Understanding the Reliability of Solder Joints Used in Advanced Structural and Electronics Applications

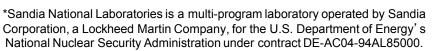
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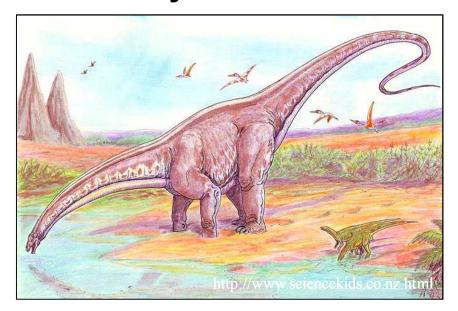






Thank You!

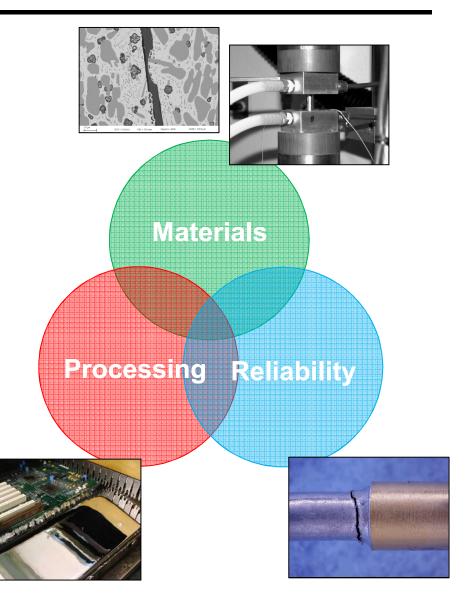
- I would like to extend my sincere gratitude to the American Welding Society for the honor of presenting the 2016 Comfort A. Adams Lecture to you, today.
- Looking back in history ...



This is the first time that the Comfort A. Adams lecture has addressed soldering technology.

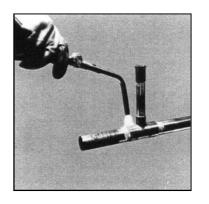
Outline

- Introduction
- Materials
 - Solderability
 - Mechanical performance
- Processing
- Reliability
 - Interface reactions
 - Solder joint fatigue
 - Empirical studies
 - Computational modeling
- Summary



Introduction

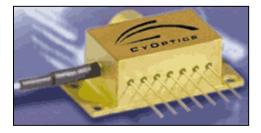
 Soldering technology has two general categories: structural applications and electronic applications.





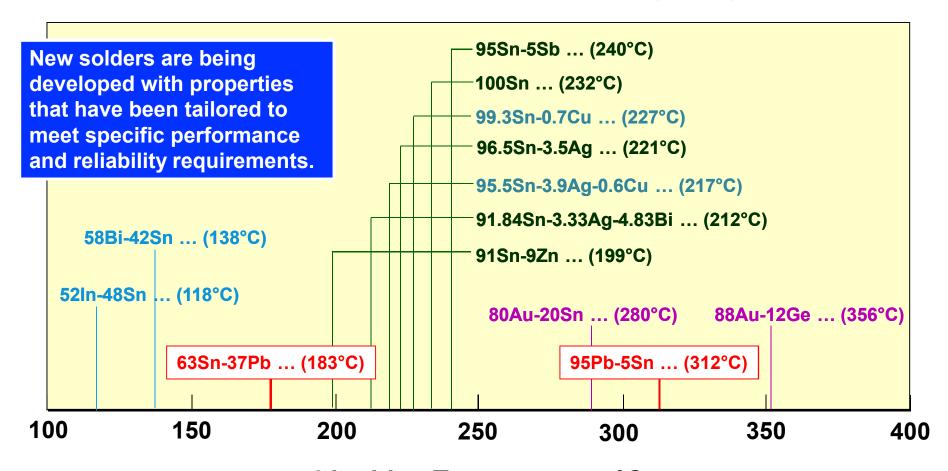




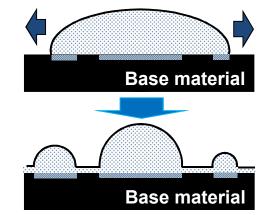


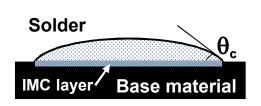
Electronic applications

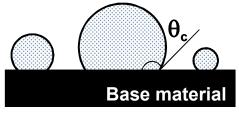
A solder filler metal is defined as a metal alloy having a liquidus temperature that is less than 450°C (842°F).

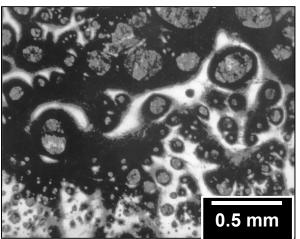


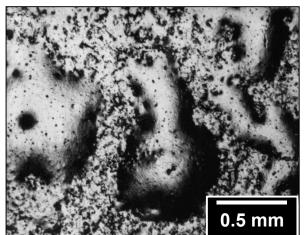
- Solderability refers to the capability of the molten filler metal to wet-and-spread on the base material surface.
 - Wetting: form a metallurgical bond
 - There are three solderability conditions:











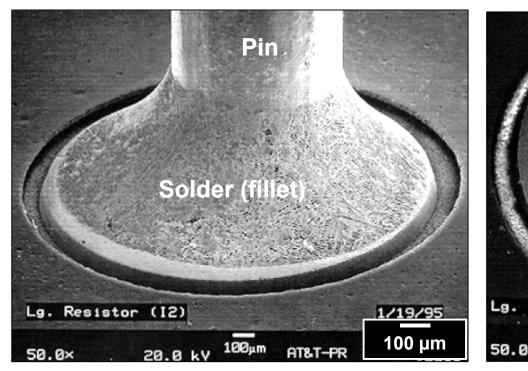
0.5 mm

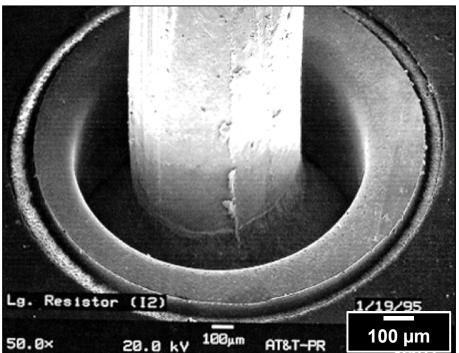
Wetting

Non-wetting

Dewetting

Solderability controls the final joint geometry (fillet size, voids, etc.), which in turn, affects long-term reliability.

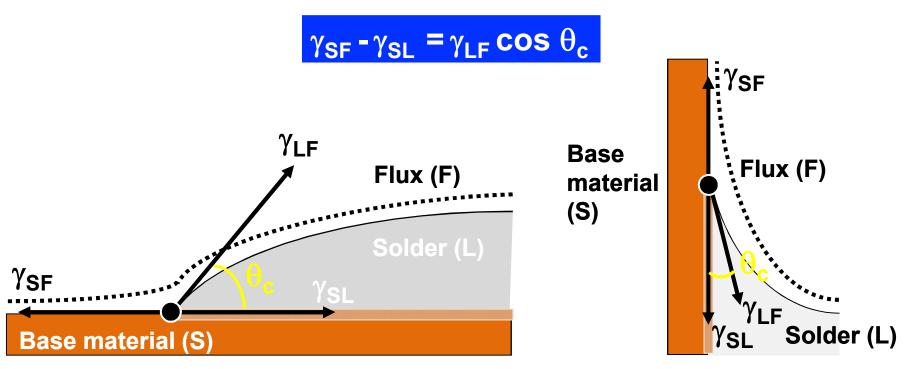




58Bi-42 Sn solder

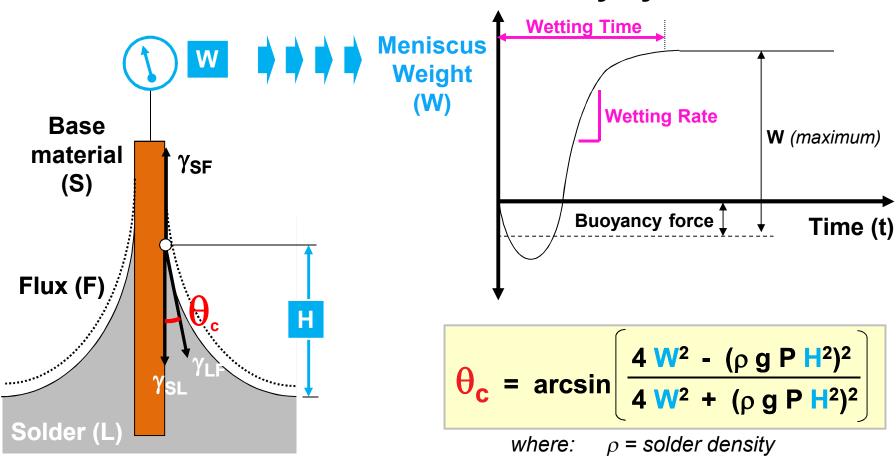
 In this instance, the flux was not properly engineered for such a low soldering process temperature – 160°C.

 Solderability is measured by the contact angle, θ_c, which at equilibrium, is calculated from Young's equation:



- Solderability is optimized by *minimizing* θ_c :
 - Minimize γ_{LF}
 - Maximize γ_{SF} γ_{SL}
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• Meniscus height-wetting balance test assesses θ_c for the base material / flux / solder alloy system.



