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# Direct Observations of Corrosion Cracking in a TEM

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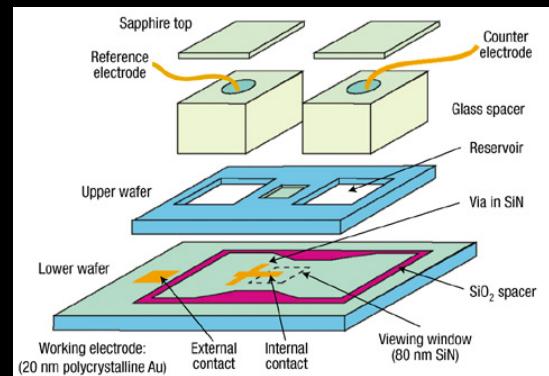
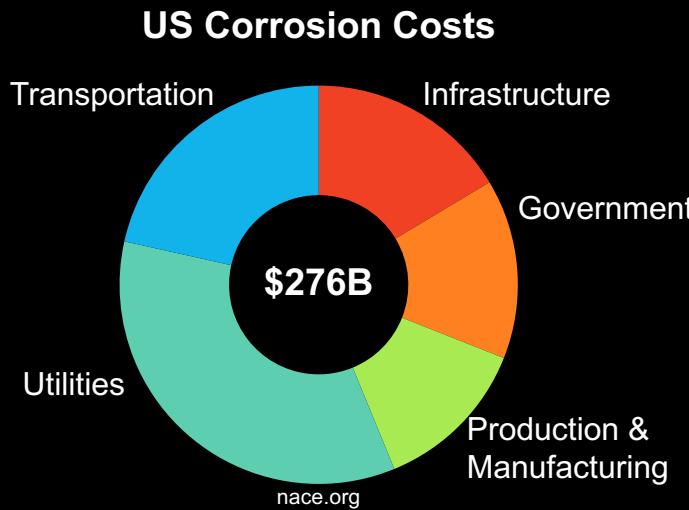
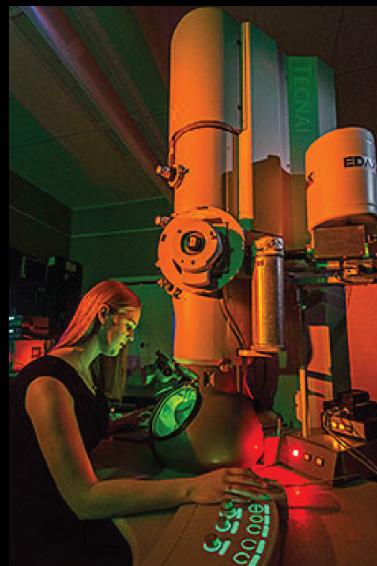


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# Motivation



- Costly
  - R&D
  - Managing failures
- Complex
  - Environmental: solution composition, pH, temperature, pressure, flow rate, microbes
  - Material: composition, microstructure, surface treatments
- Experimental limitations
  - Macro-scale
  - Ex-situ
- TEM
  - Sub- $\text{\AA}$  spatial resolution
  - Sub-ms temporal resolution
- In-Situ TEM Experiments
  - Chip-based systems
  - Commercial holders
  - Very few metals corrosion experiments

