

# Atomistic and Coarse-Grained Simulations of PNIPAM

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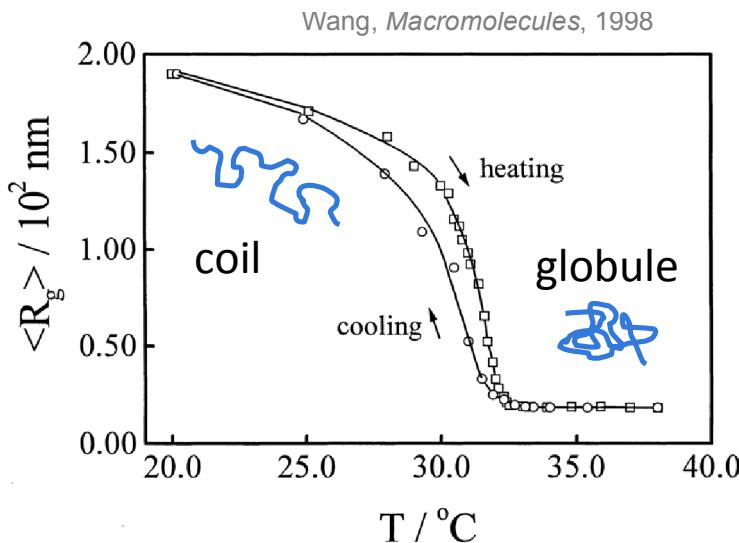
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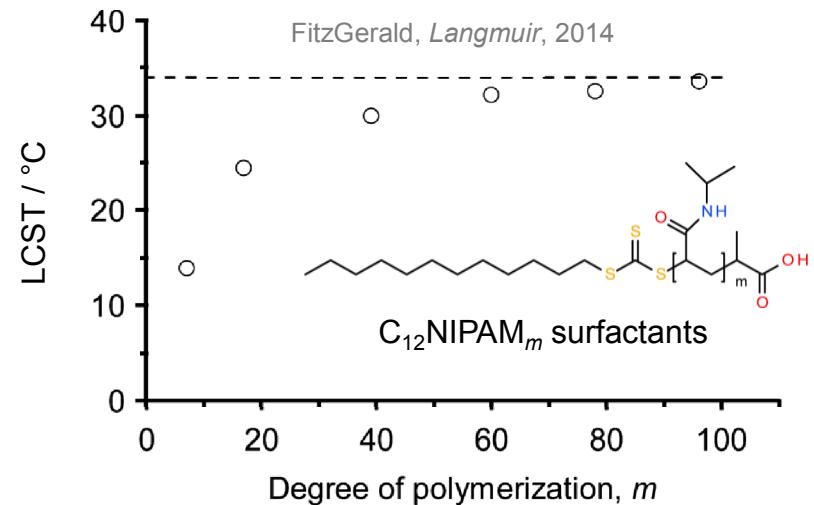
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# PNIPAM LCST & Copolymer

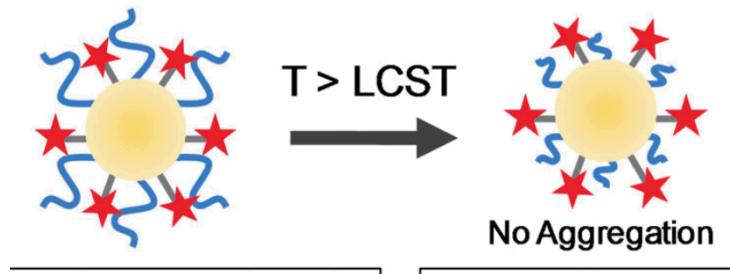
PNIPAM has LCST at  $\sim 32\text{C}$



PNIPAM copolymers

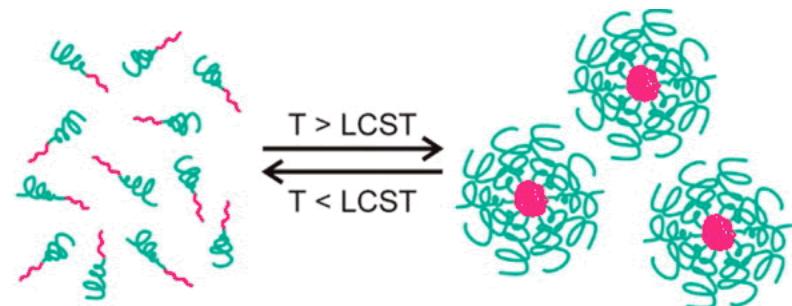


Responsive surface coatings  
(e.g., sensors, catalysis)



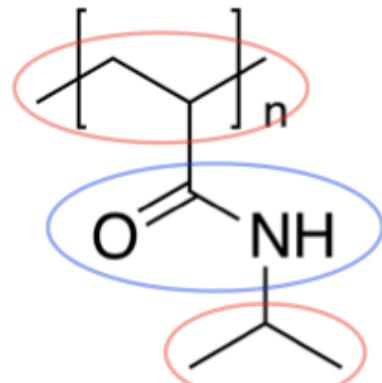
Gibson, *Chem. Soc. Rev.*, 2013

Responsive self-assemblies



Lee, *Macromolecules*, 2013

# PNIPAM



hydrophobic

hydrophilic

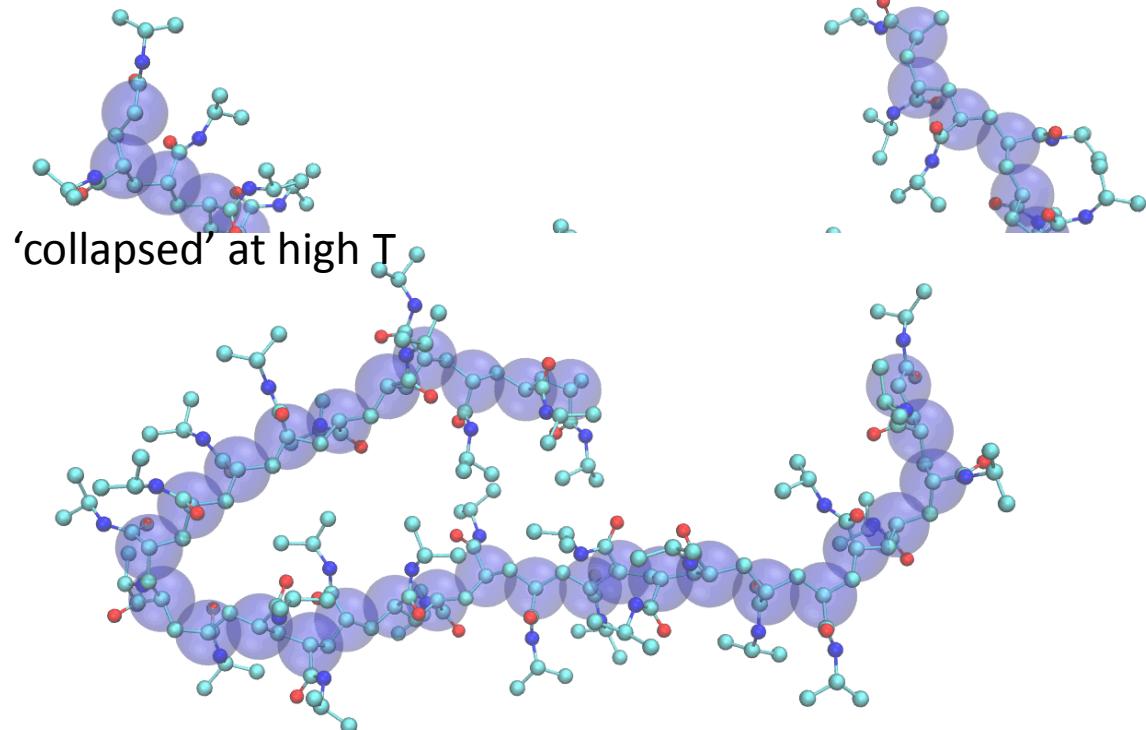
hydrophobic

$N = 30$

Single chain in water

LCST

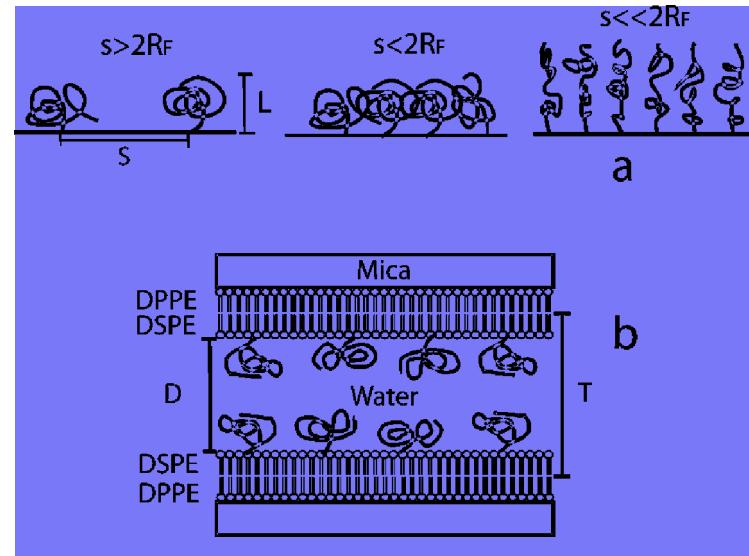
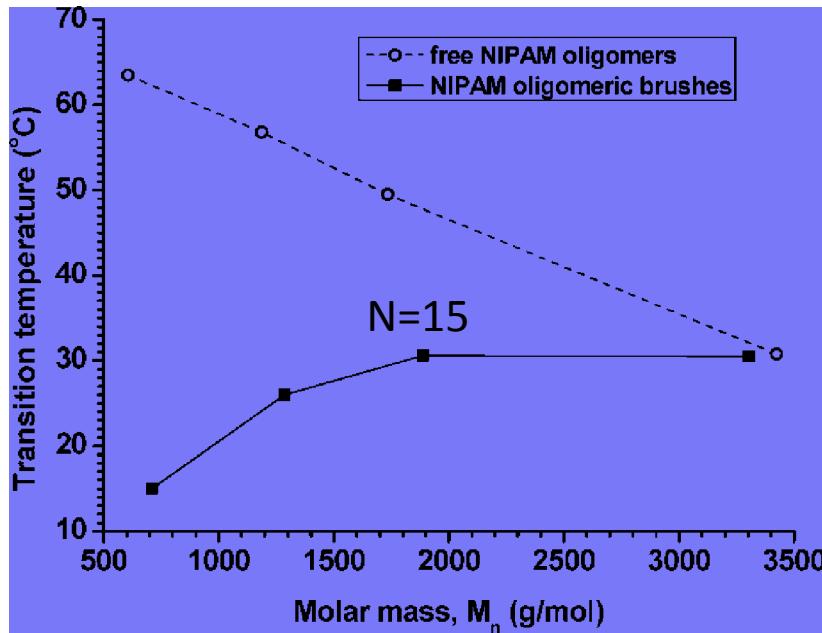
extended at low T



