

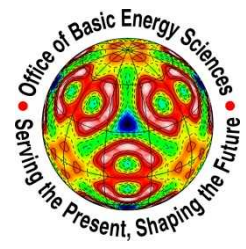
Characterization of Microtubules for Use in Multi-Scale Nanomaterials

Adrienne C. Greene, Marlene Bachand, Andrew Gomez, and George D. Bachand

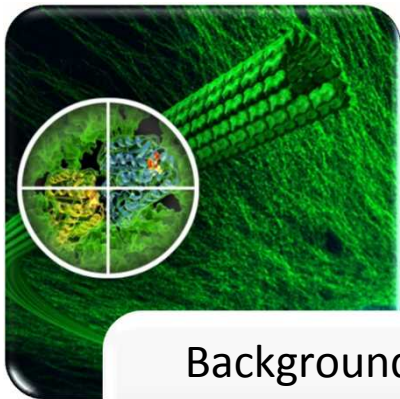
Center for Integrated Nanotechnologies
Sandia National Laboratories

MRS Fall Meeting

December 1st, 2015



Outline for Today's Talk



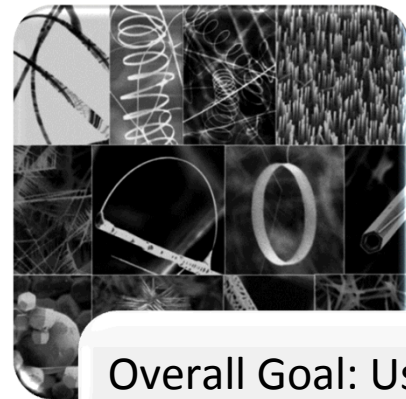
Background

- Biopolymers
- Self-assembly at multiple length-scales



Building Microtubule Arrays

- Characterize assembly

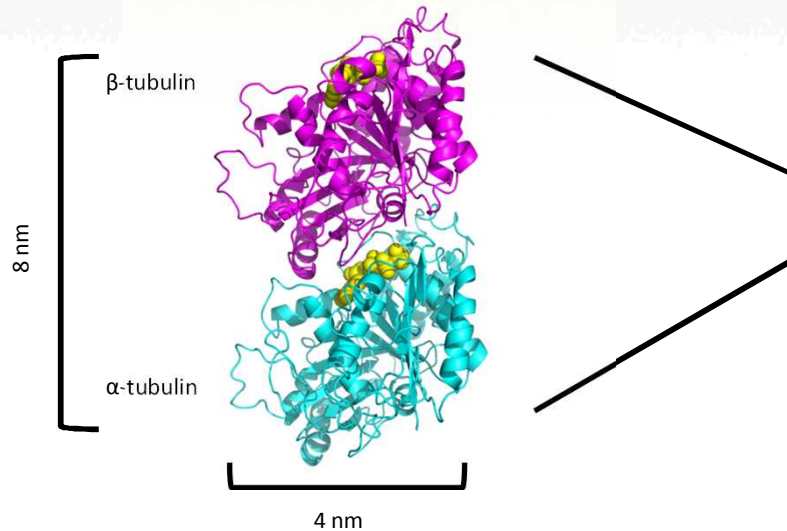


Overall Goal: Use biological components as templates for nanomaterial synthesis

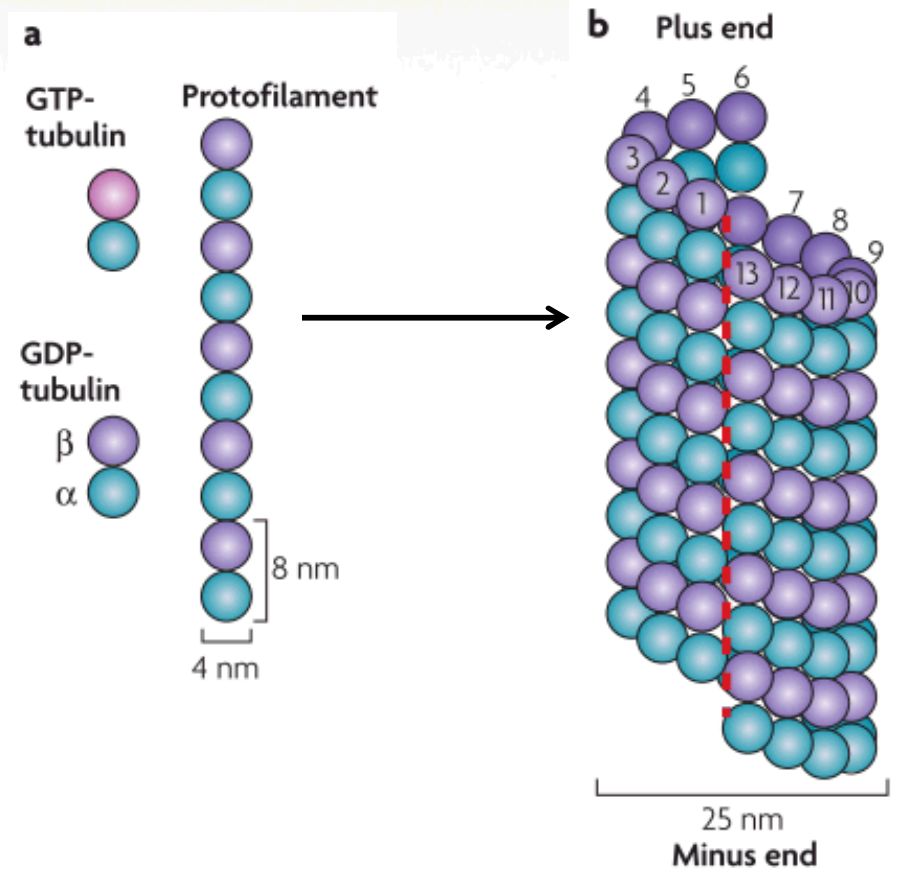


Self-Assembly of Microtubules

Microtubules (MTs) have unique self assembly processes ranging across multiple length scales