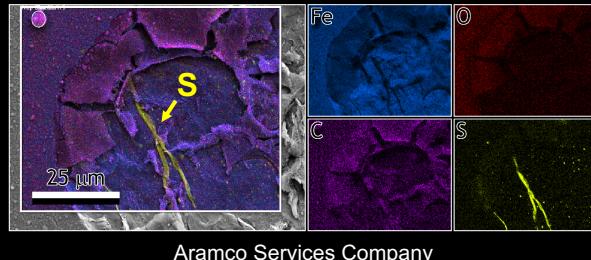


MJ Williamson, et. al., Nature Mat. 2, 2003

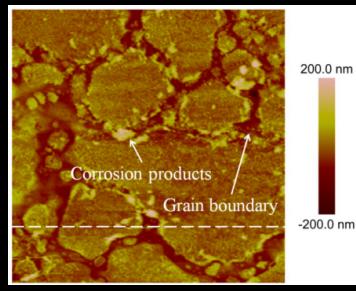
# Background

## ▪ Macro-Scale/Ex-Situ Corrosion

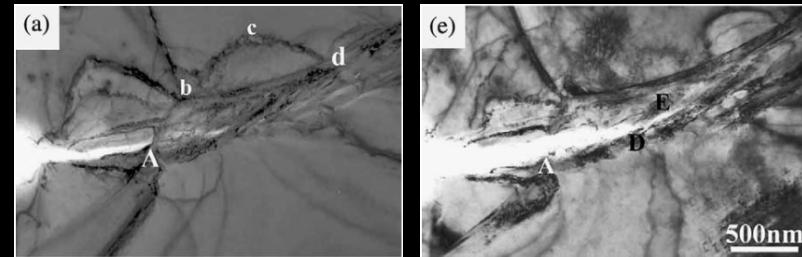
Pitting linked to impurities



GB preferential etching



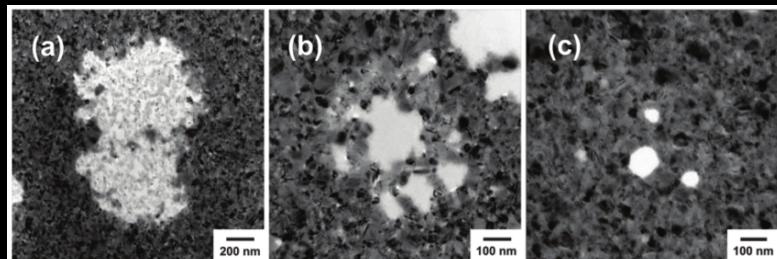
Corrosion-enhanced plasticity



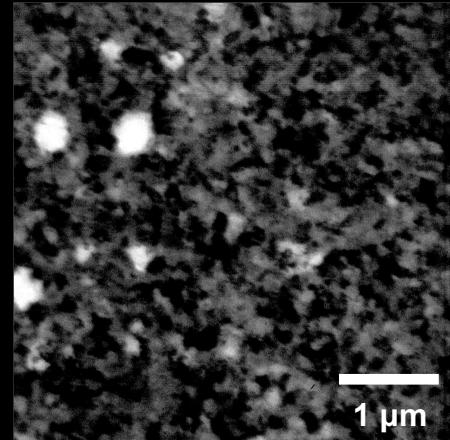
## ▪ In-Situ TEM Corrosion

- Highly localized events
- Model materials: Al, Cu, Fe

nc-Cu in 6 M NaCl soln.



nc-Fe in acetic acid



# In-Situ TEM Corrosion



- Here we present a methodology to directly investigate the onset of corrosion at the nanoscale using in-situ TEM corrosion techniques
- Results from used pipeline steel in a sweet, aqueous environment



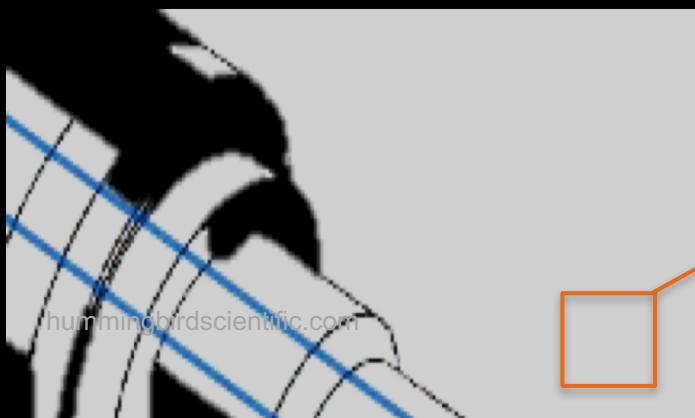
Chemical Composition of Pipe Steel (wt%, balance Fe)		
C	Ni	N
0.160	0.107	0.009
Mn	S	Al
0.710	0.021	0.003
Cu	Mo	Pb
0.345	0.017	0.002
Si	Sn	V
0.167	0.017	0.002
Cr	P	Ti
0.116	0.016	0.001

