



Solution Chemistry, Substrate, and Processing Effects on the Chemical Heterogeneity in PZT Films

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Collaborators

Sandia National Labs

Microscopy

- Paul Kotula
- Bryan Gauntt
- Ping Lu
- Bonnie McKenzie
- Amy Allen
- Michael Rye

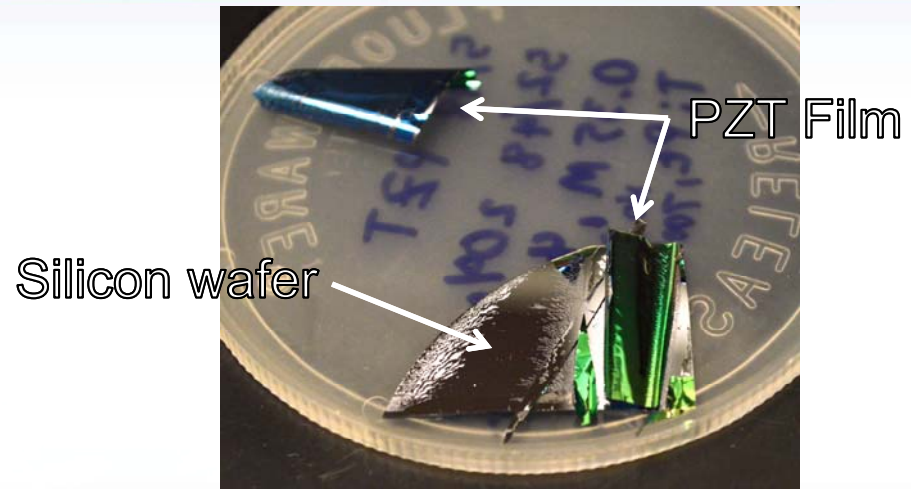
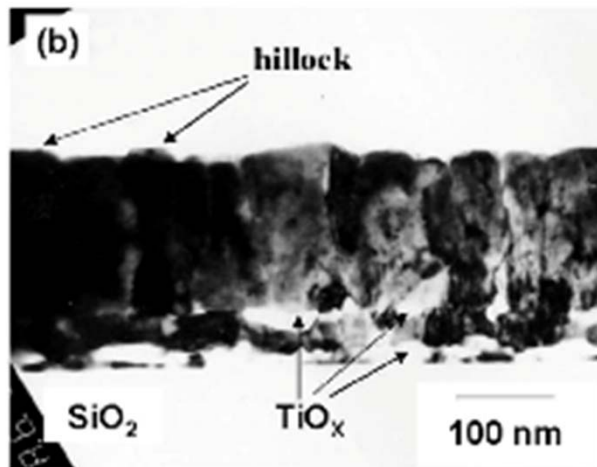
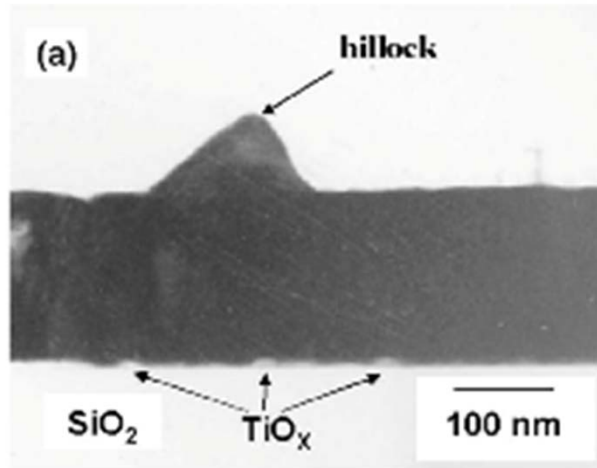
Solution/Film Preparation

- Chris T. Shelton
- Geoff Brennecka
- Dara V. Gough
- Erik Spoerke
- Kelsey Meyer
- Harlan Brown-Shaklee



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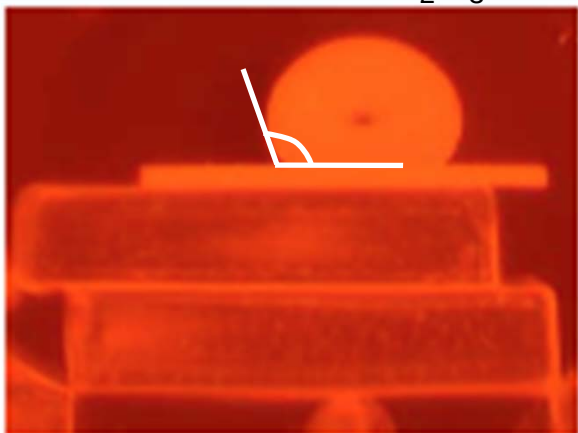
Issues with Platinized Silicon Substrates



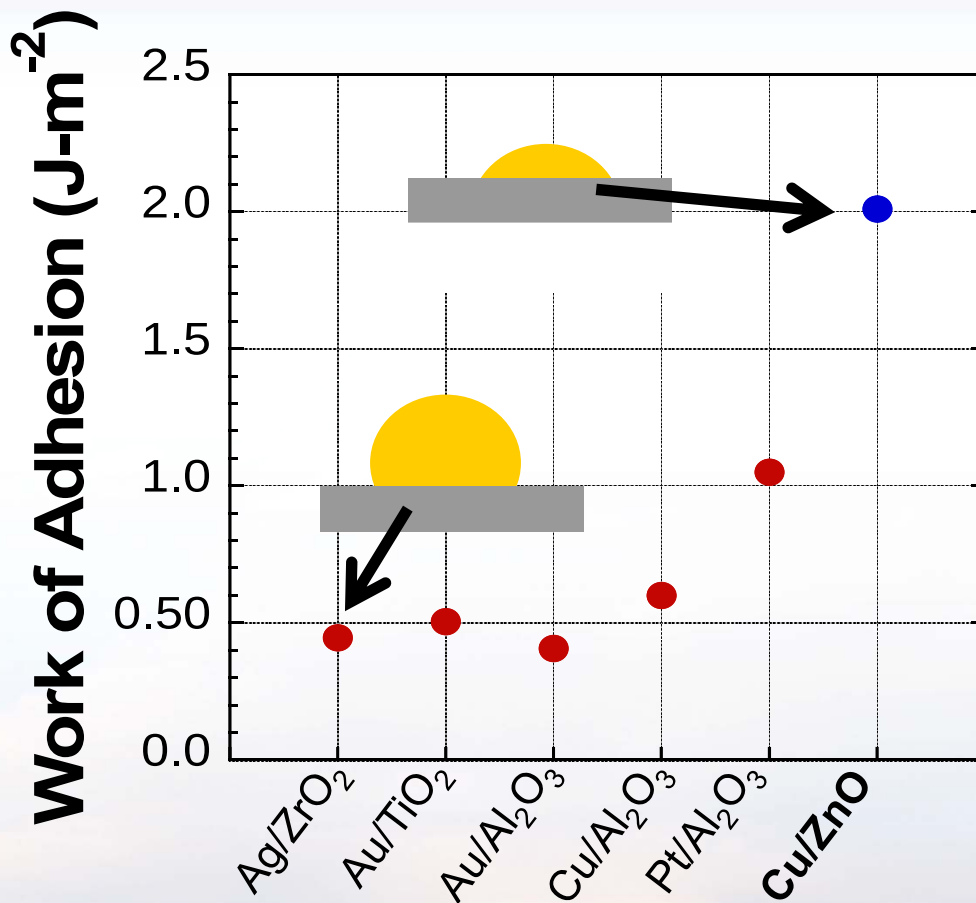
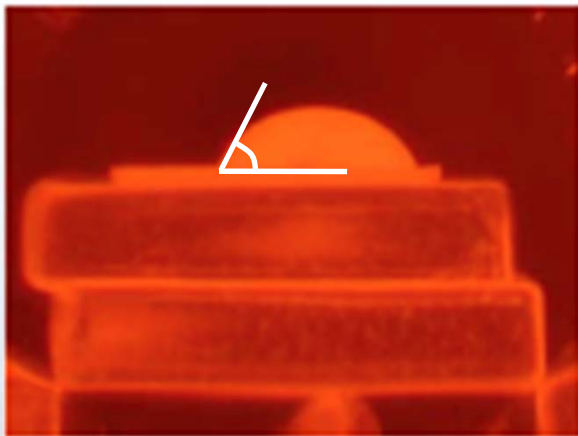
- Titanium 'adhesion' layers by far the most common
- Titanium expands upon oxidation causing hillocks
- Adhesion layers fail with large temperature swings ($>700^\circ$)

Metal Wetting and Adhesion

Molten Cu on Al_2O_3



Molten Cu on ZnO

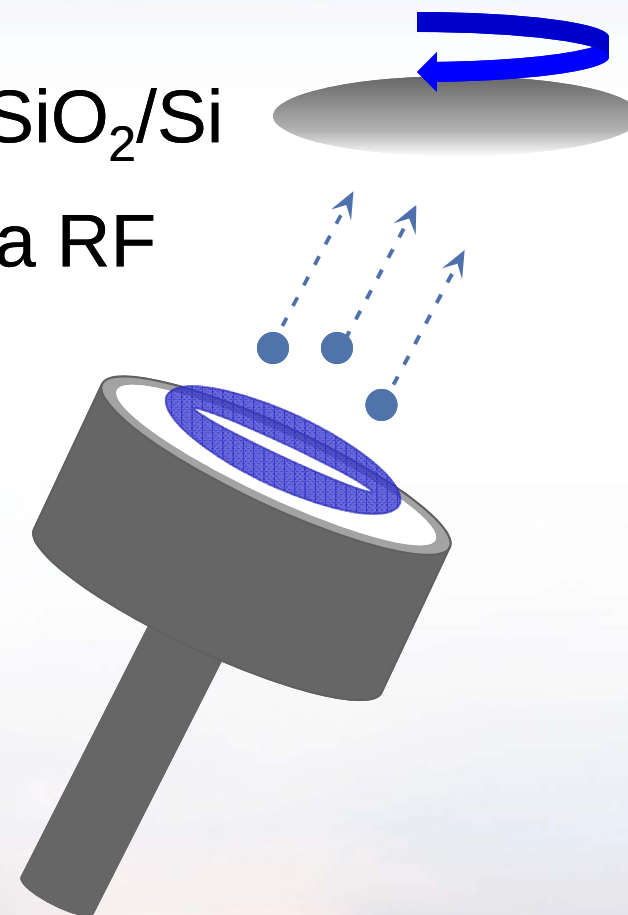


Can we use ZnO as an adhesion layer for Pt?

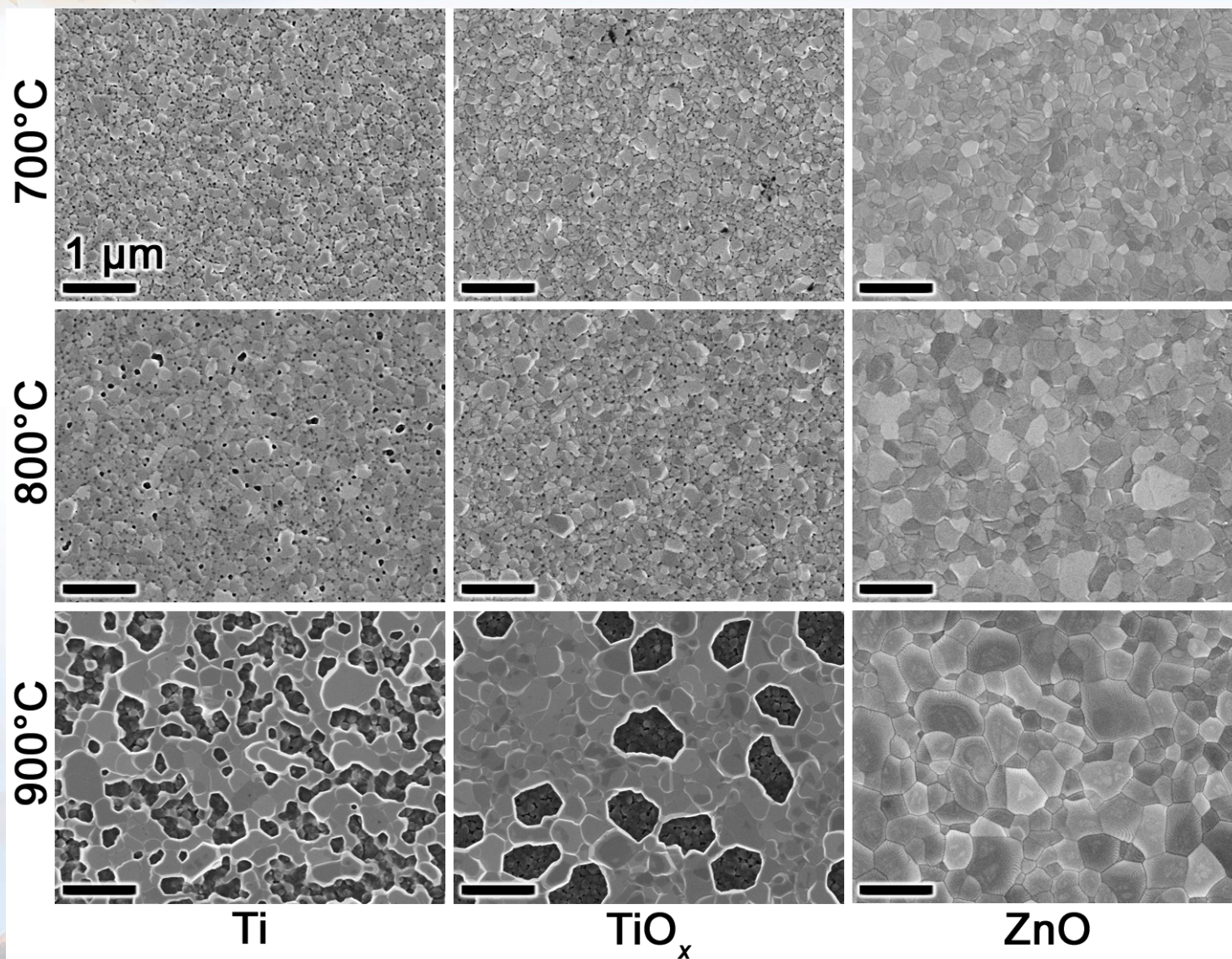
Cu and Pt both $Fm\bar{3}m$ metals

Pt/Si Substrate Preparation

- 100 nm Pt/40 nm buffer/400 nm SiO₂/Si
- Ti, TiO_x, ZnO, and Pt prepared via RF magnetron sputtering
 - 30° off-axis geometry
 - Ti: Ar atmosphere
 - TiO_x: Ar sputter, O₂ anneal
 - ZnO: 1:1 Ar:O₂ atmosphere
 - Pt: Ar Atmosphere



Electrode Temperature Stability



SNL IMO PZT CSD Film Preparation

$\text{Zr}(\text{Obu})_4 \cdot \text{BuOH} + \text{Ti}(\text{OiPr})_4$
Blend 5 min

Acetic Acid / Methanol
Chelating agent / Solvent

$\text{Pb}(\text{OAc})_4 + \text{La}(\text{OAc})_3 \cdot \text{H}_2\text{O}$
Dissolve at $\sim 90^\circ\text{C}$

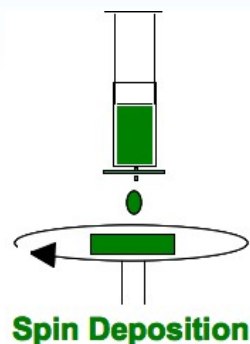
Acetic Acid / Methanol
Adjust Molarity

Spin Coat Deposition
4000rpm/30sec

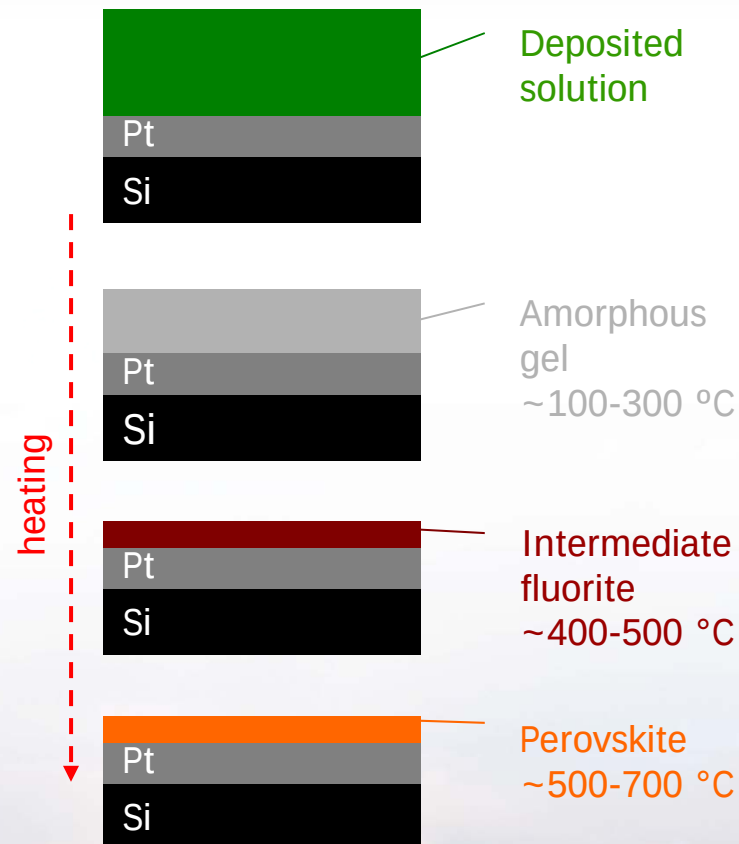
Pyrolysis

Crystallization

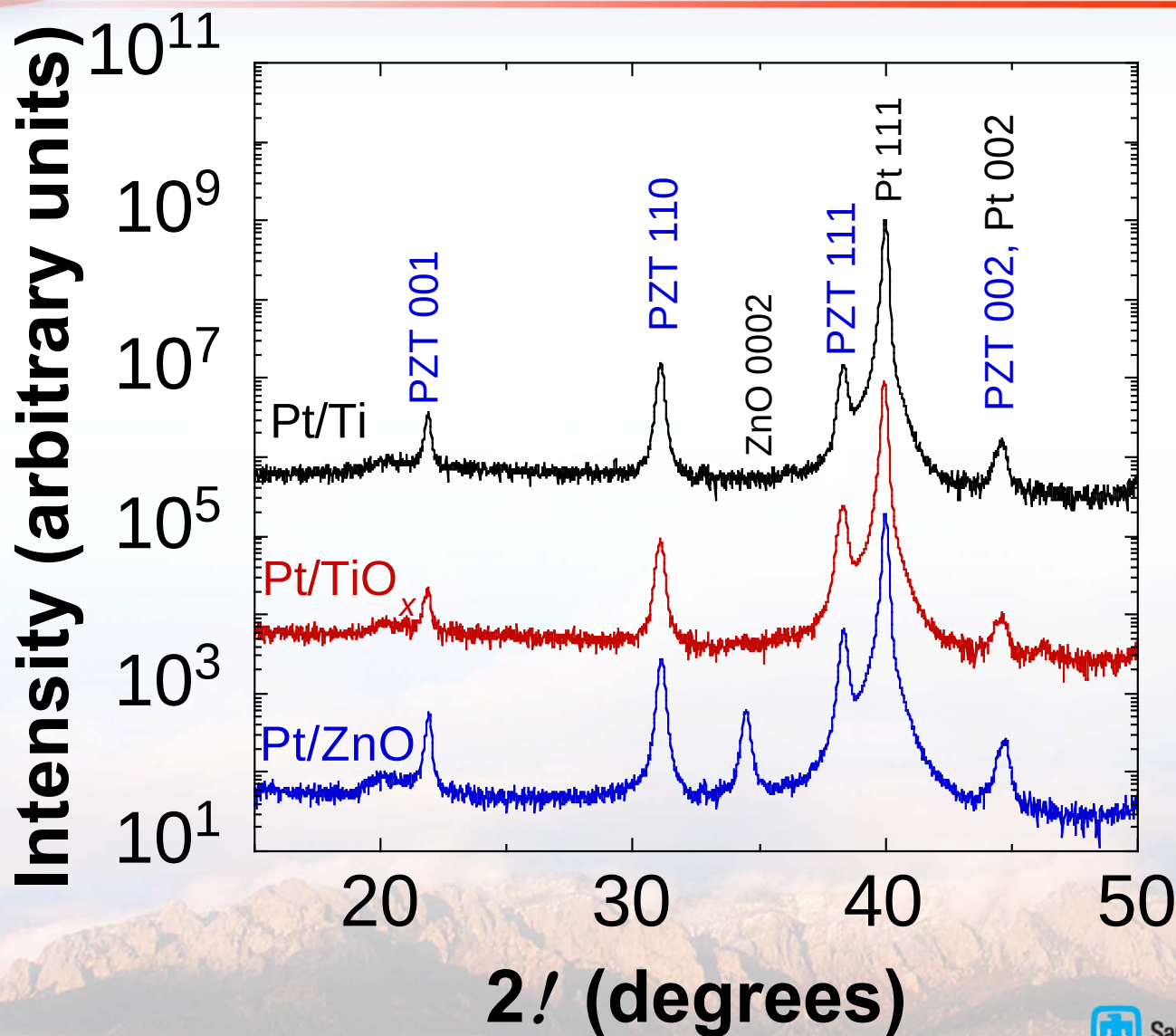
Sputter Pt electrodes



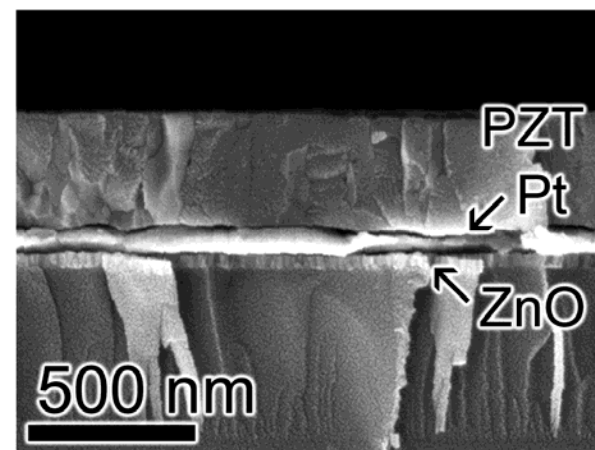
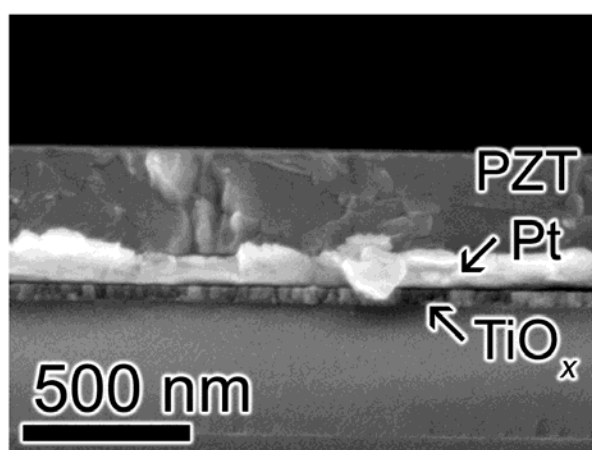
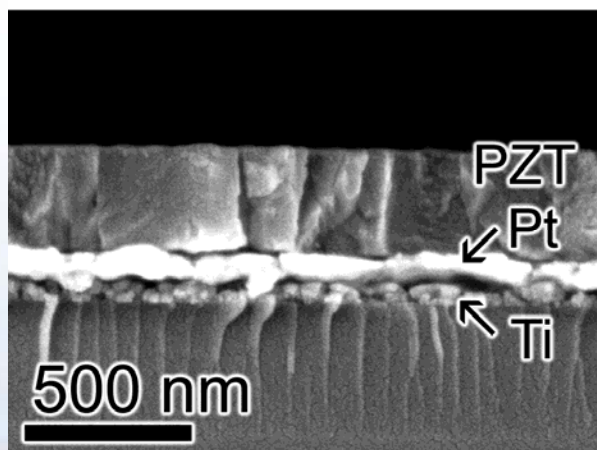
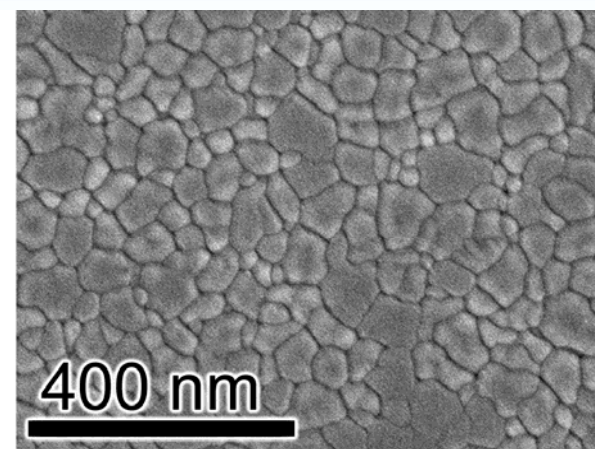
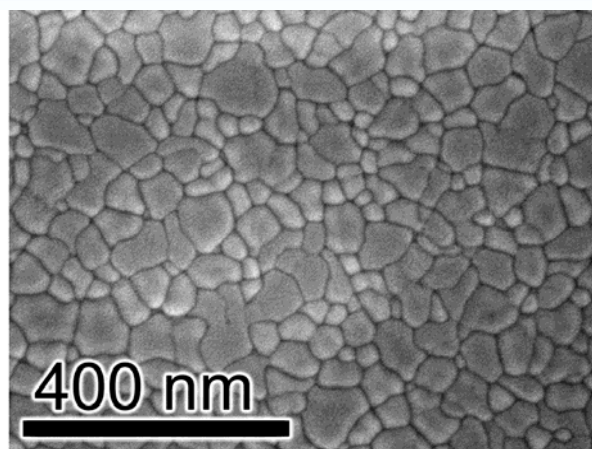
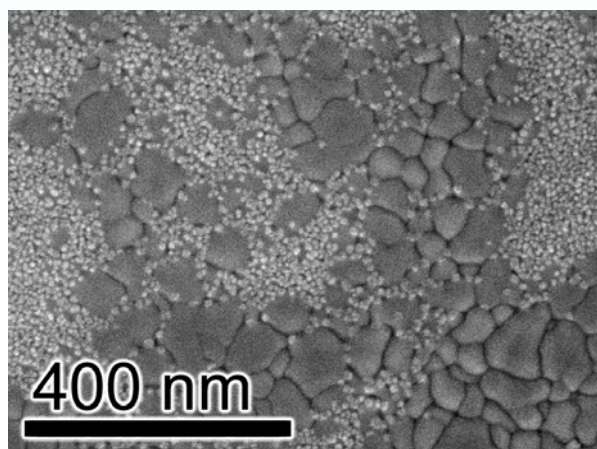
multiple
depositions



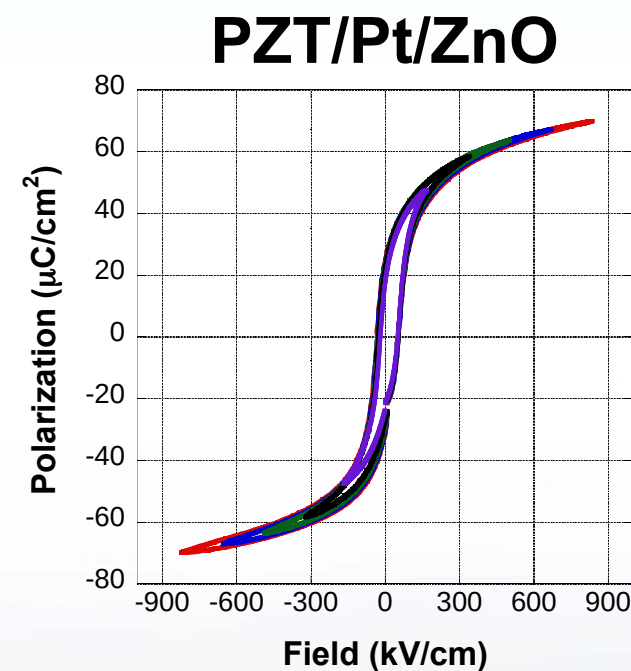
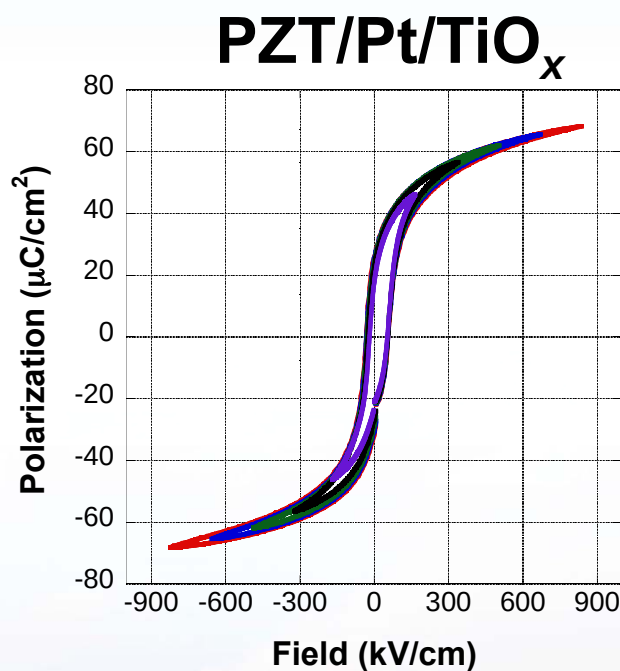
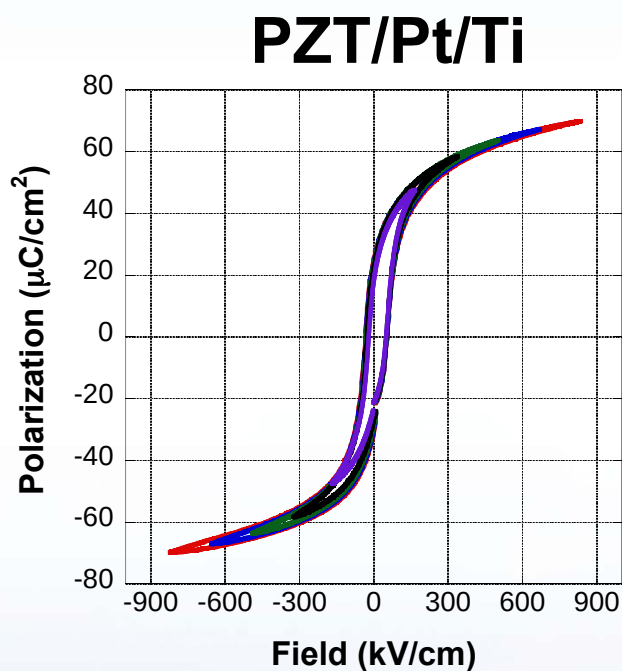
Film Phase/Crystallographic Properties



