Accepted Manuscript

The HITRAN2016 Molecular Spectroscopic Database

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PII: \$0022-4073(17)30107-3 DOI: 10.1016/j.jqsrt.2017.06.038

Reference: JQSRT 5765

To appear in: Journal of Quantitative Spectroscopy & Radiative Transfer

Received date: 23 February 2017 Revised date: 29 June 2017 Accepted date: 29 June 2017

Please cite this article as: I.E. Gordon, L.S. Rothman, C. Hill, R.V. Kochanov, Y. Tan, P.F. Bernath, M. Birk, V. Boudon, A. Campargue, K.V. Chance, B.J. Drouin, J.-M. Flaud, V.I. Perevalov, R.R. Gamache, J.T. Hodges, D. Jacquemart, A. Perrin, K.P. Shine, M.-A.H. Smith, J. Tennyson, G.C. Toon, H. Tran, V.G. Tyuterev, A. Barbe, A.G. Császár, T. Furtenbacher, J.-M. Hartmann, J.J. Harrison, A. Jolly, T.J. Johnson. T. Karman, I. Kleiner, A.A. Kyuberis, J. Loos, O.M. Lyulin, S.T. Massie, S.N. Mikhailenko, N. Moazzen-Ahmadi, H.S.P. Müller, O.V. Naumenko, A.V. Nikitin, O.L. Polyansky, M. Rey, M. Rotger, S.W. Sharpe, K. Sung, E. Starikova, S.A. Tashkun, J. Vander Auwera, G. Wagner, J. Wilzewski, P. Wcisło, S. Yu, E.J. Zak, The HITRAN2016 Molecular Spectroscopic Database, Journal of Quantitative Spectroscopy & Radiative Transfer (2017), doi: 10.1016/j.jgsrt.2017.06.038

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Highlights

- HITRAN2016 molecular spectroscopic database is described.
- Dynamic web interface at www.hitran.org is introduced
- HITRAN Application Programming Interface is introduced
- Substantial extent of the amount and quality of the data highlighted
- Many new spectroscopic parameters are now available in HITRAN



The HITRAN2016 Molecular Spectroscopic Database

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Keywords: HITRAN; Spectroscopic database; Molecular spectroscopy; Molecular absorption; Spectroscopic line parameters; Absorption cross-sections; Collision-induced Absorption; Aerosols

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Abstract

This paper describes the contents of the 2016 edition of the HITRAN molecular spectroscopic compilation. The new edition replaces the previous HITRAN edition of 2012 and its updates during the intervening years. The HITRAN molecular absorption compilation is composed of five major components: the traditional line-by-line spectroscopic parameters required for high-resolution radiative-transfer codes, infrared absorption cross-sections for molecules not yet amenable to representation in a line-by-line form, collision-induced absorption data, aerosol indices of refraction, and general tables such as partition sums that apply globally to the data. The new HITRAN is greatly extended in terms of accuracy, spectral coverage, additional absorption phenomena, added line-shape formalisms, and validity. Moreover, molecules, isotopologues, and perturbing gases have been added that address the issues of atmospheres beyond the Earth. Of considerable note, experimental IR cross-sections for almost 300 additional molecules important in different areas of atmospheric science have been added to the database.

The compilation can be accessed through www.hitran.org. Most of the HITRAN data have now been cast into an underlying relational database structure that offers many advantages over the long-standing sequential text-based structure. The new structure empowers the user in many ways. It enables the incorporation of an extended set of fundamental parameters per transition, sophisticated line-shape formalisms, easy user-defined output formats, and very convenient searching, filtering, and plotting of data. A powerful application programming interface making use of structured query language (SQL) features for higher-level applications of HITRAN is also provided.

1. Introduction

This article describes the data and software that have been added, modified, or enhanced in the HITRAN (**High** Resolution **Trans**mission) compilation since the previous edition of 2012 [1] (hereafter called HITRAN2012 in the text). The line-by-line portion and the absorption cross-sections of the new edition, hereafter called HITRAN2016, have now been cast into an underlying relational database structure as opposed to the long-standing fixed-length ASCII record format. A user-friendly powerful internet tool, HITRAN*online* (accessible through www.hitran.org), is provided to the user to query, filter, and plot sections of the data and to retrieve outputs in a host of convenient formats (see Hill et al. [2]).

The HITRAN compilation is composed of several components that include (1) line-by-line spectroscopic parameters for high resolution molecular absorption and radiance calculations (from the microwave through the ultraviolet region of the spectrum); (2) infrared absorption cross-sections (generally representing absorption by molecules that have very dense spectra or many low-lying vibrational modes); (3) collision-induced absorption datasets; (4) tables of aerosol refractive indices; and (5) global data that apply in a general manner to the archive. The updates to these five portions of HITRAN as well as the new underlying system of data structure with accompanying internet user interface will be discussed in the following sections.

The new edition of the HITRAN database substantially increases the database's potential to accurately model radiative processes in the atmosphere of the Earth and other planets. Apart from improving the accuracy of the existing parameters, more parameters have been introduced, and new bands and isotopologues added. Two gases (C₂N₂ and COCl₂) are introduced to the database for the first time. Finally, a very drastic expansion of the cross-sectional part of the database is featured in this new edition.

It is necessary to call attention to some specifics of the HITRAN database. The units used throughout HITRAN editions including this one do not strictly adhere to the SI system for both historical and application-specific reasons. Thus cm⁻¹ (reciprocal centimeter) is seen throughout, as is atm (atmosphere) for pressure (in SI units of Pascals, 101325 Pa = 1 atm). Also, the symbol ν is used throughout for line position in cm⁻¹, thereby dropping the tilde ($\tilde{\nu}$) that is the official designation of wavenumber. The HITRAN unit for intensity is traditionally expressed as cm⁻¹/(molecule cm⁻²) rather than simplifying to the equivalent cm/molecule.

1.1 Overview of parameters (including new ones).

The traditional and probably most applied portion of HITRAN is the line list of highresolution spectral parameters. These line-by-line parameters reflect values of individual transitions between energy levels of rovibronic states that are required by high-resolution radiative-transfer codes. These parameters are shown in Tables 1.1 and 1.2. Table 1 gives an overview of parameters that are traditionally provided in the ".par" format as per HITRAN2004 [3] formalism as well as broadening and shift parameters due to the pressure of H₂, He and CO₂ (see Wilzewski et al. [4], Hill et al. [2] and Kochanov et al. [5] for details). Table 2 provides information on the parameters required for non-Voigt line-shape representations (see Wcisło et al. [6], Hill et al. [2] and Kochanov et al. [5] for details). Ideally, they would arise from theoretically-consistent calculations; however, since the emphasis in HITRAN is to provide the user with the most accurate values available, they are often the values obtained from controlled laboratory measurements when quantum-mechanical calculations are not yet of comparable accuracy. Nevertheless, theoretical calculations and semi-empirical formulae are widely used in HITRAN to both interpolate and extrapolate needed parameters for transitions missing from the limited observed set.

Section 2 presents a description of changes made to the line-by-line portion of HITRAN. It is organized in the order of the molecules in HITRAN (a sequential number related to the entry of the molecule into HITRAN). The discussions in the molecular sub-sections give an overview of the addition of new bands, replacement of line positions and/or intensities, and new or modified line-shape parameters where applicable. There are also citations to more detailed studies of the modifications.

Table 1 Parameter identifiers in HITRANonline

Parameters from the HITRAN2004 format in HITRANonline Molecule ID — molec_id HITRAN integer ide molecule	, ,
Molecule ID – molec_id HITRAN integer ide molecule	, ,
molecule	, ,
molecule	, ,
	entifying the
	entifying the
Isotopologue – local_iso_id HITRAN integer ide	
ID isotopologue of a pa	
υ cm ⁻¹ nu Transition wavenum	ıber
S cm ⁻¹ / sw Transition intensity,	weighted by
(molecule cm ⁻ isotopologue abunda	ance
2)	
A s ⁻¹ a Einstein A-coefficien	
γ_{air} cm ⁻¹ atm ⁻¹ gamma_air Air-broadened Lorent	
at half maximum (H	,
coefficient (for Voig	
γ_{self} cm ⁻¹ atm ⁻¹ gamma_self Self-broadened Lore	
coefficient (for Voig	gt lineshape)
E" cm ⁻¹ elower Lower state energy	
n_air Temperature-depend	dence exponent
for γ_{air}	
δ_{air} cm ⁻¹ atm ⁻¹ delta_air Air pressure-induced	d line shift
V'b — global_upper_quanta Upper-state "global'	' quanta in
HITRAN2004forma	ıt .
V" b – global_lower_quanta Lower-state "global"	" quanta in
HITRAN2004 forma	at
Q' b - local_upper_quanta Upper-state "local"	
HITRAN2004forma	
Q" b - local_lower_quanta Lower-state "local"	
HITRAN2004forma	
<i>Ierr</i> b ierr Ordered list of indic	es corresponding

			to uncertainty estimates of transition parameters
Iref ^b		iref	Ordered list of reference identifiers
irej		nei	for transition parameters
<i>a</i> '		gn	Upper state statistical weight
<i>g</i> '		gp	
g "		gpp	Lower state statistical weight
Metadata an	nd other specia	l parameters	
Transition ID	_	trans_id	Unique integer identifying the
			transition
Global	_	global_iso_id	Global integer ID identifying
Isotopologue		· ·	the isotopologue (unique across the
ID 1 2			whole database)
qns'	_	statep	Upper state quantum numbers in
1 "		1	HITRAN <i>online</i> format
qns''	_	statepp	Lower state quantum numbers in
-1~		~PP	HITRANonline format
par line	_	par_line	Complete representation of the
.par mic		par_mic	line in the HITRAN2004 160-
			character format
			character format
	for broadening	by new perturbing	character format g species, X ^r , and line mixing
		by new perturbing	Y
	for broadening	by new perturbing	Y
parameters			g species, X ^r , and line mixing
parameters			species, X, and line mixing Lorentzian HWHM coefficient
parameters Y _x			Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber <i>X</i>
parameters Y _x		gamma_X	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber <i>X</i> Temperature-dependence exponent
γ_x n_χ		gamma_X n_X	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X
γ_x n_x		gamma_X	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent
γ_{x} n_{x} n_{self}	cm ⁻¹ atm ⁻¹	gamma_X n_X n_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self}
γ_{x} n_{x} n_{self}	cm ⁻¹ atm ⁻¹	gamma_X n_X n_self delta_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift
γ_{x} n_{x} n_{self}	cm ⁻¹ atm ⁻¹	gamma_X n_X n_self delta_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift
γ_{x} n_{x} n_{self}	cm ⁻¹ atm ⁻¹	gamma_X n_X n_self delta_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence
γ_{x} n_{x} n_{self} δ_{air}	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air}
γ_{x} n_{x} n_{self} δ_{air}	cm ⁻¹ atm ⁻¹	gamma_X n_X n_self delta_self deltap_air	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence
γ_{x} n_{x} n_{self}	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air deltap_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence coefficient for δ_{self}
γ_{x} n_{x} n_{self} δ_{self} δ_{self}'	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air deltap_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence
γ_{x} n_{x} n_{self} δ_{air} δ_{self}'	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air deltap_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence coefficient for δ_{self}
parameters γ_{x} n_{x} n_{self} δ_{air} δ'_{self} $\delta'_{H_{2}}$	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air deltap_self deltap_H2	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence coefficient for δ_{self} Linear temperature dependence coefficient for δ_{self}
parameters γ_{x} n_{x} n_{self} δ_{air} δ'_{self}	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air deltap_self	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence coefficient for δ_{self} Linear temperature dependence coefficient for δ_{self} Pressure-induced line shift due
γ_{x} n_{x} n_{self} δ_{self}	cm ⁻¹ atm ⁻¹ - cm ⁻¹ atm ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹ cm ⁻¹ atm ⁻¹ K ⁻¹	gamma_X n_X n_self delta_self deltap_air deltap_self deltap_H2	Lorentzian HWHM coefficient (for Voigt lineshape) for broadening by perturber X Temperature-dependence exponent for γ_X Temperature-dependence exponent for γ_{self} Self-induced pressure line shift Linear temperature dependence coefficient for δ_{air} Linear temperature dependence coefficient for δ_{self} Linear temperature dependence coefficient for δ_{self}

			broadened environment
Y_{self}	cm ⁻¹ atm ⁻¹	y_self	First-order (Rosenkranz) line coupling
,			coefficient within Voigt profile; air-
			broadened environment

"The third column header, identifier, denotes the names of the variables used in the new relational database structure. ^b These parameters are given in Table 1 in the way they appear in the ".par" file. However, one should note that they are stored in a flexible and substantially more detailed way in the database. The names containing *X* correspond to the foreign broadeners, where *X* is the name of the perturbing species (CO₂, H₂, He). Unless explicitly given, the parameters imply transition between lower and upper states; double primes (") and primes (') are for lower and upper energy states, respectively.

Table 2 Parameter identifiers in HITRANonline for non-Voigt parameters

Parameters fo	Parameters for non-Voigt parameters							
Parameter	Units	Identifier ^a	Description					
$\gamma_0^{SDV}(X; 296)$	cm ⁻¹ atm ⁻¹	gamma_SDV_0_X	Air-(Self-) broadened Lorentzian					
		_296	HWHM coefficient in 296K (for SDV					
			profile)					
$\gamma_2^{SDV}(X; 296)$	cm ⁻¹ atm ⁻¹	gamma_SDV_2_X	Speed-dependence of the SDV					
		_296	halfwidth in 296K; air-(self-)					
			broadened case					
$\delta_0^{SDV}(X;296)$	cm ⁻¹ atm ⁻¹	delta_SDV_0_X	Air-(Self-) induced pressure line shift					
		_296	in 296K (for SDV profile)					
$\delta'^{SDV}(X;296)$	cm ⁻¹ atm ⁻¹ K ⁻¹	deltap_SDV_X_296	Linear temperature dependence					
			coefficient for $\delta_0^{SDV}(X; 296)$					
$n^{SDV}(X; 296)$	-	n_SDV_X_296	Temperature dependence exponent					
			around 296K for $\gamma_0^{SDV}(X; 296)$					
$Y_X^{SDV}(296)$	cm ⁻¹ atm ⁻¹	Y_SDV_X_296	First-order (Rosenkranz) line coupling					
			coefficient within SDV ^b profile; air-					
			(self-) broadened case					
γ_X^G	cm ⁻¹ atm ⁻¹	gamma_g_X	Air-(Self-) broadened Lorentzian					
) '		HWHM coefficient in 296K (for					
			Galatry profile)					
β_X^G	cm ⁻¹ atm ⁻¹	beta_g_X	Dicke narrowing parameter for the					
			air- (self-) broadened Galatry line					
			profile					
Y	1							
$\gamma_0^{HT}(X;T_{ref})$	cm ⁻¹ atm ⁻¹	gamma_HT_0_X_T	Speed-averaged HTP halfwidth in					
			temperature range around $T=T_{ref}^{d}$ due					
			to perturber <i>X</i>					
$n^{HT}(X;T_{ref})$	_	n_HT_X_T	Temperature dependence exponent in					
			the range corresponding to $T=T_{ref}$ for					

			$\gamma_0^{\scriptscriptstyle HT}(X;T_{\scriptscriptstyle ref})$
$\gamma_2^{HT}(X;T_{ref})$	cm ⁻¹ atm ⁻¹	gamma_HT_2_X_T	Speed-dependence of the HTP
12 \			halfwidth in temperature range around
			$T=T_{ref}$ due to perturber X .
$\delta_0^{HT}(X;T_{ref})$	cm ⁻¹ atm ⁻¹	delta_HT_0_X_T	Speed-averaged line shift of the HTP
0 \ rej /			in temperature range around $T=T_{ref}$
			due to perturber <i>X</i>
$\delta'^{HT}(X;T_{ref})$	cm ⁻¹ atm ⁻¹ K ⁻¹	deltap_HT_0_X_T	Linear temperature dependence
, rej			coefficient for $\delta_0^{HT}(X;T_{ref})$ in
			temperature range around $T=T_{ref}$
$\delta_2^{HT}(X;T_{ref})$	cm ⁻¹ atm ⁻¹	delta_HT_2_X_T	Speed-dependence of the HTP line
2 \ rej /			shift in temperature range around
			$T=T_{ref}$ due to perturber X
$\nu_{VC}^{HT}(X)$	cm ⁻¹ atm ⁻¹	nu_HT_X	Frequency of velocity changing
70 ()			collisions in the HTP formalism
$\kappa^{HT}(X)$	_	kappa_HT_X	Temperature dependence of
			$v_{VC}^{HT}(X)$
η^{HT}	_	eta_HT_X	Correlation parameter in HTP
			formalism
$Y^{HT}(X; 296)$	cm ⁻¹ atm ⁻¹	Y_HT_X_296	First-order (Rosenkranz) line coupling
•			coefficient within HTP; air-(self-)
		X Y	broadened case

The third column header, identifier, denotes the names of the variables used in the new relational database structure. The names containing the X literal correspond to air- or selfbroadening

With the recent advances in both laboratory spectrometers and the power of theoretical treatments such as ab initio calculations, the accuracy and the access to weak, yet important, transitions have had a big impact on this edition of HITRAN.

Table 3 provides an overview of changes and additions to the database with respect to the HITRAN2012 edition.

Table 3. Molecules and isotopologues represented in line-by-line portion of HITRAN.

b SDV stands for speed-dependent Voigt.

HT stands for the Hartmann-Tran profile [7,8].

^d For the four temperature ranges and corresponding T_{ref} values consult Weisło et al. [6] and Sections 1.2 and 2.45 below.

Molecule	Isotopo -logue ^a	HITRAN2012 Spectral Coverage (cm ⁻¹)	HITRAN2016 Spectral Coverage (cm ⁻¹)	HITRAN2012 Number of Transitions	HITRAN2016 Number of Transitions	Non-Voigt line shape for at least some transitions?	Broadeni ng by H ₂ , He and CO ₂
	161	0 – 25711	0 – 25711	142 045	146 878	Yes	
	181 ^b	0 - 19918	0 - 19918	39 903	39 903	103	
40 H O	171 ^b	0 – 19946	0 – 19946	27 544	27 544		
(1) $\mathbf{H_2O}$	162	0 - 22708	0 - 22708	13 237	56 345		
	182	0 - 3825	0 - 10729	1 611	10 664		
	172	1234 – 1599	0 - 10703	175	6 313		
	262	c	0 - 12812	c	32 184	2	
	626	345 – 12785	158 – 14076	169 292	173 024	Y	
	636	406 – 12463	332 – 13735	70 611	70 577	7	
	628	0 – 9558	1 – 12678	116 482	127.850		
	627	0 – 9600	0 - 12727	72 525	77 941		
go.	638	489 – 6745	2 - 9213	26 737	43 782		
(2) CO₂	637	583 – 6769	9 - 8062	2 953	25 175		
	828	491 - 8161	482 - 8163	7 118	10 522		
	827	626 - 5047	491 - 8194	821	15 878		
	727	535 - 6933	535 - 6933	5187	6 518		
	838	4599 – 4888	2245 – 4751	121	2 916		
	837	c	549 – 4915	c	4 190		
	737	c	575 - 3615	c	1 501		
	666	0 – 6997	0 - 6997	261 886	289 340		
	668	0 - 2768	0 - 2768	44 302	44 302		
(3) O_3	686	1 - 2740	1 - 2740	18 887	18 887		
	667	0 - 2122	0 - 2122	65 106	65 106		
	676	0 - 2101	0 – 2101	31 935	31 935		
	446 ^b	0 – 7797	0 - 7797	33 074	33 074	Yes	
	456	5 – 5086	5 - 5086	4 222	4 222		
(4) N_2O	546	4 – 4704	4 - 4704	4 592	4 592		
	448	542 – 4672	0 - 4672	10 364	116 694		
	447	550 – 4430	550 – 4430	1 705	1 705		
	26 36	3 – 8465	3 - 14478	1 019	1 344	Yes	Yes
	36	3 - 6279	3 – 12231	797	1 042	Yes	Yes
(5) CO	. 28	3 - 6267	3 - 12205	770	920	Yes	Yes
(3) 00	27	3 - 6339	3 - 10295	728	800		Yes
	38	3 – 6124	3 - 8078	712	674		Yes
	37	1807 – 6197	3 – 8168	580	601		Yes
Υ. ΄	211	0 - 11502	0 - 11502	336 829	313 943		
(6) CH ₄	311	0 - 11319	0 - 11319	72 420	77 626		
(0) C114	212	15 - 6511	7 - 6511	54 550	54 550		
	312	959 – 1695	959 – 1695	4 213	4 213		
	66	0 - 15928	0 - 57028	1 787	15 263	Yes	
(7) O_2	68	1 - 15853	1 - 56670	875	2965	Yes	
	67 ^b	0 - 14538	0 - 14537	11 313	11 313	Yes	

(8) NO	46 56 48	0 - 9274 $1609 - 2061$ $1602 - 2039$	0 - 9274 $1609 - 2061$ $1602 - 2039$	103 701 699 679	103 701 699 679		
(9) SO ₂	626 ^b 646 ^b	0 - 4093 0 - 2501	0 - 4093 0 - 2501	72 460 22 661	72 460 22 661		Yes Yes
(10) NO ₂	646	0 - 3075	0 - 3075	104 223	104 223		
(11) NH ₃	446 456	0 - 7000 0 - 5180	0 - 10349 0 - 5180	45 302 1 090	65 828 1 320		Yes Yes
(12) HNO ₃	146 156	0 - 1770 0 - 923	0 - 1770 0 - 923	903 854 58 108	961 273 58 108	Y	
(13) OH	61 81 62	0 - 19268 0 - 329 0 - 332	0 - 19268 0 - 329 0 - 332	30 772 295 912	30 772 295 912		
(14) HF	19 29	24 – 46985 13 – 47365	24 - 32352 13 - 20830	10 073 24 303	8 090 11 920	Yes	Yes Yes
(15) HCl	15 17 25	8 - 34250 8 - 34240 5 - 33284	8 – 20321 8 – 20219 5 – 15266	11 879 11 907 29 994	8 892 8 908 17 762	Yes Yes	Yes Yes
	27 19 11	5 - 33258 13 - 16034 13 - 16032	5 – 15247 13 – 16034 13 – 16032	29 911 3 039 3 031	17 690 3 039 3 031		Yes
(16) HBr	29 21	7 – 8781 7 – 8778	7 – 8781 7 – 8778	1 455 1 455	1 455 1 455		
(17) HI	17 27	10 – 13908 5 – 7625	10 – 13908 5 – 7625	3 161 1 590	3 161 1 590		
(18) ClO	56 76	0 – 1208 0 – 1200	0 - 1208 0 - 1200	5 721 5 780	5 721 5 780		
(19) OCS	622 624 632 623 822	$\begin{array}{c} 0 - 4200 \\ 0 - 4166 \\ 0 - 4056 \\ 0 - 4164 \\ 0 - 4046 \end{array}$	0 - 7822 0 - 7796 0 - 6660 0 - 6631 0 - 4046	15 618 6 087 3 129 2 886 1 641	18 264 6 846 3 275 3 005 1 640		Yes Yes Yes Yes
(20) H₂CO	126 136 128	0 - 3100 0 - 117 0 - 101	0 - 3100 0 - 117 0 - 101	40 670 2 309 1 622	40 670 2 309 1 622		
(21) HOCI	165 ^b 176 ^b	1 - 3800 1 - 3800	$1 - 3800 \\ 1 - 3800$	8 877 7 399	8 877 7 399		
(22) N ₂	44 45 ^b	$ \begin{array}{r} 11 - 9355 \\ 11 - 2578 \end{array} $	11 - 9355 $11 - 2578$	1 107 161	1 107 161		
(23) HCN	124 134 125	0 - 3424 $2 - 3405$ $2 - 3420$	0 - 17586 2 - 3405 2 - 3420	2 955 652 646	58 108 652 646		

