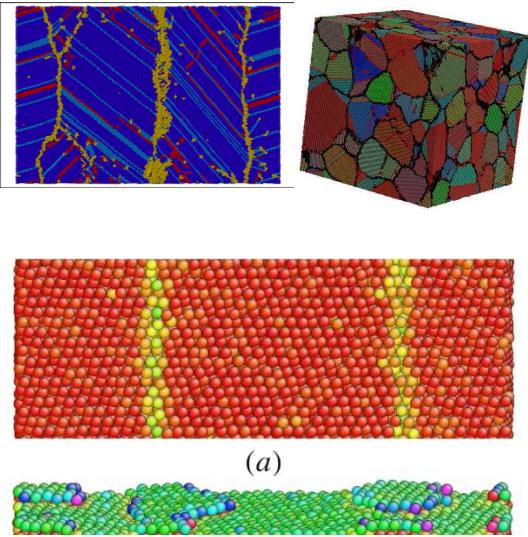


Simulating Grain Boundaries at the Atomic Scale: *More Complicated Than You Think*



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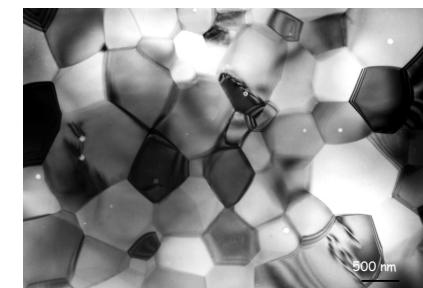
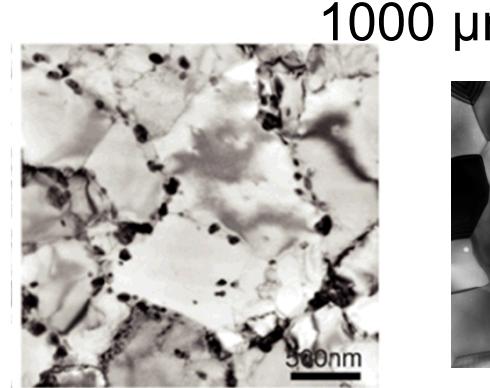
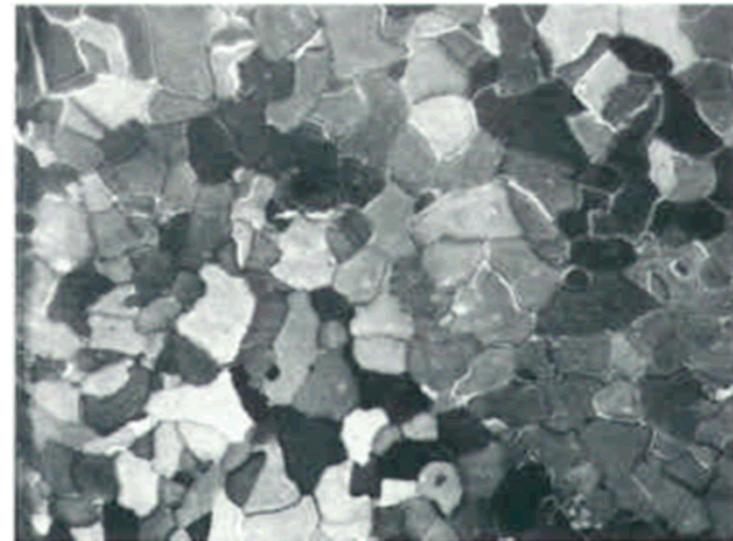
U.S. DEPARTMENT OF
ENERGY



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Why does anyone care about grain growth?

- Grain-level microstructure strongly influences a wide range of materials properties
 - Strength
 - Hall-Petch relationship: $\sigma_y = \sigma_0 + \frac{k_y}{\sqrt{d}}$
 - Toughness and Fracture
 - Corrosion resistance
 - Electrical conductivity
 - Magnetic susceptibility
 - ...
- *Controlling the microstructure and relating the microstructure to properties are central problems in materials science.*

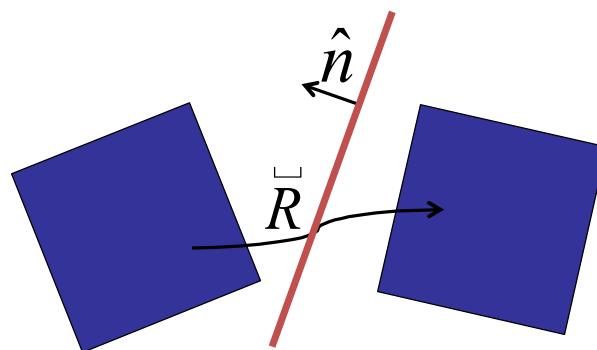


Evolution of grain microstructure is a highly complex multi-scale modeling problem

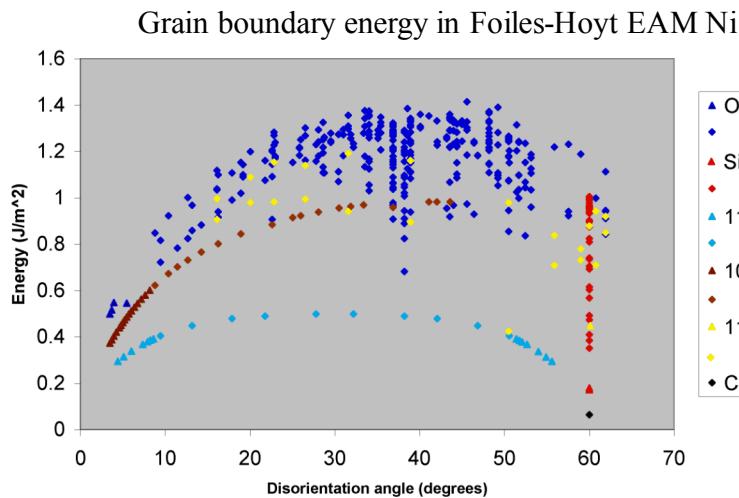
- Atomic-scale
 - Boundary properties are determined by atomic-scale structure and dynamics
 - Energy – changes in *atomic-level* bonding/coordination in the boundary
 - Motion – local *atomic-level* rearrangements at the boundary
 - Time-scale: picoseconds – nanoseconds
- Meso-scale
 - Grain sizes: ~10 nanometers - ~100 micrometers
 - Need to consider the 3-D network of grain boundaries
 - Time-scale: seconds to hours
- Conventional strategy
 - Determine the properties of grain boundaries with atomic-scale methods
 - Evolve the grain structure with meso-scale simulations that incorporate the boundary properties – energy, mobility

What is the big deal about determining grain boundary properties?

- *“We hold these truths to be self-evident, that all grain boundaries are **NOT** created equal, ...”* - apologies to Thomas Jefferson
 - There is a **5-dimensional** space of macroscopic grain boundary structure
 - The properties vary throughout this 5-D space in an, at best, partially understood manner
 - And this does **NOT** even consider the effects of **temperature, alloying, impurities, second phases, applied stress, ...**
 - For a given macroscopic configuration, multiple microscopic (atomic-level) grain boundary structures may be present in equilibrium

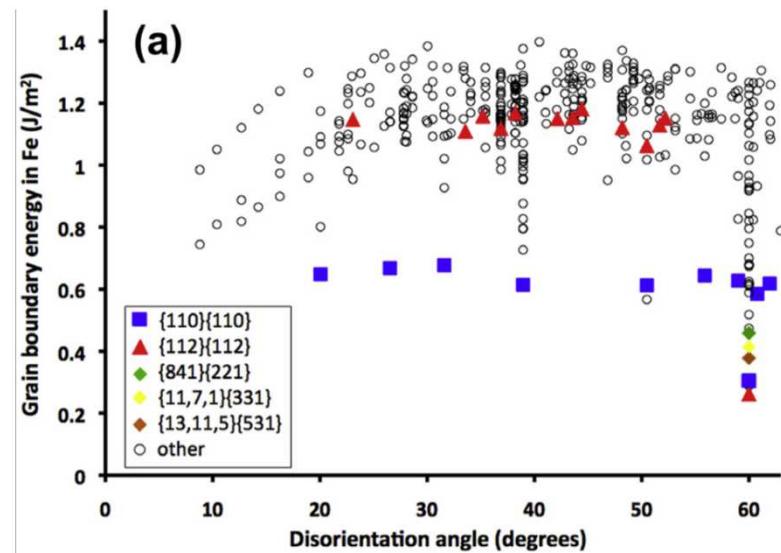


Grain Boundary Properties have been computed for large catalogs of elemental boundaries



- Similar data has been generated for BCC metals
 - Ratanaphan, Olmsted, Bulatov, Holm, Rollett & Rohrer, *Acta Mater* 88, 346 (2015)

- Energy variations computed for boundaries in FCC metals
 - Olmsted, Foiles, & Holm, *Acta Mater* 57, 3694 (2009)
- Numerical parameterization of these energies has been developed
 - Bulatov, Reed & Kumar, *Acta Mater* 65, 161 (2014)



Overview of atomistic simulation methods: 'Twitterverse' Version



- Representation of energy/forces in terms of atomic positions
 - Better: Computed from *approximate* solution of quantum mechanics of the electron ground state (ie. "DFT")
 - Typical: Interatomic potential
 - Reduced Order Model of the Born-Oppenheimer Surface
- Energy minimization (Molecular Statics)
 - Optimize atomic positions to find an energy minimum
 - *Hopefully*, global minimum but often use local minimization methods
- Molecular Dynamics
 - Follow classical Newtonian trajectory of atoms – " $F = ma$ "
 - Let the atoms "*do what they want*" subject to imposed boundaries
- Equilibrium Monte Carlo
 - Adjust atomic arrangements to sample a thermodynamic ensemble

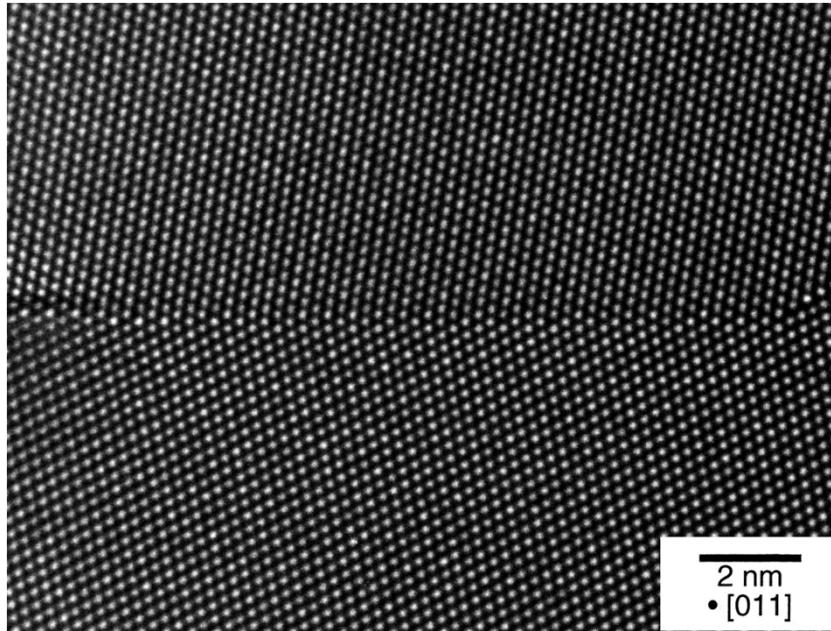
Molecular Dynamics:

What could possibly go wrong?

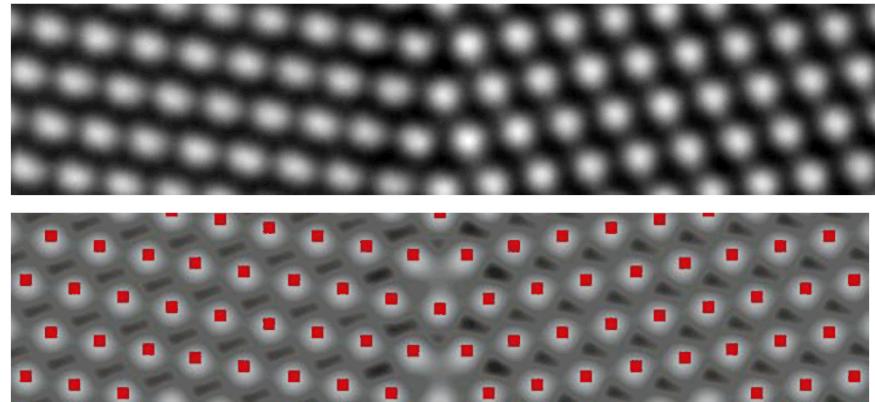


- Deviations from Born-Oppenheimer Approximation
 - Highly dynamic events, charge states, ...
- Inadequate Interatomic Potentials
 - Imperfect knowledge of Born-Oppenheimer surface
 - Errors in experimental data
 - Deficiencies of *ab initio* database
 - Assumed potential form – model form error
 - Transferability or lack thereof
- Quantum Mechanical Effects
 - Zero-point energies of light elements
 - Debye Temperature often above room temperature
- Limited time scales
 - High rates and high driving forces
 - Infrequent events
 - Sampling errors - ,metastable states
- Structural approximations
 - Where are the atoms, *really*?
 - Simplification of geometries
 - Boundary conditions
- Multi-component systems
 - Composition and structure coupled
 - Compositional variation - equilibrium or kinetic?
- Information extraction for higher-scale models
 - Millions of coordinates -> 'Physics'
 - Identification of dominant effects

MD can reproduce structure of high-symmetry boundaries in Al



- High-resolution TEM of $\Sigma 11(1\ 1\ 3)/[-1\ 1\ 0]$ boundary in Al



- Top: close-up of boundary
- Bottom: Simulated TEM image based on computed structure using EAM potential for Al
 - Red dots show column positions