PZT Film Microstructure



Dielectric and Ferroelectric properties

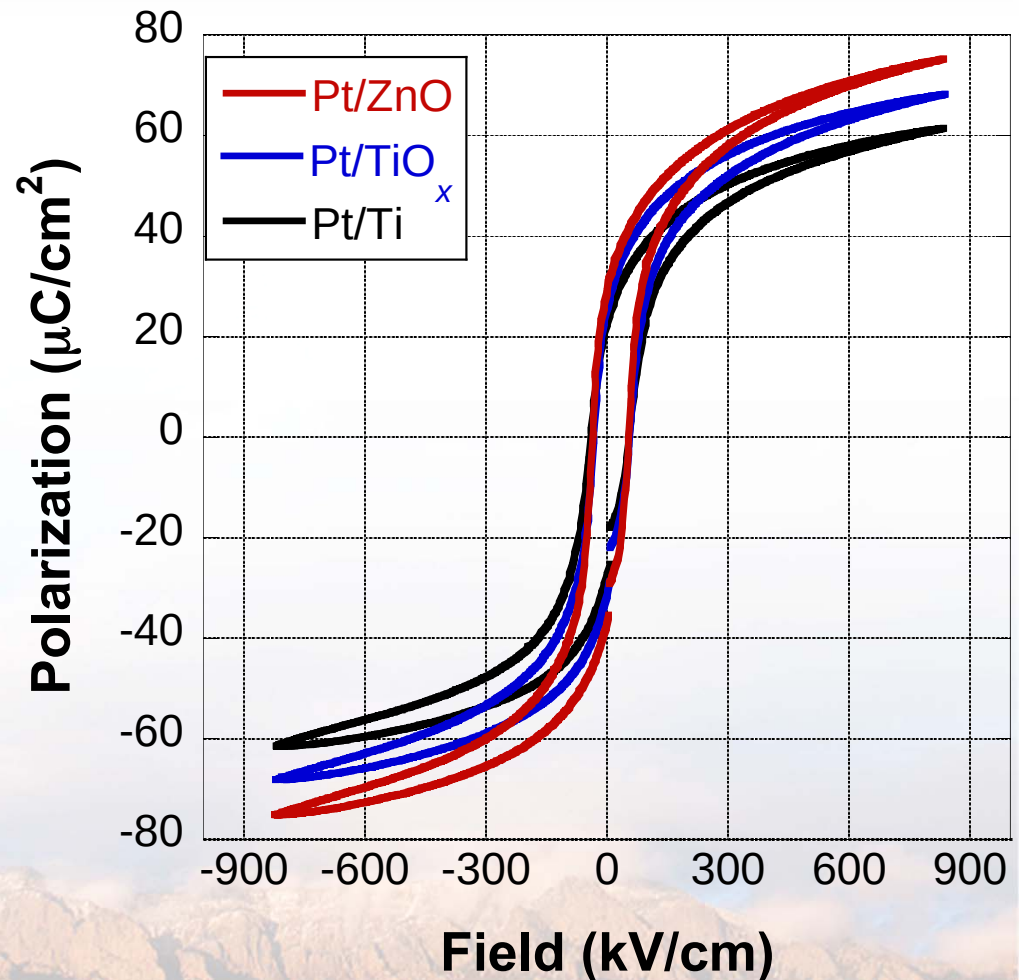


- Measurements conducted at 10 Hz
- All films saturate well
- Low leakage

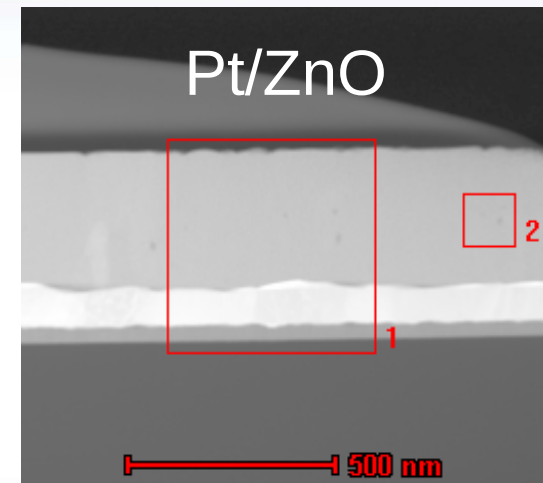
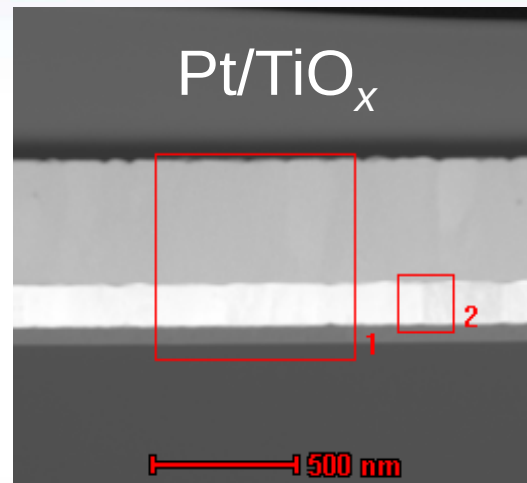
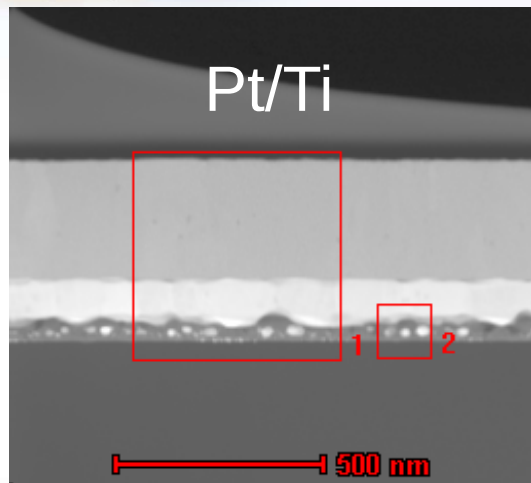
PZT Films Ferroelectric Properties: Pt/ZnO Substrates Superior

	$P_{\max}$ ($\mu\text{C}/\text{cm}^2$)	P_r ($\mu\text{C}/\text{cm}^2$)	E_c (kV/cm)	K
Pt/Ti	61.5	23.8	47.0	1440
Pt/TiO _x	68.2	27.0	44.1	1620
Pt/ZnO	75.3	32.2	43.8	1740

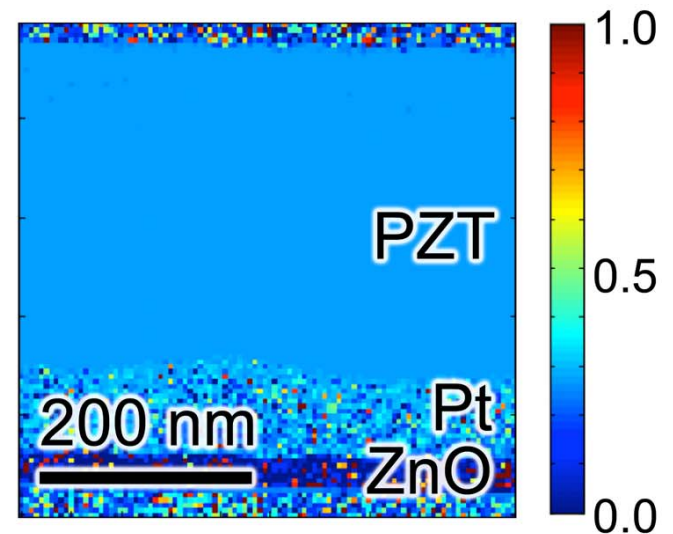
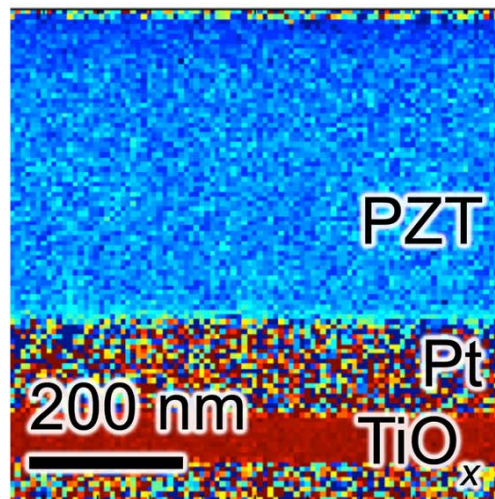
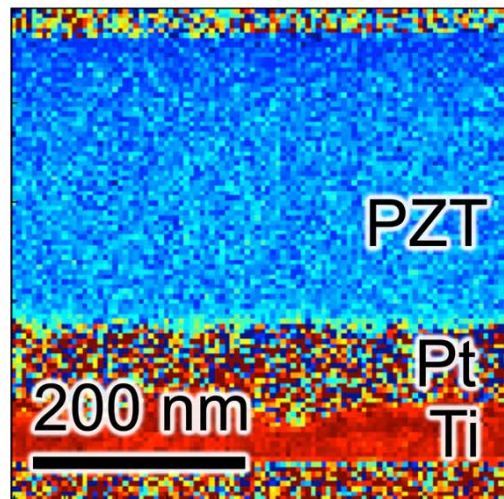
- Both P_r and K increase on ZnO-buffered substrates
- Coercive field decreases slightly for PZT on ZnO-buffered substrates



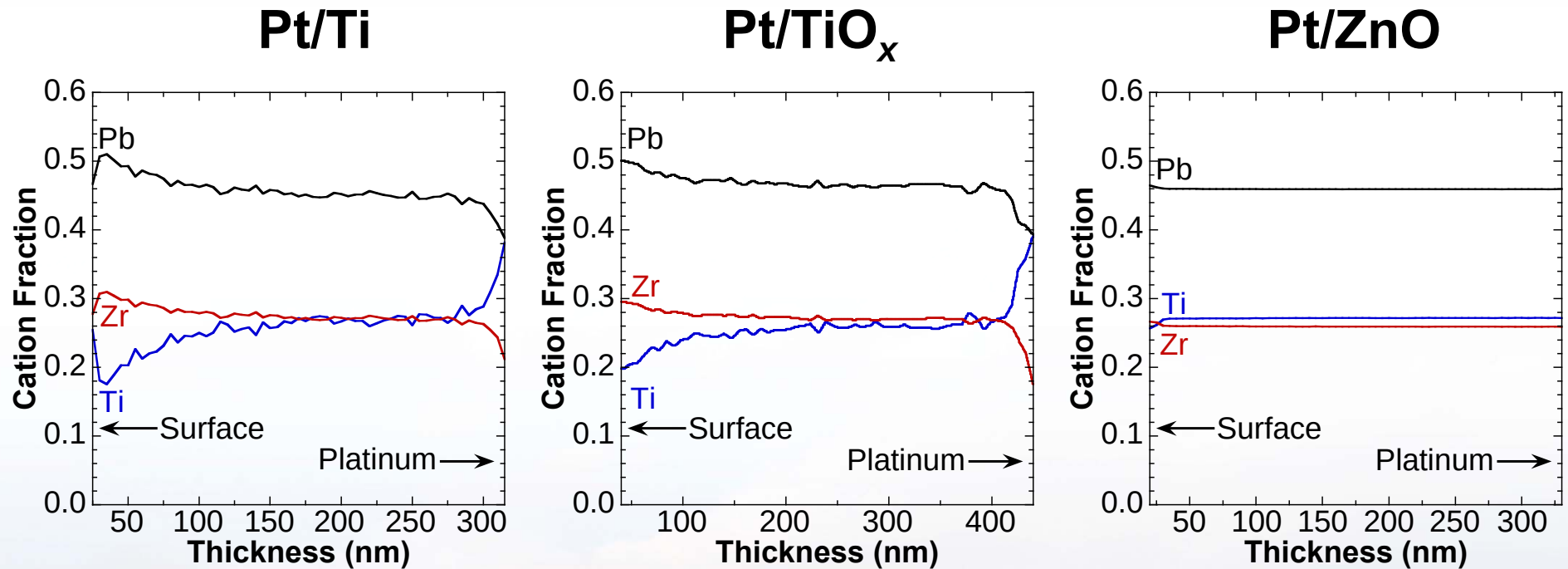
PZT Composition Analysis: No Composition Gradient on Pt/ZnO



Titanium Distribution

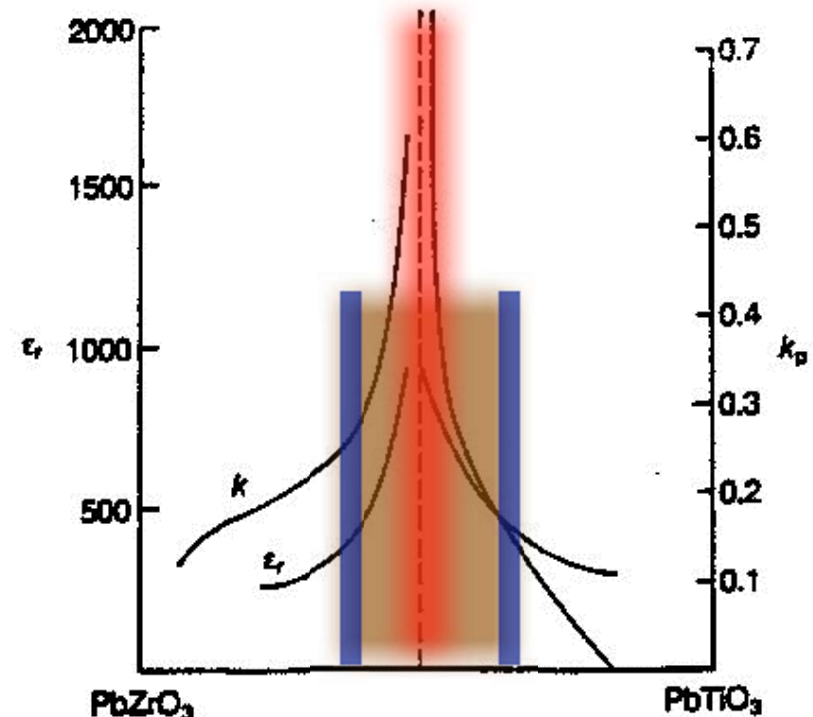
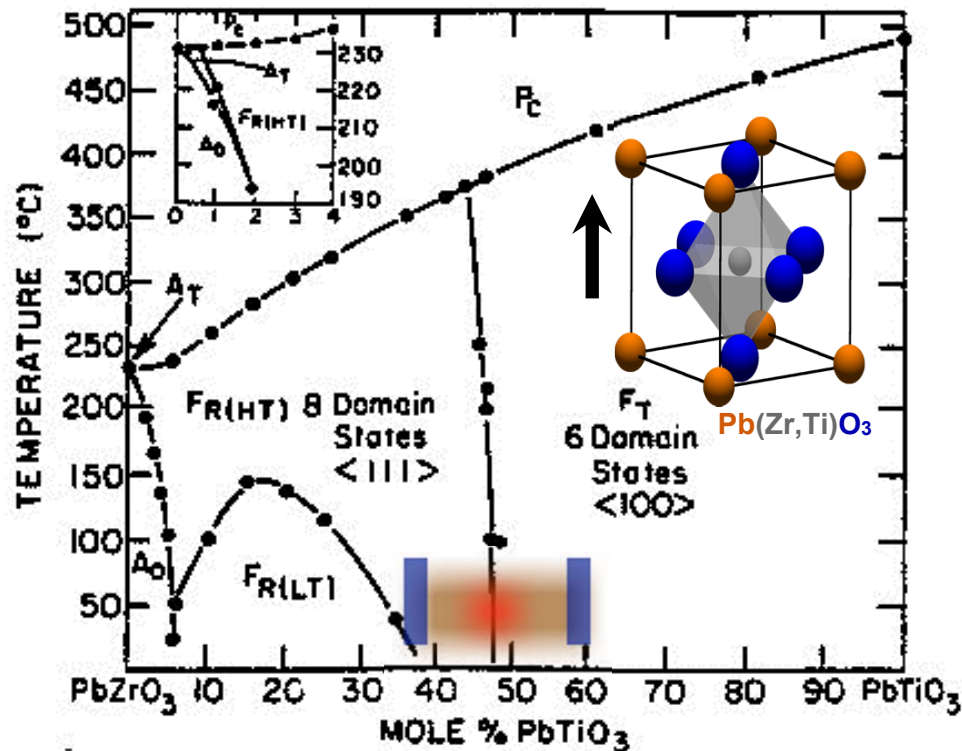


PZT Composition Analysis: No Composition Gradient on Pt/ZnO



***Substantially improved chemical
homogeneity in films on Pt/ZnO substrates***

Chemical Homogeneity Effects on Electronic Properties

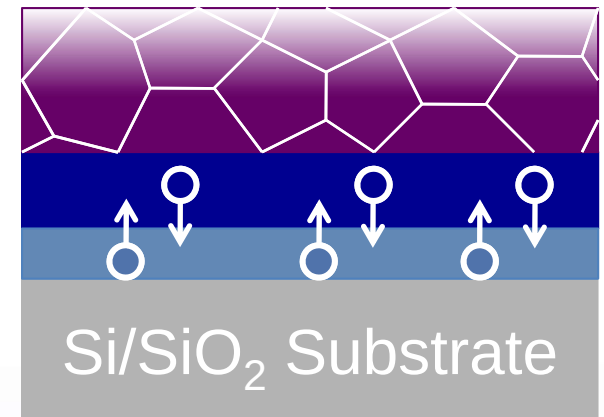
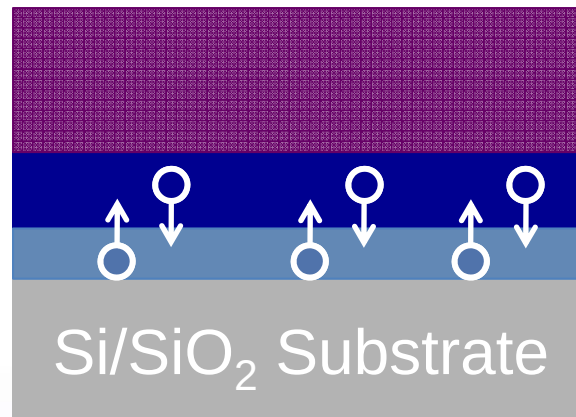
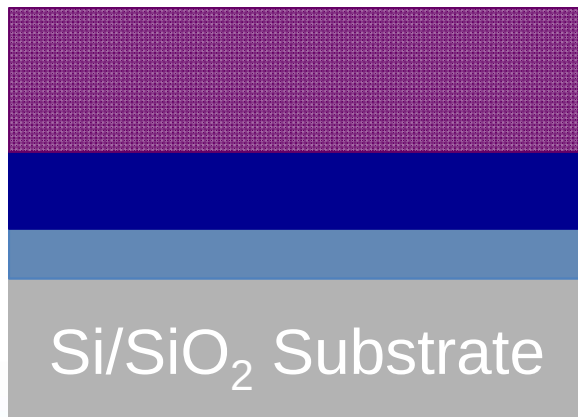


Good enough
(too many to
mention)

Better
(Bastani and Bassiri-
Gharb via MOE)

Better
(Calame and Muralt via MOE,
Shelton *et al.* via IMO)

Our Model

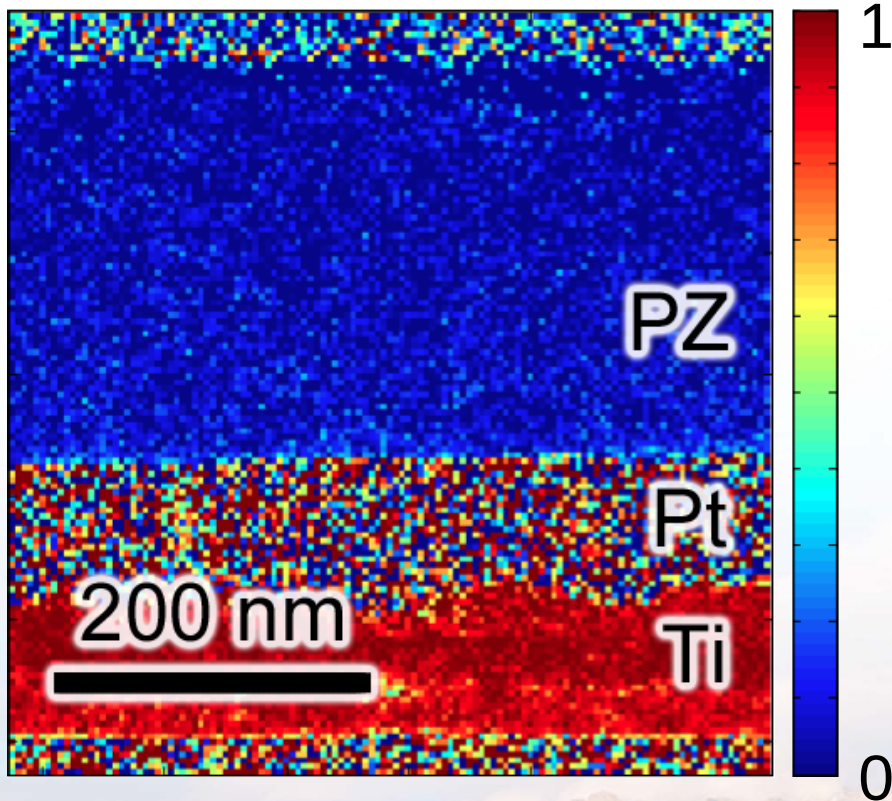


Hypothesis:

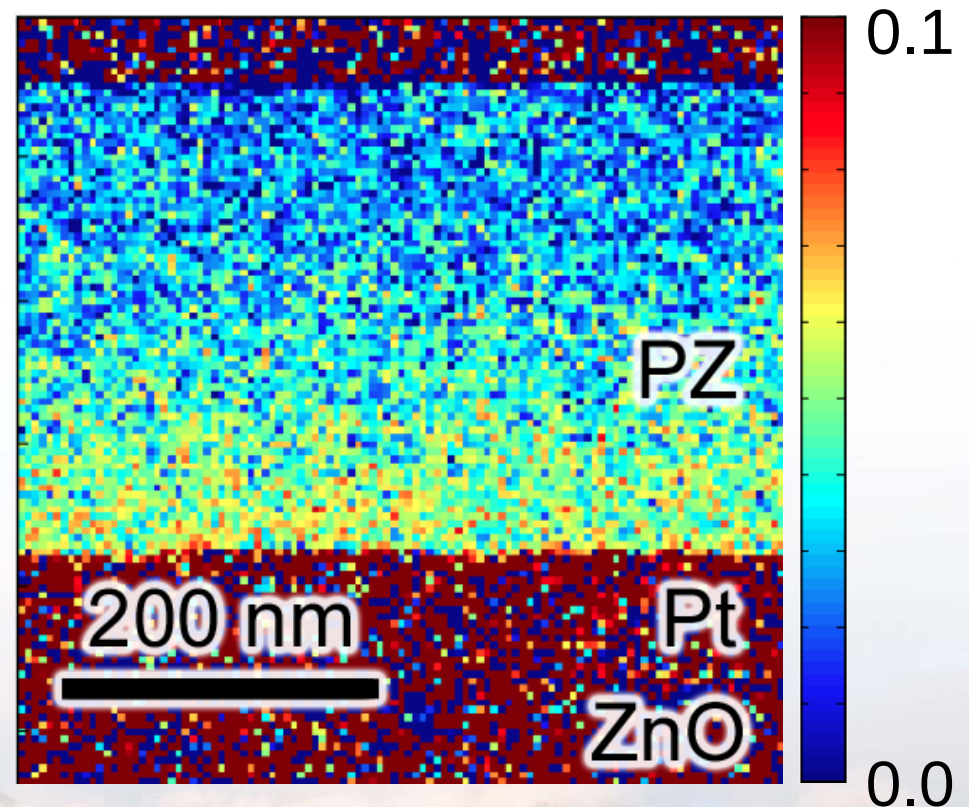
Titanium diffuses through platinum and to film/Pt interface

Test for Gradient Mechanism: PbZrO_3

PZ/Pt/Ti

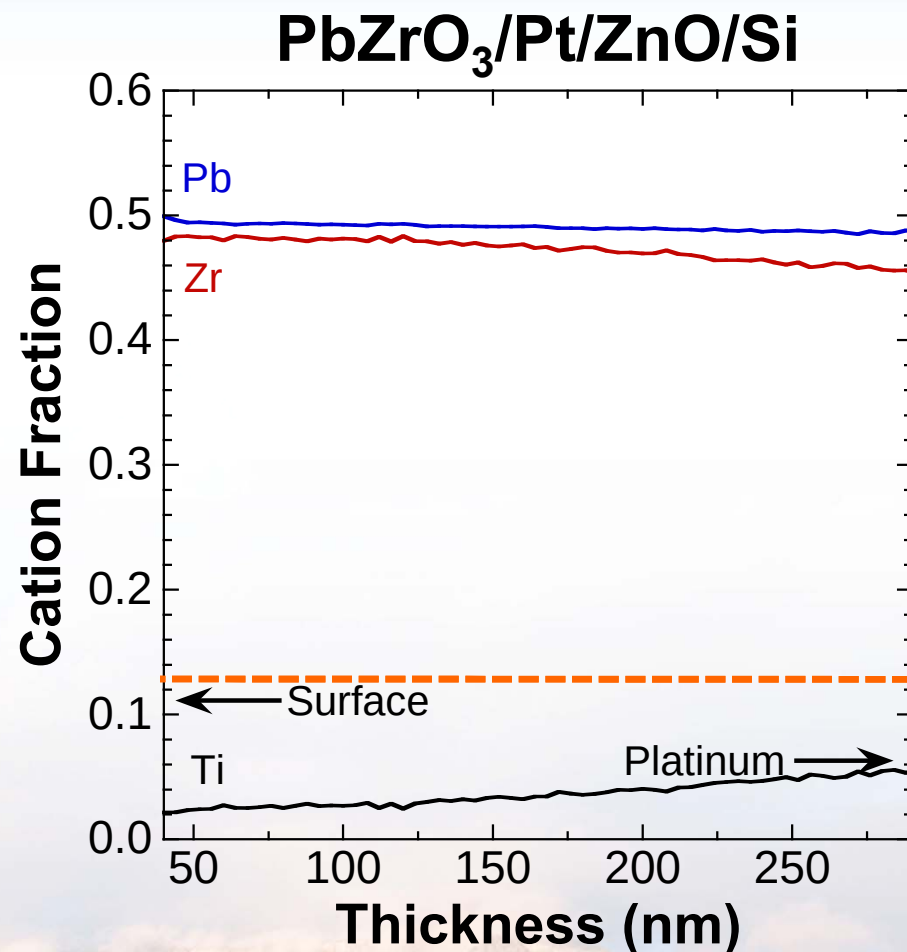
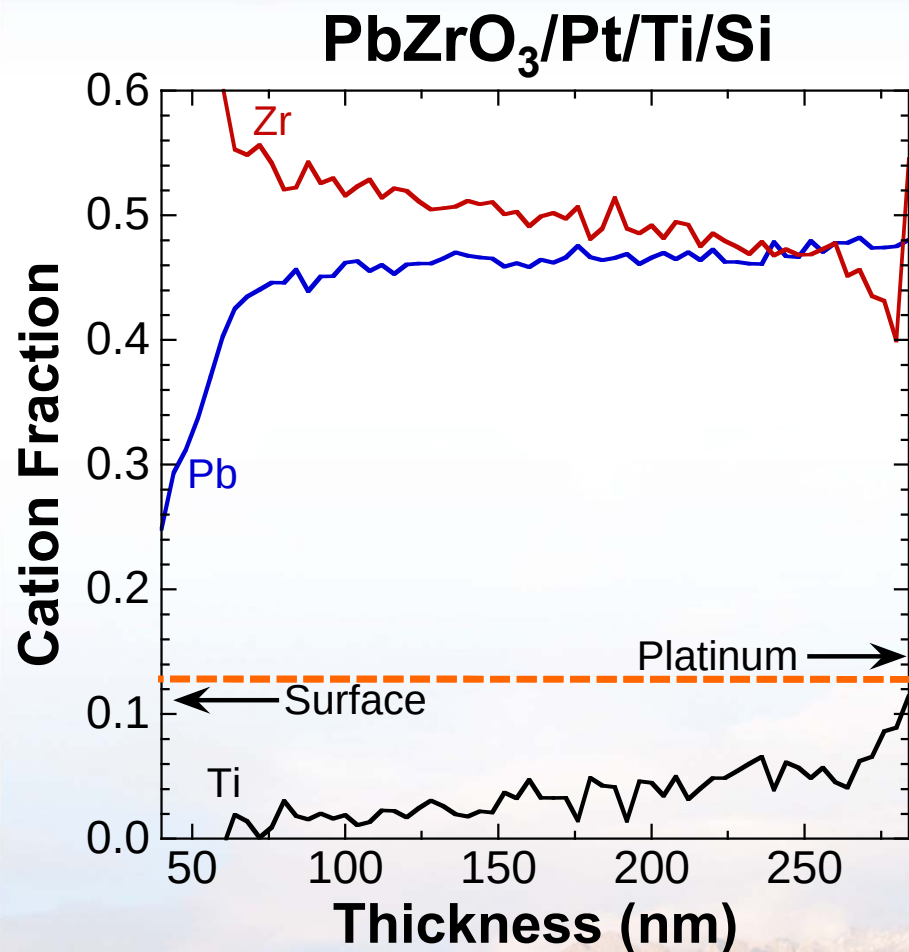


PZ/Pt/ZnO



Note Scale Difference

Mechanism: Ti Diffusion from Adhesion Layer: Test with PbZrO_3



Less Ti and smaller gradient on Pt/ZnO

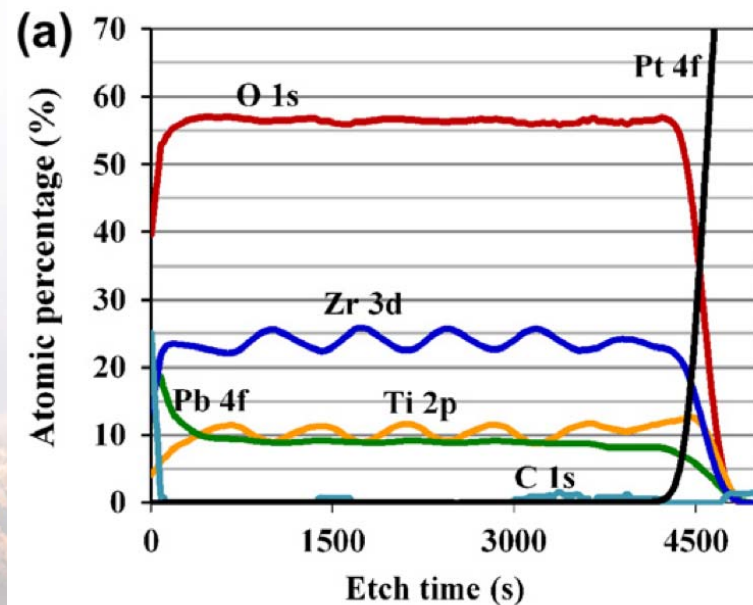
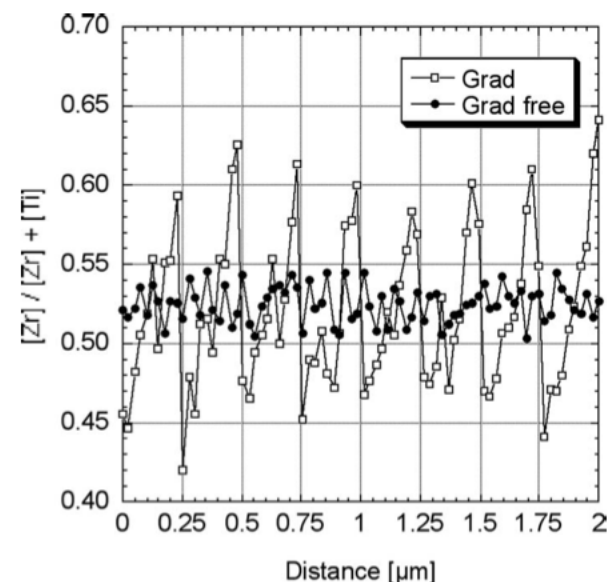
Solution Chemistry Effects?

