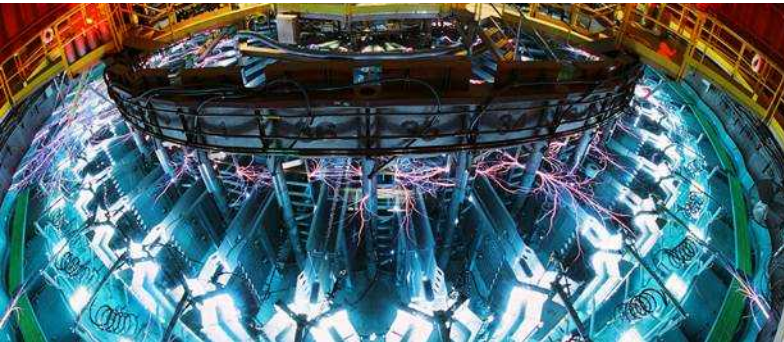


*Exceptional service in the national interest*



# Silicon Anode Gas Generation Questions



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Josh Lamb, Eric Allcorn, Chris Orendorff

# ICL During Formation in Model System Sandia National Laboratories

Electrolyte: EC:DEC 3:7 (wt%) 30 wt% FEC, 1.2 M LiPF<sub>6</sub>

Scan rate: 0.2 mV/s

cut off voltage: 2 V – 0.005 V

Cycle #: 11

Sample area: ~0.33 cm<sup>2</sup>

Charge consumption  
[mC/cm<sup>2</sup>]

Cycle 1

total	-90.53
cathodic	-112.34
anodic	21.81

2-0.29 V	-64.07
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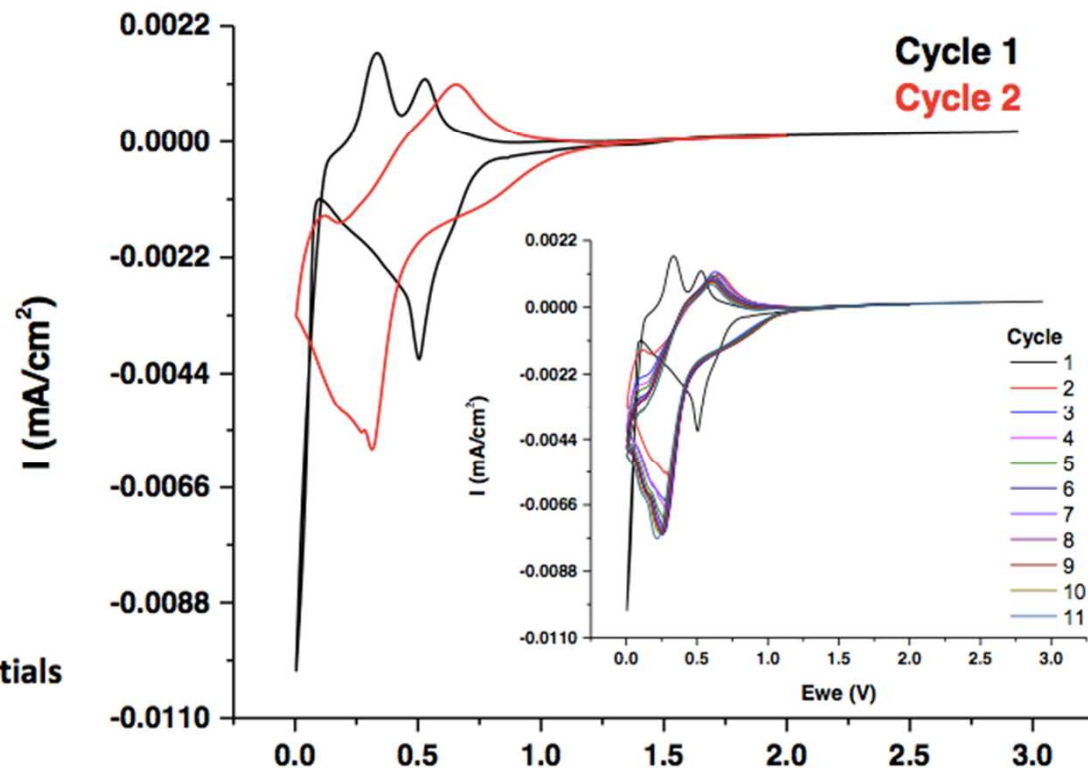
0.29-0.2 V	-45.51
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0.2-2V	18.96
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**No surface passivation.**