“Great, kid! Don’t get cocky!” – Han Solo

Boundaries in non-FCC metals are harder to predict

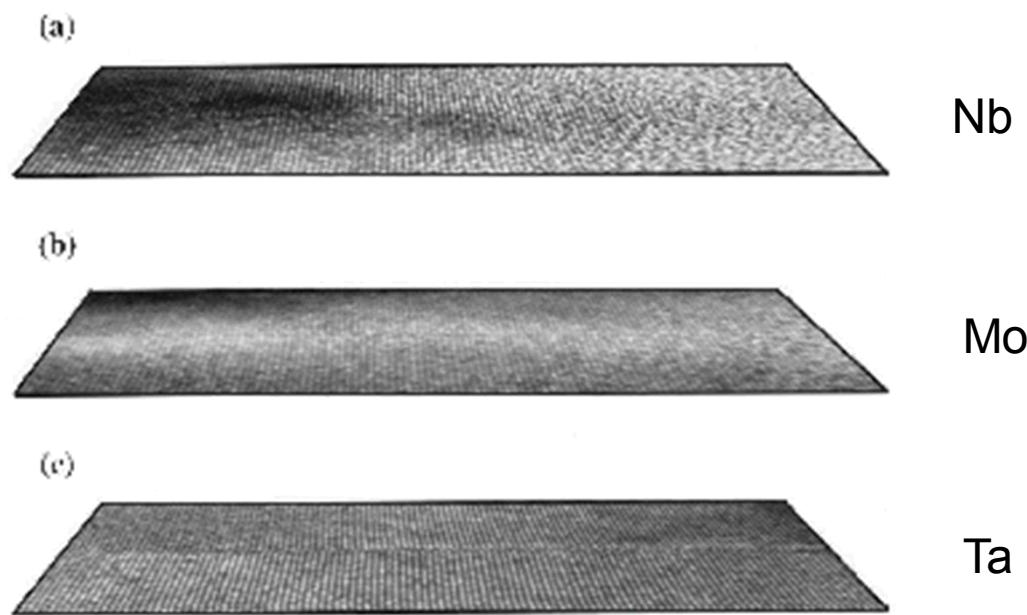
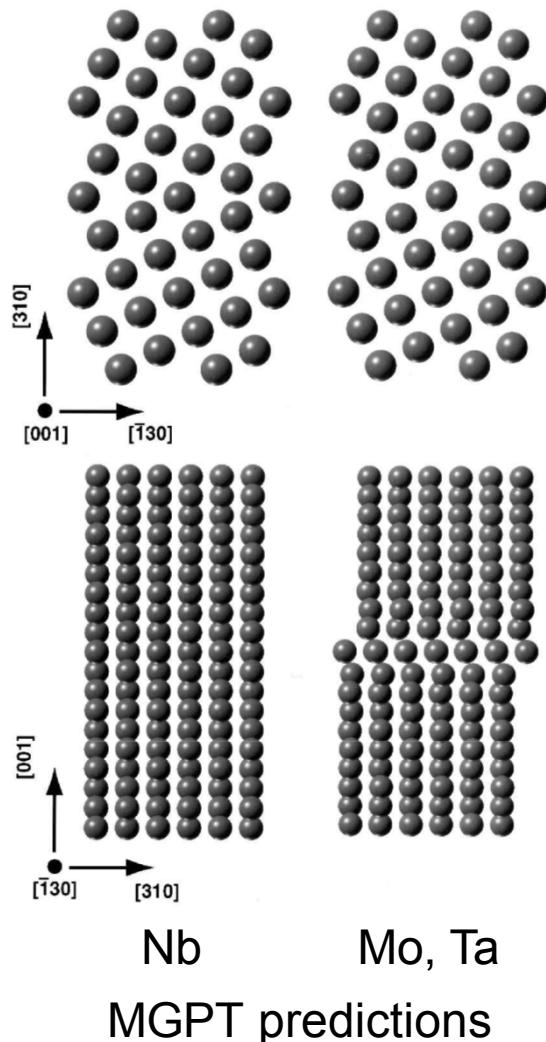
- Central transition metals have partially filled d-bands
 - d-band bonding has an angular character reflecting d-orbital symmetry
 - The trends in equilibrium structures of central transition metals can be rationalized in terms of these interactions
 - David Pettifor’s structure maps
- Commonly used potential forms like EAM and Finnis-Sinclair do **NOT** incorporate angular dependence
- *Potentials with and without angular terms predict different structures!*

71° [10], which is the nearest-neighbor bond angle in the bcc structure. Model 2 contains a significant number of right angles at the boundary, whereas the other two models do not.

Grain boundaries forming (310) twins in Nb were prepared by diffusion bonding two precisely oriented, to within $\pm 0.1^\circ$, Nb single crystals with flat polished (310) surfaces which were misoriented by 180° about [310] rel-

Predicted
structures for
(310) twin in Nb

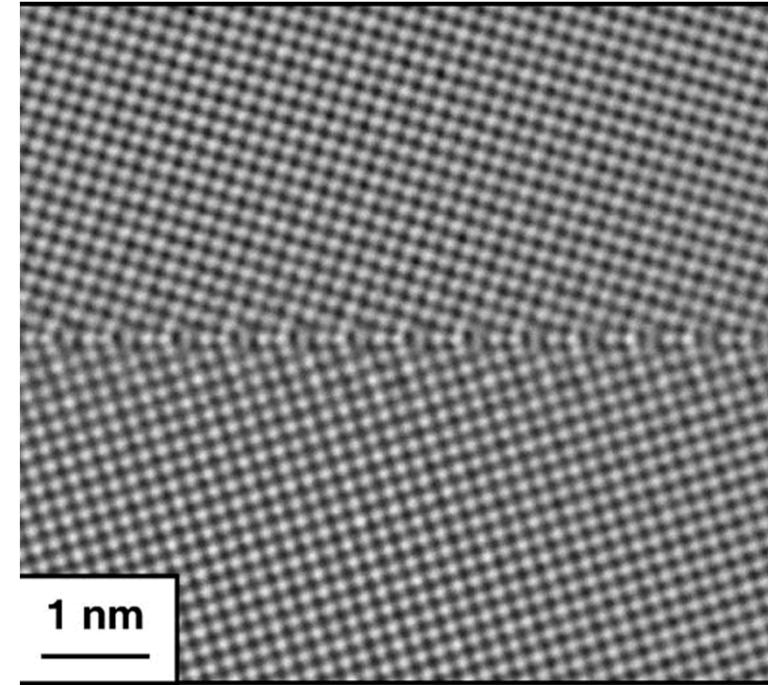
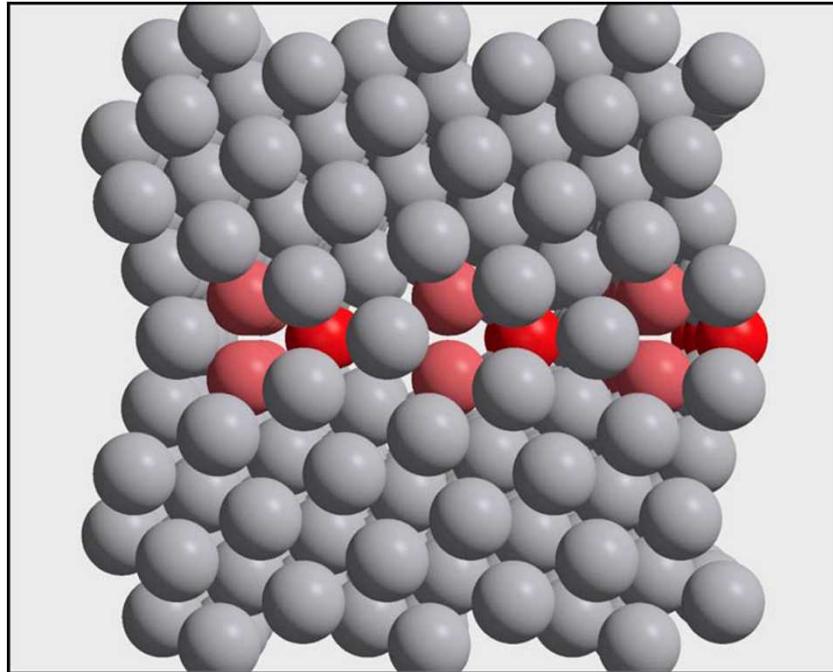
With sufficiently accurate potentials the right structures are predicted



- Modified Generalized Pseudopotential Theory (MGPT) qualitatively predicts which elements possess a shift along tilt axis
 - EAM and FS potentials fail this test

Cu in Al – bulk substitutional alloy

Just replace Al by Cu near the boundary, right?

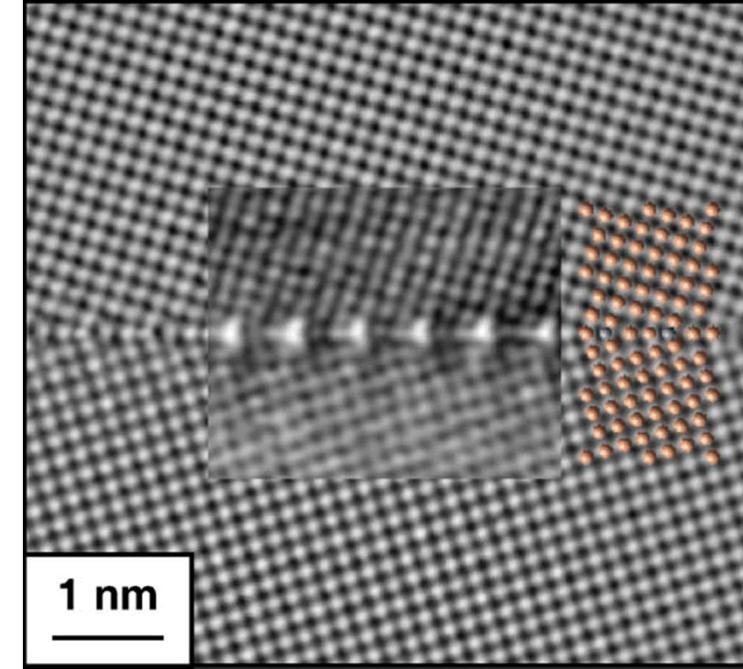
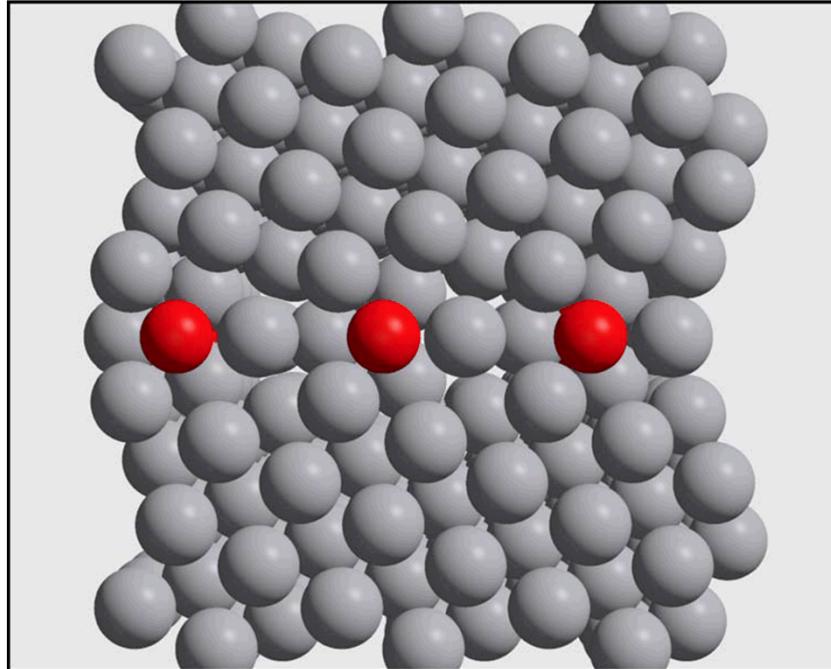


DFT predicted substitutional energies for Cu near Al $\Sigma 5$
(310)/[001]

HRTEM shows a 3-atom repeat along boundary as opposed to 2-atom repeat in calculations

Cu is substitutional in bulk Al

but interstitial at this boundary

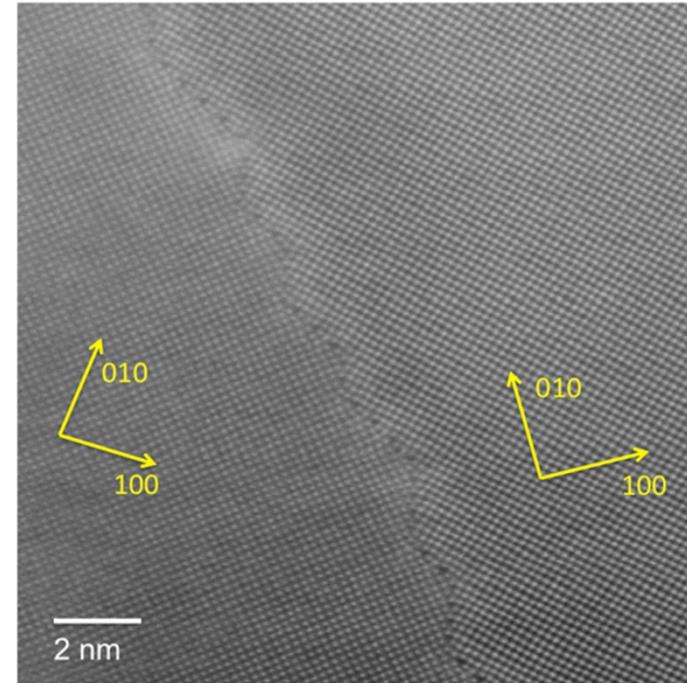
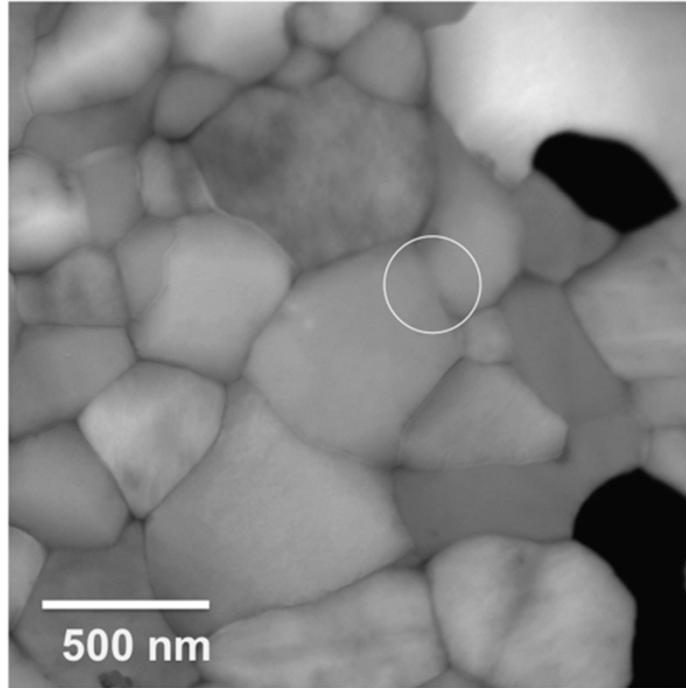


DFT predicts that the segregation to the interstitial site is stronger than to substitutional sites

Interstitial Cu agrees with the HRTEM image

Be careful where you assume the atoms are!

