

Understanding Cu-Al Limited-Volume Diffusion Towards Lifetime Prediction for Cu Wire Bonds

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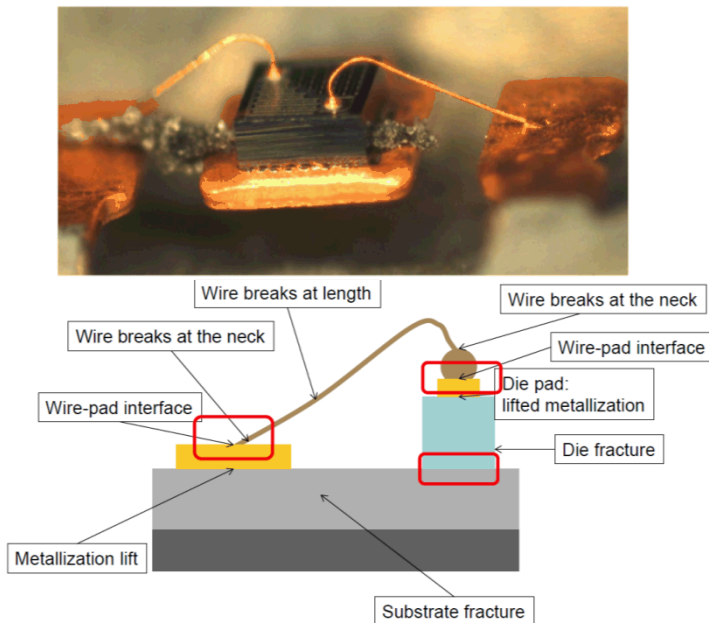
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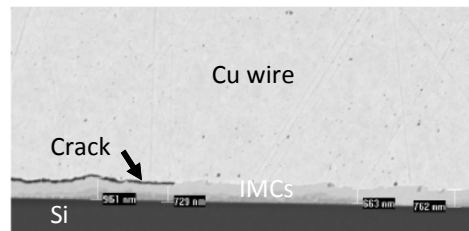
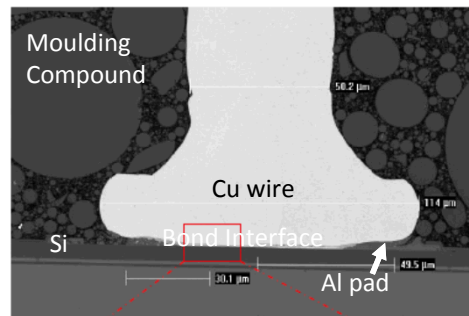
Copper Wire Bonds in Electronic Components

Motivation

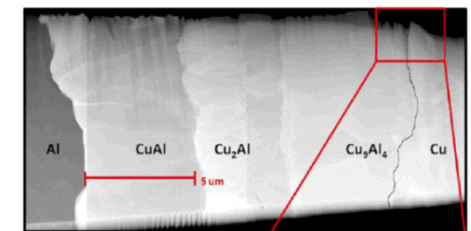
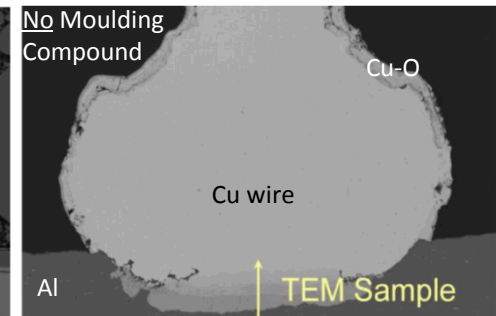
- ◆ Cu wire bonds are replacing Au wire bonds. The knowledge base, needed to assess acceptability and lifetime of Cu wires bonded to aluminum (Al) pads for high reliability products, is lacking.
- ◆ Cu & Al react to form intermetallic compound (IMC) of varying types/thicknesses across a single bond. Non-uniformity of the IMC layers is not well understood.



B. Paskaleva, et al., "Device Reliability TGC-XIV: Fundamentals behind the Growth of Cu-Al Intermetallics," oral presentation at the Spring TGC-XIV. Mar 15-16, 2016. Huntsville, AL.



G.M. O'Halloran, et al. "Planar Analysis of Copper-Aluminum Intermetallics," *Proceedings of the 39th International Symposium for Testing and Failure Analysis*. San Jose, CA. Nov 3-7, 2013.

