

Comparing hydrocarbon C-H stretch infrared emission during direct injection with and without combustion in an optical diesel engine.

SAND2017-4476C



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National Meeting of the Combustion Institute

04/24/17 - 04/26/17



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Acknowledgements

- Wayne State University
- Istituto Motori
- IFPEN
- Sandia National Labs, CRF



ISTITUTO MOTORI
Consiglio Nazionale delle Ricerche

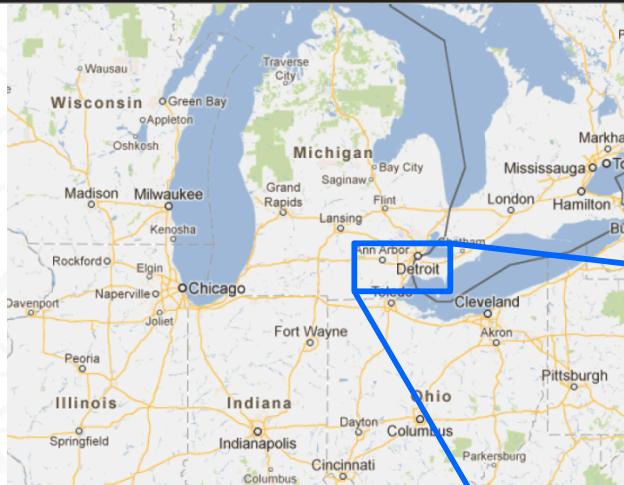


Experiments were conducted at the Combustion Research Facility, Sandia National Laboratories, Livermore, CA. Support for this research comes from the U.S. Department of Energy, Office of Vehicle Technologies. Sandia is a multi-mission laboratory operated by Sandia Corporation, a Lockheed Martin Company for the United State Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000. The authors thank Dave Cicone for his assistance maintaining the research engine.

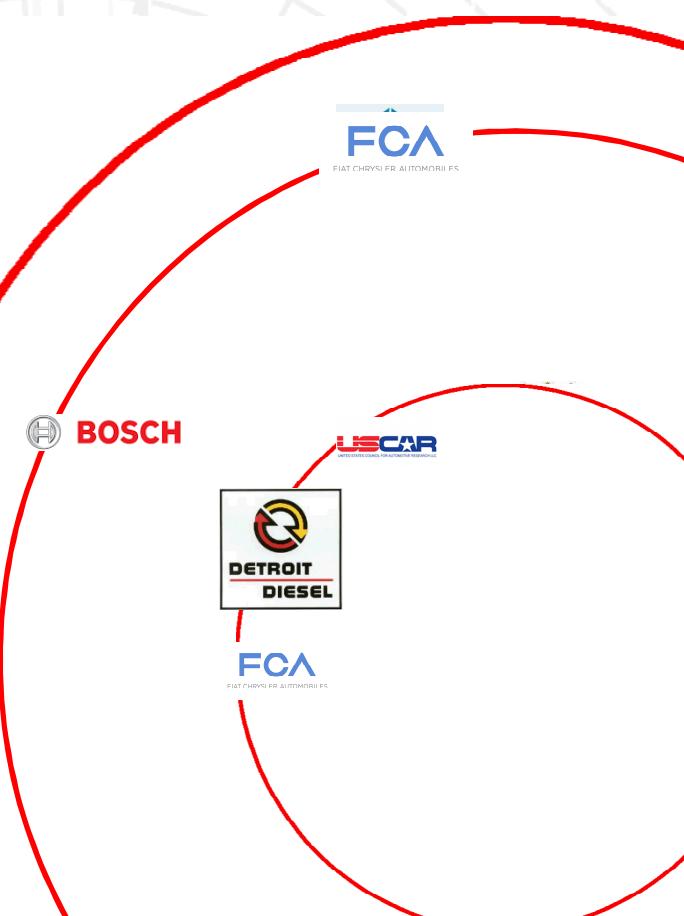
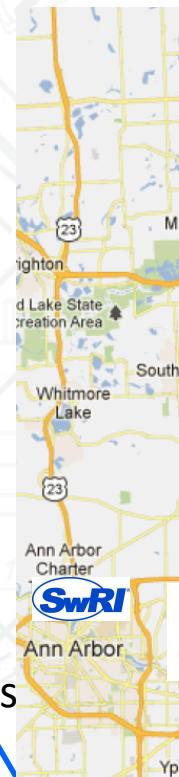


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Opportunity is waiting for you in Detroit



GM World Headquarters, Detroit, 3.5 miles
Ford World Headquarters, Dearborn, 8.9 miles
FCA Powertrain Division, Redford, 12.4 miles
USCAR office, Southfield, 13.9 miles
Detroit Diesel, Dearborn Heights, 13.8 miles
US Army TACOM, Warren 15.5 miles
FCA US, Auburn Hills, 16.9 miles
Robert Bosch LLC, Novi, 26.9 miles
Toyota Technical Center, Ann Arbor 36.6 miles
SWRI Ann Arbor Tech Center, Ann Arbor, 37 miles

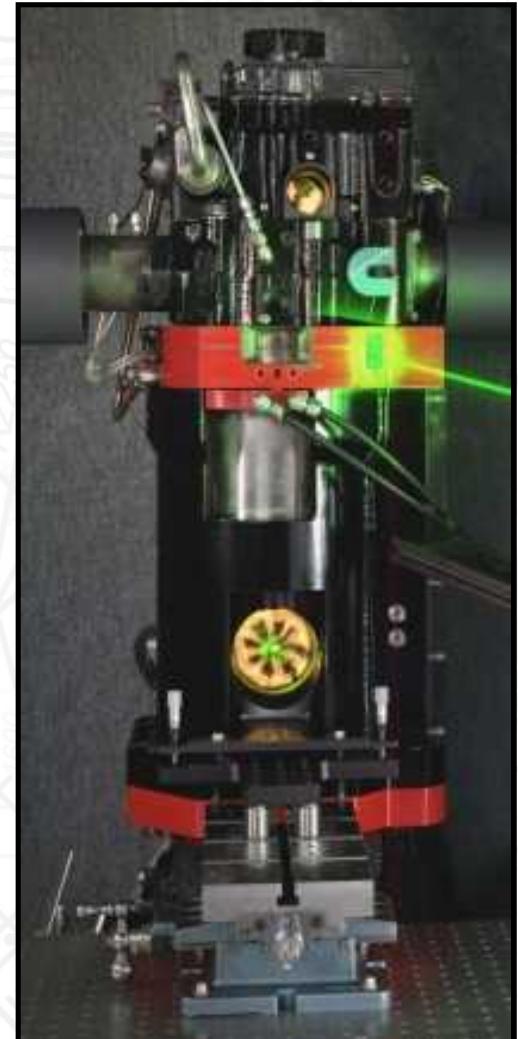




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Objectives of this Presentation

- What is the problem (or problems)?
 - Air pollution contributions from vehicles
- What is the means of attack?
 - Regulation
 - Optical Combustion Diagnostics
- Why Infrared?
 - What could we find?
 - Did we find what we were looking for?
 - How can we be sure?
- What's next?



