



# **Nanoporous Materials for Environmental and Energy Applications**

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&

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## Technical Focus

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- Technical Focus:  
**Chemical study of confinement and reactivity of ions and molecules in micro- and nano-porous materials**  
- zeolites, metal-organic frameworks (MOFs), clays and amorphous silicas
- The study of **Structure-Property Relationships** on the nanoscale enabling the informed testing of the materials for a wide range of interests.
- Structure analysis of host-guest systems on the nanoscale has led to strong collaborations with staff scientists at the APS for synchrotron data collection & Pair Distribution Function (PDF) analysis
- **Strong Interdisciplinary Teams** working in concert.  
diverse programs require diverse teams  
chemists, engineers, computational modelers  
national labs, universities and industry
- Mentoring: postdocs, young staff and collaborating young professors,  
strong support of women and minorities into the sciences



## Program Development & Management - Nenoff

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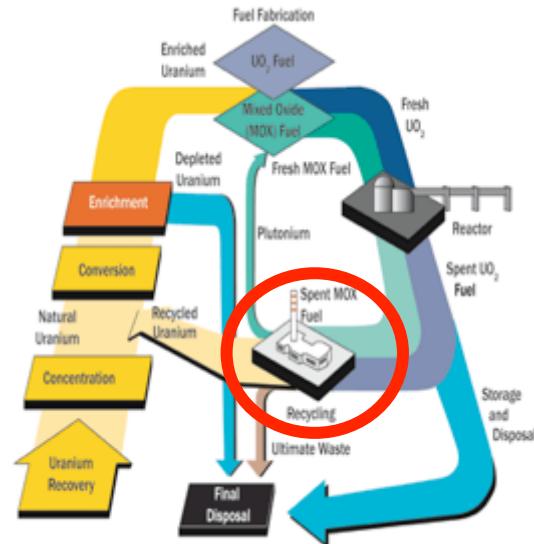
- Raised \$35+ Million raised in individual PI grants  
150+ publications, 13 US Patents, 4 book chapters,  $h$ -index = 33
- Developed strong relationships at DOE, resulted in *long running* programs
  - DOE/EERE/Office of Industrial Technologies (Advanced Materials)
  - DOE/EERE/Hydrogen
  - DOE/Environmental Management
  - DOE/Office of Nuclear Energy

*Building* customers with ARPA-E, DOE/BES, and DOE/Fossil Energy (NETL)

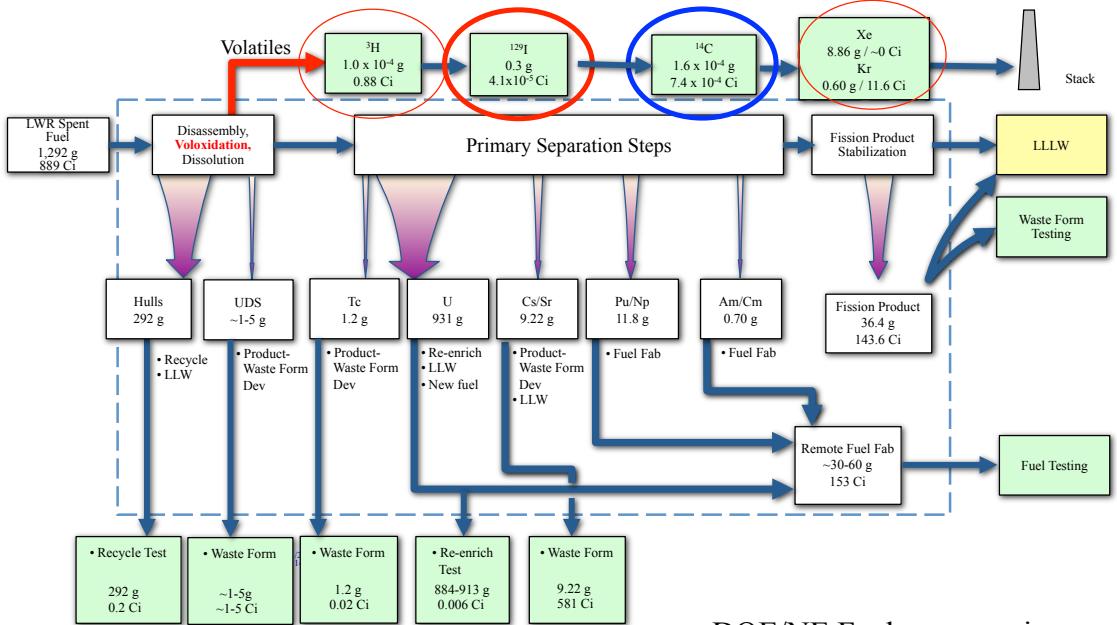
- Research Group works in **TRL 1 - 9**
- Led 2 different multiyear, multimillion \$ CRADAs
  - SNL, BP, Temec, Coors Ceramics, University Cincinnati:  
Zeolite Membranes for HC isomer feedstock purification
  - SNL, Goodyear Chemicals, University Cincinnati:  
Membranes for Separations of High Volume C<sub>4</sub>/C<sub>5</sub>/C<sub>6</sub> Mixtures

