

Progress in Overcoming Materials Challenges with sCO₂ RCBC's

SAND2016-10554C

**Matthew Walker, Alan Kruizenga,
Elizabeth Withey**

**Sandia National Laboratories
Livermore, CA**



Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000. SAND2014-15095PE

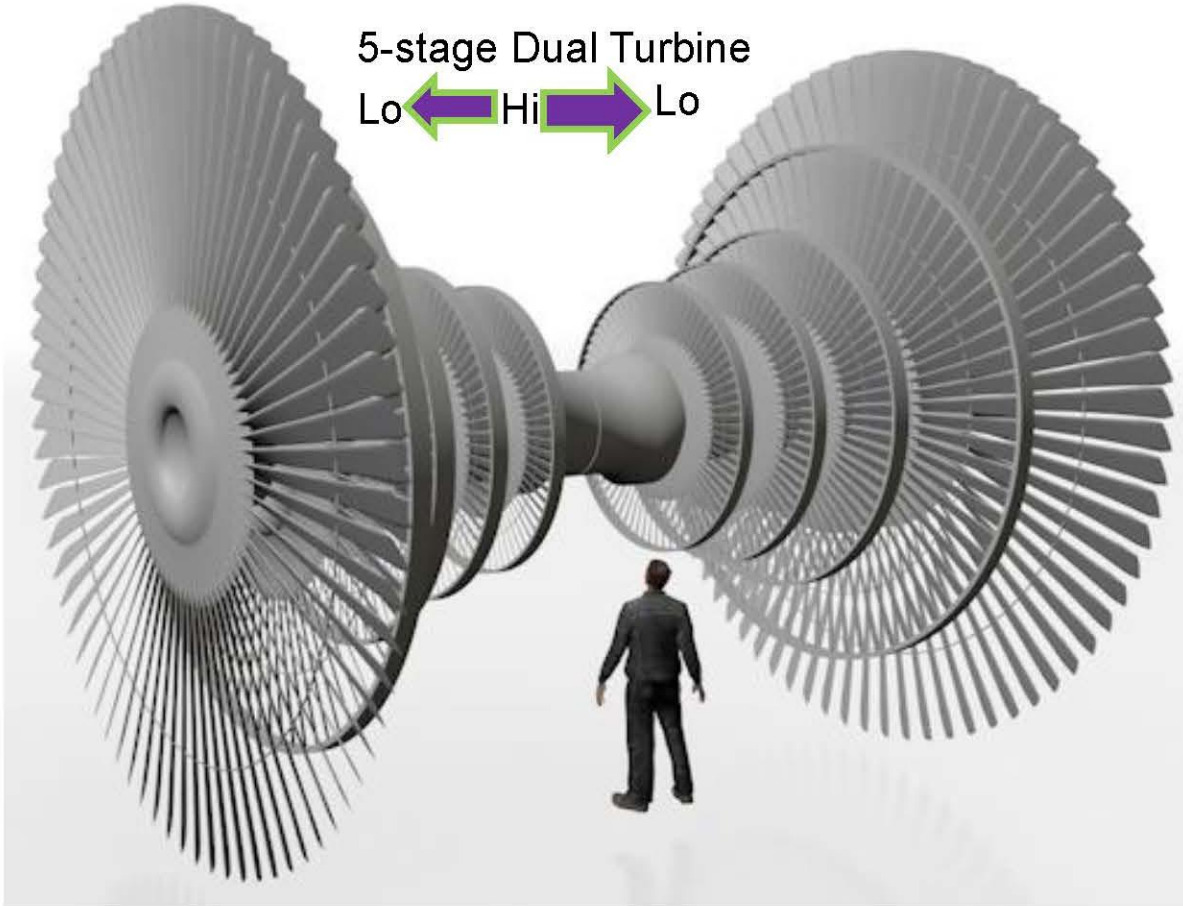
MS&T16

OCTOBER 23 – 27, 2016
SALT PALACE CONVENTION CENTER | SALT LAKE CITY, UTAH USA



Transformational Energy System

5-stage Dual Turbine
Lo ← Hi → Lo



20 meter Steam Turbine (300 MWe)
(Rankine Cycle)

Comparison

- Rankine efficiency is 33%
- Supercritical CO₂ (sCO₂) potential to surpass 40% efficiency
- Greatly reduced cost for sCO₂ compared to the cost of conventional steam Rankine cycle
- sCO₂ compact turbo machinery is easily scalable



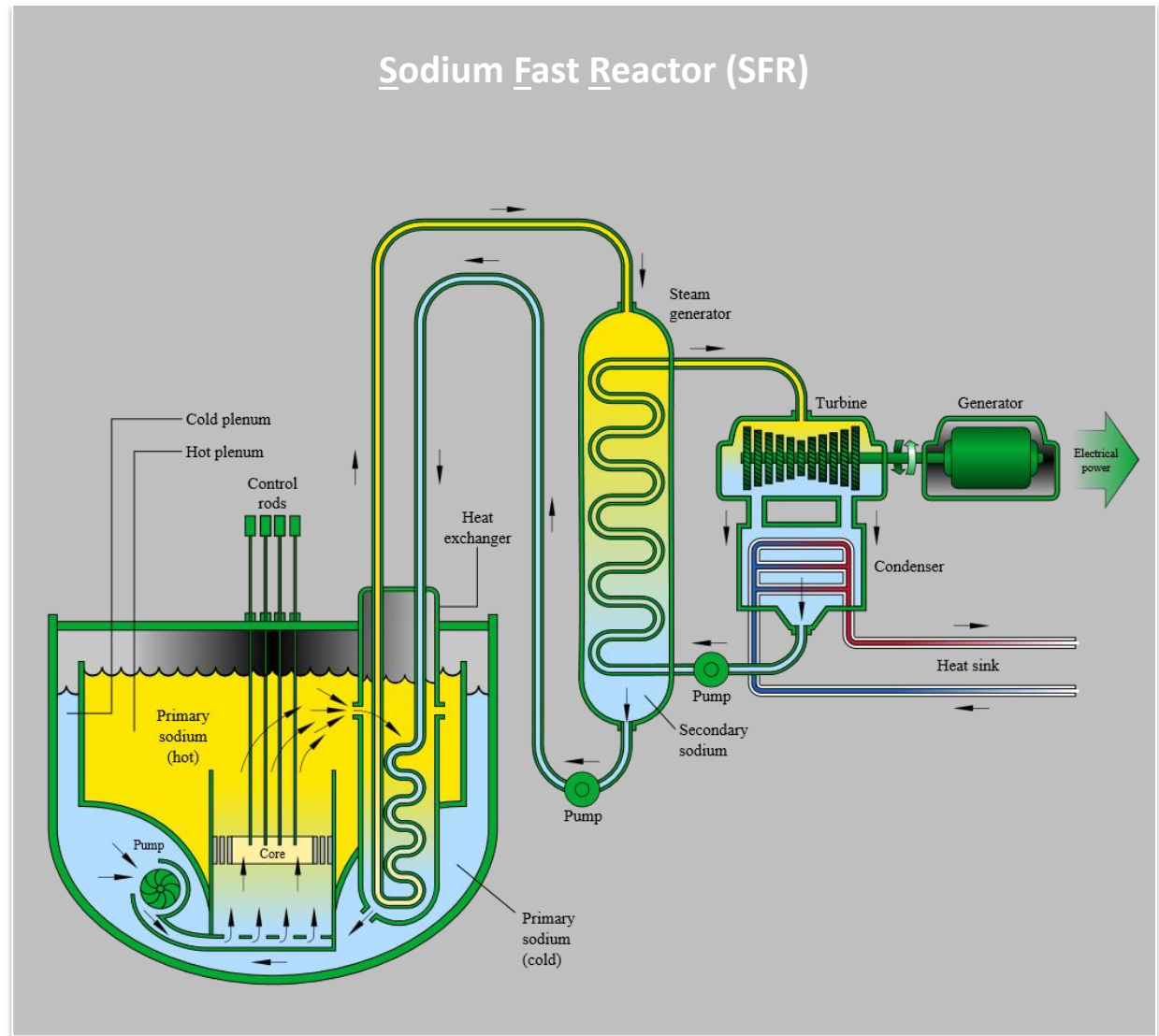
3-stage Single Turbine

Hi → Lo

1 meter sCO₂ (300 MWe)
(Brayton Cycle)