- Microtubule filaments are biopolymeric self-assemblies of $\alpha\beta$ tubulin dimer units
- Tubulin dimer units polymerize into protofilaments which then assemble into microtubule filaments
 - ~13 protofilaments assemble into hollow filaments (25 nm diameter, 10s of microns in length)

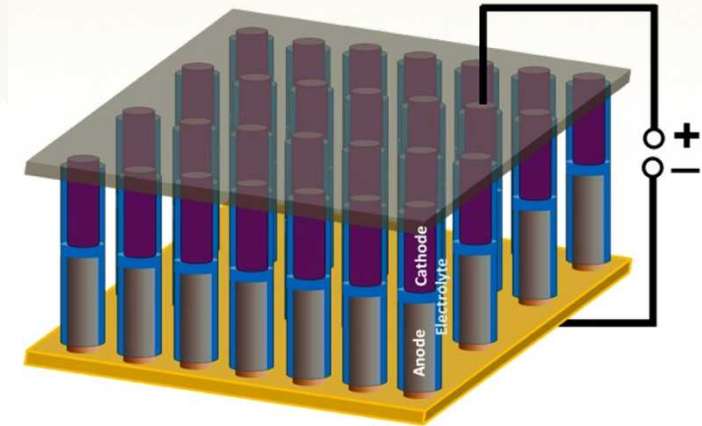


Microtubules as Templates for Nanomaterial Synthesis

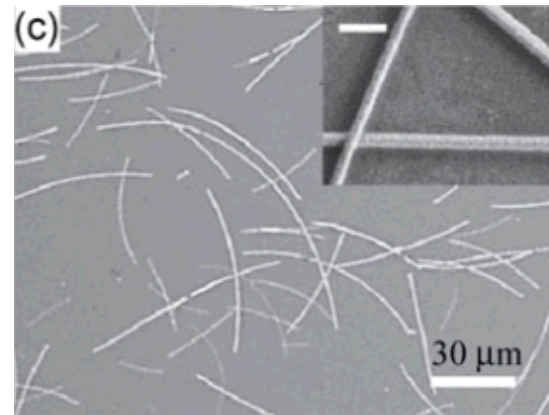
Nanowire synthesis is of increasing interest to the biosensing, MEMS, NEMS and nanobattery fields

Microtubules have attractive properties as templates for synthesis of nanomaterials

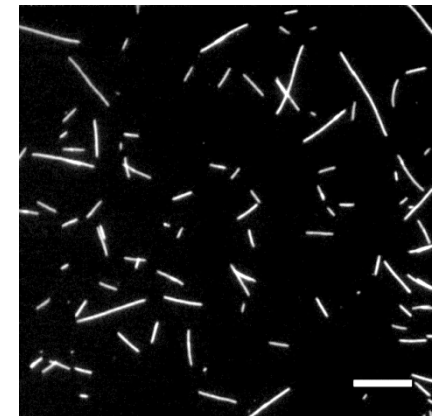
- Similar size scale
- Monodisperse diameter (25nm)
- Tunable length
- Versatility in chemical functionality to different amino acids
- *Simple and quick polymerization dynamics*
 - *Nanowire synthesis typically uses extreme conditions and takes many hours*