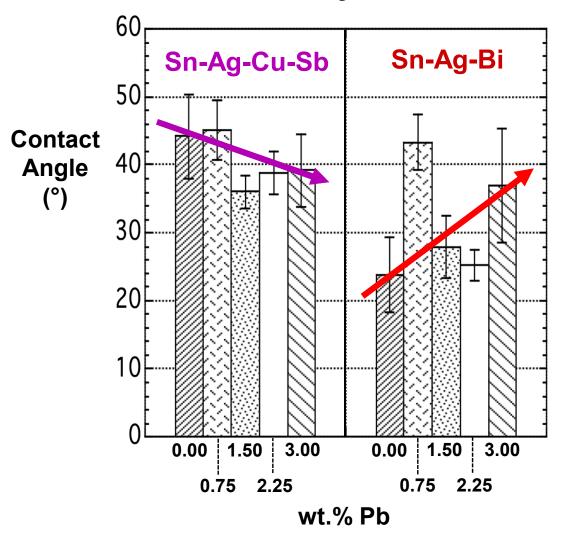
g = accn. due to gravity
P = sample perimeter

• Typical θ_c and γ_{LF} values are listed below:

Solder (wt.%)	Temp. (°C)	γ _{LF} (dyne/cm)	θ _c (°)
95.5Sn-3.9Ag-0.6Cu	260	497±16	40±1.0
	245	444±17	39±1.0
	230	485±27	42±1.4
96.5Sn-3.5Ag	260	460±30	36±3
60Sn-40Pb	260	380±10	17±4
58Bi-42Sn	215	310±50	37±7
Substrate: OFHC Cu; Flux: RMA ,1:1 IPOH			

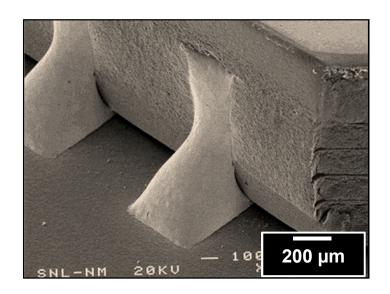
- Bi and Pb additions lower θ_c by reducing γ_{LF} .
- Temperature has a limited effect on either θ_c or γ_{LF} .

Rules-of-thumb are not always hard-and-fast ...



Pb additions do not always reduce the contact angle

◆ At room temperature, solder alloys are performing at temperatures that are equivalent to conditions in the combustion chamber of an operating jet engine !!!





It's HOT ... even when it's not ...

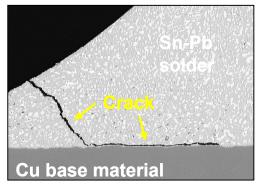
The solder joint is a "system."

Base materials Base material #1 Solder (filler metal) **Interfaces (reaction layers) Base material #1** Base material #2 20 µm Solder joint EHT = 20.00 kV Signal A = BSD **Base material #2** 200 µm EHT = 20.00 kV Signal A = BSD $WD = 9.4 \, \text{mm}$

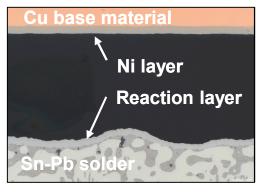
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- The base materials are stronger than the filler metals.
 - Solder joint failures occur in the filler metal or at an interface.

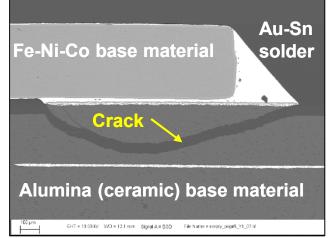


Solder (bulk) failure



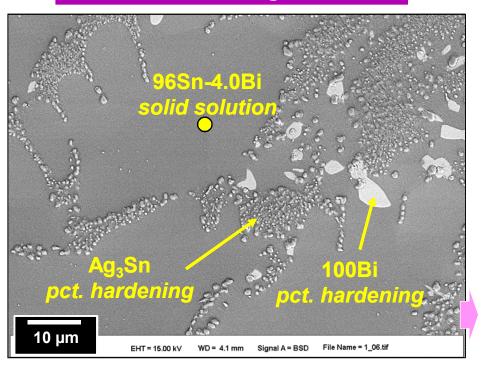
Interface failure

 The exception occurs under the combination of large loads, a high-strength solder, and a ceramic base material.

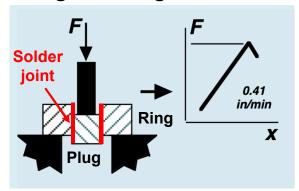


- Solders are engineered using the same strengthening mechanisms that are found in high-performance alloys:
 - Precipitation hardening
 - Solution strengthening

91.84Sn - 3.33Ag - 4.83Bi



Ring-and-Plug Shear Test



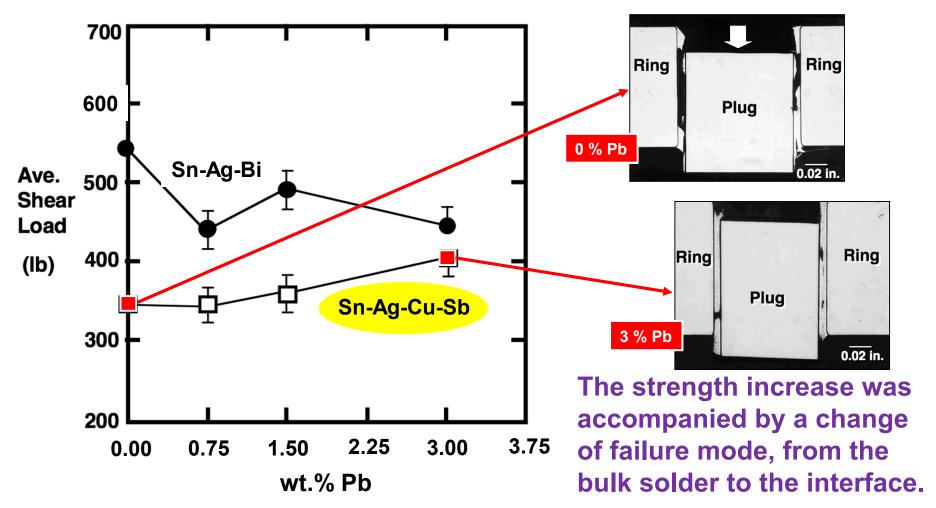
 Solder
 Stress (psi)

 Sn-Pb
 5840

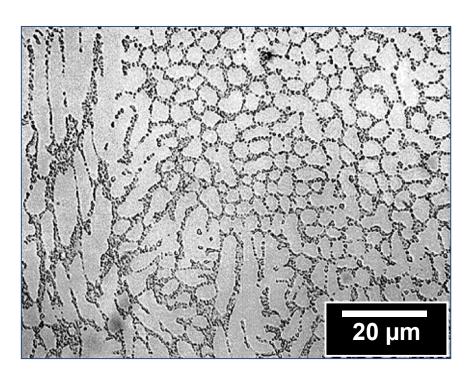
 Sn-Ag
 7970

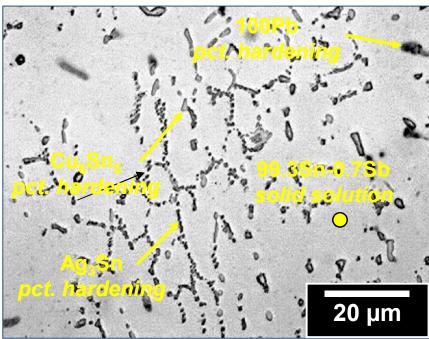
 Sn-Ag-Bi
 11800

- Alloy additions can have unexpected effects.
 - Additions of 0 3.75 wt.% Pb were made to two alloys.



Sn-Ag-Cu-Sb: The addition of Pb broke up the Ag₃Sn particle network, thereby enhancing the precipitation hardening effect.

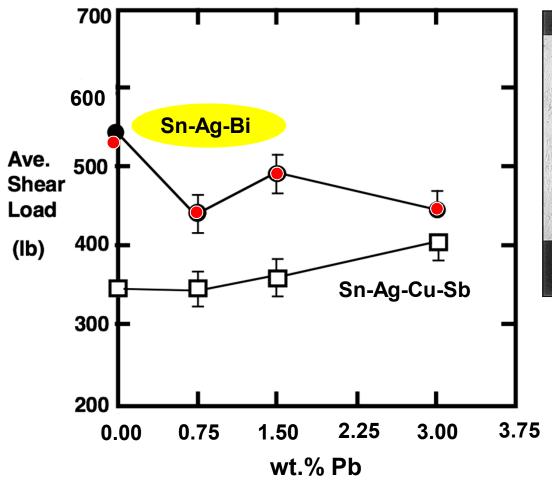


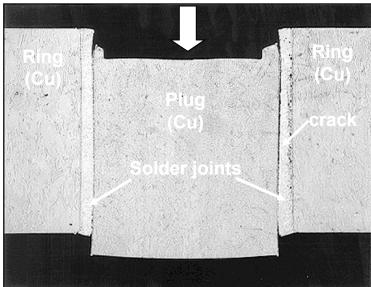


0.00 % Pb

3.00 % Pb

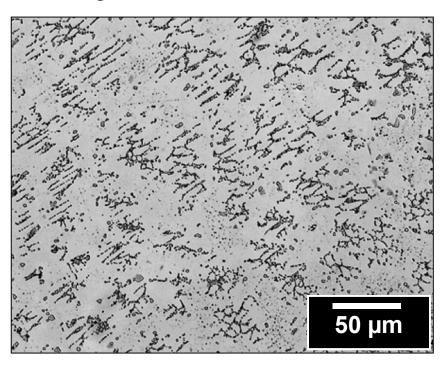
 Sn-Ag-Bi: The fracture path remained along the solder/base material interface for all Pb additions.

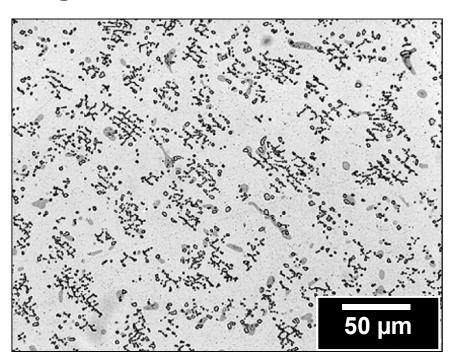




The nominal strength remained greater than the ≈ 400 lb. level above which failure occurs typically at the interface.

 Sn-Ag-Bi: The Pb additions did not affect the alloy microstructure on the larger size scale.



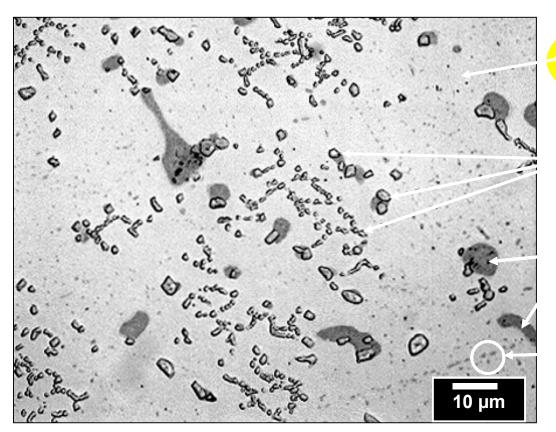


0.00 % Pb

3.00 % Pb

But ...