Origins in SFR Development



Sandia is a Leader in sCO₂ System Development

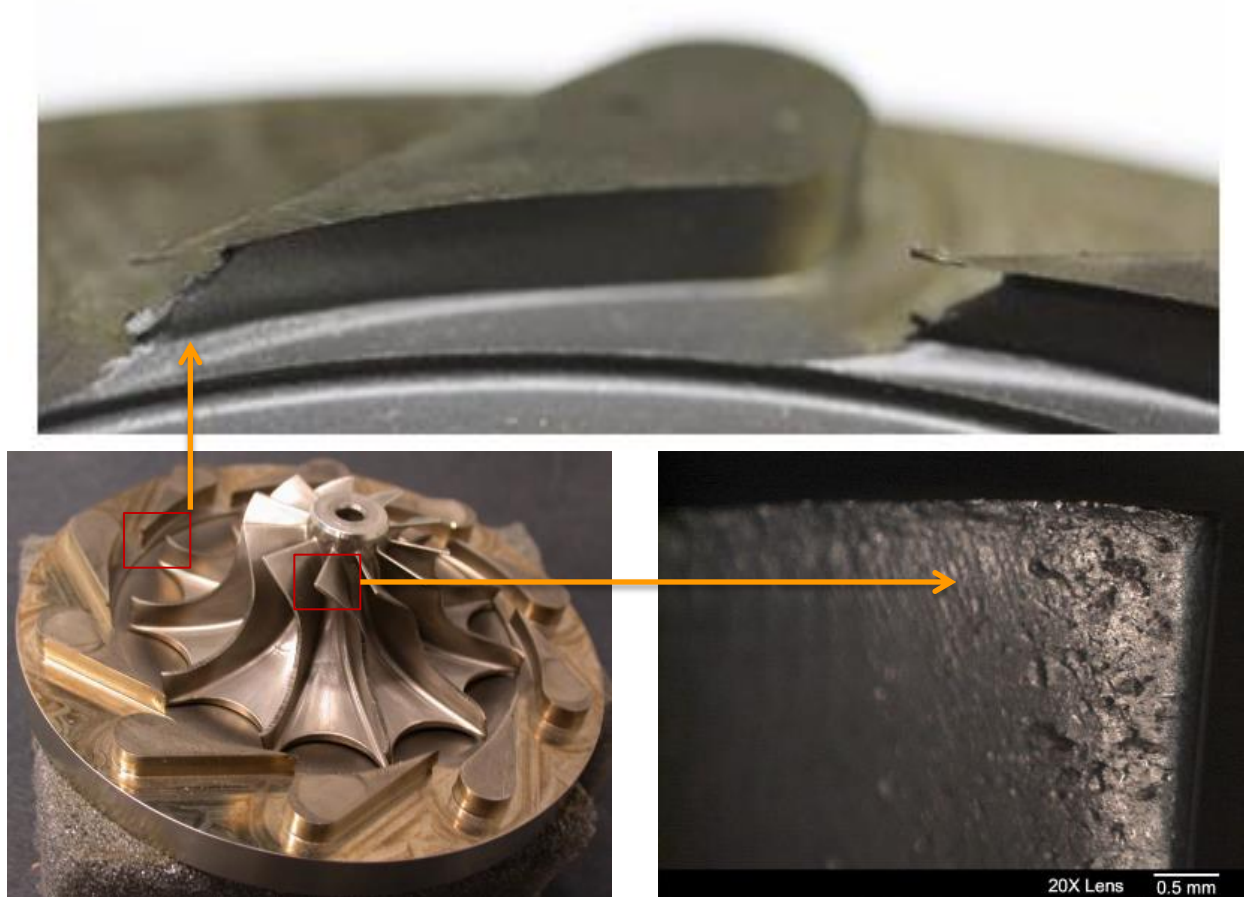
- **Component Development with Manufacturers**
- **Component Testing Platforms**
- **System Testing and Integration**
- **System Economics Modelling**
- **Materials Development**