Sandia 2.34 L Optical Engine



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Vehicle Emissions harm health

UCLA academics exposed mice to smog and found that breathing emissions can turn the good cholesterol (HDL) into bad (LDL).

... after two weeks of exposure to vehicle emissions, mice showed oxidative damage in the blood and liver -- damage that was not reversed after a subsequent week of receiving filtered air. Altered HDL cholesterol may play a key role in this damaging process, they said. The mice exposed to emissions for two weeks "had a much-decreased ability to protect against oxidation and inflammation induced by low-density lipoprotein (LDL) cholesterol... We suggest that people try to limit their exposure to air pollutants."

Image: Los Angeles (1980)

(2013) Arteriosclerosis, Thrombosis and Vascular Biology

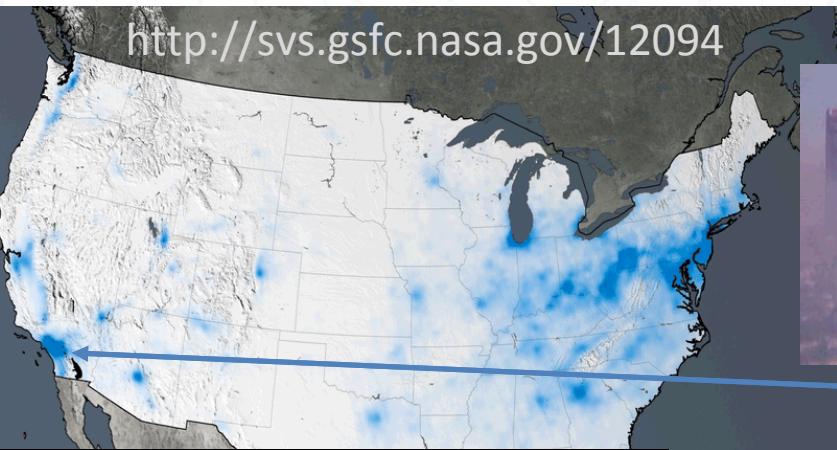
Other problems include: fuel stock sustainability, transportation infrastructure optimization, engine/fuel co-optimization, ...



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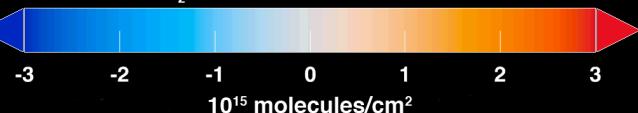
Measuring Impacts

<http://svs.gsfc.nasa.gov/12094>



Los Angeles (1980 to 2015)

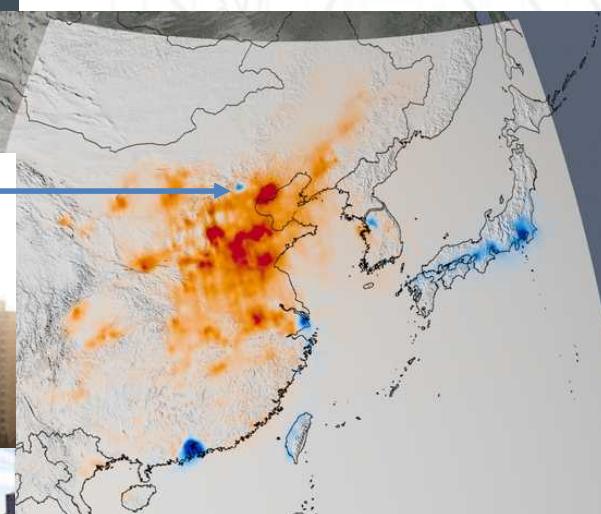
NO₂ Absolute Trend 2005-2014



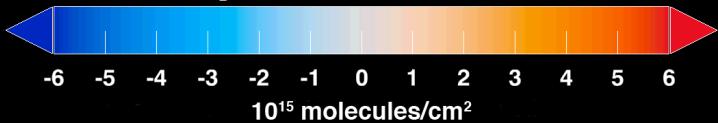
Beijing (typical)



Beijing (two weeks of driving prohibited)