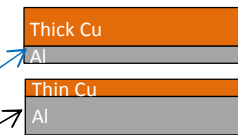
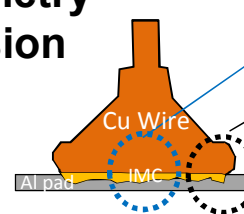


Rene Rongen, et al. "Lifetime Prediction of Cu-Al wire Bonded Contacts for Different Mould Compounds," *Proceedings of the 64th Electronic Components and Technology Conference*. Lake Buena Vista, FL. May 27-30, 2014.

Copper Wire Bonds in Electronic Components

Motivation

- ◆ Cu wire bonds are replacing Au wire bonds. The knowledge base, needed to assess acceptability and lifetime of Cu wires bonded to aluminum (Al) pads for high reliability products, is lacking.
- ◆ Cu & Al react to form intermetallic compound (IMC) of varying types/thicknesses across a single bond. Non-uniformity of the IMC layers is not well understood.
- ◆ **Need to understand effects of bond geometry and aging on Cu-Al limited-volume diffusion and resulting joint degradation.**

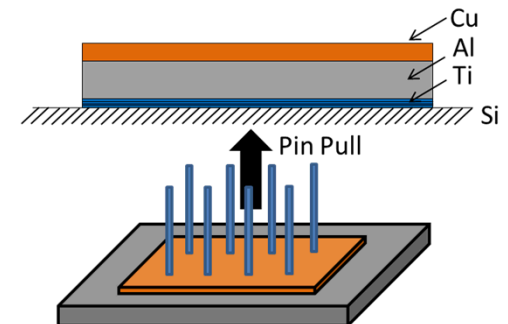


Varying Cu:Al Ratio

<u>Cu</u>	<u>Al</u>
2.0μm	0.2μm
2.0μm	2.0μm
0.2μm	2.0μm

Approach

- ◆ Age thin film stacks of varying Cu:Al ratio between 0-300 days at 50°C-300°C. Determine thin film integrity via mechanical testing (pin pulling). Determine Cu-Al solid-state interface reactions in multiple aging conditions. Focus on:
 - Room Temperature (baseline)
 - 0-100 days at 150°C



Leg work to establish diffusivity of Al into Cu as a function of aging temperature, time, and varying Cu:Al ratio.

Approach

- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.

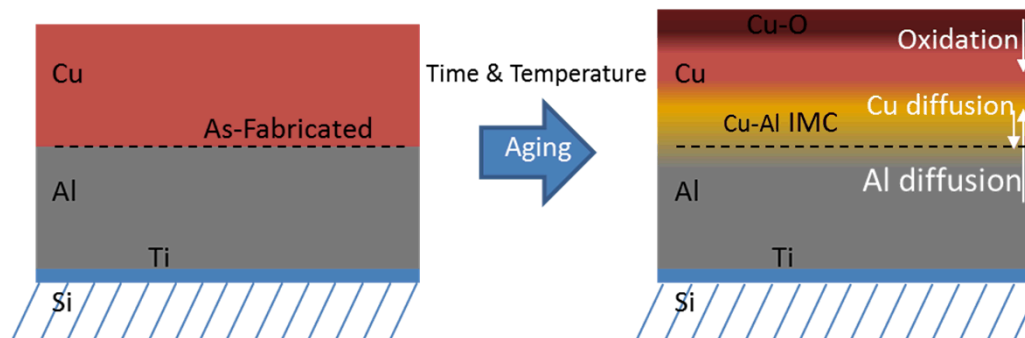
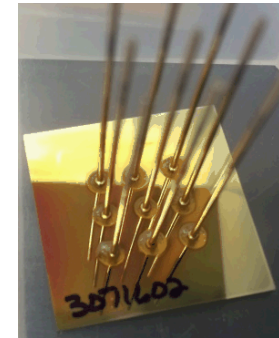
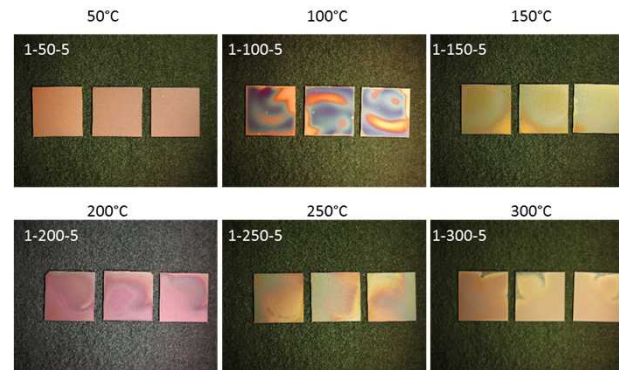
Aging parameters

