

**LA-UR-17-30525**

Approved for public release; distribution is unlimited.

**Title:** MPA-11: Materials Synthesis and Integrated Devices; Overview of an Applied Energy Group

**Author(s):** Dattelbaum, Andrew Martin

**Intended for:** Departmental Seminars and Group summary presentations

**Issued:** 2017-11-16

---

**Disclaimer:**

Los Alamos National Laboratory, an affirmative action/equal opportunity employer, is operated by the Los Alamos National Security, LLC for the National Nuclear Security Administration of the U.S. Department of Energy under contract DE-AC52-06NA25396. By approving this article, the publisher recognizes that the U.S. Government retains nonexclusive, royalty-free license to publish or reproduce the published form of this contribution, or to allow others to do so, for U.S. Government purposes. Los Alamos National Laboratory requests that the publisher identify this article as work performed under the auspices of the U.S. Department of Energy. Los Alamos National Laboratory strongly supports academic freedom and a researcher's right to publish; as an institution, however, the Laboratory does not endorse the viewpoint of a publication or guarantee its technical correctness.

# MPA-11: Materials Synthesis and Integrated Devices

## Overview of An Applied Energy Group



**Andrew M. Dattelbaum**  
**Group Leader**

November 10, 2017



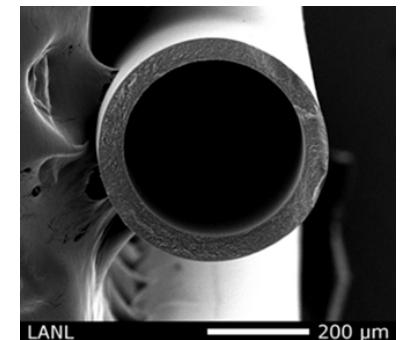
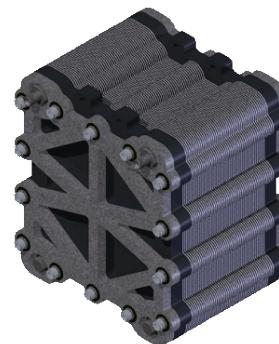
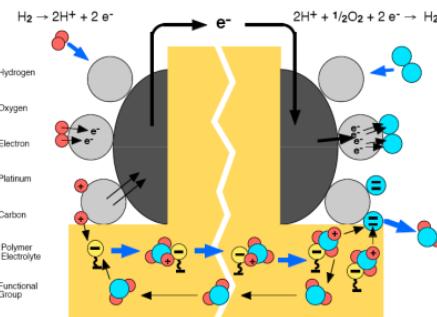
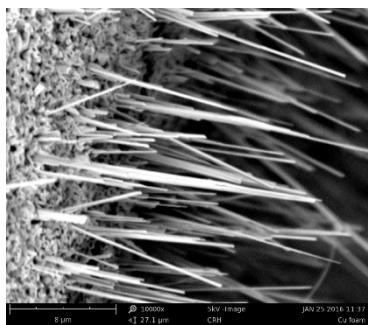
Operated by Los Alamos National Security, LLC for the U.S. Department of Energy's NNSA

# Materials Synthesis and Integrated Devices (MPA-11): An Applied Energy Group

- Our mission is to provide innovative and creative chemical synthesis and materials science solutions to solve materials problems across the LANL missions.
- Our group conducts basic and applied research in areas related to energy security as well as problems relevant to the Weapons Program.

## Areas of Specialization:

- Polymer Electrolyte Fuel Cells
- Acoustics Technology
- Weapons Aging and Safety
- Actinide Chemistry/Materials
- Polymer membranes for gas separations
- Nanomaterials synthesis of 2D materials
- Electrochemical-based gas sensor and devices



# Los Alamos National Laboratory Overview

## People

- ▶ 7,200 permanent employee
- ▶ ~3500 technical staff members
- ▶ 350 Post doctoral researchers
- ▶ 1,600 students

## Place

- ▶ 35 miles northwest of Santa Fe, NM
- ▶ 1,000 individual facilities,
- ▶ 47 technical areas
- ▶ 8.2 million square feet

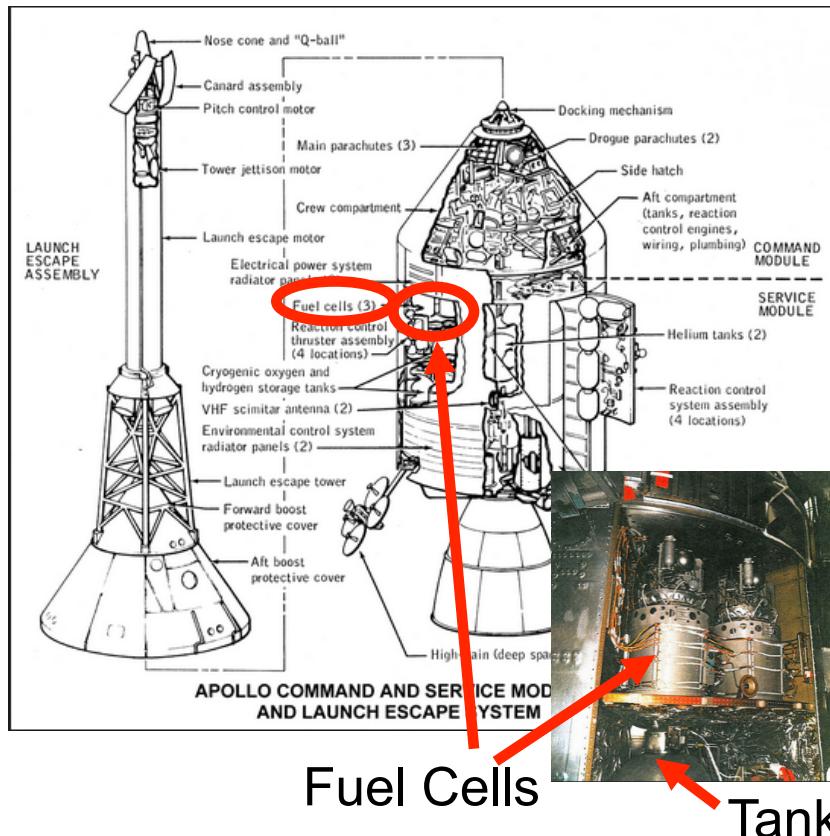
## Budget (FY16)

- ▶ ~ \$2.45 billion budget
- ▶ 65% Weapons program
- ▶ 4% DOE Office of Science
- ▶ 2% Energy programs
- ▶ 10% Work for others



# Fuel Cells – Space Applications

- Fuel Cells found their first applications in space
  - Fuel cells first flew on Gemini space missions (PEM)
  - Apollo missions were powered by two/three fuel cells (Alkaline)
  - Space Shuttle had three fuel cells (Alkaline)



General Electric PEM fuel cell (test version) used on the two-astronaut Gemini spacecraft during seven missions in 1965-66.

# The Beginning of the DOE Fuel Cell Program...

**1970s**

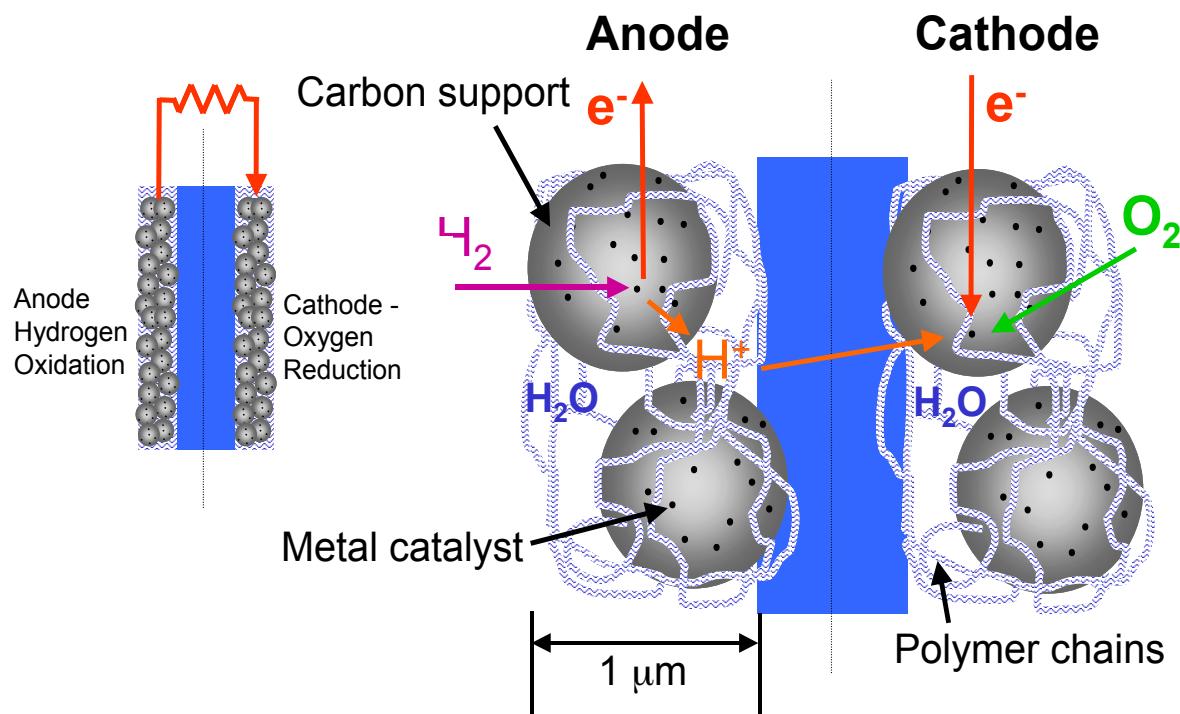
A group from labs, government and industry met at Los Alamos to set the foundation for DOE fuel cell programs



Lab researchers taught scientists around the world how to fabricate fuel cell electrodes. Group from GM relocated to Los Alamos.

# History of the LANL Fuel Cell Program

## LANL Enabling Breakthrough Thin Film Electrode

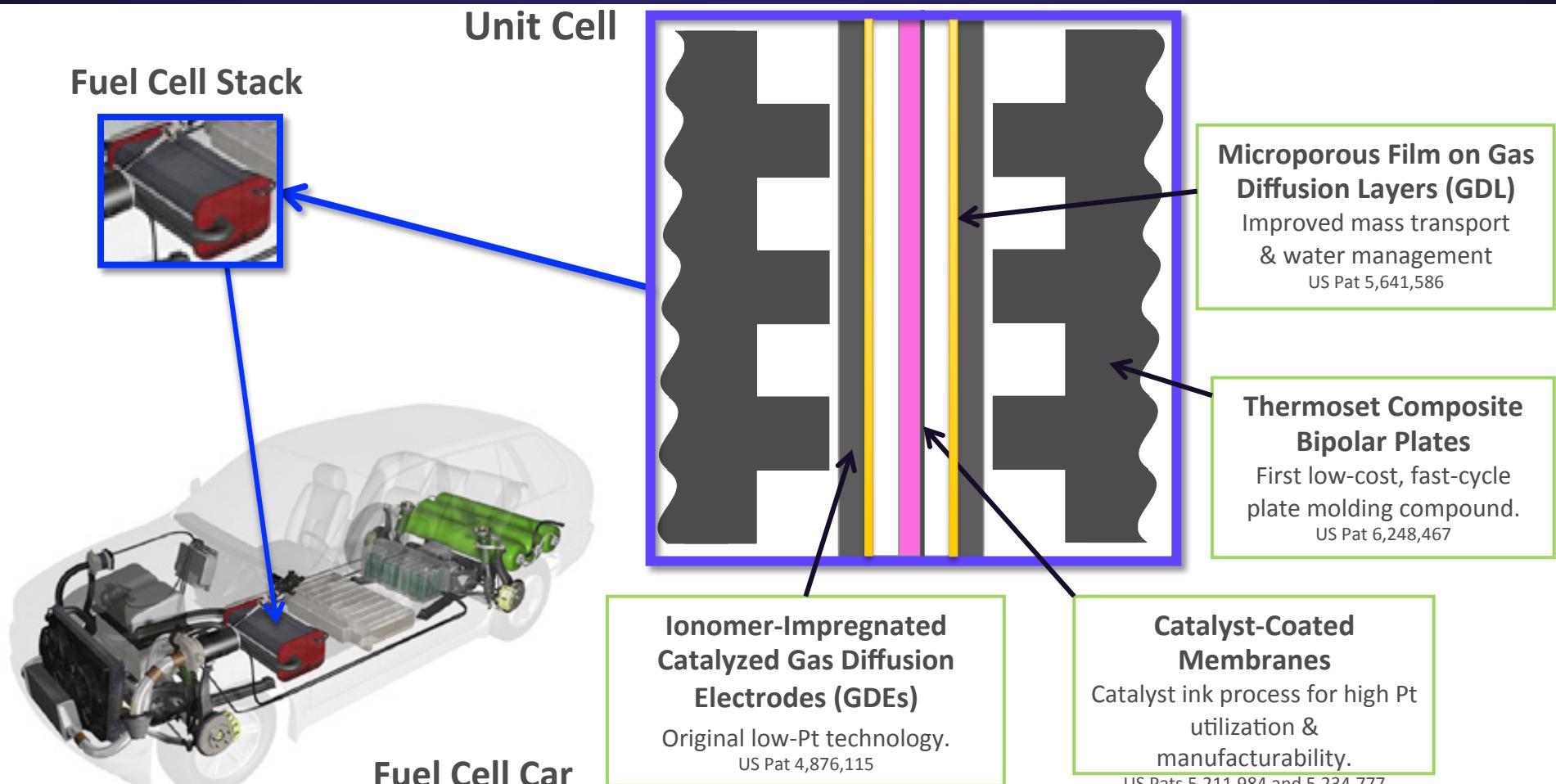


**An electrochemically active reaction site must have reactant access to catalyst, available electronic and ionic conduction paths, and manage water**

US Patents #4,876,115, #5,211,984 and #5,234,777

- ▶ One of longest running non-weapons programs at LANL (since 1977)
- ▶ The current DOE Fuel Cell Technologies Office (FCTO) program grew from the original Los Alamos program
- ▶ Program is primarily focused on polymer electrolyte membrane (PEM) technology for transportation
- ▶ LANL has the largest portion of the DOE EERE FCTO Fuel Cell Component R&D budget of any organization (NL or other)

# LANL-Led Innovation in Fuel Cells



LANL's innovation in fuel cells technology has played a critical role in the technical viability of fuel cell stacks for FCEVs.

# Fuel Cell Electric Vehicles (FCEVs) are on U.S. Roads TODAY!



*Hyundai Tucson*



*Toyota Mirai*



*Honda Clarity*

Commercial  
FCEVs are here  
today!

Can reduce total GHG  
emissions

50-90% vs. today's gasoline  
vehicles

More than 1,100 FCEVs  
sold or leased in the U.S.