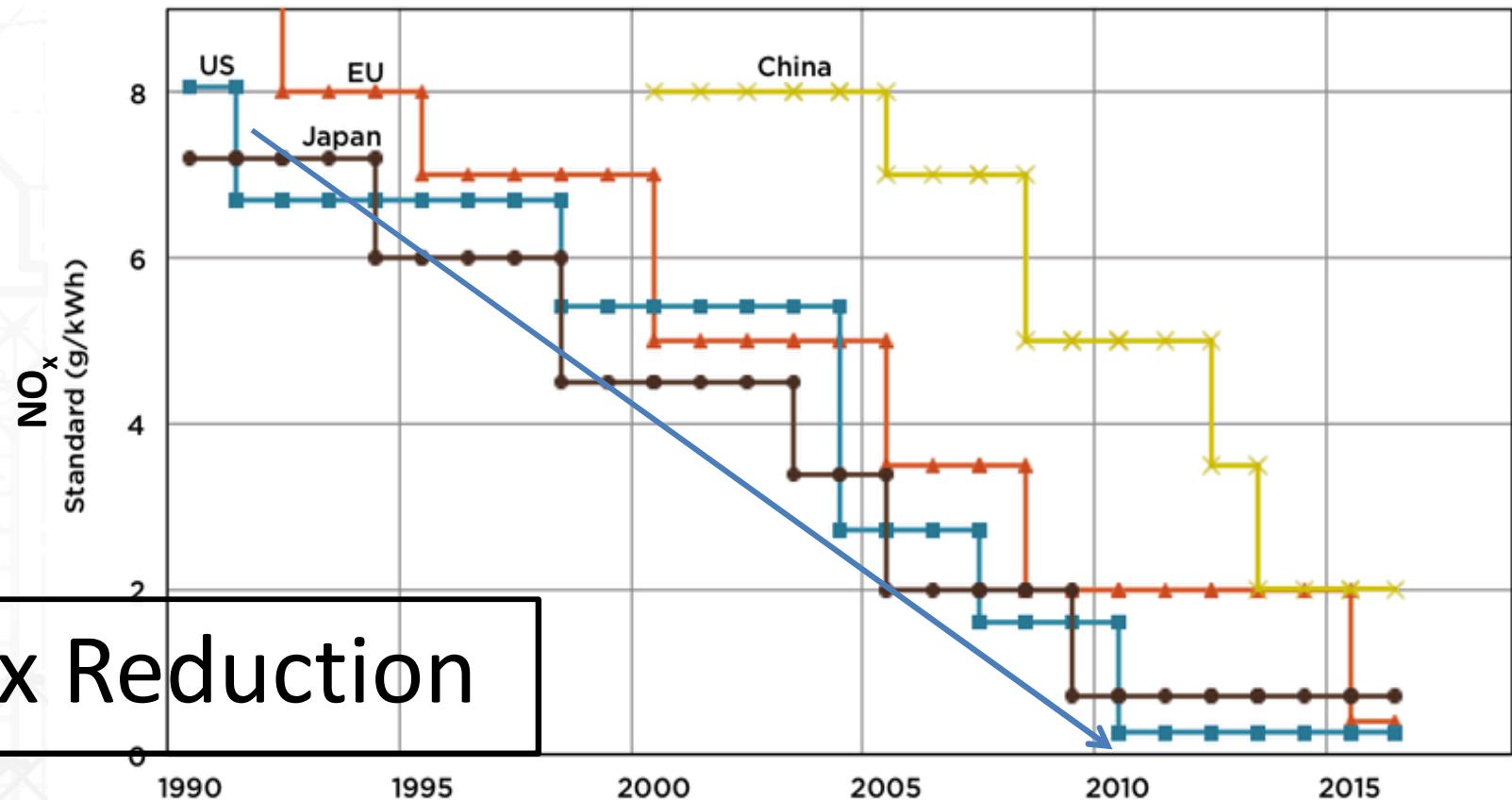


NO₂ Absolute Trend 2005-2014





Vehicle Emissions Regulation



http://transportpolicy.net/index.php?title=File:NOx_and_PM_standards_for_heavy-duty_engines.png

“It seems unlikely that the internal combustion engine will be legislated out of existence, but the quality of the engineering research required to meet the ever more stringent legislation, while delivering a competitive product is very high indeed.” -Nick Collings

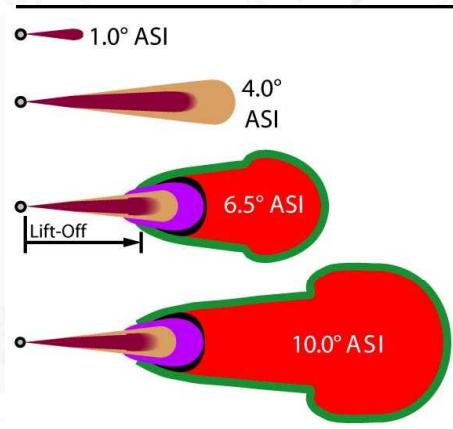


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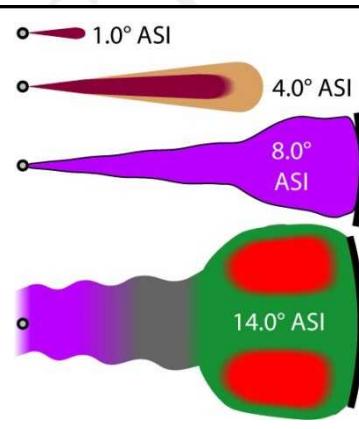
Investigate where/why pollution is created?

Develop the science to explain in-cylinder spray, combustion, and pollutant-formation processes for both conventional diesel and LTC that industry needs to design and build cleaner, more efficient engines

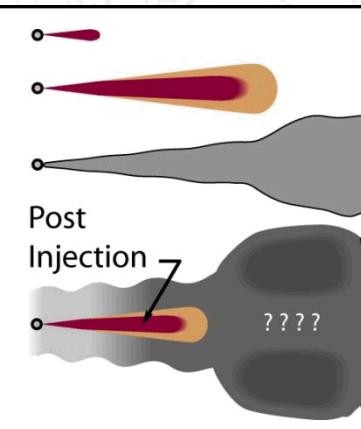
1997: Conventional Diesel
(Single Injection)



2012: LTC Diesel
(Single Injection)



2013+: Multiple Injection
(Conventional & LTC)



Legend for conventional diesel:

- Liquid Fuel (Maroon)
- Pre-ignition Vapor Fuel (Orange)
- First-Stage Ignition (H₂CO, H₂O₂, CO, UHC) (Purple)

Legend for LTC diesel:

- Intermediate Ignition (CO, UHC) (Grey)
- Second-Stage Ignition of Intermediate Stoichiometry or Diffusion Flame (OH) (Green)

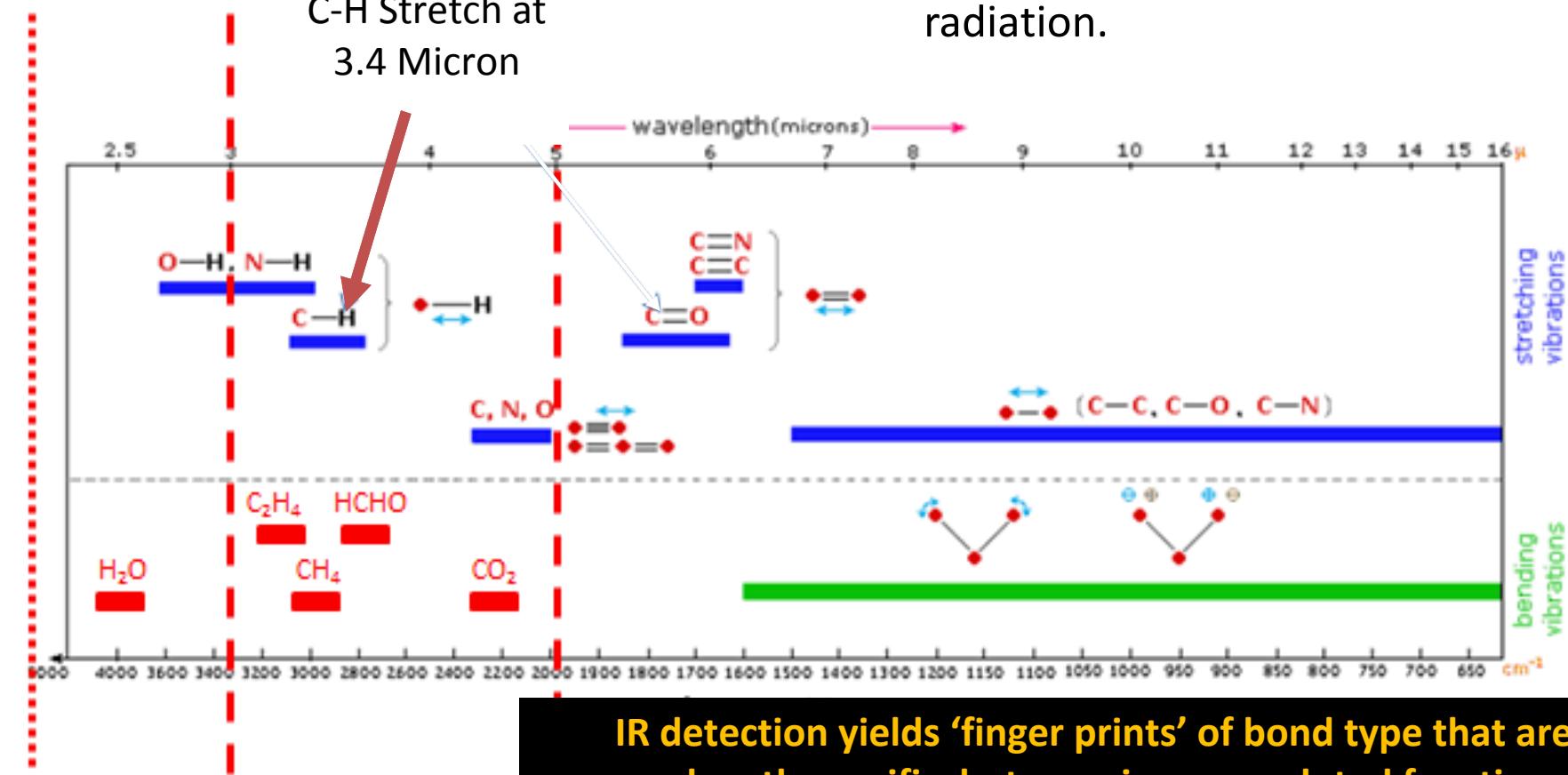
Legend for multiple injection:

- Second-Stage Ignition of fuel-rich mixtures (Black)
- Soot or Soot Precursors (PAH) (Red)

UHC – Unburned Hydrocarbons, LTC – Low Temperature Combustion, PAH- polyaromatic hydrocarbon

Musculus, M. P. B., Miles, P. C., & Pickett, L. M. (2013). *Conceptual models for partially premixed low-temperature diesel combustion*. *Progress in Energy and Combustion Science* (Vol. 39, pp. 246–283). Elsevier Ltd. doi:10.1016/j.pecs.2012.09.001

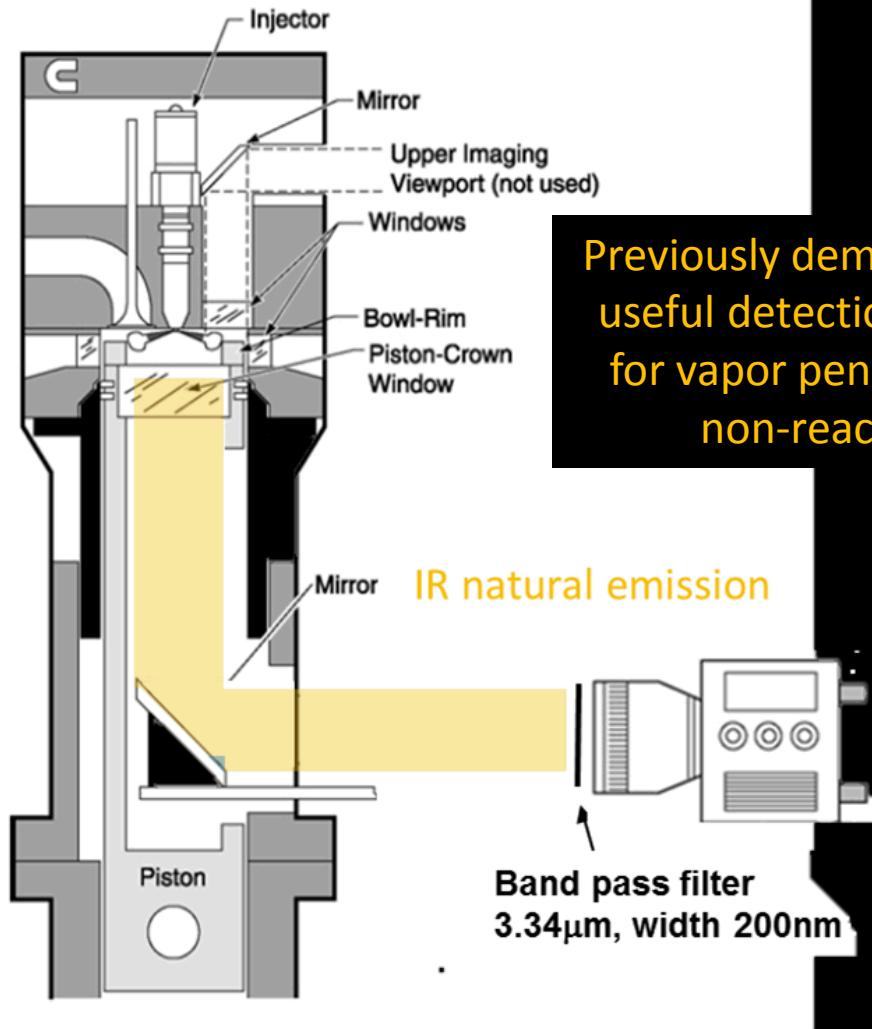
WHY IR? Prior art - diagnostics



IR detection yields 'finger prints' of bond type that are wavelength specific, but remain a convoluted function of temperature, concentration, and path length