(24) CH₃Cl	215	0 - 3198	0 - 3198	107 642	110 462		
(24) CH3CH	217	0 - 3198	0 - 3198	104 854	109 113		
(25) H_2O_2	1661 ^b	0 – 1731	0 – 1731	126 983	126 983		
	1221	604 – 9890	13 - 9890	12 613	22 866		Yes
(26) C₂H₂	1231	613 - 6589	613 - 6589	285	285		Yes
	1222	1 - 789	1 - 789	7 512	7 512		Yes
~ **	1221	706 - 3001	225 - 3001	43 592	54 460		
(27) C_2H_6	1231	725 - 919	285 - 919	6 037	7 107		
(28) PH ₃	1111	0 - 3602	0 - 3602	22 190	22 190		
	260	606 2002	725 2002	1 60 702	1 (0 700	,	
(29) COF₂	269	696 – 2002	725 – 2002	168 793	168 793	*	
	369	686 – 815	686 – 815	15 311	15 311		
(30) SF_6	29	580 – 996	580 – 996	2 889 065	2 889 065		
	121 ^b	2 – 11330	2 - 11330	36 561	36 561		
(31) H_2S	141 ^b	5 - 11227	5 - 11227	11 352	11 352		
	131 ^b	5 - 11072	5 - 11072	6 322	6 322		
(32) HCOOH	126	10 – 1890	10 – 1890	62 684	62 684		
(33) HO ₂	166 ^b	0 – 3676	0 – 3676	38 804	38 804		
(34) O	6 ^b	68 – 159	68 – 159	2	2		
~~~~	5646	763 – 798	763 – 798	21 988	21 988		
(35) <b>ClONO</b> ₂	7646	765 - 791	765 – 791	10 211	10 211		
(36) <b>NO</b> ⁺	46	1634 – 2531	3 – 2531	1 206	1 270		
(30) 110			3 - 2331	1 200	1 270		
(37) <b>HOBr</b>	169	0 - 316	0 - 316	2 177	2 177		
(37) 11001	161	0 - 316	0 – 316	2 181	2 181		
C TT	221	701 – 3243	620 – 3243	18 097	59 536		
(38) $C_2H_4$	231	2947 – 3181	614 - 3181	281	18 095		
(39) <b>CH₃OH</b>	2161 ^b	0-1408	0 - 1408	19 897	19 897		
(39) CH3OH	2101	0-1408	0 – 1408	19 697	19 097		
(40) <b>CH₃Br</b>	219	794 – 1706	794 - 1706	18 692	18 692		
(40) <b>CH3DI</b>	211	796 – 1697	796 – 1697	18 219	18 219		
(41) <b>CH₃CN</b>	2124	890 – 946	890 – 946	3 572	3 572		
(42) <b>CF</b> ₄	29	594 – 1313	582 – 1519	60 033	842 709		
(43) <b>C</b> ₄ <b>H</b> ₂	2211	0 - 758	0 – 1303	124 126	251 245		
(44) <b>HC₃N</b>	1224	0 - 760	0 - 760	180 332	180 332		
7	11	15 - 36024	15 - 27185	4 017	3 480	Yes	
(45) $H_2$	12 ^b	3 – 36406	3 – 36406	5 129	5 129	Yes	
	22	1 - 2586	1 - 2586	1 088	1 088		
	24	1 – 1359	1 – 1359	396	396		
(46) <b>CS</b>	32	1 – 1331	1 – 1331	396	396		
	23	1 – 156	1 – 156	198	198		

(47) <b>SO</b> ₃	26	0 - 2825	0 - 2825	10 881	14 295
(48) $\mathbf{C_2N_2}$	4224	c	200 – 307	c	71 775
(49) <b>COCl₂</b>	265	c	793 – 900	С	164 436
	267	c	800 - 892	c	145 477

^a Abbreviated code for isotopologues

The line-by-line parameters (as well as the portion of HITRAN devoted to cross-sections) have now been cast into a relational database structure, described in Section 6.1. This underlying structure has many advantages that were not available in the previous fixed-length format of previous HITRAN editions. The database is now able to add many more parameters (becoming necessary not only for planetary atmosphere applications, but also to satisfy sophisticated remote-sensing requirements), is not limited to constraints of field length, and can accommodate functional forms for parameters. It also allows for a much more robust way for the managers to check for errors, whether they be by transcription, violated selection rules, incorrect quantum assignments, etc.

# 1.2 New line-shape representations

Shapes of individual transitions in molecular spectra at atmospheric conditions are mainly determined by collisions and the Doppler effect. In the simplest case these two effects can be handled with the Voigt profile, where the collisions are described just as a single-parameter exponential relaxation. In the previous editions of HITRAN, the line shapes, with a very few exceptions, were represented with the Voigt profile. It turns out, however, that this model is too simple to reproduce atmospheric spectra with accuracy at the sub-percent level (see for instance

^b Although spectral ranges and amount of lines is unchanged with respect to HITRAN2012, there are changes to spectral parameters of lines for these isotopologues

^c Not included in HITRAN2012

Refs. [9,10]). For some light molecules, in particular for molecular hydrogen (important for research on dense atmospheres of gas giants), the non-Voigt effects are much more pronounced [6]. At this point the database provides Voigt (for every transition), and speed-dependent Voigt, Galatry [11] and Hartmann-Tran [7,8] profiles parameters where they had been measured and validated.

Ab initio line-shape models, originating from first principles, are far too complex to be simply implemented in a line-by-line spectroscopic database. Therefore, a compromise between simplicity of the model and the fidelity of experimental spectra representation had to be found. In recent decades, a number of phenomenological models that take into account both speeddependent effects and velocity-changing collisions had been advanced. Recently, Tran et al. developed a computationally efficient algorithm [7,8] for calculating one of them, the partiallycorrelated speed-dependent hard-collision profile [12] with quadratic speed dependence [9]. An IUPAC (International Union of Pure and Applied Chemistry) task group recommended the use of this profile for high-resolution spectra representation [13] and call it the Hartmann-Tran (HT) profile. Ngo et al. [14] demonstrated that the HT profile well reconstructs the line shapes of the most important atmospheric molecules. Therefore, the HT profile has been adopted as the default beyond-Voigt profile in HITRAN. For measurements with sufficiently high accuracy, we encourage spectroscopists to report not only the basic Voigt-profile parameters, but also the HT parameters. The details about the HT parametrization adopted in HITRAN are given in Weisło et al. [6]. Apart from new parameters introduced to represent this profile, one should note that these parameters are also provided at four different temperature ranges: 0-100K, 100-200K, 200-400K and T>400K. The reference parameters are given at following reference temperatures: 50K, 150K, 296K and 700K respectively.

It should be noted that even if the parameters of some more advanced model (preferably HT) are now provided in HITRAN for a growing number of transitions, the Voigt-profile parameters are still given for *every* transition in HITRAN. Therefore, for users not interested in making use of sophisticated line shapes, the present version of HITRAN remains compatible with the previous ones.

## $2.1.\ H_2O\ (molecule\ 1)$

The HITRAN2016 edition has undergone a substantial revision and expansion of the database of water vapor.

In HITRAN2012 [1] a very large expansion of the dynamic range of the line intensities for non-deuterated isotopologues of water vapor was implemented, thanks to the ab initio calculations from the BT2 line list [15] and, in selected spectral intervals, Lodi et al. [16] for the principal isotopologue and from Lodi and Tennyson for H₂¹⁸O and H₂¹⁷O [17]. The ab initio results were replaced with high-quality experimental or-semi-empirical data wherever possible. In HITRAN2016, we have done a similar expansion of the dynamic (and spectral) range of the singly-deuterated isotopologues. In addition, the D₂¹⁶O isotopologue makes its debut in the database. D₂O has a very low natural abundance on Earth but is measurable in the atmospheres of other planets, notably Venus which has an enhanced deuterium content [18] raising the importance of D₂O. Also, for fitting D-enriched HDO laboratory spectra, a good D₂O line list is essential. For this reason, D₂O transitions have been included using a lower intensity cutoff of  $10^{-32}$  cm⁻¹/(molecule-cm⁻²) once isotopic abundance is accounted for.

A dedicated paper detailing the update of the water vapor dataset in HITRAN2016 along with atmospheric and laboratory validations is planned. Here we summarize only the most important details.

#### 2.1.1. Line positions and intensities

This update is informed by the work of an International Union of Pure and Applied Chemistry (IUPAC) task group which produced systematic sets of empirical energy levels (and hence transition frequencies) for all the stable isotopologues of water [19–22]. In HITRAN2012 non-deuterated isotopologues already benefited from the availability of the IUPAC dataset of empirically-derived energy levels (and transition wavenumbers derived from them), and we extend this to the deuterated species here. However, some important caveats of that compilation have to be noted.

- a) The accuracy of some of the transitions generated from the IUPAC set may be inferior to individual high-accuracy experiments. Therefore, just like in HITRAN2012, we have given preference to the experimental line position data from previous HITRAN editions if the uncertainty code was 5 (0.00001-0.0001 cm⁻¹) or higher. Comparisons with atmospheric spectra have shown that this was a correct choice.
- b) The datasets constructed in Refs. [19–22] do not include experimental data from the papers published after them. Quite a few new levels have become available over the years and some reassignments were in order (see for instance Mikhailenko et al. [23] and Liu et al. [24]). We partially updated the IUPAC datasets here. In particular, for the HD¹⁷O and HD¹⁸O species, levels from Kyuberis et al. [25] were used. For H₂¹⁸O and H₂¹⁷O, and to a lesser extent for H₂¹⁶O, levels from Mikhailenko et al. [23], [26] and Liu et al. [24] were used. Unfortunately, the IUPAC and Mikhailenko et al. quantum assignments often differ and complete matching would require a substantial amount of time in the future. An update of the IUPAC energy levels is in progress which will endeavor to resolve these assignment issues.

Many of the updated water vapor lines use line positions generated from the IUPAC energy levels and theoretical transition intensities, based on a high-accuracy *ab initio* dipole moment surface [16]. The methodology developed by Lodi and Tennyson [17] involves using several calculated line lists to identify reliable theoretical predictions. Lodi and Tennyson's intensity data were used in their entirety for H₂¹⁷O and H₂¹⁸O in HITRAN2012. Subsequent measurements and analysis by Regalia et al. [27,28] gave good agreement with these intensities. The present update relies heavily on Lodi-Tennyson style calculations for all isotopologues, supplemented with high-quality experimental data where available. This approach has proven to work well in application to atmospheric and laboratory spectra (see for instance Campargue et al. [29] and Ponomarev et al. [30]).

Given the reliance of these theoretical intensities, it is important to understand the systematic errors which *ab initio* calculations may show. In the same issue of this journal, Birk et al. [31] give an extensive intercomparison of *ab initio* calculations with high-quality experimental data. The experimental intensities used in the intercomparison are all included in the HITRAN2016 update. Much of the data show agreement between *ab initio* and experiment within 2%. However, for some bands, notably those involving excitation of the stretching modes, there are larger offsets of up to 8% which can be attributed to systematic errors in the *ab initio* dipole moment calculations. Furthermore, in the  $v_1$  fundamental, differences show a systematic dependence on wavenumber,  $\Delta K_a$  and  $\Delta J$ . Other vibrational bands involving the changes in  $v_1$  quanta also showed larger systematic errors. The theoretical calculations also have problems in predicting local resonances accurately, although these resonances start to appear more frequently in the NIR part of the spectrum. The information on systematic differences between the ab initio

theory and experiment for  $H_2^{16}O$  can be used to estimate uncertainties for minor isotopologues for which no high-accuracy experimental intensities are available.

An important finding of this work is that for an individual vibrational band an intensity-independent offset appears in graphical representations of differences between experiment and theory vs. intensity. This information can be used to validate experiment as well as theory. The constant offset can also be used to predict weak line intensities where experimental data are unavailable by scaling *ab initio* values.

Figure 1 compares theory and experiment for the 1850-2280 cm⁻¹ and 2390-4000 cm⁻¹ regions. The agreements for the fundamental bands (001)-(000), (010)-(000), the overtone (020)-(000), the hot bands (030)-(010), (020)-(010), (011)-(010), (001-010), (100)-(010) are all excellent, mainly within 2%. The bands involving changes in the  $v_1$  quanta, (100)-(000) and (110)-(010), however, show large scatter and an offset around -2%. The scatter includes the systematic differences of +5% to -13%, which becomes obvious when plotting the differences against wavenumber or lower-state energy and color coding  $\Delta K_a$  and/or  $\Delta J$  [31].

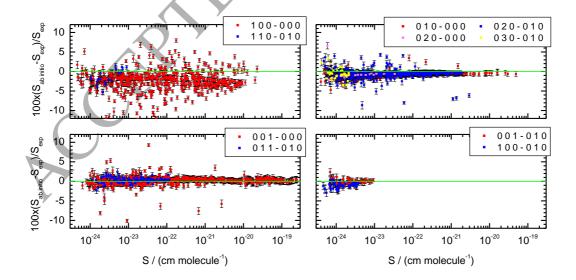
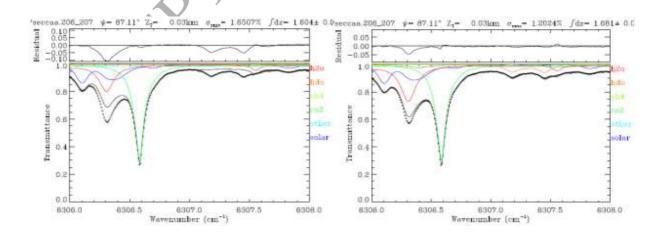


Figure 1. Percentage *ab initio* - experimental vs. intensity for bands in the  $1850-2280 \text{ cm}^{-1}$  and  $2390-4000 \text{ cm}^{-1}$  regions of  $\text{H}_2^{16}\text{O}$ . The error limit of the experimental data is 1% [31].

The 2016 update provides comprehensive line lists for the six main isotopologues of water: H₂¹⁶O, H₂¹⁸O, H₂¹⁷O, HD¹⁶O, HD¹⁸O and HD¹⁷O, as well as the newly-added isotopologue D₂¹⁶O. The use of variational calculations to provide the underlying line lists guarantees the completeness of the lists for the intensity cutoff employed. This completeness leads to a significant expansion of the number of important transitions of the deuterated isotopologues into the NIR (near-infrared) region. These line lists are presented and analyzed in articles published in this issue [25]. Figure 2 shows an example where atmospheric retrieval from a Total Carbon Column Observing Network (TCCON) [32] spectrum benefits from inclusion of HDO lines in the NIR. It is also important to note a significant increase in coverage for HD¹⁷O and HD¹⁸O which were poorly represented in previous HITRAN releases.

The variational (calculated) intensities described above were then replaced with intensities from available high-quality experiments, many of which were already in the HITRAN2012 database. The details will be given in a dedicated paper, but some of the experiments that represent a substantial bulk of new experimental intensity data are described below.



**Figure 2.** Spectral fits to the high- $H_2O$  Darwin (Australia) spectrum using the HITRAN2012 linelist (left panel) and the new linelist (right panel). The improvement in the residuals is due to the addition of HDO lines (three in this plot) denoted by the orange trace. Note the slight change of scale in the residuals on the left and right sides.

## 2.1.1.1 DLR experiments

An extensive work in the spectral ranges  $1850 - 2280 \text{ cm}^{-1}$  and  $2390 - 4000 \text{ cm}^{-1}$  was carried out by Loos et al. [33,34] and in the range 4190 – 4340 cm⁻¹ by Birk et al. [31]. In Figure 3 we will refer to these, and the experiments from the same authors that are already in HITRAN, as DLR. In Loos et al. [33,34] line positions, intensities, self- and air-broadened line-shape parameters, their temperature dependence as well as Rosenkranz line coupling coefficients were retrieved from numerous Fourier-Transform transmittance measurements of self- and airbroadened water vapor at 296 K as well as air-broadened water vapor measurements at low and high temperatures. During the analysis, a large effort was undertaken to give consolidated error bars. In the analysis, a multispectrum fitting approach applying a quadratic speed-dependent hard collision model based on the Hartmann-Tran profile [7,8] and extended to account for line mixing in the Rosenkranz first order perturbation approximation [35] was used. Line positions, intensities and self-broadening, self-speed-dependence and self-shift parameters, as well as in some cases self-line coupling coefficients, were retrieved from pure water-vapor measurements of total pressures up to 20 mbar. Air-broadening, speed-dependence, pressure shift parameters, Dicke narrowing and line mixing coefficients as well as temperature dependence parameters were obtained from air-broadened measurements at ambient temperature and total pressures from 30 to 1000 mbar as well as low and high-temperature measurements at 100 mbar total pressure. The intensities of lines with retrieved line parameters range from  $3\times10^{-26}$  to  $3\times10^{-19}$ cm⁻¹/(molecule cm⁻²). In the 4190 – 4340 cm⁻¹ region, several Fourier-Transform transmittance

spectra of pure and air-broadened water vapor at 296K as well as low and high temperatures were measured and analyzed by Birk *et al.* [31]. These measurements were dedicated to water vapor parameters to be used in TROPOMI/S5-P [36] retrievals. The analysis was also based on a multispectrum fit using the HT profile. Line positions, intensities, self- and air-broadened line-shape parameters including speed-dependence and Dicke narrowing parameters as well as their temperature dependence were retrieved in the analysis.

# 2.1.1.2. MiKaMo16 experimental set

Recently, Mikhailenko et al. [23] have compiled a database of measurements in 5850-8340 cm⁻¹ region from the Grenoble-Tomsk collaboration reported in Refs. [37–41] supplemented with measurements from Sironneau and Hodges [42] and for strong lines from Toth [43] and for very weak lines from ab initio work of Partridge and Schwenke [44,45]. This database is here referred to as MiKaMo16. Unfortunately, due to some differences in assignments with the variational Lodi-Tennyson-like line lists described above, it was hard to take full advantage of this line list. However, tests against TCCON spectra have shown that this line list is superior to any other line list in the region of 7000-8340 cm⁻¹. It was therefore used in HITRAN2016 in that region as is. This, however, creates some consistency issues with assignments of energy levels throughout the database, but substantial improvement of atmospheric retrievals justifies this approach.

As examples of complexity of the update, Figures 3 and 4 show the flow diagrams of the construction of position-intensity line lists of the principal isotopologue (only up to 8340 cm⁻¹ for simplicity) and for  $H_2^{18}O/H_2^{17}O$  respectively. Note that the line list from Lodi et al. [16] for the principal isotopologue is limited to  $10^{-28}$  cm⁻¹/(molecule cm⁻²); therefore intensities for weaker lines still originate from HITRAN2012/BT2.

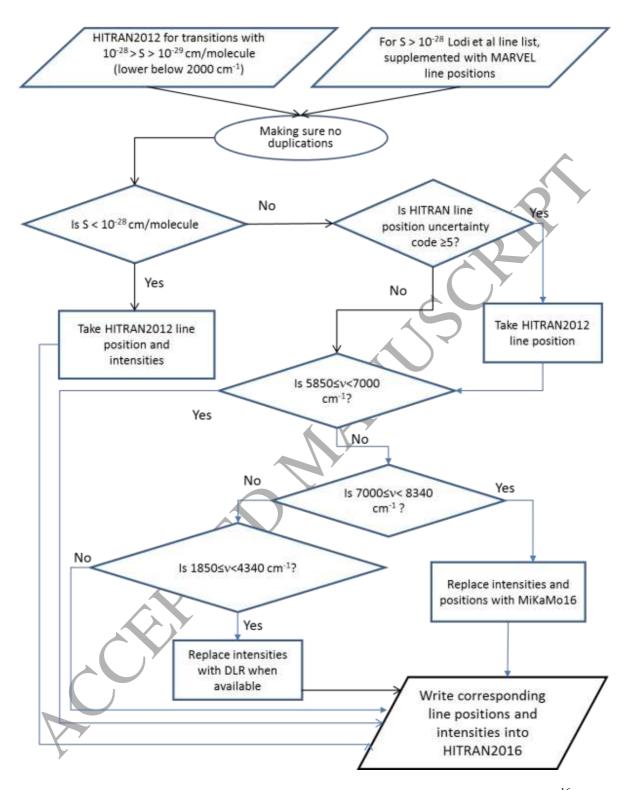


Figure 3. Flow diagram for the construction of line positions and intensities for  $H_2^{16}O$  below 8340 cm⁻¹. See text for details.

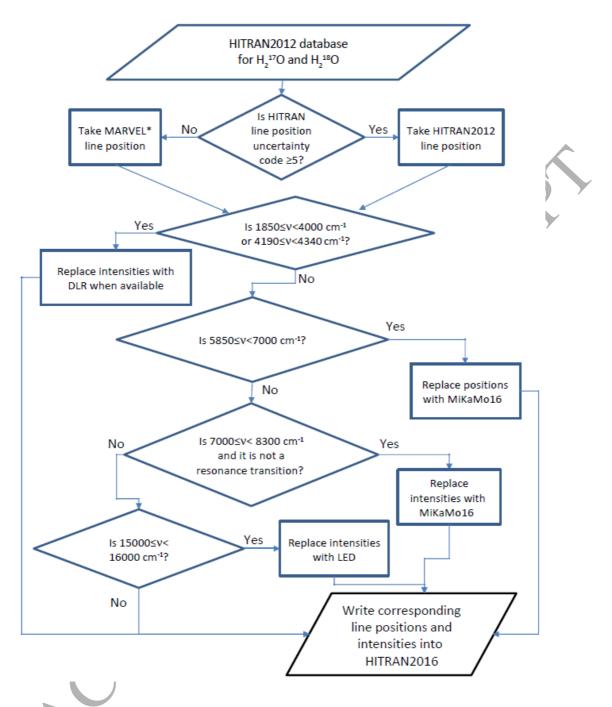


Figure 4. Flow diagram for the construction of line positions and intensities for  $H_2^{18}O$  and  $H_2^{17}O$  below 8340 cm⁻¹. The asterisks indicate that the IUPAC levels were slightly updated in this work. LED refers to the experimental data from Ref. [26]. See text for details.

Whenever a rotational quantum number could not be determined unambiguously, the index of symmetry (1, 2, 3, and 4 as defined in the BT2 work [15]) accompanied with a negative sign was

used. Note that 1 and 2 indicate para states, whereas 3 and 4 indicate ortho states. For the case of unassigned vibrational quanta, a "-2" label has been adopted.

# 2.1.2 Line-shape parameters for the $H_2^{16}O$ , $H_2^{18}O$ and $H_2^{17}O$ line lists

The air-broadened half widths (in Voigt profile formalism) for the first three isotopologues of water in HITRAN2008 and, with some modifications, in HITRAN2012 were derived using the procedure "Diet" described by Gordon et al. [46]. This procedure continues to be used in this edition with experimental outliers identified and removed. One notable example is the removal of a large portion of the half widths measured in Jenouvrier et al. [47] from the experimental data that feeds the "Diet". Indeed, some of the recent evaluations of atmospheric spectra showed that many data in Jenovrier et al. are questionable (see for instance Armante et al. [48]). However, although the removal of this extensive dataset from the updated Gordon et al. [46] procedure, as was done in the GEISA database [49] and the MiKaMo16 linelist, have improved the residuals in some of the lines, some of the residuals became worse than in HITRAN2012. The main problem here is due to the fact that, while there are definite outliers in the values reported in Jenouvrier et al. [47], it is the most extensive study of the air-broadened line widths in the 4200-6600 cm⁻¹ region. In fact, for many transitions no other measurements exist. Removal of these measurements from the experimental dataset has invoked the use of the approximations of different levels (some are very crude) from the work of Jacquemart et al. [50] for many of the transitions. The only solution in this case was to filter the Jenouvrier et al dataset. This was done through direct comparisons with the TCCON spectra for some of the transitions and crosscomparison with similar rotational transitions but from other bands, accounting for vibrational dependence from Eq. (1) of Jacquemart et al. [50]. After this filtering procedure, about 1400 measurements (out of almost 5000) from Jenouvrier et al. [47] were retained.

It is also important to emphasize that measurements by Birk and Wagner [51] in the  $v_2$  band region were given a priority and were written into the database directly as was done in HITRAN2012.

Since the modeling of water vapor absorption lines using a Voigt model is no longer sufficient in many applications and the HT profile provides considerably higher accuracy, experimental HT profile parameters of Birk et al. and Loos et al. [31,33,34] in the spectral ranges  $1850 - 2280 \text{ cm}^{-1}$ ,  $2390 - 4000 \text{ cm}^{-1}$  and  $4190 - 4340 \text{ cm}^{-1}$  were accommodated wherever possible. The experiments and analysis procedures applied are described briefly in section 2.1.1.

# 2.1.3 Line-shape parameters for the deuterated isotopologues

The addition of the line-shape parameters for  $D_2O$  follow a similar procedure as the other isotopologues; however, the available data for air-broadening of  $D_2O$  are considerably less complete than for the other isotopologues. There are some measurements for air-broadening of  $D_2O$  [52–55] as well as Complex Robert-Bonamy calculations [56]. These data amount to several hundred lines from the measured data and just over 550 lines from the theoretical calculations. Using these data, average values as a function of J'' were determined and extrapolation to J''=50 was done using a third-order polynomial fit.

Using these data, the half widths and some line shifts have been added to the  $D_2O$  lines in the database using a priority scheme. First, the measured values were added; if the measurement did not exist for the transition in question, the theoretical value was used. If neither exists, the *J*-averaged value was used. This algorithm allowed half widths to be added to all  $D_2O$  lines in the HITRAN database.

For self-broadening parameters of  $D_2O$ , the work of Gamache *et al.* [57] was used. These data are an extensive set of calculations for  $D_2O$  rotational band transitions extended to 6536

lines using the partner transition rule [58]. Using these data, a set of J-averaged half widths was determined for J=0-50 [57]. These data have been added to the algorithm that adds line-shape data to the water lines in the HITRAN database.

#### 2.1.4 Future plans

Recently, Lampel et al. [59] have identified important absorption features in atmospheric spectra due to water vibration-rotation transitions in the near ultraviolet around 363 nm. Laboratory measurements of water vibration-rotation transition intensities only extend to 25470 cm⁻¹ (393 nm) [60] although multiphoton spectra do provide some information on energy levels in the region of question [61,62]. Future updates will look to extend coverage to water absorption in the near ultraviolet; these data are important for upcoming satellite missions, including TEMPO [63] which will record spectra at these wavelengths.

At the intensity limit assumed for  $D_2^{16}O$ , the even rarer isotopologues  $D_2^{17}O$  and  $D_2^{18}O$  should also be visible. Line lists for these two species have also been prepared as part of the work on deuterated water in Ref. [25]; these data will be added in the forthcoming updates to HITRAN2016.

We will be populating the database of the HT line-shape parameters after evaluating existing literature values and new measurements as they become available.

# 2.2. CO₂ (molecule 2)

Accurate and comprehensive line lists for all naturally abundant isotopologues of carbon dioxide are required by remote-sensing missions dedicated to monitor the concentration of carbon dioxide in Earth's atmosphere. The recently launched OCO-2 mission [64–66], together

with several other space and ground based projects (GOSAT [67], AIRS [68], ASCENDS [69], TCCON [32], NDACC [70]) are dedicated to explicitly monitor the atmospheric CO₂ content. These experiments aim not only to look at overall CO₂ concentration and its variation, but also wish to pinpoint where CO₂ is being produced (sources) and where it is absorbed (sinks). This activity is clearly vital to monitoring and essential for eventually controlling the CO₂ content of the atmosphere[71]. A successful retrieval of CO₂ concentration requires validated line lists with transition intensities given at sub-percent accuracy, line positions accurate to 0.0001 cm⁻¹ or better, and beyond-Voigt-profile line-shape models [65,72,73].

Determination of isotopic ratios of carbon in Earth's samples and astrophysical objects remains crucial for modeling geophysical processes. For example, quantification of ¹⁴C in fossil fuels can provide information about the sources of human-related contribution to the total CO₂ concentration in the terrestrial atmosphere. This can be done with recently developed cavity-enhanced laser spectroscopic techniques in the IR [74–76]. However, these measurements require a priori simultaneous knowledge of reliable line intensities of many isotopologues. Precise determination of ¹³C/¹²C and ¹⁶O/¹⁷O/¹⁸O ratios is also vital, for instance, in understanding processes of formation of radiation fields in the Martian atmosphere, which is 96% rich in carbon dioxide [77].

A summary of the carbon dioxide line list in the HITRAN2012 database and comparison to HITRAN2016 is given in Table 4. The HITRAN2012 database was considerably improved with respect to its previous 2008 edition. However, several issues related to spectral completeness, inconsistency of multiple data sources, and insufficient accuracy of line intensities, still remained unsolved. The majority of entries in the 2012 version of the HITRAN database were taken from the effective Hamiltonian calculations included in the 2008 edition of the CDSD-296 database

[78].

**Table 4**. Comparison of HITRAN2016 and HITRAN2012 line lists for isotopologues of carbon dioxide.

		HITRAN2012	HITRAN2016			
ISO abundance	Number of lines	Spectral region (cm ⁻¹ )	Q(296K)	Number of lines	Spectral region (cm ⁻¹ )	Q(296K)
626 0.984204	169292	345.936-12784.056	286.94	173024	158.301-14075.298	286.094
636 1.1057×10 ⁻²	70611	406.834-12462.046	578.41	70577	332.649-13734.963	576.644
628 3.9470×10 ⁻³	116482	0.736-9557.398	609.48	127850	1.472-12677.181	607.713
627 7.3399×10 ⁻⁴	72525	0.757-9599.317	3552.70	77941	0.757-12726.562	3542.610
638 4.4345×10 ⁻⁵	26737	489.678-6744.158	1229.10	43782	2.945-9212.609	1225.270
637 8.2462×10 ⁻⁶	2953	583.593-6768.643	7162.90	25175	9.086-8061,741	7140.024
828 3.9556×10 ⁻⁶	7118	491.688-8160.439	324.21	10522	482.813-8162.743	323.424
728 1.4718×10 ⁻⁶	821	626.438-5046.875	3776.40	15878	491.181-8193.172	3766.044
727 1.3685×10 ⁻⁷	5187	535.384-6932.980	11002.00	6518	535.383-6932.693	10971.91
838 4.4440×10 ⁻⁸	121	4599.239-4887.290	653.50	2916	2245.898-4750.068	652.242
837 1.653×10 ⁻⁸	N/A	N/A	7615.20	4190	549.472-4914.496	7593.900
737 1.5375×10 ⁻⁹	N/A	N/A	N/A	1501	575.852-3614.084	22129.96
646 radioactive	N/A	N/A	N/A	41610	426.445-7928.788	2033.353

Note: ISO is the AFGL shorthand notation for the isotopologue, abundance is the terrestrial value assumed by HITRAN, and Q(296) is the partition sum at 296K.

For less abundant isotopologues, obtaining high-quality experimental data is not trivial. Therefore fits of the effective Hamiltonian or the effective dipole moment [78], were based on only four major isotopologues ¹²C¹⁶O₂, ¹³C¹⁶O₂, ¹⁶O¹²C¹⁸O and ¹⁶O¹²C¹⁷O, for which measured spectroscopic parameters were available. As a result, several spectral gaps were present in HITRAN2012 (see for instance discussion in Refs. [79,80]) which represent regions where experimental data were unavailable. For similar reasons, no entries were included in the database for the ¹⁸O¹³C¹⁷O, ¹⁷O¹³C¹⁷O, and ¹⁶O¹⁴C¹⁶O isotopologues (837, 737, and 646 in old AFGL notation). An overview of this problem is displayed in Figs. 5 and 6, where CO₂ rovibrational spectra from HITRAN2012 and HITRAN2016 are compared for different isotopologues.

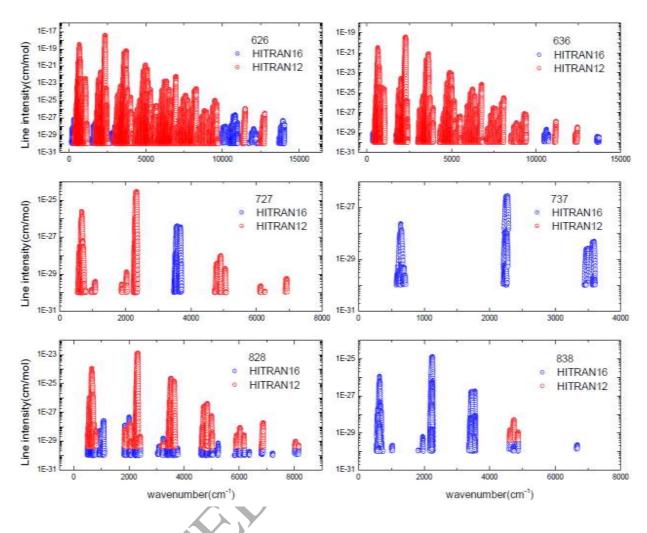


Figure 5. Overview of the line lists of stable symmetric isotopologues of carbon dioxide in HITRAN2012 and HITRAN2016

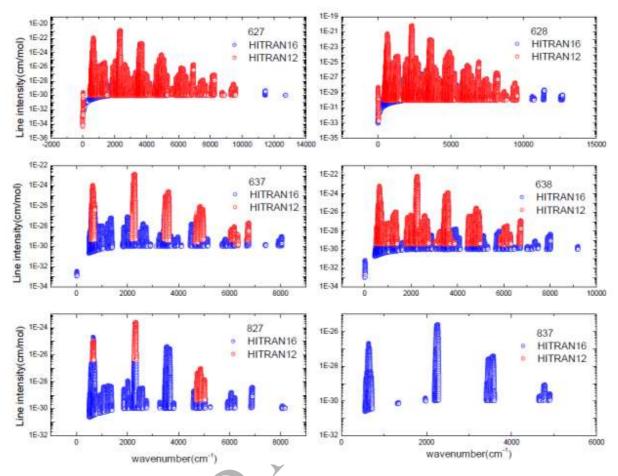


Figure 6. Overview of the line lists of stable asymmetric isotopologues of carbon dioxide in HITRAN2012 and HITRAN2016

Wherever possible, the effective Hamiltonian fits were extrapolated to the trace isotopologues, using a method of isotopic substitution [81]. In the 2012 edition, multiple data sources caused sporadic discontinuities in intensity patterns of ro-vibrational lines [82–84]. Furthermore, a high percentage of line intensities in HITRAN2012 have stated uncertainty of 20% or worse (HITRAN uncertainty index equal to 3). Although this assessment has been proven to be overly pessimistic in many cases [82,83,85–87], the uncertainty budget, especially for the Effective Hamiltonian calculations, was still too high for precise measurements of atmospheric CO₂ concentration. The most accurate entries in HITRAN2012 were taken from NASA JPL measurements by Toth et al. [88–90] and covered the 1.6-μm and 2.06-μm spectral

regions, which are used in remote-sensing measurements. The stated 1-5% accuracy of these experimental line intensities (HITRAN uncertainty index equal to 7 and 6), has been confirmed by a number of comparisons [82,83,91]; nonetheless the rigorous requirements for part-permillion resolution in measurements of CO₂ atmospheric concentration were not achieved.

Since the 2012 release of the HITRAN database, a large number of experimental and theoretical studies have been devoted to improve the knowledge of line positions, line intensities and line shapes of CO₂ isotopologues. For a comprehensive review of measurements and theoretical models see Ref.[92] and references therein.

Theoretical line lists (denoted as "Ames") for 12 stable and one radioactive (646) isotopologue of CO₂ were published by Huang et al. in the 0-20000 cm⁻¹ spectral region and for temperatures below 1500K [79,93,94]. Room-temperature line lists (denoted as "UCL-IAO") for 13 CO₂ isotopologues were also calculated by Zak et al. [82,83,85,91] in the 0-8000 cm⁻¹ spectral region. Both of these latter studies contained intensities computed with *ab initio* dipole moment surfaces and semi-empirical line positions, based on a fitted potential energy surface for the Ames effort and on the effective Hamiltonian calculations for UCL-IAO. A major advantage of the variational approach used in the Ames and UCL-IAO line lists is that it should give similar accuracy for all isotopologues. This allows coverage of spectral regions currently not probed by experiments for rare isotopologues. UCL-IAO also provides uncertainty estimates of line intensities, based on a purely theoretical methodology [17]. Such a reliable analysis allows for the detection of ro-vibrational resonance interactions, which significantly lower the accuracy of theoretical line positions and intensities. Using this method, the lines identified as unreliable have been replaced with the data from CDSD-296 [92] and, in several cases of interpolyad resonance interactions (asymmetric isotopologues), with the experimental data from Lyulin et al.

[95], Karlovets et al. [96,97] and Campargue et al. [98].

Recently, there have been a number of high-precision near-IR spectroscopic measurements which provide rigorous tests of theoretical line intensities based on effective dipole moment surface and ab initio calculations [84–87,99]. Particularly, in the 1.6-µm and 2.0-µm spectral regions, the UCL-IAO line lists have been experimentally verified as accurate to the sub-percent level. Figure 7 compares the UCL-IAO and Ames line lists to HITRAN2012 (Toth et al. blue open circles [100]) for the 20012 -00001 band and to state-of-the-art experiments including more recent Fourier-transform measurements [101] and advanced high-resolution laser measurements [101-104]. A number of comparisons here suggest that the UCL-IAO study models line intensities more accurately than the Ames study. Note that more recent results from Ames, which are available from their website (www.huang.seti.org), give closer agreement with UCL-IAO. From Figure 7 it is clear that there is a 1-3% average difference in line intensities between the new and the previous version of HITRAN for this band. The independent experiments from the National Institute of Standards and Technology [104], and the University of Naples II [103] confirm, however, a conservative 0.5% accuracy of line intensities for this band in HITRAN2016. This level of accuracy could potentially satisfy even the most stringent requirements of current remote-sensing missions. Interestingly, although line intensities for this band and the 20013 -00001 and 30013 - 00001 bands probed by OCO-2 originate from the same source (Toth et al. [100]), the agreement between UCL-IAO and HITRAN2012 is substantially better for the OCO-2 bands.

For wavenumbers greater than 8000 cm⁻¹, the majority of the line parameters were taken from CDSD-296 [92]. As we have already mentioned above, both HITRAN2012 and CDSD-296 have several spectral gaps, in particular in the wavenumber region greater than 8000 cm⁻¹.

Recently, several experimental studies of the carbon dioxide spectra in the high-frequency region have been performed [105–108]. The measured line intensities allowed determining the absent effective dipole moment parameters for several series of transitions. Using these effective dipole moment parameters and an effective Hamiltonian from Ref. [109], the line positions and intensities for the principal isotopologue were generated and included into HITRAN2016 covering the 9800-10500 cm⁻¹ and 11600-12400 cm⁻¹ wavenumber gaps.

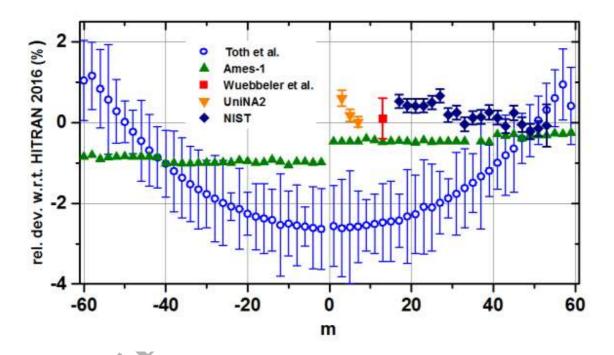


Figure 7. Comparison of line intensities between HITRAN2016, HITRAN2012 (Toth et al. [88]), and other accurate experimental and theoretical sources for the 20012 - 00001 band (2- $\mu$ m band) of  $^{12}C^{16}O_2$ : measurements Wübbeler et al. [102], NIST [104] and UniNA2 [103], and theory Ames-1 [79]. The zero relative deviation line corresponds to HITRAN2016 line intensities (in this case from Zak et al. [82]). The running index m equals -J, J, and J+1 for the P, Q, and R branches, respectively.

Line positions were updated with respect to the previous version of the database. The majority of lines come from the effective Hamiltonian calculations included in UCL-IAO line lists, which are based on the fits to the observed line positions collected from the literature and published in the latest, 2015 release of the CDSD-296 database [92]. These fits were completed and updated

with recent, accurate measurements performed on isotopically-enriched samples of CO₂. Uncertainties in the fitted line positions depend on the quality of the experimental data and vary from 0.001 cm⁻¹ to 10⁻⁹ cm⁻¹. For asymmetric isotopologues, a number of bands are affected by strong interpolyad anharmonic resonance interactions. The effective Hamiltonian model does not include this type of interaction for the asymmetric isotopologues. Hence in such cases, line positions were taken directly from measurements [80,95–98,110,111].