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# PNIPAM Oligomers



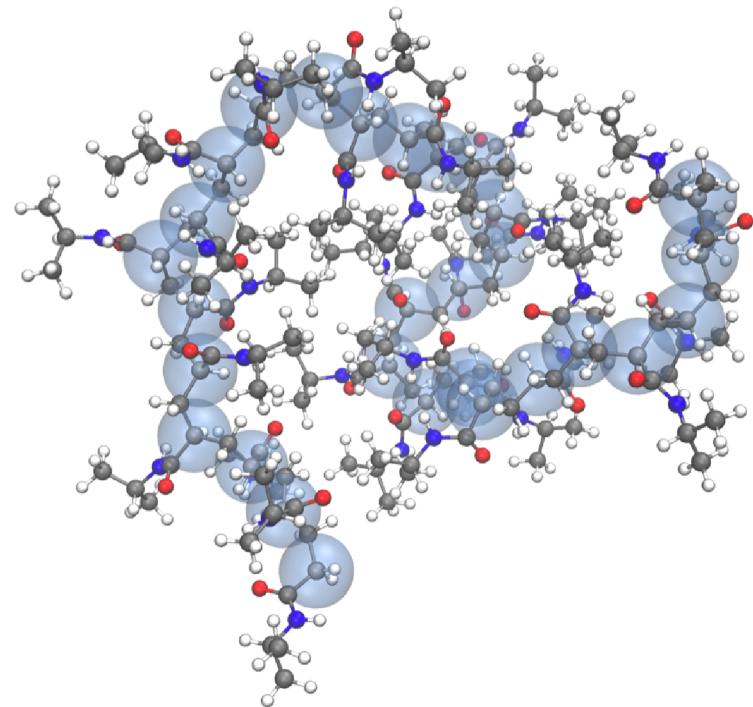
Shan et al. Macromolecules 2009  
PNIPAM coated nanoparticles  
Chain length dependent transition

Winnick & Leckband et al. Langmuir '07  
SFA measurements of grafted PNIPAM.  
At low grafting densities, short PNIPAM  
does not collapse above the LCST.

# Oligomer simulations

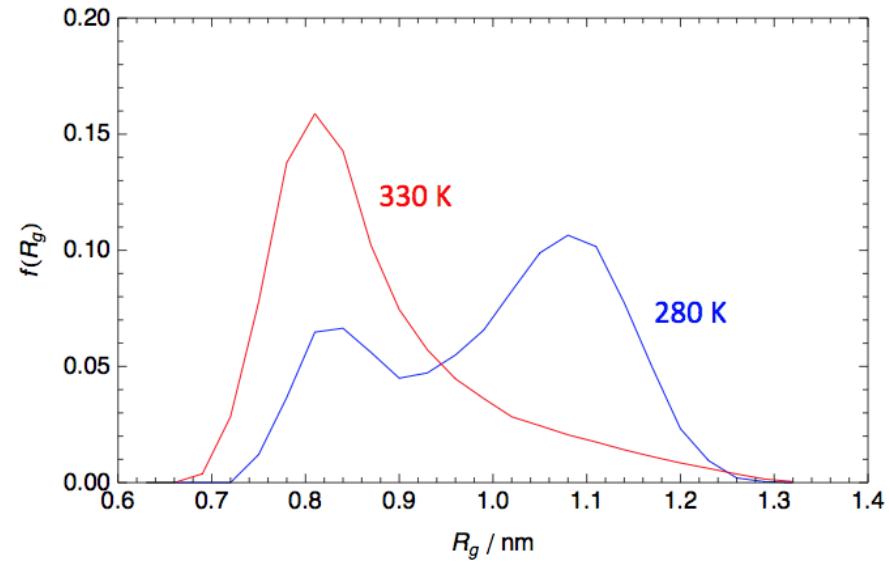
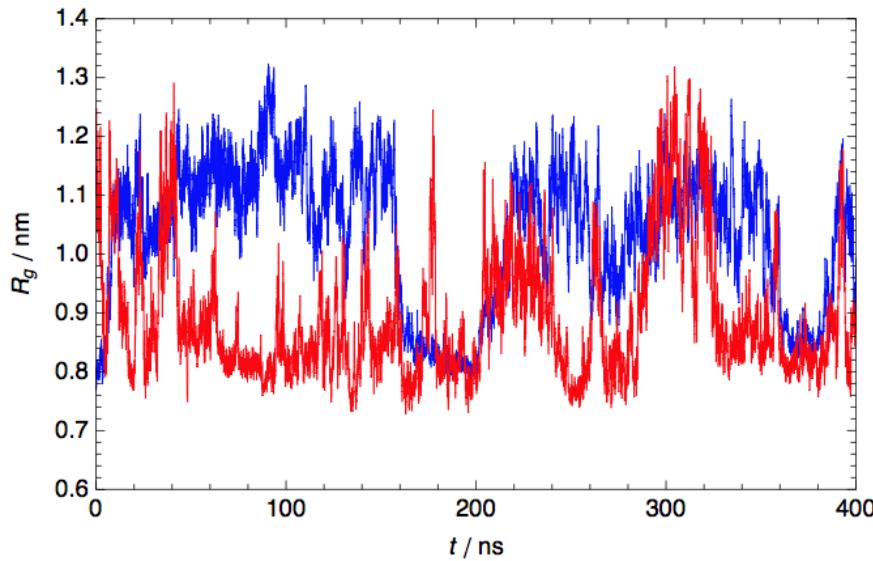
Single chain in water

- OPLS with modified dihedrals (Siu, *JCTC*, 2012)
- TIP4P/2005 water model
- Gromacs 4.6.5
- **100-400 ns production in NPT ensemble**
- 2 fs time step
- Canonical thermostat
- Parrinello-Rahman barostat
- Short-range nonbond with 1.0 nm cutoff
- Long-range electrostatics with PME
- Cubic box with PBC, 4.0-9.0 nm



# PNIPAM dynamics

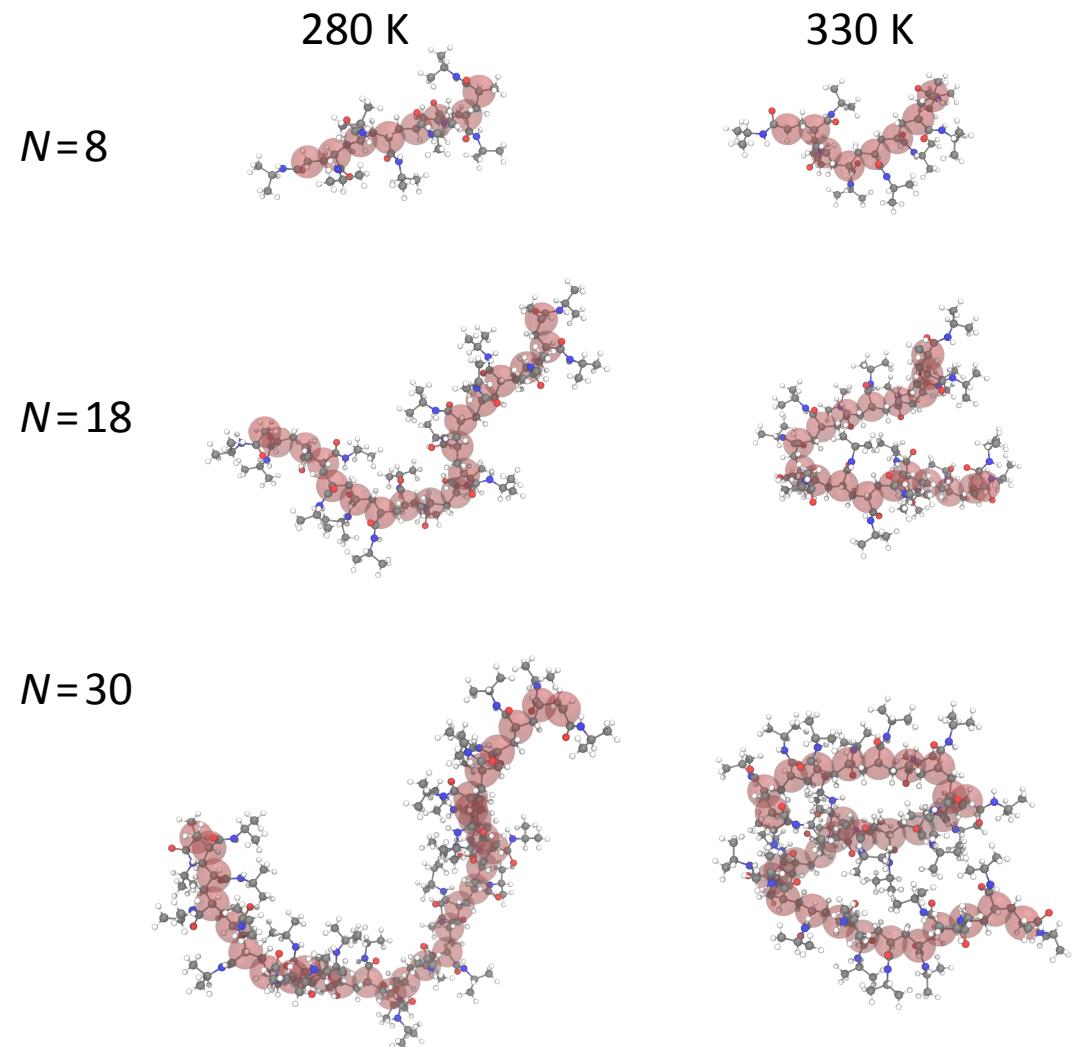
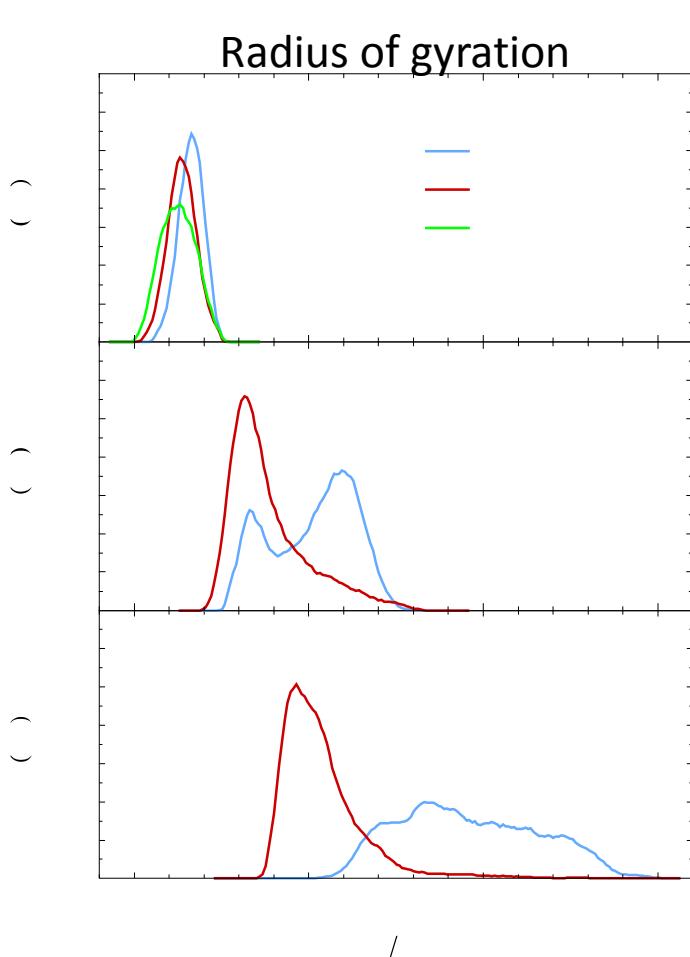
N=18 400 ns simulation. Yes, have to run *very* long to get distributions accurately.