■ Data from 2008_Jansons_IER

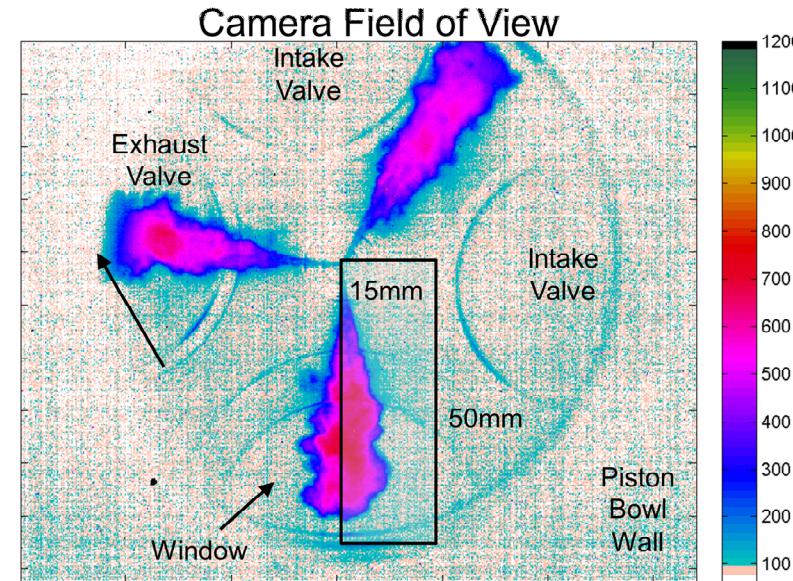
Methods



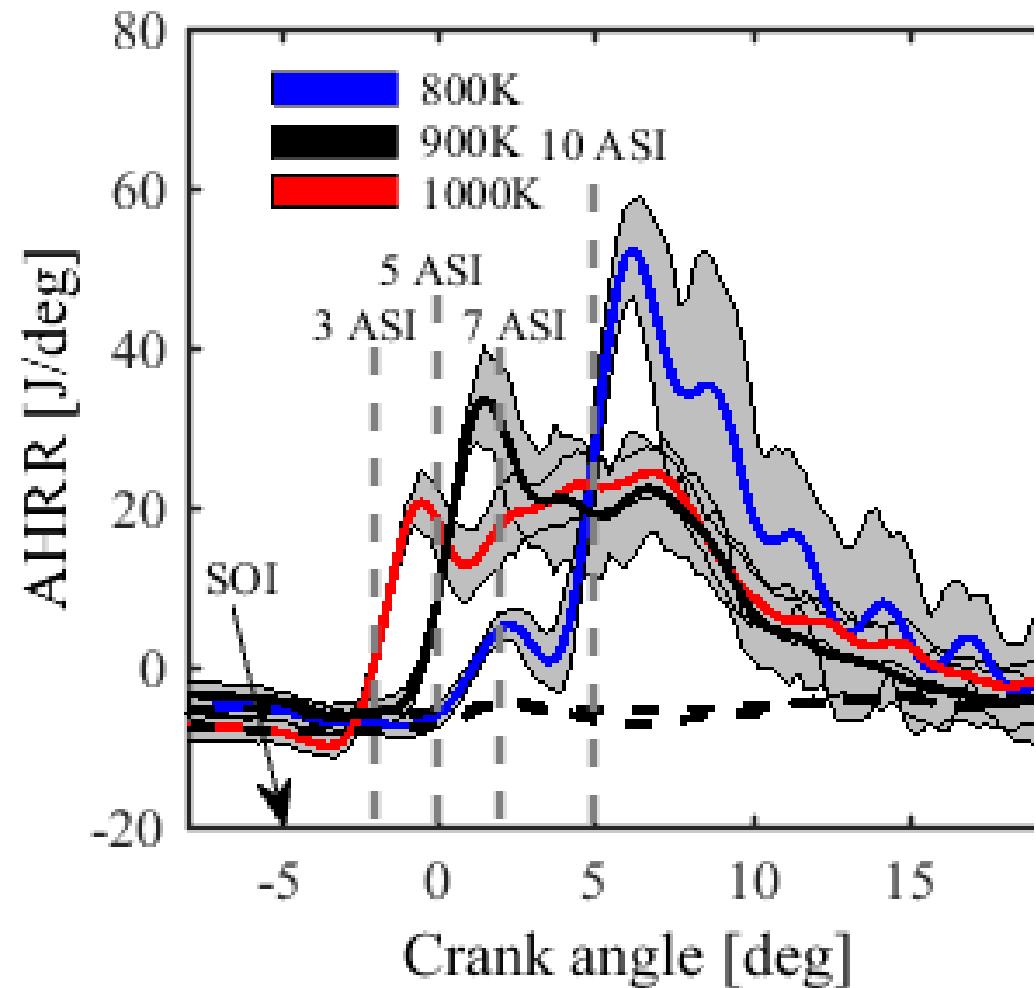
Previously demonstrated useful detection of fuel for vapor penetration, non-reacting

Infrared

- MW-IR (InSb) 3-5 μ m (TELOPS)
- Lens: 50mm IR f/2.3
- Frame rate: 1 image / cycle
- Exposure: 5 μ s
- Scale: 4.2 pix per mm



Experimental Conditions

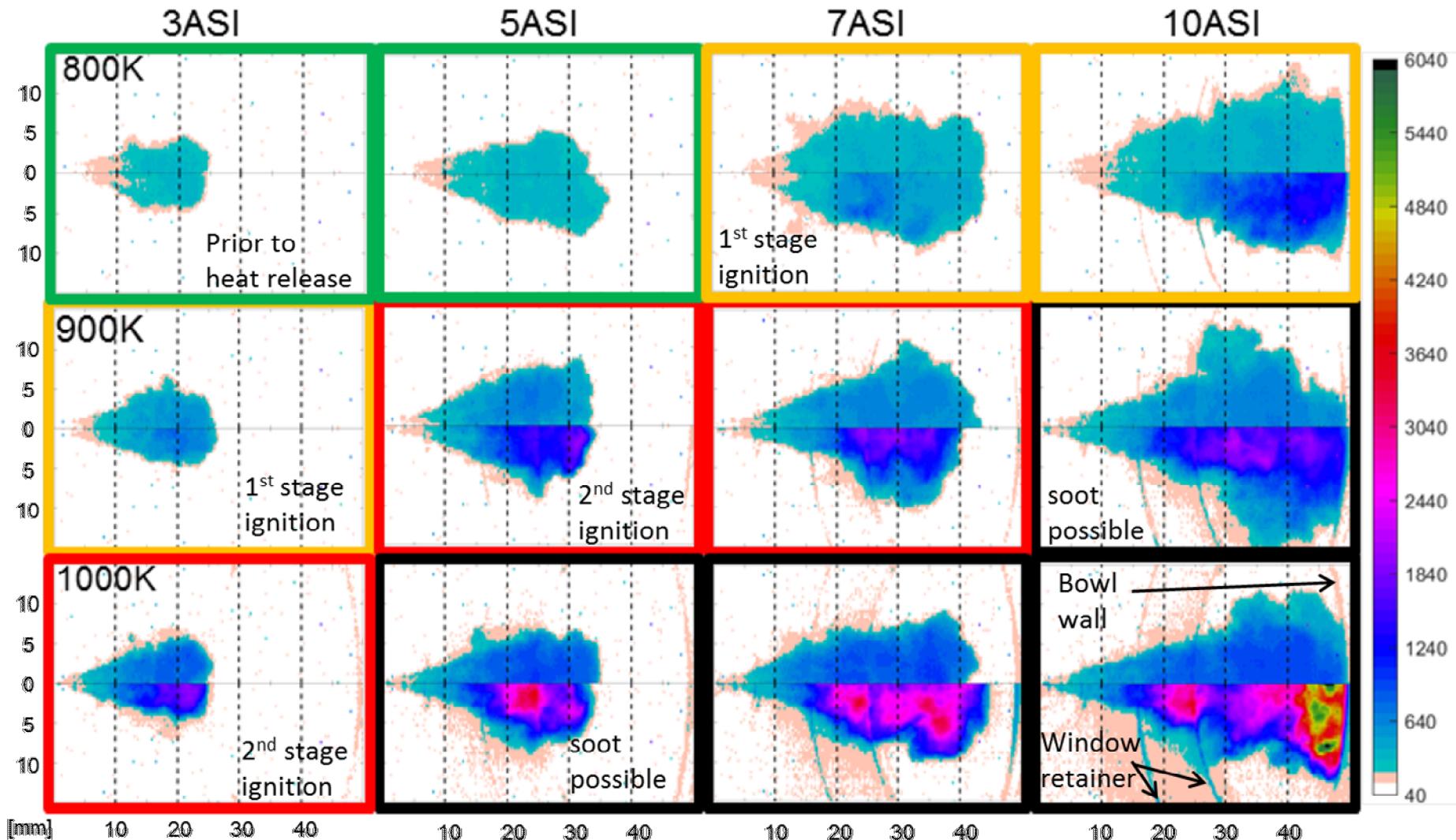
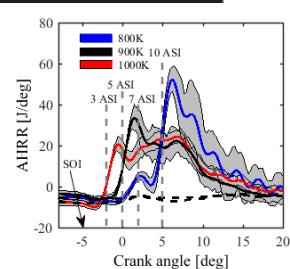