Silver Nanowires



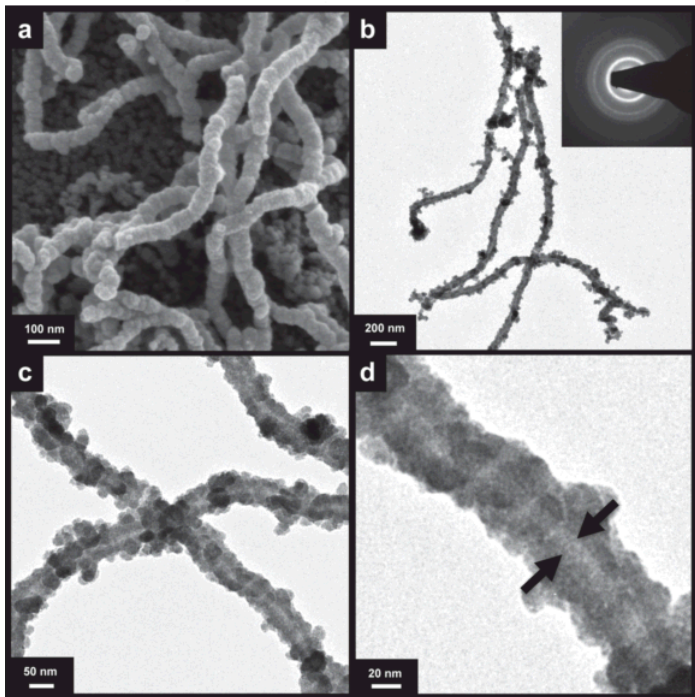
Microtubules



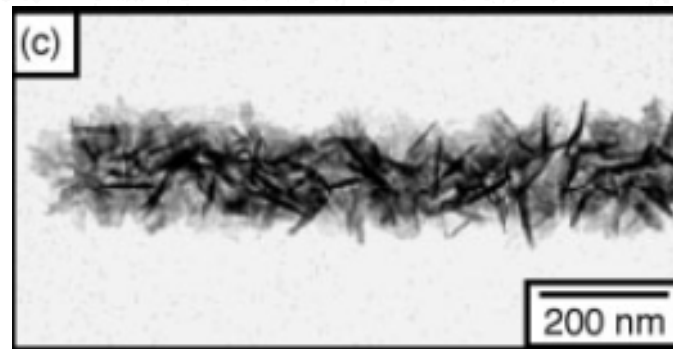
Microtubule-Templated Nanowires

Microtubules have been used as templates for a variety of nanowire synthesis

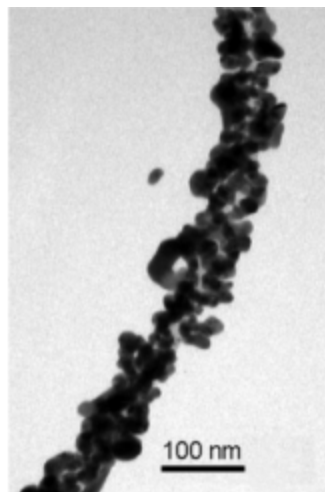
CdS Templated on Microtubules



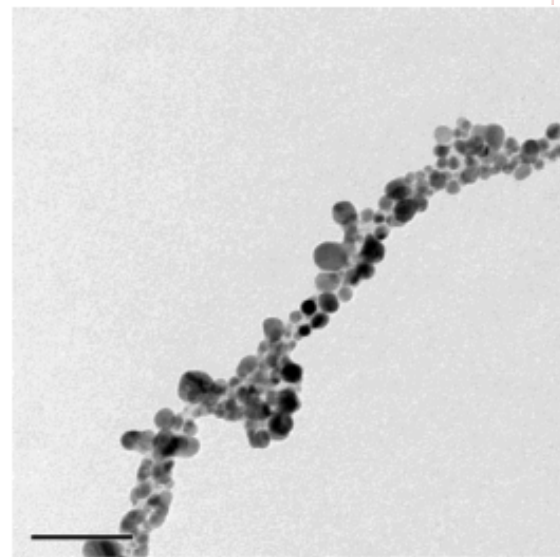
FeO(OH) Templated on Microtubules



Au-Templated on Microtubules



Ag-Templated on Microtubules



Spoerke, E., et al. *Part and Part Syst Characterization*. **31**(8): 863-70.

Boal, A., et al. *Adv Funct Mat*. **14**(1): 19-24.

