

Exceptional service in the national interest



Growth of BGaN Alloys on AlN

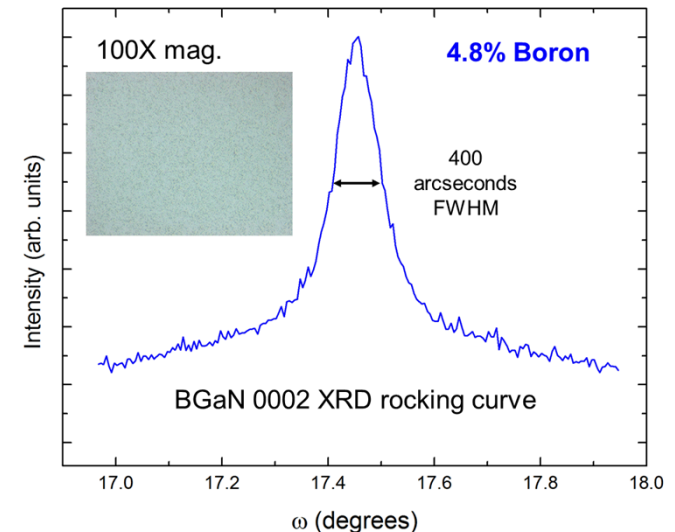
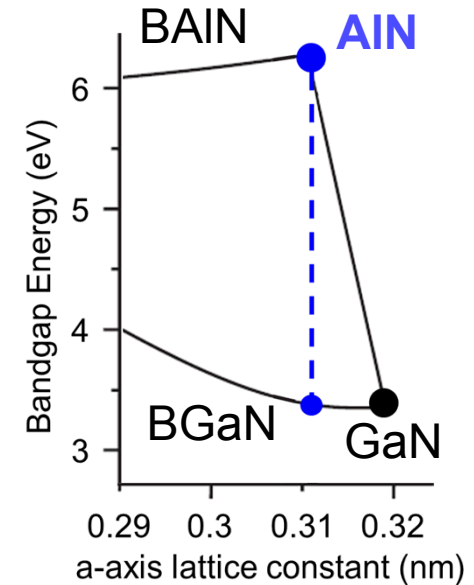
Michael W. Moseley,

Daniel D. Koleske, Stephen R. Lee, Andrew A. Allerman

Sandia National Laboratories,
Albuquerque, New Mexico, USA, 87123

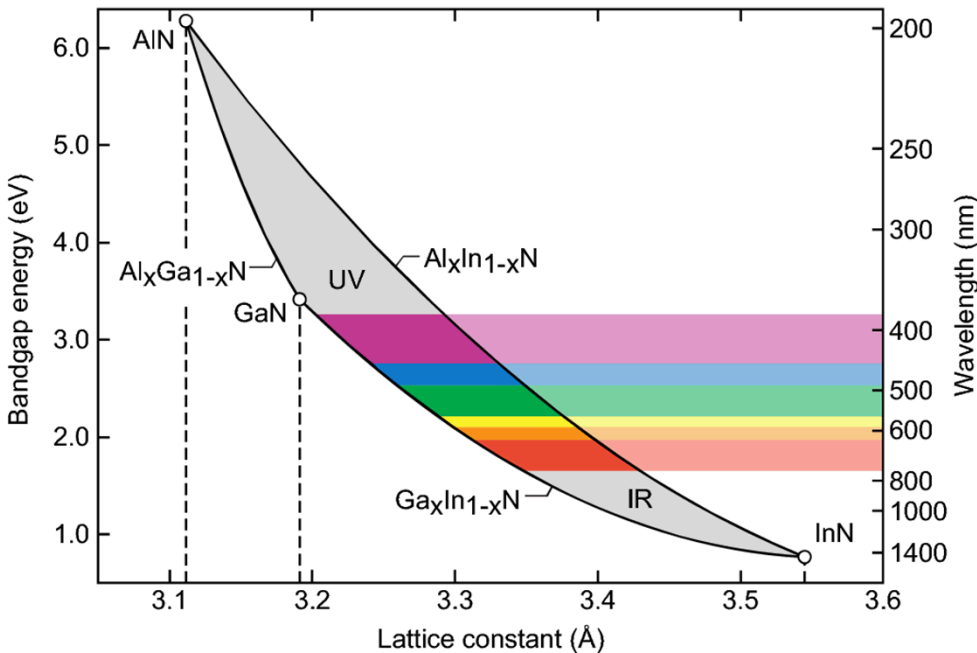
Outline

- **Background**
 - Dislocation generation
 - Opportunities for lattice-matched epitaxy
- **Investigation of BGaN alloys**
 - Add TEB flow to GaN growth
 - Accurate calculation of B%
- **Explore effect of carrier gas**
 - Morphology
 - Crystal structure
- **Lower temperature/pressure growth**
 - 5% boron mole fraction



Group-III Nitrides - Challenges

Substrate choice and trade-offs



Substrate	Mismatch to GaN	Additional notes
Sapphire	14%	Cheap , electrically and thermally insulating
SiC	3.1%	Expensive , electrically and thermally conductive
Si (111)	17%	Cheap, mature, forms amorphous SiN layer
Bulk GaN/AlN	0%	Expensive , maturing technology

No native substrate for AlGaIn alloy growth

M. F. Schubert, Ph.D. Thesis, Virginia Polytechnic Institute and State University, 2009.
 J. Wu, *J. Appl. Phys.* **106**, 011101 (2009)
 G. Namkoong et al., *Appl. Phys. Lett.* **87**, 184104 (2005).
 A. Janotti and C. G. Van De Walle, *Rep. Prog. Phys.* **72**, 126501 (2009).
 G. Namkoong et al., *Thin Solid Films* **517**, 6508 (2009).
 L. M. Sorokin et al., *Tech. Phys. Lett.* **37**, 326 (2011).

Group-III Nitrides - Dislocations

- **Dislocations generated from heteroepitaxy**

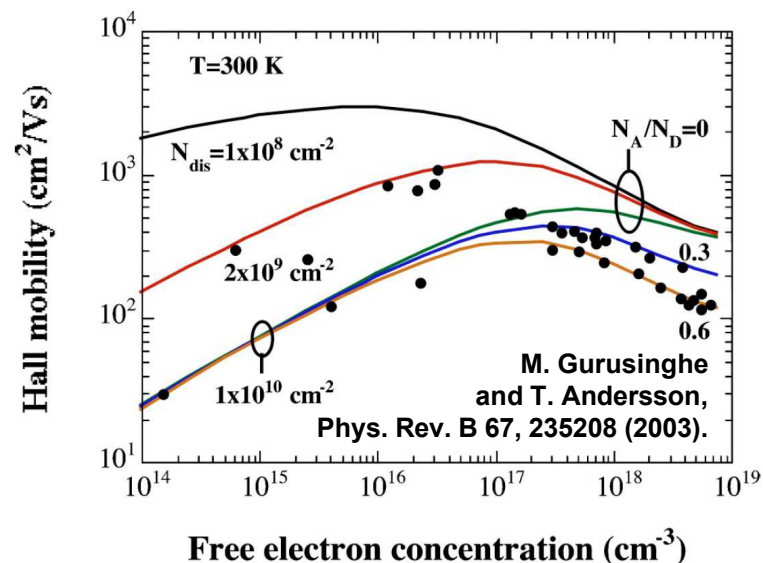
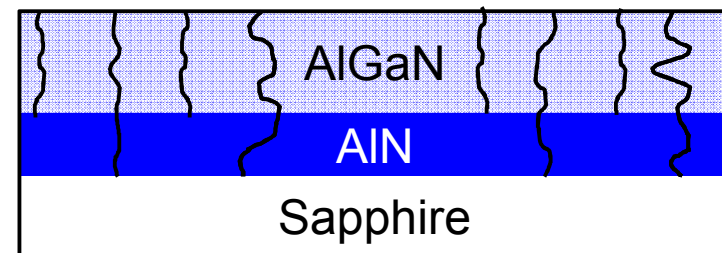
- Lattice mismatch
- Impact device performance

- **Act as compensating centers**

- Scatter charge carriers
- Reduce carrier mobility
- Reduce minority carrier lifetime