# Applications

Reprocessing: capture on nonburnable volatile fission products and lesser actinides



Source: U.S. Nuclear Regulatory Commission



DOE/NE Fuel reprocessing Scheme, ORNL

Legacy, Accident or Produced rad aqueous waste requiring highly specific ion capture

*Fukushima Daiichi*  
Nuclear Power  
Plant explosion 2011  
 $I^{129}$ ,  $I^{131}$  volatile  
gas released;  
 $Cs^{135}$ ,  $Cs^{137}$  &  $Sr^{90}$   
aqueous released  
([www.IAEA.org](http://www.IAEA.org))

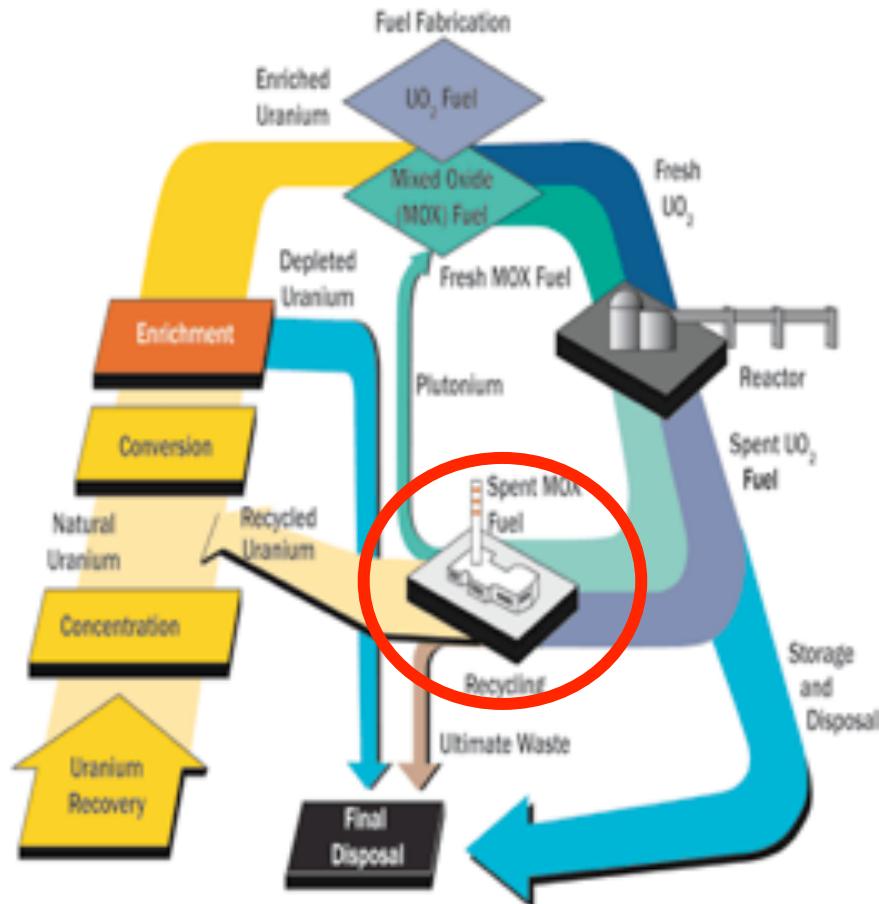




# Applications: Fission Gases

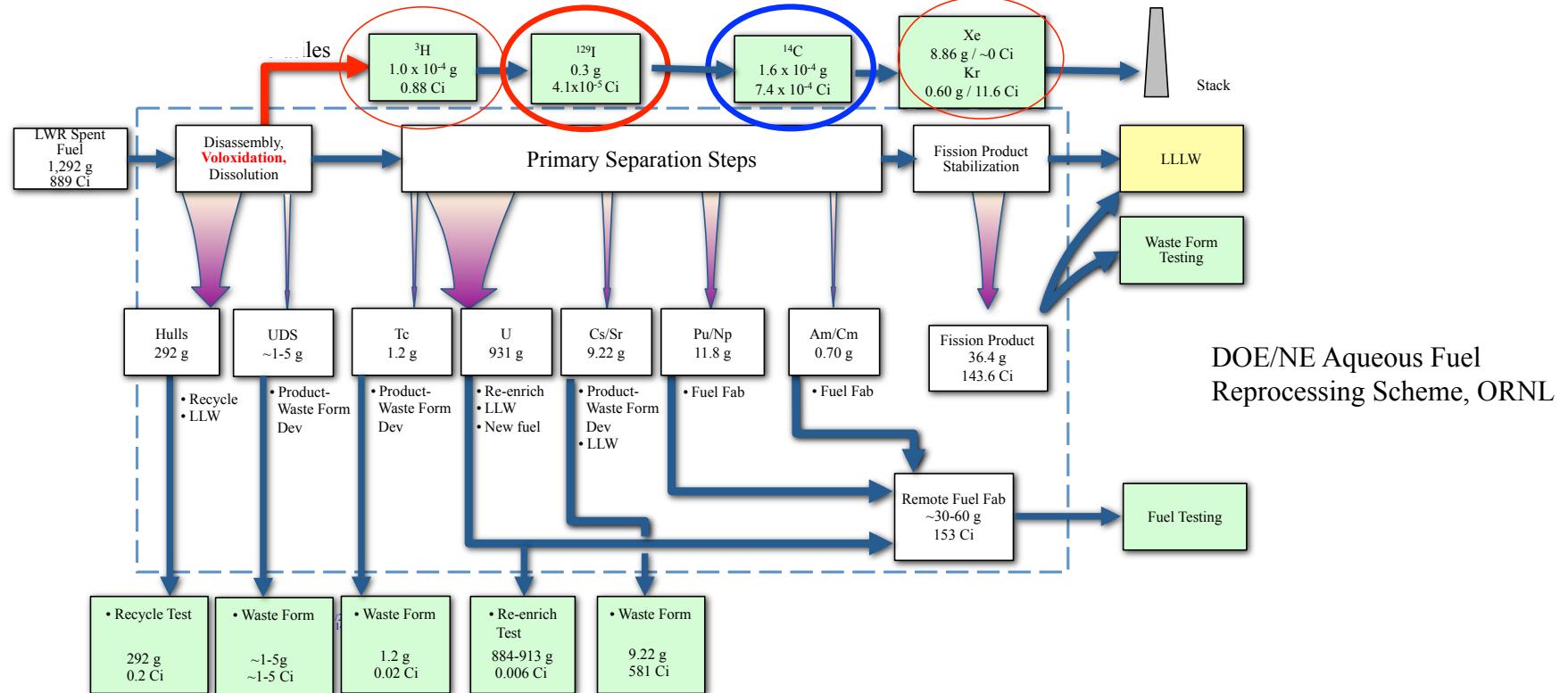
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Reprocessing: capture on nonburnable  
volatile fission products and lesser actinides

