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# Investigation of Tank 241-AW-104 Composite Floating Layer

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**Key Words**: AW-104, grab sample, composite floating layer, tributyl phosphate, tridecane, degradation products of hydrocarbons, carbonate, SEM, PLM, SVOA, TOC

Abstract: Seven grab samples and one field blank were taken from Tank 241-AW-104 (AW-104) on June 2, 2017, and received at 222-S Laboratory on June 5, 2017. A visible layer with brown solids was observed floating on the top of two surface tank waste samples (4AW-17-02 and 4AW 17 02DUP). The floating layer from both samples was collected, composited, and submitted for chemical analyses and solid phase characterization in order to understand the composition of the floating layer. Tributyl phosphate and tridecane were higher in the floating layer than in the aqueous phase. Density in the floating layer was slightly lower than the mean density of all grab samples. Sodium nitrate and sodium carbonate were major components with a trace of gibbsite and very small size agglomerates were present in the solids of the floating layer. The supernate consisted of organics, soluble salt, and particulates.

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Date

# **Investigation of Tank 241-AW-104 Composite Floating Layer**

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Date Published February 2018



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# **Executive Summary**

Tank 241-AW-104 was last sampled for solids-waste core samples in 2001 and had its last large-scale waste transfer in 2003. A visible layer with brown solids was observed floating on the top of the 4AW-17-02 and 4AW-17-02DUP samples (surface tank waste) taken in June 2017. The floating layers from both samples were collected, composited, and analyzed for semivolatile organic analysis and solid phase characterization in order to understand the composition of the floating layer. A liquid sample collected from the mid-section of the aqueous phase of 4AW-17-02 was also submitted for semivolatile organic analysis to serve as a comparison to the floating layer sample. This document reports the composition of the composite floating layer based on the semivolatile organic, density, and solid phase characterization results.

Tributyl phosphate and tridecane (a tentatively identified compound) were found in the composite floating layer sample. A trace of tributyl phosphate, tridecane, and a number of unknown hydrocarbons were present in the aqueous phase of sample 4AW-17-02. The concentration of tributyl phosphate and the estimated concentration of tridecane were much higher in the floating layer than in the aqueous phase. Density in the composite floating layer was slightly lower than the mean density of all eight grab samples. Both results complemented each other and showed high content of tributyl phosphate and tridecane in the floating layer.

Total organic carbons from the small organic acids (including acetate, formate, glycolate, and oxalate) were calculated based on the concentration of small organic acids and were compared against the total organic carbons determined either by the persulfate oxidation or the furnace oxidation. The total organic carbons from the small organic acids for samples 4AW-17-02 to 4AW-17-08, relative to the total organic carbons, ranged from 54 – 62% indicating the presence of other organic species. The ratios did not have a wide variation, indicating other organic species were present from the top of the tank supernate to the bottom of the supernate.

Sodium nitrate and sodium carbonate were major inorganic components with a trace of gibbsite, and very small size agglomerates (<10 µm) were present in the solids of the floating layer based on solid phase characterization. Agglomerates were possibly made of gibbsite, thermonatrite, nitratine, natrophosphate, hematite, cejkaite/clarkeite, and metal-containing materials. Sodium hydroxide and organic coating particulates (uranyl nitrate tributyl phosphate complex) were also found in the solids of floating layer. The yellow supernate (with the presence of soluble uranyl nitrate) of the floating layer consisted of organics, soluble salt, and fine-grained particulates. The presence of carbonate indicated the chemical reactions of dissolved carbon dioxide with hydroxide ion or ongoing oxidation of organics. The presumed formation of uranyl nitrate tributyl phosphate complex (uranyl nitrate is soluble in tributyl phosphate) resulted from a reaction similar to what was used to extract uranyl nitrate from the acidified spent fuel in the plutonium uranium extraction process, which used a mixture of tributyl phosphate and kerosene.

The re-formation of a floating layer over time was observed on 4AW-17-02 and 4AW-17-02DUP reflecting chemical reactions occurred. Subsampling of the re-formed floating layer for small organic acids and total organic carbons has been scheduled and will be included in a revision of this report when the results are available.

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#### LIST OF TERMS

#### **Abbreviations and Acronyms**

AW-104 Tank 241-AW-104

EDS energy dispersive spectroscopy (spectrometry)

GC gas chromatography IC ion chromatography

ID identification

MS mass spectrometry or matrix spike

Na not applicable ND not determined OH free hydroxide

PCB polychlorinated biphenyl
PLM polarized light microscope(py)
PUREX Plutonium and Uranium Extraction

Q qualifier
QC quality control
RI refractive index
RT retention time

SOA small organic acids SD standard deviation

SEM scanning electron microscopy SPC solid phase characterization SVOA semivolatile organic analysis

TBP tributyl phosphate TC total carbon

TIC total inorganic carbon or tentatively identified compounds

TOC total organic carbon

TOC<sub>SOA</sub> total organic carbon from small organic acids.

TOC<sub>p</sub> total organic carbon determined by the persulfate method. TOC<sub>f</sub> total organic carbon determined by the furnace method.

WAI Wastren Advantage, Inc.

WHL Wastren Advantage, Inc., Hanford Laboratory

XRD X-ray diffraction

#### Units

 $\begin{array}{ccc} g & & gram(s) \\ hr & & hour \\ L & & liter \end{array}$ 

μg microgram(s)
μm micrometer
M molarity
mg milligram
mL milliliter
% percent

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#### 1 INTRODUCTION

Seven grab samples and one field blank were taken from Tank 241-AW-104 (AW-104) on June 2, 2017 (RPP-PLAN-61252, *Tank 241-AW-104 Grab Sampling and Analysis Plan – Fiscal Year 2017*), and received at 222-S Laboratory on June 5, 2017. Sample 4AW-17-01 was not taken, and a visible layer with brown solids floating on the top of the 4AW-17-02 and 4AW-17-02DUP samples was observed (RPP-RPT-60261, *Final Report for Tank 241-AW-104 Grab Sampling 2017*). The floating layer from both samples was collected, composited, and submitted for chemical analyses and solid phase characterization (SPC) in order to understand the composition of the floating layer. Due to limited availability of the composite floating layer sample, only semivolatile organic analysis (SVOA) and SPC were performed.

Tank AW-104 entered service in 1980 when it first received dilute caustic solution (LA-UR-97-311, *Waste Status and Transaction Record Summary (WSTRS)*). Since then, AW-104 also received waste from various tanks including Tank 241-AY-101 (1981); decladding waste from the Plutonium Uranium Extraction (PUREX) facility (1982-1983); Tank 241-AZ-101 supernatant waste (1983); and transfers of PUREX miscellaneous waste (1982 and from 1986 to1991) (RPP-RPT-58671, *Derivation of Best-Basis Inventory for Tank 241-AW-104 as of May 1, 2015*). In addition, AW-104 received periodic additions of flush water through its history (LA-UR-97-311). Waste in AW-104 tank was also transferred to other tanks, such as dilute noncomplexed waste to Tank 241-AW-105 in 1982 and 1983, supernatant waste to Tank 241-AW-102 as 242-A Evaporator feed (1982 – 1989), and a transfer to 241-AW-102 in support of Evaporator Campaign 01-10 (2001) (RPP-RPT-58671). Tank AW-104 has not had any waste transaction since 2003 except some loss due to evaporation (RPP-13639, *Caustic Limits Report for Period Ending March 30<sup>th</sup>, 2017*). The tank was last sampled for solids/core samples in 2001 and for two grab samples in 2003 (RPP-58671).

Tank AW-104 contains supernatant liquid, saltcake, and sludge (RPP-RPT-58671) with approximately 289 inches of supernatant depth (RPP-PLAN-61252). A uniform liquid surface with no visually discernable crust or solids layer was observed from images taken during in-tank inspection in 2009 (RPP-13639). The supernatant waste phase is suspected to contain at least two separate layers of liquid based on samples taken in 2003, and a secondary liquid phase was suspected near the surface of the supernatant liquid (RPP-PLAN-61252). Twenty inches of supernatant liquid was left in the tank after a waste transfer in 2001. This supernatant waste had a lower density than the waste pumped into the tank in 2003 and is thought to have moved to the top layer in the tank (RPP-PLAN-61252).

In order to ensure the waste remains in compliance with the corrosion mitigation requirements, liquid grab samples were collected from AW-104 on June 2, 2017, in accordance with RPP-PLAN-61252. In addition, sample data may be needed for a compatibility assessment to support future waste transfer(s) from AW-104 and for modelling to support mission planning.

This document reports the investigation of the composite floating layer from the two surface tank samples (4AW-17-02 and 4AW-17-02 DUP) taken on June 2, 2017.

#### 2 SAMPLES

#### 2.1 THE FIRST FLOATING LAYER

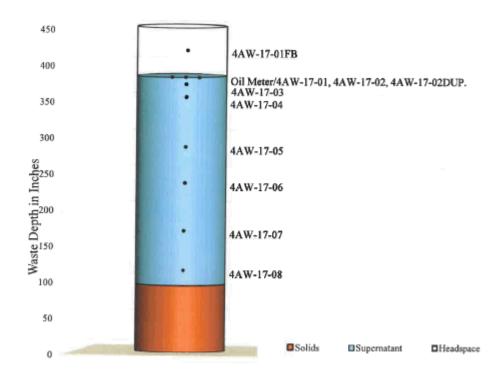
Seven grab samples and one field blank were taken from AW-104 on June 2, 2017, and received at 222-S Laboratory on June 5, 2017. Figure 1 taken from RPP-PLAN-61252 Figure 3-1 is provided to illustrate the location where the AW-104 grab samples were taken. The exact location of the samples taken is in the Field Work Supervisor data sheet per the relevant sampling work order. Although three grab samples (4AW-17-01, 4AW-17-02, and 4AW-17-02DUP) were originally scheduled to be collected at the surface of the supernatant liquid, the 4AW-17-01 grab sample was not successfully collected.

When they were received at the 222-S Laboratory, samples 4AW-17-02 and 4AW-17-02 DUP appeared as yellow liquid with light cloudy to brownish solids suspended in the liquid or floated on the top of the liquid (Figure 2, left). A thin dark layer remained on the surface of both 4AW-17-02 and 4AW-17-02 DUP samples 60 hr after receipt at the laboratory (Figure 2, right).

Subsampling of the dark floating layers of both 4AW-17-02 and 4AW-17-02 DUP was performed in the 222-S Laboratory 11A hot cell by Wastren Advantage, Inc., (WAI) Hanford Laboratory (WHL) personnel per the client's instruction (RPP-PLAN-61252). Approximately 1 g of each top layer was taken from 4AW-17-02 and 4AW-17-02 DUP to make samples S17T019207 and S17T019217, respectively. Aliquots of sample S17T019217 were transferred into the sample jar of S17T019207 on July 12, 2017, in 222-S Laboratory room 2B by the WHL personnel to make the composite sample (S17T023360, a composite from the floating layer, see Appendix A, page 34). Subsequently, S17T023360 was split into two subsamples, S17R000148 and S17R000149.

Only 0.5 mL of S17T023360 was used to determine the density by WHL (on July 12, 2017) due to inadequate sample amount in S17T023360. After the density measurement, this 0.5-mL aliquot of S17T023360 was assigned a new identification number (S17R000149) for SPC analysis. Only polarized light microscopy (PLM) and scanning electron microscopy - energy dispersive spectrometry analysis (SEM-EDS) were performed on S17R000149; no X-ray diffraction (XRD) was performed due to inadequate sample material. The remainder of S17T023360 was assigned a new identification number as S17R000148, and then submitted for SVOA, small organic acids (SOA) by ion chromatography (IC), and total inorganic carbon/total organic carbon (TIC/TOC) by persulfate oxidation method. Only SVOA analysis was performed on S17R000148; SOA and TIC/TOC were not performed due to inadequate sample amount of S17R000148.

A liquid sample collected from the mid-section of the aqueous phase of 4AW-17-02 (S17T023353) was also submitted for SVOA analysis in order to compare the SVOA results in the aqueous phase against the SVOA results of the floating layer.



Notes: Riser 14 is not in the center of the tank. The diagram above is not intended to represent horizontal position.

Figure 1. Approximate Sample Locations for AW-104 2017 Sampling Event (Taken from RPP-PLAN-61252, Figure 3-1).

# 2.2 SECOND FLOATING SAMPLE AFTER COLLECTING THE FIRST FLOATING LAYER

Samples 4AW-17-02 and 4AW-17-02 DUP have been largely undisturbed since the subsampling of the top layer taken in late June of 2017. A second subsampling of both 4AW-17-02 and 4AW-17-02 DUP was attempted on October 2, 2017, for a viscosity study; it was discovered that a visible floating layer had reformed on both samples (Figure 3). Per client instruction, the second floating layer will be collected from both samples (4AW-17-02 and 4AW-17-02 DUP) and will be composited to make sample S17R000246. Aliquots of sample S17R000246 with a new identification (ID) number, S17R000247, will be submitted to WHL for anions and SOAs by IC, TIC/TOC, and total carbon (TC)/TOC by furnace (see Appendix A, page 42). Aliquots of sample S17R000246 will also be submitted for PLM, SEM, and/or XRD analyses pending the availability of sample S17R000246. This report will be revised to include the results from the second floating sample and further observations of the floating layer.

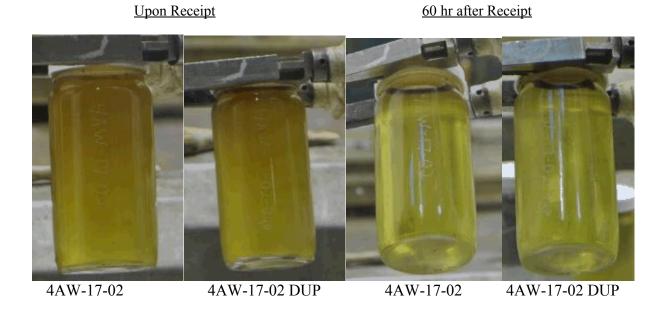


Figure 2. 4AW-17-02 and 4AW-17-02 DUP upon Receipt and 60 hr after Receipt.

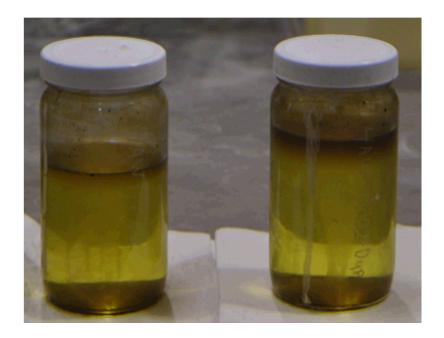


Figure 3. Photos of 4AW-17-02 (on the left) and 4AW-17-02 DUP (on the right) on October 2, 2017.

Table 1. Laboratory Sample Identification Numbers and Analyses for the First Floating Layer.

Sample Identification / Actions	Grab	Samples	
Sumple ruentmention / rections	4AW-17-02	4AW-17-02 DUP	
Sample ID of Organic Layer	S17T019207	S17T019217	
Sample ID for Composite of Organic Layer	S17T023360		
Split the Composite of Organic Layer and New Laboratory ID	S17R000148	S17R000149	
Requested Analyses	SVOA, anions and SOA, and TIC/TOC	Density, PLM, and SEM	
Liquid Sample from Mid-section of Aqueous Phase for SVOA	S17T023353 for SVOA analysis	na	

Table 2. Laboratory Sample Identification Numbers and Analyses for the Second Floating Layer (results not available in revision 0 of this report).