- **Act as current leakage sites**

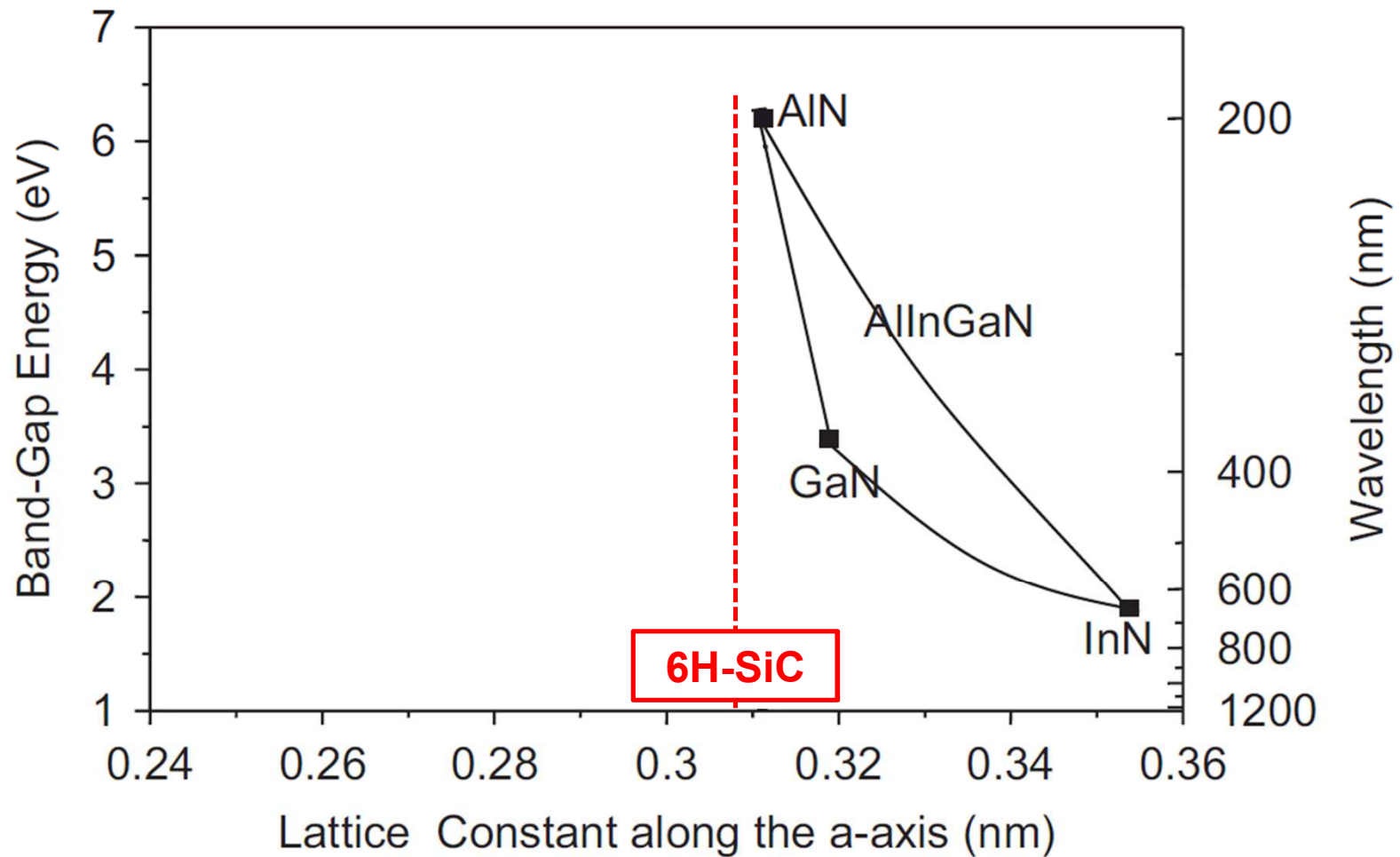
- Premature breakdown voltages
- Increases resistive losses



Can we prevent dislocations via novel materials?

AlInGaN materials system

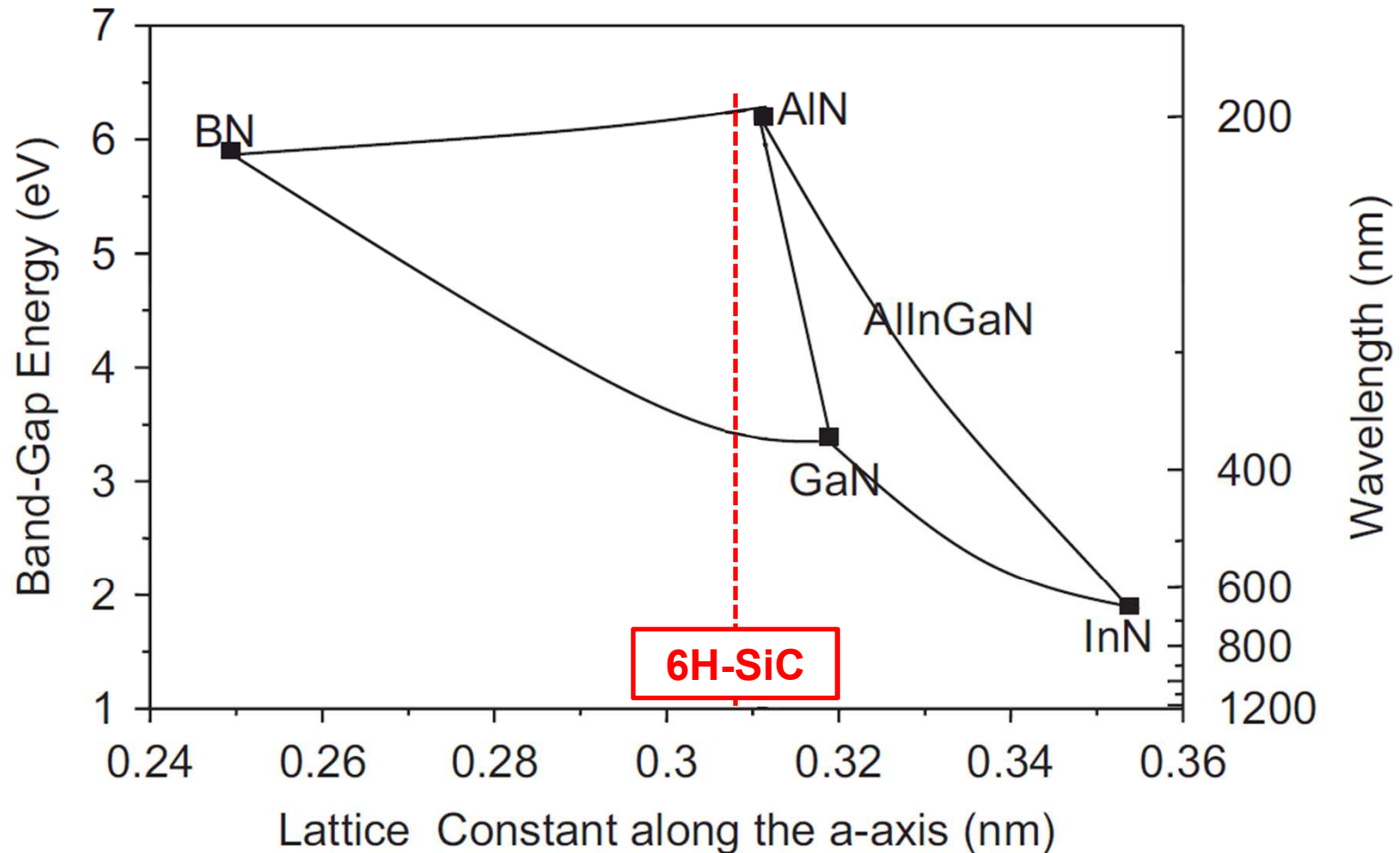
Familiar territory, well explored



after Ougazzaden, J. Cryst. Growth (2007).

Boron-containing nitride alloys

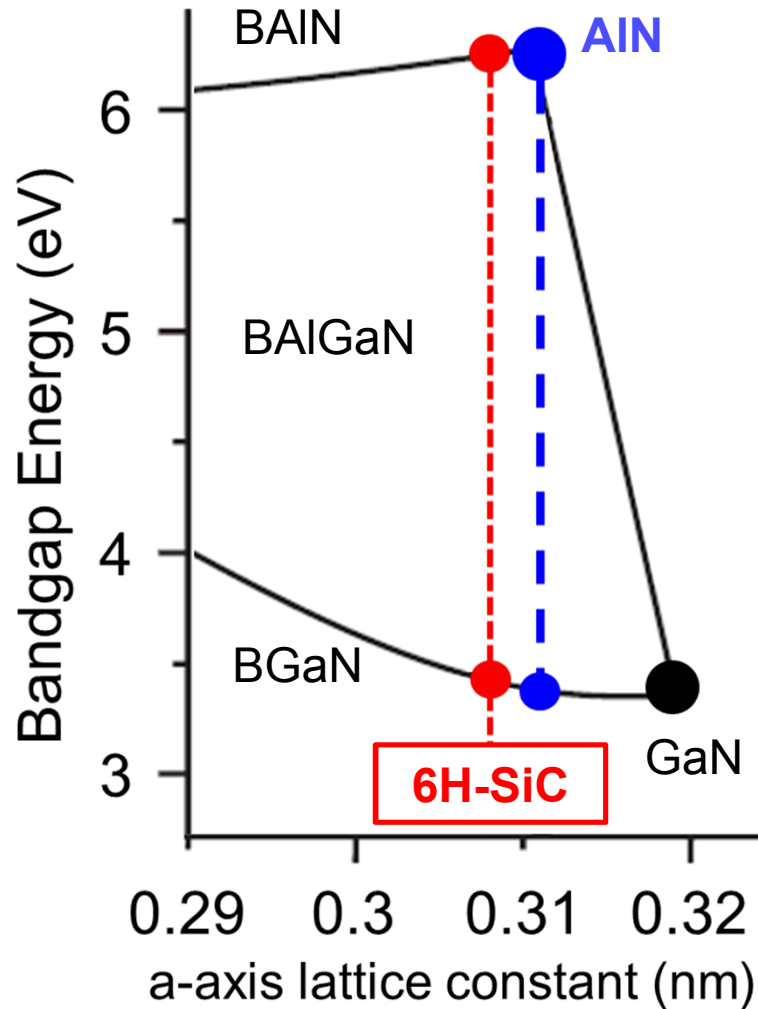
New opportunities for wide-bandgap lattice engineering!



after Ougazzaden, J. Cryst. Growth (2007).