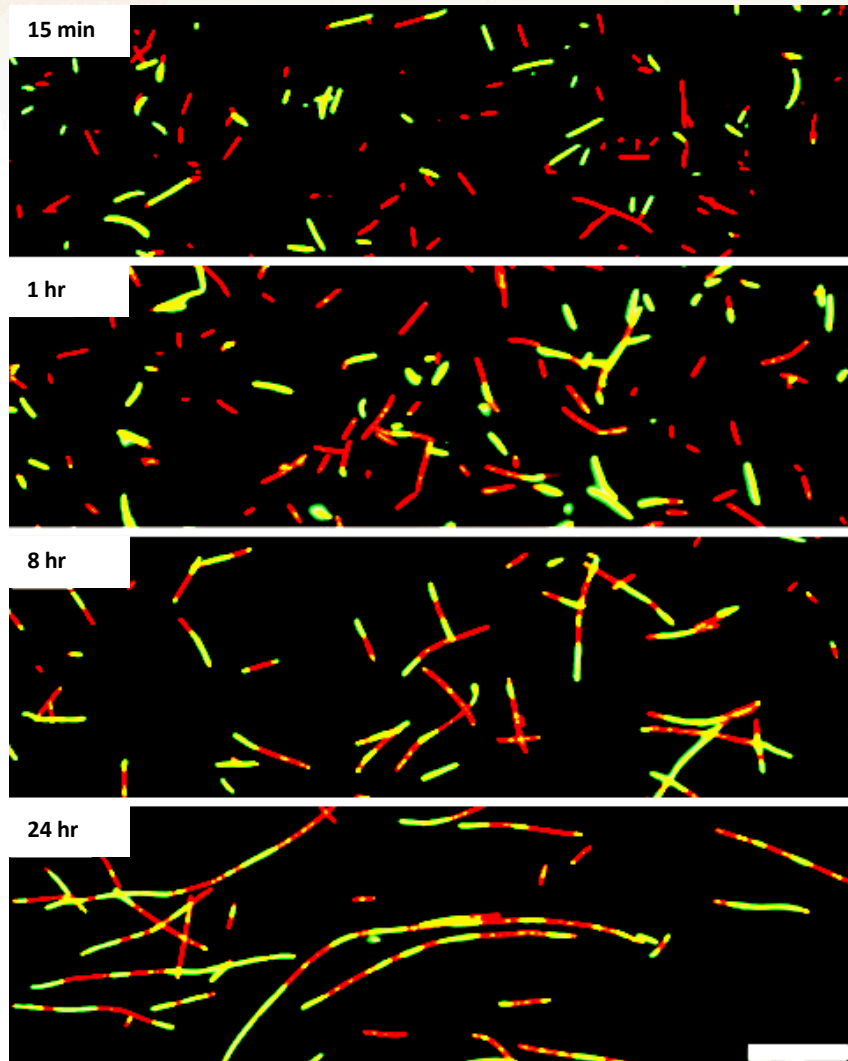
Zhou, J., et al. *Small*. **4**(9): 1507-1515.

Behrens, S., et al. *Chem Mater*. **16**(16): 3085-90.



Sandia
National
Laboratories

MTs Self Assemble Using a Unique Growth Mechanism

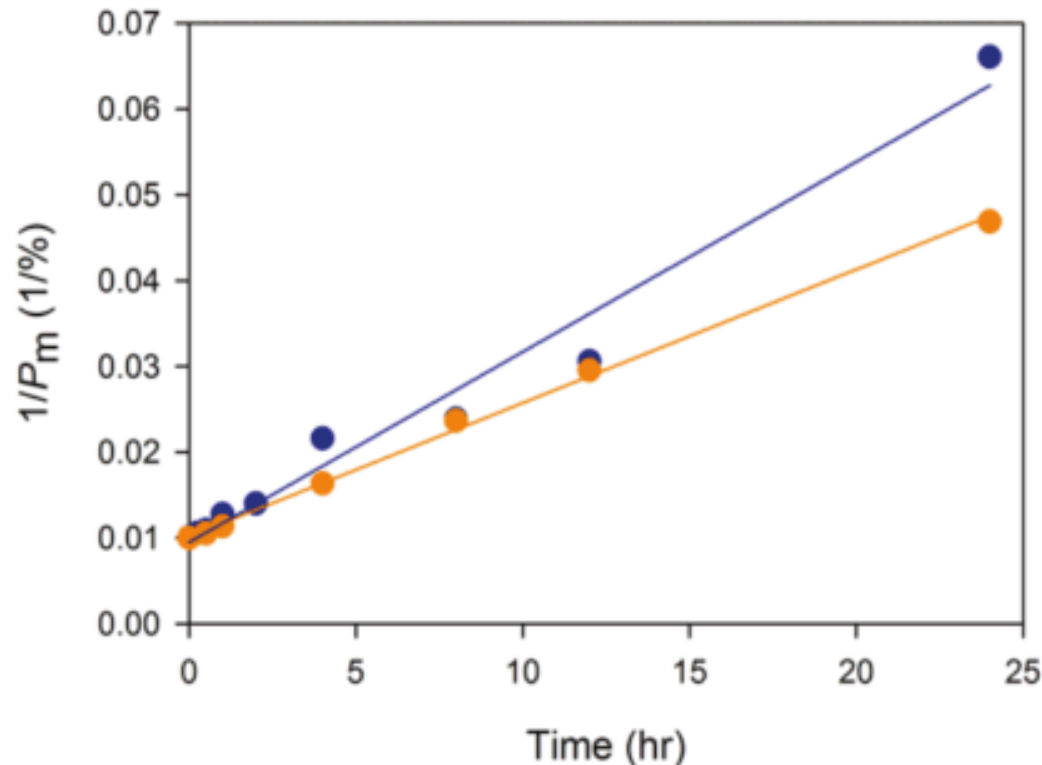


- Microtubule filaments fuse end-to-end as quickly as 15 minutes
- These fusion events increase over time, up to several weeks
- Microtubule fusion occurs using a unique secondary self-assembly
 - This is not a polymerization or nucleated growth mechanism



Reaction Rate of Self-Assembly can be Calculated

End-to-end self-assembly (or fusion) follows a **second-order rate function**, suggesting that there are two “reactants” involved in the assembly.