# Experimental Setup

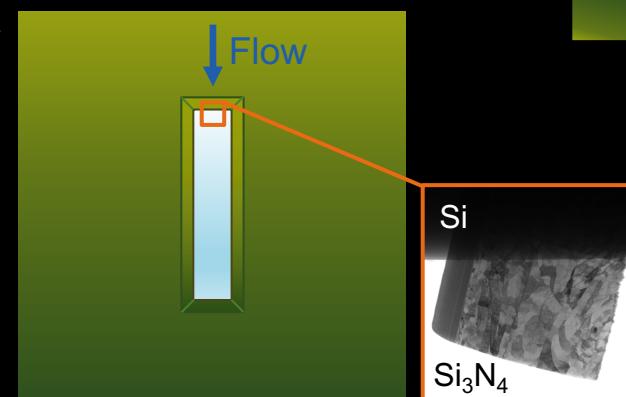
- 1018 steel from used Aramco pipeline
- Deionized H<sub>2</sub>O with 1390  $\mu\text{M}$  Na<sub>2</sub>SO<sub>4</sub> and 6  $\mu\text{M}$  CO<sub>2</sub>
- 2  $\mu\text{L}/\text{min}$  flowrate
- Low-dose STEM at 300 keV



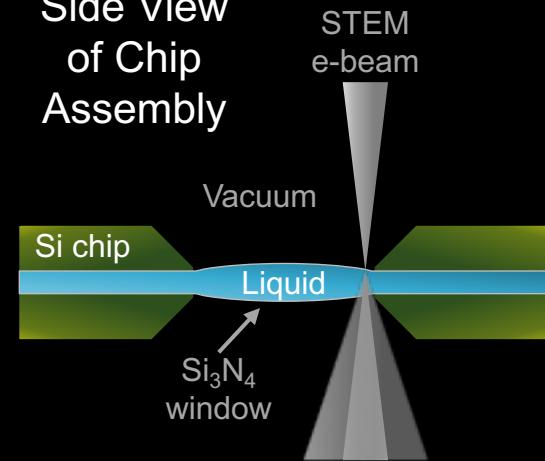
Liquid Flow TEM Holder



Top View of Chip Assembly



Side View  
of Chip  
Assembly



FIB lift-out sample  
affixed to Si chip at  
Si<sub>3</sub>N<sub>4</sub> window edge

# Electron Beam Effects

- Radiolysis of water<sup>[1]</sup>
  - Products include:  $e_{aq}^-$ ,  $H^\cdot$ ,  $HO^\cdot$ ,  $HO_2^\cdot$ ,  $OH^-$ ,  $H_3O^+$ ,  $H_2$ ,  $H_2O_2$
  - Dose-dependent yield
- Heating<sup>[2]</sup>
  - Only a few degrees can increase reaction kinetics
- Charge accumulation<sup>[3]</sup>