Temperatures: 50, 100,

150, 200, 250, 300°C

Time: 0, 5, 10, 25, 50, 100,

150, 200, 300 days

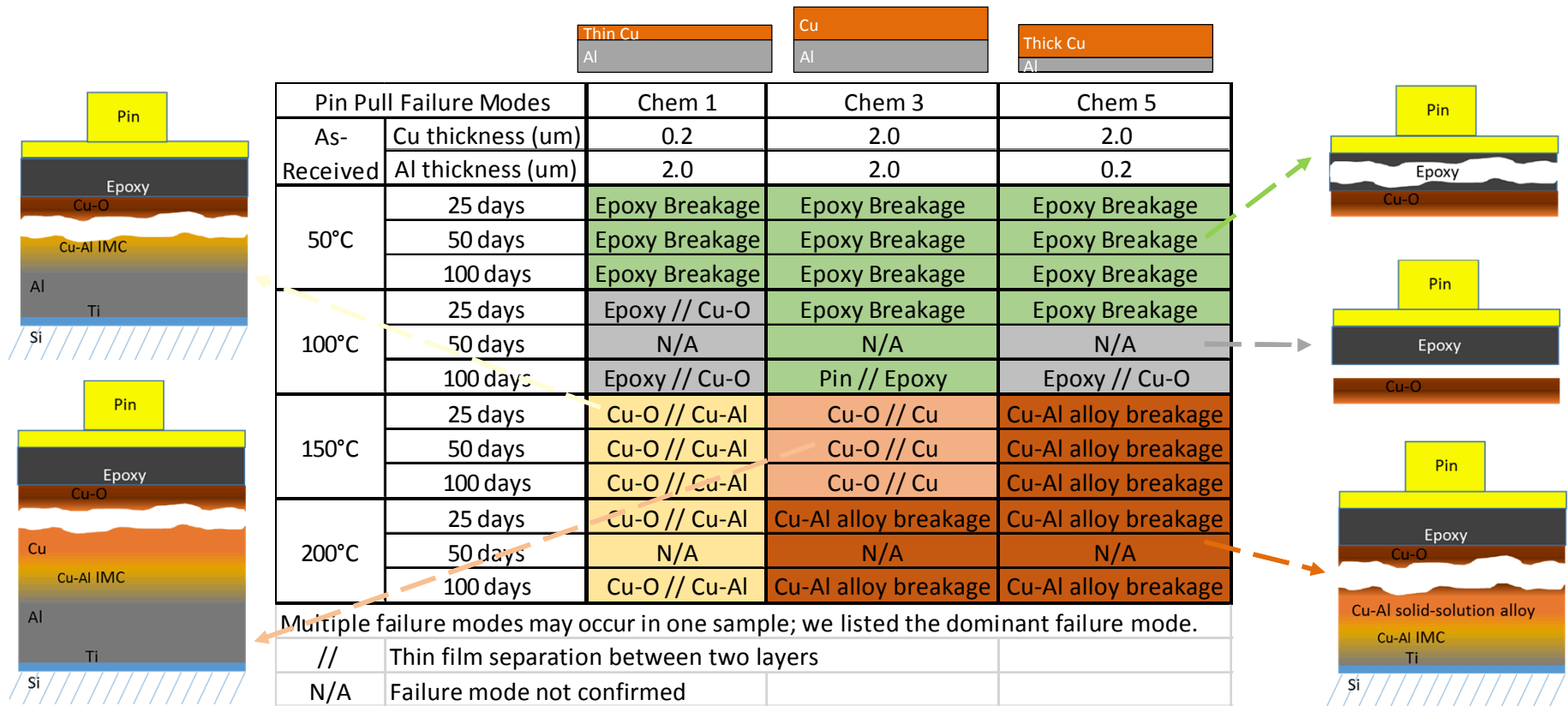


In some cases, Ti and Si may diffuse towards the film top surface.