Sample Identification / Actions	Grab Samples				
Sample recitification / /rectons	4AW-17-02	4AW-17-02 DUP			
Sample ID of Organic Layer	S17T015732	S17T015773			
Sample ID for Composite of Organic Layer	S17R000246				
Composite of Organic Layer and New Laboratory ID	S17R000247				
Requested Analyses	Anions and SOA, and TC/TOC	Density, PLM, and SEM			

#### 2.3 ANALYSES

Tables 1 and 2 summarize the requested analyses for the composite floating layer sample (S17R000148 and S17R000149) and for the future composited sample (S17R000247) collected from the second formation of the floating layer. The composite floating layer sample (S17R000149) was analyzed for SPC including PLM and SEM/EDS. Chemical analyses were requested for the composite floating layer sample (S17R000148), including anions and SOA by IC, density, SVOA by gas chromatography/mass spectrometry (GC/MS), TIC/TOC by the persulfate method, and/or TC/TOC by the furnace method. However, IC, TIC/TOC, and TC/TOC were not performed on the composite floating sample (S17R000148) due to inadequate sample material.

One aqueous sample (S17T023353) was extracted by the continuous liquid-liquid methylene chloride extraction method under both basic and acidic conditions. The composite floating layer sample (S17R000148) was diluted with an applicable amount of methylene chloride for a direct injection. Aliquots of extracts obtained from the continuous liquid-liquid extraction or aliquots of the methylene chloride diluted floating layer sample were injected into a GC/MS system for SVOA analysis.

Analyses were performed by WHL per their analytical procedures (see Table 3) for anions and SOAs by IC, density (in g/mL), SVOA, TIC/TOC by persulfate oxidation and coulometric detection, and TC/TOC by the furnace method. Process Chemistry of the 222-S Laboratory performed SPC including PLM and SEM-EDS per the applicable procedures listed in Table 3.

Table 3. Analytical and Solid Phase Characterization Procedures.

Analyte(s) or Method	Procedure Number	Procedure Title
Anions and Small Organic Acids	LA-533-166	Ion Chromatographic Analysis of Anions and Small Organic Acids on DIONEX Model ICS 3000
Density	LA-510-112	Determination of Density for Free Liquid Samples
Free Hydroxide (OH)	LA-211-102	Determination of Free OH <sup>-</sup> /OH <sup>+</sup> Using Metrohm Titrando
SVOA, Liquid-Liquid Extraction	LA-523-115	Semivolatiles and PCBs from Aqueous Samples Using Semimicro Continuous Liquid-Liquid Extraction
SVOA	LA-523-135	Semivolatile Organics by Gas Chromatography/Mass Spectrometry, Based on SW-846, Method 8270D
TIC/TOC	LA-342-100	Determination of Carbon by Hot Persulfate Oxidation and Coulometric Detection
TC/TOC by Furnace	LA-344-105	Determination of Carbon in Solutions by Combustion and Coulometry
Solid Phase Characterization	Procedure Number	Procedure Title
PLM	ATS-LT-519-107	222-S Laboratory Polarized Light Microscopy
SEM, Preparation	ATS-LT-161-104	SEM Sample Preparation Procedure
SEM	ATS-LT-161-103	222-S Laboratory Technology Procedure for the ASPEX Explorer Scanning Electron Microscope

PCB = polychlorinated biphenyl

#### 3 RESULTS AND DISCUSSION

S17R000148 contained a darker brown solid residue with a brownish yellow supernate. S17R000149 was a clear, brownish yellow supernate and did not appear to contain residues of

solids (after density study and solids were excluded). Residues of solids (trace amount) were visible in the original container of S17R019217.

#### 3.1 SEMIVOLATILE ORGANIC RESULTS

Semivolatile organic results and the associated quality control (QC) results are provided in Appendix B. The liquid sample (S17T023353, aqueous phase of 4AW-17-02) had no particulate formation during the base-neutral step of the extraction, but a fine, white precipitate formed during the acidic step, and then settled in the methylene chloride layer. The final extract appeared completely clear following filtration through sodium sulfate.

The majority of the routine QC samples associated with the liquid sample extraction met their method acceptable criteria with some exceptions. Low responses of internal standards were found for perlyene-D12 in the liquid sample and for acenaphthalene-D10 in the matrix spike (MS) sample due to matrix effects. Acid surrogates (2-fluorophenol, phenol-D6, and 2,4,6-tribromophenol) were eliminated due to nitration by the matrix. Benzylbutyl phthalate and bis (2-ethylhexyl) phthalate were detected in the blank, but were below the reporting limits. The liquid sample was extracted outside the 7-day holding time requirement as the request for this analysis was added after sample receipt (RPP-PLAN-61252, 17-CCN-10, Characterization Change Notice).

Note that recoveries of acid surrogates in sample S17R000148, which was not extracted by methylene chloride extraction, failed low; however, there was no evidence of nitration of acid surrogates.

Table 4 summarizes SVOA results including the detected target compounds and tentatively identified compounds (TICs) for the composite floating layer sample (S17R000148) and the aqueous phase of 4AW-17-02 (S17T023353) (RPP-RPT-60261 and Appendix 10). Only one target analyte, tributyl phosphate (TBP) and one TIC, tridecane, were found in the floating layering. A trace amount of TBP and small amounts of phenols, tridecane, and tetradecane were present in the aqueous phase of 4AW-17-02. Two phthalates found in the aqueous phase could be contaminants acquired during the analytical process because they were also present in the blank. Another 18 unknown TICs were also found in the aqueous phase, but were not detected in the floating layer, possibly due to a much higher detection limit caused by the much smaller sample size and dilute/direct injection to GC/MS without the concentration step in sample preparation. A further evaluation of the TIC's spectra was performed (Appendix C). The majority of these TICs were unknown hydrocarbons which may be the degradation products of hydrocarbons.

The hydrocarbon-based compound (butanoic acid) was most likely the degradation product of the hydrocarbons that converted to the acid form due to methylene chloride extraction at low pH. Note that the qualitative acetic results by GC/MS method (0.01 mg/mL) was much lower than the acetate result determined by the IC (1.45 mg/mL, see RPP-RPT-60261, page 72), possibly due to either low acetate solubility in methylene chloride or loss of acetate during the concentration step after methylene chloride extraction.

The estimated tridecane concentration was approximately 20 times higher in the floating layer than in the aqueous phase, and the TBP concentration was also much higher in the floating layer than in the aqueous phase, indicating the floating layer of 4AW-17-02 consisted of mainly hydrocarbons and TBP, while the aqueous phase consisted of mainly degradation products of hydrocarbons.

Table 4. Semivolatile Results on Composite Floating Layer and Aqueous Phase. (2 pages)

S17R000148, Composite Floating Layer				4AW-17-02 (S17T02335	53), Aqueo	ous Phas	e
Compounds	μg/L	Q	RT min	Compounds μg/L Q		Q	RT min
Tri-n-butyl phosphate	123000	$J^1$	13.21	Tri-n-butyl phosphate	815	$\mathbf{J}^1$	13.17
na	na	na	na	Pentachlorophenol	550	$J^{1,b}$	14.29
na	na	na	na	2,4-Dinitrophenol <sup>b</sup>	1110	$J^{1,b}$	12.22
na	na	na	na	4,6 Dinitro-2-methylphenol <sup>b</sup>	552	$J^{1,b}$	13.11
na	na	na	na	4-Nitrophenol <sup>b</sup>	709	$J^{1,b}$	12.31
na	na	na	na	Butylbenzylphthalate	564	$\mathbf{J}^1$	18.03
na	na	na	na	N-Nitrosodimethylamine	989	$\mathbf{J}^1$	4.00
na	na	na	na	bis (2-Ethylhexyl) phthalate	603	$\mathbf{J}^1$	19.03
Tentatively Identifie	d Compound	s, μg/L	RT min	Tentatively Identified Compounds, μg/L			RT min
na	na	na	na	Acetic acid	11710	JNT <sup>2</sup>	2.83
na	na	na	na	Butanoic acid	8985	$JNT^2$	4.65
na	na	na	na	Unknown 1, unknown hydrocarbon	25762	5.34	5.34
na	na	na	na	Unknown 2, unknown hydrocarbon	34590	JNT <sup>2</sup>	5.5
na	na	na	na	Phosgene oxime	4862	$JNT^2$	5.57
na	na	na	na	Unknown 3, unknown hydrocarbon	26106	JNT <sup>2</sup>	5.89
na	na	na	na	Unknown 4, unknown hydrocarbon	4352	$JNT^2$	6.07
na	na	na	na	Unknown 5, unknown hydrocarbon	10281	JNT <sup>2</sup>	6.12
na	na	na	na	Unknown 6, possible hydrocarbon	14076	JNT <sup>2</sup>	6.33
na	na	na	na	Unknown 7, unknown hydrocarbon	27705	$JNT^2$	6.38
na	na	na	na	Unknown 8, unknown hydrocarbon	14461	$JNT^2$	6.54
na	na	na	na	Unknown 9, unknown hydrocarbon	13137	JNT <sup>2</sup>	6.79

Table 4.	Semivolatile	Results on	Composite	Floating	Laver and A	queous Phase.	(2 page	(2
I WOIC II	Comment of the comment	I LUDWING OIL	Composite		na, ci ama ii	queous i iiusei	(- pas-	~,

na	na	na	na	Unknown 10, 4-Penten-2-ol	17072	JNT <sup>2</sup>	7.26
na	na	na	na	Unknown 11, 2-Hydroxy-3-	5710	$JNT^2$	7.40
				pentanone			
na	na	na	na	Unknown 12, Heptane, 2-Bromo-	8400	$JNT^2$	7.98
na	na	na	na	Unknown 13, 1-Propene, 3-	13708	$JNT^2$	8.41
				propoxy-			
na	na	na	na	Unknown 14, Heptane, 4-methyl-	9276	$JNT^2$	8.52
na	na	na	na	Unknown 15, 1-Propene, 3-	5564	$JNT^2$	8.63
				propoxy-			
na	na	na	na	Unknown 16, similar	14816	$JNT^2$	8.98
				hydrocarbon to compound at RT			
				8.41			
na	na	na	na	Unknown 17,	5754	$JNT^2$	9.1
				Hydrazinecarboximidamine,			
				nitrate			
Tridecane	220000	$JNT^2$	10.22	Tridecane	10076	$JNT^2$	10.23
na	na	na	na	Tetradecane	8463	$JNT^2$	11.16
Sum	343000			Sum	300758		

Q = Qualifier

#### 3.2 SMALL ORGANIC ACIDS RELATIVE TO TOTAL ORGANIC CARBONS

The SOA includes acetate, formate, glycolate, and oxalate. Results of SOAs reported in RPP-RPT-60261 for AW-104 grab samples were converted to organic carbon and then summed together as TOC for SOA ( $TOC_{SOA}$ ) (shown in Table 5). The TOC results reported in RPP-RPT-60261, determined either by the persulfate oxidation method ( $TOC_p$ ) or by the furnace oxidation method ( $TOC_f$ ), were evaluated against the  $TOC_{SOA}$ . The ratio of  $TOC_{SOA}$  to  $TOC_p$  or to  $TOC_f$  provides an indication of the presence of other organic species. Table 5 summarizes  $TOC_{SOA}$ ,  $TOC_p$ ,  $TOC_f$ , and the ratio of  $TOC_{SOA}$  to  $TOC_p$  or to  $TOC_f$ .

Note that the concentrations of  $TOC_{SOA}$ ,  $TOC_p$ , and  $TOC_f$  were relatively constant throughout the AW-104 waste. The mean  $TOC_p$  and  $TOC_f$  from the grab samples was 2748 and 2906  $\mu g/mL$ , respectively. The mean  $TOC_p$  was approximately 95% of the  $TOC_f$ . The discrepancy between  $TOC_p$  and  $TOC_f$  may be due to incomplete persulfate oxidation for some organic species.

RT = Retention time

<sup>&</sup>lt;sup>1</sup> J: Estimated results. Concentration >the method detection limit, but <the estimated quantitation limit.

<sup>&</sup>lt;sup>2</sup> JNT: J is an "unknown" constituent reported for an organic analysis. N and T: Applied to compounds identified based on mass spectrometry library search. Tentatively identified compounds are not target compounds, and the results are only an estimate and not quantitative.

b: A qualifier. Matrix spike recovery outside the acceptable range.

The ratios of  $TOC_{SOA}$  to  $TOC_p$  and  $TOC_{SOA}$  to  $TOC_f$  were relatively constant throughout AW-104 tank waste supernate and typically ranged from 58 to 62% for the former and 54 – 58% for the latter. The less than 100% ratio indicated there are other organic species present, evidenced by the presence of hydrocarbons and/or degradation products of hydrocarbons identified by the SVOA results. Unfortunately, the ratio of  $TOC_{SOA}$  to  $TOC_p$  or  $TOC_f$  for the floating layer was not available due to inadequate sample material for SOA, TC/TOC, and TIC/TOC analyses. However, the floating layer, having been formed three months later following the first collection of the floating layer, will be collected and analyzed for SOA, TC, and TOC and will provide additional comparison to the aqueous phase.

Table 5. Contribution of Small Organic Acids to Total Organic Carbons on AW-104 Grab Samples.

Samples	Sample Identification	TOCsoa	TOCp	TOCf	TOC <sub>SOA</sub> / TOC <sub>p</sub>	TOC <sub>SOA</sub> / TOC <sub>f</sub>
		μg/mL	μg/mL	μg/mL	%	%
4AW-17-01	S17T015731	na	na	na	na	na
4AW-17-02	S17T015748	1664	2690	2930	62	57
4AW-17-02 DUP	S17T015774	1641	2760	2870	59	57
4AW-17-03	S17T015781	1607	2770	2860	58	56
4AW-17-04	S17T15788	1646	2750	2850	60	58
4AW-17-05	S17T015795	1623	2790	2980	58	54
4AW-17-06	S17T015804	1666	2730	2920	61	57
4AW-17-07	S17T015811	1657	2700	2910	61	57
4AW-17-08	S17T015818	1652	2790	2930	59	56
Mean	na	1644	2748	2906	60	57
SD	na	19	36	41	1.3	0.9

SD = standard deviation.

TOC<sub>SOA</sub> = total organic carbon from small organic acids.

TOC<sub>p</sub> = total organic carbon determined by the persulfate method.

 $TOC_{f:}$  = total organic carbon determined by the furnace method.

Note: na: failed to collect this sample.

### 3.3 DENSITY AND FREE HYDROXIDE CONCENTRATION OF GRAB SAMPLES

Table 6 summarizes the density and free hydroxide concentration of AW-104 grab samples and the density of 4AW-17-02 composite floating layer sample. The mean and SD density for the eight grab samples was  $1.350 \pm 0.005$  g/mL and the density of the 4AW-17-02 composite floating layer with one observation was 1.338 g/mL. The aqueous phase (the brown liquid excluded the solid residue) in the composite floating layer was used for the density measurement. The density for the grab samples was measured in the 11A hotcell using a 5-mL sample size.