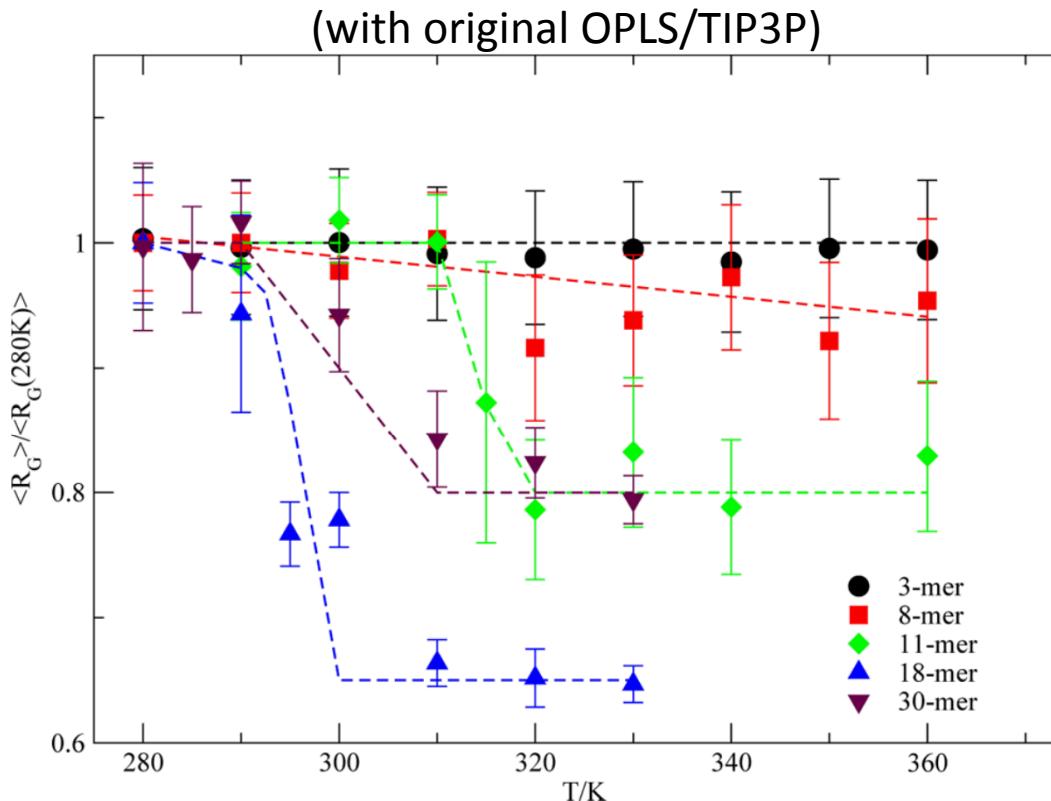
Hydrogen bonding dynamics results in slow variation in structure and demands very long simulation times.

Equilibration determined in terms of **distribution** of  $R_g$ .  
Adds to motivation for coarse-grained model.

# Oligomer Structure



# Oligomer LCST



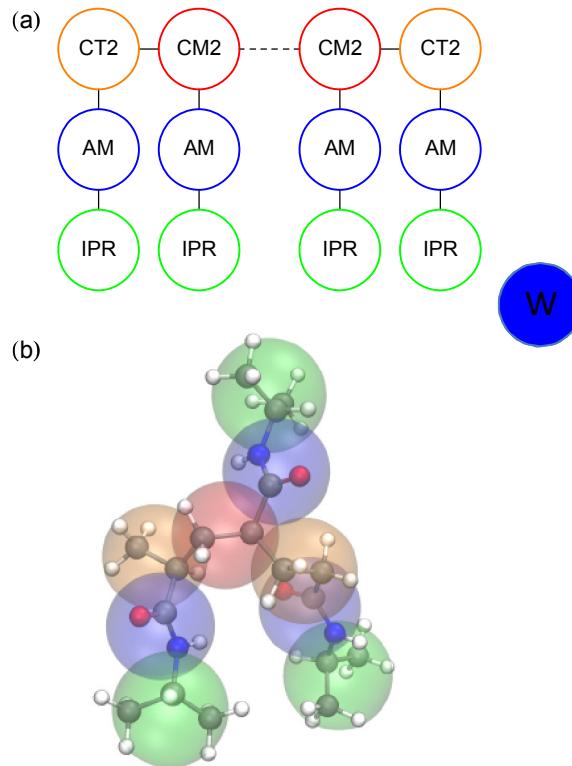
Minimum length required for a single chain to bend  $\Rightarrow$   
Minimum length required for *single* chain LCST

This corresponds to the experimental behaviors seen.

# Coarse-graining PNIPAM

**Need** CG model to treat the interesting PNIPAM systems.

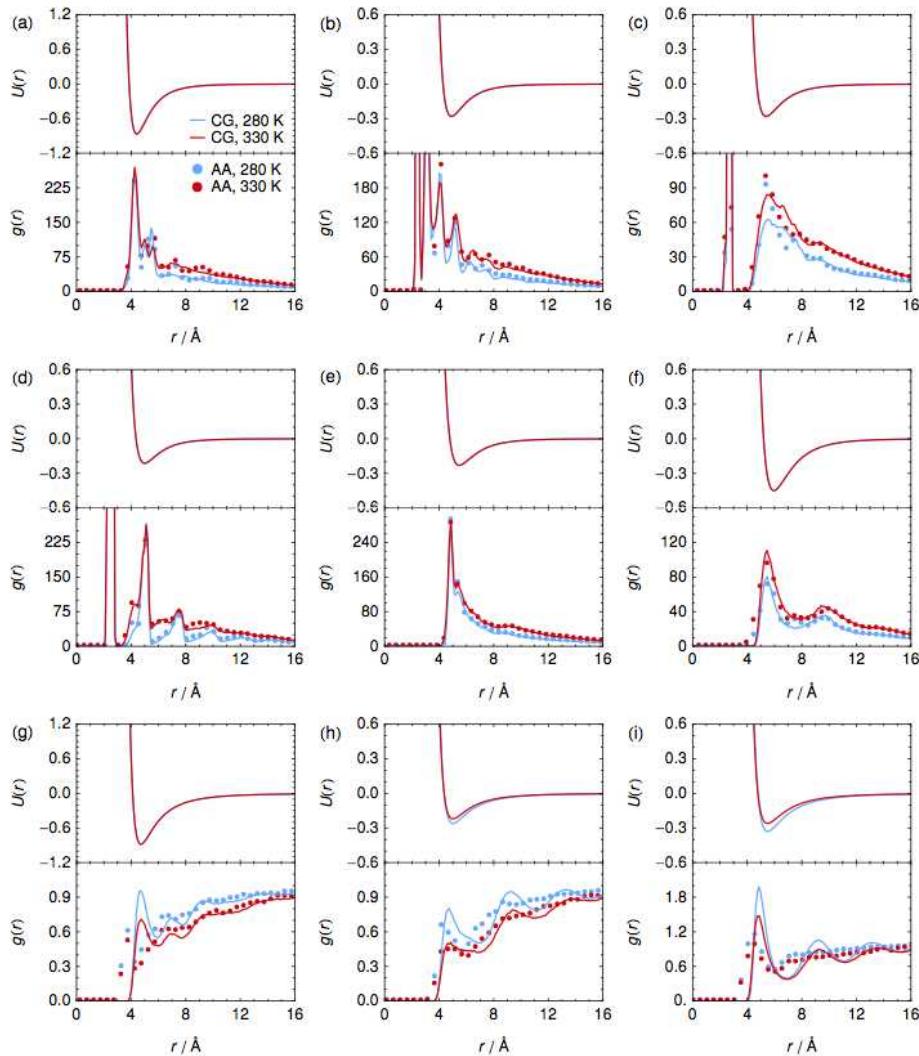
- relaxation/equilibration slow
- gels are large
- micelles slower



