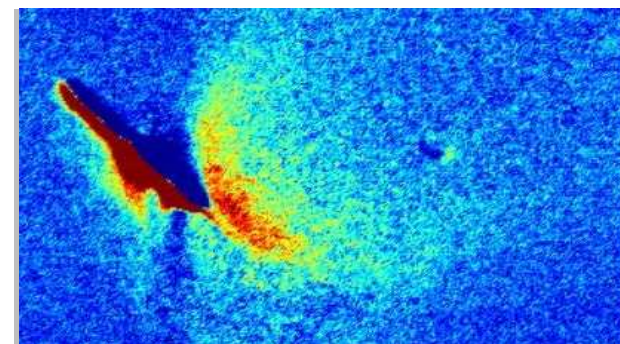
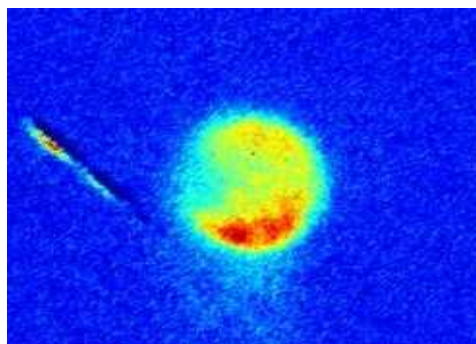
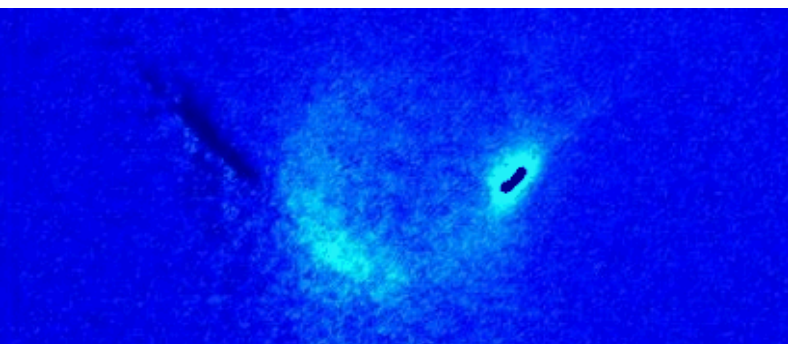


Exceptional service in the national interest



Inelastic Collisions of NO and NO₂

Casey Sugie
Mills College
Summer 2015

Crossed-beam velocity map ion imaging with NO and NO₂

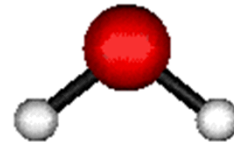
- observing collisions between NO and NO₂ molecules to study the energy transferred during the collision
- laser beams used to excite and ionize state selected NO that has gained internal energy from a collision
- detect NO⁺ ions in a velocity map image

How are molecules energetically different than atoms?

Translational Energy



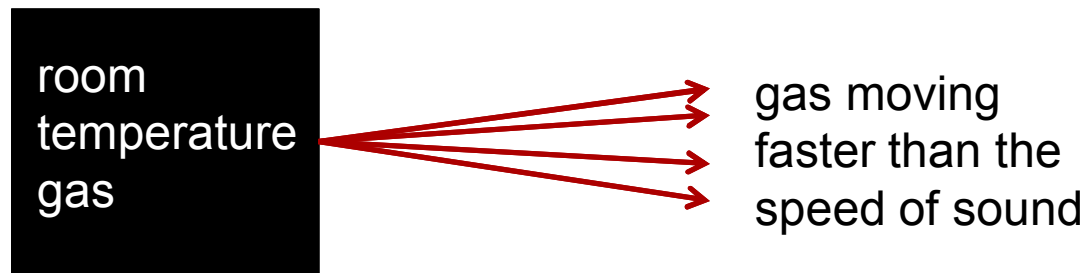
Internal Energy



How does this effect energy transfer in crossed-beam collisions?