The density measurement in the hotcell was not observed, but it was speculated that only the liquid phase was used. The density in the floating layer appeared slightly lower than the density in the aqueous phase, indicating higher concentrations of organics in the floating layer. Density of tridecane and tributyl phosphate is 0.756 g/mL and 0.973 g/mL, respectively, indicating that the organics were not the sole components in the floating layer.

The lower density found in the floating layer agreed with an earlier observation of a layer of lower density supernate which is thought to have moved to the top layer in the tank (RPP-PLAN-61252). Both SVOA and density results complemented each other and showed high content of tributyl phosphate and tridecane in the floating layer.

The concentrations of [OH] of eight grab samples (supernate samples throughout the tank) did not have a wide variation with a higher trend noted near the top (29,400 to 30,000  $\mu$ g/mL, 1.73M to 1.76M). The concentration of [OH] in the floating layer was not available, but was suggested by SEM-EDS along with nitrate and carbonate by PLM.

Table 6. Density and Free Hydroxide Concentration of AW-104 Grab Samples.

Samples	Sample Identification	Density	[OH]			
Samples	Density; [OH]	g/mL	μg/mL			
4AW-17-02	S17T015732; S17T015748	1.348	30000			
4AW-17-02 DUP	S17T015773; S17T015774	1.346	29700			
4AW-17-03	S17T015780; S17T015781	1.348	29500			
4AW-17-04	S17T015787; S17T015788	1.342	29700			
4AW-17-05	S17T015794; S17T015795	1.356	29400			
4AW-17-06	S17T015803; S17T015804	1.351	29400			
4AW-17-07	S17T015810; S17T015811	1.357	29400			
4AW-17-08	S17T015817; S17T015818	1.349	29500			
	Mean	1.350	29575			
	SD	0.005	198			
4AW-17-02 Top Layer	S17R000149	1.338	ND			
SD: standard deviation						
ND: not determined due to inadequate sample available for analysis						

## 3.4 POLARIZED LIGHT MICROSCOPY

The organic layer of 4AW-17-02 and 4AW-17-02 DUP was composited to make up sample S17R000149, which is part of sample delivery group 20172604. After the density and pH tests had been completed, S17R000149 was analyzed by PLM.

The first PLM analysis was performed on July 12, 2017, by S. Bolling and H. Meznarich on sample S17R000149, which had been prepared by L. Dinh. After reviewing the SEM data, sample S17R000149 was re-prepared by J. Eitelman and analyzed by PLM on October 31, 2017, by S. Bolling and M. LaMothe. Both analyses were performed at the 222-S Laboratory using the procedure listed in Table 3.

The primary goal of analyzing this sample was to look for solids forming in the floating layer sample. The PLM is a qualitative technique. Estimates of relative abundance of phases can be made, but these are very rough estimates.

# **Description:**

Sample S17R000149 was described as having dual phases: yellow supernatant liquid ( $\sim$ 3.5 to 16.8 µg/mL Uranium-238 in the grab samples reported in RPP-RPT-60261, possibly causing the yellow color by the presence of soluble uranyl nitrate) as the bottom layer with a brownish-yellow band on top. On July 12, 2017, the dark band made up approximately 20% of the sample. This dark band contained the majority of the solid material found in the sample. On October 31, 2017, only a very thin layer of the dark yellow/brown band was left (partially due to subsample for SEM analysis). The remainder of the sample was the yellow supernate. Table 7 summarizes the abundance of crystals observed on S17R000149 from both dates using PLM.

The majority of the material found in the first analysis (mounted in mother liquor) consisted of clusters of isotropic spheres, some of which were coated with particulate (Figures 4 and 5). These spheres were found in both layers although there were fewer present in the supernate than were found in the dark band. This spherical material was originally identified as cancrinite. However, the SEM data did not support this conclusion. Upon further evaluation, it was decided that these isotropic spheres are, in all probability, organic in nature and most likely tiny droplets of water immiscible organics, e.g., uranyl nitrate TBP complex.

Table 7. Abundance of Crystals Observed using PLM

	Chemical Formula	S17R000149 (Dark Band)	S17R000149 (Yellow Supernate)	S17R000149 (Dark Band)	S17R000149 (Yellow Supernate)
Analysis Date		July 12, 2017	July 12, 2017	October 31, 2017	October 31, 2017
Sodium Nitrate	NaNO <sub>3</sub>	N/A	N/A	Major	N/A
Sodium Carbonate	Na <sub>2</sub> CO <sub>3</sub> •H2O	N/A	N/A	Major	N/A
Gibbsite	Al(OH) <sub>3</sub>	N/A	N/A	Trace	Trace
Fine-grained Particulate	N/A	Major	Major	Minor	Major
Organic Material	N/A	Major	Major	Trace	Major

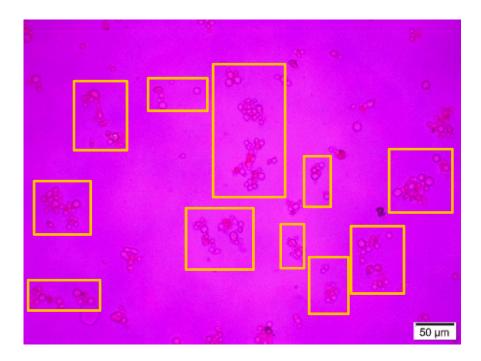


Figure 4. (Photo S17R000149a, Upper Layer, Mounted in Mother Liquor), Crossed Polars with the Red I Compensator Plate in Place, Showing Isotropic Spheres (Peach Rectangles), Some of which are Coated with Particulate.



Figure 5. (Photo S17R000149f, Upper Layer, Mounted in Mother Liquor), Uncrossed Polars, Showing Isotropic Spheres (Peach Rectangles), Some of which are Coated with Particulate.

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The second analysis of the upper layer was very different than the original analysis. In the first analysis, very few solids were found in either layer. In the second analysis, the dark band contained a considerable amount of solid material. Sodium nitrate (light green circles shown in Figures 6 and 7) and sodium carbonate (yellow rectangles shown in Figure 6) were major phases while gibbsite (blue circles in Figure 7) was present in trace amounts. Black shapes found in Figure 6 were micelles which formed when the mother liquor was mounted in refractive index (RI) 1.55 oil. Also a few isotropic spheres, most likely uranyl nitrate TBP complex [UO<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub>·2TBP] (peach rectangle shown in Figure 7), were found.

The results from the second analysis of the supernatant layer more closely matched those from the first analysis. Figures 8a and 8b show the isotropic spherical clusters, uranyl nitrate TBP complex (peach rectangles), found in the yellow supernate portion of the sample S17R000149 and a small gibbsite particle (blue circle). The other circles were micelles that formed when mother liquor was mounted in RI oil due to the amphiphilic surfactant property of TBP. The yellow supernate contained isotropic spherical clusters as the major phase and a trace amount of gibbsite.

Samples were mounted in RI oils, water, and/or mother liquor in the second analysis. When mounted in water, all of the solids in both layers dispersed. A mount of the thin dark band in mother liquor was attempted, but the mother liquor quickly dried. It was necessary to add 1.550 RI oil. After adding the RI oil, it was noted that very small 4-quadrant, lightly birefringent, spherical shapes formed. These particles, suspected to be uranyl nitrate TBP complex, were prevalent in addition to sodium nitrate (Figure 9). Based on this observation, it appears that over time, sodium nitrate, sodium carbonate, and uranyl nitrate TBP complex precipitated or formed in sample S17R000149. The presence of carbonate could be the result of chemical reactions between the dissolved carbon dioxide and hydroxide ion or the ongoing oxidation of organics. The formation of uranyl nitrate TBP complex was produced by a chemical reaction similar to that used to extract uranyl nitrate from the acidified spent fuel in the plutonium extraction process, which used a mixture 30% TBP and a hydrocarbon solvent (kerosene). Note that uranyl nitrate is soluble in TBP/hydrocarbon diluent.

As mentioned previously, the yellow supernate contained primarily isotropic spherical clusters, which dispersed when mounted in water. There were fewer clusters and these clusters were smaller (<5 µm) when mounted in RI oils than when mounted in mother liquor, which indicated there was also some dissolution or dispersion of the spheres in the RI oils. Using the Becké line test, it was determined that the RI of these spheres was greater than that of the mother liquor, but less than 1.45. This material is most likely organic in nature (i.e., RI of tridecane is 1.425).

The second PLM results indicated that the solids darker band of the floating layer consisted of major nitrate and sodium carbonate and traces of small size gibbsite. The yellow supernate in the floating layer consisted of organics, fine-grained particulates, and soluble salts. Precipitation occurred in the subsample over time, with inorganic crystals concentrating in the floating layer, e.g., uranyl nitrate TBP complex in the organic-rich floating layer.

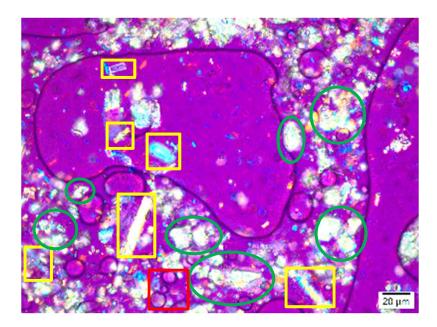


Figure 6. (Photo S17R000149aa, Upper Layer, Mounted in 1.55 RI Oil), Crossed Polars with the Red I Compensator Plate in Place, Showing Sodium Nitrate (Light Green Circles), Sodium Carbonate (Yellow Rectangles), and Micelles (Red Square).

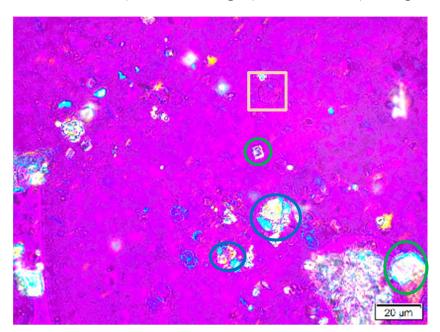


Figure 7. (Photo S17R000149ee, Upper Layer, Mounted in 1.55 RI Oil), Crossed Polars with the Red I Compensator Plate in Place, Showing Sodium Nitrate (Light Green Circles), Gibbsite (Blue Circles), and Isotropic Sphere (Peach Rectangle).

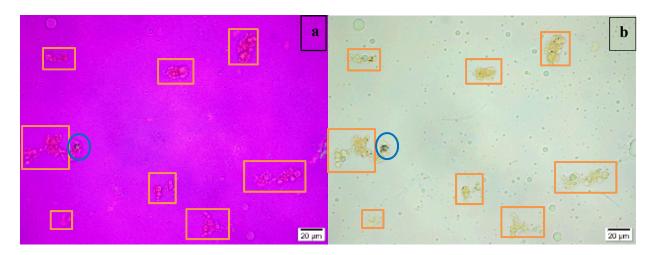


Figure 8. 8a and 8b. Photo S17R000149pp, Bottom Layer, on the Left, Mounted in 1.45 RI Oil with Crossed Polars and the Red I Compensator Plate in Place. Photo S17R000149qq, (on the Right), the Same View but with Uncrossed Polars. Isotropic Spheres (Peach Rectangles) and Gibbsite (Blue Circle). Micelles can be seen in the Background.

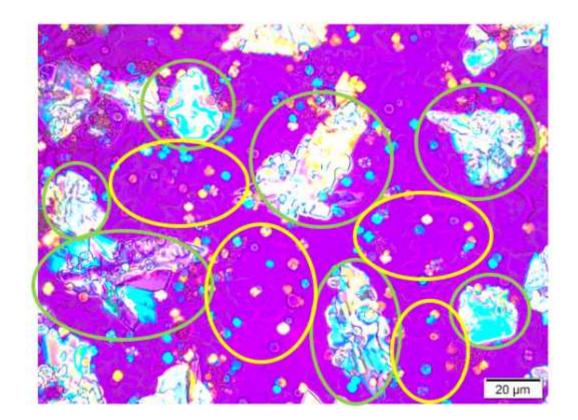


Figure 9. (Photo S17R000149ee, Upper Layer, Mounted in 1.55 RI Oil after Mother Liquor Evaporated), Crossed Polars with the Red I Compensator Plate in Place, Showing Sodium Nitrate as it Dissolves (Light Green Circles), and Very Small Lightly Birefringent Particles, Suspected to be Uranyl Nitrate TBP Complex (Yellow Circles).

# 3.5 SCANNING ELECTRON MICROSCOPY – DISPERSIVE ENERGY SPECTROMETRY

The top layer of sample S17R000149 was pipetted off and filtered prior to SEM-EDS analysis (Figures 10 to 16) or filtered without applying vacuum (Figures 17 and 18). Numerous elements were found in this sample. It seems to follow a pattern of most tanks in that the particle size is very small (5-10 μm) and agglomerate. The compositions of the majority of these particulates/agglomerates possibly included gibbsite (Al(OH)<sub>3</sub>), thermonatrite [Na<sub>2</sub>CO<sub>3</sub>·H<sub>2</sub>O], hematite (Fe<sub>2</sub>O<sub>3</sub>), potassium or sodium chloride (KCl or NaCl), cejkaite [Na<sub>4</sub>(UO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>], clarkeite [Na(UO<sub>2</sub>)O(OH)·H<sub>2</sub>O], uranyl nitrate TBP complex [UO<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub>·2TBP], nitratine (NaNO<sub>3</sub>), natrophosphate [Na<sub>7</sub>F(PO<sub>4</sub>)<sub>2</sub>·19H<sub>2</sub>O], and sidorenkite [Na<sub>3</sub>Mn(PO<sub>4</sub>)(CO<sub>3</sub>)] (Figures 10 – 17). Soluble uranyl nitrate diffused into the TBP-containing floating layer and formed uranyl nitrate TBP complex (Figures 10, 12, 14, and 16). Higher carbon in the area-scan and the diffuse nature of the particulate indicated particulates coated with organics (Figure 16). In addition, sodium hydroxide (NaOH) was identified when the sample was filtered without applying the vacuum (Figure 18).

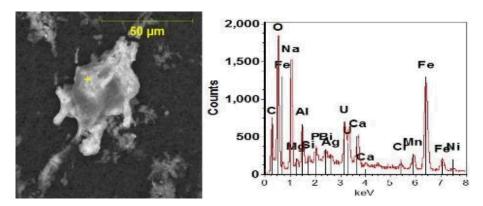


Figure 10. S17R000149 (Top Layer); Spot-Scan, EDS on the Marked "+" with a Chemistry Consistent with Hematite (Fe<sub>2</sub>O<sub>3</sub>), Cejkaite/Clarkeite [Na<sub>4</sub>(UO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>] /[Na(UO<sub>2</sub>)O(OH)·H<sub>2</sub>O], Uranyl Nitrate TBP [UO<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub>·2TBP], and Other Sodium-rich Compounds. Minor Amounts of Mixed Metal Oxides are Present.