Faceting of $\Sigma 5$ grain boundary in Fe

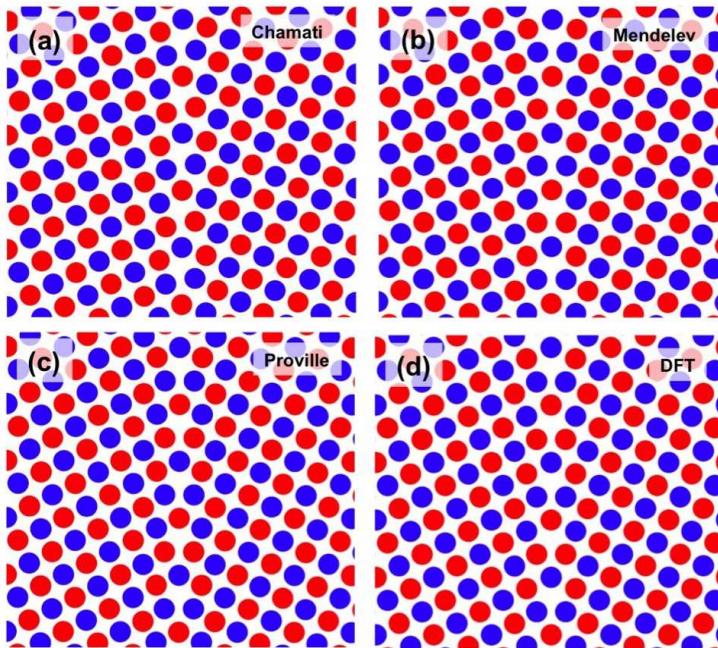


HAADF-STEM images of Fe film

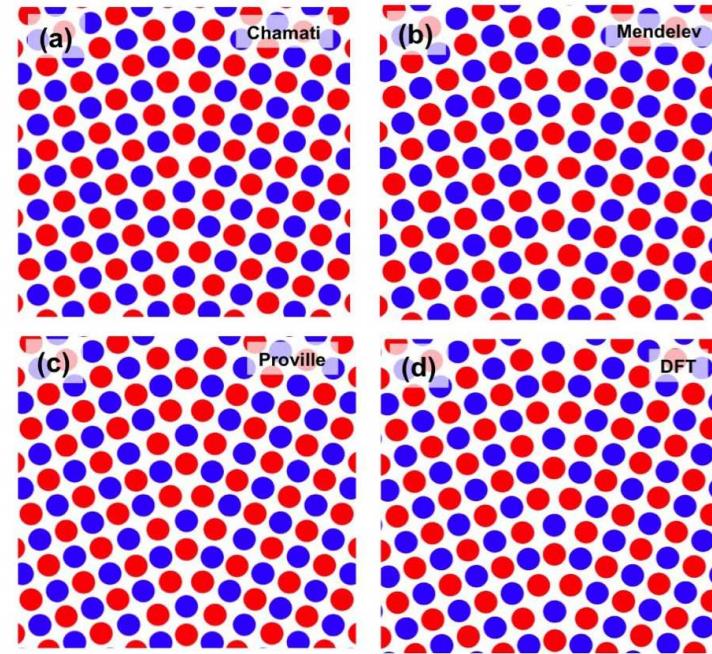
- Overall grain structure
- Atomic resolution detail of the $\Sigma 5$ boundary

Faceting of $\Sigma 5$ grain boundary in Fe: Atomistic prediction for facet geometries

{310} facets



{210} facets

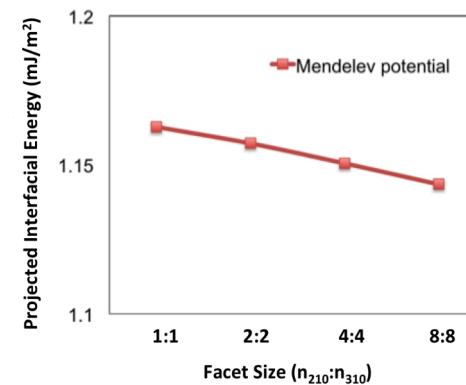
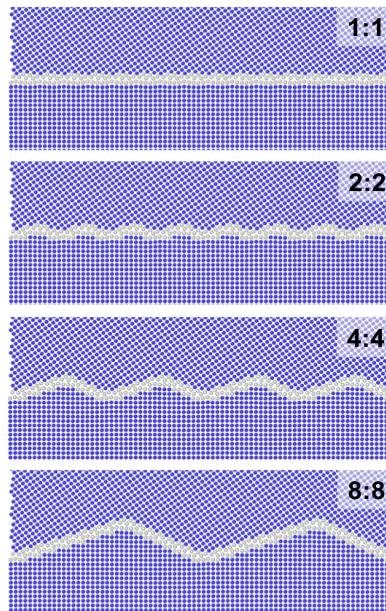


- Subtle differences between structures computed with different empirical potentials
 - Frames (a) - (c)
- DFT (approximate quantum mechanical calculations) are in frame (d)

Faceting of $\Sigma 5$ grain boundary in Fe: Atomistic prediction for facet length

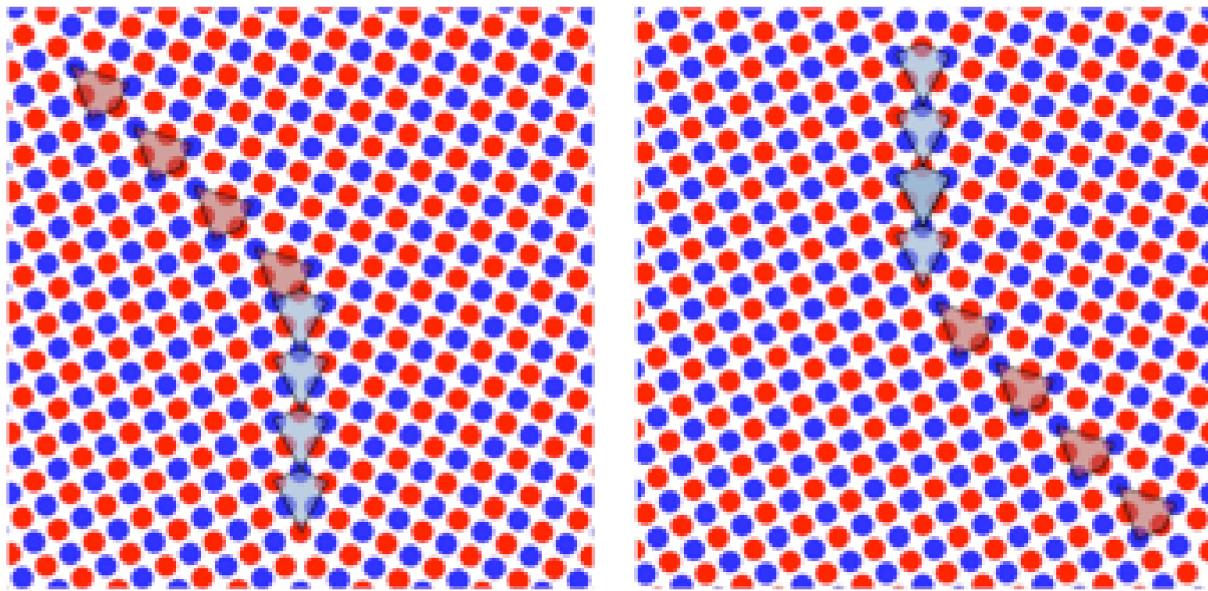
Equilibrium facet size depends on

- Energies of facet planes
 - Variation of boundary energy with boundary plane orientation
- Junction energies
- Elastic interactions between junctions
 - Junctions have dislocation-like strain fields



Computed energies using empirical potentials predict that facets should grow arbitrarily large

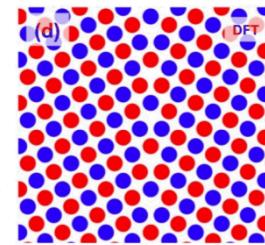
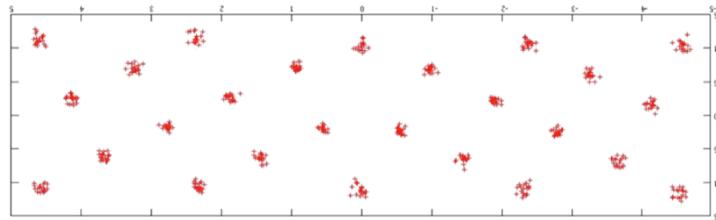
Faceting of $\Sigma 5$ grain boundary in Fe: Atomistic prediction for junction geometry



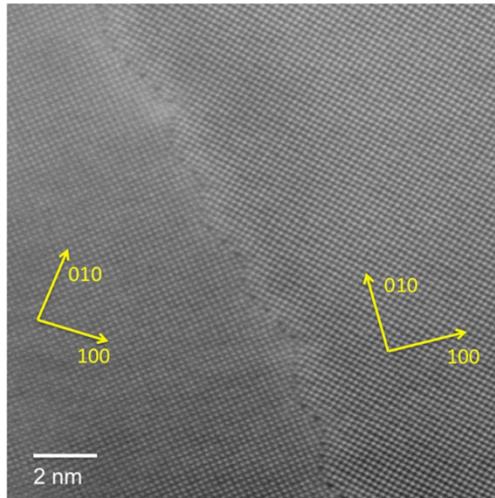
- Predicted structures of the junctions based on simulation of boundary using an empirical potential (Mendelev)
- Computed junction structure consistent with DFT calculations for $\Sigma 5$ (710) boundary which contains both of these junctions

Do Calculations and Experiment Agree?

Structures of individual facets **OK**

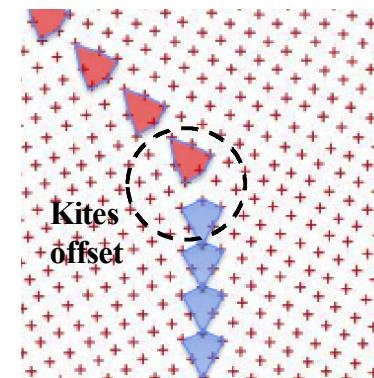


Equilibrium Facet Length **WRONG**



Junction Structure **WRONG**

Experimental Junctions
 $b=(1/5)(120)$ and $(1/5)(310)$



Relaxed Periodic
Atomistic Structure

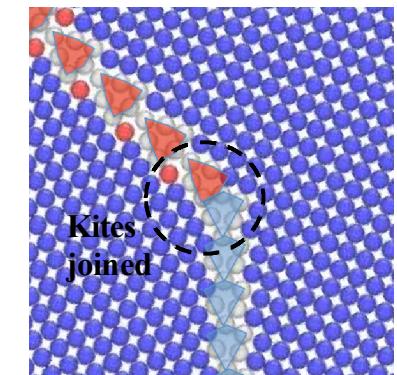
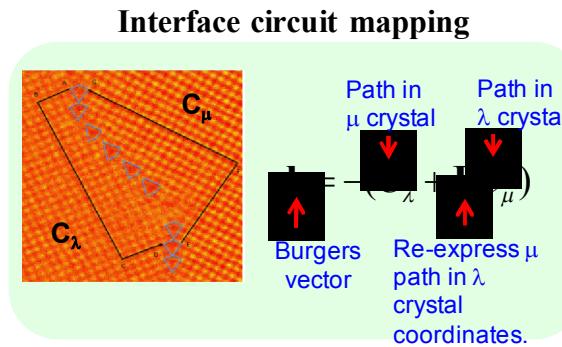
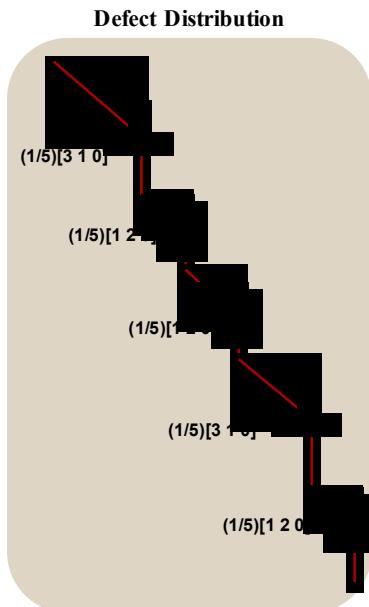


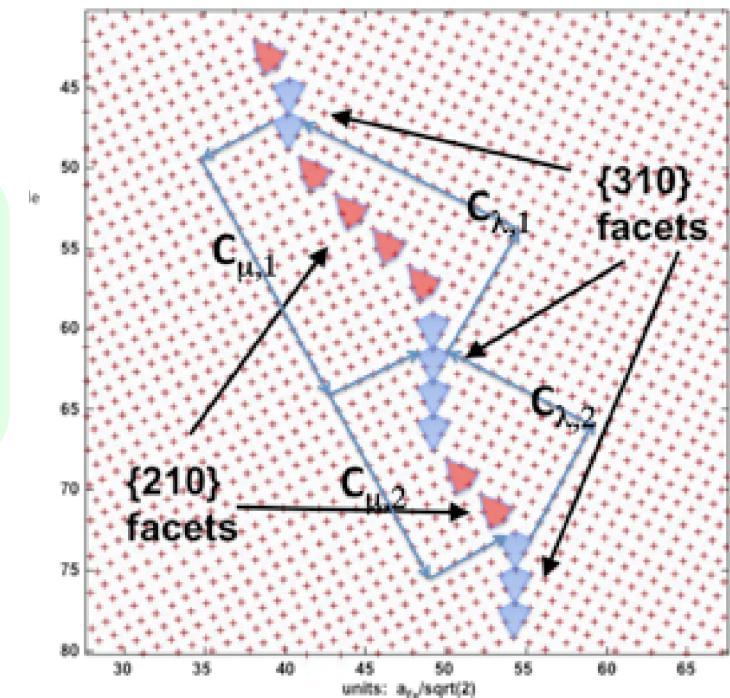
Figure 3

What went wrong!?!

