

Used Fuel Disposition Campaign

Isotope Chemistry and Source Term

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Presentation Outline

- Isotope chemistry conceptual model
- Decay and ingrowth
- Isotope partitioning
- Source term framework

Used Fuel Disposition

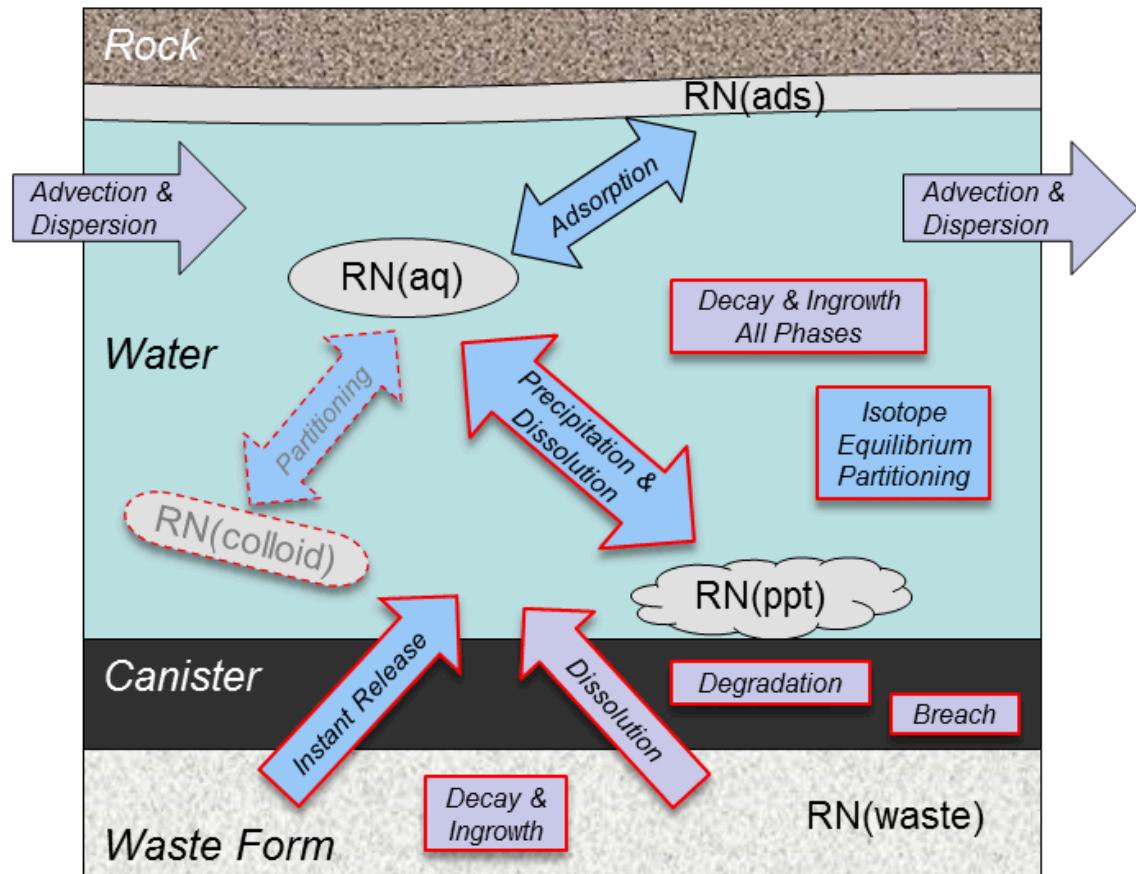
Isotope Processes

■ Isotope chemistry processes

- Decay and ingrowth
- Solubility
 - Precipitation, sorption, aqueous
- Partitioning
 - Element-based
- Solubility
 - Maximum entropy
 - Precipitation, sorption, aqueous

■ Source term

- Canister
 - Degradation framework
 - Breach
- Waste form
 - Decay and ingrowth
 - Instant release
 - Multiple dissolution models
- Isotope release
 - Multiple isotopes
 - Congruent release of RNs