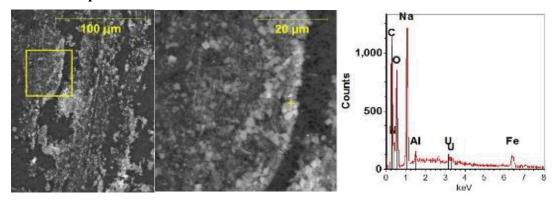


Figure 11. S17R000149 (Top Layer), Spot-Scan of Particulate, EDS on the Marked "+" with a Chemistry Consistent with Nitratine (NaNO<sub>3</sub>) and/or Thermonatrite [Na<sub>2</sub>CO<sub>3</sub>·H<sub>2</sub>O]. Trace Amounts of Hematite and Clarkeite/Cejkaite are Present.

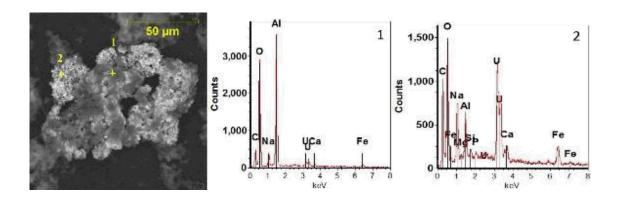


Figure 12. S17R000149 (Top Layer), Spot-Scan of Two Particulates (1) and (2), EDS on the Marked "+" with a Chemistry Consistent with Gibbsite (Al(OH)3) and Trace Amounts of Cejkaite/Clarkeite for Particulate (1), with Cejkaite/Clarkeite, Uranyl Nitrate TBP Complex, Aluminum-rich Components, Smaller Amounts of Hematite, and Metal Oxides for Particulate (2).

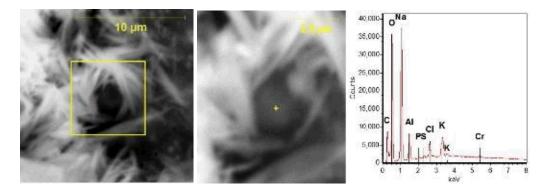


Figure 13. S17R000149 (Top Layer), Spot-Scan of an Interstitial Space, EDS on the Marked "+" with a Chemistry Consistent with Sodium-rich Salts. Smaller Amounts of Aluminum-containing Components, Potassium Salts, and Possible TBP are Present.

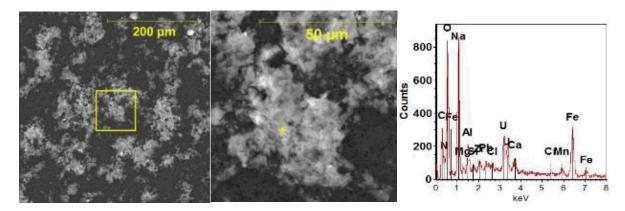


Figure 14. S17R000149 (Top Layer), Spot-Scan, EDS on the Marked "+" with a Chemistry Consistent with Hematite, Clarkeite/Cejkaite, Uranyl Nitrate TBP, Nitratine, and Sidorenkite [Na<sub>3</sub>Mn(PO<sub>4</sub>)(CO<sub>3</sub>)]. Small Amounts of Metal Oxides are Present.

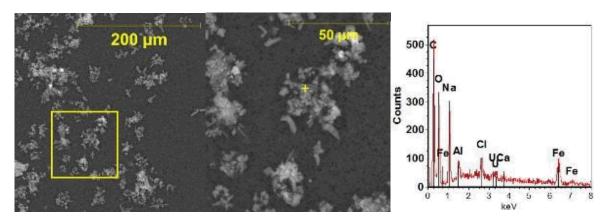


Figure 15. S17R000149 (Top Layer), Spot-Scan, EDS on the Marked "+" with a Chemistry Consistent with Thermonatrite or Sodium-organic Salts. Small Amounts of Hematite, Clarkeite/Cejkaite, and a Chloride-containing Compound are Present.

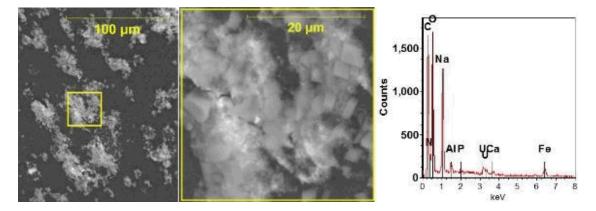


Figure 16. S17R000149 (Top Layer), EDS of an Area-Scan with a Chemistry Consistent with Thermonatrite, a Small Amount of Hematite, Clarkeite/Cejkaite, and Uranyl Nitrate TBP Complex. The Higher Carbon Content and Diffuse Nature of the Particulate Indicate the Presence of Organic Coating the Particulates.

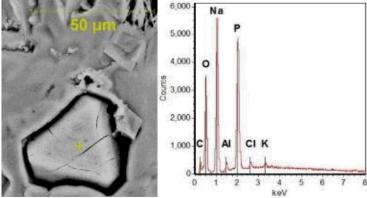


Figure 17. S17R000149 (Top Layer), Spot-Scan, EDS on the Marked "+" with a Chemistry of Natrophosphate [Na<sub>7</sub>F(PO<sub>4</sub>)<sub>2</sub>·19H<sub>2</sub>O]. Trace Amount of Potassium Chloride is Present.

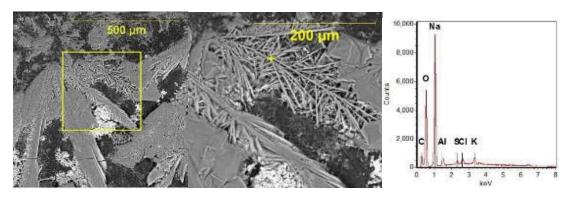


Figure 18. S17R000149 (Top Layer). Spot-Scan, EDS on the Marked "+" with a Chemistry Consistent with Sodium Hydroxide (NaOH). Small Amounts of Potassium Chloride and Aluminum-containing Components are Present.

#### 4 SUMMARY

One composited sample taken from two surface AW-104 grab samples (4AW-17-02 and 4AW-17-02DUP) was analyzed for SVOA, PLM, and SEM-EDS. The composite sample had dual phases: yellow supernatant liquid (consistent with the presence of uranyl nitrate which is soluble in TBP) as the bottom layer with a brownish-yellow band on top, reflecting an appearance similar to the floating layer in the parent samples (4AW-17-02 and 4AW-17-02DUP). This dark band contained the majority of the solid material found in the sample. Tributyl phosphate and tridecane (a TIC) were found in the composite floating layer sample. A trace of TBP, tridecane, and a number of unknown hydrocarbons were present in the aqueous phase of sample 4AW-17-02, indicating a different composition in higher amounts of organics in the floating layer than in the aqueous phase. The TOC<sub>SOA</sub> from the small organic acids for samples 4AW-17-02 to 4AW-17-08, relative to the TOC<sub>p</sub> or TOC<sub>f</sub>, ranged from 54 – 62%, indicating the presence of other organic species. However, the ratios did not have a wide variation, indicating other organic species were present from the top of the tank to the bottom of the supernate. The lower density found in the floating layer agreed with an earlier observation of a layer of lower density supernate which is thought to have moved to the top layer in the tank.

Sodium nitrate and sodium carbonate were major inorganic components; a trace of gibbsite and very small size agglomerates (<10 µm) were also present in the solids of the floating layer. The yellow supernate of the floating layer consisted of organics (presence of micelles formed by amphiphilic surfactant of TBP), soluble salt, and fine-grained particulates (uranyl nitrate TBP complex). The presence of carbonate via the dissolved carbon dioxide reacting with hydroxide ions or the ongoing oxidation of organics appears to indicate that chemical reactions occurred. The presumed formation of uranyl nitrate TBP complex pointed to a chemical reaction similar to that used to extract uranyl nitrate from the acidified spent fuel in the plutonium uranium extraction process, which used a mixture of TBP and hydrocarbon solvent. The re-formation of a floating layer over time was observed on 4AW-17-02 and 4AW-17-02DUP reflecting chemical reactions occurred. Further evaluation of the re-formation of a floating layer over time is needed and will be reported in the future.

# 5 QUALITY CONTROL

All testing carried out under this test plan followed ATS-MP-1032, 222-S Laboratory Quality Assurance Project Plan. All data entries specified in the test procedure were retained in controlled laboratory notebooks which were tracked in the project file: HNF-N-832-1, "PSEM Instrument Notebook," HNF-N-395-2, "PLM Sample Log," and HNF-N-841-1, "Studies of Tank Samples e.g., C-105 Dissolution, AN-101 Floating Solids." The Solid Phase Characterization Analyses Tracking form documenting the logbooks used for SPC is filed in the project file.

#### 6 ACKNOWLEDGEMENTS

The authors gratefully acknowledge the following contributors to this project. Their support is much appreciated.

- Liem Dinh and Jonathan Eitelman, for sample preparation of polarized light microscopy.
- Margaret LaMothe, for her support on polarized light microscopy.
- Dan Hansen, for his evaluation of spectra for tentatively identified compounds.
- Doug Kraft, for verification of spreadsheet and preparation of documentation for the SmartPlant Foundation.
- Peggy Puryear, for her editorial support.

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# APPENDIX A

# SAMPLE BREAKDOWN DIAGRAM

(15 pages, including cover page)

241-AW 104 Grab 2017-06 Group No.: 20171600 Charge Code: 002.1020

**TSCA Regulated PCB** 

#### **LOGIN NOTES:**

1. Customer: TCD\Corrosion Chemistry Control\Compatability Assessment

Sampling Mode: Liquid Grab

Riser: 17

**Customer POC: Sam Diedesch** 

PC: S. Snyder

Analytical Due Date: For (OH, pH, IC- Nitrate, Nitrite) - within 10 working days of receipt of last sample.

(TIC/TOC, TC/TOC, and Density)- within 15 working days.

Analytical Due Date for Format VI data (remaining analyses): 60 calendar days from receipt

2. QC login comments at Group level:

NO DUP or SPK on the FB

DUP, SPK/Batch on OH, IC-Anions/Org Acids, ICP- Metals, TIC/TOC, TC, TOC, and ICP/MS

DUP/Batch on Density, TGA, pH, GEA, Sr-90, Pu-241 and Pu/Am.

MS/MSD per batch for PCB, and Hg

- 3. Holding times are applicable for this project.
- 4. The TSCA PCB status: TSCA Regulated for PCB (Concentration: Liquids= 2.04ug/mL; Solids= 2.50E+01ug/g)

241-AW 104 Grab 2017-06 Group No.: 20171600 Charge Code: 002.1020

**TSCA Regulated PCB** 

### **REQUIRED DETECTION LIMITS:**

OH: 17 μg/mL NO2: 50.6 μg/mL NO3: 310 μg/mL PO4: 950 μg/ml PCB: 23 μg/L

#### **HOLDING TIMES:**

IC-NO3,NO2 and PO4: 48 hours. IC-Cl, F, Br, SO4: 28 days.

pH: Analyze immediately.

TOC: 28 days

Hg by CVAA: 28 days

Metals by ICP/AES or ICP/MS: 6 months

PCB Water (7days), Wastewater (14 days) Solids (14 Days)

241-AW 104 Grab 2017 -06 Grab Samples Riser: 17 Group 20171600

**TSCA Regulated PCB** 

# **DO NOT REFRIGERATE**

4AW-17-01FB Field Blank



S17T015725 Breakdown Pig HC Archive-HC



S17T015726 DOSE HC ICP/AES- Total metals Samp Amt HC pH Ni-63 Ni-Tracer



S17T015727
DOSE HC
IC-Anions/org
Samp Amt HC
TIC/TOC
TC by furnace
TOC by furnace



S17T015728
Americium
Americium Separation
DOSE HC
GEA
GEA Seperation
Plutonium
Plutonium-241
Samp Amt HC
Se-79

241-AW 104 Grab 2017 -06 Grab Samples Riser: 17 Group 20171600

**TSCA Regulated PCB** 

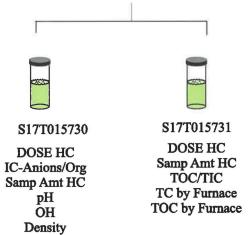
# **DO NOT REFRIGERATE**

4AW-17-01 Grab Sample (Total)



S17T015729

Breakdown Pig HC Archive-HC Density-HC



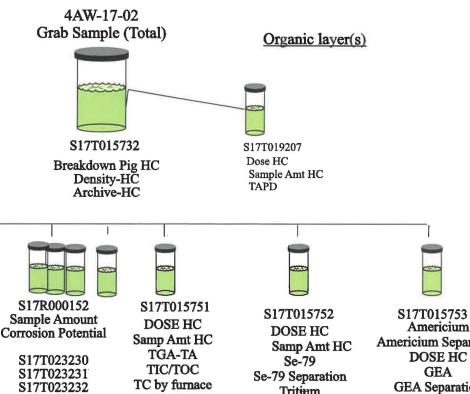
Organic Surface Sample

36 of 62

#### SubSurface Grab DO NOT REFRIGERATE

241-AW 104 Grab 2017 -06 **Grab Samples** Riser: 17 Group 20171600

**TSCA Regulated PCB** 



S17T015748 DOSE HC Samp Amt HC IC-Anions/Org MS Actinides **MS Fission Products 2** Ms Antimony MS Tc-99 OH pH Density

S17T015749 S17T019208 S17T0192018 DOSE HC Samp Amt HC Org Ext **PCB** S17T015750

**PCB 222S** 

**PCB** 

S17T023230 S17T023231 S17T023232 Sample Amount DOSE HC SVOC S17T023353

SVOC

EXT

TOC by furnace **ICP-Metals** Ni-63 Ni-63 Separation Ni-Tracer Hg-Hg Dig

S17T018477

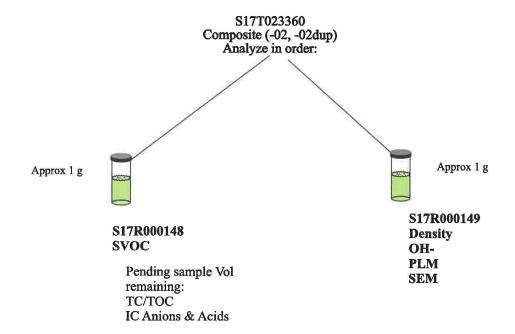
Tritium **Tritium Separation** C-14 C-14 Separation Tc-99-nonPeroxide Tc-99 nonperoxide Sep Plutonium-241 Separation **Tc-99 Separation** 

Americium Americium Separation **GEA Separation** I-129 I-129-Separation Plutonium Plutonium Separation

241-AW 104 Grab 2017 -06 Grab Samples Riser: 17 Group 20171600

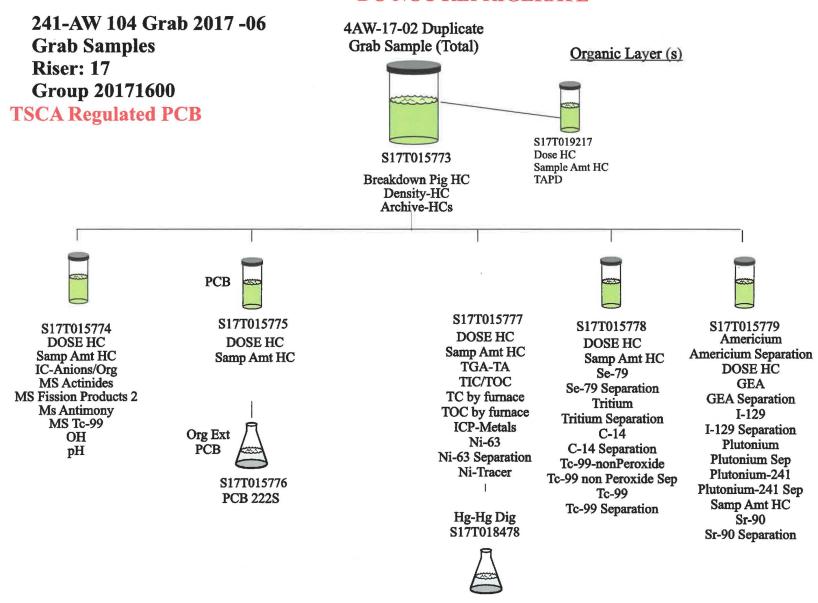
4AW-17-02/02 Dup Organic Layer Composite