- Simulations are Experiment are for **DIFFERENT STRUCTURES**
 - Simulation: **Ideal** $\Sigma 5$ misorientation
 - Experiment: **Real-world** $\Sigma 5$ misorientation
 - $\Delta\Theta = 2.4 \pm 0.8$



Interfacial dislocations required by Frank-Bilby equation



Secondary grain boundary dislocations convert computed junction to observed

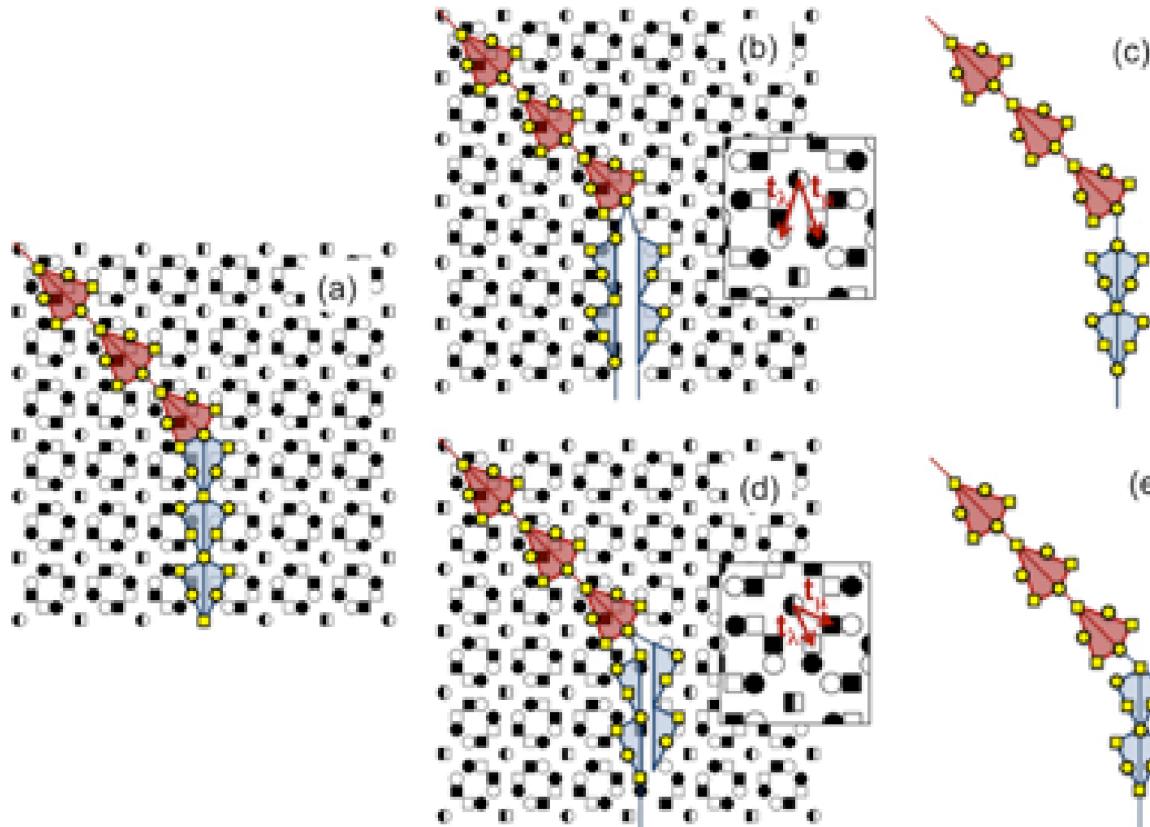
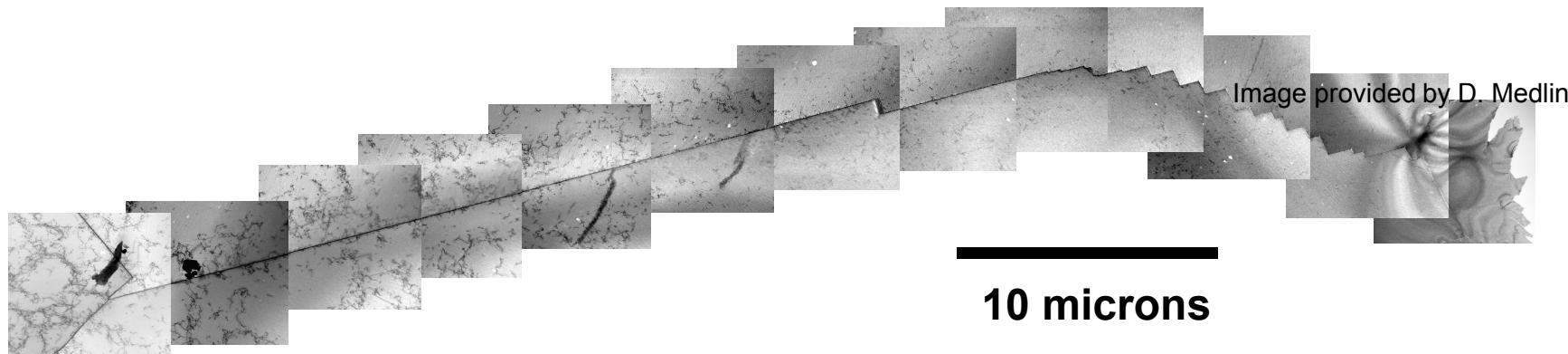


Figure 16

H in GB Engineered Materials

Grain boundary engineered materials offer a promising route to mitigate hydrogen embrittlement

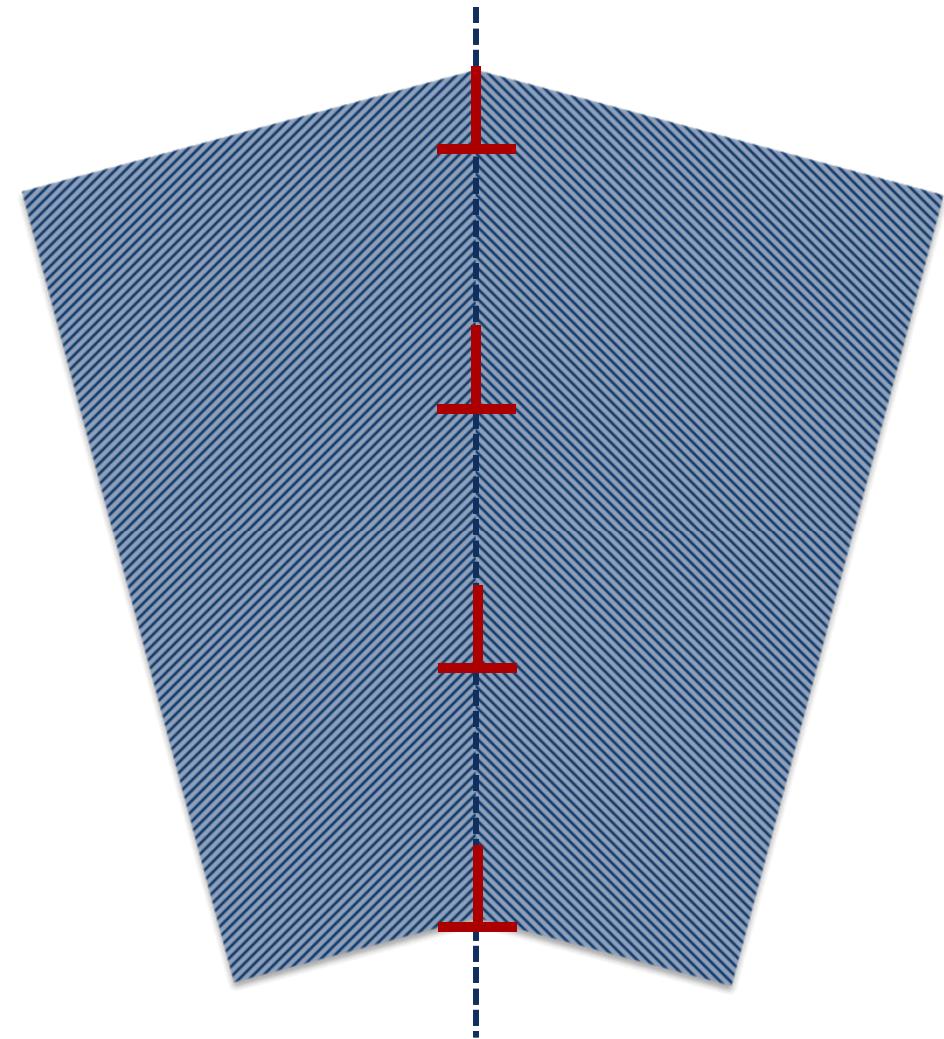
- Recent work by Bechtle[†], and Oudriess[‡], demonstrate the benefit of grain boundary engineered materials in reducing embrittlement
- While H segregation at ideal coherent twins is thought to be small, what is the role of interfacial defects for H interactions?



Misoriented Twin Grain Boundaries

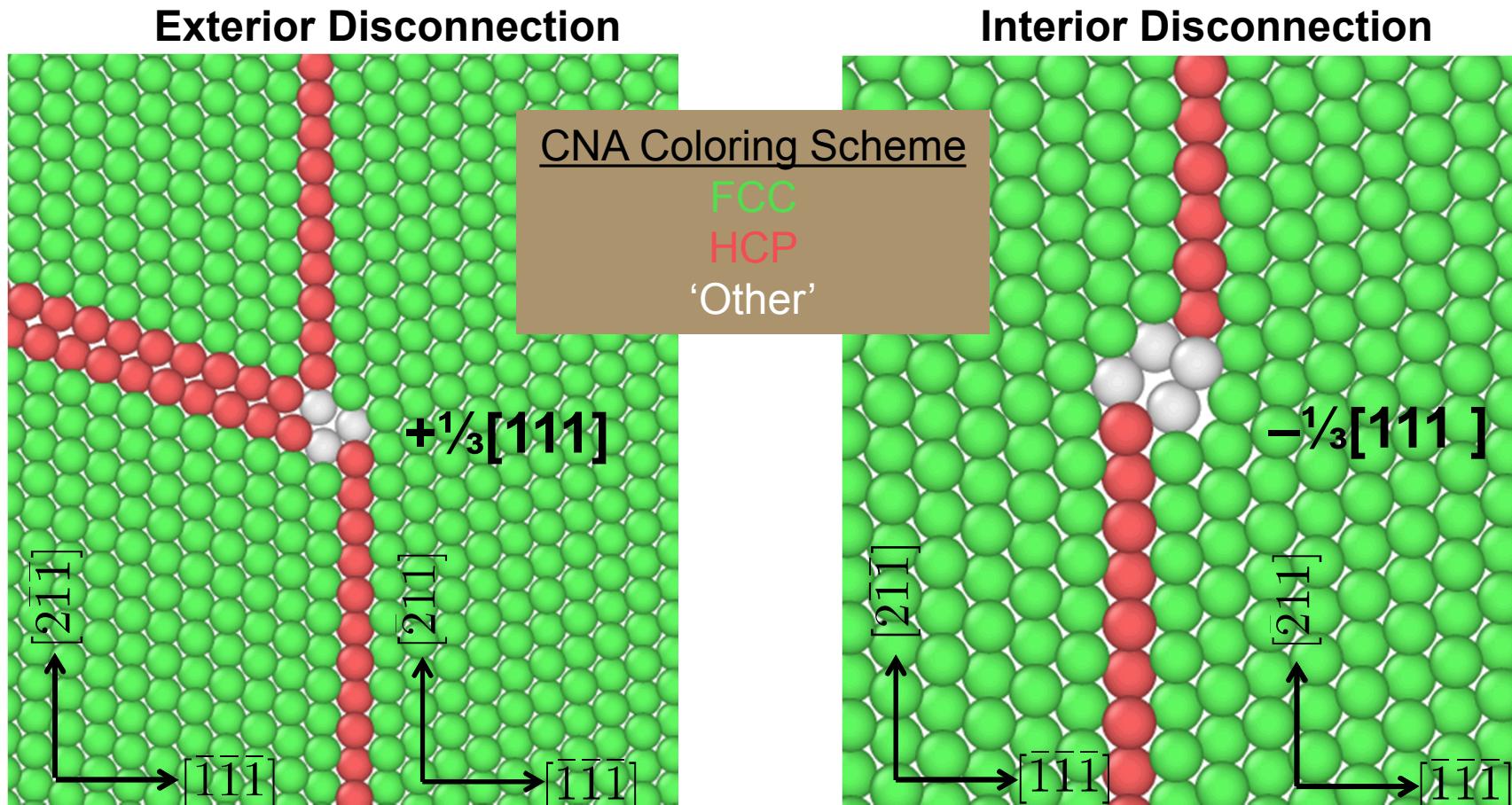
Definition: Misoriented GBs are produced by a symmetric rotation of grains about $(111)\langle 110 \rangle$ (coherent) twin $(-15^\circ < \theta < +15^\circ)$

- Misoriented GBs are generated by disconnections that come in two classes:
 1. Exterior
 2. Interior
- This terminology, due to Marquis & Medlin,[†] refers to the decomposition of the $\pm\frac{1}{3}\langle 111 \rangle$ disconnection.
 - *Exterior* disconnections disassociate and emit extended stacking faults
 - *Interior* disconnections retain compact core



