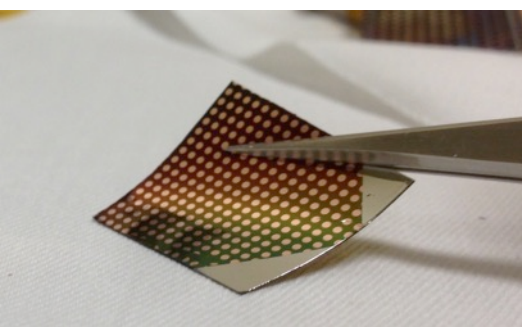
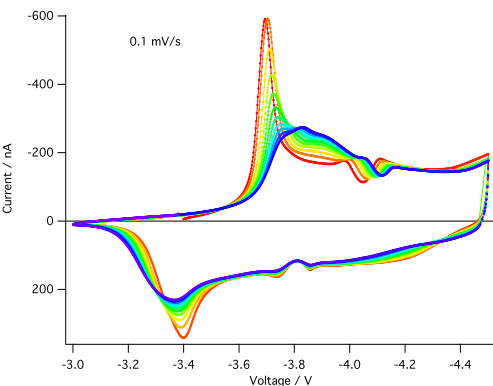




Solid-state rechargeable Li-ion batteries prepares at room-temperature over flexible substrates

Farid El Gabaly, N. Ware, E. Fuller, S. Vitale,
F. Leonard, A. Talin

*Sandia National Laboratories
Livermore, CA*



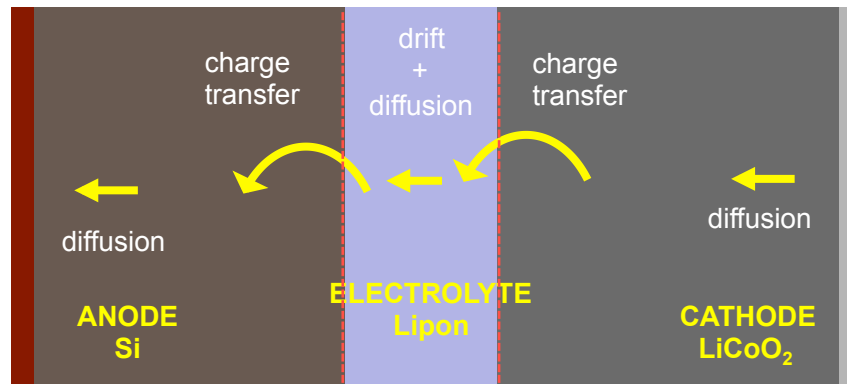
*Exceptional
service
in the
national
interest*



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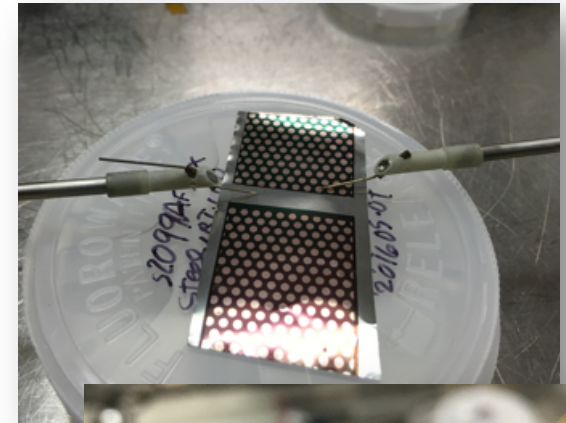
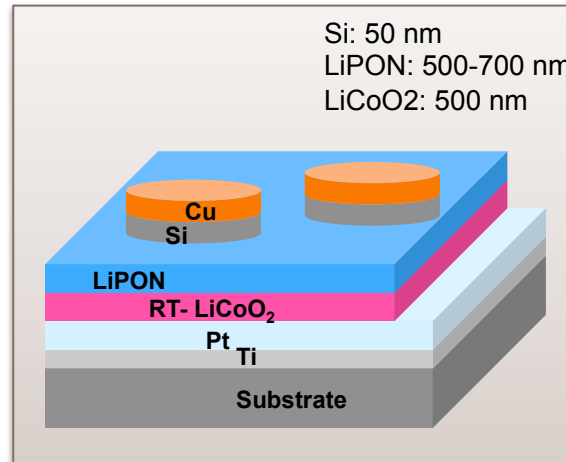
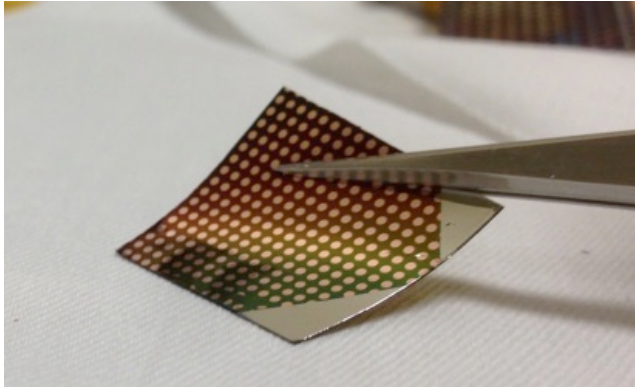
Thin-film Solid State Li-ion Batteries

- Solid state Li-ion batteries (SS-LIBs) provide long cycle life, very low self-discharge rates, and integrability with a variety of materials
- One of the most stable thin-film SS-LIBs has LiCoO_2 cathode, LiPON solid state anode, and LiPON or Si anodes

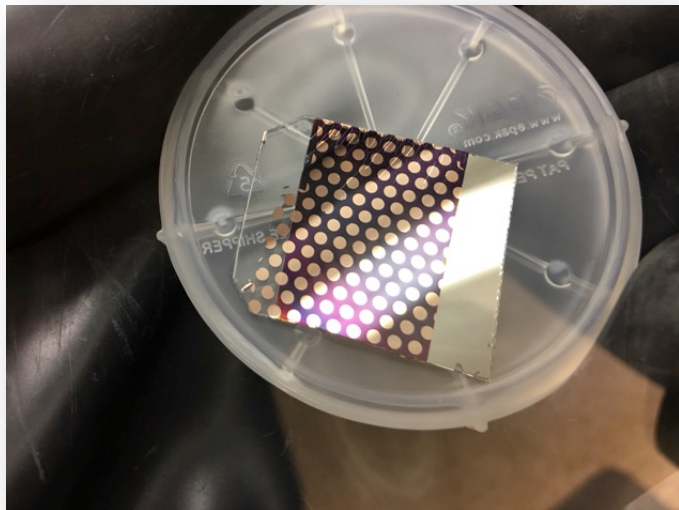


RT-LCO batteries

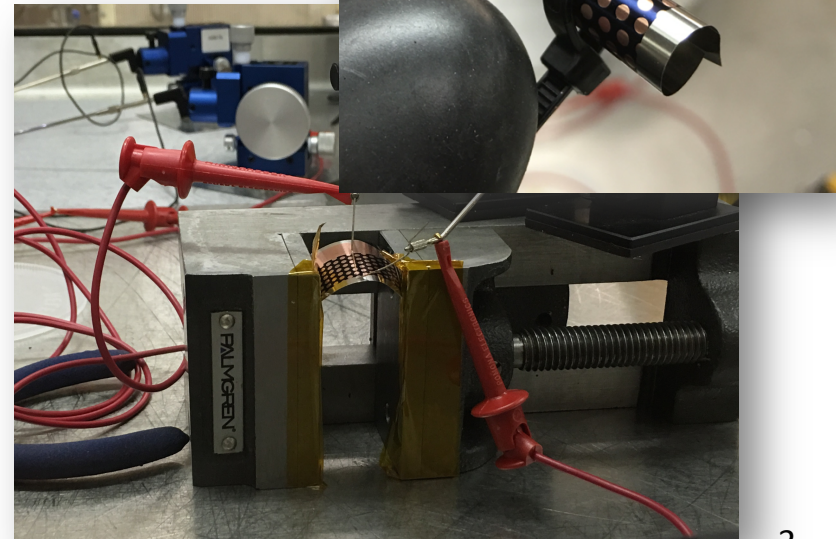
PMMA -Poly(methyl methacrylate),
acrylic, 0.125 mm thick