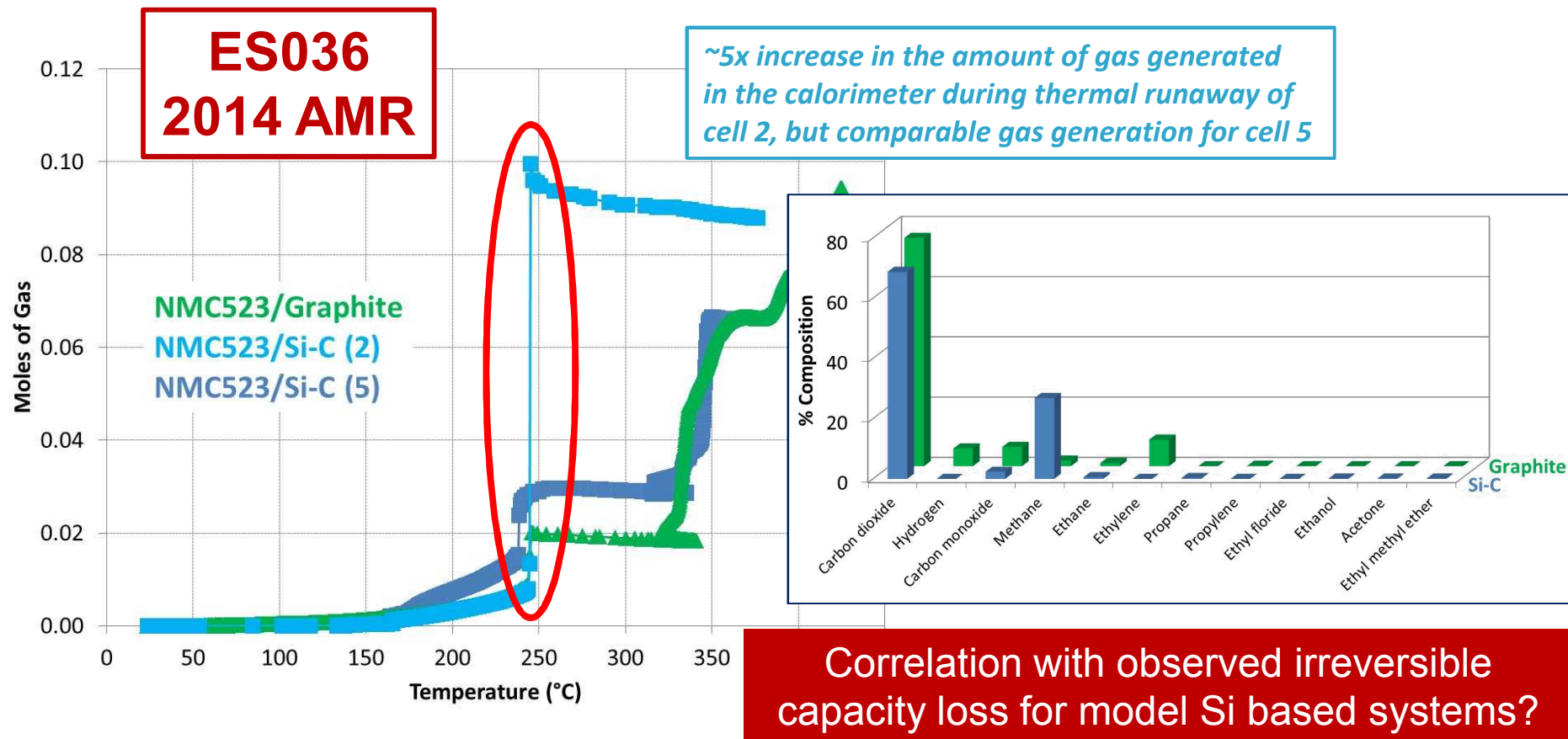
**Reduction shifted to lower potentials**

**Possible scenarios:**



Significant losses observed during formation cycling in model systems (copper and silicon). Electrochemical reduction of binder/electrolyte? Can this be correlated to gas generation seen in full electrode ensembles?

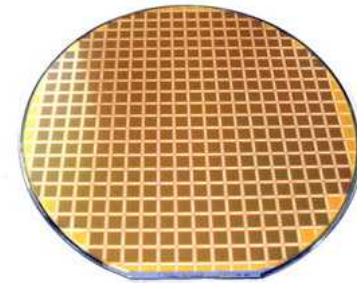
# Abuse Response of Silicon Anodes



*Difference in gas generation attributed to the differences in surface reactivity and surface products generated at the anode/electrolyte interface*

# Approach

1. Spin-coat binder and carbon phases on both Cu and p-Si wafer substrates.
2. Carry out electrochemical measurements of model thin-film electrodes in baseline electrolyte(s).
3. Ex situ evaluations for gas generation using DSC/TGA and grab samples during electrochemical and abuse evaluations.
4. Perform similar treatment for electrode ensembles to determine correlation between model system and industry relevant electrodes.



Understand  
relationship between  
model system and  
multi-particle  
ensemble behavior?