Rates of fusion can be calculated by:

1. Determining the amount of fusion over time
2. Taking the inverse to determine average rates of fusion

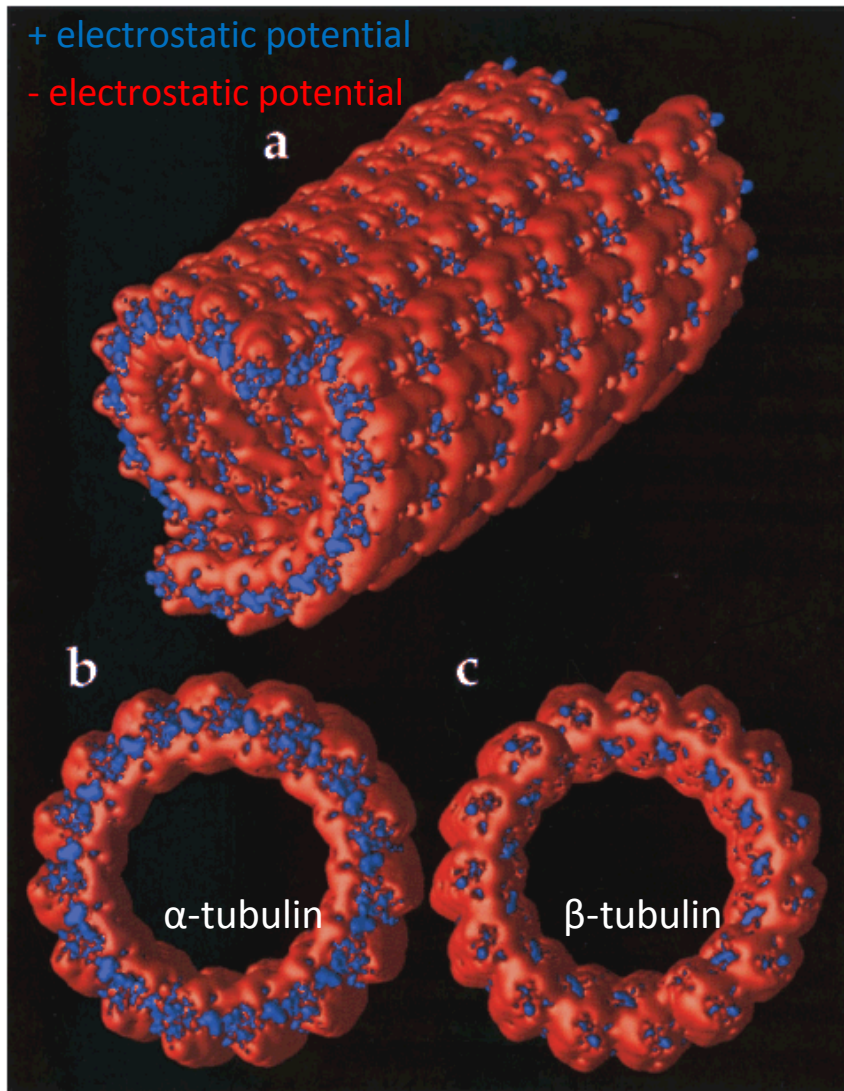
This provides an assay to probe different factors that can alter the rates of fusion