Polycarbonate, 0.5 mm thick

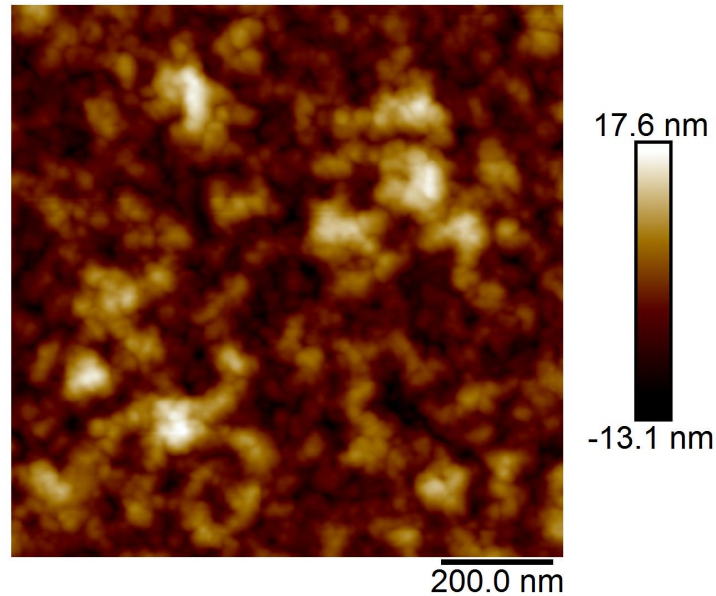


Thin stainless steel

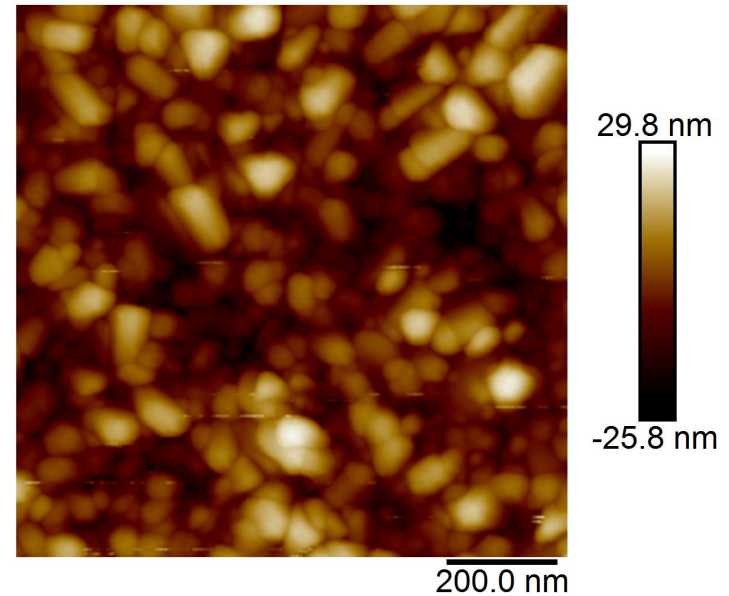


AFM LCO's

LiCoO_2 thin film annealing

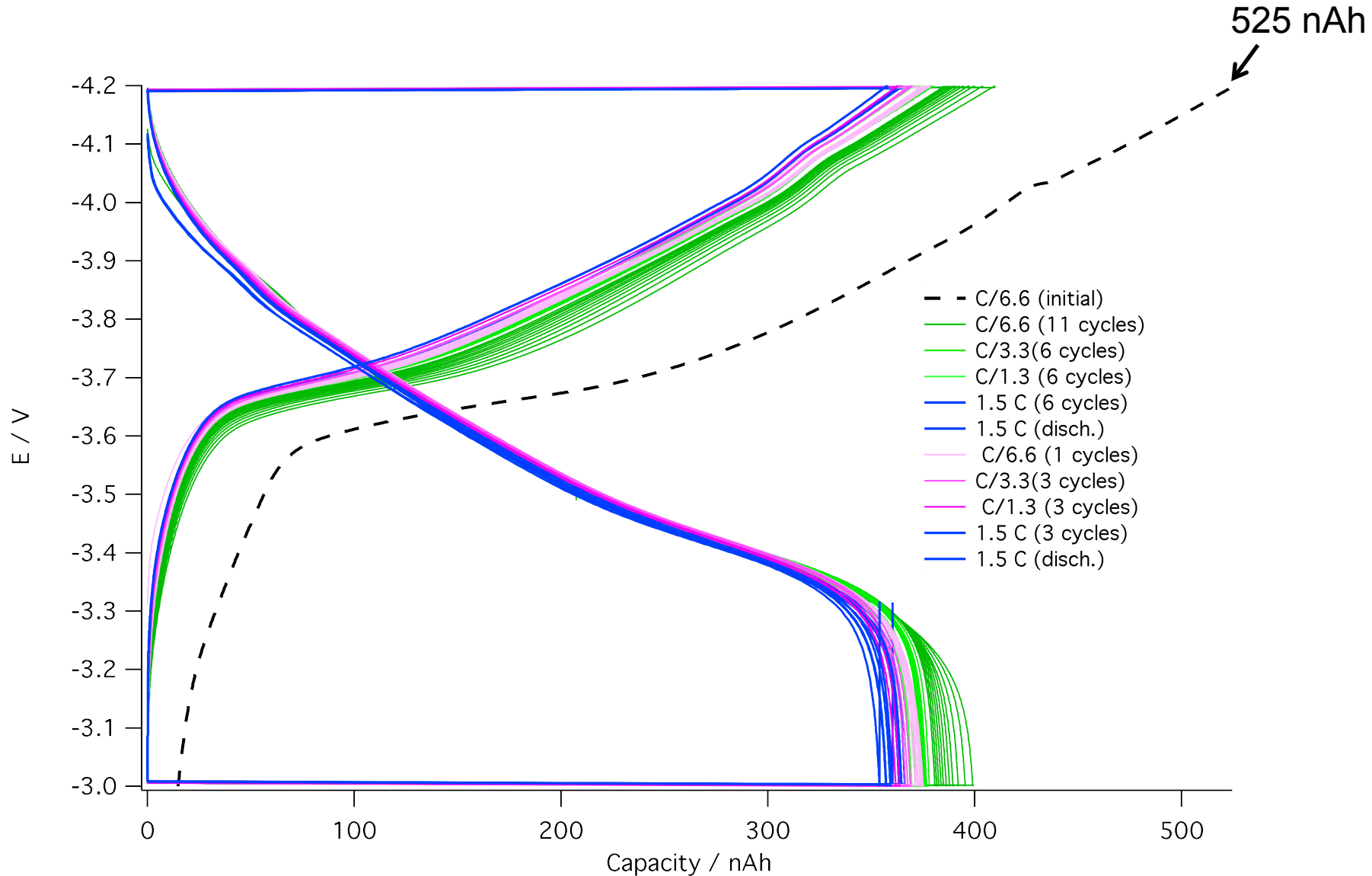


RT-LCO



HT-LCO

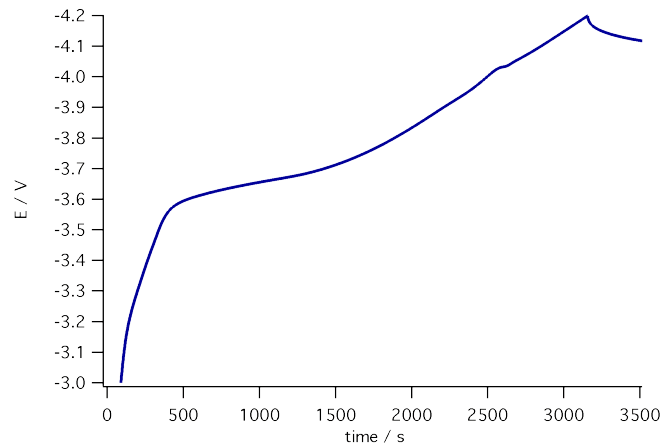
Cycling test (41 cycles, 15 days)