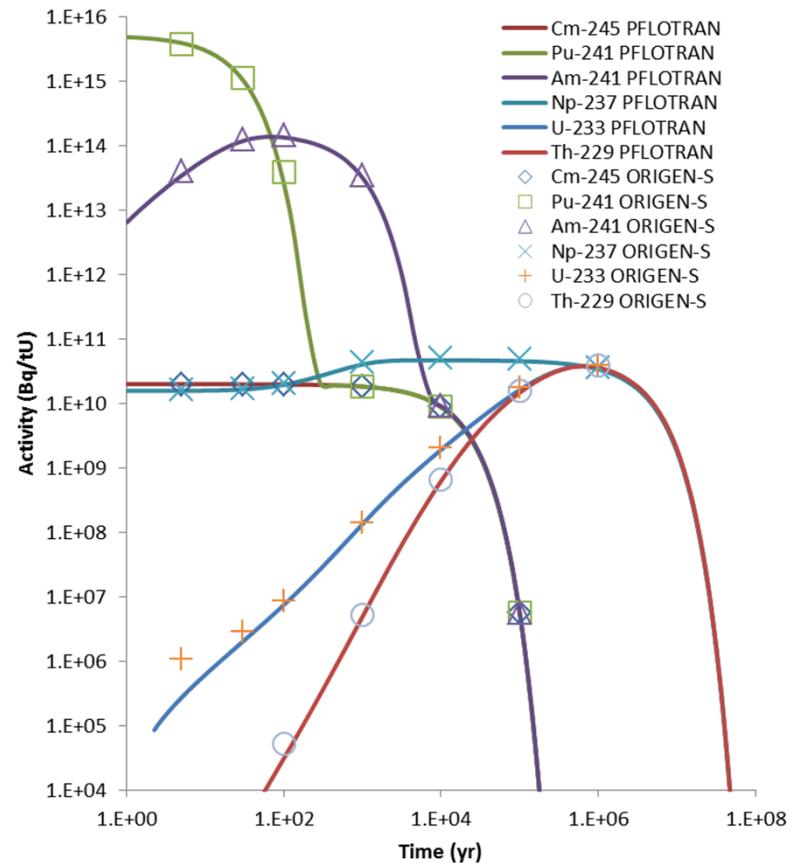
Red = area of major advances this year
Underline = new feature this year

■ 3-Generation Analytical Solution

- Concentration of daughter is a function of:
 - Decay constants and initial concentrations of daughter, parents, and grandparents
 - Duration of time step
- Accounts for multiple parents and grandparents
- Limitation
 - Grandparents may not experience significant ingrowth