Open Issues:

- This model cannot explain 'saw-tooth' profiles observed by other groups*
- What roles do solution chemistry and processing conditions play in gradient formation?
- More experiments necessary to isolate controlling variables
 - Substrate: Pt/Ti vs Pt/ZnO
 - Chemistry: IMO vs 2-MOE[‡]
 - Pyrolysis: 300, 350, 400° C
 - Heating Method/Rate: 15, 50, 200° C/s

*Amanuma, *et al.*, *Appl. Phys. Lett.* **65**, 3140 (1994)
Calame and Murali, *Appl. Phys. Lett.* **90**, 062907 (2007)
Bastani and Bassiri-Gharb, *Acta. Mater.* **60**, 1346 (2012)

[‡]Based upon: C.D.E. Lakeman and D.A. Payne, *JACerS*, **75(11)**, 3091 (1992)



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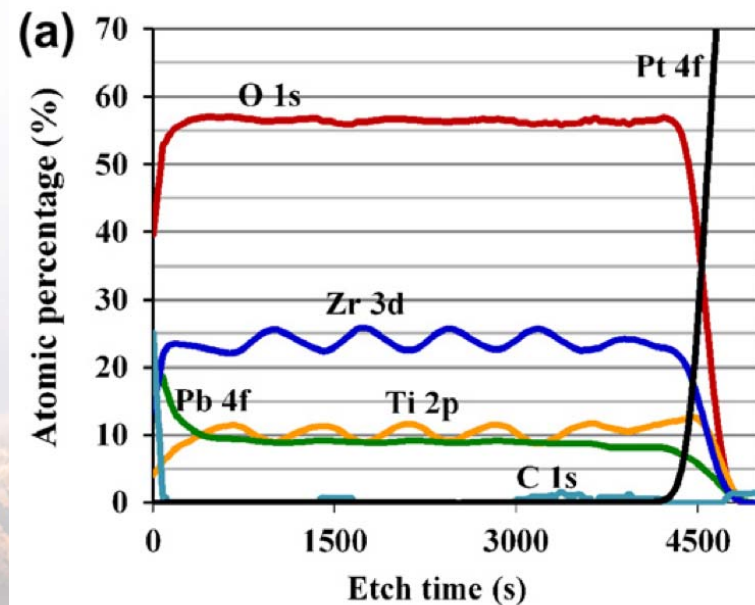
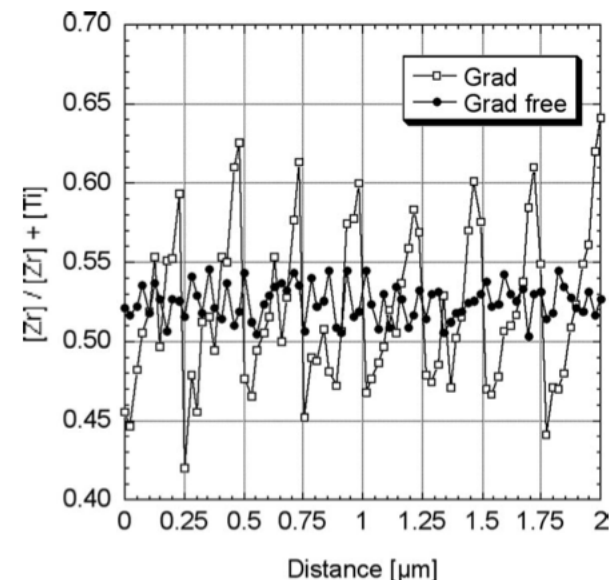
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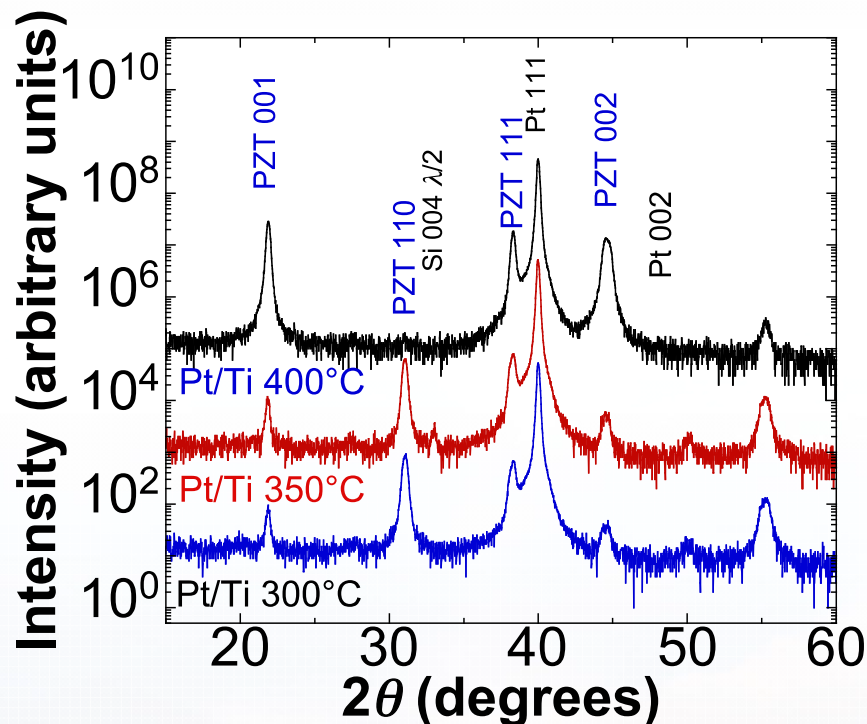
[‡]Based upon: C.D.E. Lakeman and D.A. Payne, *JACerS*, **75(11)**, 3091 (1992)



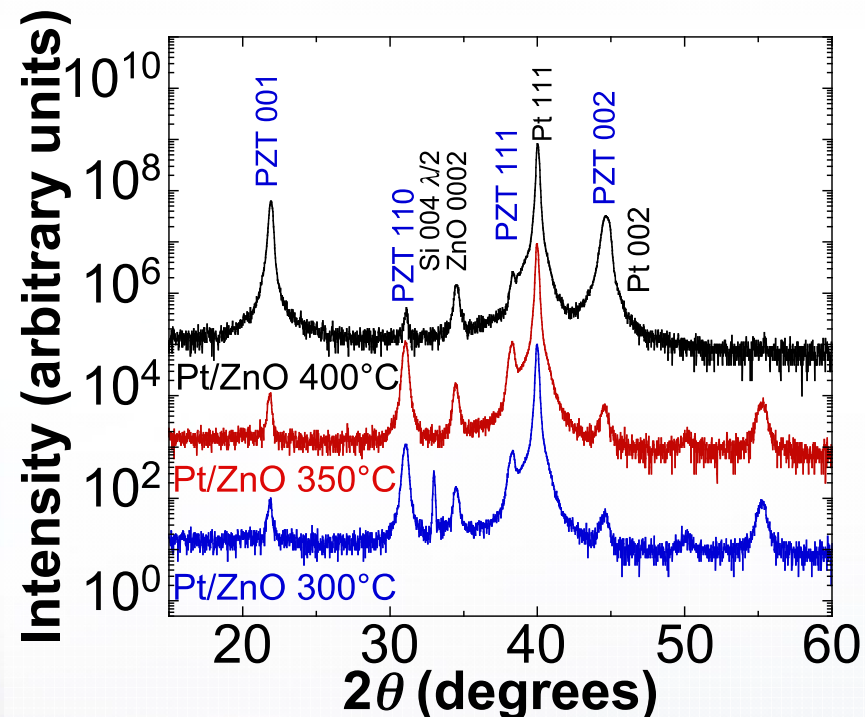
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Pyrolysis Effects: Texture

IMO on Pt/Ti



2-MOE on Pt/ZnO

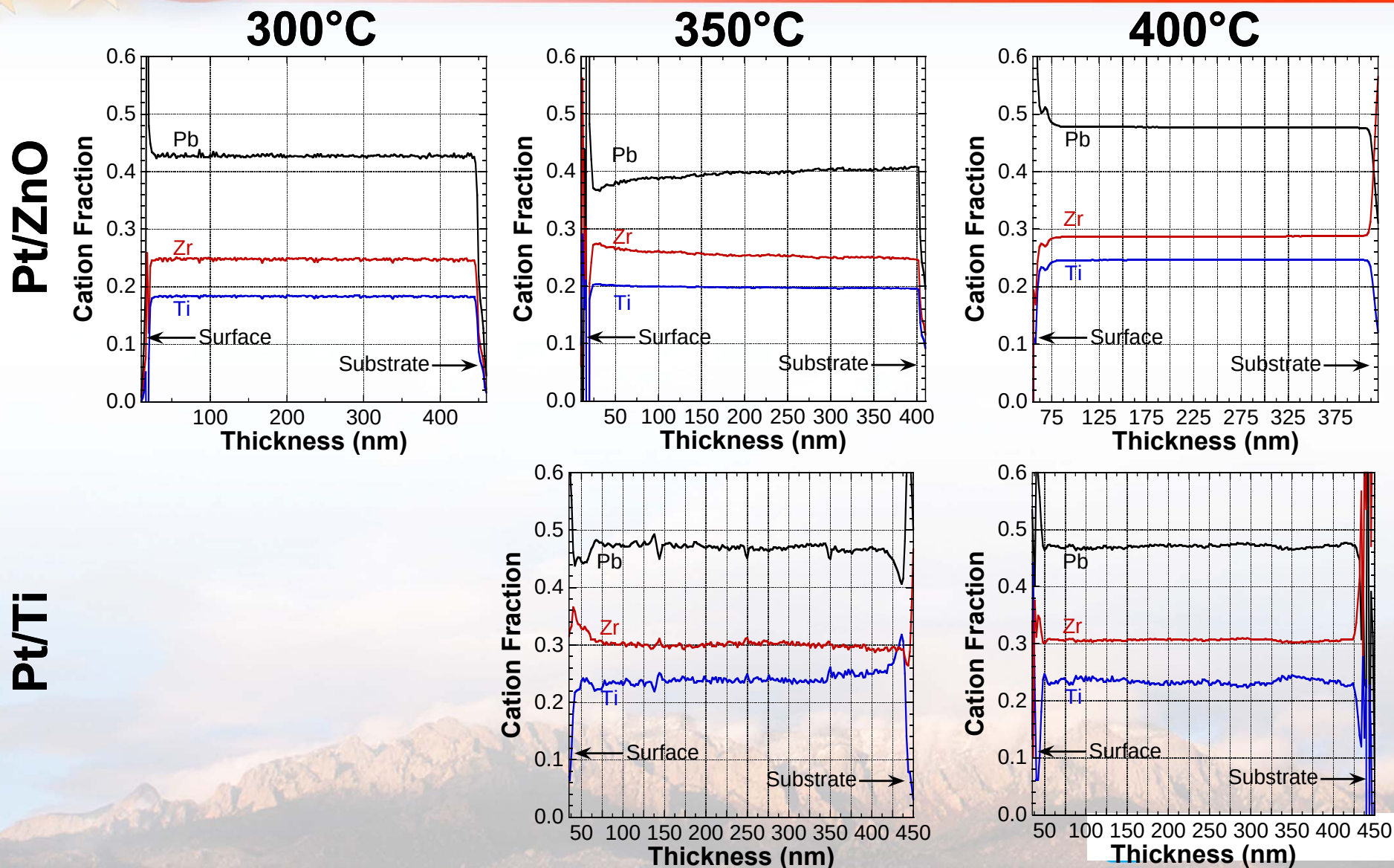


- ***Pyrolysis temperature effects texture***
- ***No apparent difference with substrate or chemistry***
- ***Consistent with previous work****

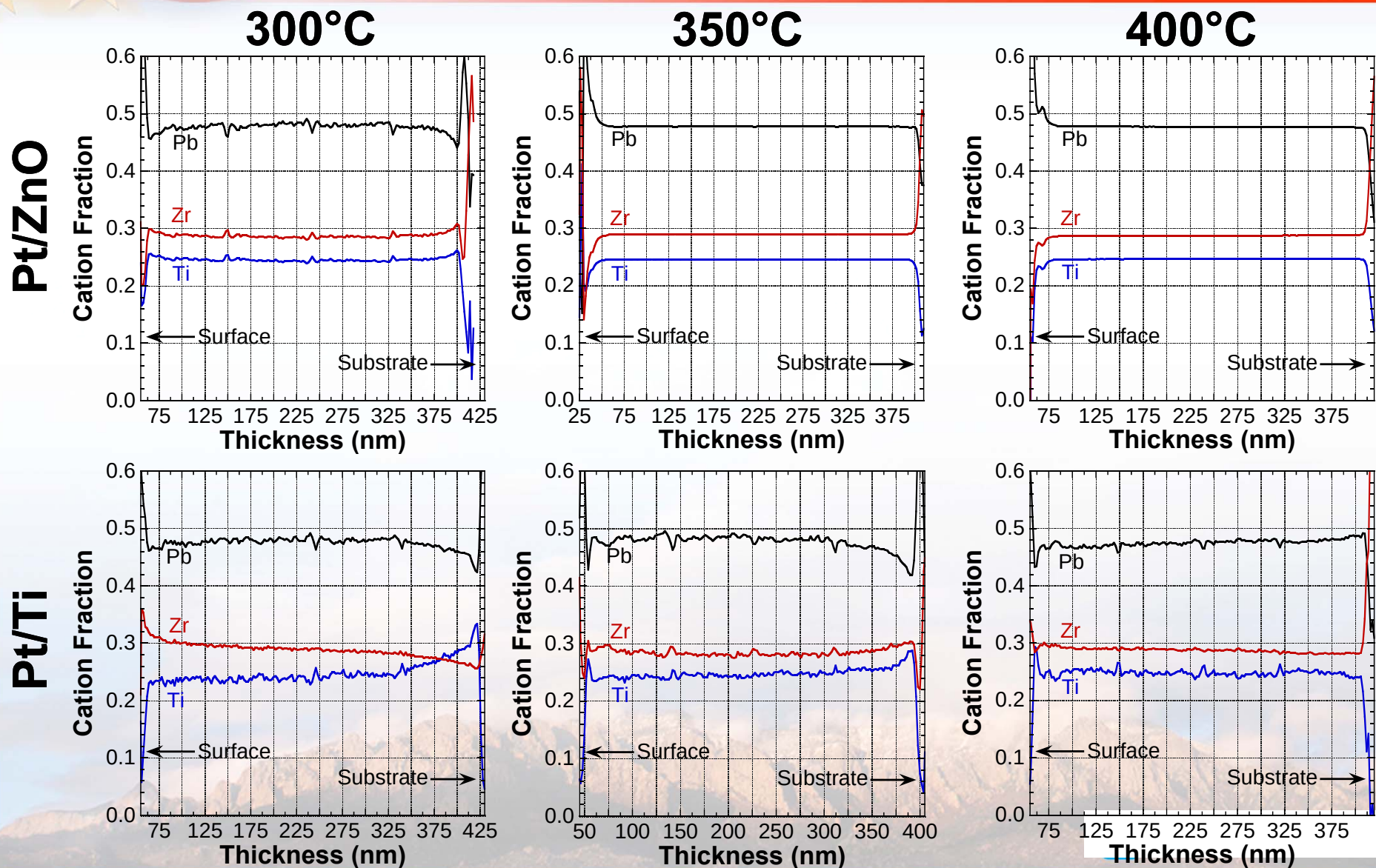
*Chen and Chen, JACerS, **81** (1) 97-105 (1998)

*Brennecka, et al., Proc. 14th US-Japan, 136 (2009)

Unfired Chemical Distribution: IMO Films



Unfired Chemical Distribution: 2-MOE Films



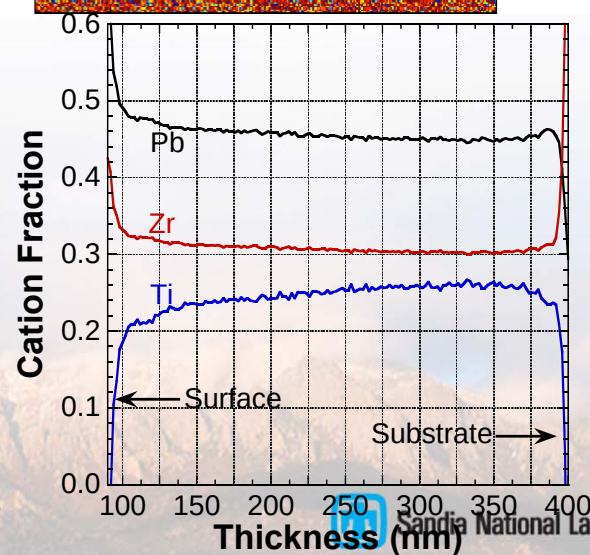
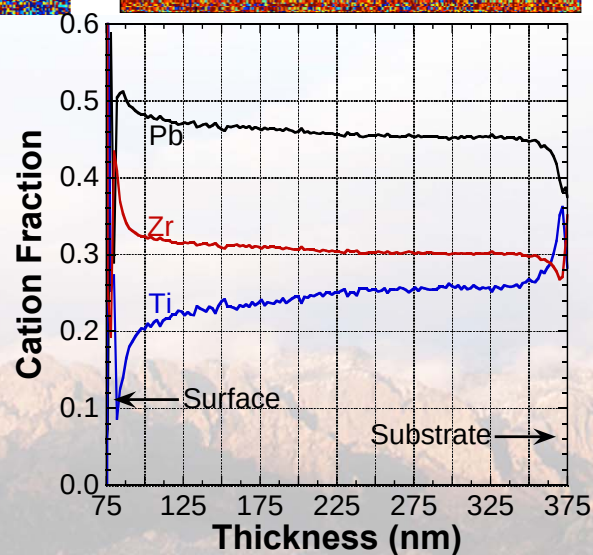
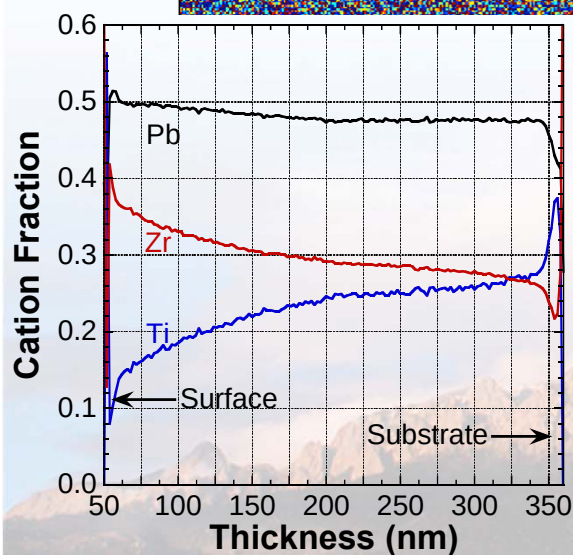
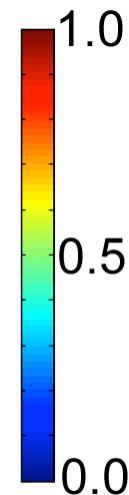
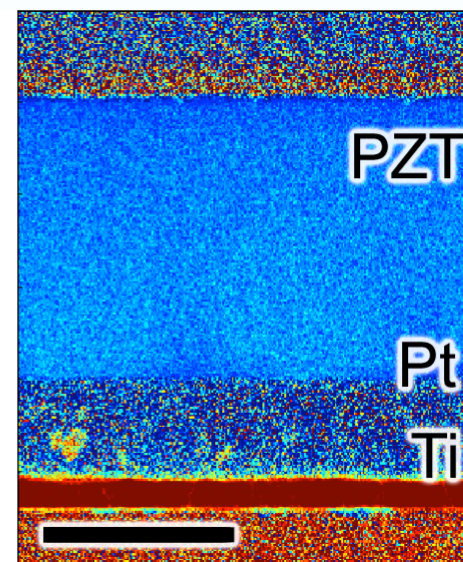
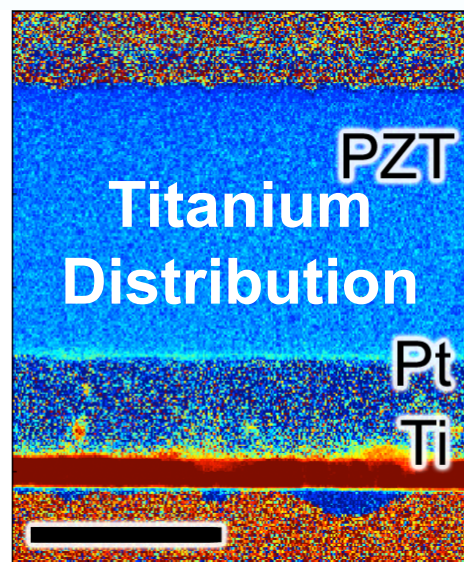
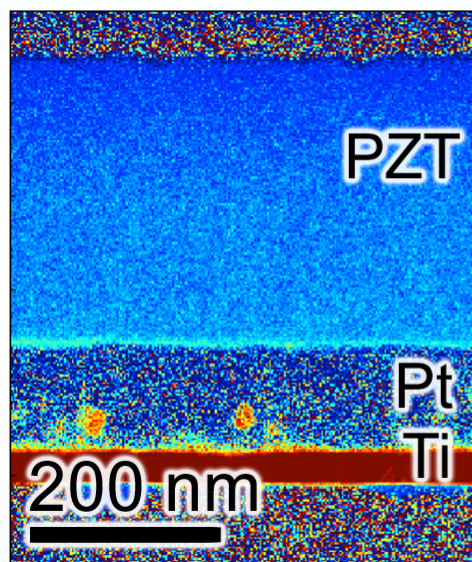
Solution Chemistry Effect: IMO films

300°C

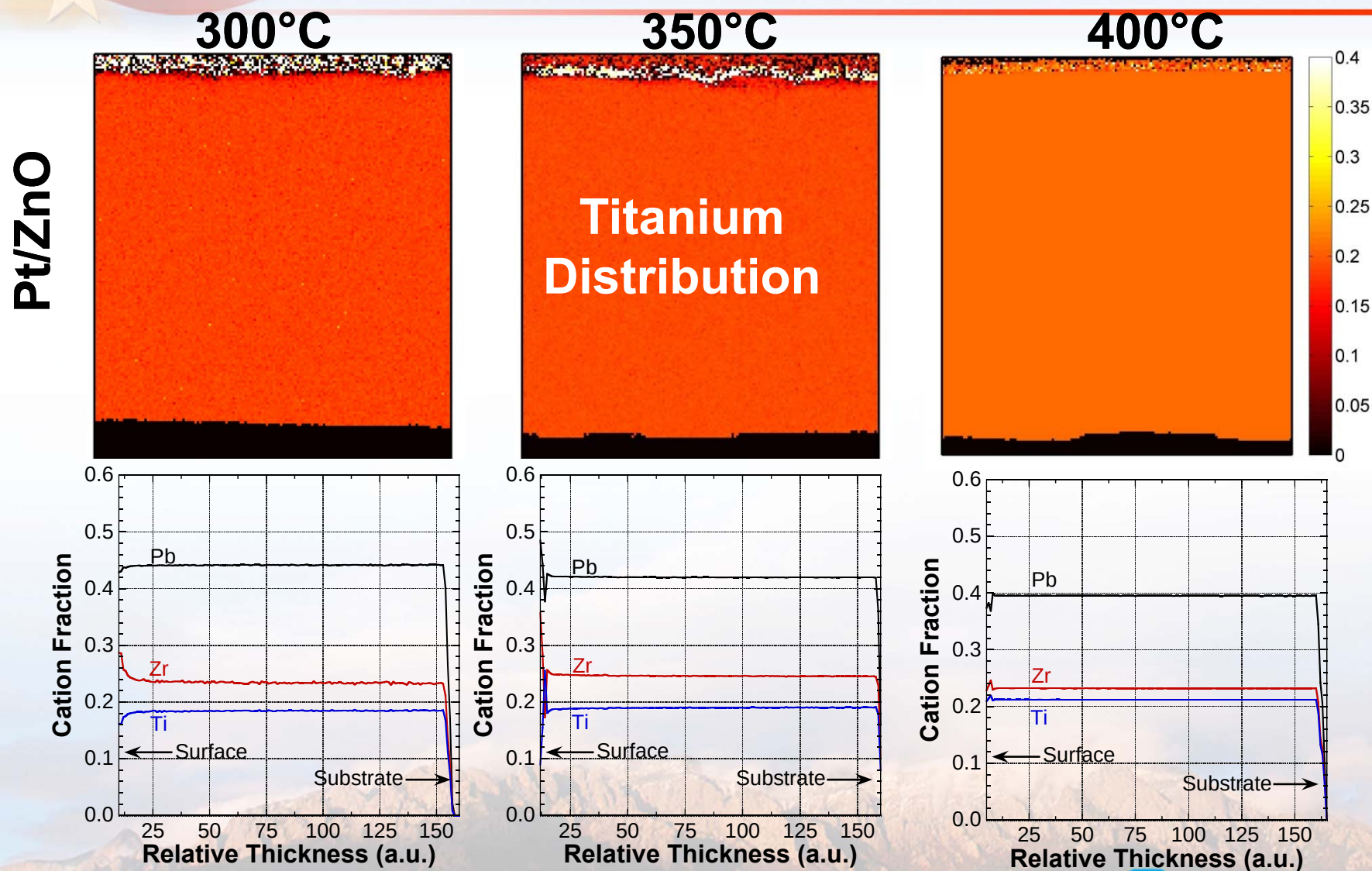
350°C

400°C

Pt/Ti



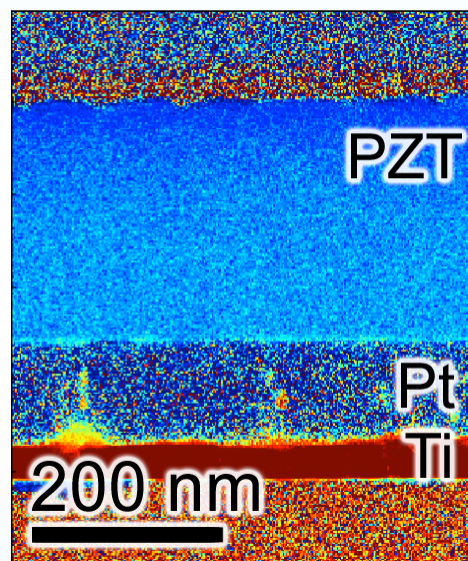
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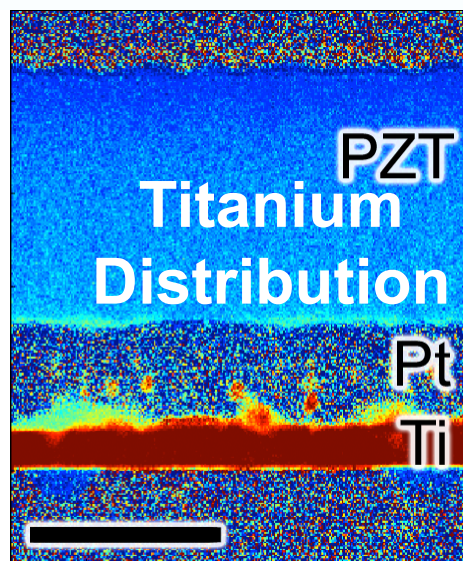
Solution Chemistry Effect: 2-MOE films