250 kW sCO₂ RCBC



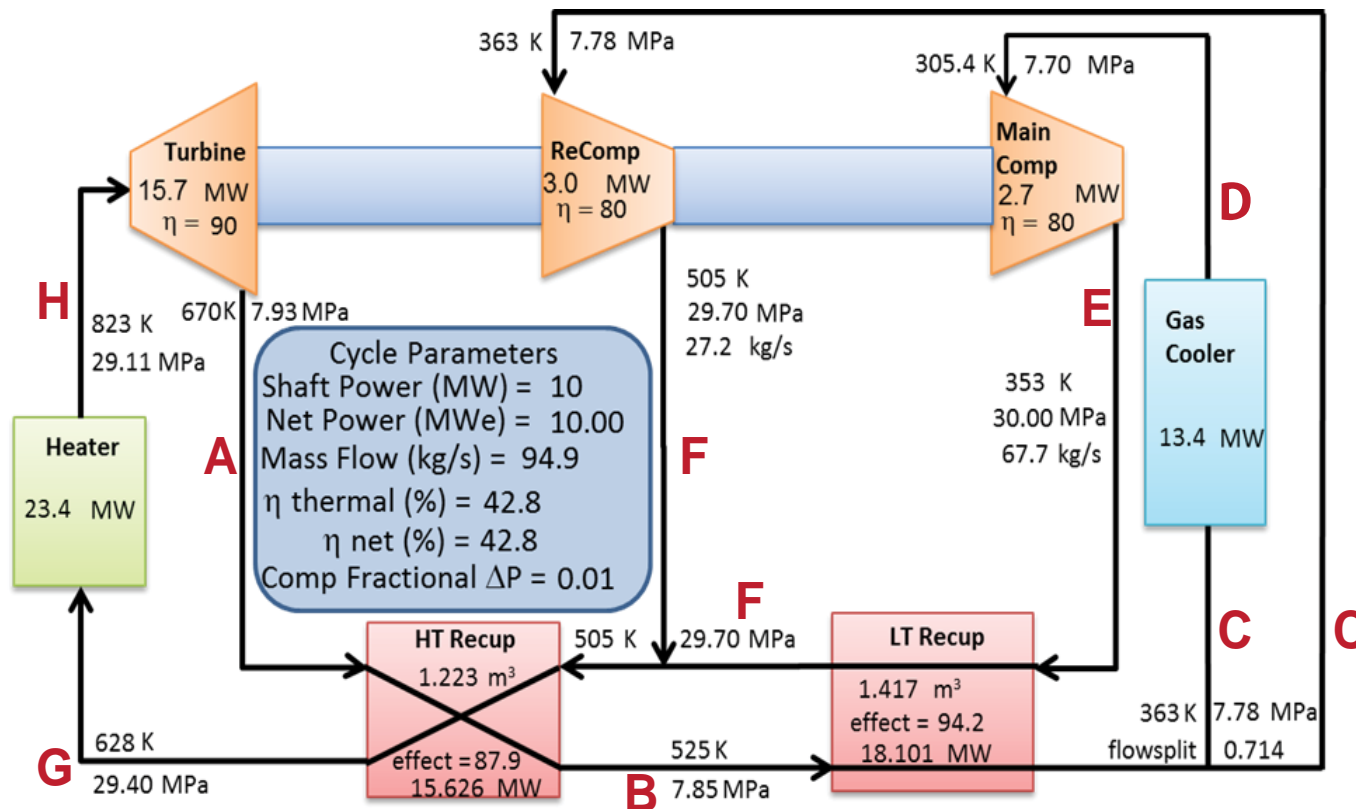
Materials Issues for sCO₂ Systems

Understanding and Resolving Turbine Degradation



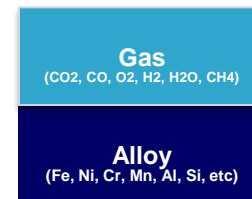
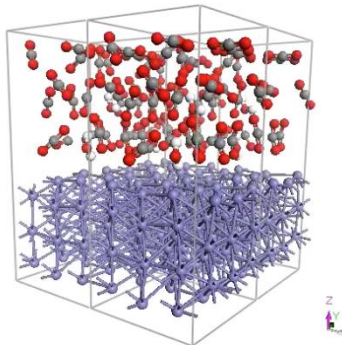
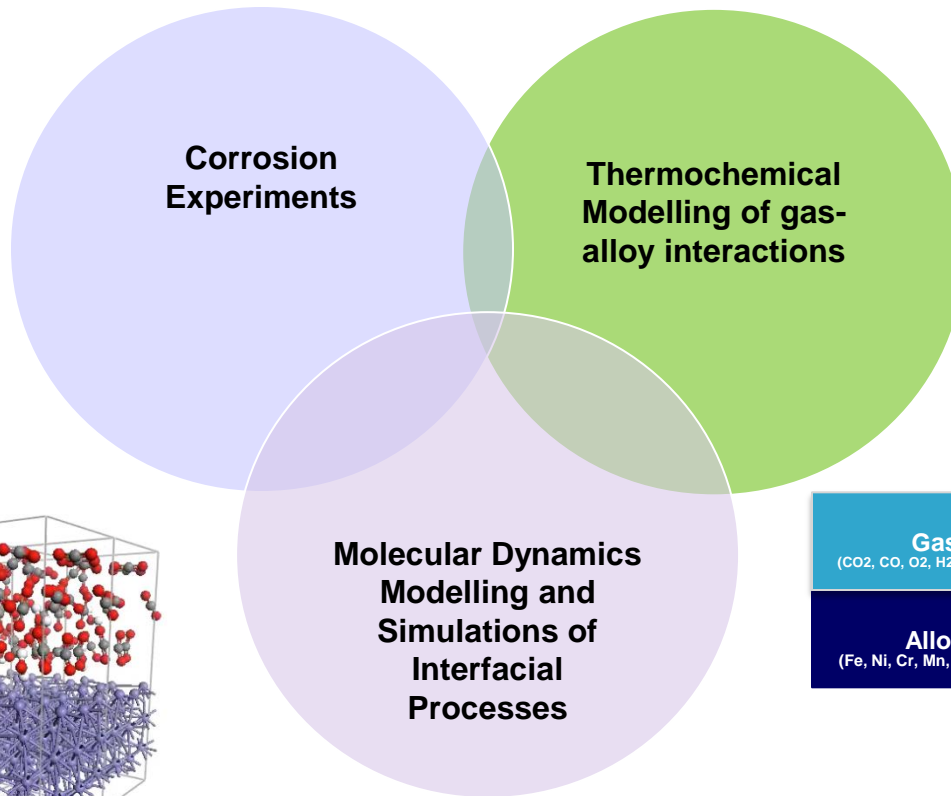
Materials Issues for sCO₂ Systems