- ... The Pb additions altered the small-scale phase distributions within the Sn-Ag-Bi microstructure.
 - The Pb scavenged Bi from the solid-solution phase, thus reducing the solid-solution strengthening effect.



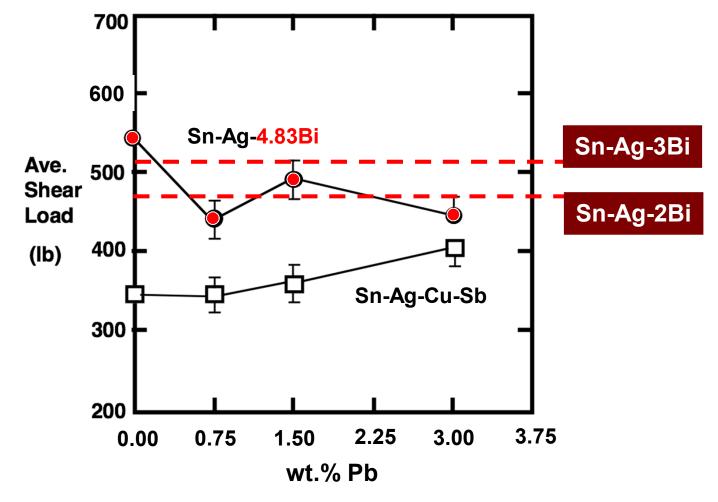
96.5Sn-3.5Bi Solid solution

Ag₃Sn

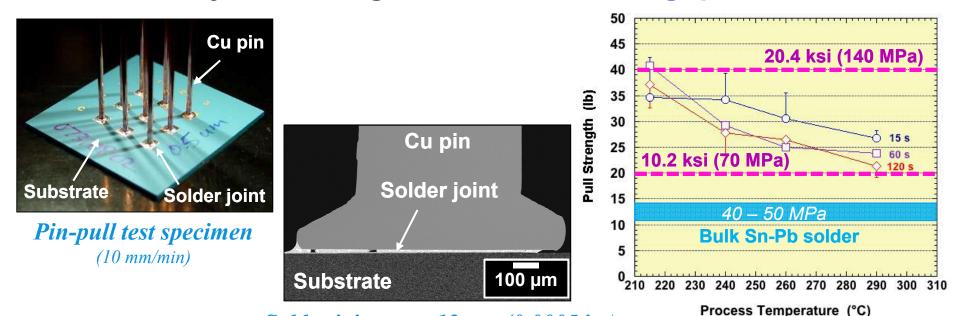
64Pb-33Bi-3Sn

63Sn-24Pb-13Bi

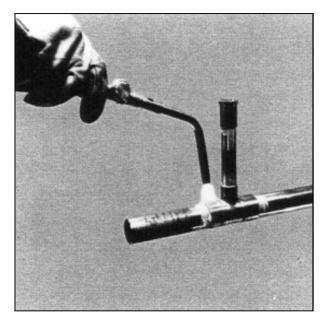
Ring-and-plug shear tests performed on Sn-Ag-XBi
 (X = 2, 3 wt.%) confirmed a similar drop of strength.



 Bulk solder strength is not always the controlling factor of joint strength ... there is also gap thickness.

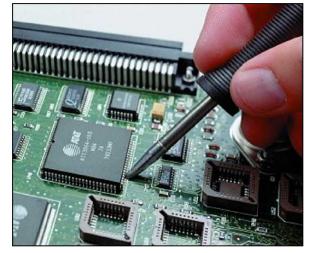


- Solder joint gap: 12 µm (0.0005 in.)
- The joint geometry constrains deformation in the solder.
- Constitutive (computational) modeling takes account of this effect: bulk and joint mechanical properties.



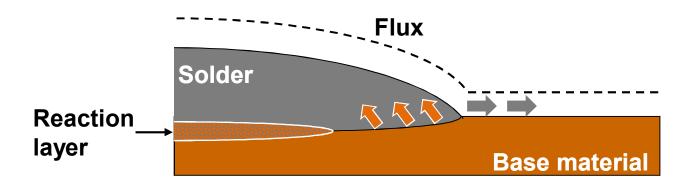






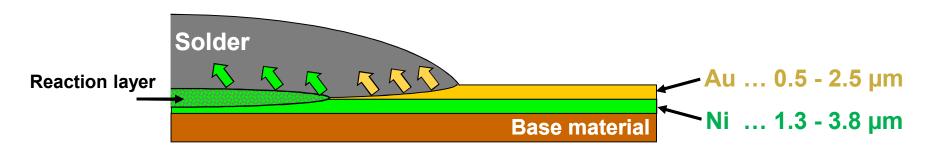
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- Processes are based upon solderability performance.
 - Solderability refers to the capability of the molten filler metal to wet-and-spread on the base material surface.
 - Wetting: formation of a metallurgical bond (reaction layer).
 - Spreading: spontaneous coverage by the molten alloy.

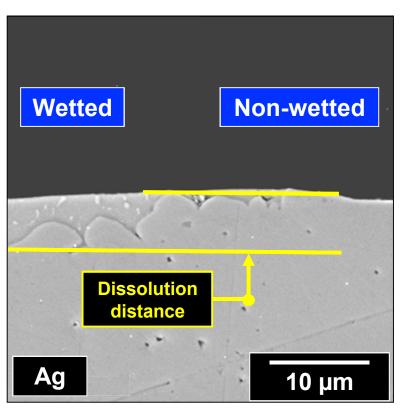


- There are three consequences of the wetting process:
 - Base material dissolution
 - Changing the solder alloy composition
 - Reaction layer formation

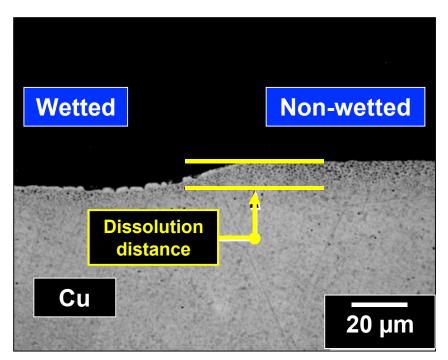
- Surface finishes are used to augment the solderability of a base material.
- The surface finish typically has these layers:
 - Solderable finish: serves as the "stand-in base material" to which is made the metallurgical bond (e.g., Ni)
 - Protective finish: protects the solderability of the solderable finish and is fully dissolved into the molten solder (e.g., Au).



 Base material dissolution results in a loss of structure (or solderable layer) cross section:

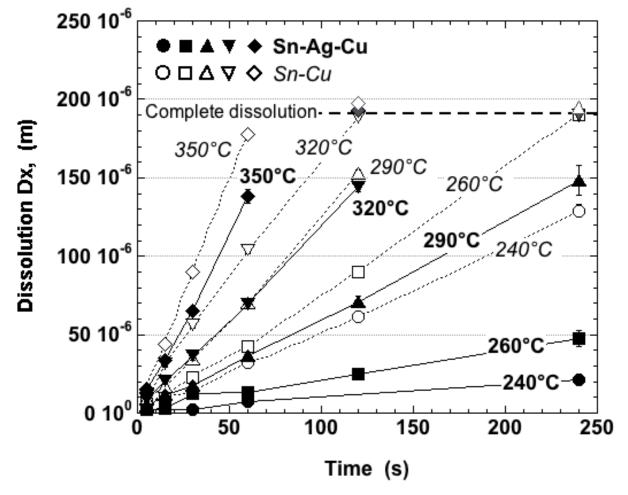


Sn-Ag-Cu, 290°C, 60 s

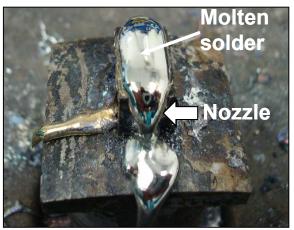


Sn-Pb, 300°C, 60 s

◆ The silver (Ag) in the Sn-Ag-Cu solder slowed the rate of Ag base material dissolution versus Sn-Cu solder.

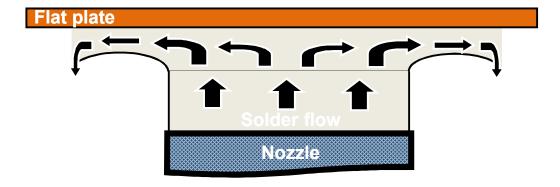


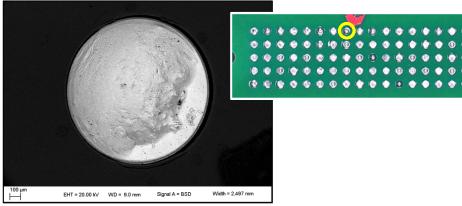
- The soldering process is a fluid mechanics problem.
 - The solder fountain is used to assemble connectors and other large form-factor components to printed circuit boards.



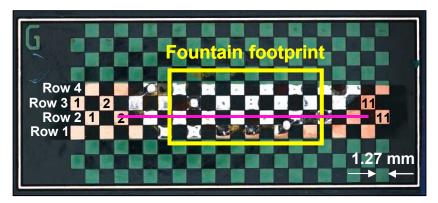
"Solder Fountain"

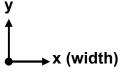
 Skips and other "unexpected" defects were observed in the solder coating.

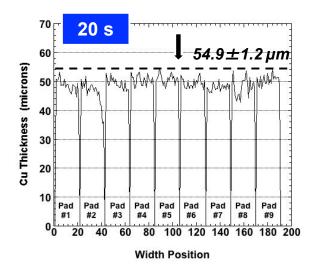


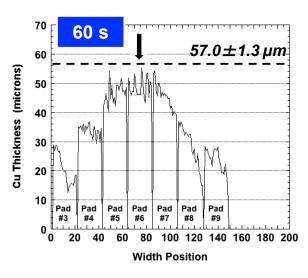


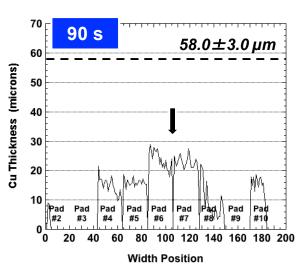
 A circuit board "grid" was used to investigate the spatial (x, y) behavior of an impinging, molten solder fountain.











