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Preparation and Characterization of UO_2 Epitaxial Thin Films



Bonnie Klamm, Ryan F. Hess, Khalid
Hattar, Timothy J. Boyle, Diana
Perales, and Remi Dingreville



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Outline

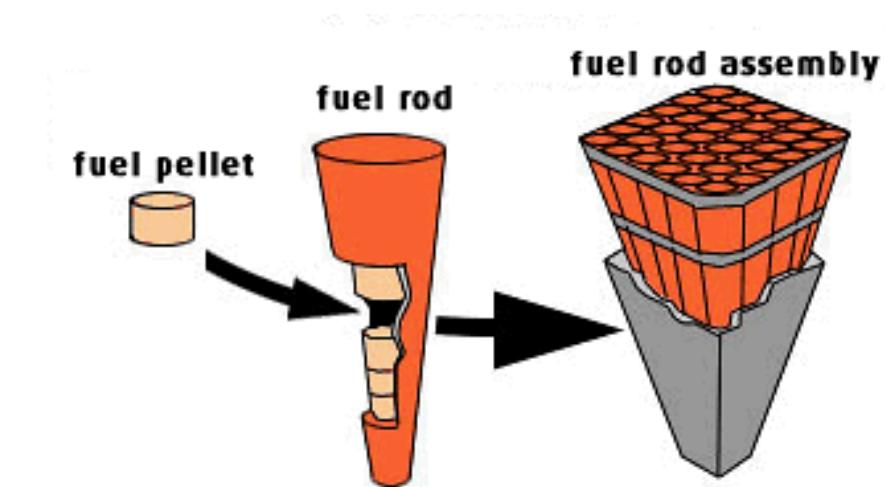
1. Recreate surface chemistry

- UO_2 -Zircaloy interface (Pellet Cladding Interface)
 - Profilometry – film thickness
 - Transmission Electron Microscopy – microstructure information
 - Powder X-ray Diffraction – phase identification