Identifying Material Options for 550°C, 10 MW System Components



Materials Issues for sCO₂ Systems

Understanding the Influence of Gas Chemistry on Alloy Corrosion



← Equilibrium gas chemistry, $f(T, P)$

← Oxide formation, $f(T, P)$

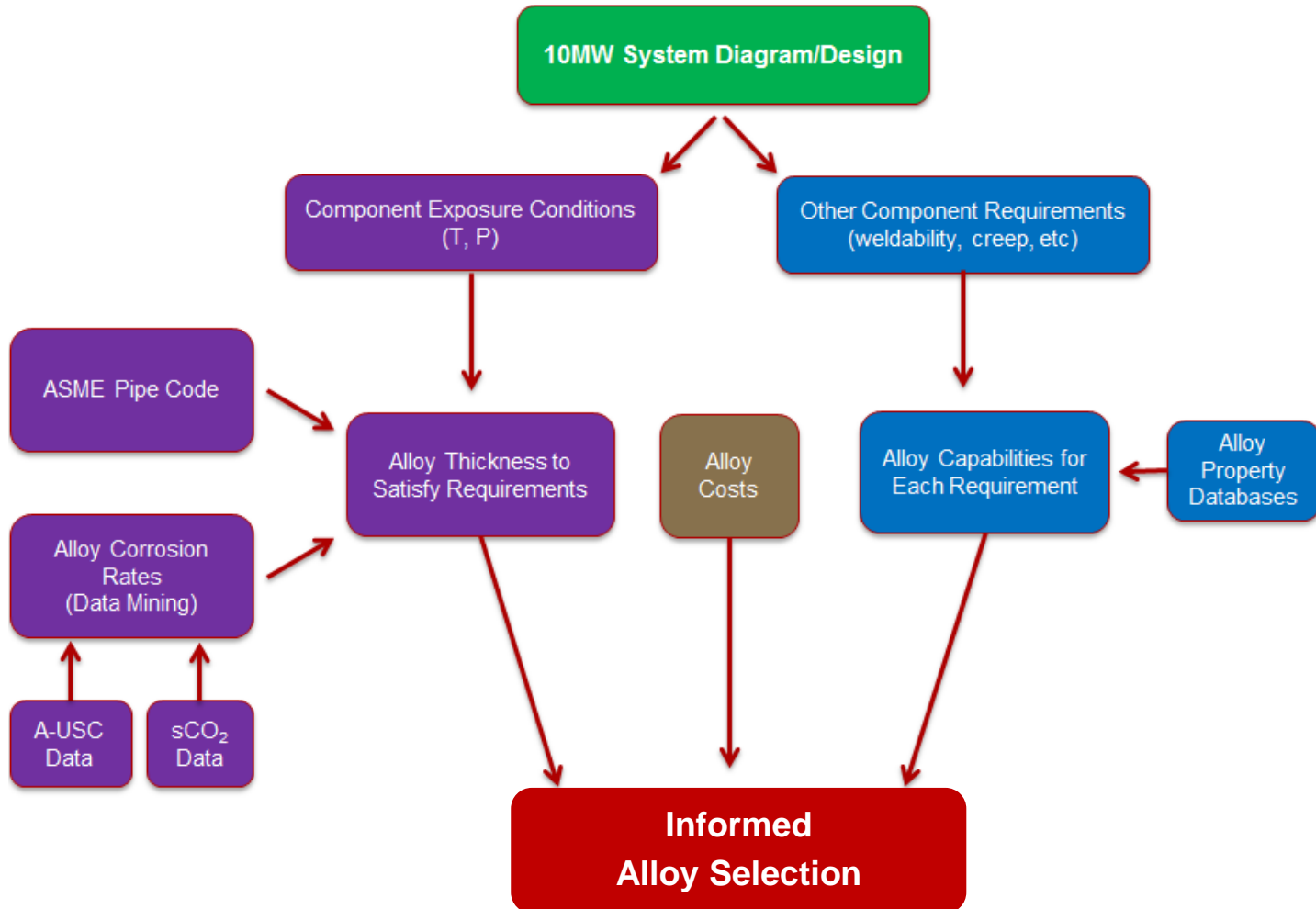


Materials Issues for sCO₂ Systems

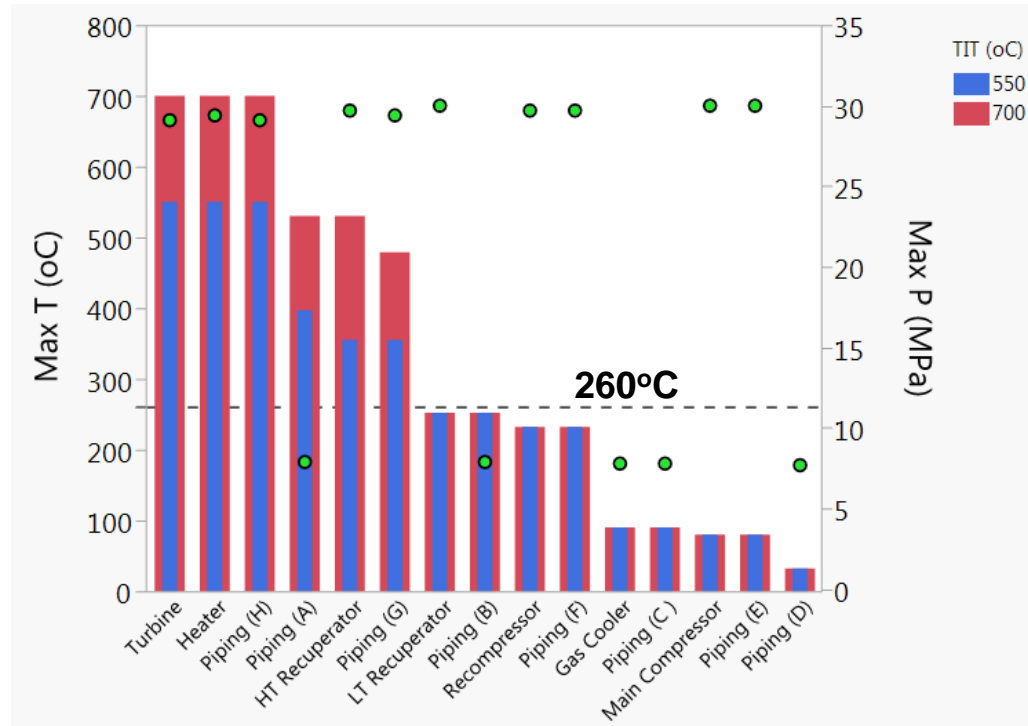
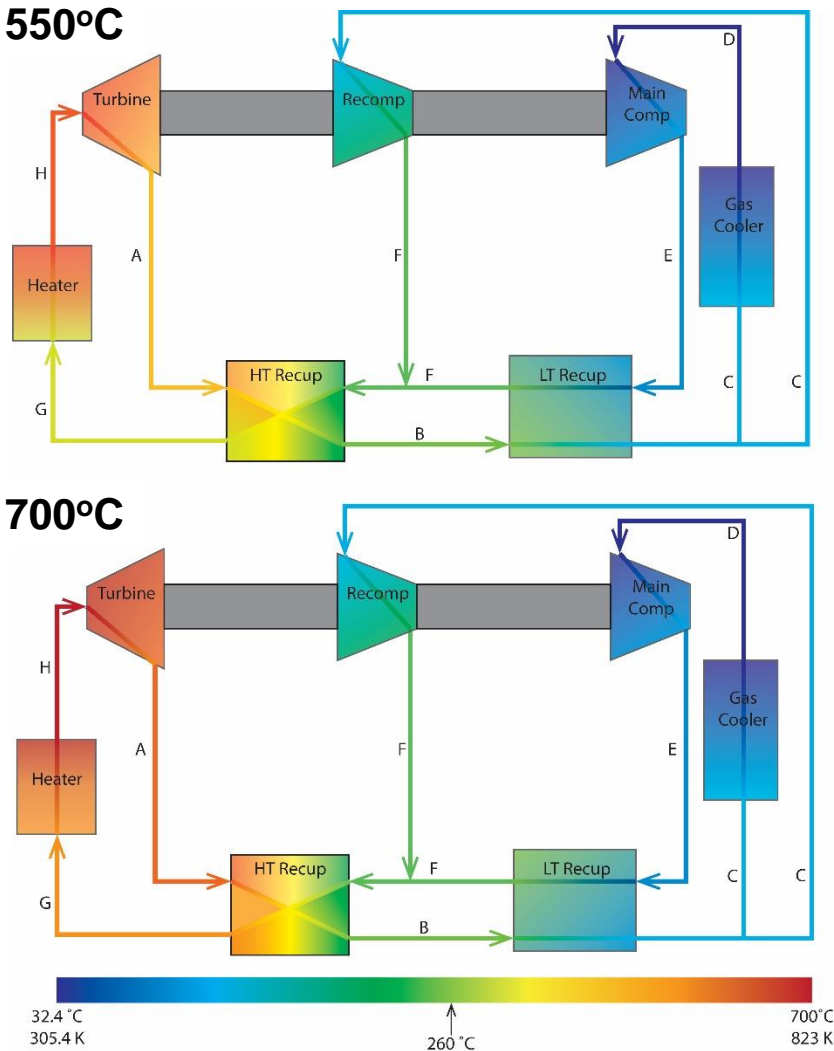
- Identifying gas foil bearing behavior in CO₂
- Understanding and resolving turbine degradation
- Identifying materials options for 10 MW RCBC system components
- Understanding the influence of gas chemistry on alloy corrosion



Component Materials Selection



Component Exposure Conditions



Identifying Candidate Alloys

Alloy	Category	Code Qualifications				Raw Material Cost (\$/Lb)	
		(Y, N)	ASME B31.1		BPVC		
			Category	Max T (°C)	Category		Max T (°C)
API 5L	Carbon Steel	Y	Seamless Pipe and Tube (X65)	427		\$0.91	
Grade 22	Ferritic	Y	Seamless Pipe and Tube (P22, T22, FP22)	593		\$1.04	
Grade 91	Ferritic	Y	Seamless Pipe and Tube (P91, T91)	649		\$1.30	
410	Ferritic	Y	Seamless Pipe and Tube (TP410)	371		\$1.23	
E-Brite	Ferritic	Y	Seamless Pipe and Tube (TPXM-27)	343		\$1.75	
304H	Austenitic	Y	Seamless Pipe and Tube (TP304H)	816		\$1.75	
310S	Austenitic	N				\$2.37	
310H	Austenitic	Y	Seamless Pipe and Tube (TP310H)	816		\$2.37	
316L	Austenitic	Y	Seamless Pipe and Tube (TP316L)	816		\$1.83	
316H	Austenitic	Y	Seamless Pipe and Tube (TP316H)	816		\$1.83	
316FR	Austenitic	N				\$1.83	
316LN	Austenitic	N				\$1.83	
316	Austenitic	Y	Seamless Pipe and Tube (TP316)	649		\$1.83	
347H	Austenitic	Y	Seamless Pipe and Tube (TP347H)	816		\$1.82	
347HFG	Austenitic	N				\$1.82	
AL-6XN	Super-Austenitic	Y	Seamless Pipe and Tube (N08367)	427		\$2.77	
800H	Super-Austenitic	Y	Seamless Pipe and Tube (N08367)	816		\$2.64	
600	Ni-Base Chromia-Forming	Y	Seamless Pipe and Tube (N06600)	649		\$4.18	
617	Ni-Base Chromia-Forming	Y	Seamless Pipe and Tube (N06617)	816		\$5.38	
625	Ni-Base Chromia-Forming	Y	Seamless Pipe and Tube (N06625)	649		\$5.60	
690	Ni-Base Chromia-Forming	Y		BPVC, Sec II, Part D	650	\$4.04	
230	Ni-Base Chromia-Forming	Y		BPVC, Sec II, Part D	900	\$5.86	
282	Ni-Base Chromia-Forming	N				\$5.17	
PE16	Ni-Base Chromia-Forming	N				\$3.43	
718	Ni-Base Chromia-Forming	N				\$5.29	
740	Ni-Base Chromia-Forming	N				\$6.27	
740H	Ni-Base Chromia-Forming	Y		BPVC, Code Case 2702-1	800	\$6.18	
HR120	Ni-Base Chromia-Forming	Y		BPVC, Sec II, Part D	900	\$4.01	
214	Ni-Base Alumina-Forming	N				\$4.17	
247	Ni-Base Alumina-Forming	N				\$9.06	

- ✓ Corrosion Data
- ✓ Code Qualified

Elements	Raw Material Price July 2016 (\$/Lb)
Nickel	\$4.65
Cobalt	\$11.67
Molybdenum	\$6.99
Copper	\$2.20
Niobium	\$34.00
Iron	\$0.90
Chromium	\$3.78
Tungsten	\$13.00
Titanium	\$1.15
Manganese	\$0.75
Hafnium	\$75.00
Tantalum	\$59.88
Vanadium	\$8.77
Aluminum	\$0.75



Alloy Cost to Satisfy Strength Requirements

Minimum wall thickness calculation (Equation 7 in B31.1-2014):

$$t_m = \frac{PD_o}{2(SE + Py)} + A$$

t_m : Minimum wall thickness (inches)

P: Internal pressure (ksi) - **used 4.35ksi (30 Mpa)**

D_o : Outer diameter (inches) - **used 4 inches**

S: Allowable stress given as function of temperature (ksi) - **used values in code**

E: Weld joint efficiency - **used 1 (assumed seamless pipe no welds)**

y: used values from table 104.1.2

A: Additional thickness (corrosion allowance, etc.)