1. implicit solvent model does not work
2. ?Koga JPCB 2016: water + LJ polymer
3. Started with CG model of Shinoda, DeVane, Klein
  - 0) Two parameter sets for 280K and 330K
  - 1) New bead types for AM & IPR and CM2
  - 2) Match experimental thermodynamic data for analogous chemical compounds at 300 K.
  - 3) CG bond parameters were fit to atomistic distributions for PNIPAM single chains at 280 K and 330 K.
  - 4) Nonbonded parameters (from step 2) were adjusted to yield better agreement with RDFs for PNIPAM single chains at 280 K and 330 K.  
(Iterative Boltzmann + Tweaking)

# Lots of cross-term calculations

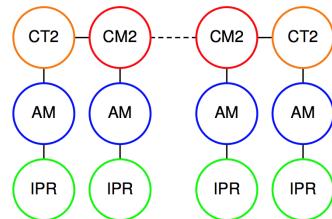
- (a) AM–AM
- (b) AM–CM2
- (c) AM–IPR
- (d) CM2–CM2
- (e) CM2–IPR
- (f) IPR–IPR
- (g) AM–W
- (h) CM2–W
- (i) IPR–W



280 K

330 K

points are atomistic data



# Adjustments

The starting LJ parameters are for  $\sigma_{AM} = \sigma_{AM-AM}$

$\sigma_{AM-AM}$  reflects a bead size for hydrogen bond formation between AM & AM.

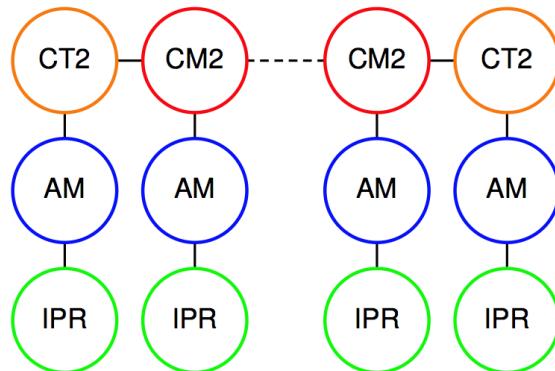
For AM-X, where X is hydrophobic,

cannot use Berthelot combination rules

must adjust  $\sigma_{AM-X}$  to be larger

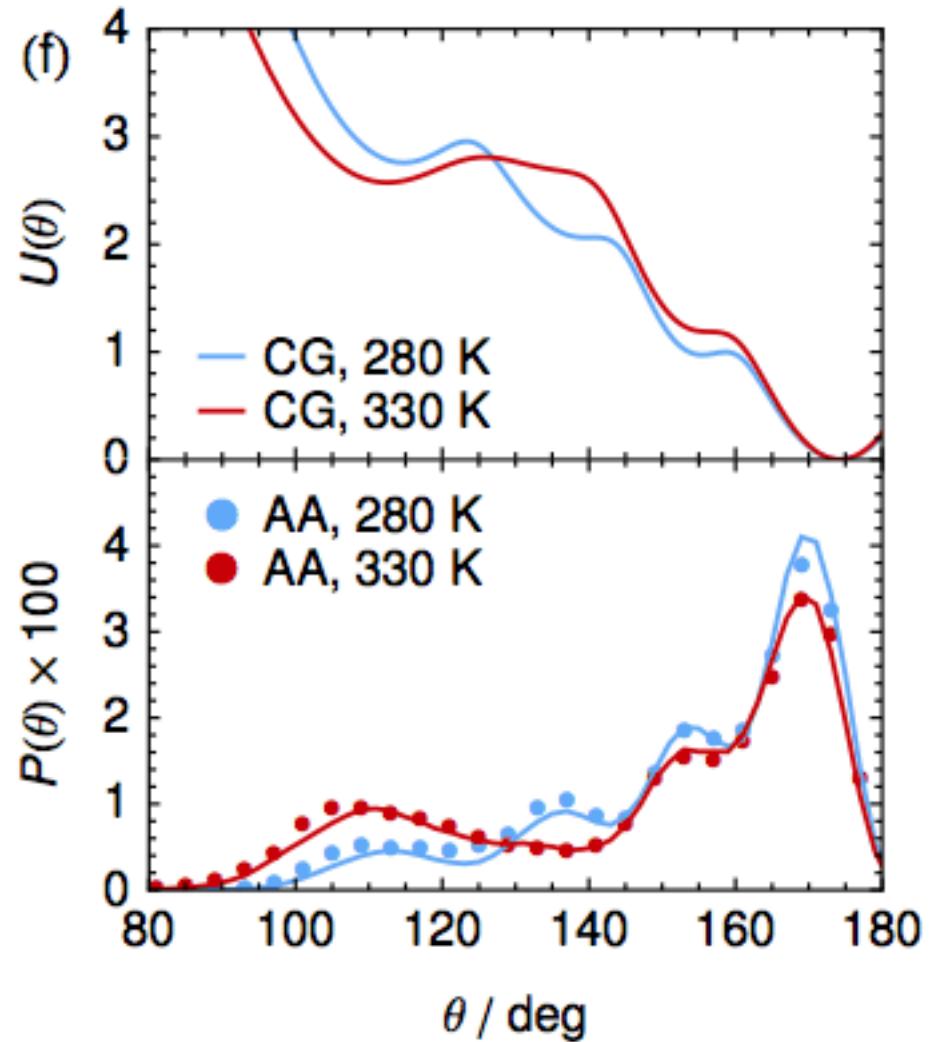
to get correct rdf peak positions

$\varepsilon_{AM-X}$  was decreased to maintain the same interaction strength at long distances.



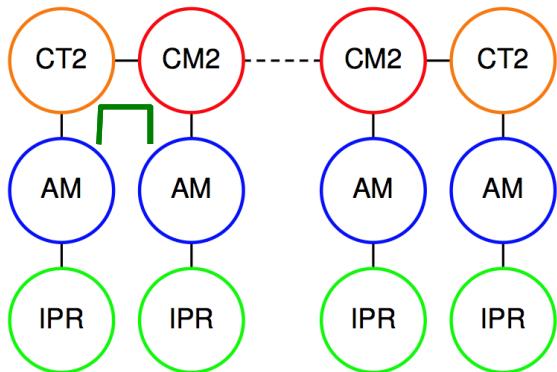
# More T-dependence

Backbone CM2-CM2-CM2 angle  
Can match with CG for each T.

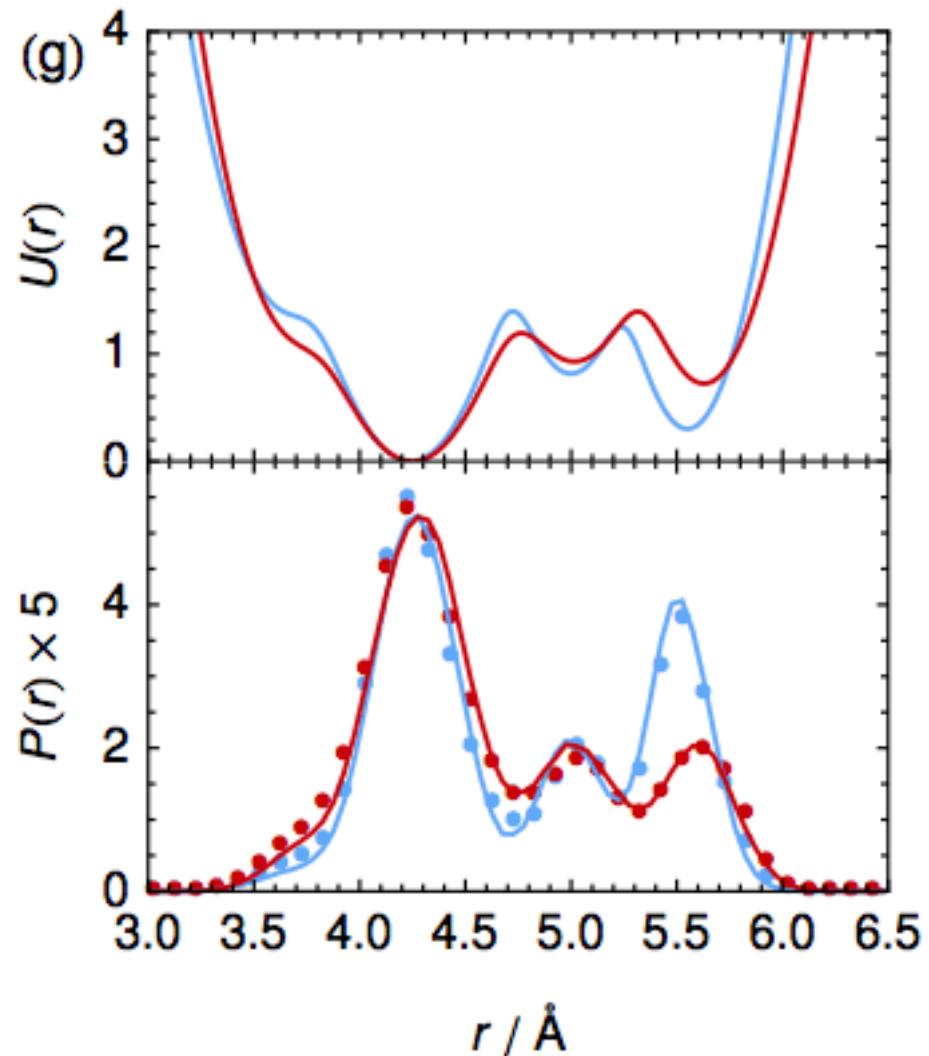


# More T-dependence

AM-AM 1-4 interaction



tacticity



# CG PNIPAM Model

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## Summary points

- lots of parameters
- interdependent and sensitive
- need some parameters at two temperatures
- better model would incorporate hydrogen bonding,  
but may have fewer atoms/CG bead

But it works!