- if colliding beam is an atom, the atom can only gain or lose translational energy
- if colliding beam is a molecule, it can gain or lose internal and/or translational energy

Creating internally cold molecules through supersonic expansion



- essentially all internal energy is converted to translational energy
- all gases coming out of this system will have the same kinetic energy

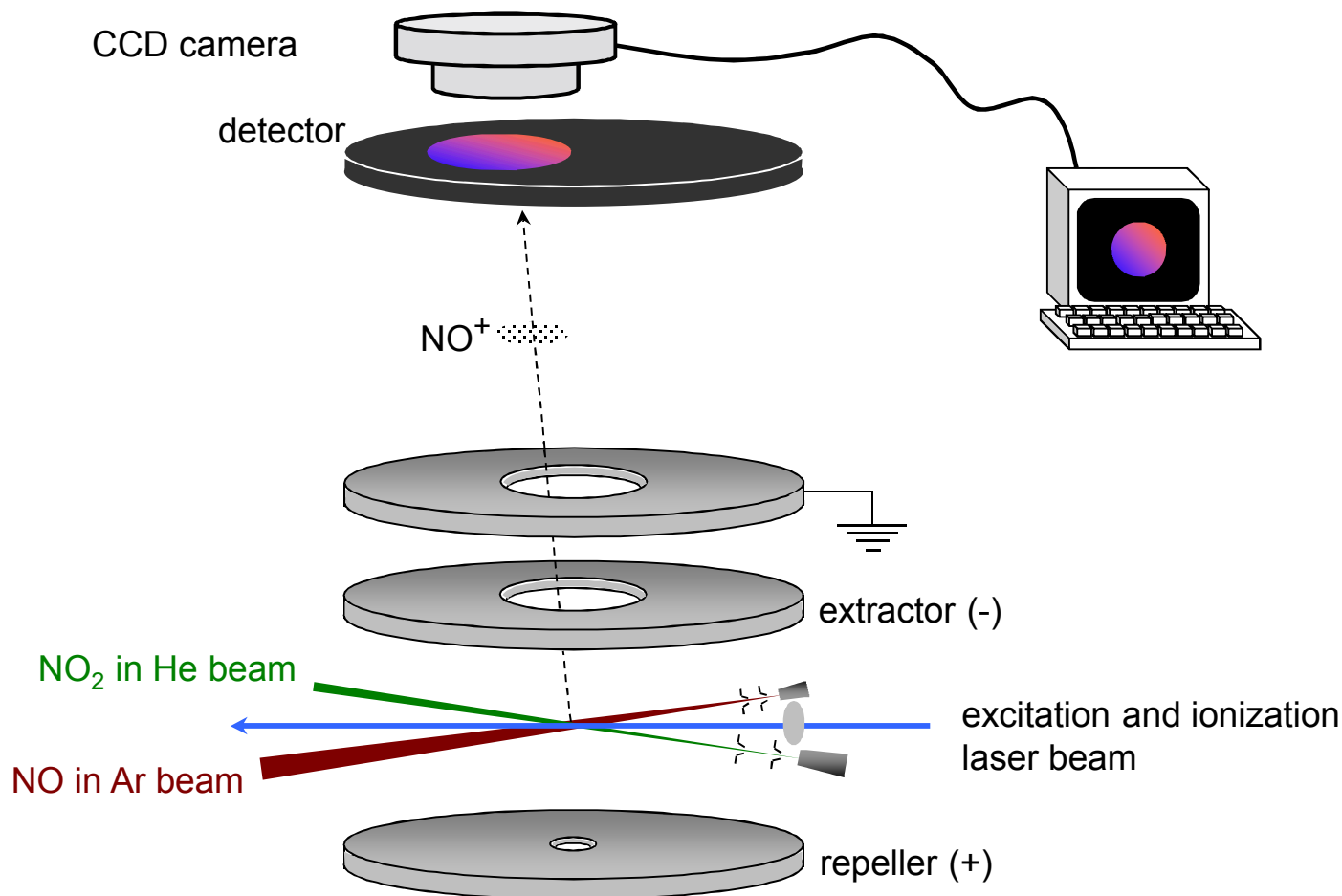
$$KE = \frac{1}{2}mv^2$$

- lighter gas molecules will travel faster