# Why Fuel Cells for Transportation Applications?

## Hydrogen Enables U.S. Energy Security

### Petroleum Diversification

~ 92% of Transportation

### Hydrogen Production

Natural Gas Steam Reforming

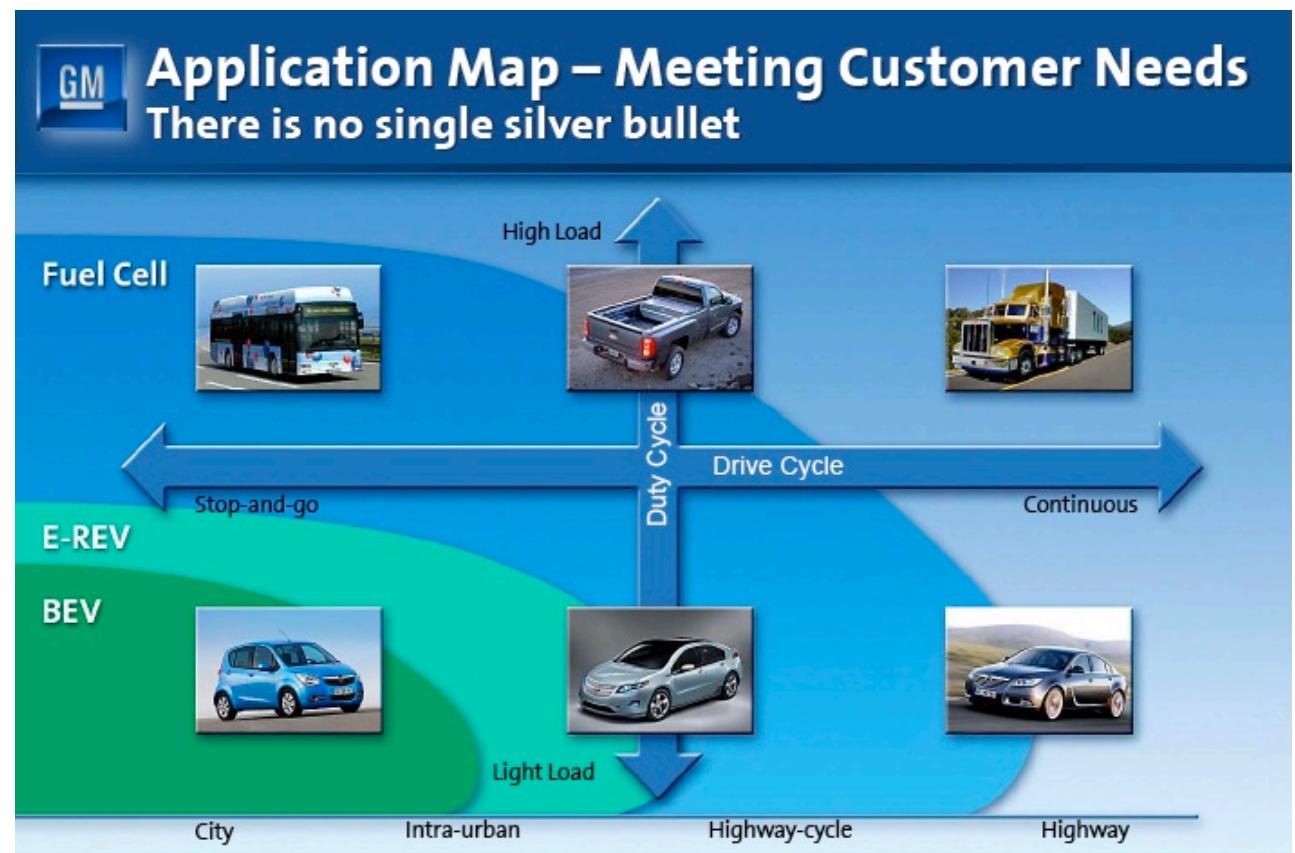
Longer-term: Renewables  
(wind, solar)

### Green House Gas Emissions

>50% with H<sub>2</sub> from

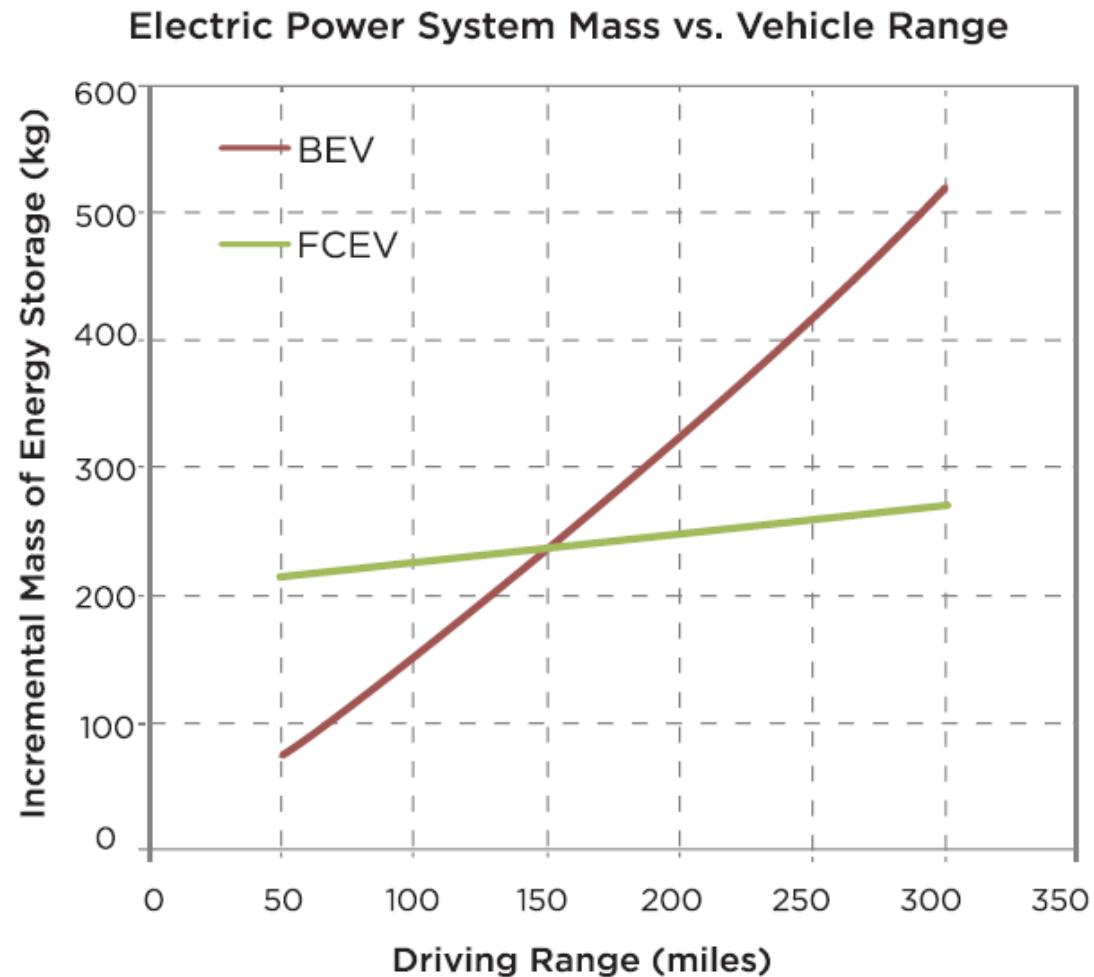
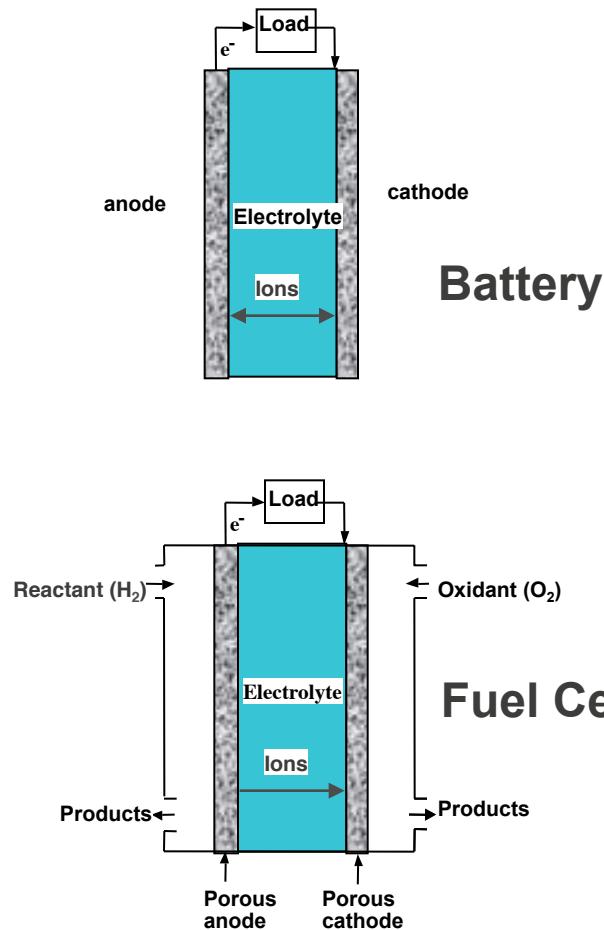
Distributed Natural Gas

>90% with H<sub>2</sub> from  
renewable sources



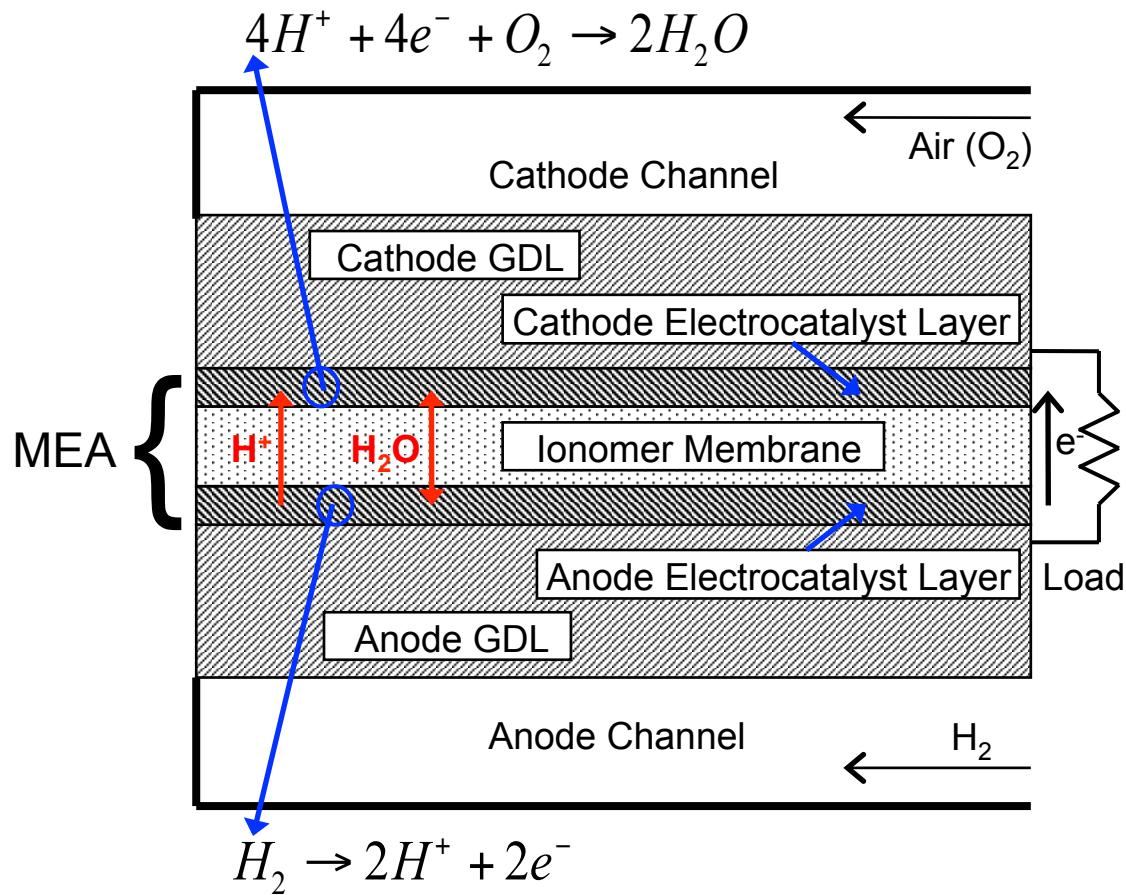
Source: [http://hydrogen.energy.gov/pdfs/13005\\_well\\_to\\_wheels\\_ghg\\_oil\\_idvs.pdf](http://hydrogen.energy.gov/pdfs/13005_well_to_wheels_ghg_oil_idvs.pdf)

# Fuel Cell vs. Battery: A Range Advantage



**Advantages:** energy density (up to  $>6000$  Whr/kg based on fuel, batteries up to 150 Whr/kg), refuel versus recharge

# The Components of a Polymer Electrolyte Membrane Fuel Cell (PEMFC)



## Materials

Membrane: **Perflourosulfonic acid (Nafion)**  
Durability, cross-over

Catalysts:

Anode: **Pt**  
Cathode: **Pt**  
Durability to cycling

Catalyst Support: **N/A**

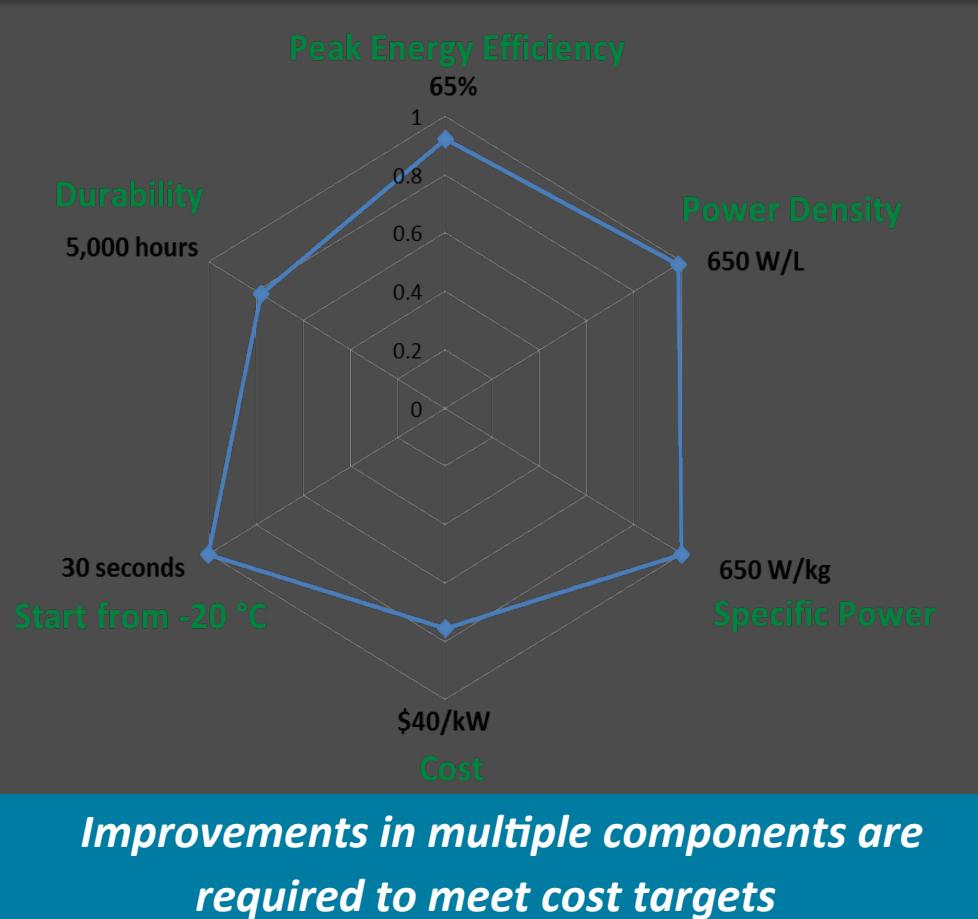
Gas Diffusion Layer: **Carbon fiber with carbon micro-porous layer**

Gaskets: **PTFE**,  
Durability/Sealing

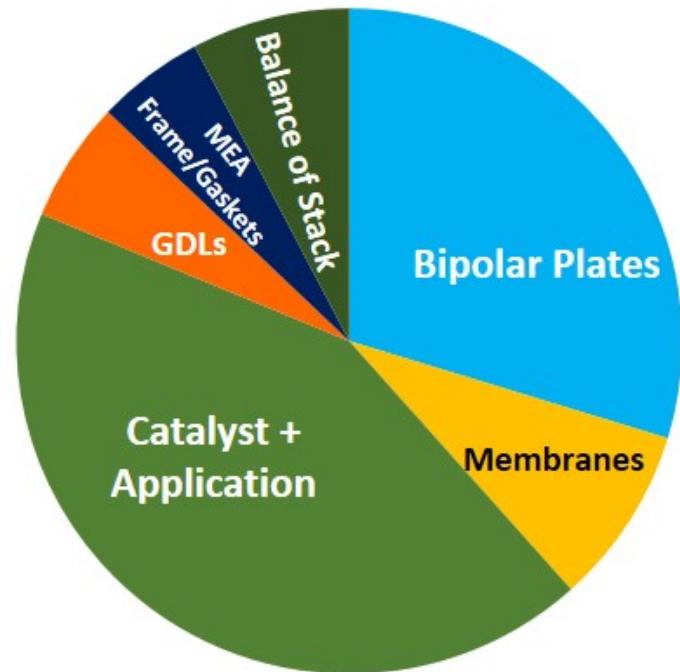
Bipolar Plate: **Coated Metal**  
Corrosion resistance  
Contact resistance

# Fuel Cell Challenges

Durability and Cost are the primary challenges to fuel cell commercialization and must be met concurrently

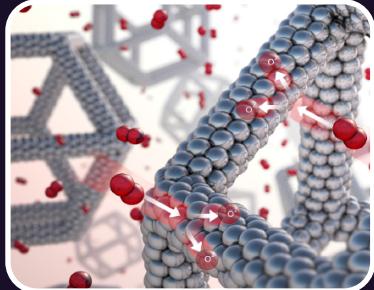


## PEMFC Stack Cost Breakdown\*



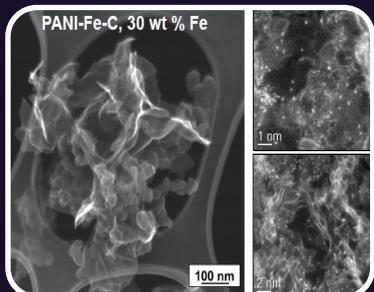
Catalyst cost is projected to be the largest single component of the cost of a PEMFC manufactured at high volume.