The uncertainty codes for the line positions were transferred from CDSD-296 to HITRAN2016. The uncertainty code 3 (0.001-0.01 cm⁻¹) was given for the line positions in the 9800-10500 cm⁻¹ and 11600-12400 cm⁻¹ wavenumber regions. Partition functions in the current release of the database are based on the direct summations taken from the variational calculations of Huang et al. [79]. On average, the new partition functions agree excellently with those of CDSD-296 [83]; however they do not agree perfectly with those in HITRAN2012 (from TIPS [112]) and three previous editions of the database and differ at 296K by about -0.3%. Although this difference is marginal, it could have an effect in the applications where sub-percent accuracy is required.

As stated above, the radioactive isotopologue ¹⁴CO₂, 646, has been added to the database. This is the first edition of HITRAN where radioactive species have been incorporated (also for CO, see Section 2.5). All lines of the 646 isotopologue were taken from the UCL line lists given in Ref. [83]. Due to issues with what constitutes a so-called natural terrestrial abundance of radioactive species (which is part of the traditional definition of intensities in HITRAN, see Eq. (1) of the Definitions and Units documentation in HITRAN*online*), line intensities for these type of species are given for unit abundance; a 10⁻²⁷ cm⁻¹/(molecule·cm⁻²) cut-off value for the intensity has been applied. This cut-off produced 41610 lines in the *J* range 0 to 114. Vibrational

assignments for the 646 isotopologue were based on isotopic shifts of energy levels and respective assignments for the 626 and 636 isotopologues, and hence should be regarded as provisional. An abundance-scaled intensity cut-off of  $10^{-30}$  cm⁻¹/(molecule·cm⁻²) is used for all stable isotopologues. Note that, for the time being, data for the radioactive isotopologues are provided as static files rather than through the HITRAN*online* interface.

Uncertainties of line intensities were informed by theoretical error analysis, which classified lines as reliable, intermediate, or unreliable. Bands with reliable lines stronger than 10⁻²³ cm⁻¹/(molecule·cm⁻²) (for unit abundance) were assigned HITRAN uncertainty code 8 (i.e. accuracy of 1% or better). Line intensities of reliable parallel bands weaker than 10⁻²³ cm⁻¹/(molecule·cm⁻²) were given an uncertainty code 7 (i.e. accuracy 1 - 2 %). Reliable perpendicular bands weaker than 10⁻²³ cm⁻¹/(molecule·cm⁻²) and intermediate lines were marked with HITRAN uncertainty code 6 (i.e. accuracy 2 - 5%). So-called unreliable lines were taken from the effective dipole moment calculations [92] and experiments. Typical intensity uncertainties for these entries range between 5 and 20%.

It is important to point out that an intensive study of the 1.6-µm and 2.06-µm bands that includes non-Voigt lineshapes and line mixing has been published by the OCO-2 spectroscopy support group ABSCO (ABSorption COefficient tables for the OCO-2 mission) [84,86]. The data were fit using a multi spectrum fit procedure which, among other things, enables retrieval of the lineshape parameters using the speed-dependent Voigt (SDV) profile as well as line mixing. These are very good experiments and it is debatable whether to use them for the strong and weak bands in place of UCL-IAO parameters described above. Indeed the ABSCO team have validated (using TCCON spectra) the cross-sections generated using results of Refs. [84,86] and found them to be the most efficient [113]. However, achieving high-precision results in nuanced

correlations, with line mixing and model assumptions that can create discontinuities in inter-band comparisons, is difficult. At the moment, HITRAN cannot provide users with tools that can be used to generate cross-sections from the works of Devi et al. [84] and Benner et al. [86]. The usable products of the ABSCO effort are absorption coefficients (available upon request from the ABSCO group) rather than spectral parameters, which are available in the publications. Moreover, these absorption coefficients are empirically scaled by the factors of 0.6% and 1.4% for the 1.6-μm and 2.06-μm bands respectively, due to lingering data and/or model biases (the use of partition function HITRAN 2012/TIPS is up to 0.3% of this factor). After these studies an update of the multi spectrum fitting code with CDSD partition functions was done. Additional methods to adjust the intensity distribution closer to the UCL list by scaling experimental conditions within the uncertainties are under evaluation. The intensities of the band at 2.06 μm are already within 0.7% of Zak et al. [82], indicating that the additional 0.7% scaling of ABSCO cross-sections may be unrelated to intensities. These issues will be considered for future editions of the database.

For wavenumbers greater than 8000 cm⁻¹, two sources of the line intensities are used: CDSD-296 [92] and the newly-generated line list in the 9800-10500 cm⁻¹ and 11600-12400 cm⁻¹ regions based on the new measurements [105–108]. The uncertainty codes of the CDSD-296 line intensities were transferred to HITRAN2016. Based on the uncertainties of the line intensity measurements in the 10700-10860 cm⁻¹ region [106], we use uncertainty code 5 (5%-10%) for the line intensities of the 3003i-00001 (i=1,2,3,4) series of bands and based on the uncertainties of the line intensity measurements in the 10000-10300 cm⁻¹ and 11600-12400 cm⁻¹ [107] wavenumber regions we use uncertainty code 3 (>20%) for the line intensities of the 4003i-00001 (i=1,2,3,4,5) and of the 6001i-00001 (i=1,2,3,4,5,6,7) series of bands.

The Voigt line-shape parameters throughout the entire database were calculated using the predictive routine of Gamache and Lamouroux explained in Refs. [114–116]. For the line mixing, we now provide a code from Lamouroux et al. [117] which has been updated to operate with HITRAN2016. We note that Lamouroux et al [117] line mixing coupled with the HITRAN2012 data has worked really well and in fact produced residuals hardly exceeding 1% when applied to the TCCON data in Ref. [113], although slightly inferior to the ABSCO cross-sections in the 2.06-µm region.

#### 2.3. $O_3$ (molecule 3)

The spectroscopic database for ozone is important for atmospheric and environmental applications due to its critical roles in UV shielding, radiative transfer and health. It is also considered a possible bio-marker for future observations of exoplanets [118]. In order to understand the impacts of ozone on the atmosphere and to monitor ozone a precise knowledge of its line parameters at all wavelengths is required.

The ozone parameters for this edition of HITRAN have been updated in two steps. First, the parameters of spectroscopic models, line positions, intensities, and lower-state energy levels resulting from analyses of experimental spectra have been introduced into the S&MPO (Spectroscopy and Molecular Properties of Ozone) information system [119]. The S&MPO offers various tools for spectral simulations and for comparisons with experimental records. At the initial stage of the work, line lists were generated using empirically-fitted model parameters of the effective Hamiltonians (EH) which considered nearby strongly interacting levels and of the corresponding effective dipole transition moments (ETM). This procedure was necessary to obtain complete sets of transitions as only a limited number of unblended lines could be

accurately measured in congested ozone spectra. State-of-the-art studies of ozone laboratory spectra in the infrared have been recently reviewed by Barbe et al. [120] and Mikhailenko [121]. Note that most of the compilations of ozone data in the infrared accessible via the VAMDC (Virtual Atomic and Molecular Data Center) portal [122,123] are also based on the S&MPO line lists. This is the case of recent improvements in HITRAN and also in GEISA [49] for which the last release essentially contained the earlier 2015 update from the S&MPO.

In the second step, the line lists were validated by simulating atmospheric spectra as discussed in Section 2.3.3.

The changes in ozone line parameters implemented in HITRAN2016 are summarized in Table 5. A total of 75694 lines for  $^{16}O_3$  have been added or updated. Note that there are 289340 lines of the principal isotopologue of ozone in total in the HITRAN2016 edition. The present release of the ozone data contains 29006 new lines and 46688 updated lines of  $^{16}O_3$  (a graphical overview can be seen in Fig. 8).

Table 5. HITRAN2016 ozone update summary: principal isotopologue ¹⁶O₃

Number of Transitions	Region [cm ⁻¹ ]	Integrated intensity, S _V [cm ⁻¹ /(molecule·cm ⁻² )]
1781	281.14 - 377.51	$1.255 \times 10^{-21}$
619	295.42 - 446.63	$2.646 \times 10^{-23}$
2558	598.26 - 804.86	$1.942 \times 10^{-21}$
6	656.12 - 725.68	$4.937 \times 10^{-25}$
1779	899.45 – 1176.42	3.233×10 ⁻²¹
3604	911.32 – 1215.49	$6.100 \times 10^{-20}$
3457	921.37 – 1211.67	$5.148 \times 10^{-21}$
4379	926.56 – 1193.79	$1.651 \times 10^{-19}$
3966	940.22 - 1229.80	$3.141 \times 10^{-21}$
20	956.13 – 1170.40	$1.916 \times 10^{-23}$
2964	956.68 – 1235.44	$6.852 \times 10^{-21}$
	Transitions 1781 619 2558 6 1779 3604 3457 4379 3966 20	Transitions       [cm-1]         1781       281.14 - 377.51         619       295.42 - 446.63         2558       598.26 - 804.86         6       656.12 - 725.68         1779       899.45 - 1176.42         3604       911.32 - 1215.49         3457       921.37 - 1211.67         4379       926.56 - 1193.79         3966       940.22 - 1229.80         20       956.13 - 1170.40

030 - 001	15	957.33 – 1201.66	$2.254 \times 10^{-24}$
002 - 010	467	1302.52 - 1435.34	$3.750 \times 10^{-23}$
101 - 010	1637	1347.52 - 1438.73	$9.393 \times 10^{-22}$
030 - 010	369	1366.12 – 1475.95	$2.268 \times 10^{-23}$
200 - 010	102	1366.62 - 1541.76	$5.305 \times 10^{-24}$
002 - 000	6486	1856.83 - 2280.54	$1.019 \times 10^{-19}$
200 - 000	6122	1909.98 - 2340.81	$3.020 \times 10^{-20}$
101 - 000	7067	1921.44 – 2328.94	1.194×10 ⁻¹⁸
030 - 000	105	2022.87 – 2272.37	4.092×10 ⁻²²
021 - 000	1482	2335.99 – 2440.14	4.360×10 ⁻²²
120 - 000	2095	2395.22 – 2616.83	3.063×10 ⁻²²
013 - 010	2459	2889.23 - 3090.28	$4.735 \times 10^{-21}$
112 - 010	2046	2910.00 - 3121.94	$3.816 \times 10^{-22}$
023 - 010	1355	3564.72 – 3657.05	$3.676 \times 10^{-22}$
013 - 000	2673	3584.62 – 3791.77	5.713×10 ⁻²¹
122 - 010	620	3584.75 – 3751.75	$3.530 \times 10^{-23}$
112 - 000	3056	3600.25 – 3831.27	$8.306 \times 10^{-22}$
400 - 010	4	3623.87 – 3653.87	$6.743 \times 10^{-25}$
311 – 100	729	3739.97 – 3826.22	$2.398 \times 10^{-23}$
005 - 100	508	3742.91 – 3726.13	$1.660 \times 10^{-23}$
104 - 100	51	3752.69 - 3863.55	$8.192 \times 10^{-25}$
005 - 001	278	3807.31 - 3917.54	$9.849 \times 10^{-24}$
311 – 001	436	3810.30 - 3946.73	$9.846 \times 10^{-24}$
104 - 001	950	3820.17 – 3894.94	$2.218 \times 10^{-22}$
005 – 010	848	4128.33 – 4228.24	$2.371 \times 10^{-23}$
311 - 010	1018	4129.14 – 4228.62	$3.444 \times 10^{-23}$
104 - 010	66	4146.09 – 4228.28	$6.310 \times 10^{-25}$
023 - 000	1192	4268.37 – 4357.21	$1.928 \times 10^{-22}$
122 - 000	724	4284.15 – 4454.93	$4.359 \times 10^{-23}$
400 - 000	8	4325.56 – 4398.58	$5.896 \times 10^{-25}$
213 - 100	63	4384.72 – 4423.50	$1.417 \times 10^{-24}$
321 - 100	126	4429.78 – 4466.82	$5.750 \times 10^{-24}$

114 - 001	306	4452.15 - 4508.72	$1.313 \times 10^{-23}$
213 – 010	503	4757.08 - 4825.80	$2.346 \times 10^{-23}$
104 - 000	1093	4802.98 – 4978.61	$7.789 \times 10^{-23}$
005 - 000	1514	4806.33 – 4938.21	$5.300 \times 10^{-22}$
311 - 000	1053	4827.65 – 4928.49	$3.450\times10^{-22}$
Total	74759	281.14 – 4928.49	1.5895×10 ⁻¹⁸

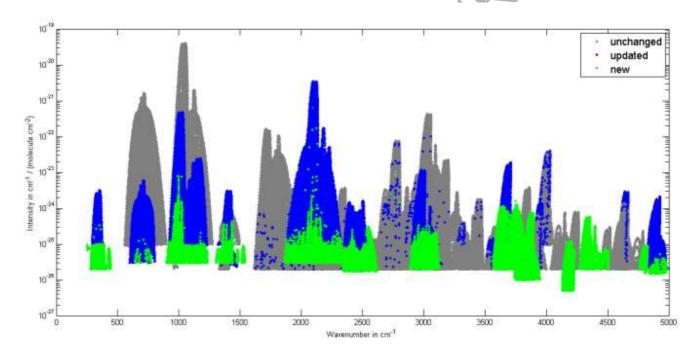


Figure 8. Log intensity diagram for the  $^{16}O_3$  lines in the 200-5000 cm $^{-1}$  HITRAN2016: every point corresponds to a single transition. Newly included lines are colored green, updated lines blue and unchanged HITRAN2012 lines grey (intensities are given in cm $^{-1}$ /molecule·cm $^{-2}$  at 296 K).

# 2.3.1 Cold bands

# 2.3.1.1, 5-µm region: Tetrad {(002),(101),(200),(030)} -(000)

The 5-µm range is of particular importance for the retrieval of atmospheric ozone from ground-based and satellite observations, as this is the second strongest absorbing range in the ozone spectrum after the 10-µm bands. For this reason, the consistency of ozone intensities

between the 5-um and 10-um regions as well as with UV ozone absorption is a major issue that has been a subject of many discussions ([124,125] and references therein). The line parameters in previous HITRAN versions for the strongest bands, i.e. the first triad  $v_1+v_3$ ,  $2v_1$ ,  $2v_3$ , in the 5um interval were from the work by Flaud et al [126]. Subsequently, the FTS spectra were revisited by Barbe et al. [127] though the corresponding data had not been included in HITRAN. Recently new measurements and analyses have been performed by the GSMA (Groupe de Spectrométrie Moléculaire et Atmosphérique) team from Reims University [120,128]. In 2015, a major update of line positions and intensities based on the list generated together with the Laboratory of Theoretical Spectroscopy of the Institute of Atmospheric Optics (Tomsk) was introduced in the S&MPO [119]. For the 2v₁ and 2v₃ bands, the difference in the integrated intensities, S_V, exceeded 6% between S&MPO and HITRAN2012. The analysis [128] covering the  $1856-2273~\text{cm}^{-1}$  region also included the  $3v_2$  band, which is coupled with  $v_1+v_3$  via an accidental Coriolis interaction. Another resonance accounted for in this list explains the anomalously strong  $\Delta K_a = 5$  lines that have been detected in atmospheric observations by Goldman et al [129]. Janssen et al [125] reported a comparative study of the sensitivity of FTIR ground-based measurements of the atmospheric ozone concentration with respect to line parameters included in the HITRAN2012, GEISA2015 and S&MPO databases in 10- and 5-µm regions. They have concluded that "only S&MPO gives an entirely consistent result at the  $\pm$  1% level". Consequently, we include the S&MPO line positions and intensities in the present HITRAN release. This should improve both intensity distribution in the ro-vibrational lines of the 5-µm bands and their consistency with the 10-µm range. The line-shape parameters for all the lines were estimated using the semi-empirical procedure described in the HITRAN2008

paper. This model largely relies on a slightly revised empirical model for the fundamental bands from Ref. [130].

Another spectral interval where S&MPO and HITRAN2012 ozone data differ significantly are for the  $2v_2+v_3$ ,  $v_1+2v_2$  bands in the 2335 - 2617 cm⁻¹ range. Previous HITRAN versions included the MIPAS list [131] for  $2v_2+v_3$  and older data from Flaud et al. [126] for  $v_1+2v_2$ . S&MPO data in this range were initially based on the analyses of Barbe et al [132]. Subsequently, new Reims FTS measurements with a better signal-to-noise ratio [120] suggested an empirical intensity scaling factor of 0.86 for these bands. The resulting line list is provided for this release of HITRAN. Recently, accurate *ab initio* dipole moment surfaces of ozone were constructed by Tyuterev et al [133] that allowed theoretical intensities to be obtained from variational nuclear-motion calculations which used the potential energy surface of Ref [134]. The shape of the  $v_1+2v_2$  band given by these calculations is qualitatively more consistent with the S&MPO data incorporated in the new list of HITRAN2016 than HITRAN2012 (see Figure 9).

$$2.3.1.3.\ 3584 - 3831\ cm^{-1}$$
 range:  $\{(013), (112)\} - (000)\ bands$ 

The S&MPO contains data for the  $v_2+3v_3$  and  $v_1+v_2+2v_3$  bands in the range 3584 – 3831 cm⁻¹ based on a larger sample of assigned transitions [135] than the previous releases of HITRAN (see Table 4 of Babikov et al. [119]). The parameters can be found in the corresponding sections of the online S&MPO site [119], and 5729 transitions are now included that makes the integrated intensity of the  $v_1+v_2+2v_3$  band nearly 5 times stronger (Table 5) than in the previous release.

$$2.3.1.4.4268 - 4399 \text{ cm}^{-1} \text{ range: } \{(023), (122), (400)\} - (000) \text{ bands}$$

In the range 4200– 4525 cm⁻¹, only the  $3v_1+v_3$  [136] and  $2v_1+2v_2+v_3$  [137] cold bands were included in the previous releases of HITRAN, whereas the  $2v_2+3v_3$ ,  $v_1+2v_2+2v_3$  and  $4v_1$  bands in the 4268-4399 cm⁻¹ range were missing, although they correspond to a total integrated intensity of  $2.4\times10^{-22}$  cm⁻¹/molecule cm⁻². The corresponding analyses [121][138] are as yet unpublished, but the line parameters can be found at the online S&MPO site [119] and are now incorporated in HITRAN2016.

 $2.3.1.5.4802 - 4929 \text{ cm}^{-1} \text{ range: } \{(005), (311), (104)\} - (000) \text{ bands}$ 

Line parameters for the  $5v_3$ ,  $3v_1+v_2+v_3$  and  $v_1+4v_3$  bands around  $4900 \text{ cm}^{-1}$  are calculated for the HITRAN database using the spectroscopic parameters of Flaud et al. [139]. These bands were then re-analyzed using new spectra recorded in Reims [120] resulting in a new set of the Hamiltonian parameters for the strongly coupled  $\{(005),(104),(311)\}$  vibration states. The corresponding line lists have been generated using these improved spectroscopic parameters. Unpublished Hamiltonian parameters can be found in the S&MPO system at <a href="http://smpo.iao.ru/1408x675/en/lev/par/1/19/">http://smpo.iao.ru/1408x675/en/lev/par/1/19/</a>. The (S&MPO-HITRAN2012) differences in the line positions are up to  $0.065 \text{ cm}^{-1}$ . The intensity ratios for three cold bands are shown in Fig. 10.

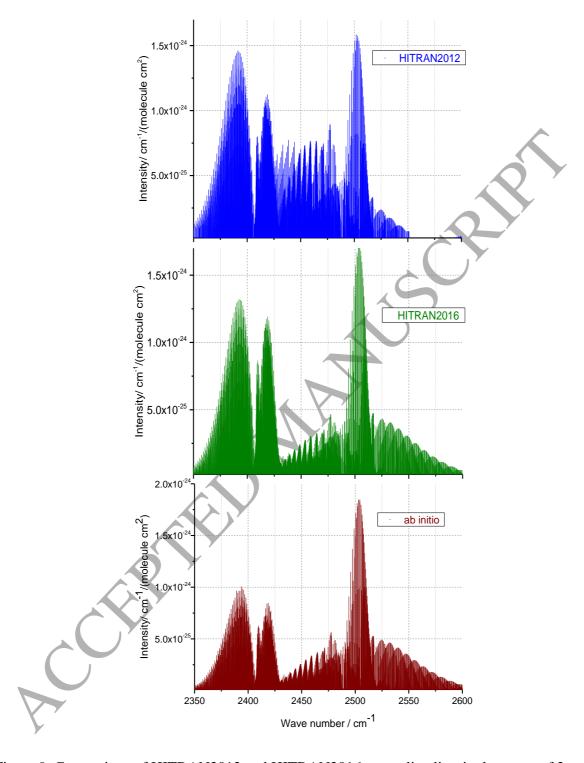


Figure 9. Comparison of HITRAN2012 and HITRAN2016 ozone line lists in the range of  $2\nu_2+\nu_3$  and  $\nu_1+2\nu_2$  bands with recent ab initio calculations using the DMS of Tyuterev et al. [133] that provides qualitative confirmation of the rotational distribution of intensities in the  $\nu_1+2\nu_2$  band given by HITRAN2016.

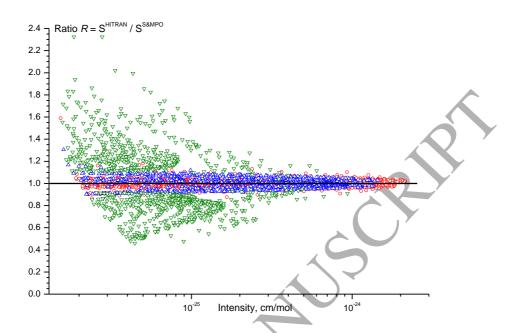


Figure 10. Intensity ratios for the  $5v_3$  (red circles),  $3v_1+v_2+v_3$  (blue triangles) and  $v_1+4v_3$  (olive triangles) bands between the previous HITRAN2012 line list and S&MPO data that are incorporated in the new HITRAN2016 release (color can be seen in online version of paper).

#### 2.3.2 Hot bands

The ozone spectrum in the range 281-447 cm⁻¹ is dominated by hot  $v_3$ - $v_2$  and  $v_1$ - $v_2$  bands whose intensities were underestimated in the previous HITRAN releases compared to the S&MPO line list [120] which was generated using the empirical transition moments parameters of Birk et al[140]. The S&MPO parameters are confirmed by the recent *ab initio* intensity calculations [133] and the corresponding update is thus included in the present HITRAN release.

 $2.3.2.2.\,598 - 1541$  cm⁻¹: hot bands towards the tetrad region

The unpublished version of the newly determined tetrad {(002),(101),(200),(030)} parameters by Tyuterev et al. [128] can be found at the S&MPO online site (http://smpo.iao.ru/1396x659/en/lev/par/1/6/). Using this set, 14 hot bands in the range 598 –

1541 cm⁻¹ that go from the (010), (020), (001) and (100) states to the tetrad were re-calculated and are now included in the present HITRAN release with a significantly augmented number of transitions. The transition moment parameters of Ref. [128] for {(101),(030)} – (010) bands and those extrapolated from cold bands [141] were used.

$$2.3.2.3.\ 2889 - 3122\ cm^{-1}\ range:\ \{(013),(112)\} - (010)\ bands\ ;\ 3564 - 3752\ cm^{-1}\ range:\ \{(023),(122)\} - (010)\ bands$$

In the ranges  $2889 - 3122 \text{ cm}^{-1}$  and  $3564 - 3752 \text{ cm}^{-1}$  the  $\{(013),(112)\} - (010)$  and  $\{(023),(122)\} - (010)$  hot bands are updated using the same upper-state EH parameters as for the corresponding cold bands and the ETM parameters of Mikhailenko et al. [142]

$$2.3.2.4.\ 3623 - 4229\ cm^{-1}$$
: hot bands to the V=5 triad  $\{(005), (311), (104)\}$ 

New line lists for 14 hot bands in the range 3623 – 4229 cm⁻¹ corresponding to the transitions from the (010, (001) and (100) states to the updated [120] triad {(005),(311),(104)} of five vibrational quanta are also included.

2.3.2.5. 4384 – 4826 cm⁻¹: missing hot bands to (213),(321) and (114) upper states

The coupled vibrational states  $\{(213),(321),(114)\}$  have been re-analyzed in Ref [143] with the FTS spectra recorded in Reims. Using these upper-state parameters we now provide four associated hot bands [119] in the range 4384 - 4826 cm⁻¹ that were missing in previous HITRAN versions. The ETM parameters were extrapolated from the corresponding cold bands: for  $2v_2+v_2+3v_3-v_1$  and  $v_2+v_2+4v_3-v_3$  from Ref [144], for  $2v_2+v_2+3v_3-v_2$  from Ref [145], and for  $3v_2+2v_2+v_3-v_1$  from Ref [137].

### 2.3.3 Validations and empirical corrections

As mentioned in the introduction to section 2.3, the second step was a validation of effective model calculations by a comparison of simulated absorption cross-sections with observed

spectra. To this end, high-resolution laboratory FTS spectra of Reims and Kitt-Peak (up to 5700 cm⁻¹) were used alongside MkIV balloon atmospheric spectra [146], using similar procedures to those described by Toon et al. in [147]. The MkIV balloon spectra cover 650-5650 cm⁻¹ simultaneously in every spectrum, so every window is fitted in every spectrum, making it easier to quantify window-to-window biases.

No major problems were seen with the initial O₃ data set prepared for HITRAN2016, at least for strong and medium-strength bands. The new line list provides improved spectral fits in the 2100 cm⁻¹ region as compared with HITRAN2012 and also better consistency with other bands in terms of the retrieved O₃ amounts in the 2000-2200 cm⁻¹ region. But it was found that empirical corrections were needed to some line positions to reduce residuals between observed and simulated cross-sections. These corrections concern a limited sample of about 0.1% transitions and are essentially of two types. First, small line position corrections are typically of the order of ~10⁻³ cm⁻¹ up to 10⁻² cm⁻¹ in a few cases. The most extreme deviations were corrected already for HITRAN2016, but additional corrections will be done in the near future. Secondly, there is an indication that pressure-induced line shift parameters may need to be added or extended in the future.

It is known [120] that for the fundamental bands the effective model fits reproduce observed spectra within experimental accuracy ( $\sim 10^{-4}$  cm⁻¹ for line positions), but the precision of the calculations gradually decreases for higher wavenumber ranges. The RMS of available line position fits increases on average to  $\sim 0.001$  cm⁻¹ in the 3000 cm⁻¹ region, to  $\sim 0.002$  cm⁻¹ around 4000 cm⁻¹ and to  $\sim 0.003$  cm⁻¹ in the 5000 cm⁻¹ range, primarily because of accidental resonances. A typical example is a sharp resonance of the (111) state interacting with the "dark" (040) state for [J, $K_a$ ]= [44, 4] that was not included in the effective model. This resonance

produces a perturbation of  $0.048 \text{ cm}^{-1}$  in line positions and mixing of the line intensities between coupled transitions. The upper-state energy level and six corresponding P, Q, R- transitions have been empirically corrected both for the cold  $v_1+v_2+v_3$  and hot  $v_1+v_2+v_3-v_2$  bands. The major part of the relatively large line position corrections ( $\sim 0.02-0.12 \text{ cm}^{-1}$  for about ten transitions) to previous HITRAN line lists occurs near 2.5- $\mu$ m and corresponds to the analysis of  $\{(103),(004),(310)\}$ -(000) bands by Perrin et al [148]. This region is extremely complicated to model because of the strong coupling of 9 bands including some "exotic" resonances, for instance $\{(103),(004)\}$  with (310) and  $\{(211),(202),(032)\}$ . Analyses of new spectra recorded in Reims for these bands are in progress as outlined in Refs. [120,121], but are not yet finalized. In the meantime, empirical corrections have been applied to corresponding series of levels, mostly around 4000 cm⁻¹, and to associated transitions. This process removes some noticeable residuals between simulated and observed absorption cross-sections.

Relative intensities of various bands play a key role in terms of the consistency of retrieved O₃ amount from different spectral windows. From this point of view, the HITRAN2016 line list is certainly much better than its predecessors. For example, the HITRAN2016/S&MPO linelist produces 8% larger retrieved amounts around 2182 cm⁻¹ than previous line lists, bringing this region into better consistency with others. This improvement can be attributed to the major update of the tetrad {(101),(200),(002),(030)} bands [120,128] in the 5-µm region and confirms the conclusion of Janssen et al [125] concerning the internal consistency of the S&MPO list that is now incorporated in HITRAN2016. The 3000 cm⁻¹ region yields the lowest retrieved O₃ amounts, about 5% lower than average, in fits to MkIV limb spectra and Kitt Peak lab spectra [146]. This indicates possible errors in line widths or intensities, since the errant line positions were corrected.

#### 2.3.4 Future work

There remain several issues for future work. The problem of absolute ozone intensities is still an important challenge [120,124] requiring very precise measurements and ab initio calculations with absolute uncertainties better than 1-3%. This concerns particularly the isotopic species. New data on ¹⁷O- and ¹⁸O- containing isotopologues will be produced in the near future. Modeling of the complicated 2.5  $\mu$ m region [120,121,148] must be significantly improved. In general, for the spectral intervals beyond the fundamental bands, the effective models do not provide experimental accuracy for all lines. In order to obtain line position accuracy that is better than 0.001 cm⁻¹ in the FTS range, which is important for atmospheric applications, we could extend experimental determination of ro-vibrational levels and systematically replace transition frequencies by available empirical (E - E V/h values as described in Refs. [149,150] where this approach was used for the analysis and line lists of CRDS spectra. Further extension of ozone spectral data to highly-excited states near the dissociation threshold (Ref. [151] and references therein) would also be important for the understanding of the dynamics of ozone formation and depletion [152] and for interpretation of satellite observations involving non-LTE effects in the upper atmosphere.

# 2.4. $N_2O$ (molecule 4)

The  $N_2O$  line list has been updated using the recently-published line list for the  $^{14}N_2^{18}O$  ( $^{14}N^{14}N^{18}O$ ) isotopologue [153]. In the immediate future, we will also add (for the first time) the line list that has already been constructed for the  $^{15}N_2^{16}O$  ( $^{15}N^{15}N^{16}O$ ) isotopologue based on Ref. [154]. Both line lists were generated using the results of the global modeling of the line positions and intensities within the framework of the method of effective operators. The polyad model of effective Hamiltonian (EH) [155] was used in both cases. The parameters of the

effective Hamiltonians were fit to the line positions collected from the literature. The global RMS of the line position fit for  $^{15}N_2^{16}O$  is 0.0011 cm⁻¹ and for  $^{14}N_2^{18}O$  is 0.0016 cm⁻¹. The uncertainties in the fitted line positions depend on the quality of experimental data and vary from 0.0046 cm⁻¹ to  $2\times10^{-6}$  cm⁻¹.

The line intensities were calculated using fitted sets of the effective dipole moment parameters for a particular isotopologue where possible. Note that the line intensity measurements for rare isotopologues of  $N_2O$  are very scarce. Therefore, based on the theoretical results of Ref. [156] (in which contributions due to isotopic substitution to the values of the effective dipole moment parameters were shown to be negligible) the majority of the line intensities were calculated using the effective dipole moment parameters of the principal isotopologue [157]. The partition sums were taken from Ref. [112] for  $^{14}N_2^{18}O$  and calculated using direct summation for  $^{15}N_2^{16}O$  [154].

It is well known that the nitrous oxide EH polyad model does not work in a totally satisfactory manner because of the existence of a number of Coriolis and anharmonic interpolyad resonance interactions (see, for example, Refs. [158,159]). But usually these kinds of interactions take place in the high wavenumber region for the weak bands. Adopting an intensity cutoff of  $10^{-29}$  cm⁻¹/molecule cm⁻² at 296 K , including the isotopic abundance factors  $1.986 \times 10^{-3}$  for  $^{14}N_2^{18}O$  and  $1.32 \times 10^{-5}$  for  $^{15}N_2^{16}O$ , overcomes the problem of the interpolyad resonance perturbations in practice because the most affected transitions are omitted. Nevertheless, line parameters for the very weak bands have to be used with caution.

For the  $^{14}N_2^{18}O$  isotopologue, Toth's [160] lines given in HITRAN2012 are retained and only new line parameters from the line list of Ref. [153] are added to HITRAN2016. Line positions from the line lists of Refs. [153,154] are assigned uncertainty code 6 (0.000001-

0.00001 cm⁻¹) in MW region and 3 (0.001-0.01 cm⁻¹) in the IR region. For the line intensities from these line lists, the uncertainty code 3 (>20%) is adopted.

Line-shape parameters as well as Rosenkranz [35] line-mixing coefficients for the main isotopologue lines belonging to the  $v_3$  ro-vibrational band are taken from the work of Loos et al. [161]. These parameters originate from a multispectrum fit of air broadened Fourier-Transform transmittance measurements at ambient temperature. The experimental data of Loos et al. was fitted using a quadratic speed-dependent Voigt model based on the Hartmann-Tran profile [7,8,14], extended to account for line mixing in the Rosenkranz first-order perturbation approximation [35]. As a result, air-broadening, air-speed-dependence, air pressure shift and line-mixing parameters were retrieved in the spectral range 2184.8 to 2251.6 cm⁻¹ for P- and R-branch lines up to J''=40 and 39, respectively.

#### 2.5. *CO* (*molecule* 5)

In order to improve the quality and consistency (between bands and isotopologues) of the intensities of the lines of carbon monoxide in HITRAN, a new piece-wise dipole moment function (DMF) was created [162] using the direct fit approach that was developed in Ref. [163], supplemented with new ab initio calculations [162]. In order to ensure the best results, new experiments were carried out in Ref. [162] for the 4-0 band and the first measurements of the 6-0 band were performed to add into the direct fit.

The line intensity and position calculations were performed using this new DMF in conjunction with the wavefunctions calculated from the experimentally-determined potential energy function (PEF) from Coxon and Hajigeorgiou [164]. Calculations were carried out using version 8.0 of LEVEL [165].

The new dipole moment function has proven to be a substantial improvement over the one used previously. In this edition of the database it was used to calculate HITRAN intensity values in all the bands of all isotopologues. Figure 6 of Li et al. [162] shows a comparison of the available experimental data in different bands of the principal isotopologue, with new calculations, HITRAN2012, and ab initio calculations. It is clear that intensities from Li et al agree much better with experimental values than previous data. This also becomes important for the consistency of intensities between the isotopologues (due to the use of the same dipole moment function for all of them); previously a lot of lines for the principal isotopologues had intensities from accurate laboratory measurements whereas minor isotopologue intensities were still calculated with outdated dipole moment functions. Finally, the new work allows one to add a few more bands and also line lists for radioactive isotopologues of carbon monoxide, which, as we mentioned in the CO₂ section, are provided in static files rather than through HITRANonline interface.

Apart from the intensities, the Li et al line list features: a) line positions calculated based on the potential energy function from Ref. [164]; b) air- and self-broadening parameters from a sophisticated semi-empirical approach, c) air-induced shift obtained for all the lines by extrapolating measurements carried out in the 2-0 band by Devi et al. [166], where, the Hartmann and Boulet [167] approach was used for extrapolating; d) widths (and their temperature dependencies) and shift due to pressure of H₂ and CO₂ important for planetary and combustion research were also provided employing semi-empirical approaches similar to the ones used for air and self pressure-induced values.

We have adapted the Li et al line list for this edition of the HITRAN database with a few notable changes:

- 1) Since the original Li et al line list was aimed at high-temperature applications, it was truncated here with an intensity cutoff of  $10^{-31}$  cm⁻¹/( molecule ·cm⁻²) (in natural abundance of the isotopologues) for all the lines except those in the MW region where a cutoff of  $10^{-42}$  cm⁻¹/( molecule ·cm⁻²) was used to satisfy some astrophysical applications.
- 2) It was found that there was a minor compatibility issue between LEVEL and PEF from Coxon and Hajigeorgiou [164]. This resulted in a growing (with rotational quantum number) deviation of the calculated line positions from their real values. This has now been fixed.
- 3) The line positions from the high-accuracy experiments that employ best calibration standards, including frequency combs, were used wherever available. In the first overtone, data from Pollock et al. [168] was used, in the second overtone data from Refs. [169,170] were employed, while in the fifth overtone Tan et al line positions were taken when available [171].
- 4) When high-quality experimental measurements of the air- and self-broadening and shift parameters were available in HITRAN2012, they were taken in place of the semi-empirical values from Li et al. This includes the values for the speed-dependent Voigt (with line mixing) profile for the first overtone from Devi et al. [166,172].
- 5) Helium-broadening parameters, their temperature dependence and pressure shift due to He were added. The broadening parameters were taken from Sinclair et al. [173] who provide both experimental and values obtained from a fit to a semi-empirical model. We used experimental values where available and semi-empirical values for all other transitions. Although these measurements were carried out only in the fundamental band, they were used here for all lines ignoring vibrational dependence. The temperature dependence of this broadening was taken to be 0.6 for all the lines based on the average value of the measurements by Picard-Bersellini et al. [174]. Finally, the pressure induced shifts are based on measurements of Luo et al [175] in the fundamental band and extrapolated wherever possible to other bands using the Hartmann and Boulet [167] approach.