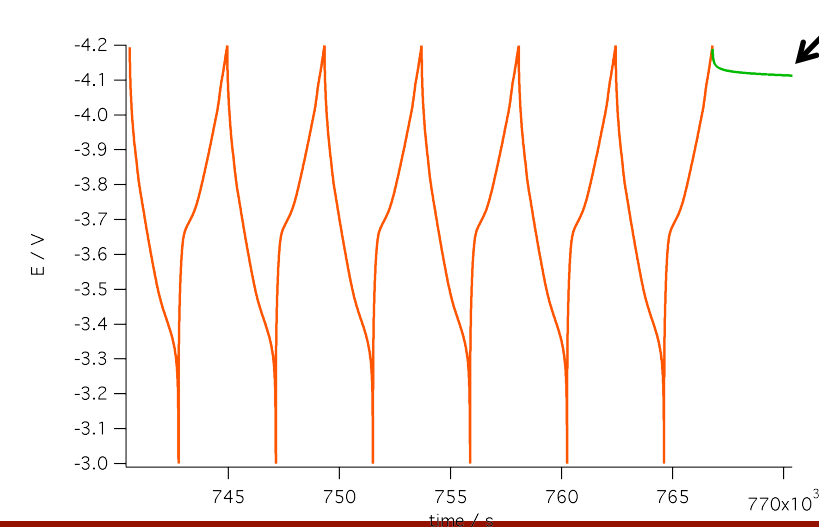
Theoretical capacity: 546 nAh

Maintaining the charge

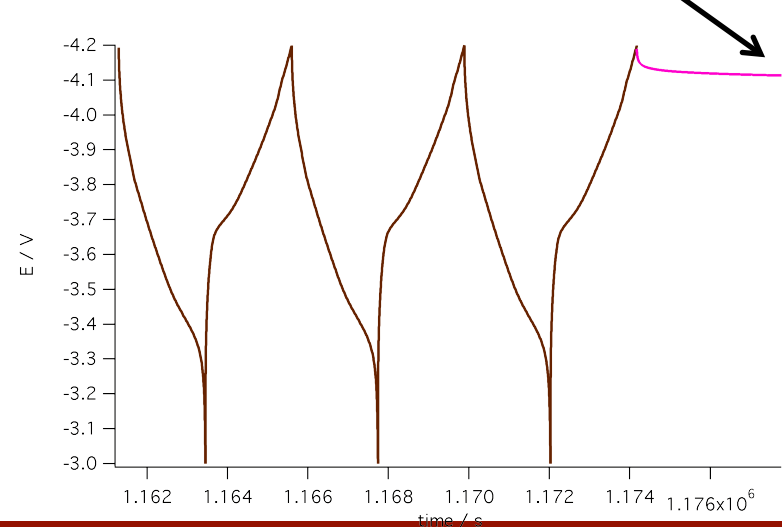
After the first charge at C/6.6:
4.12V after 1h



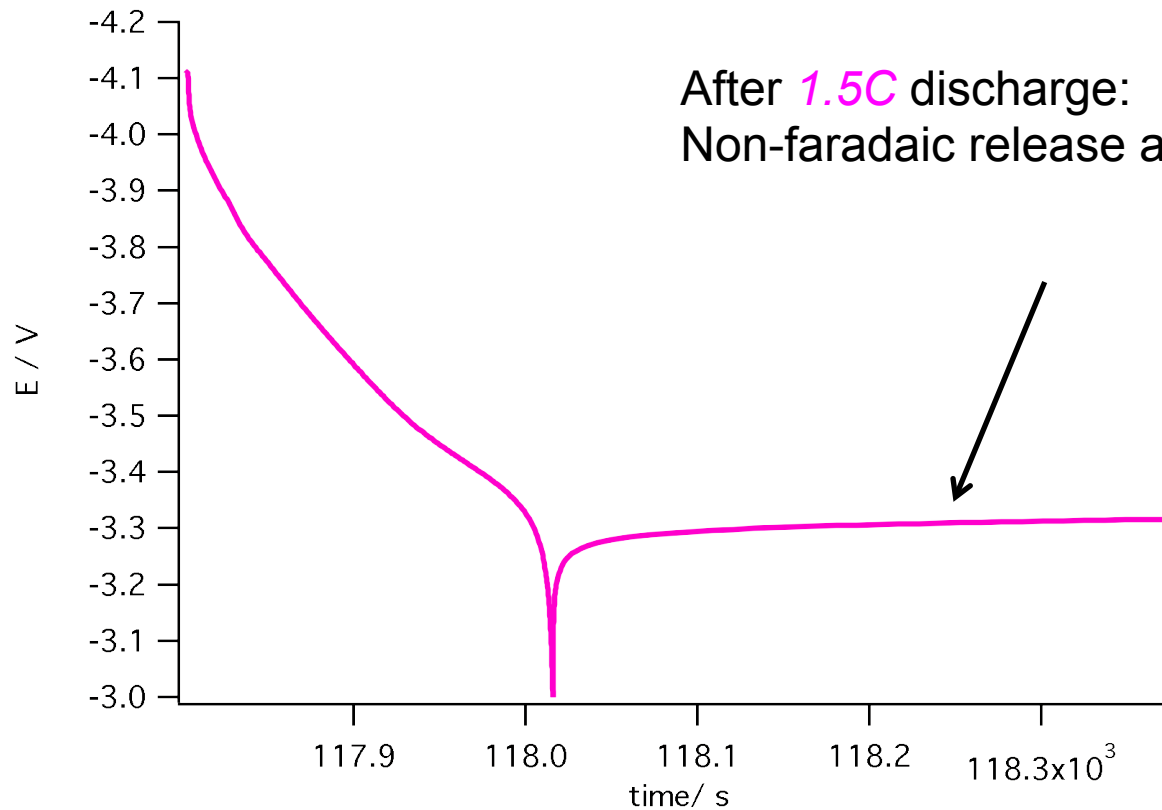
After 30 cycles
ending at 1.5 C rate:
4.113V after 1h



After 40 cycles
ending at 1.5 C rate:
4.114V after 1h



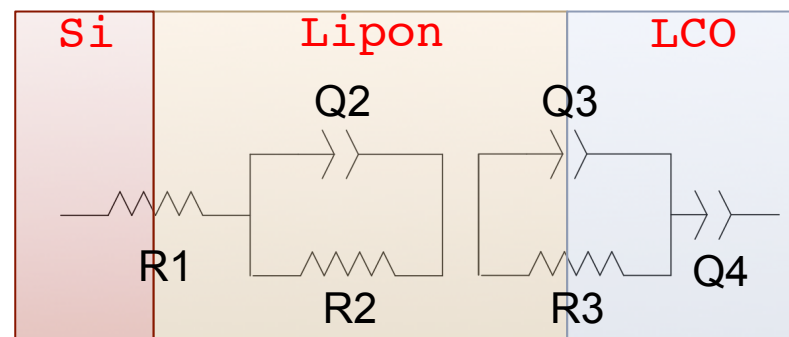
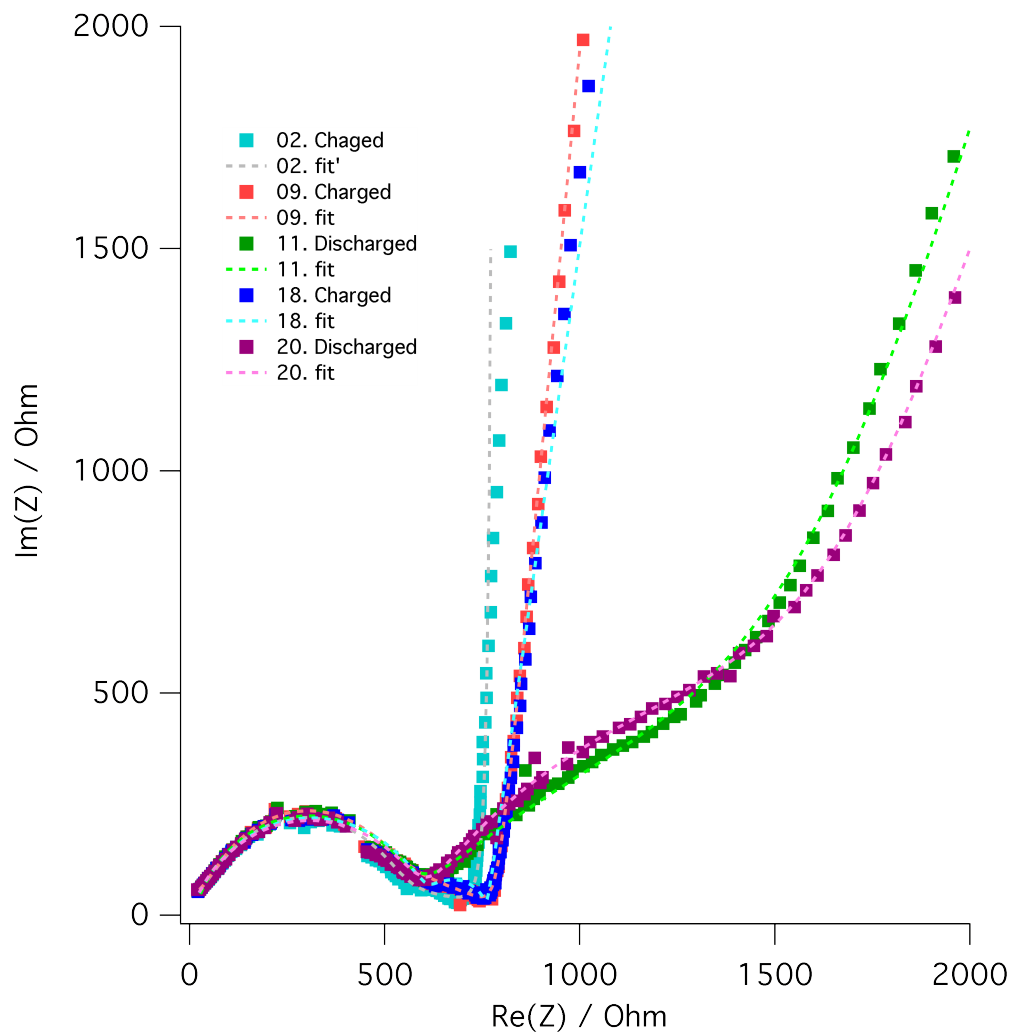
Maintaining the charge