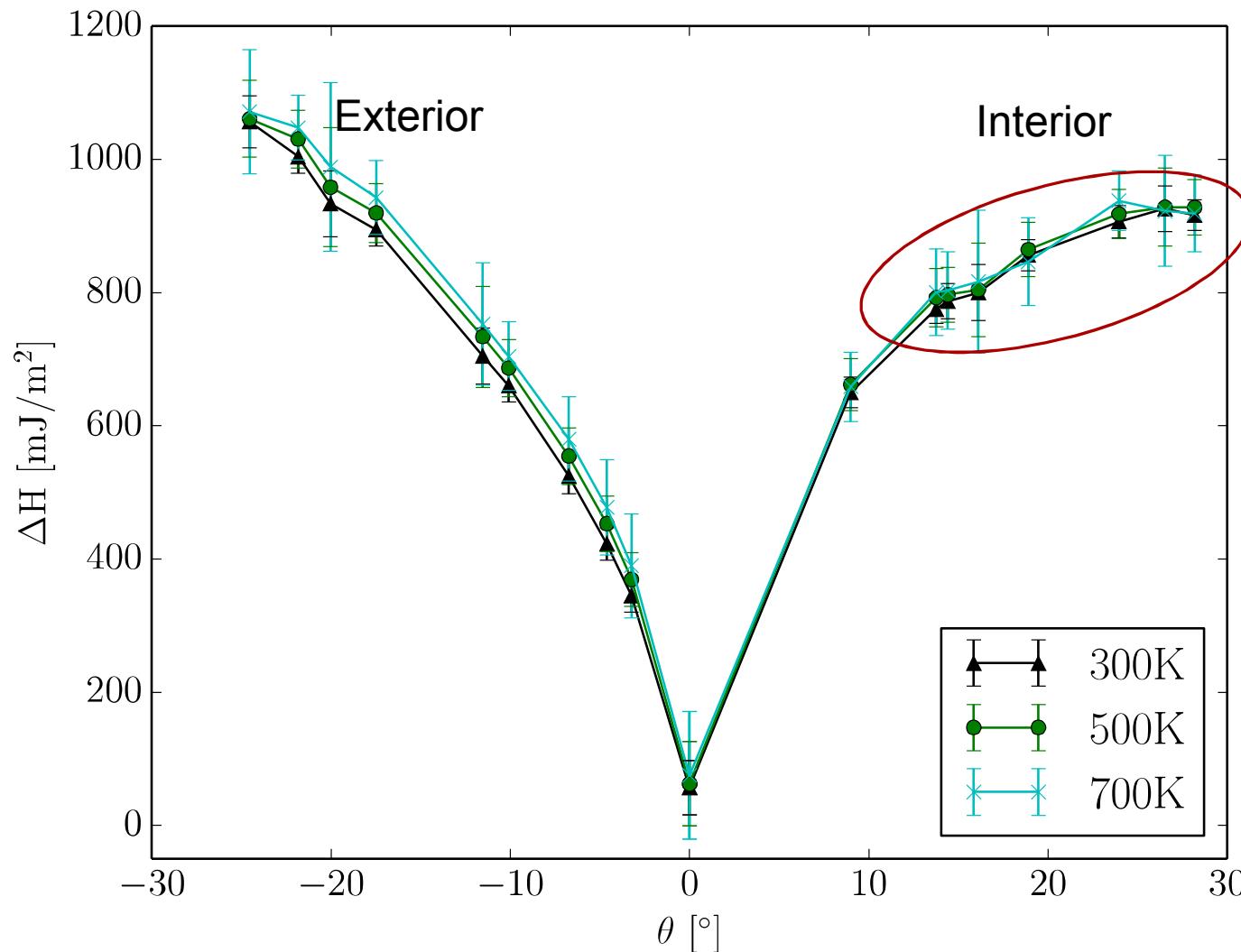
[†]Phil. Mag. Lett. 85 (8), 387–394, 2005

Fundamental Structural Difference Between $\pm\frac{1}{3}\langle 111 \rangle$ Disconnections



Superposition of these disconnections allows for the rotation of the grains or GB plane

Asymmetric Enthalpy Dependence



Temperature Dependence of GB Structure

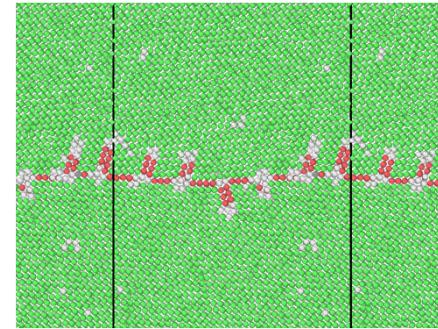
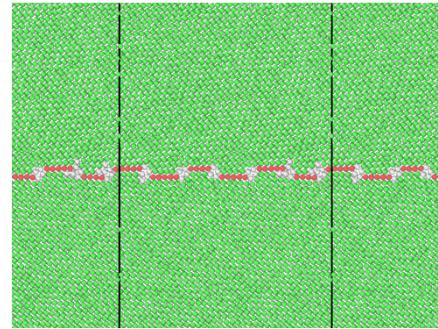
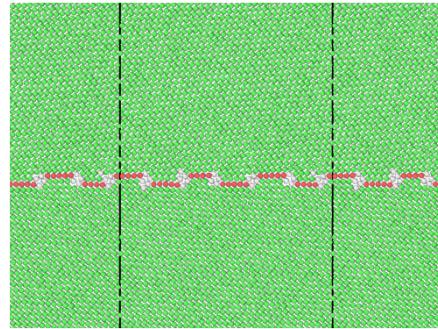
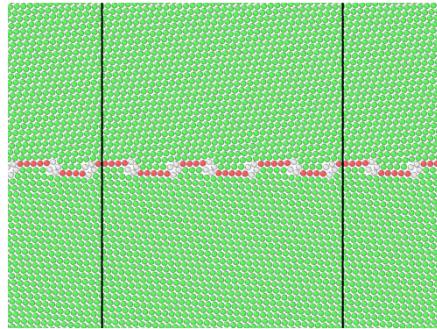
Interior ($\theta = +9.0^\circ$)

0K

300K

500K

700K



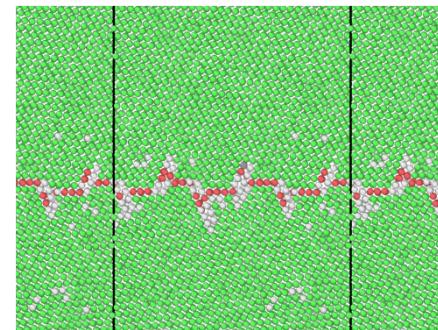
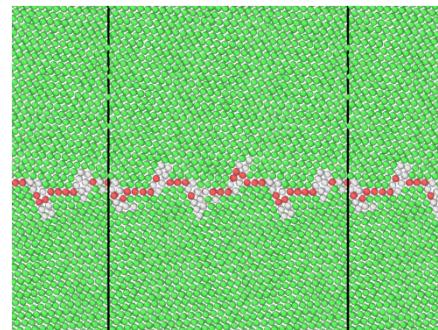
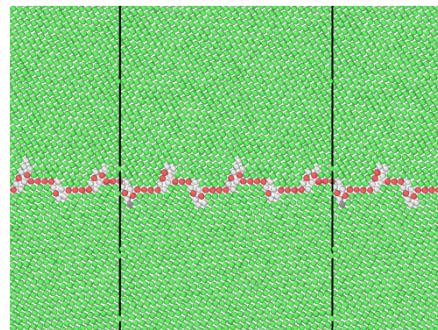
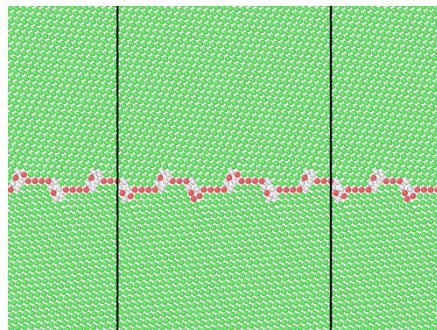
Exterior ($\theta = -10.1^\circ$)

0K

300K

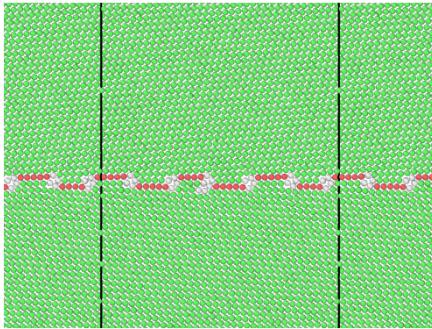
500K

700K

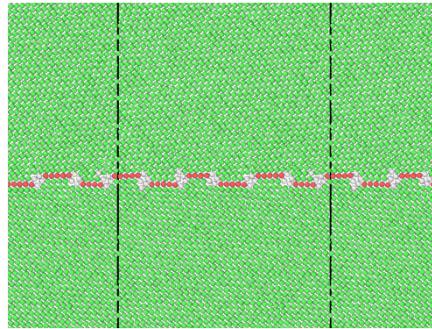


Stability of Extended Stacking Faults in Interior GBs

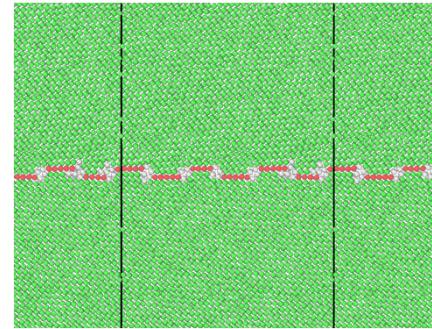
0K→100K



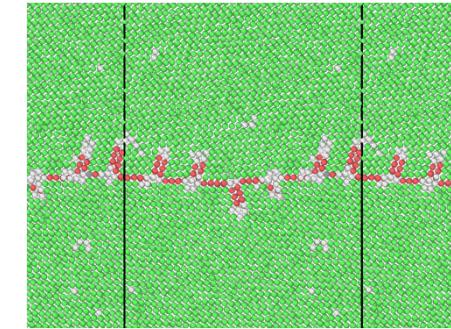
0K→300K



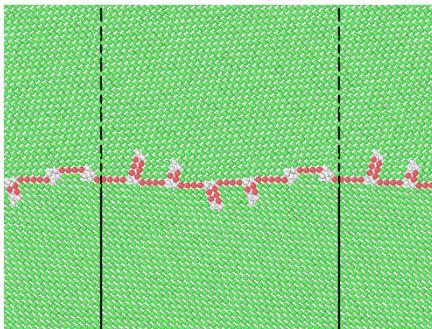
0K→500K



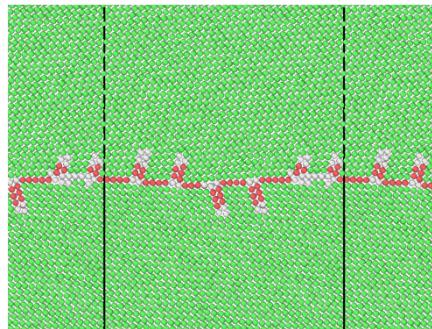
0K→700K



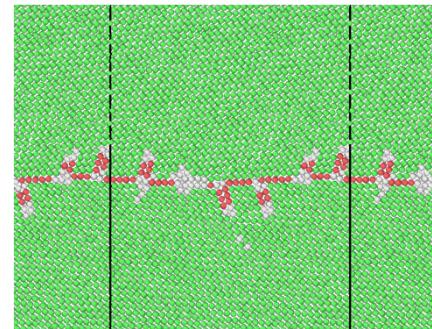
700K→100K



700K→300K



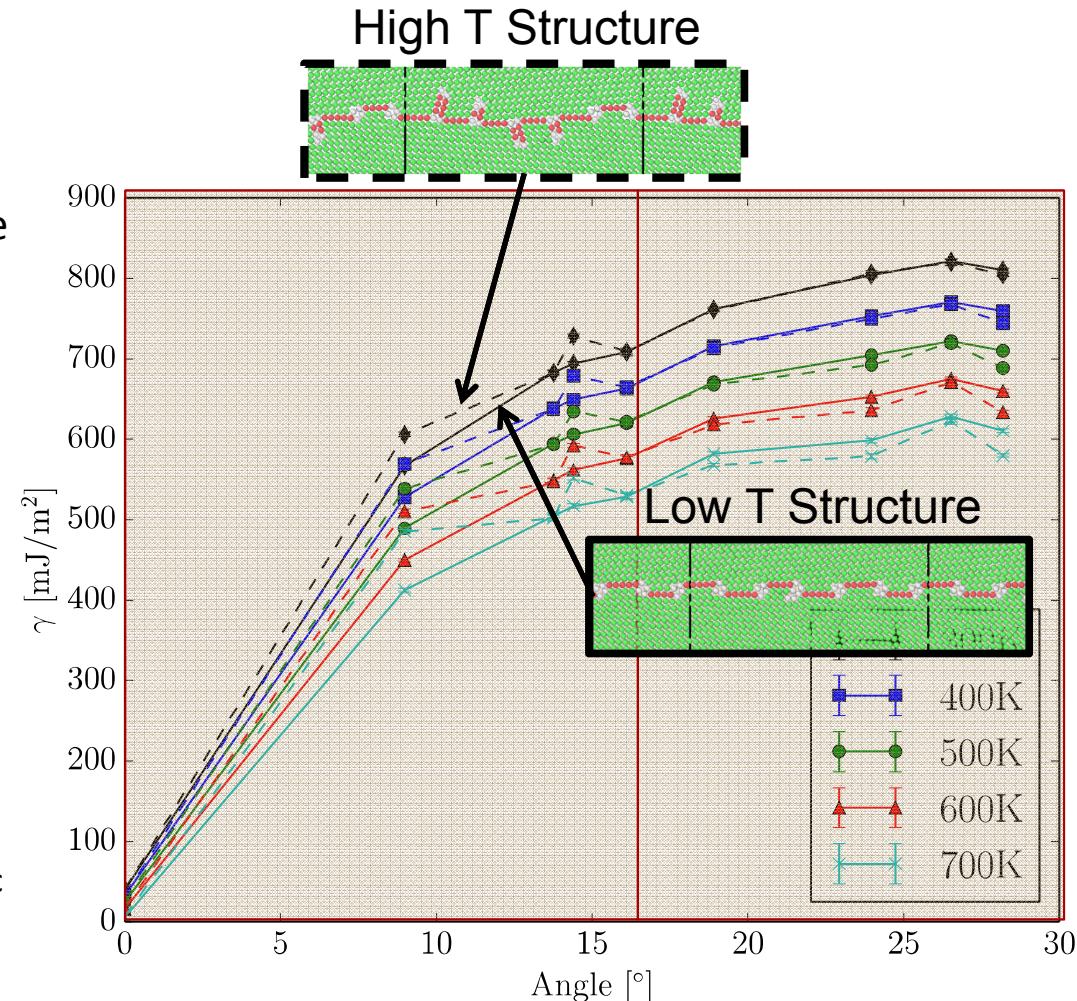
700K→500K



Faults are frozen in upon cooling, but are they stable?