[1] S Le Caer, Water 3, 2011

[2] N de Jonge & FM Ross, Nature Nanotech 6, 2001

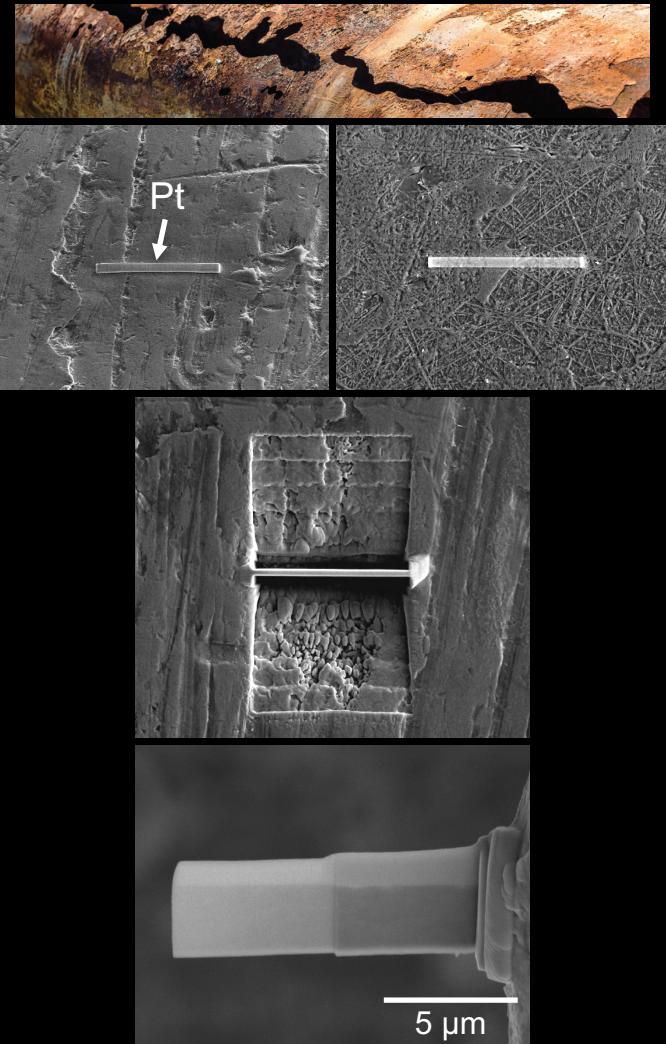
[3] TJ Woehl, et.al., Ultramicroscopy 127, 2013

# Sample Preparation

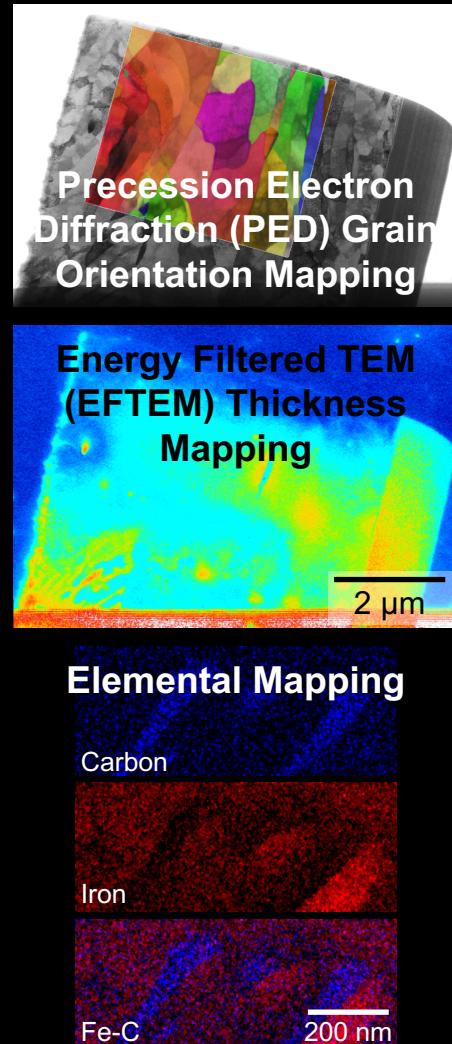


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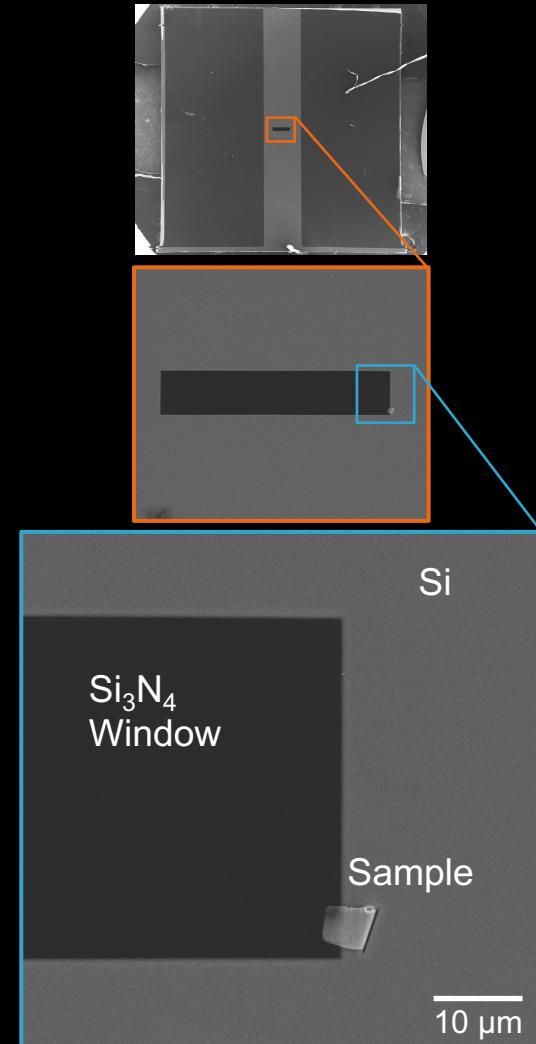
## FIB Lift-Out