After 1.5C discharge:

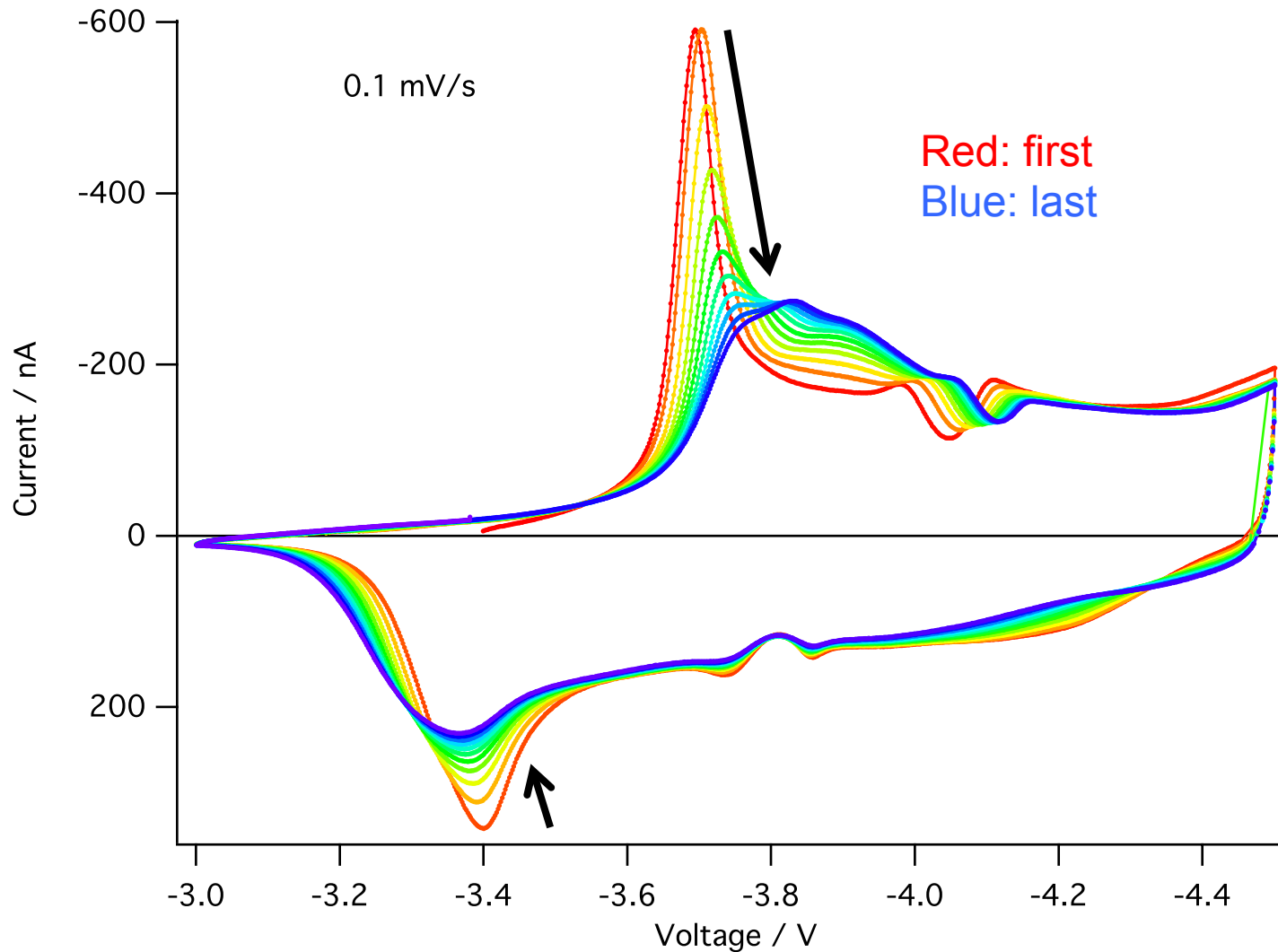
Non-faradaic release and plateau at 3.32V after 1h

EIS equivalent circuit



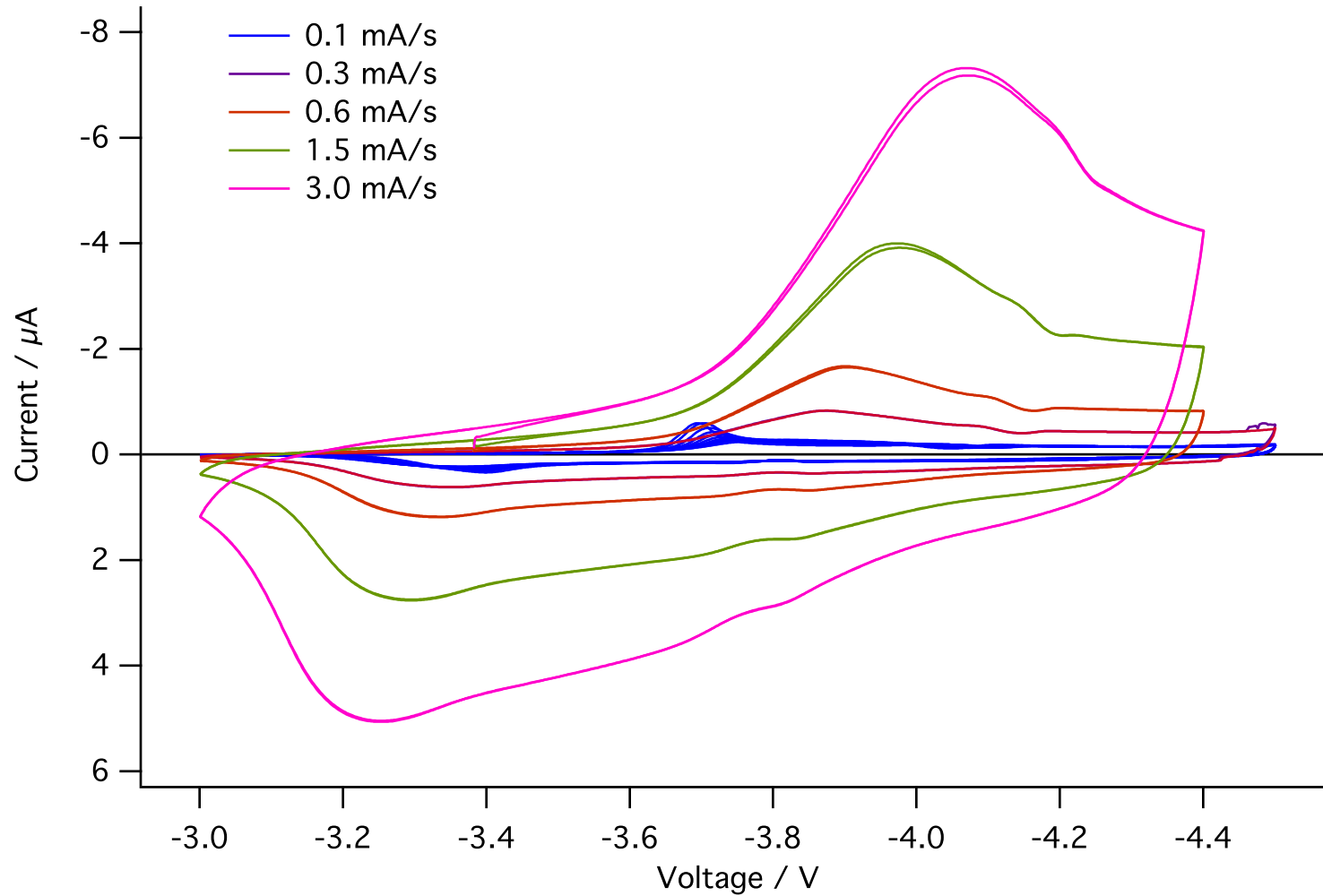
Seq.#	Cycles	R1 / Ohm	R2 / Ohm	R3 / Ohm
02.	1 (4.1 V)	84.43	5,866	1,146
09.	30 (4.1 V)	17.58	4,081	3,898
11.	31 (3.0 V)	22.96	6,078	8,957
18.	40 (4.1 V)	16.36	6,212	1,374
20.	41 (3.0 V)	10.8	5,928	10,973