29

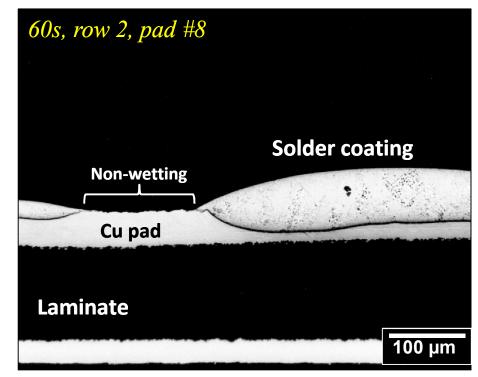
The extent of base material dissolution had two scales:

- "Large size scale" ... ≈ 10 mm
- "Small size scale" ... ≈ 0.5 mm

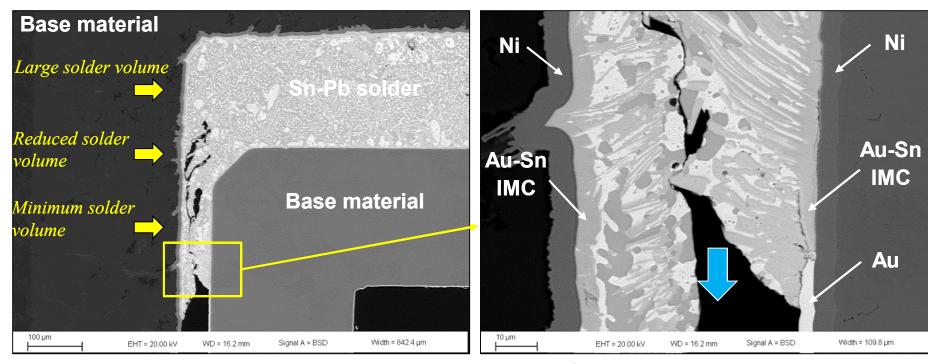
Large size scale

□ Inner Pads 50 Cu Dissolution (microns) 40 30 20 10 90 100 Immersion Time (s)

Small size scale

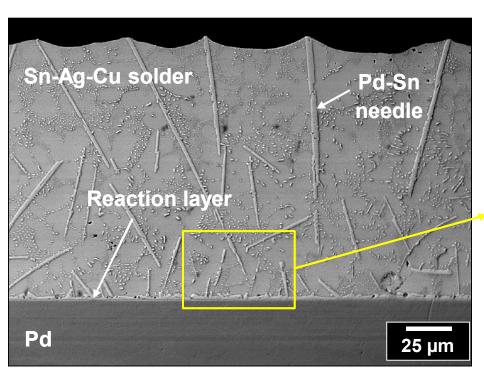


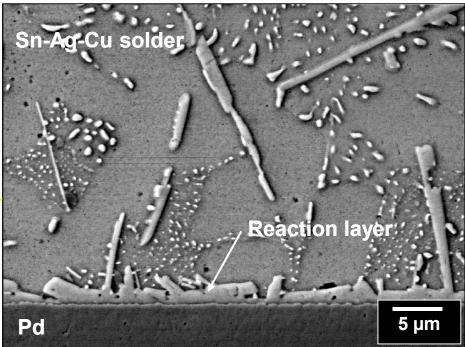
- Base material dissolution can change the solder composition and thus, its melting properties.
 - Increasing the solidus temperature (T_s) and/or liquidus temperature (T_l) can lead to constitutional solidification.



 An excessively thick Au protective finish caused the Sn-Pb solder to solidify prior to filling the gap.

- Base material dissolution can change the solid solder alloy composition:
 - There is a change to the solder's bulk mechanical properties and its interface microstructure.





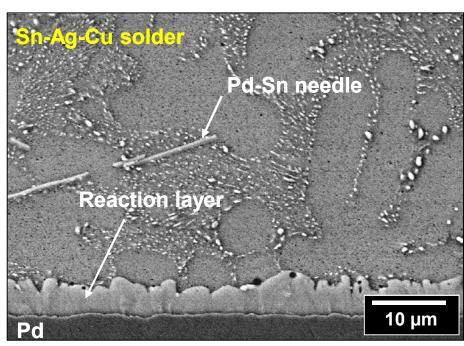
Sn-Ag-Cu, 350°C, 5s

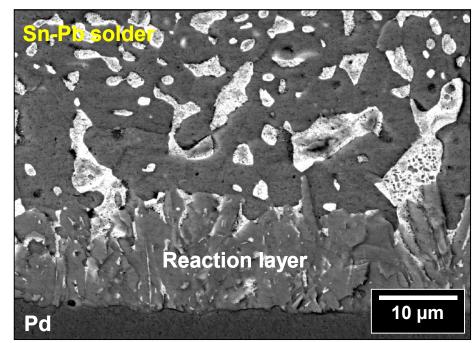
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- Interface reaction layer composition and thickness are a function of the base material and solder alloy.
 - Sn-Ag-Cu solder

Sn-Pb solder

290°C, 5 s.

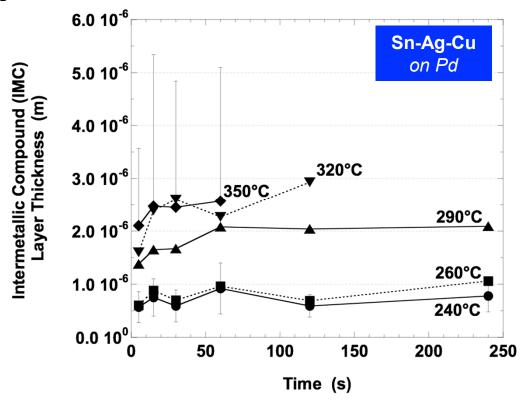




Sn-Ag-Cu: 290°C, 5s

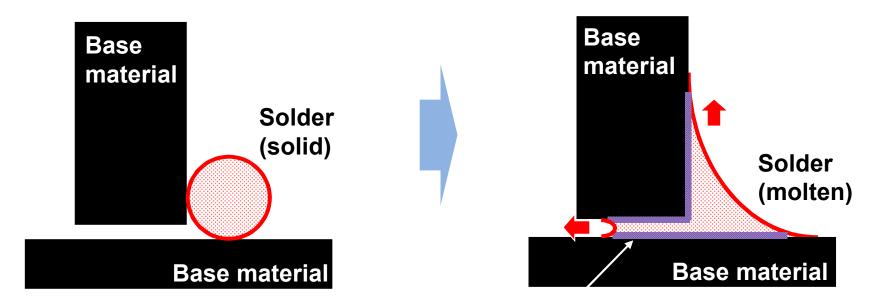
Sn-Pb: 290°C, 5s

 "Liquid" interface reactions: Layer growth tends to quickly level-off as a function of time at temperature.



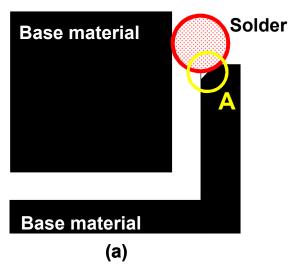
 Reaction layer thicknesses of ≤ 5 μm generally do not pose a significant risk to performance or reliability.

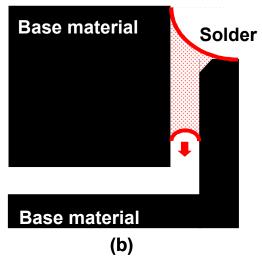
- Solderability is also about spontaneous spreading.
- Spreading is a function of the solder and base material compositions; flux; as well as joint geometry.

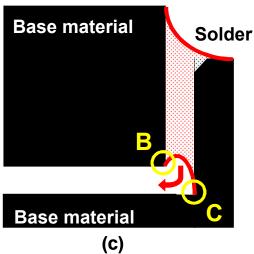


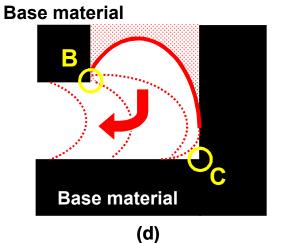
Reaction layer

Molten solders "do not like" to turn corners.



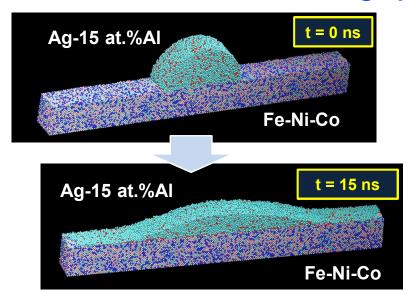


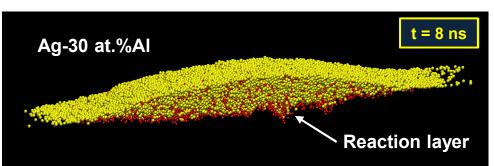


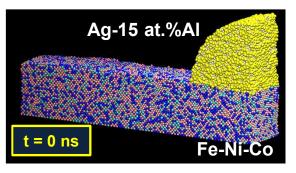


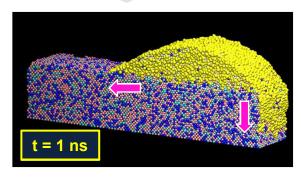
Processing

- Molecular dynamics simulations are being developed that model interface reactions as well as liquid flow.
 - This effort is being spearheaded in brazing technology.





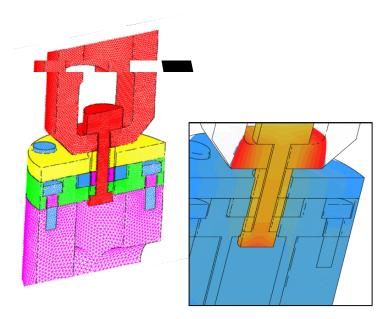


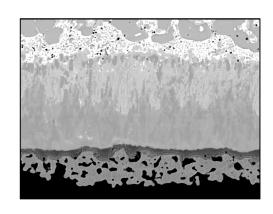


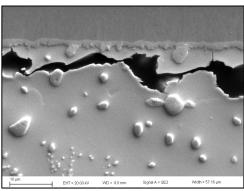
The models can now take into account the effects of geometry.