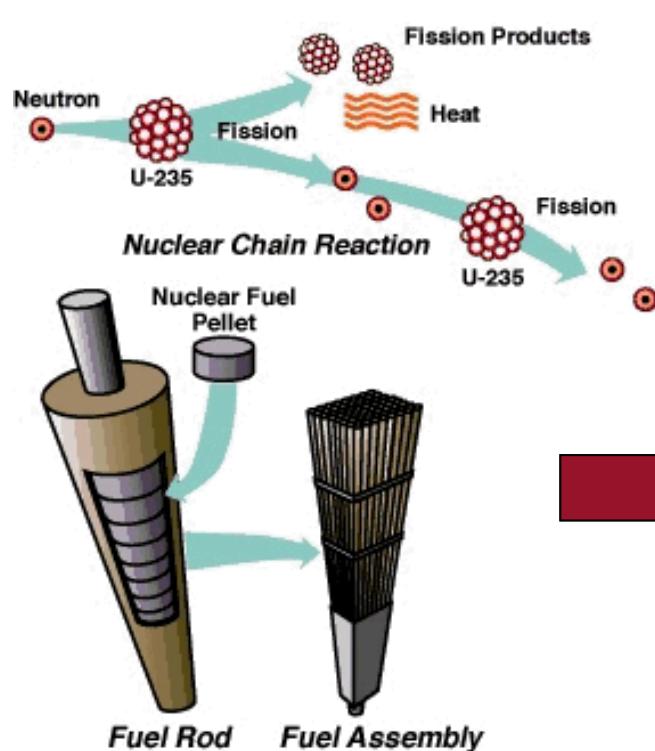
2. Simulate radiation damage

- In-situ ion bombardment
- Compare to INL spent fuel

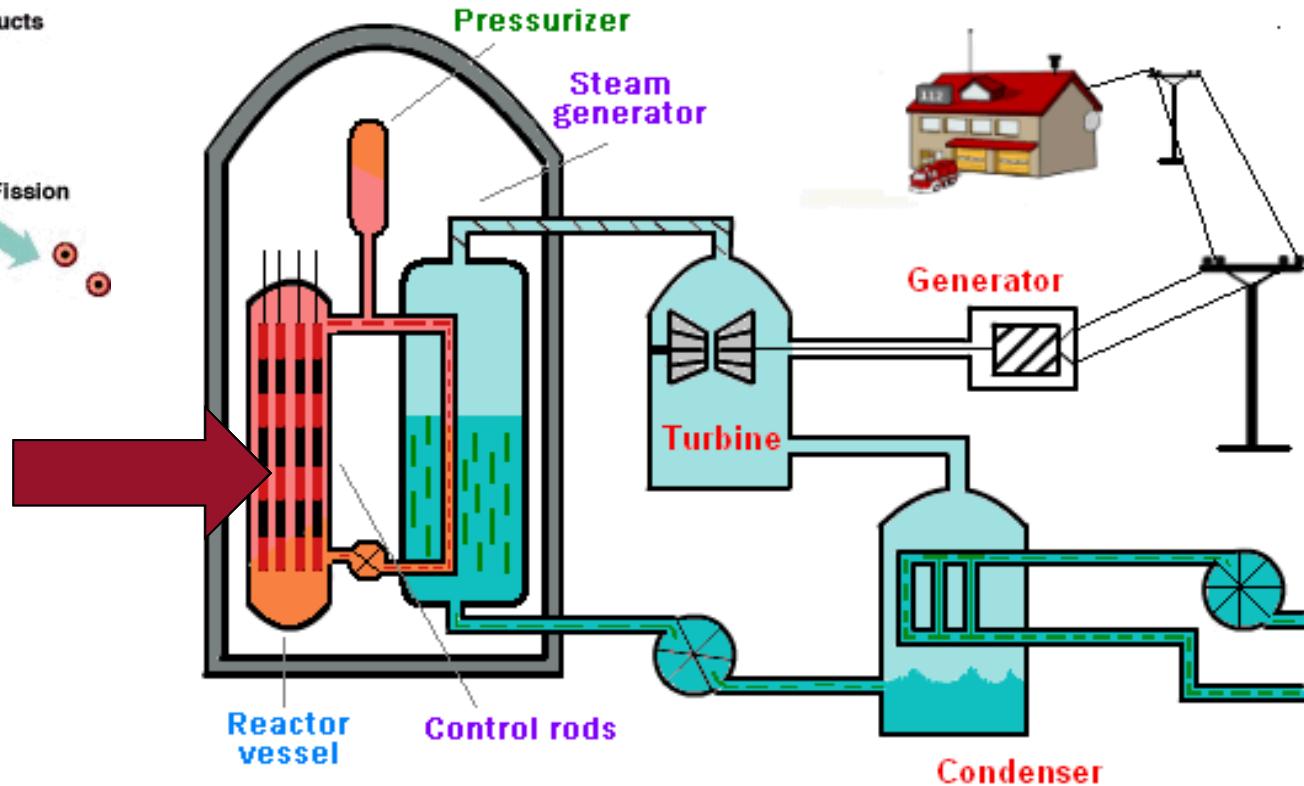
3. Characterize damage



Nuclear Reactor Overview