#### 2.6. *CH*₄ (molecule 6)

The state-of-the art of methane data in the previous HITRAN releases has been described by Brown et al. [176]. Since then, several new studies have been in progress or have been published both for cold and room-temperature spectra [177–183] and hot spectra measurements [184,185], although many experimental spectra have not yet been fully analyzed. Toon et al. [147] have reported some deficiencies in available databases including HITRAN2012 [1] by simulations of solar occultation spectra, acquired by the JPL MkIV Fourier transform spectrometer from balloon, covering the 650–5650 cm⁻¹ region. These deficiencies have been addressed in this new edition, which includes superior spectral parameters and new lines and bands for ¹²CH₄ and ¹³CH₄. At this time no update was made to CH₃D. An overall brief picture of the update is given in Table 6, while the details are given in the subsequent sections. Note that only regions where revision to HITRAN2012 was carried out are shown. In the regions not mentioned in the table, HITRAN2012 line parameters have been retained.

Table 6. Scheme of HITRAN2016 update of ¹²CH₄ and ¹³CH₄. All regions not given in the table remain identical to the HITRAN2012 edition

Range (cm ⁻¹ ):	Positions	Intensities	Line shapes
0-1370	HIT12, with many hot bands removed. P1–P0 (v ₄ band only), P1–P1, P2–P2, P3–P3, P2-P1 and P3–P2 from MeCaSDa [186], ¹³ CH ₄ v ₂ band restored from HIT08	P1–P0 (v ₄ band only), P1– P1, P2–P2, P3–P3 and P3–P2 from MeCaSDa [186], P2-P1 from HIT08	HIT12 when available or algorithm from Ref. [176]
1370-4000	HIT12 with many line positions fixed using HIT08, Tyuterev et al. [187] and empirical adjustments of HIT12 to properly fit KP spectra. Several missing lines	HIT12	HIT12
4000-4315	restored from HIT08. HIT08 with many lines improved using some of the preliminary results from Birk et al. [188], and occasionally HIT12	HIT08 with many lines improved using some of the preliminary results from DLR Birk et al. [188]	HIT08 and in some cases HIT12

	HIT12 mixed with Devi et al [189]	HIT12 mixed
4499-4630		with Devi et al
	27 1 2	[189]
4670-5300	Nikitin et al. [190] with $10^{-27}$ cm ⁻¹ /(molecule·cm ⁻² ) cutoff	Lyulin et al.
	was applied in 4800-5000 cm ⁻¹ region	[191]
5300-5550	Nilstin et al. [102]	Lundin et el
3300-3330	Nikitin et al. [192]	Lyulin et al.
5550 5055	COCATE 1: 4 [102] C   12CH   Od   1   1   1   1   1   1   1   1   1	[191]
5550-5855	GOSAT list [193] for ¹² CH ₄ . Other isotopologues HIT12	GOSAT list
		[193], or
		algorithm from
		Ref. [176]
5855-6250	GOSAT list [193] with some lines from Nikitin et al [194]	GOSAT list
	and HIT12 based on validation for ¹² CH ₄ . For ¹³ CH ₄	[193], Lyulin et
	Starikova et al. [195].	al. [191] or
		algorithm from
		Ref. [176]
7920-8250	CRDS measurements Béguier et al [196]	Algorithm from
		Ref. [176]
9028-	FTS measurements Béguier et al [197]	Algorithm from
10435		Ref. [176]

Note: HIT16 = new HITRAN edition, HIT08 = HITRAN2008 edition [198], HIT12 = HITRAN2012 edition [1], KP = Kitt Peak FTS lab spectra.

# 2.6.1. Hot bands in the spectral range below 1370 cm⁻¹.

The HITRAN2012 database exhibited some noticeable problems, including those with hot band lines of ¹²CH₄ in the low wavenumber range. The line list in this spectral region relies on calculations resulting from the effective Hamiltonian and dipole moment parameter fits, since extensive experimental line assignments have not been performed for many years. However, extrapolation errors have been introduced in some small spectral regions that were insufficiently well characterized. Recently, a new global fit of methane lines including high-temperature emission data in the 1100–1500 cm⁻¹ region was published [185]. This study included a huge number of new assignments and allowed a much more reliable modeling of hot-band lines. It has resulted in a determination of effective Hamiltonian parameters. The results were first used to update the MeCaSDa database [186] of calculated methane lines. Figure 11 compares the

resulting computed cross-sections with the HITRAN2012 and ExoMol [199] databases. This plot, in logarithmic scale, clearly displays that there was likely a problem in HITRAN2012 but better agreement of the new calculated line list with Exomol.

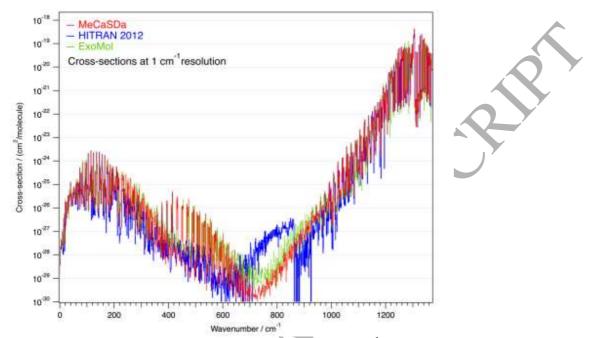


Figure 11. Comparison of cross-sections calculated at 1 cm⁻¹ resolution between different databases: MeCaSDa [186] (used for the present HITRAN update), HITRAN2012, and ExoMol [199].

It is well known that measurement and modeling of line intensities in high-temperature spectra are quite difficult, particularly in non-Boltzmann conditions. In order to avoid large uncertainties in extrapolated intensities, as was the case for some hot bands in the previous HITRAN editions, we have checked new lists against *ab initio* calculations. The line lists for hot bands resulting from new experimental spectra analyses by Amyay et al.[185] were compared to first-principle variational calculations of line intensities by Rey et al. [200,201], based on the *ab initio* dipole moment and potential energy surfaces of Nikitin et al. [202–204] as reported in the TheoReTS database [205]. An example of comparison is given in Figure 12. Those bands that were in good intensity agreement were considered sufficiently reliable to be incorporated in this new HITRAN release. Some other bands require further investigation and were not included.

Note that, at this stage, these qualitative comparisons did not involve line assignment or line position validations.

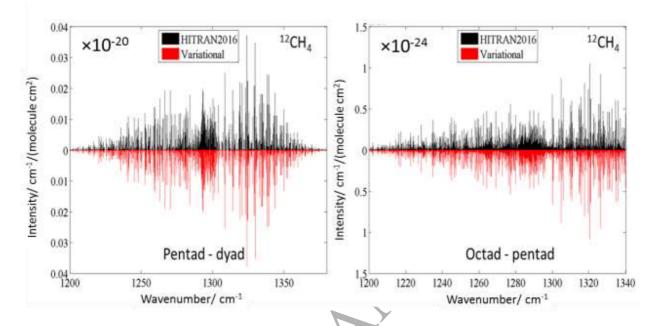


Figure 12. Comparison of line intensities for hot bands of ¹²CH₄ in the 7 to 8 µm range using the HITRAN2016 updated list based on experimental spectra analyses, Amyay et al. [185] (upper panels in black) with variational calculations, Rey et al. [201] based on ab initio dipole moment [202] and potential energy [203,204] surfaces using the line list included in the TheoReTS database [205] (lower panels upside down in red).

This new calculated line list was thus used to replace some of the HITRAN2012 lines in order to generate the HITRAN2016 update, for the P1–P0 (Dyad–GS, for the v₄ band only), P1–P1 (Dyad–Dyad), P2–P2 (Pentad–Pentad), P3–P3 (Octad–Pentad) and P4–P3 (Tetradecad–Octad) transitions (a schematic of the polyads can be seen in Fig. 2 of Ref. [5]). Table 7 shows the spectral regions and the number of lines that exist in MeCaSDa and what proportion of them was adapted for HITRAN2016. It is important to mention that validations against theoretical [205] and laboratory [185] spectra suggested adapting the P2-P1 line list from MeCaSDa, with intensities adapted from HITRAN2008 wherever possible.

It should also be noted that, recently, new highly-accurate measurements of pure rotation lines were performed in the Dunkerque group [206]. These lines have been included in the global fit leading to the effective Hamiltonian parameters used for the present update.

Table 7. Number of calculated  12 CH₄ lines taken from the MeCaSDa database [186] for some transitions, up to a maximum J value. The difference for the number of lines in the two databases is due to a different intensity cut-off with  $I_{min}$  down to  $10^{-40}$  cm/molecule in MeCaSDa. The weakest transitions were not retained for the present HITRAN update that includes  $I_{min} = 10^{-25}$  for P3-P2 and  $10^{-30}$  cm/molecule for other hot bands.

Transition	P1-P0 (v ₄ )	P1-P1	P2-P2	17	
			$< 300 \text{ cm}^{-1}$	D2 D2	
				P3–P2 1370 cm ⁻¹	
				1570 CIII	
$J_{ m max}$	30	30	30	30	
# lines (MeCaSDa)	13315	54677	457620	2175406	
# lines(HIT16)	9662	9721	10701	950	

# 2.6.2. 1370-4000 cm⁻¹ region

This region mainly includes the v₂ band, the Pentad bands and relatively weak bands at the lower part of the Octad. As discussed by Toon et al. [147], HITRAN2012 data were plagued with erroneous line positions for lines of intermediate strength, many of which had been substantially superior in the HITRAN2008 edition of the database. It was found that a large number of these lines had been assigned line position uncertainty code 1 (i.e. between 0.1 and 1 cm⁻¹) and originate from the model used in Daumont et al. [207]. For HITRAN2016, wherever possible in the Pentad region, these transitions were reverted to HITRAN2008 and, where this was not possible, line positions from Tyuterev et al. [187] were used. Unfortunately not all problematic line positions were easy to identify on a global scale by just the uncertainty index. Validations against Kitt Peak FT spectra recorded at multiple thermodynamic conditions have shown that more errors associated with line positions remained although the offending lines were not labeled

with a low uncertainty code. The most obvious outliers had been identified and fixed either by employing Ref. [187] or adjusting line positions to fit the low-pressure KP experimental spectra. This provided a significant improvement of the residuals, including those discussed in Toon et al. [147].

2.6.3.  $4000 - 4630 \text{ cm}^{-1}$  (correspond to strongest Octad bands).

Just as for the Pentad region, HITRAN2012 data had substantial errors in line positions for lines of intermediate strength, while better values actually existed in the HITRAN2008 edition of the database. However, unlike the Pentad region, the intensities of lines in the Octad region were also often inferior to those from HITRAN2008. Also, unlike the Pentad region, reverting line positions of HITRAN2012 lines with uncertainty code 1 to the HITRAN2008 values did not always work. One should also keep in mind that HITRAN2008 was not better than HITRAN2012 everywhere in this region. In addition to lacking ¹³CH₄ lines in that region, HITRAN2008 contained many duplicated transitions. Often a transition would appear as an empirical unassigned line as well as predictions from Hilico et al [208]. With that being said, some observable transitions were also missing in HITRAN2008.

The following procedure was therefore developed:

1) In the 4000-4315 cm⁻¹ region, the HITRAN2008 line list was used as the starting point. Many unassigned empirical lines were then given assignments and lower-state energies from corresponding predicted Hilico et al [208] lines, while predicted lines (usually with inaccurate line position and intensity) were removed. In the region 4190-4315 cm⁻¹, parameters of lines with the worst remaining residuals (with respect to low-pressure Kitt Peak FTS spectra) were then replaced with the preliminary results of Birk et al. [188],

which used quantum assignments of yet unpublished theoretical work [209] that was not always the same as in HITRAN2012.

- 2) In the 4315-4499 cm⁻¹ region, HITRAN2012 parameters were retained.
- 3) In the 4499-4630 cm⁻¹ region, a mixture of HITRAN2012 with parameters from Devi et al (where available) was used [189].

Figure 13 shows fits to TCCON spectra measured from Park Falls, WI, in March 2005 at a solar zenith angle of 58 deg., in windows used by TCCON to retrieve CO. The left panels show fits to the region occupied by the CO P-branch. The right panels show fits in the R-branch region. The CO lines are less than 10% deep and so are mainly hidden by stronger CH₄ and solar lines. The upper panels show fits using HITRAN 2012 CH₄; the lower panels with the HITRAN 2016 CH₄ linelist. The rms spectral fits and the peak residuals are significantly reduced in both windows with the HITRAN2016 linelist. The retrieved CH₄ amounts (written over the top right of each panel) are much more consistent between the two windows using the HITRAN 2016 linelist (0.1%), than when using HITRAN 2012 (2%). It implies that the 2016 CH₄ intensities are much more self-consistent across these two regions than in HITRAN 2012. With that being said there is still large room for improvements. A more consistent Octad list in terms of line parameters and assignments is planned to be worked out for the next HITRAN update.

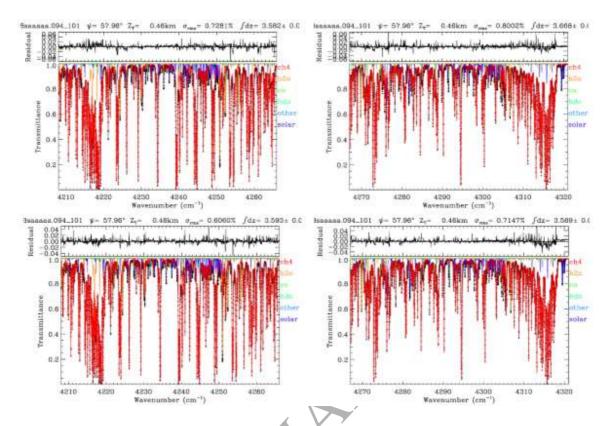


Figure 13. Fits to TCCON spectra measured from Park Falls, WI, in March 2005 at a solar zenith angle of 58 deg., in windows used by TCCON to retrieve CO. The left panels show fits to the region occupied by the CO P-branch. The right panels show fits in the R-branch region. The upper panels show fits using HITRAN 2012 CH₄; the lower panels with the HITRAN 2016 CH₄ linelist. Note different scales.

# 2.6.3. Tetradecad range 4800 – 6250 cm⁻¹.

Contrary to the low-energy range, in the Tetradecad range and above, positions and intensities are mostly experimental values.

The Tetradecad of methane contains 14 bands with 60 strongly coupled sub-bands with very congested spectra difficult to analyze because of numerous resonance perturbations. The previous HITRAN2012 line list was mainly based on the analysis of Refs. [210,211], which included about 3000 assigned lines. This was only a partial assignment with relatively large uncertainties for about 70% of weak sub-bands. In HITRAN2016, major updates have been included in four spectral intervals. Long-path Fourier Transform spectra (L= 1600 m) in the

lower Tetradecad interval 4800-5300 cm⁻¹ recorded in Reims have been re-analyzed by Nikitin et al. [190]. The list includes transitions with line positions adjusted according to their measured values and transitions with calculated line positions, all intensities being derived from the model fit. For HITRAN2016 in the 4800-5000 cm⁻¹ region, transitions with intensities greater than 10⁻²⁷ cm⁻¹/(molecule·cm⁻²) were retained.

In the next interval, 5300-5550 cm⁻¹, the list contains mostly observed positions and intensities. The analysis is in progress [192], but a preliminary empirical line list is included in the present release because the previous versions were quite poor in this range.

In the interval 5550-6250 cm⁻¹, a recent version of the GOSAT line list [193] is included. In the 2v₃ region (5855-6250 cm⁻¹), the GOSAT line list [193] was supplemented with predominantly experimental line parameters of the WKLMC list (Wang-Kassi-Leshchishina-Mondelain-Campargue) obtained from high-sensitivity laser measurements in Grenoble [212]. The previous GOSAT assignments [211] included in the HITRAN2012 list have been considerably extended in the recent work [194] based on new analyses in the 5855-6250 cm⁻¹ range. The modeling of these complicated spectra was accomplished using an approach combining *ab initio* calculations with effective Hamiltonian (EH) fits. Initial values of the EH parameters were first computed from the molecular potential energy surface of Nikitin et al. [203] using irreducible tensor techniques [213,214] via the high-order contact transformation method as described by Tyuterev et al. [187]. This information has been only partially adapted for HITRAN2016. Indeed, the validations against experimental and atmospheric spectra (similar to those done in Toon et al. [147]) showed that the GOSAT line list [193] produces better residuals than the one from Ref. [194], although the latter one is substantially more complete in terms of both amount of lines and proportion of assigned lines. In the end, the GOSAT line list

[193] was supplemented with some lines from Ref. [194] and occasionally with HITRAN2012. For instance, very accurate line positions measured with frequency comb in Zolot et al [215] were retained. In the future, a more sophisticated combination of the GOSAT line list [193] and the one from Ref. [194] will be performed. For broadening a combination of parameters from Lyulin et al. [191], GOSAT line list and HITRAN2012 was used.

# 2.6.4. Higher wavenumber ranges

In the Icosad range (6300 – 7500 cm⁻¹), the WKLMC list provided empirical values of the lower-state energy levels. Recently, on the basis of global variational *ab initio* spectra predictions [180], about 13000 transitions of 108 new bands have been assigned [216] in the WKLMC list at 80 K and 296 K. The assigned experimental list at 80 K as well as calculated ones are provided as Supplementary Material of Rey et al. [216]. These assignments will be transferred and extended in the WKLMC list at 296 K and included in the next update of HITRAN.

Figure 14 gives an overview comparison between the HITRAN2012 and HITRAN2016 lists of ¹²CH₄ above 7920 cm⁻¹. New room-temperature measurements by Cavity Ring Down Spectroscopy (CRDS) in the 7920-8250 cm⁻¹ interval have allowed better characterizing of the 1.25-μm methane transparency window [21]. In the Triacontad region (8250-9028 cm⁻¹), line parameters relying on FTS measurements by Brown [42] remain unaltered with respect to HITRAN2012. Above 9028 cm⁻¹, new FTS line parameters are adopted in the region of the 3*v*₃ band (9028-9200 cm⁻¹) [22]. From the same FTS spectra, the first empirical line list was retrieved in the Tetracontad region (9520-10435 cm⁻¹) [22]. It is worth mentioning that between 7920 cm⁻¹ and 10923 cm⁻¹ no assignments or empirical lower-state energy values are provided

except for a few tens of transitions in the region of the  $3v_3$  band near  $9046 \text{ cm}^{-1}$ . Complementary information in this case can be provided by variational ab initio predictions [31]. The estimated error for line positions in the best first-principle calculation in the 1- $\mu$ m range is on average ~ 0.3 – 2 cm⁻¹; that is not sufficient for high-resolution applications. However, intensities are in qualitatively good agreement with observations, see Fig. 2.6.5. Recent ab initio intensity results [217] for the strongest  12 CH₄ lines below 7600 cm⁻¹ have confirmed accurately measured intensity values to 1-3 % on average for the "stable" transitions non perturbed by accidental resonances. Predicted spectra [31] include lower-state energies for all lines and can be recalculated for various temperature conditions that make it suitable for low- and medium-resolution modeling of observations in planetary atmospheres similar to those reported for Titan in Ref.[45].

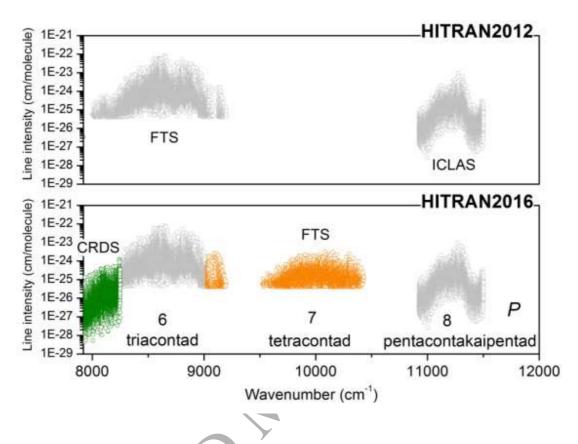


Figure 14. Overview comparison of the HITRAN line list of  12 CH₄ above 7920 cm⁻¹ in the 2012 and 2016 editions. In this region, line parameters all have an empirical origin: CRDS between 7920 and 8250 cm⁻¹ [196], FTS by Brown in the 8250-9028 cm⁻¹ interval [218] by Béguier et al. in the 9028-10923 cm⁻¹ interval [197], and ICLAS between 11000 and 11500 cm⁻¹ [219]. The different polyad and corresponding quantum numbers, P, are indicated. (The polyad number P is equal to  $2(V_1 + V_3) + V_2 + V_4$ , where  $V_i$  are the normal mode vibrational quantum numbers).

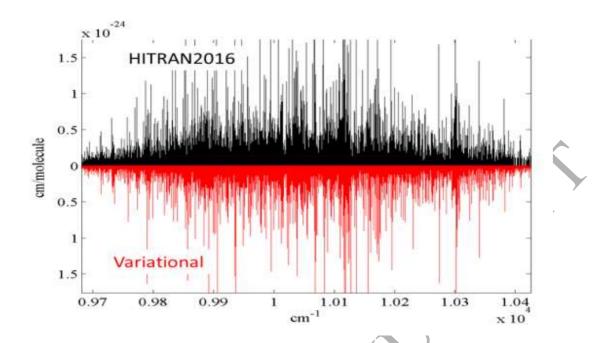


Figure 15. Comparison of unassigned experimental line list of ¹²CH₄ in the Tetracontad range (Béguier et al. [197]) included in HITRAN2016 (upper panels in black) with variational calculations, Rey et al. [205] based on ab initio dipole moment [202], and potential energy [203,204] surfaces using the line list included in the TheoReTS database [205] (lower panel upside down in red).

Finally, in the 10923-11502 cm⁻¹ spectral range, the HITRAN lists reproduce line parameters derived by Intracavity Laser Absorption Spectroscopy (ICLAS) [220]. For the ICLAS data [176,220], empirical lower-state energy values were derived by the two-temperature-method [221].

# 2.6.5. ¹³CH₄ isotopologue

In HITRAN2012, the entire  $v_2$  band of  $^{13}CH_4$  (centered at about 1550 cm $^{-1}$ ) was accidentally dropped from the compilation. This band is now restored back from HITRAN2008 [198].

For the ¹³CH₄ isotopologue, new measurements and analyses in the 3750-4700 cm⁻¹octad range have been reported by Brown et al. [222] for temperatures between 80 and 296 K. This included over 4700 positions and 3300 intensities for assigned observed transitions. The line list based on the analysis and spectra fits of this range contains 9500 calculated lines. It is available

as a supplementary material attached to Ref. [222] and will be incorporated in the next HITRAN update.

First assignments of the strongest bands at the high-wavenumber boundary of the ¹³CH₄ Tetradecad (5852-6200 cm⁻¹ region dominated by  $2v_3$ ) have been reported by Starikova et al. [195]. Differential Absorption spectra (DAS) recorded in Grenoble [223] at 80 K and 296 K have been used for this analysis. For the 80 K spectra, about 2900 lines of 3717 observed lines were assigned. The theoretical model in both spectral intervals, 3750-4700 cm⁻¹ and 5852-6200 cm⁻¹, was based on the combined approach [187] using ab initio predictions for line positions and intensities [224] with subsequent EH fits, which is similar to that for the main ¹²CH₄ isotopologue as outlined in Section 2.6.6. Supplementary materials of Ref. [195] contain a partially-assigned experimental DAS list at 80 K. For the present HITRAN2016 release, the assignments were extended for the 296 K list. Note that many observed absorption features in the congested spectrum could have contributions from various overlapping transitions. As the assignment criteria, the ratio  $R = I_{re-calc}/\{I_{obs}(296K)\}$  was used as well as the coincidence of  $J_{low}$ obtained theoretically and via the empirical 2T-method. Here I_{re-calc} stands for intensities recalculated from the 80 K DAS list via the Boltzmann factor. In doubtful cases, but with consistent  $J_{low}$  values, we have kept empirical line parameters. For big R deviations and inconsistent  $J_{low}$  values, a default value  $E_{low} = 333.3333$  cm⁻¹ was applied. The self- and air- line broadening were calculated following the method described in the GOSAT paper [211].

Figure 16, giving the overview of  13 CH₄ lines in the  $2\nu_3$  range, shows that most of the strong and medium lines in the HITRAN2016 release have consistent assignments.

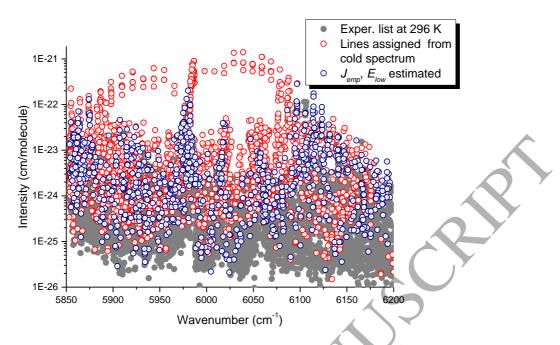


Figure 16. Overview of assigned lines [195] in the experimental  $^{13}CH_4$  DAS list [223] included in HITRAN2016 in the 1.66- $\mu$ m range.

An extension of this work using the approach of Ref. [216] is in progress for the Icosad range of  $^{13}\text{CH}_4$  [225].

#### 2.6.6. CH₃D and other isotopologues

The parameters for this singly-deuterated methane molecule have not been updated for the 2016 edition. We note a recent paper by Bray et al. [226] which indicated the need for updating the intensities of CH₃D in the pure rotational band. We are considering this work for a future update of HITRAN.

#### 2.6.7 Line-shape parameters for methane

Our knowledge of pressure broadening of methane by air, N₂, H₂, He, CH₄, and other gases remains incomplete, particularly for the near-infrared and for weaker transitions in all spectral regions. Basic coefficients of Lorentz broadening (widths, shifts, temperature dependences) all vary as a function of the transition quantum numbers, and values obtained for the fundamental bands are not easily applicable to the very complicated polyads having multiple underlying

vibrational states. Available measured values for Lorentz pressure broadened line-shape parameters from the previous edition of HITRAN were transferred into the new compilation, but most of the lines have been given crudely-estimated air- and self-broadened half widths, as described by Brown et al. [176]. Good theoretical models, confirmed by measurements, must be implemented in order to have accurate values, not the estimates used here for 99% of the present database. Since the release of the 2012 database, additional methane line-shape measurements have become available in several spectral regions for ¹²CH₄ [179,181,183,189,227–232], and CH₃D [226,233,234], and these will be added in the future updates to HITRAN2016.

The recent study of Mendonca et al. [235] illustrates the growing evidence that for remote-sensing applications requiring the highest accuracies of line parameters to model the observed methane spectrum, Voigt lineshapes are inadequate. It is well known that collisional line mixing can significantly affect absorption spectral shapes, and line mixing cannot be neglected for accurate retrievals of atmospheric CH₄ abundance. The HITRAN2012 database [1] had included the line-mixing parameters for CO₂, CO and O₂, however without providing guidance or tools for users to implement these parameters in their calculations. The new structure of the HITRAN database [236] and the HITRAN Application Programming Interface (HAPI) [5] provide the potential for the inclusion of line-mixing parameters in the database and provide tools for their implementation.

A theoretical approach proposed to model line-mixing effects for CH₄ by Tran et al. [[237–239] has been successfully applied to calculate absorption in the  $v_3$ ,  $v_4$  and  $2v_3$  bands. The same model has been used to simulate the spectra in the  $v_2+v_3$  band as a test to demonstrate whether it can be applied to other vibrational bands. Comparisons with spectra measured under different pressure conditions [189] show that line mixing only weakly affects the absorption spectral line

shape of this band under these pressure conditions. Therefore, more calculations for other vibrational bands under large pressure conditions are needed to demonstrate if this model can be applied to various vibrational bands. While the work of Tran et al. ([239] and references therein) represents significant progress in understanding non-Voigt line shapes in methane, additional laboratory studies of line mixing, speed dependence and narrowing are required to provide the basic parameters for future methane compilations in the spectral regions most utilized for remote sensing.

#### 2.7. O₂ (molecule 7)

An extensive update of the HITRAN oxygen line list for the new edition makes use of the following significant progress on: (1) a self-consistent set of energy levels and line positions for the entire HITRAN oxygen line list; (2) lineshape, line mixing, collision induced absorption (CIA), and new speed-dependent Voigt (SDV) formalism for the A-band; (3) HT profile parameters for the B- bands. We also added the  $b^I \Sigma_g^+$  (v=3)- $X^3 \Sigma_g^-$  (v=0) and  $b^I \Sigma_g^+$  (v=2)- $X^3 \Sigma_g^-$  (v=1) bands.

### 2.7.1 Line positions and energy levels

A self-consistent set of spectroscopic constants from an updated isotopically invariant Dunham fit by Yu et al. in 2014 [240] were used to update line positions and energy levels throughout the entire HITRAN2012 oxygen line list, except for the so-called zero-frequency lines (due to Zeeman splitting of the ground state), see HITRAN1986 paper for details [241]. This updated Dunham fit was obtained by adding new microwave and infrared measurements [242,243] to the first global analysis of O₂ by Yu et al. in 2012 [244] that simultaneously fits all available experimental line positions for all the bands and isotopologues from the microwave to UV region. The new microwave measurements [242] include 324 rotational transitions in the

 $a^1\Delta_g \ v = 0$  and 1 states of the six  $O_2$  isotopologues and helped determine two more hyperfine parameters, the electric quadrupole interaction eQq and the nuclear spin-rotation interaction,  $C_I$ . The new infrared work [243] reported 1644  $b^1\Sigma_g^+ - X^3\Sigma_g^-$  transition frequencies for the six  $O_2$  isotopologues and revealed a  $0.2 \text{ cm}^{-1}$  calibration error in the  $^{17}O^{18}O \ v' - v'' = 1-0$  Raman data of Edwards et al. [245]. It resolved discrepancies in the Raman data for  $^{16}O^{17}O$ ,  $^{17}O^{17}O$ , and  $^{17}O^{18}O$ , and improved the vibrational parameterization of the ground electronic state.

A quantum number assignment error has been fixed in HITRAN2016 for 17 lines in the  $\Delta N\Delta J = PO$  branch of the ¹⁶O¹⁶O a-X(v',v'') = (0,0) band. These lines were incorrectly labeled with a shift of two in N'' and J'' in HITRAN2012, as was pointed out in Yu et al. [240].

Band-by-band comparisons of the updated line positions and lower state energies versus the HITRAN2012 compilation were performed. For all the bands in the  $a^1\Delta_g - X^3\Sigma_g^-$  system (for which the data in HITRAN originate from Refs. [7–10], systematic differences, smaller than experimental uncertainties, were found for measured lines and expected deviations were found for extrapolated lines. As depicted in the upper left panel of Figure 17, systematic differences of 0.0007 cm⁻¹ were found for the  $^{16}O^{16}O$  a-X(v',v'')=(0,0) band. Similar deviations were found for all other bands except the A-band (b-X(v',v'')=(0,0)) of  $^{16}O^{16}O$ ,  $^{16}O^{17}O$  and  $^{16}O^{18}O$ , where the data originated from Refs. [250,251] (see discussion below).

As shown in upper right, lower left and lower right panels of Figure 17, deviations are very small for one sub-group of the A-band transitions while they are large for another sub-group. The sub-group of transitions with small deviations belongs to the  ${}^{P}O$ ,  ${}^{P}P$ ,  ${}^{P}Q$  branches while the other sub-group with large deviations belongs to the  ${}^{N}O$ ,  ${}^{R}R$ ,  ${}^{R}Q$ ,  ${}^{R}S$  and  ${}^{T}S$  branches.

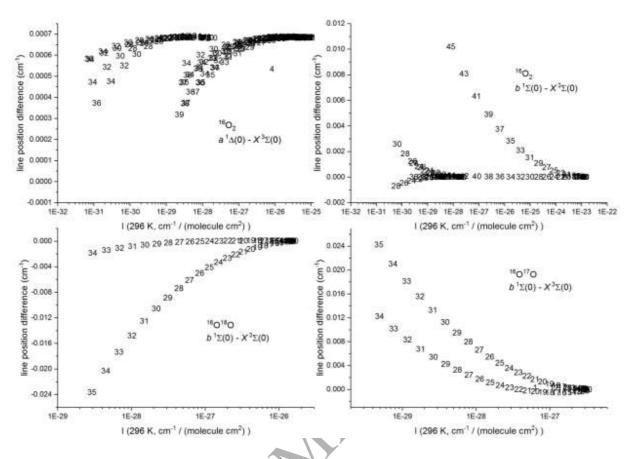


Figure 17 Comparisons of the updated positions versus the HITRAN2012 compilation. The panels show differences between listed frequencies labeled by the quantum number J" plotted vs. line intensities for the transitions for the atmospheric IR  $a^1\Delta_g - X^3\Sigma_g^-$  system of  $^{16}\mathrm{O}^{16}\mathrm{O}$  and the atmospheric  $b^1\Sigma_g^+ - X^3\Sigma_g^-$  system of  $^{16}\mathrm{O}^{16}\mathrm{O}$ ,  $^{16}\mathrm{O}^{18}\mathrm{O}$  and  $^{16}\mathrm{O}^{17}\mathrm{O}$ . Note that half of the  $b^1\Sigma_g^+ - X^3\Sigma_g^-$  magnetic dipole transitions were removed from the plots for the purpose of clarity because the spin pairs were on top of each other.

### 2.7.2. A-band region near 762 nm

The line list for the A-band in HITRAN2012 originates from the data from Long et al. [250,251] papers. Here an update of the principal isotopologue magnetic dipole allowed transitions is given that incorporates a self-consistent model of collisional lineshapes, collisional line mixing and collision induced absorption in a multispectrum fit to a wide range of experimental data sets described in Drouin et al. [252] (hereafter referred to as Drouin 2017).

The lineshape chosen was the speed-dependent Voigt profile and effects of line mixing were introduced as a fixed W-matrix derived from the theoretical line mixing formalism of Tran, Boulet and Hartmann in 2006 [253]. Strong correlation was observed between the scaled line mixing parameterization and the collision-induced absorption derived as remaining non-resonant absorption. Correlation was also observed between the line-by-line parameters correlations in the R-branch band-head region and the scaled line mixing parameterization. The self and airbroadening parameters as well as air-pressure shift parameters do not deviate significantly from HITRAN 2012, with smooth trends that closely match the polynomial expressions of Long et al. that were used in the prior compilation. Unlike HITRAN 2012, minor erratic deviations of these parameters (which were minimized in the chosen scaling of the line mixing) are retained in the compilation for self-consistency. The self-pressure shift parameters are provided for the first time in this HITRAN edition, and these values differ systematically by 0.006 cm⁻¹atm⁻¹ from The air broadening temperature dependence is also those given by Long et al. [251]. systematically shifted to higher values in comparison to HITRAN 2012. The self-broadening temperature dependence is also provided for this band for the first time in HITRAN and it is even more systematically shifted in comparison to published values from Brown & Plymate [254]. Temperature dependences of the pressure-shift are also included in the new line-by-line parameterization; there is no prior parameterization for comparisons of these values. The final fitted line-by-line parameter is the 'speed-dependence' of the pressure broadening, which replaces the Dicke parameter determined in Long's work and included in the supplemental listing of HITRAN 2012. Qualitative comparison of these two 'narrowing' parameters shows that there is a similar m dependence in each parameter set, each with a more pronounced slope at m < 0 (the P-branch). Use of the speed-dependent Voigt allows this lineshape profile to be cast

within the more encompassing Hartmann-Tran profile. Line-by-line line mixing parameters (Rosenkranz parameters) were not fitted in Drouin 2017 [252], but rather a full W-matrix formalism based on theoretical work, empirically modified to improve the multispectrum fit quality, was implemented. The empirical corrections that attenuated line mixing at high rotational excitation (see Tran and Hartman, [255]) were abandoned in favor of the removal of mixing between differing spin-states. The effects of this correction are clearly evident in the derived Rosenkranz parameters, which oscillate between even and odd J (or m) values when the spin-changing collisional mixing is allowed, but follow a smooth trend with J (or m) when they Predoi-Cross et al. [256,257] provide fitted room-temperature Rosenkranz are removed. parameters (and other lineshape parameters) that agree with the parameters of Drouin 2017. The modified line mixing parameterization significantly alters the derived collision-induced absorption, with this absorption appearing narrower and stronger, with a shape and extent that conforms more closely to the resonant band. The collision-induced absorption derived for air is provided separately in the corresponding section of the HITRAN database, see section 4 for details.

For the convenience of HITRAN users, two transformations have been applied to the lineshape/line mixing parameterization given by Drouin 2017. First, given the primary usage of the oxygen database to model the Earth's atmosphere, the foreign-broadening lineshape parameters presented in Drouin 2017 have been converted to air-broadening parameters using the binary mixing ratio  $[N_2]:[O_2] = 0.79:0.21$ . Second, the scaled, full-W-matrices for foreign and self line mixing and the associated temperature dependence were evaluated for 1 atm of air at four 'standard' temperatures and then composed into first order line-by-line Rosenkranz parameters. This representation of the line mixing is based on similar representations provided

for the fine-structure line mixing in the microwave spectrum and is both easier to tabulate as well as more efficient to compute. We note that the multidimensional oxygen A-band cross-section table (ABSCO 5.0), now in use for the reduction of OCO-2 [64] atmospheric data, utilizes the parameterization of Drouin 2017 in its native form, and was calculated with the same source code as the multi-spectrum fitting program. The differences between the ABSCO 5.0 table and the HITRAN database are due to the two cosmetic modifications indicated above, as well as the replacement of line centers with the more precise values derived from the work of Yu et al. [240] described above.

2.7.3 New addition of  $b^1\Sigma_g^+$  (v=3)- $X^3\Sigma_g^-$  (v=0) and  $b^1\Sigma_g^+$  (v=2)- $X^3\Sigma_g^-$  (v=1) bands and merging in the UV lines for  16  O₂

Atmospheric spectra can show where important bands are missing. Figure 18 shows a high-air mass ground-based Kitt Peak spectrum fitted using the HITRAN 2012 line list. It is clear that the  $b^{11}\Sigma_g^+$  (v=3)- $X^3\Sigma_g^-$  (v=0) band of O₂ at 0.58 µm is missing in the compilation. The line list for this band was calculated and while there were data to calculate line positions [258] no information was available for intensities and this atmospheric spectrum was used to scale the intensities to fit the observations). We also note that line positions used in HITRAN2016 for this band differ from those in Yu et al. [240]. This is due to the fact that the measurements in Ref. [258] were carried out using a relatively high pressure of oxygen. It is accounted for in the fit here but not completely in Ref. [240]. Broadening and shift parameters for this band were estimated using the approach described in Gordon et al. [259]. Figure 18b shows that the inclusion of this band makes the difference and it was included into HITRAN2016.

In a similar fashion it was discovered that it is important to include the  $b^1\Sigma_g^+$  (v=2)- $X^3\Sigma_g^-$  (v=1) band (basically a hot band underlying the *B*-band). Intensities for this band were also determined based on atmospheric spectra.