Cost of alloy per 1 ft length that satisfies the ASME code strength requirements:

$$Cost (\$) = \frac{\pi \times 12in \times [D_o^2 - (D_o - t_m)^2]}{1728 in^3/ft^3} \times \rho_{alloy} \times Cost_{alloy}$$

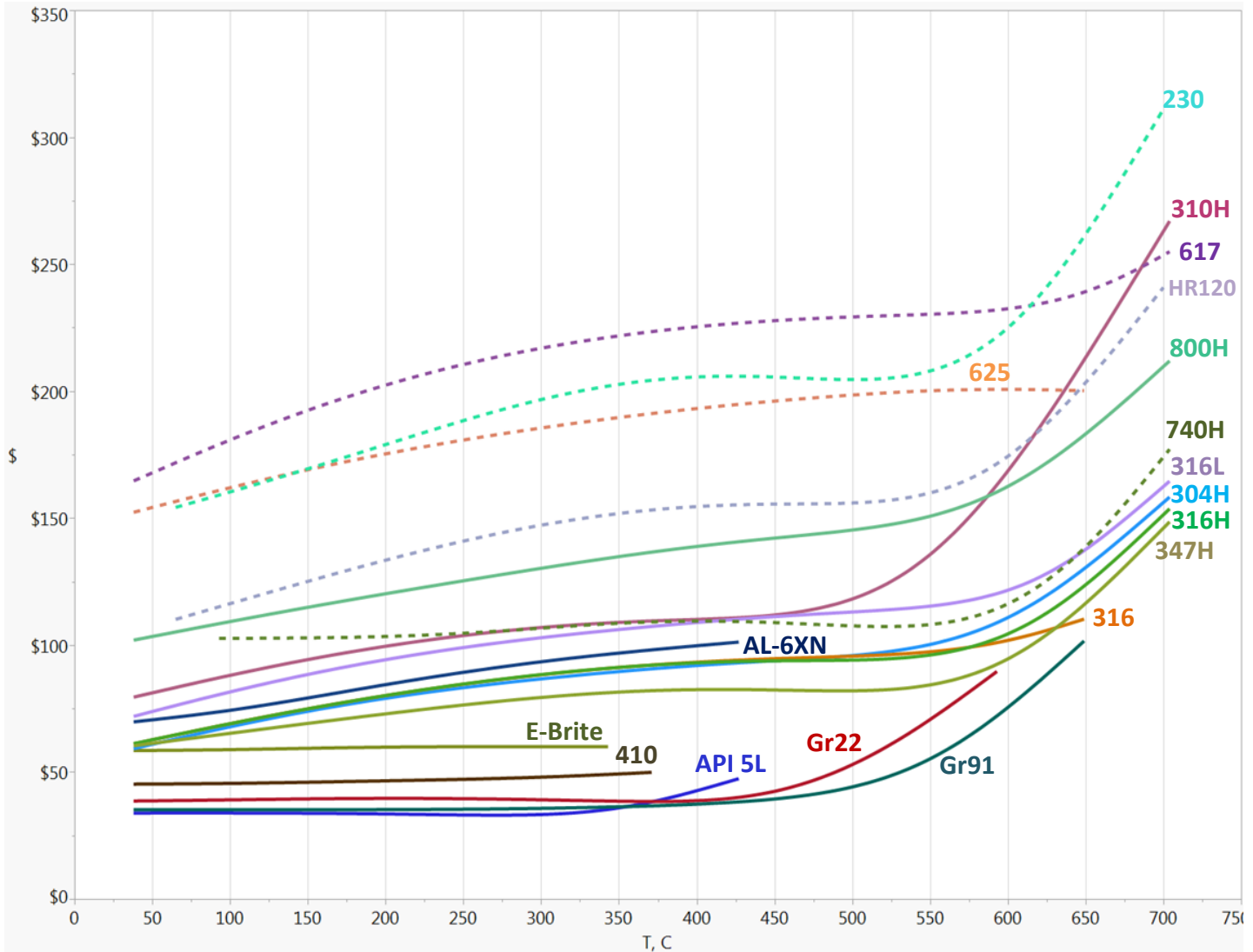
t_m : Minimum wall thickness (inches)

ρ_{alloy} : density of alloy in units of lbs/ft³

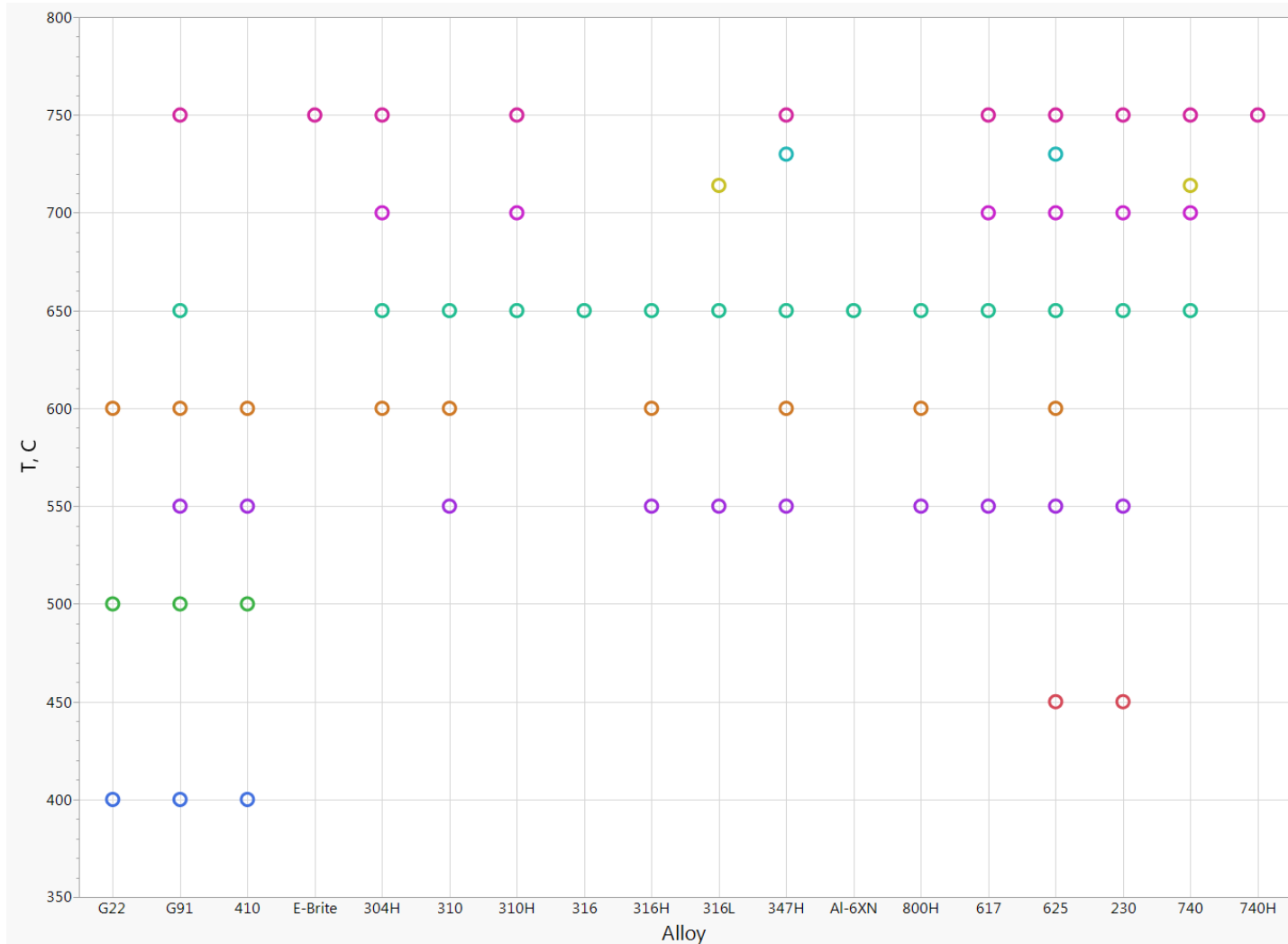
$Cost_{alloy}$: cost per lb of alloy based on raw material prices