# Approaches to Reduce Fuel Cell Costs



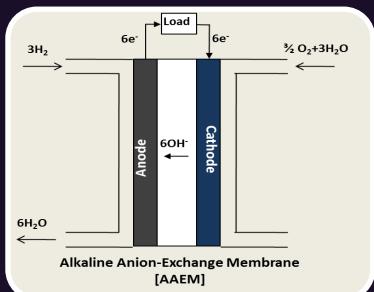
## Low Pt Cathode Catalyst Development

- Near term Approach
- ANL, BNL, LANL, NREL



## Non-PGM Cathode Catalyst Development

- Mid term
- LANL, ANL



## Alkaline Membrane Fuel Cells

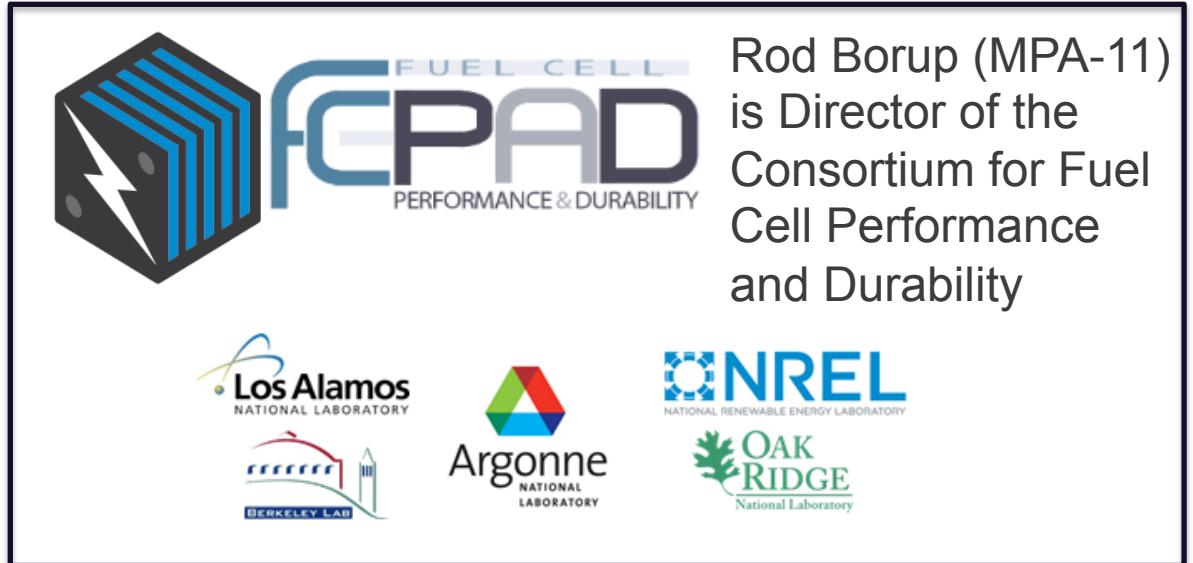
- Long-term
- LANL, SNL, NREL

# LANL has leadership roles in two DOE National Lab consortia and applies fuel cells to NNSA applications

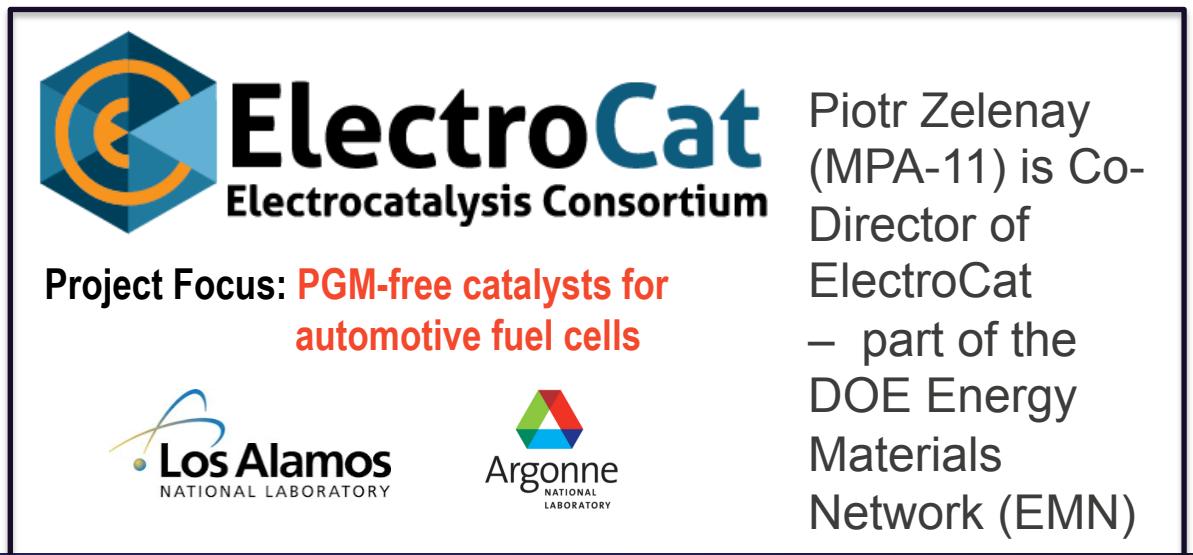
*Fuel cell system targets set to be competitive with ICEVs.  
Durability and Cost are the primary challenges to fuel cell commercialization and must be met concurrently*



MPA-11 is also testing fuel cells for stockpile and GS applications



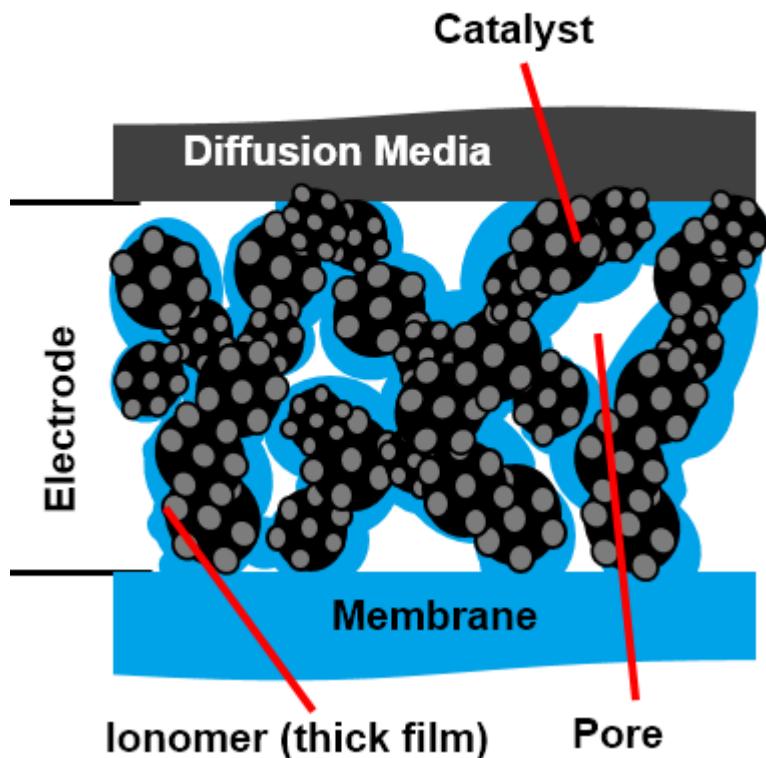
Rod Borup (MPA-11) is Director of the Consortium for Fuel Cell Performance and Durability



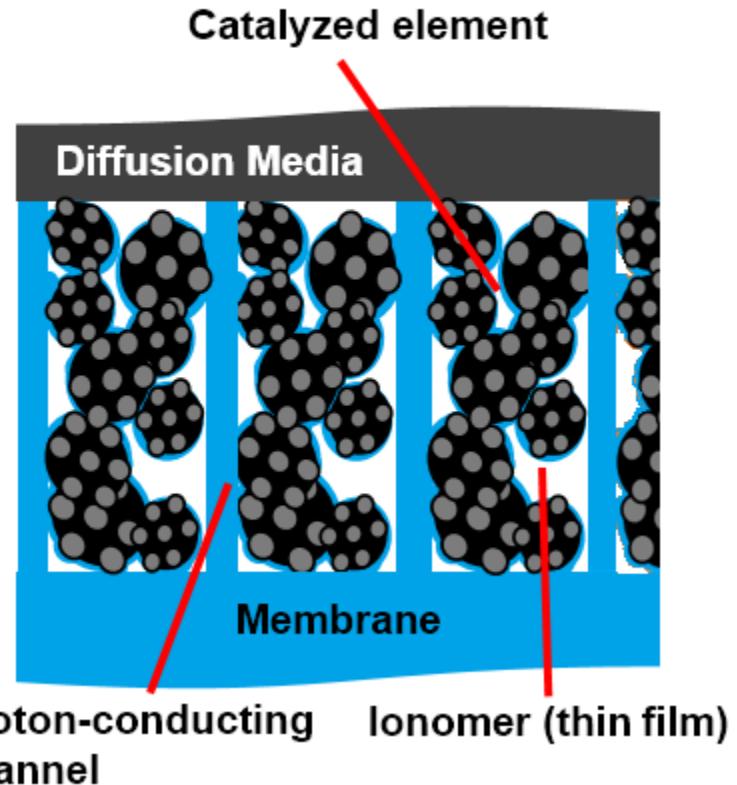
Piotr Zelenay (MPA-11) is Co-Director of ElectroCat – part of the DOE Energy Materials Network (EMN)

# FC-PAD Highlight: Novel Array Electrode

## Conventional Electrode\*



## Array Electrode\*

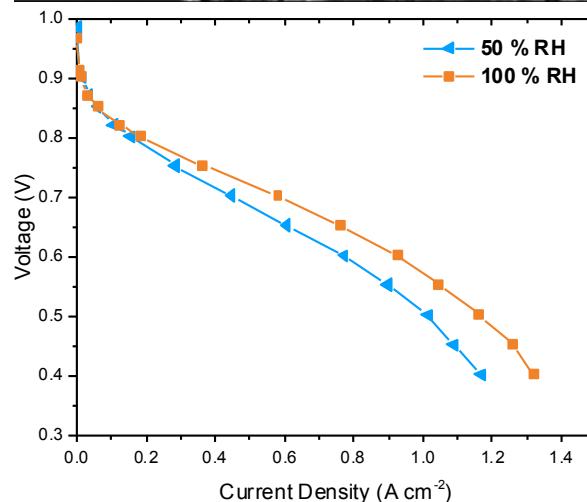
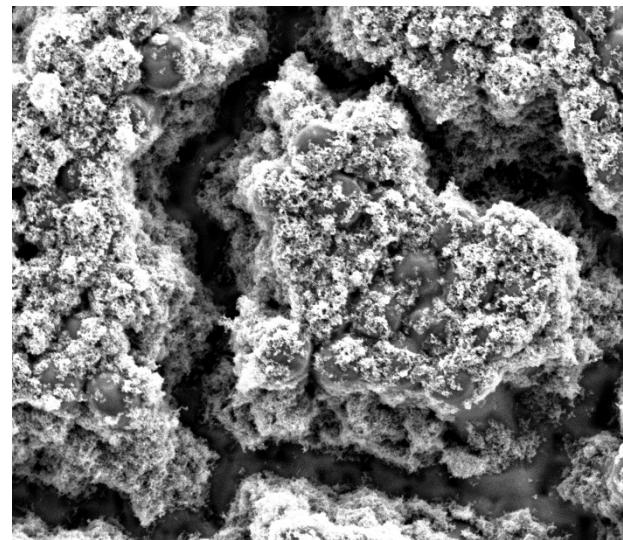
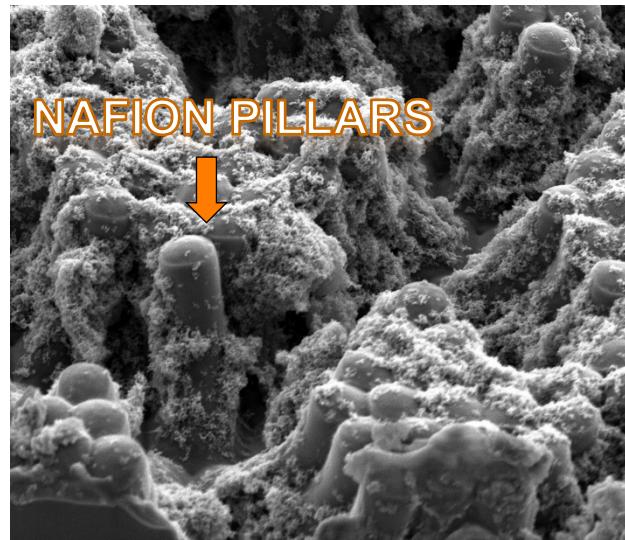


- Array electrode relies on vertically-aligned ionomer channels for long-distance  $H^+$  transport  
→ **Reduced catalyst layer sheet resistance**
- Catalyzed elements can have reduced ionomer content  
→ **Reduced  $O_2$  transport barrier**

\*not to scale

# Catalyst Deposition and Fuel Cell Testing

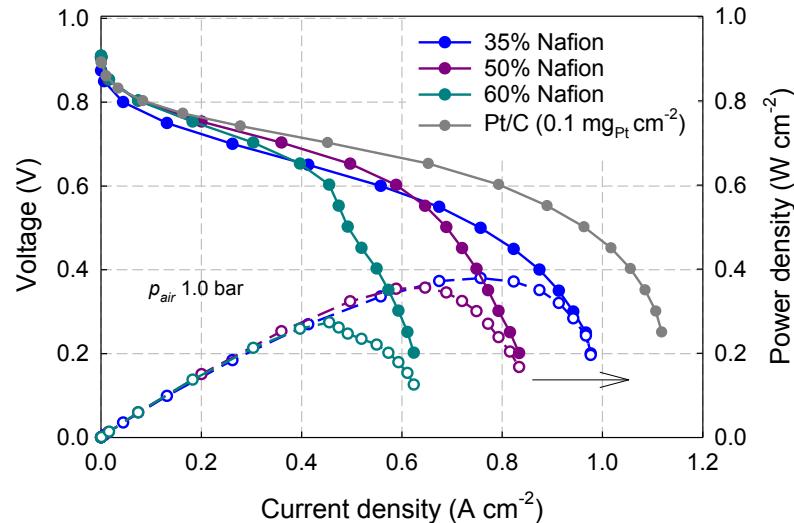
Catalyst deposited around Nafion channels (SEM)