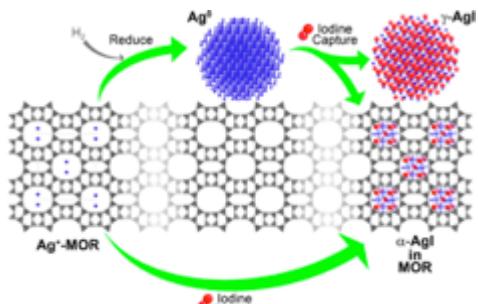


Source: U.S. Nuclear Regulatory Commission

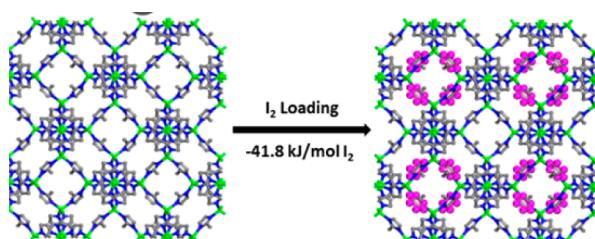
# Applications: Fission Gases



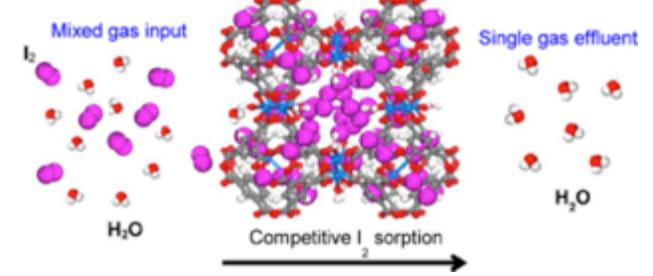
## Zeolites – I<sub>2</sub> capture/storage



## MOFs – small pore for large quantity I<sub>2</sub> capture



## MOFs – large pore for competitive I<sub>2</sub> vs H<sub>2</sub>O, and large volumes



# Applications: Rad Ion Capture

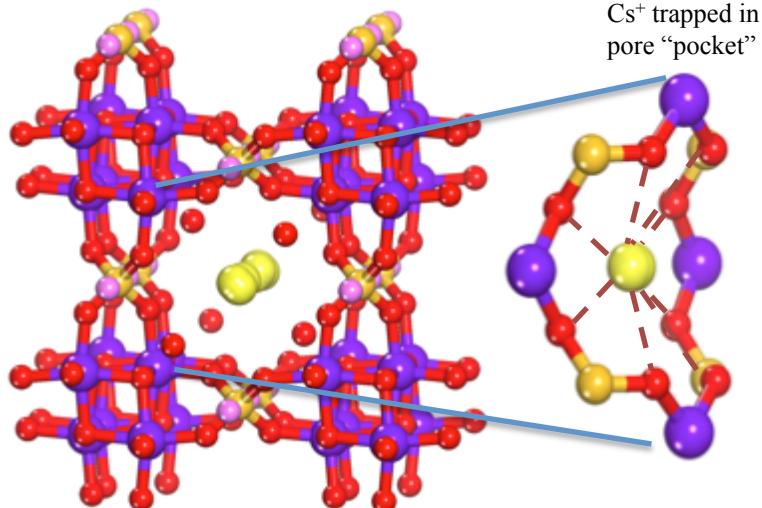
Legacy, Accident or Produced rad aqueous waste requiring highly specific ion capture.

Cs<sup>+</sup> ion capture: Crystalline Silicotitanates (CSTs)

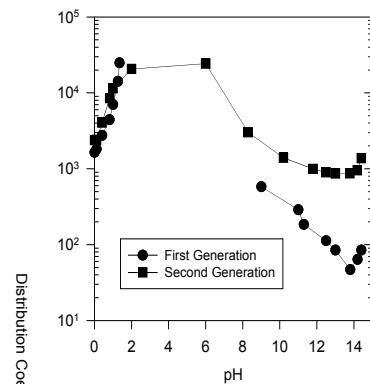
Fukushima Daiichi Nuclear Power Plant explosion 2011 I<sup>129</sup>, I<sup>131</sup> volatile gas released; Cs<sup>135</sup>, Cs<sup>137</sup> & Sr<sup>90</sup> aqueous released  
([www.IAEA.org](http://www.IAEA.org))