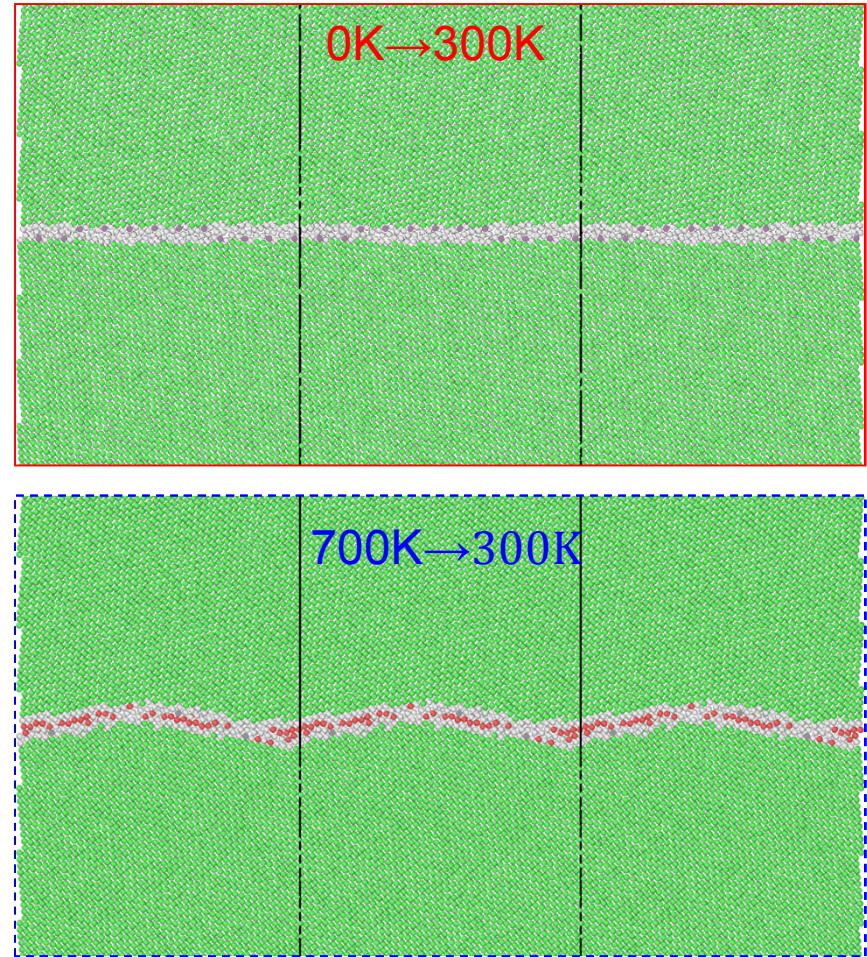
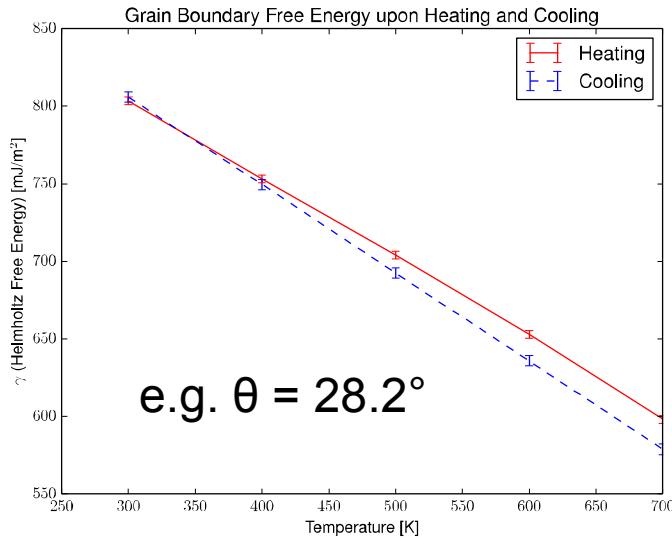
Compact Core Structure of Interior Disconnections Favored At Low Misorientation Angles

- $\theta \leq 16^\circ$
 - Low temperature structure of interior disconnections (compact cores) favored
- $\theta > 16^\circ$
 - High temperature structure favored
 - Change in character of boundaries
- Transition temperature for favorability of high temperature structures is unknown
 - Thermodynamic integration is used to calculate free energies above 300K
 - Change from low to high temperature form occurs at 500–600K in MD simulations
 - High temperature structures could form during realistic processing conditions



Faceting occurs at larger misorientation angles

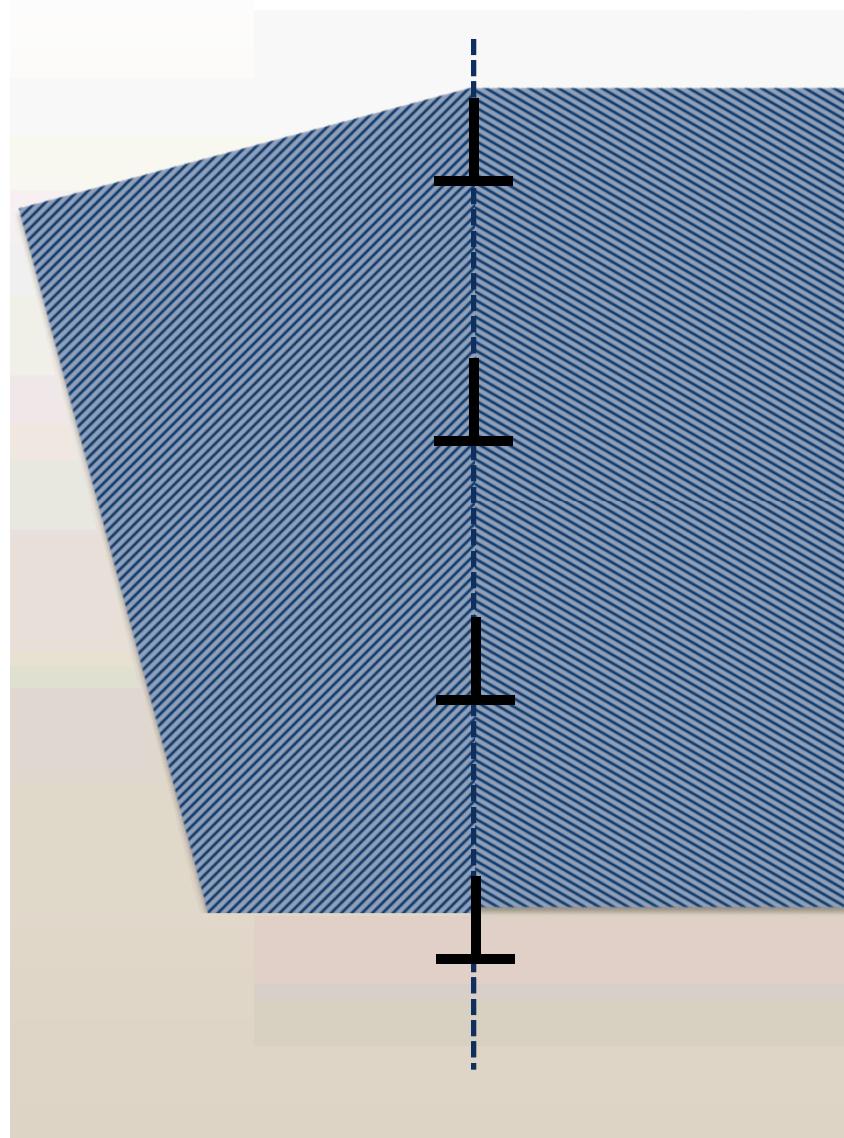
- Higher angle boundaries facet, which decreases their free energy
- Facets lie along (111) planes



Inclined Twin Grain Boundaries

Definition: Inclined GBs are produced by a change of grain boundary plane orientation

Inclination (Φ) range:
 0° $(111) \langle 110 \rangle$ (coherent)
to
 90° $(112) \langle 110 \rangle$ (lateral)



Inclined Twin Boundary Structure

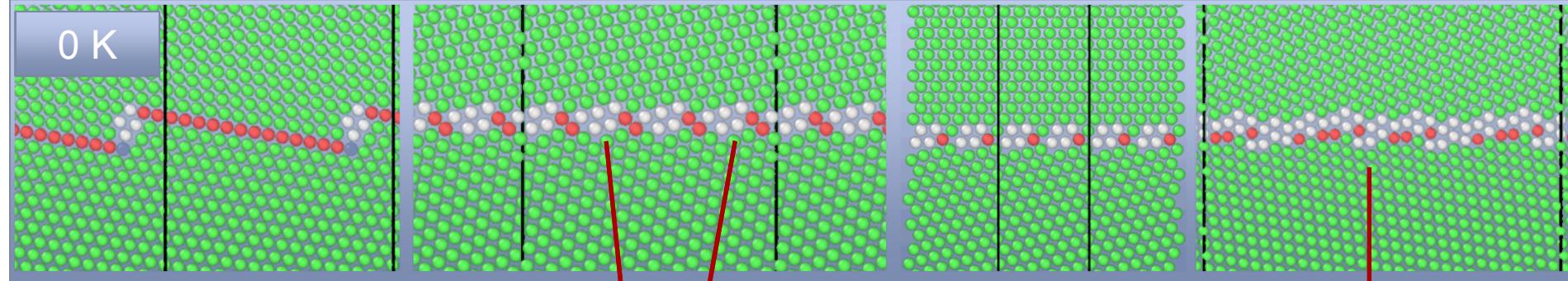
10.02°

43.31°

54.74°

81.95°

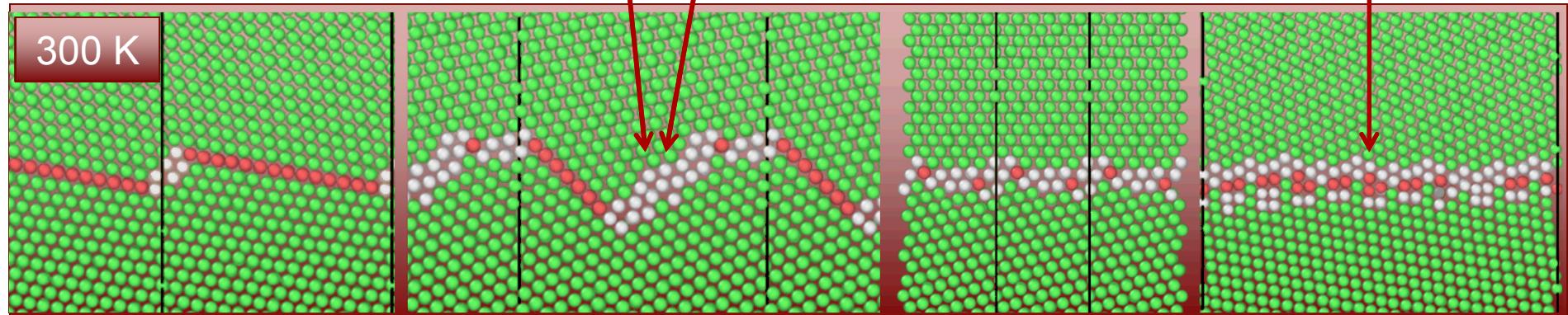
0 K



300 K

(112) units merge

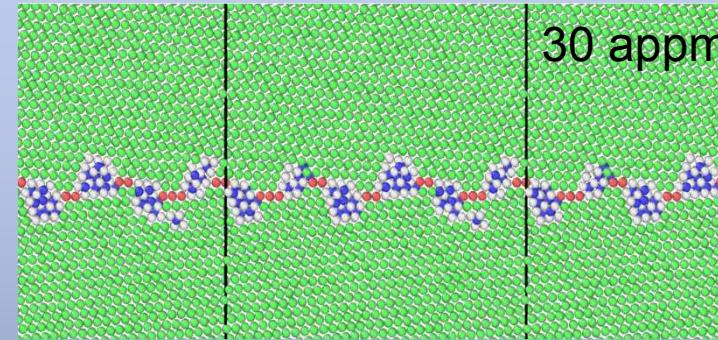
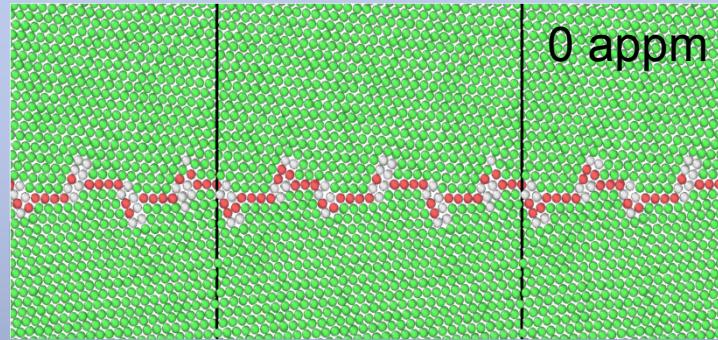
Stacking-faults grow



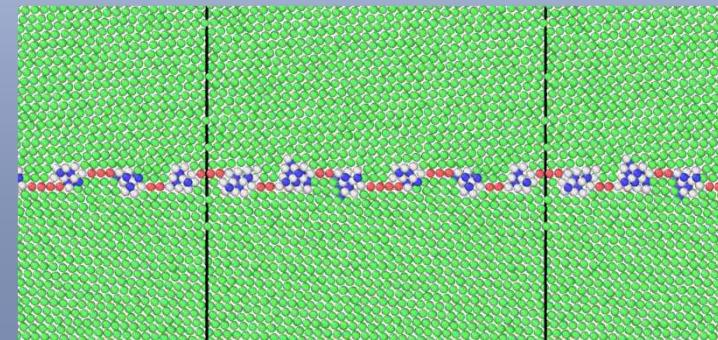
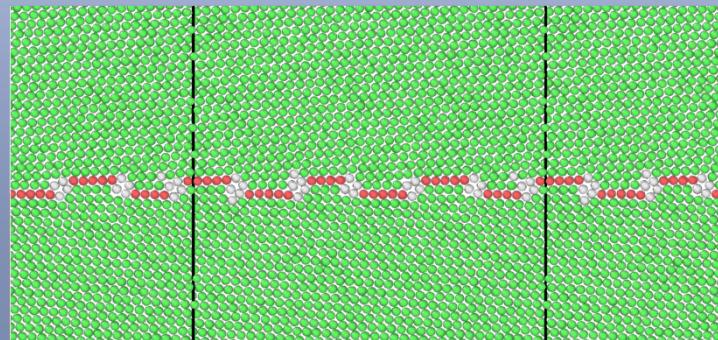
Hydrogen Segregates Defect Cores

Misoriented Twin

$\theta = -5.05^\circ$

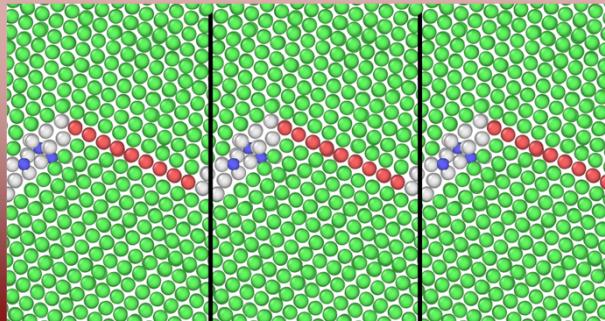
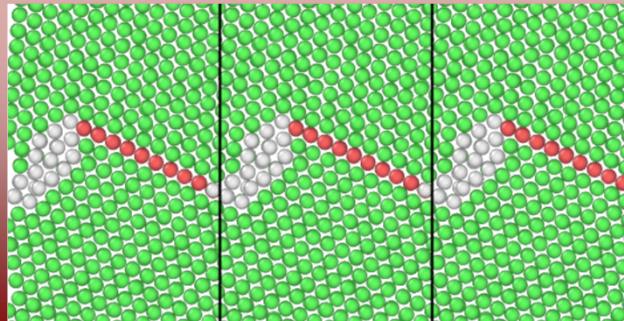