Pin pulling → determine interface strength
Cross-section → understand failure modes
Auger Profile → understand elemental diffusion

Common Failure Modes

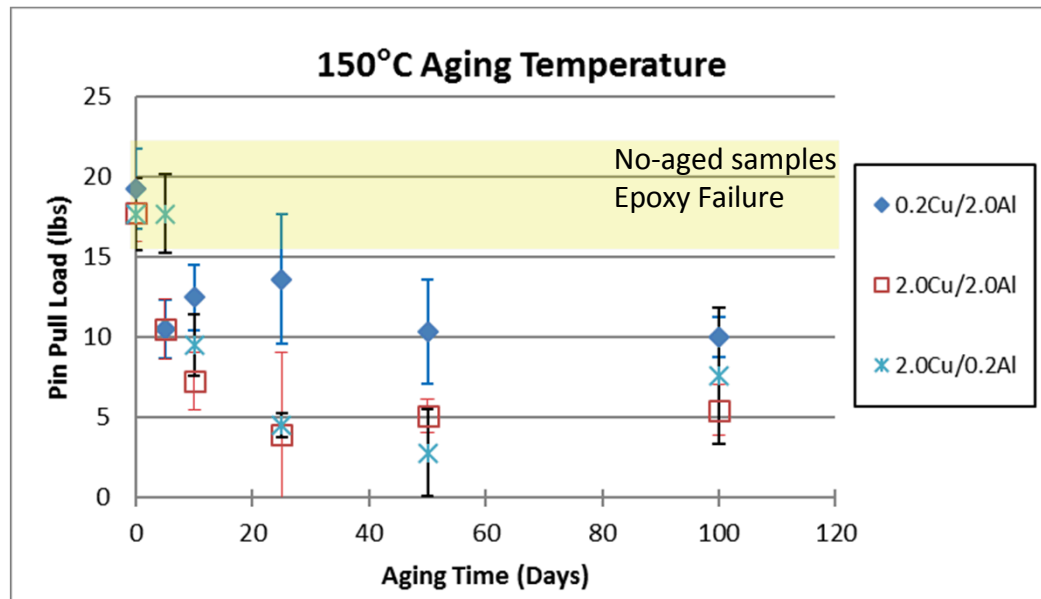
- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ◆ Observed failure modes include:



150°C Aged Thin Films

- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ◆ Failure modes - similar across 25-100 days but different for each Cu:Al ratio.

Pin Pull Failure Modes		Chem 1	Chem 3	Chem 5
As-Received	Cu thickness (um)	0.2	2.0	2.0
	Al thickness (um)	2.0	2.0	0.2
150°C	25 days	Cu-O // Cu-Al	Cu-O // Cu	Cu-Al alloy breakage
	50 days	Cu-O // Cu-Al	Cu-O // Cu	Cu-Al alloy breakage
	100 days	Cu-O // Cu-Al	Cu-O // Cu	Cu-Al alloy breakage



150°C Aged Thin Films

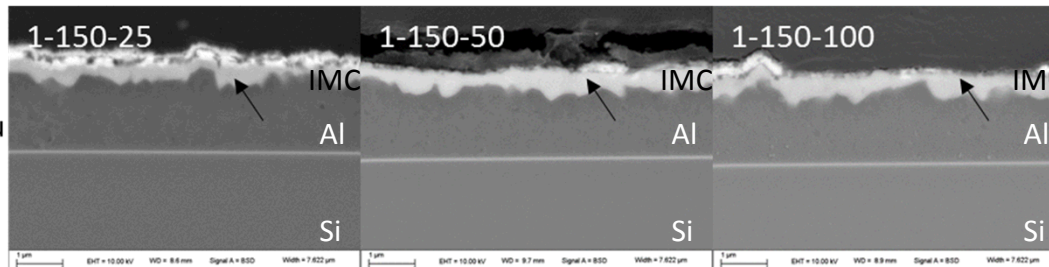
- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ◆ Failure modes - similar across 25-100 days but different for each Cu:Al ratio. Chemical potentials, driving the diffusion, are different for each Cu:Al ratio.