## Characterization

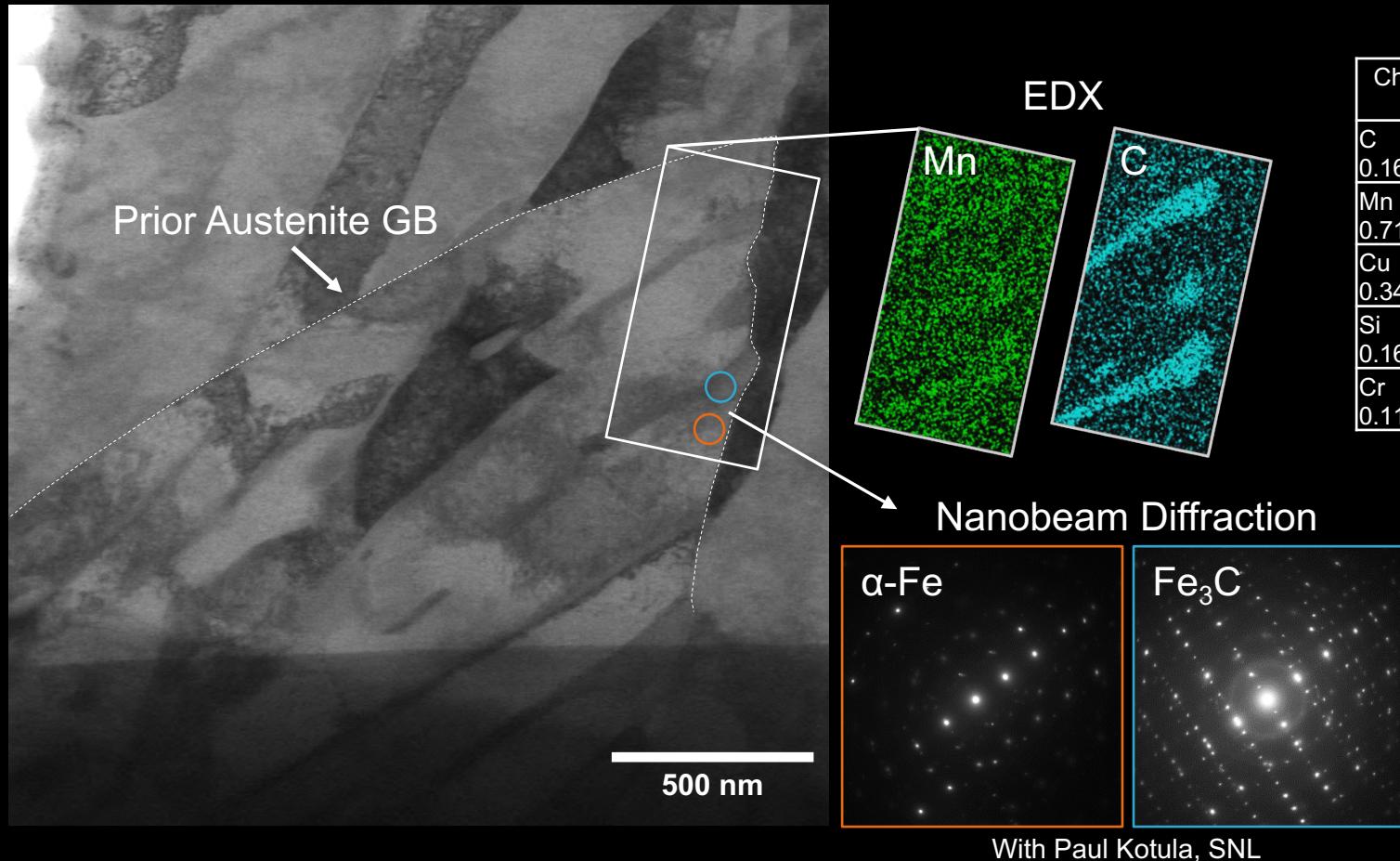


## Transfer to LC Chip

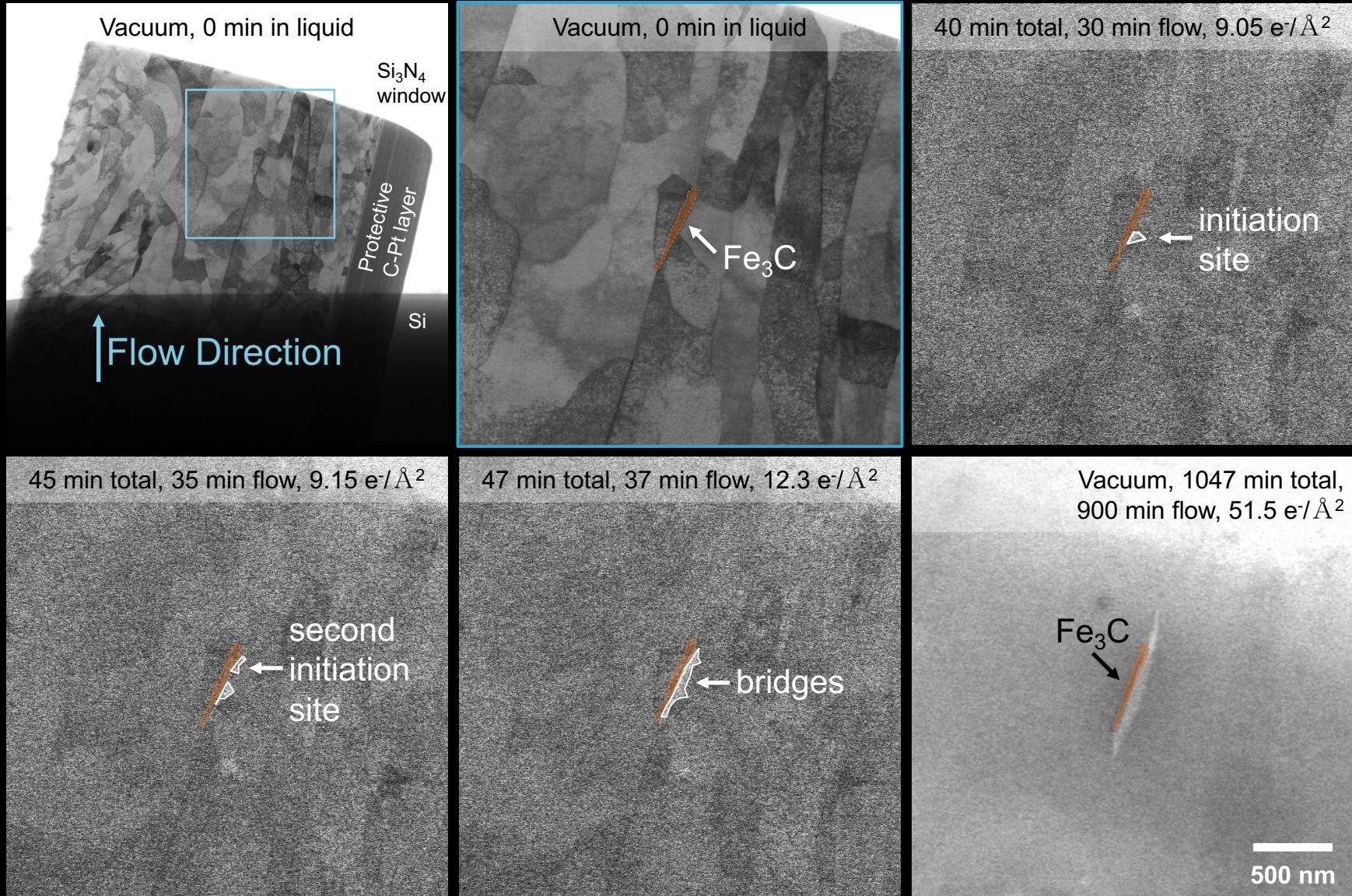


# Initial Microstructure