SS Battery s1040A: CV 0.1 mV/s

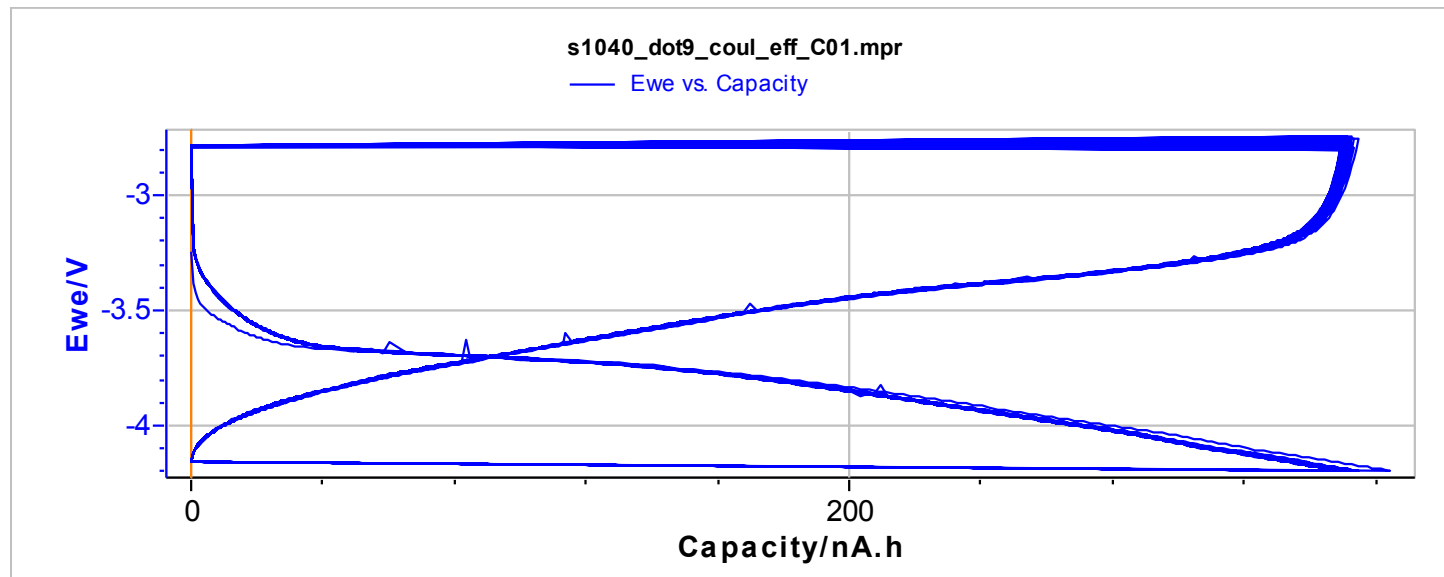
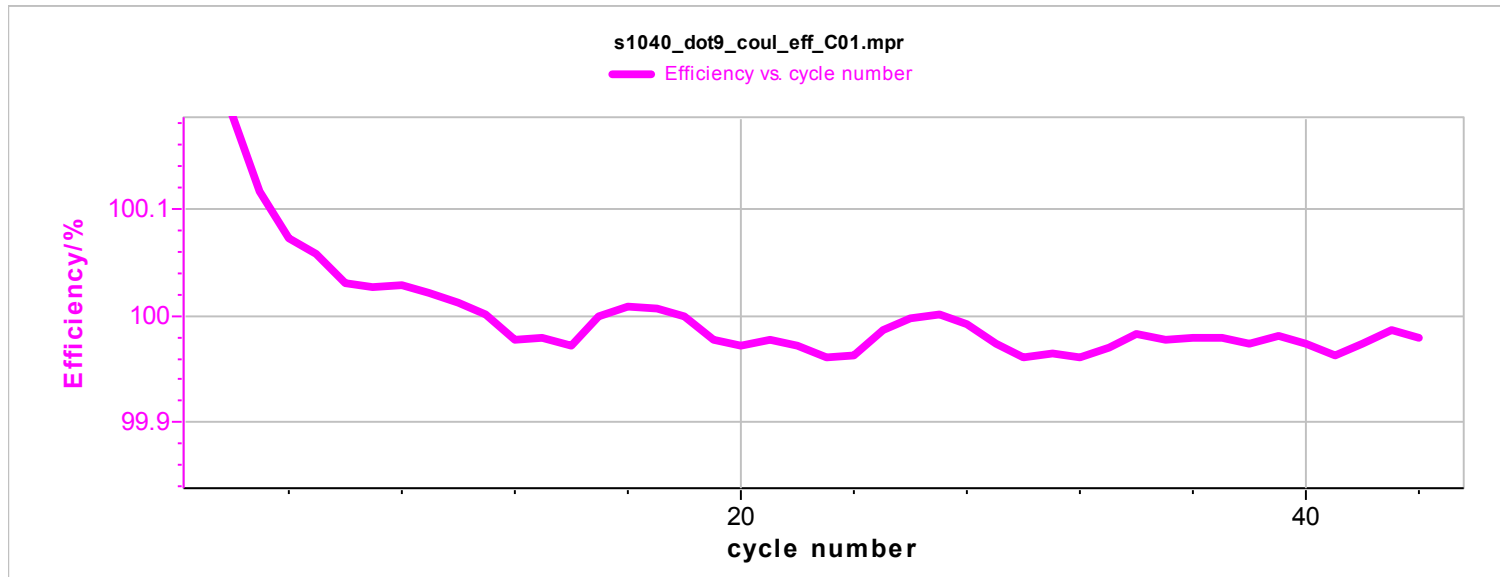