Parameter	800	900	1000
Temperature @ TDC [K]			
Density @ TDC [kg/m ³]		15.2	
Oxygen content [% vol.]		15	0
Injector rail pressure [bar]		1500	
Temperature @ IVC [K]	340	380	454
Pressure @ IVC [bar]	2.01	2.25	2.61
Injected liquid mass [mg/cycle]	3.68	3.68	3.68
Engine Speed [rpm]		1200	
Start of Solenoid Energizing [CAD ATDC]			-6.75
Start of Injection [CAD ATDC]			-5
Duration of Solenoid Energizing [CAD], [μs]		5.86	795
Injection Duration [CAD], [μs]		11	1500



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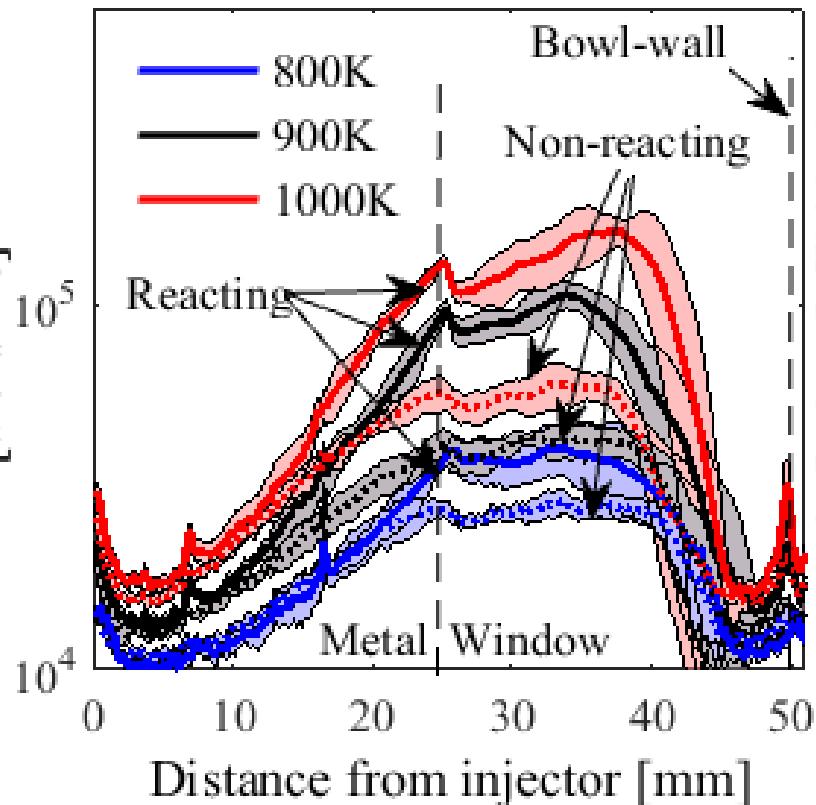
With and Without Reaction



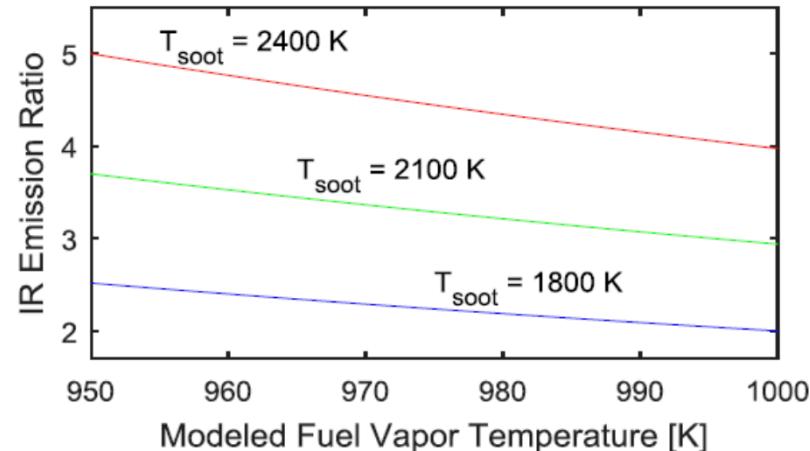
IR appears useful to detect Lift-off.

Limitations and Uncertainty

Longitudinally integrated intensity
[arb units]

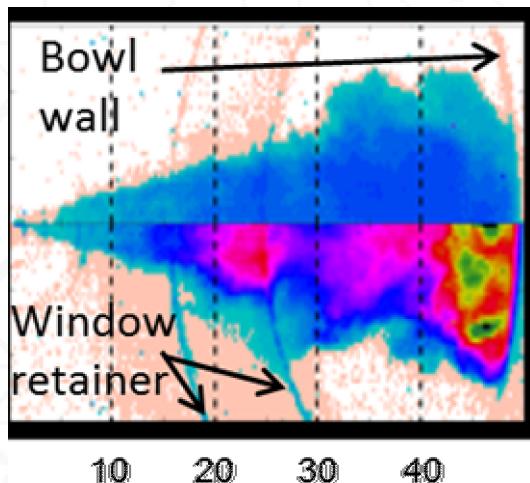


Changes in background reflection strongly influence detected signal.



Presence of high temperature soot at relevant concentrations (Spray A) likely overwhelms C-H stretch natural emission

Summary

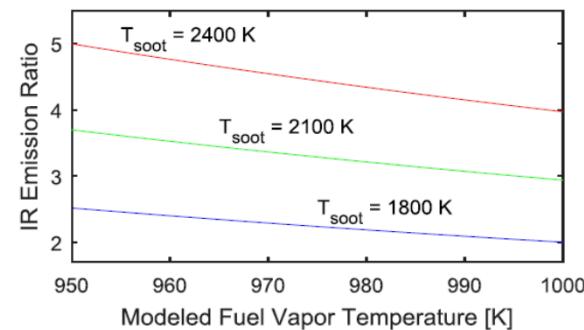
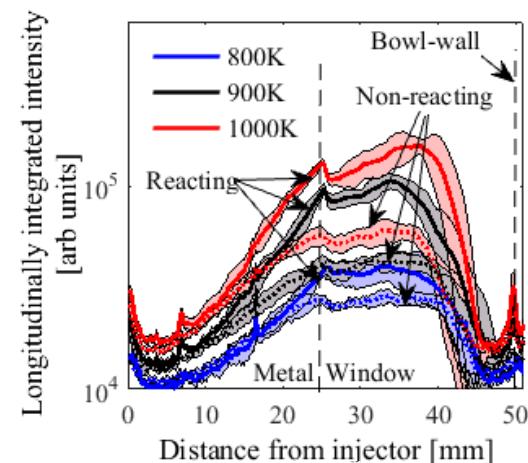


Significant limitations and uncertainties include:

- Changing/non-uniform background reflections
- grey-body radiation from soot

And not discussed here:

- changing soot deposits on window surfaces (not discussed here)
- signal self-absorbance along the line of sight (previous work)