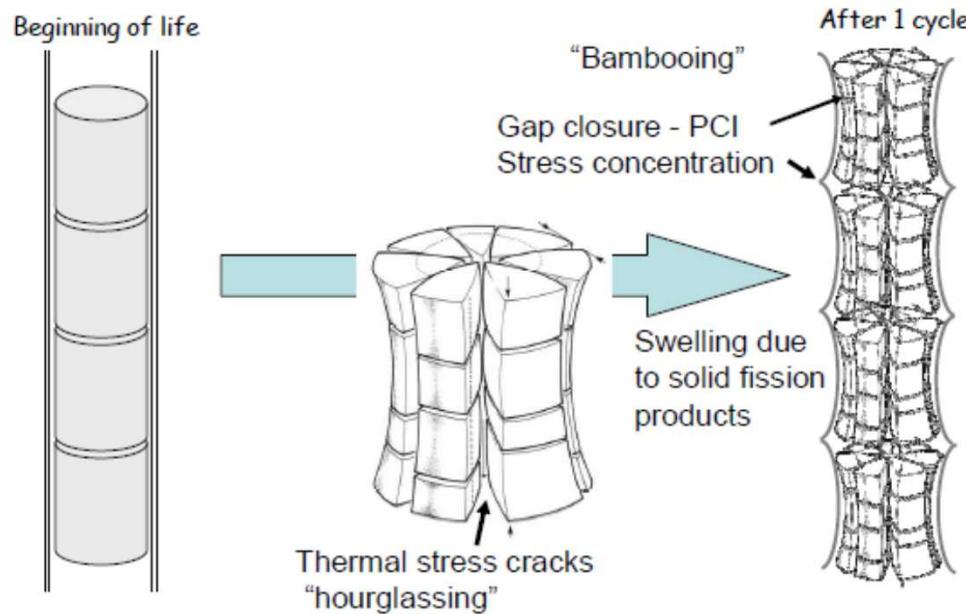


Containment Structure



Pressurized Water Reactor

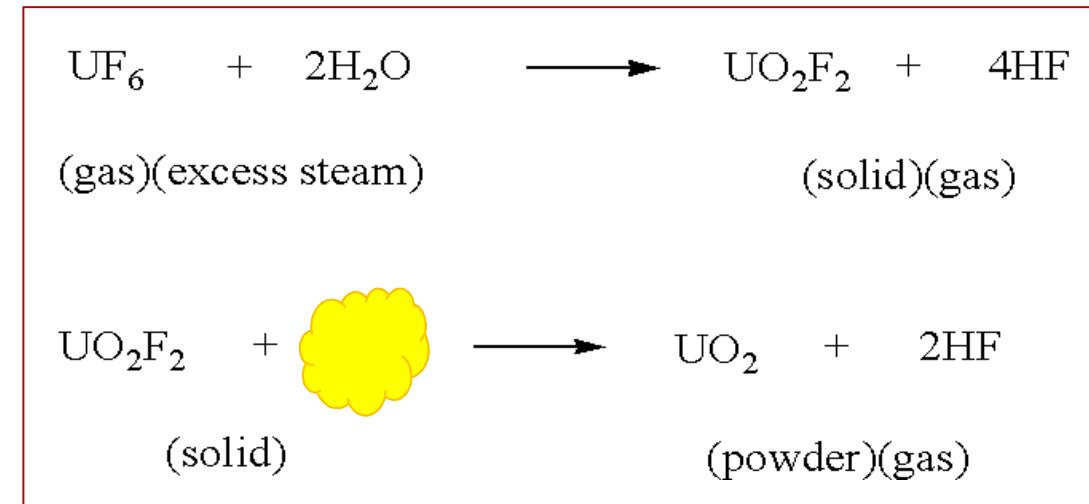
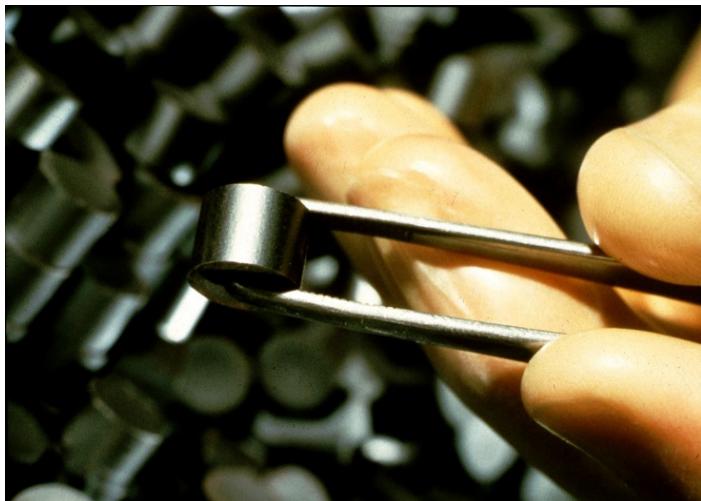
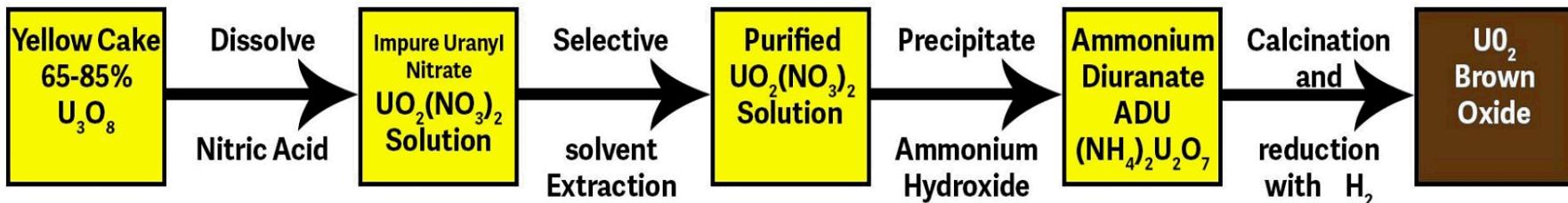
Reactor Chamber