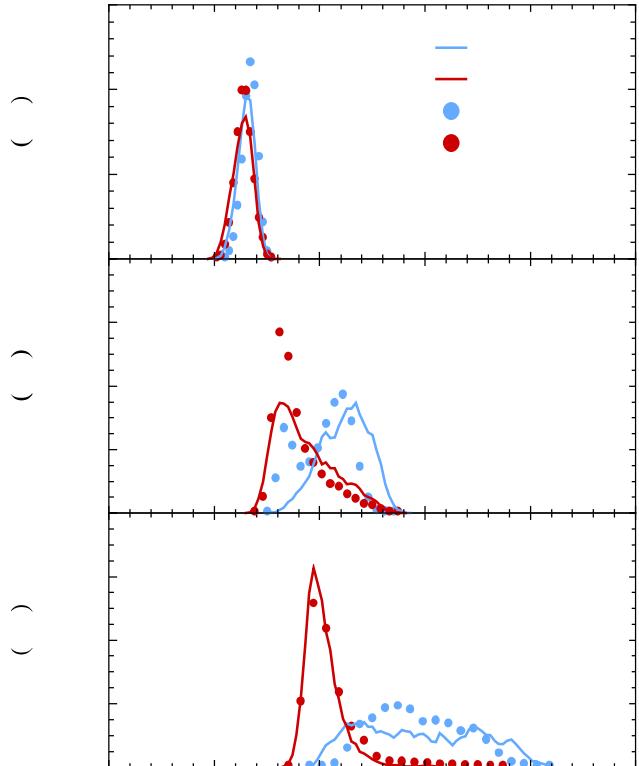


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# How good is CG model?

Radius of gyration



$N=8$

280K

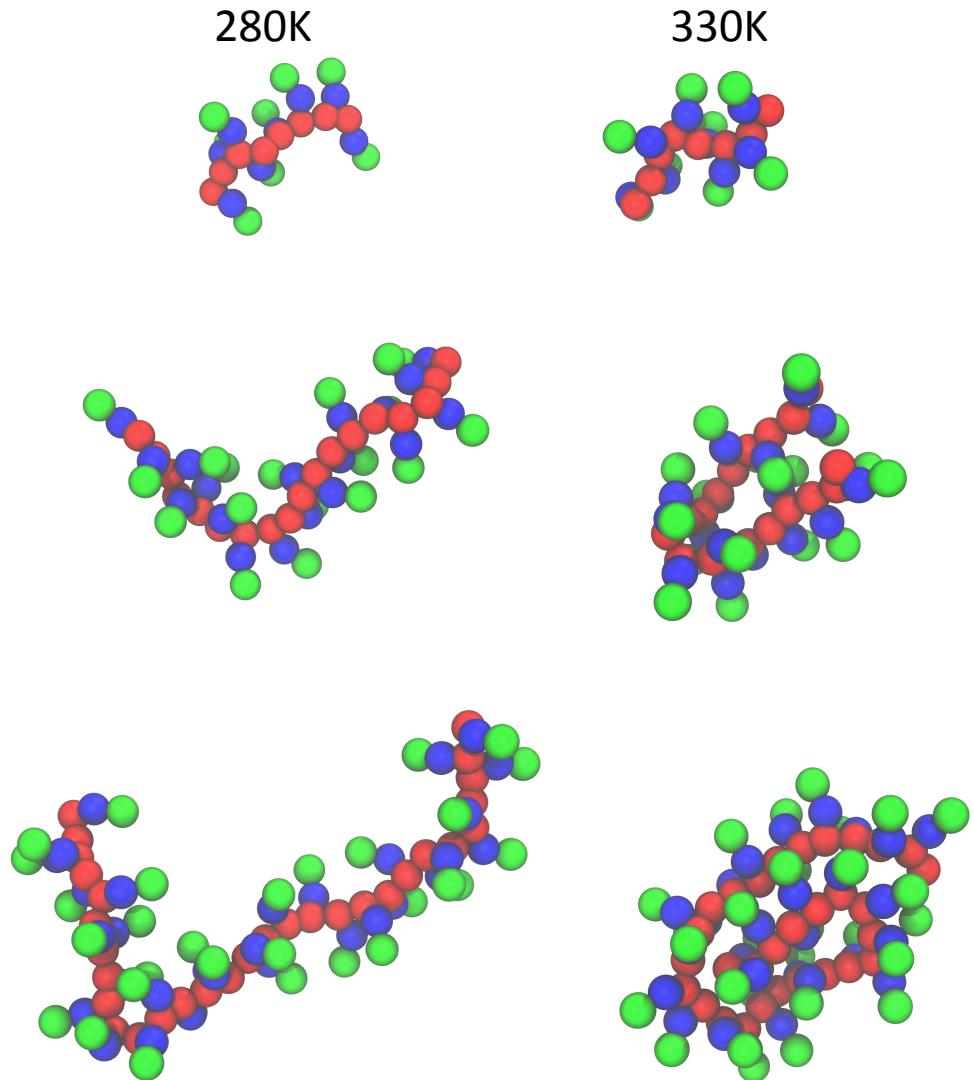
330K

$N=18$

$N=30$

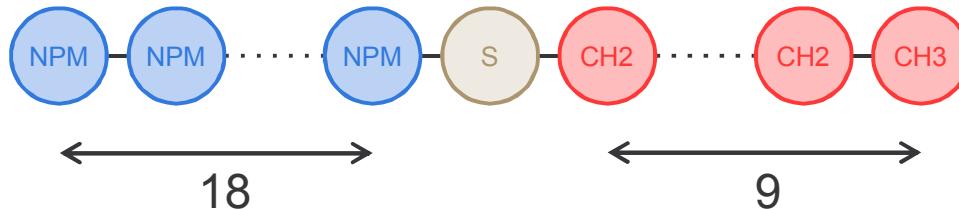
/

CG model yields correct single chain structure and  $N$  dependence.

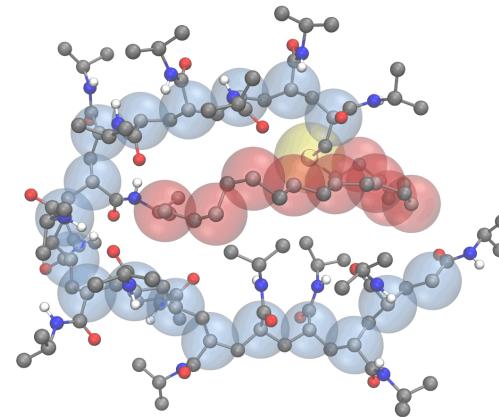


# Atomistic Simulations of Single Copolymer

PNIPAM-C18 surfactant:

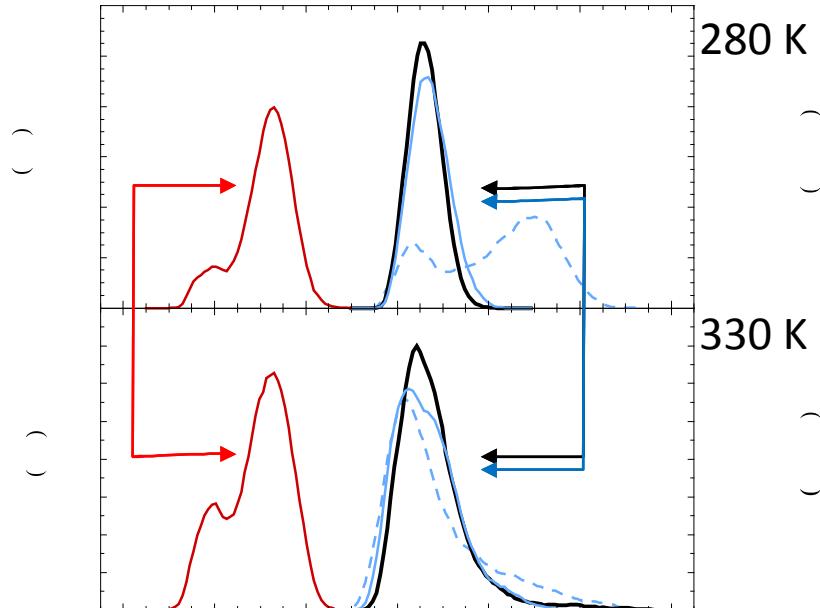


or 36 C in backbone

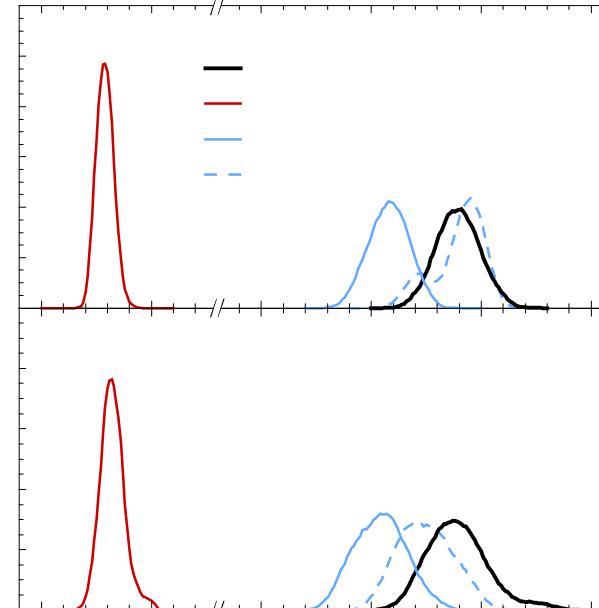


# Copolymer Structure

Radius of gyration



Solvent accessible surface area



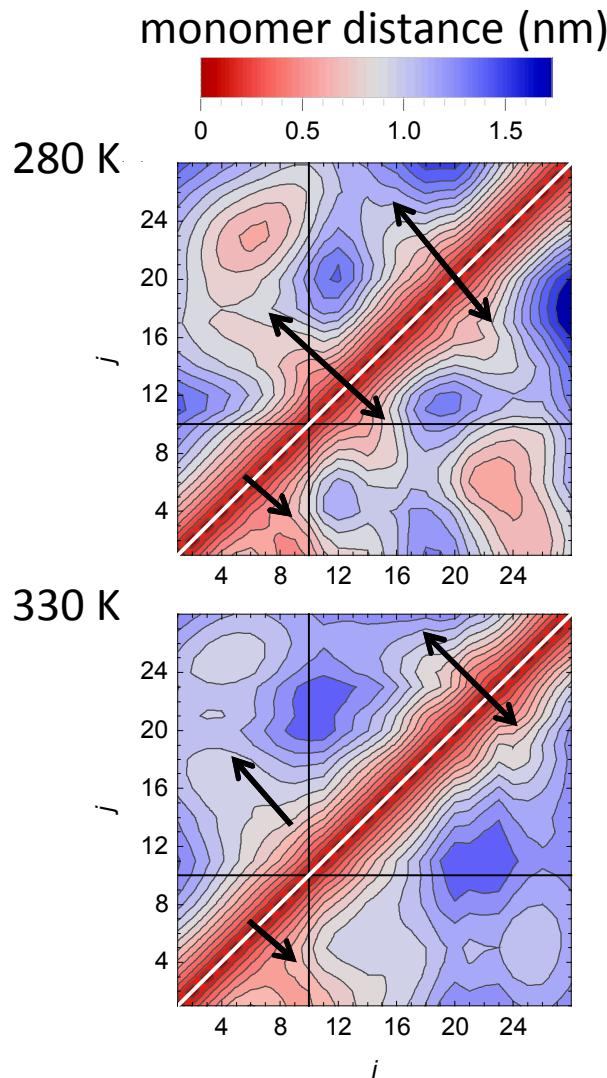
Whole Chain is compact at both T.

PNIPAM part

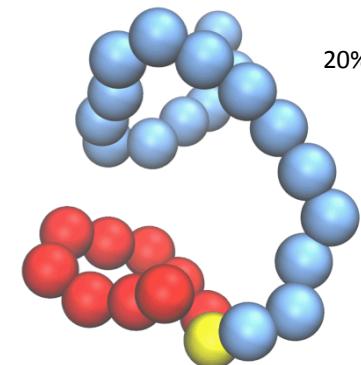
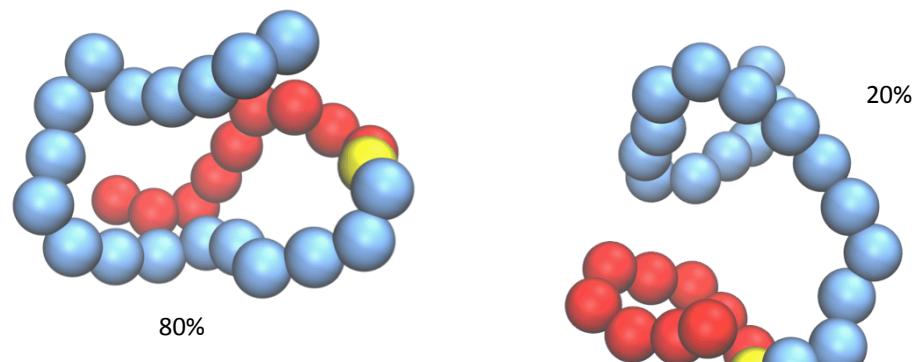
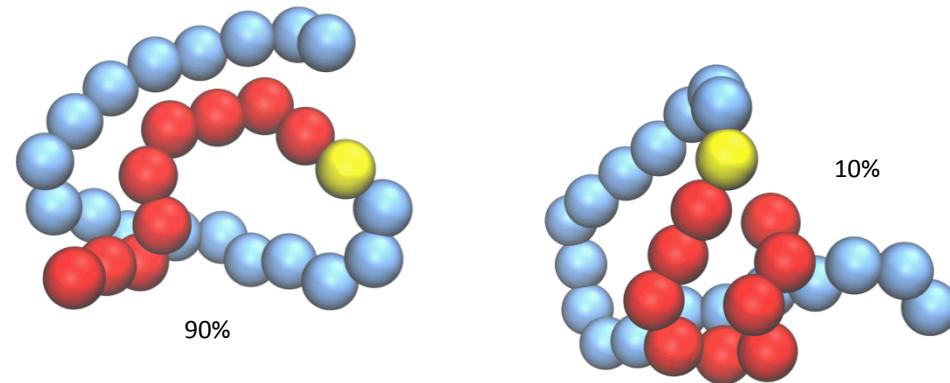
Alkane part

# Copolymer Structure

## Contact Maps

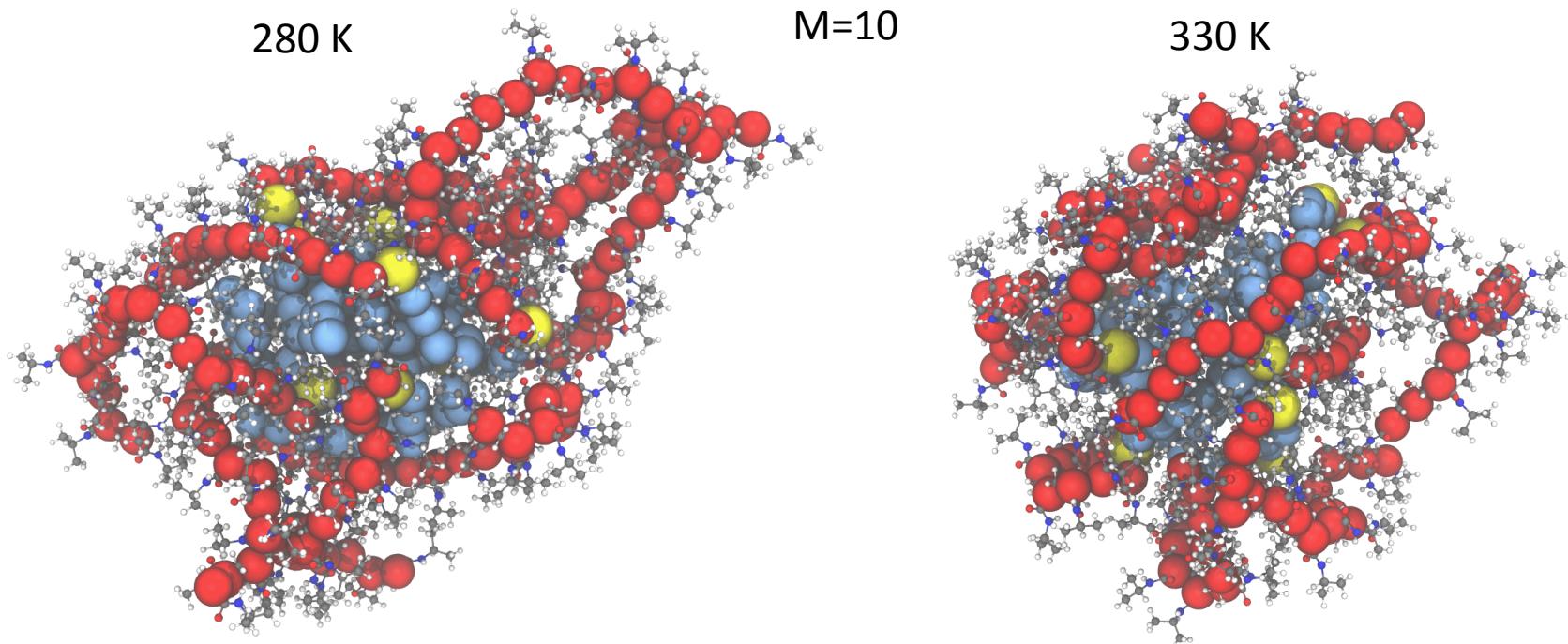


In majority structure, alkane part bends to be next to PNIPAM backbone

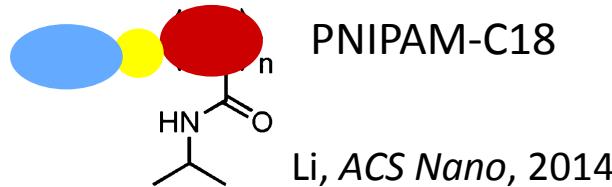


1 bead = 2 C atoms

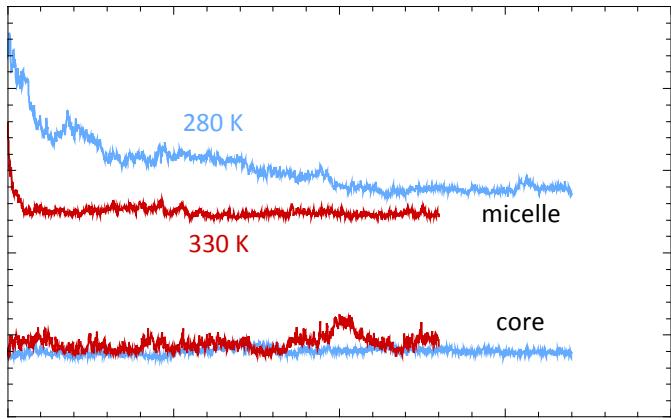
# Atomistic Micelle Simulations



Prebuilt micelles show T-responsive behavior  
Below LCST, larger PNIPAM outer region