■ Validation

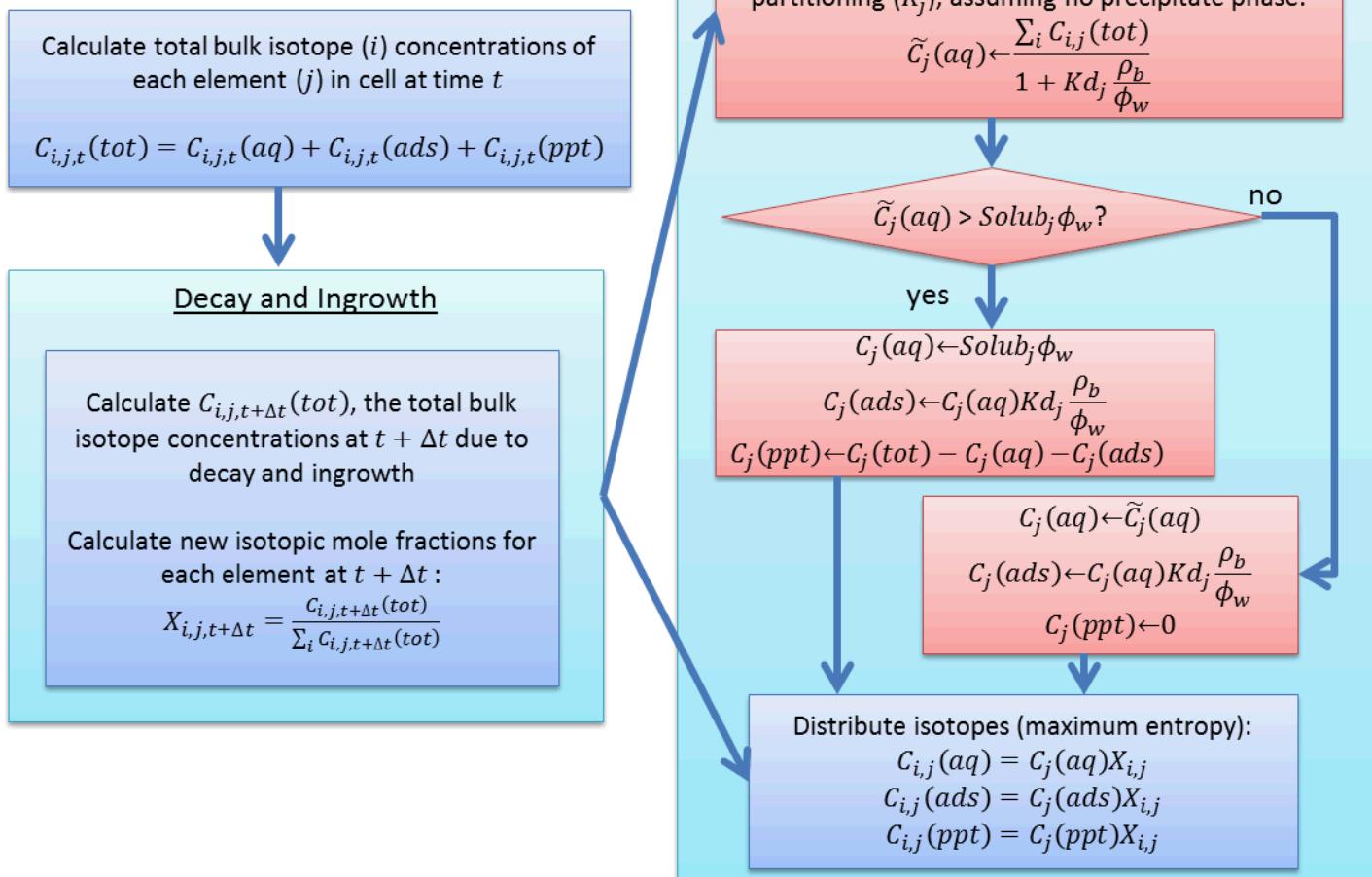
- Tested 7-generation Np series from ^{245}Cm to ^{229}Th
 - Limitation required removal of short-lived ^{233}Pa
 - Concentrations of short-lived isotopes like ^{233}Pa may be calculated on the side assuming secular equilibrium
- Results compare well to ORIGEN-S results of Anttila 2005 (pp. 152 and 158)
 - Limitation affects ^{233}U and ^{229}Th at early times due to significant ingrowth of ^{241}Pu and ^{241}Am , respectively



- **Physical and radioactive properties of isotopes of the same element are different**
 - Atomic mass, decay constant, daughter product
- **Chemical properties of isotopes of the same element are similar and are assumed to be identical**
 - Chemical reactions, Kd value, volume, volatility, etc.
- **Aqueous solubility is a property of the element**
 - The most abundant isotope(s) of each solubility-limited element must be included in the simulation to properly account for solubility-limited precipitation and dissolution
- **Thermodynamic equilibrium is enforced**
 - Precipitation and dissolution of solubility-limited elements
 - Sorption
- **Isotope concentrations are distributed to maximize entropy**
 - Isotope ratios of the same element in a given cell are forced to be identical in each phase (aqueous, sorbed, precipitate, etc.)
 - Done for simplification and for isotopic thermodynamic equilibrium