- $\alpha$ -Fe and Mn-rich  $\text{Fe}_3\text{C}$
- Pitsch-Petch OR

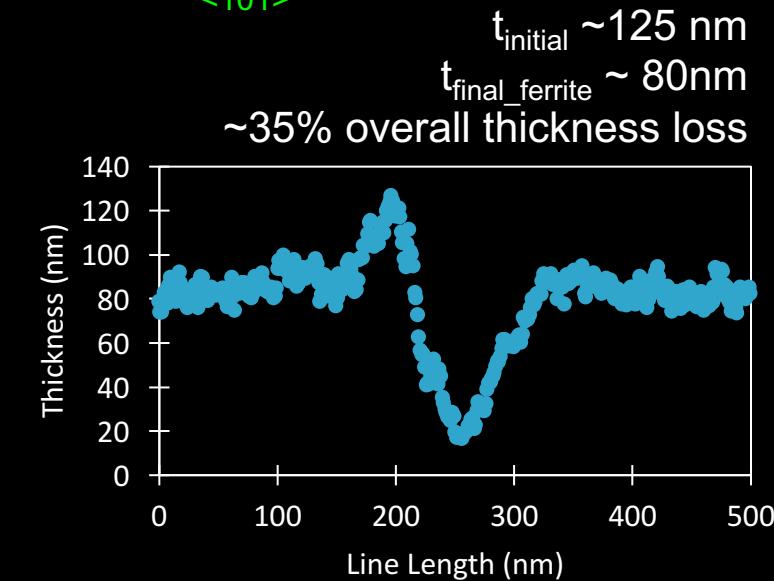
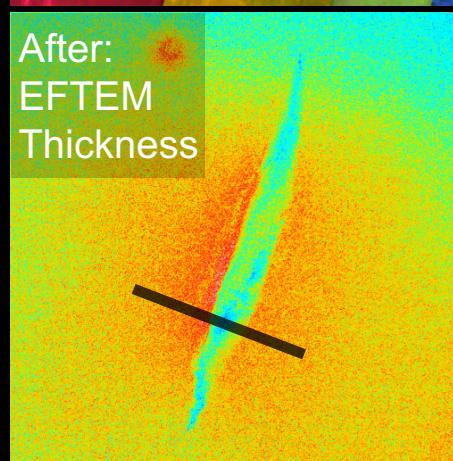