$\theta = 4.49^\circ$



Inclined Twin

$\phi = 25.24^\circ$



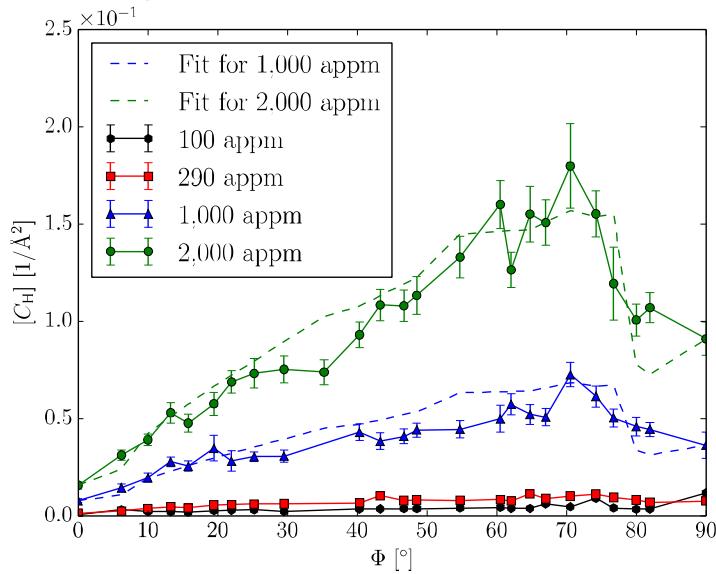
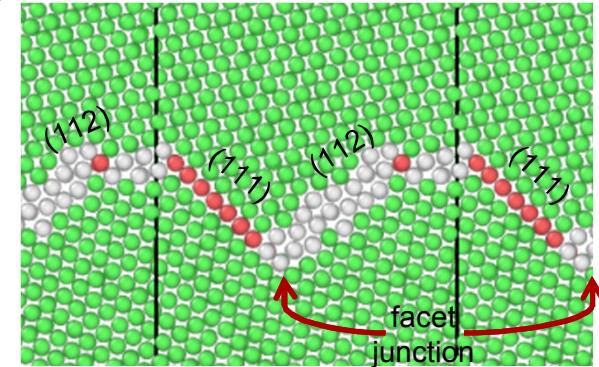
30

Analytical Model – Defect Densities

- Facet description of inclined GB employed
 - Assume H segregates to junctions and twin facets
 - $(C_H)_{LTB}$: Concentration on (112) facet
 - $(C_H)_{CTB}$: Concentration on (111) facet
 - $(C_H)_S$: Concentration at junction

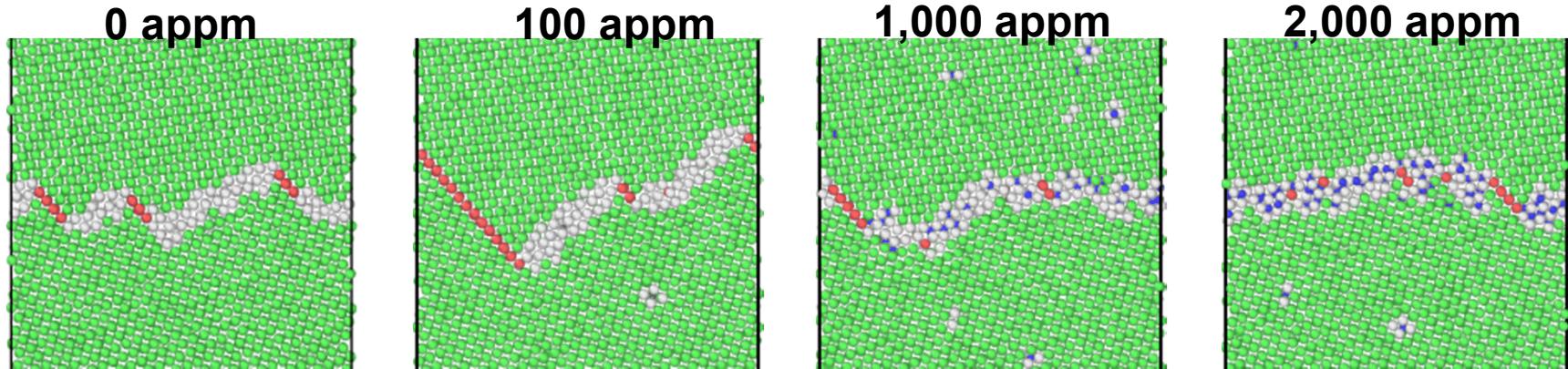
$$[C_H] = \frac{1}{A} \left\{ N_s (C_H)_S + N_{CTB} (C_H)_{CTB} + N_{LTB} (C_H)_{LTB} \right\}$$

junction
(111) Twin
(112) Twin



Φ [°]	Width [Å]	N_s	N_{CTB}	N_{LTB}	[C_H] at 500 K [$10^{-3}/\text{\AA}^2$]			
					100	290	1,000	2,000
0.00	17.384	0	8	0	0.5730	1.2362	7.6976	15.6347
6.21	1711.121	2	24	0	3.0848	2.3978	14.3273	31.1476
10.02	531.527	1	16	0	2.1946	3.7843	19.5592	39.0457
13.26	403.315	1	12	0	2.0831	4.5375	27.7196	52.9100
15.79	679.930	2	20	0	1.7180	4.0819	25.4872	47.6061
19.47	555.160	2	16	0	2.5771	5.4291	34.6411	57.5481
35.26	320.522	2	8	0	—	—	42.8791	73.8064
40.32	858.058	6	12	0	3.4301	6.6104	42.8791	93.0184
48.53	987.915	8	12	0	3.6199	8.1369	44.0075	107.9027
54.74	1063.052	10	20	0	3.9765	7.8516	44.3655	132.8865
64.76	613.753	6	8	0	3.7710	11.3393	52.0809	155.0783
70.53	392.558	4	8	0	4.5404	10.0850	72.5827	179.8277
74.21	480.783	5	4	0	8.9526	11.1193	61.4026	155.1813
76.74	570.373	6	4	0	3.9797	9.4398	50.2594	119.3123
79.98	751.691	4	2	0	3.3890	8.1005	45.8036	100.5943
81.95	934.475	4	4	1	3.4500	6.8524	44.4450	106.9386
90.00	185.053	0	0	2	11.7221	7.4934	36.2329	90.9561

Structure and H Segregation are coupled

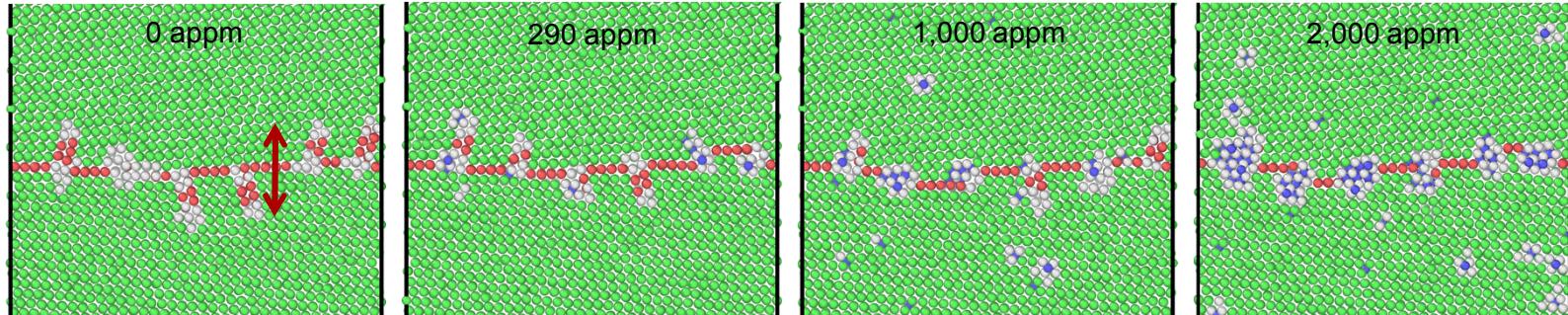


Inclined Boundary at 500 K

Increasing H concentration changes boundary structure with increasing concentration!

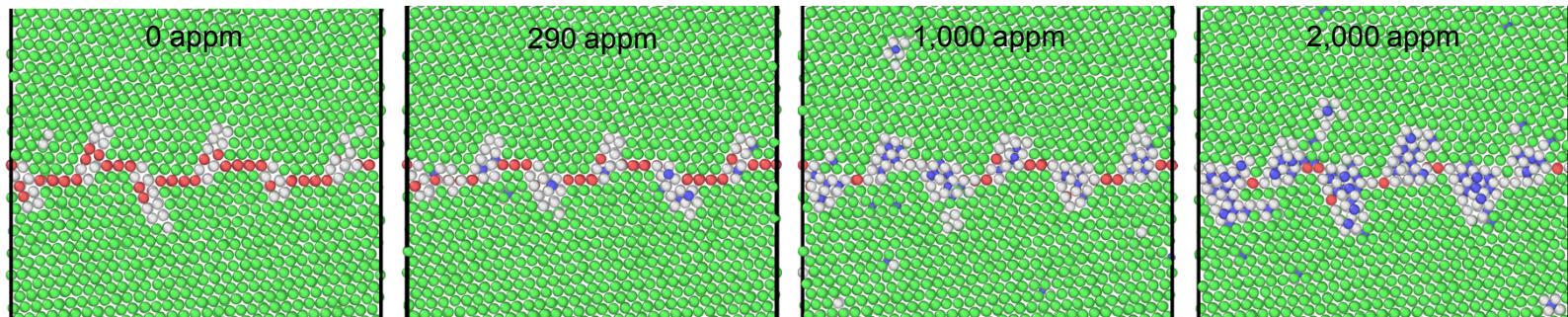
H segregation also changes in structure in some misoriented boundaries

$\theta = 4.49^\circ$



Stacking fault length reduced by increasing H concentration

$\theta = -5.05^\circ$



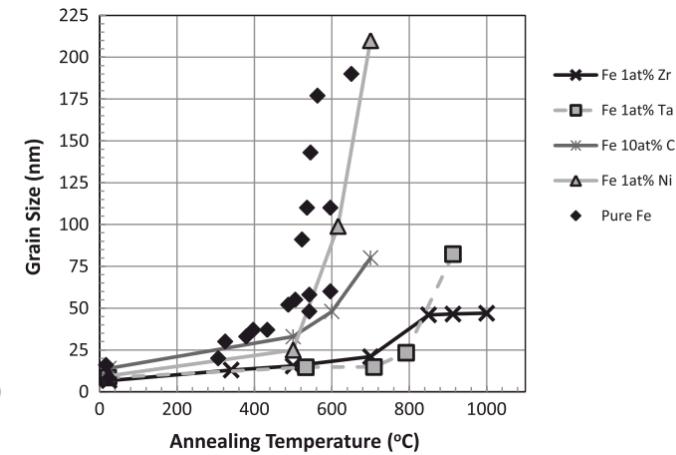
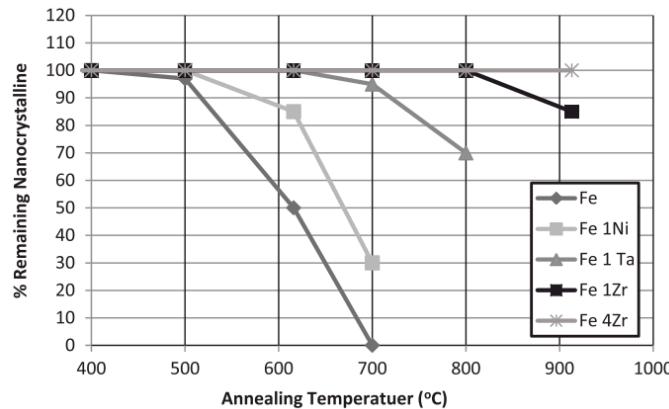
Alloying can stabilize nanocrystalline grain structures

- Sluggish growth dynamics: W – 20at. % Ti

T. Chookajorn et al., *Science*. **337** (2012)
T. Chookajorn et al., *Acta Mater.* **73** (2014)

▪ Alloy selection in Fe-based system

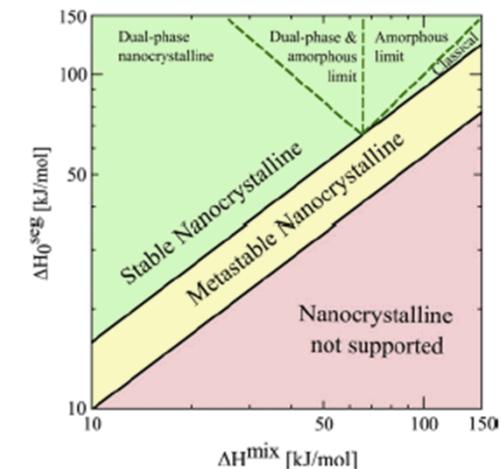
- Additions of Ta, Cr, Ni and Zr



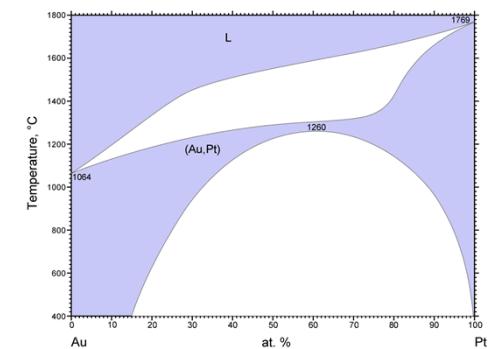
K. Darling et al., Mat. Sci. Eng. A 528 (2011)

Alloy stabilization of nanocrystals could have multiple origins

- Grain boundary segregation can strongly reduce interfacial free energy, a driving force for grain growth
 - Schuh and co-workers
- Solute segregation to boundaries may reduce the mobility of the boundaries
- Precipitation of second phase particles can act as pinning sites
 - Zener pinning
- Pt-Au is being examined as a model system to elucidate the relative importance of these effects
 - O'Brien has developed DFT-based EAM potentials for Pt-Au
 - Experimental observations of this system are in progress
 - No oxide formation & high purity