Microtubules + Free Tubulin Dimers

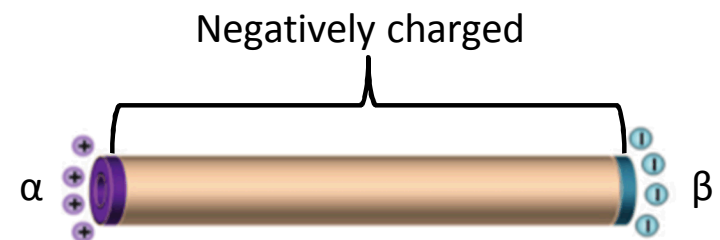
Microtubules - Free Tubulin Dimers



Microtubules are Polar and Overall (-) Charged

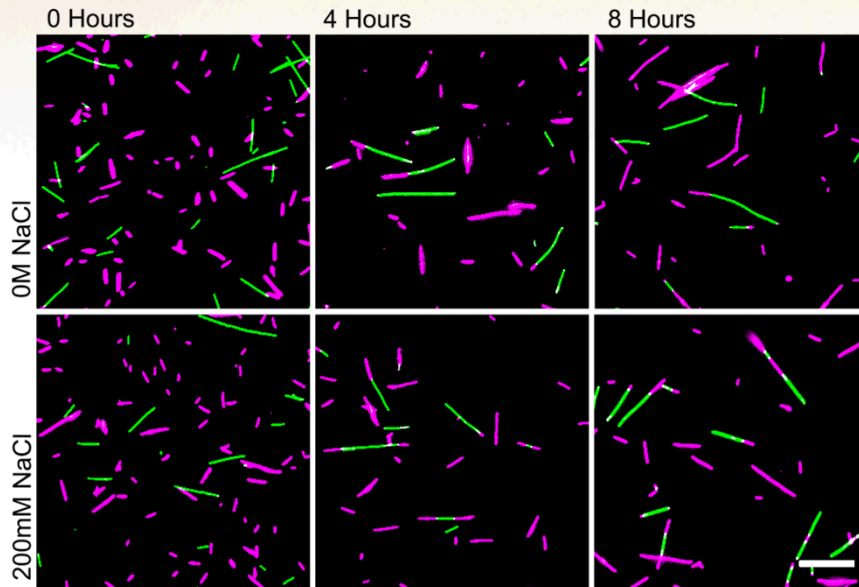


- How do electrostatics regulate fusion?
- The overall charge of a microtubule is highly negative
 - Decreasing electrostatics decreases excluded volume effect, thus increasing probability of microtubules interacting with each other
 - **A positive effect on fusion**
- Microtubules have electrostatically attractive ends (+ and -)
 - Decreasing electrostatics decreases electrostatically attractive ends
 - **A negative effect on fusion**