Cs<sup>+</sup> captured in CST cage



Distribution Coefficient of Cs on CST



UOP IONSIV™ IE-911: LDRD idea to Commercial product in 5 years



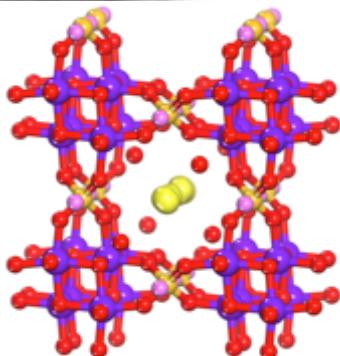
Augmentation of accumulated water processing facility (SARRY)



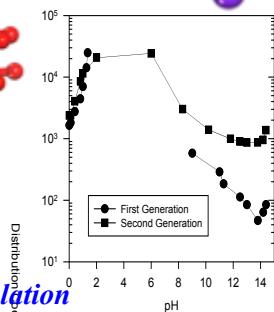
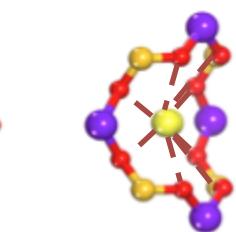
CSTs used in Cs<sup>+</sup> cleanup of 160M+ gallons from seawater in Fukushima buildings



# Example 1: Nanoporous Materials (Zeolites, Molecular Sieves & MOFs), Radiological Ion and Gas Capture



**I<sub>2</sub>/ZIF-8 MOF, Encapsulation to Waste Form**  
JACS, 2011, 133(32), 12398  
JACS, 2013, 135, 16256



## CST, Molecular Sieve:

R&D100 1996

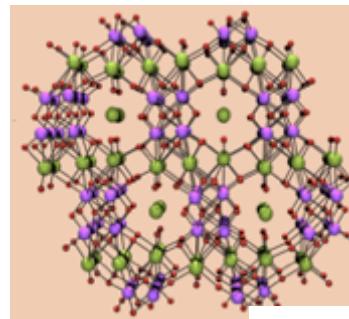
JACerS, 2009, 92(9), 2144

JACerS, 2011, 94(9), 3053

Solvent Extr. & Ion Exch, 2012, 30, 33

## CST, Cs<sup>+</sup> removal from water to Pollucite Waste Form

US Patents 6,479,427; 6,110,378



## SOMS Molecular Sieve, Sr<sup>2+</sup> getter,

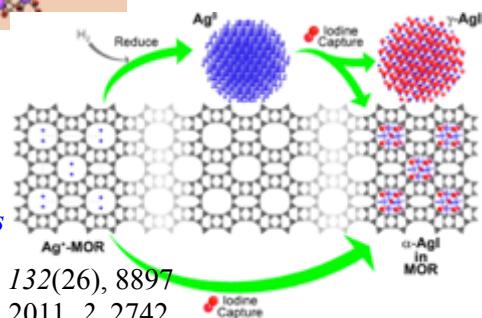
## 1-step to Perovksite WF

JACS, 2002, 124(3), 1704

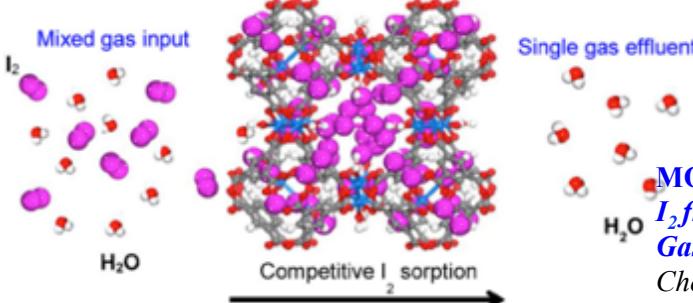
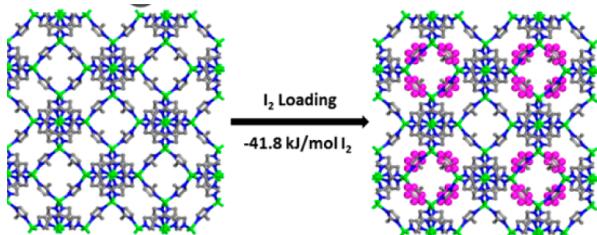
US Patent 7,122,164; 2006

