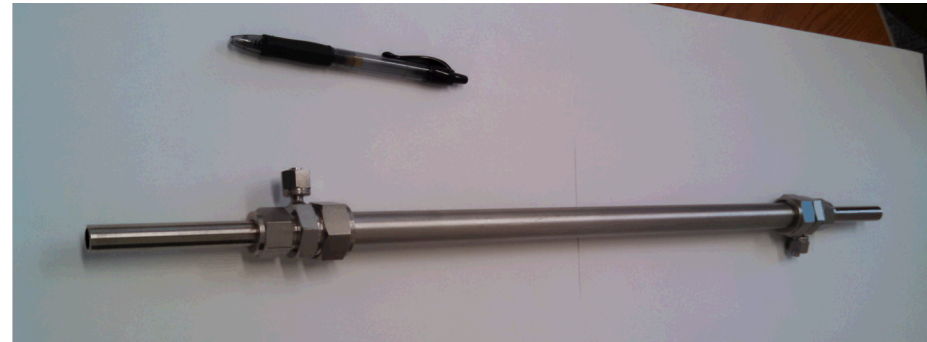
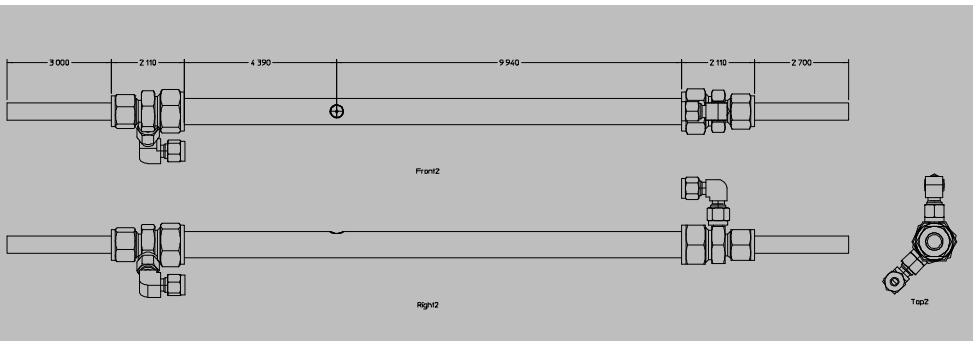


*Exceptional service in the national interest*



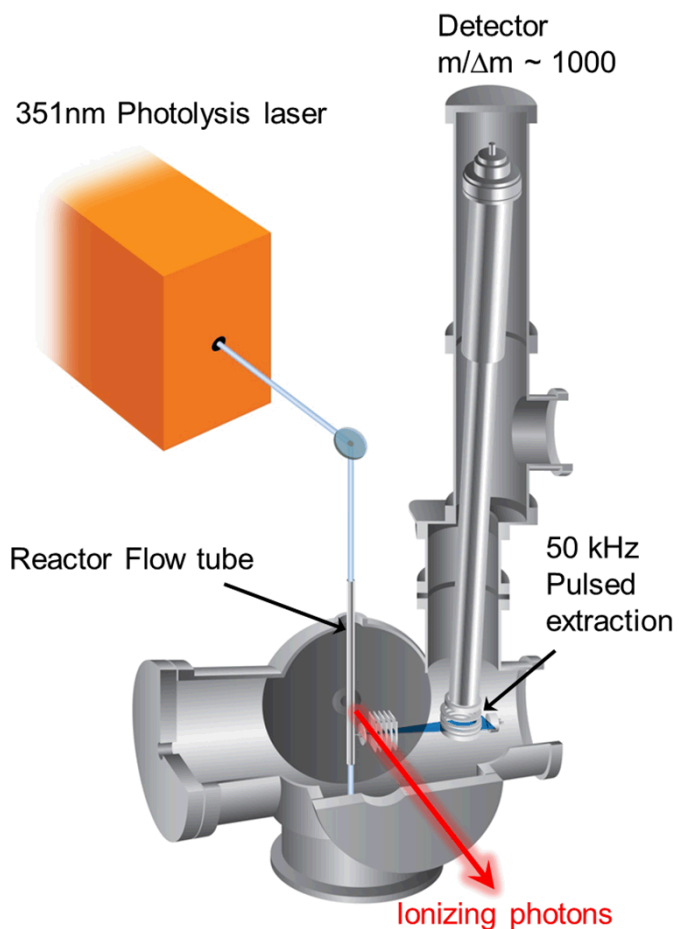
# Development of a Low-Temperature Flow Tube for Use in MPIMS