Finally, the UV bands of oxygen are not stored separately anymore and one can retrieve them from <a href="https://www.hitran.org">www.hitran.org</a> in the same manner as the other bands.

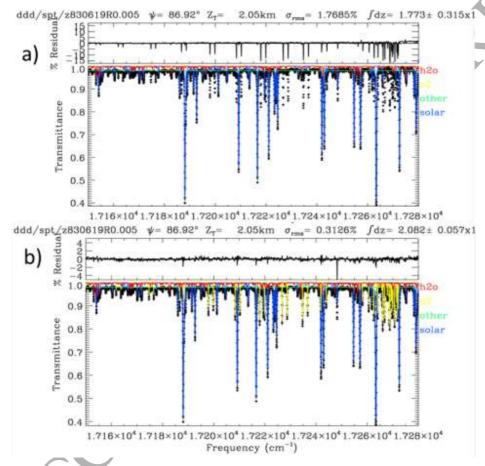


Figure 18. High-air mass ground-based Kitt Peak spectrum fitted without (a) and with (b)  $b^1\Sigma^+$  (v=3)- $X^3\Sigma_g^-$  (v=0) band of  $O_2$ .

#### 2.7.4 B-band

Recently Domysławska et al. [260] developed a dataset of the line-shape parameters for the B-band of self-broadened O₂ (the main isotopologue), based on accurate CRDS measurements

[261,262]. The dataset includes 49 lines from the P and R branches. It was observed that, at the considered level of signal-to-noise ratio, the spectra are well reproduced by the speed-dependent Voigt profile with quadratic speed dependence. In the near future we plan to incorporate these data into HITRAN within the formalism of the Hartman-Tran profile by setting the frequency of the velocity-changing collisions,  $V_{VC}$ , and the correlation parameter,  $\eta$ , to zero.

#### 2.8. *NO* (molecule 8)

Nitric oxide is an important constituent of the terrestrial atmosphere, where it is commonly detected and quantified by its fundamental IR band. The rotational spectrum plays an important role for the detection in diverse interstellar sources, but may also be used for detection in the Earth's upper atmosphere; Müller et al. [263] give a brief overview with appropriate references.

Positions and intensities for all stable NO isotopologues and for the first excited state in the pure rotation transition (1-1)of the principal isotopologue were based on Müller et al. [263] who performed a combined fit of all available rotational data together with heterodyne IR data of the fundamental band of the main species. All experimental data were carefully evaluated, a few poorly-fitting transition frequencies were omitted, and the uncertainties of some additional data were slightly adjusted as detailed in Ref. [263].

The  14 N 16 O line position and intensity data consist of millimeter and lower submillimeter data from Pickett et al. [264] and tunable far-IR data from Varberg et al. [265] in v = 0. The  $\Lambda$ -doubling transitions in v = 0 were mainly taken from Meerts and Dymanus [266]; additional data were taken from Refs. [267,268]. Further  $\Lambda$ -doubling transitions in v = 0 and 1 were taken from Dale et al. [269] and Lowe et al [270]. Also used in the fit were the v = 1 - 0 heterodyne IR data from Hinz et al. [271] and the very accurate data from Saupe et al. [272].

Millimeter and lower submillimeter data of  $^{15}N^{16}O$  were contributed by Saleck et al. [273], and tunable far-IR data by Varberg et al. [265]. The  $\Lambda$ -doubling transitions were taken from Meerts and Dymanus [266]. Millimeter and lower submillimeter data of the isotopologue  $^{14}N^{18}O$  were also published by Saleck et al. [273], and tunable far-IR data by Müller et al. [263]. Finally, millimeter data of  $^{14}N^{17}O$  and  $^{15}N^{18}O$  were taken from Saleck et al. [274].

A very accurate value of the ground-state electric dipole moment of  $^{14}N^{16}O$  was published in Ref. [268]; a value for v = 1 was determined in Ref. [275]. The strength of the magnetic dipole transition was derived from the electron spin-rotation parameter y employing Curl's relationship [276] to evaluate the g-factor of the respective isotopologue and in the appropriate vibrational state.

The data set has been extended considerably by FTIR data of  14 N 16 O reaching high v and by FTIR data of minor isotopic species from the MNRAS paper [277]. This will be used for future editions of HITRAN and HITEMP [278]. The spectroscopic model developed in this study has also been used to compute g-factors for the system [279].

#### 2.9. $SO_2$ (molecule 9)

Since the 2008 edition of HITRAN, the air-broadening coefficients of sulfur dioxide have been a constant,  $\gamma_{air}=0.1025$  cm⁻¹, for line positions lower than 3000 cm⁻¹; with  $\gamma_{air}=0.1$  cm⁻¹ for line positions larger than 3000 cm⁻¹ (see HITRAN2008 paper [198] for details). Recently, new measurements have been made of the air-broadening coefficients of SO₂ from infrared and submillimeter-wave spectra [280,281]. When analyzing these data together with previously available experiments [282–285], a slow decrease of the air-broadening coefficients with increasing ( $J''+0.2K_a''$ ) lower-state quantum number combination was observed (see the report

by Tan et al. [286] for details). The vibrational dependence of the broadening parameters is demonstrated to be very small or negligible.

A linear extrapolation (see Eq. (1)) up to  $(J''+0.2Ka'') \le 62$  (with error code 4 (10-20%)) was used, while for the transitions with (J''+0.2Ka'') > 62, the value corresponding to the experimental value with the largest (J''+0.2Ka'') was used.

$$\gamma_{\text{air}} = 0.10731 - 2.88311 \times 10^{-4} (J'' + 0.2K_a'')$$
 (1)

A complete line-by-line database of room temperature self-broadening coefficients for about  $1650 \text{ SO}_2$  transitions has been compiled by Tasinato et al. [287]. These data are used to update the self-broadening coefficients for the corresponding transitions. The error code has been set to 5, since we estimate the accuracy to be 5-6%. We also use the average values from Ref. [287] (with error code 4) of the same quantum numbers (J'',  $K_a''$ ) for all the other transitions with corresponding rotational quantum numbers. More details about this update of air and self-broadening parameters can be found in the report by Tan et al. [286].

Another update to the SO₂ data is the addition of pressure shifts, line widths and their temperature-dependence exponents in H2, He and CO2 environments for every HITRAN transition. This update described in detail by Wilzewski et al. [4]. While shifts of the SO₂ lines due to H₂, He or CO₂ were set to the default value of zero for every line because no reports of these parameters exist in available peer-reviewed sources, broadening their temperature exponents have been added based on Refs. parameters and [280,281,284,288,289]. Especially in the case of the SO₂-CO₂ system, the amount of available data used to construct the line list of broadening parameters is minimal, and future measurements of the SO₂-H₂, -He and -CO₂ systems would help to improve the present semi-empirical parameter sets.

There has been significant recent progress in the computation of extensive, accurate theoretical line lists for both room temperature [290,291] and hot [292] SO₂. The intensities in these line lists are accurate enough to provide alternatives for missing bands of both the main and the minor isotopologues; this should be considered in future updates. They can be supplemented with line positions calculated using constants from recent works from Ulenikov et al. [293–297] and references therein.

2.10. NO₂ (molecule 10)

Unchanged.

2.11. NH₃ (molecule 11)

NH₃ is a very important constituent of the atmospheres of giant planets and small, cold stars. It is also abundant in various environments of the interstellar medium and occurs in the terrestrial atmosphere. The isotopologue with ¹⁵N can be very important for studying isotopic fractionation or for studying the opacity of lines pertaining to ¹⁴NH₃.

Positions and intensities of the  $^{15}NH_3$  pure-tunneling and rotation-tunneling transitions were taken from the CDMS. It is based on rotation-tunneling transition frequencies from Belov et al. [298] and Winnewisser et al. [299], on pure tunneling data from Kukolich [300], in part corrected by Hougen [301], and from Sasada [302]. Also included are FIR data from Urban et al. [303] and Carlotti et al. [304]. Initial spectroscopic parameters were taken from a modified fit of  $^{14}NH_3$  data by Chen et al. [305]. Quantum chemical calculations and trial fits suggested that the C rotational constant cannot be determined reliably from these data. Therefore, its value was kept fixed to the  $^{14}NH_3$  value derived from Ref. [306] modified by the  $^{14}NH_3$ / $^{15}NH_3$  difference calculated from energies calculated by Huang et al. [306]. The dipole moment and its first order J and K distortion corrections were taken from Tanaka et al. [307]. The rotation-tunneling

transitions extend to J = 14 [304], and the pure tunneling transitions [302] even higher in J. The predictions should show FIR accuracy at least up to J = 14, but predictions higher than J = 15 should be viewed with increasing caution. Moreover, frequencies and possibly also intensities of the weak  $\Delta K = 3$  transitions should be viewed with some caution.

HITRAN2012 only contained ammonia transitions up to 7000 cm⁻¹. Recently, Barton *et al.* [308] analyzed Fourier Transform (FT) absorption spectra recorded at the National Solar Observatory by C. De Bergh in 1980, providing partially-assigned lists of measured line positions and intensities. The present update includes 8468 lines in the region 7400–8640 cm⁻¹ [308]; 2474 of these lines are fully assigned using the quantum numbers recommended by Down *et al.* [309]. Barton et al [310] have also analyzed and partially assigned a further spectrum covering the 8800 to 10400 cm⁻¹ region. These data were also included in HITRAN2016.

At lower wavenumbers, comparisons between the HITRAN data and comprehensive variational line lists [311,312] give good agreement below 4000 cm⁻¹. However, comparisons with these line lists suggest that the current HITRAN compilation is missing a significant number of lines in the 4000 – 7000 cm⁻¹ region. Attempts to resolve this problem using a combination of available variational line lists referred to as BYTe [311,312] and empirical energy levels provided using the MARVEL procedure [313] to give a hybrid line list referred to as "BARVEL" proved to be inferior to HITRAN2012 when compared with cross sections from PNNL [314]. It would appear that further work is needed on both the computed line list and the MARVEL dataset in this region. Work in both these directions is currently being undertaken and, in particular, a new high-accuracy *ab initio* potential energy surface [315] should provide a much better starting point for improved nuclear motion calculations.

Pressure-broadening coefficients and their temperature dependence exponent were calculated from the polynomial coefficients of Nemtchinov et al. [316] derived from the  $v_2$  band measurements. Nemtchinov et al. suggested that the rotational dependence of the air- and self-broadening half width,  $\gamma_{air}$  and  $\gamma_{self}$ , can be represented using the following polynomial:

$$\gamma^{0}(m,K) = \beta_{0} + \beta_{1}m + \beta_{2}K + \beta_{3}m^{2} + \beta_{4}K^{2} + \beta_{5}mK$$
 (2)

where m=|-J,J,J+1| for the P, Q and R branches, and  $\beta_i$  are the polynomial coefficients. This polynomial was used only for  $J \le 8$  with error code 6 for both  $\gamma_{\rm air}$  and  $\gamma_{\rm self}$ , and for the other transitions with , J>8 the extrapolated value corresponding to the largest J and K was used. For instance, when J>8 the default value,  $\gamma_{\rm air}=0.0906$  cm⁻¹/atm and  $\gamma_{\rm self}=0.5$  cm⁻¹/atm has been used for all these transitions, and the error code has been set to 5 for both  $\gamma_{\rm air}$  and  $\gamma_{\rm self}$ .

The temperature dependence exponents have also been measured in the  $v_2$  band by Nemtchinov et al. [316] but for broadening by  $O_2$  and  $N_2$  separately. We note that in the literature many researchers determine the temperature exponent to be

$$n_{\text{air}} = 0.79 n_{\text{N2}} + 0.21 n_{\text{O2}}$$
 (3)

by analogy with the standard approximation

$$\gamma_{\text{air}} = 0.79 \gamma_{\text{N2}} + 0.21 \gamma_{\text{O2}}$$
 (4)

We note however that Eq. (3) and Eq. (4) contradict each other because the variation of the air-broadening coefficients with temperature T is given by:

$$\frac{\gamma_{air}(T)}{\gamma_{air}(T_{ref})} = \left(\frac{T_{ref}}{T}\right)^{n_{air}} \tag{5}$$

Here  $T_{ref}$  is the reference temperature. Then we have:

$$n_{air} = -\frac{\ln \gamma_{air}(T_{ref}) - \ln \gamma_{air}(T)}{\ln T_{ref} - \ln T} , \qquad (6)$$

where  $\gamma_{air}$  can be defined using Eq (4). Therefore the measured half widths induced by  $O_2$  and  $N_2$  at 200, 255 and 296K can be determined from the values of  $n_{air}$  as the slope of the least-squares fits of  $-\ln \gamma_{air}^0(T)$  vs  $\ln T$ , as shown in Figure 19.

The broadening algorithm described above as well as the temperature dependence procedure is also used throughout the entire database for both isotopologues  $^{14}NH_3$  and  $^{15}NH_3$ .

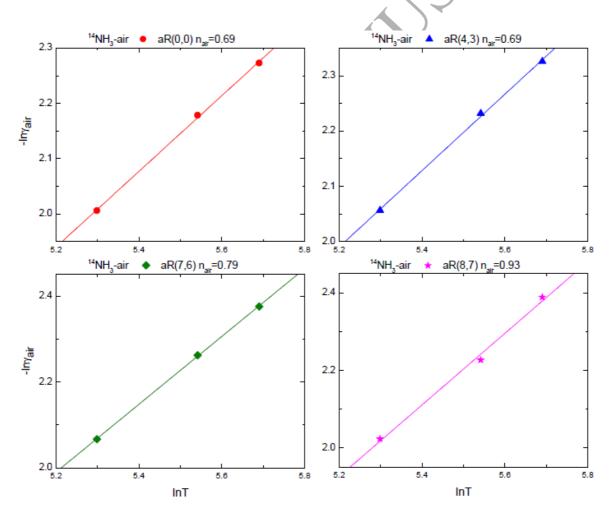


Figure 19. Air-broadening line widths of NH₃ with different temperatures for aR(0,0), aR(4,3), aR(7,6) and aR(8,7) transitions. The temperature-dependence exponent  $n_{air}$  is determined from the slope of the least-squares fits of  $-\ln \gamma_{air}^0(T)$  VS.  $\ln T$ .

Semi-empirical line widths, pressure shifts and temperature-dependence exponents of ammonia perturbed by H₂, He and CO₂ have been added to the database for the first time based on the algorithm described in Wilzewski et al. [4]. This algorithm makes use of the results from Refs. [316–320] (for H₂), Refs. [317,319–325] (for He) and Refs. [321,326,327] (for CO₂). Even though the new parameters were constructed from a relatively large number of data-reporting articles, the database would profit from future studies of NH₃ absorption lines perturbed by H₂, He and CO₂ for validation purposes as described in Ref. [4].

#### 2.12. HNO₃ (molecule 12)

For nitric acid (HNO₃), the three strongest absorption band systems, namely the  $\{v_5, 2v_9\}$ ,  $\{v_3, v_4\}$  interacting bands, and the  $v_2$  band, are located at 11, 7.6 and 5.8 µm respectively. Although two times weaker than the bands observed in the two other ranges, the 11-µm band system is of particular atmospheric interest since it coincides with a rather clear atmospheric window. As the HITRAN2012 database [1] provides a reliable description of the nitric acid spectrum at 11 µm (the line list includes the  $v_5$  and  $2v_9$  cold bands for the  $H^{14}N^{16}O_3$  and  $H^{15}N^{16}O_3$  isotopic species), as well as the  $v_5+v_9-v_9$ ,  $3v_9-v_9$ ,  $v_5+v_7-v_7$  and  $v_5+v_6-v_6$  hot bands of  $H^{14}N^{16}O_3$ ) this region is used for HNO₃ retrievals by numerous satellite or balloon-borne instruments [328–333].

However, retrievals of nitric acid at altitudes higher than ~35 km, where it is less abundant, require use of the stronger infrared signatures at 5.8 or 7.6 µm. Because the first spectral region overlaps rather strongly with water vapor absorption in atmospheric spectra, the 7.6-µm band system is preferable. However, most satellite or balloon-borne instruments avoided using this

latter region up to now for HNO₃ retrievals because of the poor quality of the available HNO₃ line positions and line intensities.

For HNO₃, the 7.6- $\mu$ m spectral region in HITRAN2012 is limited to the  $\nu_3$  and  $\nu_4$  cold bands of the main isotopologue, H¹⁴N¹⁶O₃. In addition, the quality of the line positions and intensities available therein is rather poor, mainly because the theoretical model used at that time [334] accounted only for the interactions coupling energy levels belonging to the  $\nu_3$  and  $\nu_4$  bright states but was neglecting interactions with several dark states present in the same energy range.

Therefore it was decided to update the HNO₃ line list at 7.6  $\mu$ m. First, a complete reinvestigation of the line positions for nitric acid at 7.6  $\mu$ m was performed [335]. In contrast to the previous analysis [334], the new Hamiltonian model accounted properly for the various vibration-rotation resonances and torsional effects affecting the  $v_3$  and  $v_4$  bright states and the four dark states  $2v_6$ ,  $3v_9$ ,  $v_5+v_9$  and  $v_7+v_8$ . Additionally, the  $v_3+v_9-v_9$  hot band was identified for the first time [335].

At the end of this first study [335], a new list of line positions and of preliminary line intensities was generated. This list included contribution from the  $v_3$  and  $v_4$  cold bands, the  $2v_6$ ,  $3v_9$ ,  $v_5+v_9$ ,  $v_7+v_8$  dark bands, and the  $v_3+v_9-v_9$  hot band of the main isotopologue. The final line intensities at 7.6  $\mu$ m were generated through a calibration process during which the HNO₃ volume mixing ratio profiles retrieved from the "Michelson Interferometer for Passive Atmospheric Sounding" (MIPAS) limb emission radiances in the 11- and 7.6- $\mu$ m domains were compared. This line list at 7.6  $\mu$ m, which was also validated using the available laboratory information (individual line intensities and absorption cross sections), proved to provide an improved description of the 7.6- $\mu$ m region absorption of nitric acid [336].

Table 8 compares the HNO₃ line lists at 7.6 μm given in the 2012 and 2016 releases of HITRAN showing significant changes. Finally, Fig. 20 compares the observed spectrum in the 7.6-μm spectral region (laboratory spectrum recorded in Giessen) to synthetic spectra generated using HITRAN2012 and HITRAN2016 showing the improvements brought by the new data as shown by the residuals. In conclusion, it is expected that this study should help to improve HNO₃ satellite retrievals by allowing measurements to be performed simultaneously in the 11- and 7.6-μm spectral regions.

In the HITRAN2008 paper, the several-item roadmap for the improvements to the nitric acid spectroscopy was presented. A majority of these problems have now been solved. However, the biggest challenge still remains, the lack of a line-by-line list for the v₁ band of HNO₃ centered at 3551 cm⁻¹. It is a pronounced feature that interferes with other gases absorbing at these wavenumbers. Figure 5 in the paper by Toon et al. [147] shows how the lack of spectroscopic parameters negatively affects retrievals from the Mk IV balloon observations. Low-resolution spectra of this band exist in the PNNL database [314] but at a limited set of thermodynamic conditions, while *ab initio* calculations for this molecule [337,338] do not yet allow accurate atmospheric retrievals.

Table 8. Comparison of the  $HNO_3$  line parameters in the 7.6- $\mu$ m region in the 2012 and 2016 editions of HITRAN

#### (a) HITRAN2012

Band	N	$S_{tot} $ $(10^{-18})$	v _{min} (cm ⁻¹ )	$v_{max}$ $(cm^{-1})$	$S_{min} $ $(10^{-23})$	$S_{max} $ $(10^{-21})$
ν ₃	21308	25.37	1098.376	1387.849	1.037	31.33
$\nu_4$	19584	12.78	1229.867	1387.561	1.037	18.67
Sum		38.15				

#### (b) HITRAN2016

Band	N	$S_{tot}$	$v_{min}$	$v_{max}$	$S_{min}$	$S_{max}$	
		$(10^{-18})$			$(10^{-25})$	$(10^{-21})$	
$v_3$	16408	24.94	1252.010	1394.177	4.910	32.0	
$\nu_4$	18105	9.834	1238.929	1387.081	4.020	10.7	
$2v_6$	2451	0.1194	1243.465	1348.275	4.624	3.660	
$v_5 + v_9$	13817	0.7163	1246.929	1390.071	2.081	3.543	
$v_7+v_8$	11125	0.7615	1246.422	1395.679	2.314	5.017	
$3v_9$	13894	1.177	1233.107	1388.497	4.582	2.378	
Sum	37.55						
V ₃ +V ₉ -V ₉	12106	1.408	1271.050	1394.899	5.285	1.798	

Note: N is the number of lines,  $v_{min}$  and  $v_{max}$  (cm⁻¹) are the lower and upper limits,  $S_{min}$  and  $S_{max}$  are the smallest and largest line intensity (cm⁻¹/{molecule cm⁻²} at 296 K) and  $S_{tot}$  is the sum of the line intensities.

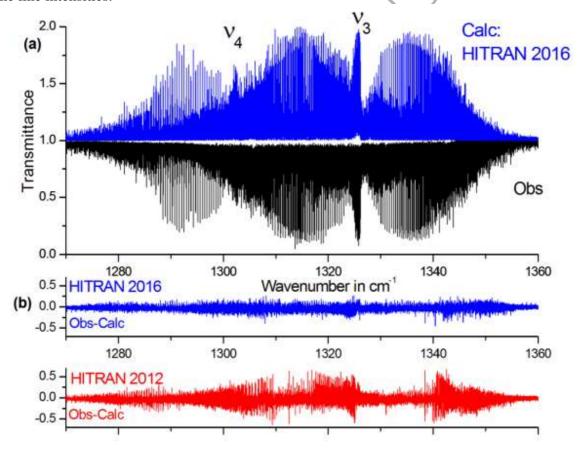


Figure 20. Overview of the Fourier transform spectrum recorded at high resolution in Giessen University in 2004 [335] and comparisons with the models performed using the HITRAN2012 and HITRAN2016 editions.

#### 2.13. *OH* (molecule 13)

The line list for the hydroxyl radical remained unchanged. The recent study by Brooke et al. [339] could be considered for updating intensities of the weak lines (this is more relevant for HITEMP [278]) in the near future.

#### 2.14. HF (molecule 14)

Transitions of HF with changes in vibrational quantum number,  $\Delta v$ , larger than 10 were removed from the database along with transitions with  $\Delta v > 8$  for DF. This removal was carried out because there were numerical issues in calculating the intensities of these transitions. Medvedev et al. [340] explain how the use of double precision in calculating overtone intensities may lead to numerical errors and that quadruple precision is needed. Additional numerical errors occur depending on the interpolation techniques used for the dipole moment functions as explained in Medvedev et al. [341].

Broadening, shifting and temperature-dependence parameters of HF due to the pressure of H₂, He and CO₂ have been added to HITRAN for the first time. The semi-empirical procedure described in Wilzewski et al. [4] was developed using results from Ref. [342] (for H₂), Refs. [343–345] (for CO₂) and Refs. [346–348] (for He). To expand and validate the present parameters, more measurements are needed while it would also be valuable to have more dedicated data for DF.

#### 2.15. HCl (molecule 15)

The intensity issue mentioned above for HF concerning the removal of high overtones of all isotopologues was likewise applied to HCl (i.e. transitions with  $\Delta v > 8$  were removed for all four isotopologues of hydrogen chloride).

Ref. [4] describes in detail the addition of H₂-, He- and CO₂-broadening, -shifting and temperature-dependence parameters of HCl to the database. Semi-empirical HCl-CO₂ broadening and shifting parameters for each HITRAN transition are based on Refs. [345,349] and sophisticated extrapolations of these results, while Refs. [350,351] were used to construct HCl-He linewidths and shifts. For the HCl-H₂ system, the new parameters are based on Refs. [352–355]. In the future, measurements of the temperature-dependence of foreign-broadened HCl spectral lines will be very helpful additions to the database.

2.16. HBr (molecule 16)

Unchanged.

2.17. HI (molecule 17)

Unchanged.

2.18. ClO (molecule 18)

Unchanged.

2.19. OCS (molecule 19)

Carbonyl sulfide is the most abundant sulfur-containing species in the free troposphere, often monitored using optical remote sensing in the 5-µm region [356]. OCS is present in the atmosphere of Venus [357]. The increasing interest for the application of very sensitive cavity ring down spectroscopy to the *in situ* detection of trace species [358] makes the near infrared range more and more attractive. However, the HITRAN2012 database provided spectroscopic information for OCS only up to 4200 cm⁻¹ [1]. Recently, Fourier transform spectra of carbonyl sulfide were recorded between 6170 and 6680 cm⁻¹ and between 7700 and 8160 cm⁻¹ using a femto/OPO laser absorption source and cavity enhanced spectroscopy [359]. The strongest bands

observed in these two regions are  $3v_1+2v_3$  (near 6640 cm⁻¹) and  $2v_1+3v_3$  (near 7812 cm⁻¹), respectively. The line positions thus measured (with an accuracy estimated to be 0.001 cm⁻¹ for not-too-weak unblended lines) were used to update a global polyad model of carbonyl sulfide (see Ref. [360] and references therein).

This global polyad model was then used to generate a list of line positions, assignments, lower state energies, and relative intensities for the main isotopologue. Because quantitative intensity information could not be retrieved from the measured spectra, the predicted relative intensities had to be scaled relying on additional information. A scaling factor was determined for each of the two spectral regions by matching calculated spectra of the  $3v_1+2v_3$  and  $2v_1+3v_3$  bands to their Fourier transform spectra, recorded previously [360]. Because predicted relative intensities only make sense for transitions involving levels belonging to the same polyads, these two scaling factors could only be applied to bands corresponding to transitions involving upper vibrational levels belonging to the same polyads as the upper levels of the  $3v_1+2v_3$  and  $2v_1+3v_3$  bands. Line intensities could therefore be obtained for cold bands matching this constraint, and hot bands involving variations of the vibrational quantum numbers identical to these cold bands (thus assuming that the vibrational dependence of the transition moments is negligible). Because isotopic substitution is not expected to significantly change the transition moments, line intensities were also obtained for the same bands belonging to \$^{16}O^{12}C^{34}S\$, \$^{16}O^{13}C^{32}S\$ and  $^{16}\mathrm{O}^{12}\mathrm{C}^{33}\mathrm{S}$ . All these predicted bands cover the ranges  $6484-6650~\mathrm{cm}^{-1}$  (region A) and 7728-7821 cm⁻¹ (region B). These two regions, involving a total of 3660 predicted lines, are presented in Figs. 13 and 14 of Ref. [359]. Note that only lines having intensities larger than  $3.0 \times 10^{-28}$ (region A) and  $6.7 \times 10^{-28}$  (region B) cm⁻¹/(molecule cm⁻²) at 296 K were considered. These intensity thresholds are slightly lower than the experimental limits. The supplementary material

of Ref. [359] also includes a list of line positions, assignments, and lower state energies for 3417 additional lines of  $^{16}O^{12}C^{32}S$ ,  $^{16}O^{12}C^{34}S$ ,  $^{16}O^{13}C^{32}S$  and  $^{16}O^{12}C^{33}S$ , covering the 6200 – 6449 cm⁻¹ and 7901 – 8120 cm⁻¹ ranges. Because of the constraint mentioned above, the relative intensities of these lines could not be scaled and were set to zero, and they were not included in HITRAN2016.

The spectroscopic information included in the new edition of HITRAN is the data of regions A and B obtained as described above, to which 10 lines belonging to the  $2v_1+2v_2^0+2v_3$  band of ¹⁶O¹³C³²S isotopologue (between 6650 and 6660 cm⁻¹) were added. The origin of this band is located near 6650.773 cm⁻¹. Relying on predictions of the global polyad model, which turned out to be as close as a few 10⁻³ cm⁻¹ to the observed positions, 17 P and R branch lines of the  $2v_1+2v_2^0+2v_3$  band of  ${}^{16}O^{13}C^{32}S$  could be observed, their positions measured in the spectra of Ref. [359] and included in the global modeling to yield improved predictions. These improved predictions are included in this edition of HITRAN. Note that the upper level of this band is the highest ever observed for ¹⁶O¹³C³²S. In the supplementary material of Ref. [359], 152 pairs of upper vibrational levels have identical assignments as a result of the combined effects of heavy mixing near avoided crossings and the assignments of the levels being based on the largest coefficient of their eigenvectors. To remove these duplicate assignments, one vibrational level in each pair was manually reassigned through careful examination of the corresponding energy level structures. Additionally, it was found that the line intensities provided in the supplementary material of Ref. [359] could be affected by the number of significant digits used when storing intermediate results. The line intensities provided in this edition of HITRAN correct this problem and are listed with the appropriate number of significant digits. The HITRAN line list for OCS

contains self- and air-broadening coefficients for all the lines calculated using the Padé approximant proposed of Koshelev and Tretyakov [361].

H₂-, He- and CO₂-broadened linewidths of OCS were added to HITRAN for the first time based on semi-empirical models derived in Wilzewski et al. [4]. The following experimental data were used to derive these models: Ref. [362] (for H₂), Refs. [362–365] (for He) and Ref. [366] (for CO₂). There are no lineshifts and temperature-dependence parameters due to H₂, He or CO₂ available in the literature.

 $2.20.\ H_2CO\ (molecule\ 20)$ 

Unchanged.

2.21. *HOCl* (molecule 21)

The  $v_2$  band region of hypochlorous acid, HOCI, was first introduced in HITRAN2004 [3] based on the line list from Vander Auwera et al. [367]. It was found that 49 lines of the principal isotopologue of HOCl in HITRAN had an incorrect vibrational assignment being attributed to the  $v_2$  band, while in reality they are  $2v_3$  lines. Also, the quantum rotational assignment of the line at 1237.62928 cm⁻¹ was incorrect. All of the assignments are now fixed, based on the correct information provided in the original publication [367].

 $2.22. N_2$  (molecule 22)

It was found that the abundance of the ¹⁴N¹⁵N isotopologue was incorrectly calculated in the original release of HITRAN2012; this issue with the intensities has been fixed in the latest edition.

2.23. HCN (molecule 23)

Hydrogen cyanide is a trace atmospheric species that can result from biomass burning [368–370]; its presence in the troposphere is being routinely monitored [371,372]. The use of HCN spectra for diagnostic applications in medicine are also being investigated [373]. HCN, and its isomer HNC (which is not yet in HITRAN), are important astronomical species with, for example, the recent tentative detection of HCN in the super-Earth exoplanet 55 Cancri e [374].

The spectrum of  ${}^{1}H^{12}C^{14}N$  at wavenumbers below 3500 cm $^{-1}$  has been included in HITRAN for some time. Here we extend the data available to higher wavenumbers. These data were based on the semi-empirical line list generated by Barber et al. [375] as part of the ExoMol project [376]. For full line lists appropriate for higher temperatures and covering both HCN and HNC the reader should consult these sources directly.

HITRAN has been updated with  ${}^{1}\text{H}^{12}\text{C}^{14}\text{N}$  lines for wavenumbers above 3500 cm⁻¹ and limited to transitions which have intensity greater than  $10^{-29}$  cm⁻¹/(molecule·cm⁻²); these criteria give a total of 55,153 lines in that spectral region. Wavenumbers for the majority of these lines were taken from the empirical energy levels of Mellau [377,378]. However, transition intensities are ones computed by Harris *et al.* [379] using the *ab initio* dipole moment and potential energy surface of Van Mourik *et al.* [380]. The Harris *et al. c*alculations are not sufficiently accurate for transition frequencies but the evidence is that the transition intensities can be considered fairly reliable [381]. An overview of the HCN line list in HITRAN 2016 is given in Fig. 21

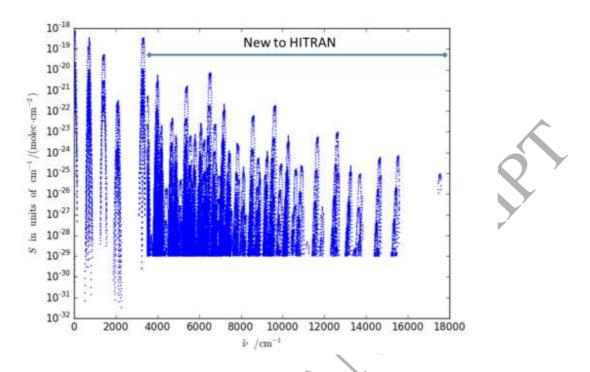


Figure 21. Overview of the transitions of HCN in HITRAN2016 with new bands revealed.

The air-broadening half widths were calculated using semi-empirical expressions derived in Yang et al. [382]. Self-broadening parameters were calculated using the procedure explained in the HITRAN2004 paper [3], see Eq. (20) there and discussion around it.

# 2.24. CH₃Cl (molecule 24)

The parameters of methyl chloride for 43147 lines (7079 of which are new), were calculated based on a recent analysis [383] Fourier transform spectra in the range of 1900-2600 cm⁻¹. The data from the supplementary file of Ref. [383] were used in the update. The error code 4 was given for line positions and the error code 5 (5-10%)- for line intensities, see Ref. [383] for details on uncertainties. We also recalculated the Einstein A coefficients for the lines which had changed intensities. In addition, a couple of issues that have been identified in the 3-μm region of CH₃Cl in HITRAN2012 were fixed. In particular, the air-broadening temperature

dependencies,  $n_{air}$ , of the lines around 3 µm used to be negative while they are supposed to be positive. And all of the lines in the 3-µm band have a shift of -0.02 cm⁻¹/atm. We compared the calculated spectra from the HITRAN database with the PNNL database [314] and the results show that such a strong shift of the lines appears to be incorrect. Therefore the shifts were set to be zero and  $n_{air}$  to be positive around 3 µm. In a similar fashion we removed the shift of -0.02 cm⁻¹/atm in the  $v_2+v_6$  band around 5 µm.

Both experimental measurements and calculations have been made of the self- and air-broadening coefficients of CH₃Cl recently, allowing for a relatively sophisticated update of these parameters in HITRAN. The available results have been collected and evaluated to update the database for CH₃Cl. The details are as follows:

For self-broadening coefficients of CH₃Cl:

1. The CH₃³⁵Cl self-broadening coefficients at reference temperature 296K for the R-branch (ΔK=0) were computed using a semi-empirical (SE) method described by Dudaryonok et al. [384]. We used these semi-empirical calculations for R-branch and P-branch (for all ΔK values) lines in all of the bands of the two isotopologues CH₃³⁵Cl and CH₃³⁷Cl considered in the database. The average value of RMS-deviations of the SE CH₃Cl self-broadening coefficients from the experimental data of Ref. [385] is 0.0408 cm⁻¹/atm. We used the ratio of the individual half widths to this value to estimate the uncertainties and determine a corresponding error code.

Only R-branch transitions for quantum number ranges of  $0 \le J \le 70$ ,  $0 \le K \le 20$  are available from Ref. [1]. The empirical relation enables evaluation of the P-branch and Q-branch broadening coefficients from those of the R-branch

$$\gamma_P(J,K) \approx \gamma_R(J-1,K) \approx \gamma_O(J,K)$$
 (7)

- 2. Experimental results collected from the literature were applied to corresponding transitions with the same J and K. These experimental values supersede the calculated ones
  - (a) Experimental results from Ref.[386] are used for the same isotopologue, (J',K'-J'',K'') transitions in all of the bands, with the error code set to 5 (5-10%).
  - (b) Experimental results from Ref.[385] are for both pure rotational (J= 6, 31, 37, 40, 45, 50) and  $v_1$  bands of CH₃Cl. The measured self-broadening coefficients for pure rotational transitions are used for the corresponding transitions in the database. Experimental results for the  $v_1$  band are used for transitions sharing the same quantum numbers throughout other bands.

### For air-broadening coefficients of CH₃Cl:

- 1. The air-broadening coefficients and associated temperature exponents for both  $CH_3^{35}Cl$  and  $CH_3^{37}Cl$  ro-vibrational transitions of R-branches ( $\Delta K$ =0) are calculated by a semi-classical (SC) approach [387]. The SC calculation are used for R-branch and P-branch (for all  $\Delta K$  values) lines in all of the bands assuming  $\gamma_P(J,K) \approx \gamma_R(J-1,K)$ .
- 2. Experimental results from Ref. [388] and Ref. [389] are used for the transitions sharing the same rotational quantum numbers.

For air-broadening temperature dependencies of CH₃Cl:

The air-broadening temperature exponents are updated using the SC calculations of Buldyreva [387]. Only the ro-vibrational transitions of R-branches and P-branches are calculated. The values for Q-branches are kept at the constant value of 0.70 for  $n_P(J,K) \approx n_R(J-1,K)$ .

2.25.  $H_2O_2$  (molecule 25)

For hydrogen peroxide,  $H_2O_2$ , the measurements of the air-broadening half widths from Refs. [390–392] were included for the corresponding transitions. The majority of the air-broadened coefficients in the database still have the same value of 0.1 cm⁻¹/atm estimated from the measurements of Malathy Devi et al. [392].

Currently the  $H_2O_2$  line list contained in HITRAN only covers wavelengths longer than 5  $\mu$ m. PNNL [314] and recent *ab initio* calculations [393,394] show strong features near 2.75  $\mu$ m associated with the OH stretch. More laboratory work is needed to obtain reliable line lists in that region.