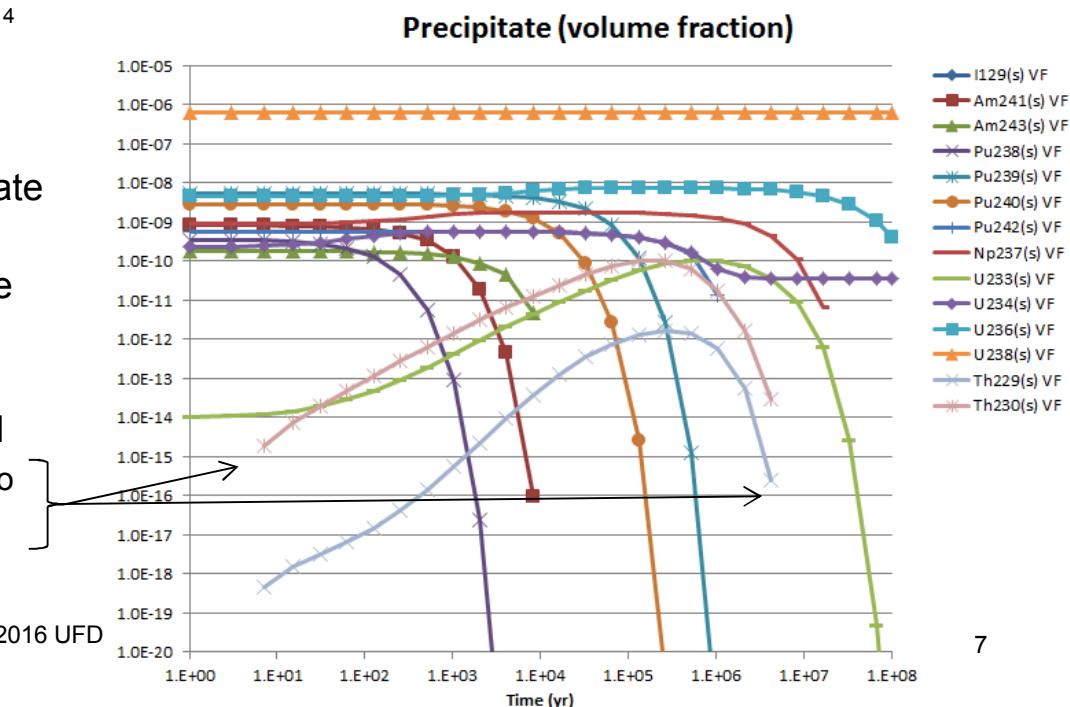
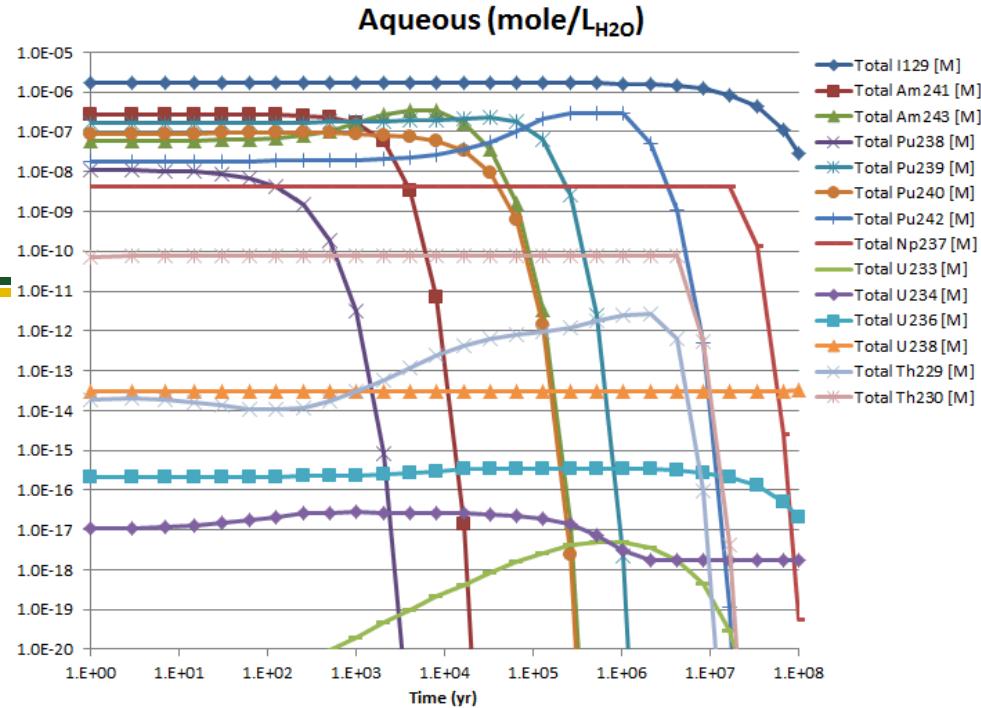
Isotope Partitioning Flow Chart