**What happens at
the microscale?**

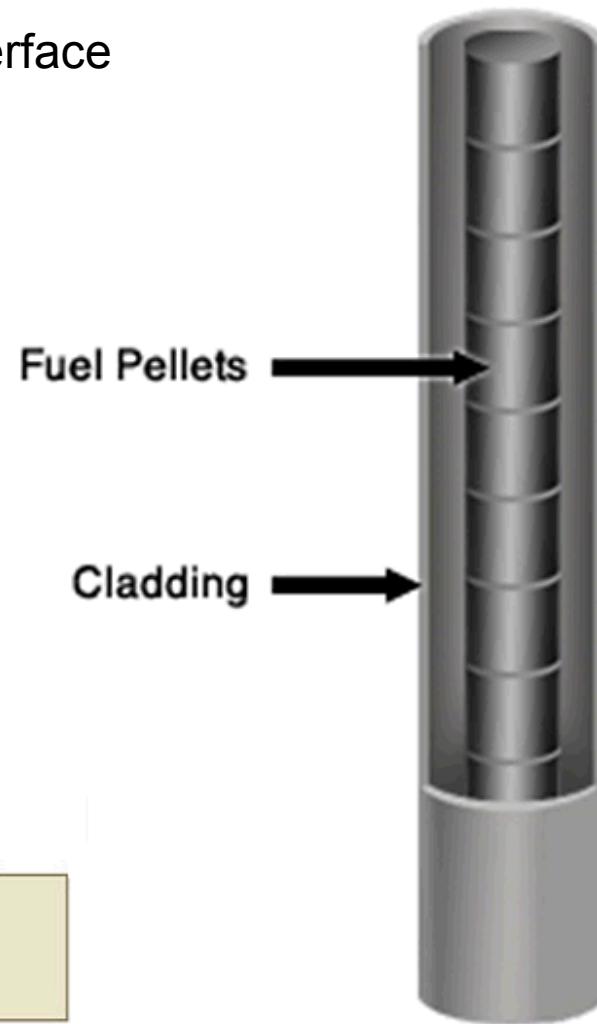
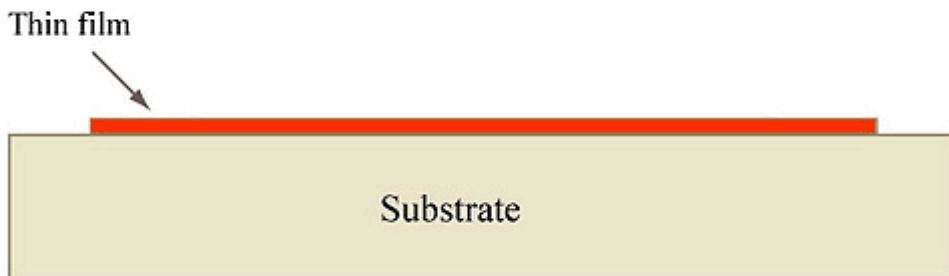
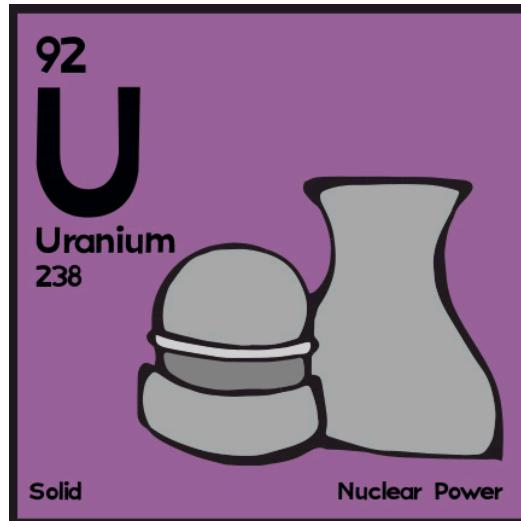
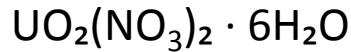


UO₂ Fabrication



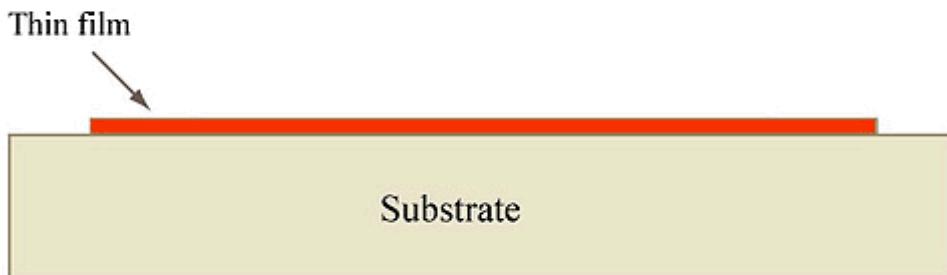
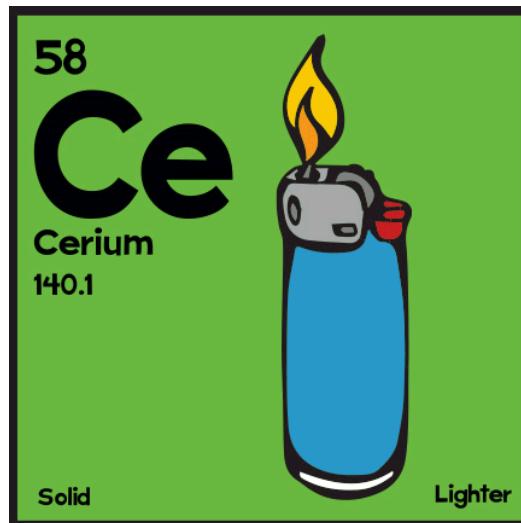
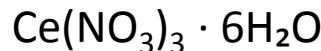
Epitaxial Thin Films