BGaN and BAlN

Expanded view:



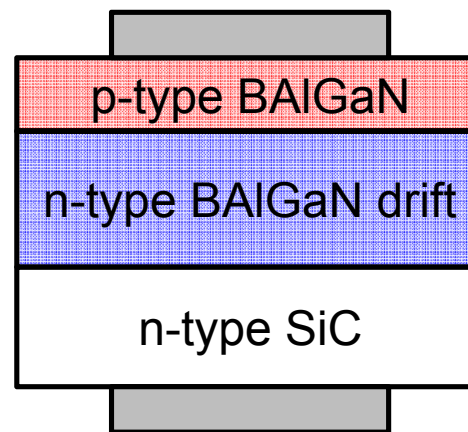
Lattice-matching opportunities:

$B_{0.08}Ga_{0.92}N$ on AlN

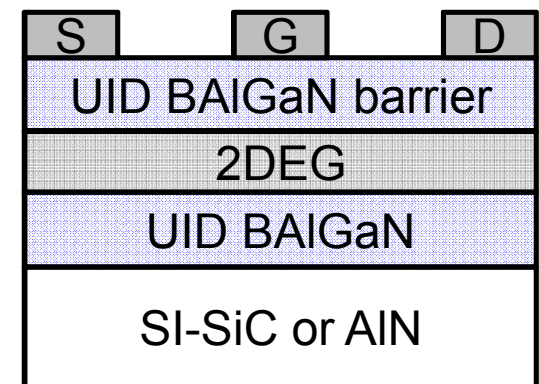
$B_{0.17}Ga_{0.83}N$ on 6H-SiC

$B_{0.05}Al_{0.95}N$ on 6H-SiC

Device prospects:



Two-terminal
vertical device

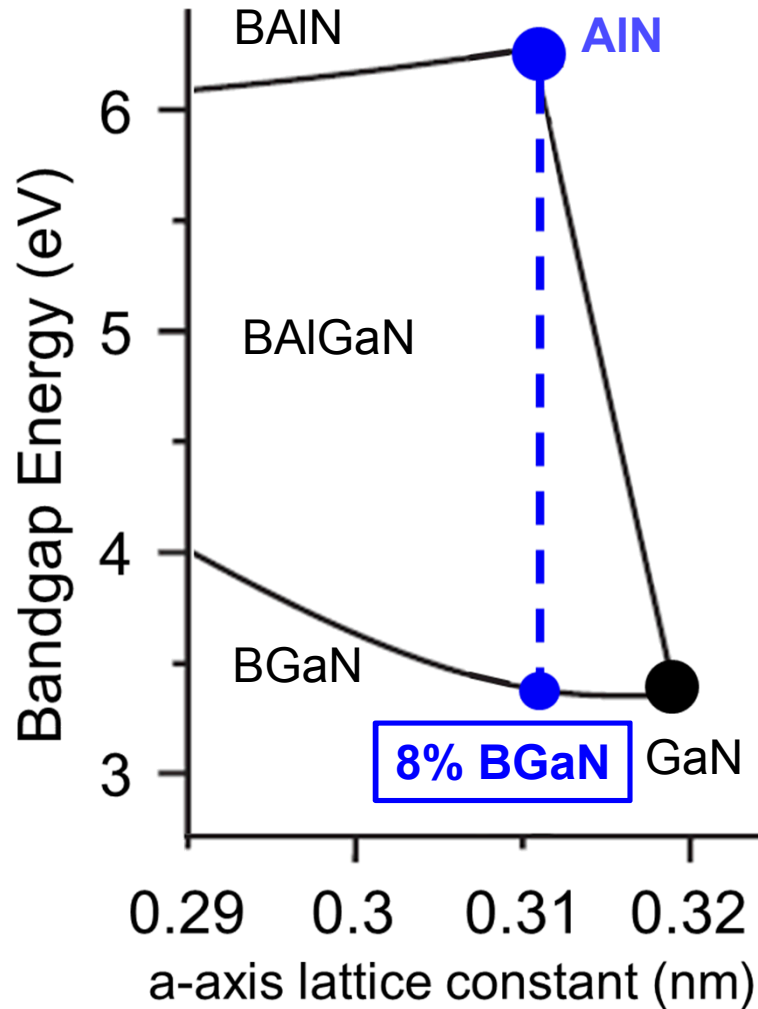


Three-terminal
lateral device

after Ougazzaden, J. Cryst. Growth (2007).

BGaN and BAlN

Expanded view:



Lattice-matching opportunities:

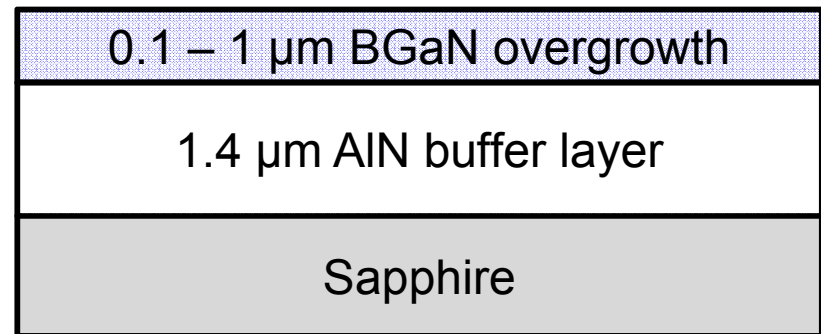
$B_{0.08}Ga_{0.92}N$ on AlN

$B_{0.17}Ga_{0.83}N$ on 6H-SiC

$B_{0.05}Al_{0.95}N$ on 6H-SiC



Growth structure:

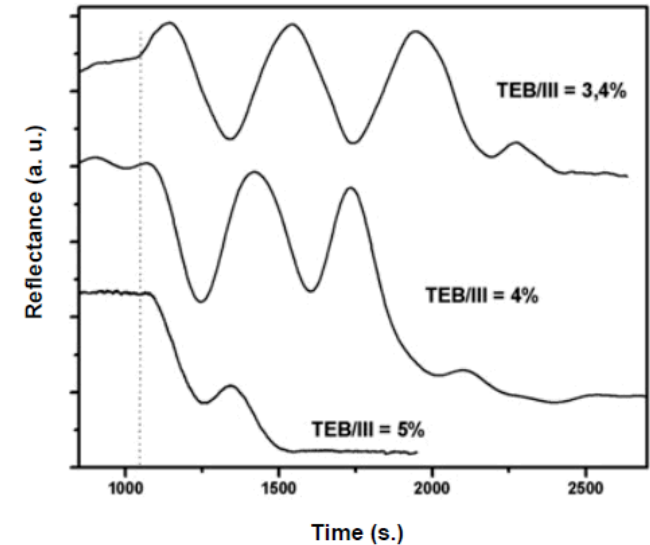


after Ougazzaden, J. Cryst. Growth (2007).

Challenges

Growth of BAlGaN alloys not yet extensively studied

- Few groups actively publishing on BGaN
- Typical B mole fractions **less than 5%**
- Simplistic calculations of B% in literature
- Rough and pitted surfaces



Growth temperature incompatibility

- Best boron nitride (BN) grown at $>1300\text{ }^{\circ}\text{C}$
- Best GaN grown at $\sim 1050\text{ }^{\circ}\text{C}$ – GaN decomposes above $1000\text{ }^{\circ}\text{C}$

Potential obstacle with incomplete miscibility of B in GaN

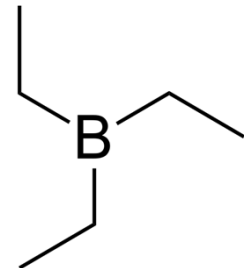
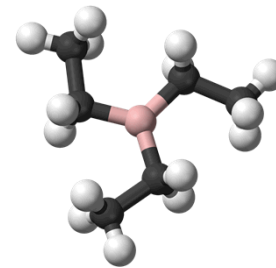
T. Malinauskas et al., Phys. Status Solidi B **252**, (2014).
 S. Gautier et al., J. Cryst. Growth **312**, 641 (2010).
 S. Gautier et al., J. Cryst. Growth **315**, 288 (2011).
 T. Takano et al., J. Cryst. Growth **237**, 972 (2002).
 A. Polyakov, et al., Journal of Elec Materi **26**, 237 (1997).

S. Majety et al., Appl. Phys. Lett. **101**, 051110 (2012).
 C. Wei and J. Edgar, J. Cryst. Growth **208**, 179 (2000).
 L. K. Teles et al., Physica E **13**, 1086 (2002).
 V. Ravindran, Thesis (2013).