■ Isotope decay and partitioning calculations



■ Isotope decay, ingrowth, and partitioning

- Example
 - Dissolve 1 gram UNF into 1 liter of water
 - Apply elemental solubility limits (M)
 - Th: 8×10^{-11} , Am: 3×10^{-7} ,
Pu: 3×10^{-7} , Np: 4×10^{-9} , U: 3×10^{-14}
 - Observe over time
- Example shows
 - Ingrowth in aqueous and precipitate phases
 - Maximum entropy occurs (isotope ratios for a given element are the same in each phase)
 - Solubility constraints are honored
 - Th(aq) is below saturation prior to 7 yr and after 5 Myr



Isotope Modeling – Order of Operations

Order of operations

1

Advection and dispersion

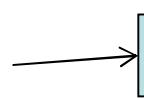
2

*Calculate new total isotope
concentrations in cell*



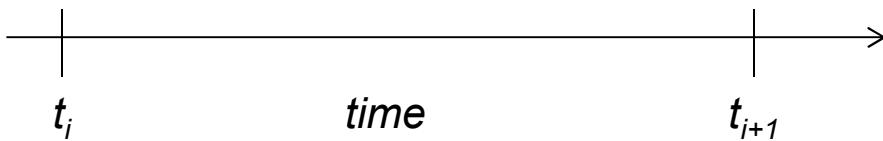
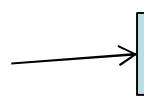
3

*Decay and ingrowth during
time step*



4

*Partition isotopes among all
phases*



■ Operator splitting

- Needed to be able to decay and ingrow isotopes in the precipitate phase

■ Likely future improvements

- Numerical solver for decay and ingrowth (to address limitation of current analytical solution)
- Colloid partitioning
- Gas phase partitioning