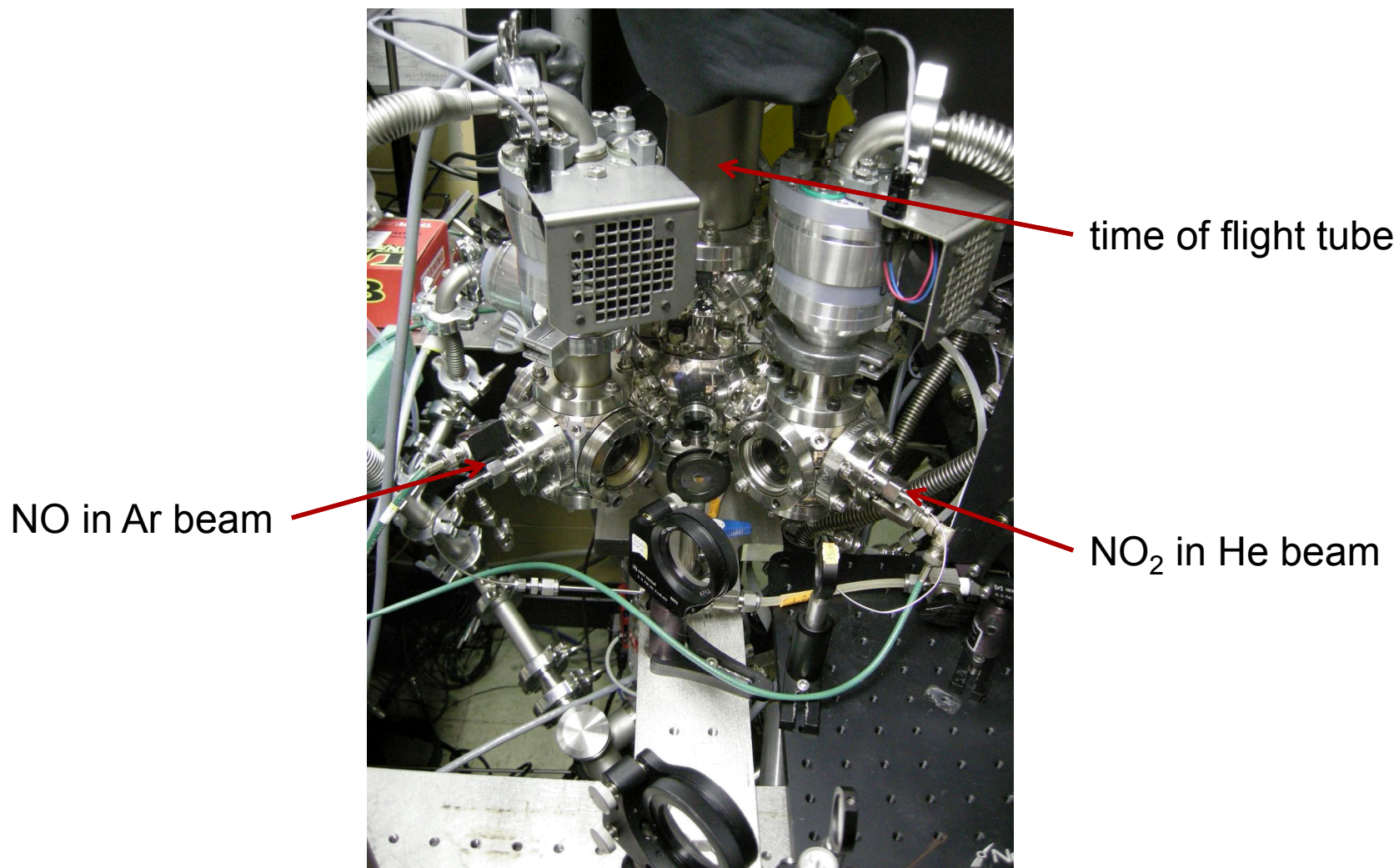
Comparing Ar and NO₂ as the colliding molecule

- masses of Ar and NO₂ are about the same (~40 and 46amu), but beam is 10% NO₂ in He, making the average mass 6.94amu
- lighter average mass of NO₂ beam means a higher velocity which makes a lot more energy available for the collision
- energy available from Ar is about 449cm⁻¹
- energy available from the NO₂ is about 1393cm⁻¹
- NO₂ has more than 3x the energy available for a collision!!!