300°C

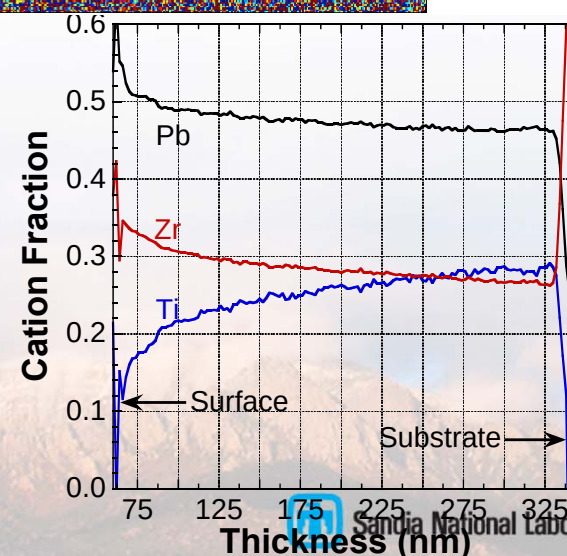
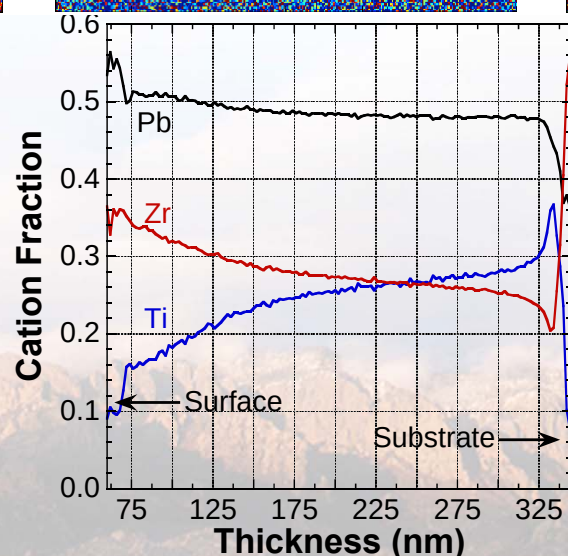
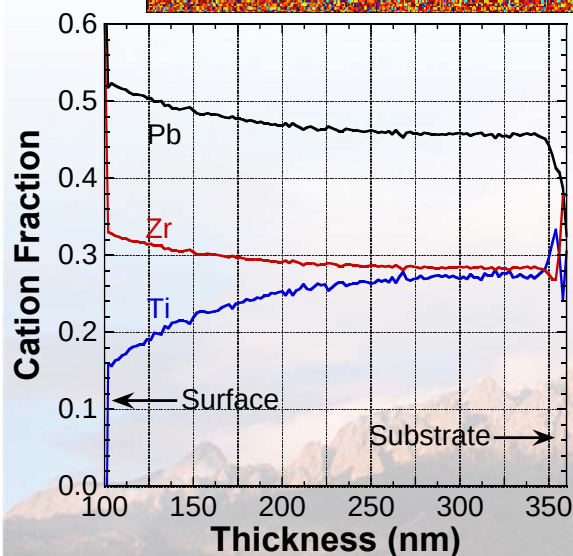
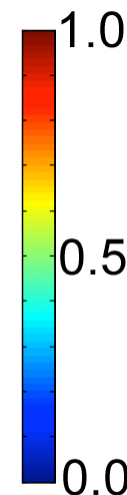
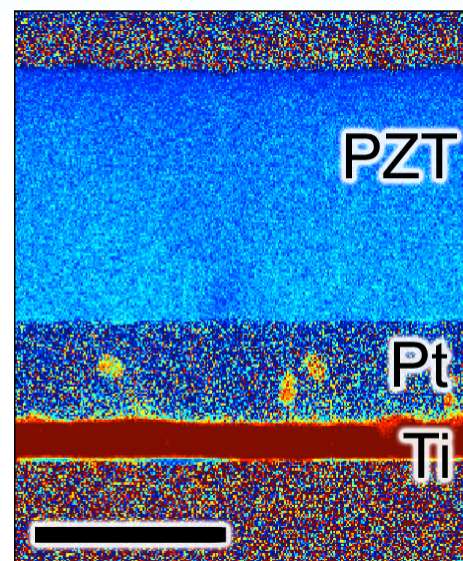
Pt/Ti



350°C



400°C



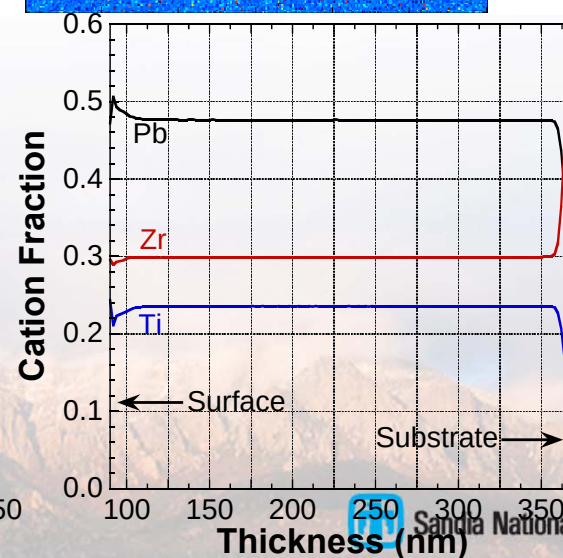
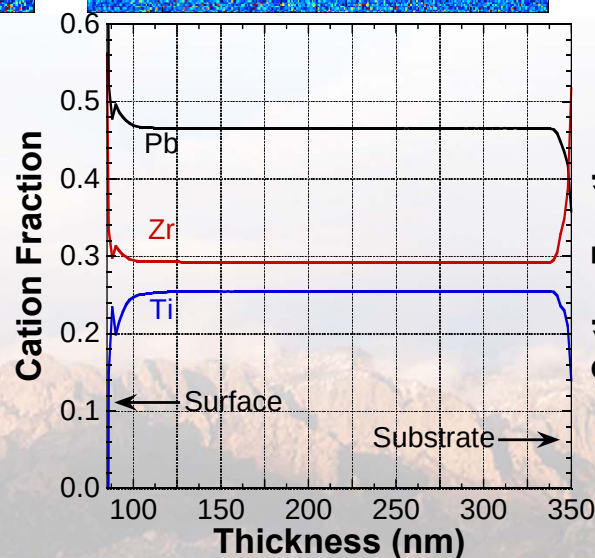
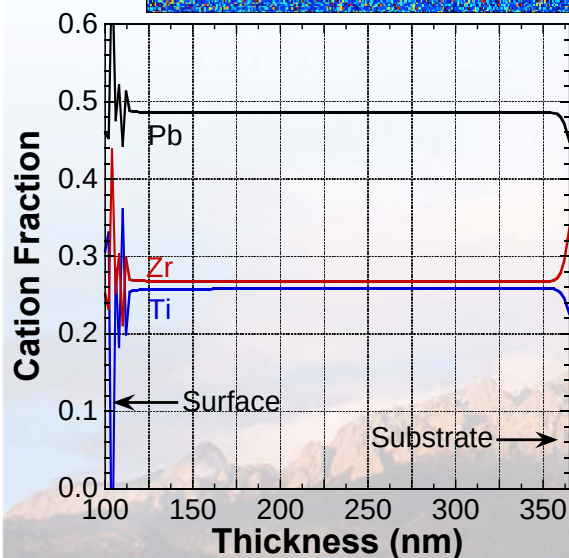
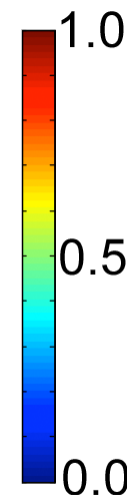
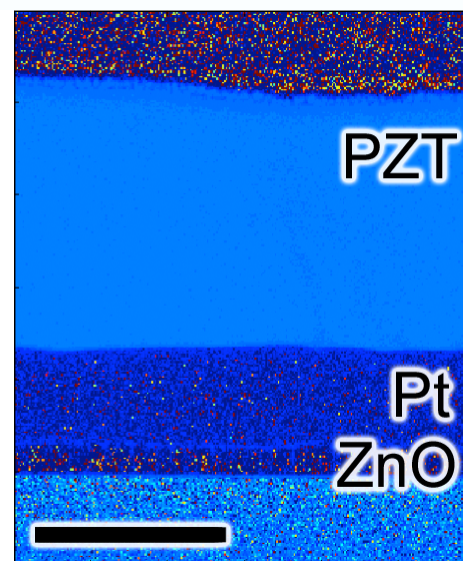
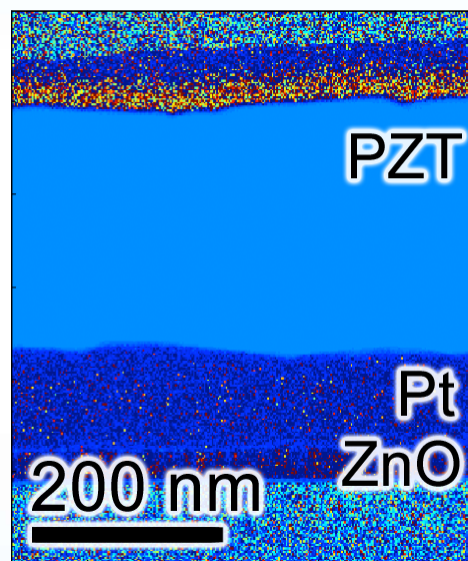
Solution Chemistry Effect: 2-MOE films

300°C

350°C

400°C

Pt/ZnO



Solution Chemistry Effects?

Open Issues:

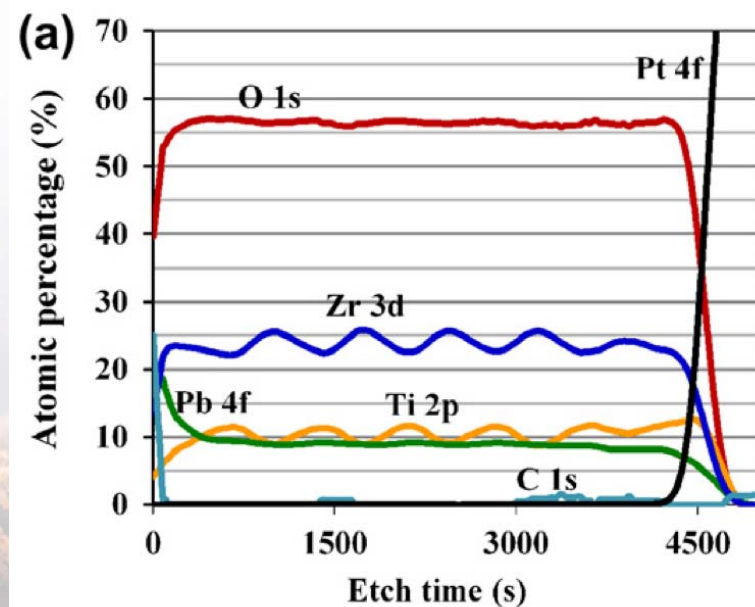
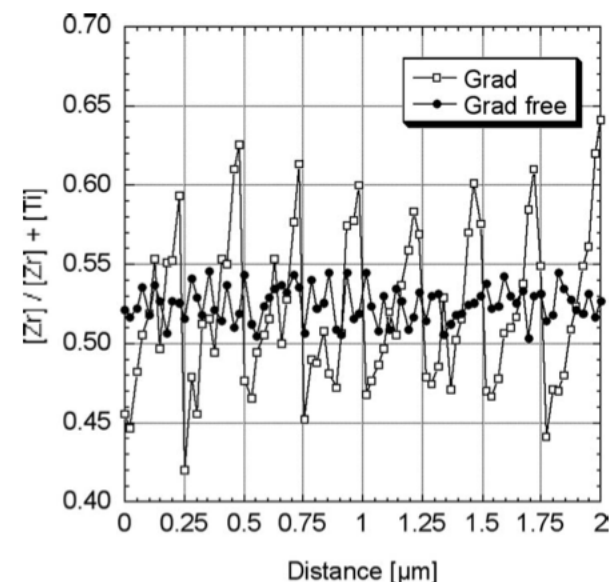
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 - Chemistry: **IMO** vs **2-MOE**[‡]
 - Pyrolysis: **300, 350, 400° C**
 - Heating Method/Rate: **15, 50, 200° C/s**
 - **RTA: Solaris 150 (Tungsten Halogen Lamp)**

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Calame and Murali, *Appl. Phys. Lett.* **90**, 062907 (2007)

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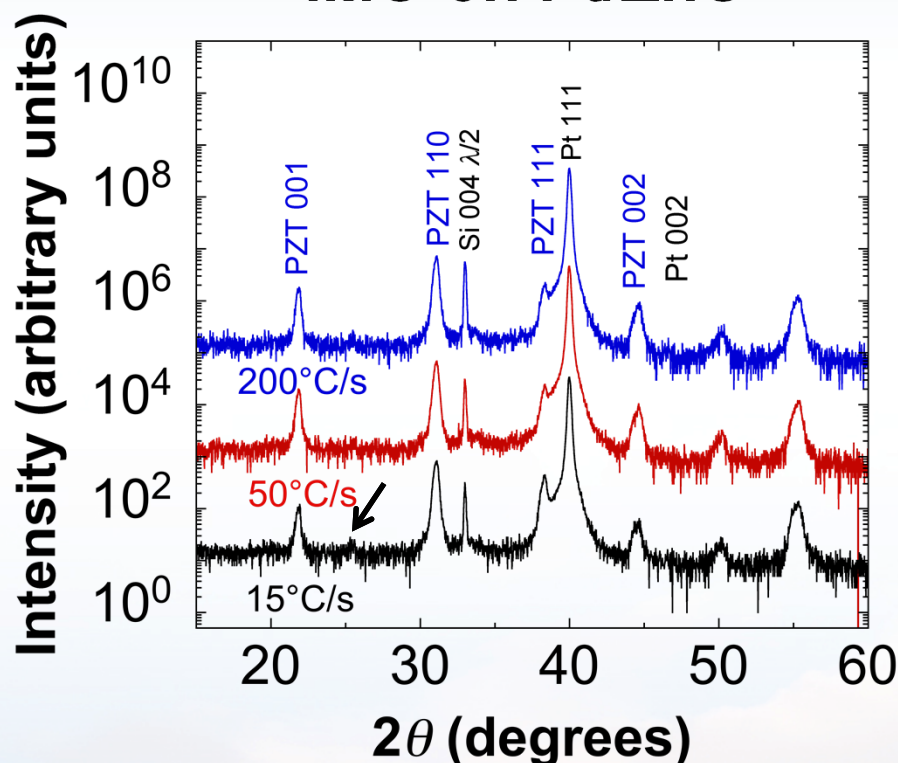
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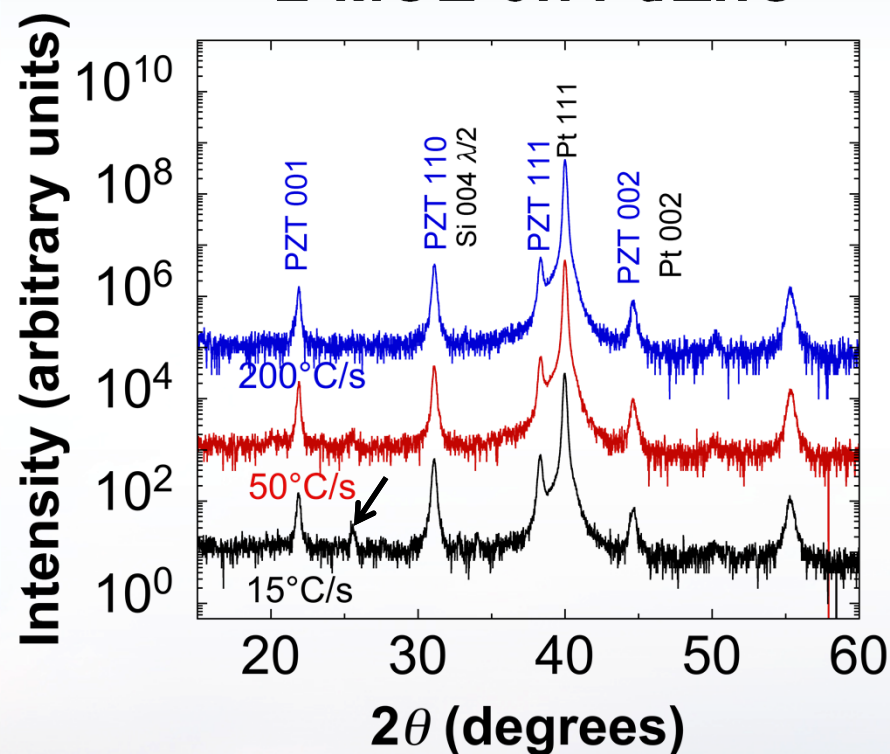
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RTA Heating Rate Effects: Texture

IMO on Pt/ZnO



2-MOE on Pt/ZnO

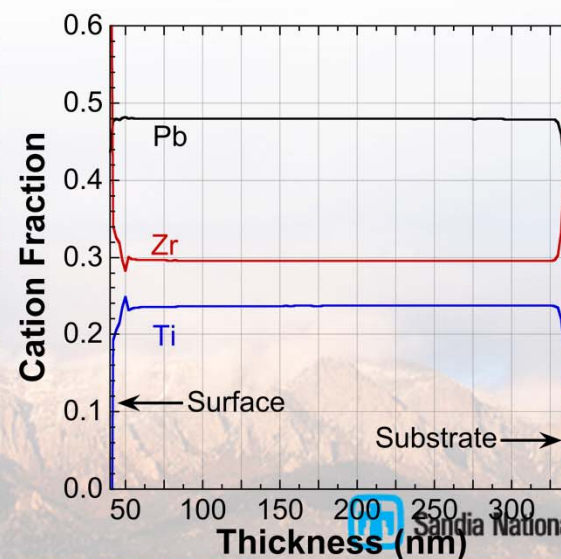
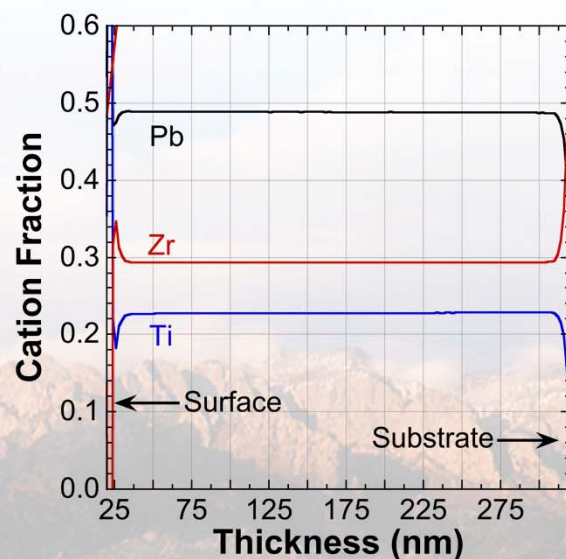
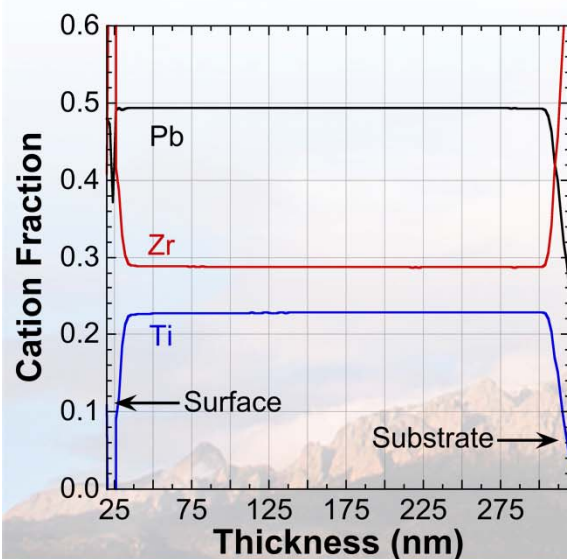
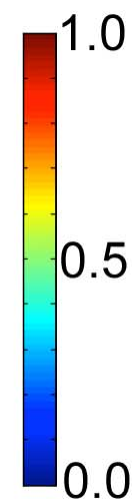
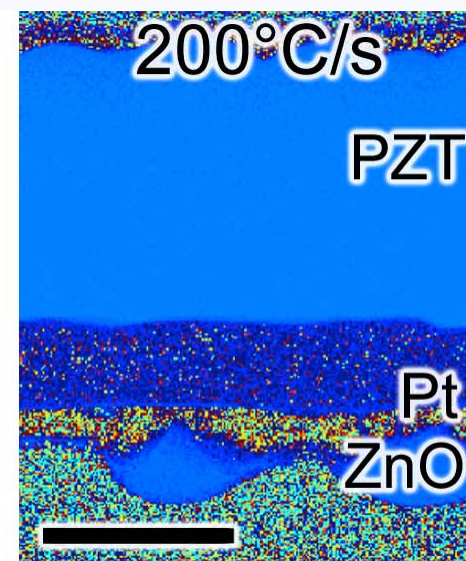
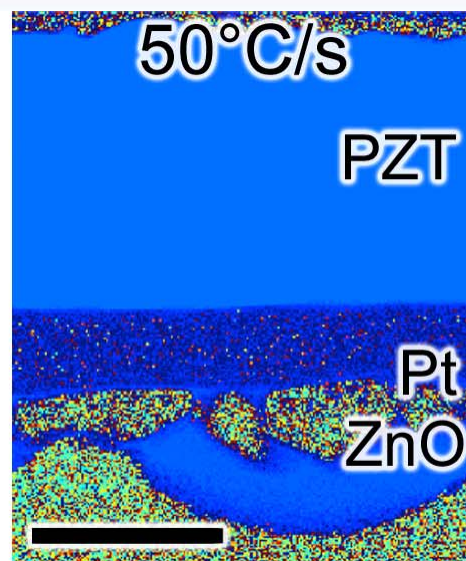
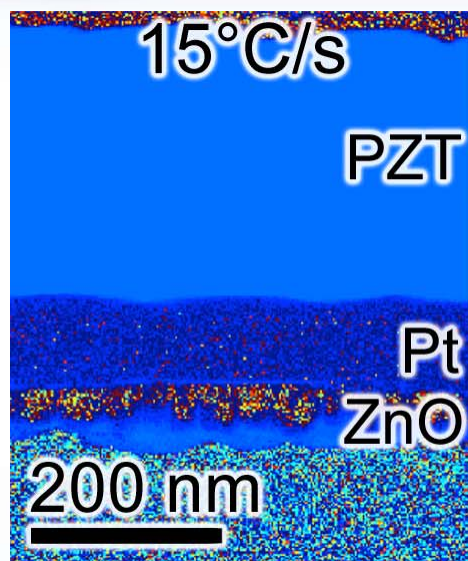


- **No apparent heating rate dependence to texture**
 - *Consistent with previous studies**
- **ZnO diffraction peaks no longer visible**
- **Additional peaks present in low ramp rate samples**

*K. Nittala and J.L. Jones, et al., *J. Appl. Phys.*, **113**, 244101 (2013)

Ramp Rate Effect: IMO Films

Pt/ZnO
Titanium Distribution

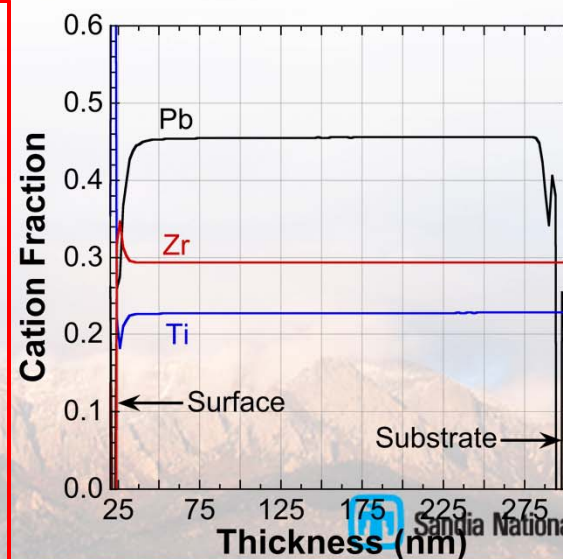
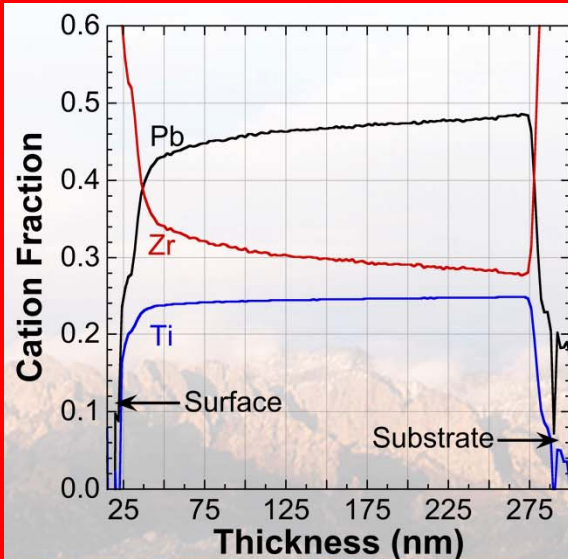
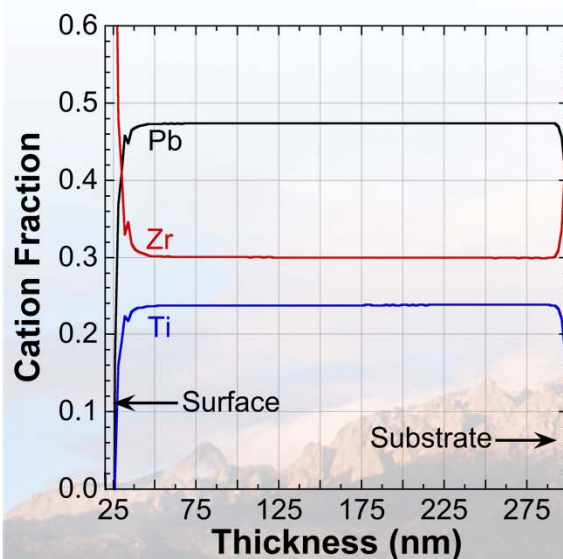
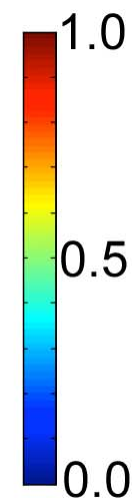
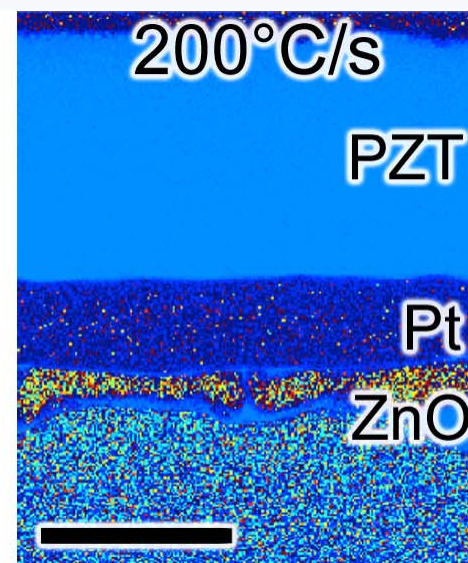
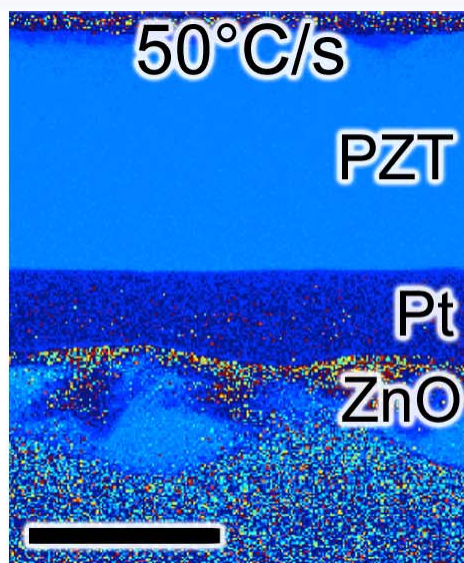
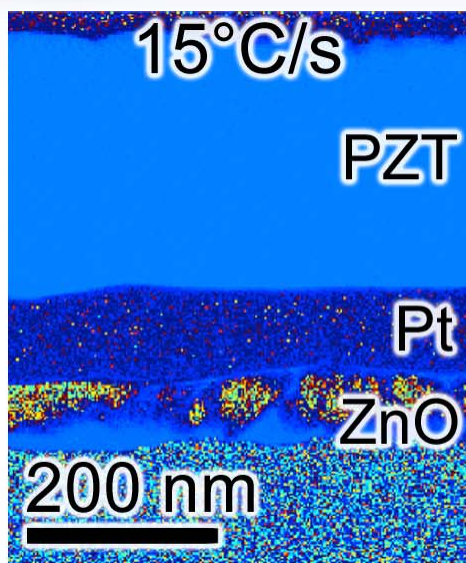


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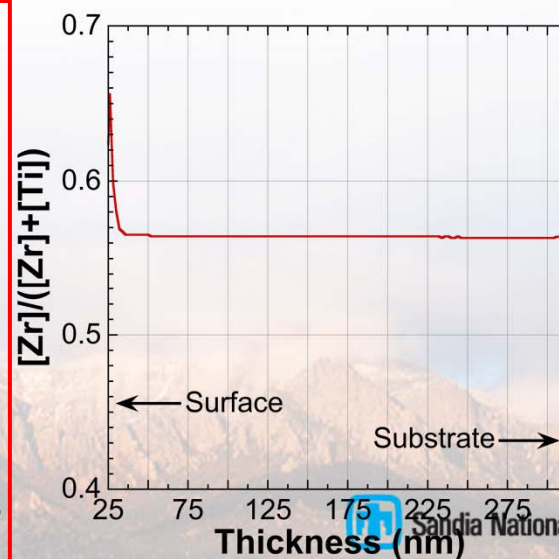
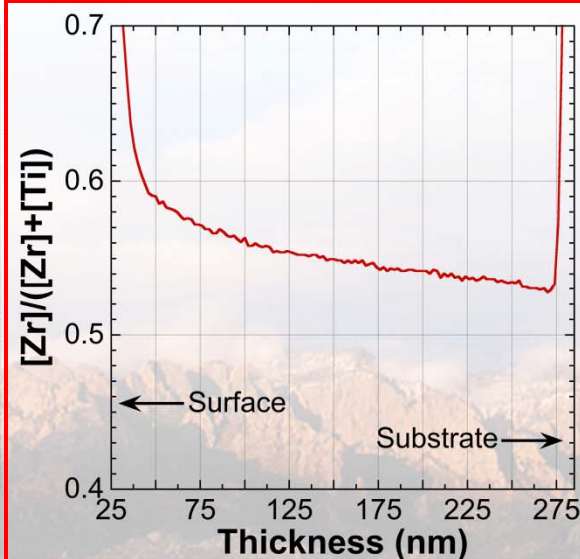
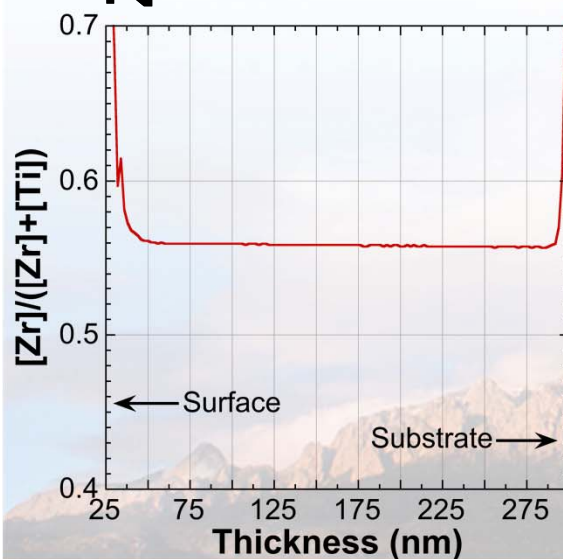
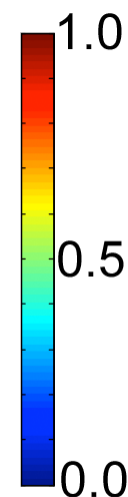
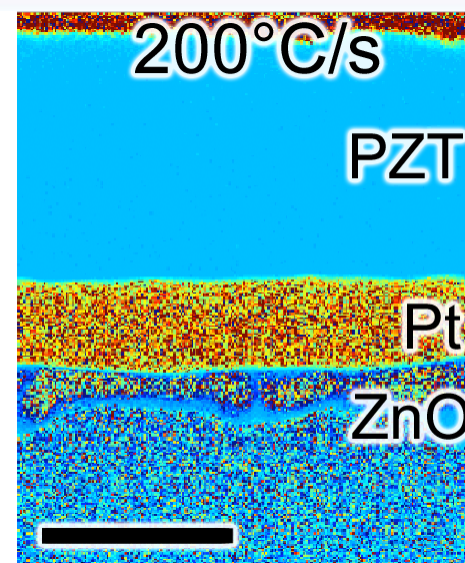
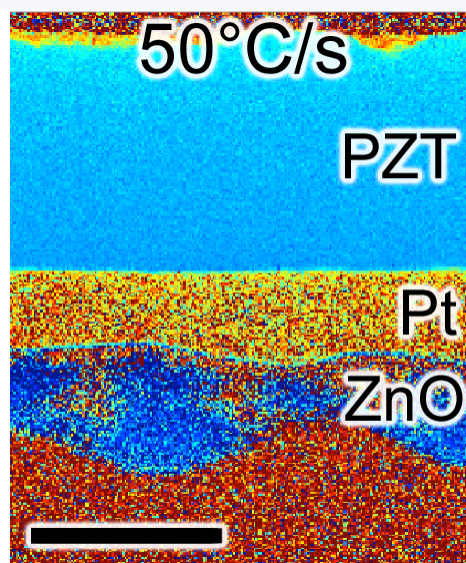
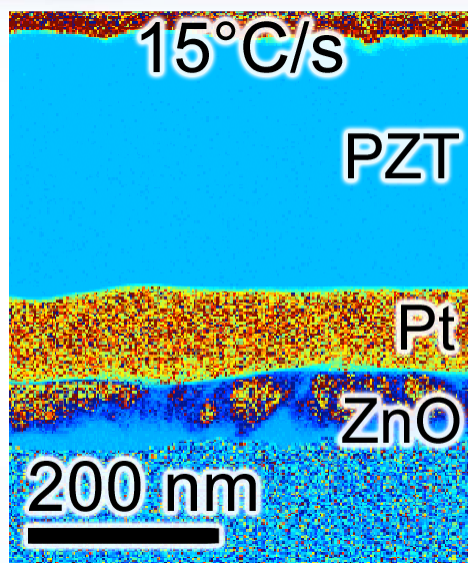
Ramp Rate Effect: 2-MOE Films