25 Days

50 Days

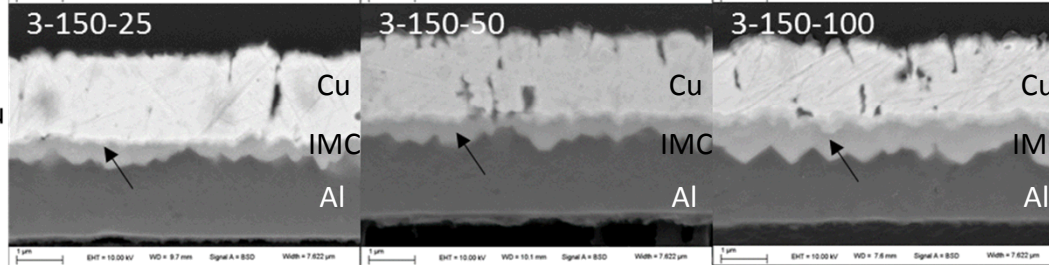
100 Days

Chem 1
0.2μm Cu
2.0μm Al



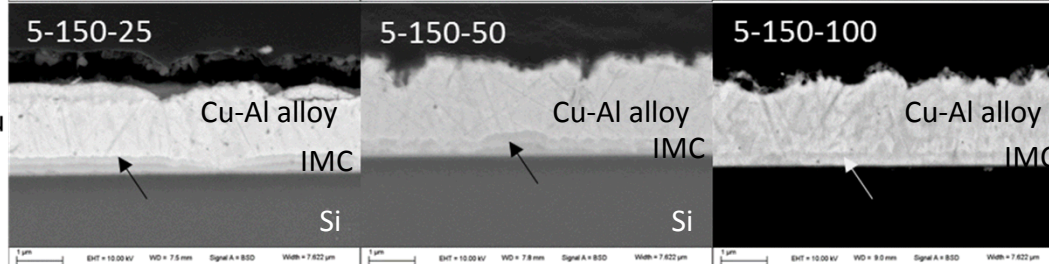
Cu completely consumed by oxidation and IMC formation

Chem 3
2.0μm Cu
2.0μm Al



Cu and Al sources not completely consumed.

Chem 5
2.0μm Cu
0.2μm Al

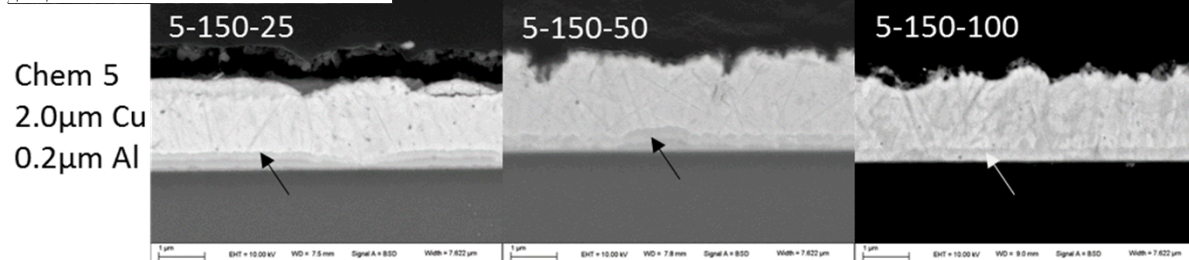
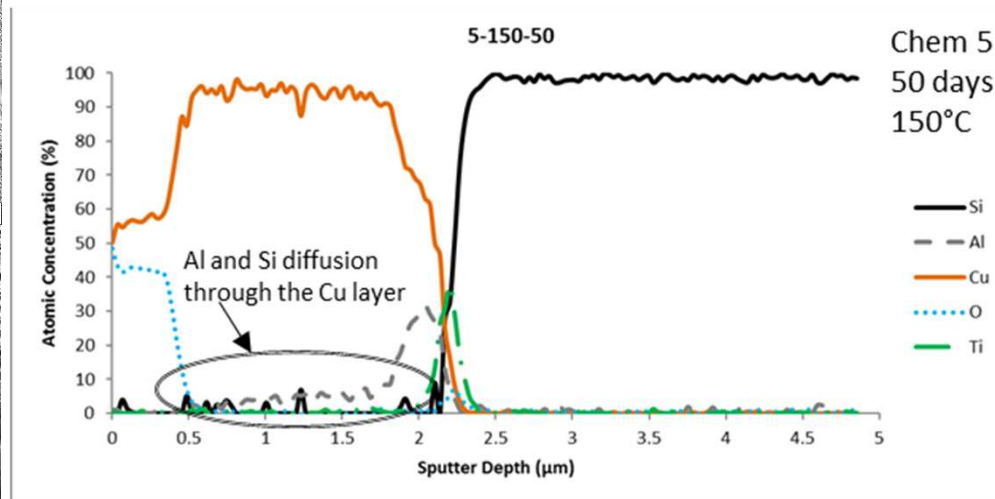
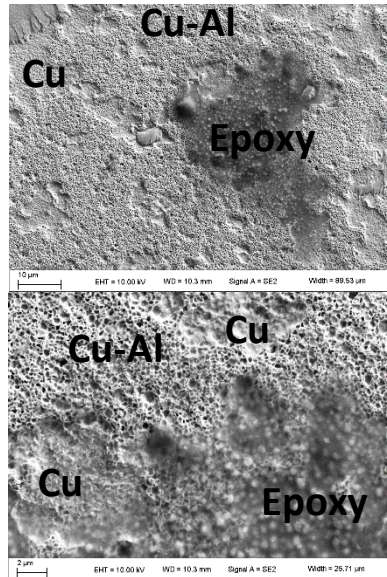