Goals

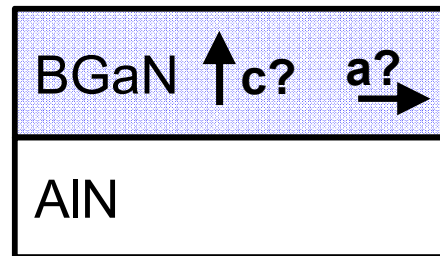
Goal: 8% BGaN lattice-matched to AlN

- Less than 8%, compressive strain
- Use triethylboron (TEB)
- Start investigation with GaN conditions + TEB
- 800 °C, 75 Torr



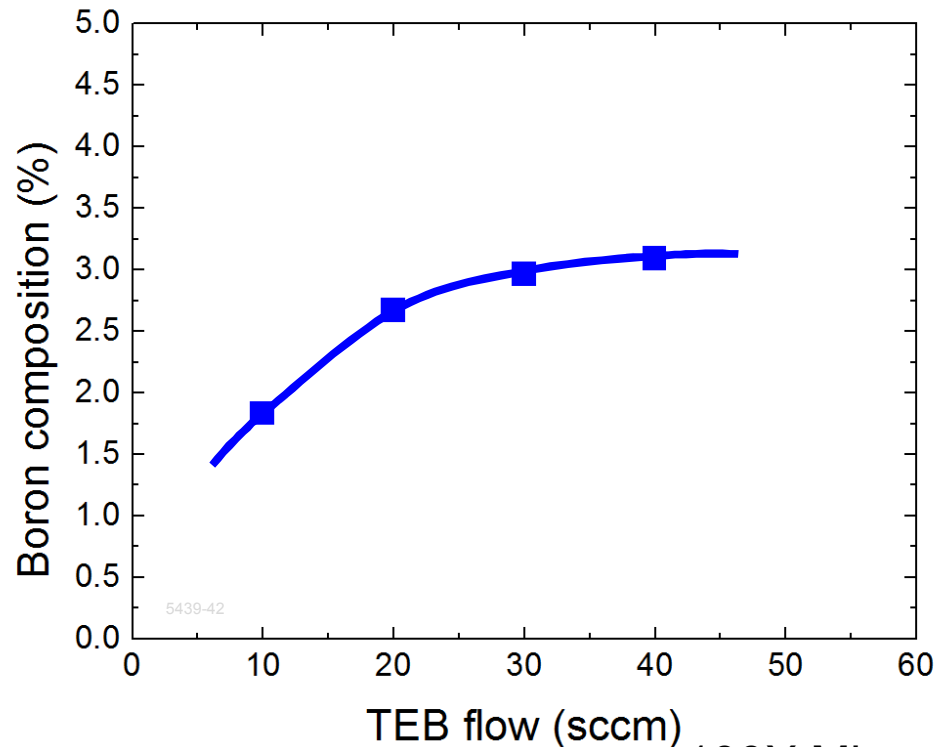
Develop accurate calculation of B% via XRD

- Calculate in-plane and out-of-plane lattice constants



INPUT										
PARAMETERS										
	a	c	a/c	c13	c33	-2c13/c33				lambda
	(Å)	(Å)	(---	(GPa)	(GPa)	(---				(Å)
GaN:	3.1889	5.1852	1.62602	103	405	-0.5086				1.5405981
AlN:	3.1113	4.9799	1.60058	108	373	-0.5791				
BN:	2.55	4.17	1.63529	74	1077	-0.1374				
2-THETA										
MEASUREMENTS										
	epilayer xrd	epilayer target	epilayer design	2theta (deg.)	h,k,m=					
	peak	composition	thickness	n=	(000n)	(000n)	(000n)	(hk.m)	(hk.m)	(hk.m)
	id:	(---	(um)		d	c	K	d	K	Kpara
					(Å)	(Å)	(1/Å)	(Å)	(1/Å)	(1/Å)
direct beam:		-----	-----	4	0.0000			0.0000		
peak 1: AlN		-----	-----	76.3927	1.245717	4.9829	5.04383	2.370022	2.6511	2.3320
peak 2: BGaN		-----	-----	73.2200	1.291657	5.1666	4.86444	2.428581	2.5872	2.2835
ANALYSIS RESULTS										
	measured a	measured c	measured BN mole fraction	unstrained a0	unstrained c0	parallel strain	perp strain	pseudomorphic coh. strain on AlN	measured relaxation on AlN	
	(Å)	(Å)	(---	(Å)	(Å)	(---	(---	(---	(%)	
AlN	3.1111	4.9829	0.990	3.1121	4.9820	-0.00031	0.00018			
BGaN	3.1772	5.1666	0.018	3.1772	5.1666	-0.00001	0.00001	-0.02080	99.9%	
ITERATE *estimated alloy										
	estimated alloy x	computed alloy x								
	1.0000	0.9899								
	0.0183	0.0183								

Initial results



Observations

- Smooth surfaces with low TEB
- B% sublinear vs. TEB flow
- Some B not incorporated
- >4% B, but rough surfaces

Yellowish clusters

- Excess boron?
- Increase in size with more TEB

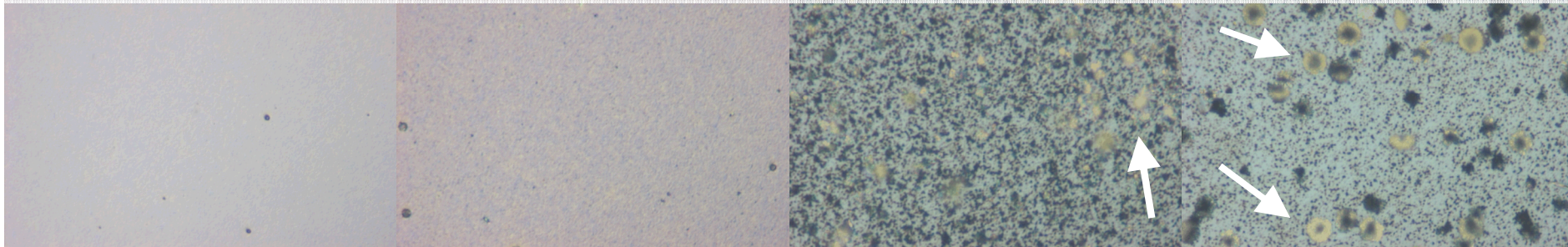
100X Microscope Images

10 sccm TEB

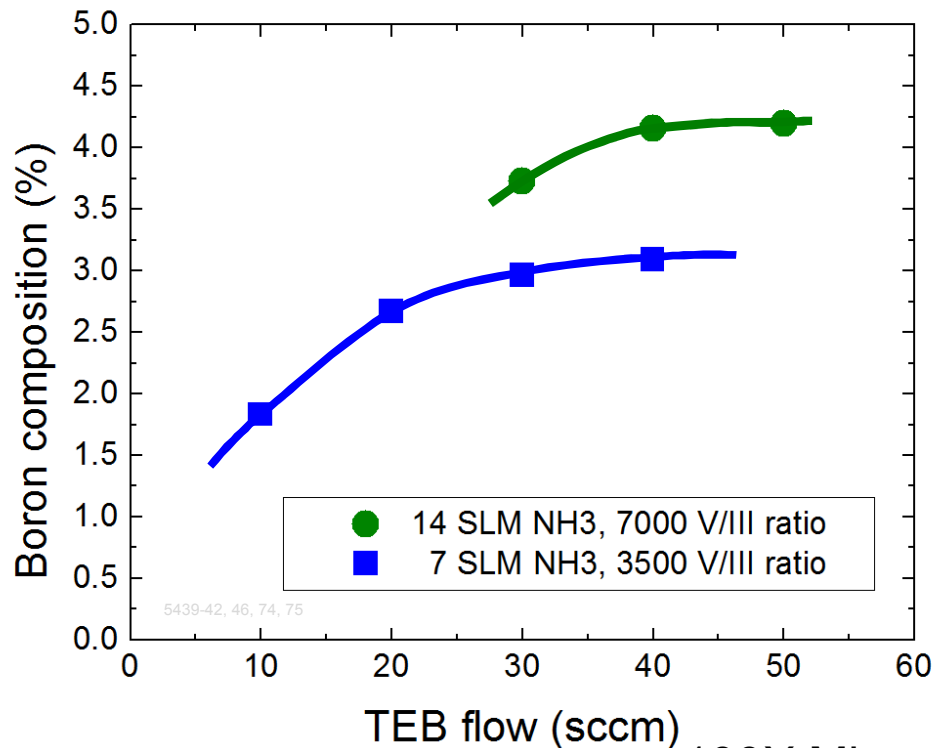
20 sccm TEB

30 sccm TEB

40 sccm TEB



Initial results



Higher V/III ratio

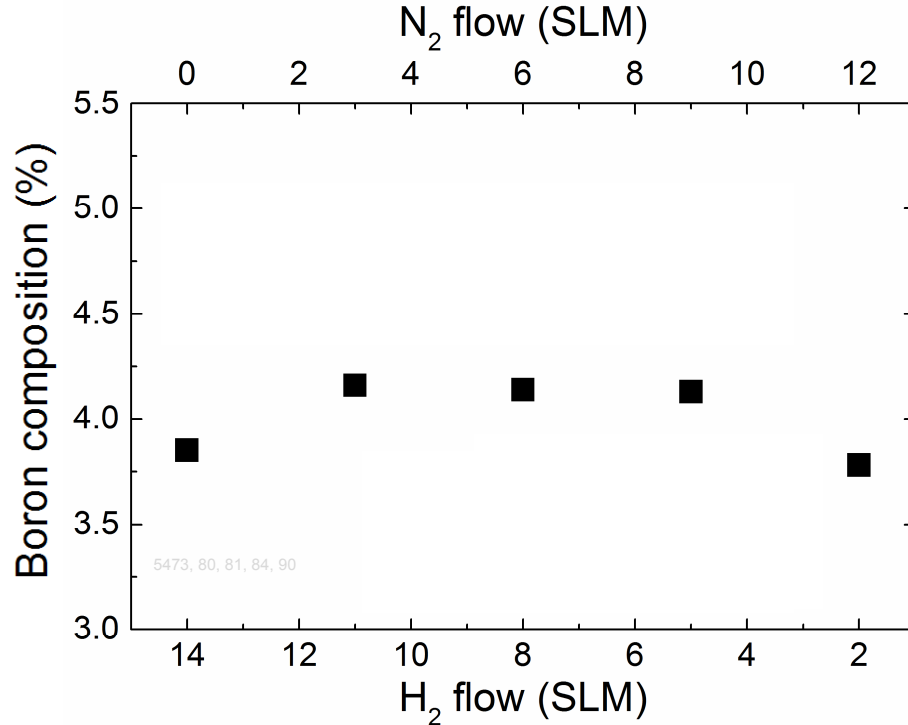
- More B incorporated
- Still sublinear
- Still exhibits yellow clusters
- >4% B, but rough surfaces