Strategy for studying the pellet cladding interface



Epitaxial Thin Films

Strategy for studying the pellet cladding interface



“Surrogate for a
Surrogate”

Differences in oxidation
states:

$\text{Ce(III)} \rightarrow \text{Ce(IV)}$ oxidation

$\text{U(IV)} \rightarrow \text{U(IV)}$ reduction

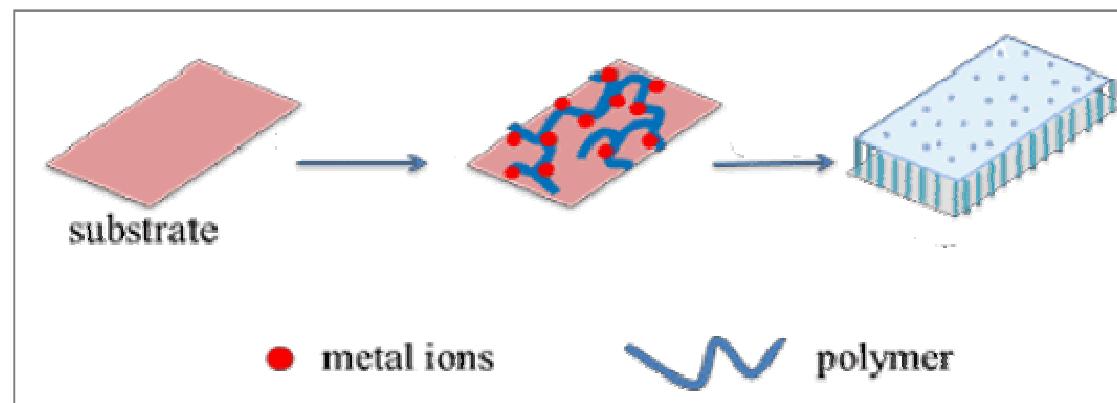
Polymer-Assisted Deposition

LETTERS

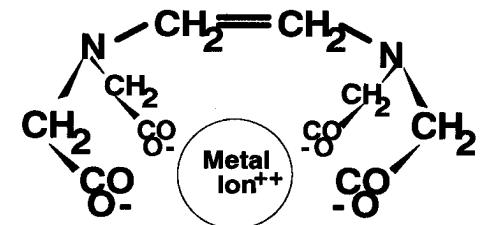
Polymer-assisted deposition of metal-oxide films

Q. X. JIA¹*, T. M. MCCLESKEY², A. K. BURRELL², Y. LIN¹, G. E. COLLIS², H. WANG¹, A. D. Q. LI³
AND S. R. FOLTYN¹

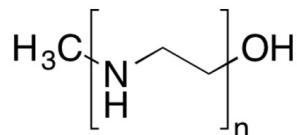
¹Superconductivity Technology Center, Division of Materials Science and Technology, and ²Structural and Inorganic Chemistry, Division of Chemistry, Los Alamos National Laboratory, Los Alamos, New Mexico 87545, USA
³Department of Chemistry, Washington State University, Pullman, Washington 99164, USA
 *e-mail: qjia@lanl.gov



- Viscosity
- Stabilization
- Homogeneity
- Uniform distribution



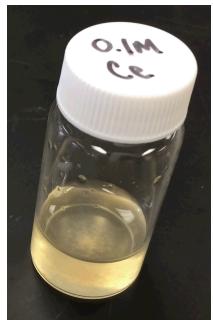
Ethylenediaminetetraacetic acid (EDTA) chelates a metal ion



Polyethylenimine (PEI)