Leah G. Dodson, Howard Johnsen,  
Mitchio Okumura, David L. Osborn



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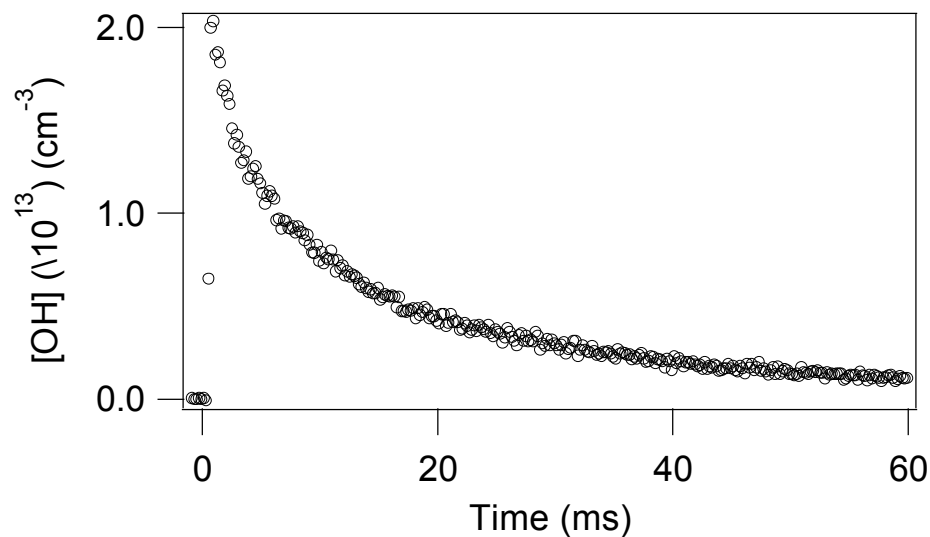
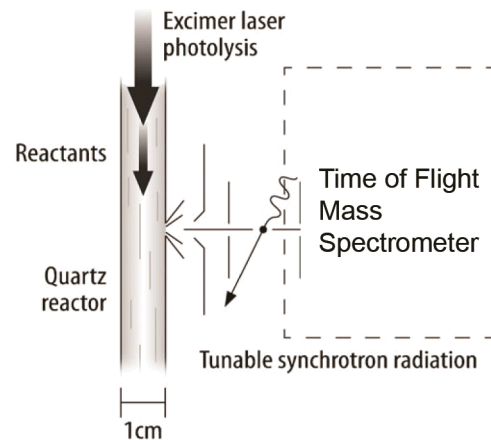
# Sandia Multiplexed Chemical Kinetics Reactor



Current capabilities:

300 – 1050 K

At 1023 K, temperature is uniform  $\pm 4$  K



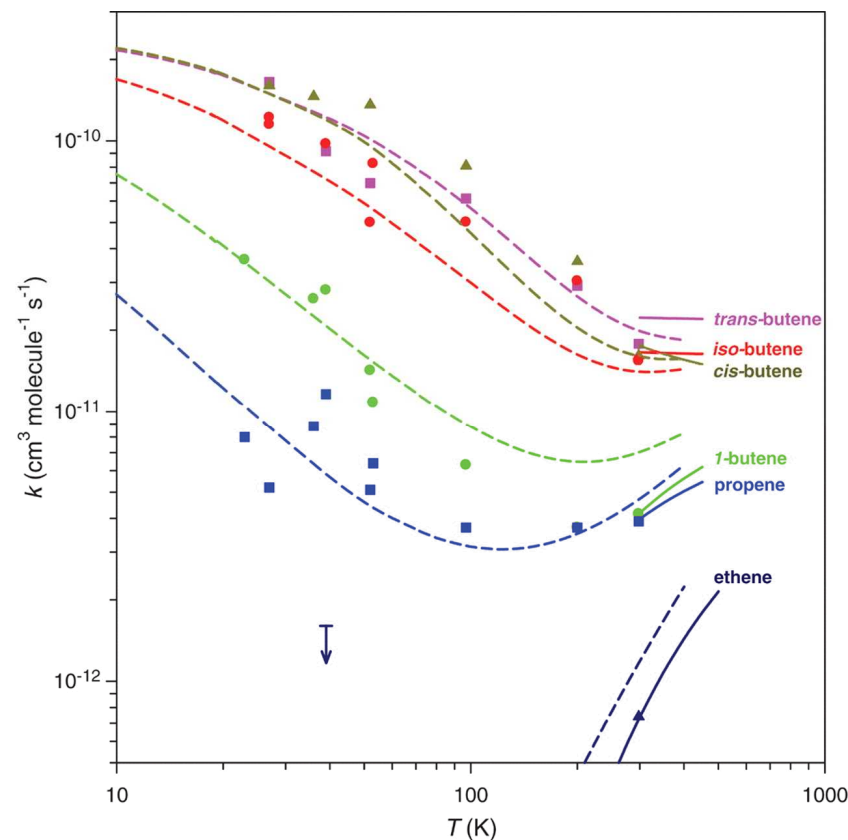
Gas Flow:

366 cm/ms

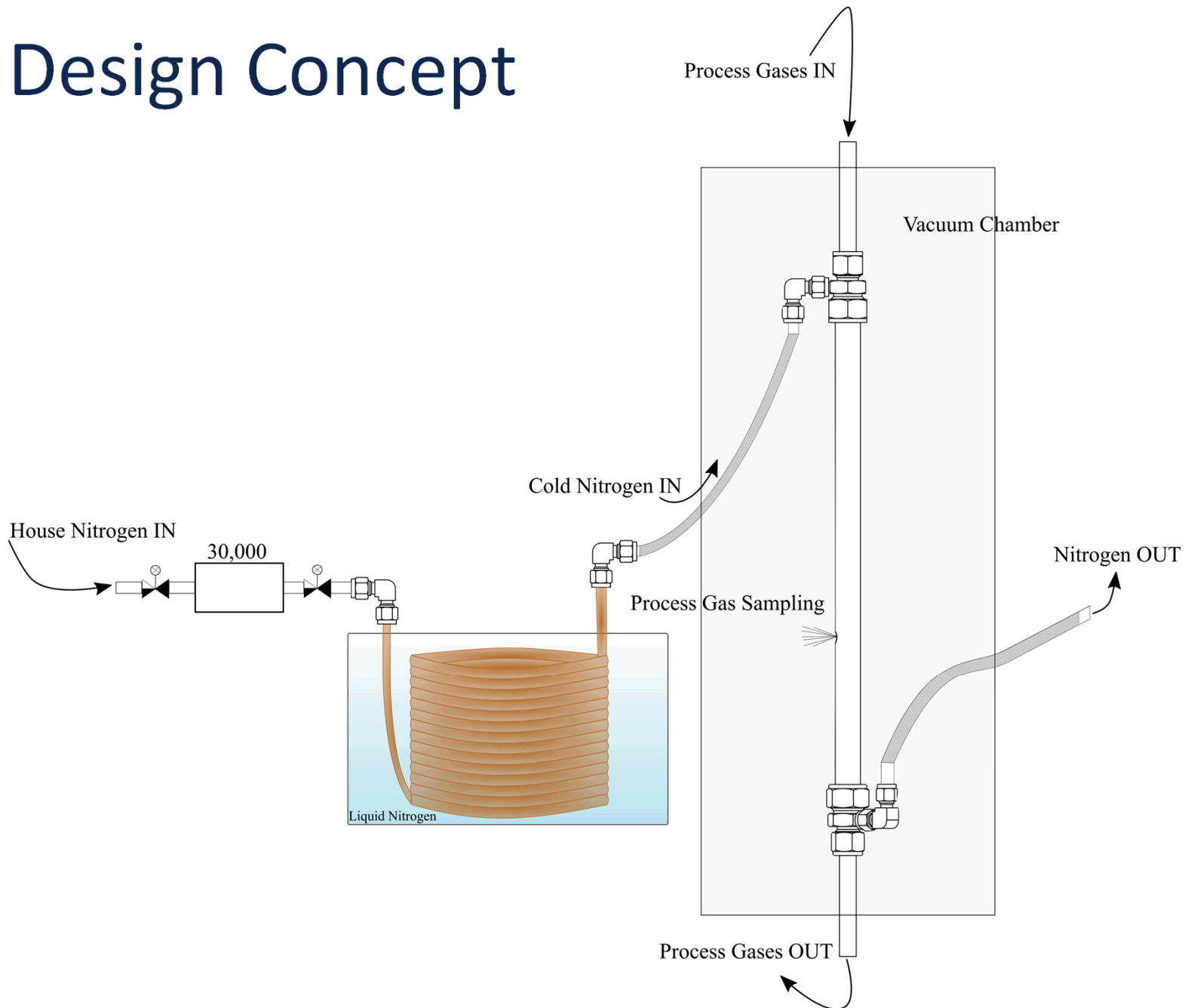
In 60 ms, gas has traveled 22 cm ( $\approx 9$  in)