BREAKAGE in Cu-Al SOLID-SOLN ALLOY
Al completely consumed by IMC formation and α-Cu solid-solution alloy

150°C Aged Thin Films – 2.0 μ mCu / 0.2 μ mAl

- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ◆ Auger diffusion profiles showed evidence of **Si** & Al diffusion towards surface.
- Breakage in Cu-Al solid solution alloy. Role of Si? Need to investigate.
- Thin Al pad = limited Al source \rightarrow large extent of joint degradation

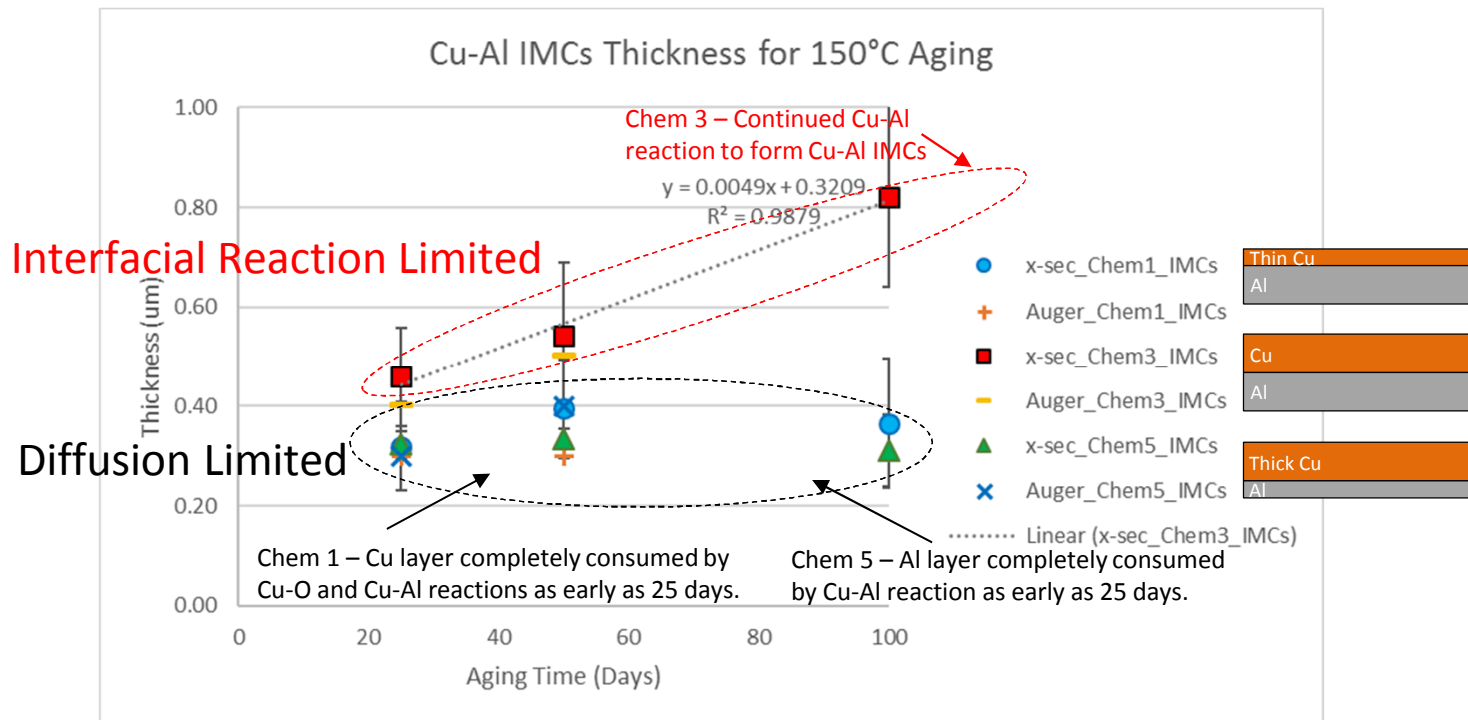
SEM of top film fracture surface



BREAKAGE in Cu-Al SOLID-SOLN ALLOY
Al completely consumed by IMC formation and α -Cu solid-solution alloy