A. Synthesis of bulk Ce(IV)O₂

0.1M Ce(NO₃)₃ · 6H₂O



0.125M EDTA

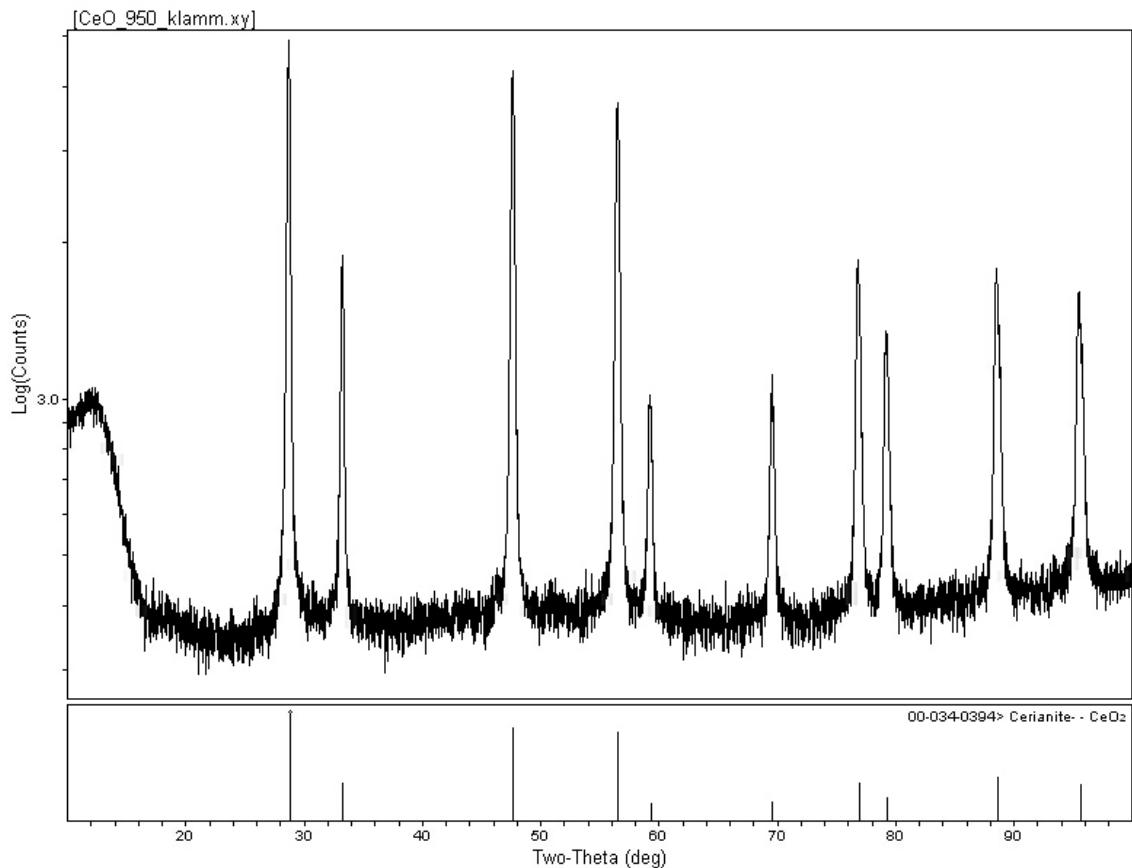
0.125M PEI



1hr
350C



1hr
1000C



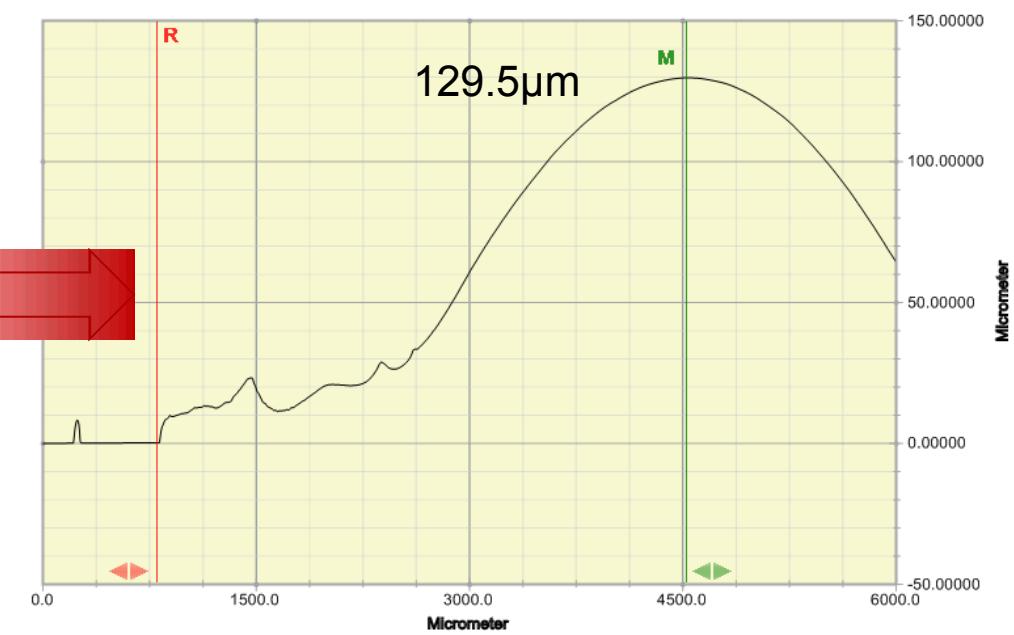
Powder X-ray Diffraction Pattern
Ce(IV)O₂

