



# Creating Robust and Reversible Cell-Gel Networks using Boronic Acid Chemistry

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## Abstract

Current biosensor technologies typically use platforms mimicking intrinsic cell detection and response pathways (i.e., antibody-based detection). These approaches require signal amplification steps, resulting in decreased discrimination between pathogenic and non-pathogenic species. In contrast, cells have evolved highly efficient discriminatory mechanisms for detecting small quantities of pathogenicity factors and evoking biological responses. Due to the inability to retain the high-level sensitivity and selectivity of these mechanisms in *ex vivo* systems, the development of cell-based biosensors (CBB) is becoming increasingly explored, but often lack longevity and stability. The model organism, *Saccharomyces cerevisiae*, offers advantages for use in CBBs due to its robustness, controllable growth cycle, and ability to be lyophilized and stored. Yeast encapsulation strategies have been explored to increase stability and longevity; encapsulation, however, is often lengthy and requires several chemical additives. We present a simple approach to directly and rapidly form reversible cell-gel networks using a synthetic peptide displaying boronic acid (BA) residues. BAs covalently bind to most saccharides, in which binding is favored at high pH and reversible by lowering the pH. We hypothesized that the BA peptide would form cell-gel networks by binding to the various polysaccharides and glycoproteins decorating the cell wall. Here, we present that through binding of the BA groups to cell surfaces, yeast cells indeed act as BA crosslinkers, creating an infinite network of chemically-linked, viable cells. Furthermore, the gelation process is versatile with other organisms and fully reversible. Overall, this unique approach to forming viable cell-gel networks provides a promising path forward toward the development of useful CBBs.

## Introduction to Boronic Acids (BAs)

BAs are Lewis acids with  $pK_a$  values tunable across physiological pH

$$R-\overset{\text{OH}}{\underset{\text{OH}}{\text{B}}} \rightleftharpoons R-\overset{\text{OH}}{\underset{\text{OH}}{\text{B}}} \text{ (dissociated form)}$$

Phenyl BAs (PBAs) are useful due to their stability and synthetic versatility

$pK_a$  can be determined through UV-vis measurements

Substituent	$pK_a$
None	8.8
4-Br	8.3
3-Cl-4-F	7.8
2-F-5-F	7.6
3-F-4-F-5-F	6.8
2-F-5-NO <sub>2</sub>	6.0

Binding affinity (K) is a complex function of:

1. BA ( $pK_a$ )
2. Saccharide
3. pH

Saccharide	K
fructose	160
sorbose	120
mannose	13
glucose	4.6
lactose	1.6
sucrose	0.67

pH	K
6.5	0.84
7.0	2.0
7.4	4.6
8.0	7.2
8.5	11

## Introduction to Gelation and Networks

Gels are comprised of infinite 3D chemically- or physically-crosslinked networks

- Chemically-crosslinked networks can form in mixtures of binding species
- An infinite 3D network requires an average of > 2 binding sites per molecule

Simple gel test – Inverted vial

- Shown is a 1% solution of peptide in H<sub>2</sub>O
- The peptide forms a physically-crosslinked network of nanofibers upon addition of salt

Swelling Equilibrium

$$\ln(1 - \phi_e) + \phi_e + \chi \phi_e^2 = \frac{M_s}{M_x} \left( \frac{\phi_e}{2} - \phi_e^{1/3} \right)$$

$\phi_e$  ≡ volume fraction solvent (e.g., H<sub>2</sub>O)

$\chi$  ≡ interaction parameter (solvent/network compatibility)

$M_s$  ≡ molar mass of solvent

$M_x$  ≡ molecular weight between crosslink points (cell size)

Rheological Measurements

- Shown are small amplitude oscillatory shear of peptide gels with variable gelator concentration