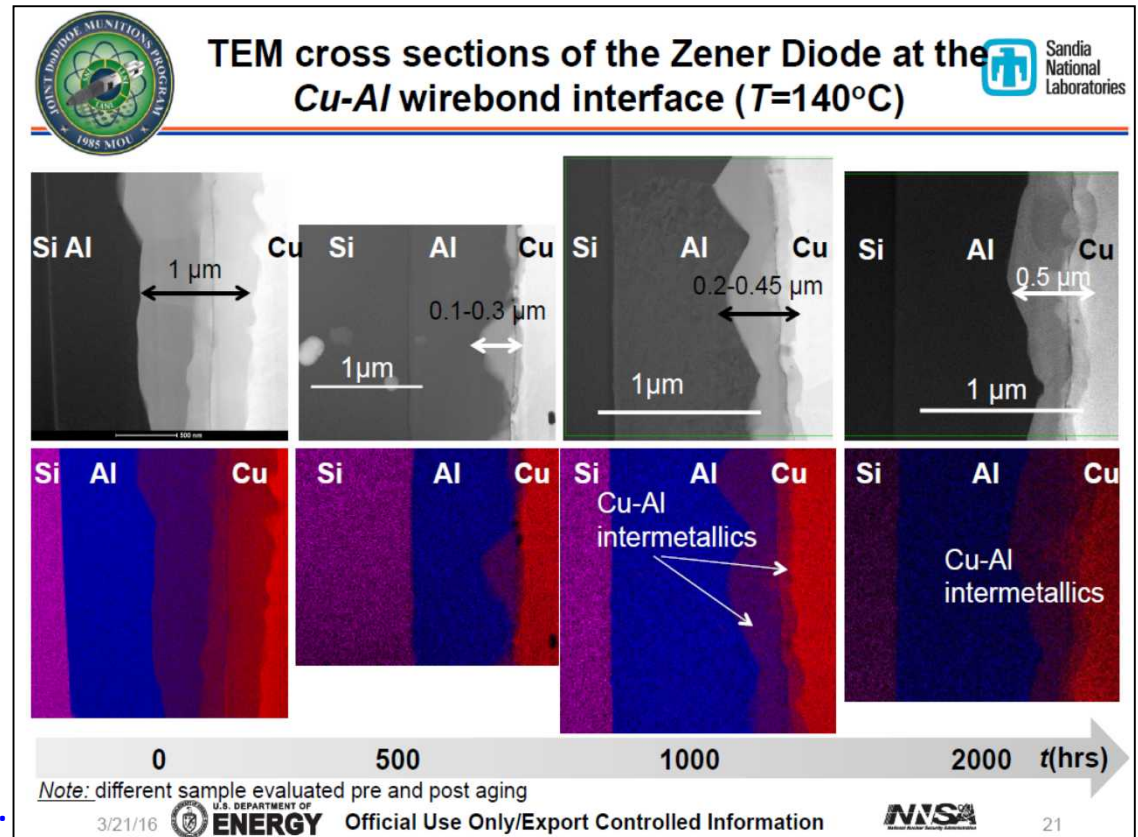
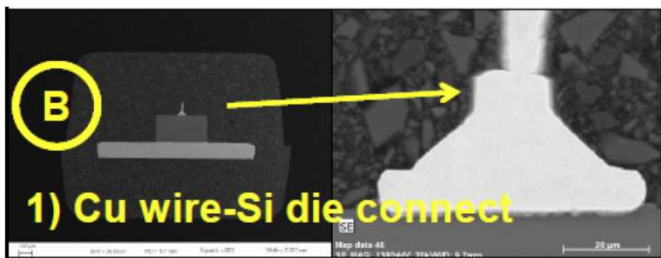
150°C Aged Thin Films – IMC thickness

- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ◆ Chem 1 & Chem 5 → Cu-Al IMCs layer thickness of 0.3-0.4 μm. The Cu and Al sources were depleted, resulting in no significant Cu-Al IMCs growth after 25 days.
- ◆ Chem 3 → Cu-Al IMCs thickness increases linearly ($R^2 = 0.99$) at the rate of ~5 nm per day between 25 and 100 day of aging at 150°C.