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# DO NOT REFRIGERATE SubSurface Grab



241-AW 104 Grab 2017 -06 **Grab Samples** Riser: 17 Group 20171600

**TSCA Regulated PCB** 



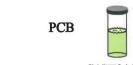
S17T015787

Breakdown Pig HC Density-HC Archive-HC

Subsurface Grab



S17T015788 DOSE HC Samp Amt HC IC-Anions/Org MS Actinides **MS Fission Products 2** Ms Antimony MS Tc-99 OH pН



**PCB** 

S17T015789 DOSE HC Samp Amt HC



S17T015790 **PCB 222S** 



S17T015791

DOSE HC Samp Amt HC TGA-TA TIC/TOC TC by furnace TOC by furnace **ICP- Metals** Ni-63 Ni-63 Separation Ni-Tracer





S17T015792

DOSE HC Samp Amt HC Se-79 Se-79 Separation Tritium

**Tritium Separation** C-14

C-14 Separation Tc-99-nonPeroxide TC-99-nonPeroxide Sep Plutionium-241 Separation

**Tc-99 Separation** 



S17T015793

Americium Americium Separaton DOSE HC **GEA** 

**GEA Separation** I-129

I-129 Separation Plutonium

Plutonium Separation

Samp Amt HC

Sr-90

241-AW 104 Grab 2017 -06 **Grab Samples** Riser: 17 Group 20171600

**TSCA Regulated PCB** 

4AW-17-03 Grab Sample (Total)



S17T015780 Breakdown Pig HC Density-HC Archive-HC

Subsurface Grab



S17T015781 DOSE HC Samp Amt HC IC-Anions/Org MS Actinides MS Fission Products 2 Ms Antimony MS Tc-99 OH pΗ



S17T015782 DOSE HC Samp Amt HC



Org Ext **PCB** 

S17T015783 **PCB 222S** 



S17T015784 DOSE HC Samp Amt HC TGA-TA TIC/TOC TC by furnace TOC by furnace **ICP- Metals** Ni-63 Ni-63 Separation Ni-Tracer





DOSE HC Samp Amt HC Se-79 Se-79 Separation **Tritium Tritium Separation** C-14 C-14 Separation Tc-99-nonPeroxide

**Tc-99 Separation** 



S17T015786

**Americium** 

Americium Separaton

DOSE HC **GEA GEA Separation** I-129 I-129 Separation Plutonium Plutonium Separation Tc-99-non-peroxide Sep Plutonium-2-1

Tc-99-non-peroxide Sep Plutonium-241 Separation Samp Amt HC Sr-90 Sr-90 Separation

241-AW 104 Grab 2017 -06 Grab Samples Riser: 17 Group 20171600

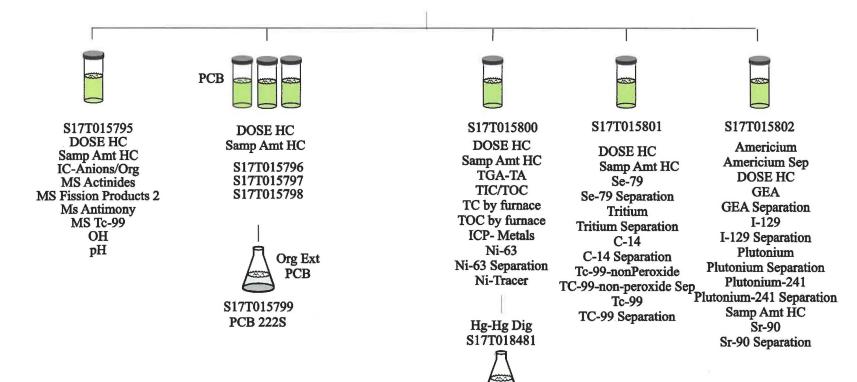
**TSCA Regulated PCB** 

Grab Sample (Total)

4AW-17-05

S17T015794

Breakdown Pig HC Density-HC Archive-HC Subsurface Grab



4AW-17-06

241-AW 104 Grab 2017 -06 **Grab Samples** Riser: 17 Group 20171600

**TSCA Regulated PCB** 



S17T015803

Breakdown Pig HC Density-HC Archive-HC

Subsurface Grab



S17T015804 DOSE HC Samp Amt HC IC-Anions/Org MS Actinides MS Fission Products 2 Ms Antimony MS Tc-99 OH

pH



S17T015805 DOSE HC Samp Amt HC



Org Ext PCB S17T015806 **PCB 222S** 



S17T015807 DOSE HC Samp Amt HC TGA-TA TIC/TOC TC by furnace TOC by furnace ICP- Metals Ni-63 Ni-63 Separation



Ni-Tracer



S17T015808

DOSE HC Samp Amt HC Se-79 Se-79 Separation Tritium **Tritium Separation** C-14 C-14 Separation Tc-99-nonPeroxide Tc-99-non-peroxide Sep

Tc-99 Tc-99 Separation



S17T015809

Americium

Americium Separation DOSE HC **GEA GEA Separation** I-129 I-129 Separation Plutonium Plutonium Separation Plutonium-241 Plutonium-241 Sep Samp Amt HC Sr-90 Sr-90 Separation

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#### DO NOT REFRIGERATE

241-AW 104 Grab 2017 -06 **Grab Samples** Riser: 17 Group 20171600

**TSCA Regulated PCB** 

4AW-17-07 Grab Sample (Total)



S17T015810 Breakdown Pig HC Density-HC Archive-HC

Subsurface Grab



S17T015811 DOSE HC Samp Amt HC IC-Anions/Org MS Actinides MS Fission Products2 Ms Antimony MS Tc-99 OH pН



S17T015812 Dose HC Samp Amt HC



**PCB** 

S17T015813 **PCB 222S** 



S17T015814

DOSE HC Samp Amt HC TGA-TA TIC/TOC TC by furnace TOC by furnace ICP- Metals Ni-63 Ni-63 Separation Ni-Tracer

> Hg-Hg Dig S17T018483



S17T015815

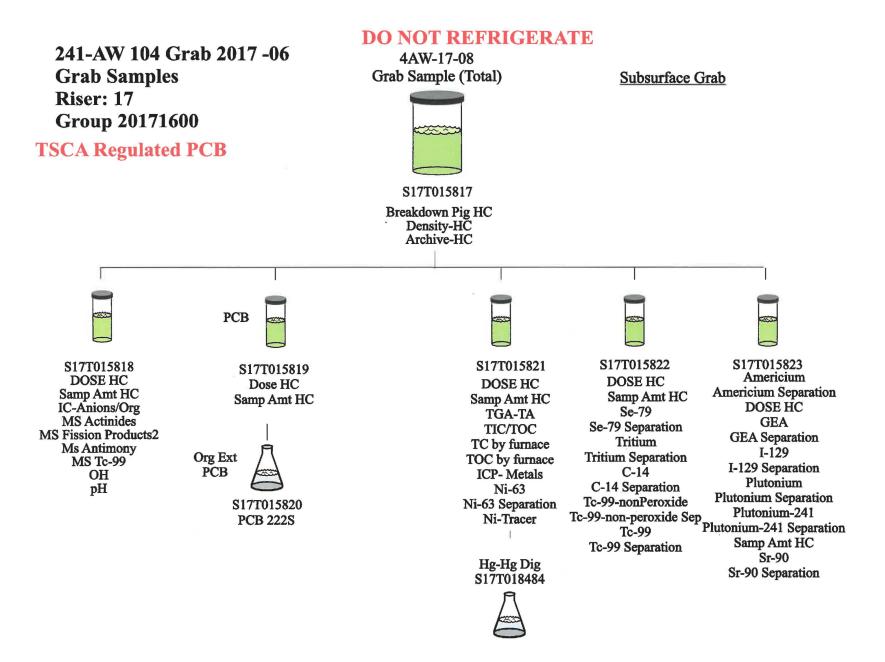
DOSE HC

Samp Amt HC Se-79 Se-79 Separation **Tritium Tritium Separation** C-14 C-14 Separation Tc-99-nonPeroxide

**Tc-99 Separation** 



S17T015816 Americium Americium Separation DOSE HC **GEA GEA Separation** I-129 I-129 Separation Plutonium Plutonium Separation Tc-99-non-peroxide Sep Plutonium-241 Separation Samp Amt HC Sr-90 Sr-90 Separation

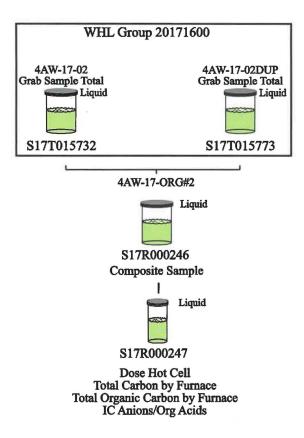


TSCA Regulated

Liquids: <4.12E-03 ug/mL Solids 1.04E-01 ug/g

for PCB

Customer: RESEARCH Project: AW104 2017 APD Group: 20172604



RPP-RPT-604410 Rev. 0

#### APPENDIX B

# SEMIVOLATILE ORGANIC ANALYSES RESULTS INCLUDING TANK 241-AW-104 FLOATING LAYER AND ASSOCIATED QUALITY CONTROL SAMPLES – ANALYTICAL PROJECT DEVELOPMENT (APD)

(12 pages, including cover page)

10/3/201/ 5:02:02PM CompleteBatch Version 3.0.13 batchreports 3.0.13

Page: 1

## LABCORE Completed Batch Report for Batch# 00075479

Analyst: Vogel, Katherine

Instrument: GCMSD10 (JANEWAY)

Method: SVOA 222S LIQUID, LA-523-135 Rev/Mod 4-2

Specification: 241-AW-104 Grab Sampling 2017-06

Prep Batch: 00074945

**Batch Comment:** 

NA

Book#: NA 1016(17

G 80		S								C00000	Limit	**		********	
Seq Ty ample	pe Sequence 1	Sample#	Assoc. Sample RepR	A	* Test	Matrix	Actual	Found	Blank	CTR	(DL/RL/UL)	Unit	Yield	Yield Unit	Flag
1	BLNK	S1707100047	0.0	0	1,2,4-Trichlorobenzene	LIQUID		<1.3770E+03			1.38E+03	ug/L			
1	BLNK	S1707100047	0.0	0	1,2-Dichlorobenzene	LIQUID		<1.3490E+03			1.35E+03	ug/L			
1	BLNK	S1707100047	0.0	0	1,4-Dichlorobenzene	LIQUID		<1.3570E+03			1.36E+03	ug/L			
1	BLNK	S1707100047	0.0	0	1-Butanol	LIQUID		<5.5520E+03			5.55E+03	ug/L			
1	BLNK	S1707100047	0 0	0	2,2-oxybis(1-Chloropro pane)	LIQUID		<1.3990E+03			1.40E+03	ug/L			
1	BLNK	S1707100047	0 0	0	2,4,5-Trichlorophenol	LIQUID		<1.3720E+03			1.37E+03	ug/L			
1	BLNK	S1707100047	0.0	0	2,4,6-Trichlorophenol	LIQUID		<1.3270E+03			1.33E+03	ug/L			
1	BLNK	S1707100047	0.0	0	2,4-Dichlorophenol	LIQUID		<1.3160E+03			1.32E+03	ug/L			
1	BLNK	S1707100047	0.0	0	2,4-Dimethylphenol	LIQUID		<6.8800E+02			688	ug/L			
1	BLNK	\$1707100047	0.0	0	2,4-Dinitrophenol	LIQUID		<4.5100E+02			451	ug/L			
1	BLNK	S1707100047	0.0	0	2,4-Dinitrotoluene	LIQUID		<7.2900E+02			729	ug/L			
1	BLNK	\$1707100047	0 (	O	2,6 bis-4Methylphenol	LIQUID		<6.1700E+02			617	ug/L			
1	BLNK	S1707100047	0 (	0	2,6-Dinitrotoluene	LIQUID		<9.2400E+02			924	ug/L			
1	BLNK	S1707100047	0.0	0	2-Butoxyethanol	LIQUID		<2.8090E+03			2.81E+03	ug/L			
1	BLNK	S1707100047	0 0	0	2-Chloronaphthalene	LIQUID		<1.4000E+03			1.40E+03	ug/L			
1	BLNK	\$1707100047	0.0	0	2-Chlorophenol	LIQUID		<1.1780E+03			1.18E+03	ug/L			
1	BLNK	S1707100047	0 (	0	2-Ethoxyethanol	LIQUID		<1.8640E+03			1.86E+03	ug/L			
1	BLNK	\$1707100047	0.0	0	2-Methylnaphthalene	LIQUID		<1.4120E+03			1.41E+03	ug/L			
1	BLNK	S1707100047	0 (	0	2-Methylphenol	LIQUID		<1.2620E+03			1.26E+03	ug/L			
1	BLNK	\$1707100047	0 (	0	2-Nitroaniline	LIQUID		<1.0040E+03			1.00E+03	ug/L			
1	BLNK	S1707100047	0 (	0	2-Nitrophenol	LIQUID		<1.2600E+03			1.26E+03	ug/L			
1	BLNK	S1707100047	0 0		Total	LIQUID		<1.1240E+03			1.12E+03	ug/L			
I	BLNK	S1707100047	0 (	0	3-Nitroaniline	LIQUID		<1.2070E+03			1.21E+03	ug/L			

Units shown for QC (BLK/BKG) may not reflect the actual units.