**Non-optimized conditions,
warrants phase-space search**

100X Microscope Images



H₂:N₂ carrier gas ratio



Morphology

- More H₂ → rough surface
- More N₂ → pitted surface

100X Microscope Images

11 L H₂ / 3 L N₂

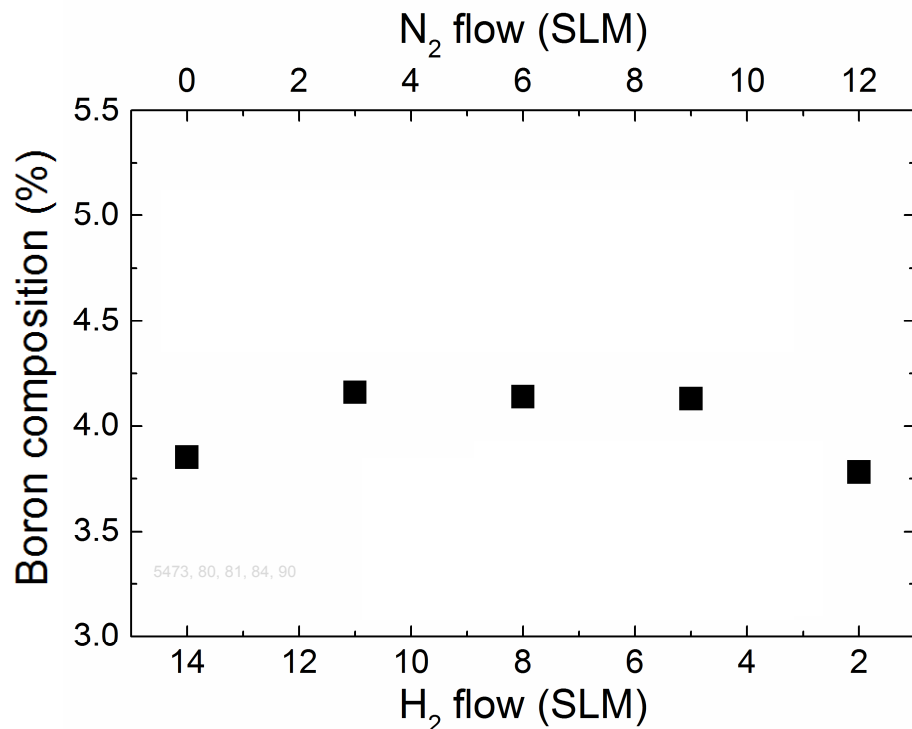
8 L H₂ / 6 L N₂

5 L H₂ / 9 L N₂

2 L H₂ / 12 L N₂

800 °C

H₂:N₂ carrier gas ratio



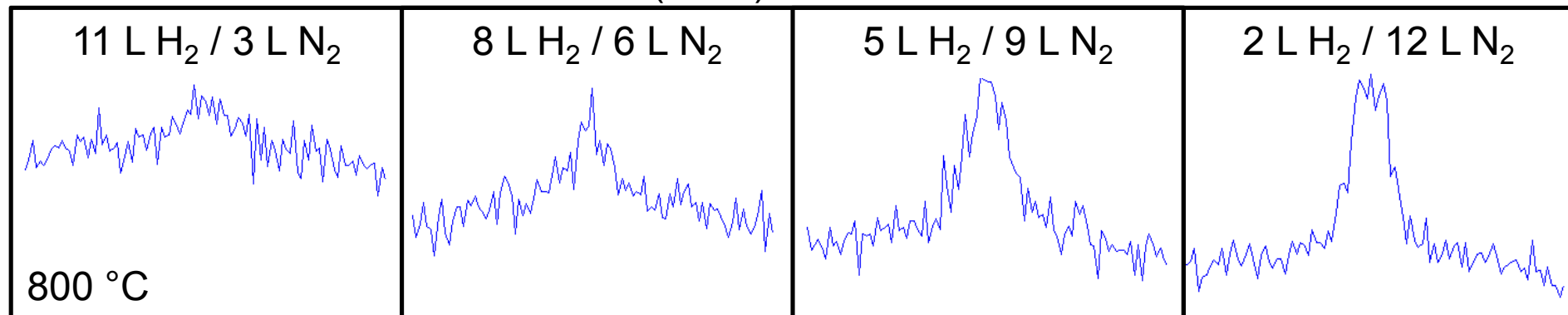
Morphology

- More H₂ → rough surface
- More N₂ → pitted surface

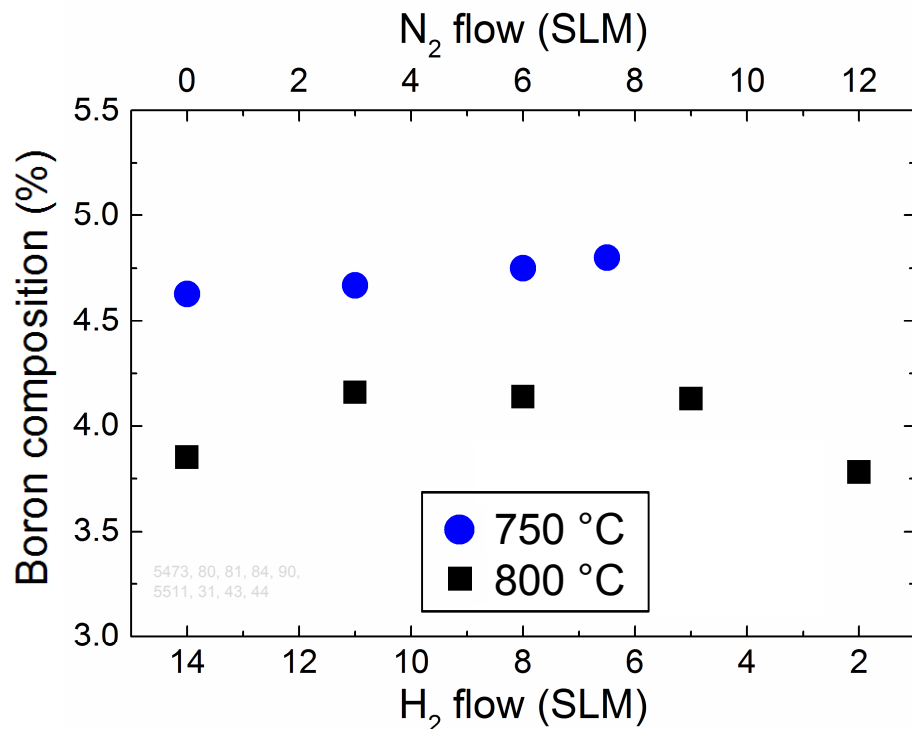
Crystal structure

- More H₂ → isolated, tilted grains
- More N₂ → coherent grains

(0002) XRD RCs



H₂:N₂ carrier gas ratio



Lower temperature

- H₂:N₂ trend still applies
- Smoother surfaces
- Higher maximum B% (**4.8%**)