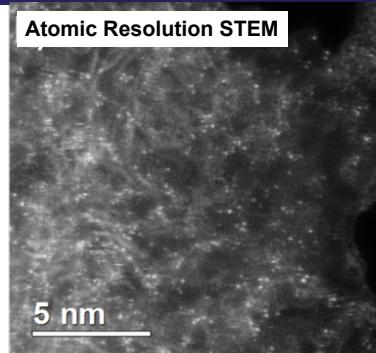
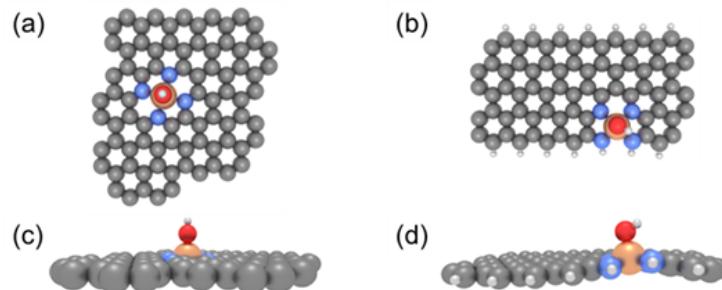
Nafion channels enable high performance under dry conditions – only ~10% difference between 50% and 100% RH performance

# Insight into Active Sites of High-Performance PGM-free Oxygen Reduction Reaction (ORR) Catalysts

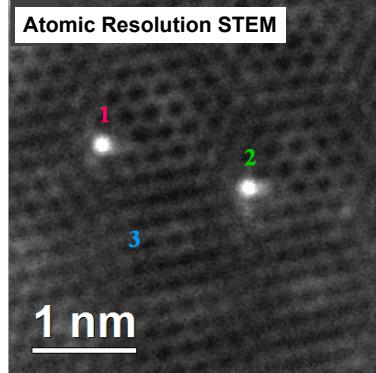
High-performance (CM+PANI)-Fe-C oxygen reduction reaction (ORR) catalyst with a hierarchical pore structure



DFT Model structures used in theoretical active-site studies  
Confirms  $\text{Fe-N}_4$  structure activity – most are likely at edge  
(C – gray; Fe – bronze; H – white; N – blue; N – red)



Advanced STEM and EELS characterization

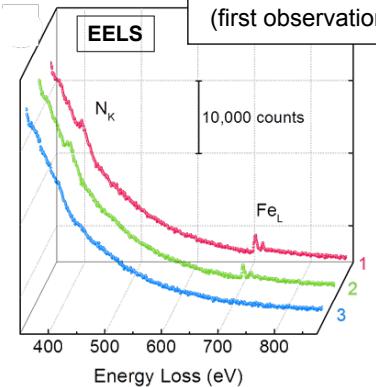


Advanced STEM and EELS characterization

N 79.5 at.%  
Fe 20.5 at.%

$\text{FeN}_4$

(first observation)



- ORR kinetics fuel cell performance matching that of a Pt catalyst at a loading of  $0.1 \text{ mg cm}^{-2}$
- Proposed catalytic site ( $\text{FeN}_4$ ) directly visualized for the first time using HR-STEM at ORNL
- Contributions of active sites associated with specific lattice-level carbon structures explored computationally and referred to the experiment

# Alkaline conditions for Fuel Cell Systems

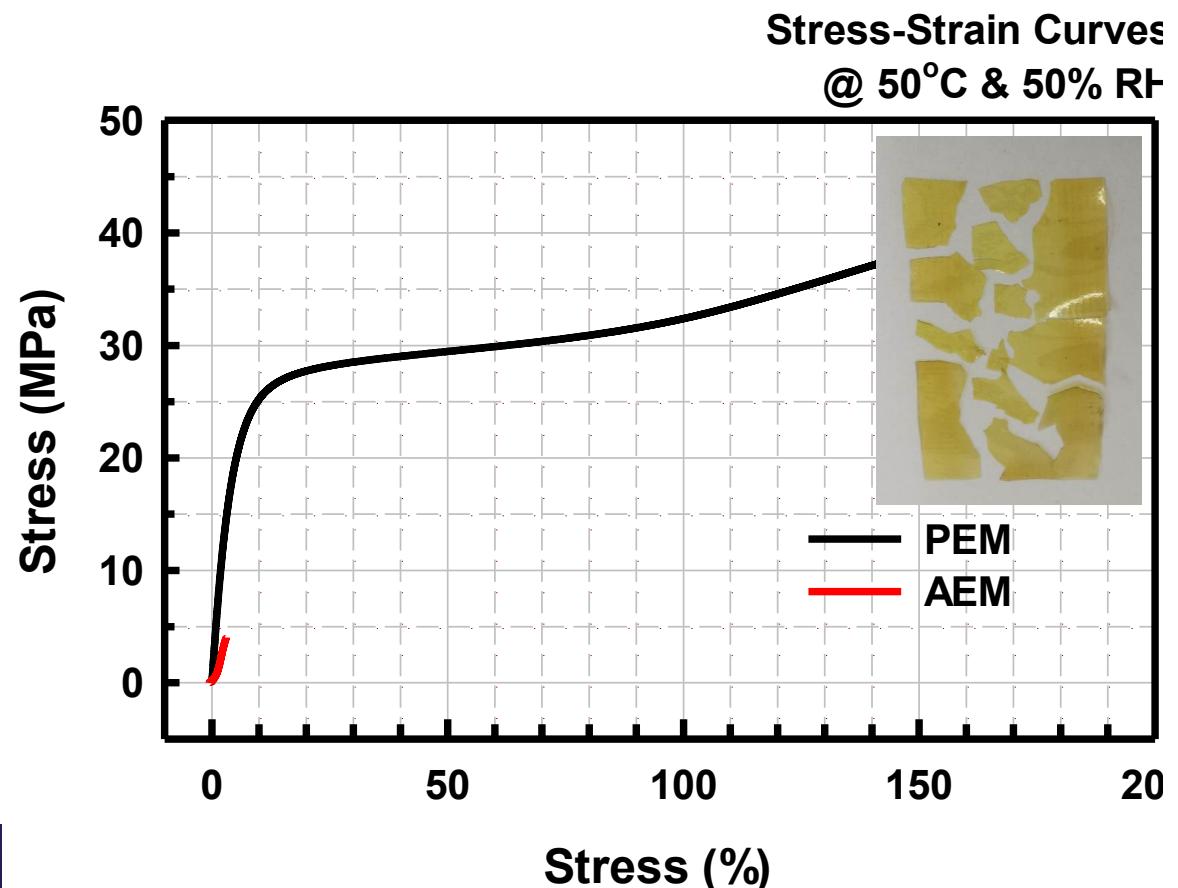
Nitrogen-containing carbon nanotubes can act as a metal-free electrode with a much better electro-catalytic activity..... than platinum for oxygen reduction in alkaline fuel cells" From Liming Dai et al. Science, 323, 760, Feb. 2009.

## Membrane Embrittlement under high pH

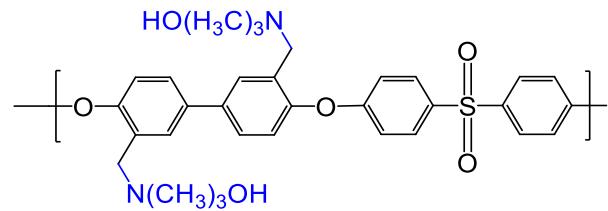
### Alkaline membrane



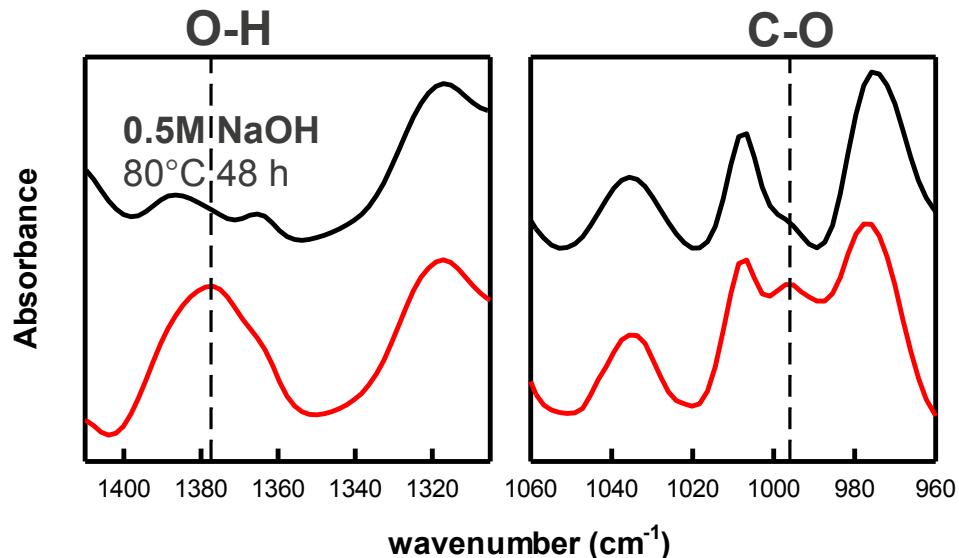
- IEC = 1.6 meq/g
- $M_n$  = 42000 g/mol (prepolymer)
- Base treatment: 0.5M NaOH, 80°C, 60 min



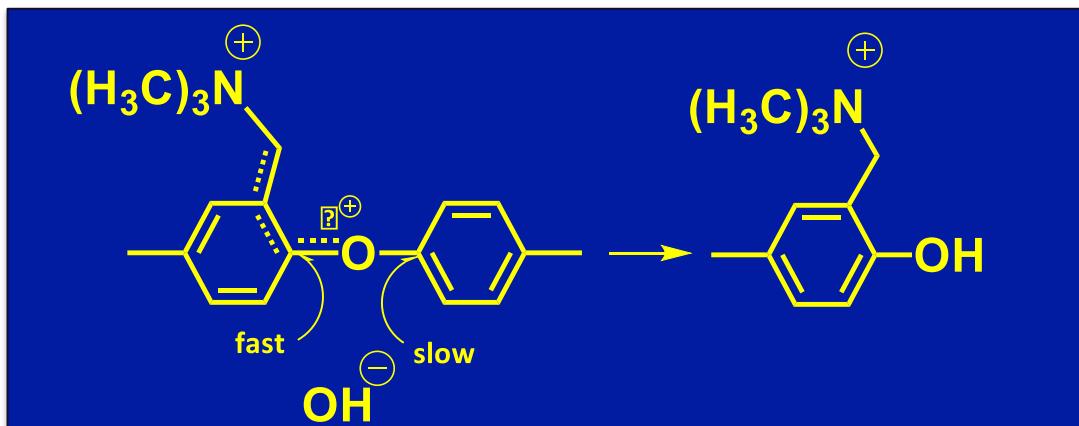
# Determination of Degradation Mechanism



Increased phenolic OH peak at 1378 and 996 cm<sup>-1</sup> after 0.5 M NaOH treatment



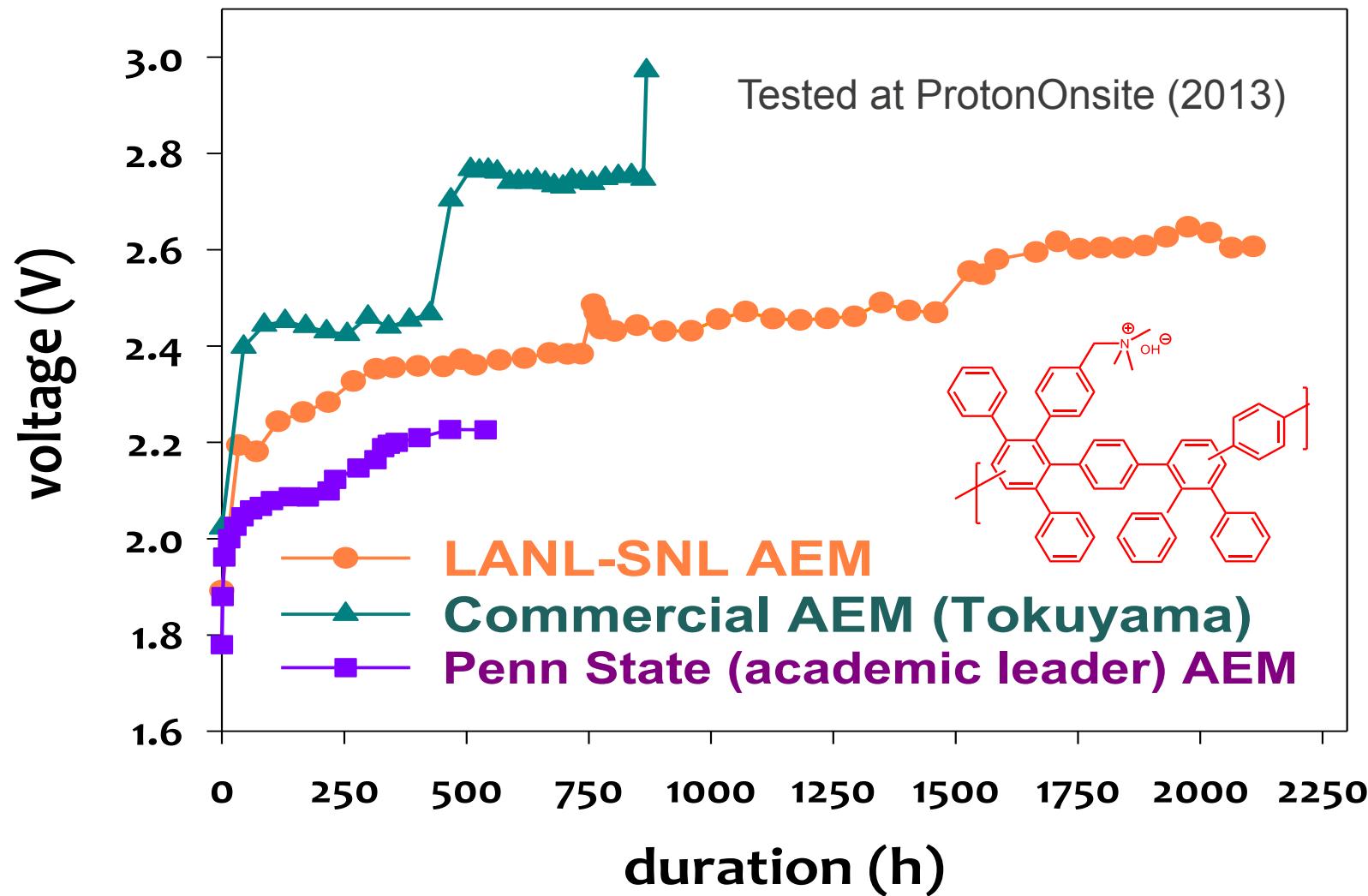
## Proposed degradation mechanism\*



**Solution:**  
Utilize Aryl-Free polymers to solve  
mechanical  
degradation  
issues

# Stable Alkaline Membranes for Water Electrolyzers: Aryl-free Membranes

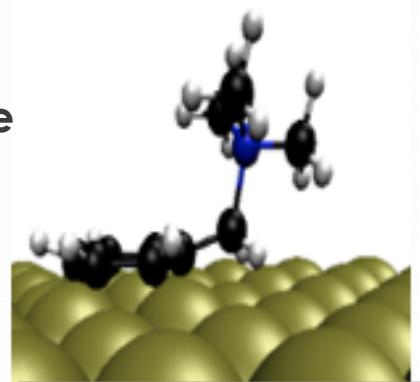
# ***In-situ Membrane Durability Test***