# Low Temperature Studies of Radical-Radical Reactions

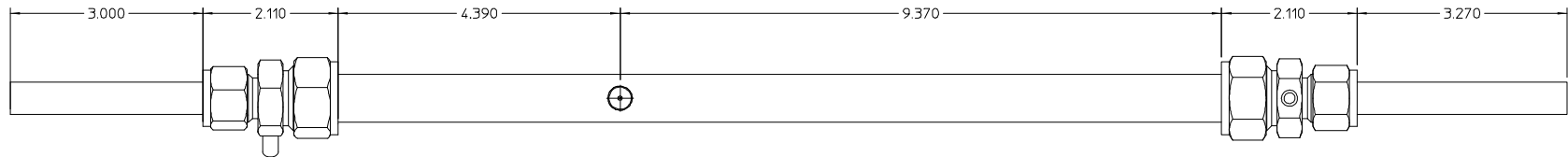
- Chemical reactions that have potential energy surfaces with multiple wells present a challenge
- The ability to measure reaction rates below room temperature complements high temperature experiments
- Example:  $O(^3P) + \text{alkene}$  – the outer transition state probed at low  $T$ , inner transition state at high  $T$



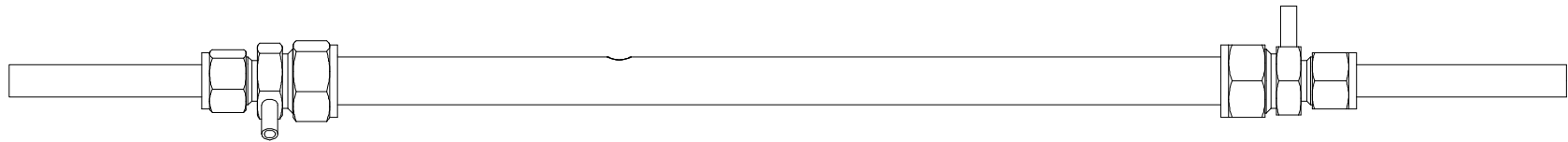
# Design Concept



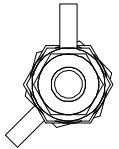
# Machine Drawings



Front1



Right1



Top1



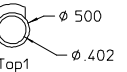
Front1



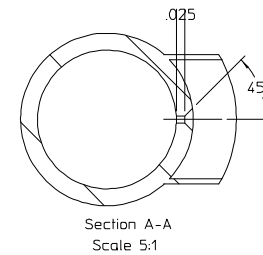
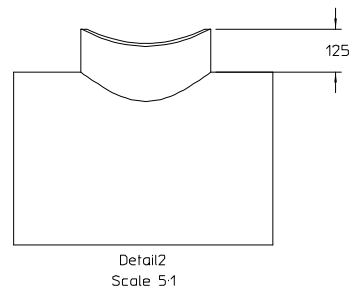
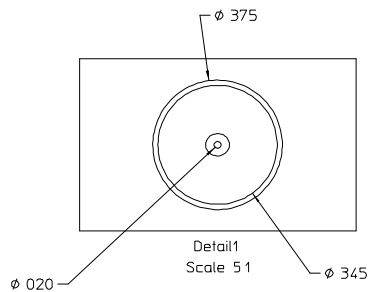
Right1