100X Microscope Images

14 L H₂ / 0 L N₂

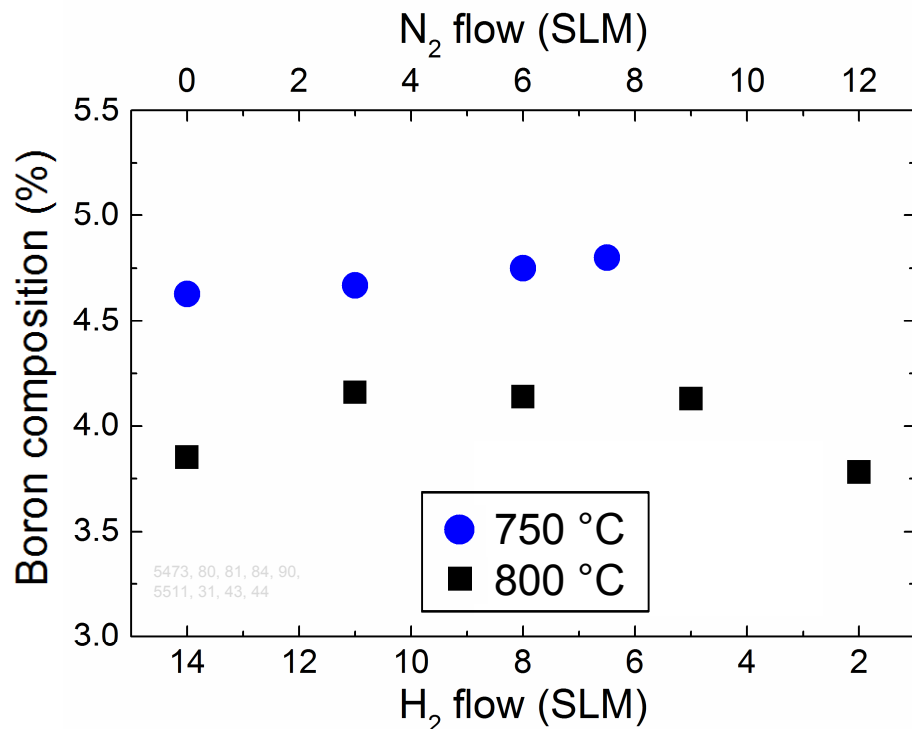
11 L H₂ / 3 L N₂

8 L H₂ / 6 L N₂

6.5 L H₂ / 7.5 L N₂

750 °C

H₂:N₂ carrier gas ratio



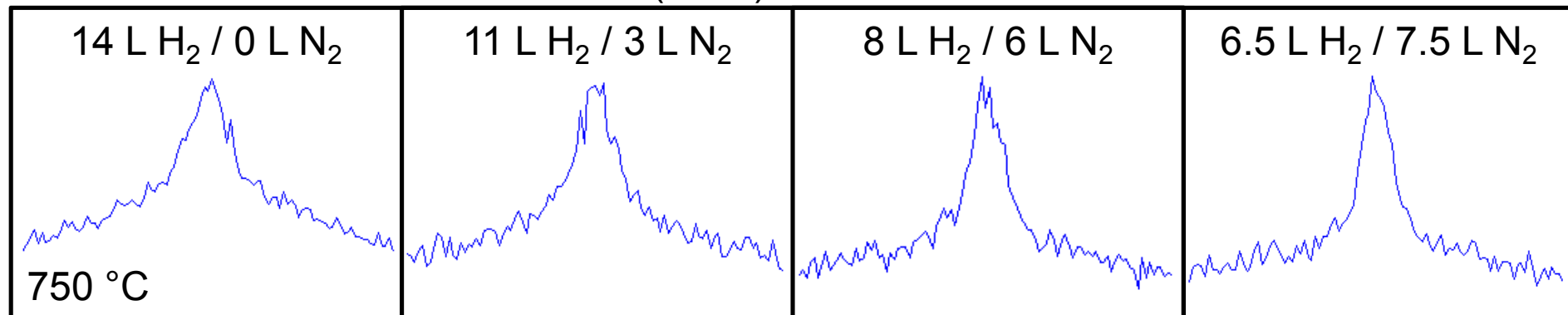
Lower temperature

- H₂:N₂ trend still applies
- Smoother surfaces
- Higher maximum B% (**4.8%**)

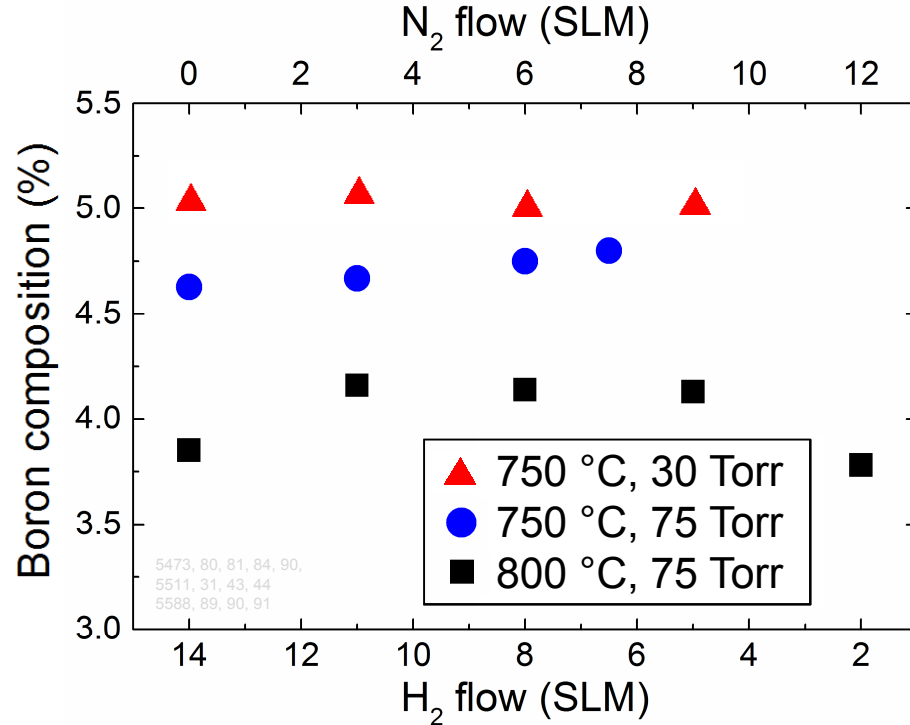
Crystal structure

- More coherent grains overall
- 400 arcsec (0002) RC FWHM

(0002) XRD RCs

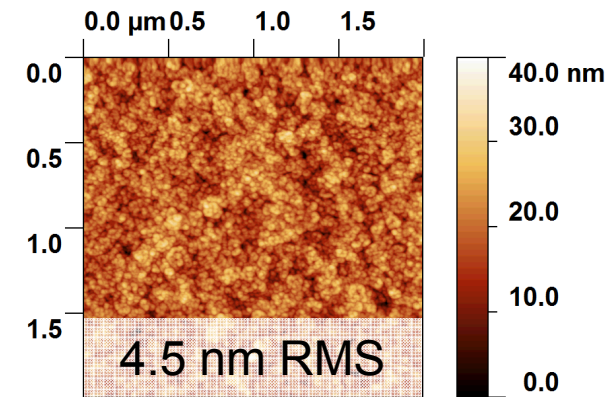


H₂:N₂ carrier gas ratio



Lower pressure

- 30 Torr vs 75 Torr
- Much smoother surfaces
- Higher maximum B% (**5%**)



100X Microscope Images

14 L H₂ / 0 L N₂

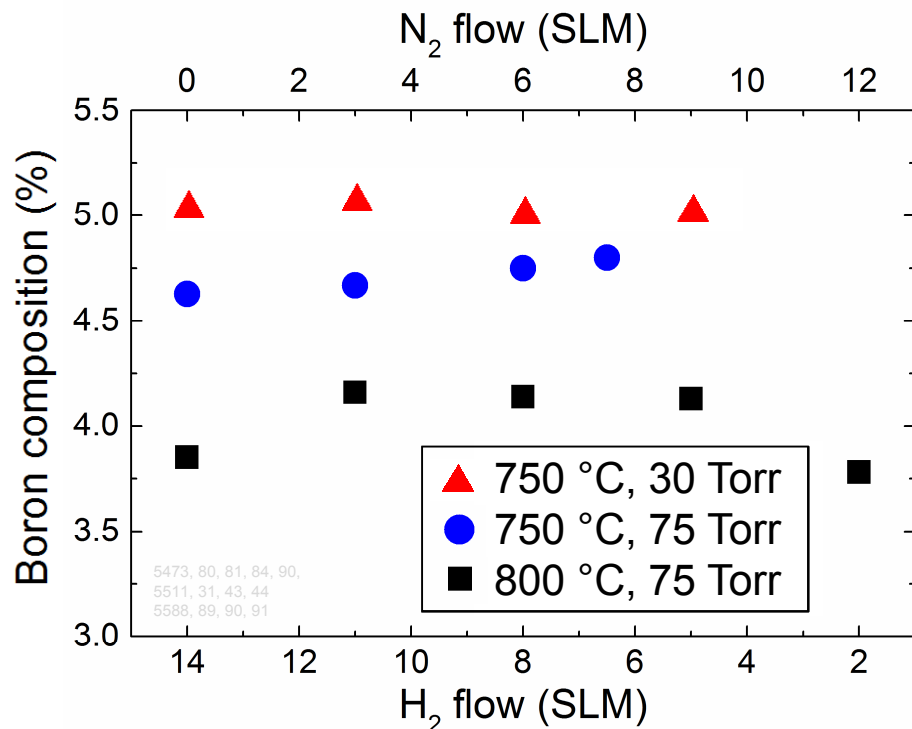
11 L H₂ / 3 L N₂

8 L H₂ / 6 L N₂

5 L H₂ / 9 L N₂

750 °C, 30 Torr

H₂:N₂ carrier gas ratio



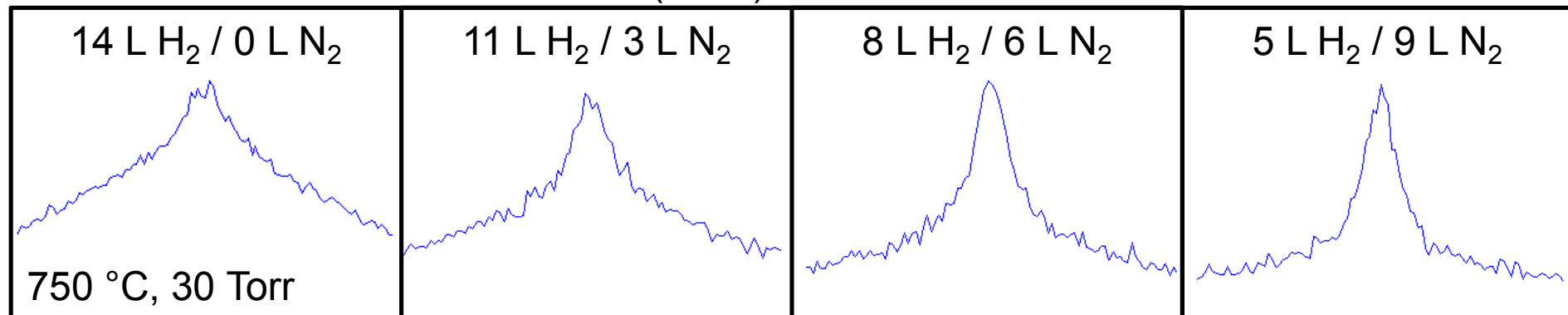
Lower pressure

- 30 Torr vs 75 Torr
- Much smoother surfaces
- Higher maximum B% (**5%**)