Pt/ZnO
Titanium Distribution



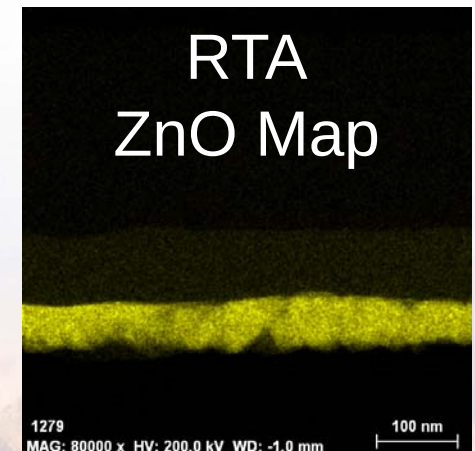
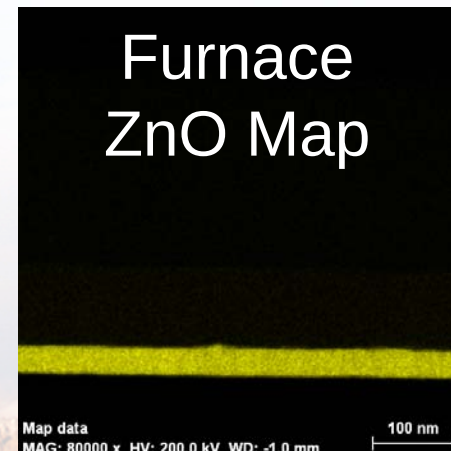
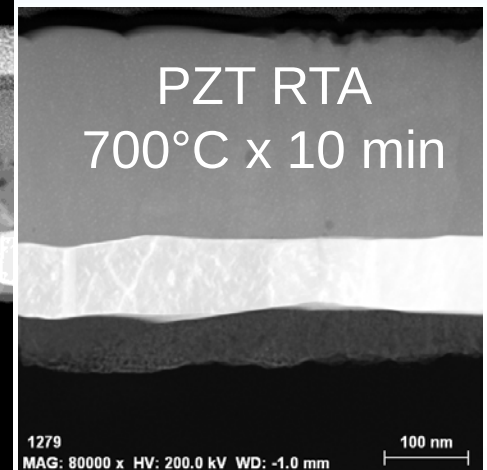
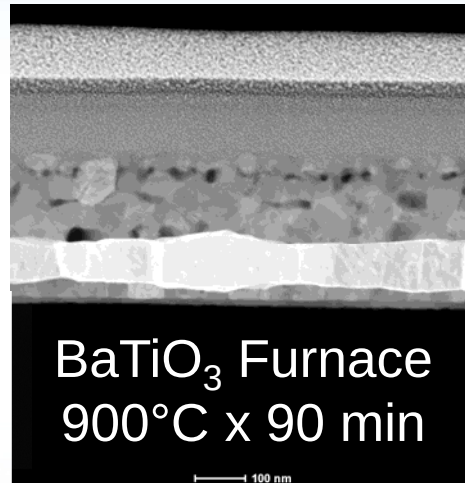
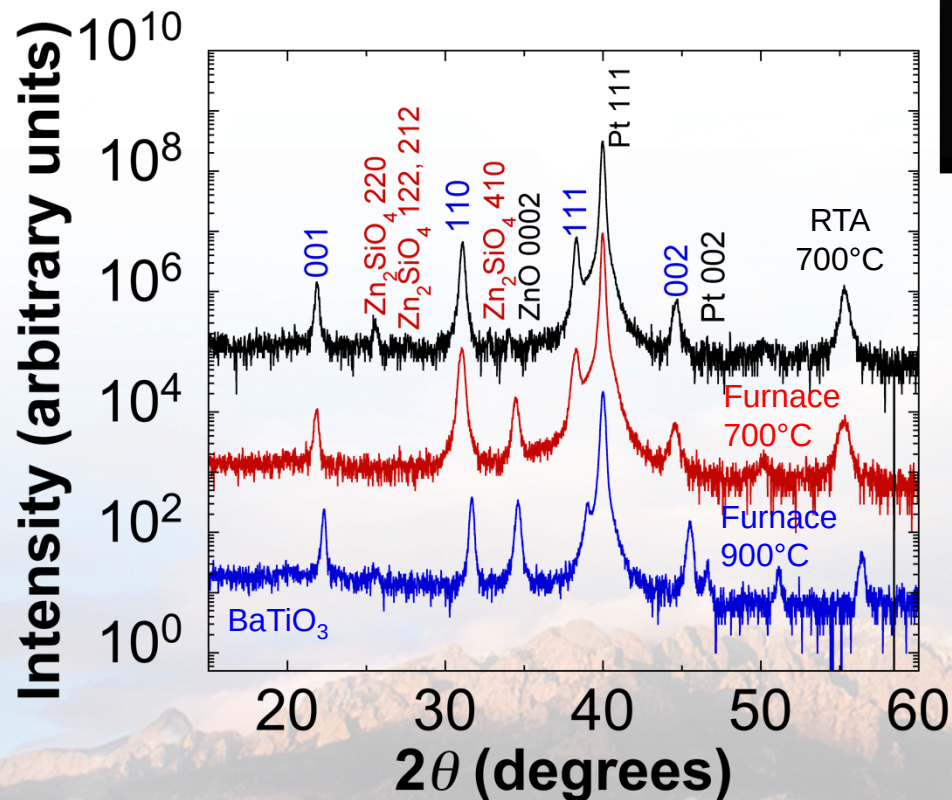
Ramp Rate Effect: 2-MOE Films

Pt/ZnO
Zirconium Distribution



Interesting Observation: ZnO-SiO₂ Reaction

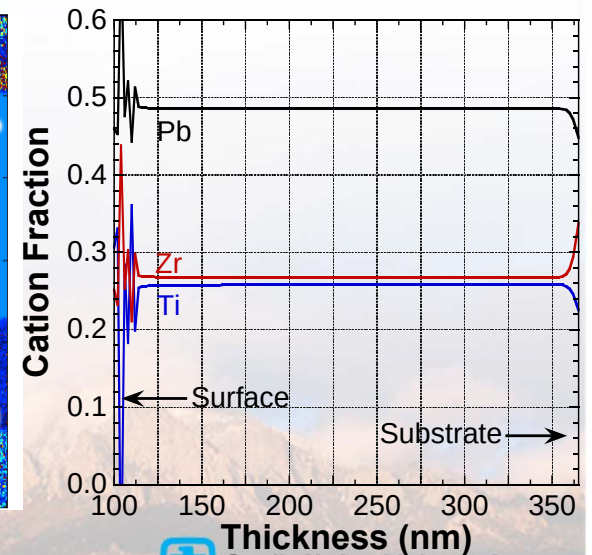
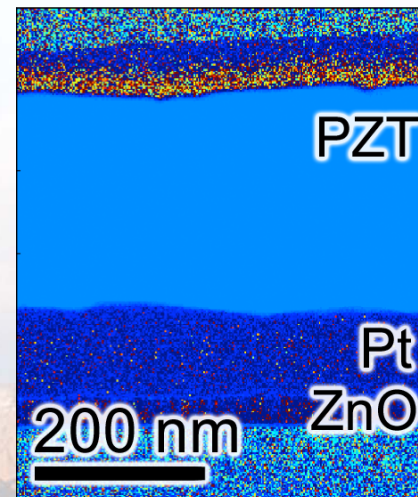
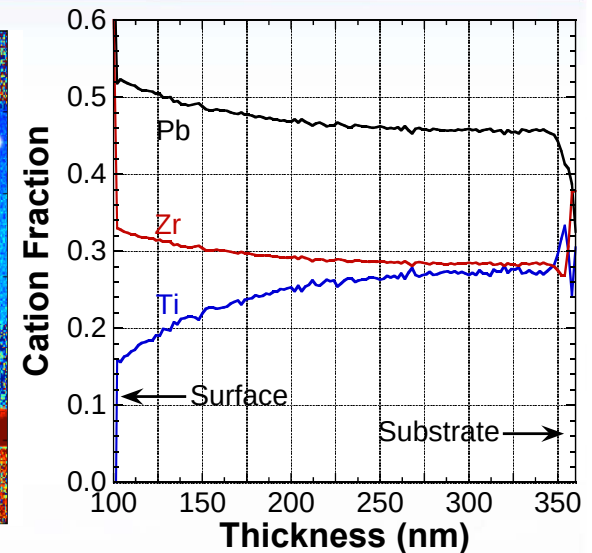
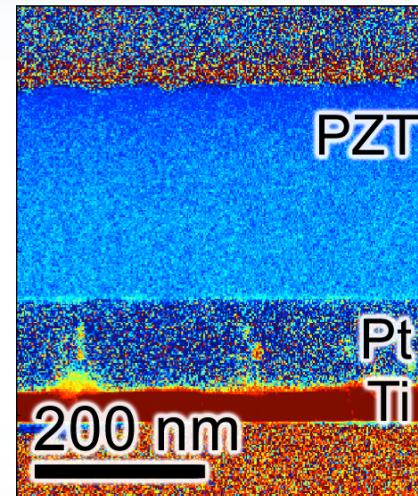
- Observe formation of secondary Zn₂SiO₄ phase in RTA samples
- Only previously observed this in films subjected to significantly higher thermal budgets



Processing of PZT Film: Lessons Learned To Date

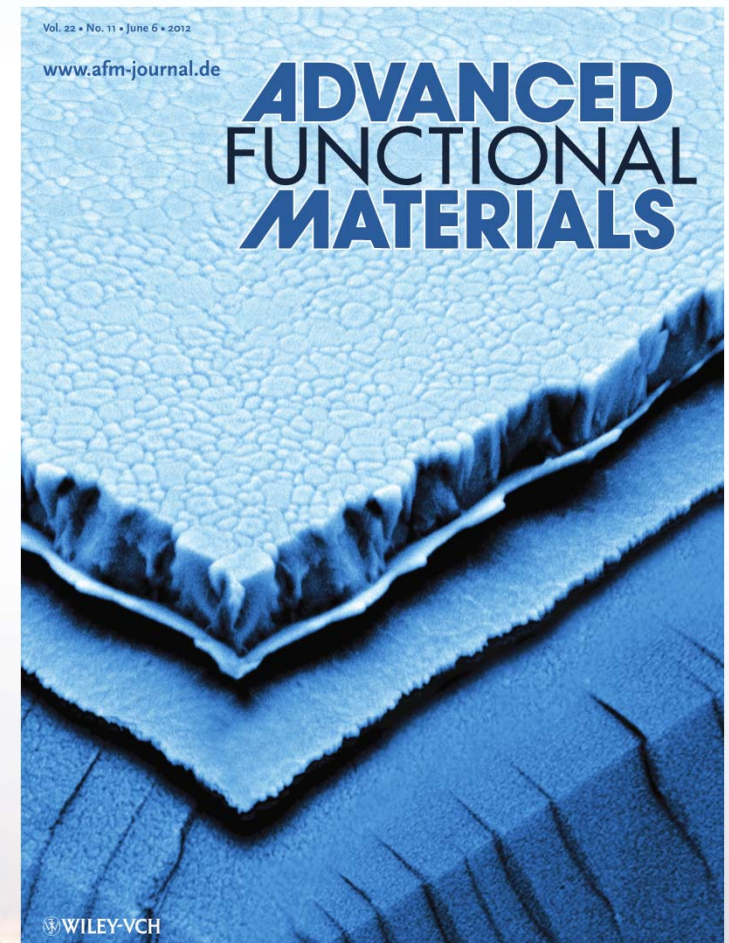
Open Issues:

- Pyrolysis temperature does not significantly effect gradient formation
- On substrates devoid of a titanium adhesion layer, no gradients form – even for 2-MOE sol-gel chemistries
- Heating rate and method **does** appear to matter
 - Gradient observed in RTA samples with 50° C/s ramp rate



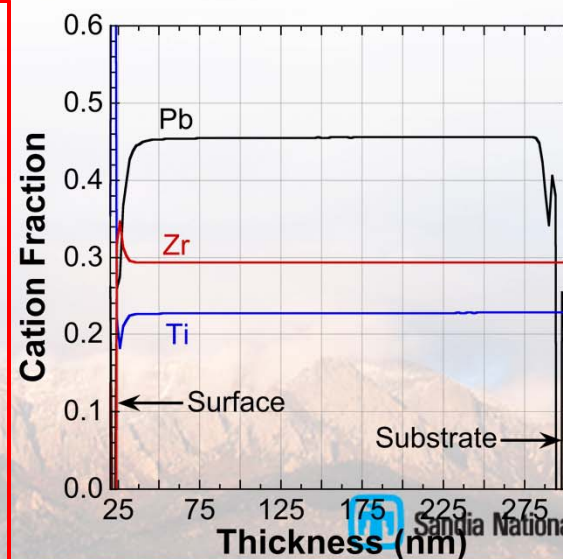
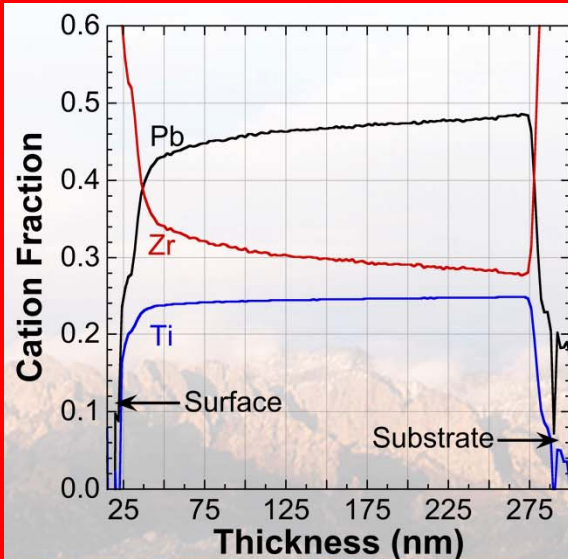
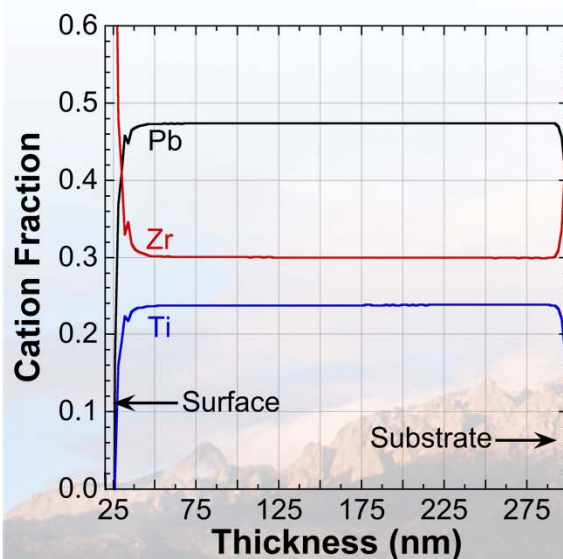
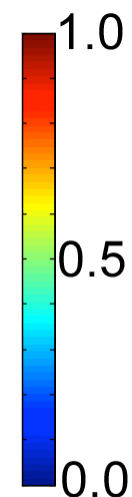
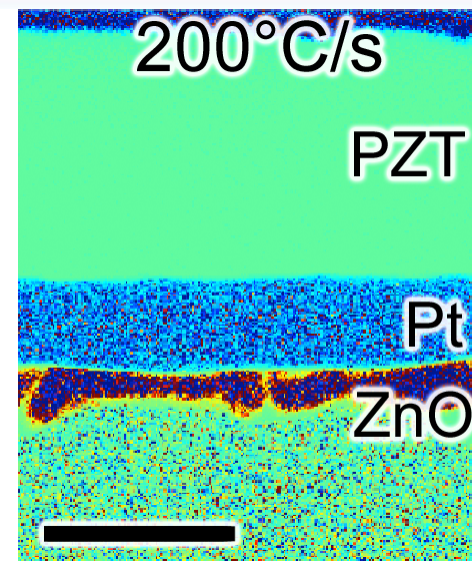
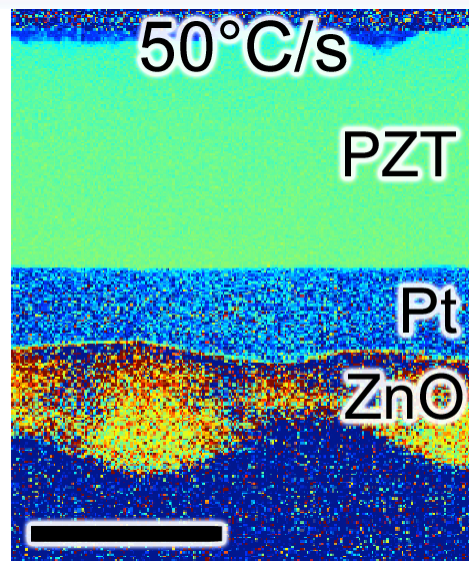
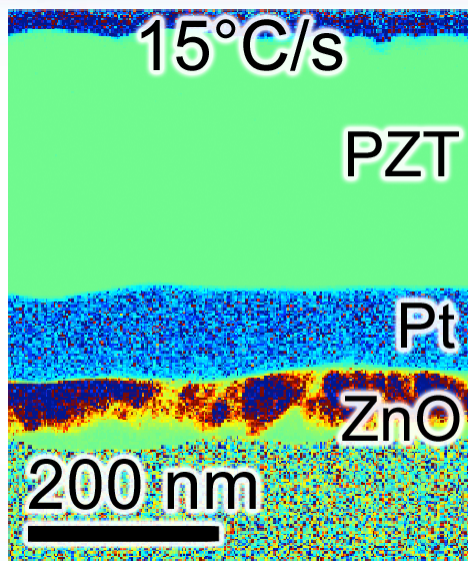
Summary

- IMO PZT films prepared on Pt/ZnO substrates possess improved electronic properties compared to traditional substrates
- ZnO as an adhesion layer enables chemically homogeneous films
- Even traditional sol-gel PZT films did not possess composition gradients on Pt/ZnO substrates: *heating rate effect?*
- *Stay tuned....*



Ramp Rate Effect: 2-MOE Films

Pt/ZnO
Lead Distribution



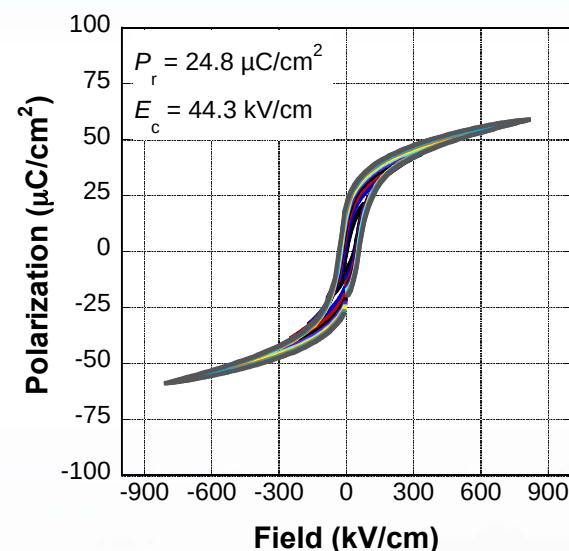
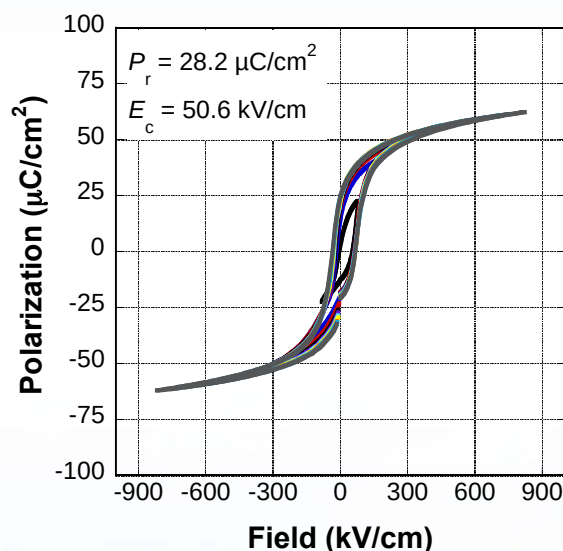
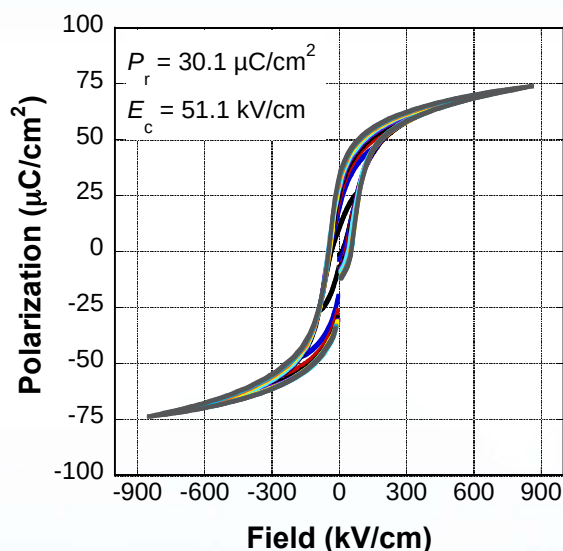
Ferroelectric Properties: IMO Films

300°C

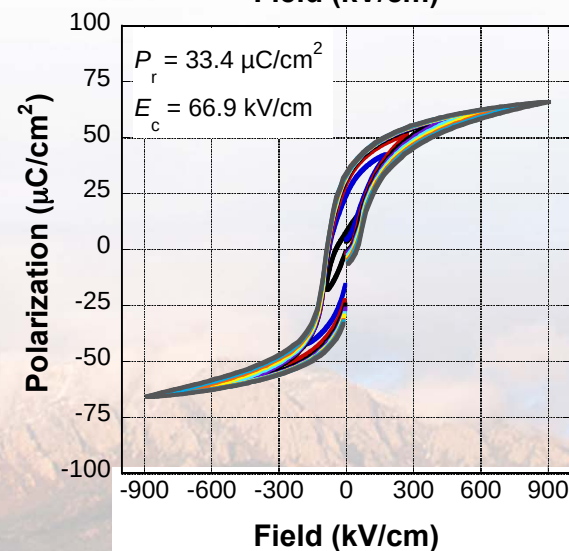
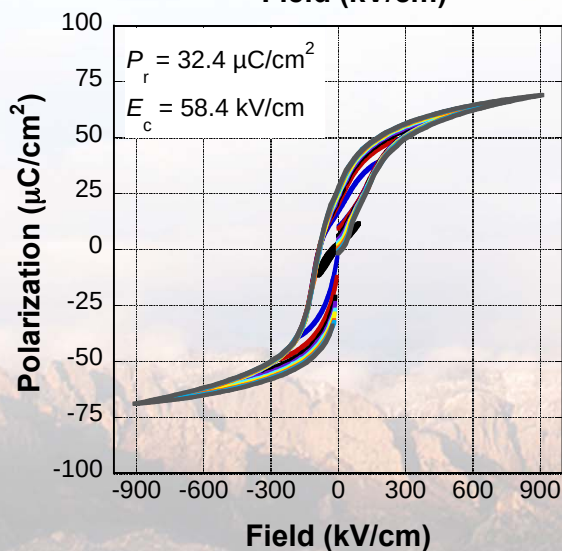
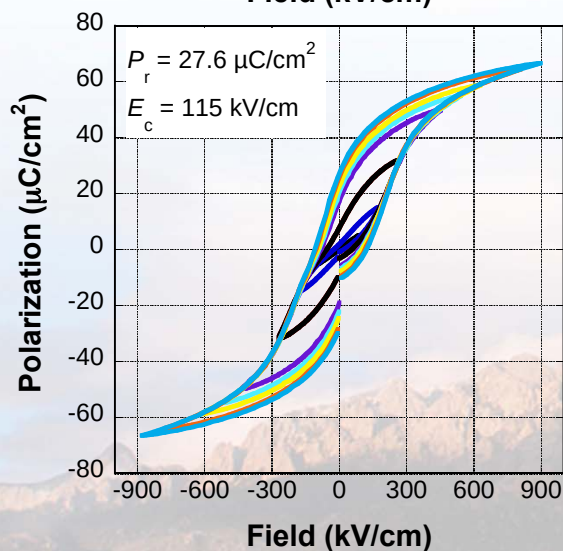
350°C

400°C

Pt/ZnO



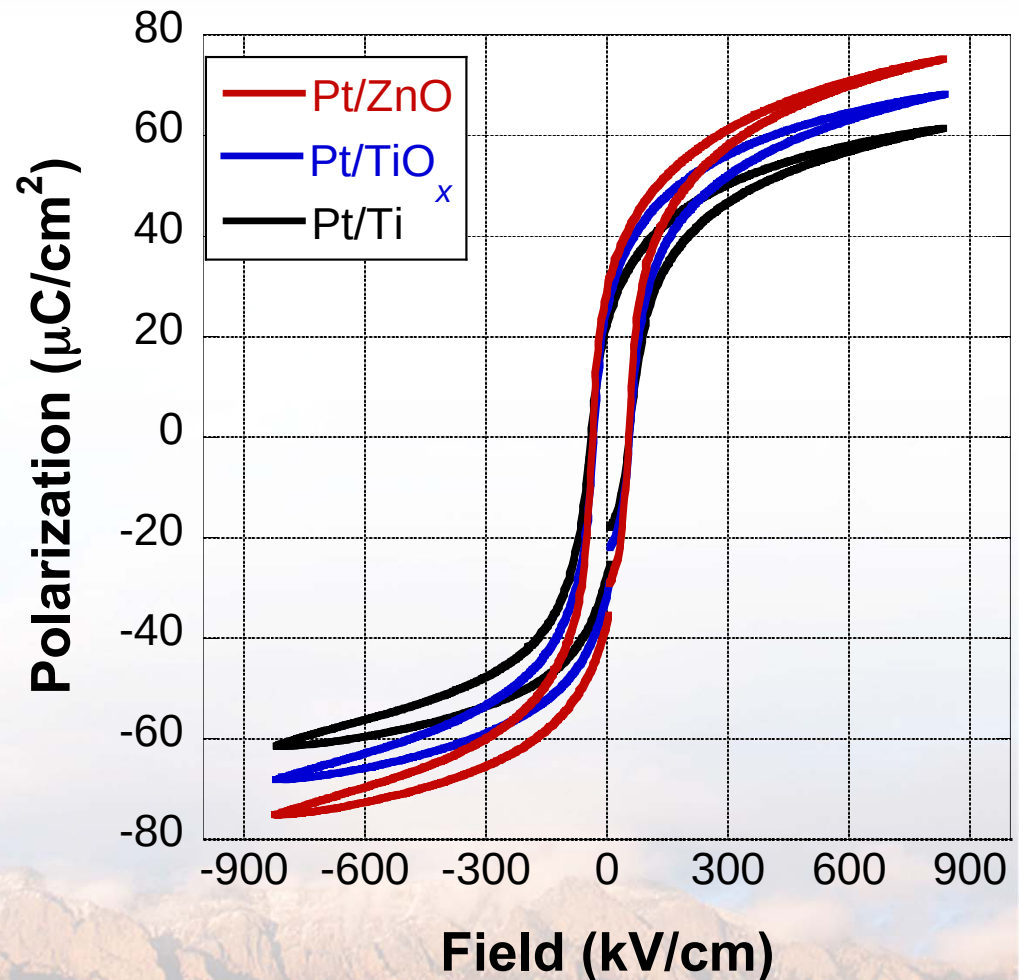
Pt/Ti



PZT Films Ferroelectric Properties: Pt/ZnO Substrates Superior

	$P_{\max}$ ($\mu\text{C}/\text{cm}^2$)	P_r ($\mu\text{C}/\text{cm}^2$)	E_c (kV/cm)	K
Pt/Ti	61.5	23.8	47.0	1440
Pt/TiO _x	68.2	27.0	44.1	1620
Pt/ZnO	75.3	32.2	43.8	1740

- Two film systems
- Performance improvements identified in both on ZnO-buffered substrates
- Mechanism?