Reliability

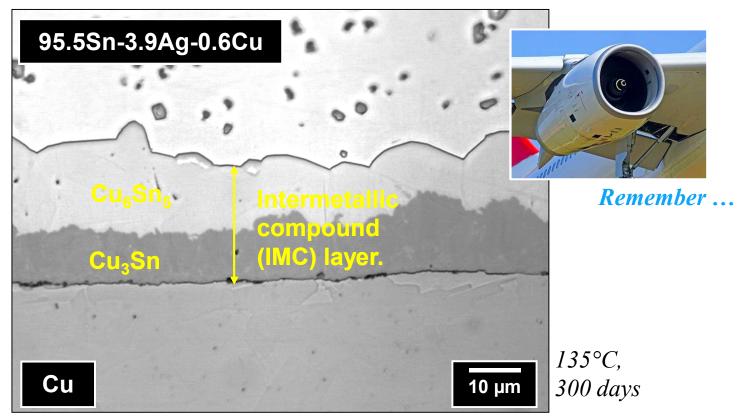
- Interface reactions
- Solder joint fatigue
 - Empirical studies
 - Computational modeling



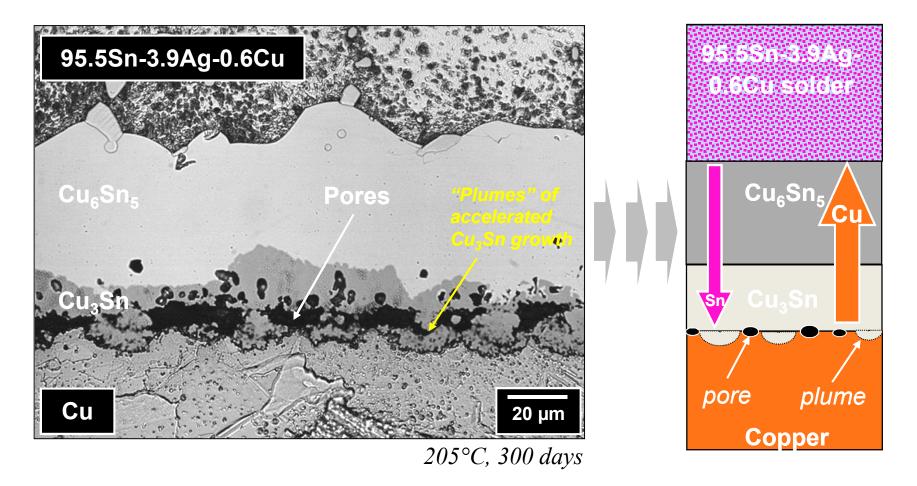




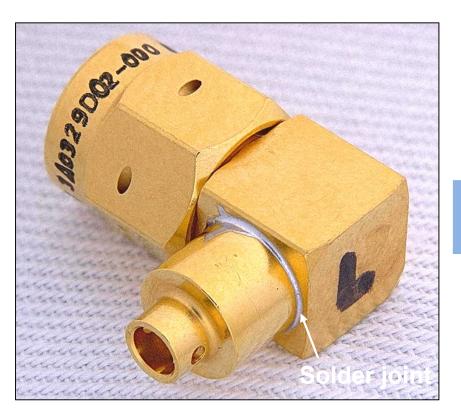
- After solidification, solid-state reactions continue to take place at the solder / base material interface:
 - Interfaces are non-equilibrium structures
 - Solid-state diffusion readily occurs in the solder.

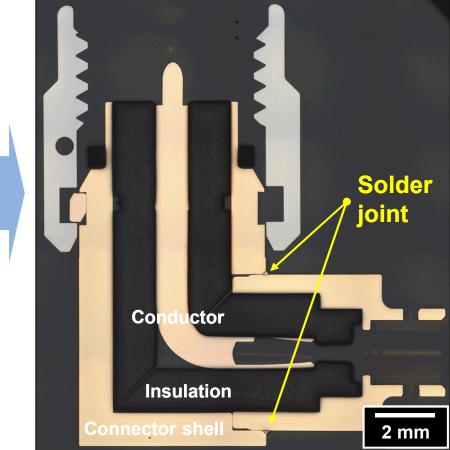


 When those reactions become excessively rapid, the interface microstructure jeopardizes joint reliability.

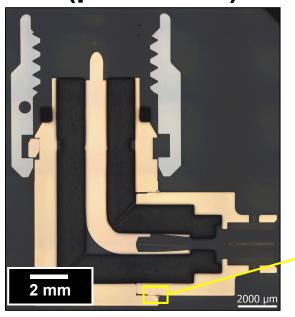


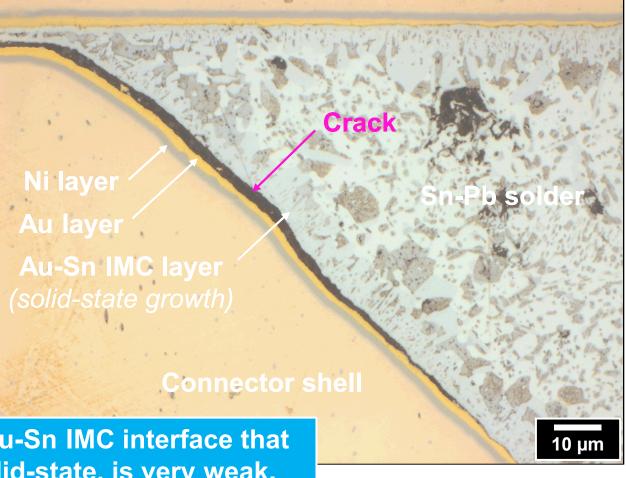
- These layered structures can accelerate failure modes.
 - Example: Sn-Pb solder joint used to assemble a connector





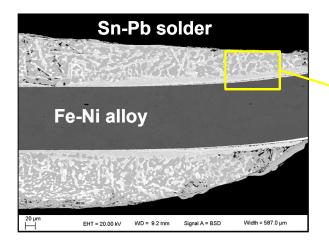
 The soldering process failed to remove the entire Au (protective) finish from the connector shell surface.

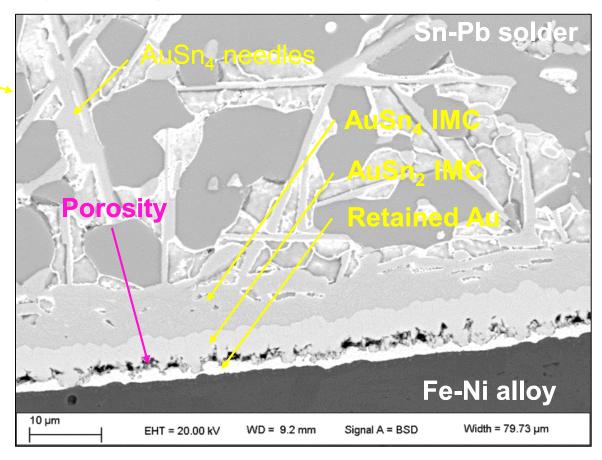




The retained Au / Au-Sn IMC interface that is formed in the solid-state, is very weak.

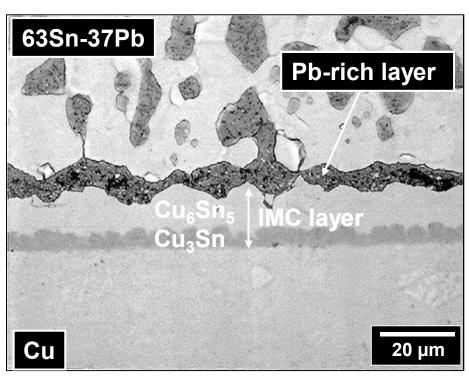
 The weak Au/Au-Sn IMC interface is due to formation of Kirkendall porosity during the solid-state reaction.

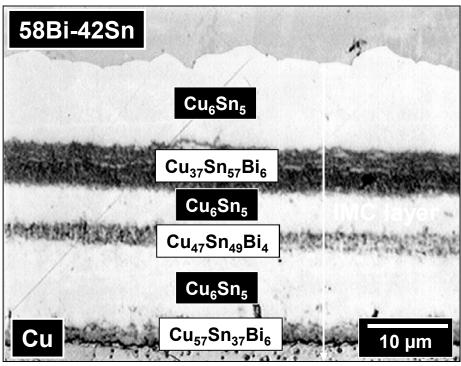




These phenomena take place at room temperature.

- Interface reactions depend upon solder composition.
 - Elements in the solder, which are rejected by the interface reaction, form a layered microstructure.



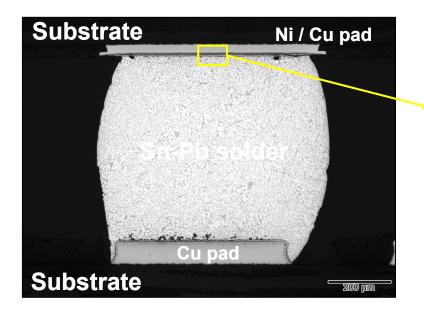


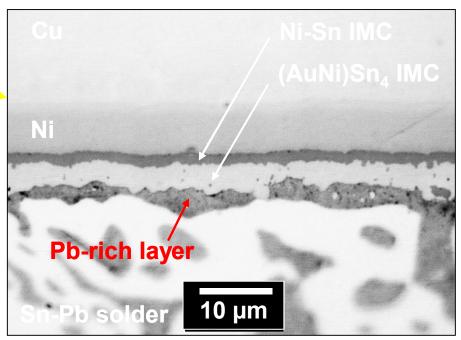
135°C, 40 days

120°C, 40 days

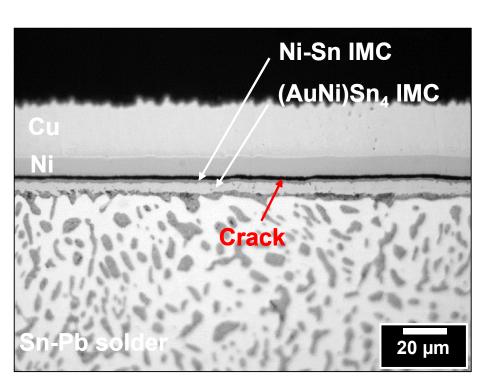
Layered interfaces create paths of easy crack propagation.