# Alkaline membrane fuel cell (AMFC) research on catalyst electrolyte interface

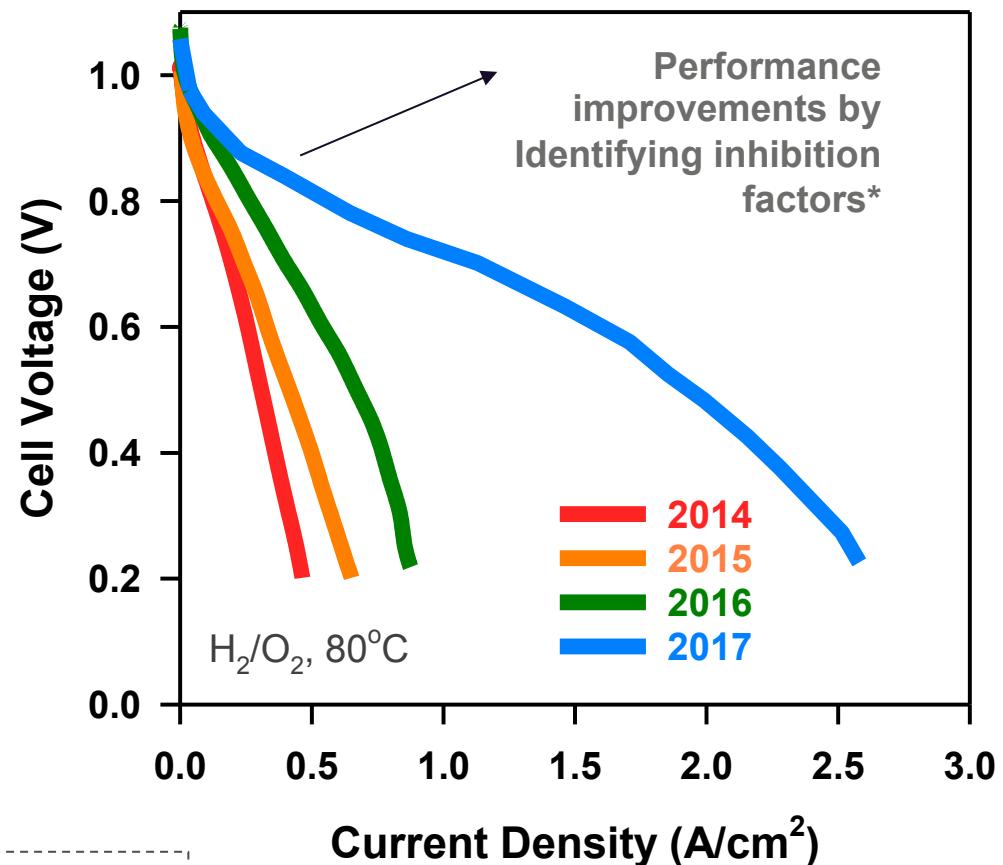
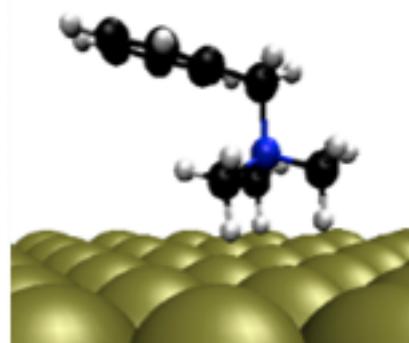
## Benzene adsorption limits AMFC performance

Matanovic et al. *J. Phys. Chem. Lett.* 2017, 8(19), 4918-4924.



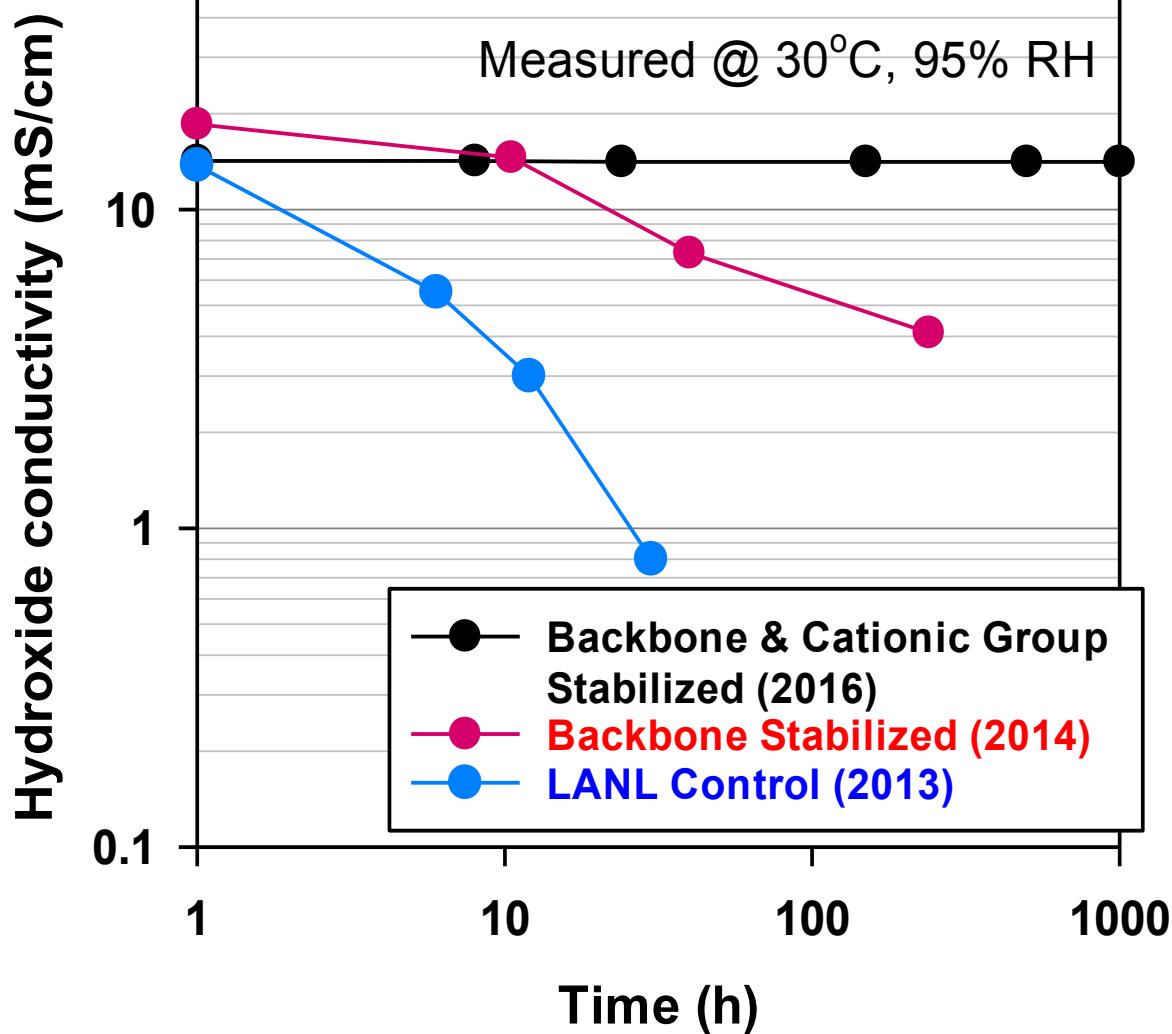
## Ammonium adsorption limits AMFC durability

Chung et al. *J. Phys. Chem. Lett.* 2016, 7(22), 4464-4469.

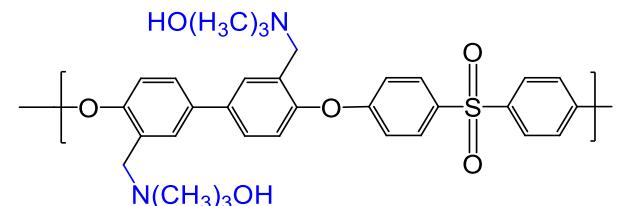


Identified performance and durability limiting factors (remove benzyl ammonium groups) of AMFC needed to reproducibly reach 1 W/ $\text{cm}^2$  peak power density

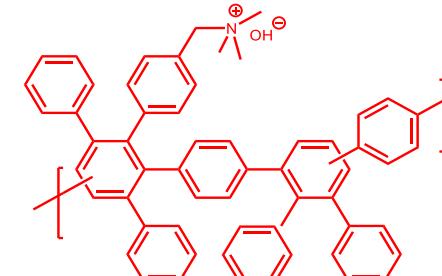
# No degradation for backbone and cationic group stabilized AEM after 0.5 M NaOH treatment at 80 °C for over 1000h.



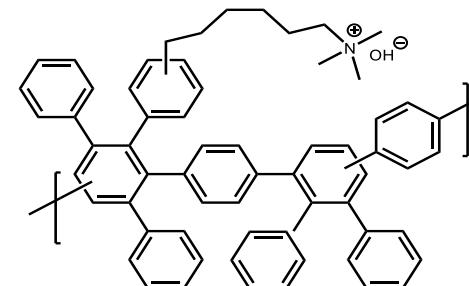
LANL control (2013)



Backbone Stabilized (2014)

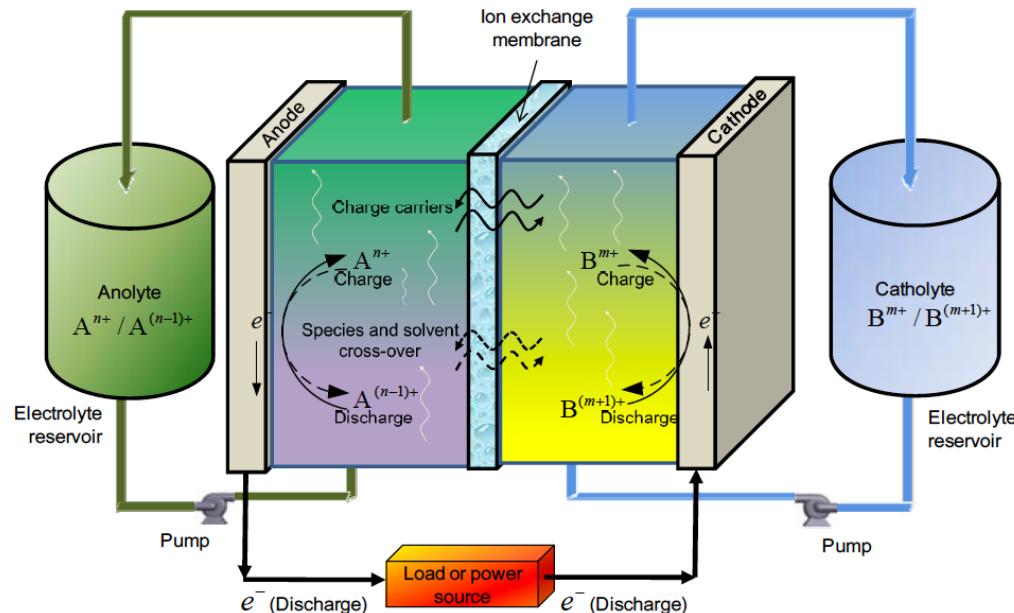


Backbone & Cationic Group Stabilized (2016)



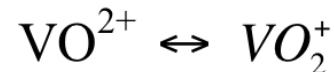
# Grid Scale Energy Storage using Flow Cells (Redox Flow Batteries)

Need to maintain a robust and resilient electricity delivery system that incorporates significant intermittent renewable sources

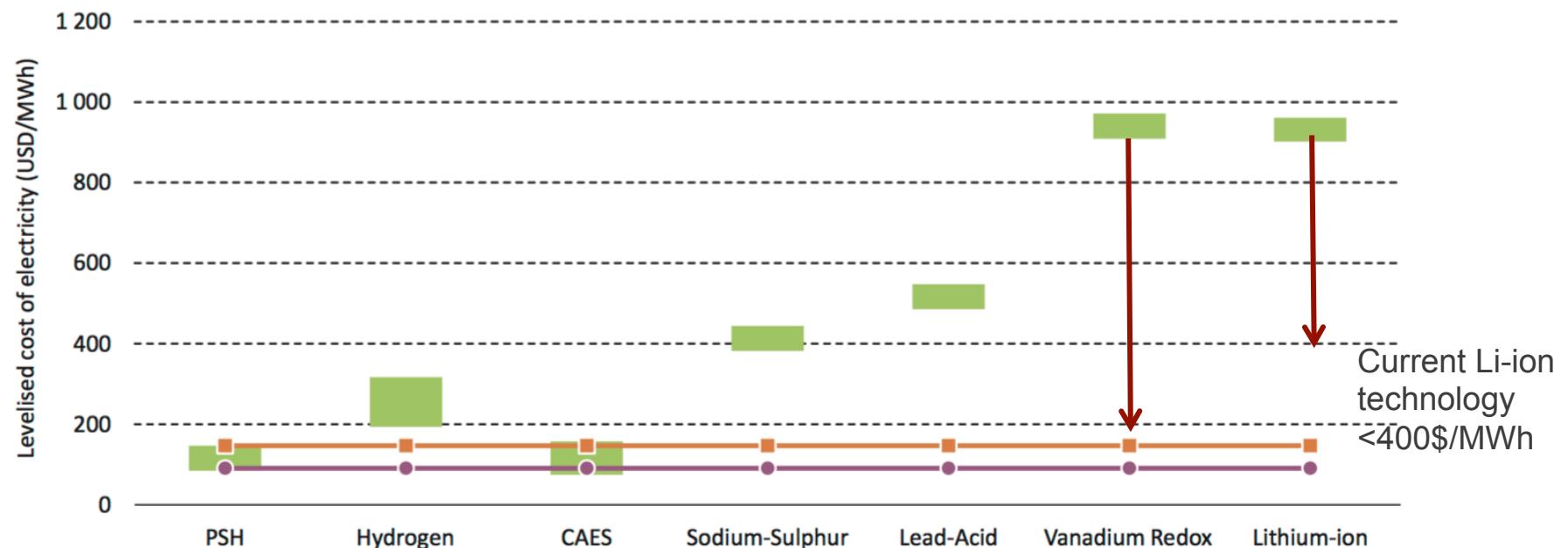


- Flow cells are uniquely scalable unlike other battery technology. Energy is determined by amount of electrolyte and size of tanks. Power is controlled by size of stack.
- Potential for high durability. Electrodes do not store energy or undergo chemical change
- Potential for high efficiency (>80% round trip efficiency possible)

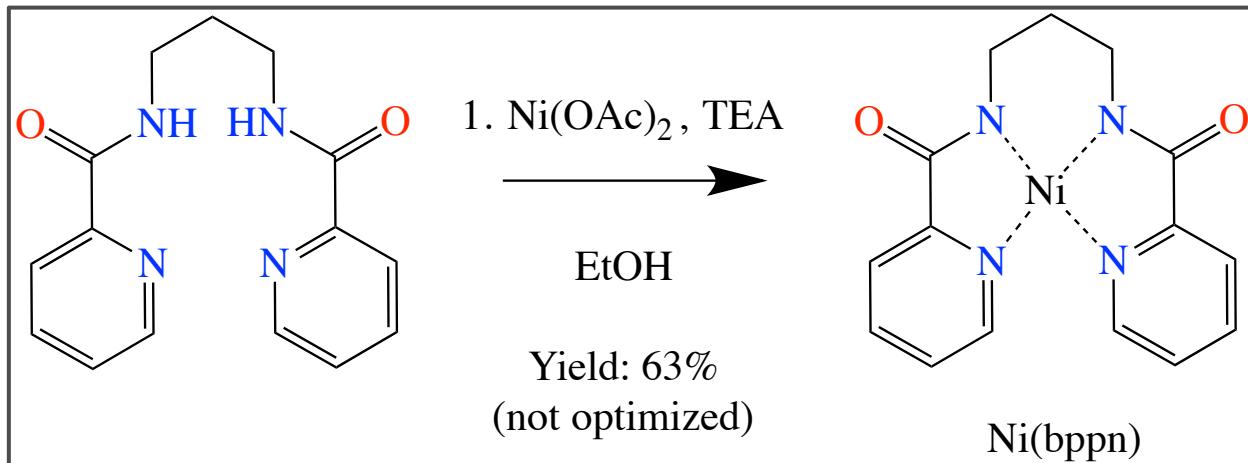
# State of the art: Vanadium redox flow batteries