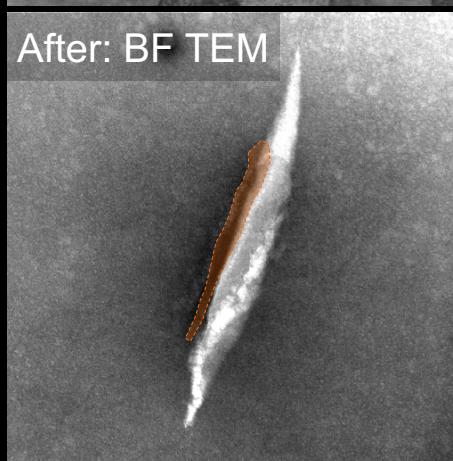
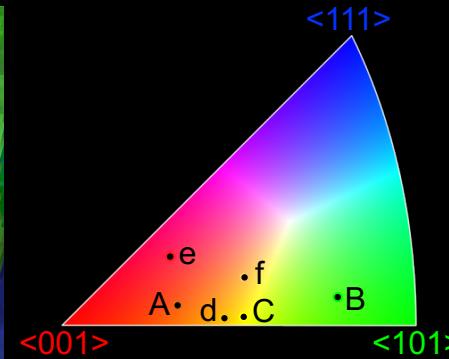
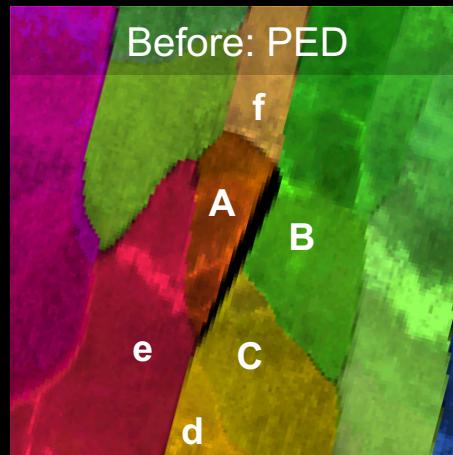
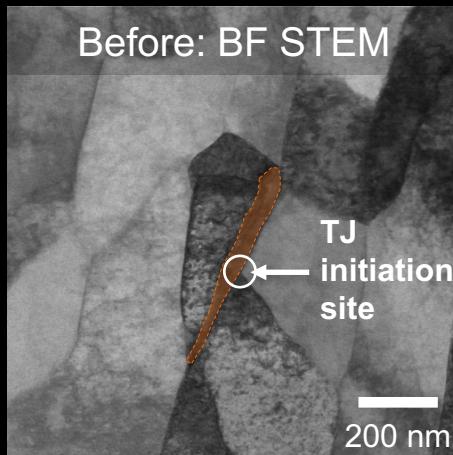