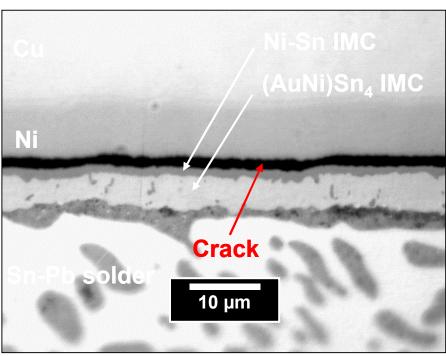
- Combinations of solid-state diffusion/reaction steps can lead to a mechanically weak, interface microstructurse.
- "Return-of-Au" phenomenon:
 - Gold protective finish that has dissolved into solder, returns to the Ni or Cu layer by solid-state diffusion.





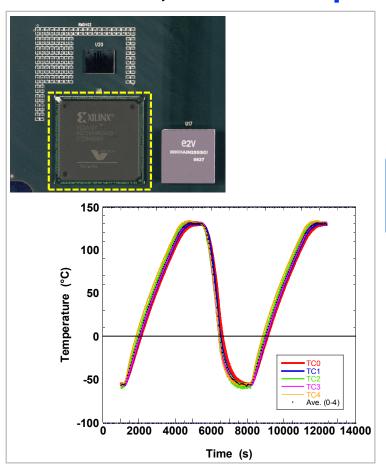
◆ The presence of the additional (AuNi)Sn₄ IMC layer increases the risk of brittle fracture under shock loads.

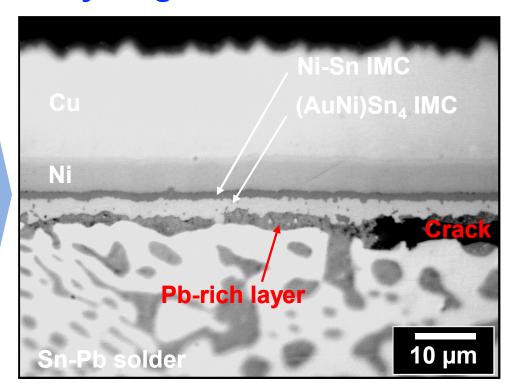




◆ The formation of a (AuNi)Sn₄ IMC layer forced crack propagation to the Ni / Ni-Sn IMC interface.

But then, under temperature cycling environments ...

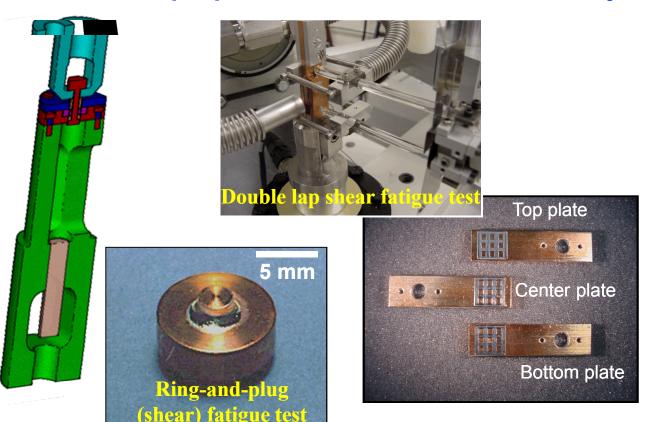




-55°C / 125°C; 15 min holds; 750 cycles

◆ The Pb-rich layer provided an easier path for the propagation of TMF cracks, causing a loss of fatigue life.

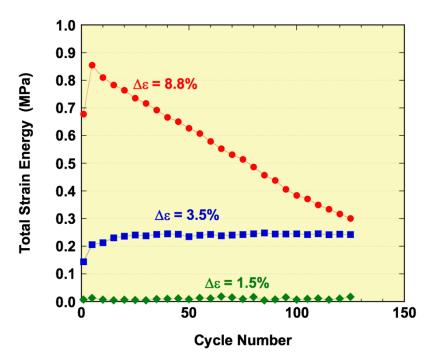
- Empirical methods are based largely upon the testing of applicable solder joint configurations.
 - Recall, there is very little correlation between the bulk solder properties and those of the solder joint structure.

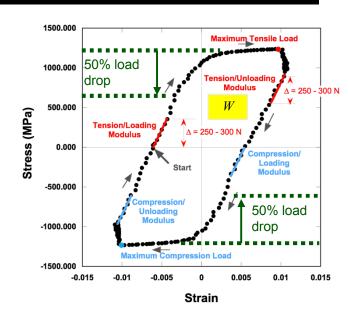


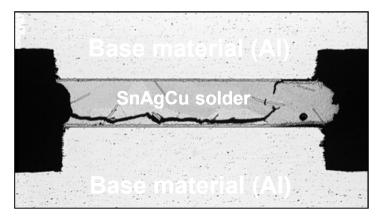




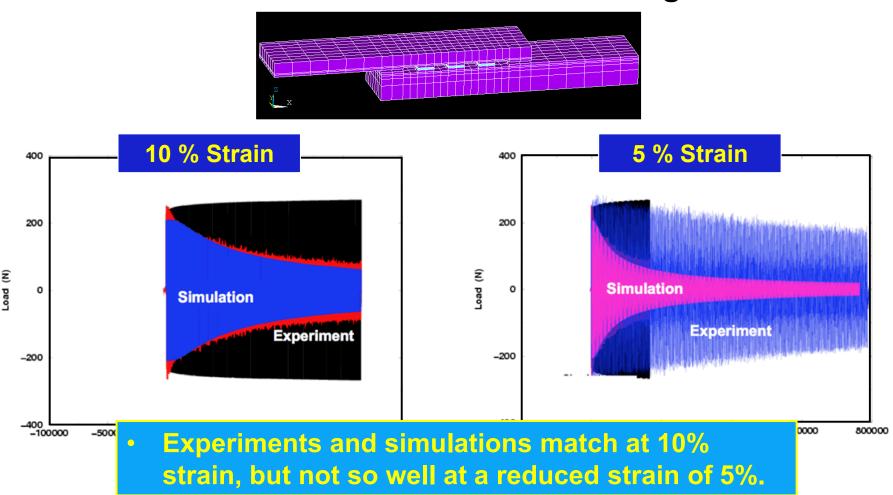
- Isothermal fatigue: The limited work has not addressed several interesting phenomena:
 - e.g., the significant improvement in fatigue life when decreasing from ≈ 9% total strain to ≈ 4%.



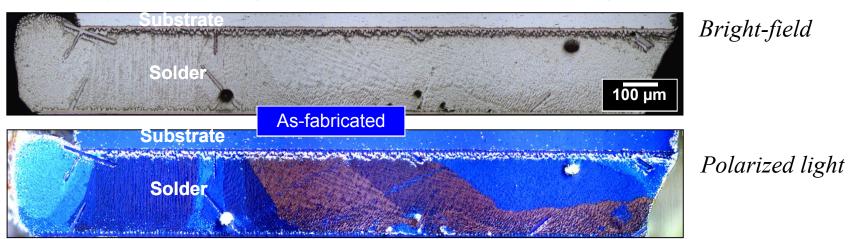


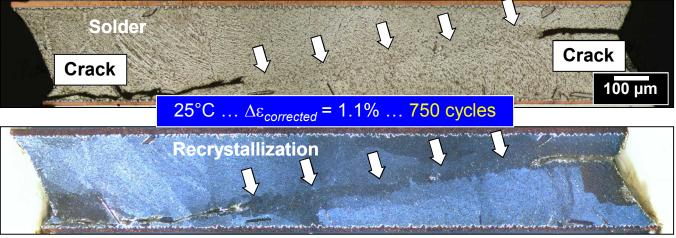


 This fatigue behavior of Sn-based solders has also been documented in constitutive modeling studies.

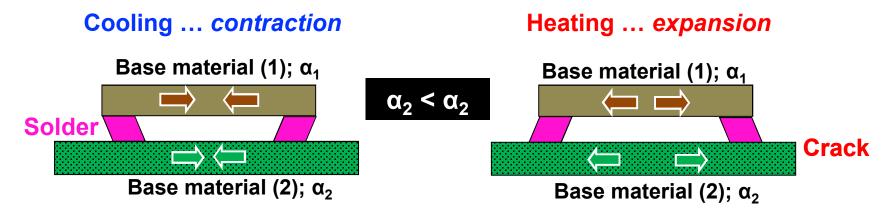


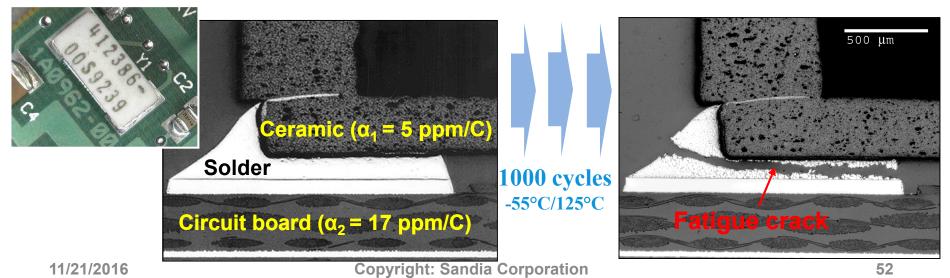
 The onset and magnitude of dynamic recrystallization likely has a significant role in the fatigue behavior.



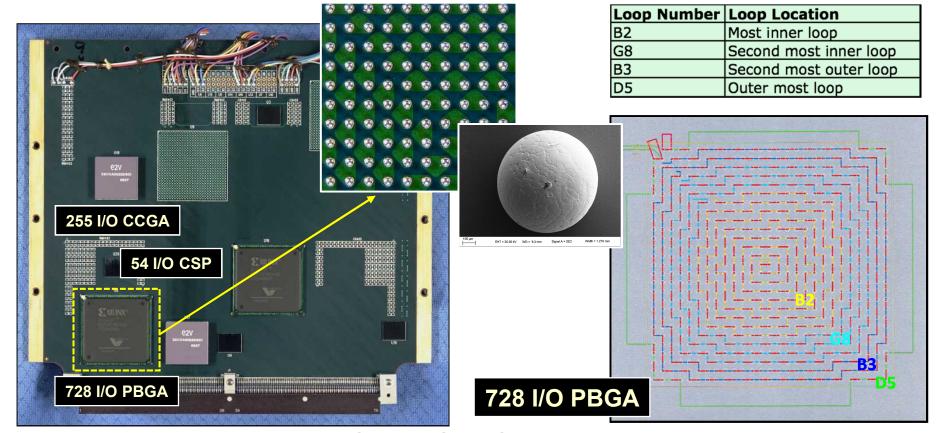


- A failure mode is thermal mechanical fatigue (TMF).
 - Combination of temperature cycling and the mismatch of base material, coefficients of thermal expansion (CTE), α .