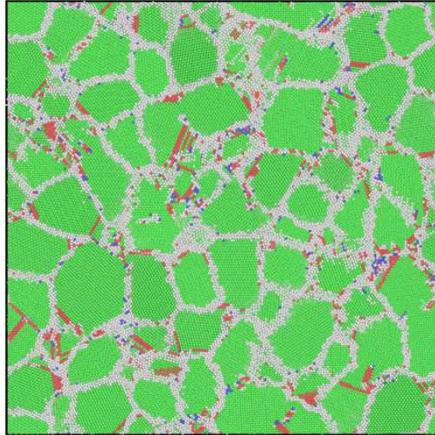


Murdoch, Schuh, Acta Mat 61 (2013) 2121-2132



Higher Au content in Pt-Au appears to slow grain growth

0.0 nsec

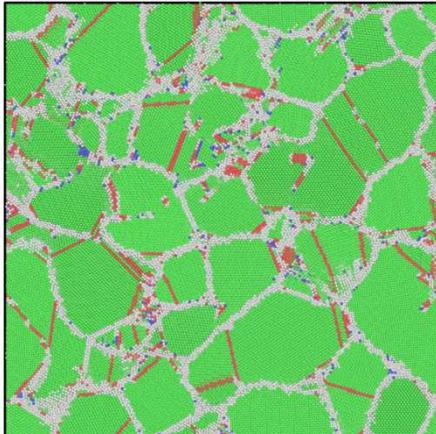


FCC
HCP
BCC
Other

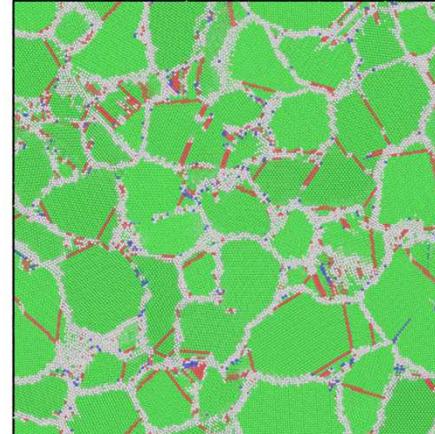
- Annealed Voronoi grain structure at 1000K
 - 600 grains
 - Computational cell size: 39 nm
- Monte Carlo simulations employed to determine initial distribution of Au

Pure Pt

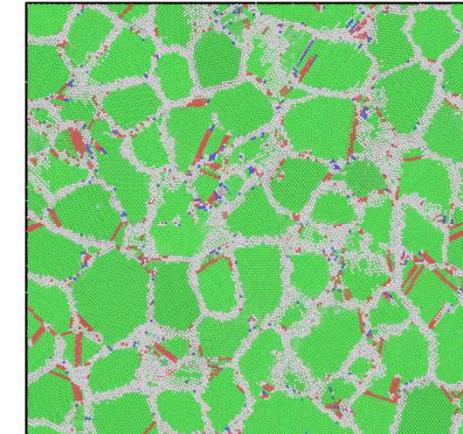
2.0 nsec



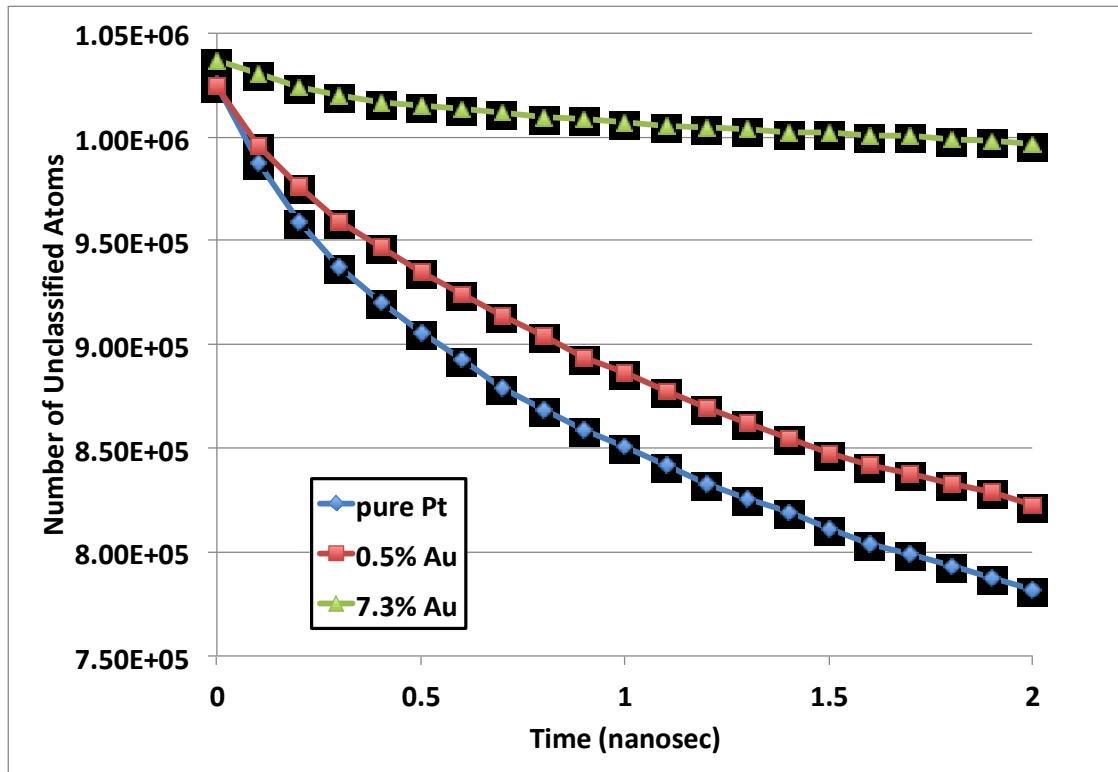
0.5% Au



7.3% Au

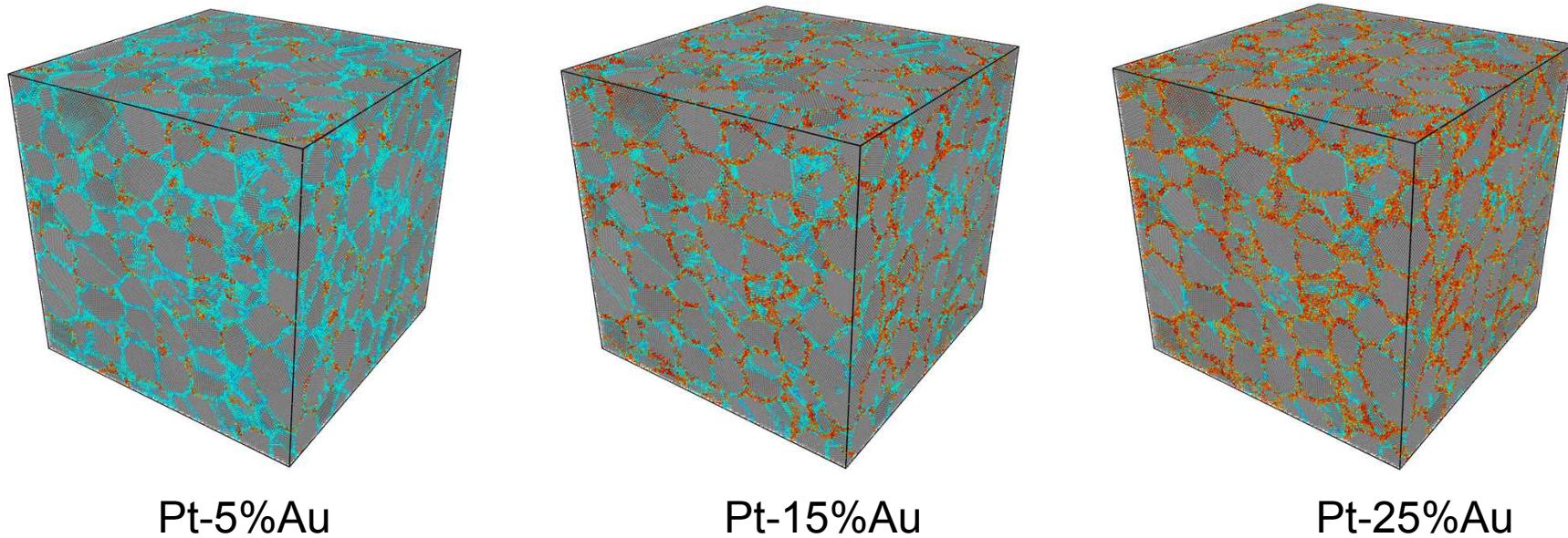


Simple analysis confirms impression: High Au slows grain growth



- Common Neighbor Analysis classifies atoms as FCC, HCP, BCC, and 'other'
- Use number of unclassified atoms as rough measure of grain boundary area

Au is located almost exclusively at boundaries and triple junctions

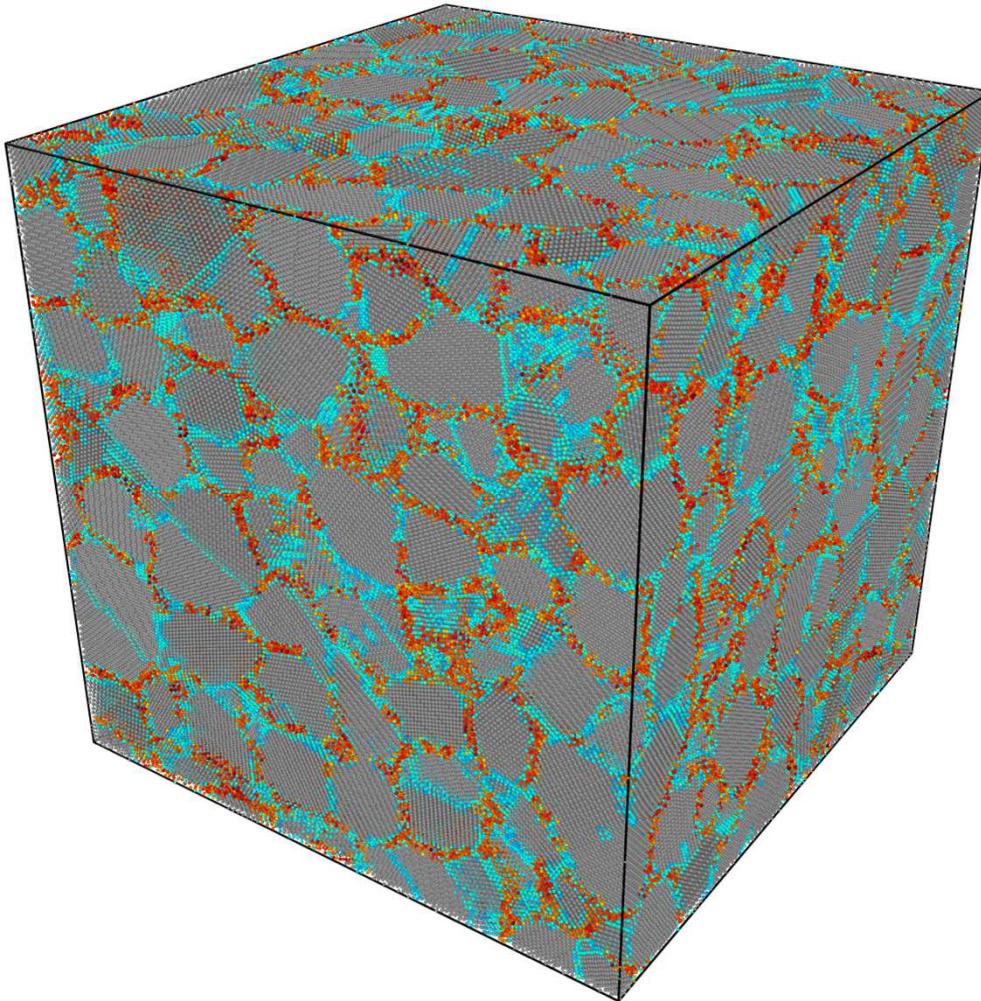


Equilibrium Monte Carlo snapshots at 500° C

- Bulk Au concentrations: $\sim 0.1\%$
- Coloring: combination of atomic element and centrosymmetry parameter
 - Gray: bulk Pt
 - Cyan: Pt interfaces
 - Orange \rightarrow Red: Au

Heterogeneous Segregation

*“ all grain boundaries are **NOT** created equal”*

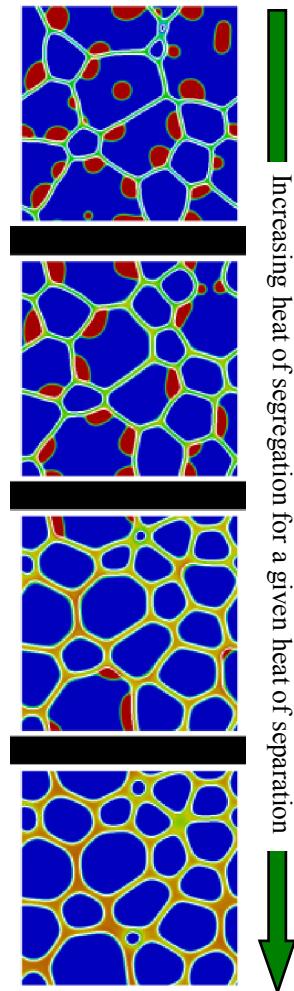


Does it matter that only some of the boundaries are segregated and so probably pinned?

- Pinning a subset of boundaries should be sufficient to stabilize the grain size
 - Holm, Foiles, *Science* 328, 1138 (2010)

Future work on Pt-Au nanocrystals

- Bi-crystal simulations of representative boundaries
 - Degree of segregation for comparable bulk compositions
 - Mobility of grain boundaries
 - Segregated & unsegregated moving insolute field
- Use results to motivate introduction of varying boundary properties into phase field simulations
 - Bare energy, heat of segregation
 - Mobility with and without segregants
- Comparison with experiment
 - *“Coming soon to a conference near you!”*



Molecular Dynamics:

What could possibly go wrong?



- Deviations from Born-Oppenheimer Approximation
 - Highly dynamic events, charge states, ...
- Inadequate Interatomic Potentials
 - Imperfect knowledge of Born-Oppenheimer surface
 - Errors in experimental data
 - Deficiencies of *ab initio* database
 - Assumed potential form – model form error
 - Transferability or lack thereof
- Quantum Mechanical Effects
 - Zero-point energies of light elements
 - Debye Temperature often above room temperature
- Limited time scales
 - High rates and high driving forces
 - Infrequent events
 - Sampling errors – metastable states
- Structural approximations
 - Where are the atoms, *really*?
 - Simplification of geometries
 - Boundary conditions
- Multi-component systems
 - Composition and structure coupled
 - Compositional variation - equilibrium or kinetic?
- Information extraction for higher-scale models
 - Millions of coordinates -> 'Physics'
 - Identification of dominant effects

Issues encountered in this talk

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