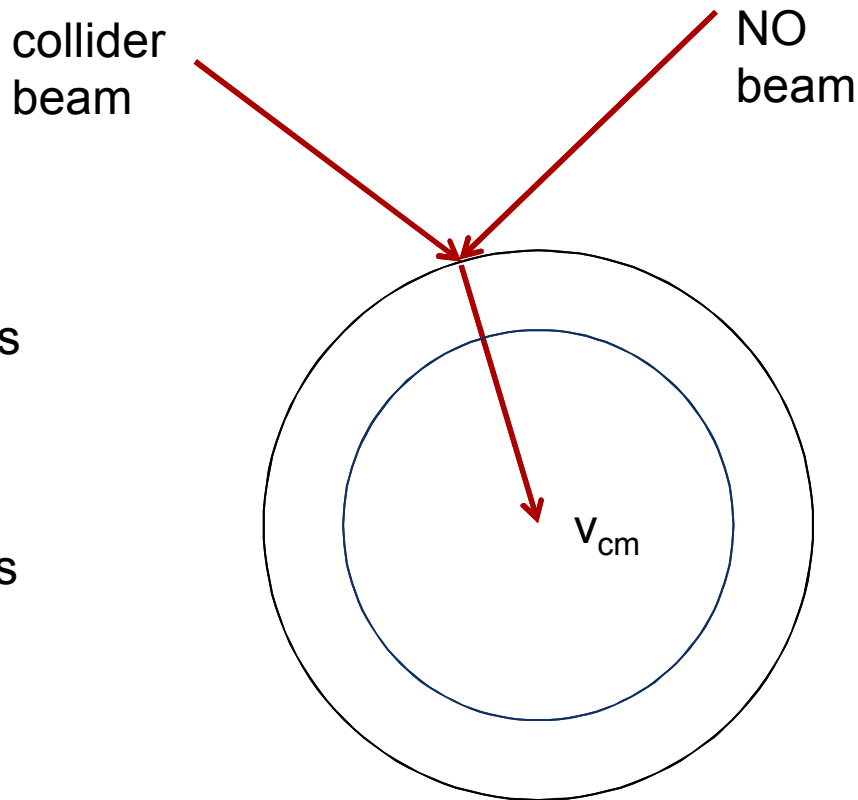
Experimental setup for crossed molecular beam collision detection