B. CeO₂ Thin Films: Drop Casting

Applying same cerium nitrate solution onto a substrate

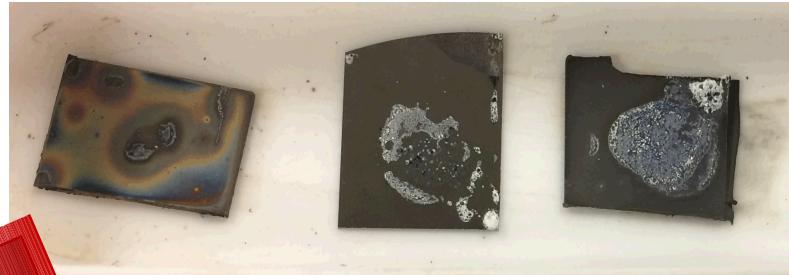
- Polymer-Metal ratio 1.25 : 1
- 5-10 layers (drops)
- 200C on hotplate (in air)

Profile of cerium nitrate on platinized Si

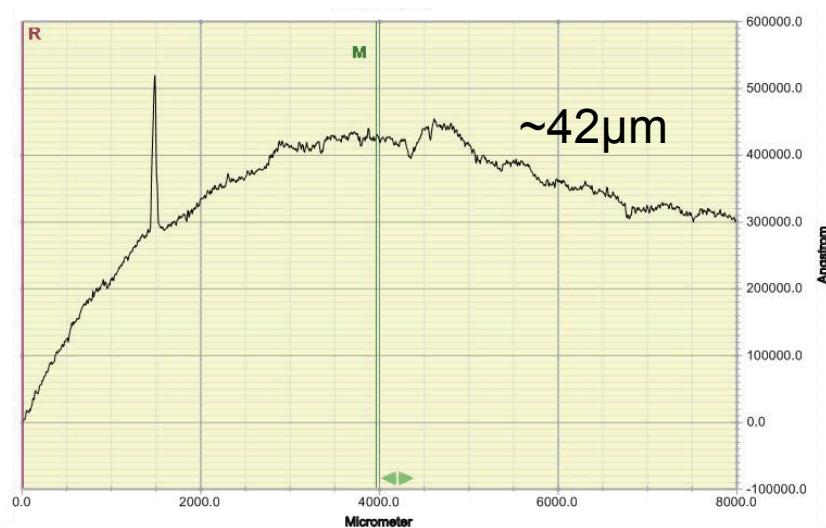


C. CeO₂ Thin Films: Spin Coating

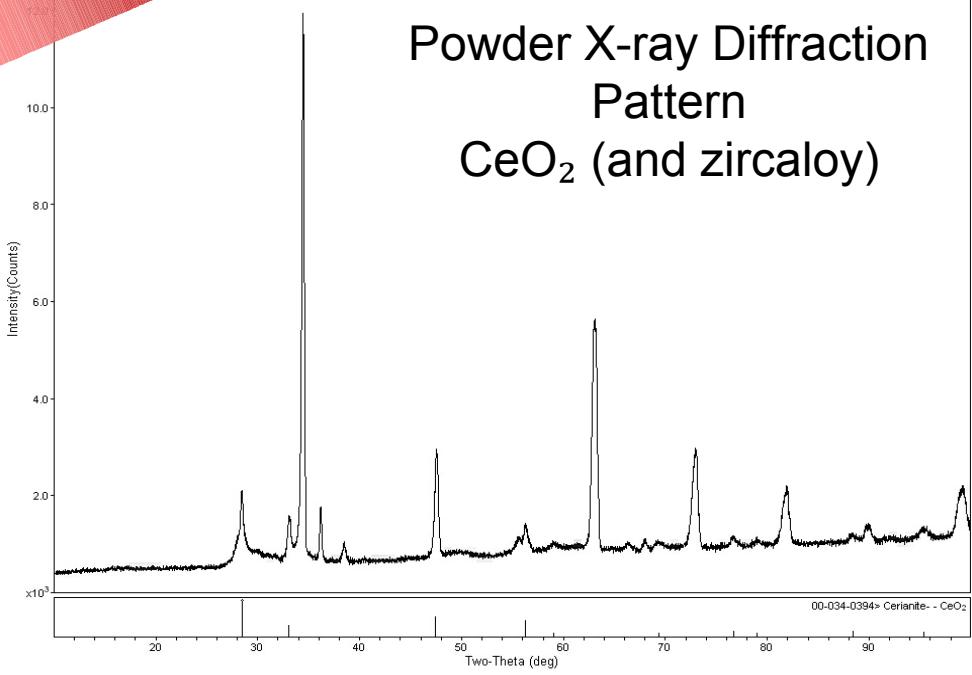
- Polymer-Metal ratio 0.625M:1M
- 4-6 layers (drops)
- Annealed 1000C in Ar