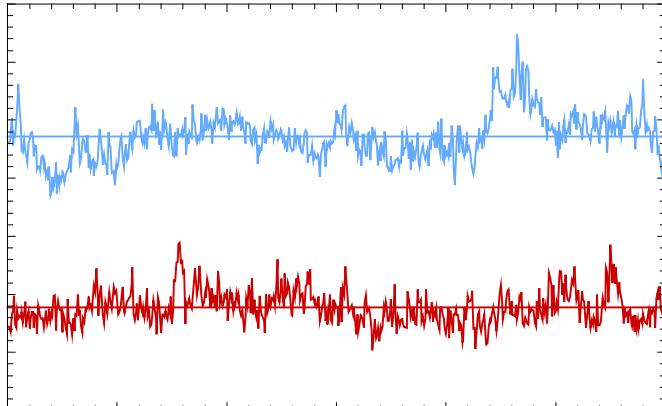


# Micelle microsecond simulations



after initial transient

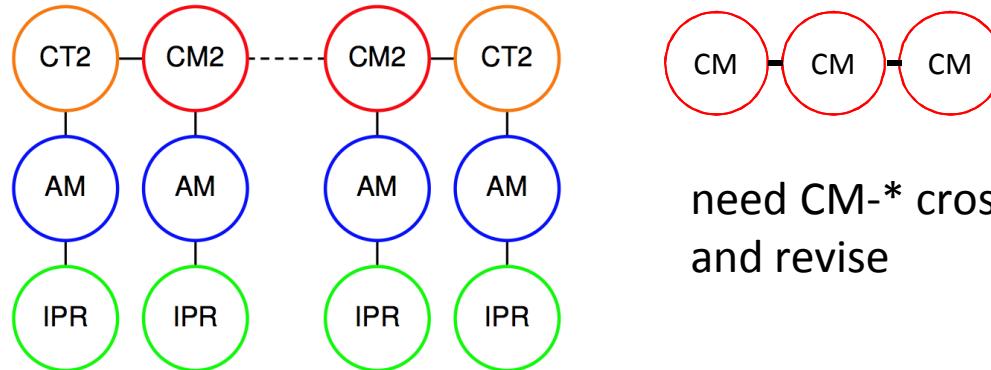
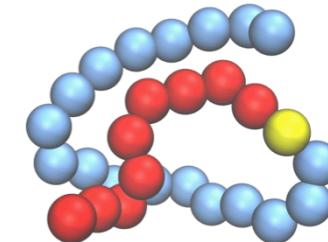
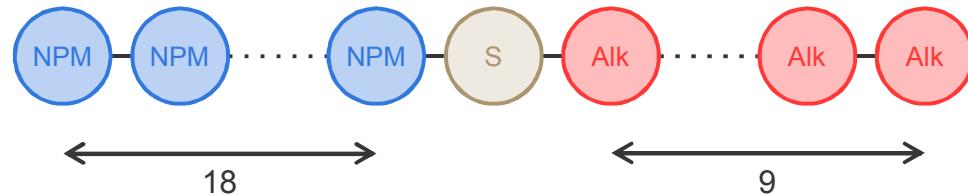
Whole Micelle



CG model are really needed to treat micelles and large systems, in general.

# CG model of Copolymer

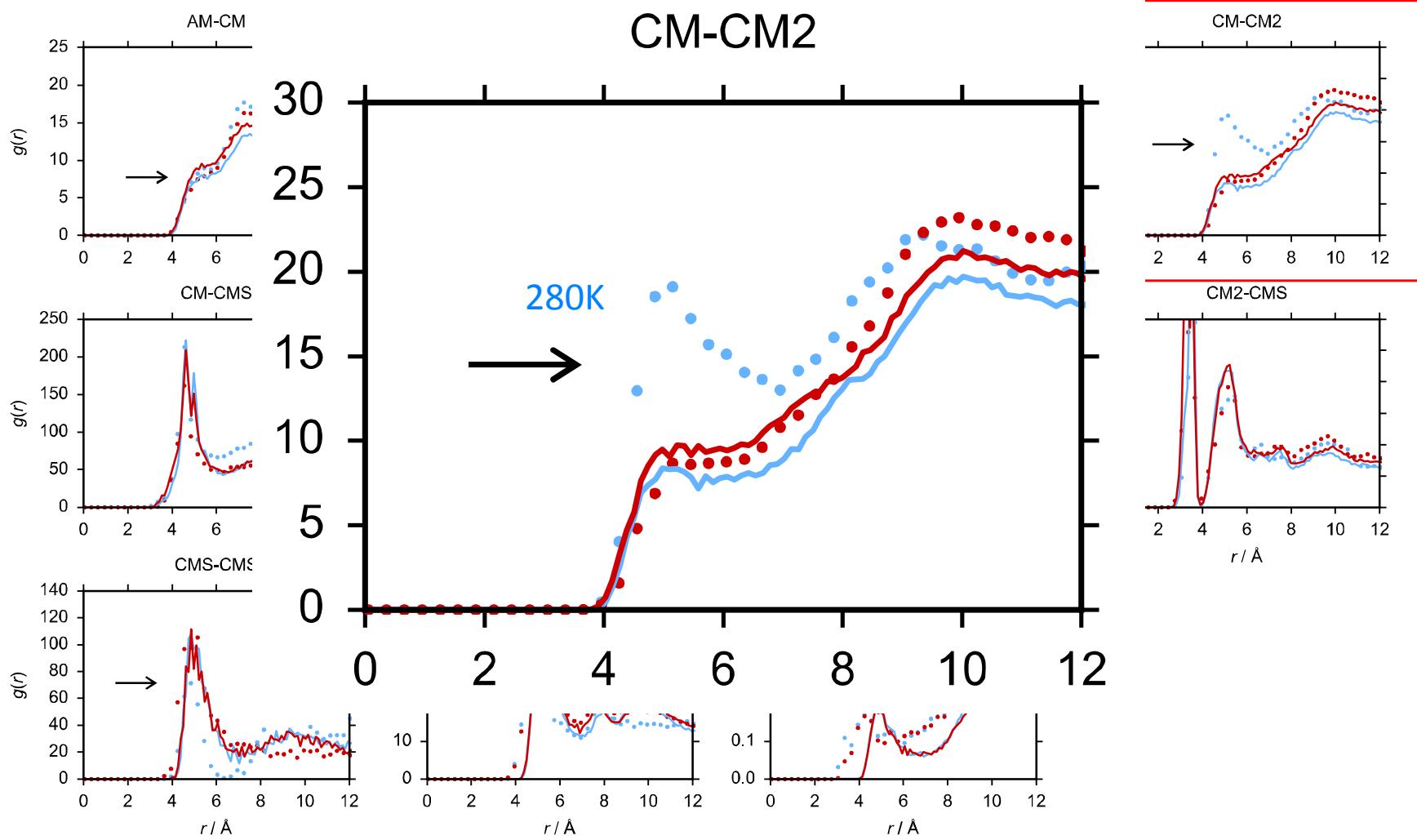
PNIPAM-C18 surfactant:



need CM-\* cross terms  
and revise

Work in Progress

# Cross-Term $g(r)$ comparison



# Tweaking nonbond parameters

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To better match peak intensities in  $g(r)$  [in kcal/mol]

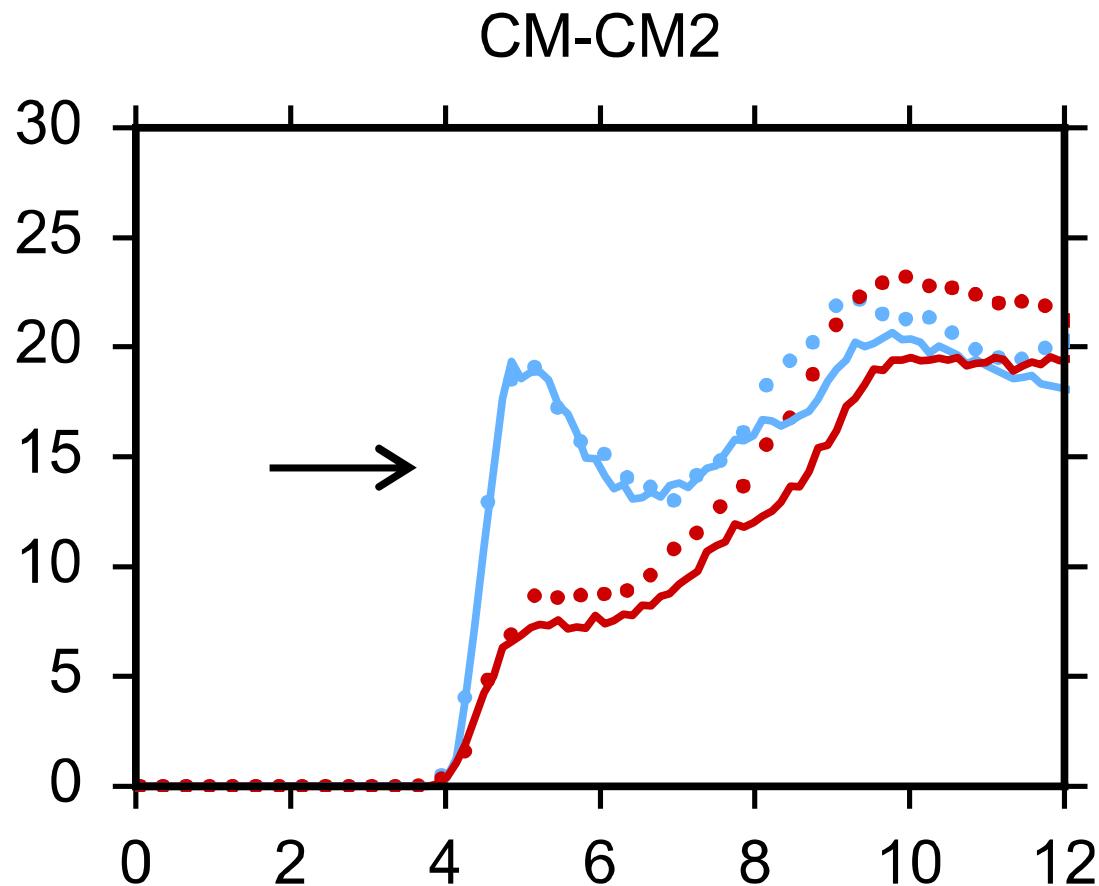
1. Decreased  $\epsilon_{\text{AM-AM}}$ : 0.39 to 0.20
2. Decreased  $\epsilon_{\text{CM-W}}$ : 0.34 to 0.28
3. Decreased  $\epsilon_{\text{CMS-IPR}}$ : 0.55 to 0.40

For better matching T dependence:

1. Increased  $\epsilon_{\text{CM-CM2}}$ : 0.2984 to 0.45/0.35
2. Decreased  $\epsilon_{\text{CM-IPR}}$ : 0.32 to 0.18/0.30
3. Decreased  $\epsilon_{\text{CMS-CM2}}$ : 0.82 to 0.30/0.40