Starting with a fresh battery

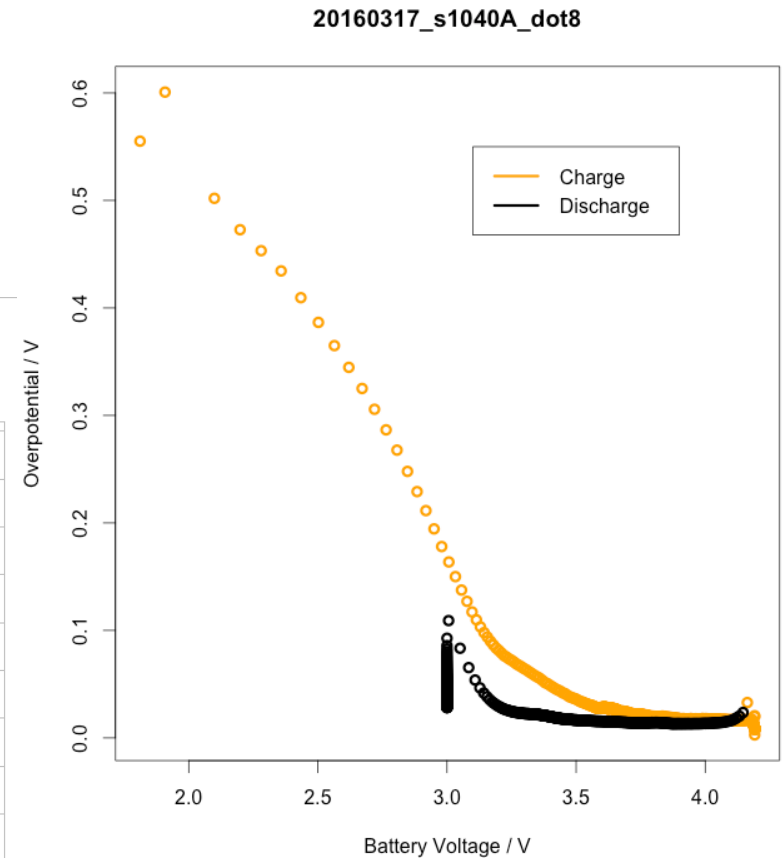
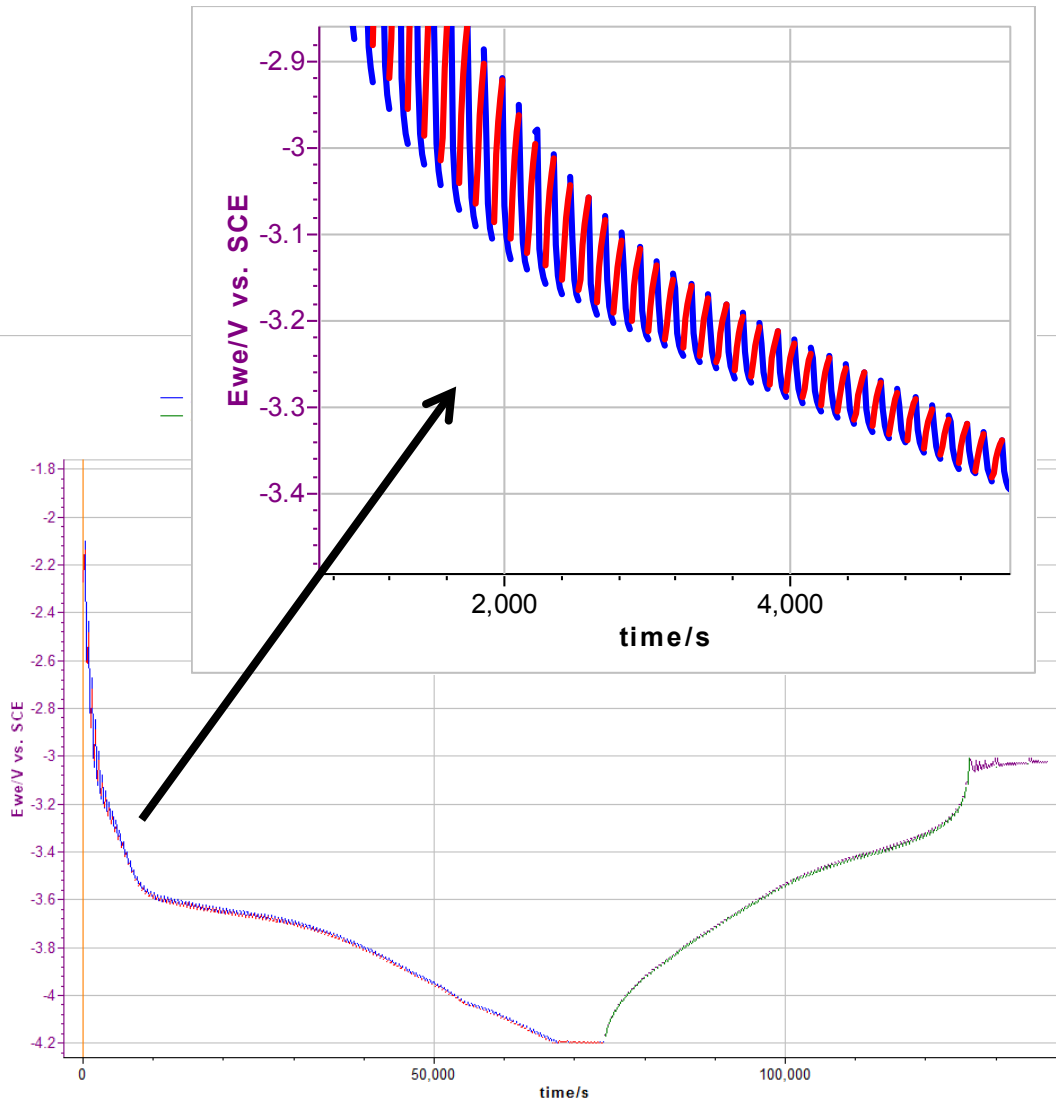
SS Battery s1040A: CVs