Crystal structure

- More coherent grains overall
- 370 arcsec (0002) RC FWHM

(0002) XRD RCs



Summary

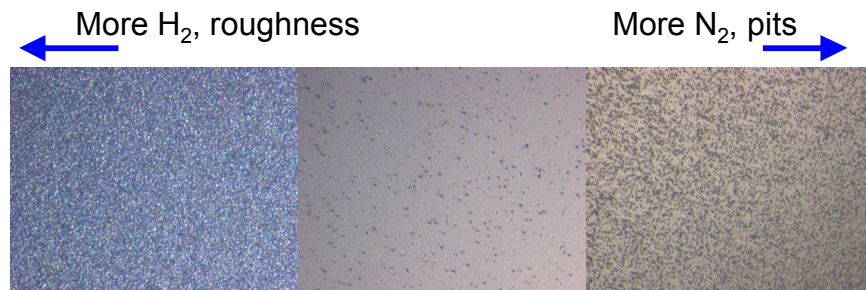
Growth of BGaN alloys on AlN

■ B mole fraction calculation

- Developed a more accurate B% estimation method
- Initial work – 2-3% B

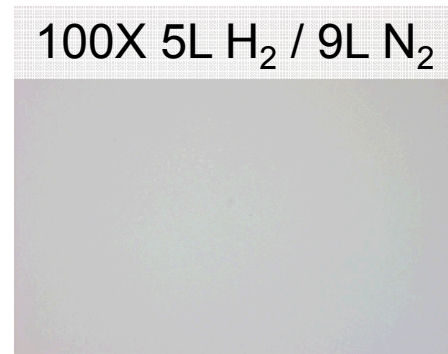
■ Explored effect of carrier gas

- $H_2:N_2$ ratio
- More $H_2 \rightarrow$ rough, tilted grains
- More $N_2 \rightarrow$ pitted, coherent grains



■ Reduced growth pressure and temperature

- Obtained 5% boron mole fraction
- Smooth surfaces
- Improving crystal quality



Exceptional service in the national interest



Growth of BGaN Alloys on AlN

Michael W. Moseley,

Daniel D. Koleske, Stephen R. Lee, Andrew A. Allerman

Sandia National Laboratories,
Albuquerque, New Mexico, USA, 87123

Exceptional service in the national interest

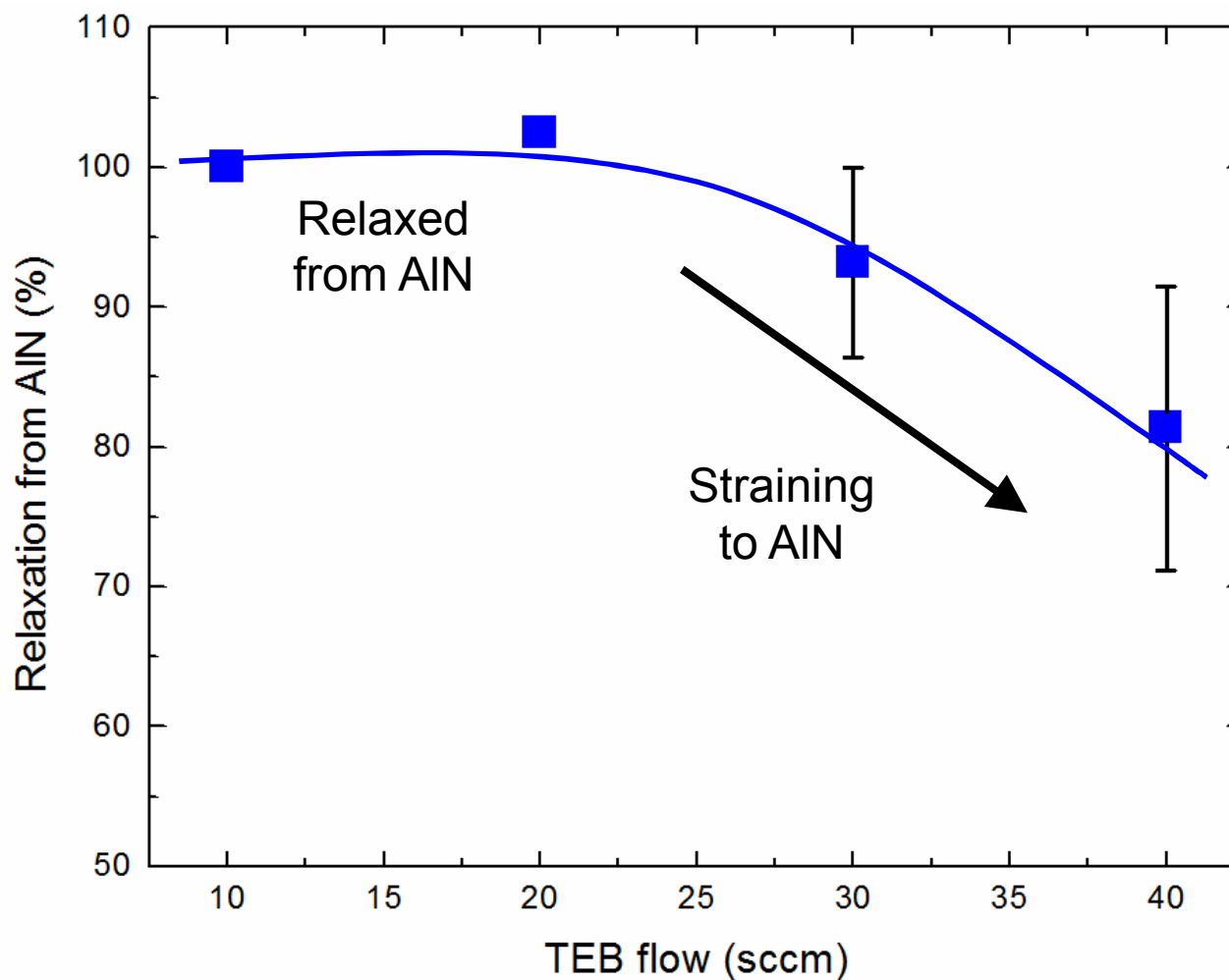


Backup figures



Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000. SAND NO. 2011-XXXXP

BGaN grown on AlN



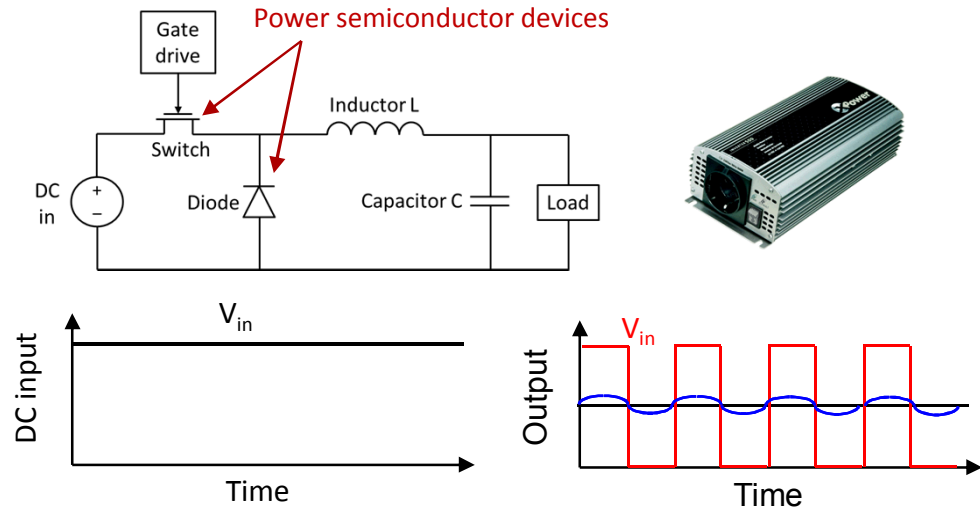
T_G : 800C
7 SLM NH₃
~150 nm thick

III-Nitride Power Electronics

Application of solid-state electronics for routing, control, and conversion of electrical power