see Detail1

see Detail2

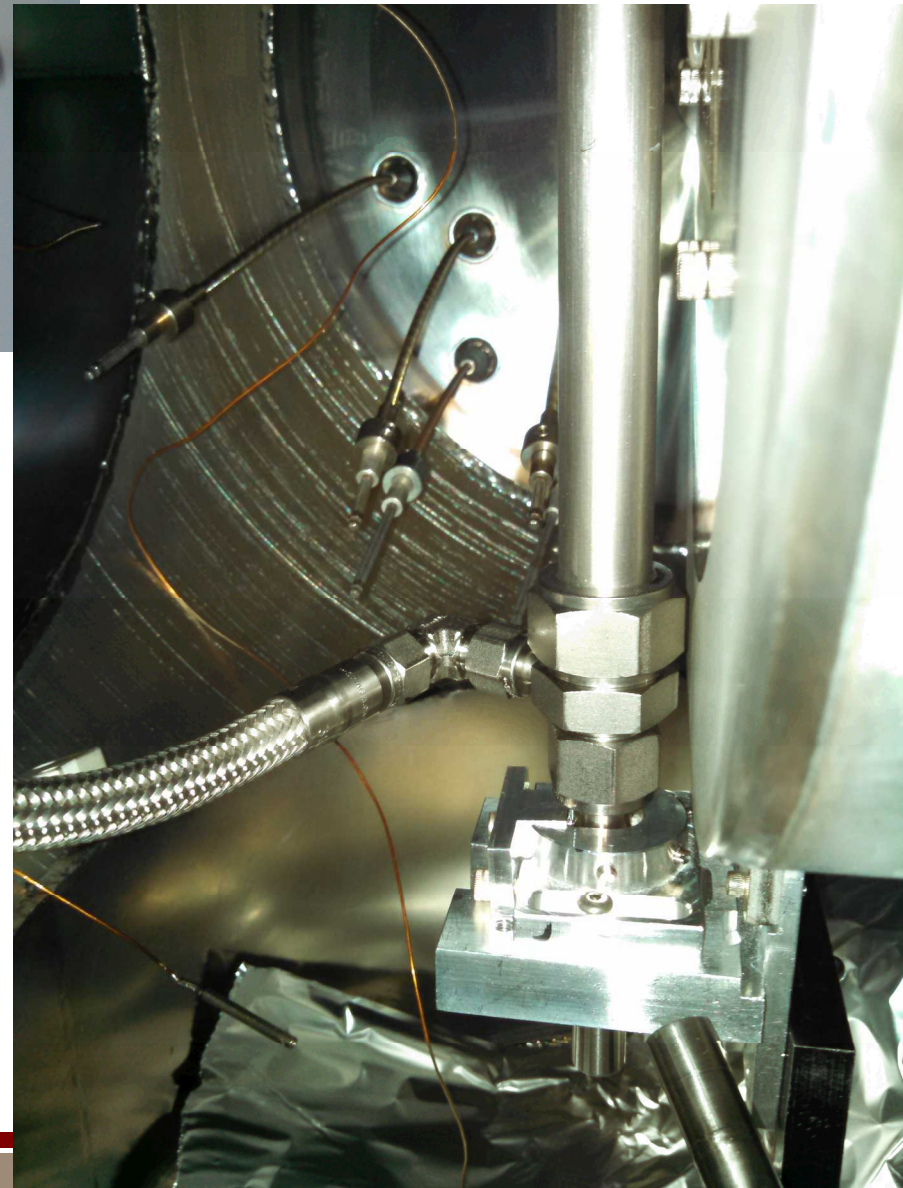