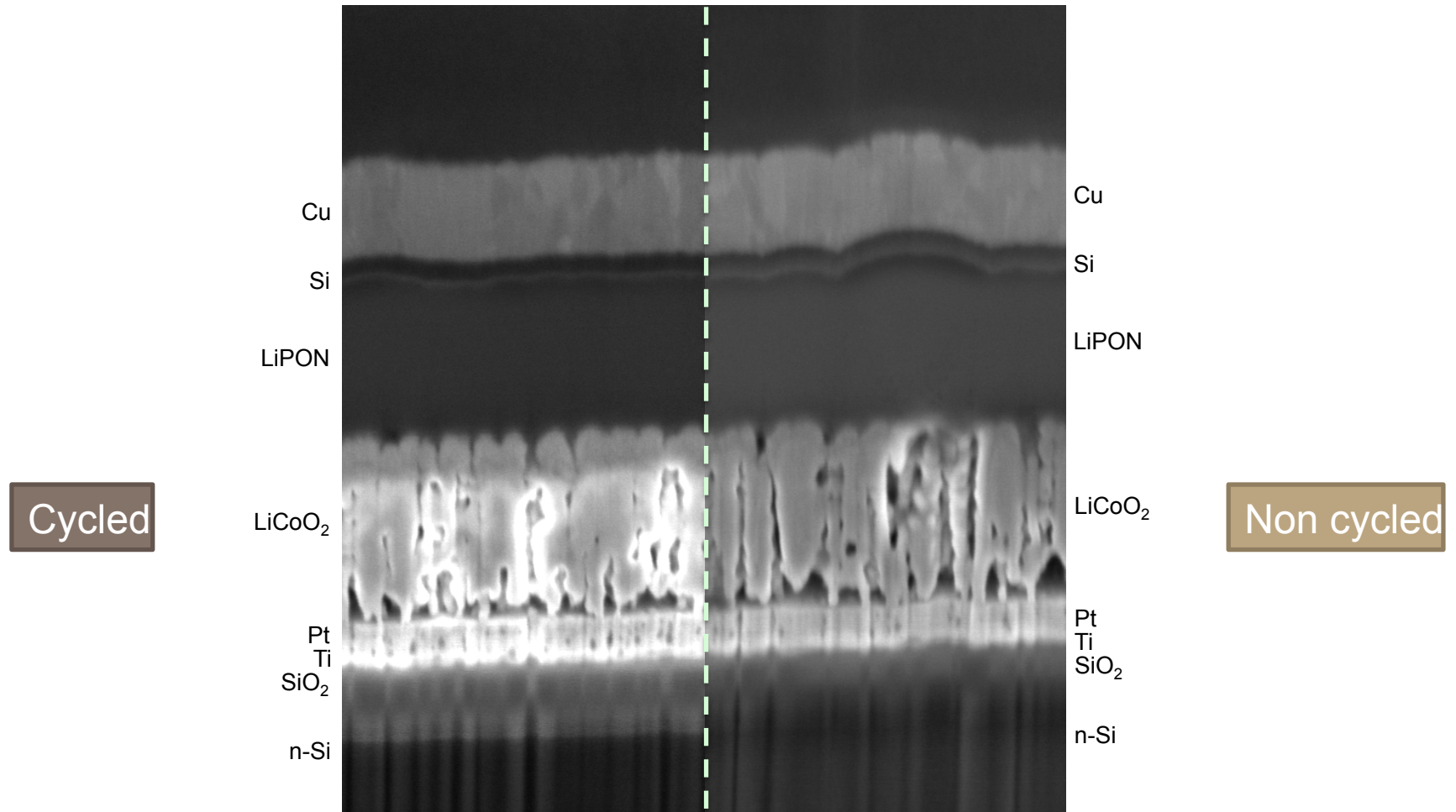
coulombic efficiency: 45 cycles



Current interrupt: Solid State Battery



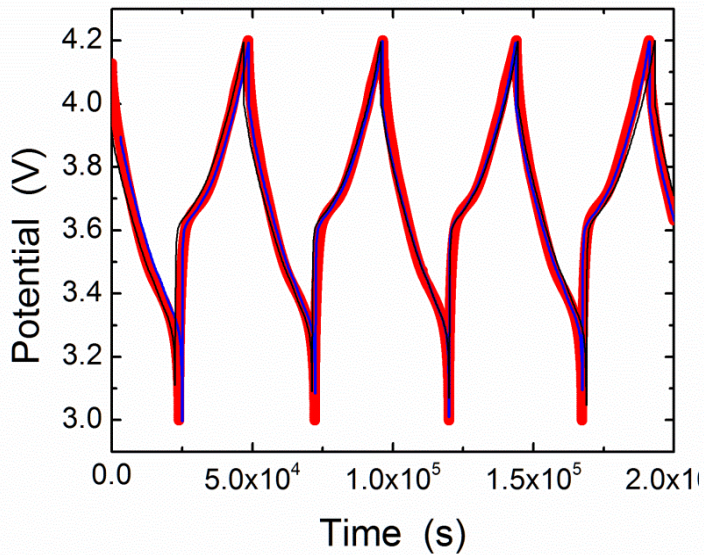
FIB/SEM s1040B cross-section



Demanding power from SS-LIBs(1/3)

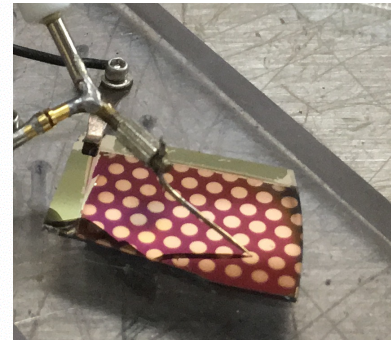
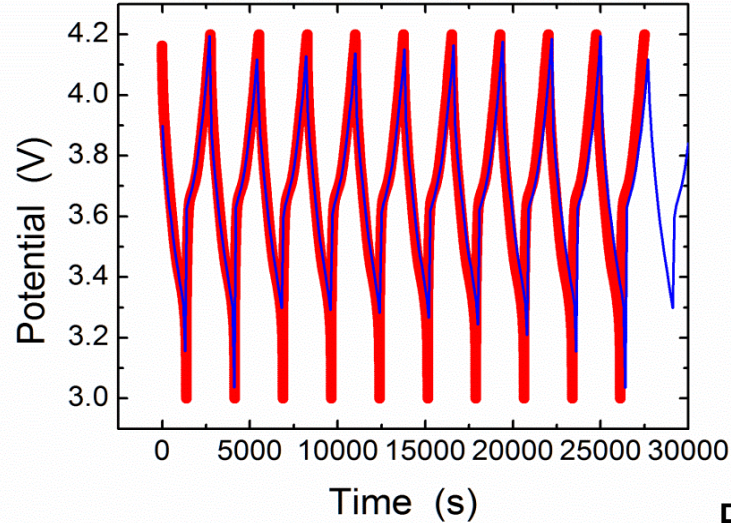
A