Impact

- Higher voltages
- Higher frequency
- Smaller size
- Higher temperature
- Reduce active cooling
- Reduce passive components



Applications

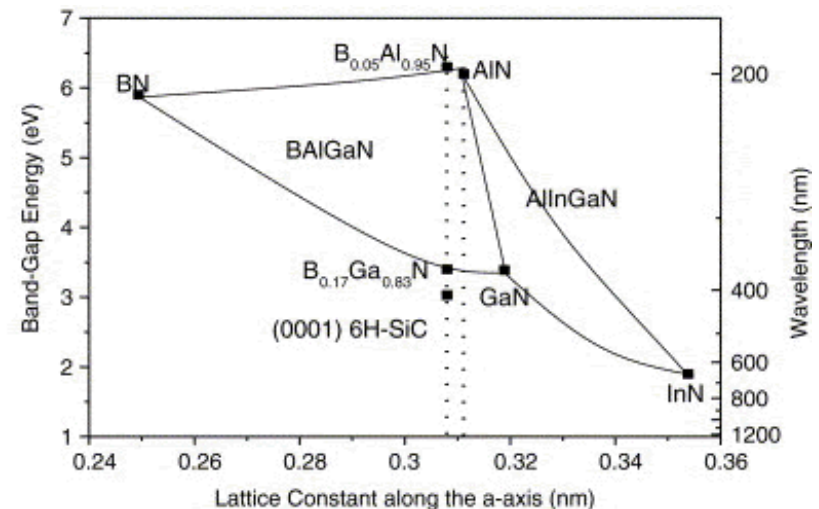
- Inverters, rectifiers, converters
- Replace aging weapons components
- Enable new systems and capabilities



S. Pearton, F. Ren, A. Zhang, and K. Lee, Mat. Sci. Eng. R **30**, 55 (2000).
U. K. Mishra et al., Proc. IEEE **96**, 287 (2007).

Group-III Nitrides - Solutions

- Investigate methods of dislocation density reduction
 - Epitaxial lateral overgrowth
 - Dislocation annihilation
- Circumvent mismatch by employing novel materials
 - Boron-containing III-nitride alloys
 - Lattice match to SiC, AlN, and even AlGaN



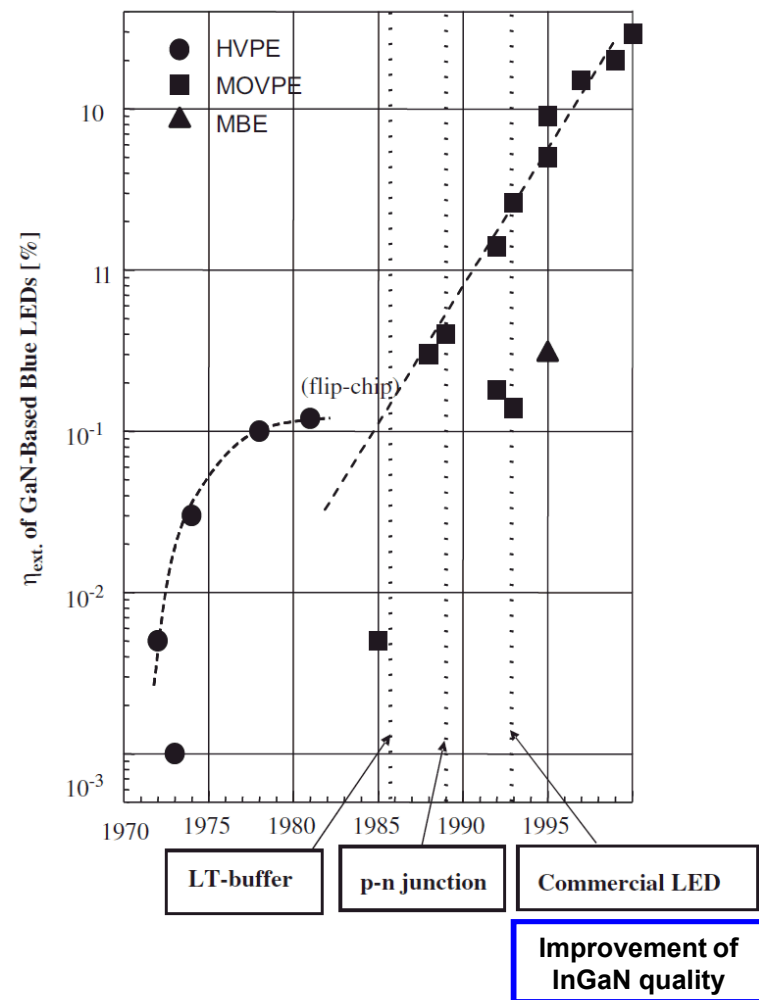
D. Cherns and M. Hawkrige, (2006).

Allerman et al., (2014).

Ougazzaden et al., (2007).

Dislocation Density Reduction

III-nitrides – device improvement follows material improvement



Buffer/nucleation layer design

Lattice-constant engineering

- Novel substrates

Migration-enhanced epitaxy (MEE)

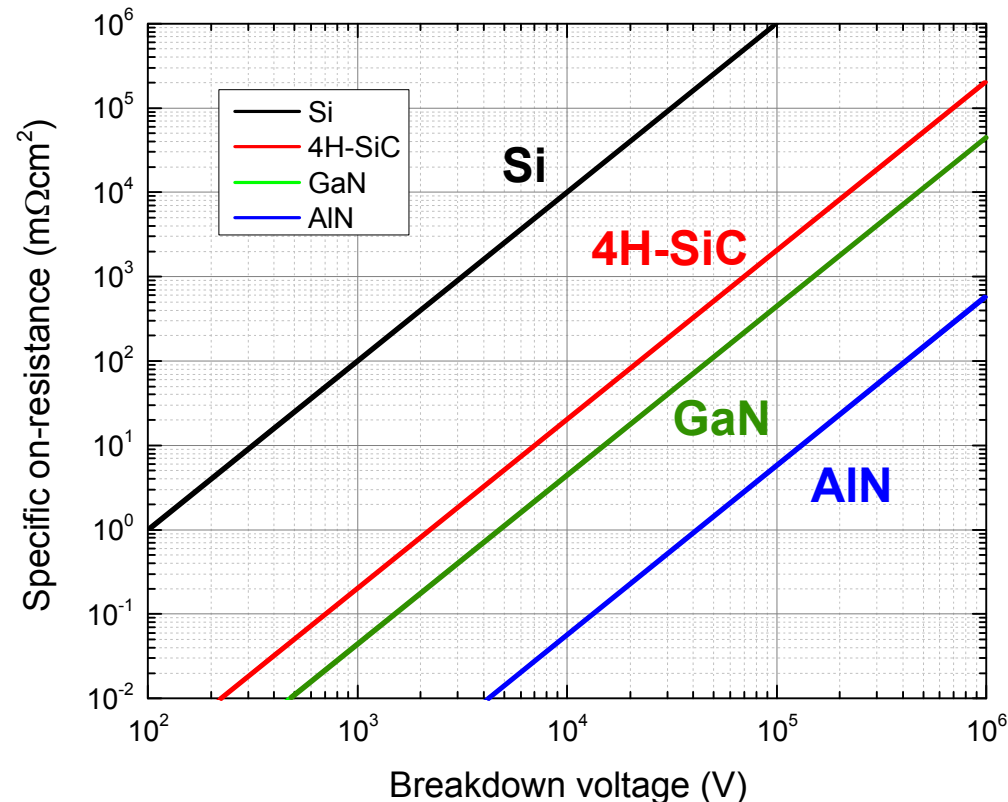
- Metal-modulated epitaxy (MME)
- Anion-modulated epitaxy (AME)
- Migration-enhanced metalorganic chemical-vapor deposition (MEMOCVD)
- etc.

I. Akasaki, "Key inventions in the history of nitride-based blue LED and LD", *J. Cryst. Growth* **300**, 2 (2007).

Group-III Nitrides - Properties

AlGaN materials system for power electronic applications

Unipolar Figure-of-Merit for Various Materials



Parameter	Si	GaAs	GaN	AlN
Bandgap, E _g [eV]	1.1	1.4	3.4	6.2
Breakdown electric field [MV/cm]	0.3	0.4	3.3	11.7
Thermal conductivity [W/cm K]	1.5	0.5	2	2.7
Saturated electron drift velocity [10 ⁷ cm/s]	1.0	1	2.5	1.8

High breakdown electric fields

High thermal stabilities

High thermal conductivities

Radiation resistant

M. F. Schubert, Ph.D. Thesis, Virginia Polytechnic Institute and State University, 2009.
J. Wu, *J. Appl. Phys.* **106**, 011101 (2009).
L. Liu and J. H. Edgar, *Mater. Sci. Eng., R* **37**, 61 (2002).

T. P. Chow, *Microelectron. Eng.* **83**, 112 (2006).
D. I. Florescu et al., *Appl. Phys. Lett.* **77**, 1464 (2000).

Future work

- **Thicker films**
 - Enabled by lower pressure/temperature
- **n- and p-type doping**
 - Lower hole activation energy for Mg
- **Increase B% to lattice match to AlN**
 - Potential miscibility gap beyond 5%