Limitations due to vanadium costs



# New Charge Carriers with high potentials and multi-electron redox properties

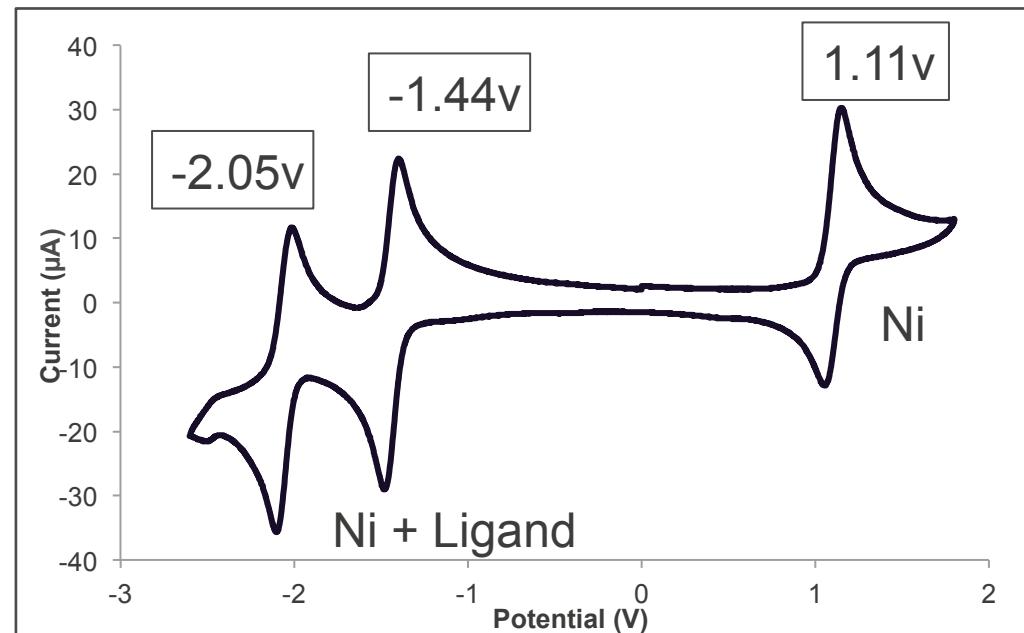


Ligand generated in good yield from inexpensive materials

Complex formation is simple and purification (crystallization) occurs in EtOH

CV in MeCN and TEABF shows greater than 3V potential is possible for the cell (ignoring overpotential and ohmics);  
2<sup>nd</sup> stored electron is 2.5v

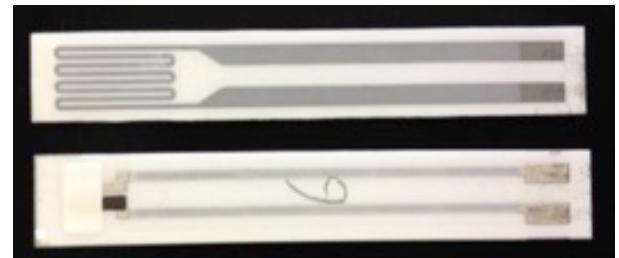
Could be used as catholyte and anolyte, eliminating need for expensive membranes



# DOE VTO-sponsored electrochemical sensor project

## Advantages:

- Significantly less complex sensor design
- Similar form factor as auto oxygen sensor
- No fugitive spacers needed / air reference channel and internal oxygen pumping cells not required
- Faster turn-on during start-up / protection against water induced thermal shock
- No pumping/diffusional channels to become blocked



# Electrochemical Sensors: $\text{NO}_x/\text{NH}_3$ Sensors for Vehicle applications

**Problem:**  $\text{NO}_x$  sensors that meet stringent vehicle requirements are not widely available:

## VDO/ NGK UniNOx Sensor

$\text{ZrO}_2$  -based multilayer sensor with 3 oxygen pumps

Complex expensive sensor is the only commercial product available today

## DOE-EERE-VT

Funding LANL to commercialize  $\text{NO}_x/\text{NH}_3$  sensors

Collaboration with ORNL National Transportation Research Center

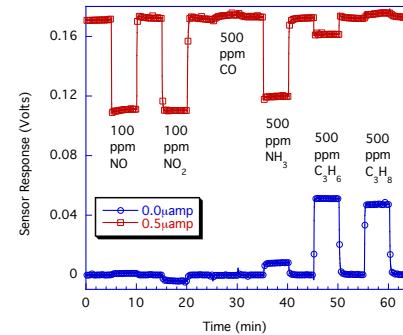
*Transform LANL patented mixed-potential technology into commercial platform*



Transform bulk LANL sensors to commercially manufacturable automotive platform.

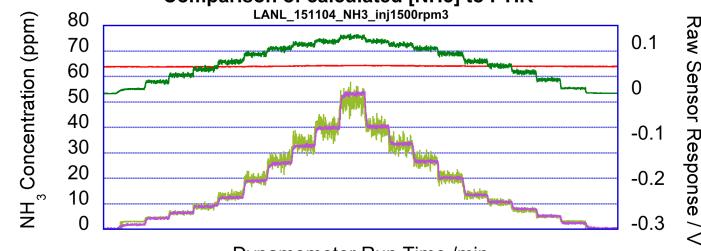


Preserving unique sensor characteristics and excellent response sensitivity/selectivity/stability



## Current Status:

- Successfully incorporated LANL technology into ESL manufactured commercial platform and demonstrated in an engine dynamometer.
- Continental is evaluating LANL technology with assistance from DOE Tech to Market funding (matching).



*Sensor tracks FTIR  $\text{NH}_3$  concentration*

# Exploiting cross-sensitivity and stability of mixed-potential sensors: Design principles of olfaction used for decomposition of gas mixtures

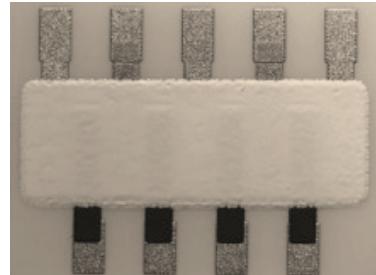
In biological systems, cross-selective receptors work together to decode odors and odor mixtures.

Use Bayesian Inference models from multiple sensors to quantitatively de-convolute the concentration of each species of interest; Collaboration with Morozov Group, Rutgers University



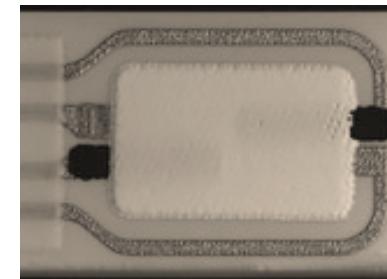
## Flexible device to refine models

Each electrode is at a different temperature than the adjacent



## Stick sensor to test models in engine exhaust

4 electrode, fixed T device for engine testing at NRTC



## Results

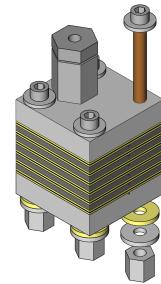
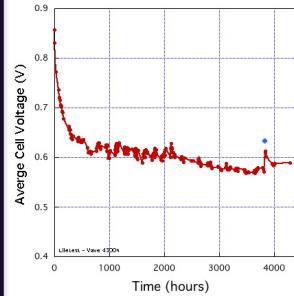
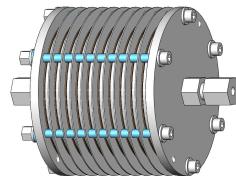
- Model estimation of  $[C_3H_8]$ ,  $[NH_3]$ , and  $[NO_2]$  with respect to  $[NO]$  in an extensive set of mixtures with max error 14.0% and average error of 1.8%.
- Predictions of each gas in 2- and 3-gas mixtures with 3.2% average error.
- Model can identify minimum number of sensors required for the application.
- Model is capable of predicting gas concentrations in complex mixtures with high accuracy.

# MPA-11 Impact on Future Stockpile Needs



Weapons Systems  
Engineering and  
Surety

Materials and technologies for  
stockpile modernization



## <sup>242</sup>Pu Production to Support Pu Science and Engineering

$\text{NpO}_2^{2+} \rightarrow \text{NpO}_3 \cdot \text{H}_2\text{O}$

$\text{Np}^{4+} \rightarrow \text{Np}^{3+}$

$\text{Np}^{3+} \rightarrow \text{Np}^{2+}$

$\text{Np}^{2+} \rightarrow \text{Np}^{0}$

$\text{Np}^{3+} + 3\text{e}^- \rightarrow \text{Np}^{2+}$

$\text{Np}^{2+} + 3\text{e}^- \rightarrow \text{Np}^0$

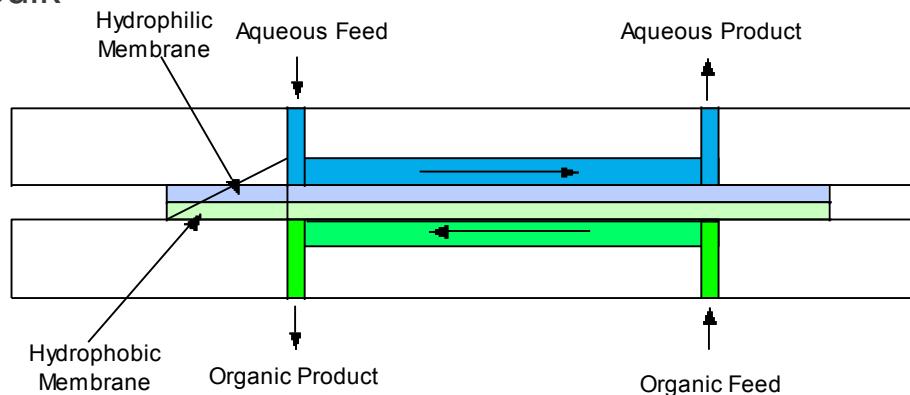
$\text{Np}^{3+} + 3\text{e}^- \rightarrow \text{Np}^{2+}$

$\text{Np}^{2+} + 3\text{e}^- \rightarrow \text{Np}^0$

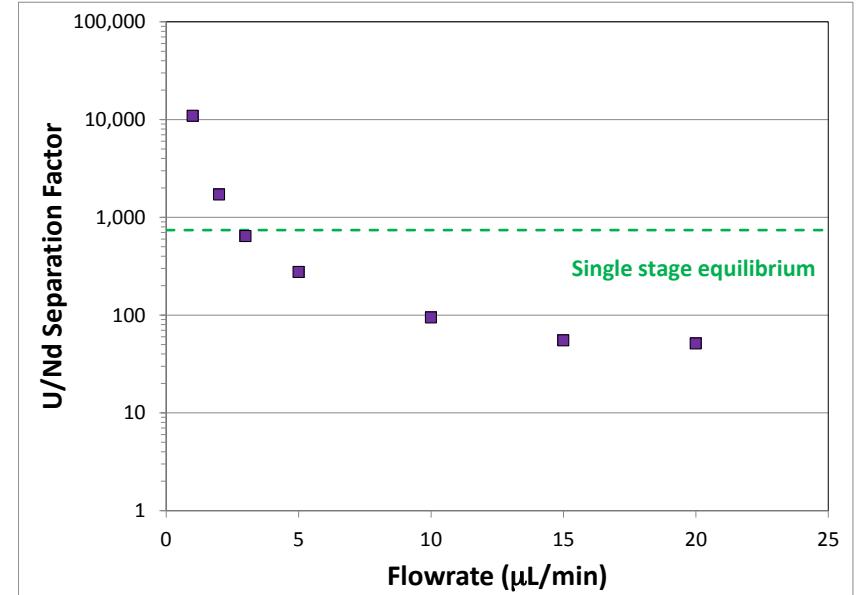
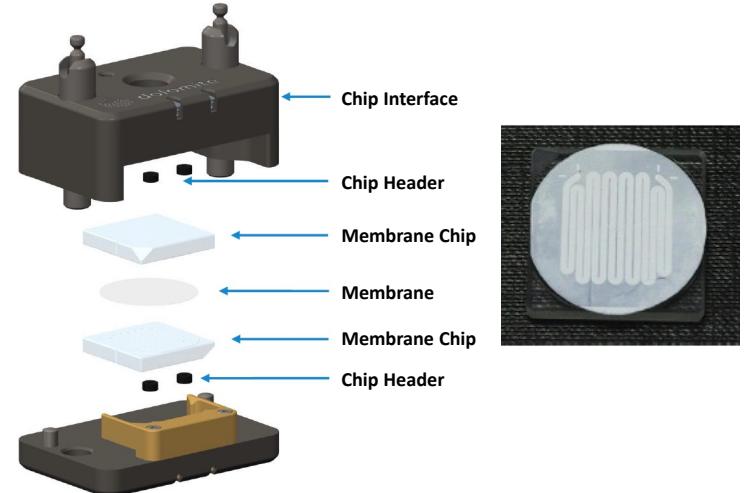
## Materials Development and Manufacturing

# Advanced Actinide Separations

- Microfluidics for actinide recovery and recycle
  - Liquid-liquid extraction processes
  - Provides rapid mass transfer and heat transfer rates
  - Reduction in-process volumes,
    - decreases criticality concerns
    - Membranes provide excellent phase separation
  - Scalability: Recycle of exotic isotopes ( $^{242}\text{Pu}$ ), up to bulk



Separation factors of U(IV) from Nd (III) of more than 50 - 10,000 in a single contact!



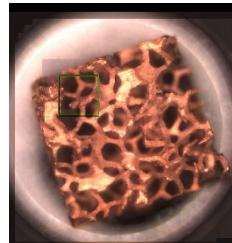
# Materials Synthesis for National Security Needs

- **Thin film synthesis and chemistry**

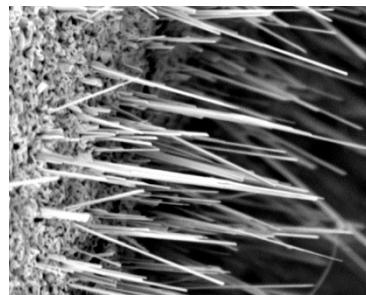
- PAD films of  $\text{PuO}_2$ ,  $\text{UO}_2$ , and  $\text{U}_3\text{O}_8$  films and associated chemistry for LEP, Surety, LDRD, AWE, DHS, NCT, and LLNL projects
- Films provide low activity source of important nuclear materials for compatibility and reactivity studies in low level radiological facilities

- **Development of fluorinated oxide nanowires on porous copper for safe storage of nuclear materials**

- Goals: hydrophobic, gas permeable, heat resistant filters for Savy cans
- Approach: grow  $\text{CuO}$  nanowires on porous copper followed by fluorination
- Collaboration with Nuclear Process Infrastructure Division (TA-55)



600°C



Porous Cu

75nm diameter  
 $\text{CuO}$  nanowires

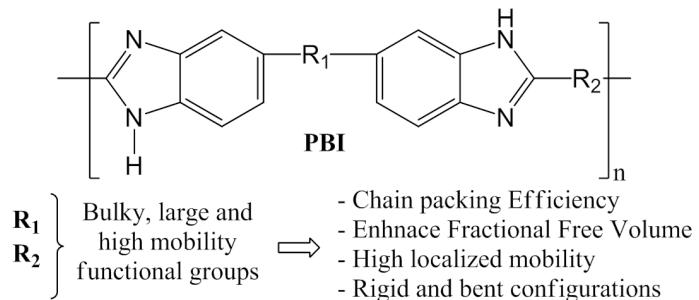
Fluorinating agent



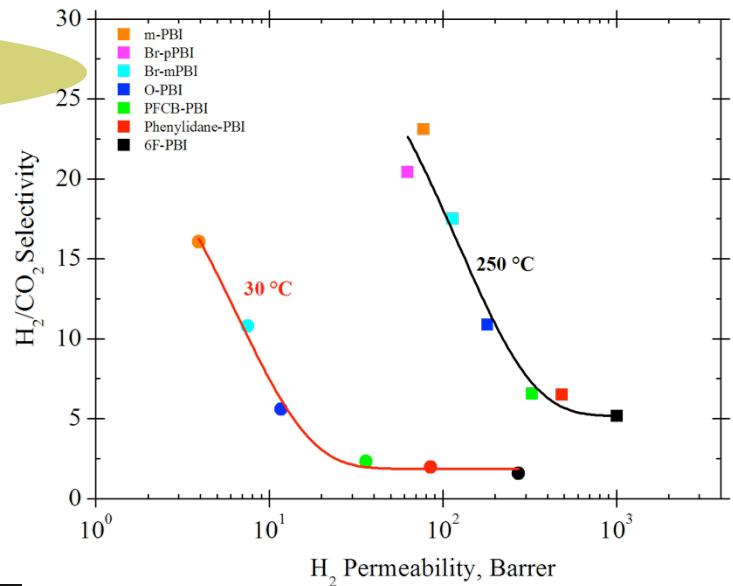
Contact angle > 120°  
on porous material