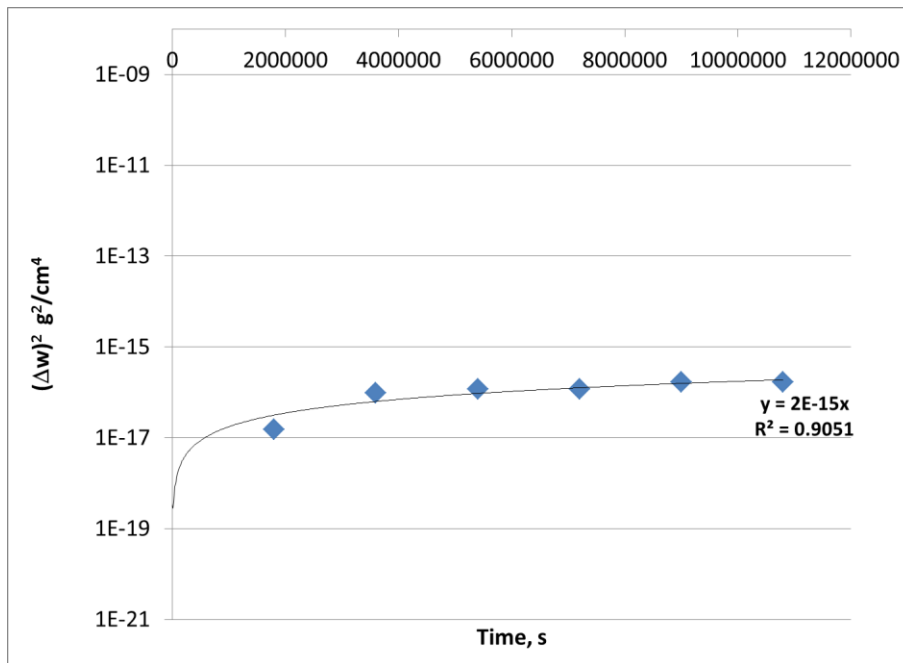
Alloy Costs (\$/ft) to Satisfy Strength Requirements



Available Alloy Corrosion Data (400-750°C, RG sCO₂, 200-250 bar)



Approach for Comparing Alloy Corrosion Rates



Parabolic fit to Cao's (2012) experimental data for 800H at 650°C

Assuming parabolic oxidation kinetics, the parabolic rate constant (k_p) can be calculated for each set of data

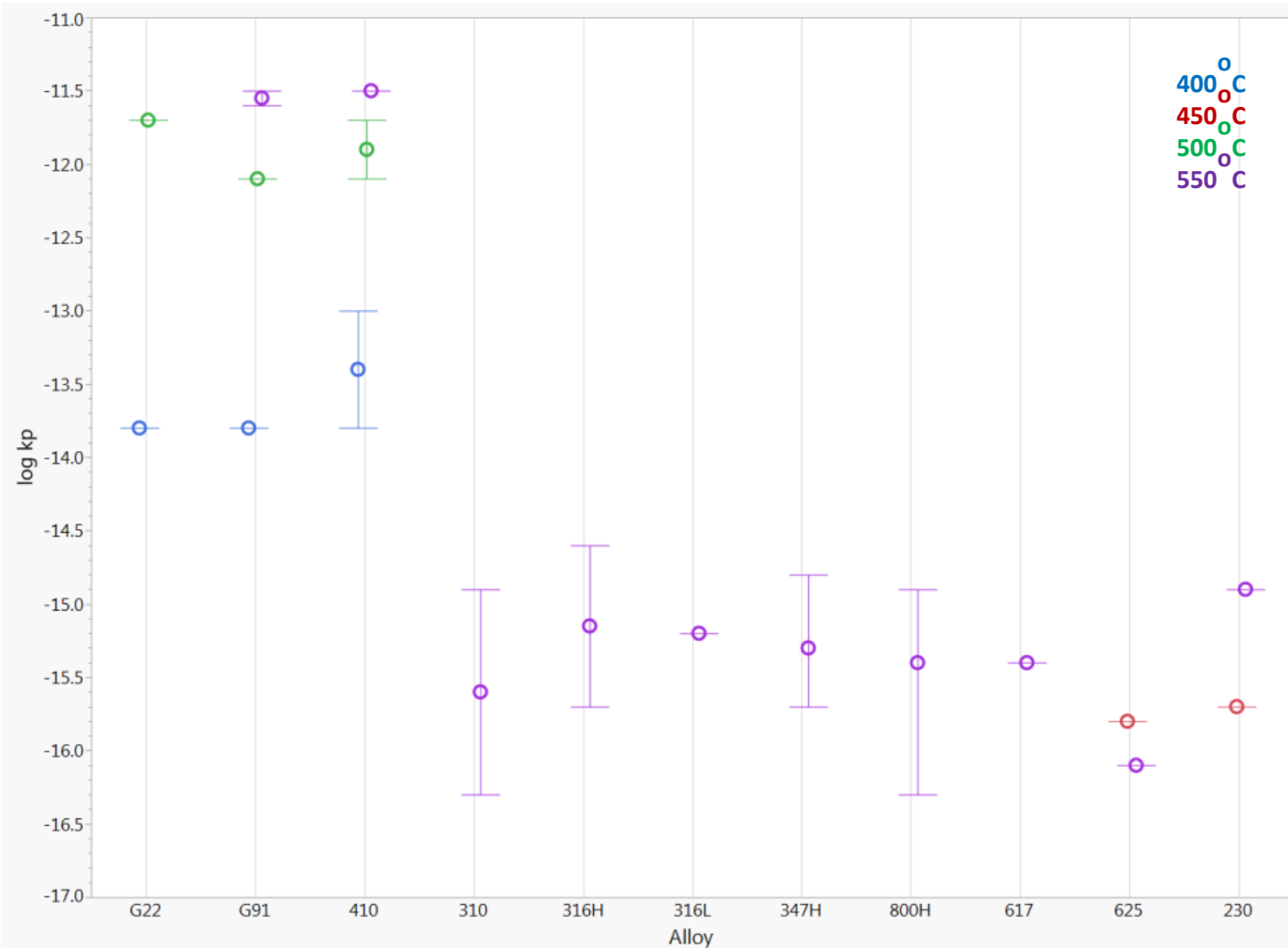
$$k_p = \frac{(\Delta m)^2}{2t}$$

Log k_p values are used as a corrosion rate comparison between alloys

Unable to accurately relate this back to alloy thickness requirements, which would be preferred