140°C Aged Test Vehicle – IMC Thickness

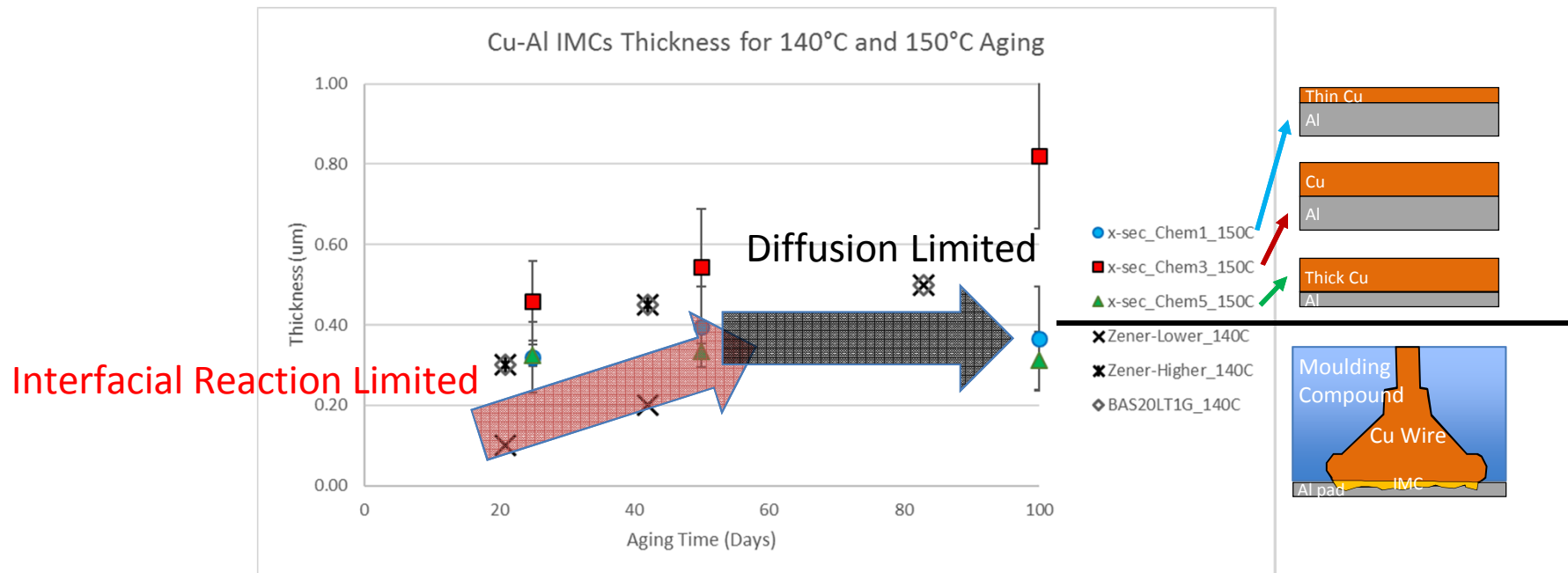
- ♦ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ♦ TEM cross-sections of the Zener Diodes at the Cu-Al wire bond interface after aging at 140°C for 500 hours (21 days), 1000 hours (42 days), and 2000 hours (83 days)



Accelerated test vehicle from JMP work by B. Paskaleva & T. Buchheit.

Comparison to Accelerated Aged Test Vehicle

- ◆ The **long-term reliability** of the Cu wires bonded to Al pads is determined by potential degradation modes resulting from solid-state reactions that occur at elevated temperature environments.
- ◆ IMC thicknesses in the accelerated test vehicles increased at
 - a high rate from 21 to 42 days → Chem 3 from 25 to 100 days.
 - Interfacial reaction controlled the growth rate of IMC.
 - a lower rate from 42 to 83 days → Chem 1 & Chem 5 aged >25 days.
 - Diffusion controlled the growth rate of IMC.



Summary

- ◆ The fundamental diffusion couple aging study allowed more systematic investigation of the effects of Cu:Al ratio and aging parameters on the microstructure evolution and IMC growth.
 - ◆ Began to build a knowledge base on how Cu wire bonds age, using thin film diffusion couples and selective comparison to accelerated aged test vehicles for validation.
 - ◆ Initially, IMC growth is controlled by interfacial reaction rate (linear with time). Over time, Cu and/or Al sources become limited and IMC growth is controlled by diffusion and is significantly slowed down.

Extreme high Cu:Al ratios, where Al becomes the limited resource showed failure due to *breakage in such Cu-Al solid solution* alloy layer with possible involvement of Si and/or Si reaction within the film stack—weakening the joint. **Practically, the implication is that “thin” pure Al pads should not be used.**
 - ◆ The materials parameters collected in the fundamental study will be used to determine diffusivity as a function of temperature, time, and Cu:Al ratios and will be used as inputs to build a lifetime prediction model of Cu wire bond aging in future work.
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Thank you for your attention!
Questions?

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