#### 2.26. $C_2H_2$ (molecule 26)

New bands of acetylene from the wavenumber regions  $13 - 248 \text{ cm}^{-1}$  and  $390 - 634 \text{ cm}^{-1}$  were added using the results of Refs. [395–397]. The new data incorporated into HITRAN2016 are based mainly on the global modeling of both line positions and intensities performed in Tomsk within the framework of the method of effective operators [395]. The new measurements performed by Jacquemart et al. [397] considerably extend the available experimental data in the 390-634 cm⁻¹ region and allow an increase in the number of the fitted effective dipole moment (EDM) parameters responsible for the line intensities in the 13-634 cm⁻¹ region. This extended set of EDM parameters has better extrapolation abilities compared to that published in Ref. [395]. Using the eigenfunctions of the effective Hamiltonian from Ref. [395] and the new set of EDM parameters, a new list of line positions and intensities was generated. This list was used to update the  $C_2H_2$  line list in the  $13 - 248 \text{ cm}^{-1}$  and  $390 - 634 \text{ cm}^{-1}$  regions.

Global modeling of the rovibrational energies of the bending vibrational states has also been performed in the Jet Propulsion Laboratory (JPL) [396] using an EH for the bending modes. The

comparisons to the observed line positions showed that the JPL calculated line positions are slightly closer to the observations for the high values of the angular momentum quantum numbers than those derived from Ref. [395]. Thus it was decided to use JPL line positions (when available) for these bands.

The line list has been completed by other parameters (self- and air-broadening coefficients, temperature exponents of air-broadening coefficients, air-shift coefficients) as described in Section 4 of Jacquemart et al. [398].

Uncertainty codes 4 (0.0001 to 0.001 cm⁻¹) for the line positions and 6 (2-5%) for the line intensities were used in the case of the observed bands. For predicted bands, an uncertainty code 3 (0.001 to 0.01 cm⁻¹) for the line positions and 4 (10-20%) for the line intensities were adopted. For the JPL line positions, the uncertainty code 5 (0.00001 to 0.0001 cm⁻¹) was used.

The labeling  $V_1 V_2 V_3 V_4 \ell_4 \ell_5 J \varepsilon$  of the ro-vibrational states has been used in the list; see Jacquemart et al. [397] for details.

In addition, it was discovered that the Q-branch in the  $v_2+2v_4+v_5$  band near 2.58 µm was missing from the database. It is now provided based on the model of Lyulin and Perevalov [395]. In our effort to expand the database to spectral line parameters relevant to the study of planetary atmospheres, we have compiled semi-empirical parameter sets of  $C_2H_2$  broadened and shifted by the pressure of  $H_2$ , He and  $CO_2$  together with the corresponding temperature dependence of the linewidth. This update was described in detail in Wilzewski et al. [4]. These new parameters were constructed from Refs. [257,399–402] for  $H_2$ , Refs. [403,404] for He and Refs. [405,406] for  $CO_2$ .

Further studies of these line-shape parameters are important to improve our current update. This is especially true for the  $C_2H_2$ - $CO_2$  system, where almost no shift parameters and no temperature-dependence values have been reported in the literature to this point.

We note a recent study by Lyulin and Campargue [407] where empirical line lists for some of the NIR bands of acetylene are provided. These will be considered for future updates of HITRAN.

#### 2.27. $C_2H_6$ (molecule 27)

Ethane ( $C_2H_6$ ) is the second most abundant hydrocarbon after  $CH_4$  in the stratospheres of the outer planets (see for instance Ref. [408]) and Titan [409]. It therefore plays an important role as a tracer of atmospheric chemistry and dynamics.  $C_2H_6$  is a direct product of  $CH_4$  photochemistry and is therefore a key constituent in photochemical modeling. In addition to the  $v_9$  fundamental of ethane, which is the strongest band seen in Titan observed in the 10- $\mu$ m terrestrial window, the weak  $v_4$  torsional band at ~35  $\mu$ m (289 cm⁻¹) has been detected on Titan by Cassini/CIRS [410]. Although the torsional band is not infrared active in the lowest order and hence very weak in a laboratory setting, the features near 35  $\mu$ m are seen relatively strong in the Cassini/CIRS spectrum of Titan. This new development in the far-infrared enables astronomers to probe the lower altitudes of Titan's atmosphere.

$$2.27.1.^{12}C_2H_6$$

Since the last update in HITRAN2012 [1] for  $^{12}\text{C}_2\text{H}_6$  at 12 µm reported by Devi et al. [411,412] and at 7 µm reported by di Lauro et al. [413], significant progress has been made for ethane in the far- and mid-infrared regions. Moazzen-Ahmadi et al. [414] studied the features in the 35-µm region both in laboratory measurements and theoretical modeling, which included the torsional fundamental  $v_4$ , and the first torsional hot band,  $2v_4$ - $v_4$ , of  $^{12}\text{C}_2\text{H}_6$  and the Q-branch of the  $^{13}\text{CH}_3^{12}\text{CH}_3$   $v_6$  band. Multiple sets of pure ethane spectra were obtained at cold temperatures down to 166 K at the Jet Propulsion Laboratory using a multipass absorption cell with 52-m path length. Starting with the earlier global analyses of ethane considering inter-vibrational

interactions among the torsional levels of  $v_3$ ,  $v_9$ ,  $3v_4$  bands [415], the observed spectra were fit by adjusting independent dipole constants and the same value for self-broadening parameter for every transition. The observed spectra were reproduced to the residuals of 1 % (the reader is referred to Figure 1 in Ref. [415]).

It was found that the modeling of the transition intensities required an expansion of the dipole moment operator to higher order; this introduced Herman-Wallis like terms. The fine positions and line intensities of the torsional band ( $\nu_4$ ) of  $^{12}\text{C}_2\text{H}_6$  at 35 µm reported by Moazzen-Ahmadi et al. [414] have been adopted in this update. The uncertainties for line positions are set to no better than 0.0003 cm⁻¹ (HITRAN code = 4), while those of the line intensities are conservatively set to be 5 to 10% (HITRAN code = 5). For the pressure-broadened widths of the torsional band ( $\nu_4$ ) at 35 µm, it should be noted that self- and N₂-broadened widths are listed in this version, for which we adopted the values calculated by using the linear expressions for  $\nu_9$  transitions reported by Devi et al. [411,412], similar to what was done for C₂H₆ in the 7-µm region in HITRAN2008 [198]. The uncertainties for widths are conservatively set to 10 - 20% (HITRAN code = 4). The uncertainties for the temperature dependence exponents of the widths are thought to be 10 - 25% (HITRAN code = 3).

In the near future two minor isotopologues of ethane could be updated in the 12  $\mu$ m region; one is for the ¹³C singly-enriched  $v_{12}$  band of ethane (¹³CH₃¹²CH₃) at 12.2  $\mu$ m measured by Devi et al. [416] and the other for the singly-deuterated ethane (C₂H₅D) at 12.5  $\mu$ m studied by Daly et al. [417]. In particular, new studies were reported for ¹³CH₃¹²CH₃ in the  $v_{12}$  band in the 12  $\mu$ m region by Devi et al. [416]-and in the 7  $\mu$ m region by di Lauro et al. [418] for the  $v_4$ ,  $v_9$ ,  $v_{10}$  and  $v_6$  +  $v_{11}$  bands between 1345 and 1557 cm⁻¹. Their intensities were reported at 130 K with partially completed theoretical modeling.

#### 2.28. *PH*₃ (molecule 28)

Phosphine (PH₃) is a molecule of interest in astronomy, and its features have long been observed and identified in the spectra of Jupiter and Saturn (e.g. Refs. [419–422]). Because of its great importance in the Earth's and other planetary atmospheres including the interstellar medium, PH₃ has been the subject of a large number of experimental and theoretical investigations. Far infrared (FIR) line parameters of PH₃ were included for the first time in the HITRAN2012 database [1]. No updates have been made for any of the spectral line parameters for the infrared (IR) bands of phosphine since the HITRAN2008 database [198]. After the release of the HITRAN2012 database, there have been a few experimental and theoretical studies reported on phosphine [423-426]. Sousa-Silva et al. [423,424] generated a line list for phosphine; their room temperature line list consists of 137 million transitions among the 5.6 million energy levels for the ro-vibrational states up to  $J_{\text{max}} = 31$  and energies up to 8000 cm⁻¹. It was augmented by a more extensive and more accurate "hot" line list capable of simulating observed spectra from astronomical bodies at higher temperatures (e.g. 1500 K). In the near infrared region that is important for remote sensing of the outer planets, especially Jupiter and Saturn, the two most extensive and comprehensive experimental measurements of phosphine to date include: (1) the line positions and intensities for the five bands ( $2v_2$ ,  $v_2$ +  $v_4$ ,  $2v_4$ ,  $v_1$  and the  $v_3$ ) in the pentad region between 1950 and 2450 cm⁻¹ [426] and (2) the measurements of spectral line-shape parameters (e.g., pressure-broadened half widths, pressure shifts, and collisional line mixing coefficients) for the  $v_2 + v_4$ ,  $2v_4$ ,  $v_1$  and the  $v_3$  bands [425].

The measurements reported in Ref. [426] show an increase in the number of individual fitted intensities for the fundamental bands  $v_1$  and  $v_3$  from 20 and 50 reported in the previous laboratory measurements of Tarrago et al. [2.2.28.11] to 184 and 409, respectively. The actual

number of measured intensities in these bands were 337 and 829, respectively; however only the best selected lines were used in the determination of the pentad band strength. Positions and intensities for more than 4000 individual transitions in the pentad were retrieved from analyses of the FTS spectra recorded at the National Solar Observatory (NSO) on Kitt Peak, and the PNNL spectra [314]. The results published in Refs. [425,426] included 53 A+A- doublet components that are split by the Coriolis interaction between the  $v_1$  and  $v_3$  bands, and nearly 150 vibrationally-mixed perturbation-allowed transitions. The predicted spectrum from the analysis, however, did not reproduce the experimental spectrum fully because the precisions of the measurements were better than the calculations (for example see Fig. 9 from Ref. [426]). To improve the calculations, a global vibrational and intensity analysis combining the pentad measurements [425,426] with data in the dyad and the octad regions should be undertaken.

It appears that the SAITY line list from the recent *ab initio* calculations [424] is in much better agreement with Ref. [426] than with the HITRAN2012 data. Figure 2.28.1 shows a comparison of the SAITY line list [424] with the phosphine data from HITRAN2012 [1] at room temperature; also shown are the recent results of Malathy Devi et al. [426].

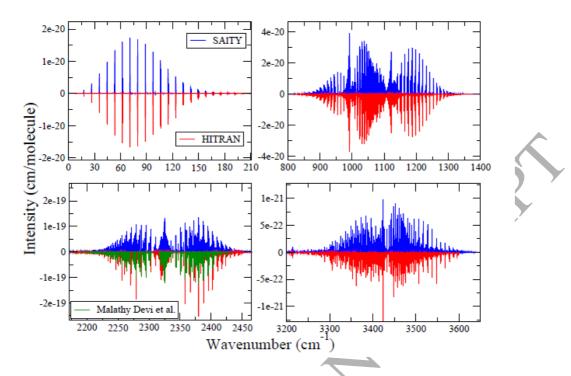


Figure 22. Comparison of the SAITY line list [424] with the phosphine data from HITRAN2012 at room temperature; also shown are the recent results of Malathy Devi et al. [426].

Recently, Nikitin et al. [427] have reported theoretical predictions of phosphine spectra based on *ab initio* dipole moment [428] and potential energy surfaces [429]. These calculations permitted a qualitatively correct agreement with observed data up to the octad range (2733-3660 cm⁻¹). An example of the comparison between these *ab initio* and experimental PNNL spectra [314] is given in Figure 23. The theoretical approach to these variational calculations is described in Rey et al. [430] and the corresponding line list and *ab initio* cross-sections in the range (0 - 7000 cm⁻¹) is available in the TheoReTS database [205].

The *ab initio* calculations described above (especially line positions) may not be of sufficient quality to model room-temperature high resolution spectrum to the HITRAN standard of accuracy. However, they are very helpful in assigning experimental spectra and invaluable in

high-temperature simulations and will be considered for inclusion in a future edition of the HITEMP database [278].

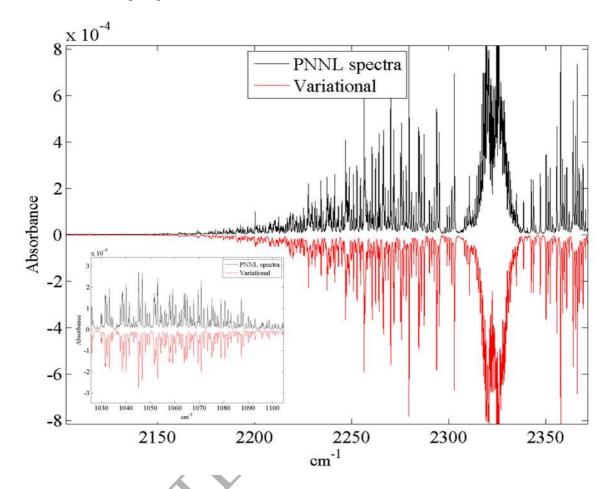


Figure 23. Comparison of experimental PNNL absorbance cross-sections [314] of PH₃ in the dyad and pentad ranges (in black) with variational calculations Rey et al. [205] (upside down in red).

The theoretical background and the set of programs used for both the energy and intensity fittings in the analysis of Malathy Devi et al. are described in Ref. [426]. Along with the positions and intensities, line-shape parameters were measured for a large number of transitions (over 840 Lorentz self-broadened half width coefficients, 620 self-shift coefficients, 185 speed dependence parameters, and collisional line mixing parameters for 10 A+A- doublet transitions) in the  $2v_4$ ,  $v_1$  and  $v_3$  bands and those measurements are reported in a separate article [425]. As

mentioned in the analysis of positions and intensities [426], the parameters in the pentad region are important for the exploration of chemistry and dynamics on Saturn using the existing Cassini/VIMS observations. The results reported in Refs. [425,426] (which have appropriate supplementary materials) are applicable to a variety of remote-sensing studies including the analyses of the near infrared data of Jupiter from the Juno mission and ESA's Jupiter Icy Moons Explorer (JUICE).

Creation of a line list of PH₃ in the pentad and octad regions is in progress and is expected to be available as an update to the HITRAN2016 database.

2.29. *COF*₂ (molecule 29)

Unchanged.

2.30. SF₆ (molecule 30)

Unchanged.

#### $2.31.\ H_2S\ (molecule\ 31)$

Line positions for 654 transitions in the (010)–(000) band were updated using experimental upper state energy levels for  $H_2^{32}S$ ,  $H_2^{33}S$ , and  $H_2^{34}S$  isotopologues reported in Ulenikov et al [431]. while the lower state energy levels were calculated using the rotational constants of Flaud et al. [432]. It appeared that the band center of this band was off by about 0.002 cm⁻¹ in HITRAN2012 while some higher-J rotational transitions were off by up to ~0.055 cm⁻¹.

In the triad region it was found that values for different parameters for some of the lines were not transferred correctly from Ref. [433] to HITRAN2000 [434]. This issue remained all the way through HITRAN2012, but has been fixed for HITRAN2016.

Intensity distribution of some of the multiplets in the NIR region was found to be inaccurate in the previous version in HITRAN for many lines where the experimental intensity for the entire multiplet were given to every component in the multiplet, rather than intensity being divided between components. This is fixed in HITRAN2016.

The air-broadening coefficients of hydrogen sulfide (H₂S) in the HITRAN database were set to a constant value,  $\gamma_{air}=0.0740~\rm cm^{-1}~atm^{-1}$  at 296 K, for most of the transitions (since HITRAN2004 [3]) based on the average value of available experimental data while just a few half widths were based on experimental values. For HITRAN2016, a thorough survey of measurements of both the air-broadening and self-broadening coefficients of H₂S has been made [435–438]. The report by Tan et al. [439] describes the results in more detail. In brief, a slow decrease of the air-(and self-) broadening coefficients with increasing ( $J^*+0.2K_a^{"}$ ) lower-state quantum number combination can be seen in Fig. 24. The vibrational dependence of the broadening parameters is neglected and we use the semi-empirical method based on the linear fit to the available experimental data to compute both air- and self-broadening coefficients of H₂S for all the isotopologues. This linear approaching value could only be used within the range probed by experiments. So for lower-state rotational quantum number  $J^*$ <12, the error code was set to 4 (between 10 and 20%) for both self- and air-broadening coefficients; for  $12 \le J^* \le 18$ , the error code was set to 3 (>20%); and for  $J^*$ >18, the value of both self- and air-broadening coefficients was set to the linear fit value corresponding to  $J^*$ =18, and the error code was 2 (estimate).

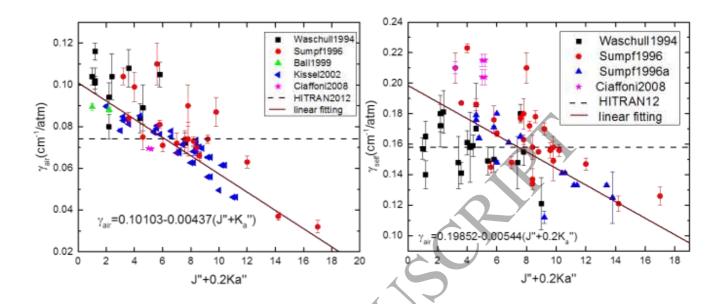


Figure 24. Left panel: Experimental air-broadening coefficients of H₂S lines. Waschull1994 stands for Ref. [435], Sumpf1996- Ref. [436], Ball1999- Ref. [440], Kissel2002-Ref. [437], Ciaffoni2008- Ref. [441]. Right panel: Experimental self-broadening coefficients. Sumpf1996s stands for Ref. [438].

### 2.32. *HCOOH* (molecule 32)

Unchanged.

### 2.33. HO₂ (molecule 33)

The air-broadening coefficients of the hydroperoxyl radical had not been modified since HITRAN2004 [3] in which they were set to a constant,  $\gamma_{air}$ =0.107 cm⁻¹/atm (based on the measurement in Ref. [442]). In the meantime, a number of different measurements of  $\gamma_{air}$  have been performed using infrared and submillimeter-wave spectra [442–447]. These measurements indicate a small vibrational dependence and a slow linear dependence with rotational quanta for  $\gamma_{air}$ .

The following approach was chosen for the air-broadening values in HITRAN2016. Experimental values were used wherever available for all the transitions with the corresponding rotational quantum numbers (in all the bands). For all other lines, a linear extrapolation with  $(J''+0.2K_a'') < 17$  was used:

$$\gamma_{\text{air}} = 0.1367 - 0.0027 \times (J'' + 0.2K_a'')$$
 (8)

For all transitions with  $(J''+0.2K_a'') > 17$ , the value corresponding to the experimental value with the largest (J''+0.2Ka'') was used. More details on the aforementioned algorithm are given in the report by Tan et al. [448].

For the self-broadening, a default estimate value of 0.3 cm⁻¹/atm was assigned to all transitions.

The values of line positions and lower-state energies for the O atom have been reverted to those given in the HITRAN1996-2000 editions [434,449]. They originate from Zink et al. [450]. The HITRAN2004 to 2012 editions used outdated values from the older version of the JPL catalogue [451] (downloaded in 2004), although the reference code mistakenly pointed to the work of Zink et al. The values have since been fixed in the JPL catalogue [451] (downloaded in 2015), and consequently we reverted back to the more accurate values of Zink et al [450].

# 2.35. ClONO₂ (molecule 35)

Unchanged.

### 2.36. *NO*⁺ (molecule 36)

The fundamental band of the NO⁺ molecule (in ground-electronic state  $^{1}\Sigma^{+}$ ) as well as its hot bands were introduced into the HITRAN1996 edition [449] and updated in 2008 [198]. As

pointed out by López-Puertas et al. [452] and references therein, the molecule is an important constituent of Earth's ionosphere. Ref. [452] actually used atmospheric observations to determine transition frequencies of its fundamental band and the associated hot band. The identification of NO⁺ in the terrestrial atmosphere by rotational spectroscopy should be possible. In addition, NO⁺ was recently identified tentatively in the interstellar medium [453].

The new edition of HITRAN adds the previously omitted pure-rotation transitions. The current prediction of the rotational spectrum corresponds to the recently created version 2 of the CDMS [454]. The data are based on a fit to rotational data [455] as well as ro-vibrational data [452,456]. The dipole moment of 0.380 D is from a quantum-chemical calculation [457].

The ground state rotational data extend to the J=8-7 transition near 32 cm⁻¹. Therefore, predictions should be reliable to at least up to 70 cm⁻¹. The entry now also contains predictions of rotational transitions in v=1; these are more uncertain than those in v=0.

# 2.37. *HOBr* (molecule 37)

Unchanged.

# 2.38. $C_2H_4$ (molecule 38)

Ethylene,  $C_2H_4$ , is both a naturally occurring compound and a tropospheric pollutant on the Earth [458]. It is also present as a by-product of methane photochemistry in the atmosphere of outer solar system bodies including Saturn [459] and Titan [409]. Optical remote sensing of ethylene in planetary atmospheres often relies on the strongest band observed in the 10-μm spectral region, namely the  $ν_7$  fundamental of the main isotopologue [460]. The extensive work carried out in this range of the ro-vibration spectrum of  $^{12}C_2H_4$  was recently reviewed by Alkadrou *et al.* [461]. The data available in HITRAN2012 for the 10-μm region corresponds to

the infrared active  $v_{10}$ ,  $v_7$  and  $v_{12}$  bands of  $^{12}C_2H_4$ , and the infrared and Raman inactive  $v_4$  band of the same isotopologue, located near 826, 949, 1442 and 1026 cm⁻¹, respectively. The information for the  $v_{10}$ ,  $v_7$  and  $v_4$  bands dates back to the 2000 edition of HITRAN [434]. It was generated relying on the frequency analysis of Willaert *et al.* [462], Cauuet *et al.* [463], Legrand *et al.* [464] and Rusinek *et al.* [465], and intensities measured for 13 lines of the  $v_7$  band observed in the range 940 – 970 cm⁻¹ [466]. Spectroscopic information obtained for the  $v_{12}$  band by Rotger *et al.* [467] was added in the 2008 edition [198].

A global frequency and intensity analysis of the 10-µm  $v_{10}/v_7/v_4/v_{12}$  band system of  $^{12}C_2H_4$  was very recently carried out using the tensorial formalism developed in Dijon for  $X_2Y_4$  asymmetric-top molecules and including the Coriolis interactions affecting the upper vibrational levels  $10^1$ ,  $7^1$ ,  $4^1$  and  $12^1$  [461]. Relying on spectroscopic information available in the literature or specifically retrieved from high-resolution Fourier transform infrared spectra, a total of 10757 line positions and 1645 line intensities were assigned and fitted with global root mean square deviations of  $2.6 \times 10^{-4}$  cm⁻¹ and 2.5 %, respectively. Using the results of that work and information available in the literature, a HITRAN-formatted list of parameters for 65776 lines belonging to the  $v_{10}$ ,  $v_7$ ,  $v_4$  and  $v_{12}$  bands of  $^{12}C_2H_4$  and covering the 620 – 1525 cm⁻¹ range was generated [461]. Comparisons of experimental spectra with spectra calculated for the same physical conditions with this line list and HITRAN 2012 are presented in Figs. 5 to 7 of Ref. [461]. They show that the modeling provided by this line list is significantly better than HITRAN 2012 for the  $v_{10}$  band, comparable for the  $v_7$  band (and for the  $v_4$  band hidden beneath it), and slightly worse for the  $v_{12}$  bands.

In the line list of Ref. [461], the vibration-rotation levels are assigned using the traditional  $(J, K_a, K_c)$  labels, where J is the quantum number associated with the total angular momentum of the

molecule, and  $K_a$  and  $K_c$  are associated with its projections along the molecular symmetry axes corresponding to the limiting prolate and oblate symmetric top approximations. As the tensorial formalism characterizes these levels using (J, C,  $\alpha$ ) labels (C is the ro-vibrational symmetry in the  $D_{2h}$  group, and  $\alpha$  is a ranking number used to sort the energy levels within a (J, C) block), a "translation" had to be carried out when generating the line list. It was very recently found that  $K_a$  and  $K_c$  labels were incorrect for a number of heavily mixed rotational levels of the  $10^1$ ,  $7^1$  and  $4^1$  upper vibrational levels. A corrected line list was therefore generated and submitted for publication [468].

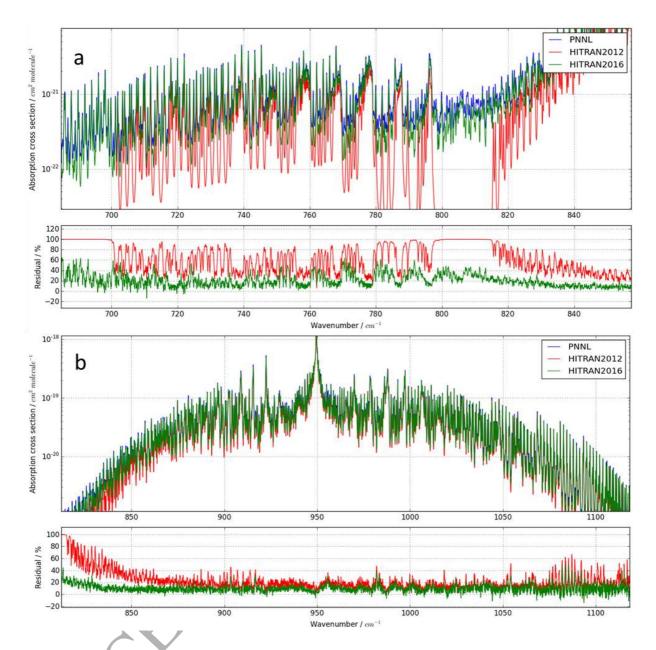
The spectroscopic information available in HITRAN for the 10-µm region of  $^{12}\text{C}_2\text{H}_4$  has been updated in the following way. Data for the  $v_{10}$ ,  $v_7$  and  $v_4$  bands were taken from the corrected line list of Alkadrou et al. [468]; however, lines having intensity smaller than  $10^{-30}$  cm⁻¹/(molecule cm⁻²) at 296 K were ignored. The line positions are given a HITRAN error code of 4 or 0 (0.0001-0.001 cm⁻¹ and larger than 1 cm⁻¹ respectively) depending on whether the corresponding J'' and  $K_a''$  values are or are not in the ranges included in the frequency analysis of Ref. [461]. Similarly, an error code of 5 (5-10%) characterizes the intensities of lines in the  $v_{10}$  and  $v_7$  bands corresponding to transitions involving J'' and  $K_a''$  values in the ranges included in the intensity analysis of Ref. [461]. The error code is set to 0 for all other line intensities. Note that errors in the degeneracies provided for the rotational levels in the  $10^1$  and  $7^1$  vibrational levels in the line lists of Refs. [461] and [468] are corrected here. Data for the  $v_{12}$  band was taken from HITRAN2012; however, 187 lines having an intensity smaller than  $10^{-30}$  cm⁻¹/(molecule cm⁻²) at 296 K have been excluded, to be consistent with the information provided for the three other bands. For all the lines, the self-broadening coefficients were calculated using the empirical expressions reported in Ref. [460], the air-broadening coefficients were generated as described in

[467], and the temperature dependence exponent n was set to 0.76, determined as described in Ref. [461].

The consistency of the C₂H₄ analyses that served for the updated line list has been confirmed by extensive variational calculations [469] based on *ab initio* potential energy [470] and dipole moment surfaces [471]. This sort of qualitative band shape validation was found to be particularly important for modeling complicated intensity perturbations involving "dark" states and for checking for possible contributions of hot bands. The corresponding variational methods are described by Rey et al. [469] and the results of *ab initio* based calculations including cold and hot bands in the range 0-6000 cm⁻¹ can be found in the TheoReTS database [205]. These calculations agree well with the new HITRAN2016 line list and experimental PNNL spectra [314] in the 10-μm region.

New bands of  $^{12}\text{C}^{13}\text{CH}_4$  are included in the new version of the database. More precisely, the new parameters which were derived from high-resolution Fourier transform spectra are those of the  $v_{10}$ ,  $v_8$ ,  $v_7$ ,  $v_4$ ,  $v_6$  bands, which absorb in the 615–1339 cm⁻¹ spectral region [472,473]. Their accuracy can be estimated to be of the order of  $0.6 \times 10^{-3}$  cm⁻¹ for the positions [473] and 4% for the intensities [472]. The broadening parameters were determined in the same fashion as for the principal isotopologue described above.

Figure 25 (generated using HAPI described in Section 6) shows that HITRAN2016 ethylene line list models experimental spectra from PNNL [314] much better than HITRAN2012.



**Figure 25**. HITRAN2016 vs HITRAN2012 when simulating PNNL experimental data in the a) 700-850 cm⁻¹, b) in the 850-1100 cm⁻¹ regions of ethylene spectra (1 atm., 296 K).

# 2.39. *CH*₃*OH* (molecule 39)

It was found that lower-state energies in the IR spectra of methanol introduced in HITRAN2004 [3] do not follow HITRAN formalism in that the lowest allowed energy level should be equal to zero. Yet the convention was observed in the MW region. The lowest energy

in the IR spectra was 128.1069 cm⁻¹ instead of zero. This has been fixed now to follow the usual convention.

2.40.  $CH_3Br$  (molecule 40)

Unchanged.

2.41. CH₃CN (molecule 41)

Methyl cyanide, also known as acetonitrile or cyanomethane, a trace species in the terrestrial atmosphere [474], has also been seen in comets [475] and in the atmosphere of Titan [476], and is an important species in various environments of the interstellar medium. Müller et al. {[3]} provided a brief overview as well as appropriate references. The  $v_4$  band of methyl cyanide was introduced into HITRAN2008 [198]. The data were based on a multispectrum analysis of this band and a preliminary modeling of the positions and intensities [477,478]. The  $v_4$  band is actually the third lowest wavelength band, with  $v_8$  and  $2v_8$  being at longer wavelengths. The prospects of detecting transitions of  $v_8$  or  $2v_8$  with the James-Webb Space Telescope or other instruments in space or in Earth's atmosphere are uncertain. The data are thought to be important for the excitation or de-excitation of the v = 0 transitions or of bands with much greater intensity, such as  $v_6$ .

The following update is planned in the immediate future, following the HITRAN2016 release. The current line position and intensity predictions for  $v_8$ ,  $2v_8$ , and  $2v_8 - v_8$  are from the CDMS [454]; the entries are actually combined CDMS and JPL entries. The predictions are based on a combined fit of rotational and ro-vibrational data involving states up to  $v_8 = 2$  that was published by Müller et al. [479]. The analysis takes into account interactions among these states as well as interactions of  $v_8 = 2$  with  $v_4 = 1$ ,  $v_7 = 1$ , and  $v_8 = 3$  and of the last three states among each other.

The  $v_8$  transition frequencies were taken from Koivusaari [480], and the  $2v_8$  transition frequencies were from Müller et al. [479]. Most of the  $v_8 = 1$  and 2 rotational data as well as some v = 0 data are also from Müller et al. [479]. Most of the additional ground-state data were published by Cazzoli and Puzzarini [481] and by Müller et al. [482]. Information of the K level structure in the ground vibrational state is based on Anttila et al. [483] and on the perturbations treated in Ref. [479]. Additional  $v_8 = 1$  and 2 data were published by Bauer et al. [484]. Parameters describing the three states around 10  $\mu$ m were taken mostly from the study of Tolonen at al. [485].

The transition dipole moments were evaluated by Müller at al. [479] as were line broadening and shifting parameters which were transferred from the  $v_4$  band [477]. The predictions should be reliable throughout, but slight caution is recommended for the highest K transitions of  $2v_8$  (K = 13 and 14).

Assignments for the  $v_4$  band are probably complete as far as the available spectral recordings are concerned. However, the preliminary analysis in Ref. [478], together with data from Ref. [479], affected the ground-state parameters substantially such that a new modeling effort will be necessary. The analysis should be completed well before the next HITRAN update, and further data may be available as well.

### 2.42. CF₄ (molecule 42)

Carbon tetrafluoride,  $CF_4$ , is a greenhouse gas with a huge global warming potential. Up to now, only the  $\nu_3$  fundamental band was present in HITRAN. However, this band represents only about 65 % of the absorption intensity at room temperature due to the presence of hot bands.

A new global fit of all CF₄ assigned lines involving the  $v_2$ ,  $v_3$  and  $v_4$  modes, as well as a separate one consisting of the  $v_1$  mode have been performed, including some new experimental data which was recently recorded in the Groupe de Spectrométrie Moléculaire et Atmosphérique, Université de Reims-Champagne-Ardenne. This study [486] will be published soon. It has resulted in new reliable effective Hamiltonian parameters for  12 CF₄, which, in turn, allow an improved calculation of the  $v_4$  and  $v_3$  fundamental bands and, also and for the first time, of the  $v_3+v_2-v_2$  hot band which represents about 17 % of the absorption intensity in this region at room temperature. The dipole moment parameters have also been corrected. This will be explained in detail in Ref. [486]. Thus, the present update completely replaces the previous data and includes the  $v_4$ ,  $v_3$  and  $v_3+v_2-v_2$  bands.

Figure 26 shows a comparison between an experimental spectrum from the Rennes group [487] and the calculation constituting the HITRAN2016 update.



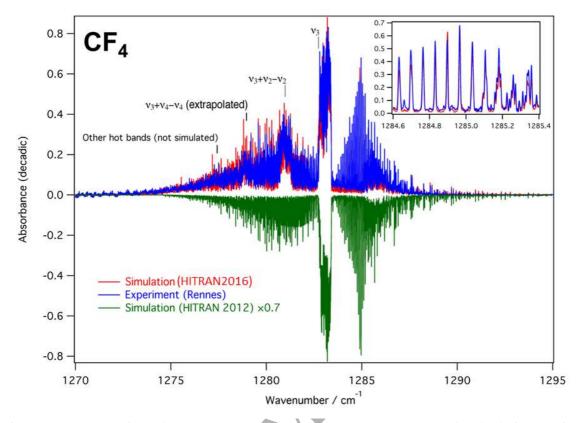


Figure 26. Comparison between an experimental spectrum [487] and calculations using the previous HITRAN2012 line list (inverted and scaled) and that constituting the HITRAN2016 update.

# $2.43.\ C_4H_2\ (molecule\ 43)$

The  $v_9$  band region (around 220 cm⁻¹) and the  $v_8$  band region (around 650 cm⁻¹) of diacetylene,  $C_4H_2$ , is important in interpreting Titan [488] observations. The  $v_9$  band region has been completely replaced with a new line list derived from a combined analysis presented in Refs. [488,489]. The updated line list provides new intensity for the lines already existing in the database but also includes numerous hot bands. The  $v_8$  band region was updated by scaling the intensities of all the lines in that region by a factor of 0.8 based on the recommendation given in Jolly et al. [489]. Finally, the  $v_6+v_8$  region (around 1200 cm⁻¹) was introduced into HITRAN for the first time based on the predictions from the global analyses. This band is important in

astrophysics as it has been observed in the atmosphere of Titan and the preplanetary nebula SMP LMC 11 with Spitzer Infrared Spectrograph [490].

 $2.44.\ HC_3N$  (molecule 44)

Unchanged.

# 2.45. $H_2$ (molecule 45)

Molecular hydrogen (H₂) is the most abundant gas in the atmosphere of gaseous giants. It was first introduced in HITRAN2012 [1]. The line list covered the quadrupole-allowed transitions for the H₂ isotopologue and dipole-allowed transitions for HD in their ground electronic states.

An update was made for singly-deuterated molecular hydrogen (HD) shortly after the release of the HITRAN2012 database. It was discovered by J. Mendrok (Lulea University of Technology, Kiruna, Sweden) that the abundance of the HD molecule was incorrectly calculated in the original release of HITRAN2012; this issue has now been fixed. The implication was that the intensities of the electric dipole transitions originally reported needed to be multiplied by a factor of two, although it did not affect the Einstein-A coefficients.

In addition, a total of 7195 electric quadrupole transitions of the HD molecule have been calculated and added to the HITRAN line list. The calculation was carried out using the energy levels from the work by Pachucki and Komasa [491] and the quadrupole moment function of Wolniewicz et al. [492].

In HITRAN2016, 527 transitions of H₂ that correspond to high vibrational overtones were removed. This was because there were numerical issues in calculating the intensities of these

transitions. Refs. [340,341] explain how the use of double precision in calculating overtone intensities may lead to numerical errors and that quadruple precision is needed.

Only a default value of the broadening parameter for the Voigt profile was given (the same for all the lines) in HITRAN2012. In HITRAN2016, a complete dataset of the line-shape parameters for the Hartman-Tran profile (HTP) is provided for self-perturbed H₂[6]. This dataset was determined from the analysis of H₂ experimental spectra. According to the new recommendation adopted in HITRAN, these line-shape parameters are stored separately for four different temperature ranges, see Ref. [6]. For the lines and conditions for which experimental spectra are not available, the line-shape parameters were extrapolated. In the near future we plan to extend HITRAN with the HTP line-shape parameters for H₂ perturbed by helium, which will be based on experimentally validated *ab initio* calculations [493].

2.46. CS (molecule 46)

Unchanged.

2.47. SO₃ (molecule 47)

Sulfur trioxide (SO₃) occurs naturally in volcanic emissions and hot springs [494]. It is also a pollutant emitted by smoke-stacks and other industrial exhausts [495]. In the terrestrial atmosphere, SO₃ rapidly forms sulfuric acid with its association with acid rain. SO₃ is a constituent in the atmosphere of Venus [496]. Recently, Zhang et al [497] have indicated that measurements of the abundances of SO₃ in the lower thermosphere of Venus can answer important questions related to sulfur chemistry. The spectroscopic study of sulfur oxides can also provide insight into the history of the Earth's atmosphere [498] and hence modeling of Earth-like planets.

SO₃ was included for the first time in the 2012 HITRAN edition. This entry was based on the experimental data of Maki and co-workers [499–502] with relative transition intensities scaled by the absolute values computed *ab initio* by Underwood et al. [503]. Also included were 25 pure-rotation transitions observed by Meyer et al. [504] for which the *ab initio* intensities of Underwood et al. [503] were used.