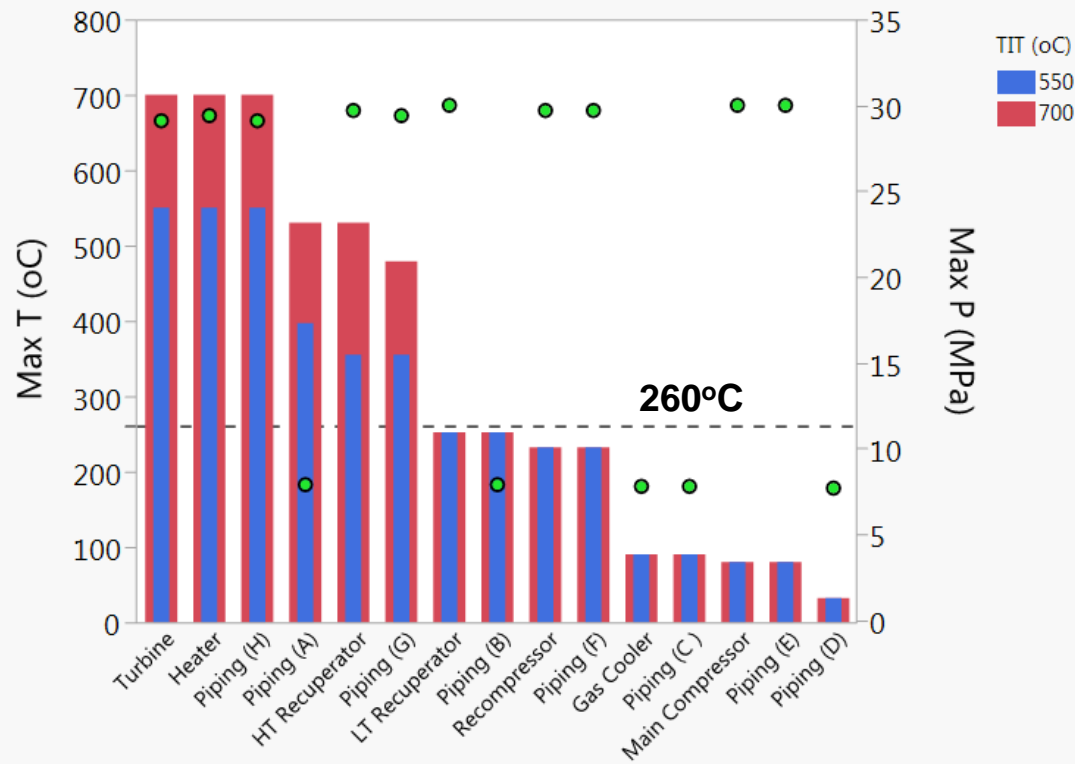
Comparing Alloy Corrosion Rates (400 – 550°C)



Comparing Alloy Corrosion Rates (600 – 700°C)



Alloy Selection for Components Based on Analysis

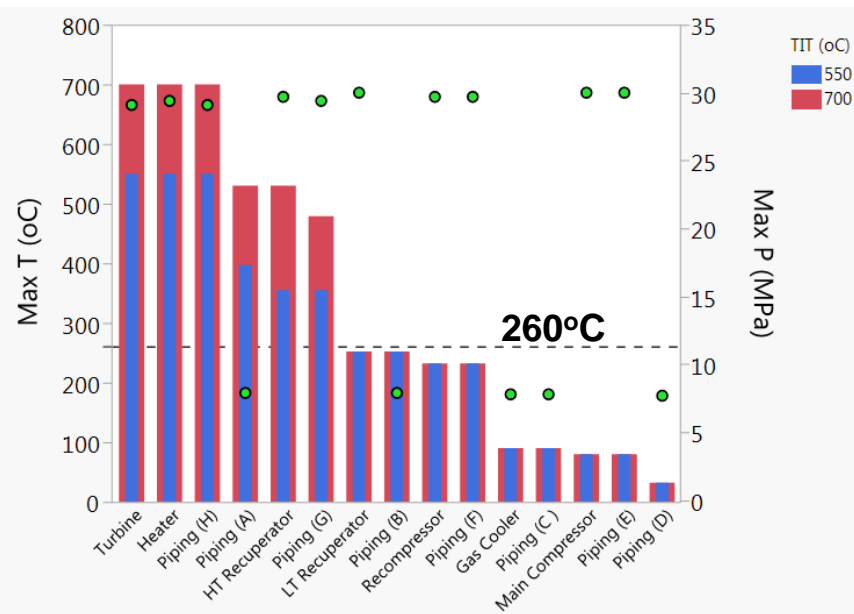


550°C TIT

- **Components @ < 400°C:**
 - Gr91 alloy recommended based on cost comparison
 - Corrosion rate comparison not available to 347H, 316H, etc. due to no data at these lower T's
- **Components @ 550°C:**
 - 347H alloy recommended
 - Significantly lower corrosion rate than Gr91 (4 orders of magnitude) at moderately increased \$ (\$75 vs \$50 per ft)
 - Similar corrosion rate to 316 alloys, but at lower \$



Alloy Selection for Components Based on Analysis



700°C TIT

- **Components @ < 260°C:**
 - Gr91 alloy recommended based on cost comparison and expected low corrosion rate (no data at these T's)
- **Components @ 500-525°C:**
 - 347H alloy recommended
 - Same reasons as for 550°C system components
- **Components @ 700°C:**
 - 740H alloy is recommended
 - 740H has \$50/ft higher cost than 347H, but has significantly lower corrosion rate (2 orders of magnitude at 650°C)
 - Corrosion data at 700°C is available for 740H, but not for 347H

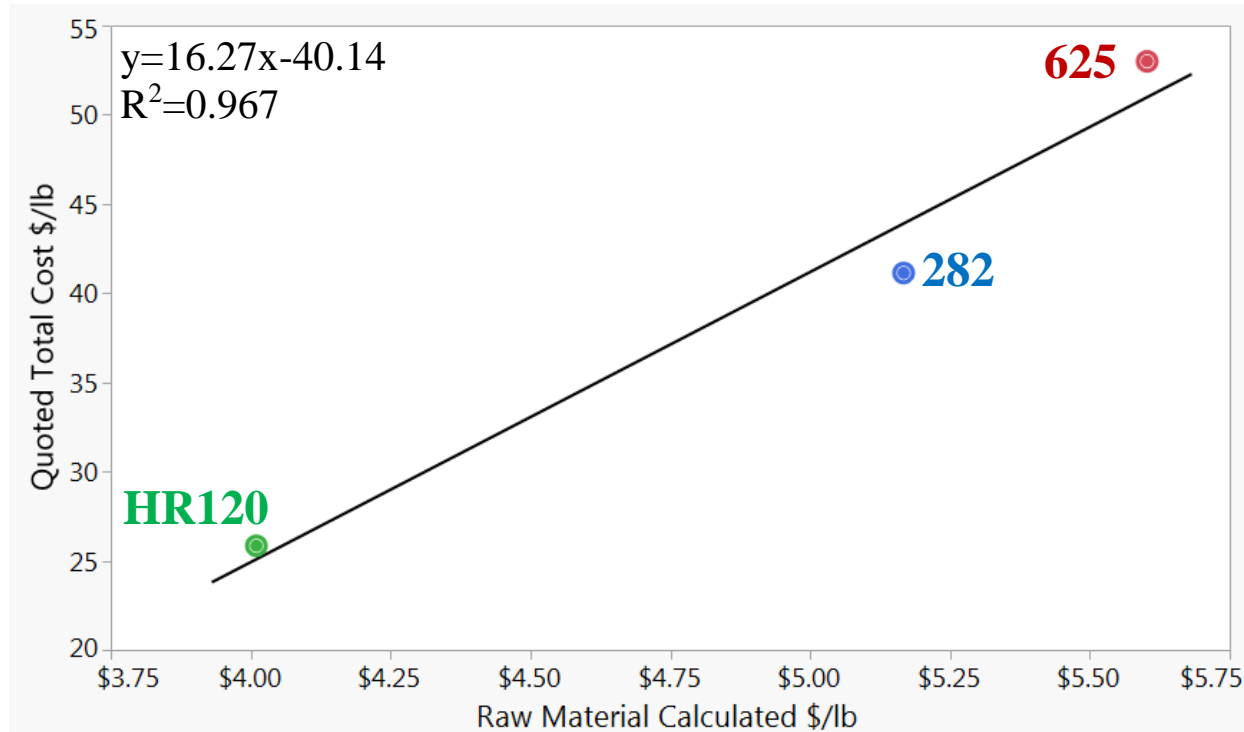


Summary and Next Steps

- **Systems level approach has been used to make informed alloy recommendations for 10MW RCBC components (550°C and 700°C TIT)**
- **Analyses revealed gaps where corrosion data is needed**
 - **Ex: 700°C for 347H, and <400°C for Gr91 alloy**
- **Future work is needed that incorporates other critical items into this process for alloy selection (manufacturability, weldability, etc.)**