## Ag-MOR Zeolite, I<sub>2</sub>(g) capture & mechanisms

JACS, 2010, 132(26), 8897  
JPC Letters, 2011, 2, 2742



**Design, Produce, Test and Commercialize Materials with Specific Selectivity to Ion/Gas by Size & Chemistry of Nanopore**

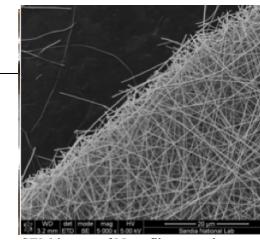
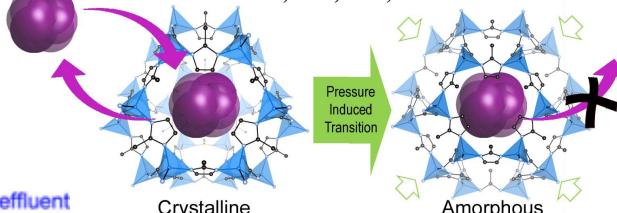


T. M. Nenoff, tmnenof@sandia.gov

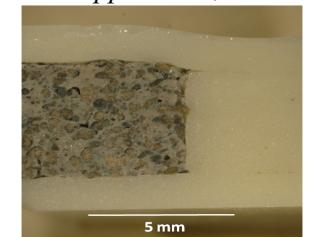
## MOF Amorphization for Gas Storage

JACS, 2011, 133(46), 18583

US Patent 9,162,914; 2015



**TiO<sub>2</sub> Nanoporous Nanofibers**  
Volatile Gas Removal  
US Patent Application, 2011



## Glass Composite Material

## GCM:

**Universal Core-Shell Iodine Glass Waste Form & Getter**  
JACerS, 2011, 94(8), 2412  
US Patent 8,262,950; 2012



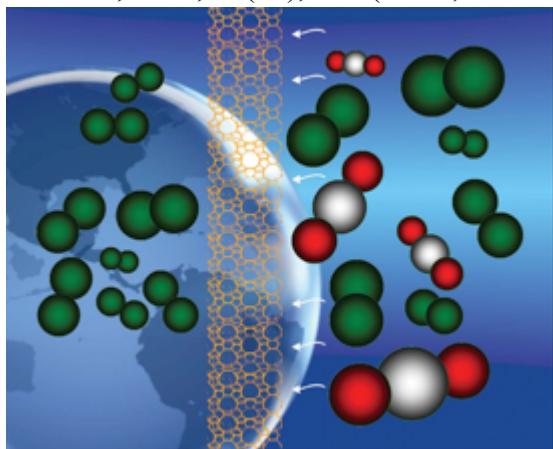
**Binder Free MOF Pelletization**  
US Patent Pending 2014

**MOF Cu-BTC: I<sub>2</sub> from Humid Gas Stream**  
Chem. Mater. 2013, 25(13), 2591

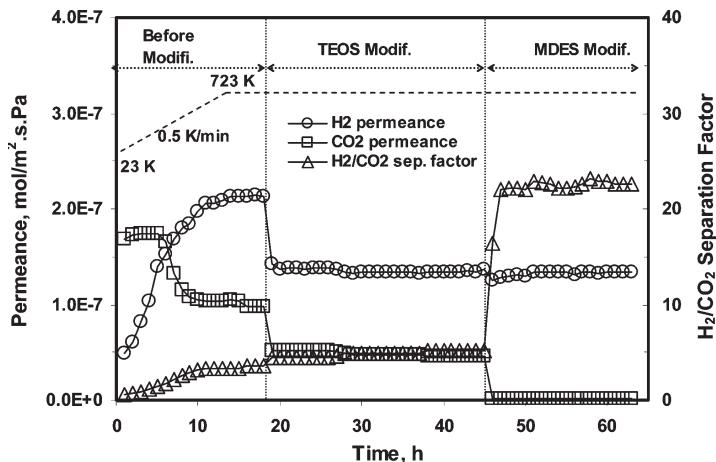
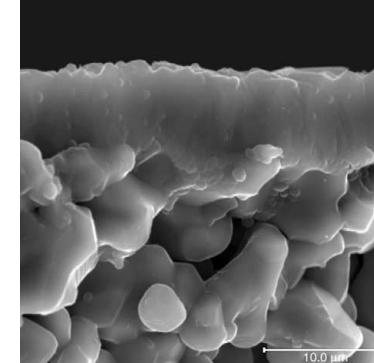
# Example 2: Zeolite/MOF Membranes for Light Gas & Hydrocarbon Separations