SO₃ is a planar, non-polar molecule but with sufficient rotational excitation that the molecule can distort and undergo pure-rotational transitions. A hybrid, empirical *ab initio* rotational line list for  $^{32}S^{16}O_3$  has been constructed for this edition of the database. For this, the ground-state rotational constants and centrifugal distortion parameters were taken from the fits to their infrared spectrum given by Ref. [502]; these were used to generate a set of vibrational ground-state rotational energies for *J* up to 85. To ensure that the extrapolation to high-*J* states is reasonable, a comparison was made between these energy levels and those calculated variationally by Ref. [505]; very good agreement was found with a root-mean-square difference of only 0.0167 cm⁻¹ for all rotational transitions involving states with *J* up to 85. Intensities (and Einstein-A coefficients) were determined using the variational calculations and an *ab initio* dipole moment surface [503]. The result is a list of 3414 pure-rotational transitions for  $^{32}S^{16}O_3$  which have been added to the 25 microwave transitions previously included. No measurements or calculations for the line-shape parameters are available; therefore the usual default values were chosen.

Recently Underwood et al. [506] presented a combined theoretical study of the spectrum of hot  $^{32}S^{16}O_3$ . This paper provided an extensive new line list with 21 billion lines for hot  $^{32}S^{16}O_3$  but no new line measurements. However, comparisons between this line list and the infrared absorption measurements suggest that the calculations overestimate the infrared line intensities

by about 30%. Interestingly, this difference appears to be rather uniform. Some direct measurement of the  $^{32}S^{16}O_3$  infrared line intensities would be welcome to inform future updates of the database.

2.48.  $C_2N_2$  (molecule 48)

Cyanogen is present in the atmosphere of Titan [507]. Its line list is now included into HITRAN for the first time based on the work of Fayt et al. [508]. The line list is for the  $v_5$  band system around 234 cm⁻¹.

### 2.49. COCl₂ (molecule 49)

Phosgene is a long-lived highly toxic synthetic gas that is widely used in pharmaceutical and polymer industries. Phosgene is a serious health hazard and was, in fact, used as a chemical weapon during World War I [509]. In addition, it has a strong greenhouse potential [510]. This molecule was detected in the atmosphere using solar occultation spectrometry [511], and using ACE-FTS [510] and MIPAS [512] retrievals.

Phosgene is a heavy molecule with small rotational constants and with a number of low-lying vibrational states. As a consequence, in order to perform a detailed and extensive analysis of its absorption region it is necessary to record spectra at low temperature. The analysis of the 11.75
µm spectral region of phosgene (which is now in HITRAN2016) was performed using a highresolution Fourier transform spectrum recorded at 169 K [513]. The  $v_5$  bands of the two
isotopologues  $CO^{35}Cl_2$  and  $CO^{35}Cl^{37}Cl$  were assigned up to very high quantum numbers and the
corresponding upper-state ro-vibrational levels were fitted to within the experimental accuracy
(~0.0002 cm⁻¹) [513]. On the other hand it was not possible to assign the corresponding hot
bands, which were modeled using extrapolated Hamiltonian constants. As a consequence, their

line positions are not accurate. As far as the line intensities are concerned, they were calculated [512] using single transition moments calibrated using the cross-section measurements from Sharpe et al. [314]. Their accuracy is estimated to be on the order of 5%.

### 3. Absorption cross-sections

Apart from the line-by-line spectroscopic absorption parameters discussed in Section 2, the HITRAN database contains information on absorption cross-sections where the line-by-line parameters are absent or incomplete. Usually this takes place for heavy polyatomic molecules (with low-lying vibrational modes) which are difficult for detailed analysis due to the high density of the spectral bands/lines, broadening effects, isomerization, and overall modeling complexity. Knowledge of the spectroscopy of these molecules is important for many purposes, including atmospheric remote sensing, radiative forcing calculations, detection of biomass burning, detection of toxic industrial chemicals (TICs), climate change monitoring, modeling planetary atmospheres, spectral calibration, etc. In fact these cross-sections have been applied in radiative-transfer codes in good approximations to the full-blown simulations that are possible with the line-by-line parameters [514–518]. It is therefore important to provide these cross-sections in HITRAN covering wide spectral ranges and at diverse thermodynamic conditions.

The cross-section files are provided in the HITRAN format described in Ref. [449] and are further described on the official HITRAN website (http://hitran.org/docs/cross-sections-definitions/). Each cross-section is given in a separate file with a header containing molecule name, chemical formula, wavenumber range, temperature, pressure, broadener, resolution of the experiment, and source. The file is completed with the value of each cross-section at equal intervals within the range, given in units of cm²molecule⁻¹. The cross-sections come from high-resolution laboratory observations, predominantly from Fourier transform spectrometers. Some

of these measurements include small negative values, which for the HITRAN-formatted crosssections were substituted by zeroes; in these cases we provide all the original cross-sections in a supplementary "Alternate" folder, because the negative values provide the user with some information about the noise level in the measurements.

The HITRAN2012 edition of the database contained cross-sections for about fifty gases. For the HITRAN2016 edition we have increased the amount of gases almost six fold by adding new data in the infrared spectral range. A dedicated paper detailing this update will be published separately. Here we only provide a brief summary and overview tables. Around 220 molecules of atmospheric, industrial, and remote sensing interests were taken from the quantitative spectroscopy PNNL (Pacific Northwest National Laboratory) database, described in Sharpe et al. [314]. This compilation was updated with 55 molecules from more recent work of Johnson et al. [519]. Twelve of these 55 molcules are new to PNNL. Spectra for these molecules will be indispensable, particularly for biomass burning detection. The PNNL database contains calibration-quality composite spectra given at atmospheric pressure and three temperatures (5°C, 25°C, and 50°C). All these cross-sections were measured using the FT-IR technique with 0.112 cm⁻¹ spectral resolution on a Bruker IFS 66v/S spectrometer.

In addition, a large portion of new molecules of atmospheric interest was taken from the 2013 review by Hodnebrog et al. [520]. The data presented in this review are more varied than PNNL considering spectral resolution, while containing all the strong spectral features important in radiative forcing calculations (on Earth). The subset of air and nitrogen broadened cross-sections from this review is for 700-750 Torr pressure range and room temperature. Some lower-temperature data for 253K from Highwood and Shine [521] were also included into HITRAN2016. Finally, more recently measured data were taken from Wagner and Birk for

BrONO₂ [522], Harrison et al for CFC-12 [523], HCFC-22 [524], CCl₄ [525], HFC-23 [526], HFC-134a [527], Reed and Hodges (2015) for ethane [528], as well as newly measured and revised data for many molecules from Refs. ([529–540]).

The summary on the new and updated compounds divided by categories is presented in Table 9. These categories are implemented in HITRAN*online* for the convenience of users to browse the data. This summary does not include those compounds that were not updated. For instance, there are no new UV cross-sections that were added; therefore molecules that had only UV cross-sections in HITRAN2012 are not included in this table.

Table 9. Categorization for the molecules in the HITRAN2016 update

Category	N _{src} ^(a)	$N_{\text{mol}}^{b}$	N _{sets} ^{c)}
Halogenated Alcohols and Ethers	27	63	106
Hydrochlorofluorocarbons	12	17	211
Hydrofluorocarbons	23	32	329
Fully Fluorinated Species	17	21	348
Chlorocarbons and Hydrochlorocarbons	6	23	336
Chlorofluorocarbons	9	11	376
Bromocarbons, Hydrobromocarbons, and Halons	7	12	127
Hydrocarbons	8	32	341
Alcohols, ethers and other oxygenated hydrocarbons	8	77	423
Nitriles, amines and other nitrogenated hydrocarbons	7	25	158
Sulfur-containing species	9	13	157
Iodocarbons and hydroiodocarbons	1	3	8
Other molecules	2	2	5

c) Number of temperature-pressure sets

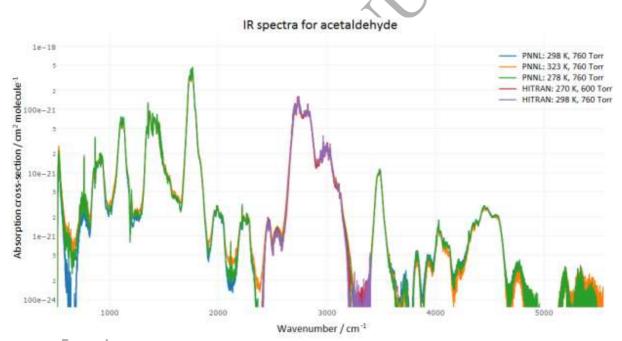


Figure 27. Update for cross-sections for acetaldehyde. Comparison of three temperature-pressure cross-section sets from PNNL [314] (update) with corresponding data from HITRAN2012 (Ref. [541]).

In Figure 27 we show an example of the update for the acetaldehyde molecule (CH₃CHO).

The new data were taken from the PNNL database [314], while the data from HITRAN2012

^{a)} Number of sources (references)

b) Number of molecules in the category. Note that some molecules can be present in more than one category

were originally from Tereszchuk and Bernath [541]. Sixteen cross-section temperature-pressure sets from Ref. [541] are presented in HITRAN2012 in the 2400-3400 cm⁻¹ spectral range. These cross-sections were calibrated using the PNNL data [541] and thus are in a good agreement with the update, in particular for absorption between  $6.0 \times 10^{-21}$  and  $1.0 \times 10^{-19}$  cm² molecule⁻¹. For this molecule, the new update extends the temperature range up to 323 K and the wavenumber range from 510 to 6500 cm⁻¹. The update includes stronger features in the mid-infrared which were absent in HITRAN2012. The data from Ref. [541] are also retained in the new HITRAN release.

In Table 10 the updates to the cross-sections of the molecules already existing in HITRAN2012 are presented. High-quality data were taken from different sources for the ranges which were either absent or covered insufficiently. Updates for molecules that were not previously in HITRAN are presented in Table 11. In both Table 10 and Table 11 we give total ranges for the wavenumber, pressure, temperature, and resolution. Each line in both of the tables summarizes all updated cross-sections for the given molecule, including those relegated to the alternate folder. If the broadener is not provided, self-broadening is assumed. Both tables contain total pressure of the sample including the buffer gas (where applicable).

Table 10. Summary for the updates for compounds already existing in HITRAN2012.

Molecule	CAS*	Wavenumber range (cm ⁻¹ )	T (K)	P (Torr)	Resolution (cm ⁻¹ )	Broadener	Reference
Acetylaldehyde	75-07-0	510-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Bromine nitrate	40423-14-1	765-1320	204-293	0-141	0.014	air	Wagner and Birk (2016) [522]
Carbon tetrachloride	56-23-5	700-860	207-297	7-760	0.01-0.03	Air	Harrison et al. JQSRT (2016) [525],
CFC-11	75-69-4	570-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
CFC-113	76-13-1	620-5000	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
CFC-114	76-14-2	600-5000	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
CFC-115	76-15-3	525-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
CFC-12	75-71-8	800-1270	189-295	7-761	0.01-0.03	Air	Harrison AMT (2015) [523]
CFC-13	75-72-9	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]*
Ethane	74-84-0	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]*, Reed et al (2015) [528]

Ethylene	74-85-1	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]*
Formaldehyde	50-00-0	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314] *
Formic acid	64-18-6	540-7200	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]*
HCFC-123	306-83-2	400-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*
HCFC-124	2837-89-0	400-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HCFC-141b	1717-00-6	500-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HCFC142b	75-68-3	500-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HCFC-21	75-43-4	600-7300	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HCFC-22	75-45-6	720-1380	190-295	7-762	0.01-0.03	Air	Harrison AMT (2016) [524]
HCFC-225ca	422-56-0	650-1400	296	700	0.5	air	Sihra et al. (2001) [542]
HCFC-225cb	507-55-1	700-1400	296	700	0.5	air	Sihra et al. (2001) [542]
Hexafluoro- ethane	76-16-4	400-6500	253-323	700-760	0.01-0.5	N ₂ , air	Highwood and Shine (2000) [521], Bravo et al. (2010b) [543], Sihra et al. (2001) [542]*, Sharpe et al. (2004) [314]
						1	Young et al. (2009) [544],
HFC-125	354-33-6	450-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HFC-134	359-35-3	350-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HFC-134a	811-97-2	750-1600	190-296	22-761	0.015	air	Harrison JQSRT (2015) [527]
HFC-143a	420-46-2	500-6500	253-323	700-760	0.03-0.5	N ₂ , air	Highwood and Shine (2000) [521]*, Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*, Le Bris and Graham. JQSRT (2015) [535]
HFC-152a	75-37-6	400-6500	253-323	700-760	0.112-0.5	N ₂ , air	Highwood and Shine (2000) [521], Sharpe et al. (2004) [314], Sihra et al. (2001) [542] *
HFC-32	75-10-5	400-6500	253-323	700-760	0.112-0.5	N ₂ , air	Highwood and Shine (2000) [521], Sharpe et al. (2004) [314], Gohar et al. (2004) [545],
Hydrogen peroxide	7722-84-1	510-7500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]*
Methylbromide	74-83-9	500-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314]*,
Phosphine	7803-51-2	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]*
Propane	74-98-6	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Tetrafluoro- methane	75-73-0	500-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Hurley et al. (2005) [546] *
Toluene	108-88-3	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Trifluoro- methylsulfur pentafluoride	373-80-8	100-6500	278-323	760	0.112-0.9	$N_2$	Nielsen et al. (2002) [547] *
m-Xylene	108-38-3	580-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
o-Xylene	95-47-6	600-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
p-Xylene	106-42-3	580-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]

[♦] Chemical Abstracts Service number

^{*} Data from these sources can be found in the alternate folder.

While the extent of the update of the cross-sectional part of the database is impressive, this section of the database is by no means complete. In the future we plan further revision of high-quality data that become available in the literature and by the facilitation of new experiments. In the near future we also plan to extend the database to include planetary-relevant cross-sections that recently became available, including those given in Refs. [548–552]. An extensive update of the UV cross-sections is also planned.

Table 11. Summary of the new compounds for which cross-sections were added in HITRAN.

Table 11. Sumn	Table 11. Summary of the new compounds for which cross-sections were added in HITRAN.							
Molecule	CAS	Wavenumber range, cm ⁻¹	T, K	P,Torr	Resolution, cm ⁻¹	Broadener	Reference	
(E)-1-Chloro-3,3,3- trifluoroprop-1-ene	102687-65-0	600-2000	295	700	0.25	air	Andersen et al. (2008) [553]	
(E)-HFC-1225ye	5595-10-8	650-2000	296	700	0.25	air	Hurley et al. (2007) [554]	
(Perfluoro-n-butyl)- ethylene	19430-93-4	650-1999	298	700	0.25	air	Andersen et al. (2012) [555]	
(perfluoro-n-octyl)- ethylene	21652-58-4	650-1999	298	700	0.25	air	Andersen et al. (2012) [555]	
(Z)-HFC-1225ye	5528-43-8	650-2000	296	700	0.25	air	Hurley et al. (2007) [554]	
(Z)-HFC-1234ze	29118-25-0	600-2000	296	700	0.25	air	Nilsson et al. (2009) [556]	
1,1,1,2-Tetrachloro- ethane	630-20-6	530-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,1,2,2-Tetrachloro ethane	79-34-5	530-6500	278-323	760	0.112	N ₂	Sharpe et al. (2004) [314]	
1,1,2,2-Tetrafluoro- 1,2- dimethoxyethane	73287-23-7	150-1900	296	700	0.25	air	Andersen et al. (2004) [557]	
1,1,2,2-Tetrafluoro-3- methoxy-propane	60598-17-6	450-3200	298		1.0		Oyaro et al. (2004) [558]	
1,1,2-Trichloro- ethane	79-00-5	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,1,2-Trifluoro-2- (trifluoromethoxy)- ethane	84011-06-3	440-3200	298	1	1.0		Oyaro et al. (2005) [559]	
1,1-Dichloroethane	75-34-3	560-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,2-Dichloropropane	78-87-5	590-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,2-Diclorobenzene	95-50-1	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,2-Difluoro-1,2- dichloroethane	431-06-1	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,3-Butadiene	106-99-0	530-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,3-Dichlorobenzene	541-73-1	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,3-Dichloropropane	142-28-9	570-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1,4-Dichlorobenzene	106-46-7	600-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1-Butene	25167-67-3	525-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1-Ethoxy-1,1,2,3,3,3- hexafluoropropane	380-34-7	400-3500	298		1.0		Oyaro et al. (2005) [559]	
1H,1H,2H,2H- Perfluorononan-1-ol	755-02-2	500-2250	296	700	0.5	air	Waterland et al. (2005) [560]	
1H,1H,2H,2H- Perfluoroundecan-1-øl	87017-97-8	600-2000	296	700	0.5	air	Waterland et al. (2005) [560]	
1-Nonene	124-11-8	550-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]	
1-Penten-3-ol	616-25-1	550-7300	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]	
2,2,2-trifluoro- acetaldehyde		400-2500	296-298	700	0.5-1.0	air	Sellevag et al. (2004) [561], Hashikawa et al. (2004) [562] *	
2,2,2-Trifluoroethanol	75-89-8	70-7250	278-323	760	0.112-1.0	$N_2$	Sharpe et al. (2004) [314], Sellevag et al. (2004b) [563]	
2,2,3,3,4,4,4- Heptafluorobutan-1-ol	375-01-9	400-4001	298		0.01	$N_2$	Sellevag et al. (2007) [533]*, Bravo et al. (2010) [564]	
2,2,3,3-Tetrafluoro- 1-propanol	76-37-9	400-4000	298				Sellevag et al. (2007) [533]	
2,2,3,4,4,4-Hexafluoro- 1-butanol	382-31-0	400-4000	298				Sellevag et al. (2007) [533]	
2,2-Difluoroethanol	359-13-7	70-4800	293		1.0		Sellevag et al. (2004) [563]	
2,3-Butanedione	431-03-8	580-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]	

2,3-Dimethylfuran	14920-89-9	510-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
2,5-Dimethylfuran	625-86-5	550-7000	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
2,6-Dimethoxyphenol	91-10-1	580-6500	323	760	0.112	$N_2$	Johnson et al. (2010) [519]
2-Carene	554-61-0	600-7300	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
2-Ethyltoluene	611-14-3	520-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
2-Fluoroethanol	371-62-0	80-7250	293-323	760	0.112-1.0	$N_2$	Sharpe et al. (2004) [314], Sellevag et al. (2004) [563]
2H, 3H-Perfluoro- pentane	138495-42-8	500-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
2-Hexanol	626-93-7	520-7250	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
2-Methylfuran	534-22-5	550-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
2-Nonanone	821-55-6	550-6500	298-323	760	0.112	N ₂	Johnson et al. (2010) [519]
2-Pentylfuran	3777-69-3	550-7300	298-323	760	0.112	N ₂	Johnson et al. (2010) [519]
2-Vinylpyridine	100-69-6	540-7300	298-323	760	0.112	N ₂	Johnson et al. (2010) [519]
3,3,3-Trifluoro- propanal	460-40-2	400-3500	298		1.0		Sellevag et al. (2004) [561]
3,3,4,4,5,5,6,6,7,7,7- Undecafluoro-heptan- 1-ol	185689-57-0	600-2000	296	700	0.5	air	Waterland et al. (2005) [560]
3,3-Demethyl-2- pentanol	19781-24-9	520-7250	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
3-Carene	13466-78-9	600-7300	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
3-Methoxyphenol	150-19-6	550-6500	323-333	760	0.112	$N_2$	Johnson et al. (2010) [519]
3-Methylfuran	930-27-8	550-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
3-Pentanol	584-02-1	570-7250	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
4-Ethyltoluene	622-96-8	600-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
4-Methyl-1-pentanol	626-89-1	520-7250	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
4-Methylvaleric acid	646-07-1	500-7300	323-338	760	0.112	$N_2$	Johnson et al. (2010) [519]
4-Penten-1-ol	821-09-0	520-7400	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
5-Nonanol	623-93-8	580-7200	298-323	760	0.112	$N_2$	1.1 (2010) [510]
Acetic acid	64-19-7	550-7100	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519] Sharpe et al. (2004) [314]
	) '						Sharpe et al. (2004) [314]
Acetol Acrolein	116-09-6 107-02-8	435-7200 540-6500	298-323 278-323	760 760	0.112 0.112	$N_2$ $N_2$	Johnson et al. (2010) [519] Sharpe et al. (2004) [314]
Acrylic acid	79-10-7	525-6500	323	760	0.112	$N_2$	
Acrylonitrile	107-13-1	550-6500		760	0.112		Johnson et al. (2010) [519] Sharpe et al. (2004) [314]
Allyl trifluoroacetate	383-67-5	600-2000	278-323 298	700	0.112	N ₂ air	Rodrigues et al. (2004) [538]
alpha-Pinene (1S)(-)	7785-26-4	600-6500	298-323	760	0.03	N ₂	Sharpe et al. (2004) [314]
Benzaldehyde	100-52-7	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314]
Benzyl alcohol	100-51-6	550-7400	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
beta-Pinene (1S) (-)	18172-67-3	600-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
bis(2-Chloroethyl) ether	111-44-4	550-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]

Bromochloro-							
methane	74-97-5	550-7100	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
bromoform	75-25-2	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
$C_2Cl_3H_3$	71-55-6	500-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
$C_4H_5F_3O_2$	383-63-1	500-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Carbon disulfide	75-15-0	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
CF ₃ CH ₂ CH ₂ OH	2240-88-2	400-4000	296-298	700	0.5	air	Waterland et al. (2005) [560], Sellevag et al. (2007) [533] *
CFC-112	76-12-0	550-6500	278-323	760	0.112-1.0	$N_2$	Sharpe et al. (2004) [314], Davis et al. (2016) [529]*
CFC-112a	76-11-9	550-6500	278-323	760	0.112-1.0	$N_2$	Sharpe et al. (2004) [314], Davis et al. (2016) [529]*
CFC-113a	354-58-5	550-1730	295-296	2.7	0.01-1.0		Davis et al. (2016) [529]*, Etminan et al. (2014) [530]
CFC-114a	374-07-2	550-1397	296		1.0		Davis et al. (2016) [529]
Chloroacetone	78-95-5	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Chlorobenzene	108-90-7	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Chloroethane	75-00-3	590-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Chloromethanol		657-2500	295	700	0.25	$N_2$	Wallington et al. (2000) [565]
Cineole	470-82-6	520-7300	323	760	0.112	N ₂	Johnson et al. (2010) [519]
cis-1,3-Dichloro- propene	10061-01-5	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Crotonaldehyde	123-73-9	500-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Diacetone alcohol	123-42-2	520-7000	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Dibromomethane	74-95-3	500-7100	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*
Dichloromethane	75-09-2	600-6500	278-323	700-760	0.112-0.25	N ₂ , air	Sharpe et al. (2004) [314],
Dichloromethanol		657-2500	295	700	0.25	$N_2$	Wallington et al. (2000) [565]
Diethylamine	109-89-7	590-7000	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Diiodomethane	75-11-6	530-7100	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Diisopropylamine	108-18-9	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Dimethyl disulfide	624-92-0	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Dimethyl ether	115-10-6	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Dimethyl sulfate	77-78-1	525-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Dimethyl sulfide	75-18-3	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Dimethyl sulfoxide	67-68-5	575-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Dimethylamine	124-40-3	550-6700	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
D-Limonene	5989-27-5	520-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
DL-Limonene	138-86-3	580-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
E-C10F18	60433-12-7	650-1500	296	700	0.5	air	Shine et al. (2005) [566]
EDB	106-93-4	570-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethyl acetate	141-78-6	565-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethyl acrylate	140-88-5	580-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethyl alcohol	64-17-5	600-7500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethyl benzoate	93-89-0	600-6500	323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Ethyl formate	109-94-4	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethylamine	75-04-7	600-7000	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethylenediamine	107-15-3	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Ethyliodide	75-03-6	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
EVE	109-92-2	570-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Fluoroether E-1	3330-15-2	450-1450	298		1.0	* 12	Oyaro et al. (2005) [559]
	2000 10 2	.50 1150					= , and et an (2005) [557]

Formic acid	64-18-6	540-7200	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Formic acid, dimer	14523-98-9	540-7200	278-298	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Furan	110-00-9	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Furfural	98-01-1	550-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Furfuryl alcohol	98-00-0	550-7300	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
•		330-7300				_	Sharpe et al. (2004) [314]
Geraniol	106-24-1	600-6500	323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Glycolaldehyde	141-46-8	570-6500	298-332	760	0.112	$N_2$	Johnson et al. (2010) [519]
Glyoxal	107-22-2	600-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Guaiacol	90-05-1	520-7250	298-323	760	0.112	$N_2$	
H-1 1201	1511 (2.2		206	700	0.5		Johnson et al. (2010) [519]
Halon-1201 Halon-1202	1511-62-2 75-61-6	500-1550 575-6500	296 278-323	700 760	0.5	air	Sihra et al. (2001) [542] Sharpe et al. (2004) [314]
riaioii-1202	/3-01-0	373-0300	216-323	700	0.112	N ₂ N ₂ ,	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314],
Halon-1211	353-59-3	500-6500	278-323	700-760	0.112-0.5	air	Sihra et al. (2001) [542]*
Halon-1301	75-63-8	510-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*
Halon-2311	151-67-7	600-1499	298		0.25	air	Andersen et al. (2012) [567]
Halon-2402	124-73-2	500-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*
HCFC-121	354-14-3	560-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
HCFC-132a	354-23-4	520-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
HCFC-133a	75-88-7	500-7100	278-323	7-760	0.01-1.0	$N_2$	Sharpe et al. (2004) [314], McGillen et al. (2015) [536]*, Etminan et al. (2014) [530]
HCFE-235ca2	13838-16-9	600-1450	298		0.25	air	Andersen et al. (2012) [567]
HCFE-235da2	26675-46-7	350-4000	296-298	700	0.25-1.0	air	Andersen et al. (2010) [568], Ryan and Nielsen (2010) [569]*, Sihra et al. (2001) [542]*
Heptafluoro- butyraldehyde	375-02-0	600-2500	296	700	0.5	air	Hashikawa et al. (2004) [562]
Hexachloro-1,3- butadiene	87-68-3	530-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Hexachlorocyclo- pentadiene	77-47-4	575-6500	323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Hexafluorobenzene	392-56-3	580-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Hexafluoropropene	116-15-4	100-6500	278-323	740-760	0.112-1.0	N ₂ , air	Sharpe et al. (2004) [314], Acerboni et al. (2001) [570]*
Hexyl acetate	142-92-7	580-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
HFC-1234yf	754-12-1	700-2000	296	700	0.25	air	Nielsen et al. (2007) [571]
HFC-1345zfc	374-27-6	650-1999	298	700	0.25	air	Andersen et al. (2012b) [555]
HFC-143	430-66-0	350-1550	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-152	624-72-6	700-3100	296	700	0.25	air	Wallington et al. (1994) [572]
HFC-161	353-36-6	350-1600	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-227ca	2252-84-8	400-1550	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-227ea	431-89-0	400-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Gohar et al. (2004) [545]
HFC-23	75-46-7	950-1500	187-295	23-763	0.015	air	Harrison JQSRT (2013) [526],
HFC-236cb	677-56-5	300-1550	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-236fa	690-39-1	350-1500	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-245cb	1814-88-6	400-1550	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-245fa	460-73-1	640-1500	296	700	0.5	air	Sihra et al. (2001) [542]

HFC-272ca	420-45-1	300-1500	296	700	0.5	air	Sihra et al. (2001) [542]
HFC-329p	375-17-7	500-4000	296	700	0.25	air	Young et al. (2009) [544]
HFC-365mfc	406-58-6	600-2000	296	700	0.25	air	Inoue et al. (2008) [573]
HFC-41	593-53-3	500-6500	278-323	700-760	0.112-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*
HFE-125	3822-68-2	400-1650	296	700	0.5	air	Sihra et al. (2001) [542]
HFE-143a	421-14-7	550-1550	296	700	0.5	air	Sihra et al. (2001) [542]
HFE-216	1187-93-5	850-1999	296	700	0.25	air	Mashino et al. (2000) [574]
HFE-227ea	2356-62-9	400-3200	298		1.0		Oyaro et al. (2005) [559]
HFE-236ea2	57041-67-5	450-3200	298		0.25-1.0	air	Andersen et al. (2010) [568]*, Oyaro et al. (2005) [559]
HFE-245fa2	1885-48-9	350-3200	296-298	700	0.5-1.0	air	Oyaro et al. (2005) [559]*, Sihra et al. (2001) [542]
HFE-263fb1	460-43-5	450-3300	296-298	700	0.25-1.0	air	Osterstrom et al. (2012) [575], Oyaro et al. (2005) [559]*
HFE-263m1	690-22-2	450-3200	298		1.0		Oyaro et al. (2005) [559]
HFE-329me3	428454-68-6	600-2000	296	700	0.25	air	Wallington et al. (2004) [576]
HFE-338mec3	56860-85-6	500-3500	296-298	700	0.25-1.0	air	Wallington et al. (2004) [576], Oyaro et al. (2005) [559]*
HFE-347mcc3	375-03-1	700-1400	298		0.03	$N_2$	Bravo et al. (2010a) [564]
HFE-347mmz1	28523-86-6	400-4000	298		0.25-1.0	air	Andersen et al. (2010) [568], Ryan and Nielsen (2010) [569]*
HFE-356mff2	333-36-8	475-3100	296-298	700	0.5-1.0	air	Oyaro et al. (2004) [558]*, Sihra et al. (2001) [542]
HFE-356mmz1	13171-18-1	450-3200	298		1.0		Oyaro et al. (2004) [558]
HFE-365mcf3	378-16-5	435-3150	298		1.0		Oyaro et al. (2004) [558]
HFE-449s1		350-1550	296-298	700	0.01-0.5	N ₂ , air	Sihra et al. (2001) [542]*, Bravo et al. (2010) [564]
HFE-7200		500-1600	296-298	700	0.01-0.5	N ₂ , air	Sihra et al. (2001) [542]*, Bravo et al. (2010) [564]
HFE-7300	132182-92-4	600-2000	298	1	1.0		Rodriguez et al. (2014) [539]
HFE-7500	297730-93-9	600-2000	298	,	1.0		Rodriguez et al. (2014) [539]
HFIP		530-3400	304-362				Godin et al. (this issue)
HFO-1243zf	677-21-4	500-1999	296-298	700	0.25-1.0	air, He	Andersen et al. (2012) [555], Gonzalez et al. (2015) [534]*
(E)-HFC-1234ze	1645-83-6	650-2000	296	700	0.25	air	Sondergaard et al. (2007) [577]
HFO-1438ezy(E)		600-1997	296		1.0	Не	Papadimitriou and Burkholder. (2016) [537]
HG-01	188690-78-0	25-2999	296		1.0	air	Myhre et al. (1999) [578]
HG'-02	485399-46-0	150-1800	296	700	0.25	air	Andersen et al. (2004) [557]
HG'-03	485399-48-2	150-1800	296	700	0.25	air	Andersen et al. (2004) [557]
HG-10	78522-47-1	25-3000	296		1.0	air	Myhre et al. (1999) [578]
HG-11		400-2000	296	700	0.25-0.5	air	Wallington et al. (2009) [579], Sihra et al. (2001) [542]*
Hydrazine	302-01-2	600-7000	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
i-HFE-7100	163702-08-7	300-1650	296	700	0.5	air	Sihra et al. (2001) [542]
i-HFE-7200	163702-06-5	400-1500	296	700	0.5	air	Sihra et al. (2001) [542]
Isobutane	75-28-5	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Isobutene	115-11-7	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Isobutyric acid	79-31-2	530-7250	340	760	0.112	$N_2$	Johnson et al. (2010) [519]
Isocyanic acid	75-13-8	500-7000	298	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Isopentane	78-78-4	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Isopentyl acetate	123-92-2	550-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]

Isophorone	78-59-1	600-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
isoprene	78-79-5	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Isopropylamine	75-31-0	600-6800	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Isovaleraldehyde	590-86-3	500-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Limonene oxide	1195-92-2	510-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
m-Cresol	108-39-4	570-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Menthol	89-78-1	580-6500	323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Methyl 2-methyl butyrate	868-57-5	520-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Methyl acetoacetate	105-45-3	520-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Methyl butyl ether	628-28-4	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl glyoxal	78-98-8	550-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Methyl iodide	74-88-4	500-6900	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl isoamyl ketone	110-12-3	550-6500	298-323	760	0.112	N ₂	Sharpe et al. (2004) [314]
Methyl isothio- cyanate	556-61-6	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl mercaptan	74-93-1	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl methacrylate	80-62-6	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl nitrite	624-91-9	520-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl pivalate	598-98-1	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl propionate	554-12-1	530-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl propyl ketone	107-87-9	520-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Methyl salicylate	119-36-8	550-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Methyl vinyl ether	107-25-5	550-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Methylethyl ketone	78-93-3	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Myrcene	123-35-3	520-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
N,N-Diethyl formamide	617-84-5	520-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314], Johnson et al. (2010) [519]
N,N-Diethylaniline	91-66-7	550-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Naphthalene	91-20-3	600-6500	323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
n-Butane	106-97-8	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
n-Butylamine	109-73-9	600-7000	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
n-C ₃ F ₇ OCH ₃	142 00 5	600-2000	295	700	0.25	$N_2$	Ninomiya et al. (2000) [580]
n-Heptane	142-82-5	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
n-Hexadecane	544-76-3	550-6500	323	760	0.112	N ₂	Johnson et al. (2010) [519]
n-Hexane	110-54-3	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
n-HFE-7100	163702-07-6	300-1520	296	700	0.5	air	Sihra et al. (2001) [542]
Nitrobenzene	98-95-3	550-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314],
Nitrogen trifluoride Nitrous acid	7783-54-2	400-6500	278-323	700-760	0.112-0.5	N ₂ , air	Robson et al. (2006) [581]*
	7782-77-6	550-6500	298	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Nonafluoropentanal	375-53-1	600-2500	296	700	0.5	air	Hashikawa et al. (2004) [562]
n-Pentadecane	629-62-9	570-6500	298-323	760	0.112	N ₂	Johnson et al. (2010) [519]
Octane	111-65-9	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Octanoic acid	124-07-2	550-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
o-Toluidine	95-53-4	550-7300	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]

Pentafluoro- propionaldehyde	422-06-0	600-2500	296	700	0.5	air	Hashikawa et al. (2004) [562]
Pentane	109-66-0	570-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
perfluoro-2-methyl-3- pentanone	756-13-8	450-2000	298		1.0	air	D'Anna et al. (2005) [582]
Perfluorobut-1-ene	357-26-6	600-2000	296	700	0.25	$N_2$	Young et al. (2009a) [583]
Perfluorobuta-1,3- diene	685-63-2	100-2600	293	740	1.0	air	Acerboni et al. (2001) [570]
D. Cl. J.	255 25 0	450, 6500	279 222	700 760	0.01.0.5	$N_2$ ,	Sharpe et al. (2004) [314],
Perfluorobutane	355-25-9	450-6500	278-323	700-760	0.01-0.5	air	Bravo et al. (2010b) [543]*
Perfluoropentane	678-26-2	500-6500	278-323	760	0.01-0.112	$N_2$	Sharpe et al. (2004) [314], Bravo et al. (2010b) [543]
PFC-1114	116-14-3	100-2600	293	740	1.0	air	Acerboni et al. (2001) [570]