10/3/201/ 3:02:02PM CompleteBatch Version 3.0.13 batchreports 3.0.13

Page: 2

# LABCORE Completed Batch Report for Batch# 00075479

Seq Type	BLNK	Sample# S1707100047	Assoc. Sample	RepR 0 0		* Test Matrix 4,6-Dinitro-2-methylphe LIQUID nol	Actual	Found Bla <2.9900E+02	ınk	CTR	Limit (DL/RL/UL) 299	Unit ug/L	Yield	Yield Unit	Flags
1	BLNK	\$1707100047		0 0	0	4-Bromophenyl-phenyle LIQUID		<8.3400E+02			834	ug/L			
1	BLNK	S1707100047		0 0	0	ther 4-Chloro-3-methylphen LIQUID ol		<1.2730E+03			1.27E+03	ug/L			
1	BLNK	\$1707100047		0.0	0	4-Chloroaniline LIQUID		<1.3820E+03			1.38E+03	ug/L			
1	BLNK	S1707100047		0 0		4-Chlorophenyl-phenyle LIQUID		<1.1520E+03			1.15E+03	ug/L			
1	BLNK	\$1707100047		0 0	0	4-Methylphenol LIQUID		0				ug/L			
1	BLNK	S1707100047		0 0	0	4-Nitroaniline LIQUID		<1.0800E+03			1.08E+03	ug/L			
1	BLNK	\$1707100047		0.0	0	4-Nitrophenol LIQUID		<7.3300E+02			733	ug/L			
1	BLNK	S1707100047		0 0	0	Acenaphthene LIQUID		<1.2980E+03			1.30E+03	ug/L			
1	BLNK	S1707100047		0 0		Acenaphthylene LIQUID		<1.5010E+03			1.50E+03	ug/L			
1	BLNK	S1707100047		0 0	0	Anthracene LIQUID		<6.2000E+02			620	ug/L			
1	BLNK	S1707100047		0.0	0	Benzo(a)anthracene LIQUID		<5.6400E+02			564	ug/L			
1	BLNK	S1707100047		0 0		Benzo(a)pyrene LIQUID		<5.4200E+02			542	ug/L			
1	BLNK	S1707100047		0 0	0	Benzo(b)fluoranthene LIQUID		<6.2200E+02			622	ug/L			
1	BLNK	S1707100047		0 0	0	Benzo(g,h,i)perylene LIQUID		<5.3000E+02			530	ug/L			
1	BLNK	S1707100047		0 0	0	Benzo(k)fluoranthene LIQUID		<6.0200E+02			602	ug/L			
1	BLNK	S1707100047		0.0	0	Butylbenzylphthalate LIQUID		<4.6400E+02			464	ug/L			
1	BLNK	S1707100047		0.0	0	Chrysene LIQUID		<6.4900E+02			649	ug/L			
1	BLNK	S1707100047		0.0	0	Cyclohexanone LIQUID		<1.3020E+03			1.30E+03	ug/L			
1	BLNK	S1707100047		0.0		Di-n-butylphthalate LIQUID		<3.7600E+02			376	ug/L			
1	BLNK	S1707100047		0 0	0	Di-n-octylphthalate LIQUID		<4.6600E+02			466	ug/L			
1	BLNK	\$1707100047		0.0		Dibenz(a,h)anthracene LIQUID		<4.5300E+02			453	ug/L			
1	BLNK	S1707100047		0.0	0	Dibenzofuran LIQUID		<1.1900E+03			1.19E+03	ug/L			
1	BLNK	S1707100047		0 0	0	Diethylphthalate LIQUID		<7.2100E+02			721	ug/L			
1	BLNK	S1707100047		0.0		Dimethylphthalate LIQUID		<8.4500E+02			845	ug/L			
1	BLNK	S1707100047		0.0	0	Diphenylamine LIQUID		<7.7300E+02			773	ug/L			
1	BLNK	S1707100047		0.0		Fluoranthene LIQUID		<4.0600E+02			406	ug/L			
1	BLNK	S1707100047		0.0	30	Fluorene LIQUID		<1.2720E+03			1.27E+03	ug/L			
1	BLNK	S1707100047		0 0		Hexachlorobenzene LIQUID		<1.0830E+03			1.08E+03	ug/L			
1	BLNK	S1707100047		0.0	0	Hexachlorobutadiene LIQUID		<1.5020E+03			1.50E+03	ug/L			
1	BLNK	S1707100047		0 0	0	Hexachlorocyclopentadi LIQUID ene		<2.6200E+02			262	ug/L			
1	BLNK	S1707100047		0 0	0	Hexachloroethane LIQUID		<1.4000E+03			1.40E+03	ug/L			
1	BLNK	S1707100047		0 0	0	Indeno(1,2,3-cd)pyrene LIQUID		<2.6200E+02			262	ug/L			
1	BLNK	S1707100047		0 0	0	Isobutanol LIQUID		<2.1570E+03			2.16E+03	ug/L			
1	BLNK	S1707100047		0 0	0	Isophorone LIQUID		<1.3490E+03			1.35E+03	ug/L			
1	BLNK	S1707100047		0 0	0	Methylphenols, Total LIQUID		0 .				ug/L			
1	BLNK	S1707100047		0 0	0	N-Nitrosodimethylamin LIQUID	1	<1.2400E+03			1.24E+03	ug/L			

Units shown for QC (BLK/BKG) may not reflect the actual units.

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# LABCORE Completed Batch Report for Batch# 00075479

Seq Ty		Sample#	Assoc. Sample Re		7 N	* Test	Matrix	Actual	Found	Blank	CTR	Limit (DL/RL/UL)		Yield	Yield Unit Flag
1	BLNK	\$1707100047			0	N-Nitrosodipropylamin			<1.3550E+03			1.36E+03	ug/L		
1	BLNK	\$1707100047			0		LIQUID		<1.0870E+03			1.09E+03	ug/L		
1	BLNK	S1707100047			0		LIQUID		<1.2330E+03			1.23E+03	ug/L		
1	BLNK	S1707100047		0 0	0		LIQUID		<1.2180E+03			1.22E+03	ug/L		
1	BLNK	S1707100047			O		LIQUID		<5.7000E+02			570	ug/L		
1	BLNK	S1707100047		00	0	Phenanthrene	LIQUID		<5.9300E+02			593	ug/L		
1	BLNK	S1707100047			0		LIQUID		<1.2120E+03			1.21E+03	ug/L		
1	BLNK	\$1707100047		0 0	0	Pyrene	LIQUID		<3.8100E+02			381	ug/L		
1	BLNK	S1707100047		0 0	0		LIQUID		<1.4620E+03			1.46E+03	ug/L		
1	BLNK	S1707100047		00	0	Tri-n-butylphosphate	LIQUID		<5.1300E+02			513	ug/L		
1	BLNK	S1707100047		0 0	0	bis(2-Chloroethoxy)met hane	LIQUID		<1.3370E+03			1.34E+03	ug/L		
1	BLNK	S1707100047		0 0	Ö	bis(2-Ethylhexyl)phthal ate	LIQUID		<4.4000E+02			440	ug/L		
1	BLNK	\$1707100047		0 0	0	ether	LIQUID		<1.2970E+03			1.30E+03	ug/L		
1	BLNK	S1707100047		0 0	0	2,4,6-Tribromophenol	LIQUID	40000	3.4800E+04			500	ug/L	87	% Recovery
1	BLNK	S1707100047		00	0	2-Fluorobiphenyl	LIQUID	20000	2.0281E+04			500	ug/L	101.4	% Recovery
1	BLNK	S1707100047		00	0	2-Fluorophenol	LIQUID	40000	3.4971E+04			500	ug/L	87.428	% Recovery
1	BLNK	S1707100047		0 0	0	Nitrobenzene-d5	LIQUID	20000	2.1330E+04			500	ug/L	106.65	% Recovery
1	BLNK	S1707100047		00	0	Phenol-d6	LIQUID	40000	3.5220E+04			500	ug/L	88.05	% Recovery
1	BLNK	S1707100047		00	O	Terphenyl-d14	LIQUID	20000	2.3408E+04			500	ug/L	117.04	% Recovery
ample !	Sequence 2														
2	LCS	S1707100048		00	O	1,2,4-Trichlorobenzene	LIQUID	133330	1.0377E+05			1.38E+04	ug/L	77.829	% Recovery
2	LCS	S1707100048		0 0	0	1,2-Dichlorobenzene	LIQUID	133330	1.0597E+05			1.35E+04	ug/L	79.479	% Recovery
2	LCS	S1707100048		0 0	Ö	1,4-Dichlorobenzene	LIQUID	133330	1.0557E+05			1.36E+04	ug/L	79.179	% Recovery
2	LCS	S1707100048		0 0	0	1-Butanol	LIQUID	533330	3.8325E+05			5.55E+04	ug/L	71.86	% Recovery
2	LCS	\$1707100048		0 0		2,2-oxybis(1-Chloropro	-	133330	1.0494E+05			1.40E+04	ug/L	78.707	% Recovery
2	LCS	S1707100048		0 0	0		LIQUID	133330	9.7900E+04			1.37E+04	ug/L	73.427	% Recovery
. 2	LCS	S1707100048			0		LIQUID	133330	1.0191E+05			1.33E+04	ug/L	76.434	% Recovery
2	LCS	S1707100048		0 0	0	2,4-Dichlorophenol	LIQUID	133330	1.0402E+05			1.32E+04	ug/L	78.017	% Recovery
2	LCS	S1707100048		00	0	2,4-Dimethylphenol	LIQUID	133330	1.0399E+05			6.88E+03	ug/L	77.994	% Recovery
2	LCS	\$1707100048		0 0	0	2,4-Dinitrophenol	LIQUID	133330	9.8310E+04			4.51E+03	ug/L	73.734	% Recovery
2	LCS	S1707100048		0.0	O	2,4-Dinitrotoluene	LIQUID	133330	1.0069E+05			7.29E+03	ug/L	75.519	% Recovery
2	LCS	S1707100048		0 0	0	2,6 bis-4Methylphenol	LIQUID	133330	1.0280E+05			6.17E+03	ug/L	77.102	% Recovery
2	LCS	\$1707100048		00	0	2,6-Dinitrotoluene	LIQUID	133330	9.8860E+04			9.24E+03	ug/L	74.147	% Recovery
2	LCS	S1707100048		0 0	0	2-Butoxyethanol	LIQUID	266670	2.0568E+05			2.81E+04	ug/L	77.129	% Recovery
2	LCS	S1707100048			0	2-Chloronaphthalene	LIQUID	133330	1.0298E+05			1.40E+04	ug/L	77.237	% Recovery
2	LCS	S1707100048			0	2-Chlorophenol	LIQUID	133330	1.0596E+05			1.18E+04	ug/L	79.472	% Recovery
2	LCS	S1707100048			0	2-Ethoxyethanol	LIQUID	133330	1.0266E+05			1.86E+04	ug/L	76.997	% Recovery
2	LCS	S1707100048		0 0		2-Methylnaphthalene	LIQUID	133330	1.0457E+05			1.41E+04	ug/L	78,429	% Recovery

Units shown for QC (BLK/BKG) may not reflect the actual units.

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# LABCORE Completed Batch Report for Batch# 00075479

Seq Type	LCS	Sample# \$1707100048	Assoc. Sample RepR		* Test 2-Methylphenol	Matrix LIQUID	Actual 133330	Found 1.0764E+05	Blank	CTR	Limit (DL/RL/UL) 1.26E+04	Unit ug/L	Yield 80.732	Yield Unit Flags % Recovery
2	LCS	S1707100048	00	-	2-Nitroaniline	-	133330				1.20E+04 1.00E+04	-	75.602	4
2	LCS	S1707100048 S1707100048	00	_	2-Nitroantine 2-Nitrophenol	LIQUID LIQUID	133330	1.0080E+05 1.0114E+05			1.00E+04 1.26E+04	ug/L ug/L	75.857	% Recovery % Recovery
	LCS				•									
2	LCS	S1707100048	0 0	0	3 & 4 Methylphenol Total	LIQUID	133330	1.0614E+05			1.12E+04	ug/L	79.607	% Recovery
2	LCS	S1707100048	0 0	0	3-Nitroaniline	LIQUID	133330	9.8710E+04			1.21E+04	ug/L	74.034	% Recovery
2	LCS	S1707100048	0 0	0	4,6-Dinitro-2-methylph	-	133330	1.0087E+05			2.99E+03	ug/L	75.654	% Recovery
2	LCS	S1707100048	0 0	0	nol 4-Bromophenyl-phenyl	le LIQUID	133330	1.0093E+05			8.34E+03	ug/L	75.699	% Recovery
	1.00	61707100040	0.0	^	ther	LIOUR	122220	1 001 (17:05			1.275104	/T	75 100	9/ Dagayamı
2	LCS	S1707100048	0 0	O	4-Chloro-3-methylpher ol	LIQUID	133330	1.0016E+05			1.27E+04	ug/L	75.122	% Recovery
2	LCS	S1707100048	0 0	0	4-Chloroaniline	LIQUID	133330	1.0265E+05			1.38E+04	ug/L	76.989	% Recovery
2	LCS	S1707100048	0 0	0	4-Chlorophenyl-phenyl	le LIQUID	133330	1.0177E+05			1.15E+04	ug/L	76.329	% Recovery
2	LCS	S1707100048	0 0	0	4-Methylphenol	LIQUID		0				ug/L	N/A	
2	LCS	S1707100048	0 0	0	4-Nitroaniline	LIQUID	133330	9.7820E+04			1.08E+04	ug/L	73.367	% Recovery
2	LCS	S1707100048	0 0		4-Nitrophenol	LIQUID	133330	1.0419E+05			7.33E+03	ug/L	78.144	% Recovery
2	LCS	S1707100048	0.0	Ó	Acenaphthene	LIQUID	133330	1.0177E+05			1.30E+04	ug/L	76.329	% Recovery
2	LCS	S1707100048	0.0	0	Acenaphthylene	LIQUID	133330	1.0303E+05			1.50E+04	ug/L	77.274	% Recovery
2	LCS	S1707100048	0.0	Ó	Anthracene	LIQUID	133330	1.0353E+05			6.20E+03	ug/L	77.649	% Recovery
2	LCS	S1707100048	0 0		Benzo(a)anthracene	LIQUID	133330	1.0268E+05			5.64E+03	ug/L	77.012	% Recovery
2	LCS	S1707100048	0 0	0	Benzo(a)pyrene	LIQUID	133330	1.0184E+05			5.42E+03	ug/L	76,382	% Recovery
2	LCS	S1707100048	0 0	0	Benzo(b)fluoranthene	LIQUID	133330	1.0704E+05			6.22E+03	ug/L	80.282	% Recovery
2	LCS	S1707100048	0 0	0	Benzo(g,h,i)perylene	LIQUID	133330	8.9980E+04			5.30E+03	ug/L	67.487	% Recovery
2	LCS	S1707100048	0 0	0	Benzo(k)fluoranthene	LIQUID	133330	1.0531E+05			6.02E+03	ug/L	78.984	% Recovery
2	LCS	S1707100048	0 0	0	Butylbenzylphthalate	LIQUID	133330	1.0269E+05			4.64E+03	ug/L	77.019	% Recovery
2	LCS	S1707100048	0 0	0	Chrysene	LIQUID	133330	1.1458E+05			6.49E+03	ug/L	85.937	% Recovery
2	LCS	S1707100048	0 0	0	Cyclohexanone	LIQUID	133330	1.0463E+05			1.30E+04	ug/L	78,474	% Recovery
2	LCS	S1707100048	0 0	0	Di-n-butylphthalate	LIQUID	133330	1.0925E+05			3.76E+03	ug/L	81.94	% Recovery
2	LCS	\$1707100048	0.0	0	Di-n-octylphthalate	LIQUID	133330	1.0849E+05			4.66E+03	ug/L	81.37	% Recovery
2	LCS	\$1707100048	0 0	0	Dibenz(a,h)anthracene		133330	8.4540E+04			4.53E+03	ug/L	63.407	% Recovery
2	LCS	S1707100048	0 0	0	Dibenzofuran	LIQUID	133330	1.0125E+05			1.19E+04	ug/L	75.939	% Recovery
2	LCS	S1707100048	0 0	0	Diethylphthalate	LIQUID	133330	9.9660E+04			7.21E+03	ug/L	74.747	% Recovery
2	LCS	\$1707100048	0.0	0	Dimethylphthalate	LIQUID	133330	1.0223E+05			8.45E+03	ug/L	76.674	% Recovery
2	LCS	S1707100048	0 0	0	Diphenylamine	LIQUID	133330	1.0344E+05			7.73E+03	ug/L	77.582	% Recovery
2	LCS	\$1707100048	0 0	Ó	Fluoranthene	LIQUID	133330	1.0468E+05			4.06E+03	ug/L	78.512	% Recovery
2	LCS	S1707100048	0 0	0	Fluorene	LIQUID	133330	1.0207E+05			1.27E+04	ug/L	76.554	% Recovery
2	LCS	\$1707100048	0 0	0	Hexachlorobenzene	LIQUID	133330	1.0132E+05			1.08E+04	ug/L	75 992	% Recovery
2	LCS	S1707100048	0 0	0	Hexachlorobutadiene	LIQUID	133330	1.0479E+05			1.50E+04	ug/L	78.594	% Recovery
2	LCS	S1707100048	0 0	0	Hexachlorocyclopenta	diLIQUID	133330	9.9650E+04			2.62E+03	ug/L	74.739	% Recovery
2	LCS	\$1707100048	0 0	O	ene Hexachloroethane	LIQUID	133330	1.0675E+05			1.40E+04	ug/L	80.065	% Recovery

Units shown for QC (BLK/BKG) may not reflect the actual units.