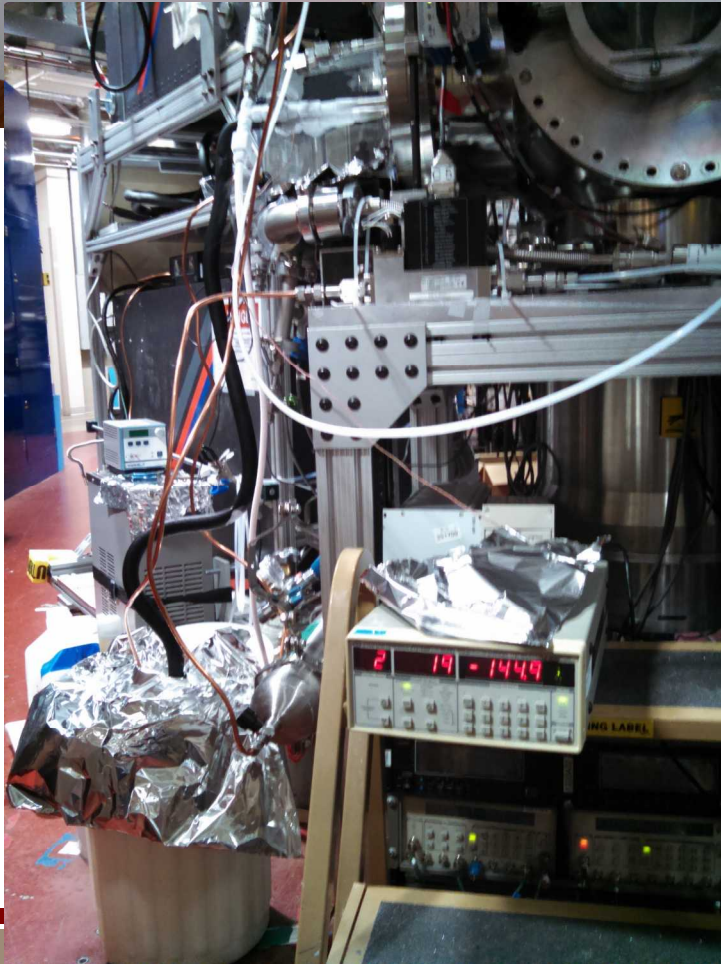
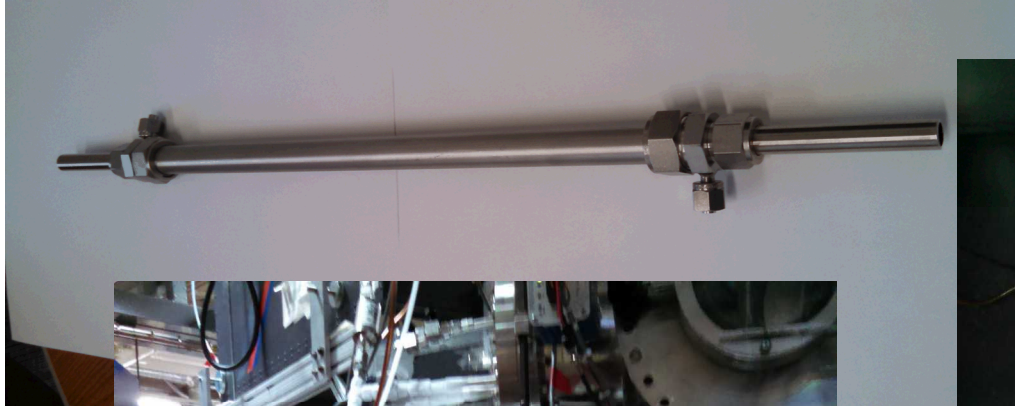