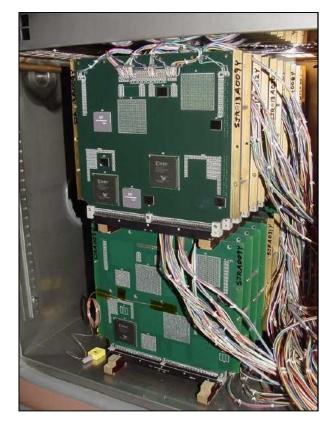




- Temperature cycling is used to assess the resistance of solder joints to thermal mechanical fatigue (TMF).
 - The test vehicle uses daisy chain loops to establish signal continuity, the metric for solder joints failure.

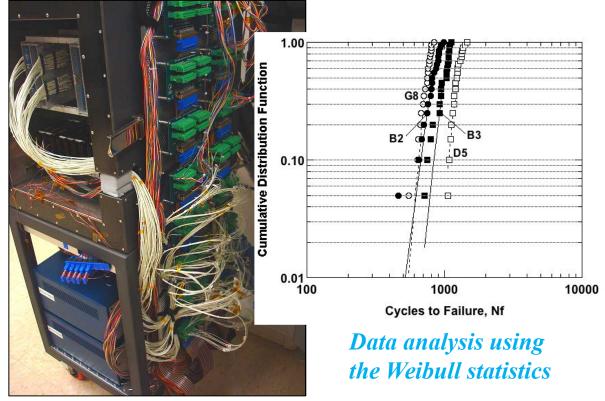


- Solder joints do not fail "cleanly" under shear strain.
- Daisy chain continuity is monitored by event detectors.

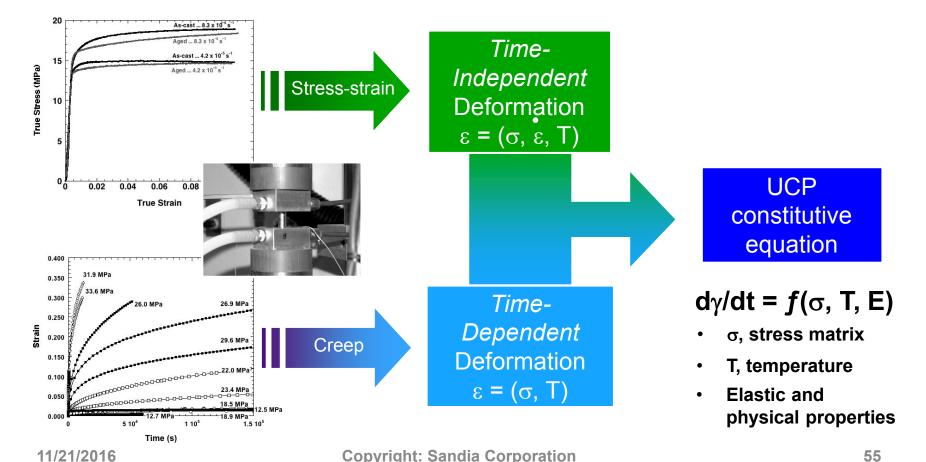


Test vehicles: wired and stacked in the thermal cycling chamber.

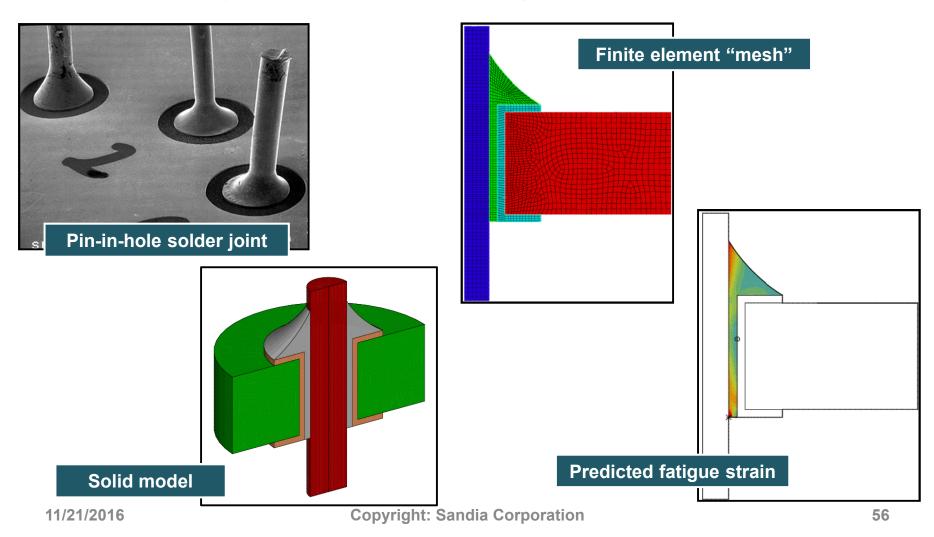
Wiring is attached to the event detectors.



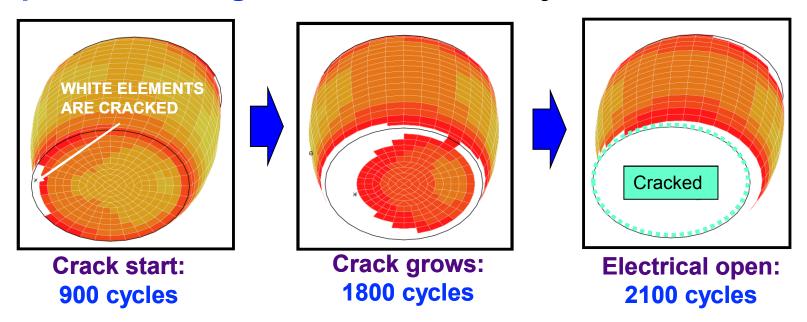
- The computational modeling approach is based upon the development of unified creep-plasticity equation.
 - Combine time-independent and time-dependent deformations



◆ The unified creep-plasticity equation is then applied to the solder joint structure using finite element models.

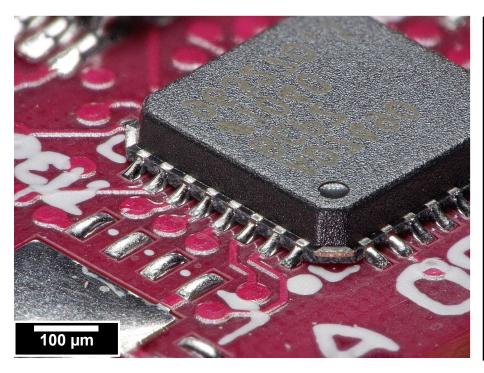


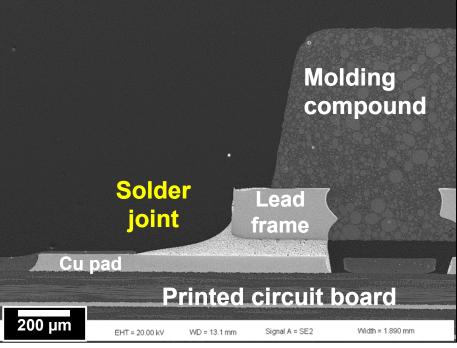
 The solder fatigue models now have the capacity to predict crack growth in the solder joint, as well.



- The advantages of the crack growth model are:
 - Document the morphology of crack propagation.
 - Determine load-bearing capacitor of partially cracked joints.
- Crack growth predictions are validated by empirical data.

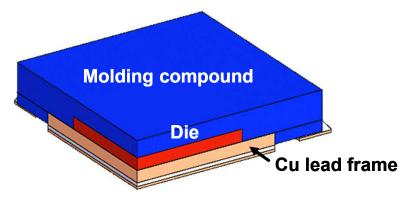
◆ The crack propagation model is demonstrated for the popular quad flat, no-lead (QFN) electronic package.



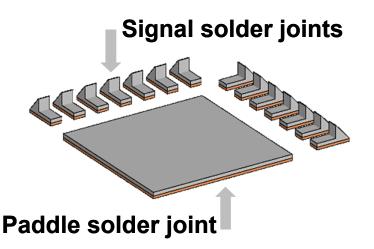


There are so many variants of the QFN package that empirical reliability testing is both cost and schedule prohibitive, which leaves computational modeling.

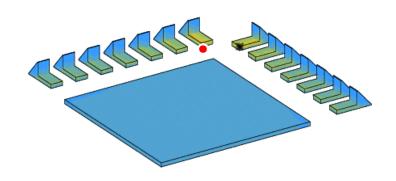
The computational model predicts crack behavior.



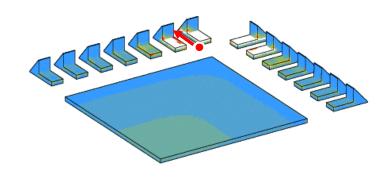
QPFN package and solder joints (quarter-symmetry)



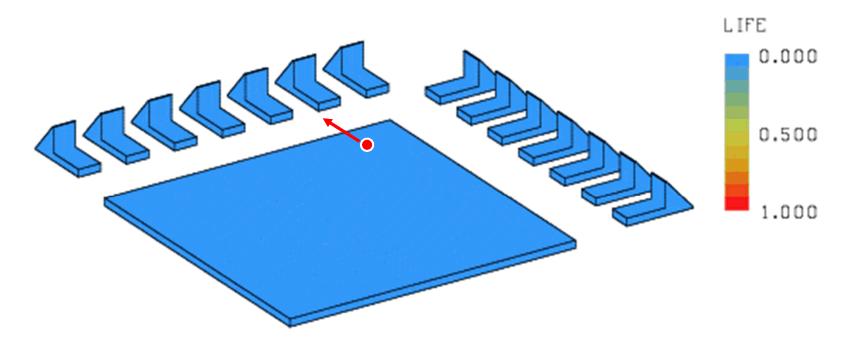
Crack initiation ... 1800 cycles



100% crack ... 7800 cycles

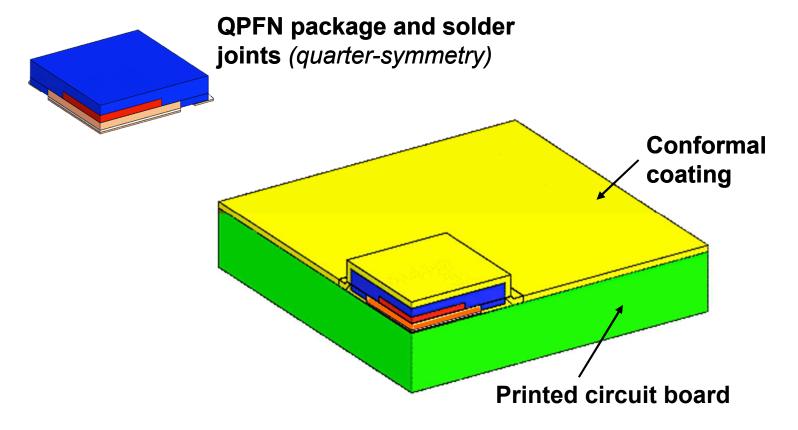


The software creates a movie of the crack propagation.