\* = Rejected Analyte C = Cancelled Analyte

N, A = Non-reported Analyte

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#### LABCORE Completed Batch Report for Batch# 00075479

Seq Ty 2	pe LCS	Sample# S1707100048	Assoc. Sample R		<b>A</b> O	* Test Indeno(1,2,3-cd)pyrene	Matrix	Actual 133330	Found 8.9870E+04	Blank	CTR	Limit (DL/RL/UL) 2.62E+03	Unit ug/L	Yield 67.404	Yield Unit Flag
2	LCS	S1707100048			0		LIQUID	266670	2.1692E+05			2.16E+04	ug/L	81.344	% Recovery
2	LCS	\$1707100048			o		LIQUID	133330	1.0117E+05			1.35E+04	ug/L	75.879	% Recovery
2	LCS	S1707100048			0	1961 10 • 10 · 10 · 10 · 10 · 10 · 10 · 10	LIQUID	155550	0			1.556.04	ug/L	N/A	70 Recovery
2	LCS	S1707100048			o	N-Nitrosodimethylamin		133330	1.0666E+05			1.24E+04	ug/L	79.996	% Recovery
2	LCS	S1707100048		00	0	N-Nitrosodipropylamine	eLIQUID	133330	1.0608E+05			1.36E+04	ug/L	79.562	% Recovery
2	LCS	S1707100048		0 0	O	N-Nitrosomorpholine	LIQUID	133330	1.0703E+05			1.09E+04	ug/L	80.275	% Recovery
2	LCS	S1707100048		0.0	0	Naphthalene	LIQUID	133330	1.0629E+05			1.23E+04	ug/L	79.719	% Recovery
2	LCS	S1707100048		0 0	0	Nitrobenzene	LIQUID	133330	1.0375E+05			1.22E+04	ug/L	77.814	% Recovery
2	LCS	S1707100048		0.0	0		LIQUID	133330	9.8200E+04			5.70E+03	ug/L	73.652	% Recovery
2	LCS	S1707100048		0 0	0		LIQUID	133330	1.0285E+05			5.93E+03	ug/L	77.139	% Recovery
2	LCS	S1707100048			0		LIQUID	133330	1.0310E+05			1.21E+04	ug/L	77.327	% Recovery
2	LCS	S1707100048			0		LIQUID	133330	1.0636E+05			3.81E+03	ug/L	79.772	% Recovery
2	LCS	S1707100048			0		LIQUID	133330	1.0488E+05			1.46E+04	ug/L	78.662	% Recovery
2	LCS	S1707100048			0		LIQUID	133330	1.0736E+05			5.13E+03	ug/L	80.522	% Recovery
2	LCS	S1707100048			0	bis(2-Chloroethoxy)met		133330	1.0385E+05			1.34E+04	ug/L	77.889	% Recovery
2	LCS	S1707100048		0 0	0	bis(2-Ethylhexyl)phthal	LIQUID	133330	1.0204E+05			4.40E+03	ug/L	76.532	% Recovery
2	LCS	\$1707100048		00	0		LIQUID	133330	1.0506E+05			1.30E+04	ug/L	78.797	% Recovery
2	LCS	S1707100048		0 0	0	2,4,6-Tribromophenol	LIQUID	400000	3.2997E+05			5.00E+03	ug/L	82.492	% Recovery
2	LCS	S1707100048		0 0	0	2-Fluorobiphenyl	LIQUID	200000	1.7837E+05			5.00E+03	ug/L	89.185	% Recovery
2	LCS	S1707100048		0 0	O	2-Fluorophenol	LIQUID	400000	3.1024E+05			5.00E+03	ug/L	77.56	% Recovery
2	LCS	S1707100048		00	Ó	Nitrobenzene-d5	LIQUID	200000	1.8605E+05			5.00E+03	ug/L	93.025	% Recovery
2	LCS	S1707100048		0 0	0	Phenol-d6	LIQUID	400000	3.1067E+05			5.00E+03	ug/L	77.668	% Recovery
2 Sample	LCS Sequence 3	S1707100048		0 0	0	Terphenyl-d14	LIQUID	200000	1.9233E+05			5.00E+03	ug/L	96.165	% Recovery
3	SAMPLE	S17R000148		0 0	0	1,2,4-Trichlorobenzene	LIQUID	N/A	<1.3770E+05			1.38E+05	ug/L		U
3	SAMPLE	S17R000148		0.0	0		LIQUID	N/A	<1.3490E+05			1.35E+05	ug/L		U
3	SAMPLE	S17R000148		0 0	0	-	LIQUID	N/A	<1.3570E+05			1.36E+05	ug/L		Ua
3	SAMPLE	S17R000148		0.0	0	1-Butanol	LIQUID	N/A	<5.5520E+05			5.55E+05	ug/L		Ū
3	SAMPLE	S17R000148		0 0	0	2,2-oxybis(1-Chloropro		N/A	<1.3990E+05			1.40E+05	ug/L		U
3	SAMPLE	S17R000148		00	Ö	2,4,5-Trichlorophenol	LIQUID	N/A	<1.3720E+05			1.37E+05	ug/L		Ü
3	SAMPLE	S17R000148		0 0	0	2,4,6-Trichlorophenol	LIQUID	N/A	<1.3270E+05			1.33E+05	ug/L		U
3	SAMPLE	S17R000148		0 0	0	2,4-Dichlorophenol	LIQUID	N/A	<1.3160E+05			1.32E+05	ug/L		U
3	SAMPLE	S17R000148		0 0	0	2,4-Dimethylphenol	LIQUID	N/A	<6.8800E+04			6.88E+04	ug/L		U
3	SAMPLE	S17R000148			0	2,4-Dinitrophenol	LIQUID	N/A	<4.5100E+04			4.51E+04	ug/L		U
3	SAMPLE	S17R000148		0 0	0	2,4-Dinitrotoluene	LIQUID	N/A	<7.2900E+04			7.29E+04	ug/L		U
3	SAMPLE	S17R000148		0 0	Ó	2,6 bis-4Methylphenol	and the second s	N/A	<6.1700E+04			6.17E+04	ug/L		Ü

Units shown for QC (BLK/BKG) may not reflect the actual units.

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## LABCORE Completed Batch Report for Batch# 00075479

Seq T	0. The contract of the contrac	Sample#	Assoc. Sample Repl			Matrix	Actual	Found	Blank	CTR	Limit (DL/RL/UL)		Yield	Yield Unit	
3	SAMPLE	S17R000148	0		2,6-Dinitrotoluene	LIQUID	N/A	<9.2400E+04			9.24E+04	ug/L			U
3	SAMPLE	S17R000148	0		2-Butoxyethanol	LIQUID	N/A	<2.8090E+05			2.81E+05	ug/L			U
3	SAMPLE	S17R000148	0		2-Chloronaphthalene	LIQUID	N/A	<1.4000E+05			1.40E+05	ug/L			U
3	SAMPLE	S17R000148	0		2-Chlorophenol	LIQUID	N/A	<1.1780E+05			1.18E+05	ug/L			U
3	SAMPLE	S17R000148	0		2-Ethoxyethanol	LIQUID	N/A	<1.8640E+05			1.86E+05	ug/L			U
3	SAMPLE	S17R000148	0		2-Methylnaphthalene	LIQUID	N/A	<1.4120E+05			1.41E+05	ug/L			Ü
3	SAMPLE	S17R000148	0		2-Methylphenol	LIQUID	N/A	<1.2620E+05			1.26E+05	ug/L			U
3	SAMPLE	S17R000148	0	-	2-Nitroaniline	LIQUID	N/A	<1.0040E+05			1.00E+05	ug/L			U
3	SAMPLE	S17R000148	0	7	2-Nitrophenol	LIQUID	N/A	<1.2600E+05			1.26E+05	ug/L			U
3	SAMPLE	S17R000148	0		3 & 4 Methylphenol Total	LIQUID	N/A	<1.1240E+05			1.12E+05	ug/L			U
3	SAMPLE	S17R000148	0		3-Nitroaniline	LIQUID	N/A	<1.2070E+05			1.21E+05	ug/L			U
3	SAMPLE	S17R000148	0	0 0	4,6-Dinitro-2-methylp nol		N/A	<2.9900E+04			2.99E+04	ug/L			U
3	SAMPLE	S17R000148		0 0	4-Bromophenyl-pheny ther	,	N/A	<8.3400E+04			8.34E+04	ug/L			U
3	SAMPLE	S17R000148	0		4-Chloro-3-methylphe ol		N/A	<1.2730E+05			1.27E+05	ug/L			U
3	SAMPLE	S17R000148	0		4-Chloroaniline	LIQUID	N/A	<1.3820E+05			1.38E+05	ug/L			U
3	SAMPLE	S17R000148	0		4-Chlorophenyl-pheny ther		N/A	<1.1520E+05			1.15E+05	ug/L			Ü
3	SAMPLE	S17R000148	0		4-Methylphenol	LIQUID	N/A	0				ug/L			U
3	SAMPLE	S17R000148	0		4-Nitroaniline	LIQUID	N/A	<1_0800E+05			1.08E+05	ug/L			U
3	SAMPLE	S17R000148	0	0 0	4-Nitrophenol	LIQUID	N/A	<7.3300E+04			7.33E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Acenaphthene	LIQUID	N/A	<1.2980E+05			1.30E+05	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Acenaphthylene	LIQUID	N/A	<1.5010E+05			1.50E+05	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Anthracene	LIQUID	N/A	<6.2000E+04			6.20E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Benzo(a)anthracene	LIQUID	N/A	<5.6400E+04			5.64E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Benzo(a)pyrene	LIQUID	N/A	<5.4200E+04			5.42E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 O	Benzo(b)fluoranthene	LIQUID	N/A	<6.2200E+04			6.22E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	(6) , /1 2	LIQUID	N/A	<5.3000E+04			5.30E+04	ug/L			Ua
3	SAMPLE	S17R000148	0	0 O	Benzo(k)fluoranthene		N/A	<6.0200E+04			6.02E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Butylbenzylphthalate	LIQUID	N/A	<4.6400E+04			4.64E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Chrysene	LIQUID	N/A	<6.4900E+04			6.49E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Cyclohexanone	LIQUID	N/A	<1.3020E+05			1.30E+05	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Di-n-butylphthalate	LIQUID	N/A	<3.7600E+04			3.76E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Di-n-octylphthalate	LIQUID	N/A	<4.6600E+04			4.66E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Dibenz(a,h)anthracen	e LIQUID	N/A	<4.5300E+04			4.53E+04	ug/L			Ua
3	SAMPLE	S17R000148	0	0 0	Dibenzofuran	LIQUID	N/A	<1.1900E+05			1.19E+05	ug/L			U
3	SAMPLE	S17R000148	0	0 O	Diethylphthalate	LIQUID	N/A	<7.2100E+04			7.21E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Dimethylphthalate	LIQUID	N/A	<8.4500E+04			8.45E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Diphenylamine	LIQUID	N/A	<7.7300E+04			7.73E+04	ug/L			U
3	SAMPLE	S17R000148	0	0 0	Fluoranthene	LIQUID	N/A	<4.0600E+04			4.06E+04	ug/L			U

Units shown for QC (BLK/BKG) may not reflect the actual units.

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# LABCORE Completed Batch Report for Batch# 00075479

Seq T		Sample#	Assoc. Sample RepR			Matrix	Actual	Found	Blank	CTR	Limit (DL/RL/UL)		Yield	Yield Unit	
3	SAMPLE	S17R000148		0 0		LIQUID	N/A	<1.2720E+05			1.27E+05	ug/L			U
3	SAMPLE	S17R000148	0 (		Hexachlorobenzene	LIQUID	N/A	<1.0830E+05			1.08E+05	ug/L			U
3	SAMPLE	S17R000148	0 (			LIQUID	N/A	<1.5020E+05			1.50E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	0	Hexachlorocyclopentad ene	LIQUID	N/A	<2.6200E+04			2.62E+04	ug/L			Ü
3	SAMPLE	S17R000148	0 (	0	Hexachloroethane	LIQUID	N/A	<1.4000E+05			1.40E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	0	Indeno(1,2,3-cd)pyrene	LIQUID	N/A	<2.6200E+04			2.62E+04	ug/L			Ua
3	SAMPLE	S17R000148	0 (	0	Isobutanol	LIQUID	N/A	<2.1570E+05			2.16E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	0	Isophorone	LIQUID	N/A	<1.3490E+05			1.35E+05	ug/L			Ü
3	SAMPLE	S17R000148	0 (	0	Methylphenols, Total	LIQUID	N/A	0				ug/L			U
3	SAMPLE	S17R000148	0 (	0	N-Nitrosodimethylamin e	LIQUID	N/A	<1.2400E+05			1.24E+05	ug/L			Ü
3	SAMPLE	S17R000148	0 (	0	N-Nitrosodipropylamin	eLIQUID	N/A	<1.3550E+05			1.36E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	Ó	N-Nitrosomorpholine	LIQUID	N/A	<1.0870E+05			1.09E+05	ug/L			Ü
3	SAMPLE	S17R000148	0 (	0	Naphthalene	LIQUID	N/A	<1.2330E+05			1.23E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	0	Nitrobenzene	LIQUID	N/A	<1.2180E+05			1.22E+05	ug/L			Ü
3	SAMPLE	S17R000148	0.0	0	Pentachlorophenol	LIQUID	N/A	<5.7000E+04			5.70E+04	ug/L			U
3	SAMPLE	S17R000148	0 (	0	Phenanthrene	LIQUID	N/A	<5.9300E+04			5.93E+04	ug/L			U
3	SAMPLE	S17R000148	0 (	0 (	Phenol	LIQUID	N/A	<1.2120E+05			1.21E+05	ug/L			U
3	SAMPLE	S17R000148	Ó (	0 0	Pyrene	LIQUID	N/A	<3.8100E+04			3.81E+04	ug/L			U
3	SAMPLE	S17R000148	0 (	0	Pyridine	LIQUID	N/A	<1.4620E+05			1.46E+05	ug/L			U
3	SAMPLE	S17R000148	0 0	Ó	Tri-n-butylphosphate	LIQUID	N/A	1.2310E+05			5.13E+04	ug/L			j
3	SAMPLE	S17R000148	0 (	0	bis(2-Chloroethoxy)met hane	LIQUID	N/A	<1.3370E+05			1.34E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	0	bis(2-Ethylhexyl)phthal ate	LIQUID	N/A	<4.4000E+04			4.40E+04	ug/L			U
3	SAMPLE	S17R000148	0 (	0	bis-(2-Chloroethyl) ether	LIQUID	N/A	<1.2970E+05			1.30E+05	ug/L			U
3	SAMPLE	S17R000148	0 (	0 0	2,4,6-Tribromophenol	LIQUID	400000	6.3200E+04			5.00E+04	ug/L	15.8	% Recovery	/
3	SAMPLE	S17R000148	0 (	0 0	2-Fluorobiphenyl	LIQUID	200000	2.1130E+05			5.00E+04	ug/L	105.65	% Recovery	,
3	SAMPLE	S17R000148	0 (	0 0	2-Fluorophenol	LIQUID	400000	<5.0000E+04			5.00E+04	ug/L			U
3	SAMPLE	S17R000148	0 (	0 0	Nitrobenzene-d5	LIQUID	200000	1.9330E+05			5.00E+04	ug/L	96.65	% Recovery	/
3	SAMPLE	S17R000148	0 (	0	Phenol-d6	LIQUID	400000	<5.0000E+04			5.00E+04	ug/L			U
3	SAMPLE	S17R000148	0.1	0 0	Terphenyl-d14	LIQUID	200000	2.1770E+05			5.00E+04	ug/L	108.85	% Recovery	/
3	SAMPLE	S17R000148	0 (	0 0	Tridecane	LIQUID	N/A	220000				ug/L			JNT