# Barrier/Membrane Material Design & Deployment: Carbon Capture and Sequestration

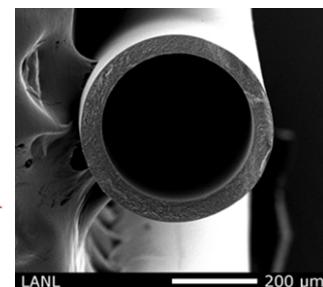
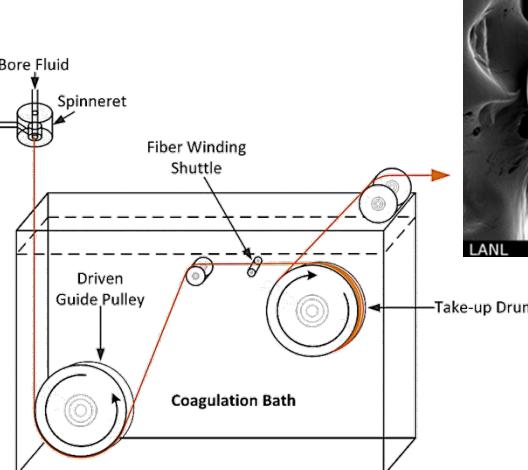
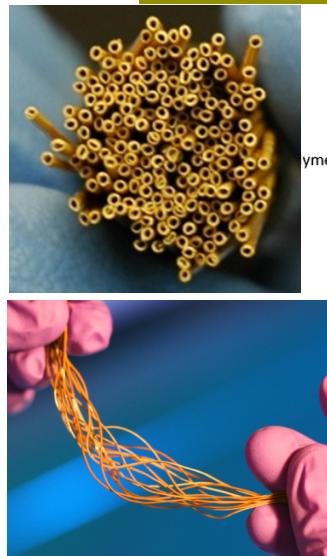
## Material Design for Controlling Small Molecule Transport and Separations



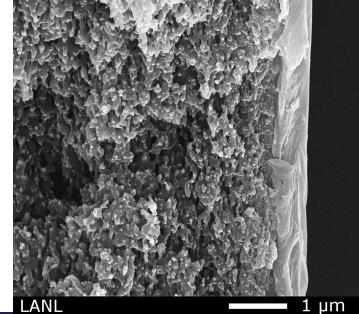
## Material Design to Tailor Structure-Property Relationships



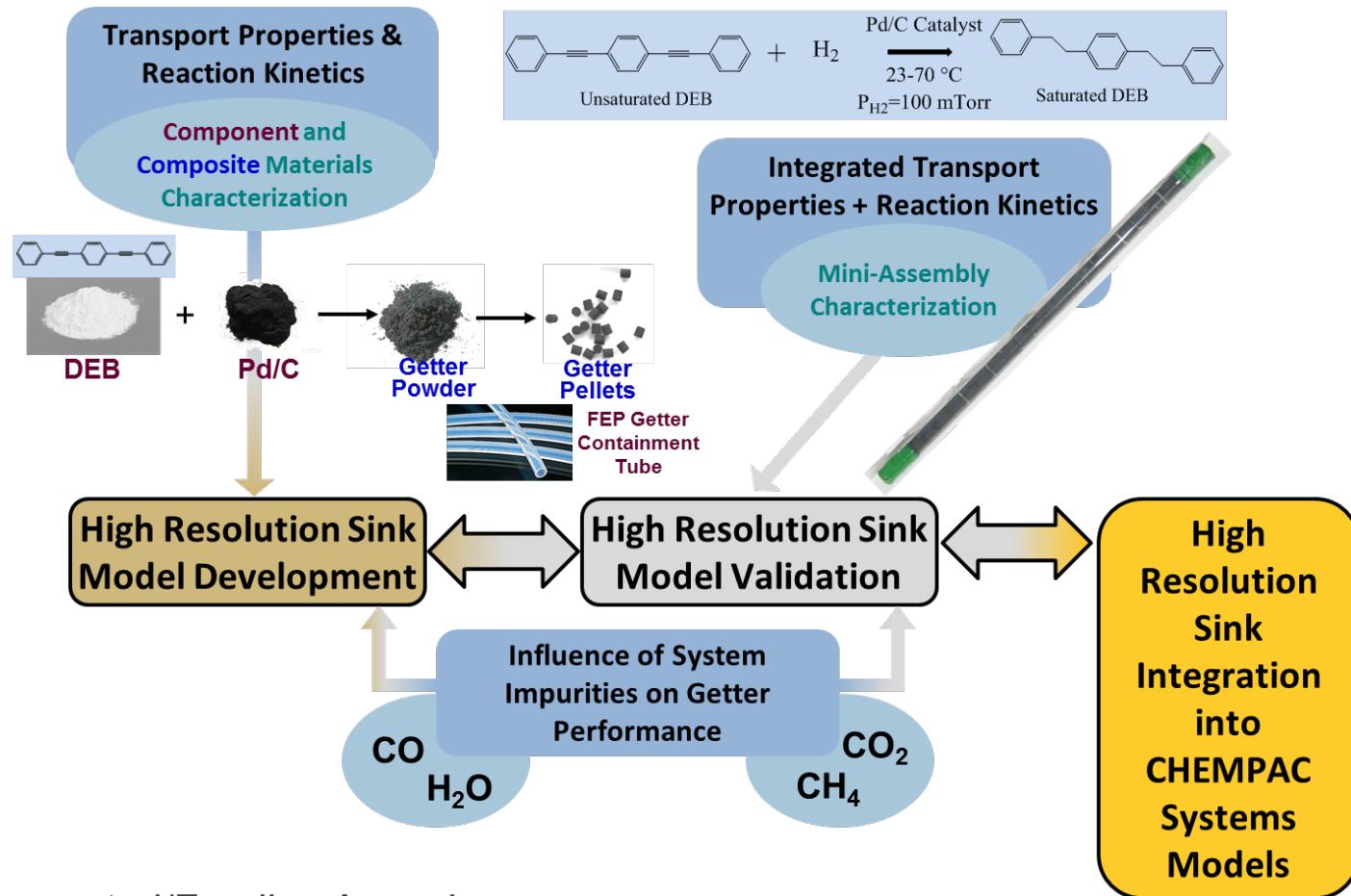
## Materials Processing to Develop an Industrially Scalable Product Platform



## Morphology Optimization to Develop High Performance Composite Structures



# Barrier/Getter/Adhesive/Structural Materials and Platform Characterization & Development for National Security Applications



## Programs Supported/Funding Agencies:

B61-12 Life Extension Program  
W76 Weapons Program  
B61 Legacy Program  
Campaign 8 - Enhanced Surveillance Campaign (ESC)



# Acoustic Sensors Team: Radial Vibration Modes of a Piezo-disk

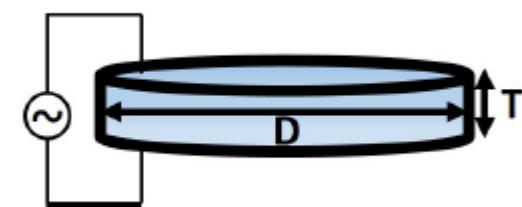
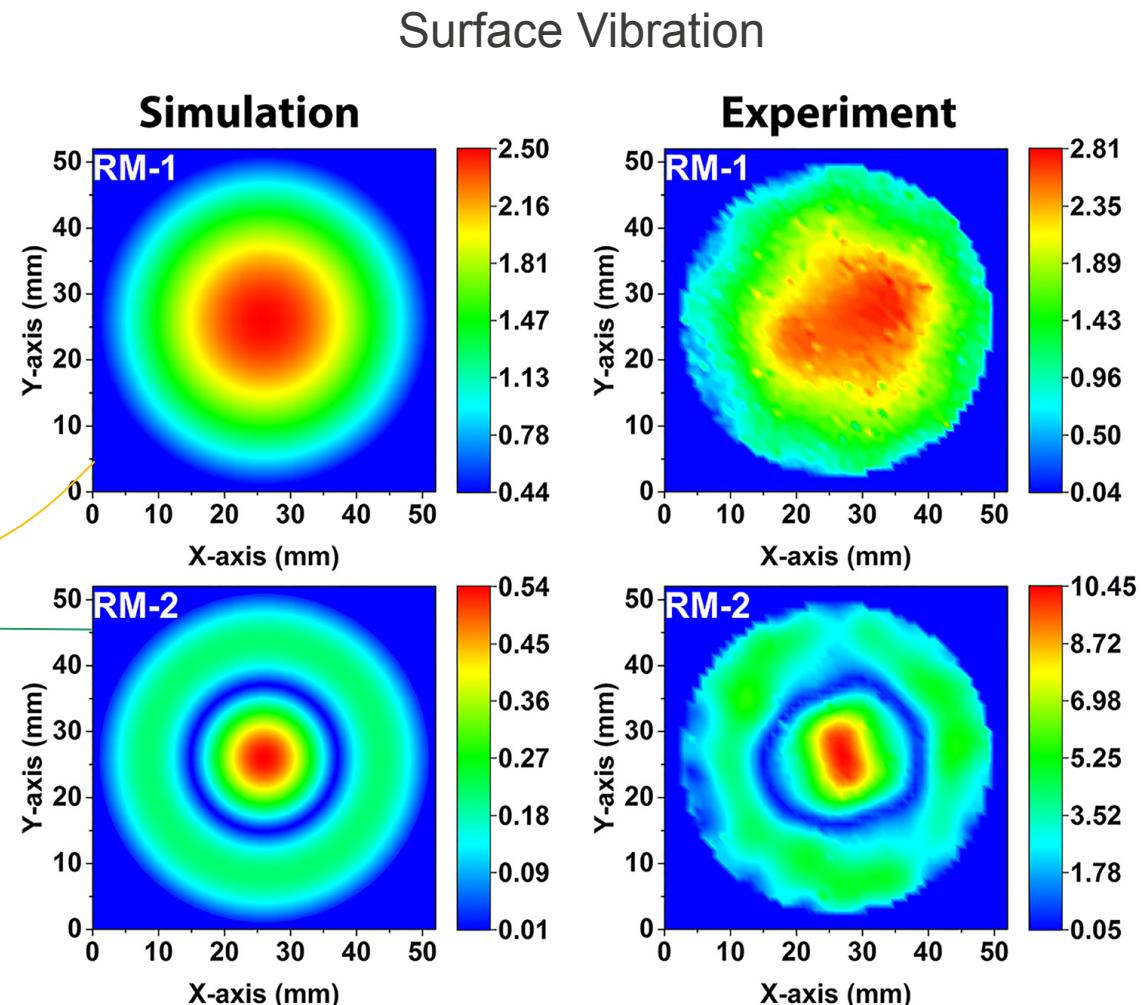
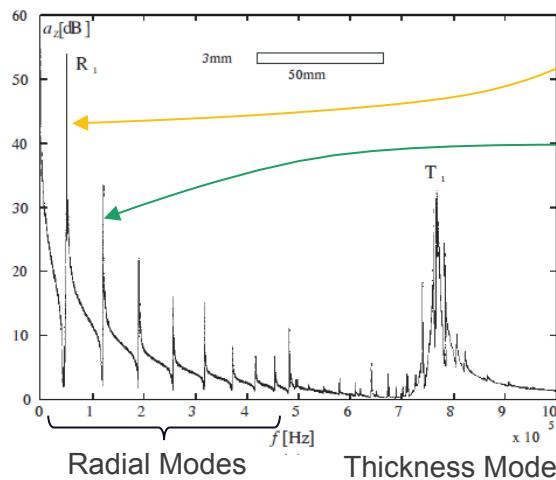


Figure 1: Schematic of a piezoelectric disc transducer.

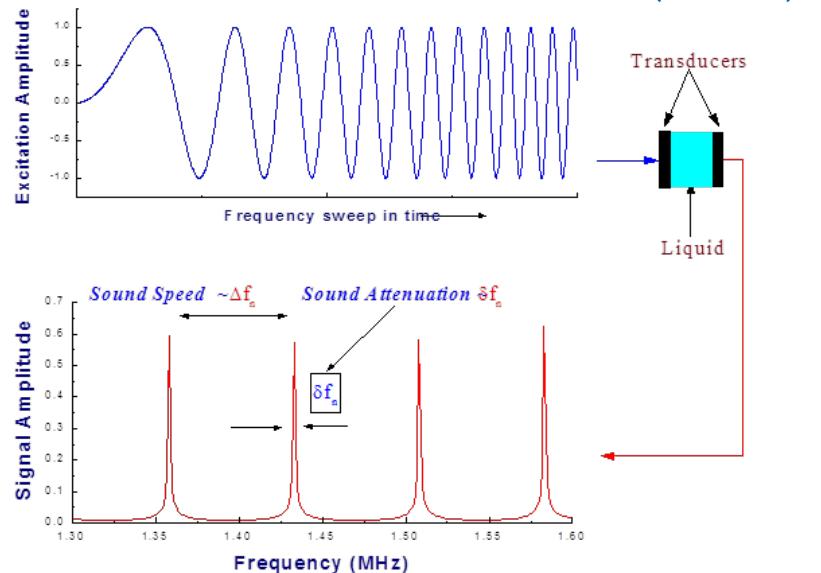


Much lower frequencies than the typical thickness mode resonance of a disk

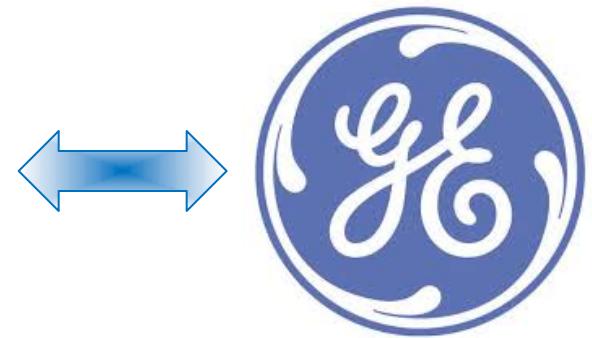
# Acoustic Sensor Technologies for Multiphase Metering

## POC: Dipen Sinha

SFAI: Swept Frequency Acoustic Interferometry



Derive multiple physical properties of fluids from a single, noninvasive measurement



Safire Meter

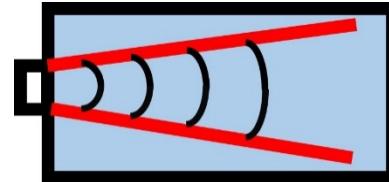


# Low-frequency Ultrasonic Bessel-like Collimated Beam Generation

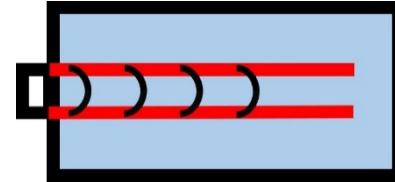
## Why Collimated beam?