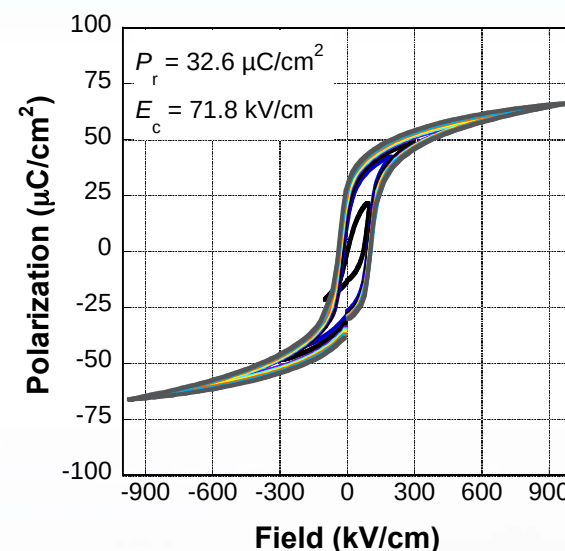
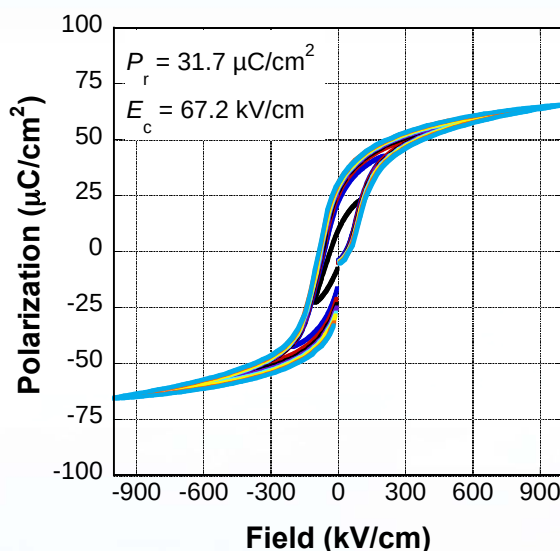
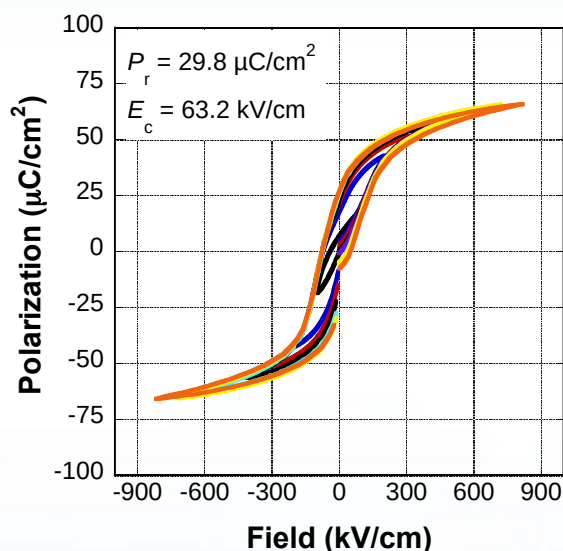
Ferroelectric Properties: 2MOE Films

300°C

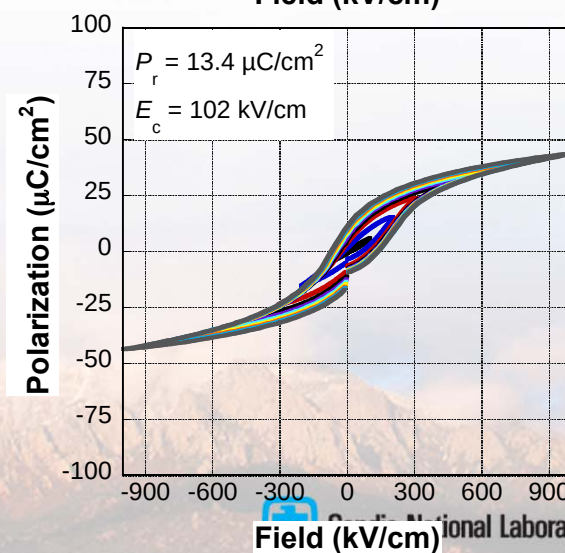
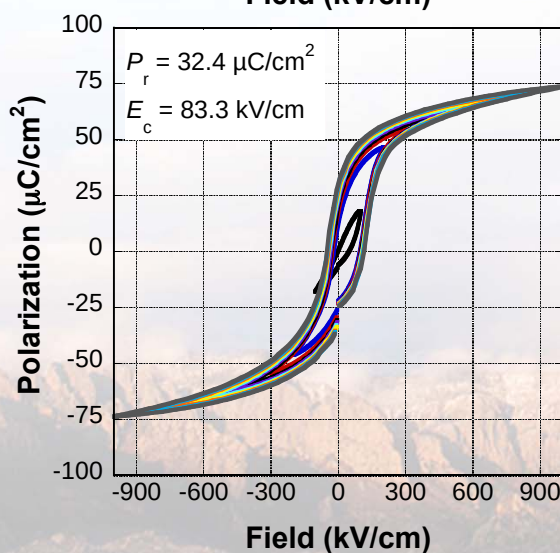
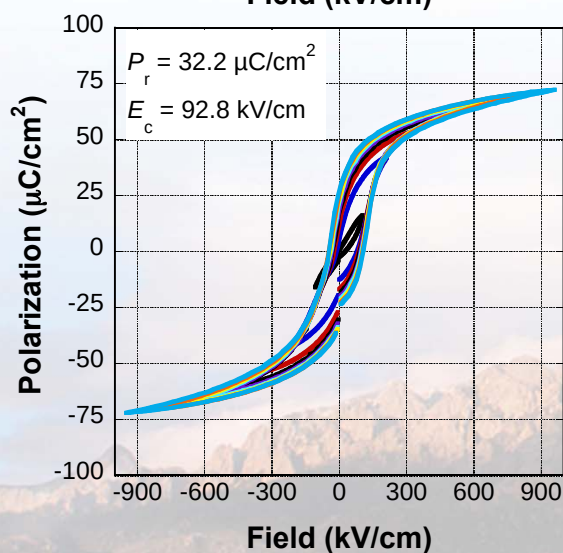
350°C

400°C

Pt/ZnO

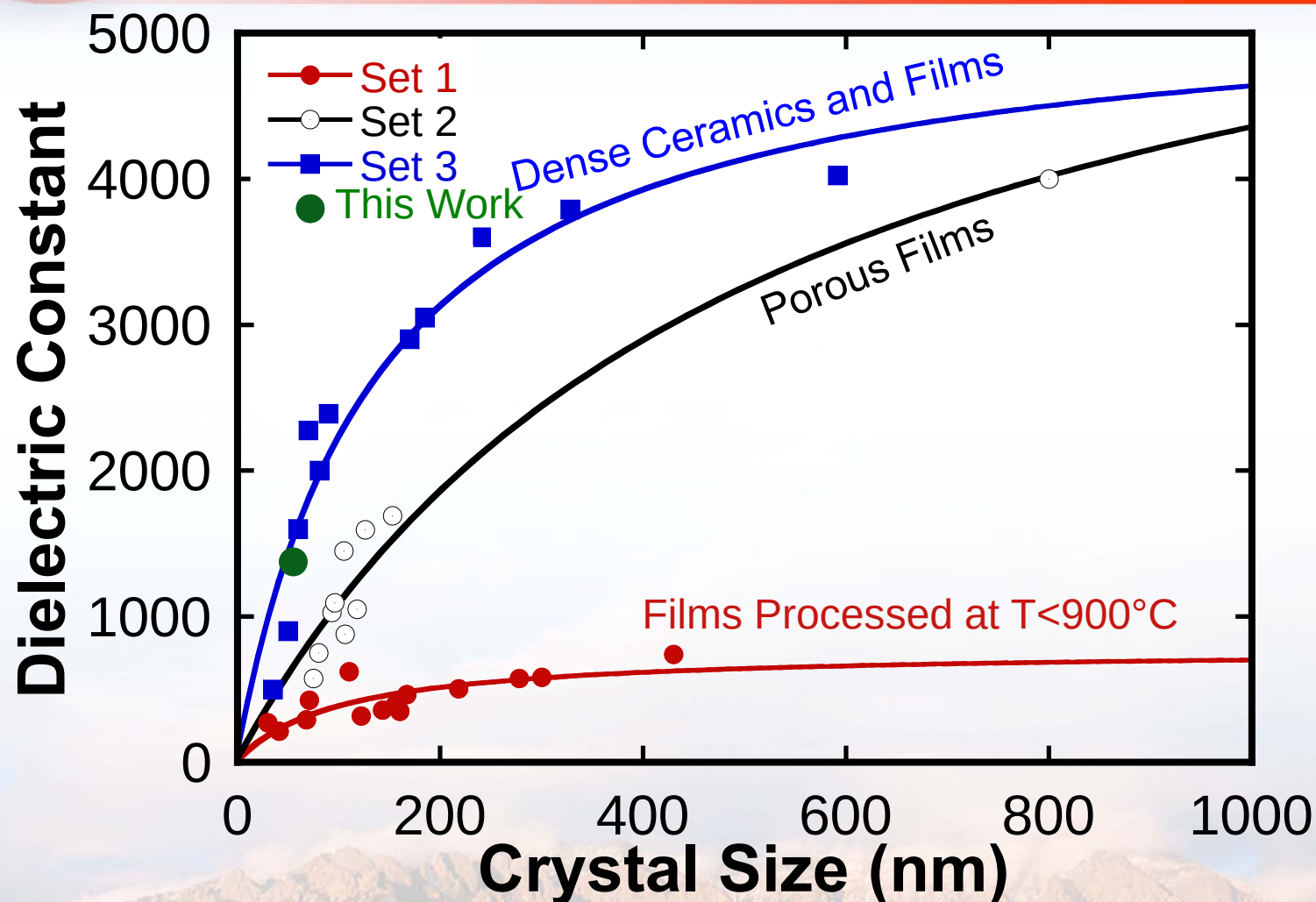


Pt/Ti



Field (kV/cm) National Laboratories

BaTiO₃ Scaling Effects



First known BaTiO₃ film with bulk properties on Si

Plot adapted from Aygün et. al. *J. Appl. Phys.* (2011)

 Sandia National Laboratories

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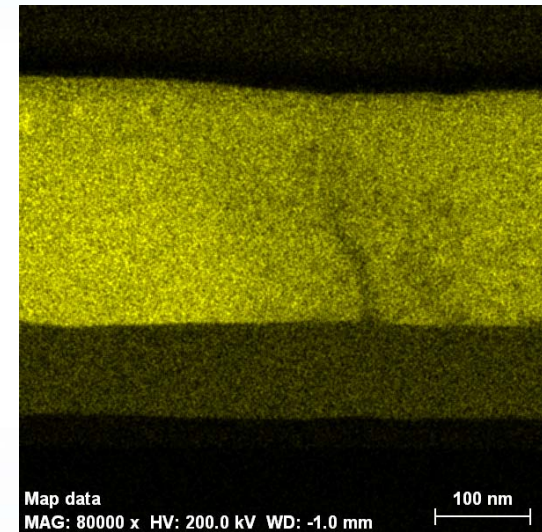
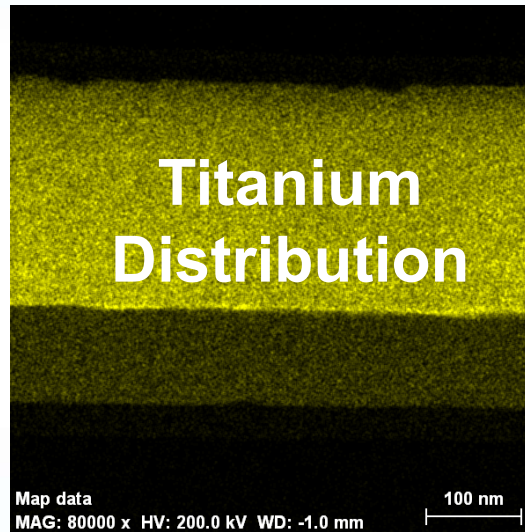
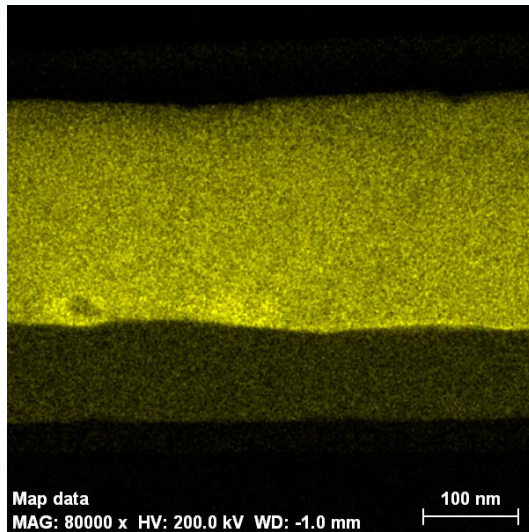
Solution Chemistry Effect: 2-MOE films

300°C

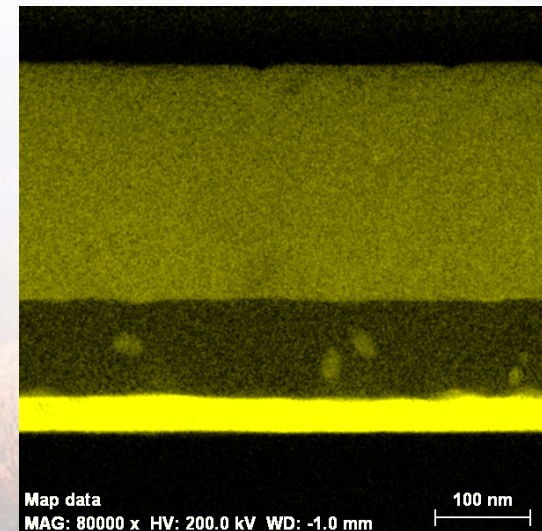
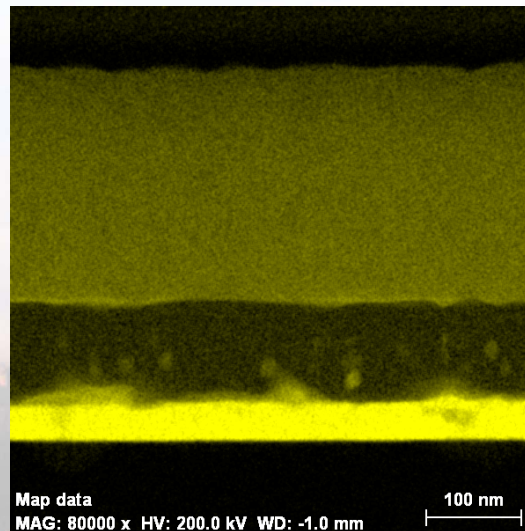
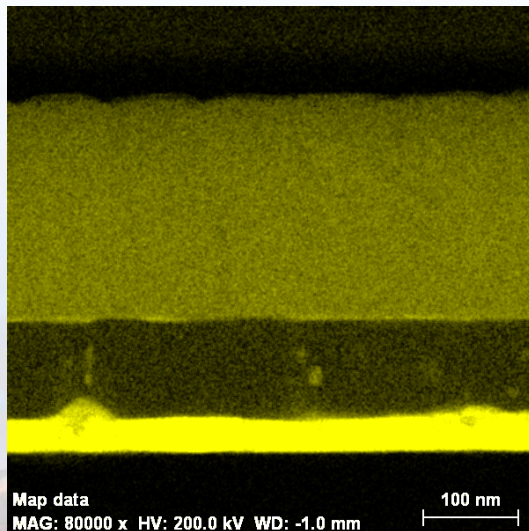
350°C

400°C

Pt/ZnO



Pt/Ti



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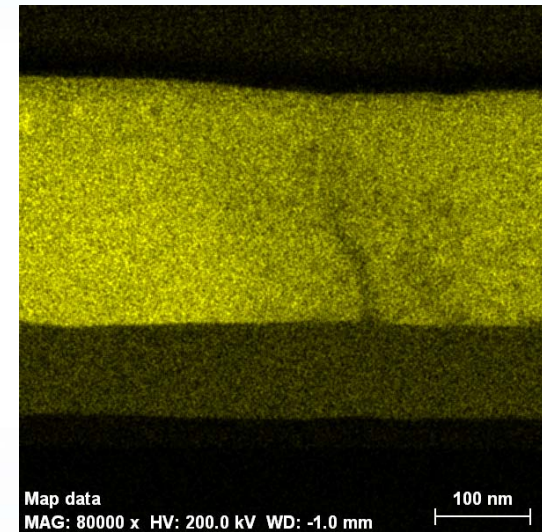
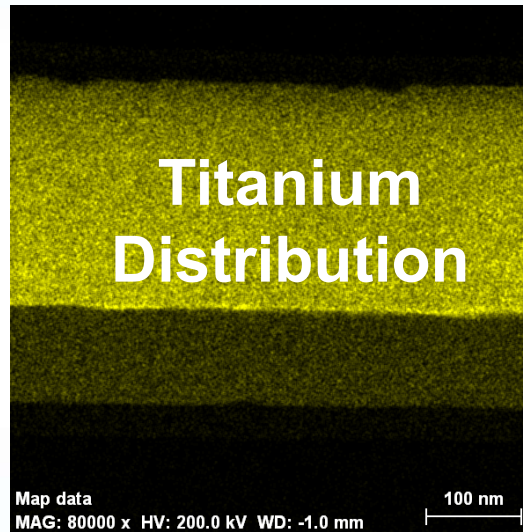
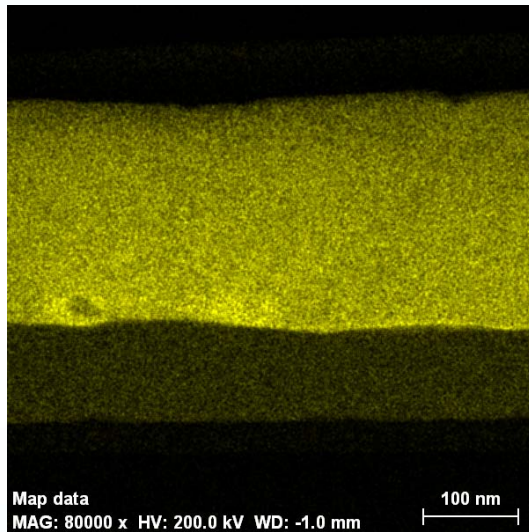
Solution Chemistry Effect: 2-MOE films

300°C

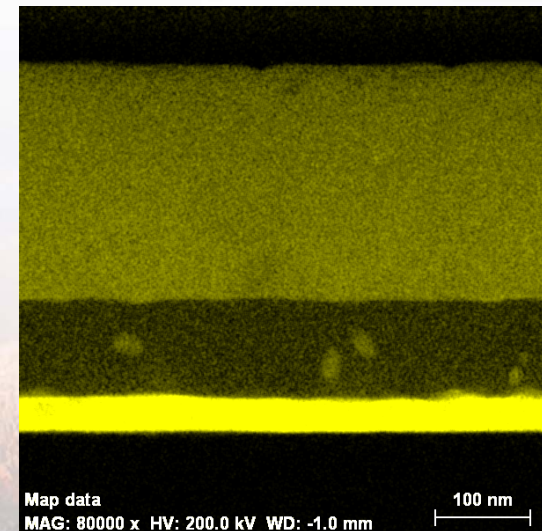
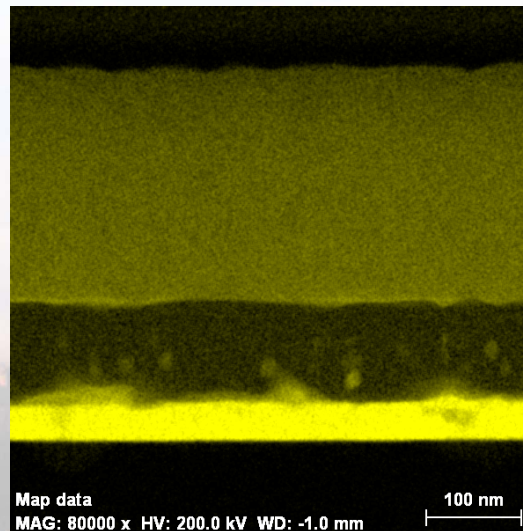
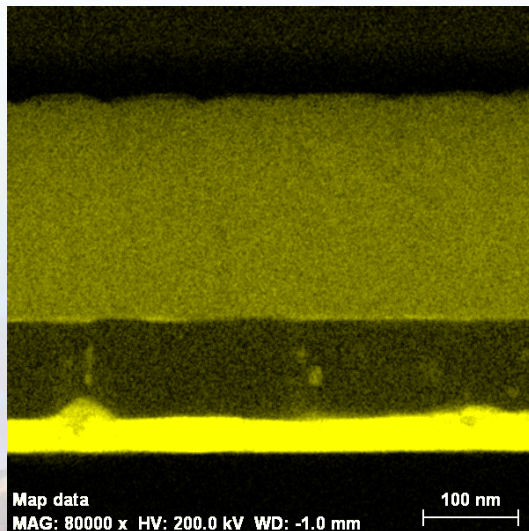
350°C

400°C

Pt/ZnO



Pt/Ti



ratories

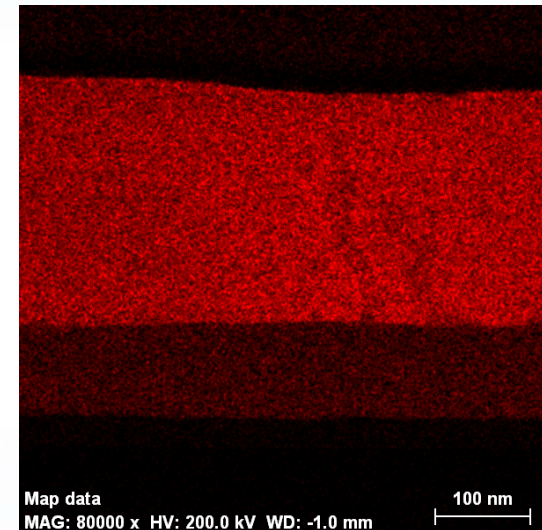
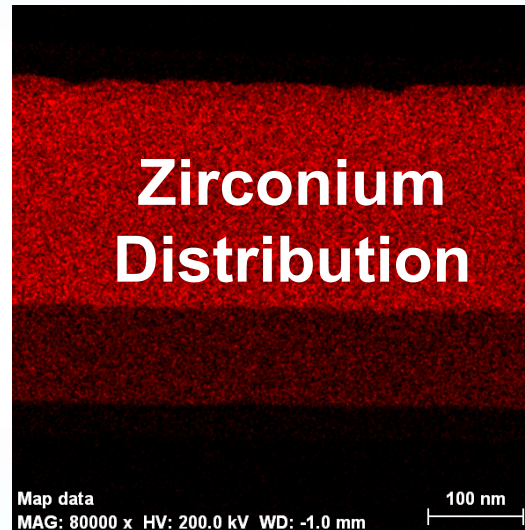
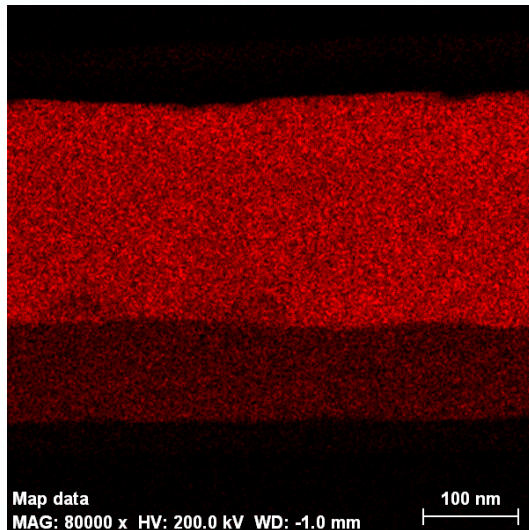
Solution Chemistry Effect: 2-MOE films

300°C

350°C

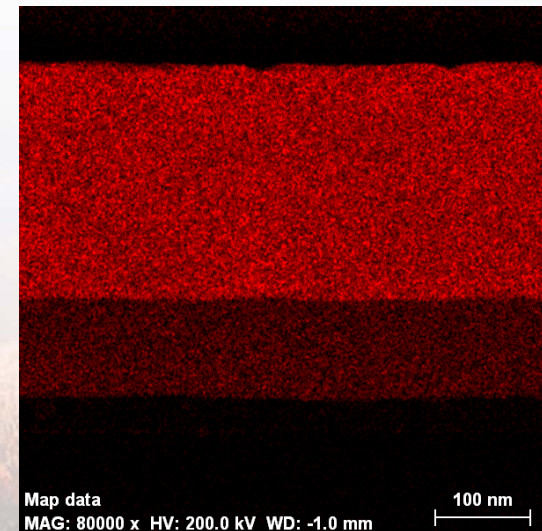
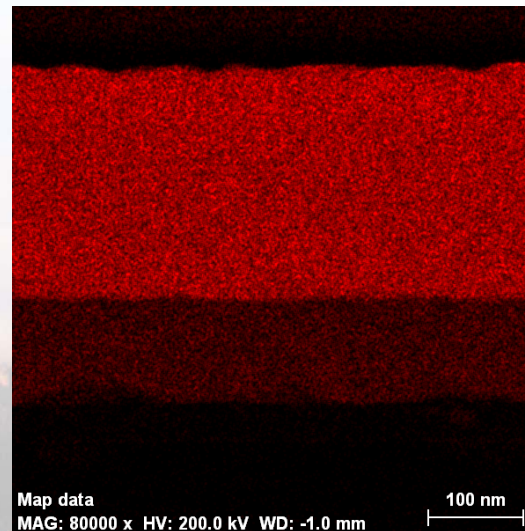
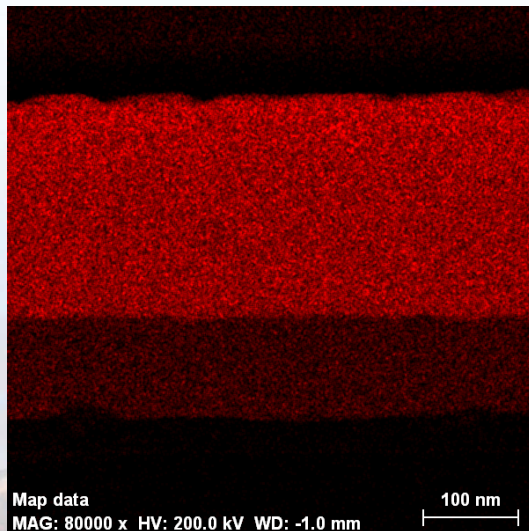
400°C

Pt/ZnO



Zirconium
Distribution

Pt/Ti



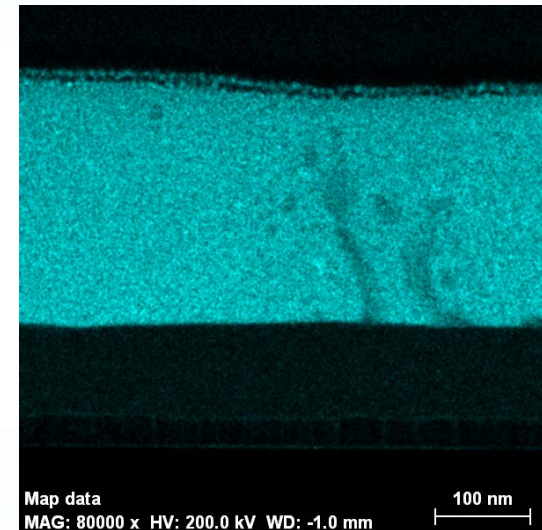
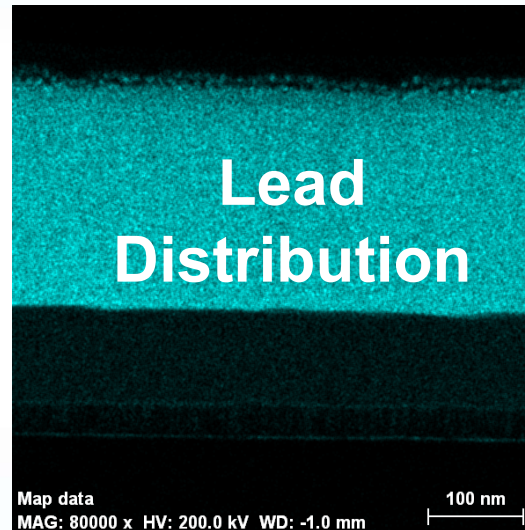
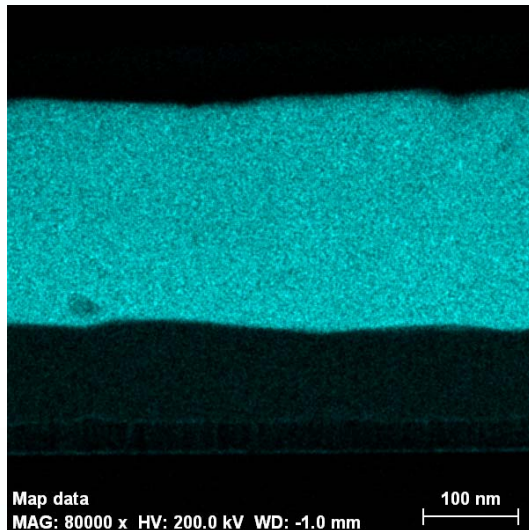
Solution Chemistry Effect: 2-MOE films

300°C

350°C

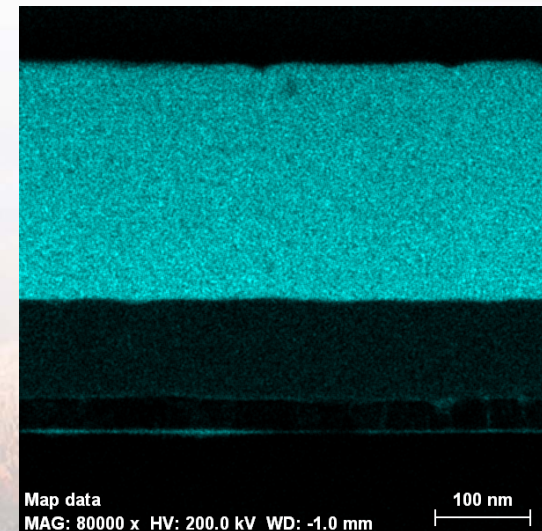
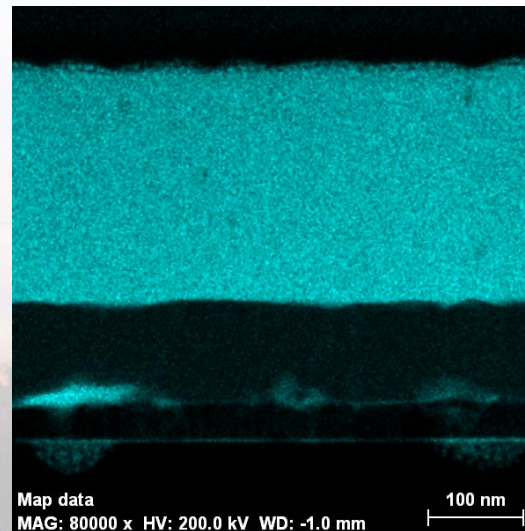
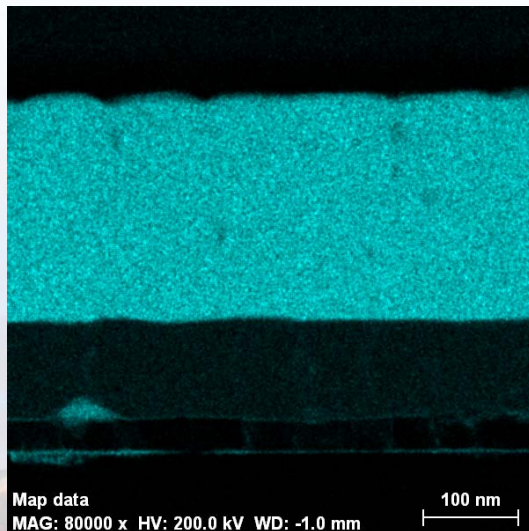
400°C