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Current Gaps

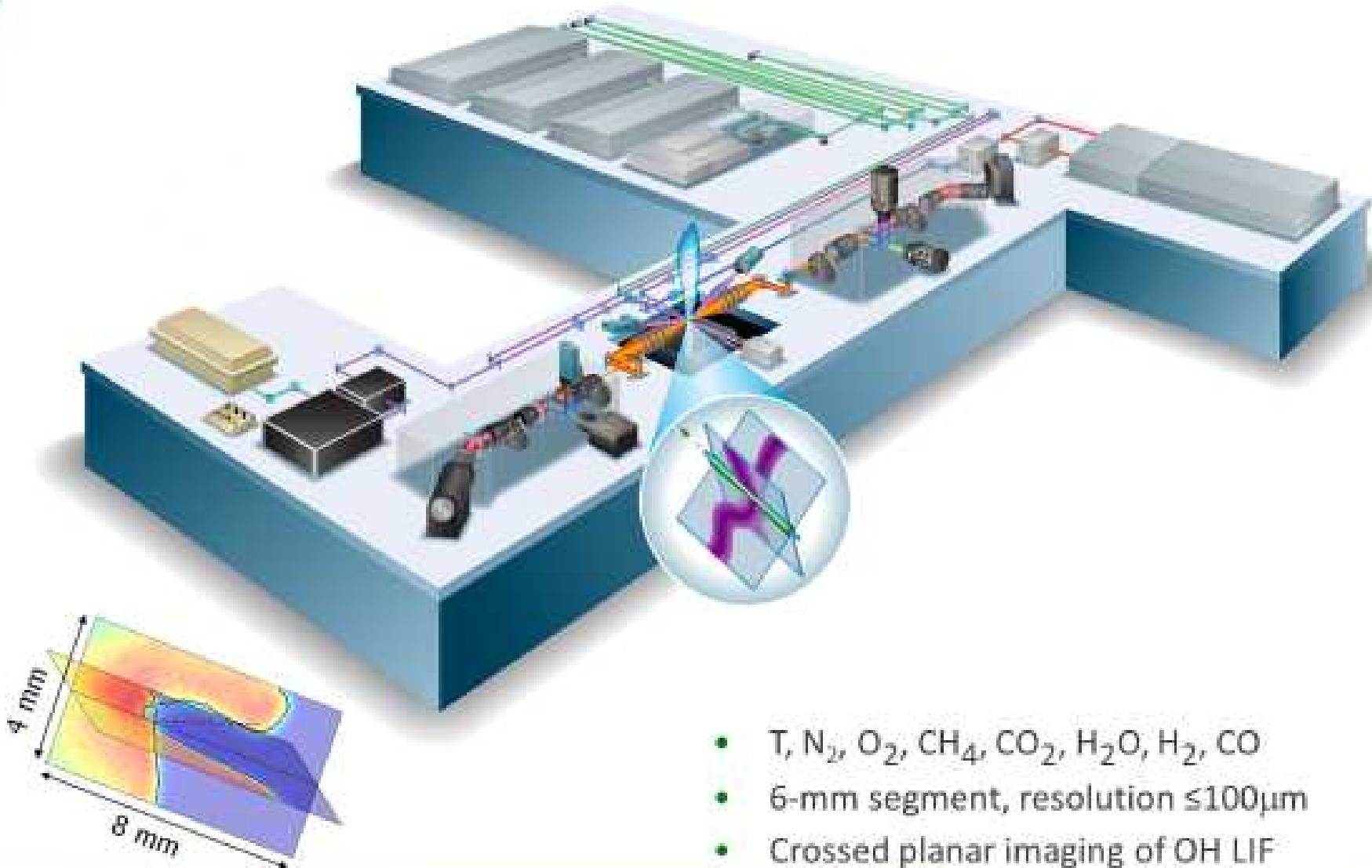
Experimental diagnostics that can help identify FORMATION vs OXIDATION of criteria pollutants in-cylinder, for the validation of CFD emissions models.

**Need all the relevant data to solve
Reynolds Transport Theorem!**

$F(c, T, p, \text{velocity}) \rightarrow (\text{change with time} = \text{transport} + \text{production} + \text{dissipation})$

Coupled Turbulence-Chemistry Problem!?
Need Lagrangians...Time Resolved...

Turbulent Combustion Laboratory





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What is (could be) next?

- Identify regions of formation/oxidation by bond type
- Can geometric transformations to recover local data? (Abel inversion?)
- Look for multi-spectral features that can de-convolve Temp and Concentration dependence (e.g. R. Hanson)
- PIV recording of mean velocity field, strain
- 'predict' next time step using experimental measured values of T, c, strain, and transport.
- Compare with next measurement to identify sources of error in modeling.



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Scientific barriers to further IR work

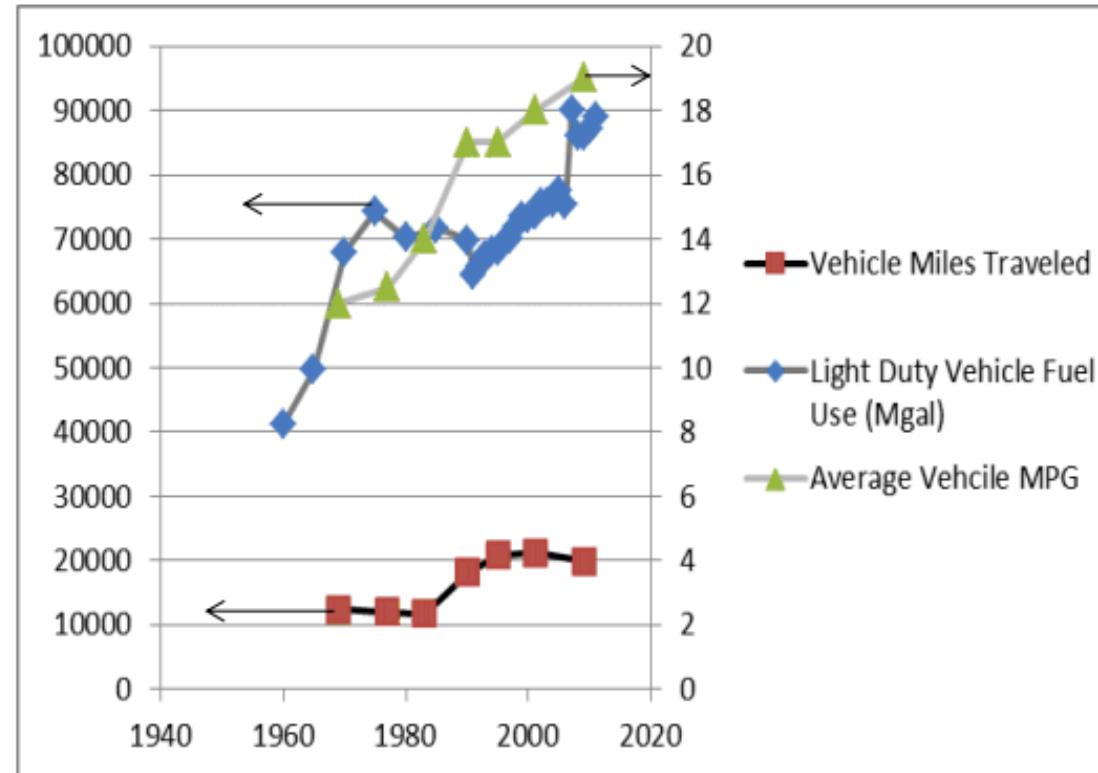
- Pressure broadening
- High temp interference e.g. OH, H₂O, CO₂
(others? un-quantified?) – any suggestions?
- Background non-uniformity(ies)
- Self-absorption (concentration and path length dependent)
- Gradient-induced beam steering between source and detector.