Conventional approach: Bessel-like patterned piezo-disc to form Bessel pattern on a transducer (**fabrication intensive technique**)

Non collimated



Collimated

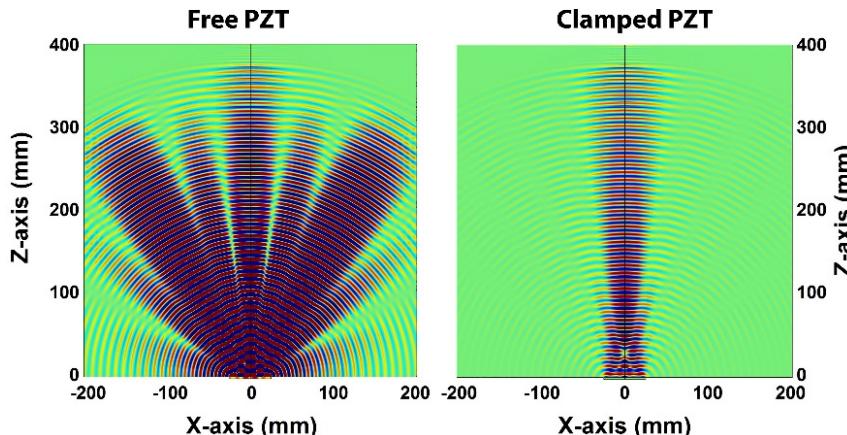


## Applications

- ✓ Biomedical Imaging (through skull)
- ✓ Imaging concrete for bore-hole integrity applications
- ✓ Characterizing highly attenuating materials

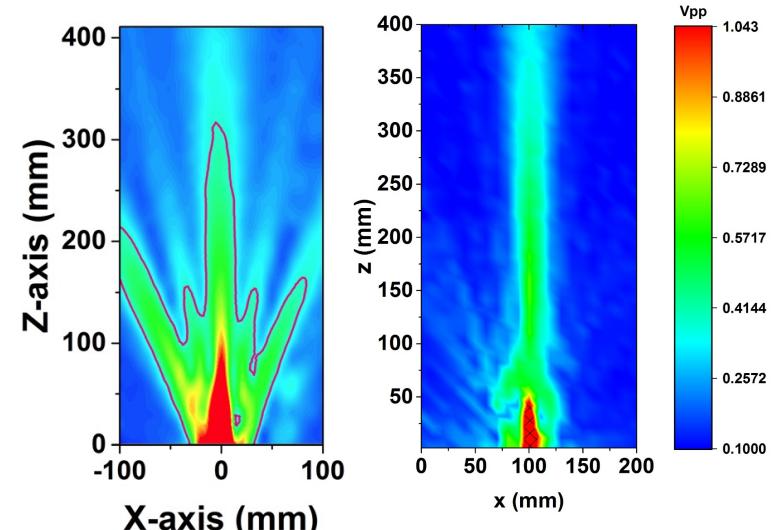
Our Approach: uses **natural vibration pattern** of radial modes of piezoelectric disc transducer

## Simulation



## Experiment in water

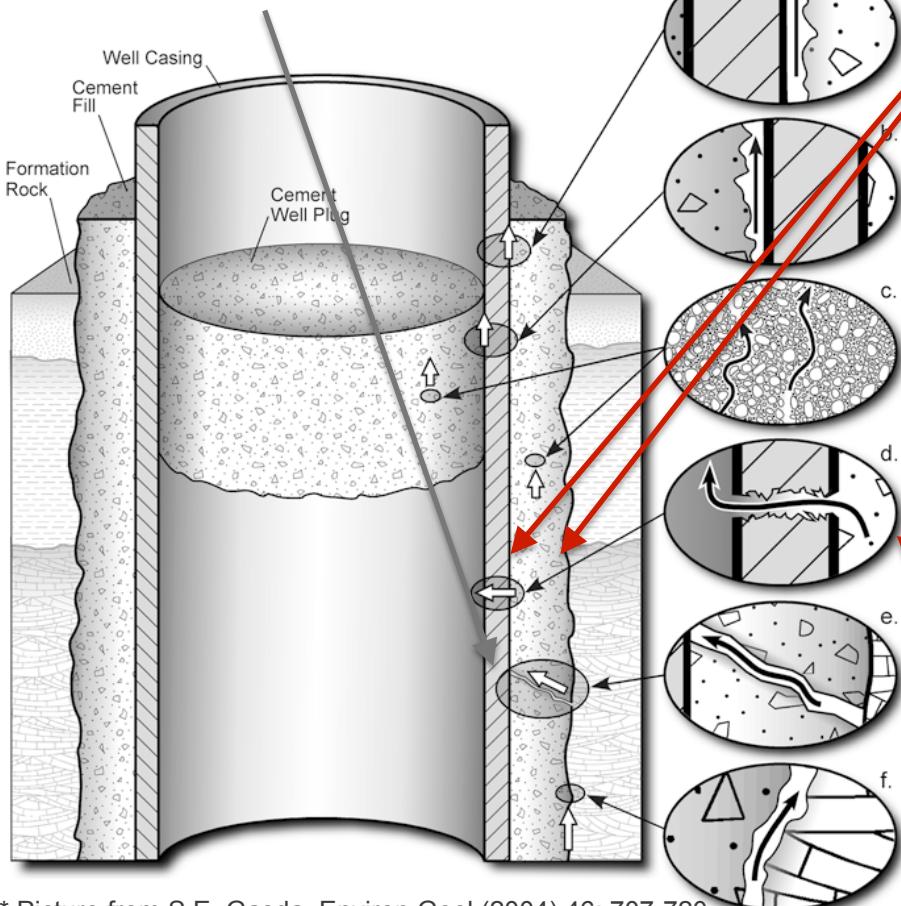
Free PZT      Clamped PZT



# SubTer Program: High-Resolution 3D Acoustic Borehole Integrity Monitoring

**The Problem:** *Defects/fracture detection beyond casing with high resolution.*

Existing ultrasonic tools work well for casing inspection



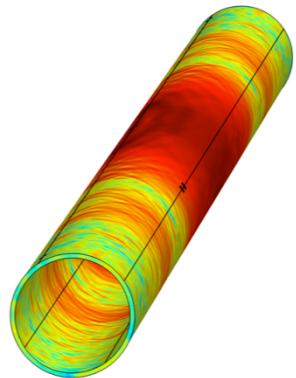
We plan to extend applicability to: (1) casing-cement interface, (2) cement-formation interface, and (3) out in the formation (up to ~ 3 meters).

*Comparison of existing techniques and the present approach*

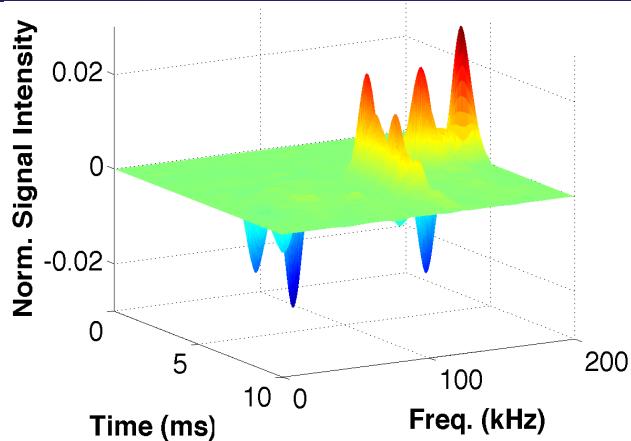
| Method   | Frequency (kHz) | Range (m)  | Resolution (mm) |
|--|-----------------|------------|-----------------|
| Standard borehole sonic probe, e.g. BARS (Borehole Acoustic Reflection Survey) | 0.3-8           | 15         | ~ 300           |
| <b>Present approach</b>  | <b>10-150</b>   | <b>~ 3</b> | <b>~ 5</b>      |
| Ultrasonic probe, e.g. UBI (Ultrasonic Borehole Imager)                        | >250            | casing     | 4-5             |

\* Picture from S.E. Gasda, Environ Geol (2004) 46: 707-720

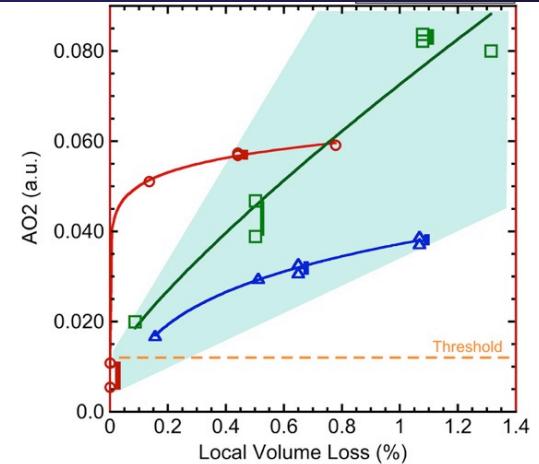
# Acoustic Sensors: Acoustic Large Area Monitoring (ALArM) of Extended Mechanical Structures



Simulated Multi-Mode Acoustic Signal Propagation in pipe



Example of Defect Detection with Freq-Timemixed-domain signal analysis

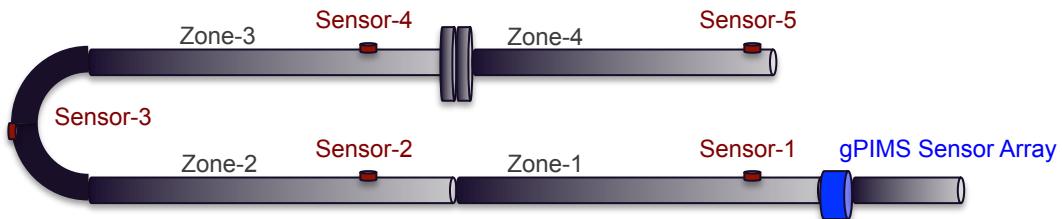


Detection of Local Vol. Loss of an extended structure, with sensitivity close to 0.1%



## Field Tests in Houston, TX (2015-2016):

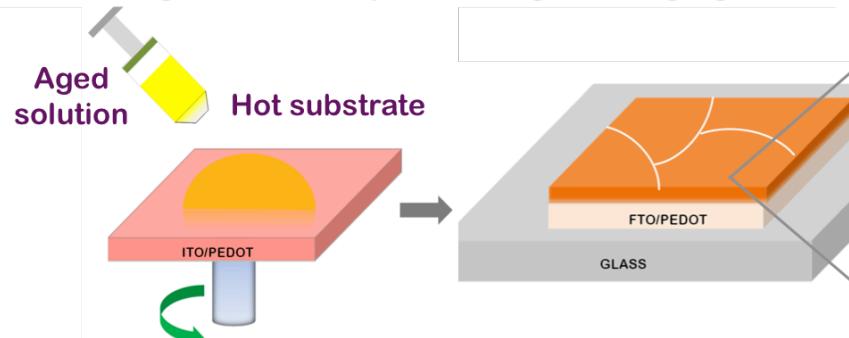
- 105'-long, 8"-Dia Sch-40 Steel pipe with 2 elbows, 8 welded joints, and 1 pair of flanges, held on 10 stands,
- ALArM system (5 single sensors, spaced 25' over 100').



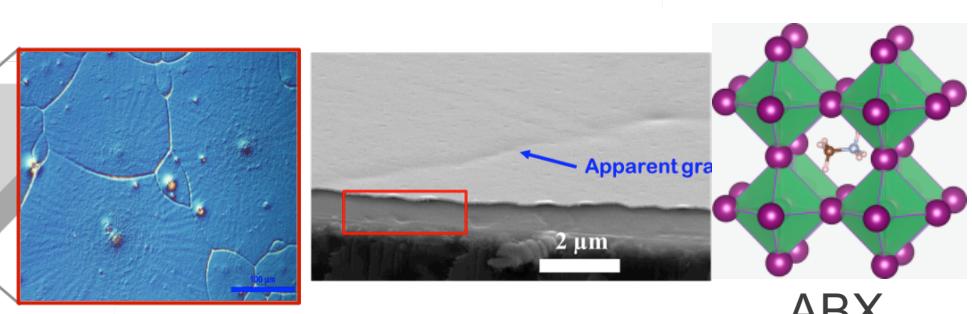
# Hybrid Perovskites for Photovoltaic and Detector Applications

## POC: Aditya Mohite, MPA-11

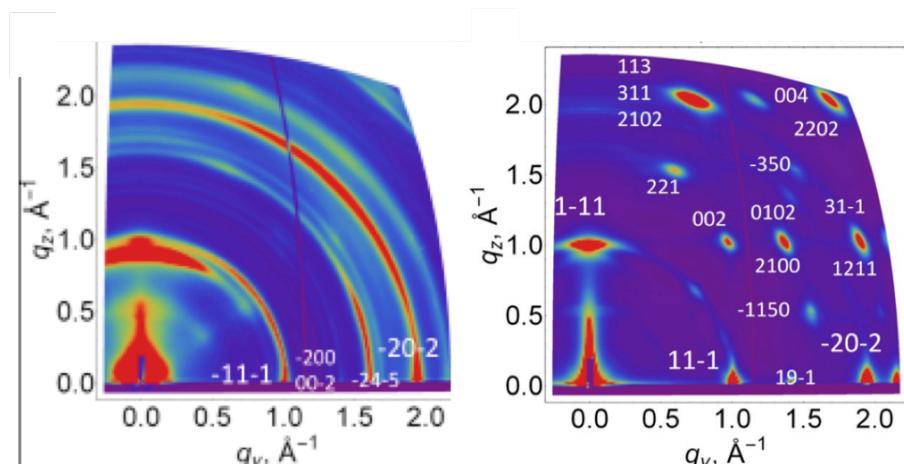
A Hot-casting method & optical image of large grains



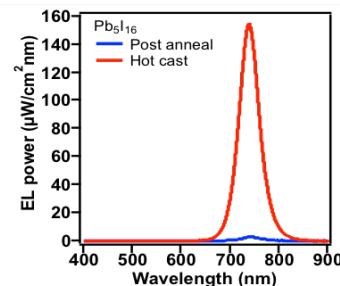
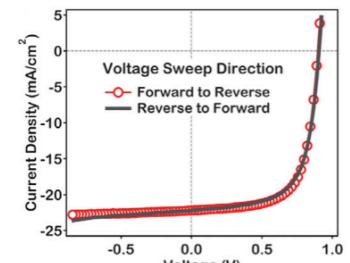
B Cross-sectional SEM shows bulk film



C Comparison of GIWAX of as-cast and hot-cast films



D High performance devices



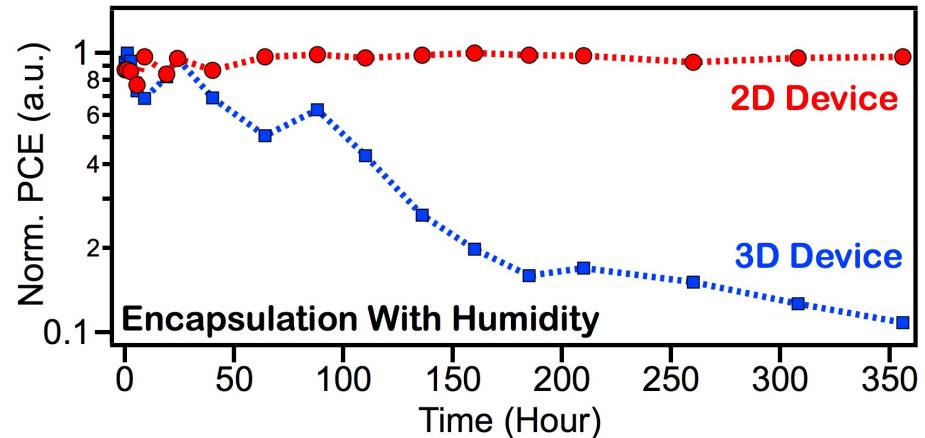
First in the world to grow large grain-size thin-films of hybrid perovskites:  
Power conv. efficiency ~ 18%; Nie et al Science et al (2015)

# 2D vs 3D hybrid perovskite optoelectronics

*High efficiency optoelectronic devices using 2D perovskites*

>12% efficient solar cells

Intrinsically stable >2000hrs operating solar cells, Tsai/Nie, *Nature* 2016



3D:  $\text{MAPbI}_3$     2D:  $(\text{BA})_2(\text{MAPb}_4\text{I}_{13})$

*Efficient intrinsic process of exciton dissociation for  $n \geq 3$ , Blancon, *Science* 2017*

Layer edge states drive exciton dissociation  
Results in longer lived free carriers leading to stronger PL