60nA = 2.64 $\mu\text{A}/\text{cm}^2 \leftarrow C/10$



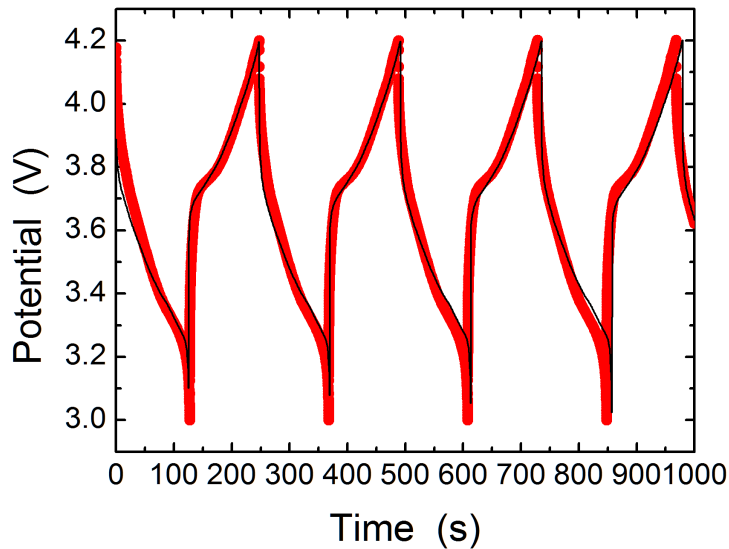
B

1 μA = 44.25 $\mu\text{A}/\text{cm}^2$



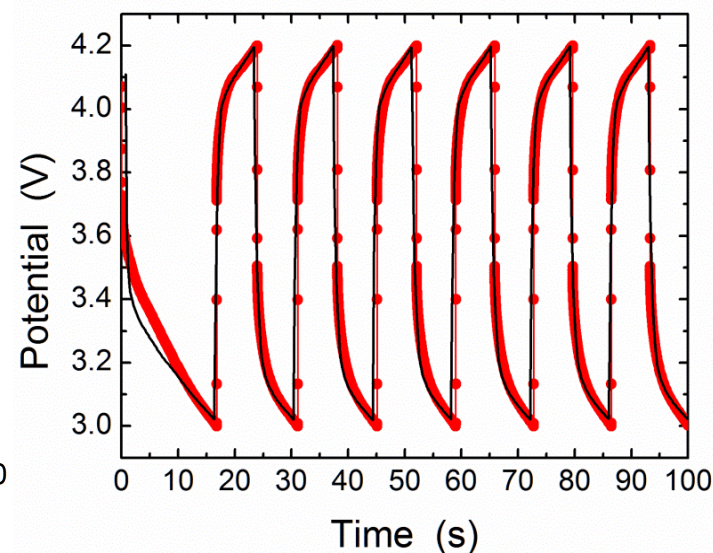
C

10 μA = 442.5 $\mu\text{A}/\text{cm}^2$



D

50 μA = 2212.5 $\mu\text{A}/\text{cm}^2$

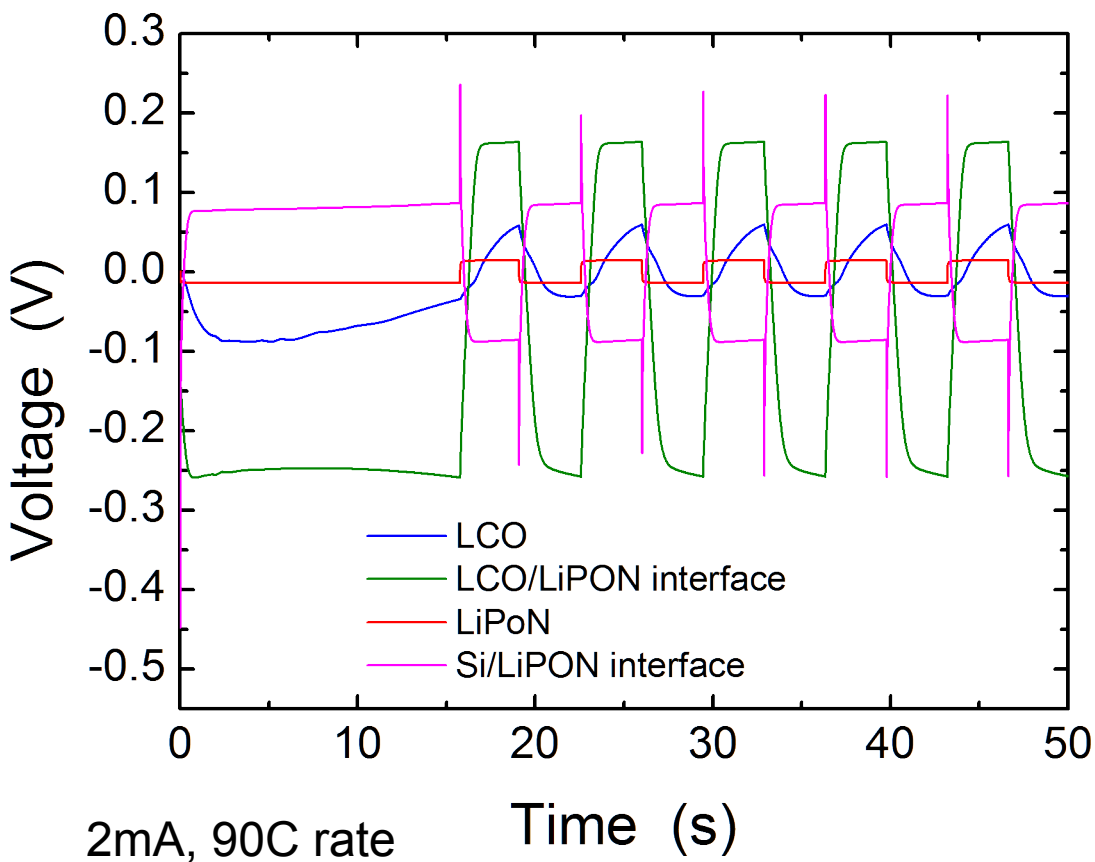


Red: experiment
Lines: model

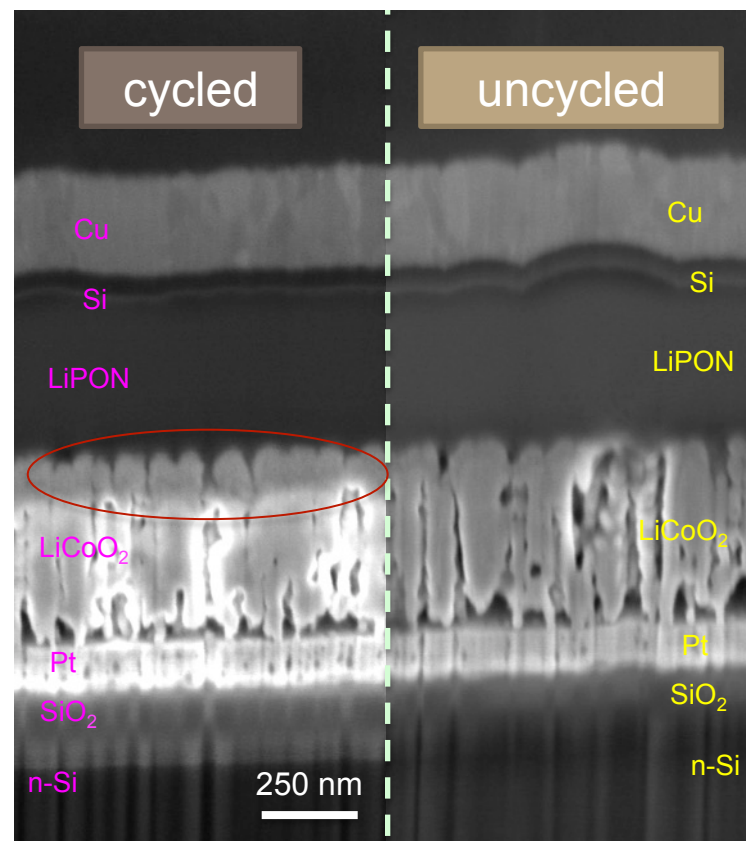
$\leftarrow 2\text{mA}$, 90C rate!
But...
capacity is 1/4

What is happening inside the battery? (2/3)

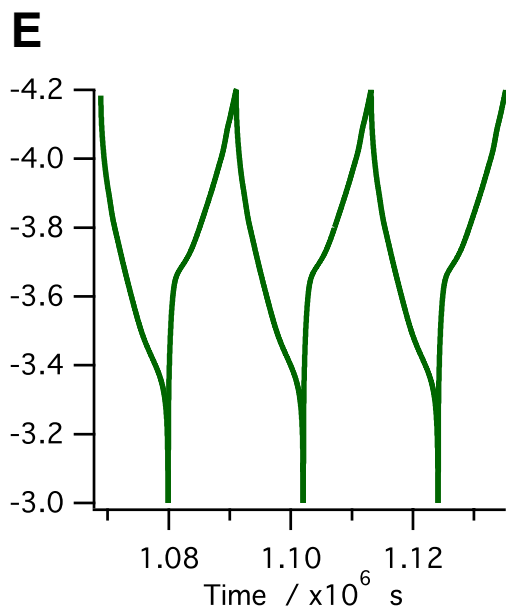
E Overpotentials extracted after fitting the experimental cycling curves with our model



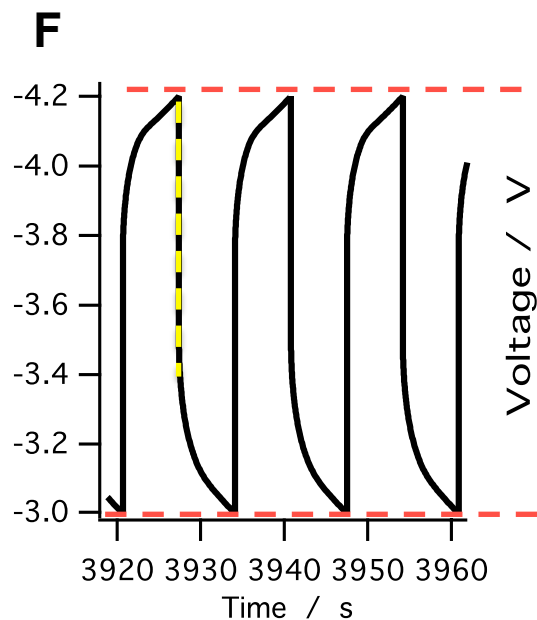
FIB/SEM cross section of the battery before and after cycling



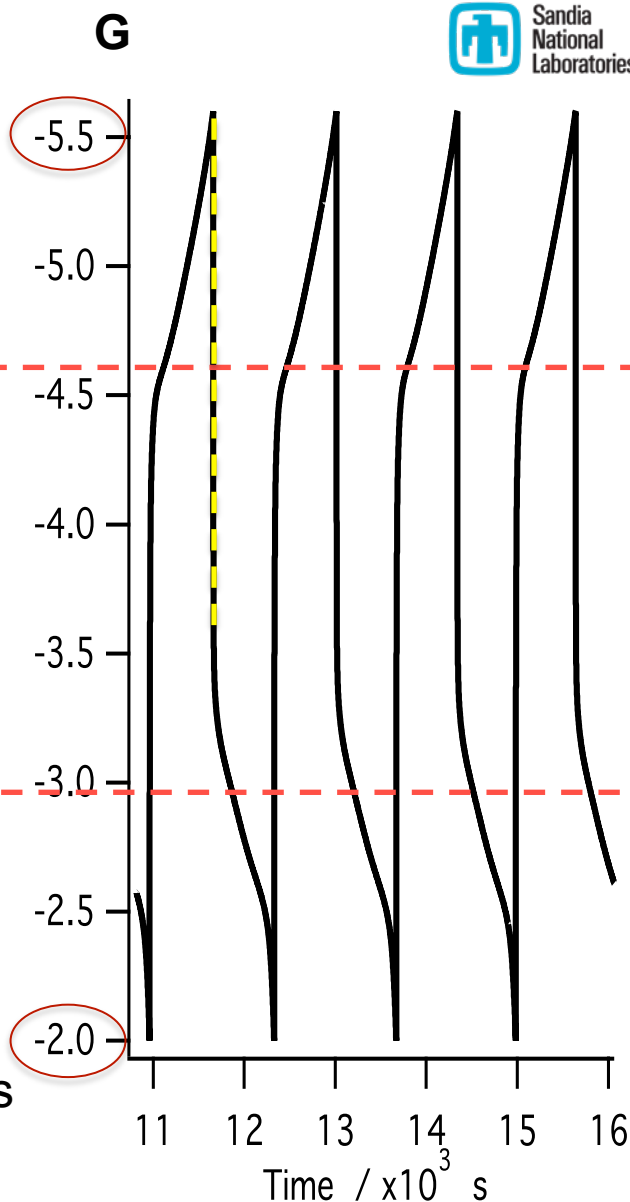
Extended voltage-range cycling



Full charge in 5h (1/5C)
Full capacity



Full charge in 40s (90C)
Low capacity due
to potential lost at interfaces



High Power + High capacity