## H<sub>2</sub> Purification

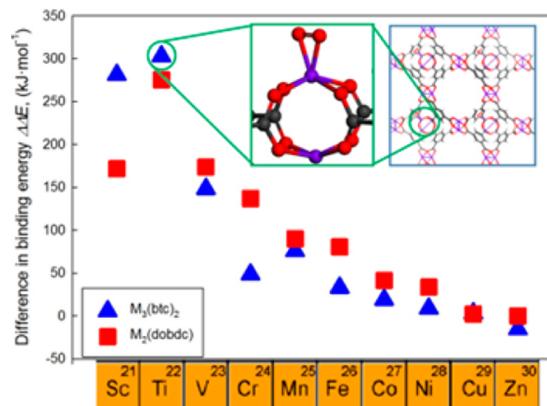
*Nature Mater.*, 2015, 7, 377; *Chem. Rev.* 2007, 107, 4078  
*MRS Bulletin*, 2006, 31(10), 735 (cover; Editor Nenoff)



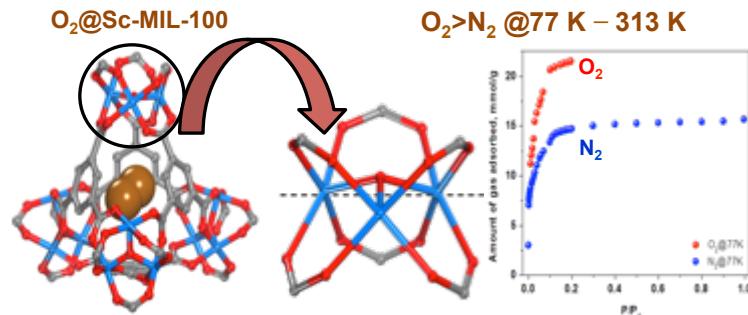
ZSM-5 Zeolite membrane for H<sub>2</sub> from CO<sub>2</sub> separations  
*Micro Meso Mater.*, 2003, 66, 181



H<sub>2</sub> Purification from Syngas  
*Langmuir*, 2009, 25(9), 4848



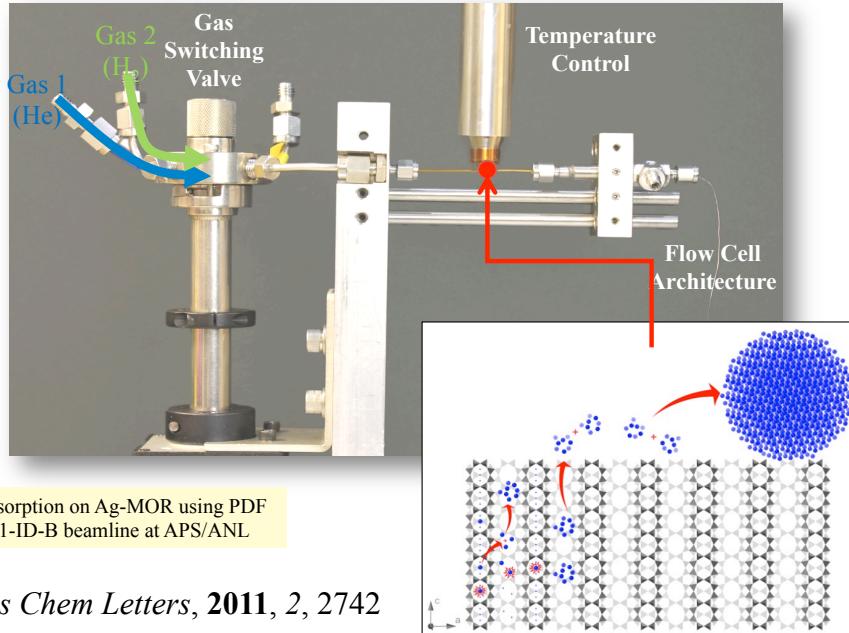
O<sub>2</sub> Separations with MOFs for Energy Efficient Oxyfuel Combustion  
*Chem. Mater.*, 2016, in press  
*Chemical Science* 2016, 18, 11528  
*J. Phys. Chem. C*, 2015, 119, 6556  
*Chem. Mater.*, 2015, 27(6), 2018