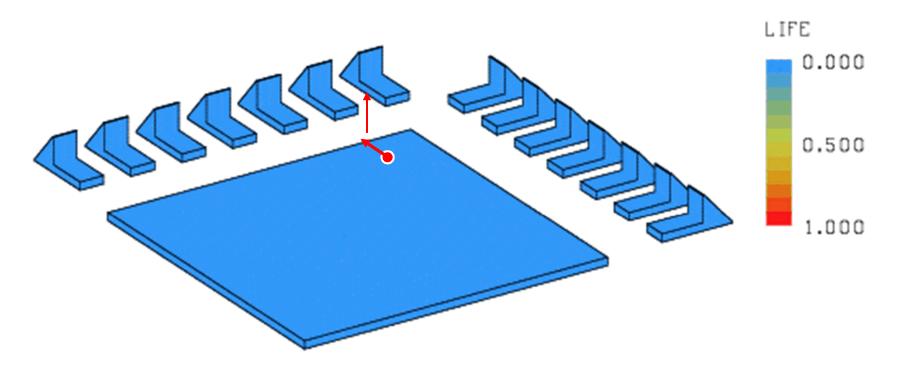
CYCLE 0.0000

- The presence of coatings that flow under the package have a significant effect on fatigue crack growth.
 - Movies visualize the change in crack propagation.



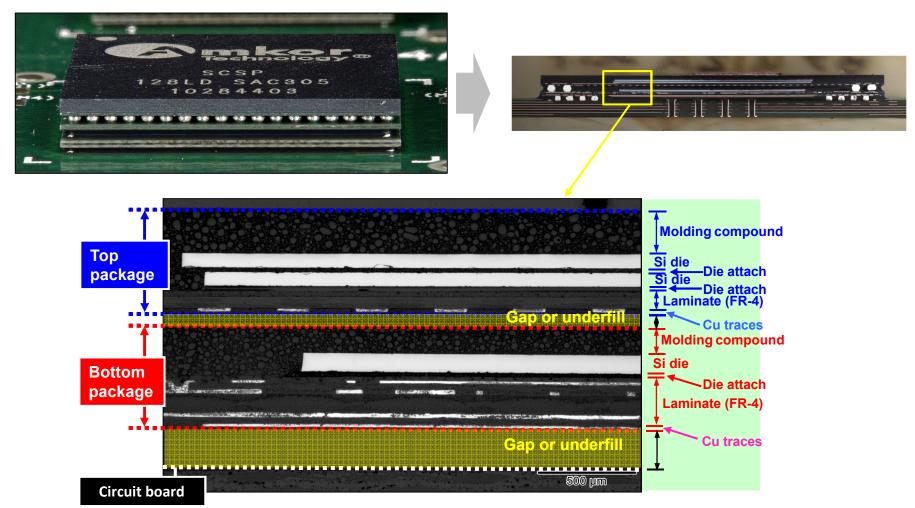
The conformal coating altered the crack path.

100% crack ... 160 cycles

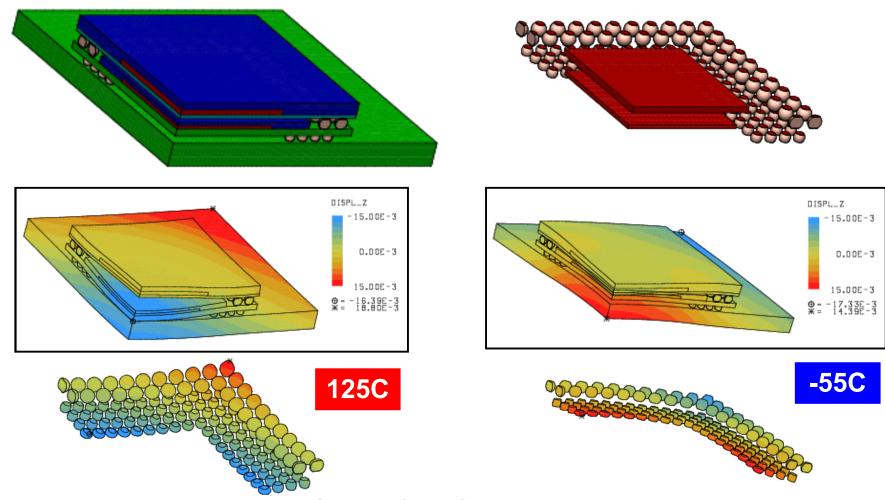


CYCLE 0.0000

 Stacked package technology presents very complex structures that are well suited for numerical models.

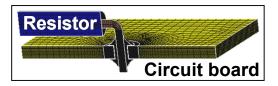


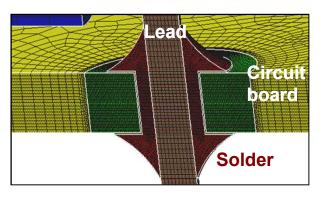
 Computational modeling predicts behaviors that are not intuitively obvious due to the complex construction.

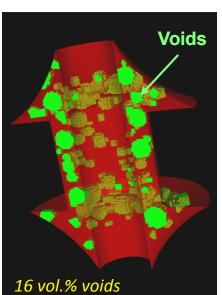


 Computational modeling predicts behaviors that are not intuitively obvious due to the complex construction.



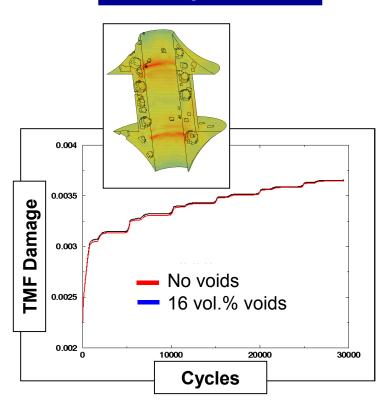






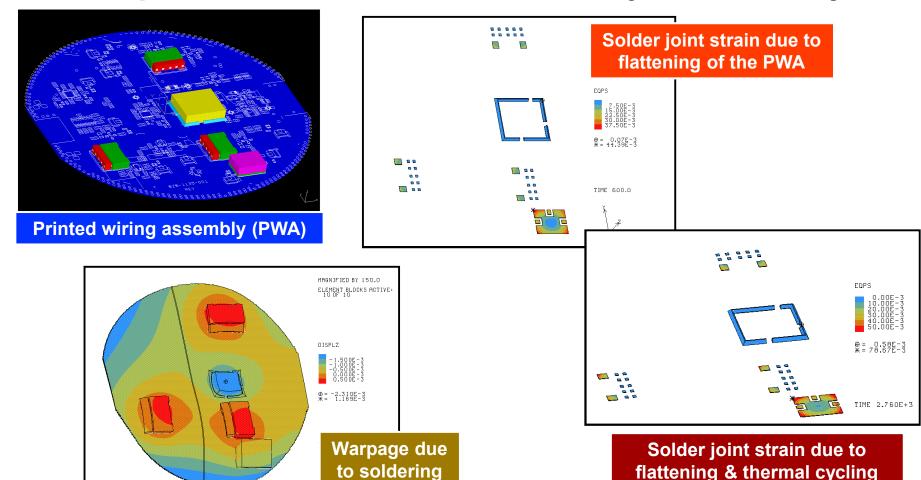
Void sizes and distribution replicated observations of an actual solder joint.

Model prediction

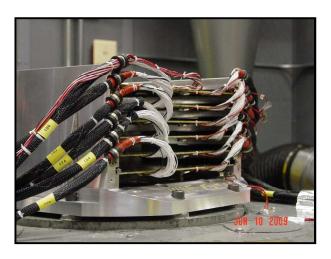


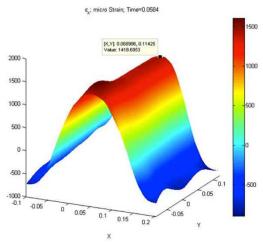
Voids of 0 – 16 vol.% content do not affect reliability.

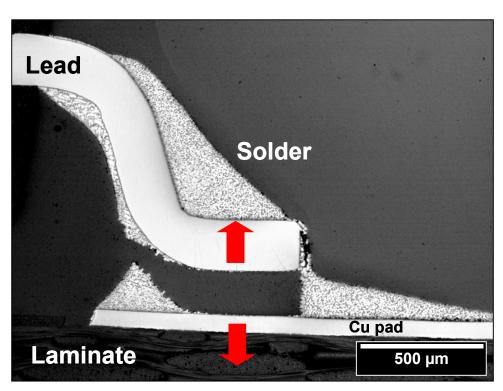
 Computational model can readily address the effects of sequential environments on solder joint reliability.



 Vibration performance (high-cycle fatigue) has not received as much attention for solder interconnections.





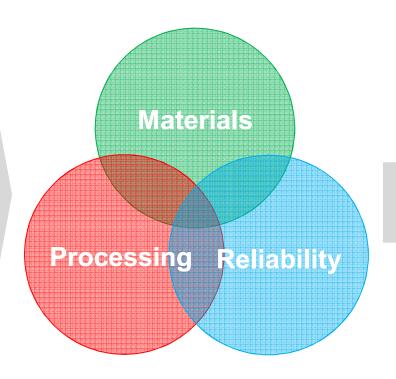


The physical and mechanical metallurgy of high-cycle fatigue is not well-understood.

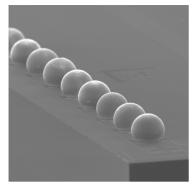
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Summary









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