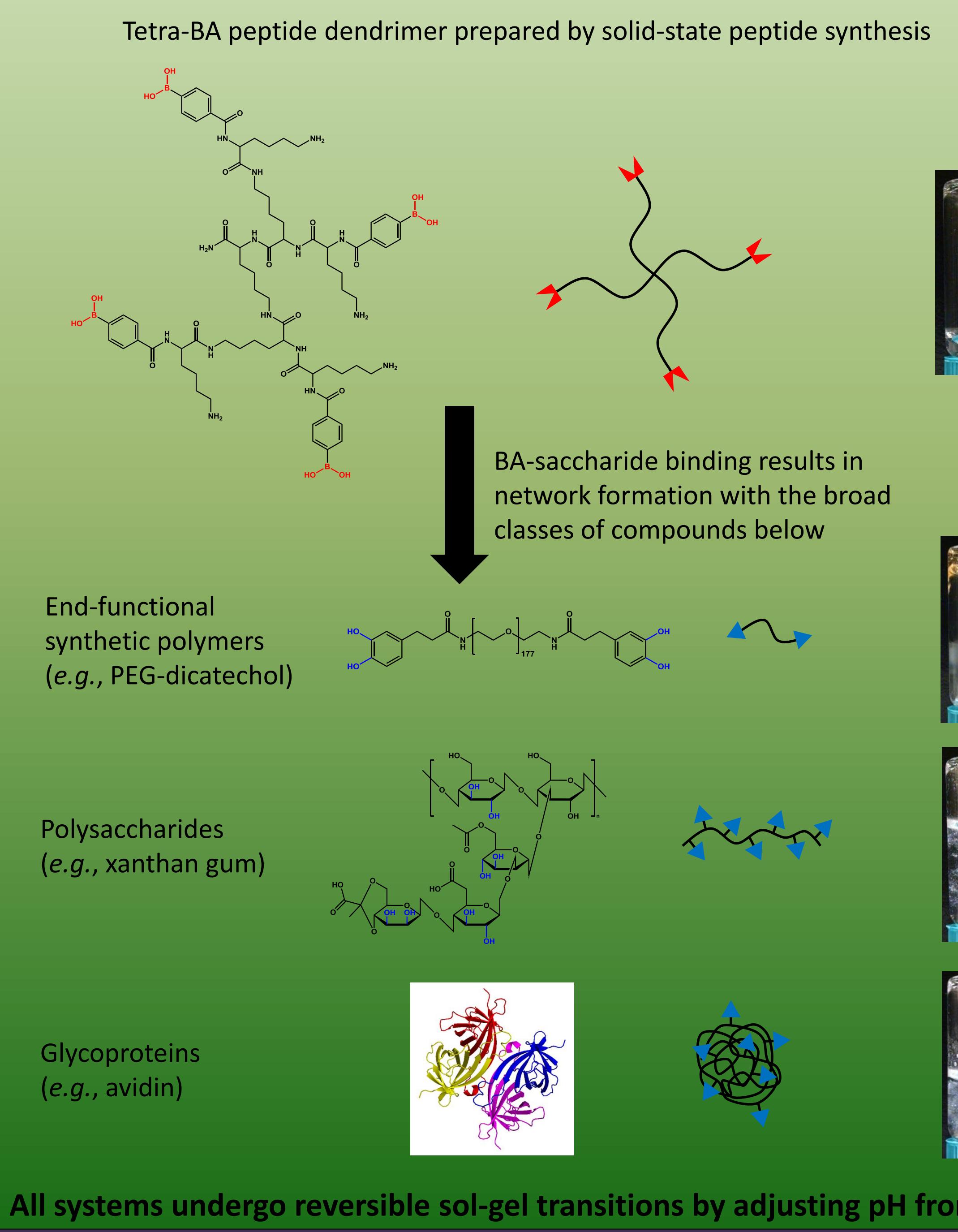
Jones, B.H., et al. *Soft Matter* 2015, 11, 3572-3580.

Jones, B.H., et al. *Chem. Comm.* 2015, DOI: 10.1039/CSCC0207F.

The amount of solvent a gel can imbibe is increased by increasing cell size

## BA-Based Gels

We have established chemistry for the preparation of multi-functional BA peptides and polymers, and have used these compounds to create reversible hydrogels from saccharide-containing (and related) entities



## Making Live-Cell Gels

The cell wall is decorated with saccharide moieties

Shown below is a diagram of a yeast cell wall

Boronic acid peptides cause cell gelation

*S. cerevisiae* can be gelled and are viable

Boronic acid binds to the cells

Boronic acid binding can be confirmed using UV-Vis

Building a boronic acid library

Synthesize different boronic acids to be compatible with different types of cells (e.g. mammalian cells)

## Conclusions/Next Steps

- Boronic acid chemistry offers a novel method for directly linking cells together in a reversible manner
- Cells remain viable and metabolically active when gelled using boronic acid peptides
- Build a library of boronic acid compounds to gel mammalian cells together
- Test compatibility of boronic acids with different cell types
- Confirm chemical binding of boronic acid compounds to cells

## Implications

- Gelling live cells can be useful for building cell-based biosensors
- Cell-gel networks can be used to study *in vitro* wound-healing