Pt/ZnO



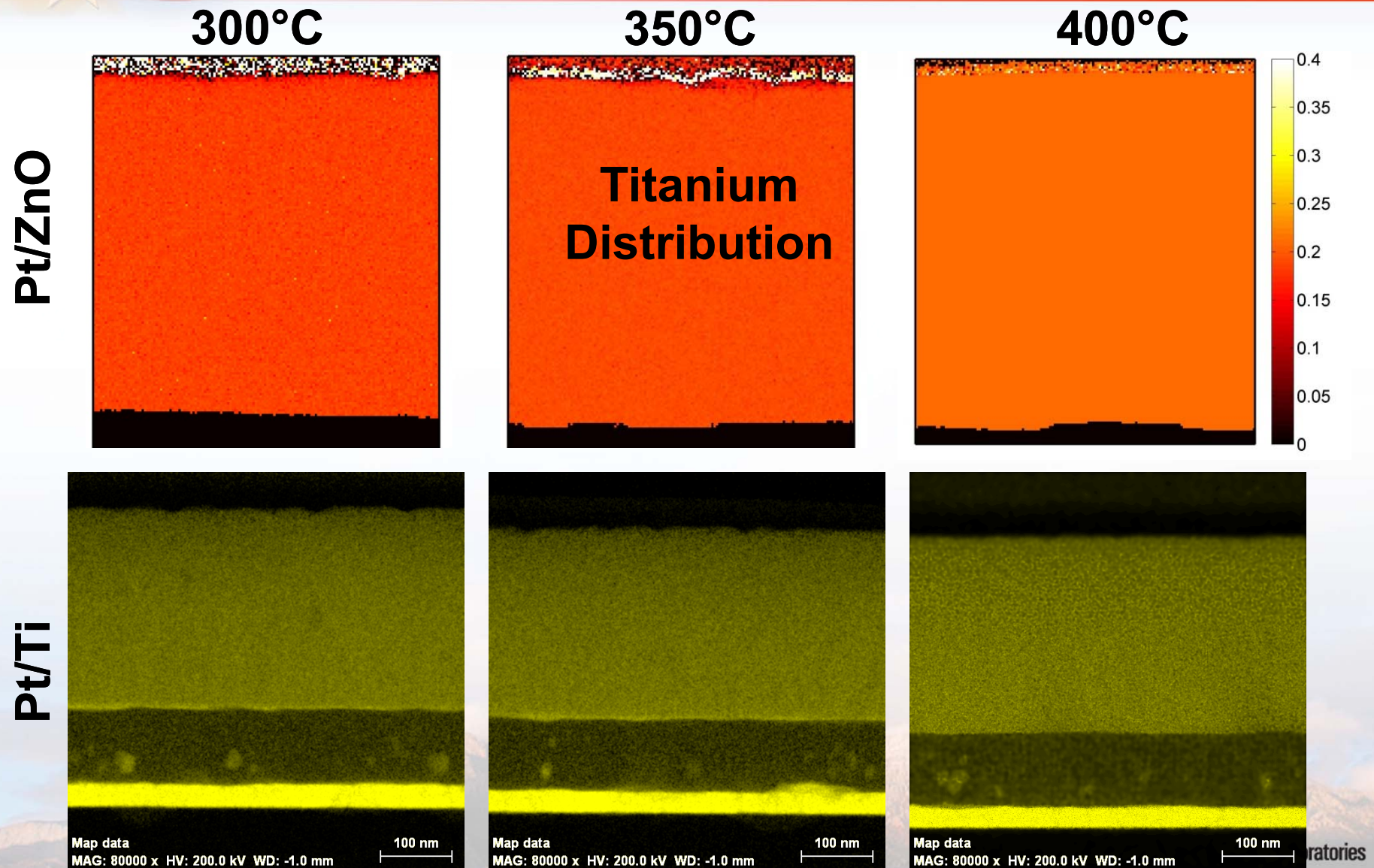
Lead
Distribution

Pt/Ti

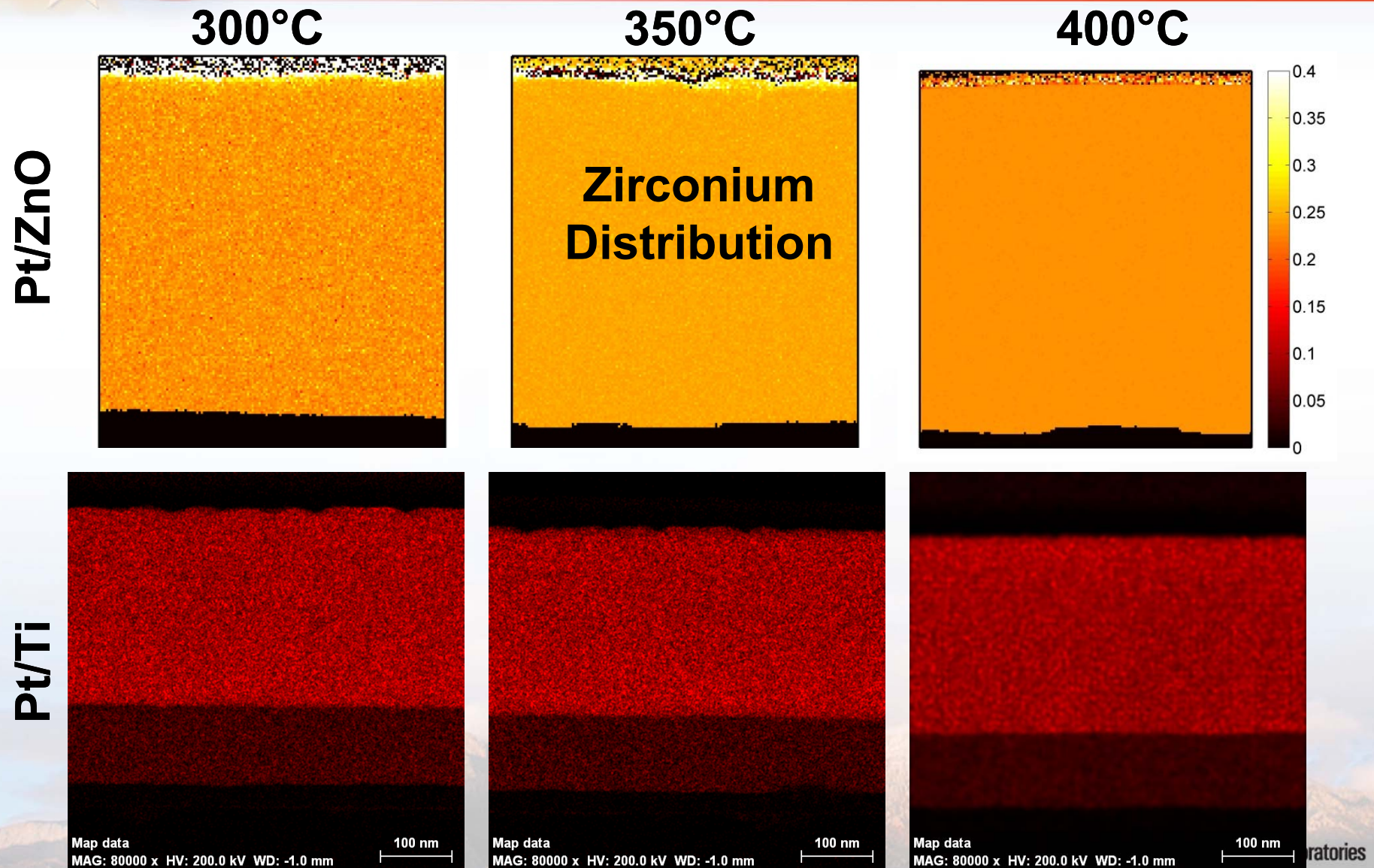


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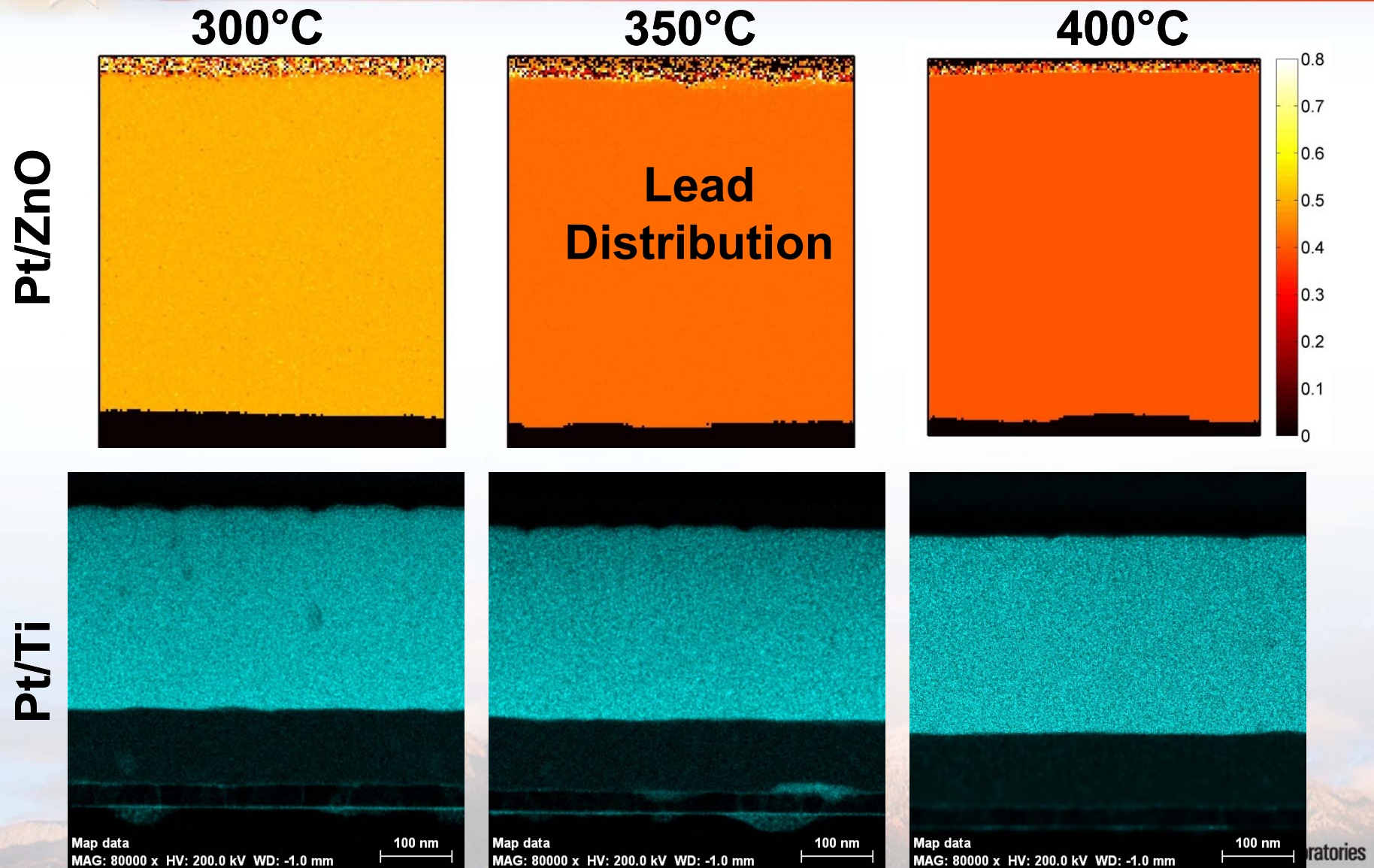
Solution Chemistry Effect: IMO Films

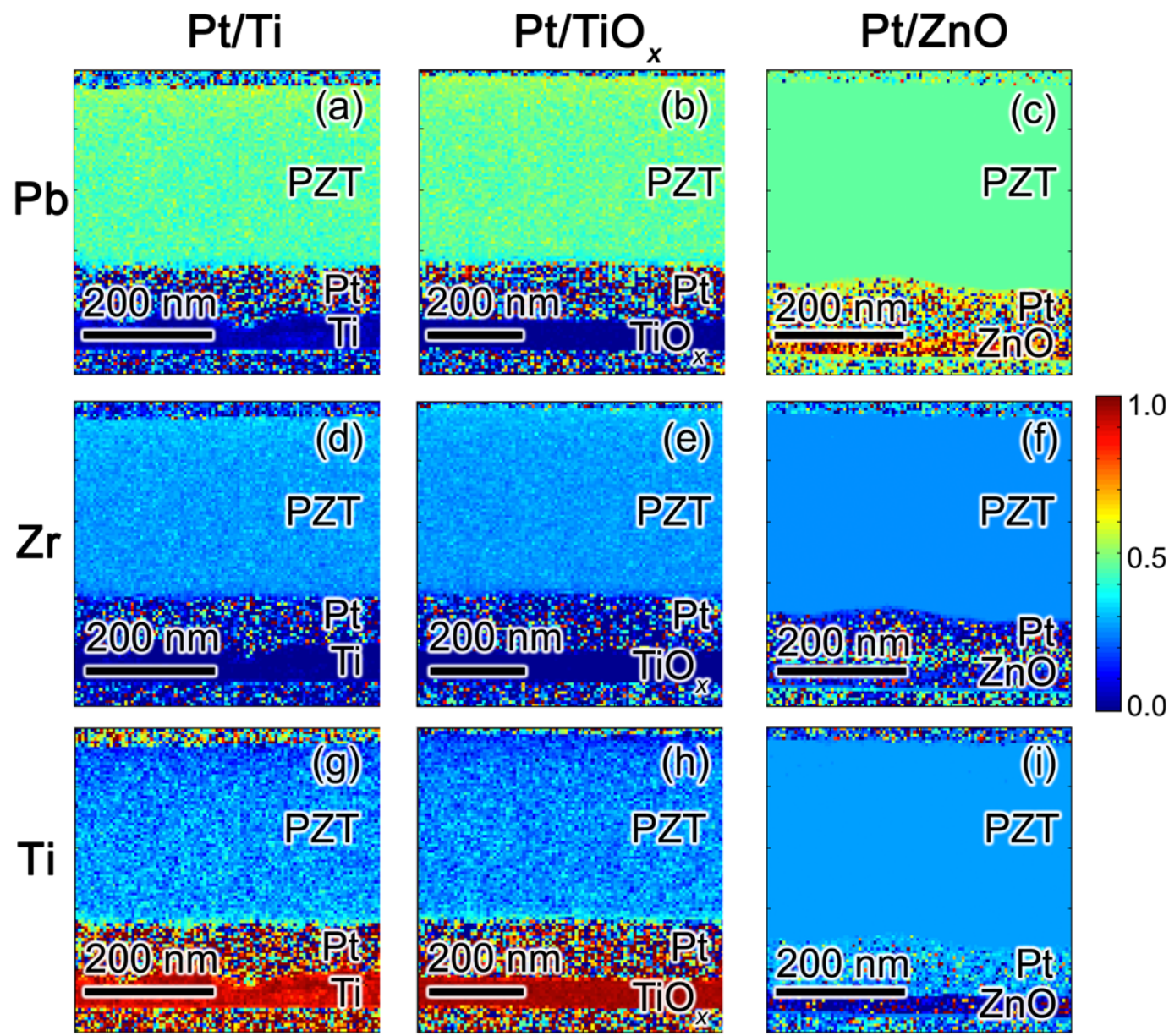


Solution Chemistry Effect: IMO Films



Solution Chemistry Effect: IMO Films





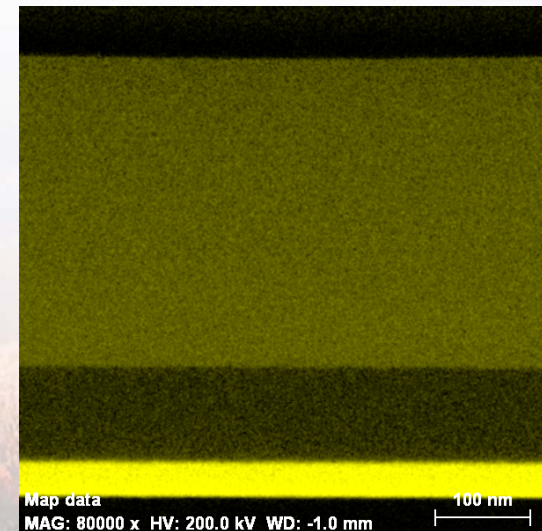
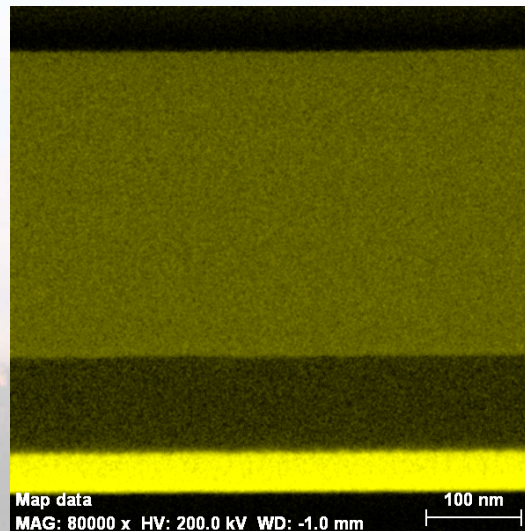
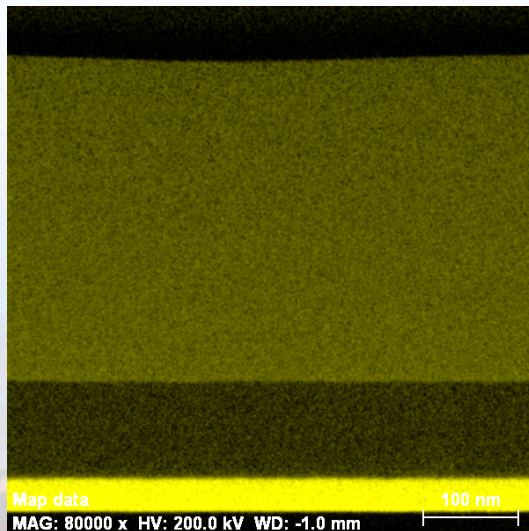
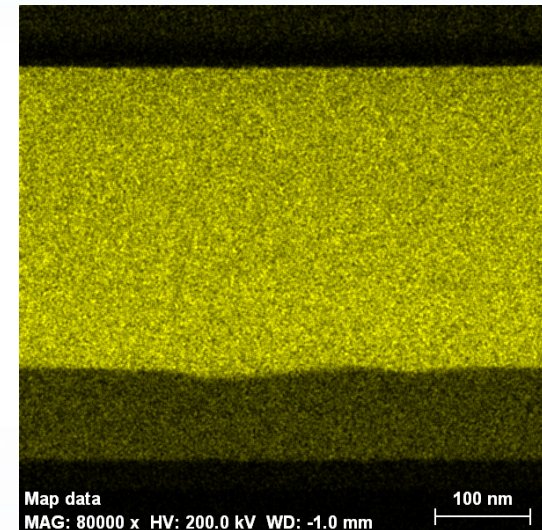
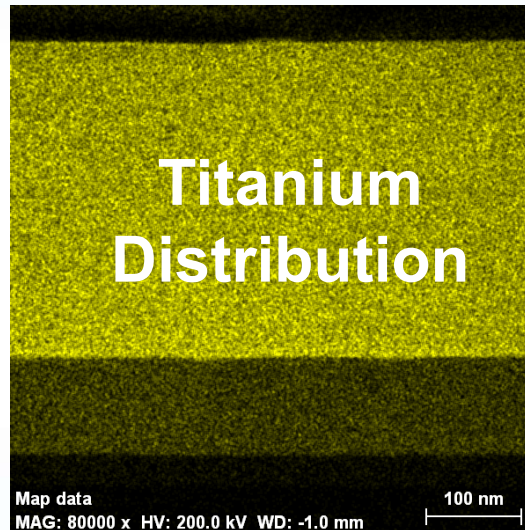
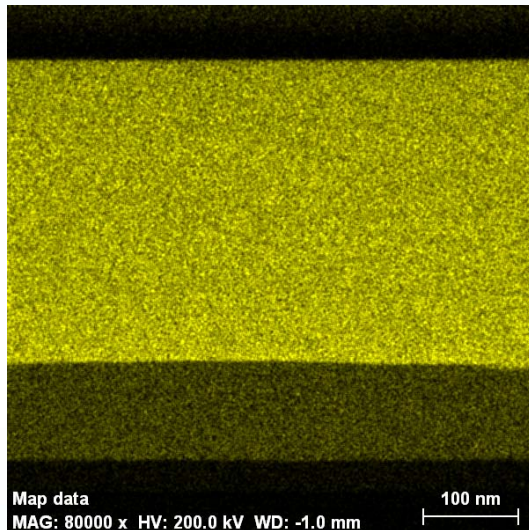


Solutions

400°C

Pt/Ti

Titanium Distribution



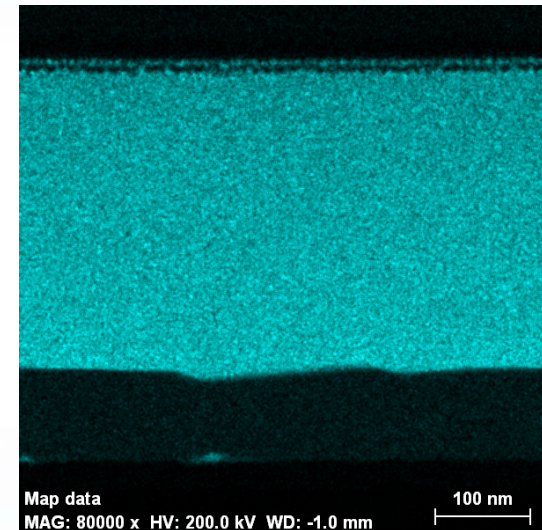
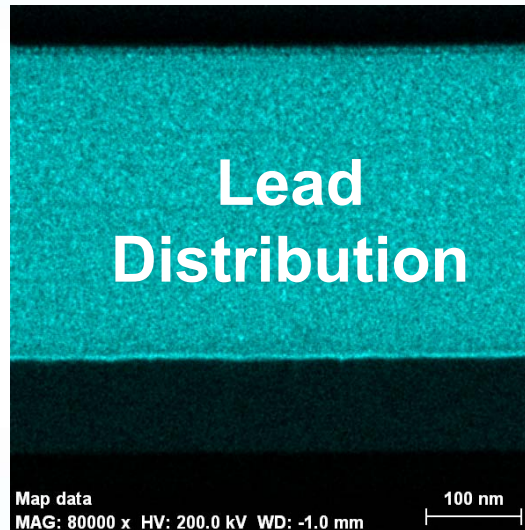
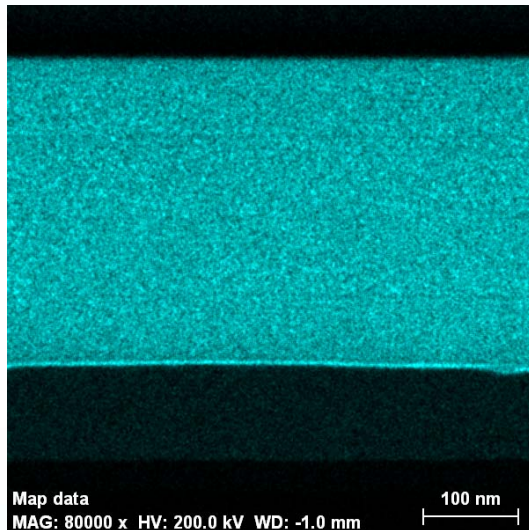
Solution Chemistry Effect: 2-MOE films Prior to crystallization

300°C

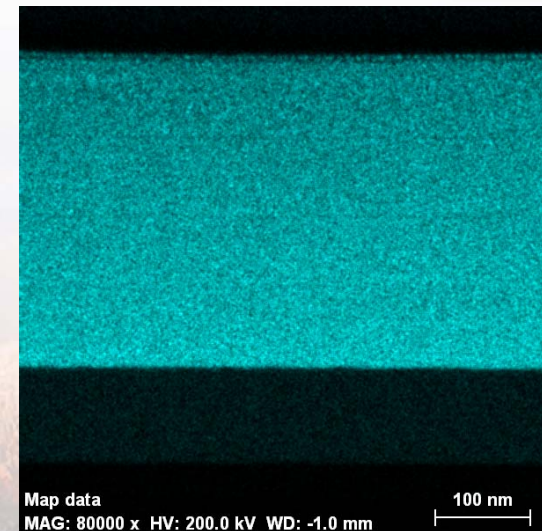
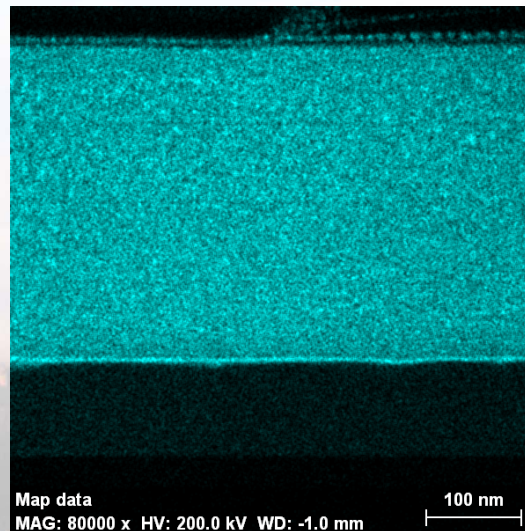
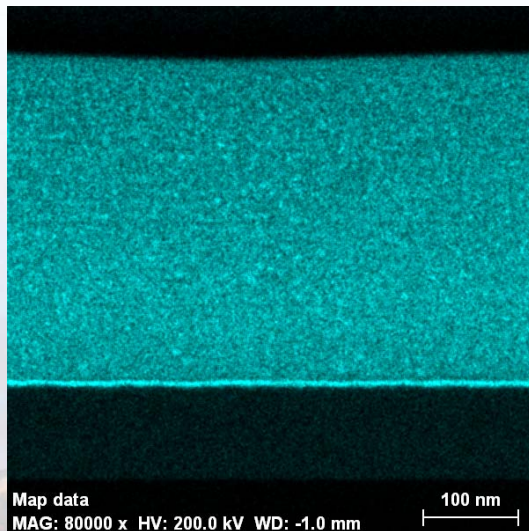
350°C