Profile of cerium nitrate on zircaloy



Powder X-ray Diffraction Pattern CeO₂ (and zircaloy)



D. Synthesis of U(IV)O₂



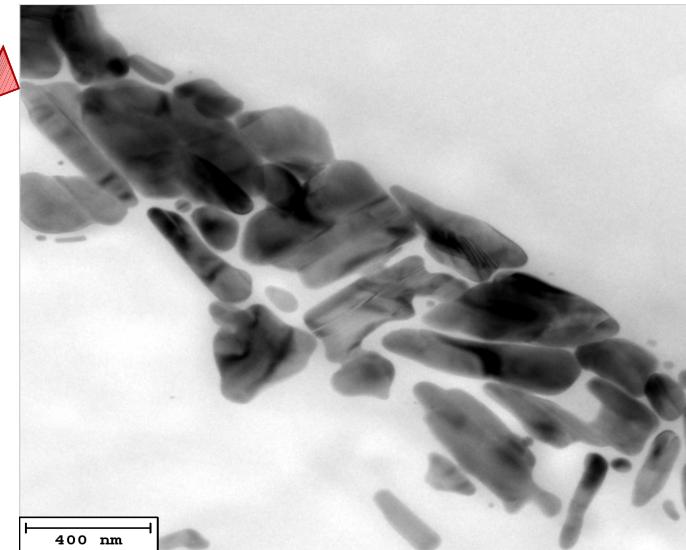
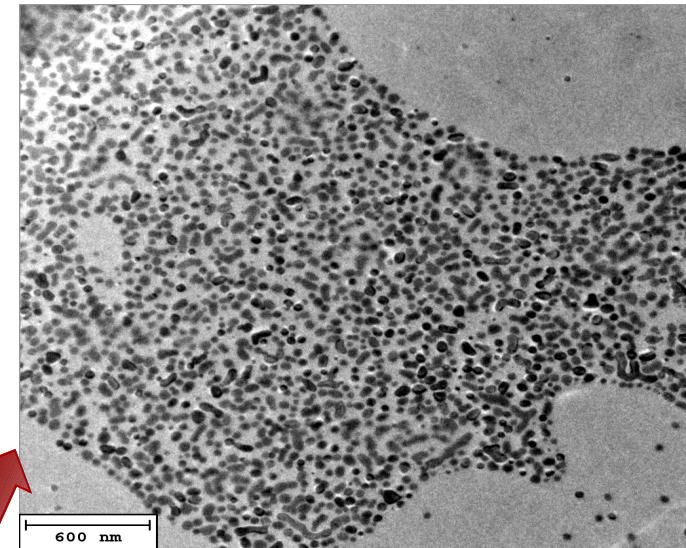
Uranyl nitrate

- Polymer-Metal ratio 0.625M : 1M
- 1000C
- Annealed in Ar and air



TEM on
Silicon
Nitride Grids

0.1M
uranium
nitrate
annealed in
Ar (top) and
in air
(bottom)



Summary

1. Recreate Surface Chemistry

- Ce(IV) oxidation state as a UO_2 surrogate
- Deposition & annealing process
- Characterize

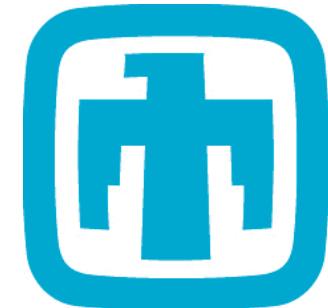


2. Future Steps

- Further characterization of films
- Mechanical testing
- In-situ TEM



Provide a surrogate thin film that will enable us to
assess the damage and integrity of nuclear systems
at the microscale



Acknowledgments

Samuel Briggs –TEM
Diana Perales and Tim Boyle –spin casting
films
Remi Dingreville –Project PI



Sandia Laboratory Directed Research and
Development program for funding



Backup Slides

Substrate

- Platinized Silica
- Zircaloy 4

	Element	Weight%
Zircaloy-2	Sn	1.2-1.7
	Fe	0.07-0.20
	Cr	0.05-0.15
	Ni	0.03-0.08
	Zr	balance + impurities
Zircaloy-4	Sn	1.2-1.7
	Fe	0.18-0.24
	Cr	0.07-0.13
	Zr	balance + impurities

Sources

- Steve Skutnik. A background on spent fuel. Neutron Economy. May 6.
https://neutroneconomy.blogspot.com/2012_05_01_archive.html
- Q. X. J, et al. *Polymer-assisted deposition of metal-oxide films*. Nature Materials. Vol 3. August 2004.
- Olander, Donald R. *Nuclear Fuels*. Engineering Journal. Vol 13. Jan. 2009