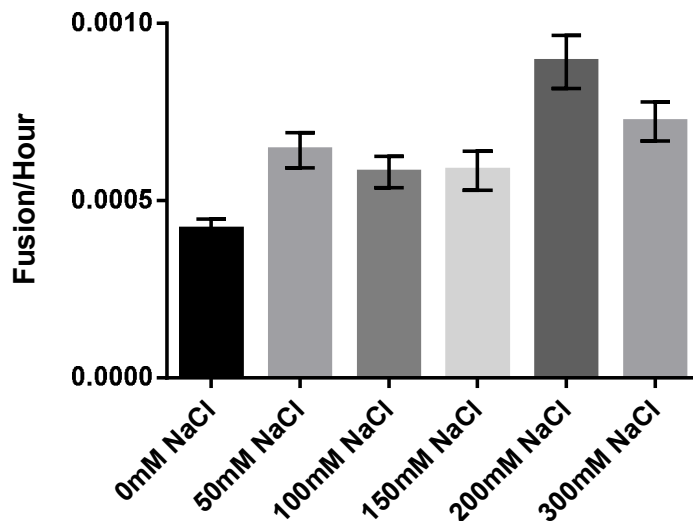


Addition of NaCl Increases Rates of Fusion

a

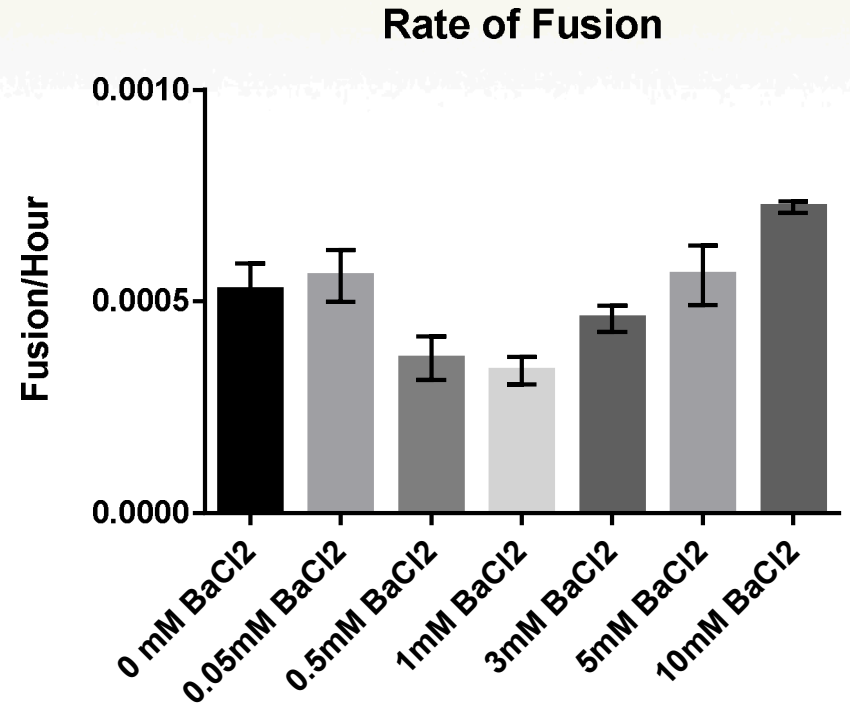
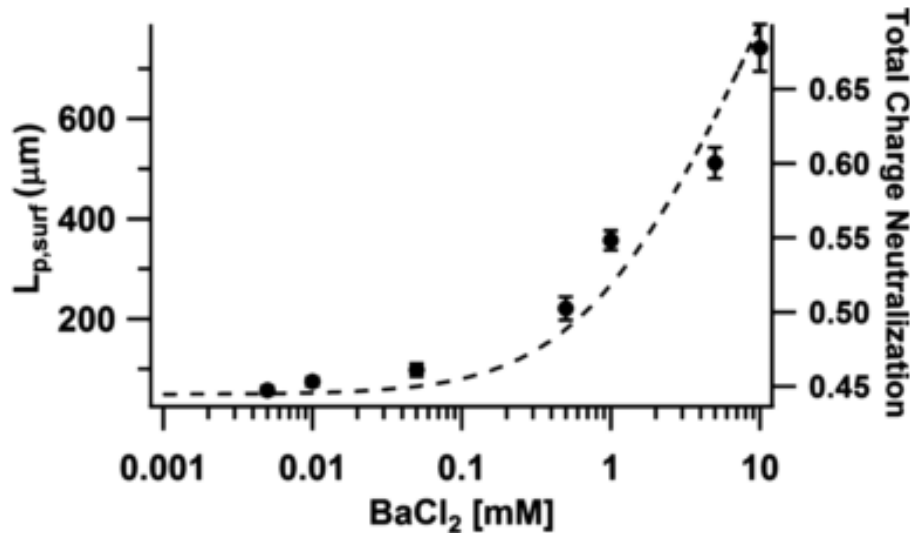


Rate of Fusion



- Addition of NaCl shields the overall negative charge of the microtubule filaments
 - Decreases the excluded volume of interaction
 - Increases probability of interaction
- Addition of NaCl increases the rate of fusion, almost in a step-wise function at **~200mM**
- Model calculations determined that microtubules are fully charge neutralized with **~217 mM NaCl**
- End-to-end fusion of microtubules must be dominated by hydrophobic interactions

Addition of BaCl_2 Alters Rates of Fusion



- Addition of BaCl_2 increases the persistence length and charge neutralization of microtubules
- Addition of BaCl_2 alters rates of fusion following charge neutralization trends

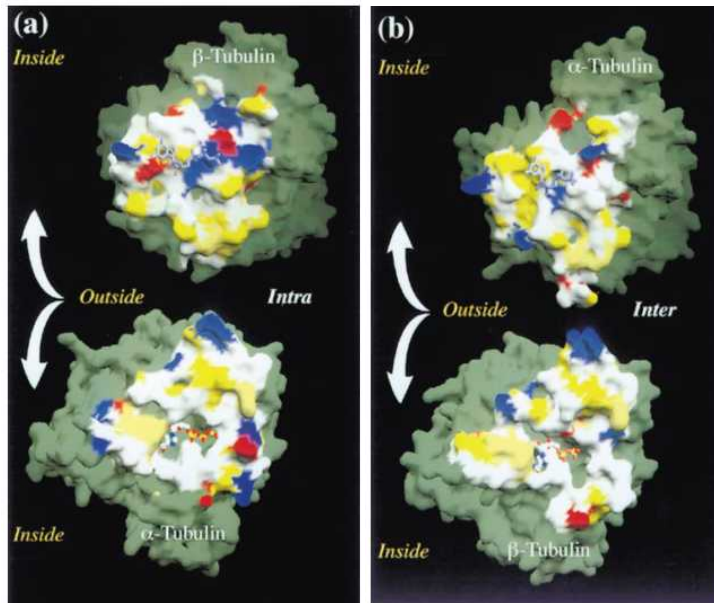


Hydrophobic Interactions Drive Longitudinal Contacts

Structural studies suggest that:

- Hydrophobicity drives monomer and dimer longitudinal contacts
- Electrostatics drives lateral protofilament interactions