Framing bigger problems...

Technologies that maintain low emissions, reduce noise and maximize system efficiency must be coupled with strategies that combat wasteful use and promote thoughtful consumption.

Jevon's Paradox:
An increase in efficiency
with which a resource is
used **tends to increase the**
rate of consumption
of that resource.

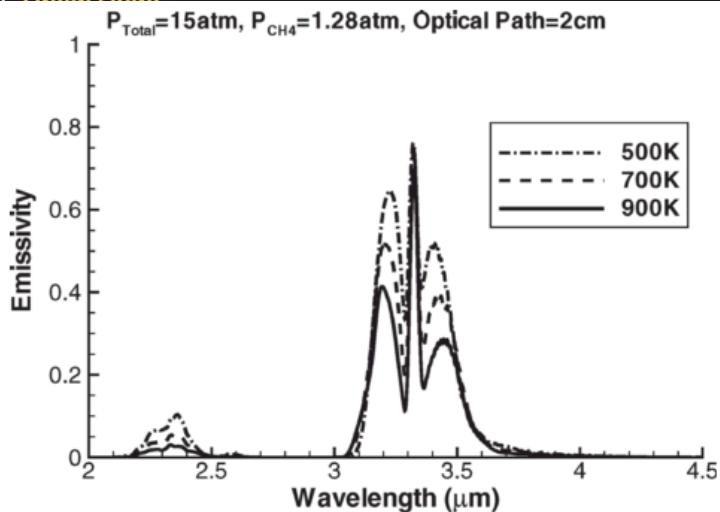
Questions?



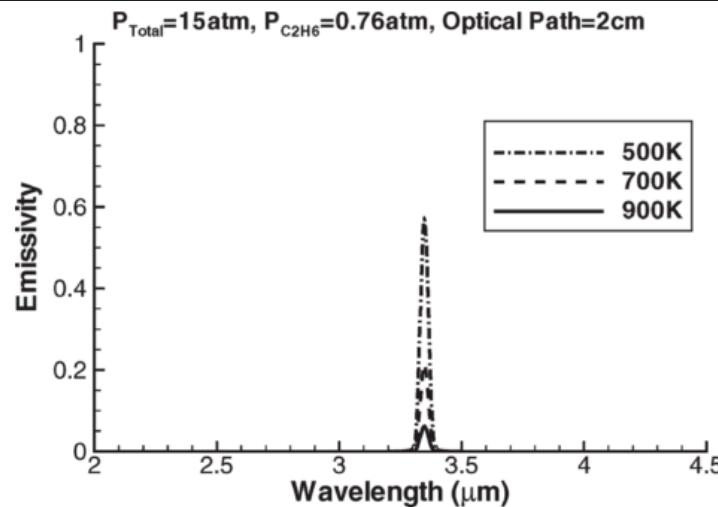


Spectral modeling of fuel and combustion products demonstrate

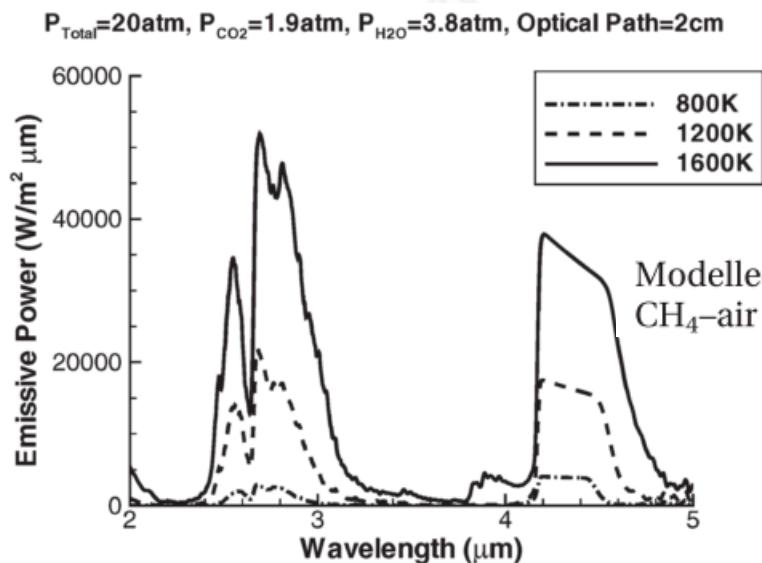
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Modelled stoichiometric CH_4 -air mixture emission spectrum



Modelled C_2H_6 -air mixture emission spectrum



Modelled emissive power of stoichiometric CH_4 -air combustion product mixture

M. Jansons, S. Lin and K Rhee Infrared spectral analysis of engine preflame emission IJER 2008



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WHY IR? Prior art - diagnostics

Species detection – 1D measures

SAE 2007-01-0644: Grosch (Laser-Lab Gottingen). Fuel absorption @3.39 μ m (using a He-Ne laser) . SI engine. Fuel concentration near spark plug. Sapphire fiber, no absorption @ 3.39 μ m.

SAE 2007-01-0639: Kakuho (Nissan), Hanson (Stanford). 2 colors Water absorption in near IR (around 1.4 μ m) + modulation. SI engine. Residual gas concentration and temperature near spark plug.

SAE 2007-01-1849: Kawahara (Obayama Univ.) CO₂ absorption at 4.26 μ m . SI engine. Residual gas concentration.

SAE 2006-01-3337: De Francqueville (IFPEN). CO₂ absorption at 4.26 μ m. SI engine. Residual gas concentration.

Species detection – 2D measures

SAE 1999-01-3494: Jansons (Rutgers). IR camera 2000Hz DI Diesel engine. Simultaneous IR emissions @ 2.2 μ m, 2.47 μ m (water), 3.42 μ m (preflame) and 3.8 μ m (soot). Qualitative images.

SAE 2000-01-1800: Jansons (Rutgers). IR camera 2000Hz SI and DI Diesel engines. Simultaneous IR emissions @ 2.2 μ m, 2.47 μ m (water), 3.42 μ m (preflame) and 3.8 μ m (soot). Qualitative images.

IJER 2008 9:215: Jansons (Wayne State), Lin & Rhee (Rutgers). IR camera 1880Hz SI and DI Diesel engines. Simultaneous IR emissions @ 2.1 μ m, 2.47 μ m (water), 3.43 μ m (preflame) and 3.8 μ m (soot). HITRANS modelling. HCHO signal weak for SI engines.

SAE 2011-01-1395: Squibb (Michigan State). FLIR Phoenix MID, 60Hz. 3-5 μ m. 20 μ s integration time. Fuel sprays and combustion visualization. In-cylinder temperature.