PFC-218	76-19-7	500-6500	278-323	700-760	0.01-0.5	N ₂ , air	Sharpe et al. (2004) [314], Sihra et al. (2001) [542]*, Bravo et al. (2010b) [543]*
PFC-318	115-25-3	550-6500	253-323	760	0.112	N ₂	Highwood and Shine (2000) [521], Sharpe et al. (2004) [314]
PFC-51-14	355-42-0	700-1400	297		0.01	N ₂	Bravo et al. (2010) [543]
PFC-71-18	.20	700-1400	297		0.01	$N_2$	Bravo et al. (2010) [543]
PFC-91-18	306-94-5	0-2000	296	700	0.5	air	Shine et al. (2005) [566]
PFPMIE	1309353-34-1	600-1500	296	700	0.25	air	Young et al. (2006) [584]
PFPO	422-05-9	490-4000	298-362		0.1		Sellevag et al. (2007) [533]*, Godin et al. (2017) [532]
PFTBA	311-89-7	550-2000	298-344		0.1		Godin et al. (2017) [532] Godin et al. (2016) [531]
Phenol	108-95-2	550-7300	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Phosgene	75-44-5	535-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Piperidine	110-89-4	510-6700	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Propionaldehyde	123-38-6	520-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Propylene	108-32-7	600-6500	298-323	760	0.112	$N_2$	•
carbonate	75 55 0	600 6700	279 222	760	0.112	N	Johnson et al. (2010) [519]
Propylenimine R124A	75-55-8	600-6700 525-6500	278-323		0.112	$N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314]
R124A R132A	354-25-6 471-43-2	510-6500	278-323 278-323	760 760	0.112 0.112	$N_2$ $N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314]
R141	430-57-9	550-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314]
sec-Amylamine	625-30-9	540-6900	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314]
Styrene (monomer)	100-42-5	600-6500	298-323	760	0.112	$N_2$	Sharpe et al. (2004) [314] Sharpe et al. (2004) [314]
•			270 323	700	0.112	N ₂ ,	Sharpe et al. (2004) [314],
Sulfuryl fluoride	2699-79-8	500-6500	278-323	700-760	0.112-0.25	air	Andersen et al. (2009) [585]*
TE-6	25291-17-2	650-1999	298	700	0.25	air	Andersen et al. (2012) [555]
Tetrachloroethylene	127-18-4	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
trans-1,2-Dichloro- ethene	156-60-5	540-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
trans-1,3-Dichloro- propene	10061-02-6	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Trichloroethylene	79-01-6	595-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Trichloromethane	67-66-3	580-7200	253-323	700-760	0.112-0.25	N ₂ , air	Highwood and Shine (2000) [521], Sharpe et al. (2004) [314]
Trichloromethanol		657-2500	295	700	0.25	$N_2$	Wallington et al. (2000) [565]
Triethylamine	121-44-8	575-7100	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Trifluoroacetic acid	76-05-1	540-7050	298	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Trimethylamine	75-50-3	600-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]

Valeraldehyde	110-62-3	600-6500	278-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Valeric acid	109-52-4	520-7100	340	760	0.112	$N_2$	Johnson et al. (2010) [519]
Vinyl chloride	75-01-4	540-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Vinyl fluoride	75-02-5	500-6500	278-323	760	0.112	$N_2$	Sharpe et al. (2004) [314]
Vinyl toluene	100-80-1	550-6500	298-323	760	0.112	$N_2$	Johnson et al. (2010) [519]
Vinyl trifluoroacetate	433-28-3	600-2000	298		0.03	air	Rodrigues et al. (2016) [538]
Z-C10F18	60433-11-6	650-1500	296	700	0.5	air	Shine et al. (2005) [566]

^{*} Data from these sources can be found in the alternate folder.

### 4. Collision-Induced Absorption

Collision-induced absorption (CIA) was first introduced into HITRAN in the 2012 edition and was largely based on the data collected in Richard et al. [586]. This section of the database is accessible through (www.hitran.org/cia). CIA refers to absorption by transient electric dipoles induced by the interaction between colliding molecules. Due to the short life-time of collision complexes, this typically leads to broad absorption features underlying the much sharper lines due to absorption by isolated molecules. As CIA arises from binary collision complexes, the intensity scales with the number densities of both the absorber *and* the perturber, i.e., with the square of the pressure at constant volume mixing ratio. To be completely explicit, the transmittance at wavenumber v,  $\tau_{abs-pert}(v)$ , is related to the collision-induced cross section tabulated in the HITRAN CIA section,  $k_{abs-pert}(v)$ , through

$$-\ln[\tau_{\text{abs-pert}}(v)] = k_{\text{abs-pert}}(v) \,\rho_{\text{abs}} \,\rho_{\text{pert}} \,L, \tag{10}$$

where  $\rho_{abs}$  and  $\rho_{pert}$  are the number densities of the absorber and perturber, respectively, and L is the path length. The CIA cross sections tabulated in HITRAN are given in cm⁵ molecule⁻² throughout [586].

Often the relevant perturber will be air, i.e., a 21:79 mixture of O₂:N₂ (the amount of argon in the atmosphere coupled with its poor efficiency as a collisional partner makes this assumption

quite reasonable). In that case, one can either add absorber- $O_2$  and absorber- $N_2$  contributions, or equivalently use the absorber-Air files which are introduced in this version of HITRAN. In the latter case, the perturber number density equals the air number density given by  $\rho_{pert} = \rho_{Air} = \rho_{O2} + \rho_{N2}$ .

### 4.1. Molecular oxygen 1.27-µm band

The CIA data for the 1.27- $\mu$ m band (corresponding to  $X^3\Sigma_g^-$  (v = 0)  $\to a^4\Delta_g$  (v = 0)) of  $O_2$  have been revised in the current release of HITRAN. In this region, CIA is stronger whereas magnetic dipole lines are weaker when compared to the oxygen A-band. Hence, under atmospheric conditions, CIA contributes significantly to the absorption in the 1.27- $\mu$ m band.

All data included for this transition are taken from the experiment by Maté *et al.* [587], recorded using a Fourier transform spectrometer. Data recorded for pure O₂ can be found in the O₂–O₂ file. These data have been corrected for an error in HITRAN2012, which accidentally contained O₂–Air spectra. Data recorded for air mixtures of 21:79 O₂:N₂ can be found in the O₂–Air file, which is described above. In the previous version of HITRAN, data for such mixtures could be found in the O₂–N₂ file, but these have been moved to the O₂–Air file to stress that these also include O₂–O₂ contributions. One should not double count the O₂–O₂ CIA by including both O₂–O₂ and O₂–Air contributions in atmospheric models. This has replaced the data of Smith and Newnham for this band [588]. The data previously provided in this band for O₂:N₂ mixtures contained magnetic dipole lines due to O₂ monomer absorption. This issue has now been resolved, as the monomer lines have been subtracted from the present data.

### 4.2. Oxygen 1.06-μm band in air

The molecular oxygen  $X^3\Sigma_g^-$  (v = 0)  $\rightarrow a^1\Delta_g$  (v = 1) transition at 1.06  $\mu$ m has much weaker magnetic dipole lines than what is observed for the 1.27- $\mu$ m band, and hence the 1.06- $\mu$ m band is

dominated by collision-induced absorption. The previous version of HITRAN already contained spectra for this transition, for different  $O_2:N_2$  mixtures, measured by Smith and Newnham [588]. These spectra have been re-analyzed by fitting spectra for different mixtures to contributions of  $O_2-O_2$  and  $O_2-N_2$  collisional pairs. This is illustrated in Fig. 28, which shows two spectra, for 50:50 and 75:25 mixtures of  $O_2:N_2$ , both at T=230 K. These spectra can be reproduced essentially using only the  $O_2-O_2$  contribution as shown. The contribution of  $O_2-N_2$  is smaller than the error bars and noise levels. The spectra determined in this way have lower noise levels than direct measurements for air mixtures, where the absorption is weaker due to the lower  $O_2$  density. These data have been smoothed further and are now available in the  $O_2$ -Air file.

### 4.3. Oxygen A band in air

Collision-induced absorption in the oxygen A band is also of atmospheric interest. In spite of the smaller contribution of CIA in this band, relative to the much stronger magnetic dipole lines, CIA is significant for high-accuracy measurements including airmass calibration of the OCO-2 instruments [589]. The HITRAN CIA database already contained data for O₂–O₂ and O₂–N₂ from Tran *et al.* [253], valid for temperatures between 200 and 300K. For convenience, we now also provide the spectrum for O₂–Air, obtained by combining the available O₂–O₂ and O₂–N₂ data. In the alternate folder, we now also include O₂–Air data from Drouin *et al.* [252] obtained in a multispectrum fitting approach. Line-by-line parameters determined in the same multispectrum fit, are also available in HITRAN2016. Consistent use of both line-by-line and CIA data accurately reproduces the total absorption under atmospheric conditions, and has been shown to reduce systematic errors in atmospheric retrievals [252].

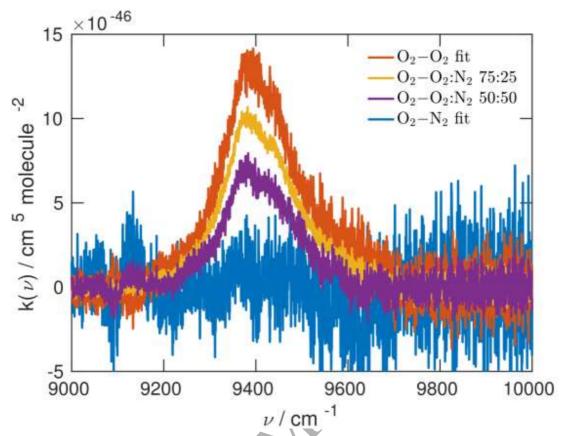


Figure 28. Raw experimental CIA data for the oxygen 1.06- $\mu$ m band at T=230 K for 50:50 and 75:25  $O_2:N_2$  mixtures, and the contributions of  $O_2-O_2$  and  $O_2-N_2$  determined from these data. For readability in gray scale, the legend is ordered vertically according to the peak intensity.

### 4.4. Planned update

This update represents a small revision of the HITRAN CIA section, and we aim to extend the CIA section with recently measured and calculated spectra in the near future. This update will include existing and new collisional pairs. We also plan to further update different spectral regions for molecular oxygen.

### 5. Aerosol refractive indices

Optical depths in the atmosphere are determined by the spatial and temporal distributions of gases, clouds, and aerosols [590]. Clouds and aerosols also take part in chemical reactions in

both the liquid and solid phases [591]. The interpretation of remote-sensing retrievals of gaseous species is limited frequently by how well one can separate gaseous opacity from that of clouds and aerosols. Knowledge of the refractive indices of the aerosols and cloud particles and their size distributions is necessary in order to specify their optical properties. HITRAN2016 contains refractive indices in the visible, infrared, and millimeter spectral ranges of many types of cloud and aerosol particles. Table 12 lists the HITRAN2016 indices.

Table 12. Refractive indices included in HITRAN2016.

Compound	Measurement Specifics	Reference
Water	27°C, 10–5000 cm ⁻¹	[592]
Water	0.67–2.5 μm	[593]
Ice	266 K, 0.04 μm – 2 m	[594]
Ice	0.67–2.5 μm	[593]
Water, ice, sodium chloride, sea salt, water soluble aerosol, ammonium sulfate, carbonaceous aerosol, volcanic dust, sulfuric acid, meteoric dust, quartz, hematite, sand	Room temperature, 0.2–40 μm	[595]
Sulfuric acid (H ₂ SO ₄ /H ₂ O)	Room temperature, 25–96% H ₂ SO ₄	[596]
Sulfuric acid (H ₂ SO ₄ /H ₂ O)	Room temperature, 75 and 90% H ₂ SO ₄	[597]
Sulfuric acid (H ₂ SO ₄ /H ₂ O)	215 K, 499–6996 cm ⁻¹	[598]
Sulfuric acid (H ₂ SO ₄ /H ₂ O)	$200-300 \text{ K}, 825-4700 \text{ cm}^{-1}$	[599]
Sulfuric acid (H ₂ SO ₄ /H ₂ O)	213–293 K, 432–5028 cm ⁻¹	[600]
Nitric acid (H ₂ SO ₄ /HNO ₃ )	Room temperature, 250–2987 cm ⁻¹	[601]
Nitric acid (H ₂ SO ₄ /HNO ₃ )	220 K, 754–4700 cm ⁻¹	[602]
Nitric acid (H ₂ SO ₄ /HNO ₃ )	213–293 K, 432–5028 cm ⁻¹	[600]
Amorphous nitric acid (NAM, NAD, NAT)	153 K, 482–7000 cm ⁻¹	[603]
NAM	179 K, 482–6002 cm ⁻¹	[603]
NAD	184 K, 482–6981 cm ⁻¹	[603]
NAD	160–190 K, 700–4700 cm ⁻¹	[604]
αΝΑΤ	181 K, 482–6989 cm ⁻¹	[603]
βΝΑΤ	196 K, 482–6364 cm ⁻¹	[603]
NAT	$160 \text{ K}, 711 - 4004 \text{ cm}^{-1}$	[605]
Burning vegetation	$525-5000 \text{ cm}^{-1}$	[606]

Burning vegetation	0.35 – 1.5 μm	[607]
Carbon flame	0.4 – 0.7 μm, 25 – 600° C	[608]
Flame soot	$0.2 - 38 \mu m$	[609]
Diesel soot	0.45 – 10 μm	[610]
Brown carbon	$0.2 - 1.2 \mu m$	[611]
Organic acids (Oxalic, malonic, succinic, pinonic, pyruvic, phthalic)	0.25 – 1.1 μm	[612]
Organic haze	0.525 nm	[613]
SOA (proxy)	0.525 nm	[614]
Minerals (clay, illite, kaolin, montmorillonite)	2.5 – 200 μm	[610]
Minerals (granite, montmorillonite)	$5-40 \mu m$	[615]
Saharan dust	0.30 – 0.95 μm	[616]
Saharan dust	$0.35 - 0.65 \mu m$	[617]
Volcanic ash	$0.45 - 25 \mu m$	[618]
SiO ₂ (amorphous)	6.6 - 487 m, 10 – 300 K	[619]
SiO ₂ (crystalline)	6.25 μm – 10 mm, 300 – 928 K	[620]
$Al_2O_3$	$7.8 - 200 \ \mu m$	[621]
FeO	$0.2 - 500 \mu m$	[622]
CaTiO ₃ (Perovskite)	2.0 – 500 μm	[623]
$Fe_2O_3$	$0.1 - 1000 \ \mu m$	[624]
Fe ₂ SiO ₄ (Fayalite)	$0.4 - 10 \; \mu m$	[625]
Fe ₂ SiO ₄ (Fayalite)	2 - 10  mm	[626]
MgAl ₂ O ₄ (annealed)	1.6 – 6825 μm	[627]
MgAl ₂ O ₄ (natural)	$2.0~\mu m - 10~mm$	[627]
Mg ₂ SiO ₄	$0.19 - 948 \mu m$	[628]
MgSiO ₃	0.2 - 500 μm	[628]
TiO ₂ (Rutile)	$0.47 - 36.2 \ \mu m$	[629]
TiO ₂ (Anatase)	$2.0 - 5843 \mu m$	[629]
TiO ₂ (Brookite)	$2.0 - 5843 \mu m$	[630]
KCl	0.22 – 166 μm	[630]
ZnS	0.22 – 166 μm	[630]
Titan Tholins	$0.02 - 920 \ \mu m$	[631]
Titan aerosol	$0.2-1~\mu m$	[632]

As discussed by Bohren and Huffman [633], the refractive index m is a function of wavelength, and has a real  $m_{real}$  and imaginary  $m_{imag}$  component:

$$m = m_{\text{real}} + i m_{\text{imag}} \tag{5.1}$$

A plane light wave of wavelength  $\lambda$  is attenuated along the propagation x axis according to

$$E = E_0 \exp(-2\pi m_{\text{imag}} x/\lambda) \exp(i2\pi m_{\text{real}} x/\lambda - i2\pi ct/\lambda), \tag{5.2}$$

with time t and the speed of light c. The imaginary refractive index  $m_{imag}$  determines the amount of light absorption in a medium, attenuating the light intensity by  $\exp(-4\pi m_{imag}x/\lambda)$  along a path of distance x.

New HITRAN2016 indices include refractive indices associated with exoplanet atmospheres. Indices (see Table 12) of materials which condense out at temperatures from 1725 K (e.g. SiO₂) to 700 K (e.g. ZnS) include most of the condensates tabulated by Wakeford and Sing [634], who calculated transmission spectral properties of clouds for hot-Jupiter exoplanets. Many of the indices have been kindly provided to HITRAN2016 by Harrald Mutschke of the Friedrich Schiller University Jena. Additional indices of materials (e.g. Mg spinels with a variety of impurities) not tabulated in Table 12 can be downloaded from the extensive Jena web site (http://www.astro.uni-jena.de/Laboratory/OCDB/index.html). An example of the new indices is presented in Figure 29, which displays the real and imaginary indices of ZnS and TiO₂, two possible exoplanet particle compositions.

# Querry ZnS Zeidler TiO2 (Rutile)

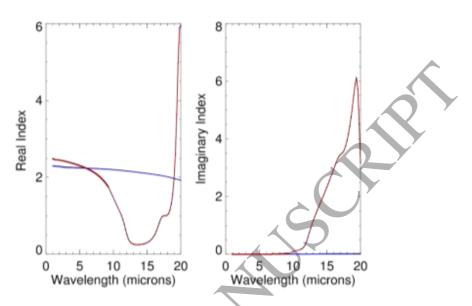


Figure 29. Comparison of the real and imaginary indices of ZnS and  $TiO_2$  from 1 to 20  $\mu m$ . Since particle extinction spectra in the infrared have a wavelength dependence which is similar to the wavelength dependence of the imaginary index, the extinction spectra of ZnS and  $TiO_2$  particles will be very different.

HITRAN2016 extends the HITRAN-RI program [635] that resides on the HITRAN website by including the exoplanet indices. HITRAN-RI is written in the IDL (Interactive Design Language) and Fortran 90 programming languages, and applies the Bohren-Huffman [633] Mie code. The user specifies in a user-edited ASCII file the particle composition (refractive index), wavelength range, and log-normal particle size distribution. The program then calculates extinction, scattering, absorption, single scattering albedo, asymmetry, and backscattering spectra of the cloud or aerosol particle. Provision is provided to generate output ASCII files of the inputs and spectra. The IDL version of HITRAN-RI provides output postscript files. Other special program features allow the user to compare two sets of refractive indices, and to calculate spectra for multiple-component aerosols. As an instructional aid, test cases can be run. PDF

versions of the original reference papers are contained in a subdirectory, while the refractive indices are stored in subdirectories in ASCII and NetCDF formats. The user can use the ASCII files to quickly look-up the real and imaginary indices at a specific wavelength, while the HITRAN-RI program accesses the netCDF files. HITRAN-RI allows the user to read in user-specified refractive indices. The source code of HITRAN-RI can also be modified and added to by the user.

#### 6. Global data and software

#### 6.1. Database structure and interface (www.hitran.org)

Since June 2015, users have been able to access the HITRAN line-by-line data not only by downloading them in the traditional ASCII 160-character (.par) format from the ftp site, but also through the easy-to-use interface HITRAN online, available at <a href="www.hitran.org">www.hitran.org</a>. Cross sections can also be accessed through this dynamic interface, while CIA and aerosols sections of the database are also accessible, though so far only as static files. This new tool (described in Hill et al. [2]) allows custom filtering and selection of molecules and their isotopologues in the wavenumber region of interest. The default output format is still .par, but users are given an opportunity to create their own, customized output formats, including the HDF5 format widely used by NASA scientists. The flexibility of the formats is achieved due to the fact that the database is cast into a relational format [236]: it is now a compilation of tables of parameters and metadata that are interconnected and searchable using Structured Query Language (SQL) (see Refs. [2] and [236] for details). This new flexible database format offers many advantages over the static fixed-width ASCII files. Indeed, we have now introduced non-Voigt line shapes [6], broadening parameters by perturbing gases other than air or self [4], and other parameters that are important for accurate atmospheric retrievals. Other advantages of the new structure include

allowing more significant figures (for instance, important for line positions of the MW transitions) and checks on data integrity. Further advantages include immediate access to the references for every parameter (including the automatic generation of bibliography in both HTML and BibTeX format), generation of interactive plots of retrieved line-by-line and cross section data (with full features such as zoom, etc.), and more-detailed filtering (such as by individual vibrational bands), etc.

HITRAN*online* has received a warm welcome from the community, with over 6200 researchers registered with it (as of the middle of June, 2017). The rate of new users signing up on <a href="https://www.hitran.org">www.hitran.org</a> has not dropped below 50 per week. Extensive documentation regarding the definition and use of HITRAN parameters is now directly accessible on the website.

There are a few new features that have been introduced to the HITRAN*online* interface since its original release. First, in order to make the selection of several dozen new special parameters less cumbersome, we "bundled" the parameters associated with selected line profiles. In the example shown in Figure 30 one can see how the Hartmann-Tran profile can be selected.

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Figure 30. Groups of related parameters such as those relating to the HT profile in different temperature regions are selected from a separate set of overlay panels in the HITRAN*online* interface. When one or more temperature groups are chosen, the three "Common parameters" are also automatically selected. Conversely, these common parameters are removed from the selection when no temperature group is chosen for this profile. Different broadening species are presented on separate panels, and may be selected from the labeled tabs at the top of the overlay.

## 6.2. HITRAN Application Programming Interface (HAPI)

The HITRAN Application Programming Interface (HAPI) is a free open source Python module (library) which provides a set of tools for working with structured spectroscopic data from different sources. It is described in detail in Ref. [5]. Here we give a brief overview of its features.

Originally created as an extension of the HITRAN *online* web interface [2] to allow users sophisticated calculations and manipulations with the HITRAN data on their computers, HAPI can handle user-supplied custom data in a flexible way and fully supports the HITRAN *online* data scheme and formats of the HITRAN [1] database.

The principal aim of HAPI is facilitating physically-sound interpretation of observations and more realistic models for a wide variety of applications such as astrophysics, planetary science, climate simulations, remote sensing, theoretical spectroscopy, and data mining. Having such tool is important in particular to prevent possible errors in radiative-transfer calculations caused by misuse of spectroscopic tools and databases (see Ref. [636] for instance).

The HAPI package can be obtained via the HITRAN*online* website (<a href="http://hitran.org/hapi">http://hitran.org/hapi</a>) or from the Zenodo community (<a href="http://zenodo.org/communities/hapi/about/">http://zenodo.org/communities/hapi/about/</a>). The current version (1.1) is distributed under a permissive open source license (MIT). Additional links can be found on the official HITRAN site.

HAPI consists of two major parts. First, it has a simple built-in database management system written in pure Python, which gives a capability of data processing and filtering. Second, HAPI incorporates a set of tools for spectra simulation accounting for the temperature, pressure, optical path length, and instrument properties. In more detail, the HAPI features are as follows:

• Support for standard HITRAN parameters given in "160-character" or ".par" format which is described in Ref. [3],

- Support for the newly added types of parameters such as foreign broadenings and shifts,
   speed dependencies for non-Voigt profiles, line mixing coefficients, etc.
- Communication with the HITRAN online web interface using the REST HTTP protocol in order to get the most up-to-date HITRAN data,
- Python implementations for line profiles including the partially correlated quadratic speed-dependent hard collision model (Hartmann-Tran profile [7,8,13,14]),
- Python implementation of the Total Internal Partition Sums (based on TIPS-2011 [112] and incorporating a large update outlined in this paper),
- High-resolution single-layer spectra simulation accounting for pressure, temperature, optical path length and instrument function. In the simplest case, the simulation is performed in a single step using a functional approach. See Fig. 25 where HAPI was used to generate cross-sections of ethylene to compare with experimental data.
- Capability of modeling the absorption cross sections broadened by a mixture of gases of astrophysical and planetary interests. The parameters for the foreign broadenings and shift are gradually being added to the HITRAN*online* system (see outline in Refs. [4]
   [5]). Figure 31 shows an example of how one can generate cross-sections of acetylene broadened by different planetary-relevant proportions of hydrogen and helium (see section 2.26 for details).

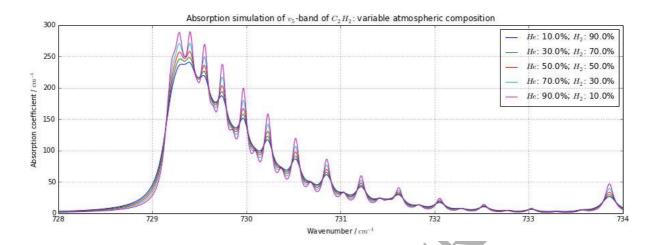


Figure 31. Cross-sections of acetylene in different environments of hydrogen and helium, generated with HAPI.

- Flexibility in the scheme of the absorption cross section calculation. Each part of the scheme given in Ref. [5] can be customized by user.
- Capability of extending the HAPI functionality by adding custom line shapes, partition sums, instrumental functions, environment dependences, and radiative-transfer code.
- Compatibility with a large set of third-party libraries in Python, C/C++, Java etc...

In the near future, in order to improve the simulation of the absorption cross sections, it is planned to include the full relaxation matrix calculation for line mixing for a number of molecules of atmospheric importance [117,238,239]. Inclusion of the line mixing in the calculation is expected to enable sub-percent accuracy (provided other parameters are known very accurately) for bands of atmospheric gases to better satisfy highly precise ground and satellite measurements.

At the current time HAPI requires the user to be familiar with Python, although most of the functions are prewritten and with the help of the manual one does not need to have advanced knowledge of that computer language. In order to extend the range of users, we plan to create a

flexible graphical user interface (GUI). This interface will give the user a possibility to simulate, plot and compare line-by-line and raw spectral data with a flexible control of scale, units, and computational parameters (just as is possible in the current Python version of HAPI). The new software will support Windows, Linux and MacOS.

### 6.3. Total Internal Partition Sums (TIPS)

Total internal partition sums (TIPS) were determined for all molecules and isotopologues in HITRAN2016 and for a number of isotopologues of molecules of interest in planetary atmospheres. The current TIPS data are calculated in one-degree steps from 1 K to  $T_{max}$ , where  $T_{max}$  is determined separately for each isotopologue based on convergence of the partition sums. The TIPS are comprised of updated calculations and values from previous studies of TIPS [112,637–639]. The updates can be grouped into several categories: improvements in *ab initio* calculations that allow direct sums over energy levels, improved  $Q_{vib}$  values using the anharmonic approximation [640], and calculations for new molecule/isotopologues in the HITRAN database. The updates that were made for isotopologues in this edition of HITRAN are listed in Table 13. The updates are detailed in an accompanying paper [641] in this HITRAN2016 special issue. Note that an error was found in Tables 1 and 2 of Ref. [641] where the values corresponding to partition sums of the  $^{16}O^{17}O$  isotopologue were a factor of six too large. This has been fixed in the official release of HITRAN2016.

Table 13. List of isotopologues whose partition sums have been updated for this edition of HITRAN. Note that there a few rare isotopologues here that are not in HITRAN.

Molecu numbe	Isotopologues
1	$H_2^{16}O, H_2^{18}O, H_2^{17}O, HD^{16}O, HD^{18}O, HD^{17}O, D_2^{16}O, D_2^{18}O, D_2^{17}O$
2	${}^{12}C^{16}O_2, {}^{13}C^{16}O_2, {}^{16}O^{12}C^{18}O, {}^{16}O^{12}C^{17}O, {}^{16}O^{13}C^{18}O, {}^{16}O^{13}C^{17}O, {}^{12}C^{18}O_2, {}^{17}O^{12}C^{18}O, {}^{12}C^{17}O_2, {}^{13}C^{18}O_2, {}^{18}O^{13}C^{17}O, {}^{13}C^{17}O_2, {}^{14}C^{16}O_2$
4	$^{14}N_2^{16}O$
5	$^{12}C^{16}O,  ^{13}C^{16}O,  ^{12}C^{18}O,  ^{12}C^{17}O,  ^{13}C^{18}O,  ^{13}C^{17}O,  ^{14}C^{16}O,  ^{14}C^{18}O,  ^{14}C^{17}O$

6	¹² CH ₄ ¹⁶ O ₂ , ¹⁶ O ¹⁸ O, ¹⁶ O ¹⁷ O, ¹⁸ O ¹⁸ O, ¹⁸ O ¹⁷ O, ¹⁷ O ¹⁷ O ¹⁴ NH ₃ , ¹⁵ NH ₃
7	$^{16}O_2$ , $^{16}O^{18}O$ , $^{16}O^{17}O$ , $^{18}O^{18}O$ , $^{18}O^{17}O$ , $^{17}O^{17}O$
11	¹⁴ NH ₃ , ¹⁵ NH ₃ H ¹⁴ N ¹⁶ O ₃ , H ¹⁵ N ¹⁶ O ₃
12	$H^{14}N^{16}O_3, H^{15}N^{16}O_3$
13	¹⁶ OH
14	H ¹⁹ F, D ¹⁹ F
15	H ³⁵ Cl, H ³⁷ Cl, D ³⁵ Cl, D ³⁷ Cl
16	$H^{79}Br, H^{81}Br, D^{79}Br, D^{81}Br$
17	$H^{127}I, D^{127}I$
20	$H_2^{12}C^{16}O$
22	$^{14}N_2$ , $^{14}N^{15}N$ , $^{15}N_2$
23	H ¹² C ¹⁴ N, H ¹³ C ¹⁴ N, H ¹² C ¹⁵ N
25	$H_{0}^{16}O_{0}$
26	$^{12}\text{C}_2\text{H}_2$
28	$^{12}\text{C}_2\text{H}_2$ $^{31}\text{PH}_3$ $^{12}\text{C}^{16}\text{O}^{19}\text{F}_2$ , $^{13}\text{C}^{16}\text{O}^{19}\text{F}_2$
29	$^{12}C^{16}O^{19}F_2$ , $^{13}C^{16}O^{19}F_2$
31	H ₂ ³² S, H ₂ ³⁴ S, H ₂ ³⁵ S
39	¹² CH ₂ ¹⁶ OH
47	$^{32}S^{16}O_{3}$
49	³² S ¹⁶ O ₃ ¹² C ¹⁶ O ³⁵ Cl ₂ , ¹² C ¹⁶ O ³⁵ Cl ³⁷ Cl

The TIPS are made available to the scientific community in the form of tables and codes that rapidly recall the TIPS for any molecule, isotopologue, and temperature. There are several codes that are distributed with this edition of HITRAN: FORTRAN codes TIPS_2017.for and BD_TIPS_2017.for and a python code TIPS_2017.py with associated dictionaries. TIPS_2017.for is a stand-alone code that queries the user for a molecule, isotopologue, and temperature and returns Q(T). BD_TIPS_2017.for is a subroutine application that users can insert into their codes to return Q(T) for a molecule/isotopologue/temperature selection. The python code uses python dictionaries to recall Q(T). The dictionaries are labeled by the molecule number and local ID number, see the TIPS article [641] for details. Details are given to develop custom python algorithms for particular molecule/isotopologue combinations. The TIPS python package was incorporated into the HITRAN Application Programming Interface

(HAPI) [5] in order to seamlessly integrate into the cross-section generating process. The tables and codes are available at faculty.uml.edu/Robert_Gamache and www.hitran.org.

#### 7. Conclusions

The improvements, expansions and new structure in the new HITRAN database release have been elaborated upon. For the line-by-line data, improved line position, intensity, and line-shape parameters for many of the previously existing molecules and isotopologues have been introduced. Many new lines/bands have been added for different molecules and their isotopologues allowing for a more complete database (for atmospheric applications) and for the expansion of the spectral coverage of the existing line lists. Additionally, HITRAN continues evolving in terms of structure and scope. A new relational database structure has been established [2] that enabled significant expansion of the amount of important spectroscopic parameters provided in the database. Literally dozens of new parameters have been introduced that allow accommodation of non-Voigt line shapes [6]; flexible representation of broadening parameters and their temperature dependencies; broadening and shift of spectral lines due to pressure of gases dominant in planetary atmospheres [4] (namely, H₂, He and CO₂). This extensive expansion of the database would have been very impractical in the old fixed-length ASCII format of previous HITRAN editions. An interface on the internet has also been established (www.hitran.org) that provides the diverse group of HITRAN users with much power to filter, extract, plot, and query the database. Even more power has been given to the user through the introduction of the HITRAN Application Programming Interface (HAPI) [5]. HAPI is currently a set of Python libraries which not only allows users to download data from the database but also to carry out: 1) single-layer absorption calculations at different thermodynamic conditions, accounting for an instrument function, new line shape parametrizations, perturbing

gas or gases, etc, 2) extended functionality provided by custom functions, line lists and partition sums.

Several new molecules and isotopologues have been added to the line-by-line portion of the compilation, which improve modeling of the absorption of the terrestrial atmosphere and beyond.

One of the main highlights of the new edition is the monumental expansion of the cross-sectional part of the database which increased (in terms of amount of gases) about six-fold and now features over 300 species. These cross-sections can also be accessible through <a href="https://www.hitran.org">www.hitran.org</a> where they are divided into categories for easy browsing.

Aerosols and collision induced absorption datasets have also been revised and extended.

The compilation is free and can be accessed through <u>www.hitran.org</u>. We recommend citing this article when using the HITRAN2016 data. If an update to a certain molecule is announced on the website and is being used then we encourage users to cite when they downloaded the data.

#### 8. Future Work

While HITRAN continues to be an international standard for reference spectroscopy for atmospheric and planetary gases, there is still substantial room for improvement. In many subsections throughout the paper we mention the avenues for improvements for individual molecules, but they are really just the tip of the iceberg.

The studies of the terrestrial atmosphere require further improvements to the existing parameters and what is more challenging: population of the dataset of the non-Voigt line-shape parameters. As was discussed above, the amount of measurements and calculations of the parameters obtained using the recommended HT profile are scarce and it will take time before a majority of

HITRAN lines will have these parameters provided. Also, we remind experimentalists and theoreticians to take extreme care when fitting to these parameters to avoid correlations.

The line mixing for many molecules is now available in HITRAN through diagonal (Rozenkranz) parametrization. Other parametrizations require provision of additional software (just as it is done in this edition in the case of carbon dioxide) that may have to be molecule or even band-specific. HAPI offers an excellent avenue for addition of this extra functionality.

At the moment only seven HITRAN gases have pressure broadening (and their temperature dependence) and pressure shift of lines by H₂, He and CO₂. We would like to extend this to all gases important for planetary research. Also, in order to aid modeling of the terrestrial atmosphere, we will add parameters associated with broadening of spectral lines by water vapor (which is sufficiently abundant to make an impact on retrievals in the tropics, for instance).

With an exception of a few diatomic molecules (namely NO, OH, HF, HCl, HBr, HI and H₂) HITRAN data are limited to temperatures encountered in the Earth atmosphere and will be deficient when modeling high-temperature environments including stellar and some planetary atmospheres. The users are encouraged to employ the HITEMP database for these applications [278]. However, there are currently only five gases that are provided in HITEMP (H₂O, CO₂, CO, NO, OH) and a major update of the HITEMP database will be released in the near future. This update will include improvement of the spectroscopy of the existing HITEMP gases but also will introduce additional molecules.

While a major update is featured here for the cross-sectional part of the database we note that it is still far from complete in terms of gases. Moreover, all of the current updates are in the IR part of the spectrum and another substantial update is needed for the UV cross-sections.

A massive revision and expansion of the CIA data is planned for the near future.

Structure and the tools provided with the database are also a very important part of the HITRAN database. We plan further improvements to the documentation for the HITRAN online interface and the development of video tutorials. Finally, development of an interactive interface is planned for HAPI, for which the video tutorials will also be created.

#### Acknowledgments

The management and development of the HITRAN molecular database has been supported by NASA AURA program grant NNX14AI55G and NASA Planetary Atmosphere grant NNX13AI59G), and NASA PDART grant(NNX16AG51G. Portions of the research described in this paper were performed at the Jet Propulsion Laboratory, California Institute of Technology, under contract with the National Aeronautics and Space Administration, Government sponsorship acknowledged.

We gratefully acknowledge the many researchers who provided data, experimental and/or theoretical: Abdulsamee Alkadrou, Lorenzo Barrett, Emma Barton, Chris Benner, Linda Brown, Jeanna Buldyreva, John Coxon, Phillip Coles, Ludovic Daumont, Jolanta Domyslawska, Andre Fayt, Photos Hadjigeorgiou, Shuiming Hu, Ekaterina Karlovets, Samir Kassi, Julien Lamouroux, Daniel Lisak, Lorenzo Lodi, Gang Li, Anwen Liu, Didier Mondelain, Tatyana Petrova, Laurence Régalia, Candice Renaud, Cyril Richard, Alexander M. Solodov, Alexander A. Solodov, Mikhail Tretyakov, Daniel Underwood, Sergey Yurchenko, Nikolai Zobov.

Those who provided independent validations are also acknowledged: Matthew Alvorado, Joshua Baraban, Chris Boone, Athena Coustenis, Jose Fonfria, Erik Kerstel, Xiong Liu, Emile Medvedev, Jana Mendrok, Eli Mlawer, Johannes Orphal, Fabiano Oyafuso, Vivienne Payne, Steven Ross, Kang Sun, Geronimo Villanueva.

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