400°C

Pt/ZnO

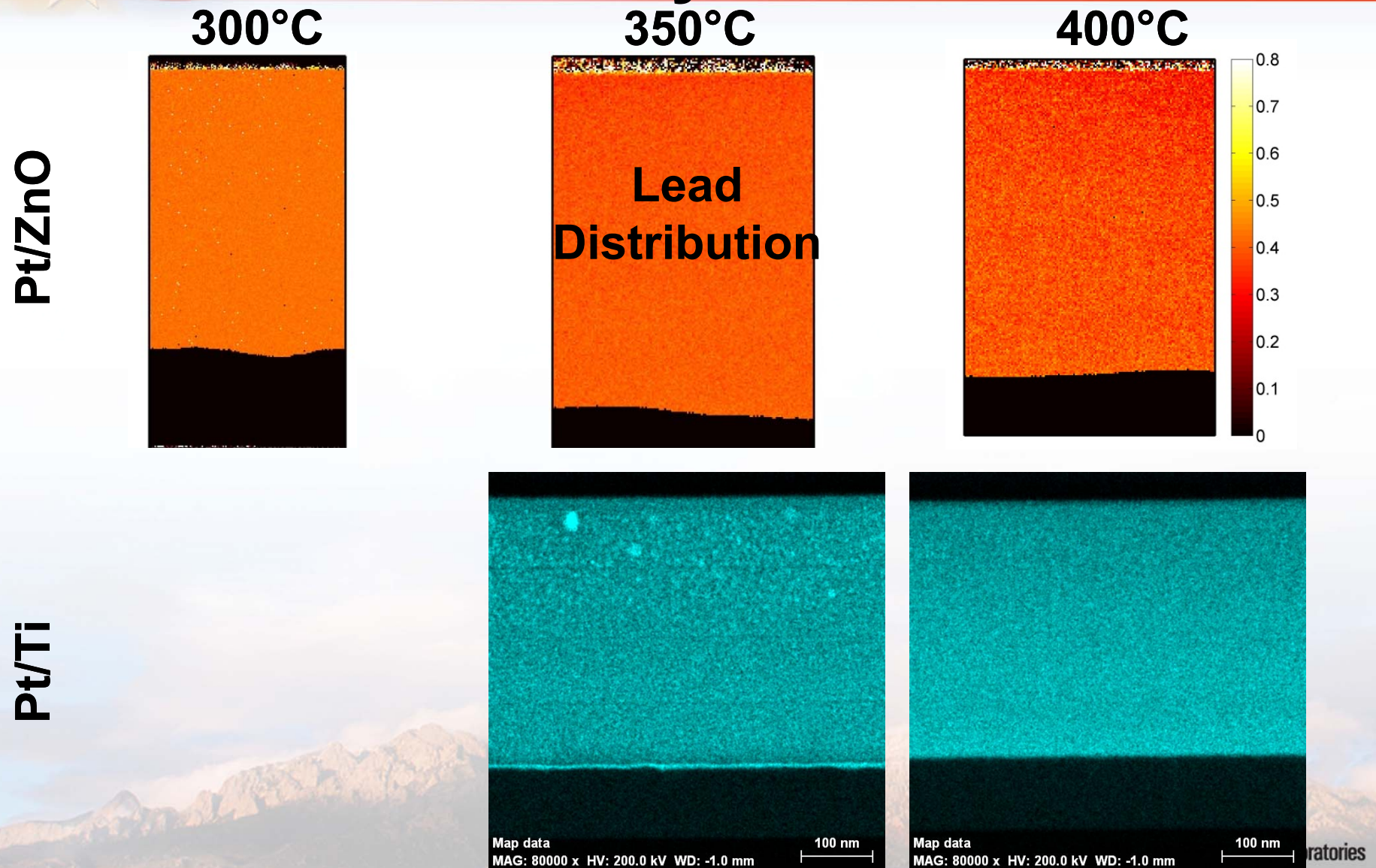


Pt/Ti

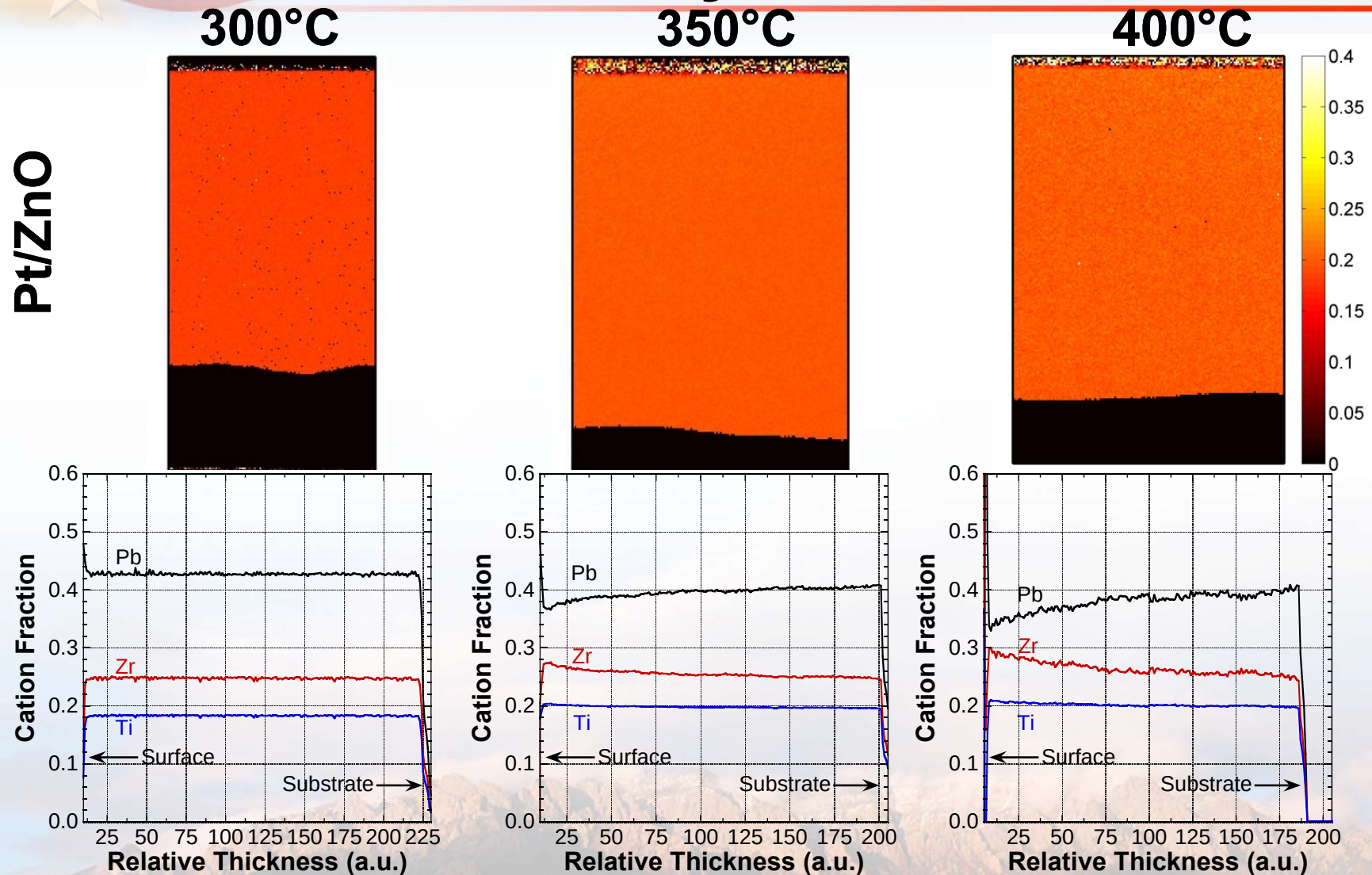


ratories

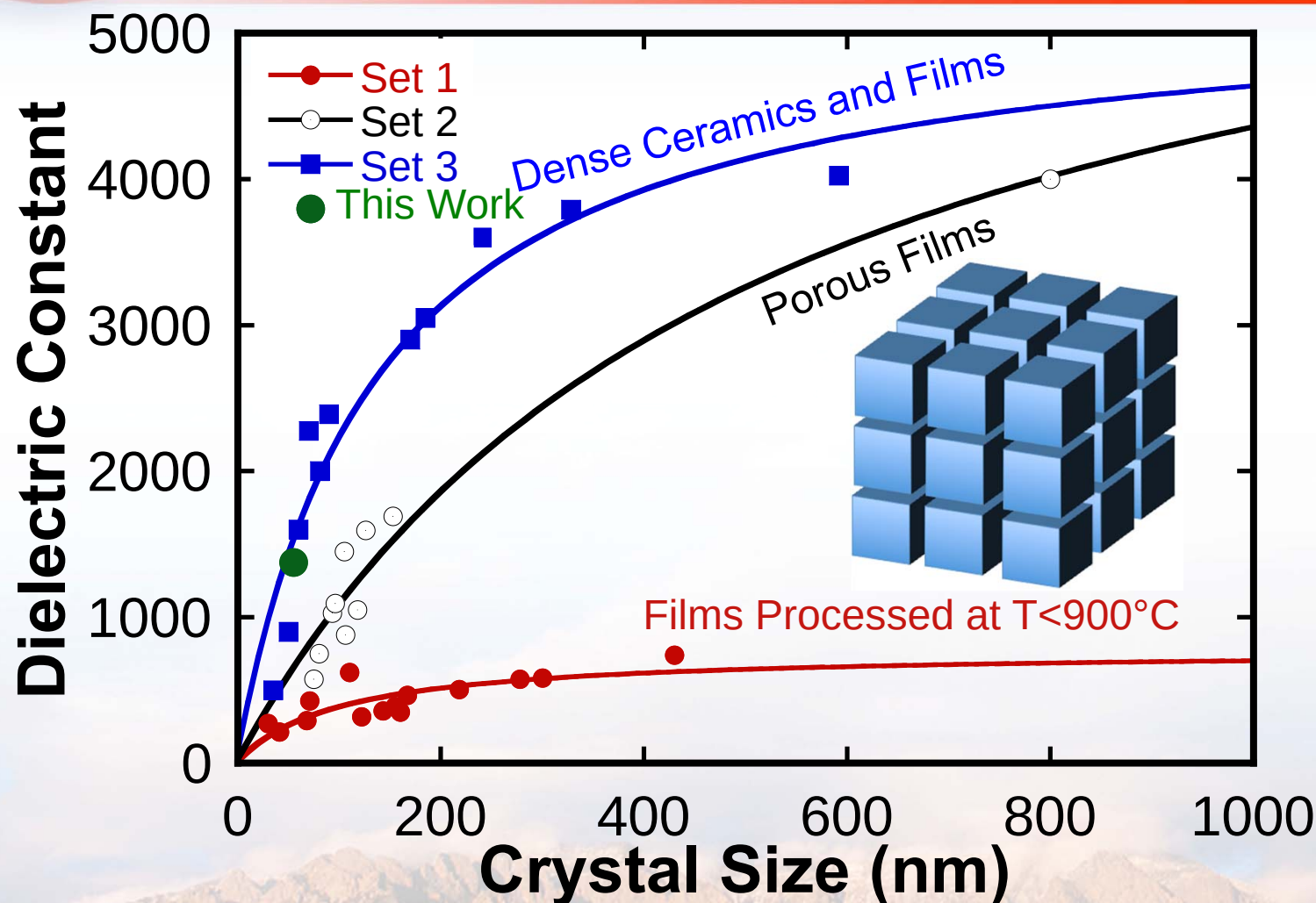
Solution Chemistry Effect: IMO Films Prior to Crystallization



Solution Chemistry Effect: IMO Films Prior to crystallization



BaTiO₃ Scaling Effects



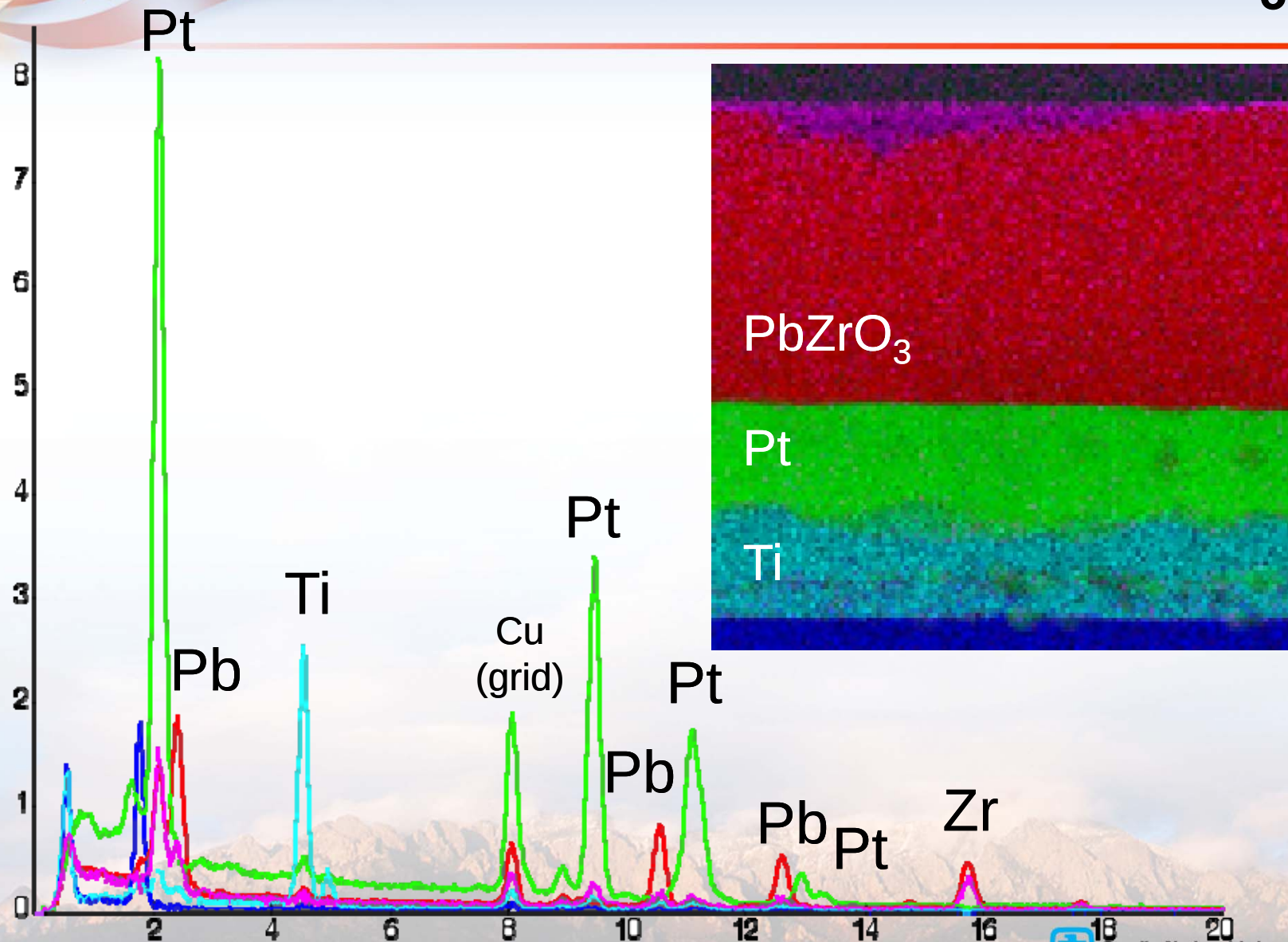
First known BaTiO₃ film with bulk properties on Si

Plot adapted from Aygün et. al. J. Appl. Phys. (2011)

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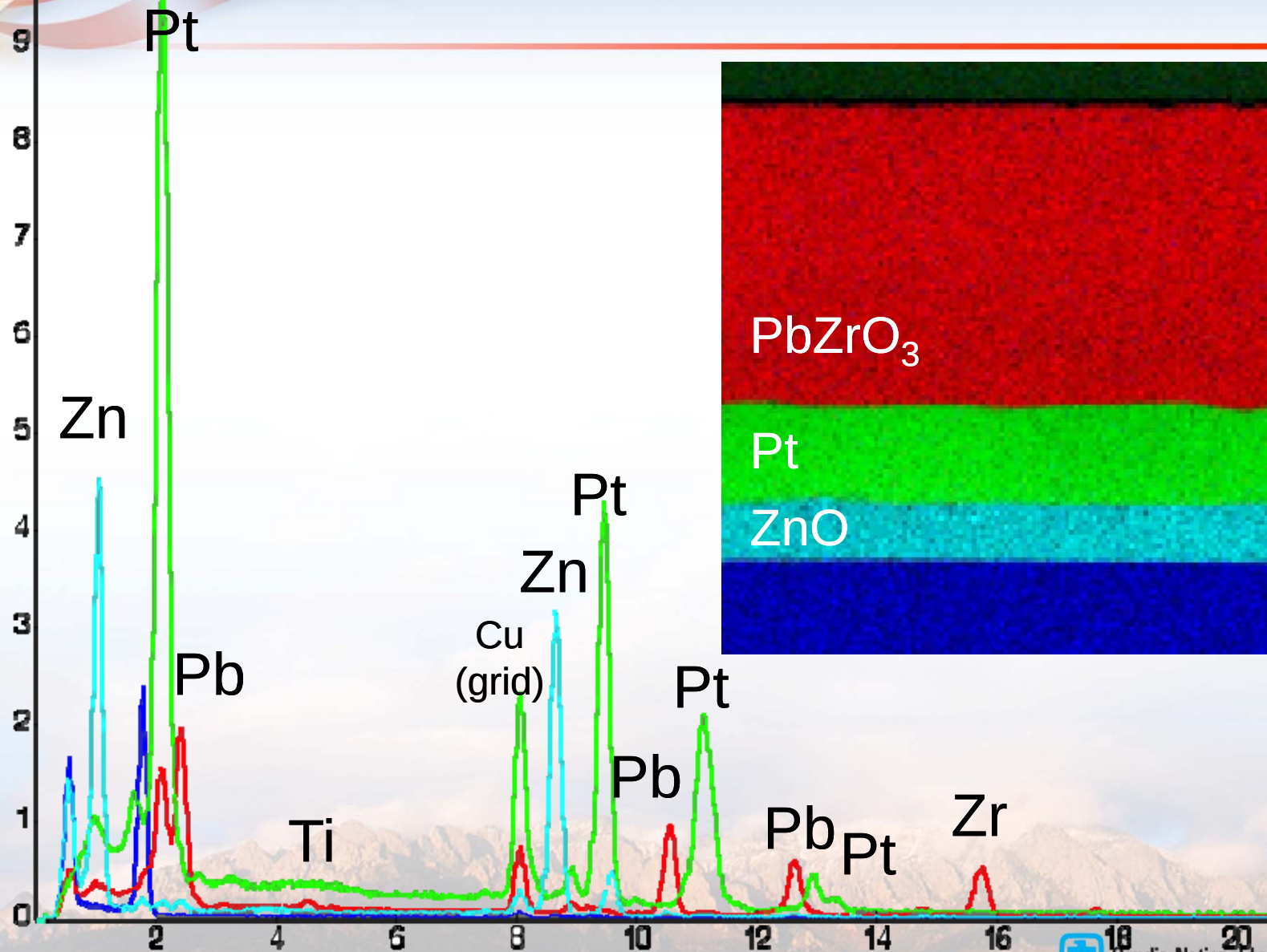
J.F. Ihlefeld, ISAF 2013, 23 July 2013

Test for Gradient Mechanism: PbZrO_3



Sandia National Laboratories

Test for Gradient Mechanism: PbZrO_3



Crystallite vs. Grain Size

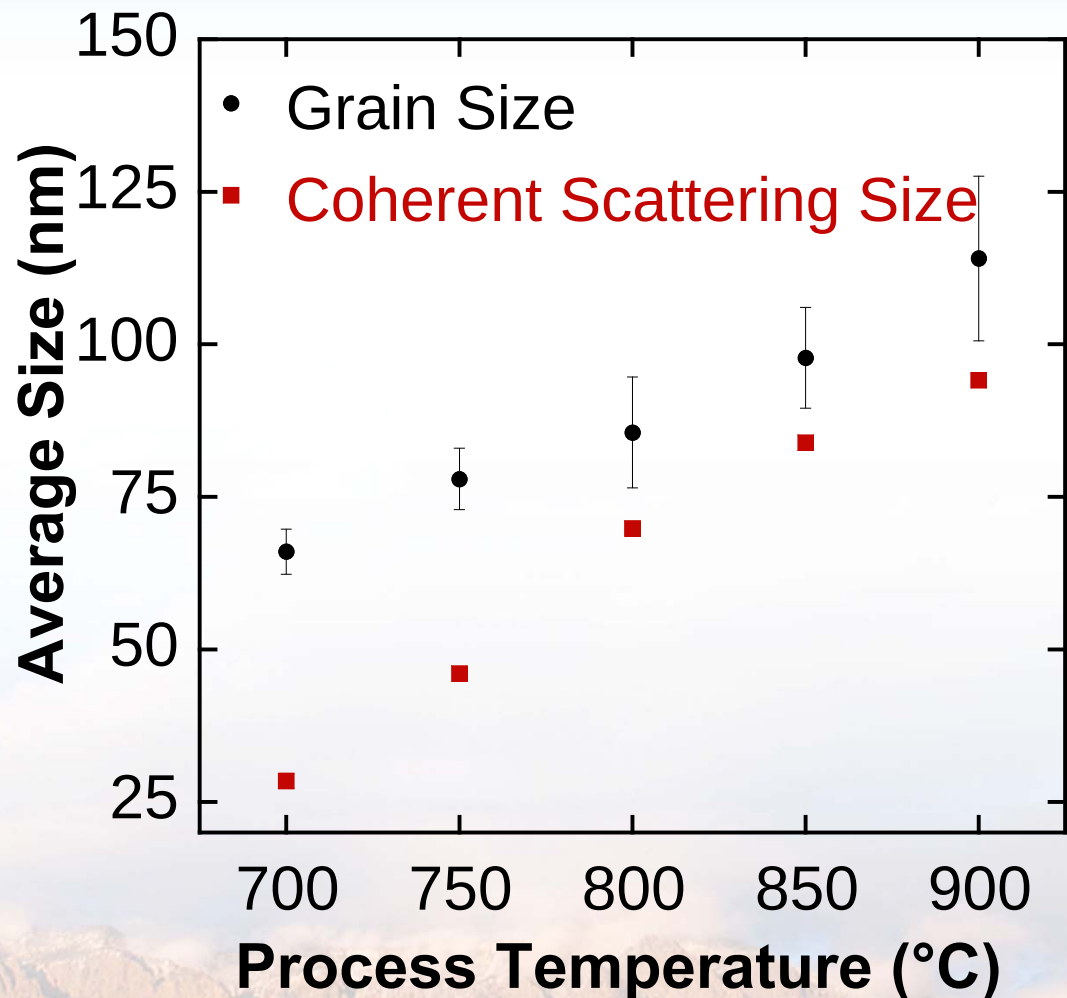
Grain Size

- AFM and FE-SEM
- 95% confidence interval

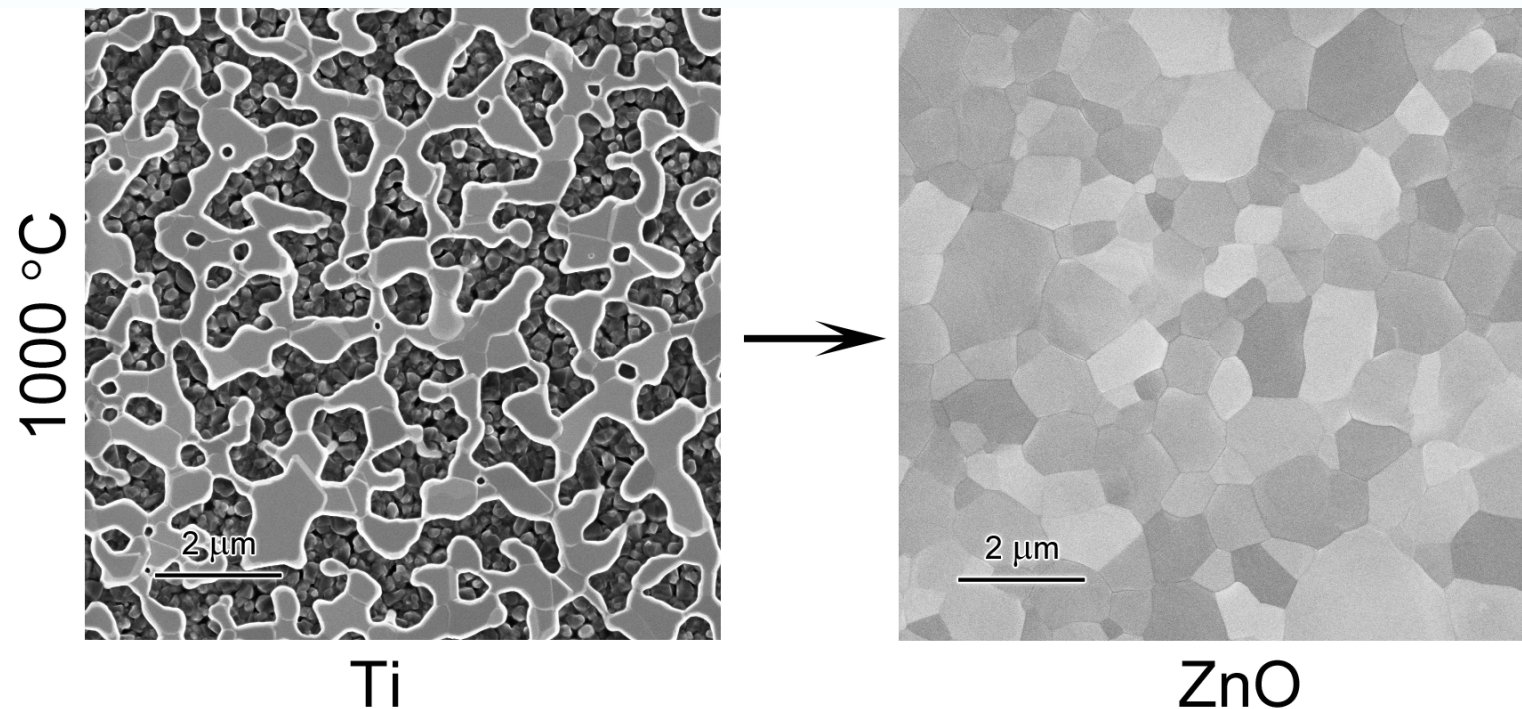
Coherent Scattering Size

- CHES
- Scherrer's Formula on 111 reflections

**Sub-grain structure
dominates dielectric
response**



Pt/ZnO as a high temperature substrate



- ZnO buffered Pt/Si is robust enough to withstand process temps in excess of 1000 °C
- Ideal candidate for materials previously best grown on single crystals:
 - BaTiO_3
 - $(\text{Ba}, \text{Sr})\text{TiO}_3$

MLC Scaling History



0603: 1600 μm X 800 μm
Most popular pre-2002 size



0402: 1000 μm X 500 μm
Most widely used size

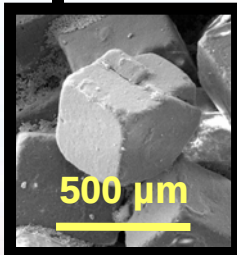


0201: 600 μm X 300 μm
1 μF in X5R rating



01005: 400 μm X 200 μm
Limited use after 2004

to-scale image
of salt grain



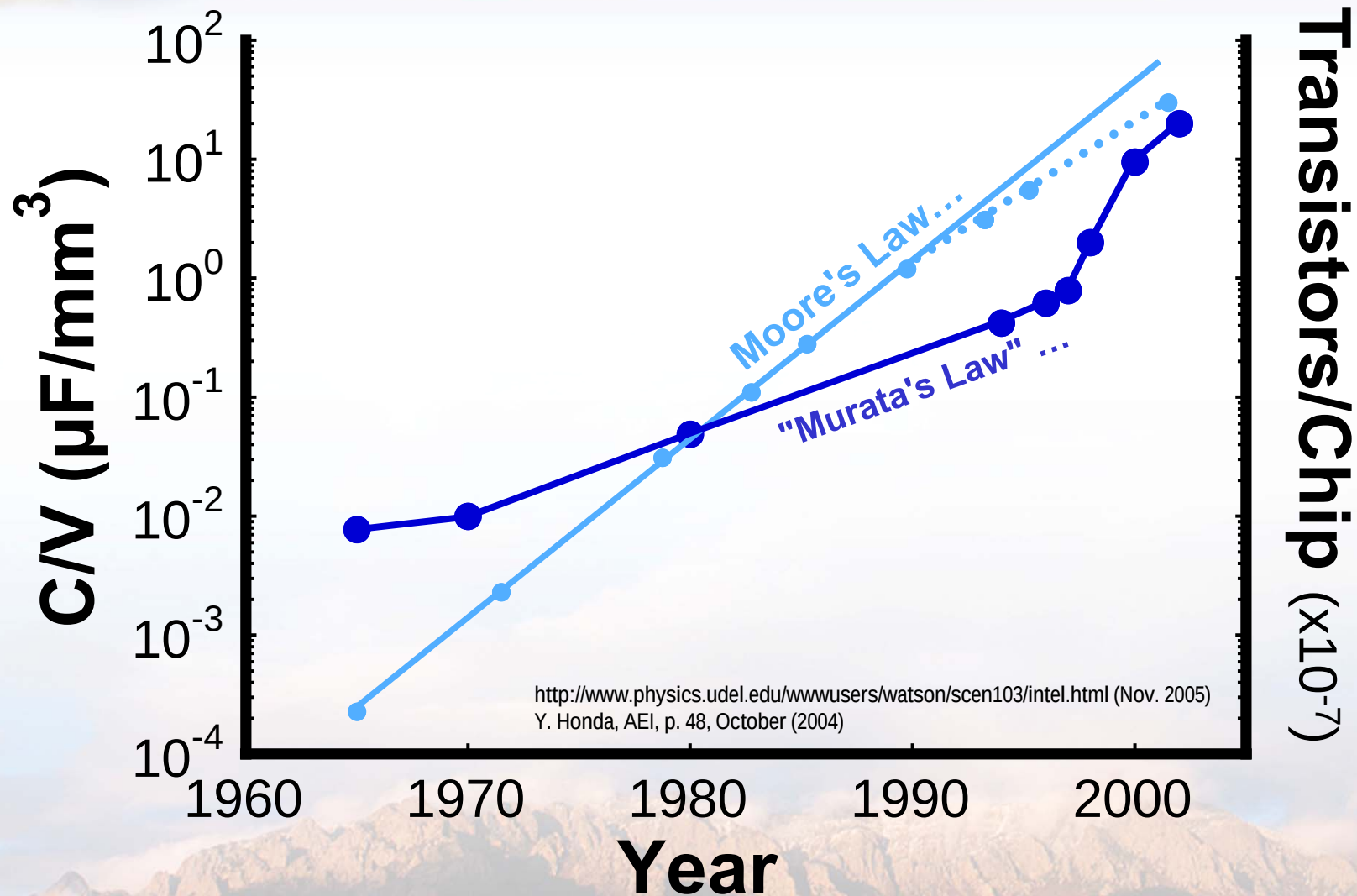
still 10^2 layers!

constant capacitance with
increasing technology node

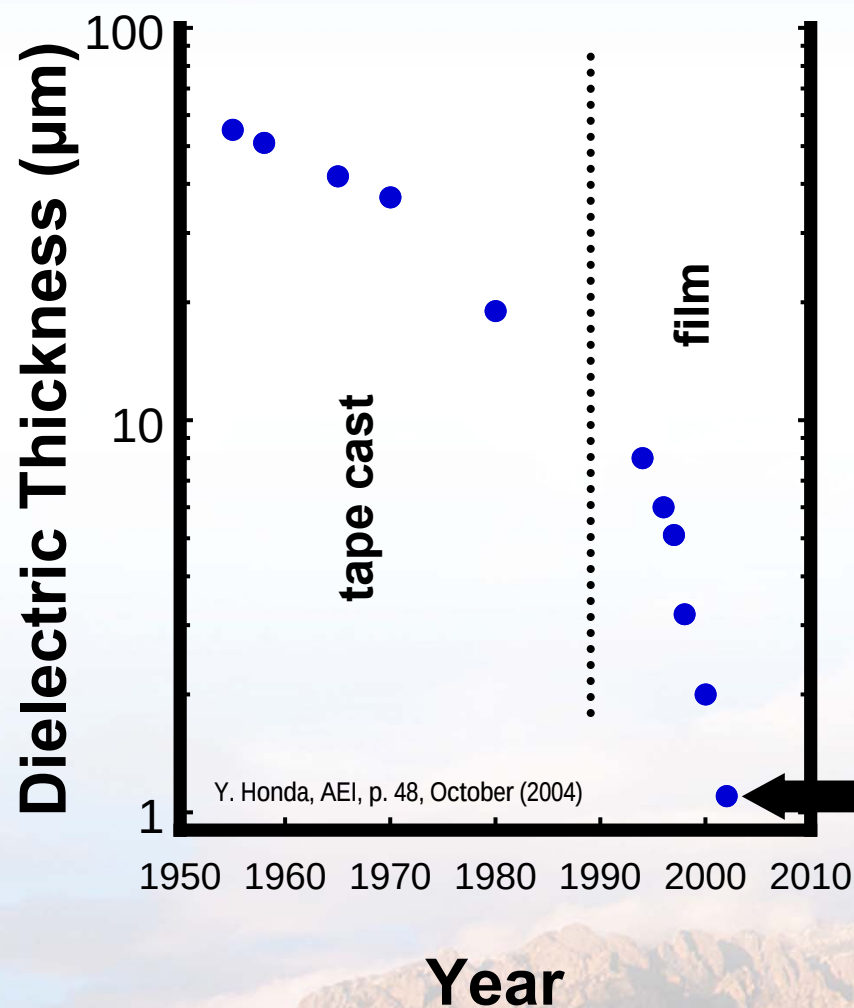


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MLC vs. Transistor Scaling



Physically, what's happening?



- Incredible scaling demonstration
- Fabricated using powder processing - **mass produced**
- Film / lamination process
- Real *Nano-technology*!