Top1

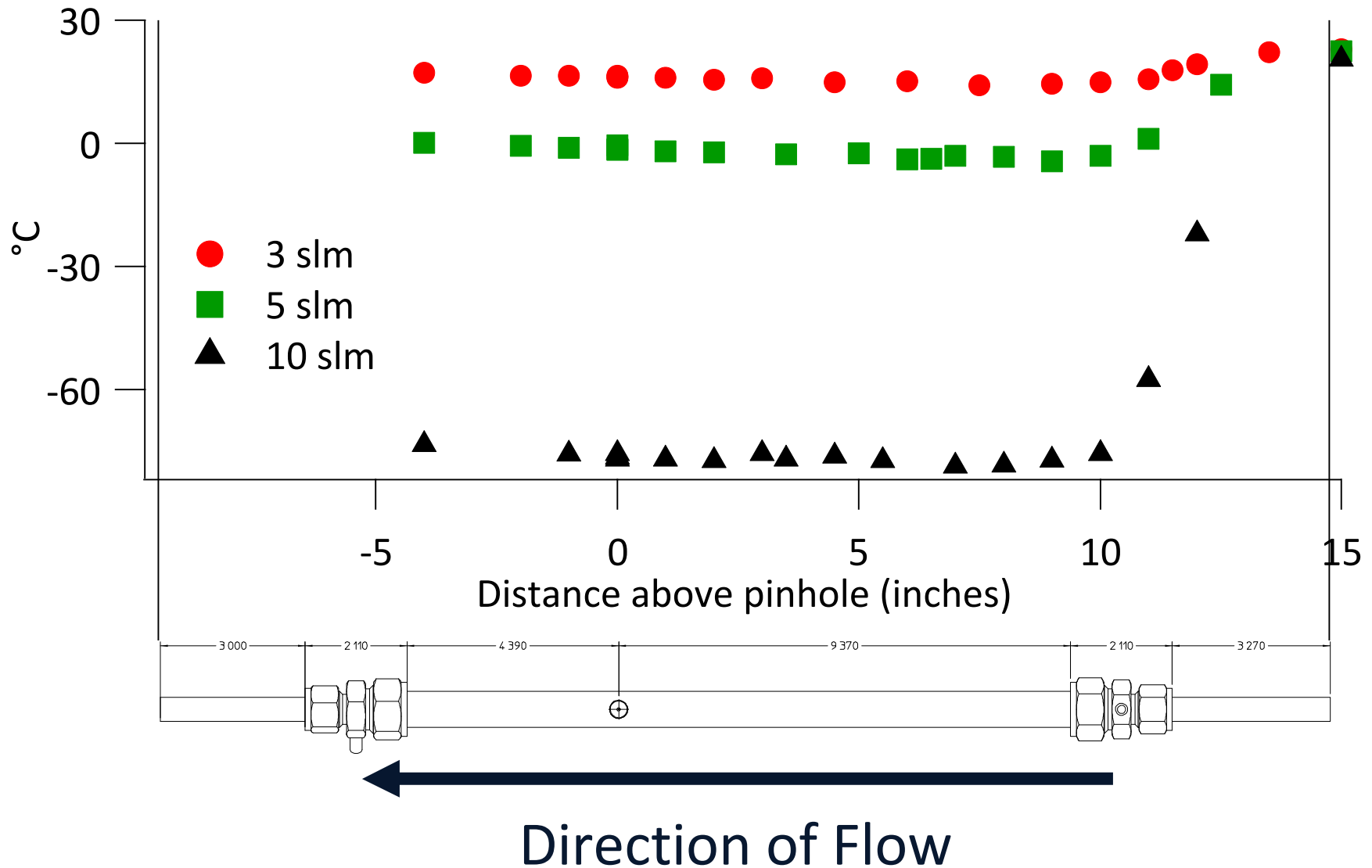




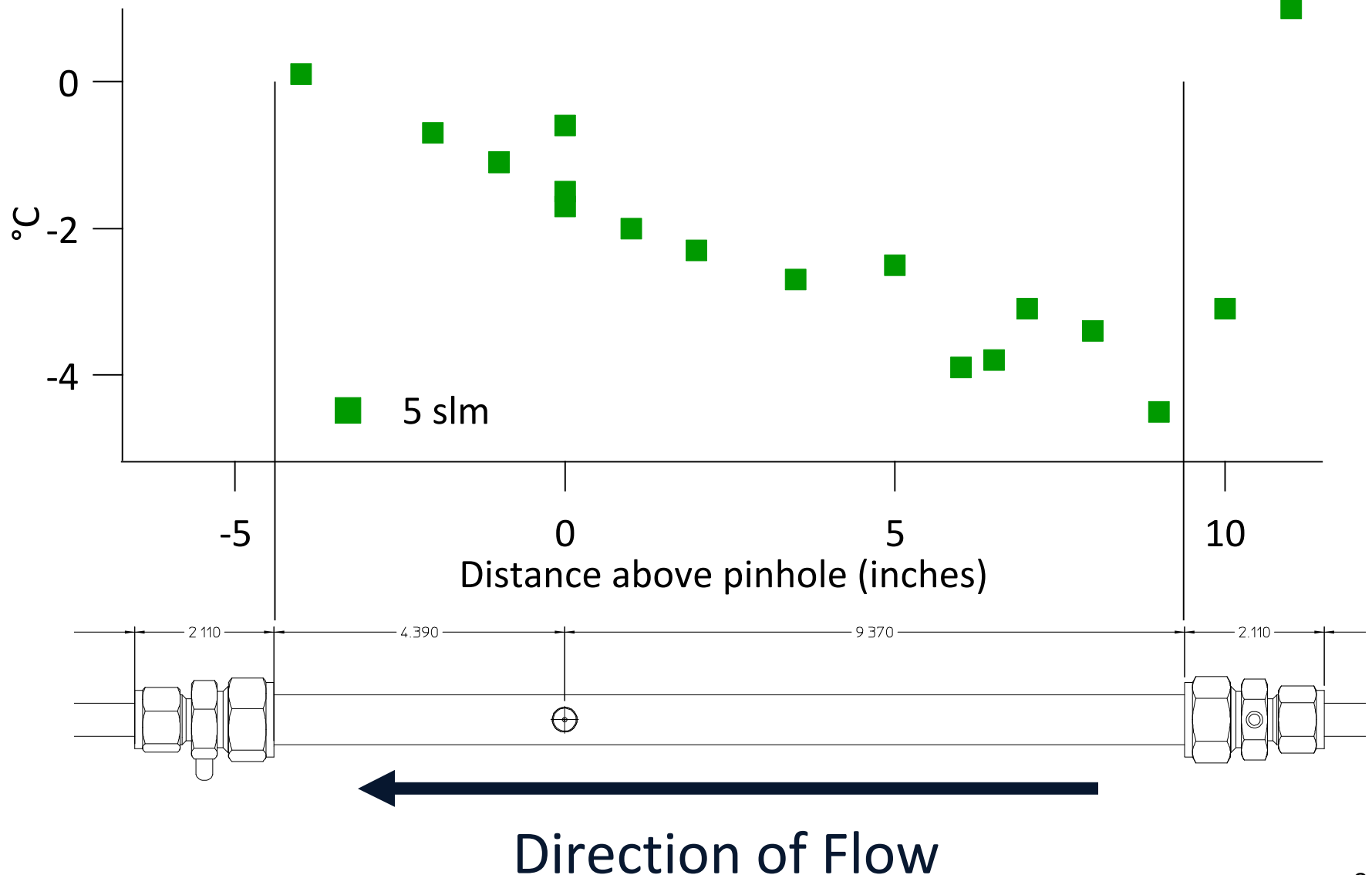
# Prototype Testing



# Overall Prototype Performance

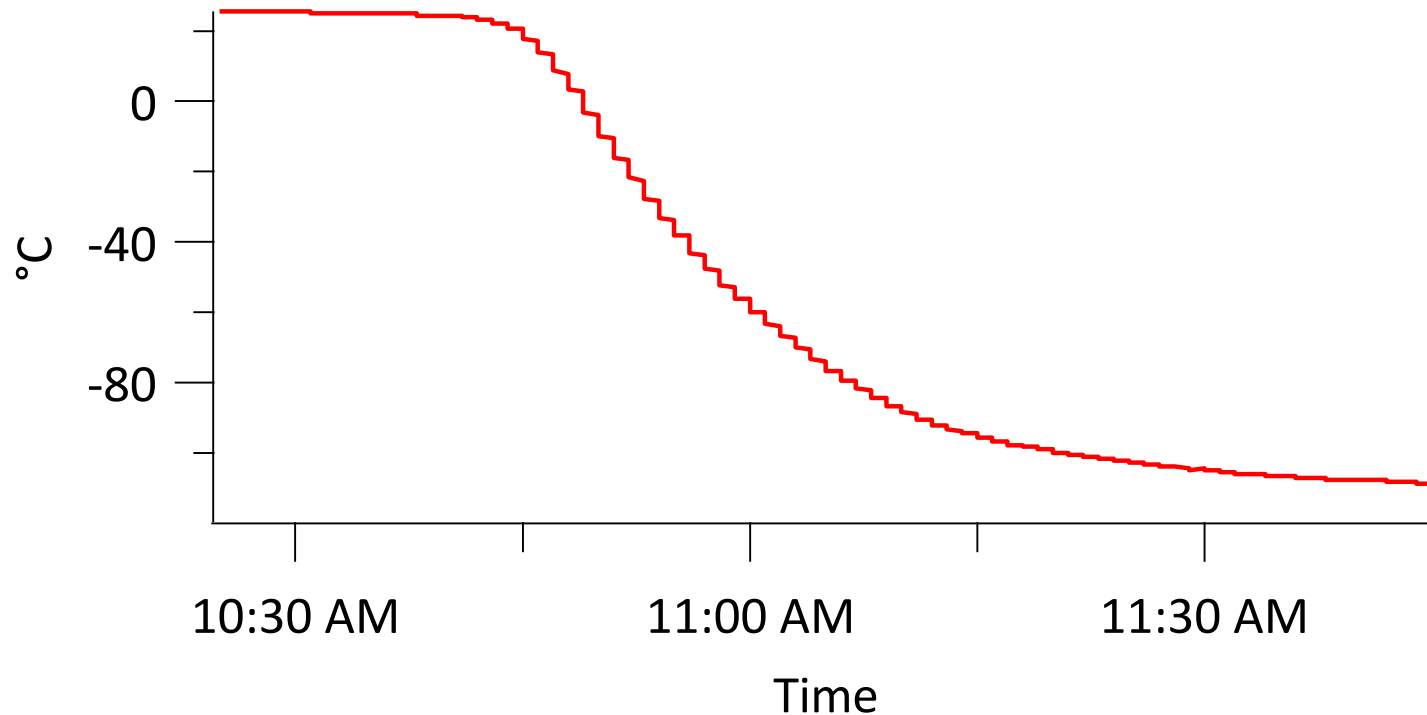


# Prototype Temperature Profile





# Equilibrium



- Temperature equilibrium is achieved after a little over an hour
- Equilibrium temperature varies by less than  $\pm 1^\circ\text{C}$

# Outlook

- Tests on prototype were a success
- Final design is being machined
- Testing of final design to begin August 10
- Experiments scheduled August 20 – 23

# Acknowledgements

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