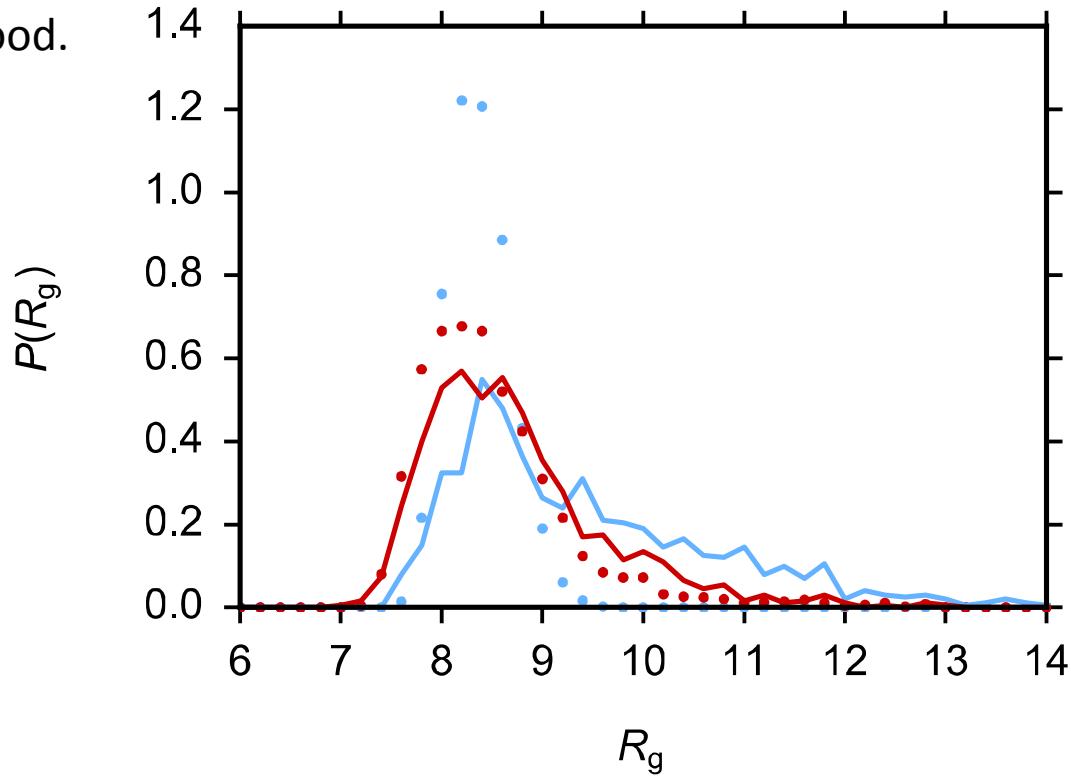
# Better Matching



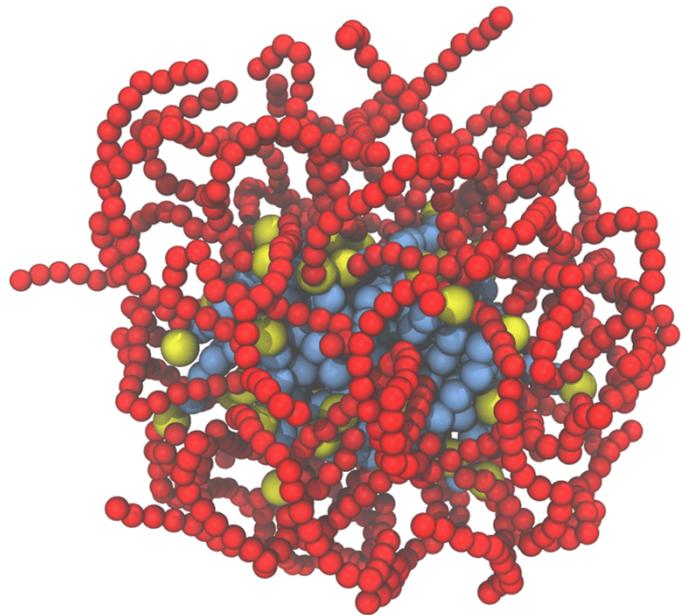
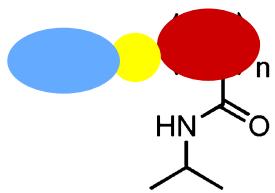
# CG Copolymer Structure

$R_g$  distribution at 330K is good.

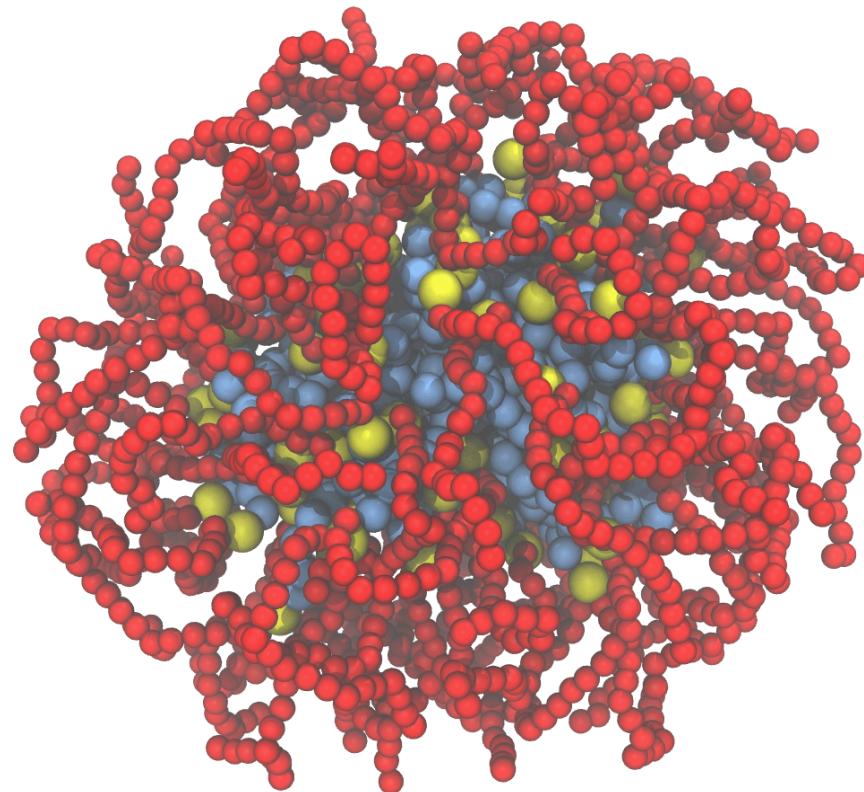
But, needs work at 280K



# CG will enable Micelle simulations.



$M = 50, 330 \text{ K}$



$M = 100, 330 \text{ K}$

# Conclusions

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## Atomistic simulations

Minimum length is required for backbone to bend.

No structural transition for short oligomers.

In copolymer single chain, the alkane part bends back to be next to nipam backbone.

In prebuilt micelles of copolymer with PNIPAM on outside,  
T dependence does appear in PNIPAM structure

## CG Model

Have PNIPAM CG model at two T (above & below LCST)

Matches atomistic structure

Copolymer model and micelles are work in progress

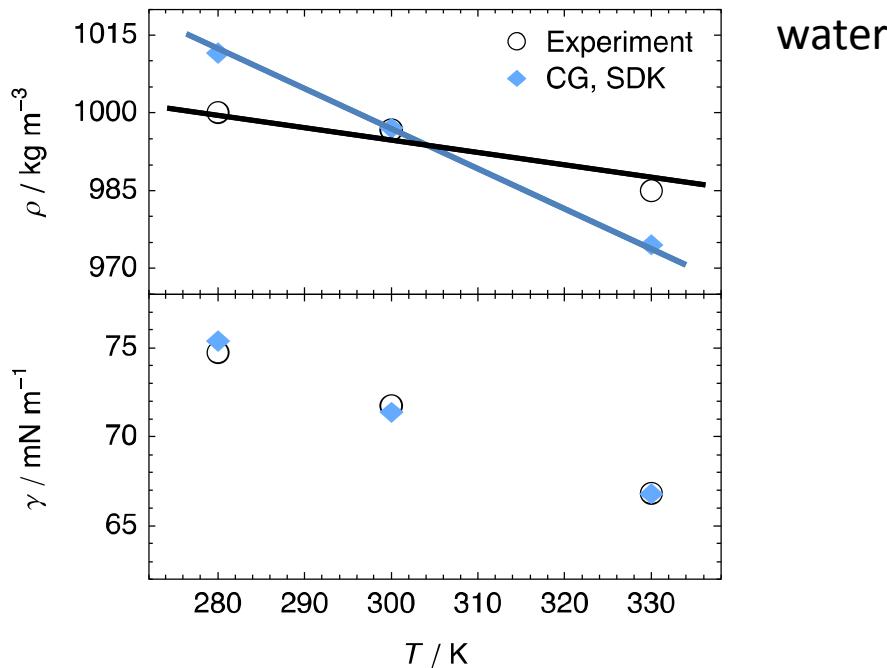
## Lessons in Coarse-Graining

CG models with many types (and T dependence) are complex.

Need a better CG water model.



# Limitations & Weakness in CG method



LJ variants of CG water do not get the density as a function T correct.

This is a critical aspect of the hydrophobic effect(s).

Koga JPCB 2016

Missing hydrogen bonding in almost all CG models.

No CG model for *multiple* waters exists that does this.

Molinari water model for a single water molecule does.

Plus need means to treat other hydrogen bonding groups, e.g. amides in PNIPAM.

What is entropy & enthalpy of PNIPAM & water as a function of T?

# Acknowledgements

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## Acknowledgements

Ashley Tucker, Lauren Abbott

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- A. K. Tucker and Mark J. Stevens, Macromolecules **45**, 6697 (2012).
- L. J. Abbott, A. K. Tucker, and Mark J. Stevens, J. Phys. Chem. B, **119**, 3837 (2015).
- L. J. Abbott and Mark J. Stevens, J. Chem. Phys. **143**, 244901 (2015).



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