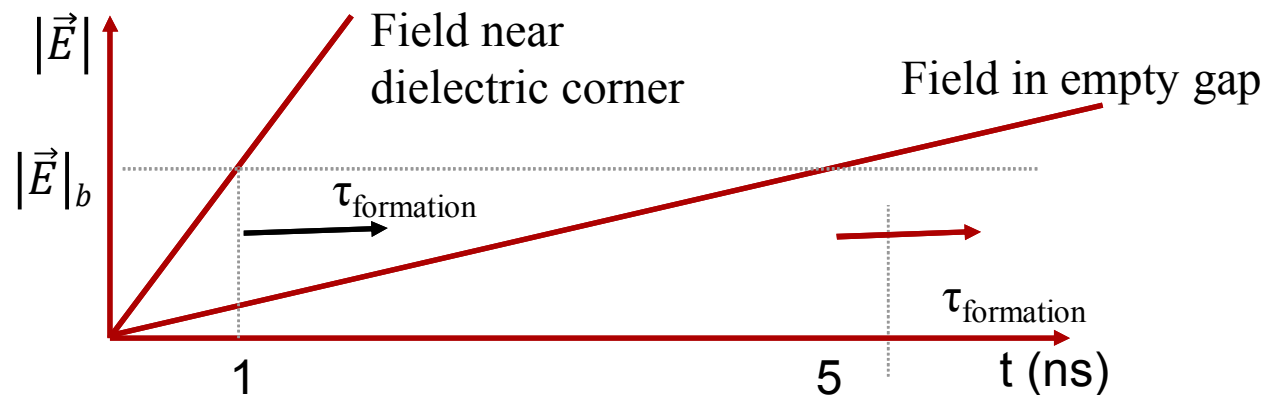
Results: Photoemission Sensitivity

- At 2.4 ns:
 - Elevated E/n near the dielectric surface results in an increase in α (net ionization). This “attracts” the streamer to the dielectric as e^- created in this region avalanche more strongly
 - With a photoemissive dielectric surface two distinct cathode directed streamers form
- At 2.7 ns:
 - The streamer travels along the dielectric surface in both cases; however, it is much more diffuse for the case without photoemission



Conclusions

- Kinetic model for breakdown in the presence of dielectric particles:
 - Townsend mechanisms (e^- - neutral collisions, ion induced SEE)
 - Ion and neutral transport (O_2^- detachment)
 - Photon transport (photoionization and photoemission)
- Dielectric particles create regions of high fields that detach O_2^- at earlier times supplying initial electrons which then initiate breakdown sooner (and more reliably) during lightning pulse



- Increased E/n near dielectric results in amplification of net ionization rate, α , preferentially directing streamer growth towards the surface even under conditions of zero quantum yield