Backup Slides



For 3 inch – Schedule 160 pipe



Alloy	T (°C)	Pressure (bar)	$\log k_p^A$ ($g^2cm^{-4}s^{-1}$)	Time range (hours)	$\log k_p^B$ ($g^2cm^{-4}s^{-1}$)	Time range (hours)	Reference
G22	400	200			-13.8	0-500	Pint 2016 ^[13]
G22	500	200			-11.7	0-500	Pint 2016 ^[13]
G91	550	200			-11.6	0-1000	Lee 2014 ^[14]
G91	550	250			-11.5	0-310	Rouillard 2011 ^[15]
G91	400	200			-13.8	0-500	Pint 2016 ^[13]
G91	500	200			-12.1	0-500	Pint 2016 ^[13]
410	400	200	-13.0	0-2000			Furukawa 2010 ^[16]
410	500	200	-11.7	0-2000			Furukawa 2010 ^[16]
410	550	200	-11.5	0-2000			Furukawa 2010 ^[16]
410	400	200			-13.8	0-500	Pint 2016 ^[13]
410	500	200			-12.1	0-500	Pint 2016 ^[13]
310	550	200			-16.3	0-1000	Lee 2014 ^[14]
310	550	200			-14.9	0-250	Kim 2014 ^[17]
316H	550	200	-14.6	0-3000	-15.8	0-1000	Lee 2015 ^[18]
316H	550	200			-15.7	0-250	Kim 2014 ^[17]
316L	550	250			-15.2	0-310	Rouillard 2011 ^[15]
347H	550	200	-15.4	0-3000	-15.7	0-1000	Lee 2015 ^[18]
347H	550	200			-14.8	0-250	Kim 2014 ^[17]
347H	550	200	-15.7	0-1000			Mahaffey 2014 ^[19]
800H	550	200			-14.9	0-250	Kim 2014 ^[17]
800H	550	200			-16.3	0-1000	Lee 2014 ^[14]
800H	550	250			-15.1	0-310	Rouillard 2011 ^[15]
800H	550	200	-15.3	0-1000			Mahaffey 2014 ^[19]
617	550	200	-15.4	0-1000	-15.1	0-200	Dheeradhada 2015 ^[20]
625	550	200			-16.1	0-1000	Pint 2014 ^[21]
625	450	200	-15.8	200-1000	-17	0-200	Mahaffey 2015 ^[22]
625	550	200	-16.1	200-1000	-15.2	0-200	Mahaffey 2015 ^[22]
625	550	200			-16.1	0-1000	Lee 2014 ^[14]
230	450	200	-15.7	200-1000	-16.8	0-200	Mahaffey 2015 ^[22]
230	550	200	-14.9	0-1000	-15	0-200	Mahaffey 2015 ^[22]

$\log k_p^A$ Derived by fitting through multiple points over the time range

$\log k_p^B$ Derived by fitting through a single point over the time range

400 to 550 °C



Alloy	T (°C)	Pressure (bar)	log k _p ^A (g ² cm ⁻⁴ s ⁻¹)	Time range (hours)	log k _p ^B (g ² cm ⁻⁴ s ⁻¹)	Time range (hours)	Reference
G22	600	200			-10.7	0-500	Pint 2016 ^[13]
G91	600	200			-10.9	0-1000	Lee 2014 ^[14]
G91	650	200			-10.7	0-1000	Lee 2014 ^[14]
G91	650	207			-10.7	0-500	Tan 2011 ^[23]
G91	600	200			-11.5	0-500	Pint 2016 ^[13]
410	600	200	-11.0	0-2000			Furukawa 2010 ^[16]
410	600	200			-12.1	0-500	Pint 2016 ^[13]
304H	650	200			-13.1	0-500	Pint 2014 ^[21]
304H	700	200			-13.5	0-500	Pint 2014 ^[21]
304H	600	200			-14.0	0-500	Pint 2016 ^[13]
310	650	200	-14.3	0-2002	-14.7	0-502	Cao 2012 ^[24]
310	600	200			-15	0-1000	Lee 2014 ^[14]
310	650	200			-14.5	0-1000	Lee 2014 ^[14]
310	650	200			-13.9	0-250	Kim 2014 ^[17]
310	650	200	-14.4	0-2000			Firouzdor 2015 ^[25]
310H	650	200			-15.4	0-500	Pint 2014 ^[21]
310H	700	200			-14.2	0-500	Pint 2014 ^[21]
316	650	200	-12.7	0-3000	-15.1	0-502	Cao 2012 ^[24]
316	650	200	-12.9	0-998	-12.5	0-117	Olivares 2015 ^[26]
316	650	200	-12.7	0-3000			Firouzdor 2015 ^[25]
316H	600	200	-14	0-3000	-15	0-1000	Lee 2015 ^[18]
316H	650	200			-13.6	0-1000	Lee 2015 ^[18]
316H	650	200			-11.8	0-250	Kim 2014 ^[17]
316L	650	200			-12.4	0-194	Lim 2008 ^[27]
347H	600	200	-15.2	0-3000	-15.1	0-1000	Lee 2015 ^[18]
347H	650	200	-13.7	0-3000	-13.7	0-1000	Lee 2015 ^[18]
347H	650	200			-12.5	0-250	Kim 2014 ^[17]
347H	600	200			-14.0	0-500	Pint 2016 ^[13]
Al-6XN	650	207	-14.5	0-3000			Tan 2011 ^[23]
Al-6XN	650	200	-14.5	0-3000			Firouzdor 2013 ^[28]
800H	650	200			-13.9	0-250	Kim 2014 ^[17]
800H	600	200			-15.2	0-1000	Lee 2014 ^[14]
800H	650	200			-15.0	0-1000	Lee 2014 ^[14]
800H	650	207	-15.0	0-3000			Tan 2011 ^[23]
800H	650	200	-14.7	0-3000			Cao 2012 ^[24]
617	650	200			-15.3	0-500	Pint 2014 ^[21]
617	700	200			-14.9	0-500	Pint 2014 ^[21]
617	650	200	-14.8	200-1000	-13.9	0-200	Dheeradhada 2015 ^[20]
625	650	200	-14.9	0-2995	-14.6	0-492	Firouzdor 2013 ^[28]
625	650	200			-15.1	0-500	Pint 2014 ^[21]
625	700	200			-14.6	0-500	Pint 2014 ^[21]
625	600	200			-15.3	0-1000	Pint 2014 ^[21]
625	650	200			-14.6	0-1000	Pint 2014 ^[21]
625	650	200	-14.9	200-1000	-14.3	0-200	Mahaffey 2015 ^[22]
625	600	200			-15.3	0-1000	Lee 2014 ^[14]
625	650	200			-14.6	0-1000	Lee 2014 ^[14]
230	650	200	-15.5	492-2995	-14.5	0-492	Firouzdor 2013 ^[28]
230	650	200			-15.2	0-500	Pint 2014 ^[21]
230	700	200			-14.9	0-500	Pint 2014 ^[21]
230	650	200	-14.8	200-1000	-13.9	0-200	Mahaffey 2015 ^[22]
740	650	200			-15.4	0-500	Pint 2014 ^[21]
740	700	200			-14.6	0-500	Pint 2014 ^[21]
740	650	200	-15.3	200-1000	-14.7	0-200	Mahaffey 2015 ^[22]

log k_p^A Derived by fitting through multiple points over the time range

log k_p^B Derived by fitting through a single point over the time range

600 to 700 °C