Crossed molecular beam velocity map ion imaging apparatus



Velocity mapping of NO after a collision

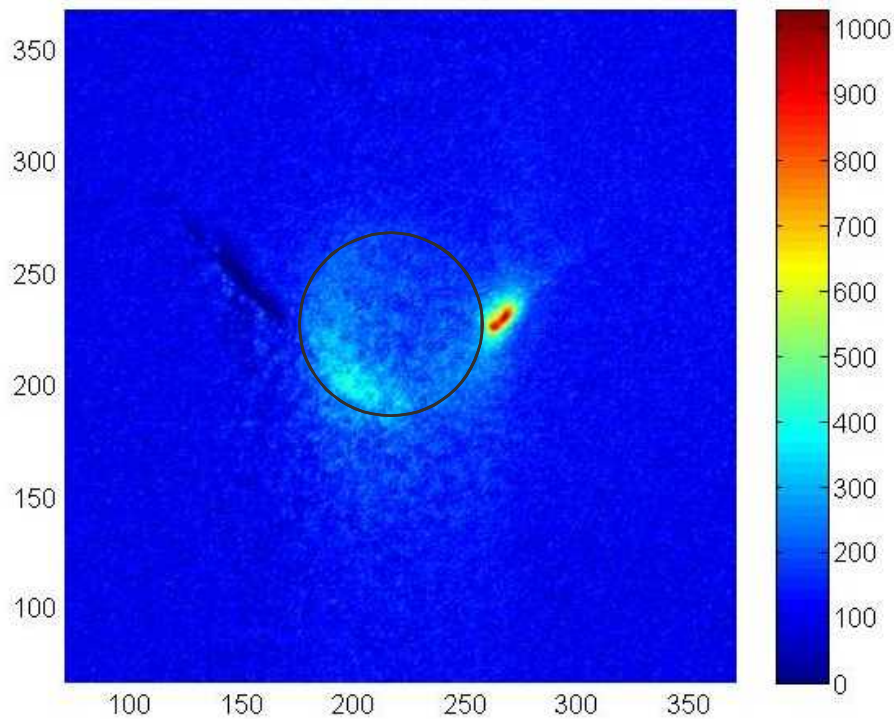


- outer ring shows velocity map at $J=0.5$

- inner ring shows velocity map at a higher J state

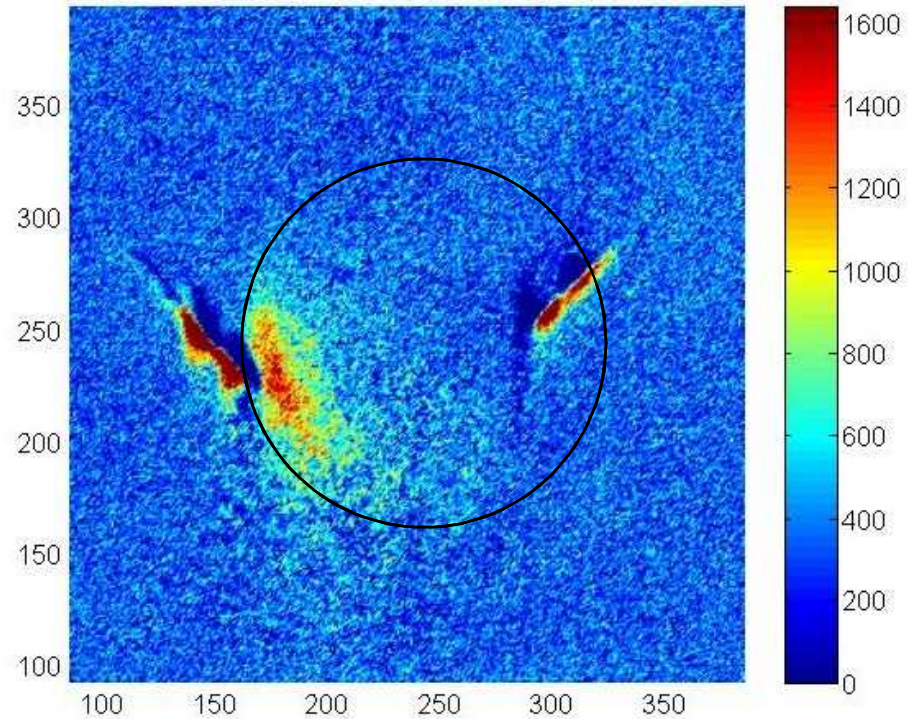
Comparing images at J=12.5

NO with Ar scattering



5 iterations

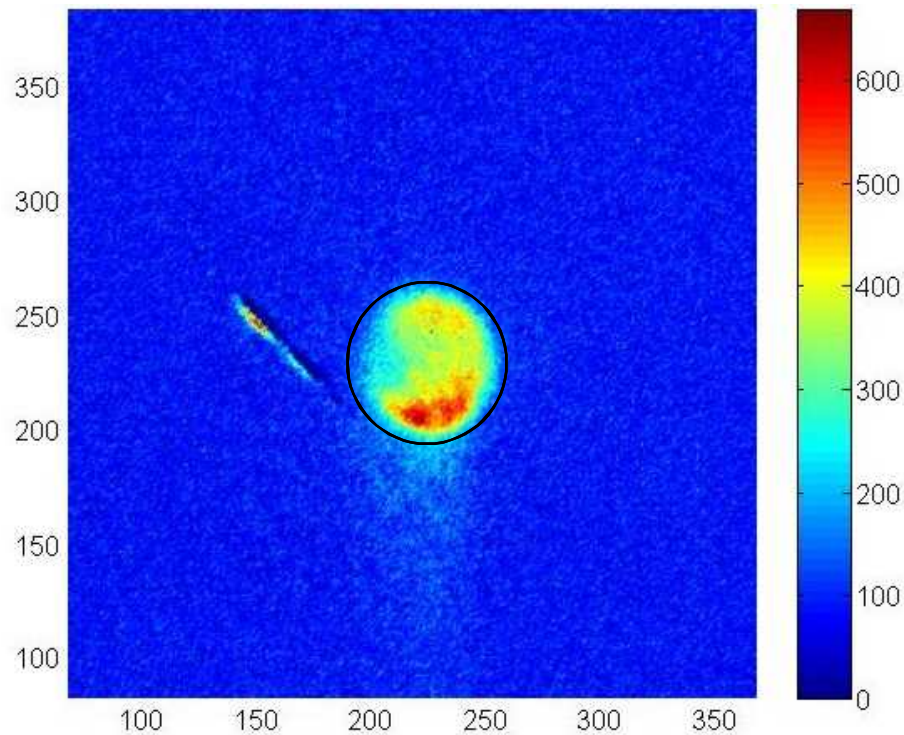
NO with NO₂ scattering



100 iterations

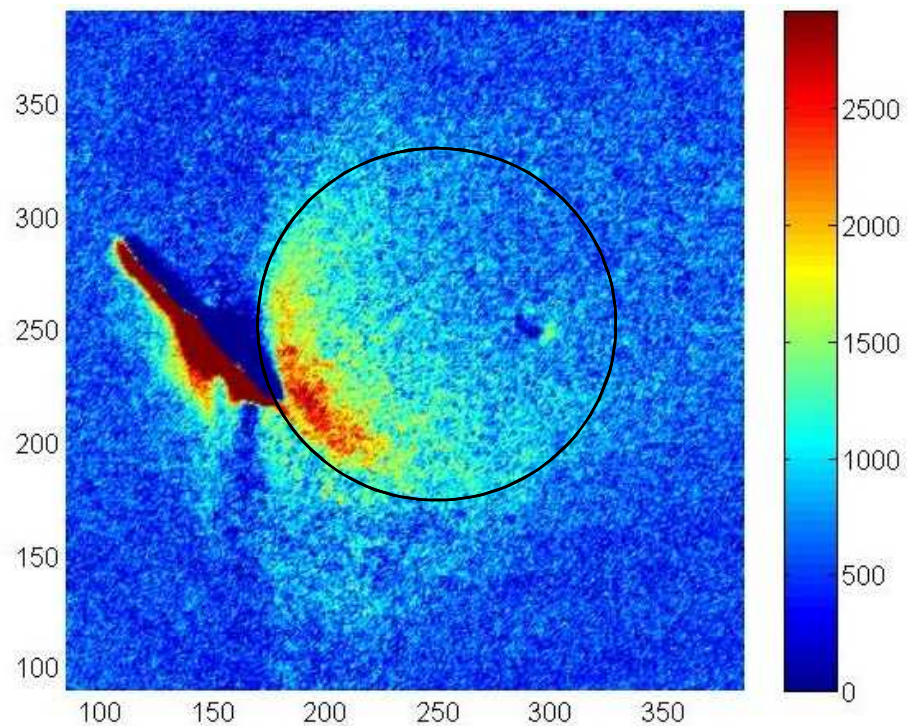
Comparing images at J=15.5

NO with Ar scattering



5 iterations

NO with NO₂ scattering



100 iterations

What can we learn from the velocity map ion images?

Difference energy of colliding beams has multiple effects on the scattering of NO

- radius of circle with Ar scattering is much smaller because less energy
- even at a high rotational state, NO is forward scattered
- the blurring effect seen in the Ar + NO image is due to the correlation between a high rotational state and more back scattering
- the blurring effect seen in the NO₂ + NO image is due to NO₂ picking up energy from the NO during the collision

Future Work

- analyze the data collected of NO and NO₂ collision energy transfer
- excite NO₂ before collision by sending in another laser

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