# Example 3: Nanoporous Zeolites for Heterogenous Catalysis and Subsequent Gas Capture

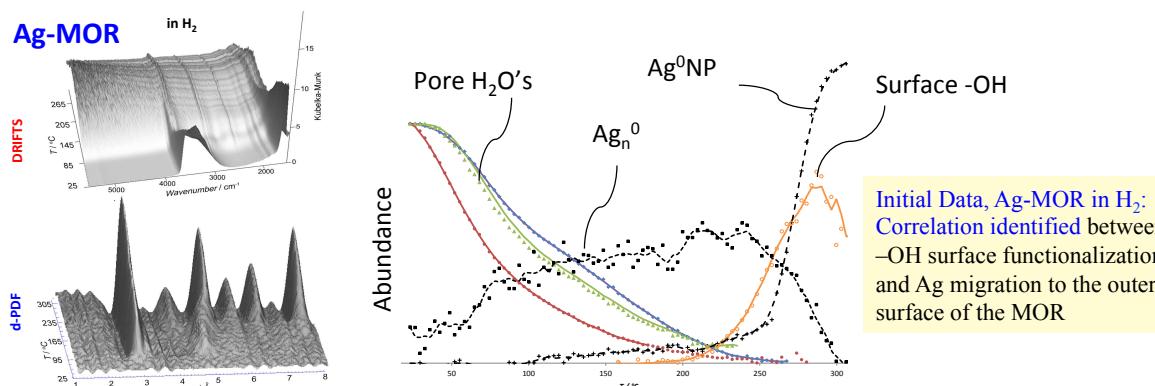
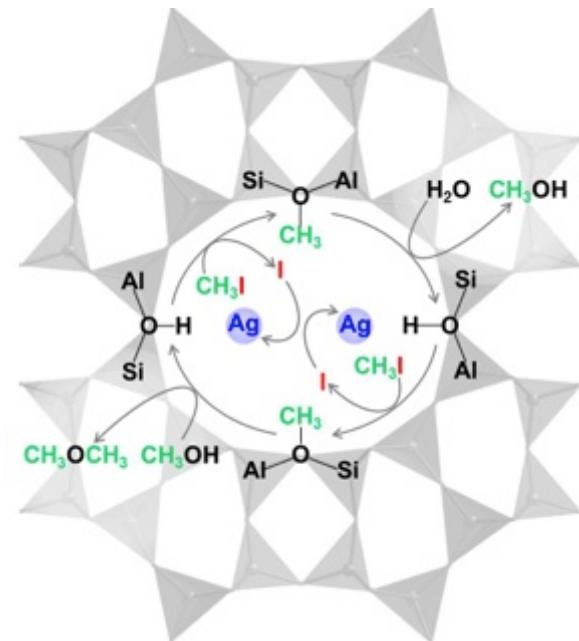
Determine Mechanism of  $\text{Ag}^+$  to  $\text{Ag}^0$  NPs in MOR



*J Phys Chem Letters*, 2011, 2, 2742

Mechanism of Iodine Capture on Ag-MOR from Acidic Humid  $\text{CH}_3\text{-I}$  Stream: Catalytic Cleaving of  $\text{CH}_3\text{-I}$  and I Capture

*Micro. Meso. Mater.*, 2014, 200, 297 (invited)



On-going: Combining complementary insights from simultaneous IR and PDF.  
For mechanism of NP catalyst formation

Chapman & Nenoff 2016, in preparation