■ Potential future improvements

- Solid solutions model for improved coupling

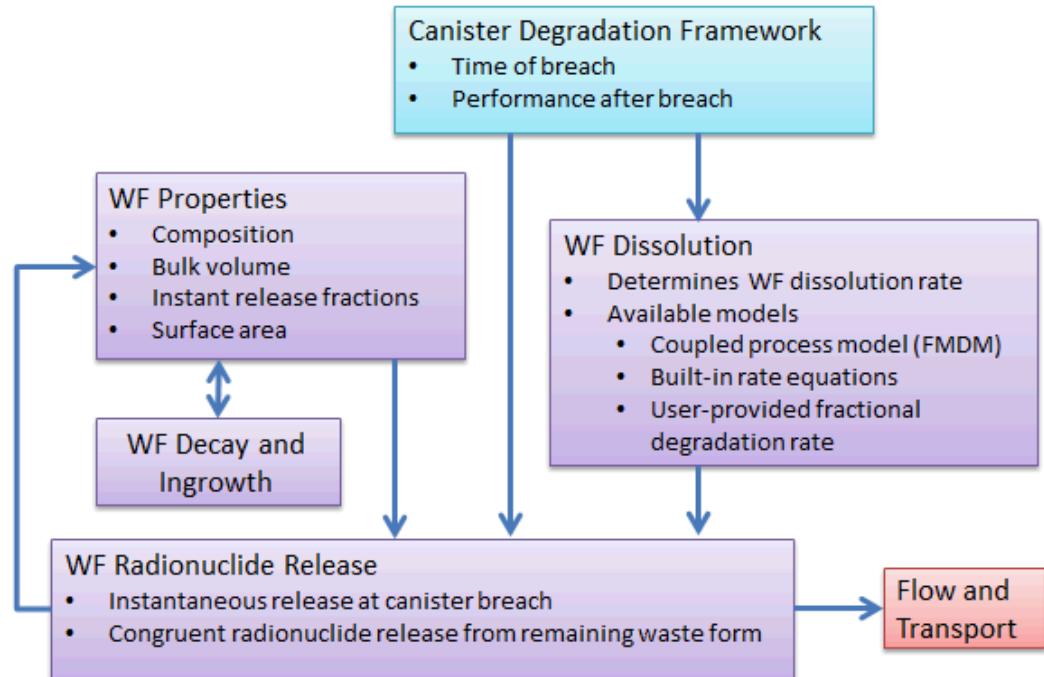
■ Canister degradation

- Degradation leading to breach
- Degradation after breach

■ Waste form

- Decay and ingrowth
- Instantaneous release
- Multiple dissolution models
- Radionuclide release
- Track remaining volume

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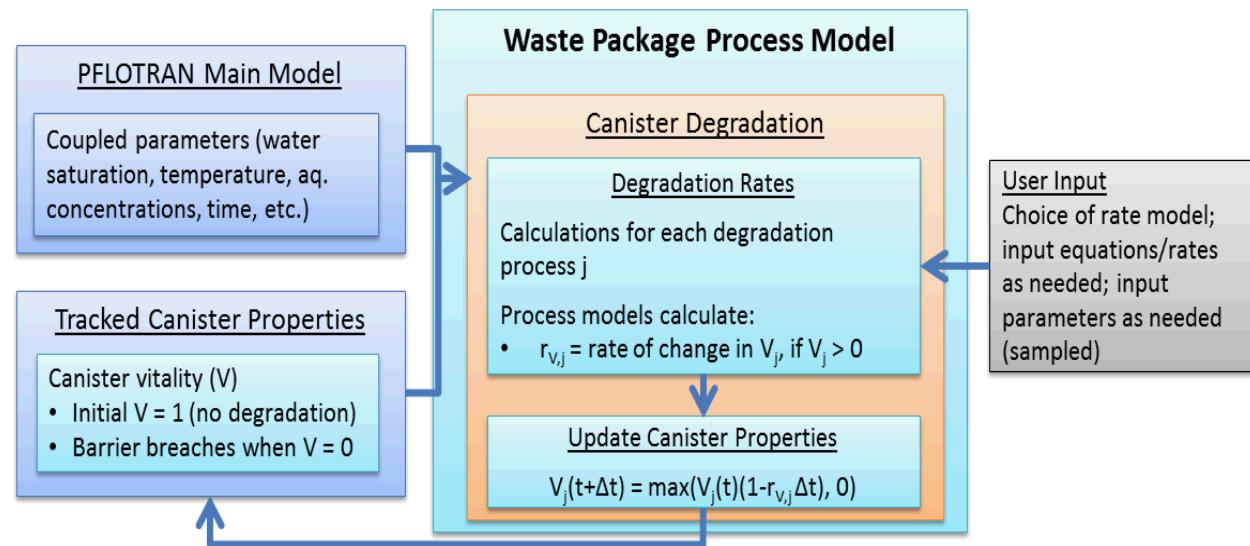
Canister Degradation Framework

■ Degradation leading to breach

- Framework allows for multiple processes to be modeled
 - General corrosion, localized corrosion, stress corrosion cracking (SCC), etc.
- Each process will effectively provide a rate of degradation of canister *vitality*

■ Canister vitality, V

- $V=1$ prior to degradation
- $V=0$ implies canister is breached
- For general corrosion,
 V = fraction of remaining canister wall thickness
- For SCC,
 V = fraction of canister thickness remaining between crack and breakthrough



■ Code

- Calculates dV/dt and V over time for each degradation process
- Process models needed; current placeholder simply samples rates from a distribution

■ Isotope processes

- Partitioning to colloids
- Partitioning to gas phase
- Decay and ingrowth for 4+ generations
- Solid solution modeling

■ Source term processes

- Canister vitality degradation process models
 - *Including early failure*
- Canister performance process models for after breach
- Improvements to waste form dissolution process models
- Neutron activation process mode