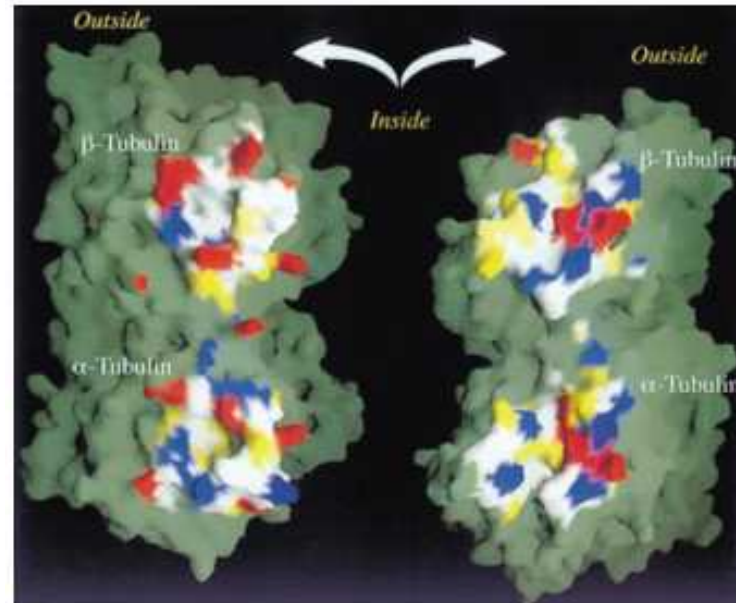
Hydrophobic interactions



Between monomers

Between dimer units

Electrostatics



Blue: + charged a.a.

Red: - charged a.a.

White: polar a.a.

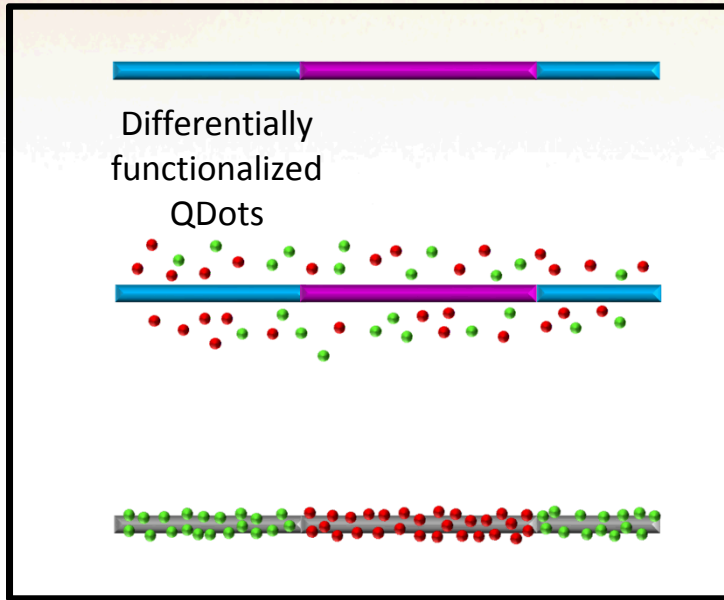
Yellow: hydrophobic a.a.

Pale yellow: Trp and Tyr

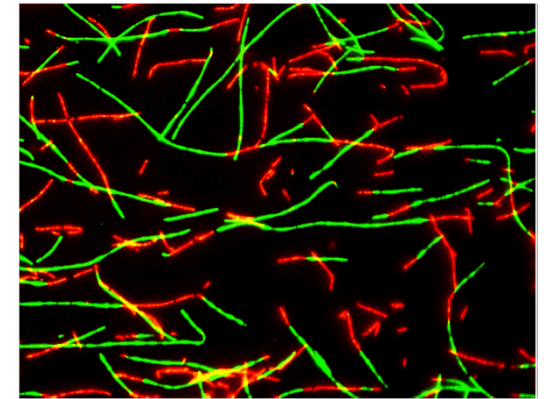


Sandia
National
Laboratories

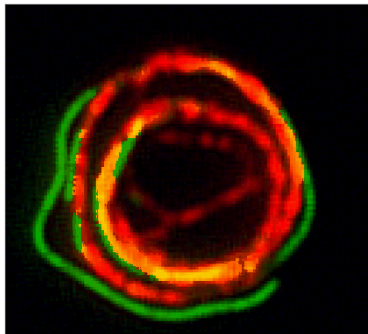
Using Microtubule Arrays to Form More Complex Structures



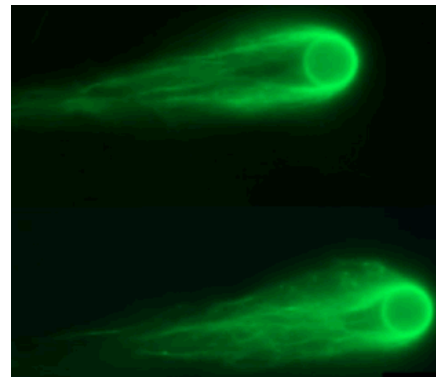
Morphology of these nanowires arrays can be tuned by altering the assembly process.



Linear Microtubule Arrays



Spooled Microtubule Arrays



Aster Microtubule Arrays



Acknowledgments

Dr. George D. Bachand
Marlene Bachand
Andrew Gomez



U.S. Department of Energy, Office of Basic Energy Sciences, Division of Materials Sciences and Engineering, Project KC0203010.