#### **Comments Section:**

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## LABCORE Completed Batch Report for Batch# 00075479

Data Flagger Status: Flagging Completed

Final Page for Batch# 00075479

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#### LABCORE Completed Batch Report for Batch# 00075479

Seq	Sample#	QC Type	Assoc. Sample	Sample Group	Method	Customer Id	Specification	Hold Time/Expiration Date
1	S1707100047	BLNK			SVOA 222S LIQUID			
2	S1707100048	LCS			SVOA 222S LIQUID			
3	S17R000148	SAMPLE		20171600	SVOA 222S LIQUID	4AW-17-02	241-AW-104 Grab Sampling 2017-06	08/22/2017 14:15

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# LABCORE Completed Batch Report for Batch# 00074945

Analyst: Fabre, Martha

Book#: see bunchshort

10/0/6/17

Instrument:

Method: ORGANIC DILUTION PREP #2, LA-523-149 Rev/Mod 3-2

Specification: 241-AW-104 Grab Sampling 2017-06

Prep Batch: N/A **Batch Comment:** 

NA

e	Sample#	Assoc. Sample RepR	A	* Test	Matrix	Actual	Found	Blank	CTR	Limit (DL/RL/UL)	Unit	Yield	Yield Unit	Flags
BLNK	S1707100047		O	Dose Rate(1)	LIQUID									
BLNK	S1707100047		O	Digest Factor(1)	LIQUID		0.001000				uL/uL			
LCS	S1707100048		O	Dose Rate(1)	LIQUID							N/A		
LCS	S1707100048		O	Digest Factor(1)	LIQUID		0.000100				uL/uL	N/A		
SAMPLE	S17R000148		0	Dose Rate(1)	LIQUID									
SAMPLE	S17R000148		0	Digest Factor(1)	LIQUID		0.000010				uL/uL			
	BLNK BLNK LCS LCS SAMPLE	BLNK \$1707100047 BLNK \$1707100047 LCS \$1707100048 LCS \$1707100048 SAMPLE \$17R000148	BLNK S1707100047 BLNK S1707100047 LCS S1707100048 LCS S1707100048 SAMPLE S17R000148	BLNK \$1707100047 O BLNK \$1707100047 O LCS \$1707100048 O LCS \$1707100048 O SAMPLE \$17R000148 O	BLNK         \$1707100047         O         Dose Rate(1)           BLNK         \$1707100047         O         Digest Factor(1)           LCS         \$1707100048         O         Dose Rate(1)           LCS         \$1707100048         O         Digest Factor(1)           SAMPLE         \$17R000148         O         Dose Rate(1)	BLNK         \$1707100047         O         Dose Rate(1)         LIQUID           BLNK         \$1707100047         O         Digest Factor(1)         LIQUID           LCS         \$1707100048         O         Dose Rate(1)         LIQUID           LCS         \$1707100048         O         Digest Factor(1)         LIQUID           SAMPLE         \$17R000148         O         Dose Rate(1)         LIQUID	BLNK         \$1707100047         O         Dose Rate(1)         LIQUID           BLNK         \$1707100047         O         Digest Factor(1)         LIQUID           LCS         \$1707100048         O         Dose Rate(1)         LIQUID           LCS         \$1707100048         O         Digest Factor(1)         LIQUID           SAMPLE         \$17R000148         O         Dose Rate(1)         LIQUID	BLNK         \$1707100047         O         Dose Rate(1)         LIQUID           BLNK         \$1707100047         O         Digest Factor(1)         LIQUID         0.001000           LCS         \$1707100048         O         Dose Rate(1)         LIQUID         0.000100           LCS         \$1707100048         O         Digest Factor(1)         LIQUID         0.000100           SAMPLE         \$17R000148         O         Dose Rate(1)         LIQUID	BLNK         \$1707100047         O         Dose Rate(1)         LIQUID           BLNK         \$1707100047         O         Digest Factor(1)         LIQUID         0.001000           LCS         \$1707100048         O         Dose Rate(1)         LIQUID         0.000100           LCS         \$1707100048         O         Digest Factor(1)         LIQUID         0.000100           SAMPLE         \$17R000148         O         Dose Rate(1)         LIQUID	BLNK         \$1707100047         O         Dose Rate(1)         LIQUID           BLNK         \$1707100047         O         Digest Factor(1)         LIQUID         0.001000           LCS         \$1707100048         O         Dose Rate(1)         LIQUID         0.000100           LCS         \$1707100048         O         Digest Factor(1)         LIQUID         0.000100           SAMPLE         \$17R000148         O         Dose Rate(1)         LIQUID         LIQUID	Sample#	Sample#	Sample#	Sample#

#### **Comments Section:**

Data Flagger Status:

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Reviewer Signature

10(5(17 Date

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# LABCORE Completed Batch Report for Batch# 00074945

Seq	Sample#	QC Type	Assoc. Sample	Sample Group	Method	Customer Id	Specification	Hold Time/Expiration Date
3	S17R000148	SAMPLE		20171600	ORGANIC DILU	ΓΙΟΝ 4AW-17-02	241-AW-104 Grab Sampling 2017-06	08/22/2017 14:15

RPP-RPT-60441, Rev. 0

#### APPENDIX C

# REVIEW OF GAS CHROMATOGRAPHY - MASS SPECTROMETRY FILE FOR TENTATIVELY IDENTIFIED COMPOUNDS

(3 pages, including cover page)

# TIC Evaluation of 170713\_AW104, file 7

The following Tentatively Identified Compounds (TICs) were evaluated using the FreeStyle program provided with the TraceFinder software. The TIC Summary report for file 7 was used to identify which peaks to evaluate. Below is a summary of the evaluation.

Retention Time (RT): 2.83 ID: Acetic Acid Evaluation: Good match, could also be ammonium acetate (spectra virtually the same).

RT: 4.65 ID: Butanoic Acid Evaluation: Good match

RT: 5.34 ID: 3-Methyl-3-nitrobut-1-ene Evaluation: ID possible match

RT: 5.50 ID: 3-Methyl-3-nitrobut-1-ene Evaluation: ID possible match, but same spectra as compound at RT 5.34. Spectra isn't unique enough to positively identify. Hydrocarbon.

RT: 5.57 ID: Phosgene oxime Evaluation: ID a good spectral match. Likely an artifact created during the extraction due to the matrix.

RT: 5.89 ID: Cyclopentane, bromo Evaluation: ID possible match, but same spectra as compound at RT 5.34 and 5.50. Spectra isn't unique enough to positively identify. Hydrocarbon. Detector may be saturated.

RT: 6.07 ID: 3-Methyl-3-nitrobut-1-ene Evaluation: ID possible match, but same spectra as compound at RT 5.34. Spectra isn't unique enough to positively identify. Hydrocarbon.

RT: 6.12 ID: 3-Methyl-3-nitrobut-1-ene Evaluation: ID possible match, but same spectra as compound at RT 5.34. Spectra isn't unique enough to positively identify. Hydrocarbon.

RT: 6.33 ID: 2-Propenoic acid, 2-methyl-, 1-methylbutyl ester Evaluation: ID not correct – missing ions. HC of some sort.

RT: 6.38 ID: 3-Methyl-3-nitrobut-1-ene Evaluation: ID possible match, similar spectra as compound at RT 5.34, but a couple of extra ions present. Spectra isn't unique enough to positively identify. Hydrocarbon.

RT: 6.54 ID: 1-Pentene, 5-nitro Evaluation: Missing ion 115. Unknown hydrocarbon.

RT: 6.79 ID: 3-Methyl-3-nitrobut-1-ene Evaluation: ID possible match, similar spectra as compound at RT 5.34, but a couple of extra ions present. Spectra isn't unique enough to positively identify. Hydrocarbon.

RT: 7.26 ID: 4-Penten-2-ol Evaluation: No - missing ions - HC of some sort.

RT: 7.40 ID: 2-Hydroxy-3-pentanone Evaluation: No - missing ions. HC of some sort.

RT: 7.98 ID: Heptane, 2-Bromo- Evaluation: No - missing ions. HC of some sort.

- RT: 8.41 ID: 1-Propene, 3-propoxy- Evaluation: No missing ions. HC of some sort.
- RT: 8.52 ID: Heptane, 4-methyl- Evaluation: No missing ions. HC of some sort.
- RT: 8.63 ID: 1-Propene, 3-propoxy- Evaluation: No missing ions. HC of some sort.
- RT: 8.98 ID: 1-Propene, 3-propoxy- Evaluation: No missing ions. Similar spectra to compound at RT 8.41, similar HC of some sort.
- RT: 9.10 ID: Hydrazinecarboximidamine, nitrate Evaluation: No missing ions. Some sort of HC.
- RT: 10.23 ID: Dodecane Evaluation: definitely an alkane, more likely Tridecane due to MW ion 184 presence.
- RT: 11.16 ID: Tetradecane Evaluation: ID is likely correct. Definitely an alkane and the MW ion is present.

Calculation

#### RPP-RPT-60441, Rev 0

#### **Calculation Checklist**

	culation e/Subje			Workbook associated with RPP-RPT-60441, Investigation of AW-104	Compo	site Floating Layer
Sco	pe of R	eview	<b>'</b> :	Entire workbook.		
Eng	ineer/A	malys	t	Huei Meznarich Munig Maryanis	Date:	1/18/18
Org	anizati	on Ma	nager	Gary Cooke Vaus A. Cole	Date:	1/18/18
Yes	No*	NA*	k			
[x]	[]	[]	a.	The objective/purpose of the calculation is clearly stated and the problem purpose statement.	is compl	etely defined by the
[x] [x]	[]	[]	b. c.	Analytical and technical approaches and results are reasonable and approp Input data are adequately described, referenced to their source, and checke		nsistency with
[x]	[]	[]	d.	original source information.  Necessary assumptions are reasonable, explicitly stated, and supported. A	ssumptic	ons requiring
[x]	[]	[]	e.	verification prior to use are clearly stated and identified/tracked using TBE For both qualitative and quantitative data, uncertainties are recognized and		
[x]	[]	[]	f.	presented in a manner to minimize design interpretations.  Mathematical derivations were checked, including dimensional consistency		
[ x ]	[]	[]	g.	Calculations are sufficiently detailed such that a technically qualified personally analysis without requiring outside information.		
[]	[]	[x]	h.	Hand and MathCAD® calculations were verified, including review that co formulae correctly interpret intended expressions, correct units are used, at appropriate.		
				No Hand or MathCAD calculations were used.		
[]	[]	[ x ]	i.	Software applications used are identified by the program name and version by Attachment A (Section 7, Use of Computer Software).	/release	number as required
				The only software used was the Hanford site standard of Microsoft Ex	cel.	
[ x ]	[]	[]	j.	Software input data is identified and/or attached/included, the input data is the calculation document.	correct,	and consistent with
[]	[]	[x]	k.	Software output is consistent with the input and with the results reported in	the calc	ulation document.
				These workbooks are not software.		
[]	[]	[x]	1.	Software verification and validation are addressed adequately in accordance TFC-BSM-IRM_HS-C-01. Software verification documentation, (typically Plan or Test Report), is included in the calculation document, or a reference Attachment A (Section 7, Use of Computer Software).	y the Sof	
				These workbooks are not software.		
[ x ]	[]	[]		Software verification show that software produces correct solution for the within defined limits for each parameter employed and the software is used		
[ x ]	[]	[]		The encoded mathematical model (method), produces a valid solution to the associated with the particular application (i.e., the methodology used is appropriated).		
[]	[]	[x]	0.	Multiple-Use spreadsheets used in the calculation are identified, verified, as accordance with TFC-ENG-DESIGN-C-32. Reference to the correspondin Plan or Spreadsheet Verification Form (for legacy spreadsheets) is included by Attachment A (Section 7, Use of Computer Software).	g Softwa	are Management
				These workbooks are single use spreadsheets.		
[ x ]	[]	[]		Single-Use spreadsheets used in the calculation are identified, verified, and calculation or other technical document as prescribed in TFC-ENG-DESIG		nted as part of the
	[]	[]	q.	Data or results presented in tables and graphs have been checked against or		urce.
	[]	[]	S.	The number of significant digits is appropriate and consistent. Limits/criteria/guidelines applied to the analysis results are appropriate and Limits/criteria/guidelines were checked against references.	referenc	ed.
[ x ]	[]	[]		Conclusions are consistent with analytical results and applicable limits.		

<sup>\*</sup> If less than the entire calculation was checked, the scope of the check should be discussed. If any blocks are checked "No" or "NA", an explanation must be provided here or attached.

[x] [x]	[]	[]	u. v.	Results and conclusions address all points in the purpose. Referenced documents are retrievable or otherwise availa reference is cited.	
[x]	[]	[]	w.	The calculation was prepared in accordance with Attachn Instructions," of TFC-ENG-DESIGN-C-10.	nent A, "Calculation Format and Preparation
[x]	[]	[]	x.	Impacts on requirements have been assessed and change of revisions to affected documents, as appropriate.	documentation initiated to incorporate
[x]	[]	[ ]	у.	All checker comments have been dispositioned.	
[x]	[ ]	[ ]	Z.	The design media matches the calculations.	
				Doug Kraft	1/18/18
			_	Checker (printed name and signature)	Date

<sup>\*</sup> If less than the entire calculation was checked, the scope of the check should be discussed. If any blocks are checked "No" or "NA", an explanation must be provided here or attached.