# Corrosion Crack Initiation



# Corrosion Crack Initiation

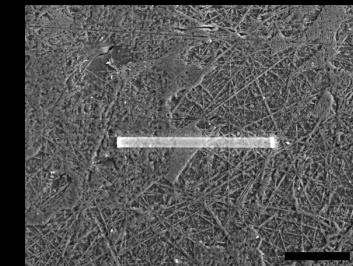
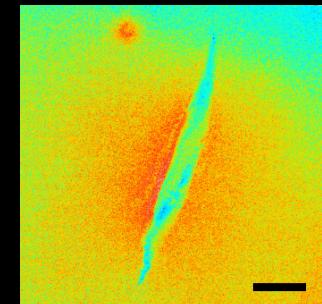
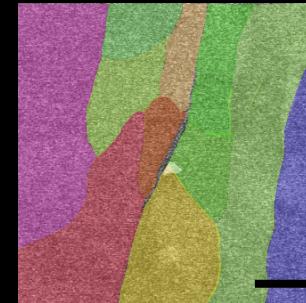
- Initiates at  $\text{Fe}_3\text{C}/\alpha\text{-Fe}/\alpha\text{-Fe}$  TJ
- Intergranular crack progression
- Results in very sharp intergranular crack,  $r \sim 5 \text{ nm}$
- Crack growth  $\sim 30 \text{ nm/h}$



# Conclusions



- Directly observed the nucleation of a corrosion crack at  $\text{Fe}_3\text{C}/\alpha\text{-Fe}/\alpha\text{-Fe}$  triple junction
- Intergranular crack grew at 30 nm/h
  - same order as field observations of near-neutral pH SCC crack growth rate,  $10^{-11} \text{ m} \cdot \text{s}^{-1}$
- Can begin to explore microstructural effects such as cementite size and spacing on corrosion rate
- Can begin to explore environmental effects, such as sweet vs. sour environments or temperature



# Acknowledgements



Steven Tanya Tim Rachael Michele



[cint.lanl.gov](http://cint.lanl.gov)

In-situ TEM techniques available at CINT include:

Heating

Cooling

Mechanical: compression, tension, bending, etc.

Electrical

Chemical: liquid, gas

Electrochemical

Laser

Ion irradiation/implantation

More to come!

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